

STUDY  
HIGH DIELECTRIC CONSTANT THIN FILM CAPACITOR MATERIALS

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**TRW SYSTEMS**

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## I. INTRODUCTION

This report covers the results of a study program conducted to investigate the high vacuum approach for thin film capacitors with high specific capacitances in order to permit reductions in the size of micro-circuits. The study was performed in accordance with the requirements of NASA Manned Spacecraft Center, General Research Procurement Branch, under Contract No. NAS 9-5592. In general, materials of the  $ABO_3$  type and having dielectric constant "k" were investigated.

The study program was separated into three phases as follows:

- Phase 1. This phase consisted of a literature survey of materials for investigation. The survey included research of material properties in both their bulk and thin film form. The results of the survey, including a bibliography, were previously reported in an interim report dated March 25, 1966. The interim report included five materials recommended for technical exploration in Phase 2.
- Phase 2. This phase was the actual study portion of the contract and was started upon receipt of NASA's approval of the materials recommended for further study. Material investigation was limited to the following materials:
- 1) Sodium niobate
  - 2) Solid solution of barium zirconate
  - 3) Solid solution of lead zirconate

Numberous depositional approaches, such as laser, electron bombardment, flash evaporation, and dual evaporation, were explored. The physics, crystallography, and chemistry of the resultant thin films were established by utilization of optical and electron microscopy, X-ray diffraction, and spectrographic analysis. The electrical characteristics of the thin films when utilized as capacitor dielectrics were established with regard to temperature coefficient of capacitance, specific capacitance, dielectric constant, frequency response, voltage breakdown characteristics and stability.

- Phase 3. This phase is to consist of fabrication and delivery of working capacitors to the NASA Manned Spacecraft Center in Houston along with appropriate drawings and process procedures utilized in the fabrication of the capacitors. This phase is to cover a two-month time spectrum and will be begun upon approval of Phase 2 by NASA.

Mr. L. A. Darling, Manager of the TRW Systems Microelectronics Department, served as program manager; and Mr. R. P. Radke was in charge of the investigation in the capacity of project engineer. Mr. Radke was assisted by thin film development engineers of the Microelectronics Department, as well as specialists in electron microscope and X-ray diffraction evaluation of thin films, as shown in Figure 1.

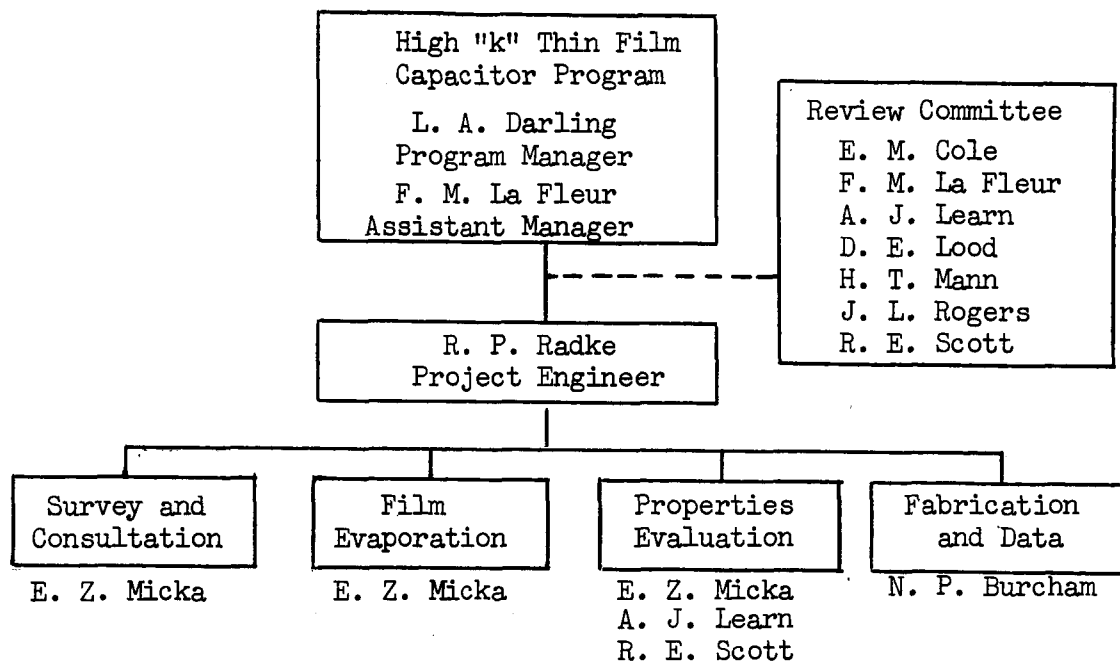


Figure 1. Program Organization

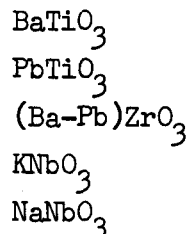
Sources used as references throughout this study are included in the bibliography at the end of this report.

## II. BACKGROUND

NASA requested in "Exhibit A" of the contract that a specified list of twenty compounds be surveyed during the Phase 1 effort. All of these plus an additional nine materials were surveyed. A list of these materials appears in Table I, along with references to their room temperature crystal structure, electrical properties, Curie point, and dielectric constant.

The previously submitted interim report included a discussion of ferroelectricity and antiferroelectricity and the bulk properties of numerous compounds and the thin film properties of the same compounds. The materials were grouped and discussed according to type (i.e., titanates, zirconates, hafnates, stannates, niobates, tantalates, and other materials).

The materials recommended in the interim report for further study in Phase 2 were:



In a communication received from Marion C. Owens, Contracting Officer for NASA Houston (reference BG731-39, C. D. Stamps), dated April 12, 1966, and received by TRW Systems on April 19, 1966, NASA authorized TRW, Inc. to proceed with Phase 2, limited to future investigation of the following materials only:

$$\begin{array}{l} \text{Sodium niobate} \\ \text{Solid solution of barium zirconate} \\ \text{Solid solution of lead zirconate} \end{array}$$

The remaining materials recommended by TRW Systems for further investigation were to be explored only in the event that investigation of the above noted compounds proved fruitless.

TABLE I. COMPOUNDS SURVEYED DURING PHASE 1

| MATERIAL   | ROOM TEMPERATURE CRYSTAL STRUCTURE | ELECTRICAL PROPERTIES | CURIE POINT | DIELECTRIC CONSTANT | TEST CONDITIONS |        | COMMENTS       |
|--|------------------------------------|-----------------------|-------------|---------------------|-----------------|--------|----------------|
|  |                                    |                       |             |                     | TEMP.           | FREQ.  |                |
| BaTiO <sub>3</sub>                                     | Tetragonal                         | Ferroelectric         | 120°C       | 1400                | 25°C            | LKC    | Thin film form |
| PbTiO <sub>3</sub>                                     | Tetragonal                         | Ferroelectric         | 490°C       | 450                 | 25°C            | LKC    |                |
| SrTiO <sub>3</sub>                                     | Cubic                              | Ferroelectric         | ~40°K       | 250                 | 25°C            | 50 cps |                |
| CaTiO <sub>3</sub>                                     | Orthorhombic                       | Paraelectric          | N/A         | 143                 | 25°C            | 50KC   |                |
| CdTiO <sub>3</sub>                                     | Orthorhombic                       | Ferroelectric         | -218°C      | 250                 | 25°C            | ---    |                |
| PbZrO <sub>3</sub>                                     | Orthorhombic                       | Antiferroelectric     | 233°C       | ~100                | 25°C            | ---    |                |
| BaZrO <sub>3</sub>                                     | Cubic                              | Paraelectric          | N/A         | low                 | ---             | ---    |                |
| (Ba <sub>0.4</sub> Pb <sub>0.6</sub> )ZrO <sub>3</sub> | ---                                | Paraelectric          | N/A         | 2600                | 25°C            | LKC    |                |
| SrZrO <sub>3</sub>                                     | Cubic                              | Paraelectric          | N/A         | low                 | ---             | ---    |                |
| CaZrO <sub>3</sub>                                     | Orthorhombic                       | Paraelectric          | N/A         | low                 | ---             | ---    |                |
| PbHfO <sub>3</sub>                                     | Tetragonal                         | Antiferroelectric     | 215°C       | 90                  | 25°C            | 10KC   |                |
| SrHfO <sub>3</sub>                                     | ---                                | Paraelectric          | N/A         | low                 | ---             | ---    |                |
| BaHfO <sub>3</sub>                                     | ---                                | Paraelectric          | N/A         | low                 | ---             | ---    |                |
| BaSnO <sub>3</sub>                                     | Cubic                              | Paraelectric          | N/A         | 17                  | 25°C            | LKC    |                |
| SrSnO <sub>3</sub>                                     | Cubic                              | Paraelectric          | N/A         | 17                  | 25°C            | LKC    |                |
| CaSnO <sub>3</sub>                                     | Orthorhombic                       | Paraelectric          | N/A         | 15                  | 25°C            | LKC    |                |
| MgSnO <sub>3</sub>                                     | ---                                | ---                   | ---         | 62                  | 25°C            | LKC    |                |
| Bi <sub>2</sub> (SnO <sub>3</sub> ) <sub>2</sub>       | ---                                | ---                   | ---         | 34                  | 25°C            | LKC    |                |
| PbSnO <sub>3</sub>                                     | ---                                | ---                   | ---         | 16                  | 25°C            | LKC    |                |
| CaSnO <sub>3</sub>                                     | ---                                | ---                   | ---         | 16                  | 25°C            | LKC    |                |
| NiSnO <sub>3</sub>                                     | ---                                | ---                   | ---         | 235                 | 25°C            | LKC    |                |
| KNbO <sub>3</sub>                                      | Orthorhombic                       | Ferroelectric         | 434°C       | ~500                | 25°C            | LKC    |                |
| NaNbO <sub>3</sub>                                     | Orthorhombic                       | Antiferroelectric     | 355°C       | 575                 | 25°C            | LKC    |                |
| Cd <sub>2</sub> Nb <sub>2</sub> O <sub>7</sub>         | ---                                | ---                   | 170°K       | 320/500             | 25°C            | ---    |                |
| Pb(NbO <sub>3</sub> ) <sub>2</sub>                     | ---                                | ---                   | 575°C       | 300                 | 25°C            | ---    |                |
| KTaO <sub>3</sub>                                      | Cubic                              | Ferroelectric         | ~13°K       | Below 500           | 25°C            | ---    |                |
| NaTaO <sub>3</sub>                                     | Orthorhombic                       | Paraelectric          | 475°C       | ---                 | ---             | ---    |                |
| WO <sub>3</sub>  | Monoclinic                         | Antiferroelectric     | -50°C       | ---                 | ---             | ---    |                |
| SbSI   | Monoclinic                         | Ferroelectric         | 22°C        | 50,000              | 22°C            | ---    |                |

--- No information available

N/A Not applicable

### III. TEST CAPACITOR GEOMETRIES

The test capacitor configuration utilized for the depositions in the earlier part of the program allowed for four capacitors per substrate, all using a common base electrode (Figure 2).

During the latter stages of Phase 2 when successful capacitor depositions were being performed, an improved test geometry was utilized (Figure 3). This particular configuration allowed for sixteen capacitors per substrate and obviously produced a greater number of units for test for any given deposition cycle. This second test pattern suffered from the shortcoming of producing "cross talk" due to the presence of common base and counter electrodes when a short in a given capacitor was obtained.

A modification of the second test pattern was devised and is the one currently in use (Figure 4). This configuration still allows for sixteen capacitors per substrate, but each now has an isolated counter electrode. In addition, the pattern has been so designed that a substantial proportion of reflective material overlaps the dielectric step, thereby facilitating the taking of interferometric thickness measurement directly on the substrate.

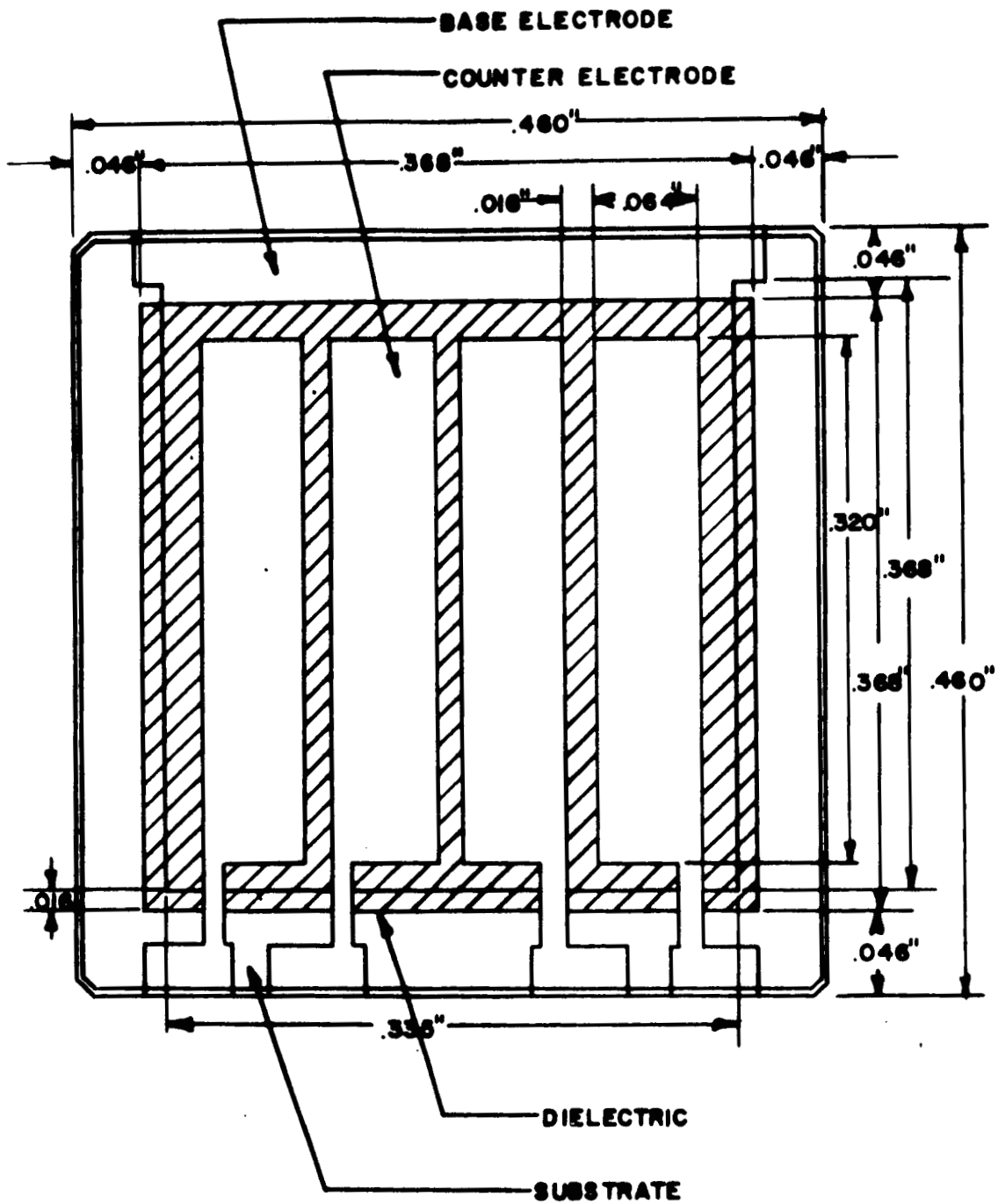


Figure 2. Capacitor Test Pattern (First)

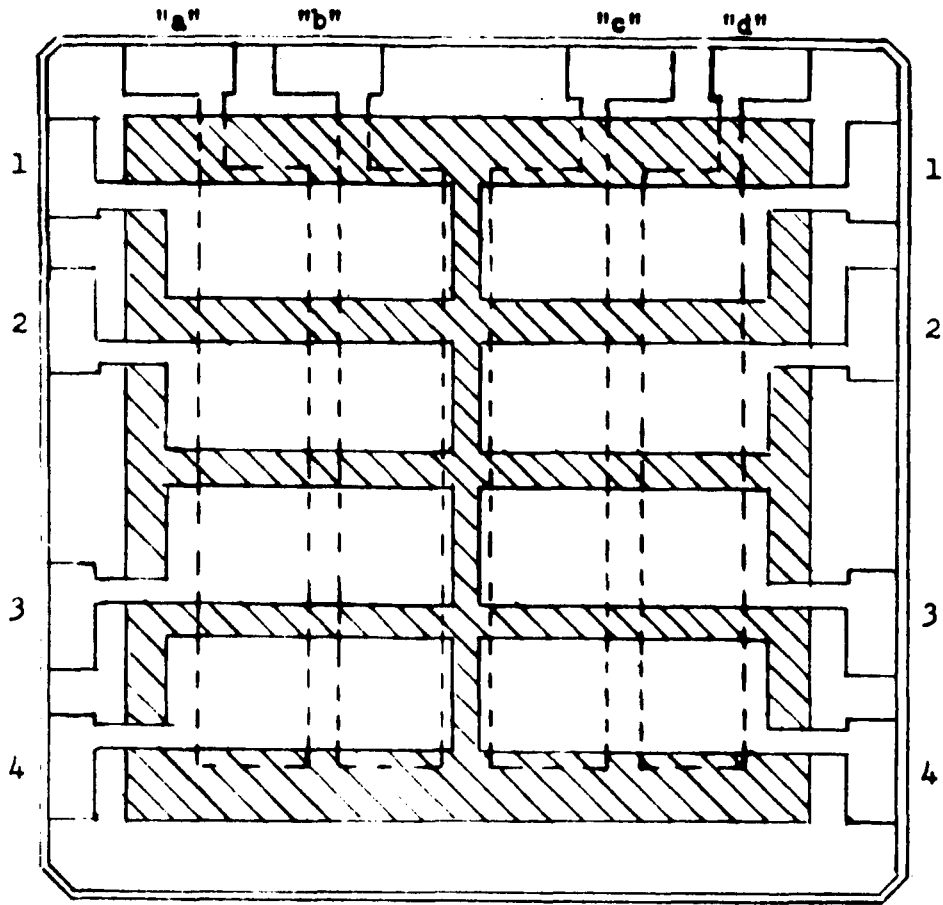


Figure 3. Capacitor Test Pattern (Second)

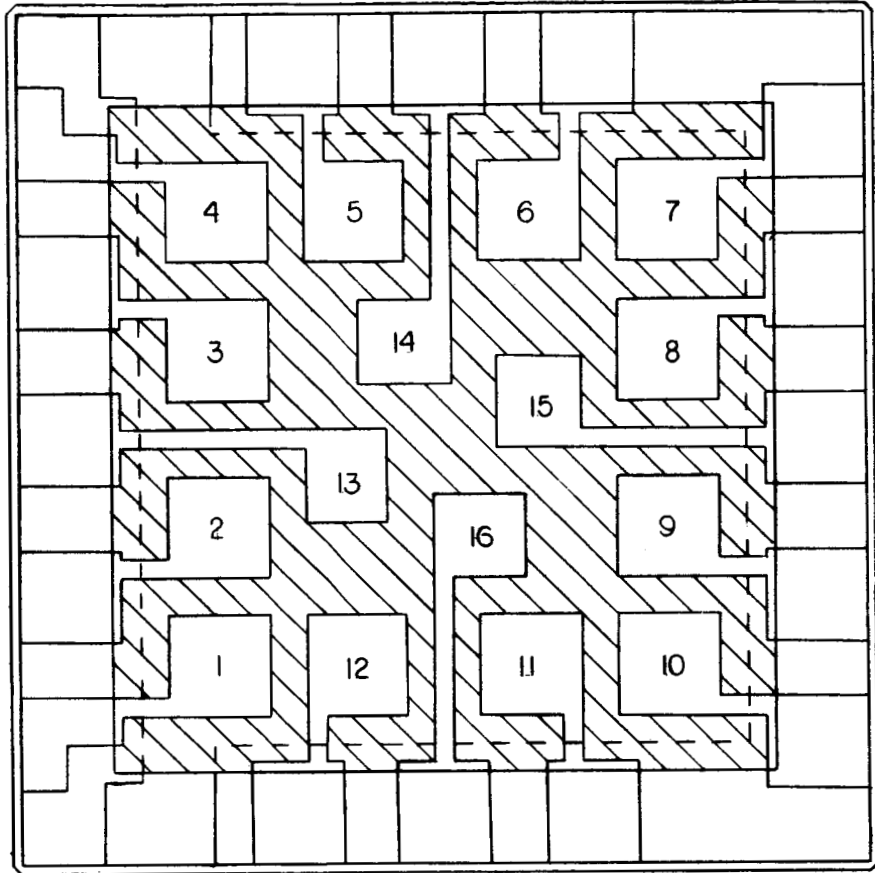


Figure 4. Capacitor Test Pattern (Final)



#### IV. DEPOSITION MODES

##### FLASH EVAPORATION

A dual flash evaporation unit was designed and constructed in the beginning months of the program (Figures 5, 6, and 7). Evaporations of barium-lead zirconate, barium titanate, and  $(\text{Ba}_{0.6}, \text{Pb}_{0.4})\text{ZrO}_3$  were performed utilizing this mechanism. Spectrographic analysis of samples prepared via this technique (i.e., flash evaporation) has shown high contamination levels from the refractory metal source material. Table II below shows a reasonably typical analysis of sodium niobate films deposited by flash evaporation.

