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Biochimica et Biophysica Acta 1768 (2007) 2400-2408



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# Membrane interactions and the effect of metal ions of the amyloidogenic fragment A $\beta(25-35)$ in comparison to A $\beta(1-42)$

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Received 17 February 2007; received in revised form 2 May 2007; accepted 3 May 2007 Available online 22 May 2007

### Abstract

 $A\beta(1-42)$  peptide, found as aggregated species in Alzheimer's disease brain, is linked to the onset of Alzheimer's disease. Many reports have linked metals to inducing A $\beta$  aggregation and amyloid plaque formation.  $A\beta(25-35)$ , a fragment from the C-terminal end of  $A\beta(1-42)$ , lacks the metal coordinating sites found in the full-length peptide and is neurotoxic to cortical cortex cell cultures. We report solid-state NMR studies of  $A\beta(25-35)$  in model lipid membrane systems of anionic phospholipids and cholesterol, and compare structural changes to those of  $A\beta(1-42)$ . When added after vesicle formation,  $A\beta(25-35)$  was found to interact with the lipid headgroups and slightly perturb the lipid acyl-chain region; when  $A\beta(25-35)$  was included during vesicle formation, it inserted deeper into the bilayer. While  $A\beta(25-35)$  retained the same  $\beta$ -sheet structure irrespective of the mode of addition, the longer  $A\beta(1-42)$  appeared to have an increase in  $\beta$ -sheet structure at the C-terminus when added to phospholipid liposomes after vesicle formation. Since the  $A\beta(25-35)$  fragment is also neurotoxic, the full-length peptide may have more than one pathway for toxicity.

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Keywords: Amyloid AB; Peptide-lipid interactions; Phospholipid membranes; Solid-state NMR; Structure; Metal interactions

### 1. Introduction

The mechanisms that govern the progression of Alzheimer's disease (AD), marked by the accumulation of neuronal plaques consisting mainly of aggregated 39–42 residue amyloid- $\beta$  peptides (A $\beta$ ), have yet to be established. A $\beta$  peptides are a product of cleavage by the  $\beta$ - and  $\gamma$ -secretases of the larger transmembrane amyloid precursor protein (APP) [1–3] and aggregate to form amyloid fibrils (reviewed in [4]). While structural models for A $\beta$ (1–40) fibrils based on solid-state NMR data have been presented [5], it is likely that soluble oligomers of A $\beta$  peptides are principally responsible for cytotoxicity [4]. Both A $\beta$ (1–40) and A $\beta$ (1–42) have been extensively studied for their cytotoxic effect, for which several

\* Corresponding author. *E-mail address:* fs@unimelb.edu.au (F. Separovic). interrelated mechanisms are suggested: metal coordination by histidines 6, 13 and 14 is implicated in oligomer formation [6], and inducing A $\beta$  insertion into phospholipids membrane bilayers [6,7]. Copper and zinc ions (Cu<sup>2+</sup> and Zn<sup>2+</sup>), found at increased levels in neuronal plaques [8,9], can also undergo Fenton-like chemistry and produce free radicals that may result in both the observed dityrosine crosslinks between intermolecular Tyr10 residues [6,10,11], lipid peroxidation [12], and the extensive oxidative stress observed in post mortem examination of AD patient brain [13–15].

Significantly, the largely hydrophobic A $\beta$ (25–35) fragment of naturally occurring A $\beta$  peptides does not have any known direct metal binding or cross-linking potential, yet aggregates much more readily than other fragments and is toxic toward cultured cortical cells as well as neuronal cells [4,16] via membrane interaction and bilayer disruption similar to the fulllength A $\beta$ (1–42) [17–19]. Further investigation of this frag-

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ment is needed due to the potential role  $A\beta(25-35)$  may play in cytotoxicity of the longer  $A\beta$  peptides.

We, therefore, examine the effect of  $A\beta(25-35)$  on mixed phospholipid bilayers in the presence of metal ions and compare structural changes of  $A\beta(25-35)$  to those of  $A\beta(1-42)$  in the presence of phospholipids. Particular attention is directed toward lipid membrane composition, which can influence the formation of  $\beta$ -sheet structure [19]. Preferential interactions of  $A\beta$  peptides with charged phospholipids as important regulators of membrane disruption have been implicated in our previous studies of  $A\beta(1-42)$  [20], together with other NMR results [21], fluorescence and atomic force microscopy [22]. Consequently, varying the lipid composition of membranes may aid in understanding the progression of AD.

Cholesterol represents another important membrane component, as sterols in general decrease membrane fluidity (or conversely increase acyl chain order) [23]. The role of cholesterol in modulating A $\beta$ /lipid interactions, however, remains controversial: cultured PC12 cells enriched with cholesterol block  $\beta$ -amyloid toxicity and appear to reduce lipid aggregation [24,25] while the cholesterol-lowering statin drugs also slow the development of AD [26]. Other less common lipids have also been found to promote peptide aggregation, e.g. A $\beta$  bound to GM1 ganglioside may seed fibril development through 'raft-like' formation [27].

