Control de la Corrosion en la Industria Aerospacial

Rumbo a los 50 Años Universidad Autónoma del Carmen Noviembre 14 – 18, 2016

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Overview



- Introduction
- NASA's Current Plans for Space Exploration
- Orion and Space Launch System
- Human Exploration
- Corrosion and Space Exploration
- What is Corrosion?
- Impact and Cost of Corrosion
- Corrosion Grand Challenges
- Corrosion in the Aerospace Industry
- NASA's Technology Roadmap
- Coatings for the Space and Launch Environments
- KSC Natural and Launch Environment

- Corrosion Control Technology Development Timeline
- Corrosion and pH
- Smart Coatings for Corrosion Detection and Control
- Microcapsules for Corrosion Indication, Inhibition, and Self-healing
- Microcapsule Response to pH Increase
- Hidden and early Corrosion Indication
- On-demand corrosion inhibition
- Self-healing
- Summary
- Acknowledgements

Introduction



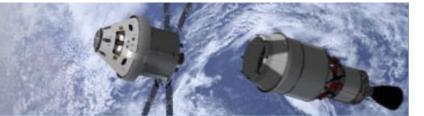
- NASA has been battling corrosion since the inception of the Space Program in 1962.
- NASA launches from the most naturally corrosive environment in North America. Corrosion conditions at the launch pads are even more severe due to solid rocket booster (SRB) exhaust products.
- NASA's Space Technology Roadmap includes corrosion control technologies as one of the areas needed to lower the cost and improve the sustainability and efficiency of its ground operations in support of future launch activities.
- NASA has developed microencapsulation technology specifically designed for corrosion control applications (U.S. Patents No. 7,790,225, 2010 and 20130017612, 2013).
- Micro containers are being used to develop a smart multifunctional coating for autonomous corrosion detection, control, and self-healing.
- This presentation provides an overview of the background and progress made to date in the development of a smart multifunctional coating to indicate corrosion at an early stage and in hidden areas, to deliver corrosion inhibitors on demand, and to self-heal mechanical damage, such as a scratch.



"NASA's Space Launch System (SLS) and Orion will allow human exploration to continue beyond the moon in ways that were once a glimmer in our minds eye. Now we are building the hardware and developing the engineering operations teams that will launch the vehicle that will one day take people to Mars"



Orion America's new spacecraft for human exploration

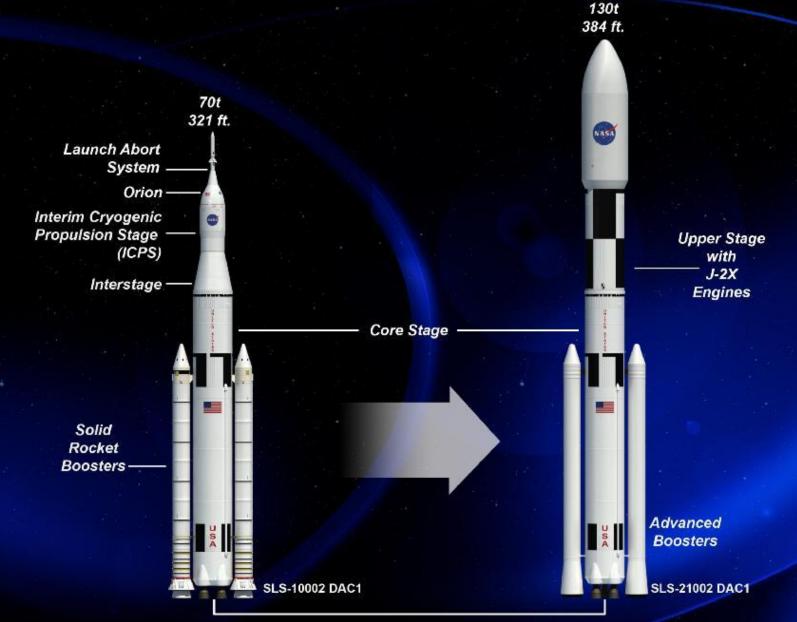


Orion



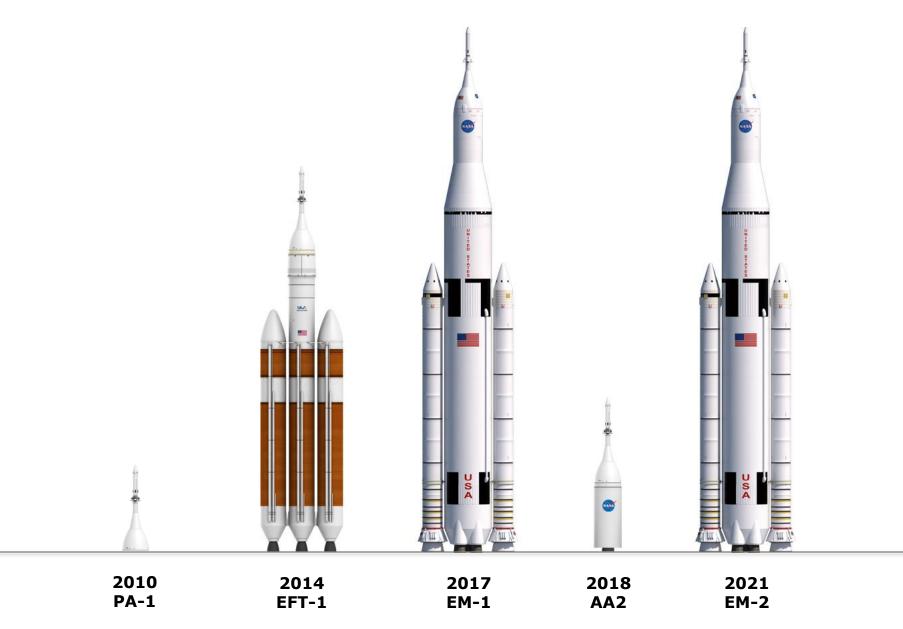
Orion/SLS combination is critical to extending human presence beyond low earth orbit

