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Preharvest foliar applications of glycine-betaine protects banana fruits from chilling injury during the postharvest stage

Luis C Rodríguez-Zapata^{1*}, Francisco L Espadas y Gil¹, Susana Cruz-Martínez¹, Carlos R Talavera-May¹, Fernando Contreras-Marin¹, Gabriela Fuentes¹, Enrique Sauri-Duch² and Jorge M Santamaría¹

Abstract

Background: Banana plantations are affected by environmental factors such as chilling injury, which reduces the quality of fruits and causes losses of up to 50% in the yield of banana and it will be more important in terms of global climate change. Chilling injury of the fruits can also occur during transport and storage at low temperatures, particularly in tropical fruits. In banana, losses of up to 20% can occur during postharvest handling. Given this situation, it is necessary to explore alternatives that might reduce chilling injury, such as the use of compatible solutes including glycine-betaine (GB).

Results: In the present work, experiments were performed to analyze the possible role of preharvest foliar applications of GB, to prevent the subsequent development of chilling injury of banana fruits during the postharvest storage at low temperatures. After 3 days of the preharvest application of 100 mM GB over banana leaves (250 ml/plant), the fruits were harvested and first stored at 10°C for 6 h and then transferred to 23°C ± 1°C until they reached commercial ripening. A second group of plants were not treated with GB during the preharvest stage, but their fruits were exposed to 10°C for 6 h before transferring them to 23°C ± 1°C until they reached commercial ripening. A control group was untreated with GB during the preharvest stage, and fruits were not exposed to low temperatures but they were kept at 23°C until they reached commercial ripening.

Conclusions: The results showed that the preharvest foliar application of GB (100 mM) to banana plants reduced the biochemical and physiological alterations caused by chilling injury on harvested fruits.

Keywords: Glycine-betaine; Chilling injury; Banana; Browning

Background

One of the factors that limits the establishment of banana plantations at certain latitudes, and has an impact on the development of plants and fruit production, is chilling damage. This loss in fruit production intensifies at the postharvest stage, because losses between 15% and 25% of production can occur at that stage [1]. Bananas are sensitive to chilling and suffer physiological damage when exposed to environment temperatures below the critical point (11°C) regardless of the type of cultivar [2,3]. It is considered that the primary cause of injuries caused by chilling are the changes in the properties of the cell membrane [4] that causes a cascade of secondary reactions,

which include the increase in the production of ethylene, increase in anaerobic respiration, and reduced photosynthesis [4-6]. Chilling-tolerant plants use evasion and tolerance mechanisms such as the synthesis of low-molecular-weight compounds called compatible solutes [4,7,8]. Compatible solutes include sugars such as trehalose and mannitol, amino acids such as proline, and amino acid derivatives such as glycine-betaine (GB) [9]. Plants capable of accumulating these osmolytes are able to survive under conditions of stress by drought, salinity, and chilling. GB is a quaternary ammonium compound that is a very effective compatible solute [10-12] and is found in a wide range of foods [13]. In plants that synthesize GB, which are known as GB accumulators (spinach, maize, and barley), this compatible solute accumulates in leaves in response to drought and salinity, as well as during

* Correspondence: lcrz@cicy.mx

¹Unidad de Biotecnología, Centro de Investigación Científica de Yucatán, Mérida 97200, México

Full list of author information is available at the end of the article

acclimation to cold [10,14,15]. Moreover, GB has been shown *in vitro* to stabilize membranes of the oxygen-evolving photosystem II complex [16,17]. GB also stabilizes the activity of ribulose 1,5-bisphosphate carboxylase/oxygenase in a transgenic cyanobacterium *in vivo* [18]. In higher plants, GB is synthesized from choline (Cho) via betaine aldehyde (BA). The first and second steps in the biosynthesis of GB are catalyzed by the enzymes choline monoxygenase (CMO) and betaine aldehyde dehydrogenase (BADH), respectively [10]. There is evidence suggesting that GB plays a role in stress tolerance in some species of plants [19]. However, to date, there are no studies on the effect of GB application on climacteric tropical fruits. It has been reported that the browning (low-temperature injury symptom) in banana peel is caused by the oxidation of polyphenols caused primarily by polyphenol oxidase (PPO) (Yang et al. [20]). Nguyen et al. [21] report that banana fruits exposed to low temperatures experience an increased activity of both phenylalanine ammonia lyase (PAL) and PPO. In chilling-sensitive plants, oxidative stress is a major component of chilling stress and active oxygen species (AOS) such as hydrogen peroxide, superoxide radicals, and hydroxyl radicals that can react very rapidly with DNA, causing severe damage to cellular proteins and lipids [22]. Plants have mechanisms to tolerate different types of stress such as chilling that also causes cellular dehydration. Commonly, plants synthesize compatible soluble compounds to retain water in the cell or to protect cellular components from chilling-induced dehydration [23]. Exogenous applications of GB have been successful in increasing the tolerance to abiotic stresses in various species including *Arabidopsis thaliana* [24-27]. Moreover, the overexpression of genes codifying for one of the enzymes involved in GB biosynthesis resulted in increased tolerance to various abiotic stresses including chilling [28-31].