In addition to metals and lipid composition, the manner in which the peptide is added to model membranes has bearing. The distinction of peptide addition made in our previous studies of  $A\beta(1-42)$  demonstrated that  $A\beta(1-42)$  disrupts model membranes when incorporated within the lipid vesicles, while peptide added after vesicle formation binds preferentially to the phospholipid headgroups [19,20]. These modes of peptide addition may sample different points in the kinetics of  $A\beta$  interactions and AD progression. Associated peptide may represent the toxic soluble  $A\beta$  oligomer interacting with a membrane surface; alternatively, the "soluble" (non-amyloidic) oligomer may actually preferentially partition into membranes and be better represented by incorporated peptide, while the associated peptide readily aggregates toward the insoluble, amyloidic species.

The lipid charge, membrane order or 'fluidity' and peptide– lipid interactions, together with metal ions, may play an important role in inducing peptide toxicity and aggregation. Therefore, we examined the effect of  $A\beta(25-35)$  on mixed phospholipid bilayers together with metal ions and compared structural changes of both  $A\beta(1-42)$  and  $A\beta(25-35)$  in the presence of phospholipids.

### 2. Materials and methods

#### 2.1. Materials

Synthetic deuterated 1-palmitoyl( $d_{31}$ )-2-oleoyl-*sn*-glycero-3-phosphocholine (*d*POPC), 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine (POPC), 1-palmitoyl-2-oleoyl-*sn*-glycero-3-[phospho-L-serine] (POPS), brain total lipid extract and brain polar lipid extract (porcine) were purchased from Avanti Polar Lipids (Alabaster, USA) and used without further purification. Cholesterol was purchased from Sigma Aldrich (St. Louis, USA). Unlabelled A $\beta$ (25–35) peptide, with sequence GSNKGAIIGLM was purchased through Auspep (Melbourne, Australia). Specifically labelled A $\beta$ (25–35) and A $\beta$ (1–42) peptides were made using standard FMOC-solid phase peptide synthesis [6], purified by reverse phase HPLC and the products identified by mass spectrometry with m/z of 1064.29 for A $\beta$ (25–35) and m/z of 4519.11 for A $\beta$ (1–42). The peptides were <sup>13</sup>C labelled with sequence: G SNK-[<sup>13</sup>C<sub>2</sub>]G [<sup>13</sup>C<sub>1</sub>]A IIG[<sup>13</sup>C<sub>1</sub>]LM for A $\beta$ (25–35), and DAEFR HDSGY EVHHQ KLVFF AEDVG SNK[<sup>13</sup>C<sub>2</sub>]G[<sup>13</sup>C<sub>1</sub>]A IIG[<sup>13</sup>C<sub>1</sub>]A IIG[<sup>13</sup>C<sub>1</sub>]A IIG[<sup>13</sup>C<sub>1</sub>]A VGGVV IA for A $\beta$ (1–42).

Note: residues of A $\beta$ (25–35) are indexed herein to be consistent with that of A $\beta$ (1–42).

<sup>13</sup>C<sub>2</sub>-Gly, <sup>13</sup>C<sub>1</sub>-Ala and <sup>13</sup>C<sub>1</sub>-Leu were purchased from Cambridge Isotope Laboratories (Andover, USA) and FMOC-succinamide (OSu) from Auspep (Melbourne, Australia). FMOC-protected amino acids were synthesized and purified for use in peptide synthesis.

### 2.2. FMOC-<sup>13</sup>C-labelled amino acid synthesis

Na<sub>2</sub>CO<sub>3</sub> solution (1 M. 30 ml) was added to 1 g each of  ${}^{13}C_1$ -Leu,  ${}^{13}C_1$ -Ala and <sup>13</sup>C<sub>2</sub>-Gly at room temperature. A 12% mol excess of Fmoc-OSu in MeCN (40 ml) was added to each reaction mixture drop-wise and the reaction stirred, with white precipitate forming overnight. De-ionised water (40 ml) was added to each solution and stirred until precipitate was dissolved. The aqueous phases were extracted with diethyl ether (1×50 ml,  $1 \times 60$  ml,  $1 \times 70$  ml) and washed with de-ionised water (10 ml) with each extraction. The aqueous phase was acidified to approximately pH 2.0 with HCl (3 M) and extracted with dichloromethane (3×70 ml). Organic phases were combined, dried over Na2SO4 and filtered. Solvent was removed under vacuum, and the white crude product was recrystallised from dichloromethane/hexane, ethyl acetate/hexane and chloroform/hexane for leucine, alanine and glycine, respectively. Crystallised product was dried under vacuum [28]. The vields of FMOC-amino acids were 2.54 g (94%) and 0.17 g (6%) from first and second recrystallisations, respectively, for  ${}^{13}C_{1-}$ Leu; 3.24 g (93%) for  ${}^{13}C_1$ -Ala; and 3.68 g (93%) and 0.27 g (7%) from first and second recrystallisations, respectively, for <sup>13</sup>C<sub>2</sub>-Gly.

