

Membrane transporter engineering in industrial biotechnology and whole cell biocatalysis

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Because they mainly do not involve chemical changes, membrane transporters have been a Cinderella subject in the biotechnology of small molecule production, but this is a serious oversight. Influx transporters contribute significantly to the flux towards product, and efflux transporters ensure the accumulation of product in the much greater extracellular space of fermentors. Programmes for improving biotechnological processes might therefore give greater consideration to transporters than may have been commonplace. Strategies for identifying important transporters include expression profiling, genome-wide knockout studies, stress-based selection, and the use of inhibitors. In addition, modern methods of directed evolution and synthetic biology, especially those effecting changes in energy coupling, offer huge opportunities for increasing the flux towards extracellular product formation by transporter engineering.

The control of flux in biochemical networks

In any complex biochemical network, all steps contribute to the control of the flux through a particular pathway or even that catalysed by a specific enzyme, but some steps exert a greater degree of control on the fluxes of interest than do others. Although all steps contribute to flux control, in devising strategies to increase such fluxes [1], it is wise to pay special attention to the particular steps that exert the greatest control over the pathway. In many cases, and especially for those systems involving xenobiotics, these steps include the cellular transporters that catalyse the influx of substrates and the efflux of products or potentially cytotoxic compounds. In other words, these steps are typically significantly rate-controlling. A quantitative measure of the extent of this rate or flux control is encapsulated in the

flux-control coefficient (Box 1). With a relative density barely greater than 1, even at 100 g.l⁻¹ wet cell concentration most of the volume of a fermentor is extracellular; thus, to maximise volumetric productivity it will be desirable to ensure that cells excrete the products of interest [2].

But doesn't stuff just diffuse into and out of cells unaided?

There is a surprisingly widespread view in the pharmaceutical industry, starting with the relevant textbooks [3], that the main means by which most xenobiotics (e.g., drugs) enter and exit cells is simply by diffusing passively across the lipid bilayer portion of cell membranes down their concentration gradients and according to their lipophilicity (log P or log D; see Glossary). If this were the case we should have little to say in this review, but it is not [4–14]. The main means by which small molecules cross biological cell membranes is through genetically encoded, proteinaceous transporter molecules, and this gives the cells, and the biotechnologist, important means by which to control and influence the process. The first thing to know, then, is qualitative [15,16]: which small molecules use which transporters? While we shall mainly consider microbes and fermentations, the principles we enunciate are general, and we recognise their role in the metabolic engineering of plants where especially vacuolar, peroxisomal, chloroplast, and root transporters can exert significant flux control.

'Unexpected' transporters for the uptake of uncharged nonelectrolyte nutrients and other small molecules

Based on the 19th-century studies of Overton, who showed a close correlation between the logarithm of the rate of

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Glossary

The logarithm of the distribution coefficient (log D): D is the ratio of the sum of the concentrations of all forms of a compound (ionised plus non-ionised) in each of two phases, typically 1-octanol and an equilibrated aqueous buffer, whose pH must be specified.

The logarithm of the partition coefficient (log P): P is a measure of the hydrophobicity of a molecule; log P is the logarithm (base 10) of the ratio of the concentration of a solute molecule in an organic solvent, usually 1-octanol, to that of the non-ionised form of the same molecule in water.

Box 1. Metabolic control and flux balance analysis

Imagine a metabolic network or pathway in which we vary the concentration of an enzyme E by an amount ΔE , with a concomitant change in the flux of interest ΔJ . In the limit of small changes, this becomes dJ/dE . By normalising these changes to the flux and enzyme concentration at the operating point (J,E) we can obtain a dimensionless quantity C_E^J , the flux-control coefficient of enzyme E on the flux of interest J, which describes in quantitative terms the extent to which that enzyme controls the flux. C_E^J is equivalent to a local sensitivity coefficient. If C_E^J is 0 then the enzyme exerts no flux control, while if it is 1 then it is completely flux-controlling. The sum of the flux-control coefficients for all enzymes on a particular flux is 1. This means that most enzymes have small flux-control coefficients, and even a 50% knockdown typically has a limited effect on flux (Figure 1). Thus, to have major effects, one should seek to use haploid organisms or homozygous diploid deletants [9,66]. A related concentration-control summation theorem shows that the sum of the concentration control coefficients = 0. Note that the flux-control coefficient is not constant – at a different operating point it would be higher or lower as flux control shifts among different parts of the network. The thesis in this review is that, where the flux-control coefficients of transporters are determined, they will often be found to be larger than those of other enzymes, providing suitable suggestions for transporter engineering.