SLS Architecture Reference Configuration



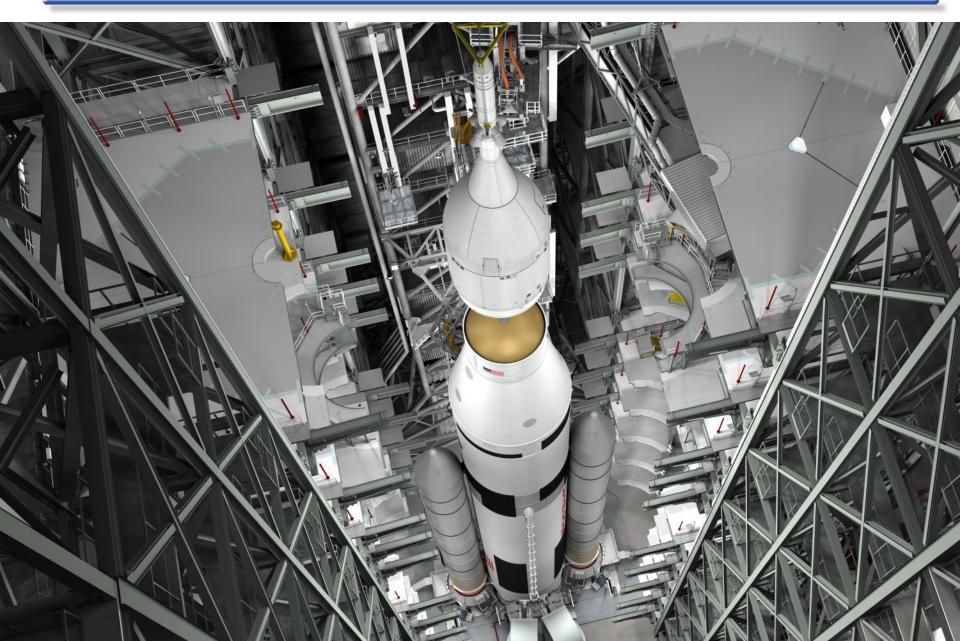
Exploration Mission Timeline





Orion/SLS are the Basis for Exploration beyond Earth Orbit

NAŚA



HUMAN EXPLORATION NASA's Journey to Mars

EARTH RELIANT MISSION: 6 TO 12 MONTHS RETURN TO EARTH: HOURS PROVING GROUND MISSION: 1 TO 12 MONTHS RETURN TO EARTH: DAYS



MARS READY MISSION: 2 TO 3 YEARS RETURN TO EARTH: MONTHS

Mastering fundamentals aboard the International Space Station

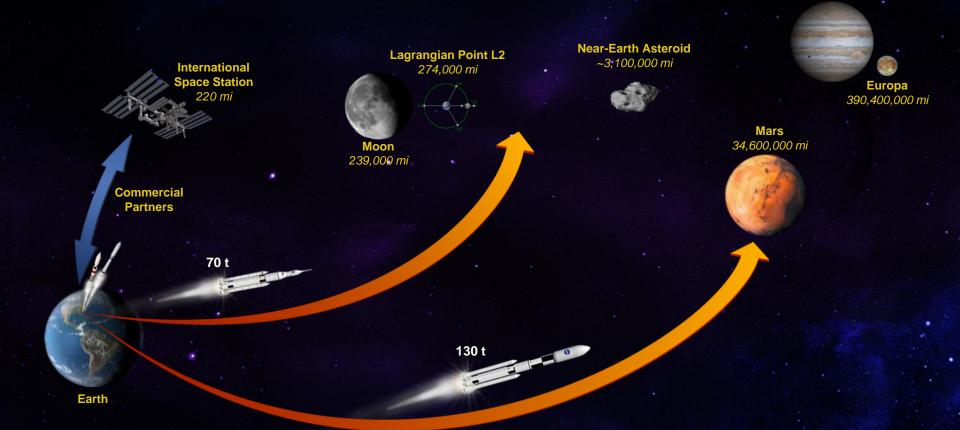
U.S. companies provide access to low-Earth orbit Expanding capabilities by visiting an asteroid redirected to a lunar distant retrograde orbit

The next step: traveling beyond low-Earth orbit with the Space Launch System rocket and Orion spacecraft

Developing planetary independence by exploring Mars, its moons and other deep space destinations

www.nasa.gov

The future of exploration



Space Launch System and Orion enable solar system exploration



Corrosion and Space Exploration

One challenge with space exploration is that equipment must withstand radical conditions, from the heat of rocket exhaust to extreme cold in space. Surprisingly, one of the most destructive forces is the corrosive effect of saltwater-laden ocean spray and fog. It rusts launch structures and equipment at the Kennedy Space Center.

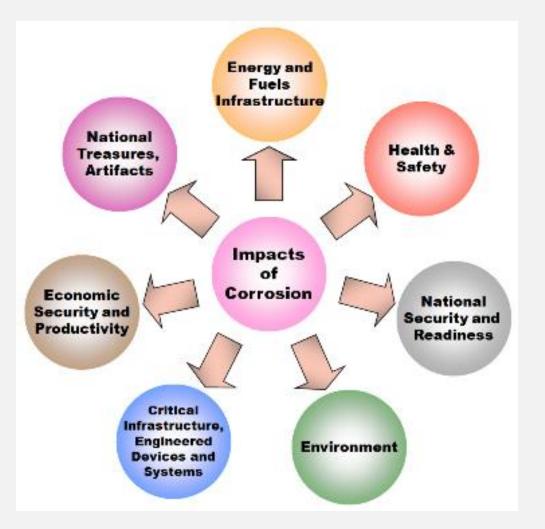


- Corrosion is the deterioration of a material due to reaction with its environment (M.G. Fontana). It literally means to "gnaw away"
- Degradation implies deterioration of the properties of the material.
- Metal atoms in nature are present in chemical compounds (i.e. minerals).
- Metals in their uncombined state
- are in a high energy state. The
- tendency is to corrode and revert
- to the low energy state. Corrosion
- has been called metallurgy in reverse.



