

Binding Cooperativity in Phage λ is Not Sufficient to Produce an Effective Switch

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ABSTRACT In the wild-type phage λ , binding of $C1$ to O_{R2} helps polymerase bound to P_{RM} transition from a closed to open complex. Activators on other promoters increase the polymerase–DNA binding energy, or affect both the binding energy and the closed–open transition probability. Using a validated mathematical model, we show that these two modes of upregulation have very different effects on the promoter function. We predict that if $C1_2$ bound to O_{R2} produced equal increase in RNAP–DNA binding constant (compared to wild-type increase in the closed–open transition probability), the lysogen would be significantly less stable.

INTRODUCTION

Although it is premature to assert that we have entered the era of synthetic biology, the groundwork for targeted design of functioning living organisms is being laid. The manipulation of DNA within an organism is, by now, a standard laboratory practice. Recent work has shown the feasibility of complete genome transplantation (1). Thus, the tools exist, but to use them effectively requires the ability to design elements of signal transduction and gene regulatory networks. While this has been done (2,3) much remains to be understood both on the level of the construction of the individual components and the design of the networks themselves. The focus of this article is on the former.

Transcriptional control plays a fundamental role in gene expression. The initiation of transcription involves a series of reactions which can be summarized into three steps (note that there are additional controls which occur in later stages of the process of transcription, but are not considered in this article):

Binding. RNA polymerase binds to promoter DNA yielding a closed RNA polymerase promoter complex.

Opening. RNA polymerase unwinds a short segment of DNA yielding an open RNA polymerase promoter complex.

Escape. After abortive cycles of synthesis and release of short RNA products, the RNA polymerase escapes the promoter and enters into productive synthesis of RNA.

The activation and repression of transcription initiation is primarily caused by regulatory proteins and the structure of DNA. Regulated recruitment (4) provides a conceptual model for this process. Considerable progress has been made in understanding the biochemistry of the various reactions in the process (5,6) and in particular, it is clear that while the

three steps are physically coupled there is considerable freedom for varying the respective energy profiles. To model these steps in the simplest way, we will treat opening and escape as a single chemical reaction with forward-reaction rate k determined by the regulatory proteins and their interaction with the DNA. Binding will be treated as a reversible reaction with an equilibrium constant K_B .

This simplification of the biochemistry allows one to develop thermodynamic models to quantify the rates of transcription initiation (5,7,8) that can be validated against experimental data (9,10). However, the combination of activators, repressors, and the above-mentioned steps implies that control of transcription initiation is a highly nonlinear process, which in turn suggests that systematic mathematical analysis may lead to a deeper understanding of this regulatory mechanism (11). Given the goal of synthetic biology, claims based on the mathematical models must be experimentally verifiable.

More is known about the phage λ machinery than any other gene regulation mechanism (4,12). After infection of *E. coli* the phage λ follows one of two pathways: lysis, where it uses the bacterial molecular machinery to make many viral copies, kills the host bacterium and leaves to infect other cells; or lysogeny, where it integrates its DNA into the bacterial DNA and divides for generations with the bacterium. The lysogen exhibits great stability, yet it induces readily when the bacteria are irradiated with ultraviolet light.

The primary objective of this article is to use the above-mentioned mathematical models to demonstrate that, in the context of the proper functioning of the phage λ induction, the binding constant K_B plays a fundamentally different role from the opening and clearing constant k . In particular, they are not interchangeable; that is, modifications in K_B cannot be directly compensated for by modifications in k and vice versa. To make this argument, we begin with a review of a simplified biological model of the phage λ switch and a precise statement of why increases in K_B are not equivalent to increases in k . After that, we recall and explain the associated

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mathematical model and relate it back to the biology. We validate the model by considering several mutants, where our model recovers experimental observations of the lysogen stability. With this justification, we make several mathematical predictions concerning the unequal role played by RNAP binding versus closed-open complex transition in transcription initiation process. These predictions are, in principle, experimentally testable.

The phage λ switch

The central controlling region for the lysogen maintenance is the right operator O_R , even though the long range cooperative binding with the O_L operator plays a crucial role in stability of the lysogen. For a more complete description of the regulatory mechanisms, the reader is referred to Ptashne (4). O_R has three subregions designated O_{R1} , O_{R2} , and O_{R3} (see Fig. 1). The O_R region also contains two disjoint promoters—right promoter (P_R) and repression maintenance promoter (P_{RM}). The promoter P_R completely overlaps O_{R1} and partially overlaps O_{R2} , while P_{RM} completely overlaps O_{R3} and partially overlaps O_{R2} . The gene *cI* (which codes for the repressor protein CI) and the gene *cro* (which codes for Cro protein) flank the O_R region. Binding of either CI or Cro dimers (CI_2 , Cro_2) to O_{R2} prevents binding of RNA polymerase (RNAP) to P_R , but it does not prevent such binding to P_{RM} . The initiation of transcription of *cro* occurs only if RNAP binds to P_R . Similarly, the initiation of transcription of *cI* occurs only if RNAP binds to P_{RM} .

The lytic pathway corresponds to a state where Cro_2 protein is bound to O_{R3} , blocking the P_{RM} promoter and thus transcription of *cI*. At the same time, RNAP is free to bind P_R , thus maintaining the transcription of *cro*. The lysogenic pathway corresponds to the state of O_R where CI_2 binds to both O_{R2} and O_{R1} , blocking the P_R promoter and hence, the transcription of *cro*. RNAP is free to bind P_{RM} , and thus, maintain the transcription of *cI*. Even though these pathways are stable, the change from lysogeny to lysis, called induction, is experimentally well documented. When the bacterial population was subjected to irradiation by UV light, the phage λ started to lyse the bacteria and emerged in ~45 min. The irradiation causes RecA protein-mediated cleavage of CI, which lowers its effective concentration (4,13–15). There are several key features that make lysogen very stable and the induction switchlike (4):

1. High level of cooperativity between CI molecules: CI forms dimers CI_2 in the solution; when bound to neigh-

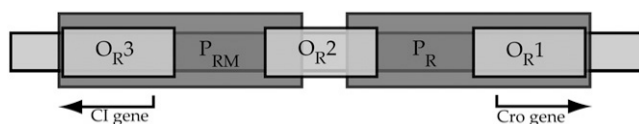


FIGURE 1 O_R region.

