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A MALST method comparison over univariate kinetic modeling for determination of shelf-life in cereal snack of dried apples

Andres Córdova^a, Cezar Quezada^a, Jorge Saavedra^{a,b,*}^a *Research Group on Chemometrics, Department of Food Engineering, Pontificia Universidad Católica de Valparaíso, Waddington 716, Valparaíso 2360021, Chile.*^b *Centro Regional de Estudios en Alimentos Saludables (CREAS), Blanco 1623 of 1402 Edificio Torres Mar del Sur II, Valparaíso 2362396, Chile*

Abstract

A Multivariate Accelerated Shelf-Life Testing (MALST) study in cereal of dried apple snack incubated at 18°C, 25°C and 35°C. Quality attributes were analyzed until 18 months. The data obtained were used to modeling univariate and multivariate kinetic, through Principal Component Analysis PCA. PCA could explain the interaction between attributes; which would imply that model reflected in terms of variability, biochemical phenomena associated with the deterioration of the product. There was a good relationship between both models in the Shelf-life estimated with the comparative advantage that MALST method could model simultaneously the complete and “all at once” phenomenon of deterioration.

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1. Introduction

Nowadays consumers demand products with superior appearance, texture, taste and flavour whilst keeping their nutritional value [1, 8]. For this purpose, the formal determination of shelf-life is a key factor in the research and development of food since provides information regarding the time that the product aptly retains its attributes [1-5]. This determination is usually performed by measurement of quality attributes [6], alternating accelerated ageing methods under extreme conditions [7]. The most usual model to this purpose is Arrhenius equation, since it relates temperature with the reaction velocity

* Corresponding author. Tel.: +0-056-032-2274219; fax: +0-056-032-2274205.

E-mail address: jorge.saavedra@ucv.cl.

(k) [2]. However, the shelf-life determination becomes complex when multiple attributes must be studied simultaneously in a food system, since each one has an own specification related with the date of expiration. For these cases, the use of multivariate statistical techniques could be more suitable [8]. In this sense, accelerated shelf life testing methods have incorporated multivariate tools, called Multivariate Accelerated Shelf Life Testing (MALST).

In this context, the objective of this research was comparatively assessing shelf-life of an absolutely new product in market which is a snack-cereal of dried apple type, through traditional univariate method and multivariate accelerated method.

Nomenclature

Aw	Activity water
Humidity	% Moisture
PCA	Principal Component Analysis
k^m	Multivariate reaction rate (Scores \cdot h $^{-1}$)
Ea^m	Multivariate Energy of activation (cal/mol)
$\alpha_{T+\delta T, T}^m$	Acceleration factor
Tc	Cut off criteria

2. Materials and Methods

Samples of a new product in the market of dried Apple snack type from an exporting agribusiness (Maule, Chile) packed in multi-laminated bags were used. They were incubated at 18 °C, 25 °C and 35 °C. Considered quality attributes were Aw (Aqua LAB at 20 °C \pm 0.3 °C), moisture (AOAC 2000-934 06), content of SO₂ (method Monier Williams), taste, color, aroma, sensory texture and color CIE-Lab (colorimeter Minolta CR-200b). Analyses were conducted over 18 months of incubation.

Univariate analysis: in this stage, univariate degradation kinetics were studied according methodologies proposed by [2, 3].

Multivariate analysis: data of quality attributes were modeled using the method proposed by [8]. This method is based on application of Principal Component Analysis (PCA) based NIPALS algorithm, modified by [9]. PCA was performed on quality attributes matrix for three storage conditions of temperature, jointly and separately. With these values, multivariate kinetic were adjusted multivariate, for each of the multivariate properties. All calculations and adjustments were made with SIMCA-P+ 12 (Umetrics, Umeå, 2009), Sigmaplot 11 (Systat Inc., 2009) and Excel 2003 (Microsoft, 2003).

3. Results and Discussion

Scores plot (Fig. 1) from PCA shows the evolution of the attributes used at three temperatures. PCA is an unsupervised exploratory tool to visualize the main variations between samples, sample clustering, and the relation between the samples and variables. PCA is a technique that, by the reduction of the data dimensionality, allows their visualization retaining as much as possible the information present in the original data. So, PCA transforms the original measured variables into new uncorrelated variables called principal components. Each principal component is a linear combination of the original measured variables. Thus, the behavior pattern underlying the studied variables could be explored.

calculation was performed with the first PC related with the time, to obtain multivariate kinetic associated with 3 conditions storage (Fig.3).

