

of the E-F-hand family of Calcium-sensing proteins. As shown in numerous studies, *in vitro* protein refolding can differ substantially from biosynthetic protein folding which already takes place co-translationally (1). One challenge is to characterize the adopted conformations of nascent chains before being released from the ribosome (2). CaM RNCs of full-length, half-length and of a structure consisting of the first E-F-hand only were synthesized *in vitro*. All constructs contained a tetra-cysteine motif site-specifically incorporated in the first N-terminal helix which is known to react with FLaSH, a biarsenic fluorescein derivative (3). As the dye is rotationally locked to this helix, the TRA decays should directly report on the rotational mobility of the investigated polypeptide chain. To investigate the scope of this procedure we analyzed TRA data of the different protein constructs free in solution as well as CaM with/without Calcium added. This enabled us to determine rotational correlation times, to choose suitable rotational diffusion models which fit the experimental data and thereby yield information about the conformational state and flexibility of the respective constructs. The feasibility of our approach to characterize structural properties and folding states of ribosome tethered, and released, polypeptide chains is finally discussed.

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### 3398-Pos Board B126

#### Refined Folding Mechanism of a Helix-turn-helix Motif

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Helix-turn-helix motifs are valuable models for elucidating the protein folding mechanism. The *de novo* designed helix-turn-helix motif  $\alpha\tau\alpha$  has been studied thoroughly in our group by infrared spectroscopy and circular dichroism, applying site-specific isotopic labels and mutations. This preliminary investigation of  $\alpha\tau\alpha$  has provided detailed site-specific information on its folding mechanism, revealing increased stability in the middle of the helices and unfolding starting from the loose turn and the ends of the helices. The hydrophobic core contributes significantly to the stability of this little protein. To get further information on the folding mechanism of  $\alpha\tau\alpha$ , provided by additional independent probes, Förster resonance energy transfer (FRET) and differential scanning calorimetry (DSC) were employed to monitor thermal unfolding. FRET allows for measuring distances within a protein during its denaturation. In  $\alpha\tau\alpha$ , the intrinsic Trp residue in position 2 at the N-terminus and EDANS attached to the C-terminus served as FRET donor and acceptor, respectively, so end-to-end distances could be obtained. Thermodynamic parameters of a protein can be directly measured using DSC. Therefore, these directly obtained data were used to verify preliminary parameters from spectroscopic measurements. Overall, the results from FRET and DSC measurements were compared to the preliminary data on  $\alpha\tau\alpha$  to refine the proposed picture of its folding mechanism.

### 3399-Pos Board B127

#### Evidence for the Formation of Dry and Wet Molten Globules During Unfolding Process of a Small Protein

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The two-state model which is generally used to describe the 'first-order like' transitions of folding and unfolding of proteins is useful in evaluating thermodynamic parameters. However the implication of two-state model, viz. only native and unfolded forms exist is questionable when the complexity of the energy landscape is recognized.

Multiple steps observed during unfolding of a protein can provide information on the forces that maintain the folded structure. Solvation of the protein core determines stability, but it is not clear when such solvation occurs during unfolding. In this study, far-UV CD measurements suggest a simplistic two-state view of the unfolding of barstar, but the use of multiple probes brings out the complexity of the unfolding reaction. Near-UV CD measurements show that unfolding commences with the loosening of tertiary interactions resulting in a native-like intermediate, N\*. FRET measurements show that N\* then expands non-uniformly to form another intermediate, I<sub>E</sub>. Spectral measurements of the single core tryptophan indicate that both N\* and I<sub>E</sub> retain

native-like solvent accessibility of the core, suggesting that they are dry molten globules. Fluorescence quenching measurements suggest that the core then becomes solvated forming a wet molten globule, I<sub>L</sub>, which precedes the unfolded form. Anisotropy decay measurements show that tight packing around tryptophan is lost when I<sub>L</sub> forms. Strikingly, the slowest step is unfolding of the wet molten globule, and involves a solvated transition state.