Therefore, in the present study, it was intended to evaluate a possible protective role of the preharvest foliar applications of GB on the subsequent physiological and biochemical disorders caused by postharvest low-temperature storage of banana fruits (cv Giant Dwarf).

Methods

Plant material

This work was conducted in banana *Musa acuminata* cv 'Giant-Dwarf' plants, grown at the nursery of CICY at Mérida Yucatán, Mexico. The fruits were harvested at the first stage of ripening, according to the Customers Services Department Chiquita Brands, Inc., Cincinnati, OH. [32].

Experimental

Three different groups of five plants each were formed. In the first group (control), distilled water (250 ml/plant) was sprayed over the first two leaves near the inflorescence for three consecutive days prior to harvesting the

fruits that were kept at 25°C until they reach commercial ripening. A second group of plants (chilled fruits) was also treated with distilled water for three consecutive days; the fruits were then harvested, but they were exposed to 10°C for 6 h before they were kept at 23°C until they reached commercial ripening. The third group of plants (GB-treated chilled fruits) was treated with 250 ml per plant of a solution of 100 mM GB, sprayed over the first two leaves near the inflorescence, for three consecutive days. The fruits were then harvested and exposed to 10°C for 6 h before they were kept at 23°C until they reached commercial ripening. In all temperature treatments, an incubator (Forma Scientific, Diurnal Growth Chamber, Marietta, Ohio, USA) under dark conditions was used. Samples of five fruits from each treatment were taken every 24 h for 8 or 9 days, to measure the various biochemical and physiological parameters.

Enzymatic activity of polyphenol oxidase (PPO)

Five grams (fresh weight (FW)) of fruit peels from the three different groups were homogenized using a polytron (Poly science model X-520, Niles, Illinois, USA) with 10 ml of 0.1 M sodium phosphate buffer pH 7 (0.1 M phosphate monobasic sodium monohydrate and 0.1 M phosphate dibasic hepta hydrated sodium) with protease inhibitors (5 $\mu\text{g ml}^{-1}$ of aprotinin, 5 $\mu\text{g ml}^{-1}$ leupeptine, and 10 mM of phenyl-methyl sulphonyl fluoride (PMSF)), 10 mM ascorbic acid, and 3% of polyvinyl-pyrrolidone (PVPP). Subsequently, the homogenate was centrifuged to 13,200 rpm for 30 min at 4°C. The specific enzymatic activity of PPO was determined according to Tamayo [33] and Arzávalo [34] and expressed as $\mu\text{mol min}^{-1} \text{mg}^{-1}$ protein.

Quantification of soluble phenols

One gram (FW) of fruit peels were crushed in a mortar in the presence of liquid nitrogen until obtaining a fine powder. The macerate was transferred to a test tube and 5 ml of a mixture of chilled ethanol (1:1 v/v) was added. The homogenate was agitated for 1 min, and it was placed in ice for 6 h. Subsequently, it was centrifuged to 10,000 rpm (Centrifuge 5415 D Eppendorff) for 10 min at 4°C; the supernatant was then separated, and this fraction was extracted for the quantification of soluble phenols. Phenol determination was made by the reaction of Folin-Ciocalteu reagent as described by Singleton and Rossi [35] with slight modifications [21].

Soluble protein quantifications

One gram (FW) of fruit peels was macerated in a mortar by adding liquid nitrogen and 1.5% of PVPP (0.015 g ml^{-1}) to obtain a fine powder. The macerate was transferred to an Eppendorf tube containing 2 ml of extraction buffer (50 mM NaCl, 1 mM of EGTA, 50 mM Tris pH 7.4, 250 mM sucrose, 10% glycerol) and protease inhibitors

(5 $\mu\text{g ml}^{-1}$ of aprotinin, 5 $\mu\text{g ml}^{-1}$ leupeptine, and 10 mM PMSF). The homogenate was centrifuged to 10,400 rpm at 4°C for 30 min. The supernatant was used to determine protein content. Spectrophotometric soluble protein quantification was made as described in Bradford [36].

Measurement of chlorophylls and total carotenoids

One gram (FW) of fruit peels was crushed in a mortar with liquid nitrogen. Chilled acetone (80%) was added (10 ml per gram of tissue). The homogenate was vacuum filtered using a funnel with filter paper. The obtained filtrate was left to stand for 30 min under chilling temperatures and darkness, and it was centrifuged for 10 min at 12,000 rpm. Spectrophotometric (Beckman Coulter DU 650, Beckman Coulter, Inc., Fullerton, CA, USA) measurements of the acetonic extract were made at three different wavelengths (470, 645, and 663 nm), and chlorophyll and carotenoids concentrations were calculated according to Wellburn [37].

