Two conformations of the integrin A-domain (I-domain): a pathway for activation?

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Background: Integrins are plasma membrane proteins that mediate adhesion to other cells and to components of the extracellular matrix. Most integrins are constitutively inactive in resting cells, but are rapidly and reversibly activated in response to agonists, leading to highly regulated cell adhesion. This activation is associated with conformational changes in their extracellular portions, but the nature of the structural changes that lead to a change in adhesiveness is not understood. The interactions of several integrins with their extracellular ligands are mediated by an A-type domain (generally called the I-domain in integrins). Binding of the I-domain to protein ligands is dependent on divalent cations. We have described previously the structure of the I-domain from complement receptor 3 with bound Mg²⁺, in which the glutamate side chain from a second I-domain completes the octahedral coordination sphere of the metal, acting as a ligand mimetic.

I-domain with bound Mn^{2+} , in which water completes the metal coordination sphere and there is no equivalent of the glutamate ligand. Comparison of the two crystal forms reveals a change in metal coordination which is linked to a large (10 Å) shift of the C-terminal helix and the burial of two phenylalanine residues into the hydrophobic core of the Mn^{2+} form. These structural changes, analogous to those seen in the signal-transducing G-proteins, alter the electrophilicity of the metal, reducing its ability to bind ligand-associated acidic residues, and dramatically alter the surface of the protein implicated in binding ligand.

Conclusions: Our observations provide the first atomic resolution view of conformational changes in an integrin domain, and suggest how these changes are linked to a change in integrin adhesiveness. We propose that the Mg^{2+} form represents the conformation of the domain in the active state and the Mn^{2+} form the conformation in the inactive state of the integrin.

Results: We now describe a new crystal form of the

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Introduction

The integrin family of cell-adhesion molecules are $\alpha\beta$ heterodimers with a generally conserved structure [1]. So far, 16 distinct α subunits (with ~1000 residues) have •been identified that pair with eight different β subunits (~750 residues). Electron microscopy indicates that integrins have an extracellular globular 'head' (~100 Å diameter) comprising the N-terminal portion of both subunits, followed by a long 'stalk' (~100 Å) between the globular head and the membrane-spanning helices, and a short C-terminal cytoplasmic 'tail' for each subunit.

Biochemical and biophysical probes have shown that activation of integrins to an adhesion-promoting state involves conformational changes in the extracellular domains. Thus, fluorescence energy transfer experiments show that the integrin $\alpha_{IIb}\beta_3$ changes shape on activation [2], and there is also a marked change in proteolytic sensitivity, consistent with a change of shape and/or rigidity [3]. Furthermore, monoclonal antibodies have been described that discriminate between active and inactive states (reviewed in [4–6]). Allostery within the extracellular domains has been vividly demonstrated by electron miscroscopic images of integrins with both

activating antibodies and ligands bound, separated by a distance of 160 Å [7]. However, the nature of the conformational changes that lead to a change in adhesiveness is not known.

Seven integrin α subunits contain a ~190-residue sequence near their N terminus that is homologous to the A-domains of von Willebrand factor, and is generally called the I-domain in integrins. This domain plays a major role in ligand binding [8–14], and undergoes conformational changes in response to receptor activation [14–16]. We have previously proposed that a conserved region found in all β subunits, and implicated in ligand binding, also adopts an A-type fold [17].

The β_2 integrins are a subfamily of integrins found exclusively on leukocytes [18]. One of them, complement receptor 3 (CR3; also known as CD11b/CD18, $\alpha_M\beta_2$ or Mac-1), is the major integrin of phagocytic cells, and is required for phagocytosis and the stable adhesion of leukocytes to endothelium and the subsequent migration into inflamed organs. These functions are mediated through binding to several physiologic ligands including iC3b (the major complement C3 opsonin),

