



ELSEVIER

Contents lists available at [ScienceDirect](http://ScienceDirect)

## Data in Brief

journal homepage: [www.elsevier.com/locate/dib](http://www.elsevier.com/locate/dib)

## Data Article

<sup>1</sup>H-NMR dataset for hydroxycoumarins –Aesculetin, 4-Methylumbelliferone, and umbelliferone

Rui Galhano dos Santos\*, João Carlos Moura Bordado, Maria Margarida Mateus

CERENA, Departamento de Engenharia Química e Biológica, Torre Sul, Instituto Superior Técnico, Av. Rovisco Pais, 1049-001 Lisboa, Portugal

## ARTICLE INFO

## Article history:

Received 18 March 2016

Received in revised form

18 May 2016

Accepted 23 May 2016

Available online 30 May 2016

## Keywords:

Coumarins

NMR

Acetone

## ABSTRACT

Herein, the integrated raw data regarding the <sup>1</sup>H-NMR, experiments of Aesculetin, 4-Methylumbelliferone, and umbelliferone, in Acetone-d<sup>6</sup> at 25 °C, are presented for further analysis and comparison purposes, for whom may be interested.

© 2016 The Authors. Published by Elsevier Inc. This is an open access article under the CC BY license (<http://creativecommons.org/licenses/by/4.0/>).

## Specifications Table

Subject area	Chemistry
More specific subject area	Structural characterization
Type of data	Figures, table
How data was acquired	The data was acquired on a Bruker Avance 400 spectrometer operating at 400 MHz
Data format	Raw
Experimental factors	Sample solutions were prepared with deuterated Acetone (Acetone-d <sup>6</sup> ). Residual acetone peak: 2.05
Experimental features	Detection temperature was set at 25 °C. Samples were scanned 16 times.
Data source location	Lisbon, Portugal, GPS: 38° 44' 10.31"N; 9° 08' 19.66"W
Data accessibility	Data is provided in the article

\* Corresponding author.

E-mail address: [rmglopes@ist.utl.pt](mailto:rmglopes@ist.utl.pt) (R.G. dos Santos).

## Value of the data

- Useful to be used as reference data for chemical shifts for other related compounds.
- Comparison between coumarins with different substitution patterns.
- Helpful in the assignment of signals of molecules containing coumarin backbone residues.

## 1. Data

The data henceforth described refers to the  $^1\text{H-NMR}$  experiments of three coumarins, Aesculetin, 4-Methylumbelliferone, and umbelliferone, in deuterated acetone.

The data disclosed regards the  $^1\text{H-NMR}$  experiments conducted with 4-Methylumbelliferone (Fig. 1), umbelliferone (fig. 2), and Aesculetin (Fig. 3), in deuterated acetone. This data may be helpful for those who intend to compare this data with other from molecules containing the same or related coumarins scaffolds. Table 1 lists all the peaks and their respective intensities.

## 2. Experimental design, materials, and methods

The coumarins used were chemical grade and purchased from Sigma-Aldrich.

The compounds were subjected to  $^1\text{H-NMR}$  measurements. The experiments were performed on a Bruker Avance 400 liquid NMR spectrometer, operating at 400 MHz. Detection temperature was set at 25 °C. The samples were loaded in a 5 mm NMR tube. The solvent peak was calibrated according to Gottlieb et al. [1].

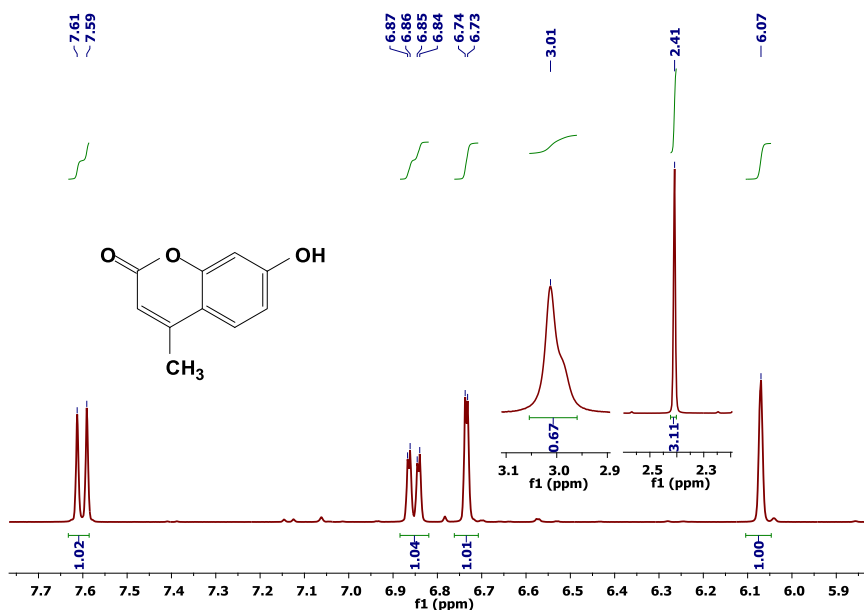


Fig. 1.  $^1\text{H-NMR}$  spectrum of 4-Methylumbelliferone (acetone- $\text{d}_6$ , 298 K, 400 MHz).