Ethylene and respiration measurement

Two fruits were placed in a 1-l glass container hermetically closed and remained in the dark at ambient temperature ($23^{\circ}\text{C} \pm 1^{\circ}\text{C}$) for 1 h. Then, 3 ml of the mixture of gases was taken from the glass container and analyzed in a gas chromatograph (Varian Star 3400 cx, Walnut Creek, CA, USA) fitted with two detectors placed in series; the first detector was a thermal conductivity detector (TCD) for CO_2 , and the second was a flame ionization detector (FID) for ethylene, using hydrogen and air as flame and helium as carrier gas. Standard mixtures of carbon dioxide (CO_2) 5,000 $\mu\text{mol mol}^{-1}$, ethylene 50 $\mu\text{mol mol}^{-1}$, and nitrogen were used for calibration. Production of ethylene and respiratory rate assessments were carried out according to McCollum and Mc Donald

[38] and Liu et al. [39] with some modifications from Chillet and Lapeyre de Bebellaire [40].

Cell membrane stability (CMS); electrolyte leakage

Cell membrane stability (CMS) was measured as electrolyte leakage according to McCollum and McDonald [38] and Prohens et al. [41]. Ten disks (10-cm diameter) of fruit peels from all treatments were placed on 25 ml of mannitol 0.4 M and incubated under agitation for 24 h at room temperature. Samples were sterilized at 121°C to break all tissues and free the electrolytes. Subsequent measurements of electrical conductivity were made with a conductimeter (Orion model 162), using 0.4 M mannitol and de-ionized water solutions as blanks.

Chlorophyll fluorescence

Fruits were exposed to darkness for 30 min prior to taking the fluorescence measurements. Photochemical efficiency of photosystem II (PSII) measured as the ratio of variable fluorescence over maximal fluorescence (F_v/F_m) was determined with a fluorometer (PEA, Hansatech, Norfolk UK) as reported in Maxwell and Johnson [42].

Statistical analysis

Each data point is the mean and SD of at least five fruits. The significant difference between treatments were detected by ANOVA with a 95% confidence level ($P = 0.05$) using StatGraphics Plus 4.1 program. For each parameter measured, ANOVA was performed on the data obtained from the last observation of the experiment.

Results and discussion

Symptoms of chilling damage

Figure 1A shows that in control fruits no change occurred in the coloring of the skin (nor on the subepidermal

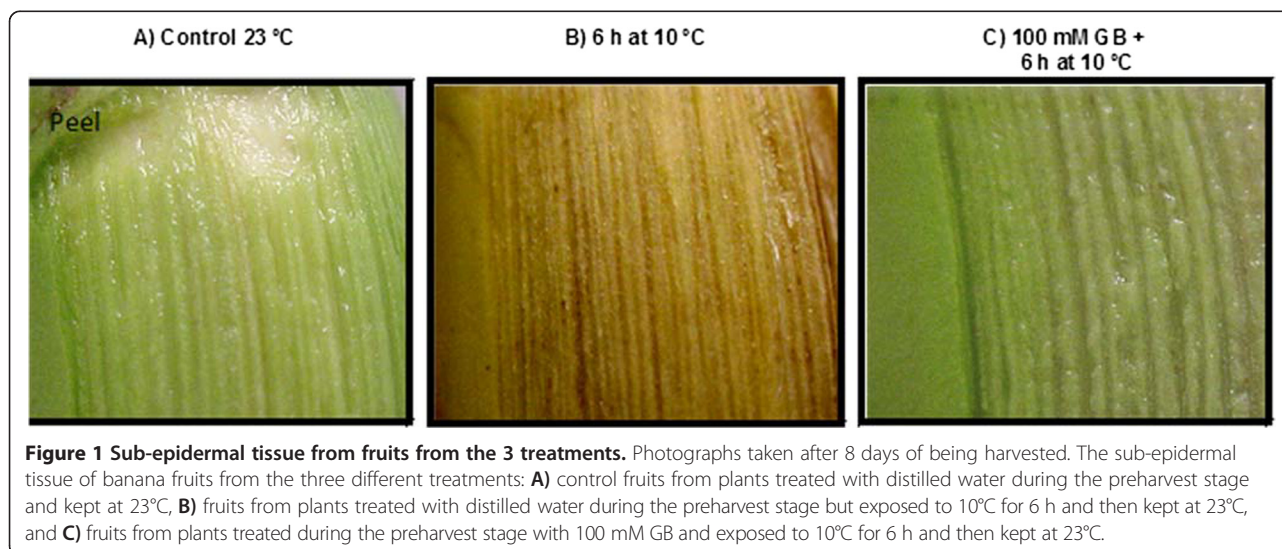


Figure 1 Sub-epidermal tissue from fruits from the 3 treatments. Photographs taken after 8 days of being harvested. The sub-epidermal tissue of banana fruits from the three different treatments: **A)** control fruits from plants treated with distilled water during the preharvest stage and kept at 23°C , **B)** fruits from plants treated with distilled water during the preharvest stage but exposed to 10°C for 6 h and then kept at 23°C , and **C)** fruits from plants treated during the preharvest stage with 100 mM GB and exposed to 10°C for 6 h and then kept at 23°C .

