

Isoprenoid biosynthesis in plant chloroplasts via the MEP pathway: Direct thylakoid/ferredoxin-dependent photoreduction of GcpE/IspG

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Abstract In the methylerythritol phosphate pathway for isoprenoid biosynthesis, the GcpE/IspG enzyme catalyzes the conversion of 2-C-methyl-D-erythritol 2,4-cyclodiphosphate into (*E*)-4-hydroxy-3-methylbut-2-enyl diphosphate. This reaction requires a double one-electron transfer involving a [4Fe-4S] cluster. A thylakoid preparation from spinach chloroplasts was capable in the presence of light to act as sole electron donor for the plant GcpE *Arabidopsis thaliana* in the absence of any pyridine nucleotide. This is in sharp contrast with the bacterial *Escherichia coli* GcpE, which requires flavodoxin/flavodoxin reductase and NADPH as reducing system and represents the first proof that the electron flow from photosynthesis can directly act in phototrophic organisms as reducer in the 2-C-methyl-D-erythritol 4-phosphate pathway, most probably via ferredoxin, in the absence of any reducing cofactor. In the dark, the plant GcpE catalysis requires in addition of ferredoxin NADP⁺/ferredoxin oxido-reductase and NADPH as electron shuttle.
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1. Introduction

Isoprenoids represent the most diverse natural product family. They are found in all living organisms, including essential metabolites such as sterols regulating plasma membrane fluidity and permeability in eukaryotes and precursor of steroid hormones in animals, dolichols and long chain polyprenols acting as carbohydrate carriers, the prenyl chains of quinones from electron transport chains (ubiquinones, men-

aquinones, plastoquinone), carotenoids and phytol from chloroplasts, as well as a multitude of secondary metabolites of less obvious function. Highest structural diversity is certainly found in phototrophic organisms (mosses, liverworts, ferns, higher plants). Isopentenyl diphosphate (IPP) (9) and dimethylallyl diphosphate (DMAPP) (10) are the universal precursors for all isoprenoids in all living organisms (Fig. 1). Two biosynthetic pathways are leading to the isoprene unit. In the firstly elucidated mevalonate (MVA) pathway, IPP is synthesized from acetyl-CoA much like fatty acids, via MVA, which is the key intermediate of this metabolic route [1]. In the more recently disclosed methylerythritol phosphate (MEP) pathway [2–4], IPP (9) and DMAPP (10) are carbohydrate derivatives, their synthesis starting from pyruvate (1) and D-glyceraldehyde phosphate (2) via 1-deoxy-D-xylulose 5-phosphate (3), 2-C-methyl-D-erythritol 4-phosphate (4), 4-diphosphocytidyl-2-C-methyl-D-erythritol (5), 4-diphosphocytidyl-2-C-methyl-D-erythritol 2-phosphate (6), 2-C-methyl-D-erythritol 2,4-cyclodiphosphate (MEcDP) (7) and 4-hydroxy-2-methylbut-2-enyl 1-phosphate (HMBPP) (8) (Fig. 1). The MVA pathway is the only pathway found in animals and fungi as well as in the cytoplasm of phototrophic organisms, whereas the MEP pathway is present in most bacteria [5,6] and in plant chloroplasts [7–9].

The last two enzymes of the MEP pathway, encoded by the *gcpE* and *lytB* genes, respectively, catalyze the conversion of MEcDP (7) into HMBPP (8) (Fig. 1) [10–13] and the conversion of HMBPP into IPP and DMAPP [14–17] by eliminations coupled with reduction steps (Fig. 1). Both enzymes are characterized upon reconstitution by a [4Fe-4S] prosthetic group [18–20], acting in its reduced [4Fe-4S]¹⁺ form as one electron donor. For catalytic activity, both enzymes have to be coupled with a system allowing the reduction of the oxidized [4Fe-4S]²⁺ cluster. This can be completed in the bacterium *Escherichia coli* by the natural flavodoxin/flavodoxin reductase/NADPH system [18,21]. The *fldA* gene encoding flavodoxin I was found to be essential in *E. coli* and supposed to be mainly involved in the reduction of the GcpE and LytB Fe/S clusters [22]. Reduction can also be performed by chemical means: e.g., with the semiquinone radical of 5-deazaflavin for the recombinant GcpE enzyme from *E. coli* [18] and *Arabidopsis thaliana* [19] or even dithionite

