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# Laser Treatment of Cu-Mo Electro-Spark Deposited Coatings

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## Abstract

The paper described properties of electro-spark deposited coatings under influence of the laser treatment process. The properties were assessed by analyzing the coating microstructure, microhardness and corrosion resistance tests and tribological studies. The tests were conducted for Mo and Cu coatings (the anode) which were electro-spark deposited over the C45 steel substrate (the cathode) and melted with a laser beam. The coatings were deposited by means of an ELFA-541. The laser processing was performed with an Nd:YAG laser. The coatings after laser processing are still distinguished by very good performance properties, which make them suitable for use in sliding friction pairs.

*Keywords:* electro-spark deposition; laser treatment; coating

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## 1. Introduction

A number of modern surface processing methods use an energy flux. The examples include electro-spark deposition and laser treatment. Electro-spark deposition (ESD) is a cheap high-energy process. Developed in the post-war period, the technology has been frequently modified. The main advantages are related to the ability of precisely selection of the area to be modified and the ability of the coating thickness selection. In this case it can be determined range from several to several dozen micrometers, good adhesion of a coating to the substrate, and finally, cheap and simple equipment for coating deposition.

The processes of coating formation on metal parts including electro-spark deposition involve mass and energy transport accompanied by chemical, electrochemical and electrothermal reactions.

Today, different electro-spark deposition techniques are used; they are suitable for coating formation and surface microgeometry formation [1–5]. Electro-spark alloying is becoming more and more popular as a surface processing technology. Electro-spark deposited coatings are frequently applied in industry, for example, to produce implants or cutting tool inserts. The coatings are deposited with manually operated equipment or robotized systems.

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Obviously electro-spark deposited coatings have also some disadvantages, however they can be easily eliminated. One of the methods is proposed laser treatment. The laser beam is used for surface polishing, surface geometry formation, surface sealing or for homogenizing the chemical composition of the coatings deposited [6–9].

Laser treatment of the electro-spark deposited Cu-Mo coatings was the object of perform studies. The properties were established base on the results of a microstructure analysis, microhardness and corrosion resistance tests and tribological studies.

## 2. Experimental

The tests were conducted for Cu-Mo coatings produced by electro-spark deposition, which involved applying Cu and Mo electrodes with a diameter of 1 mm (the anode) on the C45 steel substrate (the cathode). Here copper constitutes the core coating material in the formation of low-friction surface layers; it also compensates for the occurrence of residual stresses. Molybdenum act as the reinforcing constituents. The coating materials, i.e. molybdenum (99.8% Mo) and copper (99.2% Cu) in the form of wire ( $\phi = 1$  mm) were purchased from BIBUS Metals Sp. z.o.o. (certificate included).

The coatings were electro-spark deposited on C45 steel substrate by means of the ELFA-541 made by a Bulgarian manufacturer. Base on the analyses of the current characteristics as well as the manufacturer's recommendations, it was assumed that the parameters of the ESD operation should be as follows: current intensity  $I = 16$  A (for Cu  $I = 8$  A); table shift rate  $V = 0.5$  mm/s; rotational speed of the head with electrode  $n = 4200$  rev/min; number of coating passes  $L = 2$ ; capacity of condenser system  $C = 0.47$   $\mu$ F; pulse duration  $T_1 = 8$   $\mu$ s; interpulse period  $T_p = 32$   $\mu$ s; frequency  $f = 25$  kHz.

The subsequent laser treatment was performed with the aid of a BLS 720 laser system employing the Nd:YAG laser operating in the pulse mode. The following parameters were assumed for the laser treatment: laser spot diameter  $d = 0.7$  mm; laser power  $P = 20$  W; beam shift rate  $v = 250$  mm/min; nozzle-sample distance  $h = 1$  mm; pulse duration  $t_i = 0.4$  ms; frequency  $f = 50$  Hz.