2. Cooperative binding of CI_2 to O_{R2} and O_{R1} : binding of CI_2 to O_{R1} facilitates binding of another CI_2 molecule to O_{R2} .
3. Variable binding affinities of CI_2 and Cro_2 to different O_R regions: CI_2 has the highest affinity to O_{R1} , lower for O_{R2} , and lowest for O_{R3} , while Cro_2 has the highest affinity to O_{R3} , lower for O_{R2} and O_{R1} .
4. Cooperative binding of CI_2 to O_{R2} and RNAP at P_{RM} : that is, O_{R2} bound CI_2 increases the forward rate constant k at P_{RM} ~10-fold without having any significant effect on the binding of the RNA polymerase to the DNA (16).

We refer to the cooperativity in feature 4 as k -cooperativity. In an intriguing article, Li et al. (17) have shown that after an Arg-to-His change in the σ -subunit of RNAP, the wild-type CI_2 activates mutant RNAP by increasing K_B . We will refer to this cooperativity as K_B -cooperativity. This suggests that mutations allowing for an increase in K_B were (and are) evolutionarily accessible to the phage. It is therefore likely that k -cooperativity, as opposed to K_B -cooperativity, has been selected for functional reasons. Further support for this hypothesis is provided by the fact that not all activators increase k . In fact, in phage λ, the factor CII acting on P_{RE} promoter uses both the K_B - and k -cooperativity (18), and the CAP activation of the *lac* operon in *E. coli* uses K_B -cooperativity (19).

To investigate this hypothesis, we model the dynamics of the entire switch and study the effect of the K_B - and k -cooperativity on the stability of the lysogenic state. We show that the stability of the lysogen depends crucially not only on the fact that CI_2 interacts cooperatively with RNAP, but also on the fact that this cooperativity increases k rather than K_B . In fact, our computations suggest that increasing K_B 100-fold while abolishing k -cooperativity yields phage with lysogen that is significantly less stable than the wild-type.

THE MATHEMATICAL MODEL

We make use of a delay differential equation model developed by Santillán and Mackey (20),

$$\begin{aligned} \frac{d[M_{cI}]}{dt} = & [O_R]f_{RM}^c([CI_2]_{\tau_M}, [Cro_2]_{\tau_M}) \\ & + [O_R]f_{RM}([CI_2]_{\tau_M}, [Cro_2]_{\tau_M}) \\ & - (\gamma_M + \mu)[M_{cI}], \end{aligned} \quad (1)$$

$$\frac{d[M_{cro}]}{dt} = [O_R]f_R([CI_2]_{\tau_M}, [Cro_2]_{\tau_M}) - (\gamma_M + \mu)[M_{cro}], \quad (2)$$

$$\frac{d[CI]}{dt} = \nu_{cI}[M_{cI}]_{\tau_{cI}} - (\gamma_{cI} + \mu)[CI], \quad (3)$$

$$\frac{d[Cro]}{dt} = \nu_{cro}[M_{cro}]_{\tau_{cro}} - (\gamma_{cro} + \mu)[Cro], \quad (4)$$

which, as is explained below, tracks concentrations of *ci* mRNA, *cro* mRNA, CI protein, and Cro protein. Concentrations are denoted by square brackets; that is, $[CI]$ is the total concentration of CI protein while $[M_{cro}]$ is the concentration of *cro* mRNA.

We will use $[Cro_2]$ and $[CI_2]$ to denote the concentration of CI and Cro dimers and $[RNAP]$ to denote concentration of the RNA polymerase. The concentration of the right operator is $[O_R]$. The subscript notation $[M_{cro}]_{\tau_{cro}}$ indicates that the concentration of *cro* mRNA is evaluated at $t - \tau_{cro}$ where t is the present time. The time delays τ_{ci} and τ_{cro} are incorporated to take into account the fact that the production of the proteins from the associated mRNA and the actual process of transcription are not instantaneous.

Equations 3 and 4 are based on the assumption that the changes in protein concentrations are linear functions of the corresponding mRNA concentrations. There are two sets of positive decay constants. Since the volume of the growing bacteria increases, concentrations of all chemicals in a cell decrease. This is modeled by the decay constant μ , which is the same in all equations. In addition, each chemical species experiences a specific degradation rate denoted by γ_* . Of particular interest is the constant γ_{ci} . We will model the effect of UV light, which, as is noted earlier, lowers the effective concentration of CI dimers by an increase in the degradation rate γ_{ci} of the CI protein. The ν_* are positive translation initiation constants.

The change in concentration of mRNA is described by Eqs. 1 and 2. The nonlinear function $f_R([CI_2]_{\tau_M}, [Cro_2]_{\tau_M})$ describes the rate of transcription initiation at the promoter P_R . For the sake of clarity, the rate of transcription initiation at the promoter P_{RM} is expressed as the sum of two functions $f_{RM}^c([CI_2]_{\tau_M}, [Cro_2]_{\tau_M})$ and $f_{RM}([CI_2]_{\tau_M}, [Cro_2]_{\tau_M})$, where the first applies to the state of the operator in which CI_2 is bound to O_{R2} and the second when it is not.

Santillán and Mackey's (20) construction of these functions is based on the work of Ackers et al. (7) and begins with expressions of the probability of binding of RNAP to the promoter in the presence or absence of the regulatory proteins. The probability of a particular macroscopic state s of the operator takes the form

$$\mathbb{P}_s([CI_2], [Cro_2]) = \frac{K_B(s)[Cro_2]^{\alpha_s}[CI_2]^{\beta_s}[RNAP]^{\gamma_s}}{\sum_i K_B(s_i)[Cro_2]^{\alpha_i}[CI_2]^{\beta_i}[RNAP]^{\gamma_i}}, \quad (5)$$

where

$$K_B(s) = e^{\frac{-\Delta G_s}{RT}}, \quad (6)$$

and the summation in the denominator is taken over all possible states. Since ΔG_s denotes the binding energy of the state, $K_B(s)$ determines the equilibrium constant for the biochemical reaction that results in binding of the regulatory proteins and/or RNAP to the DNA in a closed form. The right (O_R), the left (O_L) operator (each of which has three subdomains), and the three promoters (P_R , P_{RM} , and P_L)

are included in the model of Santillán and Mackey (20). Therefore the state s of the operator is a description of which of the nine sites are empty or occupied by which of the three possible molecules CI_2 , Cro_2 , or RNAP.

