# NASA CONTRACTOR REPORT

NASA CR-61101.



# MANNED FLYING SYSTEMS (MFS) CRYOGENIC ENGINE

Prepared Under Contract No. NAS8-20082

F. B. Tatom and L. M. Bhalla

NORTHROP SPACE LABORATORIES Huntsville, Alabama

GPO PRICE \$_	
CFSTI PRICE(S) \$	
Hard copy (HC) _	.3.00
Microfiche (MF)	.75
ff 653 July 65	

For

NASA - GEORGE C. MARSHALL SPACE FLIGHT CENTER Huntsville, Alabama

September 1965

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By

F. B. Tatom and L. M. Bhalla

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#### ABSTRACT

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The results are presented of a preliminary investigation to compare the performance of a cryogenic (liquid oxygen and liquid hydrogen) propulsion system with an earth storable (nitrogen tetroxide and a 50:50 mixture of hydrazine and unsymmetrical dimethylhydrazine) propulsion system for the Manned Flying System (MFS). The mass of the propulsion system necessary to meet the mission requirements of the MFS, as outlined in the Apollo Extension System Studies, is used as the basis for comparison. A semi-empirical mathematical model is developed to predict the mass of the propulsion systems. The results, which are presented in both tabular and graphical form, indicate that the cryogenic system offers no apparent advantage over the earth storable system under the present state-of-the-art.

#### FOREWORD

The investigation described in this report was requested by Mr. Lynn L. Bradford of the Systems Concepts Planning Office, Aero-Astrodynamics Laboratory, George C. Marshall Space Flight Center. The study was carried out by Mr. F. B. Tatom and Mr. L. M. Bhalla of the Huntsville Department of Northrop Space Laboratories, under Contract No. NAS8-20082, Appendix F-1, Schedule Order 4, Technical Directive No. 2. Work commenced on May 3, 1965 and ended on July 19, 1965, with a total of five man weeks expended.

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# DEFINITION OF SYMBOLS

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# A. English Letters

Symbol	Definition	<u>Units</u>
С	General structure factor	
C <sub>fw</sub>	Thrust-to-weight ratio	
Cpf	Structure factor for propellant fuel	
Cpo	Structure factor for propellant oxidizer	
C <sub>sf</sub>	Structure factor for storage of propellant fuel	
C so	Structure factor for storage of propellant oxidizer	
с <sub>р</sub>	Heat capacity	Btu/lb_°F
F	Thrust produced by one engine	<sup>lb</sup> f
g	Acceleration of gravity on earth's surface	ft/sec <sup>2</sup>
h <sub>c</sub>	Latent heat generated by change of phase	Btu/1b <sub>m</sub>
Isp	Specific impulse of propellant	sec
k in	Thermal conductivity of insulation	Btu/ft°F hr
m <sub>a</sub>	Mass of accessory equipment	lb <sub>m</sub>
<sup>m</sup> c	Mass of fuel or oxidizer subjected to a change of phase	lb <sub>m</sub>
<sup>m</sup> e	Mass of the rocket engines	<sup>1b</sup> m
<sup>m</sup> f	Mass of the fuel	lb <sub>m</sub>
<sup>m</sup> in	Mass of insulation	lb <sub>m</sub>
m o	Mass of oxidizer	lb <sub>m</sub>
mps	Mass of propellant system (aboard MFS)	1b <sub>m</sub>
m	Mass of the storage system (including propellant)	lb <sub>m</sub>

# DEFINITION OF SYMBOLS

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# A. English Letters (Cont'd)

Symbol	Definition	Units
m s	Mass of the shell	1b <sub>m</sub>
<sup>m</sup> t	Total mass of system	1b <sub>m</sub>
m <sub>x</sub>	Total mass of fuel or oxidizer within container	<sup>1b</sup> m
<sup>m</sup> f	Mass flow rate of the fuel for each engine	lb <sub>m</sub> /sec
т <sub>о</sub>	Mass flow rate of the oxidizer for each engine	lb <sub>m</sub> /sec
m <sup>p</sup>	Mass flow rate of propellant for each engine	lb <sub>m</sub> /sec
ne	Number of engines	
n m	Number of missions	
P p	Design pressure for a propellant container	psia
P pmax	Maximum design MFS propellant tank pressure	psia
P pmin	Minimum design MFS propellant tank pressure	psia
P <sub>s</sub>	Propellant storage pressure	psia
Q <sub>s</sub>	Total heat flow from the shell	Btu
q	Rate of heat flow from the shell	Btu/hr
r <sub>i</sub>	Inner radius of insulation	ft
ro	Outer radius of insulation	ft
rs	Radius of the shell	ft
т <sub>с</sub>	Temperature for change of phase	°R
<sup>T</sup> fr	Fusion temperature of propellant	°R
T <sub>i</sub>	Temperature of propellant or inside insulation	°R
<sup>T</sup> io	Initial storage temperature	°R

# DEFINITION OF SYMBOLS

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# A. English Letters (Cont'd)

Symbol	Definition	Units
т	Outside temperature of insulation	°Ŕ
T omax	Maximum outer surface temperature of propellant container	°R
T <sub>omin</sub>	Minimum outer surface temperature of propellant container	°R
<sup>T</sup> sat	Saturation temperature of propellant	°R
t	Shell thickness	ft
x	Mixture ratio of oxidizer to fuel	
	B. <u>Greek Letters</u>	
ΔT	Temperature difference across insulation	°F
$\overline{\Delta_{\mathrm{T}}}$	Mean temperature difference across insulation as defined by Eq. (B-8)	°F
n	Safety factor in pressure vessel design	
λ	Thermodynamic parameter defined by Eq. (C-7)	hr <sup>-1</sup>
<sup>ρ</sup> in	Density of the insulation	$1b_m/ft^3$
ρ s	Density of shell material	$1b_m/ft^3$
ρ x	Density of fuel or oxidizer	1b <sub>m</sub> /ft <sup>3</sup>
σ s	Tensile strength of shell material	psia
τ	Time	(days, hr or sec)
τ <sub>b</sub>	Engine burning time during a single mission	sec
τ m	Mission time	hr
τ s	General storage time	hr
τ ss	Propellant long-term storage time	days

#### SUMMARY

An analytical investigation has been conducted to determine whether or not the use of a cryogenic (liquid oxygen and liquid hydrogen) propulsion system for the Manned Flying System (MFS) offers any advantages over the earth storable (nitrogen tetroxide and a 50:50 mixture of hydrazine and unsymmetrical dimethylhydrazine) system presently under consideration. The mass of the necessary equipment associated with each propulsion system is used as a basis for comparison. This mass includes not only that equipment aboard the MFS but also the propellant storage system which is contained within the IEM/A, IEM/D, and/or IEM/S stages.

A semi-empirical mathematical model is developed to permit calculation or prediction of the total mass of the propulsion systems. This model utilized certain dimensionless parameters including thrustto-weight ratios and structure factors. The latter are mass parameters which express the ratio of structural mass to propellant mass. A method of evaluating these structure factors is provided.

By means of the mathematical model developed, the mass of the cryogenic propulsion systems is calculated along with the mass of the earth storable system. Such calculations reveal that the cryogenic system appears to be heavier than the earth storable system under the same mission requirements. The conclusion is reached that there is no apparent justification for using the cryogenic system in place of the earth storable system.

#### 1.0 INTRODUCTION

As part of the Apollo Extension System (AES) program, a Manned Flying System (MFS) will be used to provide rapid transportation from point to point on the lunar surface. The operational characteristics of this craft have been described in previous research efforts (refs. 1, 2, and 3). As indicated in these references, present plans call for a propulsion system consisting of five 100 lb<sub>f</sub> thrust, throttable rocket engines. The propellants for these engines would be a 50:50 mixture by weight of hydrazine ( $N_2H_4$ ) and unsymmetrical dimethylhydrazine (UDMH) for fuel, and nitrogen tetroxide ( $N_2O_4$ ) for oxidizer. These propellants, commonly referred to as "earth storables", are also used in the descent and ascent stages of the LEM vehicle. Considerable thought has been given to possible arrangements whereby the propellants required for the MFS might be supplied by using the residual propellant from the LEM/D (descent stage of LEM) and/or by using a portion of the propellants from the LEM/A (ascent stage of LEM).

