

# Preferential binding of $\alpha$ -actinin to actin bundles

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At 37°C, the  $\alpha$ -actinin-F-actin binding isotherm is anomalous. In 6.7% polyethylene glycol 6000, concomitantly with the formation of actin bundles, the binding isotherm becomes hyperbolic ( $K_{\text{diss}} = 11.3 \mu\text{M}$ ).  $\alpha$ -Actinin increases the rigidity of the networks formed by actin bundles in polyethylene glycol and by paracrystalline actin in 16 mM  $\text{MgCl}_2$  but not by F-actin. It is proposed that in the cell  $\alpha$ -actinin functions are mostly carried on by interaction with actin bundles.

$\alpha$ -Actinin function; Preferential binding; Actin bundle

## 1. INTRODUCTION

$\alpha$ -Actinin is a 200 kDa protein that crosslinks actin filaments and increases the rigidity of the actin gel. Both phenomena are quite complex and are not completely understood.

The binding isotherm of  $\alpha$ -actinin with actin, as a function of actin concentration, is anomalous. The apparent association constant decreases with an increase in actin concentration [1,2]. This phenomenon was not recognized previously because the effect of actin concentration either was not tested [3–7] or was tested at relatively high actin concentrations [8].

The critical gelling concentration of  $\alpha$ -actinin increases substantially with temperature [3], a phenomenon that led to questioning of the gelling activity of this protein *in vivo* [5]. It was found, however, that  $\alpha$ -actinin is an efficient actin gelling protein, even at 37°C, provided that either the concentration of actin is low (1.2–2.4  $\mu\text{M}$ ) [2] or the reaction mixture is supplemented with macromolecules at a concentration equivalent to that found in the cell sap [9].

We offer evidence here that the presence of a network of actin bundles, independent of the mechanism of its formation, is a prerequisite for  $\alpha$ -actinin functioning *in vivo*.

## 2. MATERIALS AND METHODS

G-actin from rabbit muscle was prepared according to Spudich and Watt [10] and further gel filtered through Sephadex G-150 [11].  $\alpha$ -Actinin from chicken gizzard was prepared according to Feramisco and Burridge [12]. The absorption coefficients used were  $A_{280}^{1\%} = 6.2$  [13] for actin and  $A_{280}^{1\%} = 9.7$  [14] for  $\alpha$ -actinin. Molar concentrations were

calculated on the basis of a molecular mass of 42 kDa for actin [15] and of 200 kDa for  $\alpha$ -actinin [14]. Centrifugation was performed at 37°C in a TL100 rotor of the TL100 Beckman centrifuge.

Protein was determined by the Coomassie blue method [16] as modified by Stoscheck [17].

[ $^3\text{H}$ ]-*N*-Ethylmaleimide-labelled  $\alpha$ -actinin was prepared and radioactivity determined as previously described [2].

The rigidity of the gels of actin was measured by the droplets method [18].

## 3. RESULTS

### 3.1. Effect of polyethylene glycol 6000 on the complex interactions of $\alpha$ -actinin with actin

It is known that, in the presence of polyethylene glycol (PEG) 6000, F-actin undergoes massive conversion into actin bundles [19]. At 12  $\mu\text{M}$  actin the boundary between filaments and bundles ranges between 6 and 7% (w/v) PEG. Addition of 0.2  $\mu\text{M}$   $\alpha$ -actinin to the system displaces the boundary toward a lower (4–6%) PEG concentration. This shows that  $\alpha$ -actinin favours actin bundling (Fig. 1).

The amount of  $\alpha$ -actinin co-sedimenting with F-actin is not influenced up to 3% PEG but increases at larger PEG concentrations. The increase is concomitant with the formation of actin bundles. Under these conditions (6.7% PEG), approximately the same amount of  $\alpha$ -actinin is sedimented by centrifugation either at  $9,900 \times g$  (actin bundles are collected) or at  $366,000 \times g$  (actin bundles plus actin filaments are collected). Thus, at this PEG concentration,  $\alpha$ -actinin is bound almost exclusively to actin bundles (Fig. 2).

The  $\alpha$ -actinin-F-actin binding isotherm is anomalous both in the absence [1,2] and in the presence of 3% PEG (Fig. 3). Under both these conditions actin is filamentous. The binding isotherm becomes hyperbolic concomitant with the formation of actin bundles (6.7% PEG). Double reciprocal plot analysis shows that a sin-

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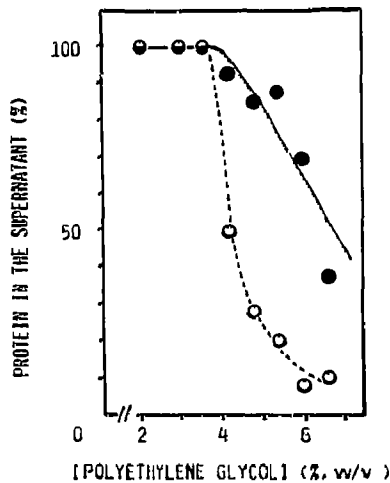