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Abbreviations: DMAPP, dimethylallyl diphosphate; GcpE, IspG, HMBPP synthase; HMBPP, (*E*)-4-hydroxy-3-methylbut-2-enyl diphosphate; IPP, isopentenyl diphosphate; LytB, IspH, HMBPP reductase; MEcDP, 2-C-methyl-D-erythritol 2,4-cyclodiphosphate; MEP, 2-C-methyl-D-erythritol 4-phosphate

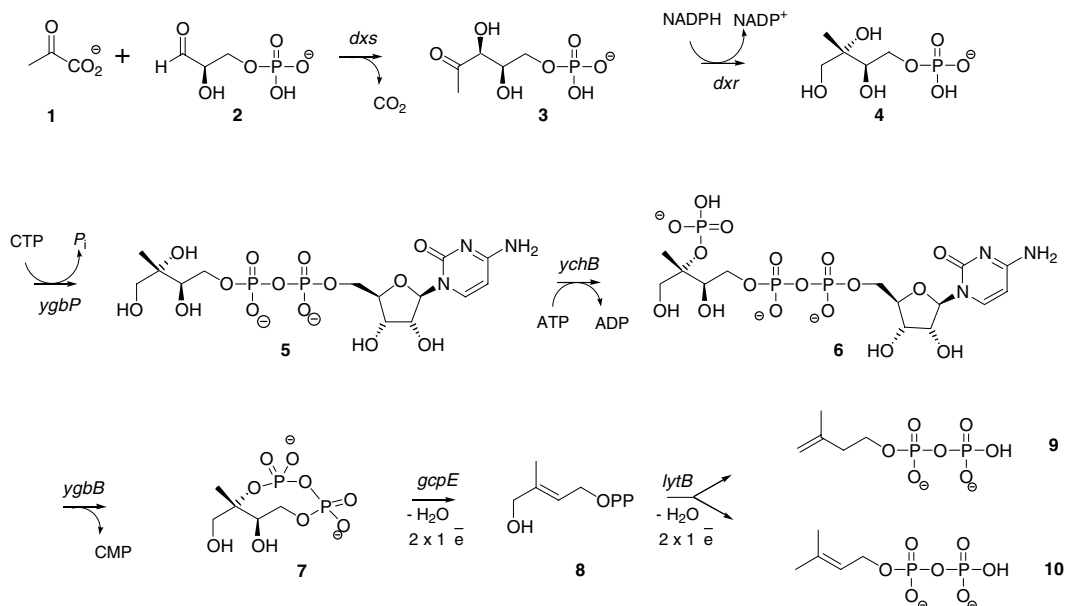


Fig. 1. MEP pathway for isoprenoid biosynthesis.

for GcpE from the bacterium *Thermus thermophilus* [23] and LytB from *Aquifex aeolicus* [15]. The plant enzyme showed, however, no activity in the presence of the flavodoxin/flavodoxin reductase from *E. coli* and NADPH [19]. In addition, flavodoxin is absent in the plastids of phototrophic organisms and in the plastid-like apicoplasts of *Plasmodium* spp. [24,25]. As the MEP pathway is closely linked to photosynthesis by its chloroplast localization and the supply of the first C₃ precursors, it seemed tempting to check whether the electron transport chain involved in photosynthesis was capable of transferring electrons to the Fe/S clusters of the last two enzymes via ferredoxin. The electron transfer from ferredoxin onto GcpE is apparently possible in the cyanobacterium *Thermosynechococcus elongatus* lacking flavodoxin [26]. LytB from *E. coli* displays some reducing activity in the presence of spinach ferredoxin/ferredoxin–NADP⁺ reductase/NADPH [27], whereas the same enzyme from *Plasmodium falciparum* was shown to be catalytically active in the presence of the same system [25]. In the present study, we show that the electron flow from photosynthesis can directly provide in the presence of light and via ferredoxin the electrons required for the GcpE catalyzed reaction without any contribution of the ferredoxin–NADP⁺ reductase and NADPH, the latter system being only required in the dark.