These probabilities need to be multiplied by an appropriate constant, $k(s)$, to incorporate the forward reaction rates of the opening and escape steps to obtain a rate of transcription initiation. Thus, for each state, the transcription initiation rate has the form

$$f_s([CI_2], [Cro_2]) = k(s) \frac{K_B(s)[Cro_2]^{\alpha_s}[CI_2]^{\beta_s}[RNAP]^{\gamma_s}}{\sum_i K_B(s_i)[Cro_2]^{\alpha_i}[CI_2]^{\beta_i}[RNAP]^{\gamma_i}}. \quad (7)$$

Though clearly a simplification, we assume that the rate constants $k(s)$ take on three values:

- k_{cro} , when RNAP is bound to P_R ;
- k_{ci}^c , when RNAP is bound to P_{RM} and CI_2 is bound to O_{R2} ; and
- k_{ci} , when RNAP is bound to P_{RM} and CI_2 is not bound to O_{R2} .

Finally, f_R is the sum of all combinations of Ackers et al. (7) with the restriction that each state s has a RNAP bound to P_R , with O_{R1} and O_{R2} unbound. Similarly, f_{RM}^c is the sum of Ackers et al. (7) for all states s which have RNAP bound to P_{RM} and CI_2 bound to O_{R2} , and f_{RM} the sum of Ackers et al. (7) for all states s which have RNAP bound to P_{RM} but CI_2 is not bound to O_{R2} .

To compare this model against experimental data requires knowledge of the above-mentioned constants. The experimentally determined values are taken from Santillán and Mackey (20) and presented in Tables 1 and 2.

INTERPRETING THE MODEL

Based on the biochemistry of the phage λ switch, the phenomenological state of lysogeny is associated with low levels of Cro and high levels of CI. Similarly, lysis is associated with low levels of CI and high levels of Cro. With this in mind, we look for equilibria of the system of Eqs. 1–4 and declare that an equilibrium for which $0 \approx [Cro] \ll [CI]$ is

TABLE 1 Estimated parameter values from Santillán and Mackey (20) (with the addition of ϕ) for Eqs. 1–4

$\mu \approx 2.0 \times 10^{-2} \text{ min}^{-1}$	$k_{cro} \approx 2.76 \text{ min}^{-1}$
$k_{ci}^c \approx 4.29 \text{ min}^{-1}$	$k_{ci} \approx 0.35 \text{ min}^{-1}$
$\gamma_M \approx 0.12 \text{ min}^{-1}$	$\gamma_{ci} \approx 0.0 \text{ min}^{-1}$
$\gamma_{cro} \approx 1.6 \times 10^{-2} \text{ min}^{-1}$	$\nu_{ci} \approx 0.09 \text{ min}^{-1}$
$\nu_{cro} \approx 3.2 \text{ min}^{-1}$	$\tau_{ci} \approx 0.24 \text{ min}$
$\tau_{cro} \approx 6.6 \times 10^{-2} \text{ min}$	$\tau_M \approx 5.1 \times 10^{-3} \text{ min}$
$K_D^{cl} \approx 5.56 \times 10^{-3} \mu\text{M}$	$K_D^{cro} \approx 3.26 \times 10^{-1} \mu\text{M}$
$[O_R] \approx 5.0 \times 10^{-3} \mu\text{M}$	$[RNAP] \approx 3.0 \mu\text{M}$
$\Delta G_{RL} \approx -3.1 \text{ kcal/mol}$	$\phi \approx 4.29/.35 = 12.26$

TABLE 2 Estimated binding energies from Santillán and Mackey (20)

$\Delta G_{O_R1}^{CI_2} \approx -12.5$ kcal/mol	$\Delta G_{O_H1}^{CI_2} \approx -11.5$ kcal/mol
$\Delta G_{O_R2}^{CI_2} \approx -10.5$ kcal/mol	$\Delta G_{O_H2}^{CI_2} \approx -9.7$ kcal/mol
$\Delta G_{O_R3}^{CI_2} \approx -9.5$ kcal/mol	$\Delta G_{O_H3}^{CI_2} \approx -9.7$ kcal/mol
$\Delta G_{O_R12}^{CI_2} \approx -2.7$ kcal/mol	$\Delta G_{O_H12}^{CI_2} \approx -2.7$ kcal/mol
$\Delta G_{O_R23}^{CI_2} \approx -2.9$ kcal/mol	$\Delta G_{O_H23}^{CI_2} \approx -2.9$ kcal/mol
$\Delta G_{O_R1}^{Cro_2} \approx -12.0$ kcal/mol	$\Delta G_{O_H1}^{Cro_2} \approx -12.0$ kcal/mol
$\Delta G_{O_R2}^{Cro_2} \approx -10.8$ kcal/mol	$\Delta G_{O_H2}^{Cro_2} \approx -10.8$ kcal/mol
$\Delta G_{O_R3}^{Cro_2} \approx -13.4$ kcal/mol	$\Delta G_{O_H3}^{Cro_2} \approx -13.4$ kcal/mol
$\Delta G_{O_R12}^{Cro_2} \approx -1.0$ kcal/mol	$\Delta G_{O_H12}^{Cro_2} \approx -1.0$ kcal/mol
$\Delta G_{O_R23}^{Cro_2} \approx -0.6$ kcal/mol	$\Delta G_{O_H23}^{Cro_2} \approx -0.6$ kcal/mol
$\Delta G_{O_{R123}}^{Cro_2} \approx -0.9$ kcal/mol	$\Delta G_{O_{H123}}^{Cro_2} \approx -0.9$ kcal/mol
$\Delta G_{P_R}^{RNAP} \approx -12.5$ kcal/mol	$\Delta G_{P_L}^{RNAP} \approx -11.3$ kcal/mol
$\Delta G_{P_{RM}}^{RNAP} \approx -11.5$ kcal/mol	

a lysogenic equilibrium and an equilibrium for which $0 \approx [CI] \ll [Cro]$ is a lytic equilibrium.

