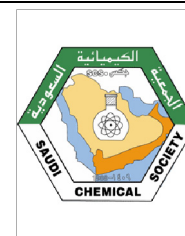




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ORIGINAL ARTICLE

1,7-Sigmatropic rearrangement in 1,2-dihydro and 1,2,3,4-tetrahydroquinoline synthesis using marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ as a catalyst



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Imino Diels–Alder;
Marine sponge/oxalic acid

Abstract Marine sponge/oxalic acid was found to be an efficient catalyst for the imino Diels–Alder reaction of synthesized *N*-aryl-substituted aldimines and various alkenes to provide 1,2-dihydro and 1,2,3,4-tetrahydro-quinolines using 1,7-sigmatropic rearrangement with induction of chirality. Mild reaction conditions, simple experimental procedure, good yields of products, and optical active induction render this new method attractive for 1,7-sigmatropic rearrangement of imino Diels–Alder reaction.

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1. Introduction

Tetrahydroquinoline and quinoline derivatives are widely found in many natural products, which exhibit various biological and pharmaceutical activities (Aslokar et al., 2004; Omura and Nakagama, 1981; Kumar et al., 2004; Smith et al., 2003; Teimouri et al., in press; Sarveswari et al., in press; Khabaz-zadeh et al., in press; Leeson et al., 1992; Katritzky et al., 1995; Kametani et al., 1986; Babu et al., 1999). For the synthesis of tetrahydroquinolines, the imino Diels–Alder reaction

(povarov) of *N*-arylimine with olefins is highly efficient (Scheme 1) (Jorgensen, 2000; Zhang et al., 2005; Kouznetsov et al., 2006; Zhou and Xu, 2008; Kamble et al., 2010; Buonora et al., 2001; Benforouz and Ahmadian, 2000; Kouznetsov et al., 1998; Kouznetsov, 2009).

Lewis acids like $\text{BF}_3\text{-Et}_2\text{O}$, montmorillonite K-10 clay, CAN, and protic acids such as TFA, *p*-TsOH, oxalic acid, I_2 , KHSO_4 , etc. (Kametani et al., 1985; Ma et al., 1999; Babu and Perumal, 1997; Zhang and Li, 2002; Yadav et al., 2001; Mahesh et al., 2004; Kouznetsov et al., 2004; Maiti and Kundu, 2006; Hadden and Stevenson, 1999; Makioka et al., 1995; Kobayashi et al., 1995; Yadav et al., 2002; Sridharan et al., 2007; Boger and Weinreb, 1987; Mellor et al., 1991; Grieco and Bahasas, 1988; Nagarajan and Perumal, 2001; Xia and Lu, 2005; Kumar et al., 2004; Anniyappan et al., 2002) and silica compounds such as $\text{HClO}_4\text{-SiO}_2$, SBA-Pr- SO_3H (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov

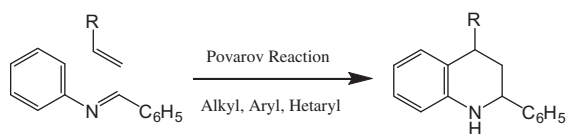
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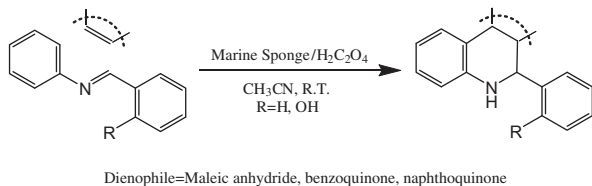
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Scheme 1 General imino diels-alder reaction.



Dienophile=Maleic anhydride, benzoquinone, naphthoquinone

Scheme 2 Marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ catalysed.

et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009) have been found to catalyze these reactions. However, most of these methods involve expensive and more than stoichiometric amounts of Lewis acids which are required due to coordination of the Lewis acids to imine nitrogen, with longer reaction times and strong acidic conditions. In recent years, great efforts have been made to develop organic reactions in an environment-friendly manner (green chemistry).