2. Materials and methods

2.1. Purification of the enzymes

The GcpE enzyme from *A. thaliana* (At-GcpE) was purified and reconstituted as already described [19]. Spinach (*Spinacea oleracea*) was purchased from a local market. Ferredoxin was purified from spinach leaves using the purification procedure described for the *Clostridium pasteurianum* ferredoxin [28]. Ferredoxin–NADP reductase from spinach leaves was purified on a ferredoxin–Sepharose affinity column as described for ferredoxin–thioredoxin reductase [29]. A final purification step was carried out on a 2′–5′ ADP–Sepharose column [30].

2.2. Preparation of thylakoids

All preparative steps were performed at 4 °C. Spinach leaves (50 g) were washed with water, deribbed and homogenized in a Waring blender in 50 mM MES buffer (pH 6.5, 250 ml) containing 330 mM sorbitol and 1 mM MgCl₂. The homogenate was filtered through several layers of surgical gauze, and the residue was discarded. After centrifugation of the filtrate (3000 × g, 1 min), the pellet was used for the preparation of spinach thylakoids as described for the preparation of pea thylakoids [31]. This final thylakoid pellet was resuspended in 50 mM HEPES (pH 7.6, 2 ml), 330 mM sorbitol and 1 mM MgCl₂ to reach a ca. 3.4 mg ml⁻¹ chlorophyll concentration. Chlorophyll was determined by the method of Arnon [32].

2.3. Enzymatic assays using thylakoids

[2-¹⁴C]MEcDP was synthesized [12] and purified by high-performance liquid chromatography [33] as previously reported. The reaction medium (97 μl) containing 920 μM [2-¹⁴C]MEcDP (0.015 μCi) and various amounts of thylakoid preparation in 50 mM Tris–HCl (pH 8) was degassed for 25 min at 4 °C in the dark under a stream of argon before adding with a gas-tight syringe the At-GcpE solution (3 μl, 1.05 μM final concentration). Each assay was incubated under irradiation with a white fluorescent tube (Osram L 18W/765, 250 μEinstein m⁻² s⁻¹) at 30 °C for 10 min. The reaction was stopped by freezing in liquid nitrogen. After thawing, losses occurring during the degassing process were evaluated by measuring the volume of the assays with a syringe. The assays were hydrolyzed using alkaline phosphatase (5 μl, 0.16 U) at 25 °C for 15 h. Two different methods were used for determining the GcpE enzyme activity. In Method 1, an aliquot (40 μl) was separated on a thin-layer chromatography silica plate. After elution with isopropanol/water/ethyl acetate (6:3:1), the radioactivity was monitored using a PhosphorImager. After scraping off the bands corresponding to (*E*)-2-methylbut-2-ene-1,4-diol (*R*_f = 0.79) and [2-¹⁴C]MEcDP (*R*_f = 0.50), the silica was directly introduced into counting vials, and the radioactivity was quantified by liquid scintillation.

In Method 2, another aliquot (30 μl) was passed through a QAE-A25 Sephadex column (0.7 cm × 1.5 cm) equilibrated with water. [2-¹⁴C]-(*E*)-2-methylbut-2-ene-1,4-diol was recovered by washing the column with water (4 ml), and radioactivity was measured by liquid scintillation after lyophilization of the sample. The background corresponding to the radioactivity of the water eluate of an aliquot (30 μl) from a similar blank experiment performed without At-GcpE was deduced from the former value. [2-¹⁴C]MEcDP was eluted with 1.5 M ammonium acetate (2 ml) and counted separately.

2.4. Enzymatic assays using ferredoxin, ferredoxin reductase and NADPH

A reaction medium (70 μl final volume) containing 23 μM [$2\text{-}^{14}\text{C}$]MEcDP (0.025 μCi), 5 mM DTT, 2.5 mM NADPH, 40 μM ferredoxin, 9.2 nM ferredoxin reductase in 50 mM Tris–HCl (pH 8) was degassed for 45 min at room temperature under a stream of wet argon before adding 2.5 μM of AtGcpE. The incubation was performed anaerobically at 30 $^{\circ}\text{C}$ for 1 h.

3. Results

The GcpE enzyme from *A. thaliana* (At-GcpE) was purified and reconstituted as already described [19]. As this enzyme contains an oxygen sensitive $[4\text{Fe-4S}]^{2+}$ prosthetic group, all tests were performed under an inert argon atmosphere.