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ICAM-1 (CD54), and the coagulation factor fibrinogen. The recombinant I-domain from CR3 has many of the binding functions of the intact receptor, and ligand binding is dependent on divalent cations. We have previously determined its structure from crystals grown in the presence of Mg²⁺ [17]. It adopts a classic 'dinucleotide-binding' or 'Rossmann' fold, and contains an unusual Mg²⁺ coordination site at its surface, which we call the metalion-dependent adhesion site (MIDAS) motif. Point mutations in the MIDAS motif abrogate cation binding and binding to many protein ligands [8-10,17]. In the Mg²⁺-containing I-domain crystal, one of the coordinating ligands is the glutamate side chain from another I-domain molecule, and we have proposed that this interaction is a mimic of a natural integrin-ligand interaction, because all known integrin ligands contain acidic residues as a key feature of their binding motifs [19].

Here we describe a new crystal form of the I-domain, grown in the presence of Mn^{2+} , in which the domain is 'unliganded', and compare it with the previously described 'liganded' structure grown in the presence of Mg^{2+} . The two structures display large differences, and we propose that they represent the conformations of the domain that exist in the active and inactive states of the integrin.

Results and discussion

Two crystal structures of the I-domain

The I-domain adopts the dinucleotide-binding or Rossmann fold, with seven α helices (labelled $\alpha 1 - \alpha 7$) surrounding a mostly parallel β sheet (labelled $\beta A - \beta F$; Fig. 1a). The MIDAS motif is located in a crevice at the top of the sheet. The motif, conserved in many members of the A-domain superfamily, comprises a DxSxS sequence (residues 140-144; single-letter code; where x is any amino acid) from the $\beta A - \alpha 1$ loop, Thr209 from the α 3- α 4 loop and Asp242 from the β D- α 5 loop. Crystallization of the domain in a minimal concentration of Mn²⁺ produced a crystal form different from that previously obtained in an excess of Mg²⁺ [17]. We determined the Mn²⁺ I-domain structure by molecular replacement and refined the model at 2.0 Å resolution (Fig. 1b). About two-thirds of the molecule has a very similar structure in both crystal forms: the central β sheet (except for β F) and the α helices (except for $\alpha 1$, $\alpha 6$ and $\alpha 7$) can be closely superimposed; residues 132-140, 167-241 and 245-265 overlap with a root mean square (rms) difference for main-chain atoms of 0.46 Å (Fig. 2a), and the two crystal forms are compared in this reference frame.

The metal coordination is different in the two structures. In the Mg^{2+} form (Fig. 3a), two serines and a threonine



Fig. 1. (a) Primary sequence of the CR3 I-domain, with secondary structure assignments. Residue numbering is according to the intact protein. MIDAS residues are shown in bold. Lower-case letters denote residues invisible in the electron-density map. The last six residues are derived from the cloning vector. (b) Stereo electron-density map $(2F_{0}-F_{c})$ of the Mn²⁺ coordination site, contoured at 1.7 o. The two phenylalanine residues (F275 and F302) which become buried in the Mn²⁺ form are shown. Water molecules are indicated by crosses. For clarity, only density within 2 Å of the selected atoms is displayed.



Fig. 2. (a) Rms difference by residue (main chain) between Mg^{2+} and Mn^{2+} I-domains. (Domains were superimposed on residues 132–140, 167–241 and 245–265 with an rms difference of 0.46 Å.) Secondary structure elements for the Mn^{2+} form are indicated along the horizontal axis. The positions of crystal contacts (interatomic distances <4 Å) for Mg^{2+} (open triangles) and Mn^{2+} (filled triangles) forms are also marked. (b) Comparison of main-chain temperature factors in the Mg^{2+} (thick line) and Mn^{2+} (thin line) I-domains.

make strong bonds to the metal via their side-chain hydroxyl oxygen atoms (bond lengths 2.0 ± 0.1 Å); two water molecules also bind directly to the metal, and two aspartates make indirect contacts via water molecules. The absence of direct bonds between Mg^{2+} and the negatively charged aspartate ligands enhances the electrophilicity of the metal, and, in the crystal, the side chain of a glutamate residue from a neighbouring molecule completes the coordination; that is, the Mg²⁺ I-domain is liganded. In the Mn²⁺ form (Fig. 3b), the metal moves by 2.3 Å; bonds are made to the two serines, as before, but the bond to Thr209 is broken and replaced by a direct bond to Asp242, reducing the electrophilicity of the metal. Water molecules complete the coordination sphere, and there is no equivalent of the exogenous glutamate (the nearest acidic side chain from another molecule is 9 Å away); that is, the Mn^{2+} I-domain appears to be unliganded.