Flux balance analysis describes a series of techniques for estimating relative metabolic fluxes without the requirement to know any of the kinetics of the participating enzymes. All it requires is knowledge of the stoichiometries of the participating reactions, the molecular identities of the reactants and products themselves, and an objective function that one is trying to optimise. Linear programming techniques can then be used to optimise the latter. The stoichiometries, including mass, charge, and energy balances, provide a very

effective series of constraints to determine the possible fluxes; however, the objective function is more problematic. Typically, biomass is used (i.e., the rate of biomass formation), with biomass being encoded as a 'molecule' with a non-integer empirical formula. It is desirable to add further constraints, for example by confining specific fluxes to restricted ranges or finding flux distributions that best correlate with expression profiles [63].

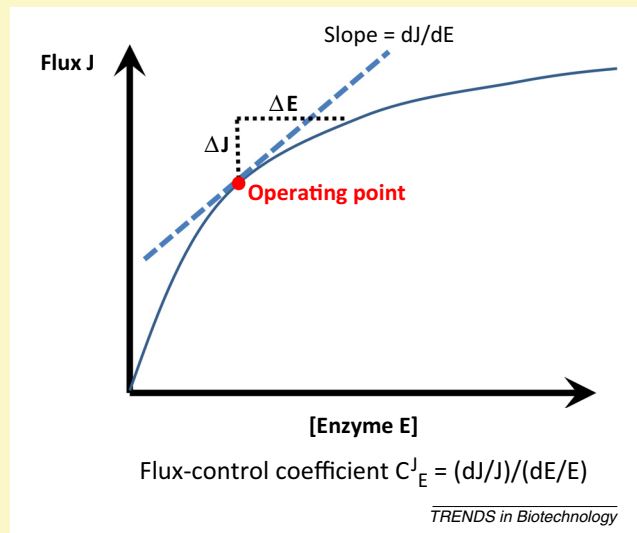


Figure 1. Flux as a function of enzyme concentration.

cellular uptake of non-electrolytes and their log P values, it had been widely assumed that small, uncharged molecules could permeate freely across the bilayer portion of biological membranes (even though it is well known that glucose and other sugars do not). However, it is now recognised that this is not at all the case, with transporters having been found (and required) for the uptake of many small, uncharged substances such as alkanes [17], ammonia (NH₃) [18], carbon dioxide (CO₂) [19,20], ethanolamine [21], fatty acids [22], glycerol [23], hydrogen peroxide (H₂O₂) [24], hydroxyurea [25], nitric oxide [26], (di)oxygen [27], urea [28,29], and even water [23,30]. The last was a finding for which Peter Agre received the 2003 Nobel Prize [31,32].

It has long been known that acetate enters cells mainly in its uncharged form, as acetic acid. This may be determined by osmotic swelling experiments [33], but these kinds of experiments do not say anything about the mechanism by which it enters: bilayer diffusion or a transporter. However, it is now known in the important amino acid producer *Corynebacterium glutamicum* that even the uptake of electroneutral acetic acid involves the use of a specific transporter [34].

Ethanol is another small nonelectrolyte of much biotechnological interest, and it is desirable to increase its export from producer cells [35]. It is not yet entirely certain which transporters are responsible for this, but the ATP-binding cassette (ABC) transporter [36] Pdr18 [37] and the glyceroaquaporin Fps1 [38] possess properties that might be consistent with such a role, although other mechanisms may also be involved [39]. While we later discuss in more detail export (efflux) transporters of molecules

not normally produced by the host, this section leads naturally to a discussion of those that are known to be involved in the secretion of metabolites that the host naturally produces.