TABLE II. ANALYSIS OF SODIUM NIOBATE FILMS DEPOSITED BY FLASH EVAPORATION

| Element        | % Present |               |
|----------------|-----------|---------------|
| Niobium        | 51. %     | 49. %         |
| Sodium         | 5.6       | 5.0           |
| Tungsten       | 15.       | 18.           |
| Iron           | 0.15      | 0.40          |
| Silicon        | 0.28      | 0.51          |
| Boron          | nil       | nil           |
| Manganese      | nil       | nil           |
| Magnesium      | 0.019     | 0.078         |
| Chromium       | trace     | nil           |
| Aluminum       | 0.046     | 0.19          |
| Copper         | 0.056     | 0.25          |
| Silver         | nil       | nil           |
| Titanium       | trace     | nil           |
| Nickel         | nil       | nil           |
| Calcium        | 0.36      | 0.31          |
| Molybdenum     | nil       | trace<br>0.04 |
| Lead           | nil       | 0.12          |
| Other Elements | nil       | nil           |

FLASH EVAPORATION APPARATUS

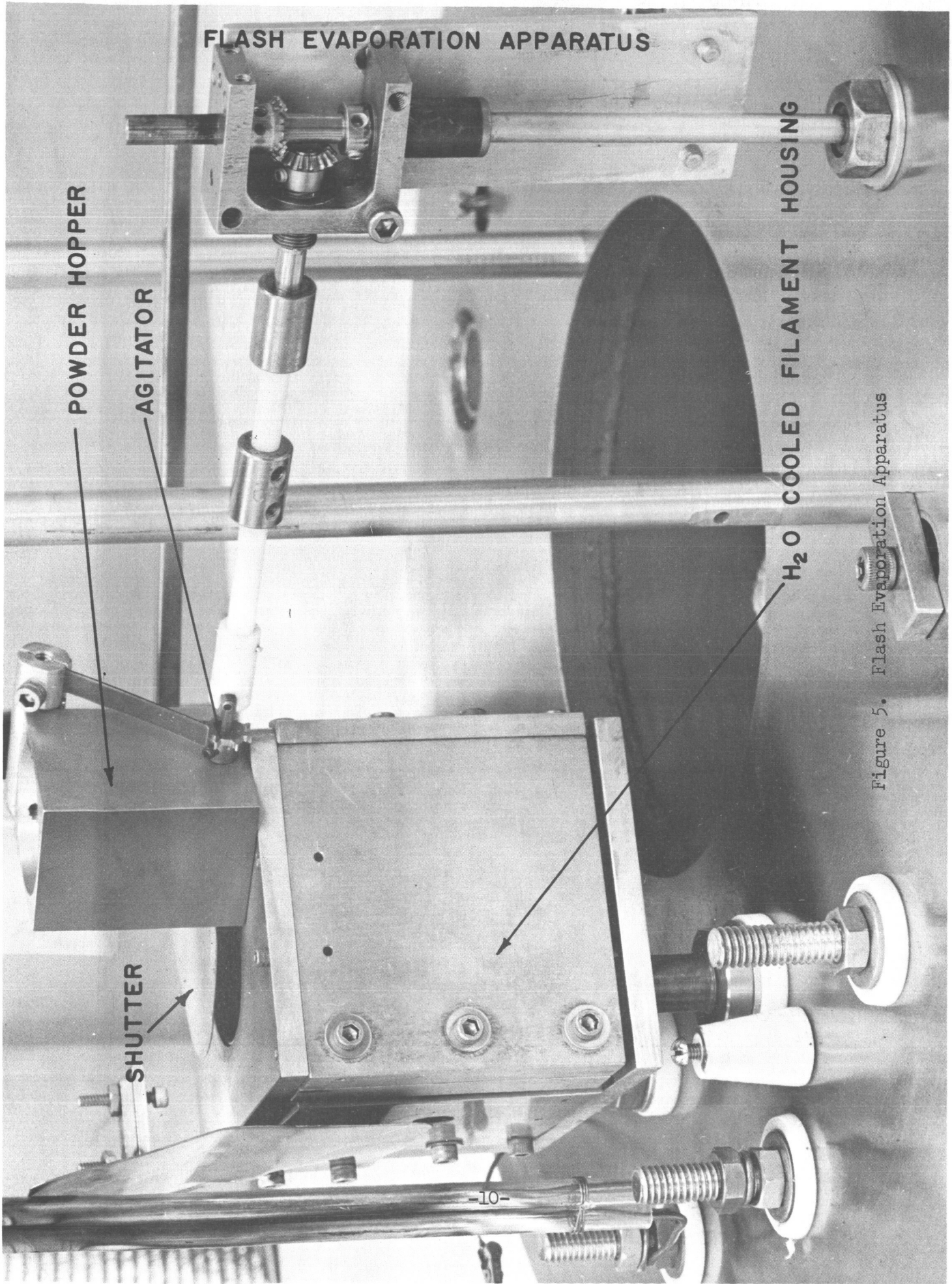
POWDER HOPPER

AGITATOR

H<sub>2</sub>O COOLED FILAMENT HOUSING

SHUTTER

Figure 5. Flash Evaporation Apparatus







Numerous attempts at reducing the contamination level of the refractory source included utilization of  $\text{Al}_2\text{O}_3$ -coated molybdenum, zirconium diboride ( $\text{ZrB}_2$ )-coated molybdenum, irridium, electron beam heating of carbon, and electron beam heating of vitreous carbon (a new material which is essentially a vitrified carbon and which exhibits extreme inertness and low thermal coefficient of expansion). The oxide-coated refractory metals were incapable of being heated to sufficiently high temperatures to allow for flash evaporation without decomposing. The irridium melted at the required temperatures. The electron beam heating of the carbon and vitreous carbon eroded the source and caused a deposit of source material to appear on the substrate.

The problem of source contamination was eventually solved by devising a unique and practical method designated as the "Molten Sphere Technique". The specifics of this technique are described later in this report.

#### ELECTRON BOMBARDMENT

Electron bombardment evaporation of  $(\text{Ba}_{0.6}, \text{Pb}_{0.4})\text{ZrO}_3$  and of sodium niobate ( $\text{NaNbO}_3$ ) was performed with a Filmtech Associates multiple source electron gun, Model TG-1 (Figure 8). A powder press was utilized to fabricate 1/4" pellets of appropriate evaporants. In order to provide for a high residual  $\text{O}_2$  in the vicinity of the substrate, a scheme was devised, constructed, and put into operation which would allow for adequate recombination of the decomposed and fractionated constituents, while still maintaining a sufficiently low pressure in the vicinity of the hot tungsten electron emitter to preclude decomposition and oxidation of the emitter (Figure 9). An example of typical results is shown in Figure 10. The pictured "slugs" are shown in various stages of evaporation. The arrows point to a tungsten wire which was inserted through the core of the slug to provide an electron drain. These electron drains prove to be a serious source of contamination

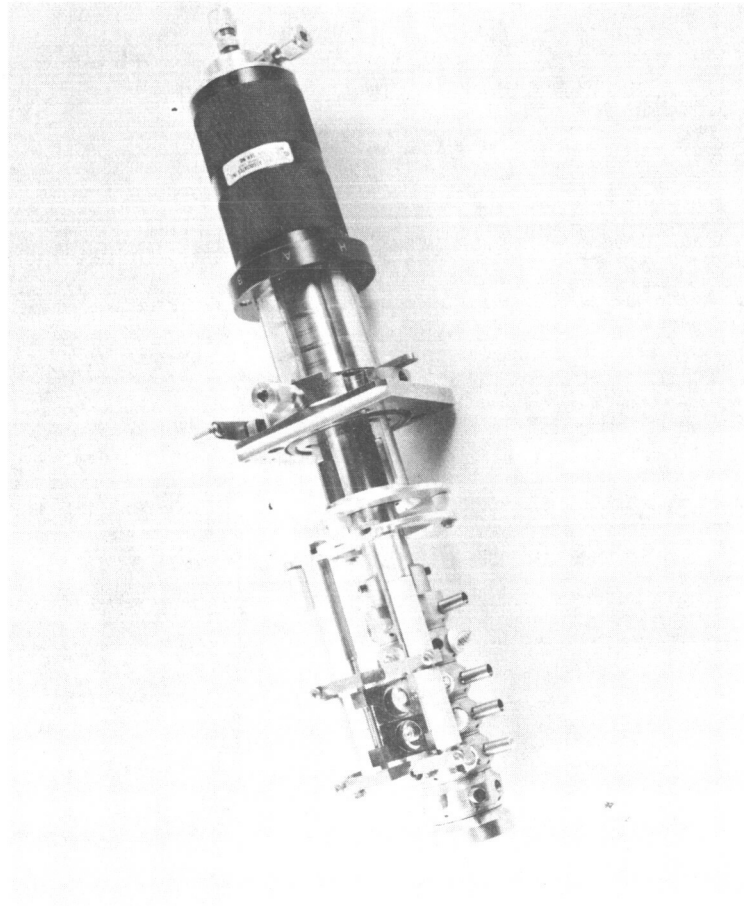


Figure 8. Multiple Source Electron Gun

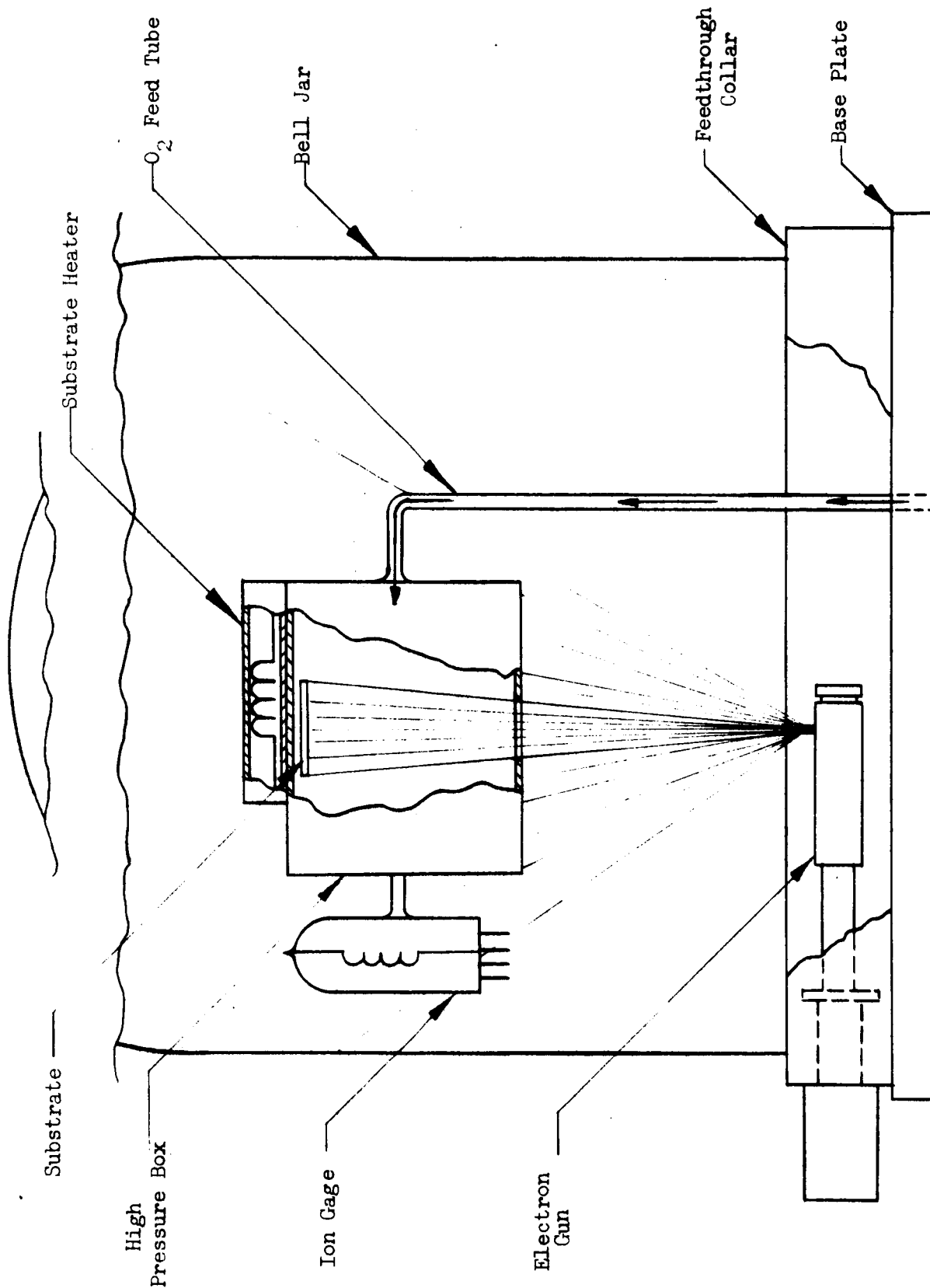


Figure 9. Differential Pressure Apparatus



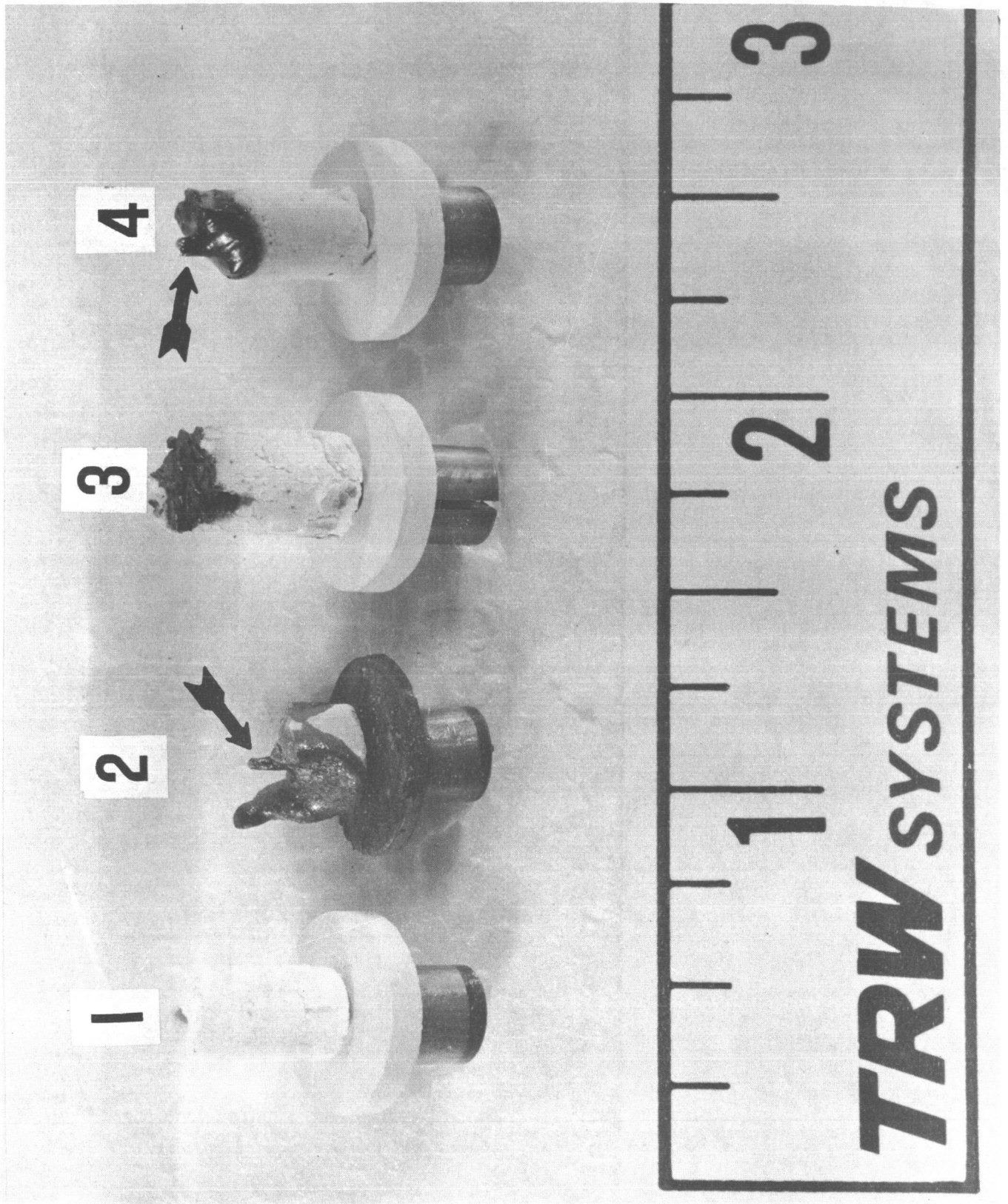


Figure 10. "Slugs" Shown in Various Stages of Evaporation



as shown in Table III. Substitution of niobium wire in place of tungsten precluded contamination by an extraneous constituent, but the technique still proved to be impractical, primarily as a result of the relatively low rate of evaporation which was maintained. Successive evaporation of as many as six slugs yielded films only in the order of 500-800 Å. This was partially due, of course, to the low mean-free path resulting from the necessity of maintaining a high residual  $O_2$  pressure.

TABLE III. SPECTROGRAPHIC ANALYSIS OF ELECTRON BOMBARDMENT DEPOSITED THIN FILM OF  $NaNbO_3$  UTILIZING TUNGSTEN ELECTRON DRAIN.

| Element        | % Present |
|----------------|-----------|
| Niobium        | 24. %     |
| Sodium         | 29.       |
| Tungsten       | 13.       |
| Iron           | 0.42      |
| Silicon        | 2.8       |
| Boron          | 0.027     |
| Manganese      | trace     |
| Magnesium      | 0.095     |
| Chromium       | 0.34      |
| Aluminum       | 0.58      |
| Copper         | 0.35      |
| Silver         | 0.018     |
| Titanium       | 0.15      |
| Nickel         | 0.089     |
| Calcium        | 2.0       |
| Other Elements | nil       |

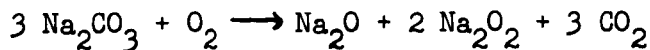
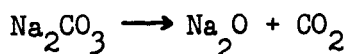
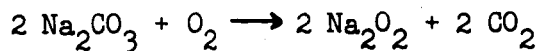
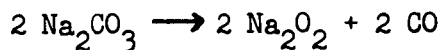
## LASER

A neodymium-doped glass laser capable of a peak pulse output of approximately 14 joules per pulse was designed and constructed within the TRW facilities. The laser utilizes a water-cooled 1/2" x 4" neodymium glass rod and is capable of pulse repetition rates of approximately three to four per minute. However, encouraging results obtained with sodium niobate capacitors deposited via the "Molten Sphere" technique precluded a concomitant investigation of laser evaporation.