### 3400-Pos Board B128

#### Volume of Hsp90 Ligand Binding and the Unfolding Phase Diagram as a Function of Pressure and Temperature

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The phenomenon of inhibitor binding to protein plays a major role in drug discovery. A complete thermodynamic picture of protein unfolding and ligand binding processes can only be obtained if both pressure- and temperature-induced protein denaturation techniques are utilized. The mechanism by which high pressure destabilizes proteins is poorly understood, but there is an agreement that pressure is an equally important fundamental thermodynamic variable as temperature. Temperature-induced protein unfolding provides information about enthalpy and entropy while pressure-induced protein unfolding - about volumetric properties of the system.

In this work, we present thermodynamic characterization of heat shock protein 90 (Hsp90) ligand binding and the unfolding phase diagram as a function of pressure and temperature. Hsp90 is an ATPase and a molecular chaperone responsible for the correct folding of client proteins. Hsp90 is essential for tumor progression because it maintains client proteins in their active forms. Inhibition of Hsp90 leads to the simultaneous degradation of these oncogenic proteins. For this reason Hsp90 has become an anticancer drug target. Hsp90 stability, ligand binding volume, enthalpy, entropy and affinity were measured by combining the Pressure shift assay, Thermal shift assay, and Isothermal titration calorimetry. The study has shown that the ligand increased both the melting pressure and melting temperature, and protein-ligand binding affinity may be correlated with binding volumes.

### 3401-Pos Board B129

#### Origin of Enthalpic Depletion Forces

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Solutes preferentially excluded from macromolecules can drive depletion attractions in important biological and colloidal association processes. The established Asakura-Oosawa theory relates depletion forces to the reduction in excluded volume and the ensuing entropy gain upon macromolecular compaction. In line with this theory, cosolute-induced protein stabilization or favored protein association are often described in terms of entropically driven "crowding", a specific manifestation of depletion interactions. In agreement, our recent experiments of peptide folding and supramolecular binding suggest that depletion forces are predominantly entropic for some cosolutes, such as polyethylene glycol polymers. Surprisingly, however, for other solutes such as polyol osmolytes, the main thermodynamic contribution is enthalpic, while the entropic change due to cosolutes can even be unfavorable. To further elucidate the molecular basis of this enthalpic depletion interaction, we have been using both simulations and analytic theory. Monte-Carlo simulations follow the association of two rod "macromolecules" in binary Lennard-Jones solutions. By dissecting the free energy change upon approach of the two macromolecules into the respective enthalpic and entropic components, we find that different cosolutes show distinct contributions to their macromolecular stabilization effect, implying different thermodynamic driving mechanisms. When considering intermolecular interactions beyond hard-cores, not all cosolutes conform to the established model, and even for these simple, nonassociative liquids, depletion forces can be completely enthalpic in nature. We discuss and analyze this newly resolved mechanism for depletion forces that originates from intermolecular interactions and solvent restructuring. Finally, a mean-field theoretical model based on the Flory-Huggins solution theory complements the simulation analysis.

### 3402-Pos Board B130

#### Statistical Mechanical Models for Analyzing the Site-Specific Folding of Helix-turn-helix Motifs

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Isotopically-edited IR spectroscopy can provide detailed site-specific information about the protein folding mechanism. With our equilibrium unfolding studies of two simple helix-turn-helix (hth) proteins using circular dichroism and <sup>13</sup>C isotopically labeled infrared spectroscopy, we incorporated this