## 3. Results and Discussion

### 3.1. Analysis of coating morphology

A Joel JSM-5400 scanning microscope equipped with an Oxford Instruments ISIS-300 X-ray microanalyzer was used to test the coating microstructure.

In Figure 1a the microstructure of electro-spark deposited two-layer Cu-Mo coating is presented. The layer thickness is approximately 8–10  $\mu$ m, and the range of the heat affected zone (HAZ) inside the (underlying) substrate material is about 10–15  $\mu$ m. In the photograph, the boundary line between the two-layer coating and the substrate is clearly visible. There are microcracks running across and along the coating. A linear analysis of the elements (Figure 1b) of the Cu-Mo coating shows that the distribution of elements is non-uniform; there are zones with greater concentrations of Cu, Mo and Fe. Analyzing the linear distribution of elements, one can see that the adhesion of the coating to the substrate is of diffusive type. There is no clear separation of components either in the Cu-Mo coating (Figure 1b). A higher content of carbon reported in the electro-spark deposited Cu-Mo coating is a result of ascending diffusion. Carbon from the C45 steel substrate travels to the electro-spark deposited technological surface layer (TSL) due to thermal interaction

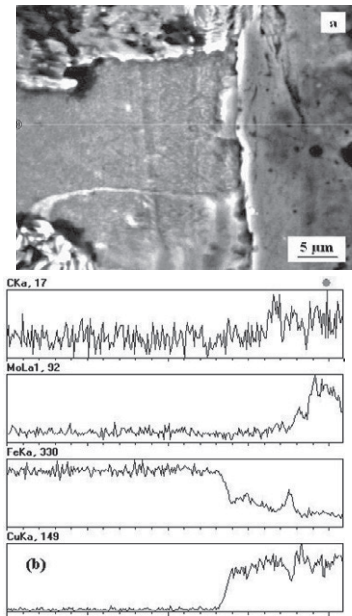


Figure 1. (a) Microstructure and linear distribution of elements; (b) in the Cu-Mo coating

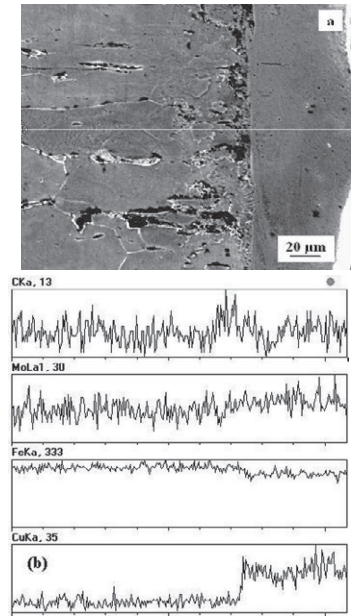


Figure 2. (a) Microstructure and linear distribution of elements; (b) in the Cu-Mo coating after laser treatment

The melting and solidifying processes during laser treatment resulted in the migration of elements across the coating-substrate interface. Laser radiation caused intensive convective flow of the liquid material in the melt pool and, in consequence, the homogenization of the chemical composition (Figure 2b). It also led to the structure refinement and highly saturated phase crystallization (Figure 2a) due to considerable gradients of temperature and high cooling rates. The technological surface layers, produced by laser alloying, were free from microcracks and pores (an effect of surface sealing), and non-continuities across the coating-substrate interface. It wasn't observed any significant change in the chemical composition of the substrate. The thickness of the fused two-layer of the Cu-Mo coating was in the 20–40  $\mu\text{m}$  range. In the heat affected zone (HAZ), which was 20–50  $\mu\text{m}$  thick, there was an increase in the content of carbon (Figs. 2b).