The change in metal coordination is linked to structural changes in the protein (Figs 3c,d and 4). In the Mn^{2+} form, the C-terminal helix, α 7, moves 10 Å up the side of the molecule. This requires a repacking of the hydrophobic face of α 7 against the side of the domain; the

helix extends by an extra turn at its C terminus and bends in the middle, wrapping around the hydrophobic core. At the N terminus of α 7, Phe302, which was completely exposed in the Mg²⁺ form, inserts into the top of the domain. Burial of this phenylalanine causes shifts in three loops and their connecting strands and helices which disrupt the MIDAS motif. First, the $\beta A-\alpha 1$ loop, which includes the DxSxS sequence, shifts so that Ser144 moves 1.5 Å further away from Thr209, making it impossible for a metal ion to bridge both residues. Second, the $\beta E - \alpha 6$ loop, which packed against the $\beta A - \alpha 1$ loop in the Mg²⁺ form, shifts by 1.5 Å, and rearrangements occur that bury a second phenylalanine (Phe275). Third, Phe275 packs against the $\beta D - \alpha 5$ loop, which flips so that the carbonyl of Gly243 rotates by 180° and its Ca moves by 4 Å; the C α of Asp242 moves by 2.5 Å so that its side chain can make a direct bond to the metal.

The temperature factors (B values) are, as expected, low for the helices and strands, and higher for some of the loops, particularly the $\alpha 1-\beta B$, $\alpha 4-\beta D$ and $\beta E-\alpha 6$ loops (Fig. 2b). The B values are generally well conserved between the two structures, despite the different crystal contacts and large structural changes. The C-terminal



Fig. 3. Structural comparisons of the Mg^{2+} and Mn^{2+} I-domains. (a) The MIDAS motif in the Mg^{2+} form. (b) The MIDAS motif in the Mn^{2+} form. The colour code is oxygen atoms (red), carbon (black), schematic backbone (grey) and the glutamate from a neighbouring molecule (gold). Water molecules are labelled w1-w3. Selected hydrogen bonds are shown as dashed red lines. (c) Stereo C α plot comparing Mg²⁺ (red) and Mn²⁺ (blue) I-domains. Regions of large change are shown with thicker lines. The metal ions are shown as spheres. The N and C termini and occasional residues are labelled. (d) Stereo close-up comparison of the Mg^{2+} (red) and the Mn^{2+} (blue) forms: C α plot, together with selected side chains (metal-coordinating residues and the two buried phenylalanines), and the main chain for residues 242-243.



Fig. 4. Schematic diagram of the Mn^{2+} I-domain (grey). Major conformational differences are shown in red (Mg^{2+}) and blue (Mn^{2+}). β strands are labelled A–F and helices 1–7.

helix, $\alpha 7$, is equally well ordered in the two structures, although the $\beta E-\alpha 6$ loop is more mobile in the Mn²⁺ form. Crystal contacts are quite evenly distributed across the surface of both crystal forms (Fig. 2a). The $\alpha 7$ helix in the Mg²⁺ structure is involved in two major contacts centred around Phe302 and Glu314, so it is possible that crystal contacts play a role in determining the position of this helix.