Marine sponges are known as a prolific source of biologically active and structurally unique metabolites. They are known to produce a large number and diversity of secondary metabolites (Rifai et al., 2005; McCaffrey and Endean, 1985). As there is no report of marine sponges of Iranian coast of Persian Gulf, we studied shallow sponges (*Desmospongia* sp.) of Qeshm Island in offshore zone which can be the source of new biological active compounds. In organic chemistry, these sponges are important and optical active source for catalytic reactions such as oxidation, reduction etc. (Sarma et al., 2005). The chiral non-racemic catalyst, marine sponge, which has optical active compounds such as alkaloids, terpenoids, is the good catalyst for induction of chirality into desired products with excellent enantioselectivities in 1,2,3,4-tetrahydroquinolines synthesis. The chiral and acidic marine sponge can activate the C–N bond for nucleophilic addition with high and predictable asymmetric induction, and is easily removed from the product. Therefore, the search continues for a better catalyst in 1,2,3,4-tetrahydroquinolines synthesis in terms of operational simplicity, with greater yields and in particular, excellent enantioselectivities.

In the course of our studies for C–C bond formation (Shushizadeh and Kiany, 2009) and development of useful synthetic methodologies, we have designed and synthesized a series of new compounds containing the 1,2,3,4-tetrahydroquinolines derivatives using 1,7-sigmatropic rearrangement of imino Diels–Alder reaction (Scheme 2).

2. Experimental

All starting materials were purchased from Merck and Aldrich Companies. The IR spectra were recorded on a Perkin-Elmer RXI infrared spectrometer. ^1H NMR spectra were recorded with a 400 MHz Bruker ft-nmr spectrometer. TLC accomplished the purity of substrates and reactions were monitored

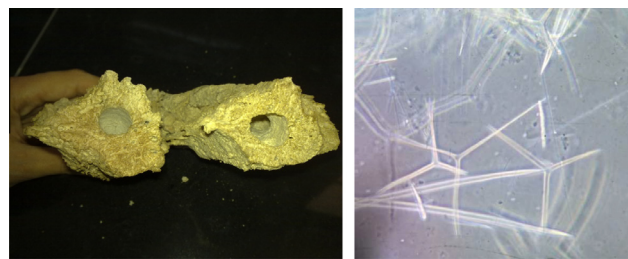


Figure 1 (a) Marine sponge (*Desmospongiae* sp.); (b) *Siliceous spicules*.

on silica gel polygram SIGL/UV254 plates. Melting points reported were determined by open capillary method using a Mettler melting point apparatus and are uncorrected.

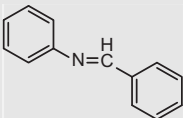
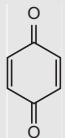
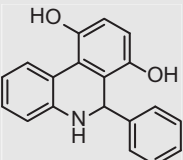
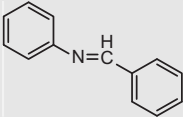
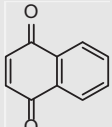
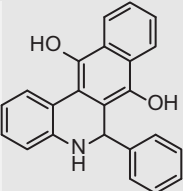
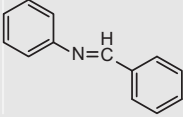
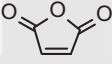
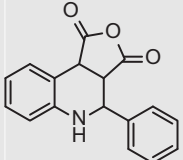
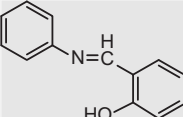
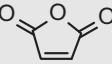
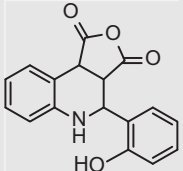
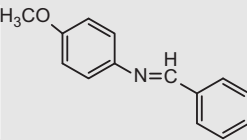
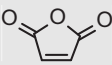
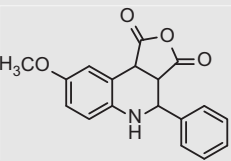
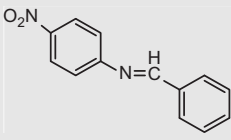
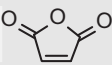
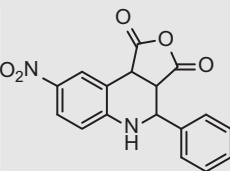
The marine sponge (*Desmospongiae* sp.) was used as solid supported with oxalic acid for transformation of acidity and chirality in reaction media and products. Samples were collected in May 2010 at a depth between 5 and 10 m in the Nakhiloo Island, Bushehr, Iran (North coast of Persian Gulf) and were washed several times using deionized water to remove extraneous and salts. They were then dried in an oven at 60 °C for 48 h. The dried marine sponge chopped, sieved and the particles with an average size of 0.5 mm were used for acid catalyzed experiments. Identification of sponges was carried out kindly by Dr. Sayed Mohammad Bagher Nabavi. The species investigated in this study are *Desmospongiae* sp. (Fig. 1a) which has siliceous (SiO_2) spicules (Fig. 1b).