#### 2.3. NMR sample preparation

For solid-state NMR studies, multilamellar vesicles (MLV) were used rather than LUV preparations. LUV typically have diameter of 100–200 nm and hence, due to motional averaging, do not reveal the full axially symmetric <sup>31</sup>P chemical shift anisotropy (CSA) and <sup>2</sup>H quadrupolar splittings ( $\Delta v_Q$ ) of liquid lamellarphase lipid membranes. MLV were prepared by co-dissolving equal amounts of *d*POPC and POPS in chloroform and methanol (9:1) solution in a round bottom flask, removing solvent under vacuum, re-dissolving the resulting lipid film in 3 mL 0.05 M Tris–HCl buffer (pH 7), and freeze–thawed several times using liquid N<sub>2</sub> and warm water. Typically 30 mg of lipid and 1.5 mL of solvent was used to make several samples. For the sterol containing samples, *d*POPC, POPS and cholesterol (1:1:1 molar) were co-dissolved in solvent to prepare the MLV as described above. Experiments were also conducted at pH 5 with no obvious difference in the <sup>31</sup>P lineshape and <sup>2</sup>H quadrupolar splittings. Hence only spectra at pH 7 are shown.

### 2.4. Incorporated A<sub>β</sub>

dPOPC/POPS (1:1) or dPOPC/POPS/cholesterol (1:1:1) were dissolved in chloroform and methanol (9:1) solution in a round bottom flask and the solvent removed under vacuum resulting in a lipid film. The film was suspended in 0.5 ml (0.05 M) Tris–HCl buffer (pH 7) and A $\beta$ (25–35) or A $\beta$ (1–42) dissolved in hexafluoroisopropanol (HFIP) (Sigma Aldrich, St. Louis, USA) was added to the suspension resulting in approximately 10% v/v HFIP. The mixture was freeze-dried overnight and the film re-suspended with distilled water and freeze–thawed for several cycles. The peptide to lipid molar ratio (P/L) was 1:30.

### 2.5. Associated $A\beta$

 $A\beta(25-35)$  or  $A\beta(1-42)$  was solubilized in HFIP and dried into a thin film. MLV preparations were subsequently added at P/L of [1:30], followed by several freeze-thaw cycles. Each sample typically contained 1 mg of peptide in 5 mg of lipid. Labelled peptide used for <sup>13</sup>C NMR studies was also freeze-dried from HFIP.

#### 2.6. Addition of metal ions

Copper and zinc ions were prepared using 0.1 M copper glycinate (CuGly<sub>2</sub>) solution or 0.1 M zinc glycinate solution (ZnGly<sub>2</sub>), respectively, by dissolving 1:2 mole ratio of metal chloride with glycine in de-ionised water. Copper and zinc glycinate solution was added to the samples to give 1:1 mole ratio of metal ions to peptide.

### 2.7. Solid-state NMR

NMR experiments were conducted on a Varian Inova 300 MHz spectrometer (Palo Alto, USA) operating at a resonance frequency of 300 MHz for <sup>1</sup>H. Static spectra were acquired at 28 °C, using a Doty (Colombia, USA) 5 mm double resonance probe. <sup>31</sup>P broad-line spectra were collected between 1.5 s recycle delays with a proton-decoupled Hahn echo pulse sequence using a 4  $\mu$ s  $\pi/2$ pulse and 40 µs echo delay, acquired with a spectral width of 62.5 kHz, and processed with 100 Hz exponential line broadening. Chemical shift was referenced using 80% phosphoric acid (H<sub>3</sub>PO<sub>4</sub>) at 0 ppm. <sup>2</sup>H broad-line spectra were collected between 0.4 s recycle delays with a solid echo pulse sequence [29] using a 6  $\mu$ s  $\pi/2$  pulse and effectual 40  $\mu$ s echo delay, acquired with a spectral width of 250 kHz, and processed as detailed below. <sup>31</sup>P  $T_2$  and <sup>13</sup>C experiments 1D cross-polarisation (CP) experiments employed a 5 mm triple resonance Doty (Columbia, USA) probe at 6 kHz magic angle spinning (MAS).  $^{31}$ P  $T_2$  measurements were made using a Hahn-echo sequence with a 5 µs  $\pi/2$ pulse and variable echo delays set at multiples of the rotor rotation period; nonlinear data were fit to a single exponential decay in all cases using gnuplot 4.0 [30]. <sup>13</sup>C CP-MAS spectra [31] were typically signal averaged over 30-k scans with 2 s recycle delays, using 55.5 kHz proton and 61.5 kHz <sup>13</sup>C field strengths for 1 ms cross-polarisation contact time, acquired with a spectral width of 30 kHz, and processed with 50 Hz exponential line broadening. <sup>13</sup>C chemical shifts were referenced externally to tetramethylsilane (TMS) at 0 ppm.

### 2.8. <sup>2</sup>H spectra dePaking, and order parameters of lipid acyl chains

<sup>2</sup>H spectra were collected with shortened pre-acquisition delays to permit *post hoc* selection of the echo maximum. Data to be used for subsequent dePaking were processed in NMRpipe [32] to DC-correct, left-shift to the echo maximum, and apodize with a Gaussian function (300 Hz decay constant)

beginning at 1.2 ms (and un-attenuated at earlier time), zero-fill to 2048 points, Fourier-transform and only zero-order phase corrected to generate static, unoriented power spectra. Oriented spectra were generated by numerically "dePaking" using Single Value Decomposition (SVD) [33] as well as a straightforward nonlinear least squares approach. The computer code was written in C and used functions freely available from the GNU Scientific Library (GSL) v1.7 [34] to handle the SVD and least squares fitting mechanics. Order parameters were calculated simply by measuring the quadrupolar splitting in the calculated 0°-oriented spectrum for each palmitoyl-CD<sub>n</sub> acyl chain position and divided by 255 kHz, the static coupling constant [35,36].