Some useful case histories from classical fermentations

A notable example of the role of transporters in improving the yield of an important fermentation product (more than 2M tonnes per annum) comes from the history of the glutamate fermentation carried out using various coryneform bacteria, notably *C. glutamicum* [40]. Following the initial discovery of the fermentative production of glutamate, various empirical treatments in the 1960s and 1970s were found to enhance the efflux of glutamate from producer strains. It was later established that this was due to a change in membrane tension that activated a mechanosensitive glutamate efflux pump encoded by a gene called *NCgl1221*, a homologue of the *Escherichia coli* *yggB* gene, now known as *mscS*, the mechanosensitive channel of small conductance. Similar efflux pumps are now known to be involved in the export of products during a variety of other amino acid fermentations [41], such as those for lysine, isoleucine, threonine, methionine, and others [2].

Why would a cell export its metabolites?

One may wonder (from an evolutionary perspective) why bacteria see fit to excrete important nutrients or metabolites, often at fast rates. The most persuasive general explanation [42] is to the effect that soil bacteria (such as corynebacteria) that have experienced drought and are hit by a raindrop (a common stress), experience truly massive osmotic stresses or turgor pressures that can only

realistically be dealt with by a virtually instantaneous excretion of internal osmolytes. This excretion is catalysed by a mechanically-sensitive, membrane-triggered osmoregulatory process which can also occur in plants [43]. Such a role for the glutamate exporter, and one may suppose for other such exporters, is consistent with the similar role of its *E. coli* homologue [44]. Indeed, since their initial discovery in bacteria, a considerable number of such mechanosensitive, turgor-regulating exporters are now known [45], including seven (in two families) in *E. coli*.

Citric acid production

In a similar vein, the large-scale (well over 1M tonnes per annum) fermentative production of the tricarboxylate citric acid by the fungus *Aspergillus niger* involves active export of the product from the producer strain using a proton-symporting transporter [46]. Dicarboxylate titre is also improved by enhancing dicarboxylate efflux transporters [47].

Biomass production

In some fermentations the biomass itself is the product, and it is of interest to know what role transporters may play in controlling growth rate more generally. By using a pHauxostat to select strains of the fast-growing yeast *Kluyveromyces marxianus* for even faster growth [48], it was possible to evolve one that could grow up to 30% faster than the starting strain. This decrease in doubling time to 52 min, apparently the fastest reported for a eukaryote, was accompanied by an increase in surface area of some 40% at essentially constant volume. These results imply that membrane processes, such as substrate uptake, were most limiting to growth rate. Indeed, 80% of the growth rate increase was ascribed to membrane processes [48]. Continuous selection is also an excellent strategy for selecting strains resistant to toxins such as solvents [49,50], especially in turbidostats [51] in which growth rate can be measured online [52].

In this context is noteworthy that a high-throughput screen [53] of heterozygous deletants of diploid *S. cerevisiae* identified 145 transporter-encoding genes that exerted significant control over growth rate (so-called high-flux-control or HFC genes) in turbidostat culture. Ninety of these genes had a haploinsufficient (HI) phenotype – that is, they reduced the maximum growth rate of yeast when present in only one copy in a diploid – while the remainder had a haploproficient (HP) phenotype, increasing the growth rate when in the heterozygous state. These HFC genes included those encoding plasma membrane transporters, but also genes specifying proteins involved in transporting ions and metabolites into subcellular organelles, especially the mitochondria and the vacuole. Amongst the HI genes were those encoding plasma membrane transporters of metals (particularly iron and zinc), organic acids (including amino acids), ammonium, phosphate, sulphate, vitamins, sugars (including glucose) and sugar alcohols, and also the aquaporin gene, *AQY1*. This group of HI genes also includes four that encode drug efflux pumps. Given the discussion of efflux transporters, above, it would seem sensible for biotechnologists and synthetic biologists to pay attention not only to transporters of important nutrients but also to those responsible for the

efflux of potentially toxic products of metabolism, such as ethanol and other biofuels.

Transcriptome-based strategies for determining transporter-mediated activities

Virtually since its inception, it has been clear that genome-wide expression profiling at the level of the transcriptome provides an excellent strategy for identifying which gene products may be pertinent for particular biological processes. This applies equally to the role of transporters in biotechnology. Thus, the availability of the *Penicillium chrysogenum* genome made it possible [54] to compare the expression profiles of low- and high-producing strains, finding a considerable enhancement in transporter expression in the high-producers, again implying strongly that enhanced transporter expression could drive increased fluxes. Although, in general terms, the expression of an individual gene does not necessarily correlate with the productivity of a fermentation, and certainly not over a wide range because of changes in the distribution of flux control (Box 1), genome-wide trawls relating expression to activity can be highly beneficial, especially for metabolic networks. This is because metabolic transformations are subject to strict stoichiometric controls (no ‘alchemy’ is allowed).

