

Available online at www.sciencedirect.com



Physics

Physics Procedia 30 (2012) 160 – 163

Procedia

^{12th} International Conference on Muon Spin Rotation, Relaxation and Resonance The magnetic phase of Lithium transition metal phosphates Li*M*PO₄ (*M*=Mn, Co, Ni) detected by μ^+ SR

Oren Ofer^{a,*}, Jun Sugiyama^b, Jess H. Brewer^{c,a}, Martin Månsson^d, Krunoslav, Prša^d, Eduardo J. Ansaldo^a, Genki Kobayashi^e, Ryoji Kanno^e

^aTRIUMF, 4004 Wesbrook Mall, Vancouver, BC, V6T 2A3 Canada

^bToyota Central Research and Development Labs. Inc., Nagakute, Aichi 480-1192, Japan ^cDepartment of Physics and Astronomy, University of British Columbia, Vancouver, BC, V6T 1Z1 Canada ^dLaboratory for Neutron Scattering, ETH Zürich and Paul Scherrer Institut, CH-5232 Villigen PSI, Switzerland ^eDepartment of Electronic Chemistry, Interdisciplinary Graduate School of Science and Engineering, Tokyo Institute of Technology, Yokohama 226-8503, Japan

Abstract

The magnetic properties of the olivine-type compounds $\text{Li}M\text{PO}_4$ (M = Mn, Co, Ni) are probed using muon spin rotation/relaxation (μ SR). These materials pose an appealing magnetic structure and a high - potential technological interest as cathode materials for future rechargeable Li-ion batteries. The LiMPO₄ family of compounds consists of a corner-sharing MO₆ octahedra of high-spin M²⁺ ions manifesting an antiferromagnetic ground state below $T_N \approx 30$ K. Additionally, these compounds belong to a class of materials exhibiting properties between two- and three-dimensional systems. A comparative study between the family members is presented.

Keywords: Li-ion batteries, magnetism, Phospho-olivines

1. Introduction

The lithium metal phospho-olivines, Li*M*PO₄ (*M*=Mn,Co,Ni) have become a current topic in recent studies due to their high potential as a positive electrode material for rechargeable Li-ion batteries. Unlike the currently used layered LiCoO₂, where highly oxidized redox couples (Co^{3+/4+}) cause chemical instability and safety issues, the Li*M*PO₄ show high stability in lithium extraction and insertion cycles[1]. One promising compound in this group is the LiFePO₄ with a stable Fe^{2+/3+}, which offers a theoretical capacity of \approx 170 mAh/g and an operating voltage of 3.45 V, compatible with the current electrolytes used. However, by replacing the Fe with Mn, Co or Ni, the successful solid-state reaction in these compounds show higher operating voltages (4.1 V, 4.8 V and 5.1 V, respectively) suggesting a higher power density. Moreover, these phosphates possess high-spin *M*²⁺ metal ions which have been the focus of electronic[2] and magnetic[3, 4, 5] calculations and measurements[6, 7, 8, 9] in order to elucidate the exchange paths between the *M* ions.

*Corresponding author.

Email address: oren@triumf.ca (Oren Ofer)



Figure 1: The orthorhombic crystal structure of $LiMPO_4$. (b) The temperature dependence of the normalized wTF asymmetry, lines are guide to the eye.

Neutron diffraction on LiNiPO₄ reveals that below $T_N = 20.8$ K, the system adopts a collinear antiferromagnetic phase, with an intermediate incommensurate phase at $T_N \le T \le 2T_N \approx 40$ K[9]. Similar behavior was observed in LiMnPO₄[7]. However, this was not observed in the isostructural LiCoPO₄ which, at $T < T_N$, the spin structure of the Co is slightly rotated from the principal crystallographic *b* axis[6]. Moreover, LiCoPO₄ was found to have isotropic domains, unlike LiNiPO₄[10]. Despite these extensive measurements, the microscopic magnetic nature of the Neél phase has, to our knowledge, have not been investigated by μ SR. Dipolar field calculations indicate that there are several possible muon sites in LiMPO₄ and therefore we would naturally expect multiple muon-spin precession frequencies below T_N . Here, following the μ SR study on LiFePO₄[11], we aim to characterize the magnetic phase of LiMPO₄ using the local probe μ SR and bulk susceptibility. Our major finding is a second Neél phase in LiMPO₄ and LiCoPO₄ observed by ZF- μ SR and bulk susceptibility. In LiNiPO₄ a single static phase is observed below T_N .

