



Table 1  
Condensation of d(TGGCCAAGCTp-Im)

Reaction mixture no.	Components of buffer (M)					pH	T (°C)	Condensation products (% yield)		
	Na-phosphate	NaCl	MgCl <sub>2</sub>	HIm	MeIm			1 day	4 days	12 days
1	0.05	0.2	0.75	-	-	7.4	6	2-3	2-3	2-3
2	0.05	0.2	0.75	-	-	5.7	0	0	0	0
3 <sup>a</sup>	0.05	0.2	0.75	-	-	7.5	6	10	12	15
4	-	0.2	0.12	0.2	-	8.0	6	0	0	0
5	-	0.2	0.12	-	0.05	8.0	6	15	30	50
6	-	0.2	0.12	-	0.1	8.0	6	47	53	61
7	-	0.1	0.06	-	0.1	7.4	6	33	42	-
8	-	0.1	0.06	-	0.1	8.0	6	46	78	91
9	-	0.2	0.12	-	0.2	8.0	6	67	73	85
10	-	0.2	0.12	-	0.2	8.0	10	61	80	84
11	-	0.2	0.12	-	0.2	8.0	20	35	40	50
12	-	0.2	0.12	-	0.2	9.0	6	40	44	54
13 <sup>b</sup>	-	0.2	0.12	-	0.4	8.0	6	0	0	0
14	-	0.2	0.12	-	0.4	8.0	6	73	75	87
15	-	0.2	0.12	-	0.4	8.0	10	71	72	83
16 <sup>c</sup>	-	0.2	0.12	-	-	9.0	6	2-3	5	12

Reaction mixture contained additional components: <sup>a</sup> 0.001 M Py; <sup>b</sup> 30% (v/v) formamide; <sup>c</sup> 0.01 M DMAP

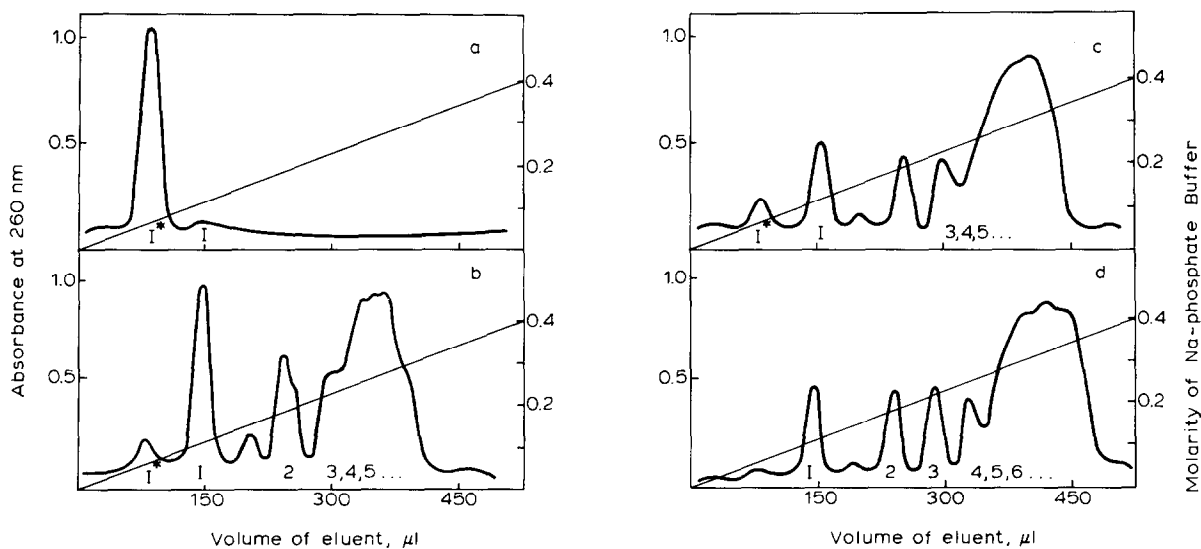


Fig. 1. Chromatography on Lichrosorb-NH<sub>2</sub> of reaction mixture no. 9 (table 1): before condensation (a); after 1 day (b); 4 days (c); 12 days (d) of incubation; 1 × 40 mm column, Na-phosphate gradient (pH 7.0) in 7 M urea, 20°C. Elution rate, 300 μl/h. Peak 1\*, d(TGGCCAAGCTp-Im); peaks 1, 2, ..., d(TGGCCAAGCTp)<sub>n</sub> where n corresponds to the number of the peak.

Yields of polycondensation products were determined as a ratio of the polycondensation products peak area to the total nucleotide peak area. The products were isolated by chromatography on Lichrosorb-HN<sub>2</sub> followed by gel-filtration on BioGel P-2. Digestion with VPDE and BAP was done as in [7], gel-electrophoresis as in [6,11]. The samples for electrophoresis contained 0.2–0.3  $A_{260}$  of the nucleotide material. The gel slabs were stained with 0.05% 'Stains-all' in 50% formamide. The temperature dependence of UV absorbance was recorded with a Cary-219 spectrophotometer in thermostatted quartz cuvettes (Helma, FRG) with 1 mm pathlength as in [8]. The nucleotide concentration/monomer was  $10^{-3}$  M.