The equilibria of this system are steady (time-independent) states of the system and thus are not dependent on delays. Notice that since both CI and Cro proteins form dimers, the right-hand side of the equations depends on the concentration of dimers. The conversion formula for computing the concentration of dimers from total concentration of monomers is

$$[CI_2] = \frac{1}{2}[CI] - \frac{K_D^{cl}}{8} \left(\sqrt{1 + 8 \frac{[CI]}{K_D^{cl}}} - 1 \right), \quad (8)$$

$$[Cro_2] = \frac{1}{2}[Cro] - \frac{K_D^{cro}}{8} \left(\sqrt{1 + 8 \frac{[Cro]}{K_D^{cro}}} - 1 \right), \quad (9)$$

and its derivation is presented in the Appendix.

Let

$$\phi = \frac{k_{cl}^c}{k_{cl}}$$

Observe that this provides a measure of the effect of O_{R2} -bound CI_2 on the forward reaction rate associated with opening and escape. In particular, $\phi > 1$ implies that the rate of transcription initiation with O_{R2} -bound CI_2 is higher than that without. We refer to this as k -cooperativity.

As indicated before, γ_{cl} represents the degradation rate of $[CI]$, induced for example by exposure to UV radiation. Since this is known to trigger induction of phage, we study the equilibria as a function of γ_{cl} . Observe that the equilibria satisfy the two equations

$$\Phi([CI], [Cro], \gamma_{cl}) = 0 \text{ and } \Theta([CI], [Cro]) = 0,$$

where

$$\Phi([CI], [Cro], \gamma_{cl}) = \frac{\nu_{cl}}{\gamma_M + \mu} [O_R] f_{RM}^c([CI_2], [Cro_2]) + f_{RM}([CI_2], [Cro_2]) - (\gamma_{cl} + \mu)[CI]$$

$$\Theta([CI], [Cro]) = \frac{\nu_{cro}}{\gamma_M + \mu} [O_R] f_R([CI_2], [Cro_2]) - (\gamma_{cro} + \mu)[Cro].$$

The intersection of these two curves in the $[CI]$, $[Cro]$ plane determines two protein concentrations at a dynamical equilibrium; the remaining two concentrations $[M_{cl}]$ and $[M_{cro}]$ can be found from Eqs. 3 and 4 with the left-hand side set equal to zero.

Observe that Θ is independent of γ_{cl} . The set $\Theta([CI], [Cro]) = 0$ is given by the solid curve in Fig. 2 *a*. According to Table 1, for wild-type phage in the absence of UV radiation, $\gamma_{cl} = 0 \text{ min}^{-1}$. The set $\Phi([CI], [Cro], 0) = 0$ is plotted in dashed representation in Fig. 2 *a*. There is a unique equilibrium, i.e., intersection point of $\Theta([CI], [Cro]) = 0$ and $\Phi([CI], [Cro], 0) = 0$, for which $[CI] = 0.528 \mu\text{M}$ and $[Cro] = 1.04 \times 10^{-5} \mu\text{M}$. This is a lysogenic equilibrium.

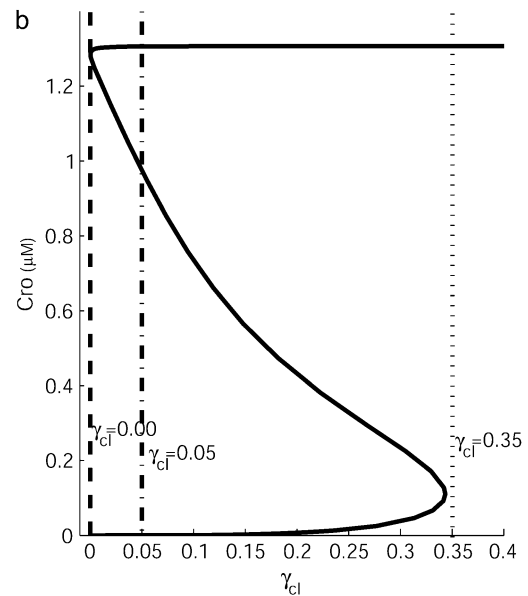
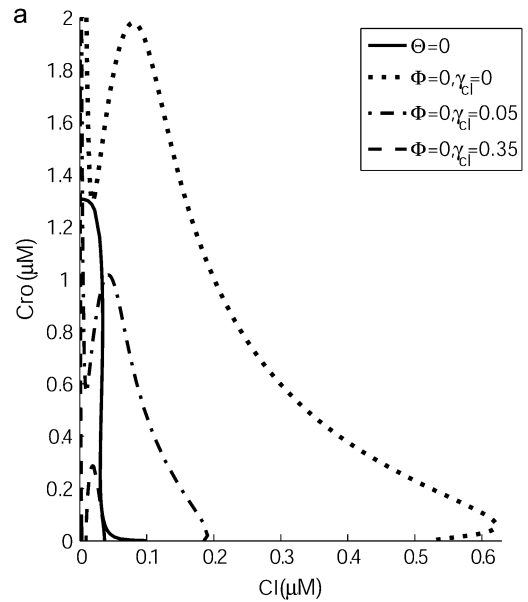


FIGURE 2 (a) Nullclines for $\Theta = 0$ (solid) and $\Phi = 0$ with $\gamma_{cl} = 0 \text{ min}^{-1}$ (dash), $\gamma_{cl} = 0.05 \text{ min}^{-1}$ (dots), and $\gamma_{cl} = 0.35 \text{ min}^{-1}$ (dash-dot). (b) Bifurcation diagram of γ_{cl} versus $[Cro]$.