## "MOLTEN SPHERE"

As an extension of the technique of electron beam heating of a refractory source (such as carbon or vitreous carbon) in order to reduce contamination, a novel and effective technique was evolved which in essence consisted of powder feeding sodium-containing compounds onto a molten and evaporating sphere of electron beam heated niobium. The technique (Figure 11) has been designated the "Molten Sphere" technique.

The sodium containing compounds investigated were selected on the basis of their likelihood of decomposing upon hitting the hot niobium sphere to a form of sodium or sodium oxide and a gas, which would be pumped from the system.  $\text{NaNO}_2$ ,  $\text{NaCl}$ , and  $\text{Na}_2\text{CO}_3$  were all attempted as the feed powder. The first two proved to be impractical, but the  $\text{Na}_2\text{CO}_3$  produced encouraging results. It is hypothesized that the sodium carbonate, upon hitting the hot niobium, decomposes in any one of the following ways:



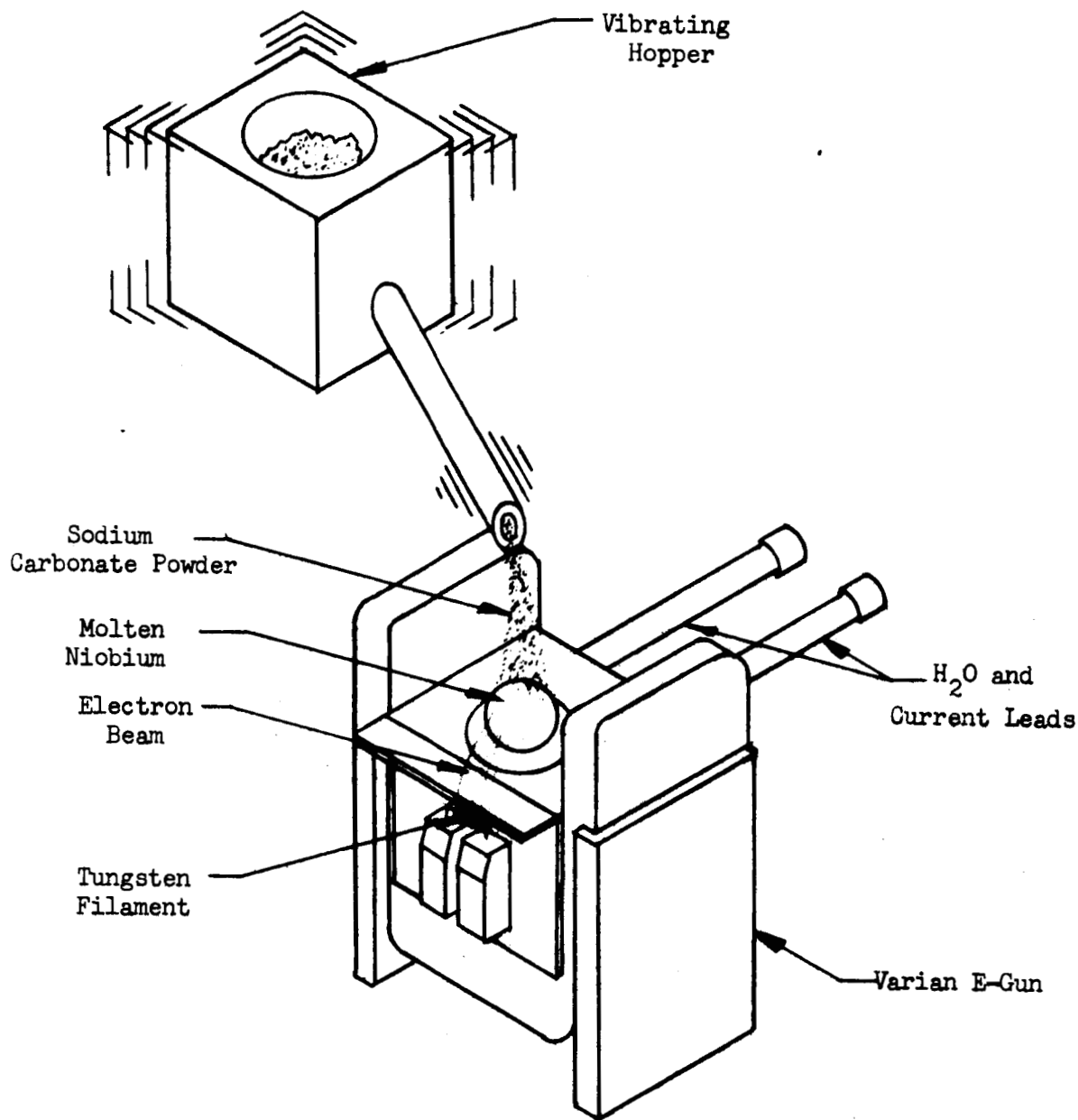


Figure 11. "Molten Sphere Technique"

Spectrographic analysis was performed on films deposited via this technique and did indeed corroborate the hypothesis that refractory metal contamination would be eliminated. The initial analyses (Table IV) indicated a disproportionately high silicon contamination. This was assumed to be resulting from the failure of the liquid nitrogen feed mechanism which caused backstreaming of cracked silicon diffusion pump oil components into the system. Spectrographic analyses of the raw materials (Table V) indicated that the source of the silicon contamination was not inherent in the parent material. A subsequent spectrographic analysis of a "Molten Sphere" evaporated sodium niobate film (Table VI) indicated a silicon contamination of only 0.1% and a reasonably good ratio of niobium to sodium (55:12, as compared with the idealized stoichiometric ratio of 56.7:14.0).

Good yields of sodium niobate capacitors were obtained with the "Molten Sphere" technique. The dissipation factors were encouragingly low (in the vicinity of 5-10% for a majority of the capacitors). The data sheets, Tables VII, VIII, and IX, show the initial readings of these capacitors.

Both optical and electron microscopy were utilized to establish insights into the crystallographic character of the evaporated thin films. Particularly good examples of the orthorhombic crystallinity are shown in Figures 12 and 13.

In order to establish what degree of repeatability could be obtained with the "Molten Sphere" technique, three additional runs (T-39, 40, and 41) were deposited in a manner exactly analogous to runs T-28, 29, and 30. The only test difference is inherent in the fact that two quartz and two sapphire substrates were utilized in each run rather than four quartz substrates. After the initial readings were taken, the substrates from all three runs

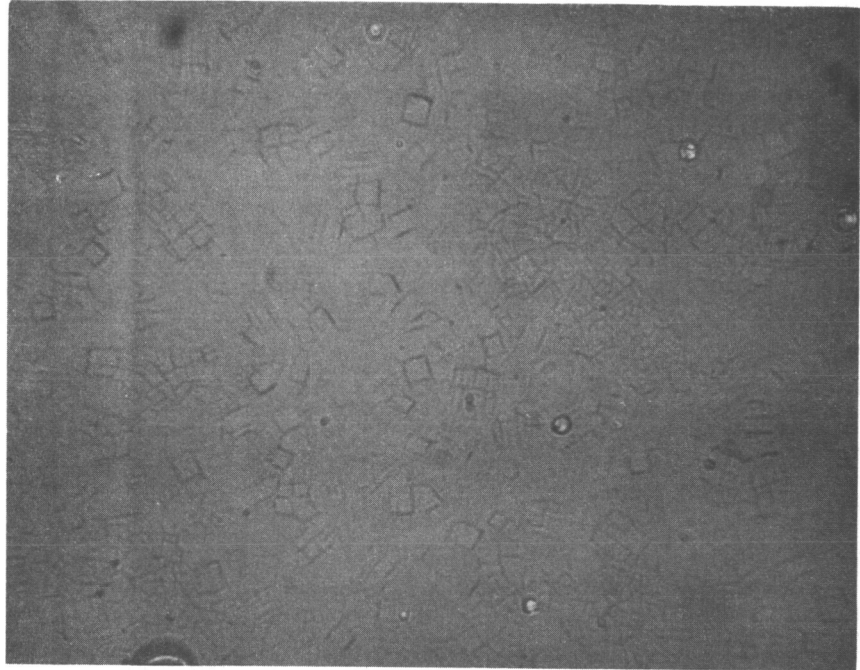


Figure 12. Orthorhombic Crystallinity  
(500X)

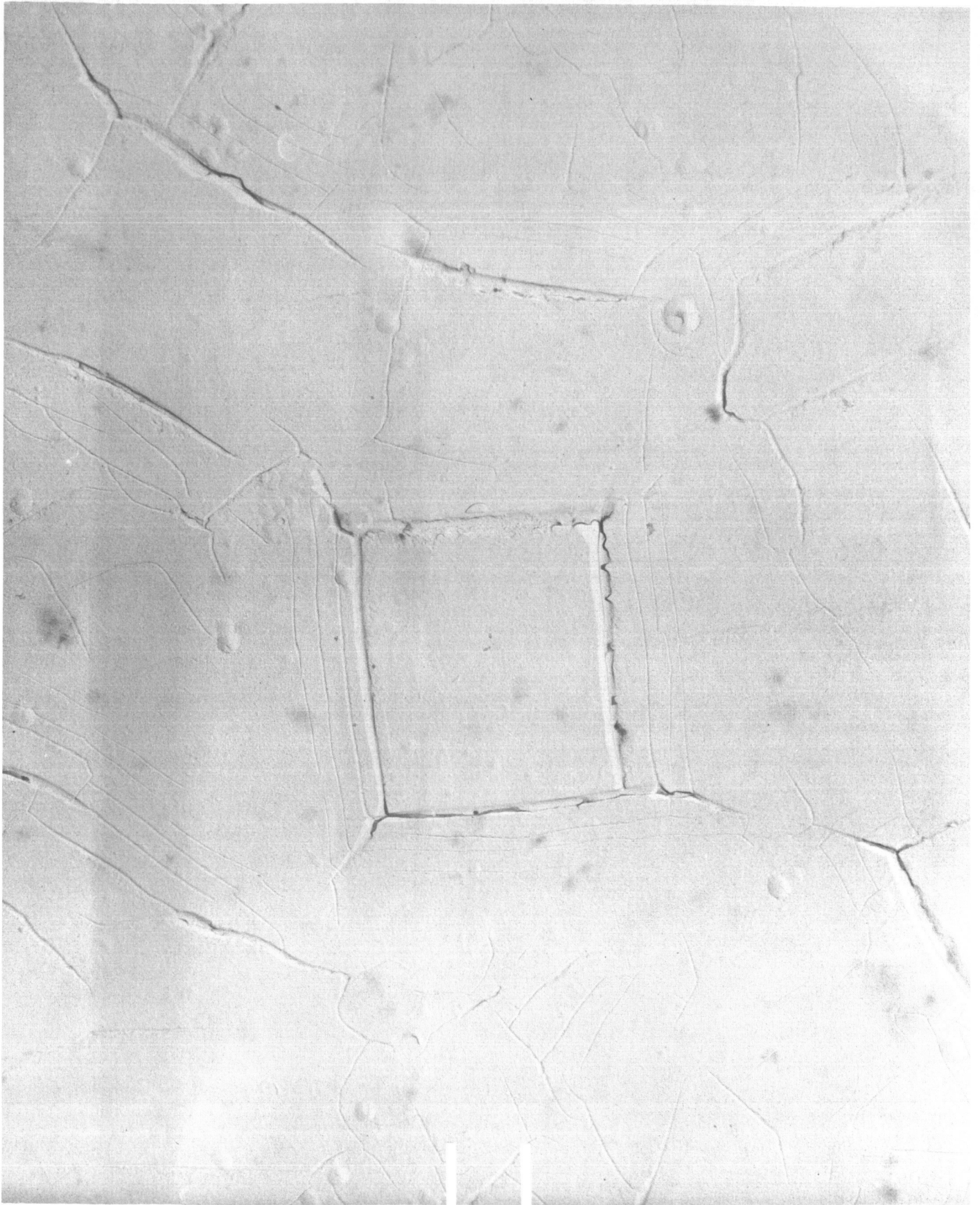


Figure 13. Orthorhombic Crystallinity  
(24,000X)

TABLE IV.  $\text{NaNbO}_3$  SAMPLES (VACUUM DEPOSITED THIN FILM ON CARBON)

| Element        | No. 1 | No. 2 |
|----------------|-------|-------|
| Niobium        | 50. % | 53. % |
| Silicon        | 4.1   | 3.7   |
| Gold           | 0.085 | nil   |
| Iron           | 0.21  | 0.28  |
| Magnesium      | 0.073 | 0.058 |
| Lead           | 0.18  | nil   |
| Aluminum       | 0.43  | 0.19  |
| Molybdenum     | 0.054 | 0.043 |
| Copper         | 0.27  | 0.25  |
| Silver         | 0.011 | nil   |
| Sodium         | 13.   | 10.   |
| Nickel         | trace | nil   |
| Calcium        | 0.28  | 0.15  |
| Other Elements | nil   | nil   |

TABLE V. RAW MATERIALS ANALYSIS

| Niobium Wire   |           | Sodium Carbonate Powder |        |
|----------------|-----------|-------------------------|--------|
| Niobium        | Remainder | Sodium                  | 43. %  |
| Titanium       | 0.056%    | Silicon                 | 0.0065 |
| Iron           | nil       | Iron                    | 0.070  |
| Aluminum       | "         | Aluminum                | 0.0028 |
| Copper         | "         | Copper                  | 0.0019 |
| Titanium       | "         | Titanium                | 0.0028 |
| Calcium        | "         | Calcium                 | 0.0051 |
| Chromium       | "         | Chromium                | 0.0071 |
| Magnesium      | "         | Magnesium               | 0.0015 |
| Other Elements | "         | Other Elements          | nil    |

TABLE VI. SPECTROGRAPHIC ANALYSIS OF  
 $\text{NaNbO}_3$  FILM, "MOLTEN SPHERE"  
 EVAPORATED

| Element        | % Present |
|----------------|-----------|
| Niobium        | 55. %     |
| Sodium         | 12.       |
| Iron           | 0.044     |
| Silicon        | 0.100     |
| Manganese      | 0.054     |
| Magnesium      | 0.033     |
| Copper         | 0.091     |
| Calcium        | 0.31      |
| Other Elements | nil       |



TABLE VII. CAPACITOR DATA, INITIAL READINGS  
(T28)

| Run Number            | "a"             |         | "b"   |         | "c"   |         | "d"   |         |       |
|-----------------------|-----------------|---------|-------|---------|-------|---------|-------|---------|-------|
|                       | Cap.            | D.F.    | Cap.  | D.F.    | Cap.  | D.F.    | Cap.  | D.F.    |       |
| Substrate Material    | T28             |         | -     |         | -     |         | -     |         |       |
| Substrate Temperature | Quartz<br>480°C |         | -     |         | -     |         | -     |         |       |
| Postevaporative Bake  | 17 hr. at 450°C |         | -     |         | -     |         | -     |         |       |
| Thickness             | 1600 Å          |         | -     |         | -     |         | -     |         |       |
| "k" (Cap. of Calc.)   | 140 (18 nf)     |         | -     |         | -     |         | -     |         |       |
| Substrate             | Capacitor       | Cap.    | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  | D.F.    |       |
| A                     | 1               | 18.6 nf | 0.068 | 17.8 nf | 0.075 | 20.3 nf | 0.075 | 20.6 nf | 0.063 |
|                       | 2               | 18.0    | 0.095 | 16.3    | 0.093 | 17.6    | 0.12  | 18.3    | 0.11  |
|                       | 3               | 17.4    | 0.13  | 16.0    | 0.12  | 17.2    | 0.13  | 17.6    | 0.12  |
|                       | 4               | 18.3    | 0.13  | 17.2    | 0.12  | 18.3    | 0.12  | 19.3    | 0.14  |
| B                     | 1               | ---     | ---   | 29.4    | 0.32  | ---     | ---   | ---     | ---   |
|                       | 2               | 16.2    | 0.14  | 14.8    | 0.13  | ---     | ---   | ---     | ---   |
|                       | 3               | ---     | ---   | 31.0    | 0.16  | ---     | ---   | ---     | ---   |
|                       | 4               | ---     | ---   | 30.6    | 0.14  | ---     | ---   | ---     | ---   |
| C                     | 1               | 14.3    | 0.077 | 14.5    | 0.073 | 14.9    | 0.082 | 14.6    | 0.084 |
|                       | 2               | 14.5    | 0.080 | 14.5    | 0.077 | 15.4    | 0.091 | 15.4    | 0.094 |
|                       | 3               | 14.4    | 0.084 | 14.4    | 0.087 | 15.2    | 0.085 | 15.6    | 0.090 |
|                       | 4               | 22.4    | 0.074 | ---     | ---   | ---     | ---   | 23.0    | 0.100 |
| D                     | 1               | ---     | ---   | ---     | ---   | 29.3    | 3.0   | 44.5    | 2.8   |
|                       | 2               | ---     | ---   | ---     | ---   | 27.0    | 3.2   | 27.5    | 8.2   |
|                       | 3               | ---     | ---   | ---     | ---   | 27.0    | 3.0   | 29.0    | 3.7   |
|                       | 4               | ---     | ---   | ---     | ---   | 31.0    | 3.4   | 32.0    | 2.6   |

TABLE VIII. CAPACITOR DATA, INITIAL READINGS  
(T29)

| Run Number            | "a"            |      | "b"   |      | "c"   |         | "d"   |         |       |
|-----------------------|----------------|------|-------|------|-------|---------|-------|---------|-------|
|                       | Cap.           | D.F. | Cap.  | D.F. | Cap.  | D.F.    | Cap.  | D.F.    |       |
| Substrate Material    | T29            |      | -     |      | -     |         | -     |         |       |
| Substrate Temperature | Quartz         |      | -     |      | -     |         | -     |         |       |
| Postevaporative Bake  | 375°C          |      | -     |      | -     |         | -     |         |       |
| Thickness             | 15 hr at 450°C |      | -     |      | -     |         | -     |         |       |
| "k" (Cap. of Calc.)   | 2400 Å         |      | -     |      | -     |         | -     |         |       |
|                       | 100 (8.5 nf)   |      | -     |      | -     |         | -     |         |       |
| Substrate             | Capacitor      | Cap. | D.F.  | Cap. | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  |
| A                     | 1              | ---  | ---   | ---  | ---   | ---     | ---   | 18.8 nf | 0.04  |
|                       | 2              | 13.0 | 0.15  | 9.8  | 0.04  | 18.8 nf | 0.05  | ---     | ---   |
|                       | 3              | 10.0 | 0.033 | ---  | ---   | ---     | ---   | 18.4    | 0.85  |
|                       | 4              | ---  | ---   | ---  | ---   | 10.0    | 0.03  | 10.0    | 0.036 |
| B                     | 1              | ---  | ---   | 12.7 | 0.22  | ---     | ---   | 20.1    | 0.32  |
|                       | 2              | 9.1  | 0.10  | 8.8  | 0.18  | 9.4     | 0.039 | 10.8    | 0.28  |
|                       | 3              | 9.1  | 0.059 | 8.4  | 0.060 | 20.2    | 0.25  | ---     | ---   |
|                       | 4              | 9.2  | 0.062 | 8.6  | 0.060 | ---     | ---   | 20.0    | 0.62  |
| C                     | 1              | 8.6  | 0.12  | 8.0  | 0.048 | 8.1     | 0.049 | 8.6     | 0.05  |
|                       | 2              | 8.0  | 0.014 | 7.8  | 0.045 | 7.5     | 0.056 | 7.7     | 0.06  |
|                       | 3              | 8.2  | 0.054 | 7.7  | 0.064 | 7.0     | 0.056 | 7.4     | 0.055 |
|                       | 4              | 8.7  | 0.049 | 8.1  | 0.044 | 7.0     | 0.053 | 7.5     | 0.049 |
| D                     | 1              | 9.0  | 0.041 | 8.1  | 0.042 | 11.9    | 0.47  | 17.0    | 1.0   |
|                       | 2              | 8.7  | 0.042 | 8.7  | 0.046 | 13.7    | 1.8   | ---     | ---   |
|                       | 3              | 8.6  | 0.043 | 8.6  | 0.040 | 8.8     | 0.057 | 9.1     | 0.070 |
|                       | 4              | 8.6  | 0.046 | 9.1  | 0.040 | 8.5     | 0.055 | 8.7     | 0.045 |