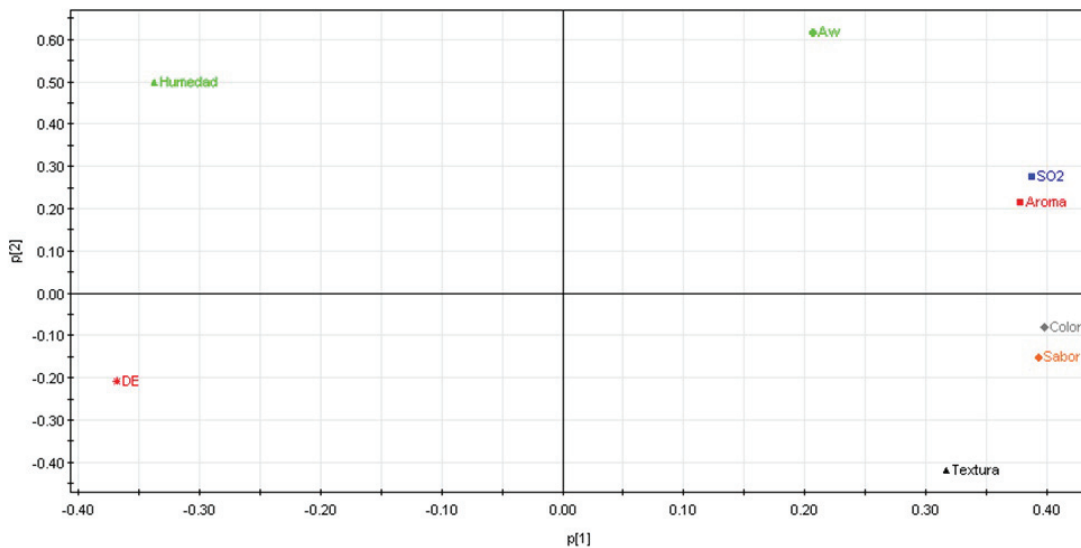


Fig. 2. Loadings Plot

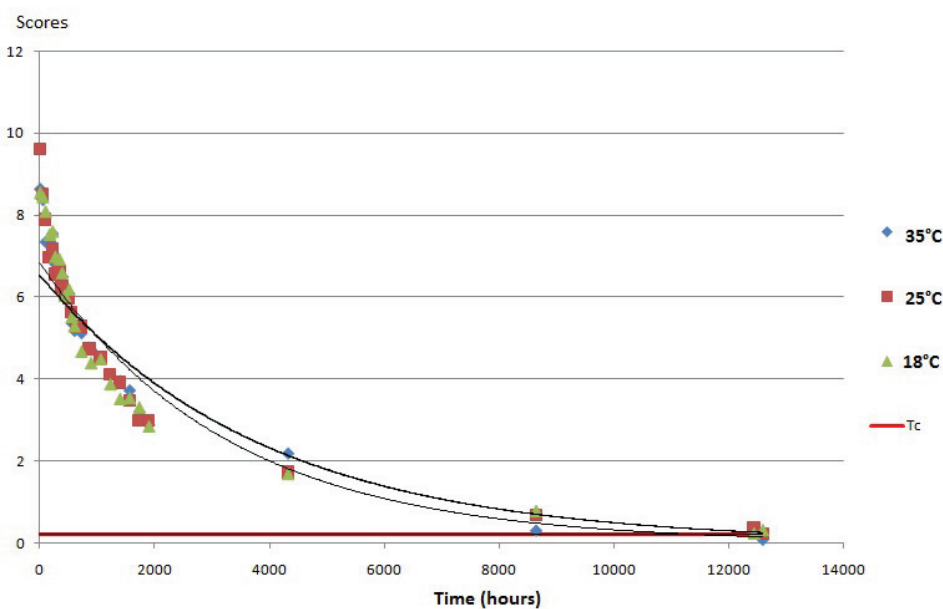



Fig. 3. Scores vs. Time plot for 3 Storage conditions (Multivariate Kinetic)

Fig. 3 shows that degradation product reflects 1st order kinetic for all storage conditions. From this graph, degradation curves were adjusted and multivariate kinetic parameters were calculated k^m (reaction rate), Ea^m (Activation energy expressed in cal/mol) and $a_T^m + \delta_{T,T}$ (acceleration factor) (Table 1). The stability

of the product is inferred, due to the minimal increase in the rate of reaction between incubation temperatures.

From the specification values of the product (not shown values) and the Loadings matrix was obtained the Cut off criteria (T_c in figure 3), which in this case was 0.2183. It was determined that shelf-life was 18.2 months for storage to 18 ° C, 18.1 months at 25 ° C and 15.5 months at 35 °C. By comparing this value with the obtained by univariate kinetics, where the degradation only contemplates humidity as attribute, shelf-life of the product was 17.4 months, existing a difference of a few weeks in estimating one method and other.

Table 1. Multivariate kinetic parameters for three storage temperatures

Storage T°	Rate constant k^m (PC 1 score hours ⁻¹)	R ² adjust		Ea^m (cal/mol)
18°C	0,00026	0,96		1563,53
25°C	0,00026	0,96	$\alpha_{25,18} = 1$	
35°C	0,00030	0,95	$\alpha_{35,18} = 1,15$	

Indeed, despite the partial increase observed in the humidity in the three storage temperatures, there was not a relationship with the attribute A_w , as had been hoped. This suggests that water migrates to inside the product through packaging; water settles so in the product, which interacts with the different ingredients, whereas the product has a high content of sugars and solutes that preclude the availability of water for degradation reactions.

Due to little difference of shelf-life at 18°C and 35°C, it could be inferred the high stability of the product and a seeming “semi-independence” of the phenomena from the temperature. In the other hand, the “self-weighting” of attribute variability approach of MALST method could be considered like an important feature at the moment of take decisions about the cut-off criteria. In this meaning, the shelf-life assessment by MALST method appears advantageous to reduce the bias of select arbitrarily the quality attributes to shelf-life determination.

4. Conclusion

MALST methodology could estimate simultaneously deterioration of quality attributes of the product, showing all the interactions occurring at the product concurrently

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