Flux balance analysis

While the counsel of perfection in genome-scale metabolic modelling includes mechanistic details of every enzymatic step, which can then be turned into an ordinary differential equation (ODE) model that may be used to model or predict all the fluxes and concentrations of interest, we very rarely have sufficient of the kinetic parameters to do this [55–58]. However, the stoichiometric constraints alluded to above mean that the methods of flux balance analysis [58–60] (Box 1) may be used to attempt to predict the fluxes of interest. As part of a strategy to minimise the number of possible flux patterns that can explain the observable data [61,62], we have found [63] that absolute transcriptomics provides a valuable surrogate for the flux through each step.

The distribution of expression levels for transporter and non-transporter genes were determined (but not shown) in a recent study [63] of a yeast strain growing at 85% of its maximum growth rate. The transporter nature of the genes was assessed (Figure 1) by the present version of the yeast metabolic network. As judged by their median levels, as well as by 5000 permutations, there is a significantly lower level of expression ($P < 0.0004$) of transporter genes (19.3 transcripts per cell) than of non-transporter genes (31.7). This is not inconsistent with the fact that cellular membranes, as 2D structures, possess a more limited amount of real estate for incorporating transporters and other membrane proteins than do the 3D intracellular spaces. The ‘surfaceome’, including SLCs, is also the most variable between different and differentiated cells [64].

Detecting relevant uptake transporter genes through genome-wide knockout analyses

Although individual genes were classically and typically discovered individually, it is now possible to extend the

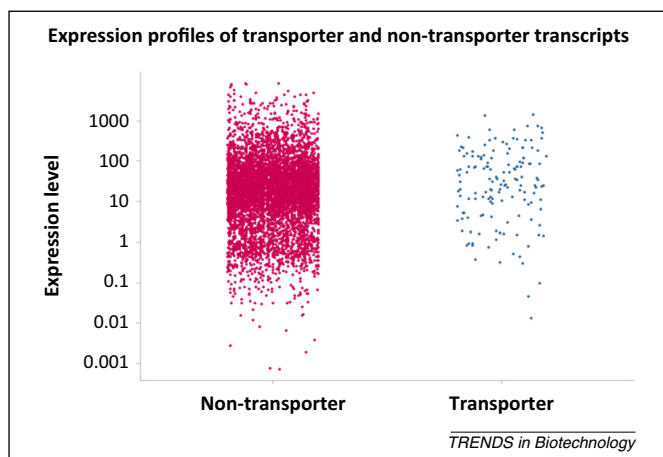


Figure 1. Expression profiles of 151 transporter and 6373 non-transporter transcripts in baker's yeast. Data from [63]. Note that fewer transport reactions in the model (327/1079, 30.3%) have associated genes (hence transcripts) than do all other metabolic reactions (1983/2255, 87.9%).

analysis of transporter roles to the genomic level (Box 2), and there are useful online databases focussing on transporters (Table 1). The sole requirements are for a suitable variation in the extent of expression of different enzymes in different strains, and this is most conveniently achieved using single gene knockouts combined with a means of selecting the phenotype of interest such as growth selection (Figure 2). Thus, in the case of *S. cerevisiae*, we were able to exploit the barcoded yeast deletion mutant collection to identify transporters for 18 out of 26 drugs tested [9]. Most had multiple transporters. For the eight where we could not detect which transporters were used, this is likely because there were simply too many transporters, and removing only one did not provide sufficient selectivity.

That study [9] used haploid strains (Box 1) and a purpose-designed microarray chip, but nowadays it is recognised that deep sequencing is much more effective and reliable [65]. Thus, in an exciting development, a near-haploid human cell line (KBM7) with a retroviral gene-trap was used to demonstrate that only a single transporter (called SLC35F2) is responsible for the uptake of the cytotoxic anticancer drug candidate sepantronium bromide (also known as YM155) into these cells [66]. Clearly, these kind of methods may be applied to any system for which cells that have or have not taken up a particular drug may be discriminated and separated (e.g., by cell sorting [67]), and then identified genetically. It is worth stressing that this kind of experiment would not 'work' – in other words, return any hits – if bilayer diffusion were the dominant mechanism of transmembrane transport. Put another way, it would indeed seem from such experiments that, for drug transport into cells, phospholipid bilayer diffusion is negligible [13,14].