Structural determinants of activation

These changes in metal coordination and associated protein movements have a striking parallel in the signaltransducing G-proteins [20–23]. For example, in the GTP-bound ('active') form of $p21^{ras}$ a Mg^{2+} ion binds directly to a serine, a threonine and two water molecules, and indirectly to an aspartate; the other coordination sites are filled by the β - and γ -phosphate oxygen atoms of the GTP. On hydrolysis of the GTP and loss of the γ -phosphate, the Mg^{2+} switches its coordination so that the bond to the threonine is lost and a direct bond to the aspartate is gained (Fig. 5). The changes in metal coordination cause large changes in the effector region which bury a previously exposed isoleucine residue and switch the molecule from an active to an inactive state. Thus, the inactive conformer of $p21^{ras}$ is characterized by reduced electrophilicity of the metal (i.e. direct coordination of an aspartate residue) and the burial of a hydrophobic residue.

Is an analogous pathway of activation operating in the I-domain? In the absence of exogenous ligand, we suggest that the Mn^{2+} form is thermodynamically favoured, owing to the burial of hydrophobic residues into the core of the protein, and that this conformation represents the inactive state. In the Mg^{2+} crystals, ligand binding (in this case provided by another I-domain molecule) and the formation of a strong bond to Thr209 provide enough energy to expel two phenylalanines from the hydrophobic core, and we suggest that this structure represents the active state. The existence of two distinct conformations is consistent with the ability of certain antibodies (that recognize epitopes on the I-domain) to discriminate between active and inactive states of the integrin [14-16]. Furthermore, the large shift of the C-terminal helix in the context of the whole integrin may radically alter the accessibility of the domain to both antibodies and protein ligands. Recent studies have shown that the ligand-binding surface is not limited to the MIDAS motif, but extends across the upper surface of the domain ([24]; P Rieu, T Sugimori and MA Arnaout, unpublished data). Indeed, the two structures have quite different surface appearances, which should provide further control of ligand affinity: for example, an acidic pocket evident in the Mg^{2+} form is largely obscured in the Mn^{2+} form (Fig. 6).



Fig. 5. Comparison of structural changes in (a) the MIDAS motif and (b) the $p21^{ras}$ -GTP complex. Movements of the metal-coordinating residues are indicated (see text).



Binding of ligands to the recombinant I-domain *in vitro* is supported by both Mn^{2+} and Mg^{2+} , but not by Ca^{2+} [9,25], recapitulating the behaviour of the intact integrin [8,26]. *In vivo*, Mg^{2+} is likely to be the cation providing the integrin-ligand bridge, because of its high concentration (~1 mM) in plasma. Our model suggests why Ca^{2+} cannot support ligand binding. The chemistry of Ca^{2+} is mostly ionic and it prefers negatively charged ligands, whereas Mg^{2+} and Mn^{2+} are smaller ions with a greater tendency to form covalent bonds [27]. Mg^{2+} and Mn^{2+} can be stably coordinated by uncharged serine and threonines residues, but Ca^{2+} cannot be thus ligated. Ca^{2+} should therefore prefer, and at high concentrations stabilize, the inactive conformation, in which an aspartate is directly coordinated instead of a threonine.

Our assignment of the Mn²⁺ crystal form as the inactive conformation appears at first sight surprising, as Mn²⁺ is well known as an activator of integrins in vitro [28]. We cannot provide a simple explanation for the apparent crystallization of the Mn²⁺ I-domain in an inactive form, but it should be pointed out that the binding affinities of active and inactive integrins for monomeric ligands differ typically by one order of magnitude ($\sim 1.5-2$ kcal mol⁻¹), so that the difference in energy between the two I-domain conformations is likely to be small compared with typical crystal lattice energies. There are differences in metal concentrations and temperature which could explain the different crystal forms (see the Materials and methods section), and we also note that in crystals of a homologous I-domain from CD11a/CD18 (LFA-1), the molecule adopts the inactive Mn²⁺ conformation described here (with the MIDAS motif unliganded), whether grown in the presence of Mg^{2+} or Mn^{2+} (our unpublished data). This shows that the metal ion is not the sole determinant of the protein conformation, and suggests that the two conformations described in this paper represent two 'snapshots' of a dynamic equilibrium existing in solution. Clearly, further biochemical studies are required to test our proposal directly.