As the parameter γ_{CI} increases, the $\Phi = 0$ curve shifts its position relative to the $\Theta = 0$ curve. When γ_{CI} is 0.00039 min^{-1} , a pair of new intersections corresponding to new equilibria appears. Plotted in dots in Fig. 2 *a* is $\Phi([\text{CI}], [\text{Cro}], 0.05) = 0$. The equilibrium with high value of $[\text{Cro}]$ and low value of $[\text{CI}]$ corresponds to a lytic state and we call it a lytic equilibrium. Observe that there are three equilibria: a lysogenic equilibrium, a lytic equilibrium, and an unstable intermediate equilibrium. Finally, the dash-dot curve represents $\Phi([\text{CI}], [\text{Cro}], 0.35) = 0$, which intersects $\Theta = 0$ in a single point corresponding to the lytic equilibrium.

Clearly, the set of equilibria changes as a function of γ_{CI} . This is indicated in the bifurcation diagram of Fig. 2 *b*, where the equilibrium values of $[\text{Cro}]$ are plotted on the vertical axis as a function of γ_{CI} . This graph allows us to describe the induction process. When no UV radiation is applied to bacterial population, $\gamma_{\text{CI}} = 0 \text{ min}^{-1}$ and the phage occupies lysogenic equilibrium. As γ_{CI} is slowly increased, the lysogenic equilibrium moves and the phage state tracks this slowly moving equilibrium. Immediately after γ_{CI} crosses the value of 0.343, the lysogenic equilibrium disappears and the state rapidly approaches the lytic equilibrium.

Therefore we define the value $\gamma_{\text{WT}}^* = 0.343 \text{ min}^{-1}$ as the wild-type induction value. The dashed lines in Fig. 2 *b* shows the values of γ_{CI} that correspond to the same dashed curves in Fig. 2 *a*.

In later sections, we make use of bifurcation diagrams such as that of Fig. 2 *b*, thus we point out some of the important features. For the parameter values $0.00039 \text{ min}^{-1} \leq \gamma_{\text{CI}} \leq 0.343 \text{ min}^{-1}$, the wild-type phage λ switch is bistable; that is, there are two stable equilibria, the lysogenic equilibrium (corresponding to the lower branch) and the lytic equilibrium (corresponding to the upper branch). Furthermore, for some initial concentrations, the state of the phage will evolve toward the lysogenic equilibrium, and for other initial concentrations, toward the lytic equilibrium.

We introduced the dimensionless parameter ϕ to have a measure of the change in the forward reaction rate associated with opening and escape. We wish to have a similar measure for the binding probabilities. When the binding of a transcription factor increases RNAP residence time on the promoter, it is reflected in the Ackers model in the cooperative increase of the binding energy of the transcription factor-RNAP pair. We denote the binding energy between CI_2 and $\text{O}_{\text{R}2}$ by $\Delta G_{\text{O}_{\text{R}2}}^{\text{CI}_2}$ and binding energy between RNAP and P_{RM} by $\Delta G_{\text{P}_{\text{RM}}}^{\text{RNAP}}$. In the absence of binding cooperation, as is the case in the wild-type phage λ , the binding energy contribution from $\text{O}_{\text{R}2}$ -bound CI and P_{RM} -bound RNAP to any state s that contains them is

$$\Delta G_{\text{ind}}(s) = \Delta G_{\text{O}_{\text{R}2}}^{\text{CI}_2} + \Delta G_{\text{P}_{\text{RM}}}^{\text{RNAP}} + \Delta G_{\text{rest}}(s),$$

where subscript *ind* stands for independent binding of the binding factors and $\Delta G_{\text{rest}}(s)$ is the binding energy of the other factors in state s .

The cooperative binding between CI_2 and RNAP is reflected in additional binding energy $\Delta G_{\text{O}_{\text{R}2\text{PRM}}}^{\text{CI}_2\text{RNAP}}$. If this energy is positive, we refer to this as K_{B} -cooperativity. We express the cooperativity in terms of the binding constant $K_{\text{B}}(s)$ (see (6))

$$K_{\text{B}}(s) = \beta K_{\text{B}}^{\text{ind}}(s),$$

where $K_{\text{B}}^{\text{ind}}(s) = \exp(-\frac{1}{RT}(\Delta G_{\text{ind}}(s)))$ and the state s independent multiplicative factor is

$$\beta = \exp\left(-\frac{1}{RT}(\Delta G_{\text{O}_{\text{R}2\text{PRM}}}^{\text{CI}_2\text{RNAP}})\right).$$

In this formulation, $\beta > 1$ represents the cooperative binding.

In summary, the k -cooperativity is manifested by the constant $\phi > 1$ and K_{B} -cooperativity by $\beta > 1$.

Model validation

To validate our biological interpretation of the equilibria of Eqs. 1–4, we model the induction scenarios for several different phage mutants which are described in the literature.

O_R323 mutant

Little et al. (21) constructed a mutant $\text{O}_{\text{R}323}$ in which the $\text{O}_{\text{R}1}$ domain was replaced by $\text{O}_{\text{R}3}$ and reported the following results:

- R1. $\text{O}_{\text{R}323}$ can lysogenize.
- R2. $\text{O}_{\text{R}323}$ has a threshold response, but at lower doses of UV radiation and at a higher level of free phage in the lysogen than the wild-type.
- R3. In the lytic state the burst size, i.e., the number of phages per infected cell, of $\text{O}_{\text{R}323}$ is lower than that of the wild-type.

This mutation is easily incorporated into the mathematical model. To replace the $\text{O}_{\text{R}1}$ binding site by the $\text{O}_{\text{R}3}$ binding site we set the binding energy of CI_2 to $\text{O}_{\text{R}1}$ to be that of CI_2 to $\text{O}_{\text{R}3}$ (-9.5 kcal/mol). Similarly, the binding energy of Cro to $\text{O}_{\text{R}1}$ is set to that of Cro to $\text{O}_{\text{R}3}$ (-12.0 kcal/mol).