Impact of Corrosion





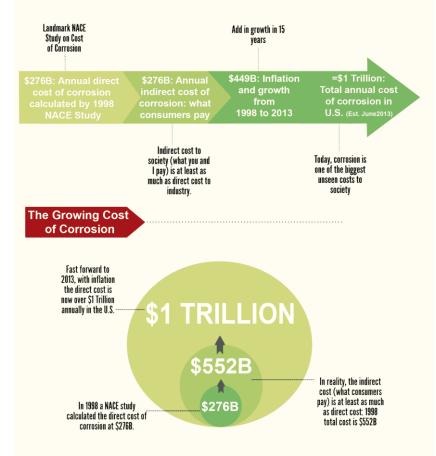


Repairs will cost about \$60 million USD and take about 2 years

Research Opportunities in Science and Engineering, Committee on Research Opportunities in Corrosion Science and **13** Engineering; National Research Council, The National Academies Press, 2011 **Cost of Corrosion**



\$1 TRILLION: Annual Cost of Corrosion in U.S.



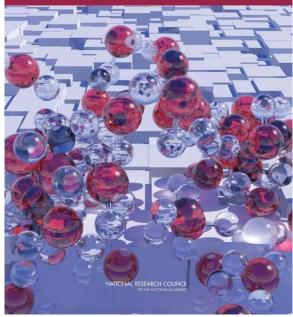
- At US \$2.2 (1.6 €) trillion, the annual direct cost of corrosion worldwide is over 3% of the world's GDP.*
- Direct costs do not include the environmental damage, waste of resources, loss of production, or personal injury.



Corrosion Grand Challenges*

- Development of cost-effective, environment-friendly, corrosionresistant materials and coatings.
- High-fidelity modeling for the prediction of corrosion degradation in actual service environments.
- Accelerated corrosion testing under controlled laboratory conditions. Such testing would quantitatively correlate with the long-term behavior observed in service environments.
- Accurate forecasting of remaining service time until major repair, replacement, or overhaul becomes necessary. i.e., corrosion prognosis.

RESEARCH OPPORTUNITIES IN CORROSION SCIENCE AND ENGINEERING



*Research Opportunities in Corrosion Science and Engineering, Committee on Research Opportunities in Corrosion Science and Engineering; National Research Council (2010)

Corrosion in the Aerospace Industry



- The aerospace industry employs high strength aluminum alloys as structural materials for airplanes because of a combination of valuable mechanical properties, strength to weight ratio, good corrosion resistance, easy recyclability, and formability.
- Aluminum needs alloying with other elements in order to improve its mechanical strength.
- This alloying of aluminum with other metallic elements has a significant effect on its corrosion resistance properties.
- Reduction in corrosion resistance occurs as a result of the heterogeneous microstructure attributed to the second phase intermetallic particles of various sizes and compositions in the aluminum alloy matrix.
- The presence of these intermetallic particles in the aluminum alloy microstructure is often responsible for localized corrosion in the form of pitting, intergranular corrosion, exfoliation corrosion, etc., depending on the alloy type.
- Aluminum alloy surfaces are coated to enhance their corrosion resistance.





Aloha Airlines Accident, 1988





Associated Press library photo • April 28, 1988 (<u>http://the.honoluluadvertiser.com/2001/Jan/18/118localnews1.html</u>)

This picture of Aloha Airlines Flight 243, was taken after one-third of the roof flew off of the aircraft while cruising at 24,000 feet. A series of small cracks, caused by corrosion, had grown simultaneously at many nearby rivet holes, the devastating results of which can be seen in this picture. This is an extreme example of the damage that corrosion can cause, and an indication of why the protection of aluminum against corrosion is so important in the aerospace industry.

Since this mishap, the aviation community has struggled to accurately predict the onset of corrosion or the extent of structural damage it induces.

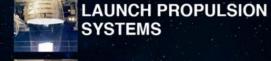
Despite advances in corrosion algorithms, computation material research, simulation, reliability and maintenance data analysis, a reliable model for prediction corrosion on aging aircraft has yet to be produced.

This incident established US civil and military programs for ageing aircraft of which corrosion was a principal area of focus.

Historically, corrosion prevention has not been appreciably designed into an aircraft.

NASA's Technology Roadmap





IN-SPACE PROPULSION TECHNOLOGIES



COMMUNICATIONS,

ROBOTICS AND



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ENTRY, DESCENT, AND LANDING SYSTEMS