tissue) within 8 days of being stored at 23°C. When fruits were exposed to 10°C for 6 h (Figure 1B), they showed obvious signs of chilling damage (browning of the subepidermal tissues). However, when GB was applied during the preharvest stage, the postharvest chilling damage effects (browning) on the fruit was prevented (Figure 1C). These results suggest that banana (cv Giant Dwarf) suffers damage when exposed to chilling (10°C) even for a short exposure of 6 h, showing uneven ripening and epidermal browning acceleration; similar results were obtained by Morrelli et al. [43], who worked with five different varieties of banana exposed to temperatures of 5°C, 7°C, and 10°C, but for as long as 7 days.

Our results also suggest that exogenous GB (when applied to leaves prior to harvesting the fruits) may have been translocated to fruits and had prevented chilling-induced subepidermal browning of fruits. No data on endogenous content of GB are available in our experiment; however, in tomato plants, it was shown that GB was translocated to meristematic tissues including flowers and fruits of tomato when applied exogenously to the leaves and that resulted in increased tolerance to chilling [44].

Total phenols and activity of PPO

In control fruits, the concentration of total phenols increased four times as the process of ripening progressed (Figure 2A). However, when fruits were exposed for 6 h to temperatures of 10°C, the contents of phenols only increased two times irrespective of whether they were treated or not with GB during the preharvest stage.

In terms of PPO-specific activity in control fruits, the PPO-specific activity remained almost constant after 8 days of being harvested (Figure 2B). In chilled fruits, the activity of PPO showed a large peak at Day 3 to later return to basal levels by Day 4 and remained low at Day 8 of being harvested (Figure 2B).

On the contrary, for chilled fruits from plants treated with GB during the preharvest stage, although they also showed an increased PPO activity in response to chilling, this was very much attenuated (i.e., 2,500 $\mu\text{mol min}^{-1} \text{mg}^{-1}$ protein in GB-treated chilled fruits vs. 6,000 $\mu\text{mol min}^{-1} \text{mg}^{-1}$ protein in non-GB-treated chilled fruits).

Our results are in line with findings of other authors in apples and bananas. In apple, some cultivars showed a positive correlation between the degree of browning and enzyme activity. In some cultivars (Classic Delicious, RI Greening, McIntosh, and Cortland), PPO activity was directly related to degree of browning while in others (Empire, Rome, and Golden Delicious), the degree of browning was related more to the final phenolic concentration. High-performance liquid chromatography analysis of the phenolics in apple showed that the types of phenolic compounds in all cultivars were similar and that no one particular compound could account for the