### 3. Results

### 3.1. Effect of $A\beta(25-35)$ and metals on cholesterol-phospholipid bilayers

The association of  $A\beta(25-35)$  with the phospholipid bilayer resulted in a small but significant reduction in the <sup>2</sup>H quadrupolar splitting but little effect on the <sup>31</sup>P CSA. These results indicated that the addition of  $A\beta(25-35)$  to the lipid bilayer resulted in more 'disorder' in the deuterated acyl region of the membrane and less effect on the phosphate headgroups (Fig. 1a–d, Table 1). Subsequent addition of  $Cu^{2+}$  after association of A $\beta$ (25–35) with the bilayer resulted in a slight increase in the <sup>2</sup>H quadrupolar splitting and a decrease in <sup>31</sup>P CSA together with  $T_2$  broadening (due to the presence of the paramagnetic copper) of the lineshape. Although the effect of changes in  $T_2$  on the lineshape are complicated since  $T_2$  is anisotropic across the <sup>31</sup>P powder pattern [37,38], the changes indicate disordering in the lipid headgroup region and restriction in phospholipid acyl chain motions (Fig. 1e, f). Similar results were observed with the addition of  $Zn^{2+}$  ions (Fig. 1g, h).

The presence of cholesterol in membrane bilayers is known to promote lipid domain or 'raft' formation and an increase in lipid order (or decrease in lipid 'fluidity') with increasing cholesterol level. If membrane disruption or destabilization is



Fig. 1. <sup>2</sup>H and <sup>31</sup>P NMR spectra of: (a, b) dPOPC/POPS bilayers with (c, d) associated Aβ(25–35), plus (e, f) copper, or (g, h) zinc ions.

Table 1

<sup>2</sup>H (CD<sub>2</sub>, CD<sub>3</sub>) quadrupolar splitting ( $\Delta v_Q$ ) and <sup>31</sup>P chemical shift anisotropy ( $\Delta \sigma$ ) for *d*POPC/POPS bilayers with 1:30 peptide/lipid ratio at 28 °C in Tris–HCl pH=7.2. CD<sub>2</sub> and CD<sub>3</sub> quadrupolar splittings ( $\Delta v_Q$ ) are taken from the 90° edges of the powder pattern

Observed nuclei		$\frac{{}^{2}\mathrm{H}}{(\Delta\nu_{\mathrm{Q}}) \mathrm{kHz}}$		<sup>31</sup> P	
Sample				$(\Delta \sigma)$ ppm $T_2$ (ms	
		$CD_2 \pm 0.3$	$CD_3 \pm 0.1$	CSA±2	±0.3
a b	dPOPC/POPS (1:1) dPOPC/POPS (1:1)+ AB(25-35) incorp	25 27	2.5 2.3	-43 -45	6.2 7.0
c	dPOPC/POPS (1:1)+ A $\beta$ (25-35) assoc.	24	2.3	-47	1.2

related to the toxic mechanism of A $\beta$ , the presence of cholesterol may play a role in influencing membrane integrity. <sup>31</sup>P and <sup>2</sup>H NMR spectra showed that cholesterol resulted in a

<sup>1</sup>P and <sup>2</sup>H NMR spectra showed that cholesterol resulted in a significantly greater <sup>2</sup>H quadrupolar splitting and narrower <sup>31</sup>P CSA (Table 2a) compared to pure *d*POPC/POPS bilayers (Table 1a) [20]. This result is consistent with the understanding in the literature of the effects of cholesterol on lipid bilayers, and in particular on the <sup>31</sup>P powder patterns, arising from the separation of lipid headgroups by intercalation of cholesterol between the lipids [36].

When A $\beta$ (25–35) was associated, an increase in <sup>2</sup>H splitting but not in the <sup>31</sup>P CSA was observed (Table 2). Following subsequent addition of  $Cu^{2+}$  ions, a significant decrease in  ${}^{31}P$ CSA (together with  $T_2$  broadening due to the strong paramagnetic effect of the copper) was observed, with only a slight increase in the quadrupolar splittings (Table 2); while Zn<sup>2+</sup> addition also had a slight increase in quadrupolar splitting but no significant effect on the phosphorus headgroup. Addition of either Cu<sup>2+</sup> or Zn<sup>2+</sup> ions to the cholesterol-phospholipid bilayer in the absence of the peptide resulted in a slight increase in the quadrupolar splitting and very little effect on the <sup>31</sup>P CSA (Table 2). A decrease in signal to noise ratio was also observed for the phosphorus headgroup in the presence of  $Cu^{2+}$  ions in comparison to the <sup>2</sup>H spectra due to the paramagnetic relaxation properties of copper, indicative of a preferential interaction of  $Cu^{2+}$  for the lipid headgroup.