experimental data into Ising-like statistical mechanical models to better understand the observed structural and energetic properties of the two proteins. Two variations of the Ising-like model were implemented: the Wato-Saito-Muñoz-Eaton (WSME) model, which can be enumerated exactly using efficient transfer matrix methods, and the Baker-Finkelstein (BF) model using a double-sequence approximation. Model parameters were optimized by simultaneously fitting the complete set of data for the whole protein as well as each helix independently to reflect what was observed through experiments. In order to give a more realistic representation of protein energy, various statistical residue-specific potential matrices were tested as the inter-residue contact energy in the model. We found that different statistical potentials varied in its success to simultaneously fit all the experimental data, however all the residue-specific matrices resulted in an improvement over considering only a single parameter for the contact energy. Both the WSME and BF models were able to reproduce the equilibrium unfolding data when analyzing the hth proteins as a whole, but the WSME model could not correctly predict the folding of only the helix when analyzed independently due to the assumptions of the model. On the other hand, the BF model was capable of reproducing the experimental data for both the whole protein and the independent helices.

### 3403-Pos Board B131

#### Towards a Test of the Aggregation Hypothesis in Huntington's Disease using $\beta$ -Hairpin Enhancing Motifs

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Huntington's Disease (HD), one of ten polyglutamine (polyQ) repeat diseases, is a devastating disorder caused by expansion of a polyQ-encoding CAG repeat from 37 or more in exon1 of the huntingtin (htt) gene. Although HD brains contain polyQ aggregates, and polyQ aggregation rates *in vivo* and *in vitro* increase with repeat length, there is a continued debate about the role of amyloid-like aggregates in HD. Recently we reported that aggregation of chemically synthesized, short (repeat length  $\sim 22$ ), simple polyQ sequences is greatly enhanced with the addition of unnatural amino acids that encourage  $\beta$ -hairpin formation in the aggregation nucleus. Here we ask whether  $\beta$ -hairpin encouraging mutations in a short polyQ version of the htt exon1 peptide also greatly enhance aggregation. We do this while confining our study to mutations that can be introduced during ribosomal synthesis. We show here that a short polyQ sequence containing (a) L-Pro-Gly instead of the previously described D-Pro-Gly and (b) a modified tryptophan zipper motif aggregates much faster than a simple polyQ sequence of similar length. This can be traced to a decrease in the critical nucleus for amyloid formation from a value of  $n^* \approx 4$  for a simple, unbroken Q<sub>23</sub> sequence to  $n^* \approx 1$  for similar length polyQ containing  $\beta$ -hairpin motifs. At the same time, the morphologies, secondary structure structures, and bioactivities of the resulting fibrils from simple and exon1 mimic polyQ were essentially identical. Importantly, incorporating these motifs into short polyQ exon1 analogs produces rapid spontaneous aggregation rates comparable to exon1 peptides with long, disease associated polyQ repeat lengths. Expression of these exon1 analogs in cells now addresses whether even short polyQ htt exon1 can be toxic if its polyQ is redesigned to promote rapid aggregation.

### 3404-Pos Board B132

#### Salt Effects on Folding of a Helical Mini Protein Villin Headpiece Subdomain HP36 Studied by Generalized-Ensemble Simulations

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Additives dissolved in solvent are important factors that affect proteins' stability and/or folding. In this study we investigated effects of salt ions in solvent on folding events of a helical mini protein HP36. Addition of low concentrations of ions should alter electrostatic interactions among charged groups, so that populations for conformational substates of proteins should be changed. Here we compared two data sets of folding simulations of HP36 with explicit water solvent. For efficient sampling of conformational space of the protein, multiconformational replica-exchange method was adopted.

Results of the present analyses suggest that addition of ions reduces the number of nonnative, nonlocal salt bridges in the protein molecule at later stages of folding at room temperature. Especially, nonnative salt bridges between Glu5 and Arg15 and/or another between Asp4 and Lys30 have been kept in the near-native conformations in pure water. Because dehydration of the

hydrophobic core of HP36 is completed only at the latest stage of folding where correct hydrophobic-core packing becomes formed, these salt bridges can prevent folding into the fully native structure of HP36 at room temperature.