### 3.2. Microhardness tests

The microhardness was determined by using the Vickers method (Hanemann tester). The measurements were performed under a load of 40 G. The indentations were made in perpendicular microsections in three zones: the white homogeneous difficult-to-etch coating, the heat affected zone (HAZ) and the substrate. The test results for the electro-spark deposited Cu-Mo coating before and after laser treatment are shown in Tables 1 and 2. Electro-spark deposition caused changes in the microhardness of the material. The microhardness of the substrate after electro-spark deposition was on average 281  $\text{HV}_{0.04}$ ; the same value was reported for the substrate before the process. There was a considerable increase in microhardness after depositing the heterogeneous Cu-Mo coating. The microhardness of the Cu-Mo coating was approx. 587  $\text{HV}_{0.04}$ , which gives increase of 110 %. The microhardness of the Cu-Mo coating in the heat affected zone (HAZ) after electro-spark treatment was 51 % higher than that of the substrate material. Laser treatment had a favorable effect on the changes in the microhardness of the electro-spark deposited of the Cu-Mo coating. There was an increase of 161 % in the microhardness of the Cu-Mo coating.

Table 1. Results of the microhardness tests for the Cu-Mo coating before laser treatment

Measured zones	Microhardness HV <sub>0,04</sub>			Mean value HV <sub>0,04</sub>
	Measurement number			
	1	2	3	
<b>Coating</b>	566	606	589	587
<b>HAZ</b>	428	437	401	422
<b>Substrate</b>	282	285	276	281

Table 2. Results of the microhardness tests for the Cu-Mo coating after laser treatment

Measured zones	Microhardness HV <sub>0,04</sub>			Mean value HV <sub>0,04</sub>
	Measurement number			
	1	2	3	
<b>Coating</b>	714	742	734	730
<b>HAZ</b>	583	612	578	591
<b>Substrate</b>	289	274	280	281

### 3.3. Tribological tests

The friction coefficient of coatings was studied at the Laboratory of Tribology, Kielce University of Technology. The coefficient of friction for Cu-Mo coating before and after laser treatment was determined using a T-01M (pin on disc type tribological tester). The tester enables continuous measurement of the friction force at a set load. The pin of  $\phi 4 \times 20$  mm was made of medium-carbon steel with a hardness of 27 HRC. The testing was performed at the following parameters: load  $Q = 10$  N, rotational speed  $n = 382$  rpm, test duration  $t = 500$  s.

In Figure 3 friction coefficient in the function of time at a load of 10N is presented. This diagram shows the Cu-Mo coating before and after modification with a laser beam. Dry friction observed in the case of the coatings resulted in the transformation of the outer layer into a surface layer. This was mainly due to the sliding stresses and speed, and the interaction with the medium. The state stabilization of the antiwear surface layer was observed. The stabilization of the friction coefficient was achieved after 80 sec. within value fluctuating at  $0.16 \div 0.18$ . In the case of a laser modified Cu-Mo coating, the friction coefficient stabilizes after 240 sec., and its value fluctuates at  $0.35 \div 0.37$ . The average friction coefficient of a Cu-Mo coating is lower than that of a laser-modified Cu-Mo coating (at the moment of stabilization).

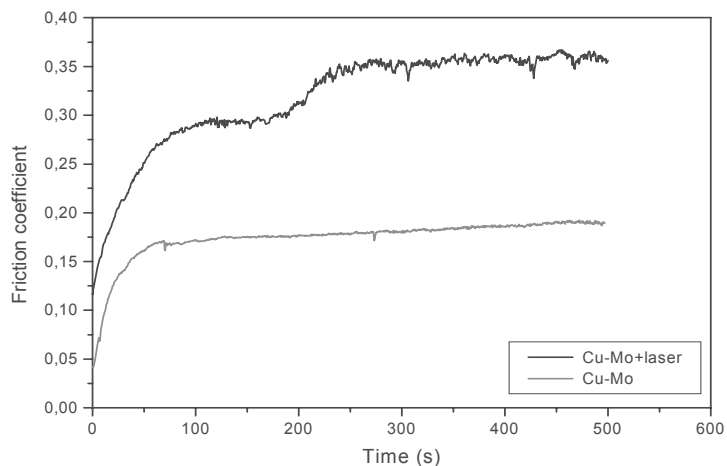


Figure 3. Relationship between friction coefficient and time

Seizure resistance was measured using a T-09 pin on disc tribotester. Two prisms and a roller constituted the friction pair. The surfaces tested were Cu-Mo coatings and C45 steel before and after laser processing. The roller had a diameter of 6.3 mm was made of carbon steel. The tests were conducted for three kinematic pairs for each material variant, so it was possible to average the data.