In theory these kinds of knockout strategies could also be used to select strains with knockouts in efflux transporters (if such exist), via their greater sensitivity to a compound, but positive selection methods for resistance are always more reliable (Box 3 and Figure 2). While it has already been noted the deletion of only one of the two copies of a gene can be sufficient to produce a significant reduction in growth rate [53], it was also found that the removal of two genes,

Box 2. Genomics approaches to transporter identification

The modern approach to detecting transporters, especially for strain improvement based on systems biology principles (Figure 1), is through genome sequencing in which particular sequence motifs can more or less reliably identify transporters, even if not always their substrates. The next step is to incorporate such transporters into genome-scale metabolic network reconstructions [58,139]. While this is most effectively done by domain experts, recent advances in methods such as text mining for systems biology [140], and other strategies [141], mean that it is becoming increasingly amenable to automation. A list of 'predictive genome-scale metabolic network reconstructions' is maintained at <http://systemsbiology.ucsd.edu/InSilicoOrganisms/OtherOrganisms>. Significantly, almost all free-living organisms analysed are known to have genes encoding hundreds of transporters [8]. Indeed, approximately one third of the reactions in the heavily curated yeast [142] and human [143,144] metabolic networks are represented by transporter reactions.

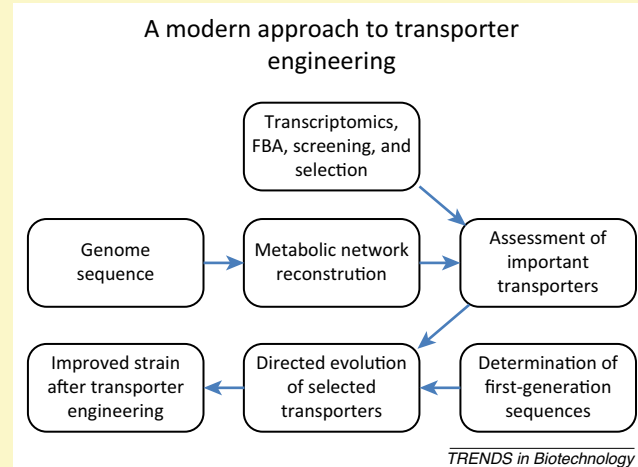


Figure 1. A modern strategy for transporter engineering in biotechnology requires first that we construct suitable metabolic networks from genomic and other data, then that we use variations in expression profiles and desirable phenotypic properties to identify qualitatively those transporters whose properties most need improving, and finally that we use the methods of intelligent directed evolution to modify their properties and expression levels appropriately.

PDR10 and *PDR12*, encoding ABC multidrug transporters, actually enhanced growth rate. Thus, further investigation of the substrate preferences of these apparently promiscuous efflux pumps might pay dividends in both biotechnology and drug design.

Genes for efflux transporters

As well as the genes for efflux transporters described above, there is considerable interest in the recognition that a chief cause of antibiotic resistance, a huge continuing and present problem [68], is the ability of microbes to pump out such molecules using 'multidrug resistance' (MDR) efflux transporters [69,70]. MDR pumps are often of wide specificity, for example, for lipophilic compounds, and increasing numbers of structures are becoming known [71]. Efflux transporters are of wider significance in medicine because, by removing toxins, they lower the intracellular concentrations. This can be good in the case of genuine toxicants [72] but less so when they encode phenotypic resistance, for example, to anticancer agents [73]. However, in biotechnology it is both desirable and possible to select for strains that are particularly resistant to