Fig. 1. In polyethylene glycol 6000  $\alpha$ -actinin promotes actin bundling. The mixtures contained F-actin ( $12 \mu\text{M}$  as monomer),  $0.5 \text{ mM}$  ATP,  $2 \text{ mM}$   $\text{MgCl}_2$ ,  $0.1 \text{ M}$  KCl,  $1 \text{ mM}$  dithiothreitol,  $10 \text{ mM}$  Tris-HCl, PEG 6000 (w/v) as indicated in the figure, with (○) or without (●)  $0.2 \mu\text{M}$  [ $^3\text{H}$ ]N-ethylmaleimide-labelled  $\alpha$ -actinin (specific activity  $12,000 \text{ dpm/nmol}$ ), pH 7.5. After 60 min of incubation at  $37^\circ\text{C}$ , the mixtures were centrifuged for 10 min at  $9,900 \times g$  to sediment actin bundles, and the supernatant solutions were assayed for protein.

gle dissociation constant ( $K_{\text{diss}} = 11.3 \mu\text{M}$ ) accounts for the binding of  $0.185 \mu\text{M}$  out of the total  $0.2 \mu\text{M}$   $\alpha$ -actinin (Fig. 3b).

3.2. The interaction of  $\alpha$ -actinin with paracrystalline actin in  $16 \text{ mM}$   $\text{MgCl}_2$

In the presence of  $0.2 \mu\text{M}$   $\alpha$ -actinin (total concentration) and of  $48 \mu\text{M}$  actin, more  $\alpha$ -actinin is bound to paracrystalline actin in  $16 \text{ mM}$   $\text{MgCl}_2$  ( $0.091 \mu\text{M}$ , Fig. 4) than to filamentous actin in  $2 \text{ mM}$   $\text{MgCl}_2$  plus  $0.1 \text{ M}$  KCl ( $0.026 \mu\text{M}$ , Fig. 3). The binding of  $\alpha$ -actinin to

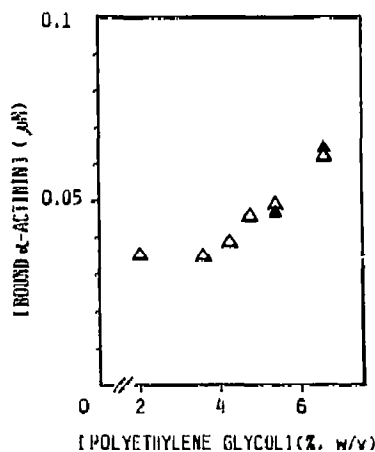


Fig. 2. Binding of  $\alpha$ -actinin to F-actin in polyethylene glycol 6000. Experimental conditions were as described in Fig. 1. After 60 min of incubation at  $37^\circ\text{C}$ , the mixtures were centrifuged for 10 min either at  $9,900 \times g$  ( $\blacktriangle$ ) or at  $366,000 \times g$  ( $\triangle$ ). The pellets were then assayed for labelled  $\alpha$ -actinin.

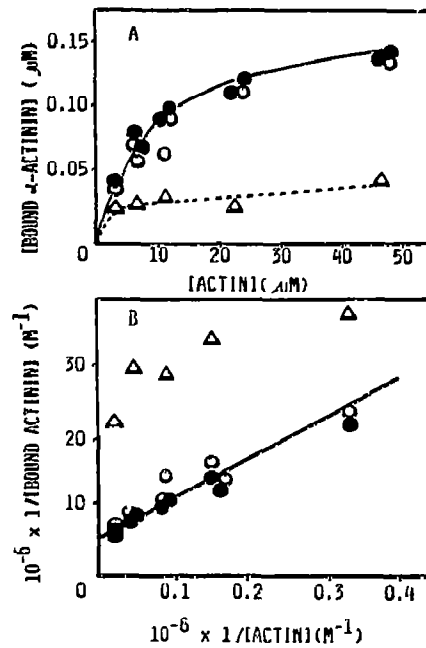


Fig. 3. Binding of  $\alpha$ -actinin to F-actin in polyethylene glycol 6000, as a function of actin concentration. (a) Actin concentration was as indicated in the figure; PEG was either 3% ( $\triangle$ ) or 6.7% ( $\circ$ ,  $\bullet$ ). Other conditions were as described in Fig. 1. After 60 min of incubation at  $37^\circ\text{C}$ , the mixtures were centrifuged for 10 min either at  $9,900 \times g$  ( $\bullet$ ) or at  $366,000 \times g$  ( $\circ$ ,  $\triangle$ ). The pellets were then assayed for labelled  $\alpha$ -actinin. (b) Double reciprocal plot of the binding of  $\alpha$ -actinin to F-actin.

paracrystalline actin, however, is not described by a simple hyperbolic function (Fig. 4).