NANOTECHNOLOGY

MODELING, SIMULATION, INFORMATION TECHNOLOGY, AND PROCESSING

MATERIALS, STRUCTURES, MECHANICAL SYSTEMS, AND MANUFACTURING

GROUND AND LAUNCH SYSTEMS

THERMAL MANAGEMENT SYSTEMS

AERONAUTICS

CHARACTERIZATION SYSTEMS HUMAN HEALTH, LIFE SUPPORT, AND

HABITATION SYSTEMS

AUTONOMOUS SYSTEMS

NAVIGATION, AND ORBITAL

DEBRIS TRACKING AND

HUMAN EXPLORATION DESTINATION SYSTEMS



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SCIENCE INSTRUMENTS, OBSERVATORIES, AND SENSOR SYSTEMS

Corrosion Control Technologies



NASA Space Technology Roadmap

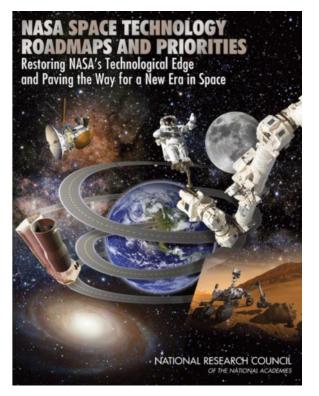


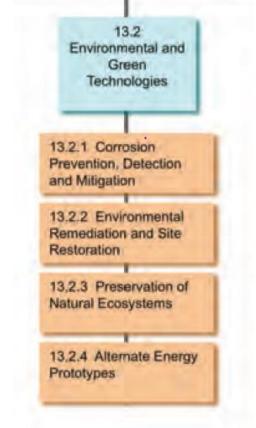
NASA Technology Benefits the Public



Since its inception in 1958, NASA has accomplished many great scientific and technological feats in air and space. NASA technology also has been adapted for many non-aerospace uses by the private sector. NASA remains a leading force in scientific research and in stimulating public interest in aerospace exploration, as well as science and technology in general.







New Materials and Coatings to Meet Changing Demands



- Evolution of aerospace and automotive materials, coatings, and applications processes motivated by:
 - Changing environmental regulations
 - Demands for better energy efficiency and performance
 - Sustainable supply chain
- Future systems require smart coatings with on-demand corrosion protection, healing, and corrosion sensing.
- Future aerospace and automotive platforms will utilize composites, new alloys and coatings with multifunctional capabilities



An artist illustration of the Vehicle Assembly Building and NASA's Space Launch System and Orion crew module on the mobile launcher, with Launch Pad 39B in the background.

Corrosion Protective Coatings



- Barrier (passive).
- Barrier plus active corrosion inhibiting components:
 - Sacrificial (zinc-rich primers)
 - Corrosion inhibitors (can have detrimental effects on the coating properties and the environment; most expensive additive; subject to progressively stricter environmental regulations)
- Smart/Active/Functional/Self-healing



Schematic of a typical multi-coating system used to protect aluminum alloys from corrosion. The conversion coating provides the first layer of corrosion protection and acts as a base for improved adhesion to the primer. Chromate and chromate-phosphate-based conversion coatings have been used for this purpose for several decades, as they are effective in inhibiting the corrosion of aluminum alloys. However, the use of chromium-containing chemicals has been limited because of harmful carcinogenic effects, and intense research efforts are in progress for finding alternatives.

Coatings for the Space Environment

The Space Environment is characterized by:

- Low pressure (vacuum)
- Atomic Oxygen (causes erosion of materials)
- Ultraviolet (UV) radiation
- Charged particles
- Temperature extremes
- Electromagnetic radiation
- Micrometeoroids
- Man-made debris

Coatings on Orion Spacecraft





Corrosion protection coating on aluminum lithium alloy (left) and heat shield (right). The heat shield protects the spacecraft from temperatures reaching 4000 degrees Fahrenheit (2204 °C)

Orion 37 days prior to 1st Test Flight

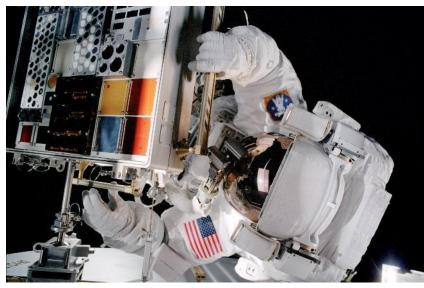




Materials Testing for Space



Materials are tested on the exterior of the International Space Station. The payload container is mounted so one side faces the Earth and the other faces space. The experiments provide a better understanding of material durability, from coatings to electronic sensors, which could be applied to future spacecraft designs.





NASA astronaut Patrick G. Forrester installs exposure experiments designed to collect information on how different materials weather in the environment of space NASA astronaut Andrew Feustel retrieves long duration materials exposure experiments before installing others during a spacewalk on May 20, 2011.

Orion Heat Shield





Textron technicians apply the Avcoat material by "gunning" the material into each of the 330,000 individual cells of the honeycomb structure

Atomic Oxygen Restoration





Interaction of the Space Shuttle with the upper atmosphere creates a corona seen at night (right photo), in part, due to atomic oxygen.

In the upper reaches of the atmosphere, about 200-500 miles, an elemental form of oxygen is created from exposure to intense solar ultraviolet light. Oxygen molecules are decomposed from O_2 into two separate oxygen atoms. This form of elemental oxygen is highly reactive and exposes a spacecraft to corrosion that shortens its life. While developing methods to prevent damage from atomic oxygen, it was discovered that it could also remove layers of soot or other organic material from a surface. Atomic oxygen will not react with oxides, so most paint pigments will not be affected by the reaction. ²⁸