Table 1. Some databases with a focus on membrane transporters

Name	Focus/organism	URL
Bioparadigms SLC tables	Human	http://www.bioparadigms.org/slc/intro.htm
<i>Caenorhabditis elegans</i> solute transporter database	<i>C. elegans</i>	http://www.wormslc.org/
Drugbank	Human/drugs	http://drugbank.ca
Human intestinal transporter database	Human/drugs	Not apparently directly online; data are downloadable from each paper's supplementary information.
Human transporter database	Human	http://htd.cbi.pku.edu.cn
Transportal	Human/drug transport	http://bts.ucsf.edu/fdatransportal/
TransportDB	Comparative genomics of transporters	http://membranetransport.org/
Transporter classification database (TCDB)	IUBMB-approved transporter classifications	http://www.tcdb.org/
Transporter database TP-search	Human/drug uptake	http://www.tp-search.jp
TransportTP	Transporter prediction	http://bioinfo3.noble.org/transporter/
Yeast metabolome database	<i>S. cerevisiae</i>	http://www.ymdb.ca/
Yeast transport protein database (YTPdb)	<i>S. cerevisiae</i>	http://ytpdb.biopark-it.be/ytpdb/
Yeti: yeast transport information	<i>S. cerevisiae</i>	http://genolevures.org/yeti.html

stresses, including stresses from organic solvents [74] and/or from high intra- and/or extracellular product titres.

Indeed, it is precisely this kind of positive selection that can be used to our advantage in biotechnology. Thus, by seeking tolerance to added compounds, efflux transporters have been found for alkanes [17,75–79], arenes [80,81], short-chain alcohols [82,83], terpenoids [78], short-chain fatty acids [84,85], and long-chain fatty acids [86,87], while those for isoprene and isoprenoids are eagerly sought [88]. Unusual efflux transporters produced by microbes for specific purposes include one for FAD in *Shewanella oneidensis* [89], while virtually all free-living aerobes must and do secrete siderophores to permit them to effect iron uptake [90–93].

Transporter-mediated osmotic stress engineering

If cells are to accumulate soluble products to high titres, there will always be the danger of significant osmotic

stresses as well as lowered water activities [94]. While these osmotic stresses can be relieved by the synthesis of so-called compatible solutes [95], such as betaine, another strategy includes their intracellular accumulation via uptake transporters [96]. *Corynebacterium glutamicum* provides an excellent example [97,98]. Note that inducing the synthesis of such compatible solutes can also be of value in the production of soluble and functional recombinant proteins [99,100].

Transporter engineering

Having established which transporters are important for the problem of interest, it is possible to improve them, typically by the methods of directed evolution [101–103]. These involve varying the primary sequence of the protein, and selecting those with improved properties, in an iterative manner. The variation in primary sequence is carried out by various forms of mutation and recombination, nowadays including the methods of synthetic biology in which we can control rather precisely which sequences are made by creating them at the DNA level by chemical synthesis [103–109]. The question then arises as to what kind of objective function we might seek (Box 4). We might wish to turn a concentrative uptake transporter into one that merely catalyses equilibration, in other words the efflux of product formed intracellularly. There is ample precedent for this loss of energy coupling, for example, in mutants of the normally concentrative lactose permease of *E. coli* [110] or of the mammalian intestinal di- and tripeptide transporter PepT1 (SLC15a1) [111], and – for influx of substances normally pumped out – of drug uptake via uncoupled variants of the LmrP ‘efflux’ transporter in lactobacilli [112]. By contrast, it was possible [113] to change a multidrug monovalent ‘efflux’ antiporter into one that used divalent ions. Thus there seems little doubt that we should be able to change the specificity [114], promiscuity [115], or detailed molecular transport pathways [116] of transporters by directed evolution as easily [117] as we can for other proteins [118]. Indeed, evidence for the selection of efflux transporters during the development of various amino acid fermentations was given

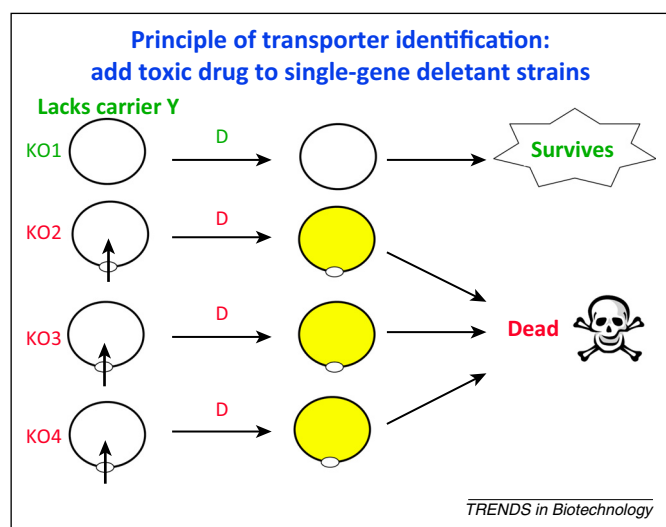


Figure 2. One approach to genome-wide identification of transporters for a toxic drug. This evaluates the enrichment of survivors when the gene encoding the transporter for an appropriate concentration of a cytotoxic drug is knocked out, relative to other strains in which other genes may or may not be knocked out.