Table 2

Effect of $A\beta(25-35)$ and cholesterol on	<sup>2</sup> H (CD <sub>2</sub> , CD <sub>3</sub> ) quadrupolar splitting
$(\Delta v_Q)$ and <sup>31</sup> P chemical shift anisotropy	$(\Delta \sigma)$ of <i>d</i> POPC/POPS bilayers

Observed nuclei		<sup>2</sup> H		<sup>31</sup> P	
		$(\Delta v_{\rm Q})$ kHz		$(\Delta \sigma)$ ppm	
dPOPC/POPS/cholesterol (1:1:1)		CD <sub>2</sub> (±0.3)	CD <sub>3</sub> (±0.1)	CSA (±2)	
a	Lipid+Cholesterol	46	4.2	-40	
b	Lipid+Cholesterol+A $\beta$ (25–35)	48	4.6	-41	
с	Lipid+Cholesterol+A $\beta$ (25–35)+Cu <sup>2+</sup>	49	4.9	-33	
d	Lipid+Cholesterol+A $\beta$ (25–35)+Zn <sup>2+</sup>	49	4.4	-40	
e	Lipid+Cholesterol+Cu <sup>2+</sup>	49	5.0	-41	
f	Lipid+Cholesterol+Zn <sup>2+</sup>	49	4.8	-43	

 $CD_2$  and  $CD_3$  quadrupolar splittings ( $\Delta\nu_Q)$  are taken from the 90° edges of the powder pattern.

### 3.2. NMR studies of $A\beta(25-35)$ with brain lipids

Since studies have shown a preference for AB(1-42) to interact with charged phospholipids [39,40], total extract and the polar extract of brain lipids were used to probe the effect of lipid composition with respect to peptide interaction and membrane stability. The experiments carried out using natural brain extracts were limited to <sup>31</sup>P NMR experiments and did not include <sup>2</sup>H NMR experiments due to the absence of deuterium labels in the lipid acyl chains. The results indicated that no significant changes were seen in the <sup>31</sup>P line shape upon addition of  $A\beta(25-35)$  to either total brain lipid extract and the polar lipid extract. While subsequent addition of Zn<sup>2+</sup> also had no effect,  $Cu^{2+}$  ions added after the association of AB(25-35) caused a decrease in bulk <sup>31</sup>P CSA together with the appearance of an isotropic peak in the polar brain lipid extract (as seen in our earlier spectra [20]). Since the presence of  $Cu^{2+}$  did not induce membrane disruption for the total brain lipid extract but did for the polar extract, it is evident that lipid composition is critical in maintaining membrane integrity in the presence of A $\beta$ (25–35) and Cu<sup>2+</sup> ions.

### 3.3. CP-MAS NMR of membrane-associated $^{13}C$ specifically labelled $A\beta(25-35)$

Structure-related information was obtained for A $\beta$ (25–35) in a phospholipid bilayer using peptide specifically labelled in the middle of the sequence and near the C-terminus. <sup>13</sup>C spectra of labelled A $\beta$ (25–35) peptide as a dry powder prior to incorporation or association of the peptide into model lipid membranes are shown in Fig. 2a. Peptide structural changes as suggested by changes in carbon chemical shift were compared with reference to polypeptide conformational studies of <sup>13</sup>C carbonyl and C( $\alpha$ ) chemical shifts using solid-state NMR [41,42]. The chemical shift of specific labelled sites of the incorporated peptide (Fig. 2b) when compared to the labelled A $\beta$ (25–35) peptide powder indicated an upfield shift for the



Fig. 2. <sup>13</sup>C-CPMAS NMR spectra of  $A\beta(25-35)$ : (a) dry peptide powder, (b) incorporated in PC/PS lipid bilayers, and (c) associated with PC/PS lipid bilayers.

carbonyl resonances (B-sheet, [41,42]) of Ala30 and Leu34 (from 174 to 173 ppm) and no significant change for Gly29 (43 ppm) from the initial random coil shift [41,42] of the dry peptide. Associated peptide (Fig. 2c) had a similar effect when added to the PC/PS membrane with a slightly more upfield shift for the carbonyls (171 ppm) when compared to the incorporated peptide. Although carbon chemical shifts were similar for both the associated and incorporated peptides,  $T_2$ measurements of the <sup>31</sup>P headgroups indicated that the phospholipid headgroup interactions for the associated species was stronger, as observed by the reduction in  $T_2$  relaxation time from  $\sim 6.2$  ms (lipid alone) to  $\sim 1.8$  ms, whereas incorporated A $\beta$ (25–35) gave a  $T_2$  of ~7 ms similar to that of pure lipids. As a result of the anisotropic nature of  $T_2$  [37,38], MAS techniques were used to measure  $T_2$  in order to determine the gross effect of the peptide on the phospholipids rather than a detailed study, which would require measurement of that anisotropy to explain the lineshape change. Interestingly, only one peak was seen by <sup>31</sup>P, suggesting that the two phospholipids were in a similar environment, although the linewidth as a result of the short NMR  $T_2$  would preclude resolving the resonances. Incorporated A $\beta$ (25–35) also caused a significant increase in the <sup>2</sup>H quadrupolar splittings and the <sup>31</sup>P CSA, indicating perturbation of both the acyl region as well as the phosphorus headgroup of the bilayer. The <sup>2</sup>H acyl chain order parameter from dePaked spectra showed an increase in order at the CD<sub>2</sub> positions nearer to the lipid headgroups with less effect in the CD<sub>2</sub> position toward the end of the chain and the terminal CD<sub>3</sub> when the peptide was incorporated in lipid vesicles compared to the associated species (Fig. 3).