2. Experimental

Powder samples of LiMPO₄ were prepared by the flux technique using Li₃PO₄ and MCl₂ as starting agents. Xray diffraction show the successful synthesis of a single phase orthorhombic olivine structure (space group *Pnma* no. 62) of LiMPO₄ revealing the crystallographic parameters which agree with previously published data. The μ SR spectra were measured on the **M20** surface muon beam line using the **LAMPF** spectrometer at TRIUMF, Canada. Approximately 500 mg powder sample were placed in a very thin 1×1 cm² Al-coated Mylar envelope, placed onto a low-background sample holder in a liquid-He flow cryostat. Subsequently, weak-Transverse field (wTF) and Zero-Field (ZF) measurements were taken at $1.7 \le T \le 50$ K. Complementary bulk dc-susceptibility measurements were performed using a Quantum Design Magnetic Properties Measurements System superconducting quantum interference device (SQUID) at temperatures between $1.7 \le T \le 400$ K under zero-field cooled and field-cooled conditions.

3. Results and Discussion

The magnetic phase transitions are probed using the wTF technique. In this technique, a small applied field of $H_{\text{TF}} = 30$ Oe is applied perpendicular to the initial muon polarization and the initial asymmetry is proportional to the volume fraction of the paramagnetic phase. The wTF asymmetry was well described by

$$A_{\rm TF}(t) = A_0 \exp(-\lambda_{\rm TF} t) \cos(\gamma_{\mu} H_{\rm TF} t + \varphi) + A_{\rm Fast} \exp(-\lambda_{\rm Fast} t)$$
(1)



Figure 2: (a) The ZF- μ SR spectra for LiMPO₄ taken at base temperature (T = 1.8 K), the solid lines are the fits to Eq. (2). (b) The temperature dependence of the muon precession frequencies (f_1 and f_2) identified in the ZF-spectra.

where the gyromagnetic ratio $\gamma_{\mu} = 135.54 \text{ MHz/T}$, $A_0 (A_{\text{Fast}})$ and $\lambda_{\text{TF}} (\lambda_{\text{Fast}})$ are the initial precessing (fast relaxing) asymmetry and relaxation, respectively. In Figure 1 we depict the normalized asymmetry ($A_0(0)$) versus the temperature. The transition to the ordered state is clearly seen by the abrupt decrease in the asymmetry as the temperature is lowered, and corresponds to 22.09 K (LiCoPO₄), 21.95 K (LiNiPO₄) and 33.504 K (LiMnPO₄) when the normalized asymmetry is 0.5. These temperatures also correlate with the transition temperature measured by the SQUID (see Fig. 3).

ZF- μ SR measurements were taken as described in Sec. 2. Figure 2a displays the time dependence of the raw ZF- μ SR data for LiMPO₄ taken at T = 1.8 K. A beat is clearly observed in the spectra, indicating more than a single frequency. Moreover, the initial asymmetry of the LiMnPO₄ is much smaller than LiCoPO₄ and LiNiPO₄. The origin of this mismatch is not understood at the moment. Nonetheless, the ZF asymmetry was well fitted, for each compound at $T \le T_M^*$ (M =Mn, Co and Ni), by a sum of two relaxing frequencies and a slowly relaxing tail,

$$A_0 P_{\text{ZF}}(t) = A_1 \exp(-\lambda_1 t) \cos(\omega_1 t + \varphi_1) + A_2 \exp(-\lambda_2 t) \cos(\omega_2 t + \varphi_2) + A_{\text{tail}} \exp(-(\lambda_{\text{tail}} t)^{\beta})$$
(2)

where $\omega_i = \gamma_\mu B_i$ is the muon frequency at its interstitial site (i = 1, 2) experiencing an internal field B_i . λ_i and λ_{tail} are the relaxation rates of each of the sites *i* or tail. The solid lines in Fig. 2a demonstrate the fit. For temperatures $T > T_M^*$, a single frequency is observed and hence the asymmetry of the second frequency is a priori set to nil, $A_2 = 0$ in Eq. (2). The temperature dependence of the frequencies $f_i \equiv \omega_i/2\pi$ for the three compounds are depicted in Fig. 2. Several points can be considered. First, as expected, the frequencies decrease monotonically with increasing temperatures, this is more pronounced in LiMnPO₄ where the frequencies at T = 1.7 K are ≈ 80 MHz. Secondly, unlike LiMnPO₄ and LiCoPO₄ with $T_{Mn}^* = 12.7 \pm 2.2$ K, and $T_{Co}^* = 16.2 \pm 1.25$ K, LiNiPO₄ experiences two frequencies for all $T \leq T_N$, thus $T_{Ni}^* = T_N$. Therefore, we suggest that the muon experiences a single static antiferromagnetic phase in LiNiPO₄ and an additional transition to a second antiferromagnetic phase below T_M^* for M =Co and Mn. We now try to observe whether these additional ordered phases can also be seen with bulk susceptibility.