TABLE . . . CAPACITOR DATA, INITIAL READINGS  
(T30)

| Run Number            | "a"             |         | "b"   |         | "c"   |         | "d"   |                 |
|-----------------------|-----------------|---------|-------|---------|-------|---------|-------|-----------------|
|                       | Cap.            | D.F.    | Cap.  | D.F.    | Cap.  | D.F.    | Cap.  | D.F.            |
| Substrate Material    | T30             |         |       |         |       |         |       |                 |
| Substrate Temperature | Quartz<br>380°C |         |       |         |       |         |       |                 |
| Postevaporative Bake  | 16 hr. at 450°C |         |       |         |       |         |       |                 |
| Thickness             | 2400 Å          |         |       |         |       |         |       |                 |
| "k" (Cap. of Calc.)   | 155 (13)        |         |       |         |       |         |       |                 |
| Substrate             | Capacitor       | Cap.    | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  | D.F.            |
| A                     | 1               | 16.8 nf | 0.25  | 16.7 nf | 0.046 | 21.0 nf | 0.32  | 20.0 nf<br>0.32 |
|                       | 2               | 16.8    | 0.45  | 17.0    | 0.29  | 18.3    | 0.11  | 18.0<br>0.151   |
|                       | 3               | ---     | ---   | ---     | ---   | 18.5    | 0.092 | 18.0<br>0.151   |
|                       | 4               | ---     | ---   | ---     | ---   | ---     | ---   | ---             |
| B                     | 1               | ---     | ---   | ---     | ---   | 16.8    | 0.1   | 15.0<br>0.03    |
|                       | 2               | 15.0    | 0.04  | 14.5    | 0.07  | ---     | ---   | ---             |
|                       | 3               | 15.2    | 0.04  | 15.0    | 0.07  | ---     | ---   | ---             |
|                       | 4               | ---     | ---   | ---     | ---   | ---     | ---   | ---             |
| C                     | 1               | 16.0    | 0.46  | 16.0    | 0.44  | 17.0    | 0.11  | 18.0<br>0.03    |
|                       | 2               | 13.0    | 0.015 | 12.8    | 0.015 | 12.4    | 0.157 | 12.7<br>0.22    |
|                       | 3               | 12.7    | 0.016 | 12.8    | 0.016 | 12.0    | 0.025 | 11.9<br>0.02    |
|                       | 4               | 13.0    | 0.065 | 13.0    | 0.05  | 12.0    | 0.025 | 12.0<br>0.02    |
| D                     | 1               | 13.0    | 0.019 | 13.0    | 0.015 | 12.0    | 0.08  | 12.0<br>0.09    |
|                       | 2               | 13.0    | 0.019 | 12.6    | 0.015 | 13.0    | 0.028 | 13.0<br>0.03    |
|                       | 3               | 12.4    | 0.009 | 12.6    | 0.015 | 12.0    | 0.015 | 12.0<br>0.04    |
|                       | 4               | 14.0    | 0.37  | 14.0    | 0.29  | ---     | ---   | ---             |

were reread at specified time intervals up to 36 days in order to get some indication as to the stability. The data is presented in Tables X through XII.

Two things that are immediately apparent are the excellent yield and dissipation factors obtained in all three of these runs. Concurrent with this, however, is noted a lower dielectric constant than was obtained in runs T-28, 29, and 30. The reasons for this reduction in dielectric constant are unknown at this time, but point up the fact that obtaining crystallinity and stoichiometry is not a routine accomplishment.

The aging data indicates degradation of a reasonably significant nature with respect to numbers of capacitors lost with time. It appears with microscopic observation that the aluminum-sodium niobate-chromium system is subject to a certain amount of instability with regard to chemical interaction between the dielectric and the electrodes. Free sodium may be the offending item, but this has not been systematically determined.

Runs T-50, 51, 52, and 53 (all utilizing the final capacitor geometry, Figure 4) were made with the intent of utilizing a technique which might provide the sodium a greater opportunity to oxidize and which might preclude the splattering of undissociated sodium carbonate onto the thin film. Essentially, the technique consisted of performing the "seeding" operations by feeding sodium oxide onto an electron beam heated carbon slab rather than onto the hot, but not molten, niobium sphere. The carbon was heated from beneath to prevent the possibility of carbonaceous contamination. High dielectric constants were obtained with this technique, as will be noted on the data sheets (Tables XIII through XVI); but, unfortunately, poor dissipation factors were universally obtained.

TABLE X(a). CAPACITOR DATA, INITIAL READINGS  
(T-39)

| Run Number                |           | "a"              |       | "b"     |       | "c"     |       | "d"  |       |
|---------------------------|-----------|------------------|-------|---------|-------|---------|-------|------|-------|
| Substrate Material        |           | T-39             |       |         |       |         |       |      |       |
| Substrate Temperature     |           | Quartz, Sapphire |       |         |       |         |       |      |       |
| Postevaporative Bake      |           | 380°C            |       |         |       |         |       |      |       |
| Thickness                 |           | 17 hr @ 450°C    |       |         |       |         |       |      |       |
| "k" (Cap. of Calculation) |           | 5500             |       |         |       |         |       |      |       |
| "k" (Cap. of Calculation) |           | 73 (3.0 nf)      |       |         |       |         |       |      |       |
| Substrate                 | Capacitor | Cap.             | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  | Cap. | D.F.  |
| A<br>(Sapphire)           | 1         | 3.26 nf          | 0.019 | 3.14 nf | 0.024 | 3.41 nf | 0.026 | 3.45 | 0.029 |
|                           | 2         | 3.22             | 0.029 | 3.06    | 0.028 | 3.35    | 0.033 | 3.40 | 0.034 |
|                           | 3         | 3.23             | 0.036 | 3.05    | 0.030 | 3.37    | 0.036 | 3.42 | 0.038 |
|                           | 4         | 3.27             | 0.032 | 3.11    | 0.029 | 3.43    | 0.030 | 3.50 | 0.033 |
| B<br>(Quartz)             | 1         | 3.26             | 0.039 | 2.96    | 0.036 | 3.51    | 0.048 | 3.50 | 0.050 |
|                           | 2         | 3.15             | 0.040 | 2.85    | 0.037 | 3.37    | 0.053 | 3.36 | 0.053 |
|                           | 3         | 3.15             | 0.075 | 2.82    | 0.047 | 3.40    | 0.050 | 3.40 | 0.050 |
|                           | 4         | 3.03             | 0.044 | 2.73    | 0.041 | 3.35    | 0.052 | 3.23 | 0.053 |
| C<br>(Quartz)             | 1         | 2.81             | 0.071 | 2.87    | 0.073 | 2.62    | 0.105 | 2.76 | 0.091 |
|                           | 2         | 3.20             | 0.072 | 3.25    | 0.068 | 2.90    | 0.110 | 3.18 | 0.084 |
|                           | 3         | 3.23             | 0.065 | 3.25    | 0.060 | 2.91    | 0.105 | 3.21 | 0.086 |
|                           | 4         | 3.31             | 0.061 | 3.37    | 0.050 | 3.00    | 0.105 | 3.21 | 0.086 |
| D<br>(Sapphire)           | 1         | 2.81             | 0.046 | 2.75    | 0.044 | 2.55    | 0.038 | 2.64 | 0.038 |
|                           | 2         | 3.21             | 0.037 | 3.13    | 0.036 | 2.84    | 0.033 | 3.00 | 0.038 |
|                           | 3         | 3.20             | 0.039 | 3.13    | 0.036 | 2.84    | 0.033 | 3.03 | 0.035 |
|                           | 4         | 3.24             | 0.040 | 3.20    | 0.037 | 2.94    | 0.035 | 3.12 | 0.038 |





TABLE X(d). CAPACITOR DATA AFTER 36 DAYS STORAGE  
(T-39)

| Run Number | "a"  |         | "b"   |         | "c"   |         | "d"   |         |
|------------|------|---------|-------|---------|-------|---------|-------|---------|
|            | Cap. | D.F.    | Cap.  | D.F.    | Cap.  | D.F.    | Cap.  | D.F.    |
|            | T-39 |         |       |         |       |         |       |         |
| A          | 1    | 3.50 nf | 0.042 | 3.37 nf | 0.041 | ---     | ---   | ---     |
|            | 2    | 3.45    | 0.055 | 3.27    | 0.052 | 4.70 nf | 0.060 | ---     |
|            | 3    | 3.47    | 0.055 | 3.25    | 0.054 | ---     | ---   | ---     |
|            | 4    | 3.45    | 0.042 | 3.30    | 0.039 | ---     | ---   | ---     |
| B          | 1    | ---     | ---   | ---     | ---   | ---     | ---   | ---     |
|            | 2    | ---     | ---   | 1.25    | 0.064 | 2.40    | 0.700 | ---     |
|            | 3    | ---     | ---   | 1.40    | 0.059 | 1.76    | 0.680 | 1.86 nf |
|            | 4    | ---     | ---   | 1.40    | 0.062 | 1.72    | 0.640 | 1.63    |
| C          | 1    | ---     | ---   | 6.00    | 0.180 | 2.80    | 0.040 | 2.90    |
|            | 2    | ---     | ---   | 3.70    | 0.045 | 3.00    | 0.052 | 3.30    |
|            | 3    | ---     | ---   | 6.00    | 0.130 | 3.20    | 0.053 | 3.40    |
|            | 4    | 3.95    | 0.045 | 3.95    | 0.044 | 3.20    | 0.046 | 3.40    |
| D          | 1    | ---     | ---   | 7.00    | 0.175 | ---     | ---   | * ---   |
|            | 2    | 3.80    | 0.037 | 3.75    | 0.038 | *       | ---   | ---     |
|            | 3    | 7.00    | 0.100 | ---     | ---   | ---     | ---   | ---     |
|            | 4    | 7.00    | 0.100 | ---     | ---   | ---     | ---   | ---     |

\*Scatched



TABLE XI(a). CAPACITOR DATA, INITIAL READINGS  
(T-40)

| Run Number                |   | "a"              |       | "b"     |       | "c"     |       | "d"     |       |
|---------------------------|---|------------------|-------|---------|-------|---------|-------|---------|-------|
| Substrate                 |   | Cap.             | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  |
| T-40                      |   |                  |       |         |       |         |       |         |       |
| Substrate Material        |   | Quartz, Sapphire |       |         |       |         |       |         |       |
| Substrate Temperature     |   | 380°C            |       |         |       |         |       |         |       |
| Postevaporative Bake      |   | 17 hrs. at 450°C |       |         |       |         |       |         |       |
| Thickness                 |   | 7560 Å           |       |         |       |         |       |         |       |
| "k" (Cap. of Calculation) |   | 93 (2.8)         |       |         |       |         |       |         |       |
| Substrate                 |   | Cap.             | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  | Cap.    | D.F.  |
| A<br>(Sapphire)           | 1 | 2.45 nf          | 0.038 | ---     | ---   | 2.33 nf | 0.033 | 2.39 nf | 0.030 |
|                           | 2 | 2.86             | 0.038 | ---     | ---   | 2.60    | 0.023 | 2.72    | 0.024 |
|                           | 3 | 2.86             | 0.038 | ---     | ---   | 2.58    | 0.023 | 2.68    | 0.024 |
|                           | 4 | 2.89             | 0.038 | 3.60 nf | 0.048 | 2.72    | 0.028 | 2.77    | 0.028 |
| B<br>(Quartz)             | 1 | 3.33             | 0.073 | 3.85    | 0.080 | 3.50    | 0.040 | 3.56    | 0.042 |
|                           | 2 | 3.49             | 0.070 | ---     | ---   | 3.29    | 0.048 | 3.36    | 0.038 |
|                           | 3 | 3.49             | 0.066 | 3.51    | 0.082 | 3.19    | 0.037 | 3.26    | 0.036 |
|                           | 4 | 3.38             | 0.063 | 3.37    | 0.078 | 3.00    | 0.032 | 3.05    | 0.036 |
| C<br>(Quartz)             | 1 | 2.57             | 0.027 | 2.54    | 0.022 | 2.29    | 0.022 | 2.48    | 0.024 |
|                           | 2 | 2.74             | 0.023 | 2.73    | 0.019 | 2.48    | 0.020 | 2.68    | 0.018 |
|                           | 3 | 2.74             | 0.023 | 2.73    | 0.019 | 2.51    | 0.024 | 2.72    | 0.026 |
|                           | 4 | 2.80             | 0.027 | 2.80    | 0.020 | 2.54    | 0.025 | 2.73    | 0.024 |
| D<br>(Sapphire)           | 1 | ---              | ---   | ---     | 0.095 | 3.20    | 0.037 | 3.27    | 0.021 |
|                           | 2 | 3.45             | 0.028 | ---     | 0.024 | 3.20    | 0.037 | 3.30    | 0.024 |
|                           | 3 | 3.45             | 0.028 | ---     | 0.024 | 3.26    | 0.026 | 3.30    | 0.024 |
|                           | 4 | 3.49             | 0.028 | ---     | 0.024 | 3.36    | 0.028 | 3.38    | 0.025 |

TABLE XI(b). CAPACITOR DATA AFTER 7 DAY STORAGE  
(T-40)

| Run Number | "B"  |         |      |      | "C"  |      |         |      | "D"  |      |         |      |
|------------|------|---------|------|------|------|------|---------|------|------|------|---------|------|
|            | Cap. | D.F.    | Cap. | D.F. | Cap. | D.F. | Cap.    | D.F. | Cap. | D.F. | Cap.    | D.F. |
|            | T-40 |         |      |      |      |      |         |      |      |      |         |      |
| A          | 1    | ---     | ---  | ---  | ---  | ---  | ---     | ---  | ---  | ---  | ---     | ---  |
|            | 2    | ---     | ---  | ---  | ---  | ---  | 2.96 nf | .031 | ---  | ---  | 5.10 nf | .040 |
|            | 3    | ---     | ---  | ---  | ---  | ---  | 2.94    | .030 | ---  | ---  | 3.05    | .032 |
|            | 4    | ---     | ---  | ---  | ---  | ---  | ---     | ---  | ---  | ---  | 3.0     | .033 |
| B          | 1    | 3.94 nf | .075 | 3.95 | .081 | ---  | ---     | ---  | ---  | ---  | 5.1     | .039 |
|            | 2    | ---     | ---  | ---  | ---  | ---  | ---     | ---  | ---  | ---  | 5.40    | .048 |
|            | 3    | 3.57    | .073 | 3.55 | .082 | ---  | ---     | 3.51 | .045 | ---  | 3.58    | .043 |
|            | 4    | 3.46    | .070 | 3.45 | .083 | ---  | ---     | 3.40 | .044 | ---  | 3.48    | .040 |
| C          | 1    | 3.00    | .050 | 3.01 | .050 | ---  | ---     | ---  | ---  | ---  | ---     | ---  |
|            | 2    | 5.72    | .24  | ---  | ---  | ---  | ---     | 5.02 | .075 | ---  | 5.05    | .075 |
|            | 3    | 5.80    | .22  | ---  | ---  | ---  | ---     | 2.96 | .034 | ---  | 2.96    | .034 |
|            | 4    | 3.26    | .049 | 3.27 | .052 | ---  | ---     | 2.96 | .034 | ---  | 2.96    | .034 |
| D          | 1    | ---     | ---  | ---  | ---  | ---  | ---     | ---  | ---  | ---  | ---     | ---  |
|            | 2    | ---     | ---  | ---  | ---  | ---  | ---     | 4.10 | .067 | ---  | 4.10    | .030 |
|            | 3    | 4.13    | .038 | 4.13 | .035 | ---  | ---     | 4.17 | .070 | ---  | 4.19    | .026 |
|            | 4    | ---     | ---  | ---  | ---  | ---  | ---     | 4.28 | .072 | ---  | 4.30    | .030 |

TABLE XI(c). CAPACITOR DATA AFTER 35 DAYS STORAGE

(T-40) "a" "b" "c" "d"

| Run Number | "a"       |        |       |        | "b"   |        |       |      | "c"   |      |      |      | "d"  |        |       |  |
|------------|-----------|--------|-------|--------|-------|--------|-------|------|-------|------|------|------|------|--------|-------|--|
| Substrate  | Capacitor | Cap.   | D.F.  | Cap.   | D.F.  | Cap.   | D.F.  | Cap. | D.F.  | Cap. | D.F. | Cap. | D.F. | Cap.   | D.F.  |  |
| A          | 1         | ---    | ---   | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 2         | ---    | ---   | ---    | ---   | 3.2 nf | 0.036 | ---  | ---   | ---  | ---  | ---  | ---  | 3.4 nf | 0.045 |  |
|            | 3         | ---    | ---   | ---    | ---   | ---    | ---   | 3.2  | 0.038 | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 4         | ---    | ---   | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | 5.4    | 0.042 |  |
| B          | 1         | 4.2 nf | 0.082 | 4.1 nf | 0.087 | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | 5.6 nf | 0.054 |  |
|            | 2         | ---    | ---   | ---    | ---   | ---    | ---   | 3.7  | 0.053 | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 3         | 3.7    | 0.075 | 3.7    | 0.092 | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | 3.7    | 0.046 |  |
|            | 4         | 3.6    | 0.072 | 3.5    | 0.085 | ---    | ---   | 3.5  | 0.048 | ---  | ---  | ---  | ---  | 3.6    | 0.049 |  |
| C          | 1         | 3.2 nf | 0.044 | 3.2    | 0.038 | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 2         | 6.8    | 0.08  | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 3         | 6.8    | 0.057 | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 4         | 6.7    | 0.054 | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
| D          | 1         | ---    | ---   | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 2         | ---    | ---   | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | ---    | ---   |  |
|            | 3         | 4.2    | 0.038 | 4.3 nf | 0.04  | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | 4.2 nf | 0.032 |  |
|            | 4         | ---    | ---   | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | ---  | 4.3    | 0.031 |  |
|            |           | ---    | ---   | ---    | ---   | ---    | ---   | ---  | ---   | ---  | ---  | ---  | 4.4  | 0.030  |       |  |

TABLE XII(a). CAPACITOR DATA, INITIAL READINGS  
(T-41)

| Run Number                | "a"              |      | "b"  |      | "c"  |      | "d"  |      |
|---------------------------|------------------|------|------|------|------|------|------|------|
|                           | Cap.             | D.F. | Cap. | D.F. | Cap. | D.F. | Cap. | D.F. |
| Substrate Material        | T-41             |      |      |      |      |      |      |      |
| Substrate Temperature     | Quartz, Sapphire |      |      |      |      |      |      |      |
| Postevaporative Bake      | 380°             |      |      |      |      |      |      |      |
| Thickness                 | 17 hr. @ 450°C   |      |      |      |      |      |      |      |
| "k" (Cap. of Calculation) | 1600 A°          |      |      |      |      |      |      |      |
| Substrate                 | 34 (4.8 nf)      |      |      |      |      |      |      |      |
| Capacitor                 | Cap.             | D.F. | Cap. | D.F. | Cap. | D.F. | Cap. | D.F. |
| A<br>(Sapphire)           | 1                | 4.75 | .045 | 4.64 | .036 | ---  | 5.62 | .041 |
|                           | 2                | 4.55 | .042 | 4.39 | .037 | 4.48 | 4.62 | .056 |
|                           | 3                | 4.50 | .042 | 4.39 | .037 | 6.62 | ---  | ---  |
|                           | 4                | ---  | ---  | 6.50 | .061 | 4.13 | 4.23 | .060 |
| B<br>(Quartz)             | 1                | 4.48 | .032 | 4.37 | .028 | 5.5  | 5.80 | .027 |
|                           | 2                | 4.22 | .024 | 4.07 | .026 | 5.2  | 5.19 | .027 |
|                           | 3                | ---  | ---  | 6.15 | .033 | 5.2  | 5.19 | .027 |
|                           | 4                | 4.30 | .036 | 4.14 | .034 | ---  | ---  | ---  |
| C<br>(Quartz)             | 1                | 3.63 | .043 | 3.52 | .043 | 3.70 | 3.76 | .074 |
|                           | 2                | 4.12 | .039 | 3.92 | .035 | 4.18 | 4.33 | .066 |
|                           | 3                | 4.12 | .039 | 3.96 | .039 | 4.30 | 4.40 | .059 |
|                           | 4                | 4.27 | .043 | 4.06 | .041 | 4.38 | 4.50 | .065 |
| D<br>(Sapphire)           | 1                | 3.86 | .105 | 6.31 | .96  | ---  | ---  | ---  |
|                           | 2                | 4.55 | .097 | 6.31 | .96  | 5.39 | 5.37 | .024 |
|                           | 3                | 4.55 | .097 | 6.31 | .96  | 5.29 | 5.28 | .031 |
|                           | 4                | 5.95 | 1.05 | ---  | ---  | ---  | ---  | ---  |