### 3.4. CP-MAS NMR of ${}^{13}C$ specifically labelled $A\beta(1-42)$

Structure-related information was obtained for  $A\beta(1-42)$  in phospholipid bilayers. <sup>13</sup>C chemical shifts of labelled  $A\beta(1-42)$  were obtained as: (i) a dry powder, (ii) when incorporated in lipid bilayers, and (iii) when associated with lipid bilayers. The  $A\beta(1-42)$  peptide was labelled at the C-terminal end region, at the same residues as for the  $A\beta(25-35)$  fragment used in the



Fig. 3. <sup>2</sup>H order parameter plot of dePaked spectra from deuterated acyl chains of *d*POPC/POPS: (i) -0-lipid order profile alone; (ii)  $-\mathbf{n}$ -lipid order profile with incorporated A $\beta$ (25–35); and (iii)  $-\mathbf{o}$ -lipid order profile with associated A $\beta$ (25–35). The order parameters are customarily assumed to correspond to the carbon position, with the widest splitting being assigned to C-2 and the narrowest to the terminal methyl.

studies. Chemical shift comparison of both the incorporated and the associated peptide relative to the peptide as a dry powder (HFIP pre-treated as discussed in Materials and methods) indicated an upfield shift, consistent with a change from  $\alpha$ helical to  $\beta$ -sheet for the carbonyl resonances (Fig. 4, Table 3) [41,42].

From our experiments, the dry  $A\beta(1-42)$  peptide displayed β-sheet propensity at Gly29 and helical propensity at both Ala30 and Leu34. We were unable to resolve Gly29  $C(\alpha)$ chemical shift due to overlapping acyl-lipid peaks when the peptide was in the presence of phospholipids. However, the  $^{13}C=O$  resonances shifted from 178 ppm, indicative of  $\alpha$ helical structure [41,42], to 173 ppm (for incorporated) and 171 ppm (for associated), indicative of  $\beta$ -sheet structure [41,42], for the A $\beta$ (1–42) peptide with POPC/POPS bilayers (Fig. 4, Table 3). The changes in <sup>13</sup>C chemical shift as seen by CP-MAS experiments upon association of  $A\beta(1-42)$  with the lipid bilayer suggest the peptide adopted a  $\beta$ -sheet conformation [41,42] toward the C-terminus where the labelled residues (29, 30 and 34) are located. This result is complementary to circular dichroism (CD) data published in earlier studies [20], where an increase in  $\beta$ -sheet conformation was also observed when  $A\beta(1-42)$  was associated with membrane vesicles. The carbonyl resonances of associated peptide have a higher upfield shift than the incorporated peptide, which may suggest A $\beta$ (1-42) adopts a more  $\beta$ -sheet-like structure when associated.

### 4. Discussion

The importance of metal-peptide interactions mediated through binding to histidine residues and promoting structural changes for both full-length  $A\beta(1-40)$  and  $A\beta(1-42)$  has been correlated to the neurotoxic properties of  $A\beta$  peptides. However, the shorter A $\beta$ (25–35) peptide fragment also exhibits neurotoxicity, but does not contain the metal coordinating histidine residues. This suggests AB can exhibit neurotoxicity through a variety of alternative pathways [16]. Lipid composition of the membrane has been shown to be important for seeding peptide aggregation, with  $A\beta$  appearing to bind preferentially to negatively charged over neutral lipids. Lipid charge distribution has been shown to be involved in signalling cell death, for example, phosphatidylserine (PS) lipid translocation from the inner to the outer membrane leaflet has been shown to occur as an apoptotic trigger in cells [43]. Also, aggregation of the peptide is affected when membrane fluidity or order is altered, for example, in the presence of cholesterol. We have examined the effect of associated or incorporated A $\beta$ (25–35) in lipid membranes of different fluidity and changes in peptide structure when added to lipid membranes using solidstate NMR. For comparison to our earlier studies of  $A\beta(1-42)$ in phospholipid bilayers [20], we used a L/P ratio of 30:1. AB(25-35) structural change was subsequently compared to full-length  $A\beta(1-42)$  in phospholipid membranes. Associated peptide could represent 'soluble' AB interacting with a membrane surface, whereas incorporated peptide could represent A $\beta$  already within a membrane bilayer.



Fig. 4. <sup>13</sup>C-CPMAS NMR spectra of labelled  $A\beta(1-42)$  associated with *d*POPC/POPS bilayer. The  $A\beta(1-42)$  sequence is given with \*(C $\alpha$ ) and <sup>#</sup>(C=O) indicating residue label position.