The bifurcation curves for the $\text{O}_{\text{R}323}$ mutation as compared with the wild-type are presented in Fig. 3. The graph shows the concentration of Cro as a function of γ_{CI} . The solid curve represents the wild-type phage, while the dot-dashed curve represents the $\text{O}_{\text{R}323}$. The lower branch on both curves corresponds to the lysogenic equilibrium and the upper branch to the lytic equilibrium.

The existence of the lower branch in the dot-dashed curve of Fig. 3 implies that $\text{O}_{\text{R}323}$ can lysogenize (compare R1). However, the induction value for the $\text{O}_{\text{R}323}$ mutant is $\gamma_{\text{O}_{\text{R}323}}^* = 0.09 \text{ min}^{-1} < 0.34 \text{ min}^{-1} = \gamma_{\text{WT}}^*$, which suggests that a lower level of UV radiation is required to induce lytic growth (compare R2). Observe that when $\gamma_{\text{CI}} = 0 \text{ min}^{-1}$ there are three equilibria in the system describing $\text{O}_{\text{R}323}$. Thus, a

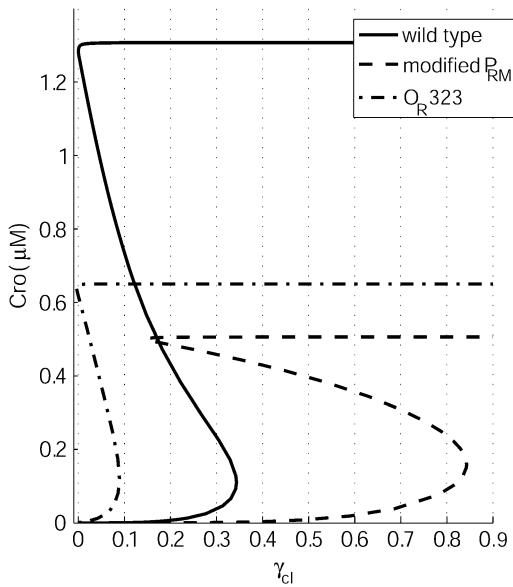


FIGURE 3 Bifurcation diagrams for wild-type and O_{R323} and P_{RM} mutants. The concentration of Cro is graphed as a function of γ_{CI} . The solid curve represents the wild-type phage, while the dot-dashed curve represents O_{R323} mutant and the dashed curve represents a phage with mutated P_{RM} binding site which resulted in having $P_{RM} = -12.5$ kcal/mol, $\phi = 4.5/35$, and $P_R = -10.5$ kcal/mol. For comparison, the wild-type values were $P_{RM} = -11.5$ kcal/mol, $\phi = 4.29/35$, and $P_R = -12.5$ kcal/mol.

stable lytic equilibrium is present even in the absence of UV radiation, and in the presence of noise, some phages can spontaneously induce and switch to the lytic state. This would manifest itself experimentally in increased number of free phages (compare R2).

Finally, it is possible that the burst size (number of phages per infected cell) is proportional to the transcription level of the lytic pathway in phage's genome, which in turn may be proportional to the level of Cro production in the lytic state. This theory is in agreement with Fig. 3 in which the Cro production in the lytic state for O_{R323} (the upper dot-dashed branch) is significantly lower than in the wild-type lytic state (the upper solid branch) (compare R3). Of course, the burst size can also be determined by energetics of the cell or by available resources, and therefore the suggested relationship between Cro production and the burst size is, at best, speculative.

P_{RM} mutant

Michalowski and Little (22) (see also (23)) obtained multiple mutants of phage λ by subjecting the P_{RM} binding site to mutagenesis. These were then compared to wild-type by three criteria: the ability to grow lytically, the ability to establish and maintain a stable lysogenic state, and the ability to undergo prophage induction. In the experiments, they were particularly careful not to affect the O_{R2} and O_{R3} binding sites. Of these isolates, they further analyzed nine which were selected because they were comparable to or more difficult to induce than the wild-type. When compared to wild-type,

these nine strains seem to share three properties: they had an equal or increased P_{RM} binding affinity; a decreased P_R binding affinity; and an increase in the k -cooperativity between CI_2 and RNA polymerase. To model such mutants we set $P_{RM} = -12.5$ kcal/mol, $P_R = -10.5$ kcal/mol, and $\phi = 4.5/35$, which should be compared to wild-type values $P_{RM} = -11.5$ kcal/mol, $P_R = -12.5$ kcal/mol, and $\phi = 4.29/0.35$. The resulting bifurcation diagrams are presented in Fig. 3. The induction parameter $\gamma_{P_{RM}}^* \approx 0.85 \text{ min}^{-1}$ for the mutation is much higher than the wild-type $\gamma_{WT}^* \approx 0.35 \text{ min}^{-1}$, implying greater stability of the lysogen.

cl - pc mutant

When a pc mutation is introduced to CI , it eliminates the k -cooperativity between CI_2 protein bound to O_{R2} and RNAP (4,24). This mutant forms lysogen in a wild-type bacteria, but suffers from a high rate of spontaneous induction and induction at very low levels of UV light.

To model this mutant we replace the k_{CI}^c in the function f_{RM}^c (see Eq. 2) by k_{CI} . This implies $\phi = 1$. The associated bifurcation curves are indicated in Fig. 4. Observe that our model predicts that the induction value is dramatically lower ($\gamma_{WT}^* = 0.34 \text{ min}^{-1}$ in wild-type, $\gamma_{CIpc}^* = 0.01 \text{ min}^{-1}$ in the mutant). In the noisy environment of a cell, we expect that this low stability threshold will yield a high spontaneous induction rate.

K_B - and k -cooperativity are not interchangeable

Our most significant prediction is that K_B - and k -cooperativity affect the stability of the lysogen differently, and thus are not interchangeable. To demonstrate, this we compare the stability of the lysogen under k -cooperativity, $\beta = 1$, $\phi = \alpha > 1$, and against K_B -cooperativity, $\phi = 1$, $\beta = \alpha > 1$, for different values of α . The analysis of the stability of the cl - pc mutant above provides the first step of this analysis. In this mutant, both $\phi = 1$ and $\beta = 1$; thus, all cooperation is abolished and our model predicts that the induction value is dramatically lower.