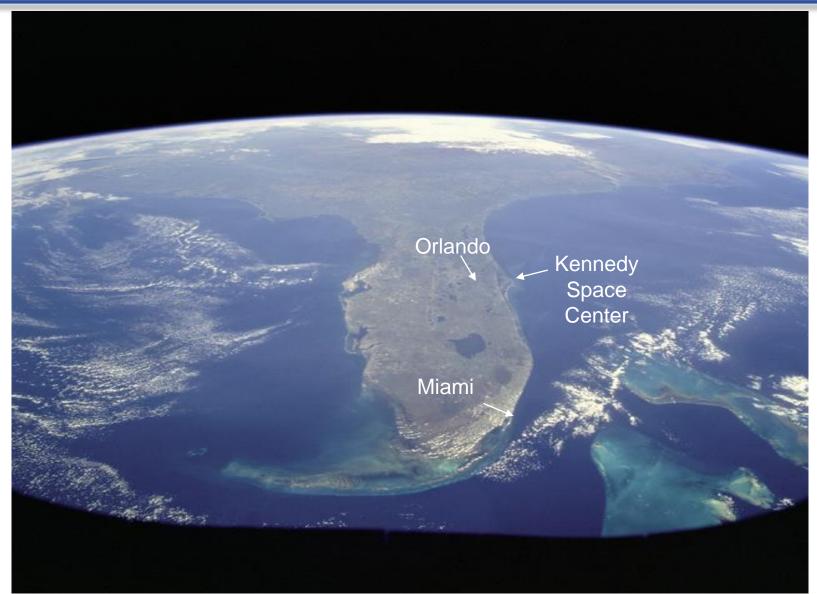
International Space Station Technology – Benefits Fine Art





The left photo was taken after the Cleveland Museum of Art's staff attempted to clean and restore it using acetone and methylene chloride. The right photo is after cleaning by the atomic oxygen technique.

Coatings for NASA's Launch Environment



The Kennedy Space Center in Florida, USA, is a special place where we launch rockets from a wild life refuge in one of the most corrosive areas in the world





KSC Natural Environment



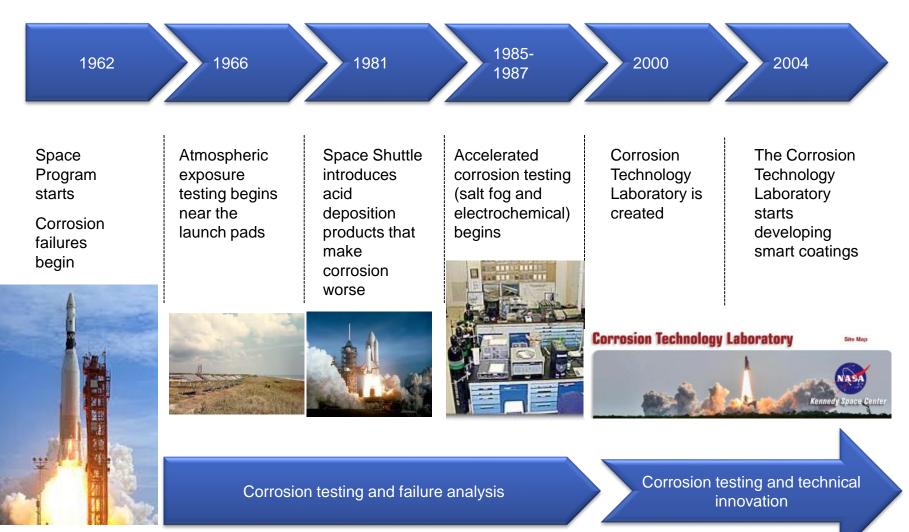


KSC Launch Environment



Corrosion Technology Laboratory at KSC Timeline





Atmospheric Exposure Testing





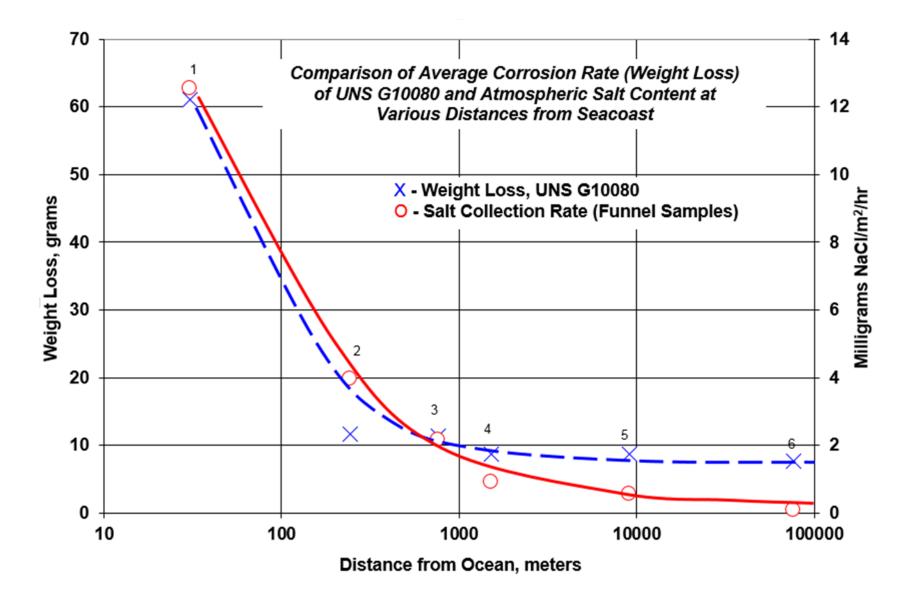
KSC Natural Environment





Changes in Corrosion Rate with Distance from the Ocean







Corrosion rates of carbon steel calibrating			
specimens at various locations*			
Location	Type Of Environment	µm/yr	Corrosion rate ^a mils/yr
Esquimalt, Vancouver Island, BC, Canada	Rural marine	13	0.5
Pittsburgh, PA	Industrial	30	1.2
Cleveland, OH	Industrial	38	1.5
Limon Bay, Panama, CZ	Tropical marine	61	2.4
East Chicago, IL	Industrial	84	3.3
Brazos River, TX	Industrial marine	94	3.7
Daytona Beach, FL	Marine	295	11.6
Pont Reyes, CA	Marine	500	19.7
Kure Beach, NC (80 ft. from ocean)	Marine	533	21.0
Galeta Point Beach, Panama CZ	Marine	686	27.0
Kennedy Space Center, FL (beach)	Marine	1070	42.0