### 3 RESULTS AND DISCUSSION

According to the optical data the conditions (table 1) chosen for polycondensation of (II) permit the formation of stable complementary duplex with  $T_m$  25–27°C and hypochromicity  $h = 20\%$ . Thus the covalent binding of the imidazolid moiety to oligonucleotide does not cause the destabilisation of duplex.

The reaction mixtures containing (II) were incubated in the conditions of table 1; the composition of the buffers, pH and temperature were varied. Analysis of the reaction mixtures showed that polycondensation of (II) into di-, tri- etc. -meric blocks is efficient only in buffers containing MeIm, the yield of products exceeding 80% (table 1: 8–10; 14; 15; fig. 1). Polycondensation of (II) in the presence of other tertiary amines is less efficient (table 1: 3,4,16); in the absence of the tertiary amines no reaction occurs (table 1: 1,2). Further investigation of the role of different tertiary amines in the polycondensation of (II) is in progress. The optimal pH of the reaction was found to be 8.0, the decrease of pH to 5.7–6.0 enhances the hydrolysis of imidazolid (II) to decanucleotide (I) (table 1: 1,2,7–9,12).

The stability of the duplex is an indispensable condition of effective polycondensation. It was proved by the decrease of yields of polymeric products at an increase from 6–20°C (table 1: 9–11,14,15). Under the conditions causing total denaturation of duplex (table 1: 13), only hydrolysis of d(TGGCCAAGCTp-Im) to d(TGGCCAAGCTp) occurs.

The condensation products were analyzed by gel-electrophoresis and enzymatic digestion. The slag-gel electrophoresis of total polymeric fraction from table 1: 8–12,14,15 showed that it consists of d(TGGCCAAGCTp)<sub>n</sub>,  $n = 2-7$  (fig. 2a). The polymers were identical to the products of carbodiimide-induced condensation of d(TGGCCAAGCTp) (fig. 2b–d) obtained as in [8]. Decanucleotide (I) formed as a result of hydrolysis of the phosphoimidazolid bond is a terminator for a growing polynucleotide chain. The degree of

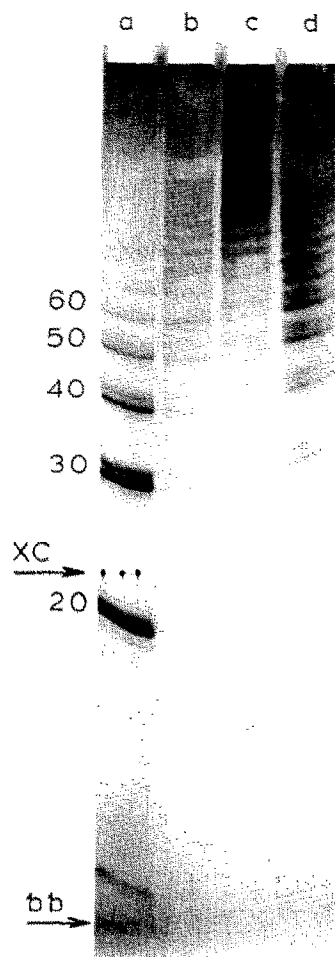
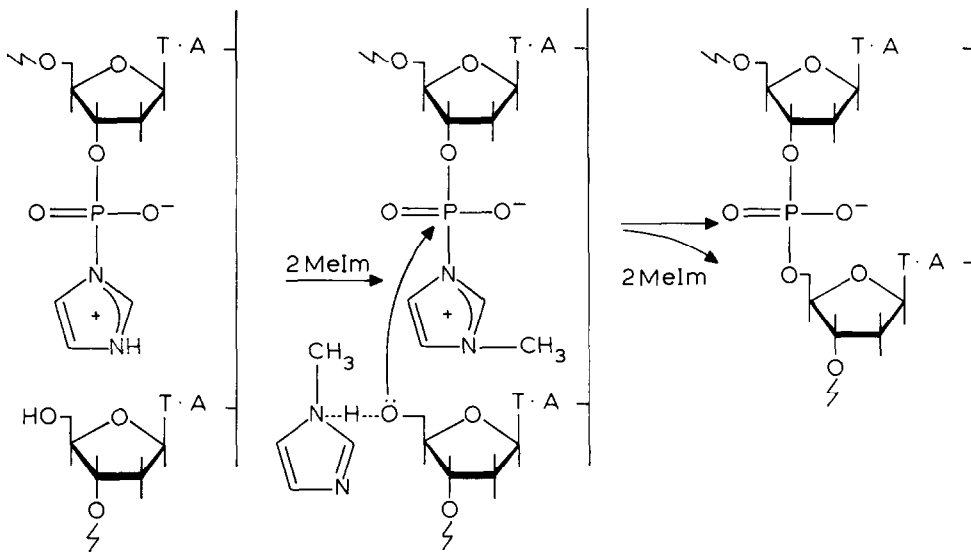