2.1. General procedure

Imines were prepared by the standard procedure of refluxing equimolecular amounts of the suitable aniline and aldehyde in ethanol for 3–4 h and filtering the solid that precipitated after cooling (Chakraborti et al., 2004; Guzen et al., 2007)¹. The suitable imine (2 mmol) was dissolved in acetonitrile (15 mL). To this stirred solution was added the solution of 2.5 mmol of dienophile and marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ (50 mg,

¹ Selected spectral data for some imines in Table 1 are given: (*E*)-*N*-benzylideneaniline (**1a**) Chakraborti et al., 2004; Guzen et al., 2007: IR(cm^{-1} , KBr): 1610 (C=N), 1520 (C=C), 1232, 1033, 790 cm^{-1} . ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 8.40 (1H, s), 7.72 (2H, d), 7.62 (2H, d), 7.51 (1H, t), 7.4–7.2 (5H, m); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 160 (HC=N), 152, 137.4, 132.2, 128(6C), 125.1, 120. (*E*)-2-((phenylimino)methyl)phenol (**1d**) Chakraborti et al., 2004; Guzen et al., 2007: IR(cm^{-1} , KBr): 3421 (–OH), 1600 (C=N), 1550, 1520 (C=C), 1222, 1033, 850, 760. ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 12.4 (1H, bs), 8.50 (1H, s), 7.8 (2H, d), 7.72 (1H, d), 7.61 (1H, t), 7.3–7.0 (5H, m); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 162 (HC=N), 158, 147.1, 132(2C), 129(2C), 125, 119(3C), 117.9, 115. (*E*)-*N*-benzylidene-4-methoxybenzenamine (**1e**) Chakraborti et al., 2004; Guzen et al., 2007: IR(cm^{-1} , KBr): 1504, 1251, 1033, 831. ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 8.20 (1H, s), 7.16 (2H, d), 6.83 (2H, d), 6.43–7.49 (5H, m), 3.72 (3H, s); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 158.4 (HC=N) 152.2, 145.6, 145.2, 144.2, 122.2, 115.5, 114.3, 112.0 (2Ar, 12C), 55.3 (OMe). (*E*)-*N*-benzylidene-4-nitrobenzenamine (**1f**) Chakraborti et al., 2004; Guzen et al., 2007: IR(cm^{-1} , KBr): 3073, 2348, 1869, 1531, 1118. ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 8.65 (1H, s), 8.38 (1H, s), 8.16 (1H, d), 8.10 (1H, d), 7.50 (1H, t), 7.62–7.11 (5H, m); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 162.2 (HC=N), 157.3, 151.0, 138.0, 134.3, 130.5, 130.0, 129.5, 127.0, 125.7, 123.6, 121.2 (2Ar, 12C).

Table 1 Marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ -catalyzed 1,2-dihydroquinolines (**3a** and **3b**) and 1,2,3,4-tetrahydroquinolines (**3c–3f**) synthesis by 1,7-sigmatropic rearrangement of imino Diels–Alder reaction.

Entry	Diene (1)	Dienophile (2)	Product (3)	Time (h)	Yield (%)	m.p. ($^{\circ}\text{C}$)	<i>cis/trans</i>
a				12	73	198	–
b				10	70	180	–
c				12	75	186	20/80
d				11	80	210	18/82
e				10	59	195	–
f				13	45	220 (dec.)	5/85

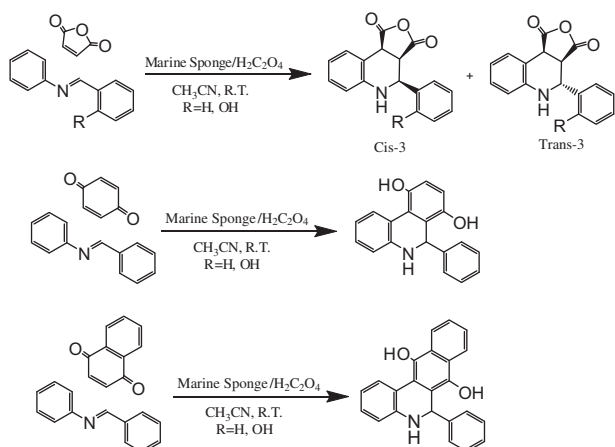
All products were confirmed by comparison with authentic samples (IR, ^1H NMR and TLC).