^aTwo-year average

* Data extracted from: S. Coburn, Atmospheric Corrosion, in Metals Handbook, 9th ed, Vol. 1, Properties and Selection, Carbon Steels, American Society for Metals, Metals Park, Ohio, 1978, p.720

39

KSC Launch Pad Environment

The launch environment at KSC is extremely corrosive:

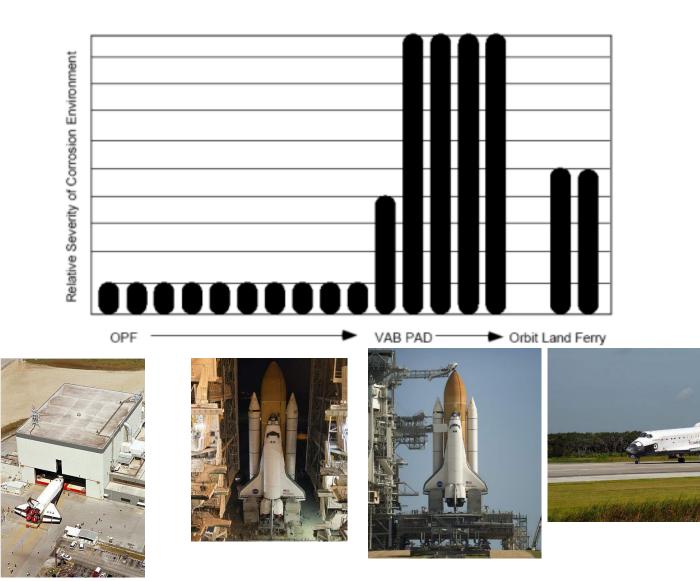
- Ocean salt spray
- Heat
- Humidity
- Sunlight
- Acidic exhaust from SRBs



Natural Salt Fog Chamber



Orbiter Flight and Ground Environment



A S

In 1981 the Space Shuttle introduced acidic deposition (70 tons of HCI) products. NASA plans to use Shuttle-derived SRB rockets in future missions.

SRB Exhaust

 $NH_4CIO_4(s) + AI(s)$

binder, Fe_2O_3

USA

 $AI_2O_3(s) + HCI(g) + H_2O(g) + NO_x(g)$

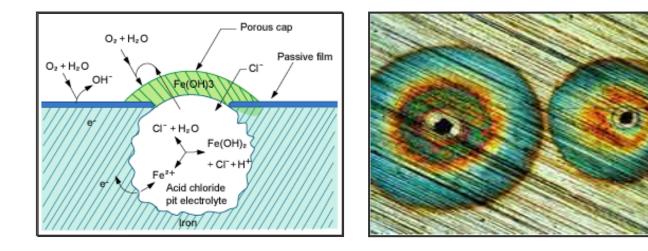
Electrochemical Nature of Corrosion

Metal is oxidized (anodic reaction); something else is reduced (cathodic reaction)

Overall Reaction:

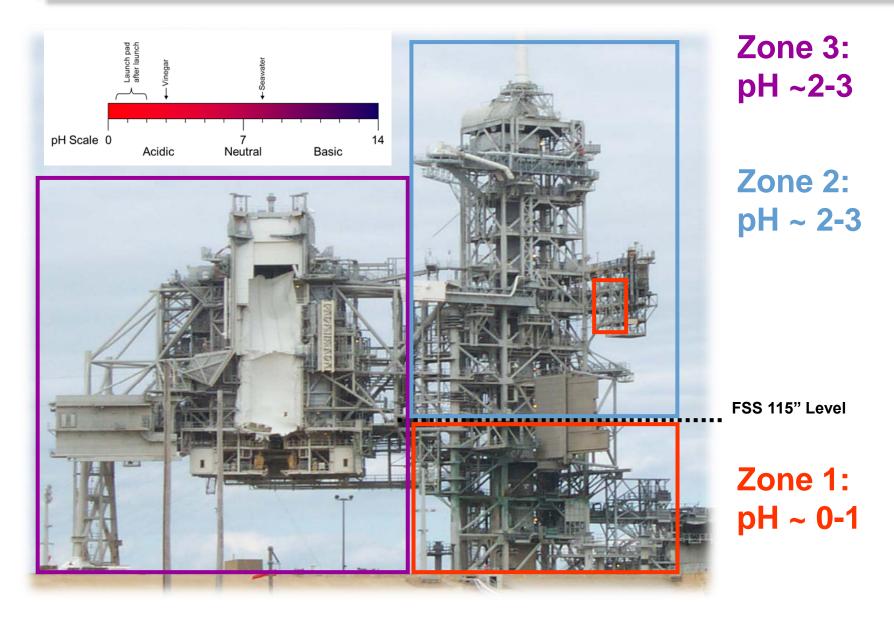
Anodic:
$$Fe \rightarrow Fe^{2+} + 2e^{-}$$