Because of their greater specific impulse, liquid hydrogen and liquid oxygen (cryogenic propellants) appear to be promising alternatives to the earth storable propellants. There is the possibility that the auxiliary power supply for the LEM/A and the LEM/S (shelter stage of LEM) will involve fuel cells using liquid hydrogen and liquid oxygen (refs. 4, 5, 6, and 7). By means of a common storage system, these cryogenic propellants might be used both to provide auxiliary power for the LEM/S and LEM/A, and to propel the MFS. Because of the cryogenic

temperature involved, however, such an arrangement would appear to require more elaborate facilities than would be necessary for the earth storables. This report presents the results of a preliminary study by the Huntsville Department of Northrop Space Laboratories concerning the relative merits of the two propulsion systems.

#### 2.1 Background

In order to obtain a more up-to-date understanding of the two propulsion systems under consideration, a short literature search was conducted. The performance and overall characteristics of both cryogenic and earth storable propulsion systems for thrusts ranging from 100,000 to 1,000,000 1b<sub>f</sub> are relatively well established. In the low thrust range from 100 to 500  $1b_f$  , however, especially for cryogenic systems, a scarcity of published material exists. For this reason, letters were written to seven aerospace companies, prominent in cryogenic propulsion, requesting information regarding their most recent work in the area of low-thrust cryogenic propulsion. A sample of these letters is provided in Appendix A, along with the responses obtained. Although limited information was available concerning low-thrust earth storable propulsion systems, several reports (refs. 2 and 3) were obtained which were both detailed and pertinent. Thus, written inquiries to the appropriate aerospace firm for this case were unnecessary.

Because of the relatively short time period involved in the study, the investigation was carried out for the most part before the answers to the written inquiries were received. Thus, certain assumptions which were made initially out of necessity, regarding propulsion system performance, may differ to some extent from the actual data provided in

the responses contained in Appendix A. Such differences, however, do not significantly affect the overall validity of the analysis.

### 2.2 Basis for Comparison

There are numerous parameters upon which the performance of a propulsion system can be judged. In the problem under consideration, a number of these parameters have already been fixed either by the mission requirements or by the system interface requirements as discussed in reference 1. These include:

> Propellant long-term storage time  $(\tau_{ss}) = 180$  days Engine thrust (F) = 100 lb<sub>f</sub> Number of engines  $(n_e) = 5$ .

Certain other quantities while not truly constant can or must be treated as such for simplicity. These quantities are presented in Table 2-1. In drawing conclusions from this study, consideration must be given to the assignment of constant values to the parameters indicated in this table.

The most logical parameter upon which to compare the performance would appear to be the total mass involved in the operation of the propulsion system. This mass would take into account, not only the mass of the rocket engines and of the propellant storage and feed system aboard the MFS, but also the mass of the MFS propellant storage system located within the IEM/A, IEM/D, and/or IEM/S stages. That system with the least mass which can meet the operational requirements

# TABLE 2-1

# Values of Important Propulsion Parameters (Assumed to be constants)

Parameter	Value
Number of missions $(n_m)$	3
Mission time $(\tau_m)$	3 hours
Minimum design MFS propellant tank pressure (Ppmin)	150 <b>psi</b> a
Maximum design MFS propellant tank pressure (Ppmax)	1000 psia
Cryogenic propellant storage pressure (P <sub>s</sub> )	50 psia
Earth storable propellant storage pressure (P <sub>s</sub> )	14.7 <b>psi</b> a
Structure safety factor (n)	1.67
Maximum outer surface temperature of propellant container (T ) omax	600°R
Minimum outer surface temperature of propellant container (T ) omin	200°R
Earth storable propellant specific impulse (I ) sp	300 <b>sec</b>
Cryogenic propellant specific impulse (I )	400 sec
Earth storable propellant mixture ratio (x)	2.0
Cryogenic propellant mixture ratio (x)	5.0

specified by the constant parameters already listed, represents the optimum propulsion system.

#### 2.3 Mathematical Model Development

Based on the discussion presented in the preceding subsections, the need arises for developing a mathematical relationship which will permit calculation of the total system mass. In the most primitive form, the total system mass,  $m_t$ , can be expressed as

$$m_{t} = m_{e} + m_{s} + m_{s}$$
(1)

where

Notice should be taken that the mass described by Eq. (1) does not include certain equipment, such as the pressurization system and the attitude rocket assembly. These items were not included because, for low thrust levels, the associated masses would remain essentially constant regardless of the type of propellant.

An exact analytical prediction of  $m_e$ , the mass of the rocket engines, would be exceedingly complex due to the nature of the technologies involved. A reasonably accurate value based on the ratio of thrust-to-weight ( $C_{fw}$ ) can be obtained empirically. Based on the

definition of this ratio,

$$m_{e} = n_{e} \frac{F}{g C_{fw}}$$
(2)

where

 $n_e = number of rocket engines$   $F = engine thrust (1b_f)$ g = acceleration of gravity on earth's surface (ft/sec<sup>2</sup>).

The factor  $C_{fw}$  is primarily a function of the nature of the propellants and the magnitude of the thrust. Figure 2-1 represents the approximate variation of  $C_{fw}$  with thrust for both cryogenic and earth storable propellants. For a thrust of 100 lb<sub>f</sub> Figure 2-1 indicates that the appropriate thrust-to-weight ratios are

$$C_{fw} = 14$$
 (earth storables)  
 $C_{fw} = 42$  (cryogenics).

Notice should be taken that, due to the lack of existing data for lowthrust cryogenic systems, considerable extrapolation is necessary in order to obtain a value of  $C_{fw}$  for cryogenic systems with thrusts in the range of 100 to 500 lb<sub>f</sub>.

The mass of the propellant systems, m<sub>ps</sub>, includes all equipment aboard the MFS which is associated with the propulsion system other than the rocket engines. The mass consists primarily of the propellant feed tanks and does not include the mass of the propellants themselves. As already noted, the pressurization system mass and attitude rocket



FIGURE 2-1 VARIATION OF THRUST-TO-WEIGHT RATIO WITH THRUST

assembly mass are constants which can be neglected in the present analysis. For convenience, the mass  $m_{ps}$  can be expressed in the form

$$m_{ps} = (C_{pf}-1) m_{f} + (C_{po}-1) m_{o}$$
 (3)

where

Now the masses  $m_{f}$  and  $m_{o}$  can be expressed as

$$m_{f} = n_{e} \int_{0}^{\tau_{b}} \dot{m}_{f} d\tau \qquad (4)$$
$$m_{o} = n_{e} \int_{0}^{\tau_{b}} \dot{m}_{o} d\tau \qquad (5)$$

where

 $\tau = time (sec)$   $\tau_b = burning time for a single mission (sec)$   $\dot{m}_f = mass flow rate of fuel for each engine (lb_m/sec)$  $\dot{m}_o = mass flow rate of oxidizer for each engine (lb_m/sec).$ 

A combination of Eqs. (3), (4), and (5) yields

$$m_{ps} = n_e ((C_{pf}-1) \int_{0}^{b} \dot{m}_f d^2 + (C_{po}-1) \int_{0}^{b} \dot{m}_o d^2)$$
 (6)

By definition

$$\dot{\tilde{m}}_{p} = \dot{\tilde{m}}_{0} + \dot{\tilde{m}}_{f}$$
(7)

and

$$\mathbf{x} = \dot{\mathbf{m}}_{o} / \dot{\mathbf{m}}_{f}$$
(8)

where

$$\dot{m}_{p}$$
 = mass flow rate of propellant (lb<sub>m</sub>/sec)  
x = mixture ratio.

Then

$$\dot{m}_{o} = \dot{m}_{p} \frac{x}{x+1}$$
(9)

and

$$\dot{m}_{f} = \dot{m}_{p} \frac{1}{x+1}$$
(10)

Thus by substitution,

$$m_{ps} = n_{e} \left[ (C_{pf} - 1) \int_{0}^{b} \frac{\dot{m}_{p}}{x+1} d + x(C_{po} - 1) \int_{0}^{b} \frac{\dot{m}_{p}}{x+1} \right]$$
$$= \frac{n_{e}}{x+1} \left[ (C_{pf} - 1) + x(C_{po} - 1) \right] \int_{0}^{b} \dot{m}_{p} d \qquad (11)$$

The mass of the storage system  $m_{ss}$ , includes the mass of all fuel and oxidizer necessary to perform " $n_m$ " missions, where for the case under consideration, as shown in Table 2-1,

$$n_m = 3$$
.