Box 3. Classical strategies for detecting the roles of particular cell membrane transporters

Originally, the determination of which transporters accounted for the uptake of particular nutrients or other compounds used classical genetic techniques, often obtaining mutants in transporter genes by selecting for resistance to cytotoxic structural analogues of those nutrients. For example, canavanine is a structural analogue of arginine that can be taken up by cells, including those of *Saccharomyces cerevisiae* [9] and humans, and is incorporated into proteins where it disrupts their function, thus proving cytotoxic. Such cytotoxic molecules, that bear structural similarities to intermediary metabolites, are known as antimetabolites, and antimetabolite molecules such as analogues of folate, nucleobases, and nucleosides (Figure 1) continue to play a major role in cancer chemotherapy [145]. In yeast, the overwhelming bulk of canavanine uptake and, in mutants, resistance to it, is determined by the arginine transporter Can1p encoded by the gene *can1*. Strains lacking this gene function are, depending upon the precise metric, more than 100-fold more resistant to the antimetabolite than is the wild type [9], and of course the gene encoding the arginine transporter is explicitly named after its ability to encode resistance to this antimetabolite.

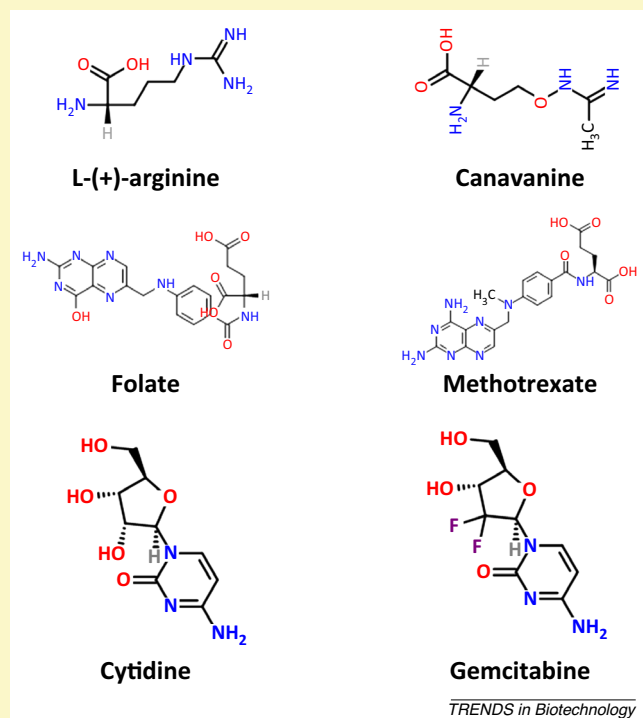


Figure 1. Some antimetabolites that bear structural similarities to natural metabolites with which they compete for uptake transport (and intracellular activity).

above. Papers showing a gain-of-function of NCgl1221 to constitutive glutamate excretion [119,120] are of special note, indicating the potential for transporter engineering.

E. coli contains a (possibly) surprising number of efflux pumps, comprising one sixth of all its transporters [117], even for sugars. Indeed, in *E. coli* there are as many as 37 MDR transporters [121], most commonly from the Major Facilitator Superfamily [122]. Arguably, the main efflux transporters are AcrB [123,124], MdfA [125], EmrE [126,127], and MtdM [122,128]. Thus, and while n-alkanes are much less cytotoxic than many other organic solvents [129], a particularly nice example of the directed evolution of a

Box 4. 'Influx' and 'efflux' transporters

Assessing the contributions of membrane proteins to the tolerance of stresses induced by fermentation or incubation conditions is also an important experimental approach to detecting 'efflux' transporters, a comment that leads us to note that thermodynamic principles mean that any transporter is theoretically reversible in its direction of operation, although for kinetic reasons connected with the Haldane relationship this may not appear to be the case. Thus, 'influx' and 'efflux' transporters refer to their normal direction of operation *in vivo*, and this is determined both by the thermodynamics and the mechanistic details of any energy coupling involved (Figure 1). While our chief interest here is in identifying cases where transporters exert significant flux control, increasing numbers of 3D protein structures for transporters are becoming available [146], and these are beginning to allow calculation of their molecular mechanisms from first principles, based on molecular dynamics [147]. This will also, in time, assist in their rational redesign.