Our structural data suggest that the I-domain can exist in two distinct tertiary forms, a high-affinity state and a low-affinity state. Because the isolated I-domain binds **Fig. 6.** Solvent-accessible surface of (a) the Mg²⁺ and (b) the Mn²⁺ I-domain, with surface charge calculated by GRASP [39]: negatively charged regions in red, positive in blue. The view is perpendicular to that shown in Figure 4.

ligands constitutively [8], quaternary constraints must be present in the intact $\alpha\beta$ integrin that hold the domain in the inactive state in the absence of an activation signal [29]. We do not yet know how the tertiary changes we have described are linked to the quaternary changes that occur on integrin activation, but their magnitude and character are reminiscent of those found in other well-studied allosteric systems [30].

Biological implications

Integrins are plasma membrane proteins that mediate adhesion to other cells and to components of the extracellular matrix. Integrin-mediated cell adhesion is regulated by intracellular signals that lead to conformational changes in the extracellular domains of these receptors. This property is essential both in development and in many functions of the immune system, but its structural basis is unknown. Integrins are $\alpha\beta$ heterodimers with a generally conserved structure. The A-domain (or I-domain) is a ~190-residue extracellular fragment from the α subunit of certain integrins that has many of the ligand-binding properties of the intact receptor. Our structural studies of this domain reveal two distinct conformations, in which a change in metal coordination is linked to large conformational changes in the protein. These changes cause a dramatic alteration in the surface of the protein implicated in binding ligand, suggesting that they are linked to a change in adhesiveness. The two conformations described here have parallels with those observed in the signal-transducing G-proteins, and we propose, by analogy, that they represent the structure of the I-domain in the active and inactive states of the integrin.

In integrins that do not contain an α subunit I-domain, a conserved region in the β subunit has been implicated in ligand binding and activation [4]. The region contains part of the metal-binding consensus sequence (DxSxS), and we have previously proposed, based on hydropathy plots, that this conserved region also adopts an I-domain fold [17]. If this is the case, then it too may undergo structural transformations of the kind described here.

The extensive biochemical data on integrins indicate that these cell-adhesion molecules exist primarily in one of two states: active and inactive. Our results show that the I-domain can exist in two distinct tertiary states, and it is tempting to suggest a two-state allosteric model of integrin activation [31]. In such a model, the integrin is in equilibrium between a low-affinity quaternary state (inactive, 'T' state) and a high-affinity quaternary state (active, 'R' state). The T state is more stable in the absence of ligand or activation signal (owing to the burial of hydrophobic groups), but the R state has a higher affinity for ligand (owing to increased electrophilicity of the metal and the presence of a complementary ligand-binding surface). The tertiary and quaternary states are linked, so that each quaternary state is associated with a unique set of stable tertiary states, 't' and 'r'. What are the available data? First, binding of monoclonal antibodies to regions of the α or β subunit not directly involved in binding ligand can activate integrins [4]. In a twostate model, these antibodies preferentially bind to and stabilize the 'r' tertiary state of an integrin domain (i.e. the conformation of that domain in the R quaternary state), which shifts the quaternary equilibrium towards the R state, in which the tertiary state of the ligand-binding domain is in the high-affinity 'r' conformation. Second, mutations in subunits that are not directly implicated in ligand binding can have dramatic effects on integrin-ligand interactions. For example, a point mutation in the conserved region of the β subunit abolishes adhesion to physiological ligands, even though the major binding site for these ligands is located on the α subunit [32,33]. In a two-state model, the mutation locks the β -conserved region into the 't' state and thus shifts the quaternary equilibrium of the integrin towards the T state, in which the α subunit I-domain is in the 't' conformation. Finally, the exquisite sensitivity of integrins to even conservative mutations [34] is reminiscent of other well-studied allosteric systems, where point mutations can radically alter the equilibrium between the T and R states [30].