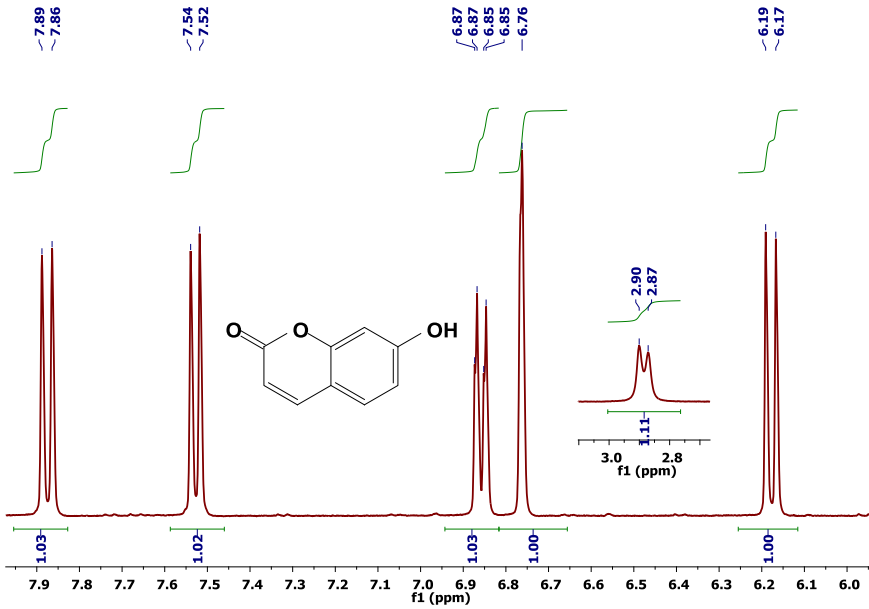


Fig. 2. <sup>1</sup>H-NMR spectrum of umbelliferone (acetone-d<sub>6</sub>, 298 K, 400 MHz).

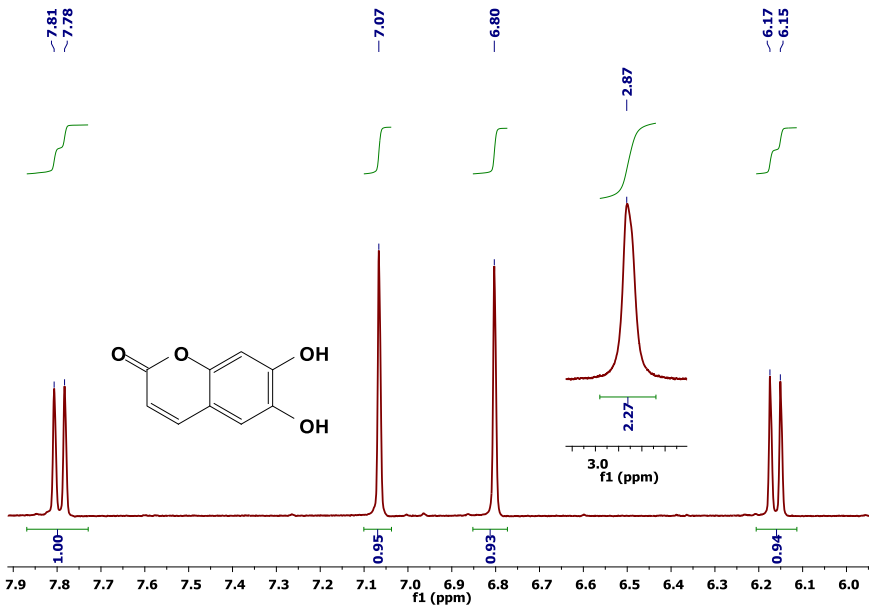


Fig. 3. <sup>1</sup>H-NMR spectrum of Aesculetin (acetone-d<sub>6</sub>, 298 K, 400 MHz).

**Table 1**

Chemicals shifts for 4-methylumbelliferone, umbelliferone, and Aesculetin [1].

<b>Compound</b>					
<b>4-Methylumbelliferone</b>		<b>Umbelliferone</b>		<b>Aesculetin</b>	
<b>ppm</b>	<b>Intensity</b>	<b>ppm</b>	<b>Intensity</b>	<b>ppm</b>	<b>Intensity</b>
2.41	2129.6	2.87	171.2	2.87	134.2
3.01	116.5	2.9	194.3	6.15	246.5
6.07	607.4	6.17	501.6	6.17	255.4
6.73	518.7	6.19	515.2	6.8	456
6.74	535.3	6.76	662.3	7.07	484.5
6.84	291.3	6.85	379.1	7.78	237.5
6.85	251.3	6.87	274.9	7.81	233.3
6.86	308.3	7.52	511.3	–	–
6.87	268.4	7.54	479.5	–	–
7.59	489.4	7.86	484.7	–	–
7.61	461.7	7.89	471.7	–	–
9.35	70.5	–	–	–	–

**Transparency document. Supplementary material**

Transparency document associated with this article can be found in the online version at <http://dx.doi.org/10.1016/j.dib.2016.05.048>.

**Reference**

- [1] Hugo E. Gottlieb, Vadim Kotlyar, Abraham Nudelman, NMR chemical shifts of common laboratory solvents as trace impurities, *J. Org. Chem.* 62 (1997) 7512–7515. <http://dx.doi.org/10.1021/jo971176v>.