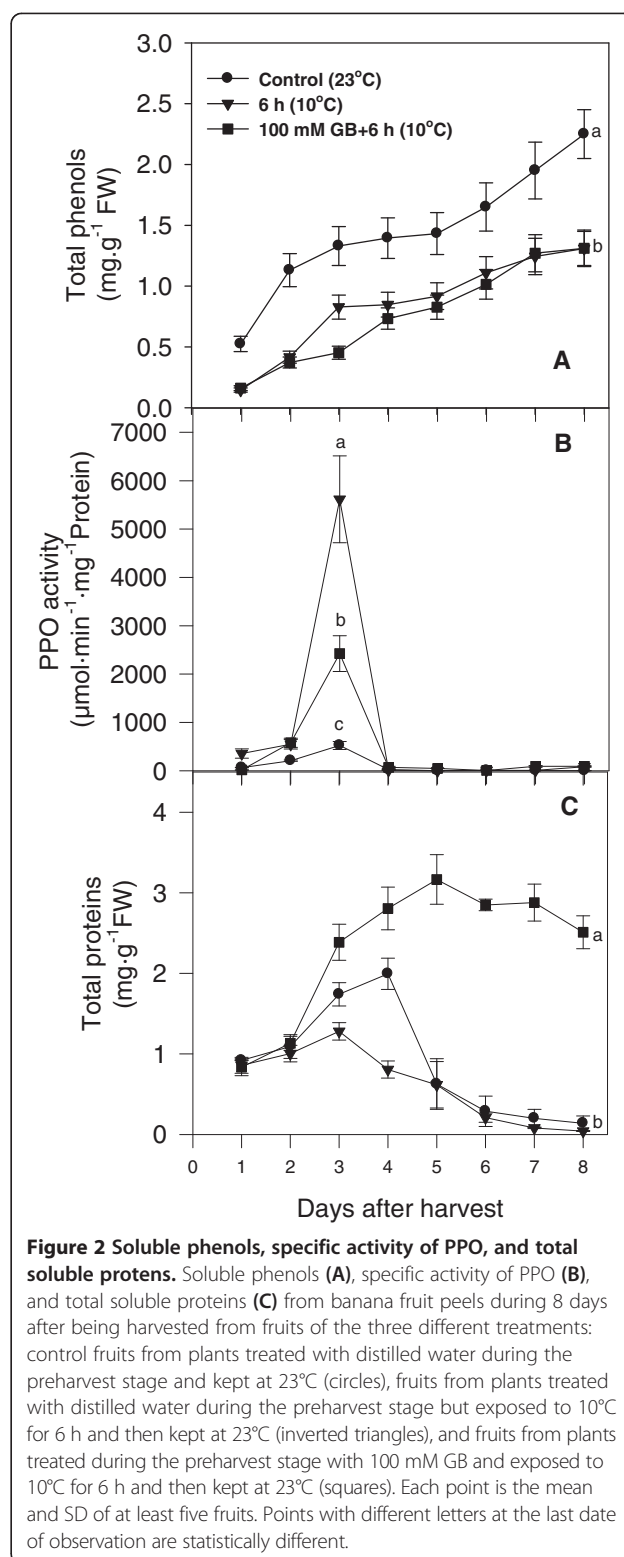


Figure 2 Soluble phenols, specific activity of PPO, and total soluble proteins. Soluble phenols (A), specific activity of PPO (B), and total soluble proteins (C) from banana fruit peels during 8 days after being harvested from fruits of the three different treatments: control fruits from plants treated with distilled water during the preharvest stage and kept at 23°C (circles), fruits from plants treated with distilled water during the preharvest stage but exposed to 10°C for 6 h and then kept at 23°C (inverted triangles), and fruits from plants treated during the preharvest stage with 100 mM GB and exposed to 10°C for 6 h and then kept at 23°C (squares). Each point is the mean and SD of at least five fruits. Points with different letters at the last date of observation are statistically different.

differences observed in browning [45]. They concluded that the increase of the phenolic compounds and the activity of enzymes could be the factor that causes the increase in the degree of browning of fruits when exposed

to low temperatures [45]; although, this was not the case in all varieties tested.

In banana, on the other hand, Nguyen et al. [21] noted that the development of chilling damage, which causes the fruit skin browning, was due to increased activities of the enzymes PAL and PPO. They found that the banana varieties Kluai Khai and Kluai Hom Tom, exposed to temperatures of 6°C and 10°C, had increased activity of PPO and a greater browning in the fruits. In contrast, Nguyen et al. [46] found that, in another variety of banana, PPO activity did not correlate with phenol content but that it was more related to PAL.

Protein content

In control fruits, protein content increased in the first 4 days after removal from the plant but declined afterwards being almost 0, 8 days after being harvested (Figure 2C). The decline in protein content was not associated to the climacteric peak as it occurred at Day 9 in those fruits (see Figure 3). Chilling caused a reduction in the protein content found in fruits 3 and 4 days after being harvested, when compared to control fruits, followed by a similar decline from Days 5 to 8 after harvest. In contrast, the protein content of chilled fruits from plants preharvestly treated with GB not only prevented the protein degradation associated with the exposure to low temperatures but also perhaps increased the rate of protein synthesis or prevented the ripening-related protein degradation, as protein content in those fruits increased about three times (from 1 to 3 mg of protein g⁻¹ FW) at Day 5 but remained high 8 days after being harvested. These results strongly point towards a protecting effect of GB on the rate of proteolysis caused by both the ripening process in control fruits and the chilling-induced proteolysis observed in chilled fruits. GB perhaps promoted the synthesis of new proteins related to the protection of the membrane or reduced the rate of protein degradation or both. There are few reports in banana documenting the changes in the pattern of proteins associated with fruit ripening, but certainly, the observed variation in the protein concentration should be related to the synthesis and hydrolysis of proteins involved in the ripening process. Brady and O'Connell [47] reported that much of the increase in early climacteric-phase protein synthesis is the result of an increase in the turnover of preexisting proteins. On the other hand, Hubbard et al. [48] mentioned that during the climacteric phase of banana fruits there is an increased synthesis of sucrose phosphate synthase (SPS).

Pigments content

Total chlorophylls

In control fruits, total chlorophyll content of peels was reduced from 55 to 20 μg g⁻¹ FW, only 1 day after being harvested, and decreases further from 20 to 0 μg g⁻¹ FW