To test the ability of K_B -cooperativity to restore the lysogen stability, we fix $\phi = 1$ and solve for the equilibria at $\beta = 10$ and $\beta = 100$. The bifurcation diagrams are presented in Fig. 4, where they can be compared against the cl - pc mutant and the wild-type (recall that for the wild-type, $\phi \approx 12$ and $\beta = 1$). Observe that when $\beta = 10$, the induction value is $\gamma_{\beta=10}^* = 0.04 \text{ min}^{-1}$, which is much lower than $\gamma_{WT}^* = 0.34 \text{ min}^{-1}$. We predict that this produces a very unstable lysogen. Even in the case of unrealistically strong K_B -cooperativity, $\beta = 100$, and the induction value is only $\gamma_{\beta=100}^* = 0.07 \text{ min}^{-1}$.

Fig. 4 clearly indicates that K_B - and k -cooperativity are not equivalent. This difference is highlighted in Fig. 5 where isoclines of the induction value γ_* are plotted as a function of β and ϕ . The deviation of symmetry across the diagonal $\beta = \phi$ indicates the extent to which K_B - and k -cooperativity fail to

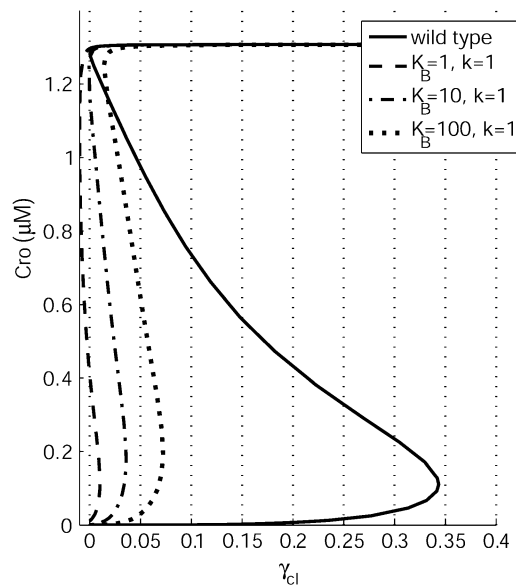


FIGURE 4 Results from eliminating positive control ($\phi = 1$) with values of $\beta = 1$ (dashed curve, cI-pc mutant), $\beta = 10$ (dotted curve), and $\beta = 100$ (dash-dot curve). We graph concentration of Cro as a function of γ_{Cl} .

be equivalent in maintaining the stability of the lysogenic state.

While Figs. 4 and 5 clearly indicate that there is a difference between K_B - and k -cooperativity, they provide no explanation for this difference. Since the interactions between the binding factors are mediated through nonlinear functions we do not expect there to be a simple, but complete quantitative description of this difference. However, there are two mathematical results that provide a partial explanation.

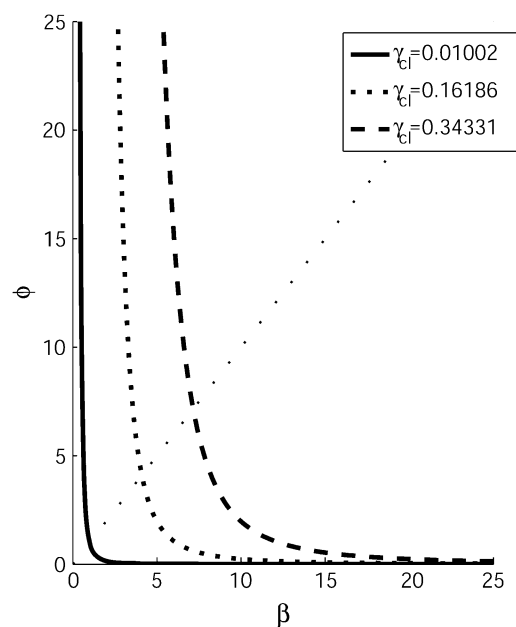


FIGURE 5 Level curves of the induction value γ_* as a function of both β and ϕ . Here $\beta > 1$ represents K_B -cooperativity and $\phi > 1$ represents k -cooperativity.

The first has to do with the rate of production of CI. Let

$$f_{RM}^{\beta, \phi}([CI_2], [Cro_2]) := f_{RM}^c([CI_2], [Cro_2]) + f_{RM}([CI_2], [Cro_2])$$

for fixed values of β and ϕ . By (Theorem 4.8 in (11)), if $\alpha > 1$, then

$$f_{RM}^{1, \alpha}([CI_2], [Cro_2]) > f_{RM}^{\alpha, 1}([CI_2], [Cro_2]).$$

This means that the rate of transcription of cI mRNA is greater under k -cooperativity than under an equal amount of K_B -cooperativity.

The second has to do with the biological fact that at low concentrations CI_2 upregulates its own transcription, while at high concentrations it downregulates its own transcription (4). In the lysogen O_R1 is almost always bound by CI_2 protein and thus, the production of Cro is very low. To produce a simple model that can be easily analyzed we assume CI_2 is always bound to O_R1 , and thus the states of interest involve the binding of CI_2 to O_R2 and O_R3 . In Example 4.11 in Gedeon et al. (11), it is proven that under these assumptions there exists a unique critical concentration κ , such that if $[CI_2] < \kappa$, then CI_2 is an activator and if $[CI_2] > \kappa$, then CI_2 is a repressor. This implies that the maximal production rate of CI mRNA occurs at $[CI_2] = \kappa$. As is shown in Example 4.13 in Gedeon et al. (11), κ is larger under k -cooperativity than under an equal amount of K_B -cooperativity. In particular, the critical concentration for the wild-type is greater than the critical concentration for the cI-pc mutant.