A novel assay was developed for At-GcpE. It relies on the non-sensitivity of MEcDP (7) towards alkaline phosphatase [12], which, in contrast, hydrolyses HMBPP (8) into (*E*)-2-methylbut-2-ene-1,4-diol, which is more easily purified on TLC silica gel plates than HMBPP. Furthermore, MEcDP is retained on anionic ion exchange column, whereas (*E*)-2-methylbut-2-ene-1,4-diol is not. These chromatographic properties were used to quantify the GcpE activity by two different analytical methods that led to the same specific activity (Table 1). According to the blank experiment of Method 1, MEcDP was not totally recovered from the silica, most probably because of its high polarity. Method 2 using a separation of the diol and MEcDP on an ion exchange column seemed thus more efficient for the recovery of the cyclodiphosphate. In Method 2, the water eluate contained (*E*)-2-methylbut-2-ene-1,4-diol as well as some free methylerythritol (less than 4%) that arose from the slow degradation of MEcDP. Corrections were accordingly introduced.

In the presence of NADPH, ferredoxin reductase (9.2 nM) and ferredoxin (40 μM), At-GcpE (2.5 μM) converted quantitatively [$2\text{-}^{14}\text{C}$]MEcDP (7) (23 μM) into HMBPP (8) after 1 h. Accordingly, no specific activity was calculated. No activity was detected when At-GcpE was tested under the same conditions, but in the absence of ferredoxin and in the sole

presence of NADPH and ferredoxin reductase as reducing system, indicating that ferredoxin is essential in the reduction process. This shows for the first time that the plant GcpE can use ferredoxin as an electron donor, and that ferredoxin might represent the reducing system coupled to GcpE in vivo. The reducing power of ferredoxin in chloroplasts is however linked to photosynthesis, ferredoxin being the final electron donor. At-GcpE was accordingly tested in the absence of NADPH and added ferredoxin reductase, but in the light and in the presence of spinach thylakoids containing some endogenous ferredoxin in order to mimic what may happen in vivo and to check whether the photosynthetic electron may reduce the Fe/S cluster flow.

The specific activity of reconstituted At-GcpE was determined upon photoactivation in the presence of increasing amounts of the thylakoid preparation. Thylakoids were quantified by their chlorophyll content determined by the method of Arnon [39]. A 160 $\text{nmol min}^{-1} \text{mg}^{-1}$ constant maximal activity was obtained in a 300–600 $\mu\text{g ml}^{-1}$ chlorophyll concentration range, already indicating saturation at a ca. 300 $\mu\text{g ml}^{-1}$ chlorophyll concentration (Fig. 2). No activity was detected when the assays were run in the dark. As plant GcpE is a plastidial enzyme, the possible presence of endogenous spinach GcpE was checked in the thylakoid preparation. A test was therefore performed without adding At-GcpE, resulting in no activity. This indicated that the thylakoid preparation was not contaminated with significant amounts of endogenous GcpE enzyme from spinach. At-GcpE was only active in the light and required the presence of thylakoids, showing the involvement of the photosynthetic electron transfer, most probably via ferredoxin, in the MEP pathway.

The bacterial GcpE is known to use flavodoxin/flavodoxin reductase/NADPH as reducing system [18]. It was therefore tempting to check whether the At-GcpE enzyme test with thylakoids applies for the *E. coli* enzyme. A thylakoid preparation (corresponding to 600 $\mu\text{g ml}^{-1}$ chlorophyll) was tested in the presence of light and in the absence of NADPH on the bacterial GcpE from *E. coli*: a ca. 30 $\text{nmol min}^{-1} \text{mg}^{-1}$ activity was recorded. The spinach thylakoids are thus also capable in the

Table 1
Conversion of MEcDP (79000 dpm) by GcpE from *A. thaliana* and *E. coli* in the presence of thylakoids (600 $\mu\text{g ml}^{-1}$ chlorophyll) and light