## 4.1. Effect of associated and incorporated $A\beta(25-35)$ and metals on membrane bilayer

Our observations of the effect of  $Zn^{2+}$  and  $Cu^{2+}$  ions on the membrane integrity showed the formation of smaller lipid structures indicative of membrane disruption by the ions [9]. Metal coordination of these ions to the histidine residues of the full-length  $A\beta(1-42)$  peptide has been shown to be important for  $A\beta$  toxicity where a decrease in peptide toxicity was seen using methylated histidine mutant peptides that bind metals with an altered coordination mode compared to wild-type peptide [6]. The A $\beta$ (25–35) fragment contains no histidine residues, thus excluding obvious potential for direct metal binding, although the effect of AB(25-35) and metal ions on bilaver integrity could still be studied. Our data suggest that  $A\beta(25-35)$  inserts less deeply into the bilayer when associated with phosphatidylcholine/phosphatidylserine (PC/PS) membranes as a dry peptide film than when incorporating peptide during vesicle formation. The association with peptide results in a decrease in <sup>2</sup>H NMR quadrupolar splitting, increase in <sup>31</sup>P CSA and a significant decrease in <sup>31</sup>P  $T_2$  relaxation time from 6.2 to 1.8 ms, which indicates decreases in lipid acyl chain order and direct perturbations of phospholipids headgroup consistent with peptide located more at the interface between the membrane and aqueous solution. Previous neutron diffraction studies of

Table 3

Comparison of <sup>13</sup>C-CPMAS chemical shifts of  $A\beta(1-42)$  specifically enriched at <sup>13</sup>C $\alpha$  of Gly29 and <sup>13</sup>C=O of Ala30 and Leu34 as a dry peptide powder and with *d*POPC/POPS lipid bilayers at 1:30 peptide/lipid ratio in Tris-HCl pH=7.2 at 28 °C.

	<sup>13</sup> Cα (ppm)	<sup>13</sup> C=O (ppm)
$A\beta(1-42)$ (dry powder)	44	178
Incorporated $A\beta(1-42)$ in lipid	43 (broad peak)	173
Associated $A\beta(1-42)$ in lipid	43 (broad peak)	171

similarly associated A $\beta$ (25–35) [44–46] also indicate at least a partial preference for the more hydrophobic membrane environment over the aqueous membrane interface, and positions the penultimate Leu34 residue ~15 Å from the centre of the ~30 Å hydrophobic distance between glycerol backbones [47]. These results together with our present study indicate the A $\beta$ (25–35) is partially inserted into the bilayer, while still interacting with the <sup>31</sup>P headgroups. Addition of Cu<sup>2+</sup> ions subsequent to peptide addition only affected the <sup>31</sup>P headgroups and reduced the <sup>31</sup>P CSA with an increase in  $T_2$  broadening, consistent with paramagnetic relaxation. Addition of Zn<sup>2+</sup> after A $\beta$ (25–35) peptide did not lead to significant changes.

In contrast, incorporated A $\beta$ (25–35) increased <sup>2</sup>H NMR quadrupolar splittings indicating an increase in lipid acyl chain order akin to the effect of cholesterol [23], and had less effect than associate peptide on <sup>31</sup>P CSA and no significant impact on <sup>31</sup>P  $T_2$  relaxation time. These results indicate that A $\beta$ (25–35) was more deeply inserted in the membrane.

### 4.2. Effect of cholesterol on $A\beta(25-35)$ insertion into membrane bilayer

The presence of cholesterol in a membrane bilayer orders the lipid acyl region while 'disordering' the headgroup due to incorporation of the sterol into the membrane. Cholesterol increased the lipid <sup>2</sup>H quadrupolar splitting of the chains and decreased the <sup>31</sup>P CSA of the headgroup as seen by Davis et al. [36]. The effect of A $\beta$ (25–35) on the lipid bilayer (Table 1c) was reduced by the presence of cholesterol (Table 2b), which suggested that cholesterol decreases membrane fluidity, the increase in lipid acyl-chain order may reduce peptide insertion. Cholesterol also tends to preserve membrane integrity and, unsurprisingly, the effect of A $\beta$ (25–35) on both the total brain lipid extract (with cholesterol) and the polar brain lipid extract

(without cholesterol) was different. For the total brain lipid extract there was no significant perturbation of the <sup>31</sup>P lipid headgroup while for the polar brain extract, there was a significant increase in <sup>31</sup>P isotropic phase with the addition of Cu<sup>2+</sup> ions. The appearance of this isotropic phase, representative of smaller lipid structures as seen in earlier studies [20] indicated that the membrane integrity was destabilized for the more polar lipid in the presence of A $\beta$ (25–35) and Cu<sup>2+</sup> ions. These data suggest that, in the presence of A $\beta$ , cholesterol may have a stabilizing role on membrane structure.