This mass can be expressed in a manner similar to that developed for

 $\ensuremath{\mathtt{m}}\xspace$  . The resulting expression is  $\ensuremath{\mathtt{ps}}\xspace$ 

$$m_{ss} = n_{m} \cdot n_{e} \left( C_{sf} \int_{0}^{\tau} \dot{m}_{f} d\tau + C_{so} \int_{0}^{\tau} \dot{m}_{o} d\tau \right)$$
$$= \frac{n_{m} \cdot n_{e}}{x+1} \left( C_{sf} \int_{0}^{\tau} \dot{m}_{p} d\tau + x C_{so} \int_{0}^{\tau} \dot{m}_{p} d\tau \right)$$
$$= \frac{n_{m} \cdot n_{e}}{x+1} \left( C_{sf} + x C_{so} \int_{0}^{\tau} \dot{m}_{p} d\tau \right)$$
(12)

where

 $C_{sf} = \text{storage system fuel structure factor } \frac{m_{sf}}{n m_{f}}$   $m_{sf} = \text{mass of fuel storage system including mass of fuel itself (lb_m)}$   $C_{so} = \text{storage system oxidizer structure factor } \frac{m_{so}}{n_{m}m_{o}}$   $m_{so} = \text{mass of oxidizer storage system including mass of oxidizer itself (lb_m)}.$ 

The total mass,  $m_t$ , can be expressed as a combination of Eqs. (1), (2), (11), and (12), yielding,

$$m_{t} = \frac{n_{e}}{g} \frac{F}{C_{fw}} + \frac{n_{e}}{x+1} \left[ (C_{pf}-1) + x(C_{po}-1) \right]_{o}^{\tau} \overset{h}{m}_{p} d\tau + \frac{n_{m} \cdot n_{e}}{x+1} \left( C_{sf} + x C_{so} \right)_{o}^{\tau} \overset{h}{m}_{p} d\tau = n_{e} \left\{ \frac{F}{g} \frac{F}{C_{fw}} + (\frac{1}{x+1}) \left[ (C_{pf}-1 + n_{m}C_{sf}) + x(C_{po}-1+n_{m}C_{so}) \right]_{o}^{\tau} \overset{h}{m}_{p} d\tau \right\}$$
(13)

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Eq. (13) represents the mathematical model developed and used in the analysis presented in this report. In using this equation, the integral term,  $n_e \int_{0}^{b_e} p^d t$ , must be carefully evaluated, and appropriate values of the structure factors,  $C_{pf}$ ,  $C_{po}$ ,  $C_{sf}$ , and  $C_{so}$ , must be selected.

### 2.4 Evaluation of Propellant Consumption

The integral,  $n_e \int_0^{\tau_b} \dot{m}_p d^{\tau}$ , represents the amount of propellant burned during a single mission which requires the engines to be operating for a burning time of  $\tau_b$ . A previous study (ref. 1) has shown that for the MFS a trajectory consisting of a vertical ascent phase, a horizontal leg, and a vertical descent phase is more desirable than a ballistic trajectory. Accordingly, the former type of trajectory has been used as a basis for comparison in the present investigation. Based on Table 2-1 from reference 1, for an altitude of 1,300 feet and a coast velocity of 900 ft/sec, the propellant consumption for the earth storable system is as follows:

$$\begin{bmatrix} n_{e} \int_{0}^{t_{b}} \frac{m_{p}}{p} d\tau \\ earth storable \end{bmatrix}$$
 = 484.74 lb<sub>m</sub>.

The mixture ratio for earth storables from Table 2-1 is

x = 2.

Thus by the definition of mixture ratio,

$$m_f = 161.58 \ lb_m$$
  
 $m_o = 323.16 \ lb_m$ 

for earth storables.

Exact evaluation of cryogenic propellant consumption can only be achieved by performing a dynamic analysis of the MFS (along the trajectory already described) in the same manner as that carried out in reference 1. Due to the shortage of time such an analysis was not feasible. Instead, as indicated in Section 2.2 and Table 2-1 of this report, the thrust and burning time for each leg of the trajectory were assumed to be the same for the cryogenics as for the earth storables. With this assumption,

$$\begin{bmatrix} \int_{0}^{t_{b}} Fd\tau \end{bmatrix}_{\substack{\text{earth} \\ \text{storables}}} = \begin{bmatrix} \int_{0}^{t_{b}} Fd\tau \end{bmatrix} \text{ cryogenics}$$
(14)

Now by definition of specific impulse

$$\frac{1}{gI_{sp}}\int_{0}^{\tau_{b}}Fd\tau = \int_{0}^{\tau_{b}}\dot{m}_{p}d\tau \qquad (15)$$

Thus

$$\begin{bmatrix} {}^{\tau}_{b} \\ {}^{n}_{e} \int {}^{m}_{p} d\tau \end{bmatrix}_{cryogenics} = \begin{bmatrix} {}^{\tau}_{b} \\ {}^{m}_{e} \int {}^{m}_{p} d\tau \\ {}^{sp (earth storables)} \\ storables \end{bmatrix}$$

Then

$$\begin{bmatrix} n_{e_{o}} \int_{m_{p}}^{\tau_{b}} d\tau \end{bmatrix} cryogenic = 484.74 \cdot \frac{300}{400}$$
$$= 363.54 \ 1b_{m} .$$

The mixture ratio for liquid oxygen and liquid hydrogen from Table 2-1

is

Then as before,

$$\begin{array}{c} m_{f} = 60.59 \quad 1b_{m} \\ m_{o} = 302.95 \quad 1b_{m} \end{array} \right\} \qquad \text{for cryogenics.}$$

It is important to note that the assumption of the cryogenic burning time and thrust being equal to their counterparts in the earth storable system permits a comparison on the basis of burning time. However, with its greater specific impulse, the cryogenic system should travel further than the earth storable system during the same time period.

### 2.5 Evaluation of Structure Factors

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The definitions of the structure factors are deceptively simple. These factors are not constants but are functions of a number of variables including:

> storage time storage environment 0 propellant mass 0 propellant density 0 propellant specific heat ο propellant latent heat of fusion 0 propellant latent heat of vaporization ο propellant storage temperature ο propellant storage pressure 0 propellant tank shape 0

- o propellant tank material density
- o propellant tank material tensile strength
- o insulation density
- o insulation thermal conductivity.

For spherical tanks the general relationship for a structure factor, C, is

$$C = (1.0 + \frac{m_{c}}{m_{x}}) \left\{ 1.0 + \frac{n_{s}n_{s}^{n} - \frac{p_{p}}{2}}{2\pi s} \left[ 3 + \frac{3n_{p}}{2\pi s} + \left( \frac{n_{p}}{2\pi s} \right)^{2} \right] + \frac{n_{s}n_{s}^{n} + \frac{n_{s}n_{s}^{n}}{2\pi s} \left[ \frac{q^{3}}{2\pi s} + \left( \frac{n_{s}n_{s}}{2\pi s} \right)^{2} \right] - \frac{q^{3}}{\left[ \frac{q^{3}}{\left[ q - 4\pi \left( \frac{3}{4\pi} \right)^{1/3} - \left( \frac{m_{s}n_{s}}{2\pi s} \right)^{2} - \left( \frac{1}{2\pi s} \right)^{2} + \left( \frac{n_{s}n_{s}}{2\pi s} \right)^{2} \right] \right\} (B-13)$$

with

$$Q = (m_x + m_c) c_p (T_i - T_i) + m_c h_c (T_{fr} \leq T_i \leq T_{sat})$$
(B-12)

where

- mass of fuel or oxidizer subjected to a change of phase  $(lb_m)$ m - total mass of fuel or oxidizer  $(lb_m)$ m, - density of fuel or oxidizer  $(lb_m/ft^3)$ ် X - density of shell material  $(lb_m/ft^3)$ ດ s n - safety factor - design pressure for a propellant container (psia) Pn - tensile strength of shell material (psi) ്ട - density of the insulation  $(1b_m/ft^3)$ <sup>r</sup> in

q - rate of heat flow from the shell (Btu/hr)

- thermal conductivity of the insulation (Btu / ft°F hr) k in mean temperature difference between inner and outer ٨T insulation temperature (°F) as defined by Eq. (B-8) - storage time (hr) T<sub>s</sub> - heat capacity (Btu/lb\_°F) с<sub>р</sub> <sup>T</sup>io - initial storage temperature (°R) T, - temperature of propellant or inner surface of insulation (°R) - temperature for change of phase (°R) T - latent heat generated by change of phase (Btu/lb\_) h - fusion temperature of propellant (°R) Tfr T<sub>sat</sub> - saturation temperature of propellant (°R).