		Ec-GcpE	At-GcpE	Blank
Final volume + phosphatase (5 μl)		95 μl	95 μl	90 μl
Method 1: TLC	(<i>E</i>)-2-methylbut-2-ene-1,4-diol	13900 dpm in 40 μl	10280 dpm in 40 μl	0
		33000 dpm total	24400 dpm total	0
	MEcDP	15300 dpm in 40 μl	20900 dpm in 40 μl	23200 dpm
		36300 dpm total	49600 dpm total	52200 dpm total
Method 2: ion exchange column	Water elution: diol	12700 dpm in 30 μl	8800 dpm in 30 μl	1000 dpm in 30 μl
		40200 dpm total	28000 dpm total	2950 dpm total
	NH ₄ OAc elution: MEcDP	8250 dpm in 30 μl	15200 dpm in 30 μl	19900 dpm in 30 μl
		26100 dpm in total	48200 dpm total	59700 dpm total
GcpE		0.0225 mg	0.0083 mg	0
Time (min)		45	15	45
Substrate (μmol)		0.061	0.061	0.061
Specific activity ($\text{nmol min}^{-1} \text{mg}^{-1}$, Method 1)		25	152	–
Specific activity ($\text{nmol min}^{-1} \text{mg}^{-1}$, Method 2)		28	155	–

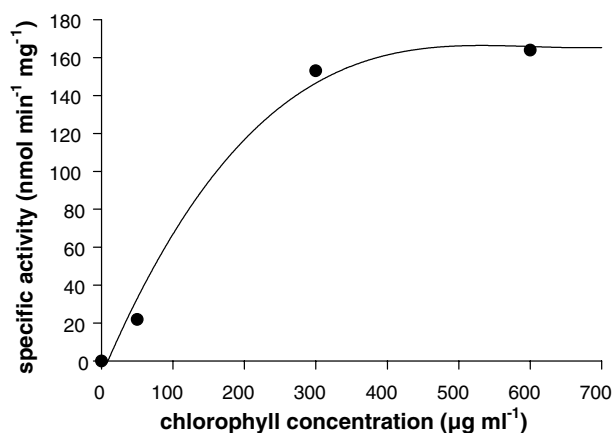


Fig. 2. Activity of At-GcpE upon illumination in the presence of increasing amounts of thylakoid under an inert N_2 atmosphere. $250 \mu\text{Einstein m}^{-2} \text{s}^{-1}$, 30°C , 10 min, At-GcpE (1.05 μM), $[2\text{-}^{14}\text{C}]\text{MEcDP}$ (920 μM , 0.015 μCi).

presence of light of acting as reducing system for the bacterial enzyme. It was previously shown that ferredoxin from spinach in the presence of ferredoxin reductase and NADPH could be used as reducing system in the absence of light on the bacterial GcpE [34].

4. Discussion

The GcpE catalyzed reaction is dependent on an electron donating system. In *E. coli*, this can be completed by the flavodoxin/flavodoxin reductase/NADPH system already present in the bacterium [18]. This bacterial redox shuttle is, however, ineffective with the *A. thaliana* GcpE [19]. In addition, flavo-

doxin is absent in plant chloroplasts, suggesting that another electron transfer shuttle is operating in these organelles.

In the present study, we show that electron flow from photosynthesis can directly provide in the presence of light and via ferredoxin the electrons required for the GcpE catalyzed reaction without any contribution of ferredoxin–NADP⁺ reductase and NADPH (Fig. 3, Part A). Indeed, the GcpE activity in the light did not require the addition of pyridine nucleotides, and no activity could be detected in the dark, suggesting that the reducing cofactors had been eliminated by washing the thylakoids. In contrast, in the absence of light, GcpE requires an electron shuttle, as shown by the activity in the presence of ferredoxin/ferredoxin reductase/NADPH. This situation prevails in the dark, and especially in non-photosynthetic tissues such as roots (Fig. 3, Part B). Indeed, this reducing system has been characterized in pea roots [35], and an active MEP pathway has been pointed out in roots of *Glycyrrhiza glabra* [36], *Catharanthus roseus* [37], *Ophiorrhiza pumila* [38] and *Daucus carota* [39], mycorrhized roots of *Hordeum vulgare* [40], many species of the Poaceae family [41,42] and *Salvia miltiorrhiza* [43]. The *dxs* and *dxr* genes are expressed in the roots of *A. thaliana* [44] and *Artemisia annua* [45]. In *Medicago truncatula*, two distantly related genes have been found: the deoxyxylulose phosphate synthase gene, DXS1, is preferentially expressed in many developing tissues excepting roots, whereas the DXS2 transcript levels are highly expressed in roots upon mycorrhizal colonization and are low in other tissues [46]. A mycorrhiza-inducible DXR has been identified by immunolocalization in arbuscule-containing cells of *Zea mays* [47].