TABLE XII(c). CAPACITOR DATA AFTER 35 DAYS STORAGE  
(T-41)

| Run Number | "a"       |           |        |       | "b"    |       |        |       | "c"    |       |        |       |
|------------|-----------|-----------|--------|-------|--------|-------|--------|-------|--------|-------|--------|-------|
|            | Substrate | Capacitor | Cap.   | D.F.  | Cap.   | D.F.  | Cap.   | D.F.  | Cap.   | D.F.  | Cap.   | D.F.  |
|            |           |           | T-41   |       |        |       |        |       |        |       |        |       |
| A          | 1         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 2         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 3         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 4         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
| B          | 1         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 2         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 3         |           | ---    | ---   | ---    | ---   | 7.6 nf | 4.2   | ---    | ---   | ---    | ---   |
|            | 4         |           | ---    | ---   | ---    | ---   | 5      | 0.038 | 5 nf   | 0.038 | ---    | 0.038 |
| C          | 1         |           | 3.6 nf | 0.031 | 3.6 nf | 0.032 | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 2         |           | 4.2    | 0.028 | 4.0    | 0.027 | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 3         |           | 4.2    | 0.029 | 4.0    | 0.027 | 5.5 nf | 0.034 | 5.5 nf | 0.034 | 5.5 nf | 0.03  |
|            | 4         |           | 4.3    | 0.03  | 4.1    | 0.026 | 5.6    | 0.032 | 5.6    | 0.032 | 5.6    | 0.032 |
| D          | 1         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |
|            | 2         |           | ---    | ---   | ---    | ---   | 5.8 nf | 0.2   | ---    | ---   | ---    | ---   |
|            | 3         |           | 5.2 nf | 0.032 | 5.2    | 0.032 | 5.6    | 0.3   | ---    | ---   | ---    | ---   |
|            | 4         |           | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   | ---    | ---   |

TABLE XIII. CAPACITOR DATA, INITIAL READINGS  
(T-50)

| Run Number                | T-50                 |       |      |      |      |      |      |      |
|---------------------------|----------------------|-------|------|------|------|------|------|------|
| Substrate Material        | Quartz               |       |      |      |      |      |      |      |
| Substrate Temperature     | 350°C                |       |      |      |      |      |      |      |
| Post Evaporative Bake     | 16 hrs in air @450°C |       |      |      |      |      |      |      |
| Thickness                 | 1500 Å               |       |      |      |      |      |      |      |
| "k" (Cap. of Calculation) | 220 (25 nf)          |       |      |      |      |      |      |      |
| Substrate                 | Cap.                 | D.F.  | Cap. | D.F. | Cap. | D.F. | Cap. | D.F. |
| A<br>(Sapphire)           | 1-5-9-13             | 7.7nf | 1.3  | ---  | ---  | ---  | 27.7 | .73  |
|                           | 2-6-10-14            | 8.9   | 1.3  | ---  | ---  | ---  | 25.6 | .35  |
|                           | 3-7-11-15            | 8.9   | 1.5  | ---  | ---  | 24.2 | ---  | ---  |
|                           | 4-8-12-16            | 6.6   | 2.0  | ---  | ---  | 22.8 | 23.2 | .36  |
| B<br>(Quartz)             | 1-5-9-13             | 4.8nf | .32  | .71  | ---  | ---  | 5.1  | 1.3  |
|                           | 2-6-10-14            | 4.67  | .33  | .50  | ---  | ---  | 16.7 | .50  |
|                           | 3-7-11-15            | 5.0   | .36  | ---  | ---  | 4.28 | ---  | ---  |
|                           | 4-8-12-16            | ---   | ---  | ---  | ---  | 4.98 | ---  | ---  |
| C<br>(Quartz)             | 1-5-9-13             | 28.6  | .32  | .37  | ---  | ---  | 31.7 | .36  |
|                           | 2-6-10-14            | 30.1  | .32  | .36  | ---  | ---  | 29.1 | .35  |
|                           | 3-7-11-15            | 29.0  | .35  | ---  | ---  | 29.3 | ---  | ---  |
|                           | 4-8-12-16            | 24.7  | .40  | ---  | ---  | 29.5 | 30.7 | .35  |
| D<br>(Sapphire)           | 1-5-9-13             | 21.7  | .45  | 1.7  | ---  | ---  | ---  | ---  |
|                           | 2-6-10-14            | 25.2  | .55  | 1.6  | ---  | ---  | 6.0  | 2.3  |
|                           | 3-7-11-15            | 26.6  | .94  | ---  | ---  | 31.5 | ---  | ---  |
|                           | 4-8-12-16            | 10.3  | 1.2  | ---  | ---  | 28.0 | 12.4 | 1.5  |

TABLE XIV. CAPACITOR DATA, INITIAL READINGS  
(T-51)

| Run Number                | T-51            |         |      |      |      |      |     |      |     |
|---------------------------|-----------------|---------|------|------|------|------|-----|------|-----|
| Substrate Material        | Quartz          |         |      |      |      |      |     |      |     |
| Substrate Temperature     | 350°            |         |      |      |      |      |     |      |     |
| Postevaporative Bake      | 16 hr. @ 450° C |         |      |      |      |      |     |      |     |
| Thickness                 | 1750 Å          |         |      |      |      |      |     |      |     |
| "k" (Cap. of Calculation) | 305 (30)        |         |      |      |      |      |     |      |     |
| Substrate                 | Cap.            | D.F.    | Cap. | D.F. | Cap. | D.F. |     |      |     |
| A<br>(Sapphire)           | 1-5-9-13        | 40.5 nf | .41  | 36.7 | .40  | 59.8 | .56 | 36.0 | .40 |
|                           | 2-6-10-14       | 21.2    | .36  | 39.6 | .44  | 68.0 | .53 | 47.0 | .49 |
|                           | 3-7-11-15       | 28.0    | .38  | 49.5 | .48  | 59.0 | .54 | 56.0 | .57 |
|                           | 4-8-12-16       | 41.5    | .49  | 55.5 | .57  | 52.5 | .49 | 63.5 | .79 |
| B<br>(Quartz)             | 1-5-9-13        | 33.2 nf | .24  | 35.5 | .26  | 36.3 | .27 | 32.7 | .29 |
|                           | 2-6-10-14       | 30.3    | .29  | 31.7 | .32  | 36.2 | .31 | 35.8 | .29 |
|                           | 3-7-11-15       | 26.6    | .28  | 31.2 | .31  | 36.8 | .28 | 40.2 | .28 |
|                           | 4-8-12-16       | 39.2    | .19  | 31.1 | .29  | 34.8 | .29 | 42.8 | .28 |
| C<br>(Quartz)             | 1-5-9-13        | 28.0    | .20  | 14.2 | .30  | 14.9 | .30 | 14.9 | .24 |
|                           | 2-6-10-14       | 26.4    | .19  | 8.9  | .28  | 15.5 | .28 | 13.2 | .31 |
|                           | 3-7-11-15       | 9.7     | .23  | 9.2  | .23  | 21.2 | .27 | 14.2 | .28 |
|                           | 4-8-12-16       | 23.8    | .20  | 9.4  | .27  | 25.3 | .24 | 22.1 | .26 |
| D<br>(Sapphire)           | 1-5-9-13        | 38.2    | .41  | 89.0 | .62  | 30.7 | .48 | 97.0 | .70 |
|                           | 2-6-10-14       | 80.5    | .57  | ---  | ---  | 20.  | .34 | 53.2 | .73 |
|                           | 3-7-11-15       | 89.0    | .50  | 34.8 | .48  | 25.3 | .40 | 67.  | .78 |
|                           | 4-8-12-16       | 81.0    | .50  | 39.0 | .53  | 28.5 | .44 | 75.0 | .63 |



TABLE XV. CAPACITOR DATA, INITIAL READINGS  
(T-52)

| Run Number                | T-52           | Cap.    | D.F. | Cap. | D.F. | Cap. | D.F. | Cap. | D.F. |
|---------------------------|----------------|---------|------|------|------|------|------|------|------|
| Substrate Material        | Quartz         |         |      |      |      |      |      |      |      |
| Substrate Temperature     | 350°           |         |      |      |      |      |      |      |      |
| Postevaporative Bake      | 16 hr. @ 450°C |         |      |      |      |      |      |      |      |
| Thickness                 | 1800 Å         |         |      |      |      |      |      |      |      |
| "k" (Cap. of Calculation) | 136 (13)       |         |      |      |      |      |      |      |      |
| Substrate                 | Capacitor      | Cap.    | D.F. | Cap. | D.F. | Cap. | D.F. | Cap. | D.F. |
| A<br>(Sapphire)           | 1-5-9-13       | 13.8    | .23  | 13.6 | .18  | 12.3 | .17  | 13.9 | .19  |
|                           | 2-6-10-14      | 14.4    | .20  | 15.0 | .19  | 11.6 | .16  | 16.3 | .18  |
|                           | 3-7-11-15      | 13.6    | .20  | 11.7 | .16  | 10.7 | .15  | 12.7 | .17  |
|                           | 4-8-12-16      | 14.1    | .18  | 13.6 | .18  | 14.1 | .20  | 14.1 | .17  |
| B<br>(Quartz)             | 1-5-9-13       | 27.2    | .34  | 27.2 | .33  | 23.2 | .36  | 27.0 | .34  |
|                           | 2-6-10-14      | 24.3    | .32  | 22.1 | .32  | 22.0 | .35  | 33.0 | .38  |
|                           | 3-7-11-15      | 25.7    | .31  | 25.7 | .31  | 20.3 | .34  | 21.6 | .36  |
|                           | 4-8-12-16      | 20.8    | .31  | 22.2 | .32  | 23.3 | .36  | 25.7 | .38  |
| C<br>(Quartz)             | 1-5-9-13       | 16.0    | .33  | ---  | ---  | 21.1 | .36  | 13.4 | .30  |
|                           | 2-6-10-14      | 15.3    | .33  | 18.4 | .33  | 15.8 | .31  | 14.5 | .30  |
|                           | 3-7-11-15      | 16.6    | .33  | 10.6 | .31  | 17.5 | .32  | 15.6 | .34  |
|                           | 4-8-12-16      | ---     | ---  | 12.8 | .31  | 19.3 | .33  | 11.8 | .29  |
| D<br>(Sapphire)           | 1-5-9-13       | 12.8 nf | .15  | 11.1 | .16  | 13.0 | .17  | 18.2 | .32  |
|                           | 2-6-10-14      | 15.2    | .18  | ---  | ---  | 14.9 | .16  | 14.4 | .17  |
|                           | 3-7-11-15      | 18.1    | .19  | 11.0 | .17  | 15.7 | .17  | 15.2 | .18  |
|                           | 4-8-12-16      | 11.6    | .17  | 13.5 | .18  | 22.7 | .70  | 19.2 | .19  |

TABLE XVI. CAPACITOR DATA, INITIAL READINGS  
(T-53)

| Run Number                | T-53              |        |         |      |        |      |      |      |        |
|---------------------------|-------------------|--------|---------|------|--------|------|------|------|--------|
| Substrate Material        | Quartz            |        |         |      |        |      |      |      |        |
| Substrate Temperature     | 350°              |        |         |      |        |      |      |      |        |
| Postevaporative Bake      | 16 hrs. at 450° C |        |         |      |        |      |      |      |        |
| Thickness                 | 2430 A°           |        |         |      |        |      |      |      |        |
| "k" (Cap. of Calculation) | 490 (35)          |        |         |      |        |      |      |      |        |
| Substrate                 | Capacitor         | Cap.   | D.F.    | Cap. | D.F.   | Cap. | D.F. | Cap. | D.F.   |
| A<br>(Sapphire)           | 1-5-9-13          | ---    | ---     | 33.9 | .97    | 37.5 | 1.4  | 28.0 | .68    |
|                           | 2-6-10-14         | 40.5nf | 1.5     | 19.8 | 1.9    | 39.8 | 1.1  | 22.0 | .88    |
|                           | 3-7-11-15         | 23.2   | 1.3     | 34.2 | 1.5    | 35.1 | 1.9  | 30.0 | 1.1    |
|                           | 4-8-12-16         | 23.3   | 1.3     | 25.3 | 1.1    | 41.5 | 1.9  | 32.5 | 1.8    |
| B<br>(Quartz)             | 1-5-9-13          | 38.6   | 1.1     | 38.0 | 1.1    | 39.0 | 1.5  | 50.5 | 1.8    |
|                           | 2-6-10-14         | 42.2   | 1.9     | 34.4 | 1.1    | 37.0 | 1.1  | 54.5 | .80    |
|                           | 3-7-11-15         | 46.0   | 1.1     | 32.8 | 1.3    | 42.3 | 1.3  | 50.0 | .86    |
|                           | 4-8-12-16         | 38.2   | 1.0     | 40.7 | 1.1    | 40.1 | 1.7  | 50.5 | 1.5    |
| C<br>(Quartz)             | 1-5-9-13          | 32.4   | .69     | 44.8 | .62    | 19.7 | .35  | ---  | ---    |
|                           | 2-6-10-14         | 30.6   | .65     | 20.2 | .44    | 19.9 | .40  | 28.7 | .60    |
|                           | 3-7-11-15         | 21.5   | .52     | 28.5 | .68    | 21.7 | .41  | ---  | ---    |
|                           | 4-8-12-16         | 20.0   | .38     | 23.7 | .45    | 24.6 | .50  | 24.5 | .47    |
| D<br>(Sapphire)           | 1-5-9-13          | 22.1   | .95 ad  | 12.8 | 1.1    | 29.6 | 1.1  | 31.5 | 1.0 ad |
|                           | 2-6-10-14         | 51.5   | .98     | 17.8 | .85 ad | 21.9 | 1.2  | 15.2 | 1.1    |
|                           | 3-7-11-15         | 68.0   | 1.15 1d | 25.0 | .90 ad | 30.5 | 1.1  | 54.0 | .80    |
|                           | 4-8-12-16         | 19.5   | 1.2     | 25.2 | .95 ad | 26.7 | 1.2  | 65.5 | .88    |

## V. CAPACITOR CHARACTERIZATION

### ELECTRICAL

The deposited capacitors were all read initially for capacity and dissipation factor and had the thicknesses of the dielectric films determined by utilization of interferometric techniques. This thickness was then used to calculate the dielectric constants.

Temperature coefficients of capacitance were run on selected samples throughout the course of Phase 2. Typical temperature coefficient curves obtained in the latter stages of Phase 2 are as shown in Figures 14.

Voltage breakdown tests were also performed on selected samples, and in general showed voltage capability only to about 3 or 4 volts (Figure 15).

### PHYSICAL

Spectrographic analysis, optical and electron microscopy, and X-ray diffraction techniques were all utilized to characterize the physical, chemical, and crystallographic makeup of the sodium-niobium-oxygen complex.

Utilizing X-ray diffraction, employing sodium niobate Standard No. 14-603 (Table XVII) and a conversion chart of d-spacings to degrees (Table XVIII), the presence of orthorhombic crystalline sodium niobate in the "Molten Sphere" evaporated films was established. Tape traces of the diffraction pattern (Figure 16) are representative of numerous films discovered. The ratio between relative intensities of the (004) reflection and the (200) reflection in the diffraction pattern provides a measure of the preferred orientation within the thin film.

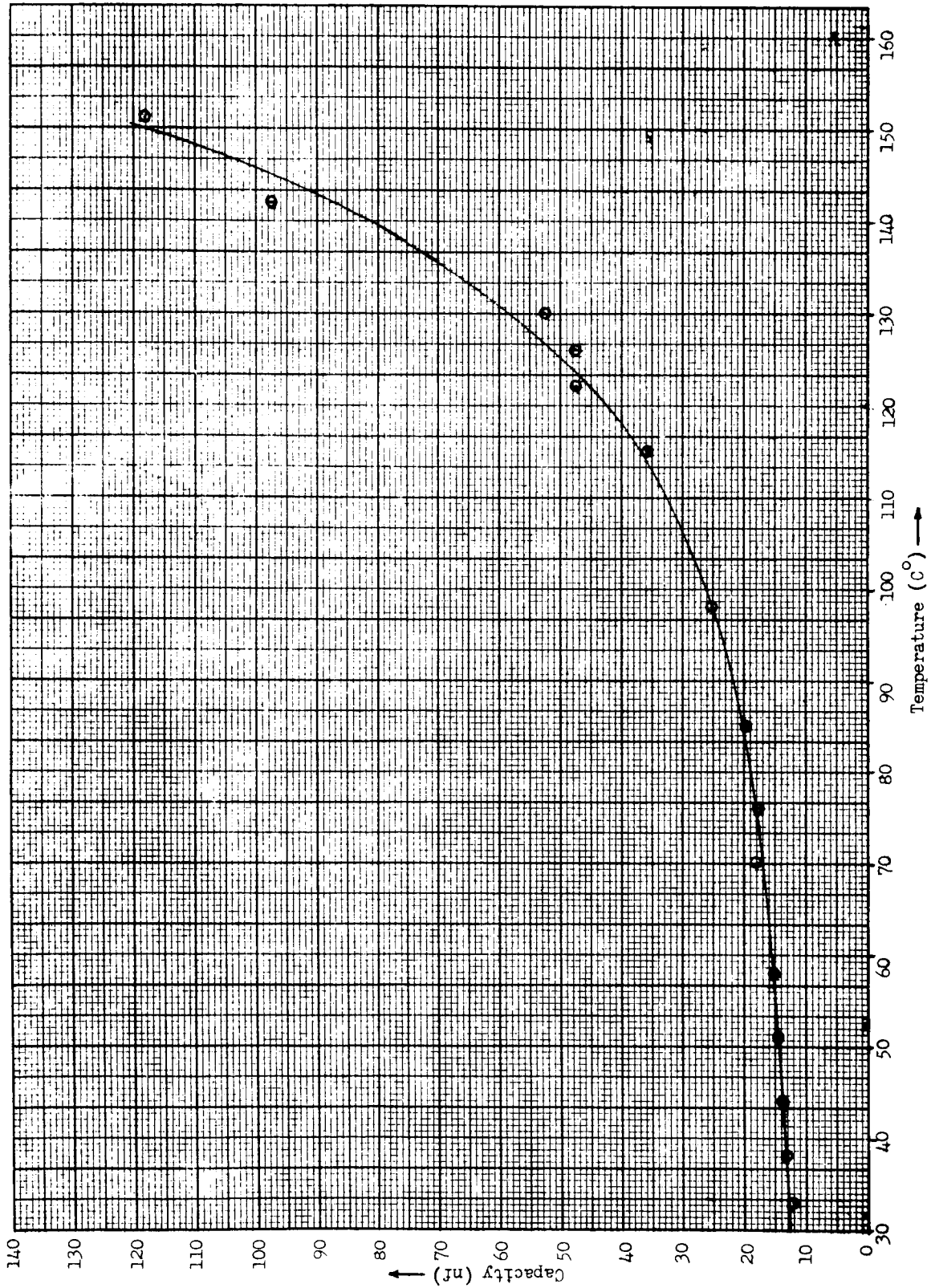


Figure 14. Capacity vs. Temperature  
Run T29A, Sample (D-3)