Cathodic: $2H_2O + O_2 + 4e^- \rightarrow 4OH^-$



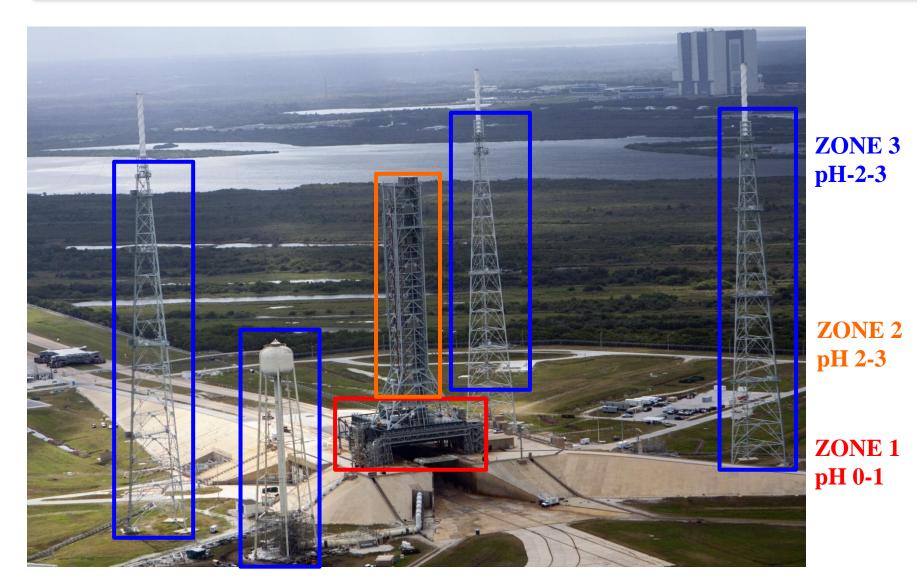
Launch Complex 39 pH Values after a Space Shuttle Launch











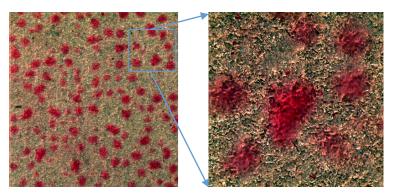
Zone 1: Direct or indirect impingement by SRB exhaust; Zone 2: Elevated temperatures and acid deposition; Zone 3: Acid exposure or other types of chemical contamination

Corrosion Protective Coatings



- Barrier (passive).
- Barrier plus active corrosion inhibiting components:
 - Sacrificial (zinc-rich primers)
 - Corrosion inhibitors (can have detrimental effects on the coating properties and the environment; most expensive additive; subject to progressively stricter environmental regulations)
- Smart (Intelligent, self-healing, active or feedback active)

A smart coating detects and responds actively to changes in its environment in a functional and predictable manner and is capable of adapting its properties dynamically. These coatings are also referred to in the literature as intelligent, self-healing, and active or feedback active, to distinguish them from barrier or passive coatings.



Smart coating responding to changing pH conditions

Smart Coatings for Corrosion Control

- The use of "smart coatings" for corrosion sensing and control relies on the changes that occur when a material degrades as a result of its interaction with a corrosive environment.
- Such transformations can be used for detecting and repairing corrosion damage.
- NASA's Corrosion Technology Laboratory is developing a coating that can detect and repair corrosion at an early stage.
- This coating is being developed using pH-sensitive microcontainers that deliver the contents of their core when corrosion starts to:
 - Detect and indicate the corrosion location
 - Deliver environmentally friendly corrosion inhibitors
 - Deliver healing agents to repair mechanical coating damage.

Microencapsulation-based Smart Coatings



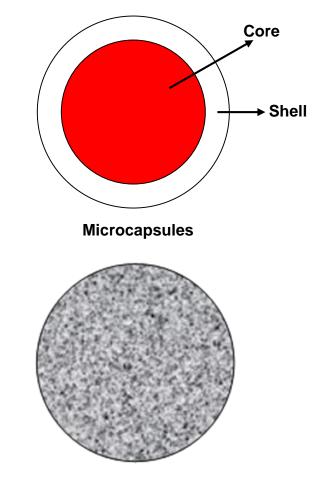
Corrosion indication, detection, and healing of mechanical damage can be achieved using microencapsulation technology

What are microcontainers?

Particles or liquid drops coated in polymers. These microcontainers can carry any material that needs protection or controlled release.

Why microencapsulate a material?

- Incorporate active materials while maintaining coating integrity
- Achieve controlled-release
- Make active materials easier/safer to handle.
- Incorporate multiple component systems.
- Prevent undesired leaching
- Versatility





Types of Feedback-Active Micro Containers for Corrosion Detection and Control

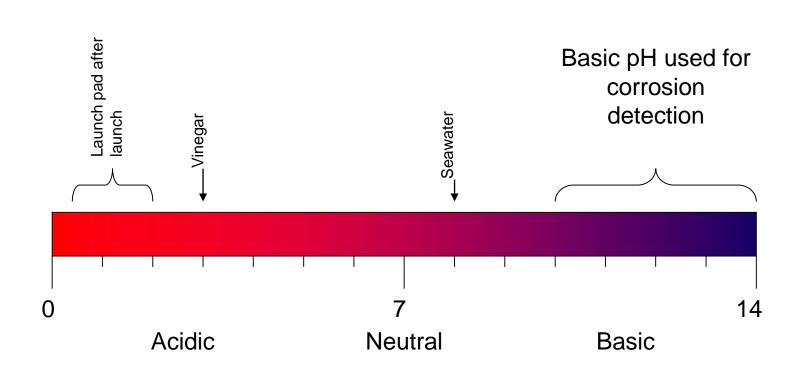
- Containers with an active ingredient-rich core and stimuli-responsive shell (microcapsules)
- Containers with an active ingredient incorporated into a stimuli-responsive matrix (micro particles)
- Containers with a porous ceramic core impregnated by inhibitor and enveloped by a stimuli-responsive polyelectrolyte (PE) shell*



*D. Grigoriev, D. Akcakayiran, M. Schenderlein, and D. Shcukin, Corrosion, 70 (2014): p.446-463.

