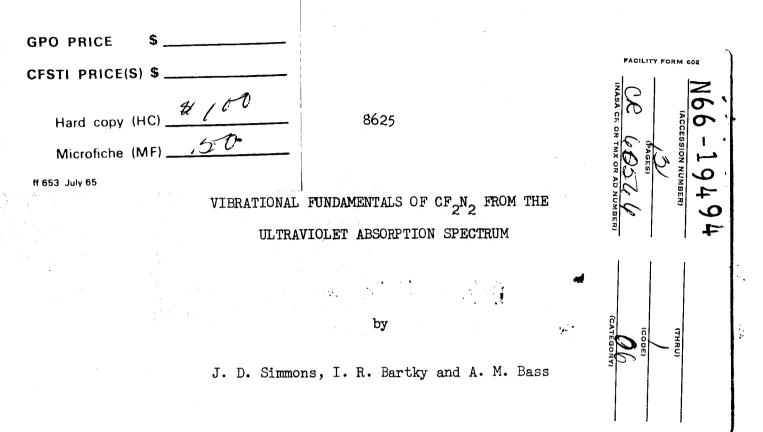


6. N

## UNPUBLISHED I RELIMINATE PATT

# NATIONAL BUREAU OF STANDARDS REPORT



Technical Report

to National Aeronautics and Space Administration Washington, D. C.



U.S. DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS

## THE NATIONAL BUREAU OF STANDARDS

The National Bureau of Standards is a principal focal point in the Federal Government for assuring maximum application of the physical and engineering sciences to the advancement of technology in industry and commerce. Its responsibilities include development and maintenance of the national standards of measurement, and the provisions of means for making measurements consistent with those standards; determination of physical constants and properties of materials; development of methods for testing materials, mechanisms, and structures, and making such tests as may be necessary, particularly for government agencies; cooperation in the establishment of standard practices for incorporation in codes and specifications; advisory service to government agencies on scientific and technical problems; invention and development of devices to serve special needs of the Government; assistance to industry, business, and consumers in the development and acceptance of commercial standards and simplified trade practice recommendations; administration of programs in cooperation with United States business groups and standards organizations for the development of international standards of practice; and maintenance of a clearinghouse for the collection and dissemination of scientific. technical, and engineering information. The scope of the Bureau's activities is suggested in the following listing of its four Institutes and their organizational units.

Institute for Basic Standards. Electricity. Metrology. Heat. Radiation Physics. Mechanics. Applied Mathematics. Atomic Physics. Physical Chemistry. Laboratory Astrophysics.\* Radio Standards Laboratory: Radio Standards Physics: Radio Standards Engineering.\*\* Office of Standard Reference Data.

Institute for Materials Research. Analytical Chemistry. Polymers. Metallurgy. Inorganic Materials. Reactor Radiations. Cryogenics.\*\* Office of Standard Reference Materials.

Central Radio Propagation Laboratory.\*\* Ionosphere Research and Propagation. Troposphere and Space Telecommunications. Radio Systems. Upper Atmosphere and Space Physics.

Institute for Applied Technology. Textiles and Apparel Technology Center. Building Research. Industrial Equipment. Information Technology. Performance Test Development. Instrumentation. Transport Systems. Office of Technical Services. Office of Weights and Measures. Office of Engineering Standards. Office of Industrial Services.

# CASE FILE COPY

<sup>\*</sup> NBS Group, Joint Institute for Laboratory Astrophysics at the University of Colorado. \*\* Located at Boulder, Colorado.

# NATIONAL BUREAU OF STANDARDS REPORT

NBS PROJECT

NBS REPORT

221-11-2210461

February 1, 1965

8625

VIBRATIONAL FUNDAMENTALS OF CF2N2 FROM THE ULTRAVIOLET ABSORPTION SPECTRUM

by

J. D. Simmons, I. R. Bartky and A. M. Bass

Technical Report

to

National Aeronautics and Space Administration

Washington, D. C.

NASA Order No. R-64

#### IMPORTANT NOTICE

NATIONAL BUREAU OF STANDARDS REPORTS are usually preliminary or progress accounting documents intended for use within the Government. Before material in the reports is formally published it is subjected to additional evaluation and review. For this reason, the publication, reprinting, reproduction, or open-literature listing of this Report, either in whole or in part, is not authorized unless permission is obtained in writing from the Office of the Director, National Bureau of Standards, Washington 25, D.C. Such permission is not needed, however, by the Government agency for which the Report has been specifically prepared if that agency wishes to reproduce additional copies for its own use.



U. S. DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS

## Preface

A manuscript based upon this report will be submitted for publication as a Letter to the Editor in the Journal of Molecular Spectroscopy.

# VIBRATIONAL FUNDAMENTALS OF CF2N2 FROM THE ULTRAVIOLET ABSORPTION SPECTRUM<sup>\*\*</sup>

by

J. D. Simmons \*\*, I. R. Bartky and A. M. Bass

### Abstract

The ultraviolet absorption spectrum of  $GF_2N_2$  has been observed and found to include bands of two types. All of the bands of one type can be analyzed in terms of four totally symmetric vibrational frequencies plus a prominent sequence of difference bands. Some bands of the second type can be interpreted as vibronically allowed fundamentals induced by the nontotally symmetric torsional vibration. This latter assignment reveals enough information to account for the difference bands of the first type. All ground state fundamental assignments are compatible with earlier infrared and Raman results.