The development of Eqs. (B-12) and (B-13) is provided in Appendix B.

The manner in which the structure factors are defined obviously requires that the minimum possible value for any such factor is 1.0. Previous studies (refs. 7 and 8) have indicated that for an 180-day period the storage system structure factor for cryogenics would be

 $2.2 \leq C_{cf} \leq 5.35$  (for liquid hydrogen) (17)

$$C_{so} = 1.11 \text{ (for liquid oxygen)}$$
 (18)

Because of the variation of the lunar environment during a period of 180 days, actual calculation of the factors C<sub>so</sub> and C<sub>sf</sub> for the earth storables using Eq. (B-13) is quite involved. Due to the physical characteristics of the earth storables, however, as shown in Table 2-2, the associated structure factors should be no greater than that for

# TABLE 2-2

# Physical Characteristics of Earth Storable and Cryogenic Propellants

# Earth Storable Propellants

Propellant	Fusion Temperature (°R)	Saturation Temperature at 14.7 psia (°R)	$\frac{\text{Density}}{(1b_m/ft^3)}$	Heat Capacity (Btu/lb_°F)
Nitrogen tetroxide (N <sub>2</sub> 0 <sub>4</sub> )	472	530	93	0.084
50:50 mixture of Hydrazine + UDMH	· 480	618	56.2	0.4503
Hydrazine (N <sub>2</sub> H <sub>4</sub> )	495	696	63.0	0.3552
UDMH	289	606	49.18	0.6285

Cryogenic Propellants

Propellant	Saturation Temperature at 50 psia (°R)	Density (1b <sub>m</sub> /ft <sup>3</sup> )	Latent heat of Vaporization (Btu/lb <sub>m</sub> )
Liquid Hydrogen	46	4.0	180
Liquid Oxygen	188	68.0	430

liquid oxygen. The assignment of a value of 1.11 for the C and C sf for the earth storables thus appears reasonable.

The propellant system structure factors  $C_{po}$  and  $C_{ps}$  involve only a short-term storage period of three hours, which is the duration of one mission. During this time period, variation of the lunar environment is negligible. Thus actual calculations of these structure factors by means of Eq. (B-13) in Appendix B are feasible for both the earth storables and the cryogenics. Because both a maximum and minimum design MFS propellant tank pressure must be considered as given in Table 2-1, two separate values of each structure factor,  $C_{po}$ and  $C_{pf}$ , will result.

In carrying out the calculations for C<sub>po</sub> and C<sub>pf</sub>, the necessity arises to select specific materials for use in the storage facilities under consideration. Accordingly, an analysis was made of the most desirable materials for such use. Based on this analysis, titanium was selected for the shell material and SI-91 was chosen for the insulation. The pertinent physical characteristics of these materials are found in Table 2-3.

Notice should be taken that cryogenic propellants are normally stored at their saturation temperature. Thus boil-off must be taken into account. Because of the form of Eq. (B-13), an optimization process is involved when boil-off occurs. An example of the necessary calculations is given in Appendix B. Obviously the minimum value of

### TABLE 2-3

### Physical Characteristics of Titanium and SI-91 Insulation

# Titanium

density  $(\cdot_s)$  - 281  $1b_m/ft^3$ tensile strength  $(\cdot_s)$  - 80,000 psi

# SI-91 Insulation

density ( $c_{in}$ ) - 7.8  $lb_m/ft^3$ thermal conductivity ( $k_{in}$ ) - 1 x 10<sup>-5</sup> Btu/hr ft °R the structure factor is the desired value. Figures 2-2 and 2-3 present the manner in which the values of  $C_{po}$  and  $C_{pf}$  for the cryogenic propellant vary with boil-off. As indicated by these figures

 $C_{pf} = 3.373$  $(P_{pf} = 1,000 \text{ psia})$  $C_{pf} = 1.373$  $(P_{pf} = 150 \text{ psia})$  $C_{po} = 1.136$  $(P_{po} = 1,000 \text{ psia})$  $C_{po} = 1.020$  $(P_{po} = 150 \text{ psia})$ 

Cryogenic Propellant

When boil-off does not occur, as is generally the case with earth storables, evaluation of the propellant system structure factors involves selection of the optimum storage temperature in the propellant tanks. A comparison of the maximum and minimum lunar temperatures provided in Table 2-1, with the boiling and freezing points of the earth storable propellants, as presented in Table 2-2, indicates that the possibility of these propellants freezing as well as boiling must be guarded against. The true thermal problem involved is transient in nature and its exact solution is too complex to attempt in this preliminary investigation. Instead, the approximate analysis presented in Appendix C may be used. Based on this analysis the optimum propellant temperatures,  $T_{io}$ , for 3-hour time periods are

$$T_{io} (N_2O_4) = 518^{\circ}R$$
$$T_{io} (N_2H_4+UDMH) = 618^{\circ}R \text{ (Saturation temperature).}$$





With these values, the propellant system structure factors are as follows:

$$C_{pf} = 1.162 \qquad (P_{pf} = 1,000 \text{ psia}) \\ C_{pf} = 1.028 \qquad (P_{pf} = 150 \text{ psia}) \\ C_{po} = 1.100 \qquad (P_{po} = 1,000 \text{ psia}) \\ C_{po} = 1.016 \qquad (P_{po} = 150 \text{ psia}) \\ \end{array} \right\} \qquad \text{Earth Storables.}$$

Notice should be taken that the calculated values of  $C_{po}$  and  $C_{pf}$  for earth storables are less than the corresponding calculated values of  $C_{po}$  and  $C_{pf}$  for cryogenics at the same propellant tank pressure.

#### 2.6 Calculation of Total System Mass

Based on the procedure described in subsections 2.4 and 2.5, values of  $n_e \int_0^{t_b} \dot{m}_p d\tau$ ,  $C_{sf}$ ,  $C_{so}$ ,  $C_{pf}$ , and  $C_{po}$  were obtained for use in Eq. (13), for both cryogenic and earth storable propellants. For the cryogenic  $C_{sf}$ , a value of 4.68 was assigned based on reference 7. For the thrust-to-weight ratios, those values listed in subsection 2.3, corresponding to 100 lb<sub>f</sub> thrust engines, were used. By means of Eq. (13), the total system mass was then calculated. The results are presented in Table 2-4. A sample of the calculations for the total systems mass is provided in Appendix D. For each propellant tank pressure, the calculated mass of the earth storable propulsion system was less than the calculated mass of the cryogenic propulsion system.

# TABLE 2-4

# Calculated Total Mass of Earth Storable and Cryogenic Propulsion Systems

a.	$m_t = 2056.31$	1b <sub>m</sub>	$(P_{p} = 1000 \text{ psia})$	$\mathbf{z}$	ormononic
b.	$m_{t} = 1900.0$	1b <sub>m</sub>	(P <sub>p</sub> = 150 psia)	5	cryogenic
c.	$m_t = 1708.35$	1b <sub>m</sub>	(P <sub>p</sub> = 1000 psia)	}	earth
d.	$m_t = 1663.22$	1b <sub>m</sub>	(P = 150 <b>psi</b> a)	5	Storable

#### 2.7 Variation of Propulsion Parameter

As noted previously, the value of the structure factor  $C_{sf}$  for cryogenic propellants used in the mass calculations was 4.68. This value is the primary reason for the cryogenic system being heavier than the corresponding earth storable system. Because of the range in possible values of  $C_{sf}$  for liquid hydrogen as indicated by Eq. (17), it is of interest to consider what effect variation of this factor has upon total system mass for cryogenics. Figure 2-4 represents this variation for a propellant tank pressure of 1,000 psia, while Figure 2-5 represents such variation for a pressure of 150 psia. In each figure, the value of the cryogenic  $C_{so}$  was 1.11. Also shown in each figure is the variation of system mass with earth storable  $C_{sf}$  for several different values of  $C_{so}$ . The points with letter designators correspond to the calculated values given in Table 2-4.

Examination of Figures 2-4 and 2-5 reveals that with a propellant tank pressure of 1,000 psia, if the cryogenic  $C_{sf}$  is less than 2.80 (with the cryogenic  $C_{so}$  held at 1.11), the cryogenic system mass will be less than the earth storable system mass given by Eq. (13), with the earth storable  $C_{sf}$  and  $C_{so}$  both equal to 1.11. Likewise, with a propellant tank pressure of 150 psia, if the cryogenic  $C_{sf}$  is less than 3.37, the cryogenic system mass will be less than the earth storable system mass. As before, this condition applies only when the cryogenic  $C_{so}$  and the earth storable  $C_{sf}$  and  $C_{so}$  are all equal to 1.11.