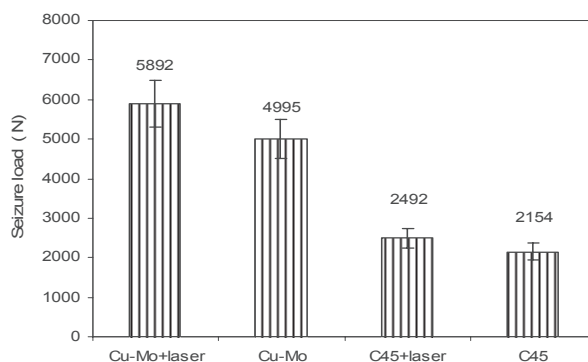


Figure 4. Average seizure load

The experiment involved submerging the specimens in pure lubricant. In this case paraffin oil was used. This oil guarantee constant lubricating ability in each carried out test. It's difficult to compare results, when lubricant oil with additional oilness improvers is applied. Some of improvers are biodegradal and changing oil properties even during short time of storage. To keeping stable and comparable results parafin oil is common use in laboratory's wearing tests. In Figure 4 an average seizure loads before and after laser processing is presented. It is clear that the laser processing operation caused an increase in the load force and, accordingly, seizure of the electrosark deposited coatings and C45 steel.

### 3.4. Corrosion resistance tests

The corrosion resistance of the Cu-Mo coating and the underlying substrate before and after laser treatment was analyzed using a computerized system for electrochemical tests, Atlas'99, produced by Atlas-Sollich. The potentiodynamic method was applied, because it is reported to be one of the most effective methods of electrochemical testing.

The cathode polarization curve and the anode polarization curve were determined by polarizing the samples with a potential shift rate of 0.2 mV/s in the range of  $\pm 200$  mV of the corrosive potential, and with 0.4 mV/s in the range of higher potentials. Samples with a marked area of 10 mm in diameter were polarized up to a potential of 500 mV. The polarization curves were drawn for samples exposed for 24 hours to a 3.5% NaCl solution so that the corrosive potential could be established. The tests were performed at a room temperature of 21°C ( $\pm 1^\circ\text{C}$ ).

Table 3. Current density and corrosion potential of the materials tested

Material	Corrosion current density $I_k$ [ $\mu\text{A}/\text{cm}^2$ ]	Corrosion potential $E_{\text{KOR}}$ [mV]
<b>C45</b>	112 $\pm$ 17.8%	-458
<b>C45+laser</b>	86.4 $\pm$ 16%	-522
<b>Cu-Mo</b>	42.9 $\pm$ 11.8%	-620
<b>Cu-Mo+Laser</b>	30.7 $\pm$ 2.6%	-629

Figure 5 shows example diagrams of the polarization curves of the surface layers. The characteristic electrochemical values of the materials under test are presented in Table 3. The electro-spark deposited coatings were reported to have similar corrosion resistance to that of the substrate material. A system with a two-layer

coating is assumed to fulfill two functions: increase corrosion resistance and wear resistance. The coatings which contained Cu acted as cathodes. Resistance to wear and corrosion depends on the quality of coatings, particularly their sealing properties.