Work supported in part by the National Aeronautics and Space Administration.

<sup>\*\*</sup> National Academy of Sciences-National Research Council Postdoctoral Research Associate at the National Bureau of Standards.

 $CF_2N_2$  has recently been prepared and characterized by Mitsch (<u>1</u>). The infrared and Raman spectra have been observed and interpreted in terms of a cyclic diazirine structure as opposed to the isomeric linear diazomethane structure (<u>2</u>). The following preliminary vibrational analysis of the electronic spectrum is offered as further support for that assignment.

The ultraviolet absorption spectrum has been observed in the 3750-3000  $\stackrel{o}{A}$  region under high resolution and with sufficient path-lengths, using a multiple reflection cell, to observe the vibronically allowed ground state fundamentals. The system contains bands of at least two distinct types; a strong group of broad bands with three intensity maxima and a much weaker group with one prominent sharp maximum.

On the basis of a cyclic difluorodiazirine with  $C_{2v}$  symmetry, there are four totally symmetric vibrations of species  $A_1$ , and five nontotally symmetric vibrations - one, two and two of species  $A_2$ ,  $B_1$  and  $B_2$ , respectively. Vibronic selection rules allow only totally symmetric fundamental and combination frequencies to appear in the electronic spectrum with the same band-type as the origin band. Any nontotally symmetric fundamental which appears with appreciable intensity due to vibronic interaction must have a band-type different from that of the origin.

All of the broad-type bands can easily be interpreted in

- 2 -

terms of four totally symmetric vibrational frequencies accompanied by a strong sequence of difference bands with the origin at 28374.6 cm<sup>-1</sup>. The assignment for these totally symmetric vibrations is summarized in Table I where the results from the ultraviolet data are compared with those from Ref. 2. The agreement for the ground state fundamentals is very good except for  $v_3$ . Here, although the deviation between the infrared and the ultraviolet data is uncomfortably large, the assignment in each case is rather convincing.

The relative intensities of the upper state progressions, in decreasing order, are  $v_1$  (4 members),  $v_4$  (2 members),  $v_3$ (1 member) and  $v_2$ . While the ground state  $v_2$  fundamental appears in the long-path experiments, it has much less intensity than does  $v_1$ , even though it has a more favorable Boltzmann distribution. The  $v_2$  upper state fundamental is not observed and is presumably overlapped by stronger bands in the expected region. Since  $v_1$  and  $v_4$  may be described as the symmetric NN stretch and symmetric  $CF_2$  deformation respectively, one would conclude from Franck-Condon considerations that the most significant change in geometry between the two electronic states is a stretching of the nitrogen-nitrogen bond which expands the ring and effects some change in the FCF angle.

It should also be noted that the nontotally symmetric  $v_5$  torsional vibration (species  $A_2$ ) which has a Raman active

- 3 -

frequency of 451 cm<sup>-1</sup> (2) also appears in the electronic spectrum by vibronic interaction. As expected from the selection rules, the ground and upper state fundamentals -- with frequencies of 448 cm<sup>-1</sup> and 327 cm<sup>-1</sup>, respectively, from the broad-type origin -are bands of the sharp type. This low frequency vibration offers a convincing assignment for the above mentioned strong sequence of broad-type difference bands which has a 120.6 cm<sup>-1</sup> interval.

The  $v_5$  assignment does not, however, account for all of the sharp bands observed. There is preliminary evidence for another transition in this region. Further work on this system is in Progress including an attempt to analyze the partially resolved rotational structure by an asymmetric rotor contour technique.

The authors are grateful to Dr. Ronald Mitsch of the Minnesota Mining and Manufacturing Company for supplying the sample and to Dr. Norman Craig for making a preprint of Ref. 2 available in advance of publication.

- 4 -

## REFERENCES

- (1). Ronald A. Mitsch, J. Heterocyclic Chem. 1, 59 (1964).
- (2). C. W. Bjork, N. C. Craig, R. A. Mitsch and J. Overend,J. Am. Chem. Soc. (in press).

IR (vapor)	Raman (liq.) Rof. 2 (cm <sup>-1</sup> )	Ultraviolet (vapor) This work	
Ref. 2 (cm <sup>-1</sup> )		v" (cm <sup>-1</sup> )	v' (cm <sup>-1</sup> )
1563	1 560	1564.0	1436.9
1282	1280		
805	804	775.0	643.3
502	500	500.2	516.6
	Ref. 2 (cm <sup>-1</sup> ) 1563 1282 805	Ref. 2 Ref. 2   (cm <sup>-1</sup> ) (cm <sup>-1</sup> )   1563 1560   1282 1280   805 804	Ref. 2 Ref. 2 v"   (cm <sup>-1</sup> ) (cm <sup>-1</sup> ) (cm <sup>-1</sup> )   1563 1560 1564.0   1282 1280 1282.9   805 804 775.0

Table I. Vibrational Assignment for the  ${\rm A_1}$  Species in  ${\rm CF_2N_2}$