Bulk susceptibility (χ) data for the three compounds are shown in Fig. 3. Due to the appearance of the second frequency in the μ SR spectra of LiCoPO₄ and LiMnPO₄ at T_M^* , we performed detailed measurements and analysis of the bulk data. At first glance, for all three compounds, χ increases towards T_N and then decreases sharply indicating the antiferromagnetic transition temperature as expected, which also correlates with the wTF- μ SR measurements with no obvious field-cooled or zero-field-cooled dependence. However, by computing $\partial \chi / \partial T$ for $T \leq T_N$, changes are



Figure 3: The bulk susceptibility of LiMPO₄.

revealed. In LiNiPO₄ and LiCoPO₄, apart from the transition, $\partial \chi / \partial T$ is smooth and no additional transitions are observed. In fact, the two compounds are very similar in their χ and $\partial \chi / \partial T$ behavior, and a small change in the slopes of $\partial \chi / \partial T$ is observed at ≈ 15 K. In LiCoPO₄ this is accompanied by an increasing deviations between the ZF and FC measurements as *T* decreases below T_{Co}^* , which is not seen elsewhere. Moreover, such deviations are not observed in LiNiPO₄ consistent with a $T_{Ni}^* = T_N$. In LiMnPO₄, χ shows a small kink at 13 K, which is clear in $\partial \chi / \partial T$, corresponding to T_{Mn}^* .

4. Acknowledgments

We thank the staff of TRIUMF for help with the μ SR experiments. We thank K. Nishiyama of KEK for discussion. JHB is supported at UBC by NSERC of Canada, and (through TRIUMF) by NRC of Canada, KHC by NSERC of Canada and (through TRIUMF) by NRC of Canada.

5. References

- [1] A. V. Murugan, T. Muraliganth, P. J. Ferreira, A. Manthiram, Inorg. Chem. 48, 946 (2009).
- [2] P. Tang, N. A. W. Holzwarth. Phys. Rev. B. 68, 165107 (2003).
- [3] D. Dai, M. H. Whangbo, H. J. Koo, X. Rocquefelte, S. Jobic, Inorg. Chem. 44, 2407 (2005).
- [4] E. Bousquet, N. Spaldin, Phys. Rev. B. 82, 220402 (2010).
- [5] K. Yamauchi, S. Picozzi, Phys. Rev. B. 81, 024110, (2010).
- [6] D. Vaknin, J. L. Zarestky, L. L. Miller, J. P. Rivera, H. Schmid, Phys. Rev. B. 65, 224414 (2002).
- [7] J. Li, W. Tian, Y. Chen, J. Zarestky, J. Lynn, D. Vaknin, Phys. Rev. B. 79 144410 (2009).
- [8] T. B. S. Jensen, N. B. Christensen, M. Kenzelmann, H. M. Rønnow H, C. Niedermayer, N. Andersen, K. Lefmann, M. Jiménez-Ruiz, F. Demmel, J. Li, J. L. Zarestky, D. Vaknin, Phys. Rev. B. 79, 092413 (2009).
- [9] D. Vaknin, J. Zarestky, J. P. Rivera, H. Schmid, Phys. Rev. Lett. 92, 207201 (2004).
- [10] B. B. Van Aken, J. P. Rivera, H. Schmid, M. Fiebig, Phys. Rev. Lett. 101, 157202 (2008).
- [11] J. Sugiyama, H. Nozaki, M. Harada, K. Kamazawa, O. Ofer, M. Månsson, J. H. Brewer, E. J. Ansaldo, K. H. Chow, Y. Ikedo, Y. Miyake, K. Ohishi, I. Watanabe, G. Kobayashi, R. Kanno, Phys. Rev. B 84 (2011) in press.