Such reductions in the cryogenic structure factor  $C_{sf}$ , as noted in reference 7, would represent considerable advances in the present state-of-the-art for cryogenic storage.

Time did not permit an extensive analysis of the effect of the variation of other important parameters on total system mass. An examination of the mathematical model as given by Eq. (13) does indicate, however, that as the number of missions increases, the mass of the earth storable systems probably increases more slowly than the mass of the cryogenic system. Also, for the cryogenic system, as the mixture ratio decreases, the mass of the system will increase. A decrease in the mixture ratio for the earth storable system, however, does not appreciably change the total system mass.

#### 3.0 CONCLUSIONS

Based on the results presented in Section 2.0, the use of cryogenic propulsion systems for the Manned Flying System does not appear to offer any improvement with regard to weight savings over the earth storable propulsion system. The basic reason for this fact is the heavier storage apparatus required for long-term storage of the cryogenic propellants. Liquid hydrogen, with its low density and low saturation temperature, is especially noticeable in this respect. The development of improved insulation might conceivably result in a cryogenic system which would be as light as the earth storable system. Such a development, however, would require considerable advances in the present state-of-the-art.

The most general conclusion that can be drawn from this preliminary investigation is that there appears to be no justification for using a cryogenic propulsion system in place of an earth storable system for the Manned Flying System. The possibility exists that a more detailed investigation might reveal advantages offered by the cryogenic system which were not uncovered in this study.

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#### APPENDIX A

### Survey of the Present State-of-the-Art for Low Thrust Cryogenic Propulsion Systems

Because of the scarcity of information concerning liquid oxygen and liquid hydrogen propulsion systems with thrusts ranging from 100  $lb_f$ to 500  $lb_f$ , written inquiries were sent to the following aerospace companies:

> Aerojet-General Corporation Sacremento, California

Jet Propulsion Laboratory Pasadena, California

Martin Marietta Corporation Denver, Colorado

Pratt and Whitney East Hartford, Connecticut

Reaction Motor Division Thiokol Chemical Corporation Denville, New Jersey

Rocketdyne Canoga Park, California

Space Technology Laboratories, Inc. Redondo Beach, California

The remainder of this appendix consists of a sample of the written inquiry, along with the responses received.

<u>S A M P L E</u>

17 May 1965

Company "X"

Dear Sirs:

Northrop Space Laboratories under Contract NAS8-20082 is presently providing support to the Aero-Astrodynamics Laboratory of NASA's George C. Marshall Space Flight Center, Huntsville, Alabama. As part of this contract, Northrop has been asked to determine the present stateof-the-art with regard to low thrust (100-500 lbs.), cryogenic fueled (liquid hydrogen and liquid oxygen) rocket engines. Any information or data with regard to such engines, which your company can provide without compromising your own proprietary interests, is requested.

Thank you,

Frank B. Tatom Senior Engineer Northrop Space Laboratories

FBT/1b

EROJET-GENERAL CORPORATION

O. BOX 1947 • SACRAMENTO, CALIFORNIA

DATE 7 June 1965 DHM:ecs

5-1042

### REPLY TO TECHNICAL INFORMATION REQUEST

P

TO: Northrop Space Laboratories Northrop Corporation Attn: Frank B. Tatom, Senior Engineer 6025 Technology Drive Huntsville, Alabama

REFERENCE(S):

AEROJET

GENERAL TIRE

GENERAL

 (a). Your Ltr., dtd. 17 May 1965, requesting information on the present state-of-the-art with regard to low thrust (100-500lbs.) cryogenic fueld (Liquid hydrogen and Liquid oxygen) rocket engines.

\_\_\_\_\_ Information requested is CLASSIFIED. Please resubmit your request through the agency monitoring the contract on which it is needed.

- Information requested was prepared under government contract. Please resubmit your request to:
- \_\_\_\_\_ Report(s) requested out-of-print, and our supply has been exhausted.
- Available from the Defense Documentation Center, as AD\_\_\_\_\_
- Available from NASA Scientific and Technical Information Facility, as,

N \_\_\_\_\_\_ X

- \_\_\_\_\_ Informàtion requested cannot be identified.
- X. Other This is to advise you that we have not published any research on this subject.

Signed:

D. T. Bedsole, Manager Technical Library Sacramento Plant

Jet Propulsion Laboratory

(No reply received)

Martin Marietta Corporation

٠

(No reply received)

Pratt & Whitney Aircraft DIVISION OF UNITED AIRCRAFT CORPORATION

July 21, 1965

Mr. Frank B. Tatom Northrop Space Laboratories 6025 Technology Drive Huntsville, Alabama

Dear Mr. Tatom:

Attached is a copy of the estimated design data on low thrust  $O_2/H_2$  engines which you requested in your letter to Pratt & Whitney Aircraft, dated 17 May 1965.

Because of the general nature of this request providing no specific application, the engines presented were not optimized but designed with representative values of chamber pressure, area ratio, and mixture ratio for low thrust engines. The design data presented utilizes a fixed bed catalyst for the ignition source.

It also includes two mixtures ratios. This is because the catalyst is limited to the temperature corresponding to a maximum mixture ratio of approximately 1.2. The upper limit varies according to the temperature of the inlet propellants. A toohigh temperature will result in fusion of the catalyst pellets. If a higher mixture ratio should be required, it can be obtained by adding additional oxygen downstream of the catalyst chamber where it will mix and ignite with the catalytic combustion gases. In this manner, the overall mixture ratio can be increased; however, it would require additional controls and a cooled thrust chamber.

The estimated engine weights include the solenoid operated inlet valves, injector, combustion chamber including the catalyst bed, and nozzles.

Page: 2 Date: July 21, 1965

We hope this information will be helpful to Northrop in determining state-of-the-art of low thrust oxygen/hydrogen engines. If there are any further questions, please do not hesitate in calling this office.

Very truly yours,

John G. Campbell

Encl

JGC/ah

cc: Mr. L. M. Bhalla

Thrust. Lb.	001	100	500	500	100	100	500	500
Chamber Pressure, psia	50	50	20	20	50	50	50	50
Vacuum Specific Impulse (95% Theoretical Shifting), Sec.	344	363	344	3,63	400	427.	400	.427
Mixture Ratio			-	-	<mark>م</mark> ا	2 2	S	S
Thrust Chamber Cooling	•	Radi	ation	•	• • • • •	Regene	rative	• • •
Fuel Flow. lbs/sec	.146	.138	.720	.690	.0415	.039	.208	. 195
Ovidizer Flow. lbs/sec	.146	.138	.720	069 <sup>.</sup>	.2085	.195	1.04	.975
Fnaine Weight, lbs	5.5	6.5	18	23	7	6	21	31
Area Ratio	01	30	10	30	10	30	01	30
length. Max. in.	11	16	22	40	10	15	20	31
Dia, Max, in.	Ņ	1		15.5	4	2	6	15

DESIGN DATA FOR LOW THRUST OXYGEN-HYDROGEN ENGINES

Space Technology Laboratories, Inc.

(No reply received)



A DIVISION OF NORTH AMERICAN AVIATION, INC.

6633 CANOGA AVENUE, CANOGA PARK, CALIFORNIA 91304

1 June 1965

IN REPLY REFER TO: 65RC8975

Northrop Space Laboratories Northrop Corporation 6025 Technology Drive Huntsville, Alabama

Attention: Mr. Frank B. Tatom Senior Engineer

Subject: Low Thrust Cryogenic Fueled Rocket Engines

Reference: (a) Northrop Request for Information dated 17 May 1965

Gentlemen:

Your reference (a) request for information concerning the present state-of-the-art with regard to low thrust (100-500 pounds) rocket engines utilizing liquid hydrogen and liquid oxygen as propellants has been received. Although we have extensive experience on engines using cryogenic propellants we regret to inform you that we do not have information or data on engines in thrust ranges as low as 100 to 500 pounds.

If additional information is desired, please contact Mr. W. P. O'Dea, 347-5651, extension 6064.