Isolated yield after column chromatography. The *cis/trans* ratio in entries c,d,f are about 20/80, 18/82 and 15/85.

10 mol% and pH = 3–4) in 10 mL acetonitrile, and stirring was continued until TLC monitoring showed the completion of the reaction (10–12 h). Then the reaction mixture was extracted in CH_2Cl_2 (2 \times 25 mL). The combined solutions were dried over anhydrous magnesium sulfate. Evaporation of the solvent afforded the products (for yields see Table 1).

3. Result and discussion

Marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ was prepared (in a 10:1 molar ratio) by adding the amounts of the suitable marine sponge powder into 2 M oxalic acid solution and stirred for 24 h at room temperature, and then the solid was filtered after drying. Then



Scheme 3 High stereo-selective products in marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ catalyze imino diels-alder reaction.

0.5 g of this solid was added into solution of 1:1 molar ratios of imine and dienophile in acetonitrile. The reaction mixture was stirred at room temperature. Several imines underwent the above rearrangement reacting with dienophile to form a series of 1,2,3,4-tetrahydroquinolines derivatives (Table 1).

For our initial studies we chose to explore the marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ -catalyzed rearrangement in 1,2,3,4-tetrahydroquinolines synthesis by the reaction of imines with dienophiles. The reaction was carried out in the presence of a catalytic amount of marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ (10 mol%) under mild reaction condition. The rearranged products were obtained with high stereo-selectivity in good to high yields (Scheme 3). The structures of the products were characterized by their melting points and spectral (^1H NMR and IR) data (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009). The products were obtained as a mixture of *trans* and *cis* isomers, favoring the *trans* diastereomer.

From these results, we propose the following possible mechanism to account for the reaction. The observed stereo-selectivity can be explained by taking into account that the Povarov reaction is known to proceed in a concerted [2+4] cycloaddition reaction between the intermediate iminium species and dienophile. Therefore, the generation of this intermediate from the reaction between diene (maleic anhydride) and marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ is expected under our conditions. The final 1,7-sigmatropic rearrangement step should take place for cyclization of benzene ring, leading to produce a mixture of *trans* and *cis* isomers, favoring the *trans* diastereomer (Fig. 2).

These results indicate that marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ can be used as an efficient and chiral catalyst in the imino Diels–Alder reaction of *N*-benzylideneanilines with dienophiles such as maleic anhydride, *p*-benzoquinone and *p*-naphthoquinone (Table 1, entries 1–6). In this reaction, *p*-benzoquinone and *p*-naphthoquinone afforded the hydroxyl products with good yields.

Selected spectral data for the products in Table 1 are given:

6-Phenyl-5,6-dihydrophenanthridine-7,10-diol (**3a**) (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009)–(entry 1): IR(cm^{-1} , KBr): 3448 (–OH), 3263 (–NH), 1508 (C=C), 754, 695 (aromatic); ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 7.9 (2H, d), 7.3–7.6 (5H, m), 7.2(2H, d), 6.9(2H, dd), 3.9 (1H, s); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 156.3 (2C–OH), 140, 133(2C), 128.5(3C), 126(2C), 123.2(2C), 121.4, 119.1(2C), 114(2C), 111, 50.8.

6-Phenyl-5,6-dihydrobenzo[*j*]phenanthridine-7,12-diol (**3b**) (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009)–(entry 2): IR(cm^{-1} , KBr): 3402 (–OH), 3260 (–NH), 1508 (C=C), 754, 698 (aromatic); ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 7.65–8.1 (4H, m), 7.35–7.6 (5H, m), 6.9–7.2 (4H, m), 3.8 (1H, s); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 151, 138 (2C–OH), 139, 136(2C), 133, 130(3C), 126.2(2C), 125(3C), 123.9(2C), 121.3(2C), 119(2C), 113.6, 51.