CONCLUSIONS

One of the common features of transcriptional control in bacteria and eukaryotes is “activation by recruitment,” where subtle interactions between the transcription factors and RNAP control the rate of transcription. The three essential steps in this process (binding, opening, and escape) coalesce in the Ackers modeling framework into two sets of constants. One set captures binding energies, while the other models the transcription initiation process which includes both opening and escape. If for some state of the operator the binding of a factor increases the binding probability of RNAP, we call it K_B -cooperativity. If, on the other hand, the factor increases the probability of transcription initiation, we call it k -cooperativity.

At the first glance it may appear that these two types of activation are interchangeable. We have shown, using an experimentally validated dynamic model of phage λ that, with respect to induction of the lysogenic state, k - and K_B -cooperativity are not substitutable. Without k -cooperativity, the lysogenic state of the phage λ switch is quite unstable and comparable to some known mutants like O_{R323} (21).

Our model produced experimentally verifiable predictions and can serve to test hypothesis about induction of phage λ mutants before they are constructed in the lab. Furthermore, the mathematical techniques and arguments used to obtain

these predictions are quite general and thus in the long run we believe that this type of analysis will prove useful for bioengineers who are trying to design novel genetic control units.

APPENDIX

Computation of binding energies

For each state of the promoters P_R and P_L , the transcription initiation rate is

$$f_s([\text{Cl}_2], [\text{Cro}_2]) = k(s) \frac{K_B(s) [\text{Cro}_2]^{\alpha_s} [\text{Cl}_2]^{\beta_s} [\text{RNAP}]^{\gamma_s}}{\sum_i K_B(s_i) [\text{Cro}_2]^{\alpha_i} [\text{Cl}_2]^{\beta_i} [\text{RNAP}]^{\gamma_i}},$$

where

$$K_B(s) = e^{\frac{-\Delta G_s}{RT}}$$

and ΔG_s is a binding state energy. We calculate these energies using the formula

$$\begin{aligned} \Delta G_s = & \sum_{X=R,L} \sum_{Y=\text{Cl}_2, \text{Cro}_2} \sum_{\nu=1}^3 \Delta G_{O_{X\nu}}^Y \Gamma_{O_{X\nu}}^Y(s) \\ & + \sum_{X=R,L} \sum_{Y=\text{Cl}_2, \text{Cro}_2} \sum_{\nu=1}^2 \Delta G_{O_{X\nu\nu+1}}^Y \Gamma_{O_{X\nu}}^Y(s) \Gamma_{O_{X\nu+1}}^Y(s) \Gamma_{O_{X123}}^{\text{Cro}_2}(s) \\ & + \sum_{X=R,L} \Delta G_{O_{X123}}^{\text{Cro}_2} \Gamma_{O_{X1}}^{\text{Cro}_2}(s) \Gamma_{O_{X2}}^{\text{Cro}_2}(s) \Gamma_{O_{X3}}^{\text{Cro}_2}(s) \\ & + \sum_{X=RM,R,L} \Delta G_{P_X}^{\text{RNAP}} \Gamma_{P_X}^{\text{RNAP}}(s) + \sum_{\nu=1}^3 \Delta G_{R_L} \Gamma_{O_{R\nu}}^{\text{Cl}_2}(s) \Gamma_{O_{L\nu}}^{\text{Cl}_2}(s), \end{aligned}$$

where

$$\Gamma_X^Y(k) = \begin{cases} 1, & \text{if molecule } Y \text{ is bound to site } X; \\ 0, & \text{otherwise} \end{cases}$$

and

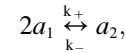
$$\Gamma_{O_{X123}}^{\text{Cro}_2}(s) = \begin{cases} 0, & \text{if Cro}_2 \text{ is bound to } O_{R1}, O_{R2}, \text{ and } O_{R3} \\ 1, & \text{otherwise.} \end{cases}$$

All ΔG_s^* values in Table 2 are computed from Darling et al. (25). The detailed explanation of how these energies have been computed can be found in Santillán and Mackey (20). The first sum includes all binding energies of transcription factors to the six binding sites on both left and right operators. The second sum includes all cooperation energies between any two adjacent factors and the third takes into account cooperativity that results from having Cro bound to all three binding sites on either O_R or O_L . It should be noted that, in the measurements by Darling et al. (25), the cooperative binding energies when Cro is bound to all three subdomains of O_R or O_L are not equal to the sum of the cooperative binding energies $\Delta G_{O_{X12}}^{\text{Cro}_2}$ and $\Delta G_{O_{X23}}^{\text{Cro}_2}$ (see Table 2). The term $\Gamma_{O_{X123}}^{\text{Cro}_2}(s)$ in the second sum guarantees that when Cro occupies all three subdomains in O_R or O_L , the cooperative energies $\Delta G_{O_{X12}}^{\text{Cro}_2}$ and $\Delta G_{O_{X23}}^{\text{Cro}_2}$ are not included in this sum. The energies $\Delta G_{O_{X123}}^{\text{Cro}_2}$ are then added in the third sum. The fourth sum adds the RNAP binding energy for the state, and the last one contributes any cross cooperation between Cl_2 molecules bound to P_R and P_L .

The differential equation model

In the differential equation model (Eqs. 1–4), the concentrations on the left-hand side denote total monomer concentration, while on the right-hand side

we have dimer concentrations $[\text{Cl}_2]$ and $[\text{Cro}_2]$. To accurately represent this, Eqs. 8 and 9 embody this dimerization. As demonstrated in Santillán and Mackey (20), these equations arose from the chemical reaction



where a_1 is a free monomer form of the protein and a_2 represents a dimer of protein a , and k_+ , and k_- are the forward and backward rate constants respectively.

In chemical equilibrium with $K_D = k_-/k_+$, we have the relation

$$[a_1]^2 = K_D [a_2]. \quad (10)$$

K_D is the dissociation constant and $[\cdot]$ represents concentration. In addition, if $[a]$ is the total monomer concentration,

$$[a] = [a_1] + 2[a_2]. \quad (11)$$

Equations 10 and 11 can be used to solve for $[a_2]$, leading to

$$[a_2] = \frac{[a]}{2} - \frac{K_D}{8} \left(\sqrt{1 + 8 \frac{[a]}{K_D}} - 1 \right),$$

from which Eqs. 8 and 9 follow.

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