Very truly yours,

NORTH AMERICAN AVIATION, INC. Rocketdyne Division

A. F. Pietrowski Chief, Program Administration Spacecraft Engines

WPO'D:fml RC65-1265 08724RC

Thickol

## CHEMICAL CORPORATION

#### **REACTION MOTORS DIVISION**

DENVILLE, NEW JERSEY 07834 Phone Area Code 201, 627--7000 TWX 201, 627--3913

28 June 1965

Northrop Space Laboratories Northrop Corporation 6025 Technology Drive Huntsville, Alabama

Attention: Mr. F. B. Tatom

Subject: Cryogenic Low Thrust Engines

Reference: Northrop letter dated 17 May 1965, same subject

Enclosure: (1) PI 2-65, Attitude Control System Technology

Gentlemen:

Thiokol Chemical Corporation, Reaction Motors Division, regrets that we are not able to furnish you information at this time on low thrust, cryogenic attitude control engines. We are presently engaged in the earth storable, bi-propellant attitude control engine area. We have several contracts for attitude control engines, however, these are with N<sub>2</sub>O4 and monomethyl.

I am forwarding you our latest report on attitude control engine and system work that is presently being conducted at RMD. I trust that you will find this report helpful in the general area of attitude control systems - unfortunately it does not cover cryogenics. If after review of this report there is additional information that we may be able to supply please do not hesitate to contact either Mr. D. Culbertson at our Southern District Office, Huntsville, Alabama, Area Code 205-881-2661, or Mr. Paul O'Dea; Denville, New Jersey, Area Code 201-627-7000.

Very truly yours,

CHEMICAL CORPORATION THIOKC Reactio Motors Division unce Turner

Director of Applications Engineering

#### APPENDIX B

#### Analytical Expression for Structure Factors

#### A. General Development

The general structure factor, C, by definition is

$$C = \frac{\frac{m_{x} + m_{s} + m_{in} + m_{c} + m_{a}}{m_{x}}}{m_{x}}$$
(B-1)

where

 $m_x = mass of useful fuel or oxidizer within container (lb_m)$  $<math>m_s = mass of the shell (lb_m)$   $m_{in} = mass of insulation (lb_m)$   $m_c = mass of fuel or oxidizer subjected to a change of phase (lb_m)$  $m_a = mass of accessory equipment (lb_m).$ 

The mass,  $m_x$ , is fixed by the mission requirements. Thus the problem is to express the masses  $m_s$ ;  $m_{in}$ , and  $m_a$  in terms of  $m_x$  and/or other known quantities.

For spherical containers the development of such an expression for m\_ proceeds as follows:

 $m_{x} + m_{c} = \frac{4}{3} \pi r_{s}^{3} \rho_{x}$  (B-2)

where

$$r_s = radius of the shell (ft)$$
  
 $\rho_x = density of the fuel or oxidizer (1b_m/ft3).$ 

Then

$$\mathbf{r}_{s} = \left(\frac{3}{4\pi}\right)^{1/3} \left(\frac{\mathbf{m}_{x} + \mathbf{m}_{c}}{\mathbf{\rho}_{x}}\right)^{1/3}$$
(B-3)

For thin-shelled spheres, the thickness t is

$$t = n \frac{\frac{P r s}{2\sigma}}{s}$$
(B-4)

where

$$\eta$$
 = safety factor (1.67)  
 $P_p$  = design pressure for the shell (psia)  
 $\sigma_s$  = tensile strength of shell material (psi)

Thus the mass of the shell can be expressed as

$$\mathbf{m}_{s} = \frac{4}{3} \pi \rho_{s} \left[ \left[ \mathbf{r}_{s} + \frac{\eta \mathbf{p}_{s} \mathbf{r}_{s}}{2\sigma_{s}} \right]^{3} - \mathbf{r}_{s}^{3} \right]$$
$$= \frac{4}{3} \pi \rho_{s} \left[ \frac{3}{4\pi} \frac{\mathbf{m}_{x}^{+} \mathbf{m}_{c}}{\rho_{x}} \right] \left[ \left[ 1 + \frac{\eta \mathbf{p}_{p}}{2\sigma_{s}} \right]^{3} - 1 \right]$$
$$= \frac{\rho_{s} (\mathbf{m}_{x}^{+} \mathbf{m}_{c})}{\rho_{x}} \frac{\eta \mathbf{p}_{s}}{2\sigma_{s}} \left[ 3 + \frac{3\eta \mathbf{p}_{s}}{2\sigma_{s}} + \left( \frac{\eta \mathbf{p}_{p}}{2\sigma_{s}} \right)^{2} \right]$$
(B-5)

The mass of the insulation can be expressed as follows:

From Fourier's Law for steady-state radial heat transfer in a sphere

$$q = 4\pi \frac{r_i r_o}{r_o - r_i} k_{in} \Delta T \qquad (B-6)$$

$$\mathbf{r}_{o} = \frac{\mathbf{r}_{i} \mathbf{q}}{\mathbf{q} - 4\pi \mathbf{k}_{in} \Delta \mathbf{T} \mathbf{r}_{i}} \tag{B-7}$$

where

For the present case, the temperature difference is not always a constant. For simplicity, however, a mean temperature difference,  $\overline{\Delta T}$  can be defined such that,

$$Q_{s} = \int_{0}^{s} q d\tau = 4\pi \frac{r_{o}r_{i}}{r_{o}-r_{i}} k_{in} \overline{\Delta T} \tau_{s}$$
(B-8)

where

 $Q_s = total heat flow from the shell (Btu)$  $\tau_s = general storage time (hr).$ 

Then for the transient case under consideration,

$$\mathbf{r}_{o} = \frac{\mathbf{r}_{i} Q}{Q - 4\pi \mathbf{k}_{in} \Delta \mathbf{T} \mathbf{r}_{i} \mathbf{\tau}_{s}}$$
(B-9)

Now

$$\mathbf{r_i} = \mathbf{r_s} \begin{bmatrix} 1 + \frac{\eta P}{2\sigma_s} \end{bmatrix}$$
(B-10)

Then

.

ŧ

$$\begin{split} \mathbf{m}_{in} &= \frac{4}{3} \pi \rho_{in} (\mathbf{r}_{o}^{3} - \mathbf{r}_{i}^{3}) \\ &= \frac{4}{3} \pi \rho_{in} \left[ \frac{\mathbf{r}_{i}^{3} \mathbf{Q}^{3}}{(\mathbf{Q} - 4\pi \mathbf{k}_{in} \ \overline{\Delta T} \mathbf{r}_{i} \mathbf{r}_{s})^{3}} - \mathbf{r}_{i}^{3} \right] \\ &= \frac{4}{3} \pi \rho_{in} \mathbf{r}_{s}^{3} (1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}})^{3} \left\{ \frac{\mathbf{Q}^{3}}{\left[ \mathbf{Q} - 4\pi \mathbf{r}_{s} (1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}}) \mathbf{k}_{in} \ \overline{\Delta T} \mathbf{\tau}_{s} \right]^{3}} - 1 \right\} \\ &= \frac{4}{3} \pi \rho_{in} \left[ \frac{3}{4\pi} \frac{(\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c})}{\mathbf{p}_{x}} \right] \left[ 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}} \right]^{3} \left\{ \frac{\mathbf{Q}^{3}}{\left[ \mathbf{Q} - 4\pi \left( \frac{3}{4\pi} \right)^{1/3} \left( \frac{\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c}}{\mathbf{p}_{x}} \right)^{1/3} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}} \right) \mathbf{k}_{in} \ \overline{\Delta T} \mathbf{\tau}_{s} \right]^{3}} \right] \\ &= \frac{\rho_{in} (\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c})}{\mathbf{p}_{x}} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}} \right)^{3} \left\{ \frac{\mathbf{Q}^{3}}{\left[ \mathbf{Q} - 4\pi \left( \frac{3}{4\pi} \right)^{1/3} \left( \frac{\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c}}{\mathbf{p}_{x}} \right)^{1/3} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}} \right) \mathbf{k}_{in} \ \overline{\Delta T} \mathbf{\tau}_{s} \right]^{3}} \right] \\ &= \frac{\rho_{in} (\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c})}{\mathbf{p}_{x}} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}} \right)^{3} \left\{ \frac{\mathbf{Q}^{3}}{\left[ \mathbf{Q} - 4\pi \left( \frac{3}{4\pi} \right)^{1/3} \left( \frac{\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c}}{\mathbf{p}_{x}} \right)^{1/3} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}}}{2\sigma_{s}} \right) \mathbf{k}_{in} \ \overline{\Delta T} \mathbf{\tau}_{s} \right]^{3}} \right] \\ &= \frac{\rho_{in} (\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c})}{(\mathbf{p} - \mathbf{m}_{x}^{-1} \mathbf{c})} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}} \mathbf{p}}{2\sigma_{s}} \right)^{3} \left\{ \frac{\mathbf{Q}^{3}}{\left[ \mathbf{Q} - 4\pi \left( \frac{3}{4\pi} \right)^{1/3} \left( \frac{\mathbf{m}_{x}^{+\mathbf{m}} \mathbf{c}}{\mathbf{p}_{x}} \right)^{-1/3} \left( 1 + \frac{\mathbf{n}_{p}^{\mathbf{p}} \mathbf{p}}{2\sigma_{s}} \right) \mathbf{k}_{in} \ \overline{\Delta T} \mathbf{\tau}_{s} \right]^{3}} \right\}$$