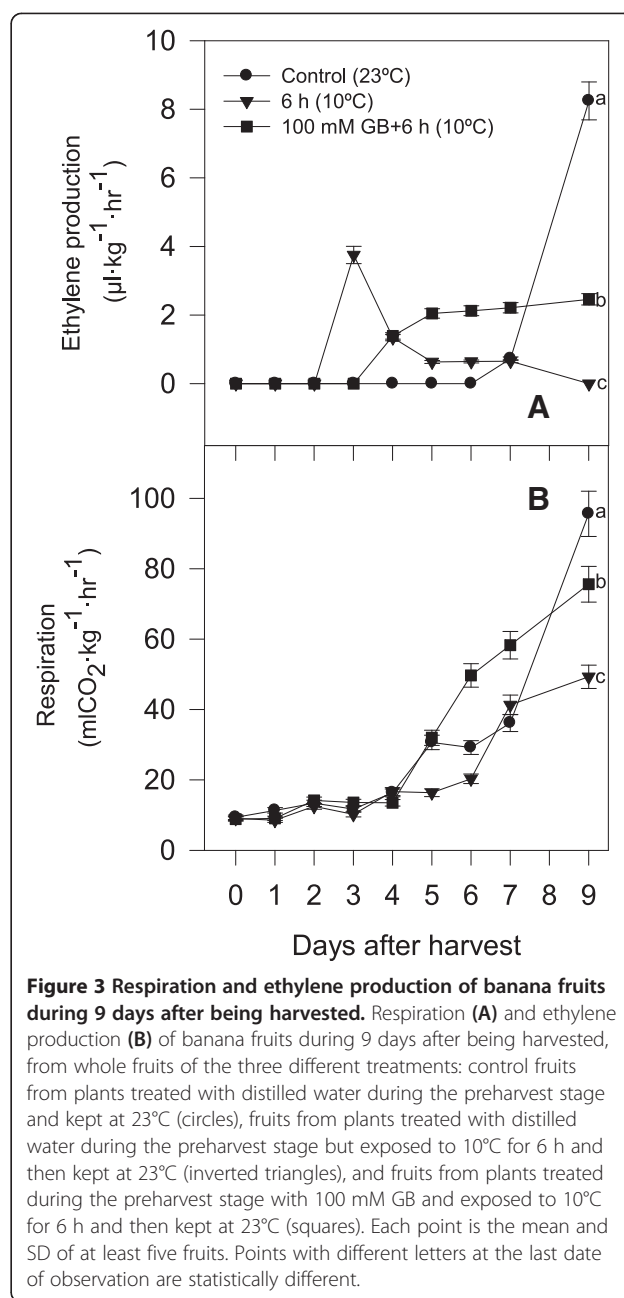
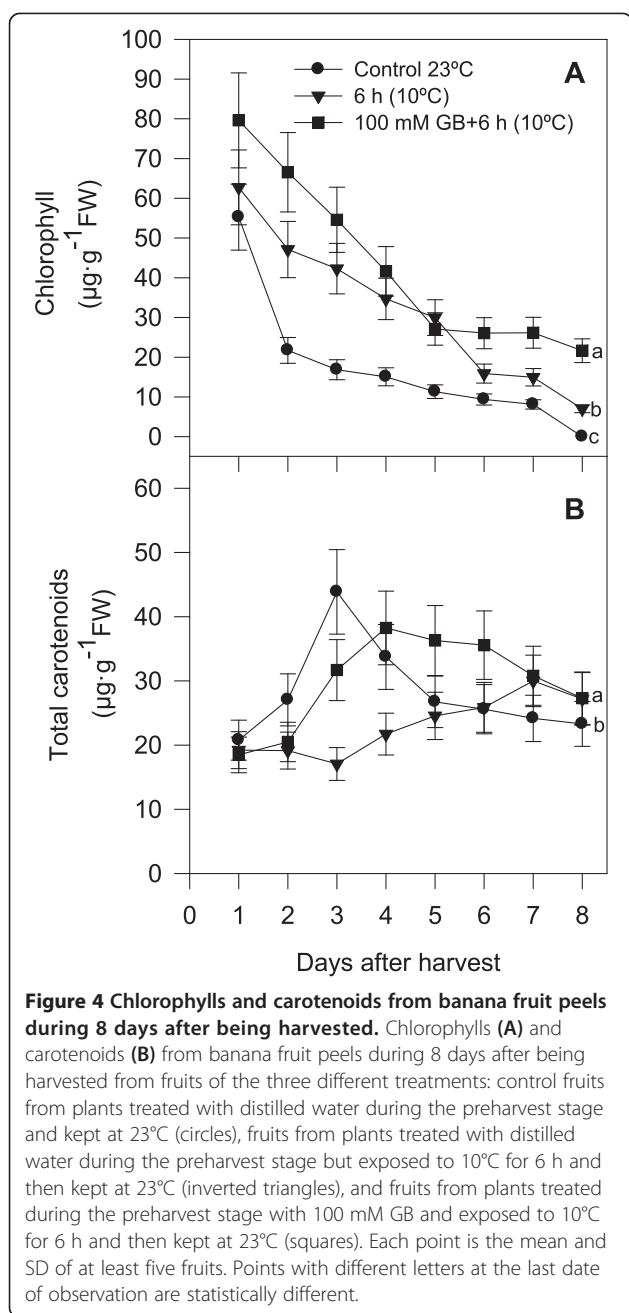