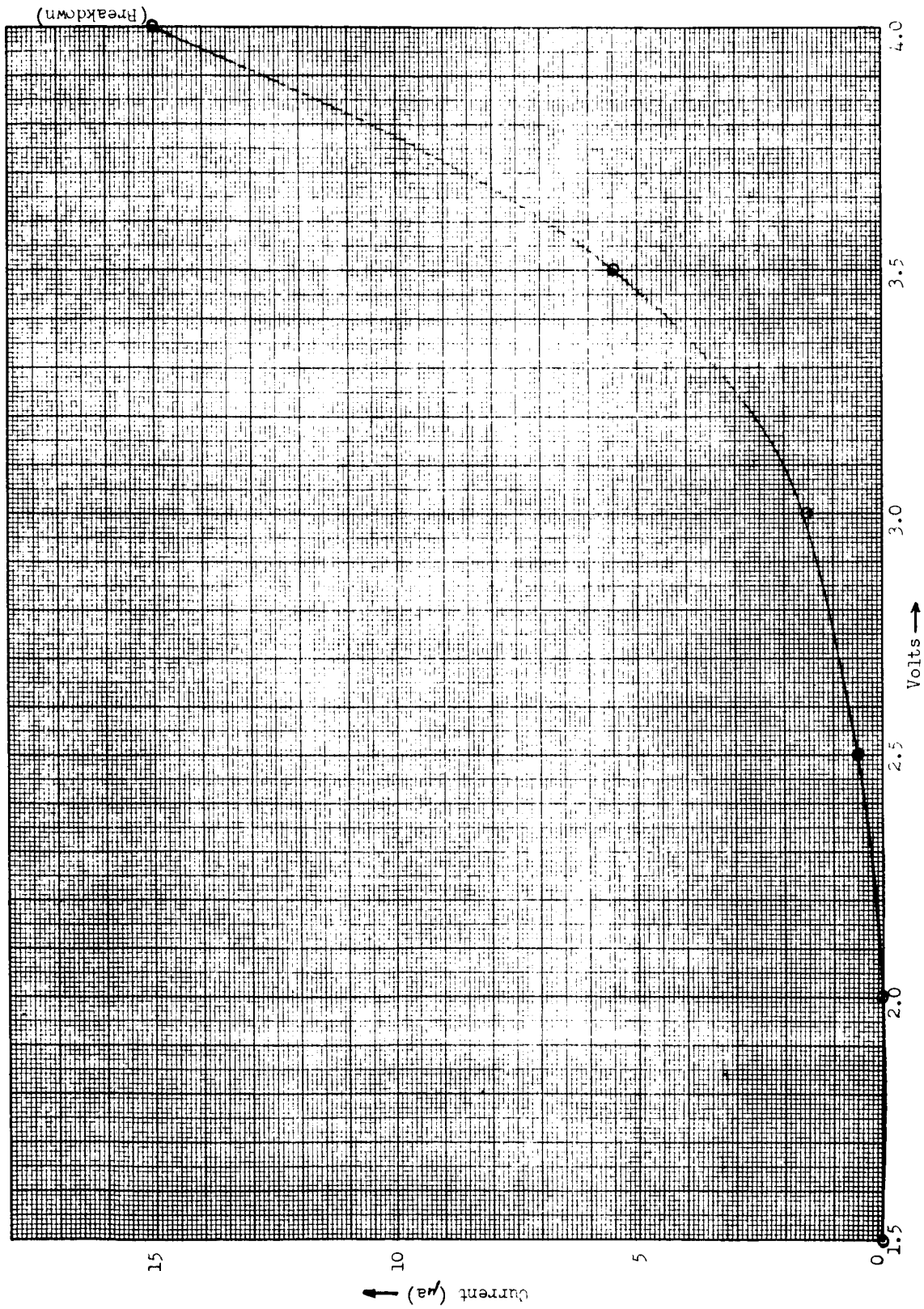


Figure 15. Current vs. Voltage  
Run HD31, Sample B-3

14-603

| d  |       | 2.74             | 1.58      | 3.93  | 3.93             | NaNBO <sub>3</sub> (AT 25°C) |         |
|--|-------|------------------|-----------|---|------------------|------------------------------|---------|
| I/I <sub>1</sub>   |       | 100              | 90+       | 90  | 90               | SODIUM NIOBATE (LUESHITE)    |         |
| Rad. CuKα <sub>1</sub> λ 1.54050 Filter Ni Dia. 114.6mm  |       |                  |           |   |                  |                              |         |
| Cut off I/I <sub>1</sub> VISUAL INSPECTION   |       |                  |           |   |                  |                              |         |
| Ref. E. WOOD, BELL TELEPHONE LABORATORIES, MURRAY HILL, NEW JERSEY   |       |                  |           |   |                  |                              |         |
| Sys. *ORTHORHOMBIC S.G. P222 <sub>1</sub> (17)**   |       |                  |           |   |                  |                              |         |
| a <sub>0</sub>   | 5.512 | b <sub>0</sub>   | 5.557     | c <sub>0</sub>  | 4x3.885          | A                            | C       |
| a  |       | β                |           | γ   | Z                | 8                            | Dx      |
| Ref. WOOD, ACTA CRYST. 4 353-362 (1951)  |       |                  |           |   |                  |                              |         |
| ε a  | n ω β | 2.30**           | ε γ       | Sign  |                  |                              |         |
| 2V   | D     | 4.44**           | mp        | Color CLEAR, COLORLESS BUT TURNS TAN ON EXPOSURE TO LIGHT |                  |                              |         |
| Ref.   |       |                  |           |   |                  |                              |         |
| *PSEUDOTETRAGONAL: Z=16, a <sub>0</sub> =2x3.921, c <sub>0</sub> =4x3.885  |       |                  |           |   |                  |                              |         |
| *MONOCLINIC: a <sub>0</sub> AND c <sub>0</sub> =2x3.921, b <sub>0</sub> =4x3.885, β=90°40'; CHANGES TO TETRAGONAL FORM AT ABOUT 3700; TO CUBIC AT ABOUT 640°C. |       |                  |           |   |                  |                              |         |
| **SAFIANNIKOFF, BULL. ACAD. ROY. SOC. COL. BELGES 5 1251 (1959)  |       |                  |           |   |                  |                              |         |
|  | d Å   | I/I <sub>1</sub> | hkl       | d Å   | I/I <sub>1</sub> | hkl                          | bkl     |
|  | 3.93  | 90               | 110       | 1.523   | 15               |                              | INDEXED |
|  | 3.85  | 90               | 004       | 1.389   | 40               |                              | BY      |
|  | 2.77  | 90               | 114       | 1.377   | 80               |                              | LGB     |
|  | 2.74  | 100              | 021(200?) | 1.302   | 50               |                              |         |
|  | 2.59  | 5                | 202,006   | 1.295   | 60               |                              |         |
|  | 2.44  | 20               | 211,115+  | 1.265   | 10               |                              |         |
|  | 2.37  | 10               | 122       | 1.242   | 20               |                              |         |
|  | 2.34  | 10               | 016,106   | 1.236   | 70               |                              |         |
|  | 2.25  | 20               | 204       |   |                  |                              |         |
|  | 2.24  | 20               | 123,213   |   |                  |                              |         |
|  | 1.96  | 80               | 220       |   |                  |                              |         |
|  | 1.94  | 60               | 008       |   |                  |                              |         |
|  | 1.78  | 5                | -         |   |                  |                              |         |
|  | 1.75  | 50               | 224       |   |                  |                              |         |
|  | 1.74  | 90               | 310       |   |                  |                              |         |
|  | 1.66  | 20               | 304       |   |                  |                              |         |
|  | 1.653 | 5                | -         |   |                  |                              |         |
|  | 1.598 | 50               | 134,028   |   |                  |                              |         |
|  | 1.583 | 90+              | 208       |   |                  |                              |         |
|  | 1.537 | 5                | 230,320+  |   |                  |                              |         |

TABLE XVII. SODIUM NIOBATE STANDARD

TABLE XVIII.  $d(\text{\AA})$  vs.  $2\theta$  for  $\text{NaNbO}_3$  at  $25^\circ\text{C}$

| $d(\text{\AA})$ | $2\theta$ (degrees) |
|-----------------|---------------------|
| 3.93            | 22.62               |
| 3.85            | 23.08               |
| 2.77            | 32.3                |
| 2.74            | 32.66               |
| 2.59            | 34.62               |
| 2.44            | 36.80               |
| 2.37            | 37.94               |
| 2.34            | 38.44               |
| 2.25            | 40.04               |
| 2.24            | 40.22               |
| 1.96            | 46.28               |
| 1.94            | 46.78               |
| 1.78            | 51.28               |
| 1.75            | 52.24               |
| 1.74            | 52.56               |
| 1.66            | 55.30               |
| 1.653           | 55.56               |
| 1.598           | 57.64               |
| 1.583           | 58.24               |
| 1.537           | 60.40               |
| 1.523           | 60.76               |
| 1.389           | 67.36               |
| 1.377           | 68.02               |
| 1.302           | 72.54               |
| 1.295           | 73.00               |
| 1.265           | 75.02               |
| 1.242           | 76.66               |
| 1.236           | 77.10               |

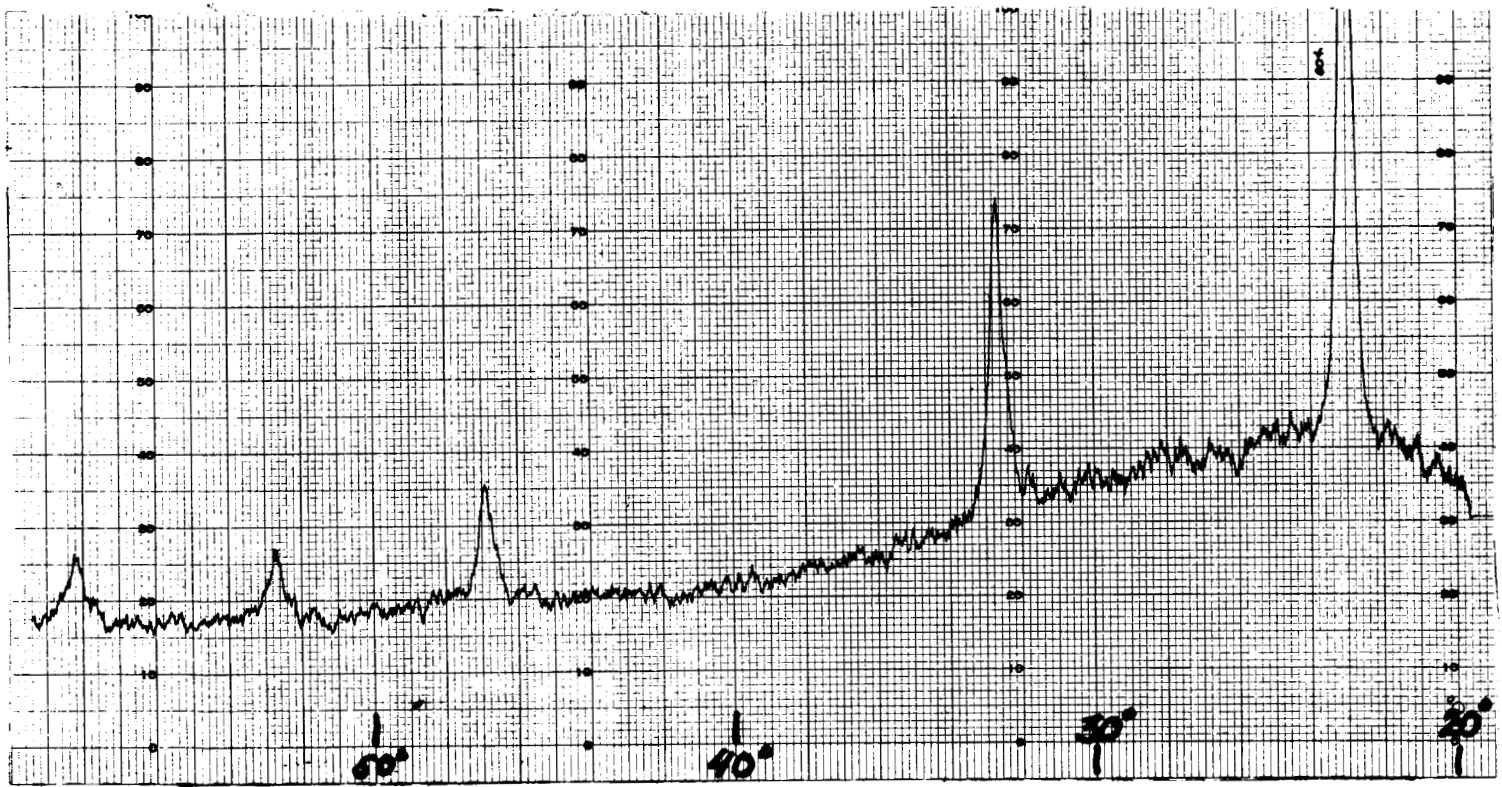


Figure 16. Diffraction Pattern  
Tape Traces



## VI. CONCLUSIONS

The "Molten Sphere" deposition approach has been shown to be one which holds high potential for successful fabrication of high dielectric constant thin film capacitors of sodium niobate. The technique is especially advantageous from the standpoint of eliminating contamination from a refractory metal source, a problem which has persistently plagued investigators attempting to evaporate compounds of the  $ABO_3$  type.

Good yield factors of  $NaNbO_3$  capacitors deposited via this approach have been obtained, but their long-range stability is open to question. Dissipation factors have been variable, ranging from good (1-10%) to questionable (10-25%). Dielectric constants have ranged from 50-150. While this does not approach the figure of 500-600 obtained with pure, single-crystalline and preferentially oriented bulk material, it nevertheless represents a significant improvement in the specific capacitance obtainable with vacuum evaporated capacitors. The data obtained to date regarding temperature coefficient of capacitance is not conclusive, but does indicate that reasonable flatness of temperature response is obtained up to about 100°C. Voltage breakdown characteristics remain as the most notable deficiency at the present time. Breakdowns in excess of 5 volts DC have been obtained only infrequently.

One primary area of concern at the present time involves the selection of appropriate base and counter electrode material. Nichrome, gold, titanium, aluminum, and chromium have all been attempted for one or the other or both electrodes during the Phase 2 investigation, but the necessity for frequent modification of the dielectric deposition parameters precluded the possibility of obtaining well integrated and conclusive data regarding electrode material. The behavior of the base electrode during the postevaporative bake

cycle and the chemical interaction between both electrodes and the dielectric are extremely complex and important considerations, and undoubtedly have a significant effect on the dissipation factors and voltage breakdown characteristics.

An important deficiency during the Phase 2 investigation had been the lack of control of the powder being fed onto the molten sphere of niobium. Frequent "dumpings" of large amounts of the powder caused relatively violent eruptions. As a result of this, particulate matter was injected into the thin film on the substrate, causing undesirable roughness. The implications of this with regard to voltage breakdown characteristics is obvious. To correct this deficiency, Bendix Balzer's vibratory feeder unit, Model No. C-20460, has been obtained and is presently being installed in the vacuum system. It is felt that this unit will alleviate the problem to a very large extent, if not entirely.

## VII. BIBLIOGRAPHY

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 1 Jona, F., and G. Shirane. Ferroelectric Crystals. International Series of Monographs on Solid State Physics. New York: Macmillan Company, 1962.
- 2 Wul, B. "Dielectric Constant of Barium Titanate at Low Temperatures," Journal of Physics (U.S.S.R.), X:1 (1946), 64-66.
- 3 Remeika, J. P. "A Method for Growing Barium Titanate Single Crystals," Journal of the American Chemical Society, 76 (1954), 940-941.
- 4 Forrester, W. F., R. M. Hinde. "Crystal Structure of Barium Titanate," Nature, 156:3954 (1945), 177.
- 5 Goodman, G. "Ferroelectric Properties of Lead Metaniobate," Journal of the American Ceramic Society, 36:11 (1953), 368-372.
- 6 Ginsburg, V. "On the Dielectric Properties of Ferroelectric (Seignetteelectric) Crystals and Barium Titanate," Journal of Physics (U.S.S.R.), X:2 (1946), 107-115.
- 7 Donley, H. L. "Effect of Field Strength on Dielectric Properties of Barium Strontium Titanate," R. C. A. Review, VIII (1947), 539-553.
- 8A Shirane, G., and R. Pepinsky. "Dielectric Properties and Phase Transitions of  $Cd_2Nb_2O_7$  and  $Pb_2Nb_2O_7$ ," Physical Review, 92 (1953), 504.
- 8B Hulm, J. K. "Low Temperature Dielectric Properties of Cadmium and Lead Niobates," Physical Review, 92 (1953), 504-505.
- 9 Blunt, R. F. and W. F. Love. "The Dielectric Properties of Barium Titanate at Low Pressures," Physical Review, 76:8 (1949), 1202-1204.
- 10 Matthias, B. T. "New Ferroelectric Crystals," Physical Review, 75 (1949), 1771.
- 11 Cook, Jr., W. R., and H. Jaffe. "Ferroelectricity in Oxides of Face-Centered Cubic Structure," Physical Review, 89 (1953), 1297-1298.
- 12 Sawaguchi, E., H. Maniwa and S. Hoshino. "Antiferroelectric Structure of Lead Zirconate," Physical Review, 83 (1951) 1078.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 13 Merz, W. J.. "The Dielectric Behavior of BaTiO<sub>3</sub> Single-Domain Crystals," Physical Review, 75 (1949), 687.
- 14 Marks, B. H. "Ceramic Dielectric Materials," Electronics 21:8 (1948), 116-120.
- 15 Roberts, S. "Dielectric Constant of Barium Titanate at High Temperatures," Physical Review, 75 (1949), 989-990.
- 16 Merz, W. J. "The Dielectric Properties of BaTiO<sub>3</sub> at Low Temperatures," Physical Review, 81 (1951), 1064-1065.
- 17 Shirane, G., E. Sawaguchi and Y. Takogi. "Dielectric Properties of Lead Zirconate," Physical Review, 84:3 (1951), 476-481.
- 18 Evans, H. T., and R. D. Burbank. "The Crystal Structure of Hexagonal Barium Titanate," Journal Chemical Physics, 16 (1948), 634.
- 19 Noland, J. A. "Optical Absorption of Single-Crystal Strontium Titanate," Physical Review, 94:3 (1954) 724.
- 20 Edwards, James W., Rudolph Speiser, and Herrick L. Johnston. "Structure of Barium Titanate at Elevated Temperatures," Journal of the American Chemical Society, 73 (1951) 2934-2935.
- 21 Shirane, G., R. Newham and R. Pepinsky. "Dielectric Properties and Phase Transitions of NaNbO<sub>3</sub> and (Na,K)NbO<sub>3</sub>," Physical Review, 96:3 (1954) 581-588.
- 22 Hulm, J. K., B. T. Matthias and E. A. Long. "A Ferromagnetic Curie Point in KTaO<sub>3</sub> at Very Low Temperatures," Physical Review, 79 (1950), 885-886.
- 23 Megan, Helen D. "Crystal Structure of Double Oxides of the Perovskite Type," Proceedings of the Physical Society, 58:2 (1946), 10, 134-152.
- 24 Wood, Elizabeth A. "Polymorphism in Potassium Niobate, Sodium Niobate, and other ABO<sub>3</sub> Compounds," Acta Crystallographica, 4 (1951), 353-362.
- 25 Tokagi, Yutaka, Etsuro Sawaguchi and Tetsuo Aioaka. "On the Effect of Mechanical Stress Upon the Permittivity of Barium Titanate," Journal of the Physical Society Japan, 3 (1948), 270-271.
- 26 Wul, B. "Dielectric Constants of Some Titanates," Nature, 156 (1945), 480.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 27 Busch, G., H. Flury and W. Merz. "Electrische Leitfähigkeit und Brechungsindex des Bariumtitanats," Helvetic Physica Acta, 21 (1948), 212-215.
- 28 Partington, J. R., G. V. Planer and I. I. Boswell. "Voltage Effects in Titanate Polycrystals," Nature, 162 (1948), 151-152.
- 29 Rhodes, R. G. "Structure of BaTiO<sub>3</sub> at Low Temperatures," Acta Crystallographica, 2 (1949), 417-419.
- 30 Rooksby, H. P. "Compounds of the Structural Type of Calcium Titanates," Nature, 155 (1945) 484.
- 31 Reddish, W. "Cyclic Variations of Capacitance, Effect of 50 c/s Voltage on High-Permittivity Ceramics," Wireless Engineer, 25 (1948), 331-337.
- 32 DeBrettville, Jr., Alexander. "Oscillograph Study of Dielectric Properties of Barium Titanate," Journal of the American Ceramic Society, 29 (1946), 303-307.
- 33 Merz, Walter J. "Double Hysteresis Loop of BaTiO<sub>3</sub> at the Curie Point," Physical Review, 91:3 (1953), 513-517.
- 34 McQuarrie, Malcom. "Time Effects in the Hysteresis Loop of Poly-Crystalline Barium Titanate," Journal of Applied Physics, 24 (1953), 1334-1335.
- 35 Schmitt, Roland W. "Barium Titanate at Low Temperatures," Physical Review, 82 (1951), 329.
- 36 Koenzig, W. "Atomic Positions and Vibrations in the Ferroelectric BaTiO<sub>3</sub> Lattice," Physical Review, 80 (1950), 94-95.
- 37 Shirane, Gen, and Roy Pepinsky. "Phase Transitions in Anti-Ferroelectric PbHfO<sub>3</sub>," Physical Review, 91:4 (1953), 812-815.
- 38 Sawaguchi, Esturo, and Tomoyoshi Kittoka. "Antiferroelectricity and Ferroelectricity in Lead Zirconate," Journal of the Physical Society Japan, 7 (1952), 336-337.
- 39 Shirane, Gen, and Akitsu Takeda. "On the Phase Transition in Barium-Lead Titanate," Journal of the Physical Society Japan, 6:5 (1951), 329-332.