Corrosion and pH

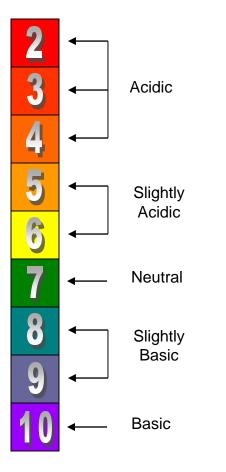


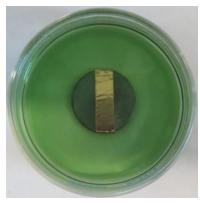


pH Scale

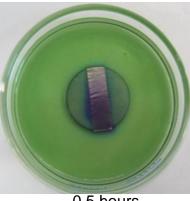
Corrosion Indication

pH changes that occur during corrosion of a metal

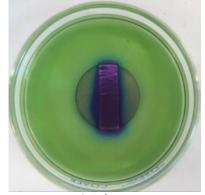




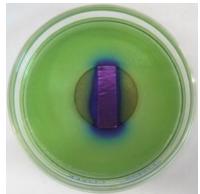
Elapsed Time: 0 hours



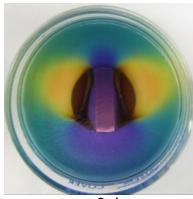
0.5 hours



1.5 hours



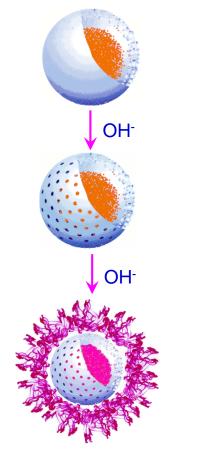
4.5 hours





pH Sensitive Microcapsules for Corrosion Indication





Microcapsule containing pH indicator (inhibitor, self healing agents)

The shell of the microcapsule breaks down under basic pH (corrosion) conditions

pH indicator changes color and is released from the microcapsule when corrosion starts

Why pH-Sensitive Micro Containers?



Benefits from corrosion sensing coatings:

- Overcoming solubility limit of indicators in solvents
- Protecting indicator from coating constituents
- Easy incorporation into different paint systems

Benefits from corrosion-controlled inhibitor release:

Smart:

- Corrosion-controlled release on demand at tailorable rate.
- Minimization of inhibitor loss by leaching or diffusion if desired.
- Inhibitor release can be maximized when corrosion occurs.
- Inhibitor delivery when and where needed.

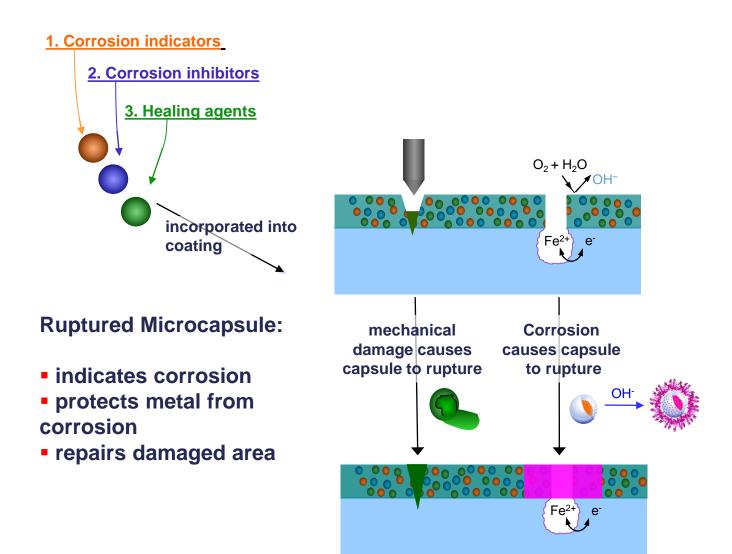
Green:

• Reduction of the detrimental impact of inhibitors on the environment.

Enabling:

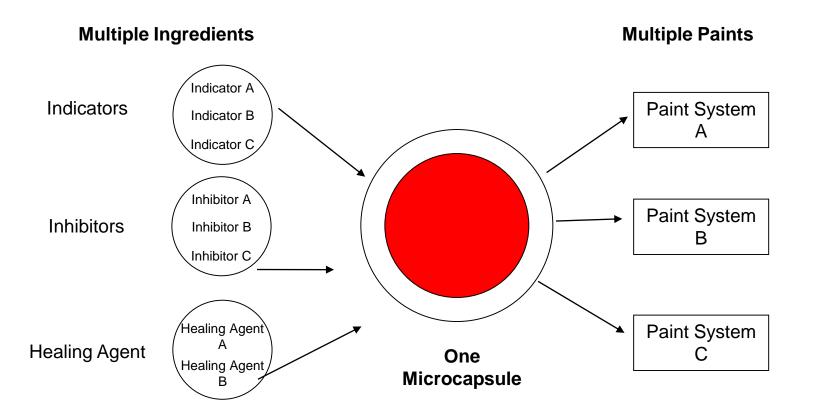
- Inhibitor isolation avoids incompatibility with other coating components.
- Allows incorporation of water-soluble inhibitor into paint formulation without blistering.

Smart Coating for Corrosion Detection and Control





Versatility: Microcapsules can deliver multiple types of contents into different paint systems. The microcapsule wall can be modified to deliver the contents at different rates.



Summary



- NASA has been battling corrosion since the inception of the Space Program in 1962.
- NASA launches from the most naturally corrosive environment in North America.
- NASA's Space Technology Roadmap includes corrosion control technologies as one of the areas needed to lower the cost and improve the sustainability and efficiency of its space exploration missions.
- Corrosion control in the aerospace industry is based on materials selection and coatings.
- The aerospace industry has relied on hexavalent chromium coatings for corrosion control of aluminum alloys.
- New corrosion control coatings are needed to replace hexavalent chromium coatings because of their toxicity and imminent ban.
- NASA is developing smart coatings as replacements for hexavalent chromium coatings for the aerospace industry.

Acknowledgements

FUNDING

NASA's Ground Systems Development and Operations (GSDO) Program. US Department of Defense (DoD). ⁵⁶