Figure 3 Respiration and ethylene production of banana fruits during 9 days after being harvested. Respiration (A) and ethylene production (B) of banana fruits during 9 days after being harvested, from whole fruits of the three different treatments: control fruits from plants treated with distilled water during the preharvest stage and kept at 23°C (circles), fruits from plants treated with distilled water during the preharvest stage but exposed to 10°C for 6 h and then kept at 23°C (inverted triangles), and fruits from plants treated during the preharvest stage with 100 mM GB and exposed to 10°C for 6 h and then kept at 23°C (squares). Each point is the mean and SD of at least five fruits. Points with different letters at the last date of observation are statistically different.

in the following 7 days (Figure 4A). In chilled fruits, the exposure to temperatures of 10°C for 6 h caused a reduction in the rate of chlorophyll degradation during the first 5 days after being harvested; however, it falls at Day 6 to values close to those of the control fruits. In contrast, in chilled fruits from plants treated with GB, the chlorophyll degradation rate was the lowest of the three treatments, remaining at values of 30 μg g⁻¹ FW from Days 5 to 8 after being harvested (Figure 4A). A protective effect of GB on chlorophyll degradation has been reported previously. Blackburn et al. [49] reported that GB can be



found in chloroplasts and prevents membrane damage caused by peroxidation associated to ROS.

Total carotenoids

In control fruits, the carotenoid content of peels increased from 20 to 45 $\mu\text{g g}^{-1}$ FW during the first 3 days but declined later reaching again values of 20 $\mu\text{g g}^{-1}$ FW, 8 days after being harvested. In chilled fruits, exposure to temperatures of 10°C for 6 h caused no damage on the carotenoid content during the first 3 days after being harvested. However, it increased from Days 5 to 8 to values close to

30 $\mu\text{g g}^{-1}$ FW. In contrast, in chilled fruits from plants treated with GB, during the first 4 days, the carotenoid content increased to values close to those of control fruits, remaining at values around 40 $\mu\text{g g}^{-1}$ FW, 6 days after being harvested and only decreased to values of 30 $\mu\text{g g}^{-1}$ FW at Day 8 (Figure 4B).

Production of ethylene and respiration

In control fruits, no ethylene was detected during the first 6 days after being harvested; ethylene increased slightly at Day 7 after harvest but showed a significant increase (eightfold) 8 days after being harvested. In chilled fruits, the exposure to temperatures of 10°C for 6 h caused an ethylene peak reaching values of 4 $\mu\text{l C}_2\text{H}_4 \text{Kg}^{-1} \text{hr}^{-1}$ as early as Day 3 after being harvested, falling again at Day 4 reaching values near 0 $\mu\text{l C}_2\text{H}_4 \text{Kg}^{-1} \text{hr}^{-1}$ at Day 9 after being harvested.

In contrast, in chilled fruits from plants treated with GB, ethylene also increased by Days 4 and 5 but remained as high as 2 $\mu\text{l C}_2\text{H}_4 \text{Kg}^{-1} \text{hr}^{-1}$ even after 9 days of being harvested (Figure 3A). This result indicates that the chilling caused a premature ethylene peak that resulted in an uneven ripening of the fruit. The preharvest application of GB mitigated this effect, and fruits, despite being subjected to chilling, did not show the premature ethylene peak observed in non-GB-treated chilled fruits. However, chilled fruits from GB-treated plants showed intermediate values of ethylene content between the non-chilled controls and those of the non-GB-treated chilled fruits.

In the case of respiration, in control fruits, low- CO_2 rates (10 $\text{ml CO}_2 \text{Kg}^{-1} \text{hr}^{-1}$) were detected during the first 4 days after being harvested; they increased slightly during Days 5 to 7, and by Day 9, they showed a significant rise to values near 100 $\text{ml CO}_2 \text{Kg}^{-1} \text{hr}^{-1}$. Chilled fruits (exposed to temperatures of 10°C for 6 h) again showed low CO_2 during the first 4 days; they did not show an early CO_2 peak as the control fruits did. Respiration rates in chilled fruits only increased to values of 30 $\text{ml CO}_2 \text{Kg}^{-1} \text{hr}^{-1}$ at Day 7 and to values of 50 $\text{ml CO}_2 \text{Kg}^{-1} \text{hr}^{-1}$ by Day 9. In contrast, in chilled fruits from plants treated with GB, CO_2 also remained low during the first 4 days, but from Day 5 onwards, it showed a continuous increase reaching values of 70 $\text{ml CO}_2 \text{Kg}^{-1} \text{hr}^{-1}$ by Day 9 after being harvested (Figure 3B). Those values were again intermediate between the high values observed in control plants and the low values observed in non-GB-treated chilled fruits. There were statistically significant differences between treatments at Day 9. GB perhaps protected proteins involved in the respiration process from chilling injury.

Under conditions of chilling stress, the rate of ethylene and CO_2 production usually increases [50–52]. Our results show that in fruit exposed to chilling, respiration declined 48% with respect to control. While the

production of ethylene showed a premature peak causing early and uneven ripening of the fruit. These disturbances in the metabolism of the fruit agree with previous reports [4,53]. According to Jiang et al. [54], in fruits stored at low temperatures, chilling injuries are due to the reduction in the ability to respond to the ethylene signal, and it has an impact on an abnormal maturation of the fruit. Banana fruit exposed to chilling are perhaps less sensitive to ethylene causing abnormal maturation. Climacteric fruit ripening is a process which is coordinated by the presence of ethylene perception by receptors.