Two biosynthetic pathways for isoprenoid biosynthesis are present in plants. The mevalonate pathway is found in the cytoplasm, whereas the MEP pathway is only present in the plastids. The latter biosynthetic route is strongly linked to photosynthesis. Of its first two precursors, glyceraldehyde 3-phos-

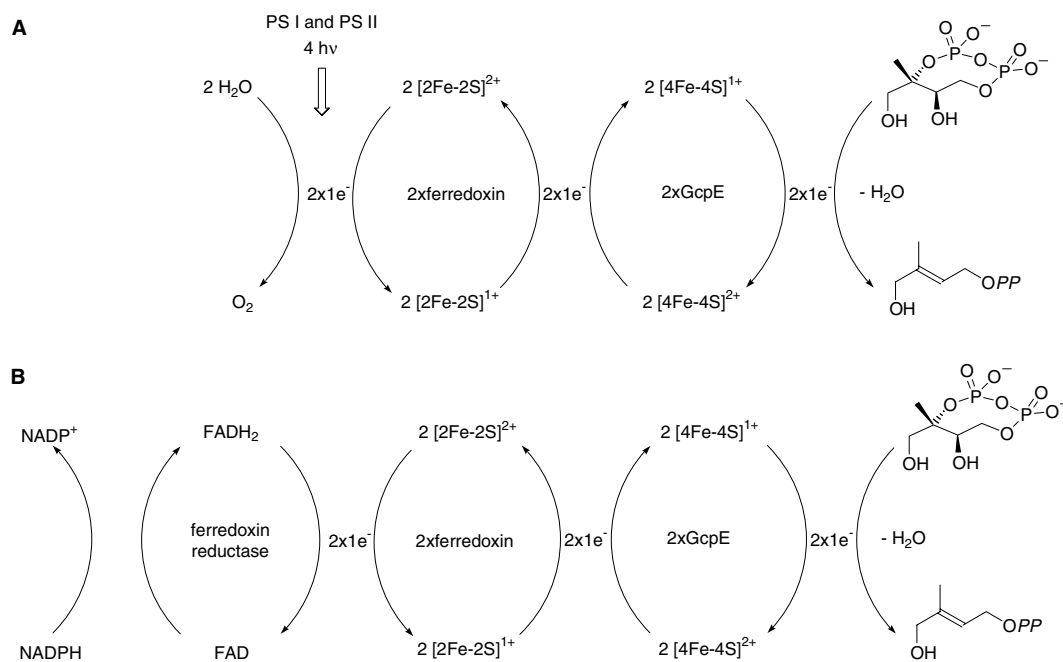


Fig. 3. MEP pathway for isoprenoid biosynthesis: hypothetical biogenetic scheme for plant GcpE activity: (A) in photosynthetic tissues in the light; (B) in photosynthetic tissues in the dark and in non-photosynthetic tissues. PS I and PS II: photosystems I and II.

phate belongs to the Calvin cycle, and pyruvate is derived from hexoses directly issued from photosynthesis. Activation of the MEP pathway by light is also documented by the rapid emission of isoprenoids known to be derived from MEP pathway upon irradiation of green tissues [48]: isoprene [49], 2-methyl-3-buten-2-ol [50] and monoterpenes [51]. $^{13}\text{C}_2$ feeding to *Quercus rubra* leaves indicated a close linkage between the photosynthetic carbon reducing pathway and the carbon source required for isoprene biosynthesis [52]. Illumination of *Quercus ilex* leaves induced emission of α -pinene and other monoterpenes, which depended on photosynthetic carbon assimilation products. In the latter study, de novo synthesis, and not mobilization of endogenous terpenes or terpene precursor pools, is responsible for this quick production [53,54]. Light dependence of isoprene and methylbutenol emission in cottonwood (*Populus deltoides*) is in part due to controls over DMAPP biosynthesis [55]. In the petals of snapdragon flowers (*Antirrhinum majus*), the MEP pathway provides IPP and DMAPP for both the plastidial monoterpenes and the cytosolic sesquiterpenes biosynthesis and operates only in the light in a rhythmic manner controlled by the circadian clock [56]. All this data strongly suggests a close linkage between photosynthesis and the MEP pathway for isoprenoid biosynthesis, not only for providing the carbon source, but also the required reducing electron flow. Light may thus directly interfere with the regulation of the MEP pathway and with the cross-talk between the cytoplasmic mevalonate and the plastidial MEP pathways [57].

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