From thermodynamics

$$Q = (m_x + m_c) c_p(T_{io} - T_i) + m_c h_c (T_{fr} \le T_i \le T_{sat})$$
(B-12)

where

$$c_p$$
 = heat capacity (Btu/lb\_m°F)  
 $T_{io}$  = initial storage temperature (°R)  
 $T_i$  = temperature of propellant or inner surface of insulation (°R)  
 $T_c$  = temperature for change of phase (°R)  
 $m_c$  = mass subjected to a change of phase (1b<sub>m</sub>)

 $h_c = heat generated by a change of phase (Btu/lb<sub>m</sub>)$  $<math>T_{fv} = fusion temperature (°R)$  $T_{sat} = saturation temperature (°R).$ 

The mass of accessory equipment,  $m_a$ , is difficult to express analytically. Furthermore, this mass is relatively small and does not appear to have an appreciable effect on the value of the structure factor. For these reasons  $m_a$  is neglected in this analysis.

By means of a combination of Eqs. (B-1), (B-5), and (B-11), the structure factor can be written

$$C = 1 + \frac{\rho_{s}(1+m_{c}/m_{x})n_{s}P_{p}}{2\rho_{x}\sigma_{s}} \left[ 3 + \frac{3^{n}}{2\sigma_{s}}P_{s} + \left(\frac{n}{2\sigma_{s}}P_{s}\right)^{2} \right] + \frac{\rho_{in}(1+m_{c}/m_{x})}{\rho_{x}} \left(1 + \frac{n}{2\sigma_{s}}P_{s}\right)^{3} \left\{ \frac{Q^{3}}{\left[Q-4\pi(\frac{3}{4\pi})^{1/3}\left(\frac{m_{x}+m_{c}}{\rho_{x}}\right)^{1/3}\left(1 + \frac{n}{2\sigma_{s}}P_{s}\right)k_{in}\overline{\Delta T}\tau_{s}\right]^{3} - 1 \right\}$$

$$= (1 + m_{c}/m_{x}) \left\{ 1 + \frac{\rho_{s} n^{P}_{p}}{2 \rho_{x} \sigma_{s}} \left[ 3 + \frac{3 n P_{p}}{2 \sigma_{s}} + \left( \frac{n P_{p}}{2 \sigma_{s}} \right)^{2} \right] + \frac{\rho_{i} n}{\rho_{x}} \left( 1 + \frac{nP_{p}}{2 \sigma_{s}} \right)^{3} \left[ \frac{Q^{3}}{\left[ Q - 4 \pi \left( \frac{3}{4 \pi} \right)^{1/3} \left( \frac{m_{x} + m_{c}}{\rho_{x}} \right)^{1/3} \left( 1 + \frac{nP_{p}}{2 \sigma_{s}} \right)^{k} i n^{\overline{\Delta T} \tau_{s}} \right]^{3} - 1 \right] \right\}$$
(B-13)

To demonstrate the procedure for using Eq. (B-13), the remaining portion of this appendix consists of sample calculations.

Sample Calculations of C for Oxygen B.

$$P_{p} = 1000 \text{ psia}$$

$$T_{io} = T_{i} = 188^{\circ}R$$

$$T_{c} = 188^{\circ}R$$

$$\rho_{x} = 68 \ 1b_{m}/ft^{3}$$

$$m_{x} = m_{o} = 302.95 \ 1b_{m}$$

$$\tau_{s} = \tau_{m} = 3 \text{ hr}$$

$$\overline{\Delta T} = T_{omax} - T_{io}$$

$$= 600 - 188$$

$$= 412^{\circ}R$$

$$k_{in} = 10^{-5} \text{ Btu/hr ft}^{\circ}F$$

$$\eta = 1.67$$

$$\sigma_{s} = 80,000 \text{ psi}$$

1

Assume

$$m_{c} = 15 \ 1b_{m}$$
.

Now from Eq. (B-13), for the case of the structure factor  $C_{po}$  for oxygen,

$$C_{po} = (1+m_{c}/m_{x}) \left\{ 1 + \frac{\rho_{s} n P_{p}}{2\rho_{x} \sigma_{s}} \left[ 3 + \frac{3nP_{p}}{2\sigma_{s}} + \left(\frac{n P_{p}}{2\sigma_{s}}\right)^{2} \right] + \frac{\rho_{in}}{\rho_{x}} \left(1 + \frac{nP_{p}}{2\sigma_{s}}\right)^{3} \left[ \frac{Q^{3}}{\left[ Q - 4\pi \left(\frac{3}{4\pi}\right)^{1/3} \left(\frac{m_{s} + m_{c}}{\rho_{x}}\right)^{1/3} \left(1 + \frac{nP_{p}}{2\sigma_{s}}\right) k_{in} \overline{\Delta T} \tau_{m} \right]^{3}}{\left[ \left[ Q - 4\pi \left(\frac{3}{4\pi}\right)^{1/3} \left(\frac{m_{s} + m_{c}}{\rho_{x}}\right)^{1/3} \left(1 + \frac{nP_{p}}{2\sigma_{s}}\right) k_{in} \overline{\Delta T} \tau_{m} \right]^{3}} \right] \right\}$$
(B-13a)

where from Eq. (B-12)

$$Q = (m_{x}+m_{c}) c_{p} (T_{io}-T_{i}) + m_{c}h_{c} (T_{fr}-T_{i}-T_{sat})$$
(B-12)

Then

$$Q = 5.430 = 6450 \text{ Btu}$$

$$C_{po} = (1 + \frac{15}{302.95}) \left\{ 1 + \frac{281.7 \cdot 1.67 \cdot 1000}{2 \cdot 68 \cdot 80,000} \left[ 3 + \frac{3 \cdot 1.67 \cdot 1000}{2 \cdot 80,000} + \left( \frac{1.67 \cdot 1000}{2 \cdot 80,000} \right)^2 \right] + \frac{7 \cdot 8}{68} \left( 1 + \frac{1.67 \cdot 1000}{2 \cdot 80,000} \right)^3$$

$$\left[ \frac{6450^3}{\left[ 6450 - 4\pi \left( \frac{3}{4\pi} \right)^{1/3} \left( \frac{317.95}{68} \right)^{1/3} \left( 1 + \frac{1.67 \cdot 1000}{2 \cdot 80,000} \right) \cdot 10^{-5} \cdot 412 \cdot 3 \right]^3 - 1 \right] \right\}$$

$$= 1.04952 \left[ 1 + .04324 \left[ 3 + .031313 + .00011 \right] + \frac{7 \cdot 8}{68} \cdot 1.03164 \left[ \frac{268 \cdot 336125 \cdot 10^9}{\left[ 6450 - (4\pi)(.62)(1.67)(1.014)(1.236)(10^{-2}) \right]^3} - 1 \right] \right]$$

= 1.04952(1+.13108 + .00000895)

= 1.187

$$T_{i} = T_{io} - \int_{0}^{\tau} \frac{qd\tau}{m_{x}c_{p}}$$
(C-3)

Then if the thermal capacitance of the insulation is neglected,

$$\int_{0}^{\tau} q d\tau = \int_{0}^{\tau} \frac{4\pi r_{o} r_{i} k_{in}}{r_{o} - r_{i}} \left[ (T_{io} - \int_{0}^{\tau} \frac{q d\tau}{m_{x} c_{p}}) - T_{o} \right] d\tau \qquad (C-4)$$

or

$$q = \frac{-4\pi r_o r_i k_{in}}{r_o - r_i} \left[ \int_0^{\tau} \frac{qd\tau}{m_x c_p} + T_o - T_{io} \right]$$
(C-5)

or

$$\frac{dq}{d\tau} = \frac{-4\pi r_o r_i k_{in}}{(r_o - r_i) m_x c_p} q \qquad (C-6)$$

Let

$$\lambda \equiv \frac{4\pi r_0 r_1 k_{in}}{(r_0 - r_1) m_c r_p}$$
(C-7)

Then

$$q = c_1 e^{-\lambda \tau}$$
 (C-8)

In order to satisfy the boundary conditions,

$$C_{1} = m_{x}c_{p} \lambda (T_{io}-T_{o})$$
 (C-9)

Thus

$$q = m_{x}c_{p} \lambda (T_{io}-T_{o}) e^{-\lambda\tau}$$
(C-10)

52

or

#### APPENDIX C

### <u>Calculation of Optimum Storage Temperature and</u> <u>Insulation Thickness for Earth Storable Propellants</u>

The actual heat transfer problem involved in the flow of heat from the propellant containers is quite complex due to its transient nature. In the analysis which follows, the thermal capacitance of the insulation is ignored. This simplification permits a closed-form solution but the accuracy of the solution is reduced. For purposed of this preliminary investigation, however, the technique appears sufficient.