Fig. 2. Electrophoresis in 20% polyacrylamide gel of reaction mixtures 5–12,14,15 (table 1) (track a); products of 1-ethyl-3-(3'-dimethylaminopropyl)carbodiimide-induced condensation of d(TGGCCAAGCTp), (tracks b–d). The numbers on the left indicate monomeric units in the oligonucleotide, XC, bb, position of marker dyes xylencyanol and bromphenol blue, respectively.

polymerization may be increased up to 10-12 by synthesizing imidazolidine derivatives of  $d(\text{TGGCC-AAGCTp})_n$  ( $n = 2-7$ ) obtained previously and submitting them to further condensation under experimental conditions (table 1: no. 10). The degradation of the products of polycondensation (table 1: 5-12,14,16) with VPDE and BAP gave only a mixture of nucleosides, thus confirming that the material formed is (3'-5')-linked.

The mechanism of the formation of interoligomeric bond is not yet known, but we

should like to emphasize that the rate of condensation of (II) is greatly enhanced by MeIm (table 1: 5,6,9,14) and is not sufficiently influenced by other tertiary amines. These results as well as some experimental data obtained in our laboratory enable us to suggest that MeIm functions as a nucleophilic [13] and at the same time general basic catalysts. On one hand MeIm transaminates the protonated form of phosphorimidazolidine (II), on the other it acts as an acceptor of proton from the neighbouring 5'-OH group:



This mechanism can operate only under the conditions that permit the formation of stable duplex and therefore provide a steric neighbourhood of reacting groups. In the case of concatameric duplex studied here the mechanism works rather well.

We have thus achieved an effective polycondensation of synthetic decanucleotide by means of imidazolidine activation of its 3'-terminal phosphate. This approach of chemical condensation of oligonucleotides may have several advantages over the enzymatic ligation:

- (i) It may be applied to a large scale synthesis of DNA-duplexes in amounts sufficient for instrumental study (e.g., X-ray, NMR) of DNA;
- (ii) The chemical method may be used in synthesis of double-stranded polynucleotides with unnatural sugar-phosphate backbones as well as modified heterocyclic bases.

#### ACKNOWLEDGEMENTS

We thank Dr L.E. Orgel (The Salk Institute for Biological Studies, San-Diego CA) for attention and information which was of great value to us.

#### REFERENCES

- [1] Naylor, R. and Gilham, T. (1966) *Biochemistry* 5, 2722-2728.
- [2] Uesugi, S. and Ts'o, P.O.P. (1974) *Biochemistry* 13, 3142-3152.
- [3] Badashkeeva, A.G., Kabasheva, G.N., Knorre, D.G., Shamovsky, G.G. and Shubina, T.N. (1972) *Dokl. Akad. Nauk. SSSR* 206, 870-873.
- [4] Bridson, P.K., Fakhrail, H., Lohrmann, R., Orgel, L.E. and Van Roode, M. (1981) in: *Origin of Life* (Wolman, Y. ed) pp. 233-239, Reidel, New York.
- [5] Shabarova, Z.A. and Prokofiev, M.A. (1970) *FEBS Lett.* 11, 237-240.

- [6] Shabarova, Z.A., Dolinnaya, N.G., Turkin, S.I. and Gromova, E.S. (1980) *Nucleic Acids Res.* 8, 2413-2429.
- [7] Shabarova, Z.A., Dolinnaya, N.G., Drutsa, V.L., Melnikova, N.P. and Purmal, A.A. (1981) *Nucleic Acids Res.* 9, 5747-5761.
- [8] Schneider, H., Lohrmann, R., Orgel, L.E., Sulston, I. and Weimann, B.J. (1968) *Science* 162, 809-810.
- [9] Jencks, W.P. (1969) *Catalysis in chemistry and enzymology*, McGraw-Hill, New York.
- [10] Juodka, B.A. (1980) *Bioorgan. Khim.* 6, 1445-1465.
- [11] Shabarova, Z.A., Volkov, E.I., Oretskaya, T.S., Turkin, S.I., Dolinnaya, N.G., Kagramanova, V.K. and Prokofiev, M.A. (1981) *Dokl. Akad. Nauk. SSSR* 258, 914-917.
- [12] Ivanovskaya, M.G., Gottikh, M.B. and Shabarova, Z.A. (1982) *Bioorgan. Khim.* 8, 940-944.
- [13] Orgel, L.E. and Lohrmann, R. (1974) *Acc. Chem. Res.* 7, 368-377.