### 3405-Pos Board B133

#### Simulation Model of Protein Transport and Stabilization by GroEL/ES Apichart Linhananta.

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In a previous communication (Linhananta et al., *Biophys. J.*, 2011, 100, 459), we reported results of a simulation model of a protein in solvents with protein-solvent contact energy parameter  $\epsilon_{PS}$ , which represents osmolytes ( $\epsilon_{PS} > 0$ ) and denaturants ( $\epsilon_{PS} < 0$ ). Here a model of a three-helix-bundle (THB) protein in solvents is confined in a cylindrical cavity that mimics the GroEL/ES chaperone. The interior is characterized by the protein-wall energy,  $\epsilon_{PW}$ , and solvent-wall energy,  $\epsilon_{SW}$ . Simulations found a substantial increase in the folding temperature from  $T^* = 4.2$  (scaled unit), for THB in vacuum, to  $T^* > 6.0$  for confined THB in osmolytes. The optimum stabilization of the native state is  $T^* = 6.6$ , for THB in osmolytes with  $\epsilon_{PS} = 0.6$ , confined by walls repulsive to THB ( $\epsilon_{PW} = 1.0$ ) and solvents ( $\epsilon_{SW} = 1.0$ ). Weight histogram analysis reveals an entropy-driven stabilization mechanism due to confinement and the osmolytes. The model is generalized to THB and solvents confined in two connected cylindrical segments. The bottom segment represents the GroEL/ES, with the interior sidewall characterized by the parameters  $\epsilon_{PW}$  and  $\epsilon_{SW}$ . The upper segment represents the exterior surrounding the GroEL/ES, with periodic boundary condition on the sidewall, where the protein and solvents can move through the channel connecting the two segments. For neutral solvents ( $\epsilon_{PS} = 0$ ) with a sidewall that is repulsive to solvents ( $\epsilon_{SW} > 0$ ) and attractive to the protein ( $\epsilon_{PW} < 0$ ), the THB protein preferentially distributes in the lower segment that represents the interior of the GroEL/ES. As the temperature increases and the protein denatures, there is an increase in the probability that the protein is found in the GroEL/ES. This highlights the roles of solvents and surface properties in the transport of unfolded proteins into the GroEL/ES.

### 3406-Pos Board B134

#### Non Local Interactions are Essential Elements of the Initiation and Guidance of the Folding Pathway of Proteins

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The rate of protein folding is determined by the rate of passage through the transition state, however major structural transition precede the TSE formation. We hypothesize that few non-local interactions are effective in the early phases of the folding transition prior to the cooperative transition. These interactions loosely stabilize few closed loops which form the folding noncontiguous nucleus, reduce the chain entropy and determine the course of the folding pathway (the "loop hypothesis"). We study the order of formation of secondary structure elements and loop closure during the early phases of the folding of *E. coli* adenylate kinase (AK) by combination of rapid mixing methods and time resolved FRET spectroscopy. We find that at the initiation of folding of the AK molecule two closed loop structures in the CORE domain reach native end to end distance within a millisecond while a third loop (the N terminal loop) is closed on the microsecond time scale. Three representative CORE domain  $\beta$ -strands have non-native end to end distance during the first 15 ms and undergo slow change (3 sec) to native distance. Along the folding pathway of AK the fast closed N terminal loop is reopened and closed again. We conclude that non local interactions are essential factor at the early phases of the folding transition and that the folding of sub-domain elements is context dependent and should be studied in the whole molecule, *in situ*.

### 3407-Pos Board B135

#### Computational Studies of the Formation of Peroxiredoxin Dimers

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The proteins in the ubiquitous peroxiredoxins (Prx) protein family play an important role in redox signaling and antioxidant defense. The biological functions of Prxs are closely related to the formation of their quaternary structures. To understand details of interactions within Prxs and their quaternary structures formations, the disassembly and unfolding processes of 1YEP (chains A and B) and 3DRN were studied as an example through molecular dynamics simulations. Hundreds of four-microsecond-long simulations using a Go-type model show that disassembly and unfolding processes are