A. General Development

As already noted

$$Q_{s} = \int_{0}^{\tau_{s}} q d\tau \qquad (B-8)$$

Now, for steady radial heat transfer

$$q = \frac{4\pi r_o r_i k_{in}}{r_o - r_i} \qquad (T_i - T_o)$$

where

 $T_{a}$  = outside temperature of insulation (°R)

T<sub>i</sub> = temperature of propellant or inner surface of insulation (°R).

Also, if no change of phase occurs,

$$(T_{io}-T_{i}) m_{x}c_{p} = \int_{0}^{1} q d\tau \qquad (C-2)$$

Now

$$T_{i} = T_{io} - \int_{0}^{\tau} \frac{m_{x}c_{p}\lambda(T_{io}-T_{o})}{m_{x}c_{p}} e^{-\lambda\tau}d\tau$$
$$= T_{io} + (T_{io}-T_{o}) (e^{-\lambda\tau} - 1)$$
$$= T_{o} + (T_{io}-T_{o}) e^{-\lambda\tau} \qquad (C-11)$$

In order to avoid both freezing and boiling for a time period of  $\tau_{m}^{},$ 

$$T_{sat} \ge T_{omax} + (T_{io} - T_{omax}) e^{-\lambda T_m}$$
 (C-12)

and

$$T_{fr} \ge T_{omin} + (T_{io} - T_{omin}) e^{-\lambda \tau_m}$$
 (C-13)

where

B. Optimum Storage Temperature for N204

For  $N_2O_4$ , as noted in Tables 2-1 and 2-2,

and

T<sub>fr</sub> > T<sub>omin</sub>.

Under these conditions logic reveals that the most desirable condition involves

$$T_{sat} = T_{omax} + (T_{io} - T_{omax})e^{-\lambda \tau}m \qquad (C-12a)$$

and

$$T_{fr} = T_{omin} + (T_{io} - T_{omin})e^{-\lambda \tau m}$$
 (C-13a)

These equations can be arranged to yield

$$T_{io} = (T_{sat} - T_{omax})e^{\lambda T_m} + T_{omax}$$
(C-14)

and

$$T_{io} = (T_{fr} - T_{omin})e^{\lambda \tau m} + T_{omin}$$
(C-15)

or

$$(T_{sat}-T_{omax})e^{\lambda \tau}m + T_{omax} - (T_{fr}-T_{omin})e^{\lambda \tau}m - T_{omin} = 0 \qquad (C-16)$$

Thus

$$\lambda = \frac{1}{\tau_{\rm m}} \ln \left[ \frac{T_{\rm omax} - T_{\rm omin}}{T_{\rm omax} - T_{\rm sat} + T_{\rm fr} - T_{\rm omin}} \right]$$
(C-17)

By means of the definition of  $\lambda$ , as given in Eq. (C-7),  $r_0$  can be expressed as

$$\mathbf{r}_{o} = \frac{\mathbf{r}_{i} \mathbf{m}_{o} \mathbf{c}_{p} \ln \left[ \frac{T_{omax} - T_{omin}}{T_{omax} - T_{sat} + T_{fr} - T_{omin}} \right]}{\mathbf{m}_{o} \mathbf{c}_{p} \ln \left[ \frac{T_{omax} - T_{omin}}{T_{omax} - T_{sat} + T_{fr} - T_{omin}} \right] - 4\pi \mathbf{r}_{i} \mathbf{k}_{in} \mathbf{\tau}_{m}}$$
(C-18)

If SI-91 insulation is used, for  $N_2O_4$ 

$$r_{o} = 1.0001 r_{i}$$
  
 $\lambda = .05217 hr^{-1}$ 

and

$$T_{io} = 518 \, ^{\circ}R.$$

C. Optimum Storage Temperature for a 50:50 mixture of  $\mathrm{N_2H_4}$  and UDMH

For this fuel, as shown in Tables 2-1 and 2-2,

and

Under these conditions, from basic thermodynamics, it is desirable to have

$$T_{io} = T_{sat}$$
(C-19)

By means of Eq. (C-13)

$$T_{fr} = T_{omin} + (T_{sat} - T_{omin})e^{-\lambda \tau}m \qquad (C-20)$$

or

$$\lambda = \frac{1}{\tau_{\rm m}} \ln \left[ \frac{T_{\rm sat} - T_{\rm omin}}{T_{\rm fr} - T_{\rm omin}} \right]$$
 (C-21)-

Then by the definition of  $\lambda$ ,

$$\mathbf{r}_{o} = \frac{\mathbf{r}_{i} \ \mathbf{m}_{f} \ \mathbf{C}_{p} \ \ln\left[\frac{\mathbf{T}_{sat} - \mathbf{T}_{omin}}{\mathbf{T}_{fr} - \mathbf{T}_{omin}}\right]}{\mathbf{m}_{f} \ \mathbf{C}_{p} \ \ln\left[\frac{\mathbf{T}_{sat} - \mathbf{T}_{omin}}{\mathbf{T}_{fr} - \mathbf{T}_{omin}}\right] - 4\pi\mathbf{r}_{i} \ \mathbf{k}_{in} \ \mathbf{T}_{m}}$$
(C-22)

With SI-91 insulation,

$$r_0 = 1.0000115 r_1$$

and

$$\lambda = 0.1321 \text{ hr}^{-1}$$
.

Based on Eq. (C-19) and Table 2-2,

 $T_{io} = 618 \ ^{\circ}R.$ 

# APPENDIX D

# Sample Calculations for Total System Mass

The calculations which follow are provided to demonstrate the procedure for calculation of total system mass as expressed by Eq. (13).

Propellants:  $N_2O_4$  and 50:50 mixture of  $N_2H_4$  and UDMH

$$m_{e} = 5$$
  

$$F = 100 \ 1b_{f}$$
  

$$m_{m} = 3$$
  

$$C_{fw} = 14$$
  

$$x = 2$$
  

$$m_{e} \int_{0}^{\tau_{b}} \dot{m} d\tau = 484.74 \ 1b_{m}$$
  

$$P_{p} = 1,000 \ psia$$
  

$$C_{pf} = 1.162$$
  

$$C_{po} = 1.100$$
  

$$C_{sf} = 1.11$$
  

$$C_{so} = 1.11$$

Then

$$m_{t} = n_{e} \left[ \frac{F}{g c_{fw}} + (\frac{1}{x+1}) \left[ (C_{pf} - 1 + n_{m} c_{sf}) + x (C_{po} - 1 + n_{m} c_{so}) \right] \int_{0}^{\tau} b \dot{m}_{p} d\tau \right]$$
(13)  
= 5  $\cdot \frac{100}{14} + \frac{1}{2+1} \left[ (1.162 - 1 + 3 \cdot 1.11) + 2(1.100 - 1 + 3 \cdot 1.11) \right]$ 484.74

$$= 35.68 + \frac{1}{3} (3.492 + 6.86) 484.74$$
$$= 35.68 + 1672.67$$
$$= 1708.35 \quad 1b_{m}.$$

MANNED FLYING SYSTEMS (MFS)

CRYOGENIC ENGINE

Research & Analysis Section Tech. Memo. #113

Prepared under Contract NAS8-20082

by

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July 1965

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