4-Phenyl-3a,4a,9b-tetrahydrofuro[3,4-*c*]quinoline-1,3-dione (**3c**) (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009)–(entry 3): IR(cm^{-1} , KBr): 3275 (–NH), 3073 (=CH), 1701 (C=C), 1542 (C=C), 758, 608 (aromatic); ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 7.2–7.5 (5H, m), 6.4–6.9 (4H, m), 4.1 (1H, d), 3.8–4.0 (1H, d), 3.75 (1H, dd); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 174, 168 (2C=O), 143(=C–N), 135, 128–126(6C), 123, 119, 115, 110, 59, 48, 44.

4-(2-Hydroxyphenyl)-3a,4a,9b-tetrahydrofuro[3,4-*c*]quinoline-1,3-dione (**3d**) (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009)–(entry 4): IR(cm^{-1} , KBr): 3449 (–OH), 3274 (–NH), 3072 (=CH), 1700 (C=C), 1541 (C=C), 757, 607 (aromatic); ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 7.2–7.6 (4H, m), 6.9–7.1 (4H, m), 5.9

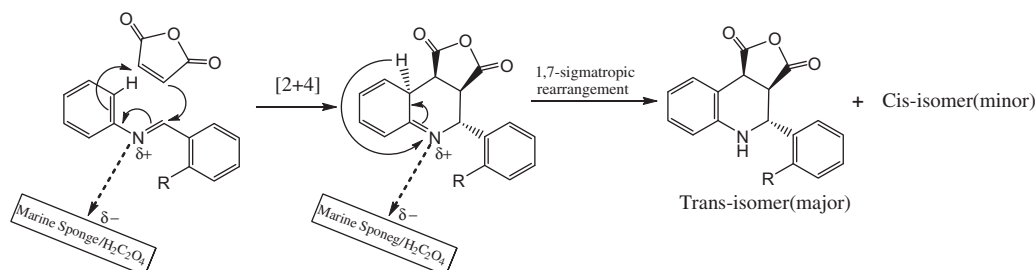


Figure 2 Possible mechanism in 1,7-sigmatropic rearrangement in marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$ catalyzed imino diels-alder reaction.

(1H, d), 4.4(1H, dd), 4.2(1H, d); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 170 (2C=O), 155(C-OH), 144(=C-N), 132, 128, 126.5, 123.8, 118(2C), 117(2C), 114.2, 109, 57, 45, 42.9.

8-Methoxy-4-phenyl-3a,4a,9b-tetrahydrofuro[3,4-*c*]quinoline-1,3-dione (**3e**) (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009)–(entry 5): IR(cm^{-1} , KBr): 3230 (–NH), 3060 (=CH), 1700 (C=O), 1550 (C=C), 758, 608 (aromatic); ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 7.65–7.2 (5H, m), 6.9–6.55 (3H, m), 5.6 (1H, d), 4.5 (1H, dd), 4.2 (1H, d), 3.7 (3H, s); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 174.8, 165 (2C=O), 152(=C–O), 135.2(=C–N), 134, 128–126(5C), 109, 102, 118, 113, 59, 47, 55, 50.

8-Nitro-4-phenyl-3a,4a,9b-tetrahydrofuro[3,4-*c*]quinoline-1,3-dione (**3f**) (Kamble et al., 2010; Mohammadi Ziarani et al., in press; Kouznetsov et al., 2007; Zhang et al., 2002; Powell and Batey, 2003; Han et al., 2006; Khan et al., 2011; Byrne and Evans, 2008; Xue et al., 2009)–(entry 6): IR(cm^{-1} , KBr): 3081, 2350, 1855, 1580, 1118; ^1H NMR (400 MHz, CDCl_3 , TMS, δ ppm): 8.3 (1H, s), 7.95 (1H, dd), 7.55 (1H, d), 7.45–6.8 (5H, m), 5.6 (1H, d), 4.53 (1H, dd), 4.3 (1H, d); ^{13}C NMR (400 MHz, CDCl_3 , TMS, δ ppm): 175, 169 (2C=O), 148(=C–N), 138.5, 134.5, 128.4–126(5C), 125(2C), 115, 105, 59.2, 46.8, 44.8.

4. Conclusion

In conclusion, we have developed a new, chiral and efficient catalyst, marine sponge/ $\text{H}_2\text{C}_2\text{O}_4$, for synthesis of 1,2,3,4-tetrahydroquinolines derivatives using 1,7-sigmatropic rearrangement of imino Diels–Alder reaction in good to high yields (45–80%).

Acknowledgments

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