Materials and methods

Crystallization

Recombinant CD11b I-domain was expressed and purified as described previously [17]. Crystals of the Mn²⁺ form grow overnight by sitting-drop vapour-equilibration of protein (12 mg ml⁻¹) against 14% PEG 3350, 0.1 M Tris pH 8.5, 10 mM β -mercaptoethanol, 1 mM MnCl₂ (initial) at 4°C. Crystals grow in the monoclinic space group C2, with a=135 Å, b=37.1 Å, c=38.5 Å, β =92.6°. Under the alkaline conditions used, most of the unchelated Mn²⁺ oxidizes and

precipitates (as MnO₂) within a few hours. The final concentration of Mn²⁺ ion in the crystallization drop was estimated with the dye Eriochrome Black T (Sigma Chemical Co., St Louis, MO), and found to be equimolar with the protein; rinsing the crystals in fresh Mn²⁺ solution prior to data collection ensured full metal occupancy. Most other metals, with the exception of Mg²⁺, are unstable or precipitate under the alkaline conditions required for crystal growth. Crystals of the Mg²⁺ form grow under very similar conditions: 12% PEG 8000, 0.1 M Tris pH 8.5, 1 mM MgCl₂, 10 mM β-mercaptoethanol at room temperature, and belong to the tetragonal space group P4₃ with unit cell dimensions a=45.7 Å, c=94.3 Å [17]. Crystals do not grow in the absence of metal.

Data collection and refinement

A Mn²⁺ I-domain crystal was transferred briefly into a freshly prepared solution containing 14% PEG 3350, 0.1 M Tris pH 8.5, 1 mM MnCl₂, 30% glycerol, and immediately frozen in a nitrogen stream at 100 K prior to data collection. Data were collected to 2.0 Å resolution on beam line X12C of the National Synchrotron Light Source at the Brookhaven National Laboratory using a MarResearch imaging plate, with a wavelength of 0.98 Å. Data were processed with DENZO and SCALEPACK [35] to an R_{sym} of 6% (27% in outer shell) and the data are 99.5% complete (12933 unique reflections; average redundancy 3.6). The structure was solved with the molecular replacement program AMoRe [36] using the Mg^{2+} structure as the search model [17]. The rotational and translational search gave one obvious solution with an initial R-factor of 45%. The structure was refined using X-PLOR [37] and manual model building into omit maps [38]. The metal ion was not included in the initial refinement, which converged at an R-factor of 30%. Many regions of the protein had poor or discontinuous electron density at this stage, especially around the MIDAS motif and helix α 7. A peak in the F_o-F_c map at the presumed location of the metal was no stronger than a water molecule peak, but inclusion of a Mn²⁺ ion into the model dramatically improved both the electron density for the MIDAS motif and the overall appearance of the map; the new position of the helix α 7 also became clear in an F_0 -F map. After rebuilding these regions the model refined smoothly to a final R-factor of 21% (outer shell=28%) for data between 12 Å and 2.0 Å resolution (F>2 σ ; the final R-factor for all data is 22%). The Mn²⁺ ion refines to a reasonable B value (26 $Å^2$), consistent with full occupancy of the site, although it is considerably more mobile than the metal in the Mg²⁺ I-domain. The final model comprises all non-hydrogen atoms of residues Asp132-Ala318 and 100 water molecules, and has good geometry: rms deviations from ideality of bond lengths and bond angles of 0.017 Å and 3.0° respectively. The Ramachandran plot shows that no nonglycine main-chain dihedral angles fall in the disallowed regions; 156 residues fall in the most favoured region, 8 in additional allowed regions, and 2 in generously allowed regions. These last two residues (Ser51 and Leu80) have good electron density. No solvent correction has been applied.

Coordinates for the Mg^{2+} and Mn^{2+} I-domains have been deposited with the Brookhaven Protein Data Bank, and are available from the authors.

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