The rigidity of the system ( $7.1 \mu\text{M}$  actin as monomer) increases from  $2.6$  to  $6.7 \text{ dyn/cm}^2$  in the transition from

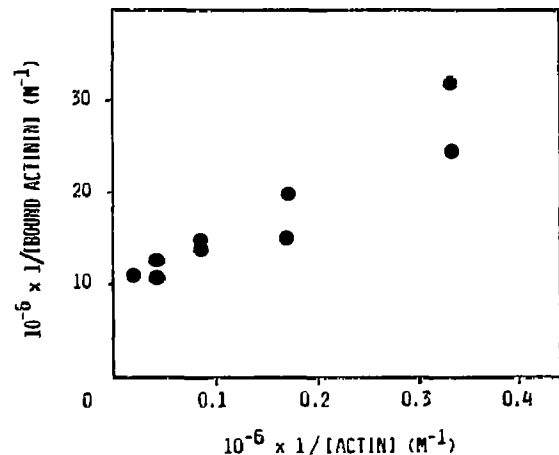


Fig. 4. Double reciprocal plot of the binding of  $\alpha$ -actinin to actin paracrystals in  $16 \text{ mM}$   $\text{MgCl}_2$ . The mixtures contained F-actin as indicated in the figure,  $0.2 \mu\text{M}$  [ $^3\text{H}$ ]N-ethylmaleimide-labelled  $\alpha$ -actinin (specific activity  $12,000 \text{ dpm/nmol}$ ),  $0.5 \text{ mM}$  ATP,  $16 \text{ mM}$   $\text{MgCl}_2$ ,  $1 \text{ mM}$  dithiothreitol,  $10 \text{ mM}$  Tris-HCl, pH 7.5. After 60 min of incubation at  $37^\circ\text{C}$ , the mixtures were centrifuged for 10 min at  $366,000 \times g$ . The pellets were then assayed for labelled  $\alpha$ -actinin.

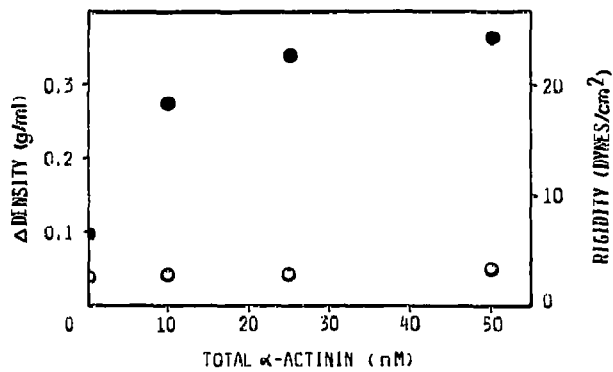


Fig. 5. Estimate of the effect of increasing  $\alpha$ -actinin concentrations on the rigidity of the network formed by paracrystalline actin. The mixtures contained F-actin ( $7.1 \mu\text{M}$  as monomer),  $0.5 \text{ mM}$  ATP,  $1 \text{ mM}$  dithiothreitol,  $10 \text{ mM}$  Tris-HCl and either  $16 \text{ mM}$   $\text{MgCl}_2$  (●) or  $2 \text{ mM}$   $\text{MgCl}_2$  plus  $0.1 \text{ M}$  KCl (○).  $\alpha$ -Actinin concentration was as indicated in the figure. After 60 min of incubation at  $37^\circ\text{C}$  and  $\text{pH}$  7.5, rigidity measurements were performed.  $\Delta$  density represents the difference between the density at which the droplets remained stationary in the complete system and in the salt solution without protein.

F-actin to paracrystalline actin. The rigidity is further increased to  $24.3 \text{ dyn/cm}^2$  when paracrystalline actin is supplemented with  $0.05 \mu\text{M}$   $\alpha$ -actinin (Fig. 5).

#### 4. DISCUSSION

In 6.7% PEG 6000, the binding of  $\alpha$ -actinin to actin bundles is described by a single dissociation constant of  $11.3 \mu\text{M}$ . This contrasts with the anomalous behaviour displayed by F-actin and characterized by the apparent decrease of the binding constant to  $\alpha$ -actinin, as a function of the increase of F-actin concentration.

The parallel arrays of actin filaments, formed either in 6.7% PEG 6000 or in  $16 \text{ mM}$   $\text{MgCl}_2$ , bind  $\alpha$ -actinin tighter than does F-actin. This is in keeping with the observation that, in the cell,  $\alpha$ -actinin is mostly associated with actin fibers [20,21]. It is likely that the arrays of filaments offer an ordered matrix of actin, which favours by bidentate binding of  $\alpha$ -actinin. The crosslinking by  $\alpha$ -actinin prevents the filaments from sliding in actin bundles. As a result, since the network of actin bundles is largely anastomosed, the rigidity of

the system is increased by  $\alpha$ -actinin, even at  $37^\circ\text{C}$ . In F-actin, at least at  $37^\circ\text{C}$ , the monodentate binding of  $\alpha$ -actinin prevails. This is indicated by the total lack of effect of  $\alpha$ -actinin on the rigidity of the network formed by F-actin.

These observations support the view that, in the cell,  $\alpha$ -actinin functions are mostly carried on by interaction with actin bundles.

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