DOCUMENT  
NUMBER

AUTHOR, TITALE, PUBLICATION

- 40 Ueda, Ryuzo. "X-ray Study on Phase Transition of Lead Zirconate,  $PbZrO_3$ ," Journal of the Physical Society Japan, 6 (1951), 209-210.
- 41 Reoberts, Shepard. "Dielectric Properties of Lead Zirconate and Barium-Lead Zirconate," Journal of the American Ceramic Society, 33:2 (1950), 63-66.
- 42 Gaaf, Ernest G. "Properties of Some High Titania Dielectric Ceramics," Ceramic Bulletin 31 (1952), 279-282.
- 43 Bunting, Elmer N., George R. Shelton, and Ansel S. Creamer. Properties of Calcium-Barium Titanate Dielectrics. National Bureau of Standards, United States Department of Commerce, Research Paper RP2025. Washington: Government Printing Office, 1949.
- 44 Coffeen, William W. "Ceramic and Dielectric Properties of the Stannates," Journal of the American Ceramic Society, 36:7 (1953), 207-214.
- 45 Cross, L. E., and B. J. Nicholson. "Ferroelectricity and Antiferroelectricity in Sodium Niobate," Research, 7 (1954), S36-S38.
- 46 Vousden, P. "The Non-Polarity of Sodium Niobate," Acta Crystallographica, 5 (1952), 690.
- 47 \_\_\_\_\_, "The Structure of the Ferroelectric Niobates and Tantalates," Acta Crystallographica, 4 (1951), 68.
- 48 Brous, J., I. Fankuchen and E. Banks. "Rare Earth Titanates with a Pervoskite Structure," Acta Crystallographica, 6 (1952), 67-70.
- 49 Günther, K. G. "Aufdampfschichten aus Halbleitenden III - V Verbindungen" (Evaporated Films of Semiconducting III - V Compounds), Zeitschrift für Naturforschung, 13a (1958), 1081-1089.
- 50 Roder, Oskar. "Gitterbildung in Aufgedampften Titanatschichten," Zeitschrift für Angewandte Physik, XII:7 (1960), 21-22.
- 51 Moll, Alexander. "Einige Eigenschaften Dünner, im Elektrischen Felde Aufgedampfter Einkristallschichten aus Barium-Strontiumtitanat" (Some Properties of Thin, Single Crystal Films of Barium-Strontium Titanate, Deposited in an Electric Field), Zeitschrift für Angewandte Physik, X:9 (1958), 410-416.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 52 Keister, F. Z. "Thin-Film Titanium Dioxide Capacitors for Microelectronic Applications," IEEE Transactions on Component Parts, CP-12 (March, 1965), 16-20.
- 53 Sekine, E., and H. Toyoda. "Experimental Study of Evaporated BaTiO<sub>3</sub> Films," Review of the Communication Laboratory (Japan), 10 (1962), 457.
- 54 Feldman, Charles. "Formation of Thin Films of BaTiO<sub>3</sub> by Evaporation," Review of Scientific Instruments, 26:5 (May, 1955), 463-466.
- 55 Bursian, E. V., and N. P. Smirnova. "Single-Crystal BaTiO<sub>3</sub> Films Grown from the Melt in an Oxygen Atmosphere," Soviet Physics, Solid State, 4:6 (1962), 1231-1232.
- 56 Frankl, D. R., A. Hagenlocher, E. D. Haffner, P. H. Keck, A. Sandor, E. Both and H. J. Degenhart. "Design of a System for Deposition of Compound Thin Films by Evaporation from Separate Sources," Proceedings of the Electronic Components Conference, AIEE (1962).
- 57 Szigeti, B. "Polarizability and Dielectric Constant of Ionic Crystals," Transactions of the Faraday Society, 45 (1949), 155-166.
- 58 Feldman, Charles. "Time Changes in Thin Films of BaTiO<sub>3</sub>," Journal of Applied Physics, 27:8 (August, 1956), 870-873.
- 59 Kibblewhite, A. C. "Noise Generation in Crystals and in Ceramic Forms of Barium Titanate When Subjected to Electric Stress," Electronic and Communication Engineer, 102 (1955), 59-68
- 60 Müller, E. K., B. J. Nicholson and M. H. Froncombe. "The Vapor Deposition of BaTiO<sub>3</sub> By a Grain-By-Grain Evaporation Method," Electrochemical Technology, 1:5-6 (1963), 158-163.
- 61 Feuersanger, A. E., A. K. Hagenlocher and A. L. Solomon. "Preparation and Properties of Thin Barium Titanate Films," Journal of the Electrochemical Society, III:12 (1964), 1387-1391.
- 62 Cummins, S. E. "Switching Behavior of Ferroelectric Bi<sub>4</sub>Ti<sub>3</sub>O<sub>12</sub>," Journal of Applied Physics, 36:6 (1965), 1958-1962.
- 63 von Hippel, A. "Ferroelectricity, Domain Structure and Phase Transitions of Barium Titanate," Reviews of Modern Physics, 22:3 (July, 1950), 221-237.

DOCUMENT  
NUMBER

## AUTHOR, TITLE, PUBLICATION

- 64 Borzyak, P. G., O. G. Sarbei and R. D. Fedorovich. "Electron Emission and Conductivity of a Silicon p - n Junction with Barium Oxide Adsorbed on its Surface," Soviet Physics - Solid State, 6:8 (February, 1965), 1783-1788.
- 65 Stekhonor, A. I., A. A. Karanyan and N. I. Astaf'ev. "Infrared Absorption Spectra of Perovskite-Type Ferroelectric Crystals," Soviet Physics - Solid State, VII:1 (1965), 119-121.
- 66 Ismailzade, I. G., and S. A. Kizhaer. "Determination of the Curie Point of The Ferroelectrics  $\text{YMnO}_3$  and  $\text{YbMnO}_3$ ," Soviet Physics - Solid State, VII:1 (1965), 236-238.
- 67 Syrkin, L. N., and A. M. El'gard. "The Electromechanical Properties of Ceramic Ferroelectrics in Strong Electric Fields and at High Pressures," Soviet Physics - Solid State, VI:11 (May, 1965), 2586-2590.
- 68 Anliker, M., H. R. Bougger and W. Kanzig. "Das Verhalten von Kolloidalen Seignetteelektrika III, Bariumtitanat  $\text{BaTiO}_3$ " Helvetica Physica Acta, 27 (1954), 99-125.
- 69 Dorin, V.A., and P. M. Tartakovskiy. "Effect of the Oxygen Liberated on Reduction  $\text{TiO}_2$ , on the Oxidation of Titanium," [sic] Physics of Metals and Metallography, 17:4 (1964), 54-58.
- 70 Wasserman, M. S., A. K. Hagenlocher and A. E. Feuersanger. Study of Thin Film Compounds Formed from Simultaneously Evaporated Constituents. Bayside, New York: General Telephone & Electronics Laboratories Inc., 1965.
- 71 Brajer, E. J., W. Jaffe, and F. Kulcsar. "Shift in the Transition Point in Barium Titanate by Partial Substitution," Journal of the Acoustical Society of America, 24 (1952), 117.
- 72 Arntz, F., and F. Chernow. "Optical and Structural Properties of Oxidized Titanium Films," Vacuum (January-February, 1965), 20-23.
- 73 Partington, J. R., G. V. Planer and I. I. Boswell. "Anomalous Dielectric Properties of Polycrystalline Titanates of the Perovskite Type," Philosophical Magazine, 40 (1949), 157-175.
- 74 Wul, B. M., and I. M. Goldman. "Dielectric Constants of Titanates of the Metals of the Second Group," Compt. Rendus (Doklady) de l'Academie des Sciences de l'URSS, 46 (1945), 139-142.



DOCUMENT  
NUMBER

## AUTHOR, TITLE, PUBLICATION

- 75 Nomura, S., and S. Sawada. "On the Anomalous Dielectric Properties of Mixtures of Lead Titanate and Strontium Titanate," Journal of the Physical Society, Japan, V (1950), 279-281.
- 76 Shirane, G., and K. Suzuki. "Phase Transition in Barium-Lead Titanate," Journal of the Physical Society, Japan, VI (1951), 274-278.
- 77 Nomura, S., and S. Sawada. "Dielectric and Thermal Properties of Barium-Lead Titanates," Journal of the Physical Society, Japan, VI (1951), 36-39.
- 78 van Santen, J. H., and G. H. Jonker. "Effect of Temperature on the Permittivity of Barium Titanate," Nature, 159 (1947), 333-334.
- 79 Nomura, S., and S. Sawada. "D. C. Resistance of Barium Titanate and Its Solid Solutions Ceramics," Journal of the Physical Society, Japan, V (1950), 227-230.
- 80 Schmidt, H. "Some Measurements on Barium Titanate," Akustische Beihfte, II:AB (1952), 83-88.
- 81 Dunst, G., M. Grotenhuis and A. G. Barkow. "Solid Solubility of Barium, Strontium and Calcium Titanates," Journal of the American Ceramic Society, 33:4 (1950), 133-139.
- 82 Shelton, G. R., A. S. Creamer and E. N. Bunting. "Properties of Barium-Magnesium Titanate Dielectrics," Journal of the American Ceramic Society, XXXI:7 (1948), 205-212.
- 83 Graf, R. G. "Effect of Impurities Upon the Dielectric Properties of Barium Titanate," Ceramic Age, 58:6 (1951), 16-19.
- 84 McQuarrie, M. C., and F. W. Behnke. "Structural and Dielectric Studies in the System (Ba,Ca)(Ti,Zr)O<sub>3</sub>," Journal of the American Ceramic Society, 37:11 (1954), 539-543.
- 85 Burfoot, J. C. "Ferroelectrics," Wireless World, (June, 1959), 256-260; (July/August, 1959), 326-332; (October, 1959), 458-462.
- 86 Wul, B. M., and I. M. Goldman. "Dielectric Constant of BaTiO<sub>3</sub> as a Function of Strength of an Alternating Field," Comptes Rendus (Doklady) de l'Academie des Sciences de l'URSS, 49 (1945), 177-180.
- 87 \_\_\_\_\_, "Dielectric Hysteresis in BaTiO<sub>3</sub>," Comptes Rendus (Doklady) de l'Academie des Sciences de l'URSS, 51 (1946), 21-23.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 88 Palandor, I. N. "Investigation of the Ferroelectric Solid Solution  $\text{Ba}(\text{Ti}_{0.95}\text{Zr}_{0.05})\text{O}_3$  at High Pressures," Soviet Physics - Solid State, VII:6 (1965), 1508-1509.
- 89 DeBretteville, Jr., A. P. "Antiferroelectric  $\text{PbZrO}_3$  and Ferroelectric  $\text{BaTiO}_3$  Phenomena," Ceramic Age, 61:4 (1953), 18-22, 92-93.
- 90A McQuarrie, Malcolm. Barium Titanate and Other Ceramic Ferroelectrics: I - Introduction," Ceramic Bulletin, 34:6 (1955), 169-172.
- 90B \_\_\_\_\_, "Barium Titanate and Other Ceramic Ferroelectrics: II - Properties of Barium Titanate," Ceramic Bulletin, 24:7 (1955), 225-230.
- 90C \_\_\_\_\_, "Barium Titanate and Other Ceramic Ferroelectrics: III - Related Materials," Ceramic Bulletin, 34:8 (1955), 256-260.
- 90D \_\_\_\_\_, "Barium Titanate and Other Ceramic Ferroelectrics: IV - Solid Solutions of Ferroelectrics," Ceramic Bulletin, 34:9 (1955), 295-298.
- 90E \_\_\_\_\_, "Barium Titanate and Other Ceramic Ferroelectrics: V - Theory," Ceramic Bulletin, 34:10 (1955), 328-331.
- 91 Rushman, D. F., and M. A. Strivens. "The Permittivity of Polycrystals of the Perovskite Type," Transactions of the Faraday Society, 42A (1946), 231-238.
- 92 Green, Jr., John P. A Method for Fabricating Thin Ferroelectric Films of Barium Titanate. Cambridge: Massachusetts Institute of Technology, Electronics Systems Laboratory, 1961.
- 93 Smith, Howard, M., and A. F. Turner. "Vacuum Deposited Thin Films Using an Optical Maser." Paper read at the 47th Annual Meeting, Optical Society of America, Rochester, New York, October 3, 4, 5, 1962.
- 94 Mason, W. P. "Aging of the Properties of Barium Titanate and Related Ceramics," Journal of the Acoustical Society of America, 27 (1955), 73-85.
- 95 Statton, W. O. The Phase Diagram of the  $\text{BaO-TiO}_2$  System, Technical Report XXXI. Cambridge: Massachusetts Institute of Technology, Laboratory for Insulation Research, April, 1950.
- 96 Cohen, M. H. "Ferroelectricity Versus Antiferroelectricity in Barium Titanate," Physical Review, 84 (1951), 369.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 97 Mason, W. P., and B. T. Matthias. "Theoretical Model for Explaining the Ferroelectric Effect in Barium Titanate," Physical Review, 74:11 (1948), 1922-1936.
- 98 Sawada, S., R. Ando and S. Nomura. "The Ferroelectric Curie Point of Tungsten Oxide," Physical Review, 82 (1951), 952-953.
- 99 Matthias, B. T., and J. P. Remeika. "Dielectric Properties of Sodium and Potassium Niobates," Physical Review, 82 (1951), 727-729.
- 100 Shirane, G., E. Sawaguchi and A. Takada. "Phase Transition in Lead Zirconate," Physical Review, 80:3 (1950), 485.
- 101 Matthias, B. T., and R. P. Remeika. "Ferroelectricity in the Ilmenite Structure," Physical Review, 76 (1949), 1886.
- 102 Linz, Jr., A. "Electrical Properties of Strontium Titanate," Physical Review, 91 (1953), 753-754.
- 103 Hulm, J. K. "Dielectric Properties of Some Alkaline Earth Titanates at Low Temperatures," Proceedings of the Physical Society (London), 63:370A (1950), 1184-1185.
- 104 Shirane, G., S. Hoshino and K. Suzuki. "Crystal Structure of Lead Titanate and Lead-Barium Titanate," Journal of the Physical Society, Japan, 5 (1950), 453-455.
- 105 Shirane, Gen, and Sadao Hoshino. "Crystal Structure of the Ferroelectric Phase in  $\text{PbZrO}_3$  Containing Ba or Ti," Physical Review, 86 (1952), 248-249.
- 106 Shirane, Gen. "Ferroelectricity and Antiferroelectricity in Ceramic  $\text{PbZrO}_3$  Containing Ba or Sr," Physical Review, 86 (April, 1952), 219-227.
- 107 Minomura, Shigeru, Tatsuyuki Kawakubo, Takehiko Nakagawa and Shoro Sawada. "Pressure Dependence of Curie Point and Tetragonal-Orthorhombic Transition Point of  $\text{BaTiO}_3$ ," Journal of Applied Physics, Japan, 3 (1964), 562-563.
- 108 Sawyer, C.B., and C. H. Tower. "Rochelle Salt as a Dielectric," Physical Review, 35 (1930), 269-273.
- 109 Cotton, G. N., and Hans Jaffe. "Piezoelectric Coefficients of Polycrystalline Barium Titanate in the Tetragonal and Orthorhombic States," Physical Review, 82 (1951), 774-775.

DOCUMENT  
NUMBER

## AUTHOR, TITLE, PUBLICATION

- 110 Murzin, V. N., and A. I. Demeshina. "Effect of Temperature on the Oscillation Spectra of Polycrystalline  $\text{BaTiO}_3$  and  $\text{SrTiO}_3$  in a Wide Spectral Range," Soviet Physics - Solid State, 6:1 (1964), 144-152.
- 111 Smolenskii, G. A., V. A. Isupov, A. I. Agranovskaya and N. N. Krainik. "New Ferroelectrics of Complex Composition IV," Soviet Physics - Solid State, 2:11 (May, 1961), 2651-2654.
- 112 Bokov, V. A., and I. E. Mylinikova. "Ferroelectric Properties of Monocrystals of New Perovskite Compounds," Soviet Physics - Solid State, 2:11 (May, 1961), 2428-2432.
- 113 Yokel, H. L., and W. C. Koehler. "On the Crystal Structure of the Manganese (III) Trioxides of the Heavy Lanthanides and Yttrium," Acta Crystallographica, 16 (1963), 957-962.
- 114 Dunne, T. G., and N. R. Stemple. "Ferroelectric Properties of  $\text{BaLi}_{2x}\text{Al}_{2-2x}\text{F}_{4x}\text{O}_{4-4x}$ ," Physical Review, 120:6 (1960), 1949-1950.
- 115 Miller, Robert C., Elizabeth A. Wood, Joseph P. Remeika, and Albert Savage. " $\text{Na}(\text{Nb}_{1-x}\text{V}_x)\text{O}$  System and 'Ferroelectricity'," Journal of Applied Physics, 33:5 (May, 1962), 1623-1630.
- 116 Bokov, V. O. "Investigation of the Temperature Dependence of Total Polarization, Coercive Force and Hysteresis Losses of Polycrystalline Solid Solutions  $(\text{Ba}, \text{Sr})\text{TiO}_3$ ,  $\text{Ba}(\text{Ti}, \text{Sn})\text{O}_3$  and  $\text{Ba}(\text{Ti}, \text{Zr})\text{O}_3$ ," Soviet Physics - Technical Physics, 3 (1958), 70-78.
- 117 Matthias, B., and A. von Hippel. "Domain Structure and Dielectric Response of Barium Titanate Single Crystals," Physical Review, 73:11 (June, 1948), 1378-1384.
- 118 Smolenskii, G. A., V. A. Isupov and A. I. Agranovskaya. "Ferroelectrics of the Oxygen-Octahedral Type with Layered Structure," Soviet Physics - Solid State, 3:3 (1961), 651-655.
- 119 Bokov, V. A., J. A. Smolenskii, S. A. Kizhaev, and I. E. Myl'nikova. "Magnetic and Electrical Properties of Ferroelectric Yttrium and Ytterbium Manganates," Soviet Physics - Solid State, 5:12 (June, 1964), 2646-2647.
- 120 Iatsenko, A. F. "The Optical Transmission Spectrum of Barium Titanate," Soviet Physics - Technical Physics, 2 (1957), 2257-2258.