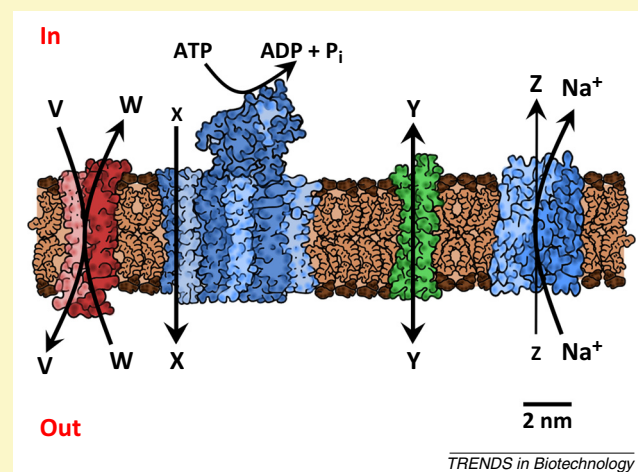


Figure 1. An illustration of four types of transporters. V and W are transported in exchange for each other – if there is a concentration gradient of one, it will drive the transport (antiport) of the other. X is transported out of the cell, potentially against its concentration gradient, by a transporter that couples its transport activity to ATP hydrolysis. Y enters and exits the cell by facilitated diffusion (it is a uniporter), while Z is taken up concentratively in symport with a sodium ion (that descends its own concentration gradient). The terms 'active' (concentrative) and 'passive' (equilibrative) are best used solely to describe the thermodynamics, with no mechanism being implied unless stated [13]. The membrane is drawn approximately to scale, with a typical *in vivo* protein:lipid ratio of 3:1 by mass. Note too that there can be a highly-intimate interaction between specific lipids and transporter function such that changing the former may affect the latter.

membrane protein for catalysing product efflux is the study of Foo and Leong [78], who evolved AcrB to drive improved efflux of the hydrocarbons n-octane and α -pinene from *E. coli* using selection against the toxicity of n-octanol (that was also presumably excreted) while Fisher *et al.* did the same for shorter-chain alcohols [82]. Mutations in several other genes, such as *lon*, *proV*, *soxS*, and *marR*, also act via AcrB to increase the solvent tolerance of *E. coli* [130,131]. Multidrug resistance transporters have also been used to export dipeptides [41] and arabinose [132] from *E. coli*, while NAD transporter engineering has been exploited to advantage in the whole cell biocatalytic production of dihydroxyacetone [133].

S. cerevisiae contains 28 members of the Major Facilitator Superfamily of multidrug efflux pumps and at least six members of the ABC multidrug transporter family

[134–136]. All these efflux pumps reside in the plasma membrane, while Vmr1p is a vacuolar membrane protein. While the importance of the plasma membrane pumps in drug resistance (notably to azoles) in pathogenic yeasts is well recognised [137], any possible role in the efflux of diesel fuels from engineered yeast seems not to have been considered [138] or, at least, not published.

Concluding remarks and future perspectives

In this short review we have sought to summarise some of the evidence that membrane transporters represent somewhat underutilised yet excellent targets for the purposes of strain improvement in biotechnology. Some of the evidence comes from more classical fermentations where such changes ‘emerged’ from undirected (mutation and selection) strain improvement programmes, while more recently there are examples of more deterministic strategies based on metabolic engineering. We anticipate many major improvements in the future as the powerful techniques of directed evolution are brought to bear on selected membrane transporters, especially those catalysing concentrative efflux of the desired product. Much as with pharmaceutical drug transporters [13], what we need now are good, predictive, quantitative structure–activity relationship models that will help to determine the activity of any transporter sequence for any drug. Such models will bring us truly closer to the era of ‘designer transporters for biotechnology’.

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