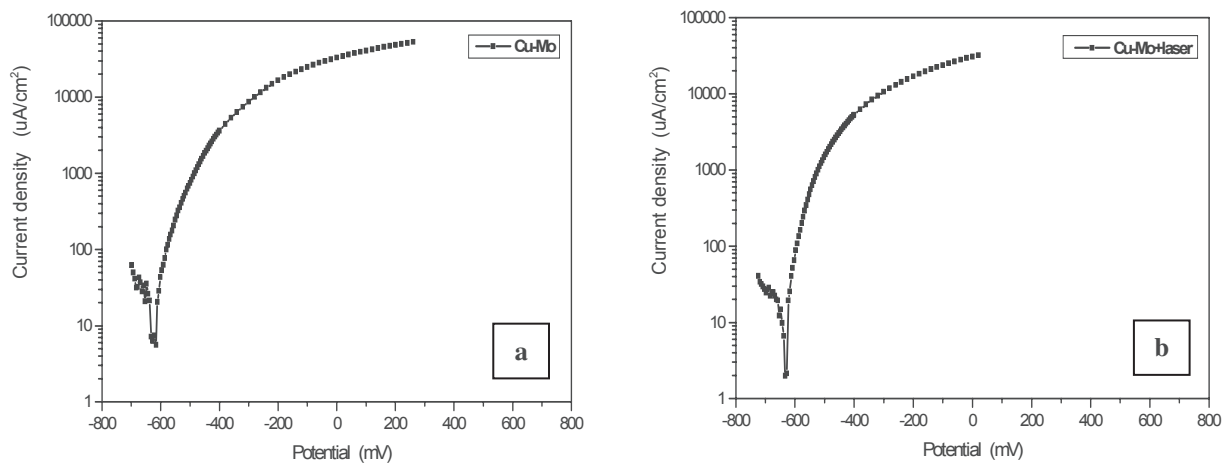


Figure 5. Curves of the Cu-Mo coating polarization: (a) before laser treatment; (b) after laser treatment

The Cu-Mo coating was reported to have the highest corrosion resistance. The corrosion current density of the coating was  $42.9 \mu\text{A}/\text{cm}^2$ , while that of the C45 steel substrate was  $112 \mu\text{A}/\text{cm}^2$ . Applying the Cu-Mo coating improved the sample corrosion resistance by approx. 162%. There was some improvement in the corrosion resistance of the electro-spark deposited coatings after laser treatment. The healing of microcracks resulted in higher density and therefore better sealing properties. The highest corrosion resistance after laser treatment was reported for the Cu-Mo coating ( $I_k=30.7 \mu\text{A}/\text{cm}^2$ ). For the C45 steel substrate,  $I_k$  was  $6.4 \mu\text{A}/\text{cm}^2$ . Thus, the corrosion resistance increased by about 30 % after laser treatment. Laser treatment improved the surface smoothness and corrosion resistance; there was a decrease in the surface roughness,  $R_a$ , from  $2.02 \mu\text{m}$  to  $1.75 \mu\text{m}$ .

#### 4. Conclusion

The following conclusions can be drawn from the analysis and test results:

- A concentrated laser beam can effectively modify the state of the surface layer, i.e. the functional properties of electro-spark coatings
- There is no change in the chemical composition of electro-spark deposited coatings after laser treatment in spite of their melting and solidification. The results of laser radiation are the homogenization of the chemical composition, structure refinement and the healing of microcracks and pores
- The average value of the friction coefficient (at the moment of stabilization) obtained during the tribological tests for a Cu-Mo coating is approximately 54% lower than that obtained for the same coating after laser modification
- After laser processing significant load force can be applied. The material seizure of the Cu-Mo coatings modified after laser treatment was increase about 18%
- Laser treatment caused a 20% increase in the microhardness of the electrospark deposited Cu-Mo coatings
- Laser radiation causes an improvement in the functional properties of the two-layer electro-spark deposited Cu-Mo coating, i.e. they exhibit higher resistance to corrosion
- In the next phase of the research, it is essential to determine the phase composition and porosity of the coatings before and after laser treatment.

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