### 4.3. Structural characteristics of membrane-associated and -incorporated $A\beta$

CP-MAS data using specifically <sup>13</sup>C labelled peptide provides information on the peptide structure in association with lipid vesicles. The data indicated that both Ala30 and Leu34 carbonyl carbons had more  $\beta$ -sheet structure than the peptide powder, more so for associated than incorporated A $\beta$ (25–35) in PC/PS bilayers, but with little such differential effect at the  $\alpha$ -carbon of Gly29 in both cases. The interaction of the native Alzheimer's peptide C-terminal fragment A $\beta$ (29–42) with neutral phospholipid bilayers by solid-state NMR has been reported [48]. The results for this longer peptide also support a  $\beta$ -sheet oligomeric association of the peptide at the bilayer interface. Interestingly, although we expected the hydrophobic transmembrane region of  $A\beta(1-42)$  to exhibit greater preference for helical structure upon membrane interaction, the CD studies reported earlier [20] together with our <sup>13</sup>C CP-MAS results from specific labels in the hydrophobic region of AB(1-42) showed a change from  $\alpha$ helical as a dry powder to  $\beta$ -sheet structure when both associated and incorporated with phospholipid vesicles. Although not in association with lipid, in amyloidic fibril form, and significantly two residues shorter, the structural model from Tycko and coworkers based on solid-state NMR data [5] also show  $\beta$ -strand conformations for residues 12-24 and 30-40 of AB(1-40).

### 4.4. Comparison of $A\beta(25-35)$ with $A\beta(1-42)$

We have shown previously that  $A\beta(1-42)$  behaves similarly to our present observation on  $A\beta(25-35)$  when associated with phospholipid bilayers, as it binds to the phosphorus headgroup causing a decrease in spin-spin  $(T_2)$  relaxation time; phospholipid headgroup binding of  $A\beta(1-42)$  was further confirmed by <sup>31</sup>P magic angle spinning SS-NMR with the appearance of a lipid–A $\beta$  bound phase together with a peptide-free lipid phase in large unilamellar vesicles [49]. This commonality suggests that the A $\beta$ (25–35) region within the full-length peptide may be involved in phospholipid interaction. Incorporation of  $A\beta(1-42)$ and  $A\beta(25-35)$  in phospholipid membranes, however, yielded diverse results. A $\beta$ (1–42) when incorporated destabilizes the bilayer, as reported previously [20], while  $A\beta(25-35)$  inserted into the hydrophobic acyl-chain region of the phospholipid membrane and enhanced membrane stability, as indicated by the absence of an isotropic peak in the <sup>31</sup>P NMR spectra.

Although more membrane associated  $A\beta$  peptide structural detail is necessary before probable toxicity mechanisms can be

developed, the weaker but similar shift toward  $\beta$ -sheet structure of A $\beta$ (25–35) as A $\beta$ (1–42), together with evidence of headgroup interactions, may reflect partial insertion of the shorter peptide and aggregation of the full-length peptide largely on the membrane surface, respectively [21], suggesting that the toxic form of A $\beta$  may reside on the bilayer surface with  $\beta$ -sheet structural propensity driven somewhat by the A $\beta$ (25–35) fragment.

### 5. Conclusion

AB(25–35) appeared to promote  $Cu^{2+}$  destabilization of the membrane of polar lipid brain extracts. Although metal ions could adhere non-specifically to amyloid oligomers, our earlier NMR results [20] also showed that  $Cu^{2+}$  alone but not  $Zn^{2+}$  destabilized PC/PS bilayers without peptide. A $\beta$ (25–35), if inserted through to the hydrocarbon-water interface of the bilayer, could allow more accessibility for Cu<sup>2+</sup> binding for the polar lipid extract than for the less fluid or more ordered membrane system, such as total brain lipid extract with cholesterol. Structural comparison of  $A\beta(1-42)$  and  $A\beta(25-35)$  at similar residues suggested that A $\beta$ (1–42) undergoes an increase in  $\beta$ -sheet structure in the Cterminal region while  $A\beta(25-35)$  maintained its conformation in the presence of phospholipids. Since both peptides are cytotoxic to cell cultures, the conformational change towards  $\beta$ -sheet may be a crucial change required for  $A\beta(1-42)$  to exhibit cytotoxicity. Interestingly,  $A\beta(25-35)$  has been shown to mediate the effects of the full-length peptide and to be homologous to the tachykinin neuropeptide family [50]. Our previous studies showed alterations to the metal coordination by AB histidine residues attenuated peptide toxicity [6]. It seems  $A\beta(25-35)$ , which lacks the histidine residues required for metal coordination, adopts a different toxic pathway that does not involve metal coordination [51]. This pathway can also be engaged by  $A\beta(1-$ 42) as an alternative to metal dependent toxicity since  $A\beta(25-35)$ is embedded in the A $\beta$ (1–42) sequence. However, the methionine is C-terminal in A $\beta$ (25–35), which may play a crucial role [51,52] and, intriguingly,  $A\beta(25-36)$  with one additional residue is not toxic to cultured neurons [51]. If the cytotoxic pathway of A $\beta$ (25–35) represents an alternate toxic pathway for the fulllength peptide, and the composition of the membrane plays a role in the advancement of dementia, then the disease could be amplified by a deficiency in maintaining 'normal' lipid composition crucial for preserving membrane order or fluidity.

### Acknowledgements

T.L.L. thanks the University of Melbourne for award of a PhD scholarship and David Hays Postgraduate Writing-up Award. K.J.B. is funded by the NHMRC. F.S. is grateful to the ARC for financial support (grant # DP0664601).

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