DOCUMENT  
NUMBER

## AUTHOR, TITLE, PUBLICATION

- 121 Last, J. T. "Infrared-Absorption Studies on Barium Titanate and Related Materials," Physical Review, 105:6 (March, 1957), 1740-1750.
- 122 Fatuzzo, E., G. Harbeke, W. J. Merz, R. Nitsche, H. Roetschi, and W. Ruppel. "Ferroelectricity in SbSI," Physical Review, 127:6 (September, 1962), 2036-2037.
- 123 Isupov, V. A., N. N. Krainik, I. D. Fridberg and I. E. Zelenkova. "Antiferroelectric Properties of Lead Orthovanadate," Soviet Physics - Solid State, 7:4 (October, 1965), 844-847.
- 124 Mason, W. P. "Barium Titanate Ceramic as an Electromechanical Transducer," Bell Laboratories Record, XXVII:8 (August, 1949), 285-289.
- 125 Rase, D. E., and Rustum Roy. "Phase Equilibria in the System BaO - TiO<sub>2</sub>," Journal of the American Ceramic Society, 38:3 (March, 1955), 102-113.
- 126 Yamashita, J., and M. Watanabe. "On the Conductivity of Non-Polar Crystals in Strong Electrostatic Field," Journal of the Physical Society, Japan, 7 (1952), 334-335.
- 127 Shirane, Gen., and Kazuo Suzuki. "Crystal Structure of Pb(Zr-Ti)O<sub>3</sub>," Journal of the Physical Society, Japan, 7 (1952), 333.
- 128 Arend, H. T. "On the Preparation of BaTiO<sub>3</sub> Single Crystals By an Evaporation Method," Czechoslovakian Journal of Physics, B10 (1960), 971.
- 129 Shirane, Gen., and Akitsu Takeda. "Phase Transitions in Solid Solutions of PbZrO<sub>3</sub> and PbTiO<sub>3</sub>(I) Small Concentrations of PbTiO<sub>3</sub>," Journal of the Physical Society, Japan, 7:1 (March, 1951), 5-11.
- 130 Wallmark, Signe, and A. Westgren. "X-Ray Analysis of Barium Aluminates," Arkiv för Kemi, Mineralogi Och Geologi, 35:12B (April, 1937), 1-4.
- 131 Shirane, Gen., and Sadao Hoshino. "On the Phase Transition in Lead Titanate," Journal of the Physical Society, Japan, 6:4 (November, 1950), 265-270.
- 132 Sawaguchi, Etsuro, Gen Shirane and Yutaka Takagi. "Phase Transition in Lead Zirconate," Journal of the Physical Society, Japan, 6:5 (September, 1951), 333-339.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 133 Argyle, J. F., and F. A. Hummel. "Dilatometric and X-Ray Data for Lead Compounds, I," Journal of the American Ceramic Society, 43:9 (1960), 452-457.
- 134 Geller, R. F., A. S. Creamer, and E. N. Bunting. "The System: PbO-SiO<sub>2</sub>," Journal of Research of the National Bureau of Standards, 13 (August, 1934), 237-245.
- 135 Matthias, Bernd T. "Ferroelectricity," Science, 111 (May, 1951), 591-596.
- 136 Dönges, E. "Über Thiohalogenide des Dreiwertigen Antimons und Wismuts," Zeitschrift für Anorganische und Allgemeine Chemie, 263 (1950), 112-132.
- 137 Van Nitert, L. G., and L. Egerton. "Bismuth Titanate. A Ferroelectric," Journal of Applied Physics, 32 (1961), 959 (Part 1).
- 138 Roberts, Shepard. "Barium Titanate and Barium-Strontium Titanate." Unpublished Doctorate's Dissertation, Massachusetts Institute of Technology, Cambridge, 1946.
- 139 Merker, L., and L. E. Lynd. "Optically Glass-Like Material," United States Patent 2,628,156 February 10, 1953.
- 140 Evans, H. T. An X-Ray Diffraction Study of Barium Titanate. Cambridge: Massachusetts Institute of Technology, Laboratory for Insulation Research, January, 1953.
- 141 Bokov., V. A. "The Problem of the Nature of the High Permittivity Observed in Certain Solid Solutions Which Exhibit Ferroelectric Properties," Soviet Physics - Technical Physics, 2 (1957), 1657-1666.
- 142 von Hippel, A., R. G. Breckenridge, A. P. deBretteville, Jr., J. M. Brownlow, F. G. Chesley, G. Oster, L. Tisza, and W. B. Westphal. High Dielectric Constant Ceramics. Cambridge: Massachusetts Institute of Technology, Laboratory for Insulation Research, August, 1944.
- 143 Hulm, J. K. "The Dielectric Properties of Some Alkaline Earth Titanates at Low Temperatures," Proceedings of the Physical Society (London), 63A (1950), 1184-1185.
- 144 Pulvari, E. F. "Ferroelectricity," Physical Review, 120:5 (1960), 1670-1673.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 145 Pulvari, C. F. Research on Ferroelectric Switching Phenomena, Technical Documentary Report No. 62-1052. A Report Prepared by The Catholic University of America. Wright-Patterson Air Force Base, Ohio: Air Force Avionics Laboratory, December, 1962.
- 146 Kittel, C. "Theory of Antiferroelectric Crystals," Physical Review, 82:5 (1951), 729-732.
- 147 Becker, W. M. "Chemical Composition, Homogeneity and Phase Conditions in Vacuum Deposited BaTiO<sub>3</sub> Films," Bulletin of the American Physical Society, 4 (1959), 184.
- 148 Shirane, G., S. Hoshino and K. Suzuki. "X-Ray Study of the Phase Transition in Lead Titanate," Physical Review, 80 (1950), 1105-1106.
- 149 Lur's, M. S. "Thin Ferroelectric Films of Pb(Ti·Zr·Sn)O<sub>3</sub>," Soviet Physics Doklady, 4 (1960), 1082-1083.
- 150 Pulvari, C. F. Mixed Bismuth Oxide Ferrielectrics. Technical Documentary Report No. AL TDR 64-124. A Report Prepared by The Catholic University of America. Wright-Patterson Air Force Base, Ohio: Air Force Avionics Laboratory, May 1964.
- 151 Khozyainov, V. T. "Theoretical Model of ABO<sub>3</sub> Type Ferroelectrics," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 527-532.
- 152 Venevtsev, Yu. N., V. N. Lyubimov, S. P. Solov'ev and G. S. Zhdanov. "Calculation of the Internal Electric Fields and Their Gradients in Perovskite Compounds With Distinctive Dielectric Properties," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 537-541.
- 153 Krainik, N. N. "Antiferroelectricity in Compounds With the Perovskite Structure," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 550-555.
- 154 Murzin, V. N., and A. I. Demeshina. "Temperature Dependence of the Dielectric Dispersion of Polycrystalline BaTiO<sub>3</sub> and SrTiO<sub>3</sub> in a Wide Spectral Range," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 602-609.

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 155 Bogdanov, S. V., and K. V. Kiseleva. "Nature of the Dielectric Properties of  $\text{SrTiO}_3 \cdot \text{Bi}_2\text{O}_3 \cdot 3\text{TiO}_2$  Solid Solutions," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 543-549.
- 156 Fritsberg, V. Ya., and B. N. Rolov. "Some Factors Affecting the Character of Ferroelectric Phase Transitions," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 556-559.
- 157 Isupov, V. A. "Towards an Explanation of Some of the Properties of Ferroelectric Materials with a Diffuse Phase Transition," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 560-564.
- 158 Lyubimov, V. N. "Space Symmetry of Electric and Magnetic Dipole Structures," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 565-566.
- 159 Shuvalov, L. A. "Crystallophysical Classification of Ferroelectrics and Its Applications," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 567-572.
- 160 Fesenko, E. G., V. S. Filip'ev and M. F. Kupriyanov. "Concerning the Crystal Chemistry of Perovskites of Complex Composition," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 576-582.
- 161 Ismailzade, I. G. "X-ray Diffraction Studies of the High-Temperature Phase Transitions in the  $\text{NaNbO}_3\text{-NaTaO}_3$  System," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 582-587.
- 162 Merts, V. I., and R. Nitshe. "Ferroelectricity in SbSI and Other Compounds of Group V, VI and VII Elements," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 588.
- 163 Sannikov, D. G. "Contribution to the Theory of Domain Wall Motion in Ferroelectrics," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 610-614.
- 164 Khuchua, N. P., and L. F. Lychkataya. "Concerning Dispersion of the Dielectric Constant of Ferroelectric Materials," Bulletin of the Academy of Sciences of the USSR - Physical Series, 28:4 (April, 1964), 615-620.



DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 165 Arthur R. von Hippel, ed., Dielectric Materials and Applications (John Wiley and Sons, Inc., and The Technology Press of Massachusetts Institute of Technology, New York, 1954).
- 166 M. M. Nehrasov and Yu. M. Poplovko, "Investigation of the Dispersion of the Dielectric Constant of Barium Titanate Type Ferroelectrics in the Microwave Region, "Bulletin of the Academy of Sciences of the U.S.S.R. - Physical Series (April, 1964), 621-623.
- 167 J. E. Burke, ed., Progress in Ceramic Science, Volume I (Pergamon Press, New York, 1961).
- 168 Maurice H. Francombe and Hiroshi Sato, ed., Single-Crystal Films (The Macmillan Company, New York, 1964).
- 169 L. Holland, Vacuum Deposition of Thin Films (Chapman and Hall Ltd., London, 1963), fifth printing.
- 170 S. Dushman, Scientific Foundations of Vacuum Technique (John Wiley and Sons, Inc., New York, 1965), third printing, second edition.
- 171 Yu. Ya. Tomashpol'skii, Yu. N. Venevtsev, and V. N. Beznozdrev, "Ferroelectricity and Magnetism in Ferroelectric - Ferromagnetic Systems," Soviet Physics - Solid State 7, 2235-2239 (1966).
- 172 E. P. Guenok and A. Yu. Kudzin, "Influence of Polar Liquid Vapors on the Dielectric Properties of Barium Titanate Single Crystals Containing Admixtures," Soviet Physics - Solid State 7, 2300-2301 (1966).
- 173 D. A. Gorodetskii and Yu. P. Mel'nik, "The Structure of Barium Oxide Films on the (110) Face of a Tungsten Single Crystal," Soviet Physics-Solid State 7, 2248-2254 (1966).
- 174 M. B. Bronfin, A. A. Zhukhovitskii, and V. A. Marichev, "The Influence of Oxide Films on the Sublimation Kinetics," Soviet Physics - Solid State 7, 2107-2109 (1966).
- 175 A. V. Baranov, L. G. Nekrasova, and N. I. Dimova, "Investigation of Impulse Breakdown in Solid Dielectrics," Soviet Physics - Solid State 7, 2030-2031 (1966).
- 176 Yu. Ya. Tomashpol'skii and Yu. N. Venevtsev, "Ferroelectricity - Magnetism in the System  $Pb_2CoWO_6 - BaTiO_3$ ," Soviet Physics - Solid State 7, 2529-2531 (1966).
- 177 B. A. Rotenberg, Yu. L. Danilyuk, E. I. Gindin, and V. G. Prokhvatilov, "Electrical and Radiospectroscopic Investigations of Barium Titanate with Admixtures of Oxides of Trivalent Elements," Soviet Physics - Solid State 7, 2465-2469 (1966).

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 178 Egon K. Müller, Basil J. Nicholson, and Gerard L'E. Turner, "The Epitaxial Vapor Deposition of Perovskite Materials," Journal of the Electrochemical Society 110, 969-973 (1963).
- 179 Bernd T. Matthias, "Ferroelectricity," Science 113, 591-596 (1951).
- 180 P. Vousden, "The Structure of Ferroelectric Sodium Niobate at Room Temperature," ACTA Crystallography 4, 545-551 (1951).
- 181 Alfred E. Feuersanger, "Some Properties of Thin Lead Titanate Films," Bulletin of the American Physical Society 11, 35 (1966).
- 182 E. K. Müller, B. J. Nicholson, G. L'E. Turner, "The Epitaxy of Barium Titanate Films by Vapour Deposition," British Journal of Applied Physics 13, 486 (1962).
- 183 A. K. Hagenlocher, "Vapor Deposition of Lead Titanate Capacitors on Oxidized Silicon," presented at the Electrochemical Society Meeting, Cleveland, Ohio, 2 May 1966.
- 184 Rolland P. Roup, "Titania Dielectrics," American Ceramic Society Bulletin 29, 160-163 (1950).
- 185 J. P. Coughlin and E. G. King, "High-Temperature Heat Contents of Some Zirconium-Containing Substances," Journal of the American Chemical Society 72, 2262 (1950).
- 186 H. F. Kay, "Report on Researches with Ferroelectric Materials," Technical Report L/T257, The British Electrical and Allied Industries Research Association, Thorncroft Manor, Dorking Road, Leatherhead, Surrey, England (1951).
- 187 G. A. Samara and A. A. Giardini, "Pressure Dependence of the Dielectric Constant of Strontium Titanate," Physical Review 140, A954 (1965).
- 188 R. M. Glaister and H. F. Kay, "An Investigation of the Cubic-Hexagonal Transition in Barium Titanate," Proceedings of the Physical Society 76, 763 (1960).
- 189 A. V. Chebkasov, "Some Physical Properties of Vacuum Annealed Polycrystalline Barium Titanate," Bulletin of the Academy of Sciences of the U.S.S.R. Physical Series 24, 1259 (1960).
- 190 J. T. Last, "Preparation of Thin Single Crystals of Barium Titanate," Review of Scientific Instruments 28, 720 (1957).
- 191 G. Teather, "Semi-Annual Report on Thin Semi-Conducting Films for Solid Circuits - Part 6: Device Applications of Ferroelectrics," ASTIA Document AD 478-220 (1965).

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 192 Michiyoshi Tanaka and Goro Honjo, "Electron Optical Studies of Barium Titanate Single Crystal Films," *Journal of the Physical Society of Japan* 19, 954 (1964).
- 193 Kazuo Kawabe and Yoshio Inuishi, "Resistivity Anomaly in Oxygen Deficient Single Crystals of BaTiO<sub>3</sub>," *Japanese Journal of Applied Physics* 2, 590 (1963).
- 194 Helen D. Megaw, Ferroelectricity in Crystals (Methuen and Co., Ltd., London, 1957).
- 195 E. Waimier and C. Wentworth, "Niobate and Tantalate Dielectrics," *Journal of the American Ceramic Society* 35, 207.
- 196 K. H. Behrndt, "Phase and Order Transitions During and After Film Deposition," *Journal of Applied Physics* 37, No. 10.
- 197 E. Krikorian and R. J. Sneed, "Deposition Parameters Affecting Epitaxial Growth of Single Crystal Films," *Transactions of the Tenth National Vacuum Symposium*, 1963.
- 198 E. Krikorian, "Formation of Single Crystal Oxide Films by Reactive Evaporation and Sputtering," paper presented at the 13th National Vacuum Symposium, San Francisco, California, October, 1966.
- 199 Charles Kittel, Introduction to Solid State Physics (John Wiley and Sons, Inc., New York, 1956), 2nd ed.
- 200 M. J. Buerger, X-ray Crystallography (John Wiley and Sons, Inc., New York, 1942).
- 201 Dr. F. Huber and W. Witt, "High Performance Thin Films for Microcircuits (Capacitors)," CR-65-419-22, Report No. 1, Electronic Components Laboratory, U. S. Army Electronics Command, Fort Monmouth, New Jersey, 30 June 1965.
- 202 A. E. Hultquist and M. E. Sibert, "Electrolytic Preparation of High Dielectric Thin Films," No. N65-19875, Lockheed Missiles and Space Co., Sunnyvale, California, 8 February 1965, Quarterly Report.
- 203 A. E. Hultquist and M. E. Sibert, "Electrolytic Preparation of High Dielectric Thin Films," N65-33872, Lockheed Palo Alto Research Laboratory Sunnyvale, California, 3 August 1965, Quarterly Report.
- 204 N. M. Tallan, et al, "Defect Structure and Electrical Properties of Some Refractory Metal Oxides," *Materials Science Research* 2, 33-67 (Plenum Press, 1965).

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

- 205 B. T. Matthias, "Ferro-Electric Properties of  $WO_3$ ," Physical Review 76 (1949).
- 206 M. A. Nicolet, "Unipolar Space-Charge-Limited Current in Solids with Nonuniform Spacial Distribution of Shallow Traps," Journal of Applied Physics 37, 4224-4235 (October 1966).
- 207 P. Kofstad and D. J. Ruzicka, "On the Defect Structure of  $ZrO_2$  and  $HfO_2$ ," Journal of the Electrochemical Society 110, 181-184 (March 1963).
- 208 D. H. Rice and R. W. Stowe, "Research on Anisotropy in Polycrystalline Dielectric Materials," Final Report, Melpar, Inc., Falls Church, Va. (June 1966).
- 209 M. J. F. Gaze and J. P. Keene, "Thin Films Formed by Electrochemical Reactions," Quarterly Report No. 8, Texas Instruments, Inc., Dallas, Texas (30 November 1965).
- 210 H. J. Degenhart and I. H. Pratt, "Preparation and Evaluation of Vacuum-Deposited Thin-Film Capacitors," Technical Report 2228, U. S. Army Signal Research and Development Laboratory, Ft. Monmouth, N. J. (July 1961).
- 211 N. N. Krainik, "Ferroelectric and Antiferroelectric Properties of Solid Solutions of  $NaNbO_3$ - $PbZrO_3$ ," Academy of Sciences, Solid State-Soviet Physics 2, 633-637 (October 1960).
- 212 J. T. Last, "Infrared-Absorption Studies on Barium Titanate and Related Materials," Physical Review 105, No. 6 (March 15, 1957).
- 213 L. E. Cross and B. J. Nicholson, "The Optical and Electrical Properties of Single Crystals of Sodium Niobate," Philosophical Magazine 46, No. 376 453-466 (May 1955).
- 214 M. H. Francombe and B. Lewis, "Structure and Phase Transitions of Ferroelectric Sodium-Lead Niobate and of Other Sodium Niobate Type Ceramics," Journal of Electronics 2, No. 4 (January 1957).
- 215 B. Lewis and L. A. Thomas, "The Ferroelectric Behavior of Some New Niobate Compounds and Their Application to Piezoelectric Devices," Solid State Physics in Electronics and Telecom. 2 (Semiconductors-part 2) 883-890 (1960).
- 216 R. F. Jannick and D. H. Whitmore, "Electrical Conductivity and Ther electric Power of Niobium Dioxide," Journal Physical Chemical Solids 27, 1183-1187 (July 19, 1965).
- 217 R. M. Glaister, "Solid Solution Dielectrics Based on Sodium Niobate," Jnl. American Ceramic Society 43, No. 7, 348-353 (July 1960).

DOCUMENT  
NUMBER

AUTHOR, TITLE, PUBLICATION

218 Donald H. Rice and Richard W. Stowe, "Research on Anisotropy in Polycrystalline Dielectric Materials," Final Report, Melpar, Inc. Falls Church, Virginia (June 1966).