Cell membrane stability

Exposure of banana fruits to chilling temperatures resulted in changes in the CMS measured as electrolyte leakage. Chilled fruits had electrolyte leakage values of 51% compared to values of 38% found in control fruits 10 days after being harvested. Chilling did not affect fruits that were previously treated with GB, as they showed a lower electrolyte leakage (29.45%) value that was even lower than that observed in control fruits (Figure 5A). This suggests that GB was protecting the integrity of the membrane perhaps by GB taking action as an osmoregulator that allowed membrane stabilization.

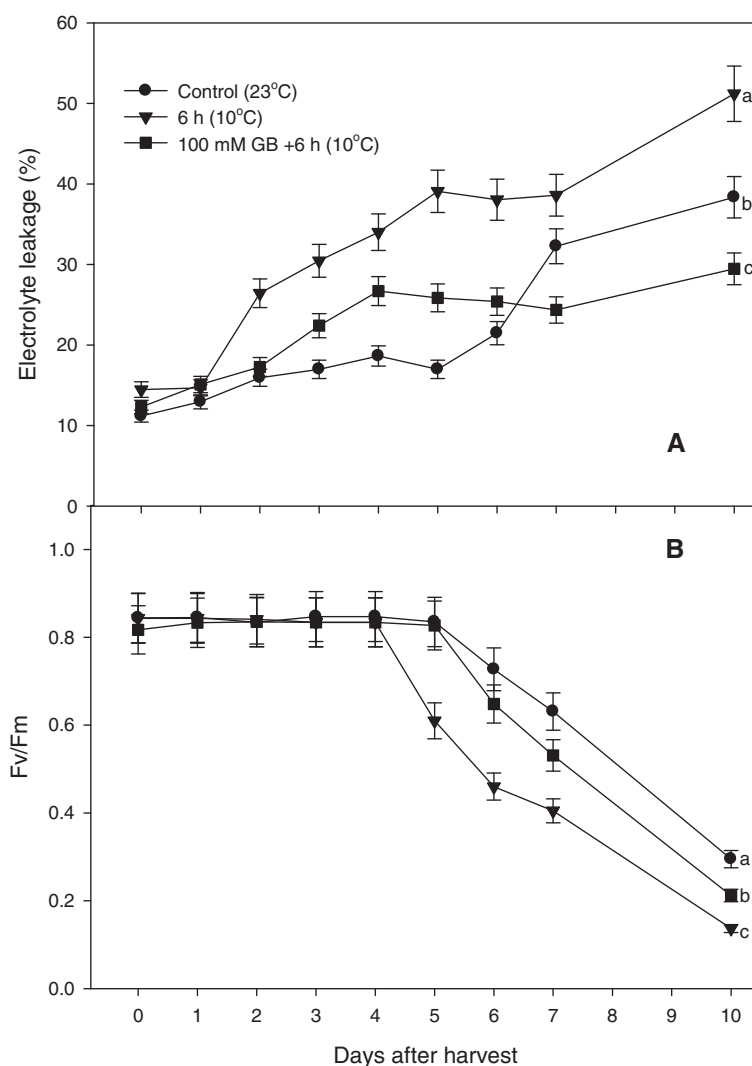


Figure 5 Electrolyte leakage and efficiency of photosystem II from banana fruit peels. Electrolyte leakage (A) and efficiency of photosystem II (B) from banana fruit peels during 10 days after being harvested from fruits of the three different treatments: control fruits from plants treated with distilled water during the preharvest stage and kept at 23°C (circles), fruits from plants treated with distilled water during the preharvest stage but exposed to 10°C for 6 h and then kept at 23°C (inverted triangles), and fruits from plants treated during the preharvest stage with 100 mM GB and exposed to 10°C for 6 h and then kept at 23°C (squares). Each point is the mean and SD of at least five fruits. Points with different letters at the last date of observation are statistically different.

Other authors also found that the exogenous application of GB allowed chilled tomato plants to maintain lower electrolyte leakage than controls [55]. More recently, Hu et al. [56] also found that exogenous GB reduced electrolyte leakage in ryegrass and contributed to ameliorate the adverse effects of salt stress.

Activity of photosystem II (F_v/F_m)

During the first 4 days after harvest, fruits from all three treatments showed no damage to the efficiency of PSII as measured by the ratio-variable chlorophyll fluorescence to maximal fluorescence (F_v/F_m). F_v/F_m values remained above 0.8 that indicate no damage to the efficiency of PSII by Day 4. In fruits exposed to chilling temperatures (10°C for 6 h), F_v/F_m did fall abruptly by Day 5 reaching values as low as 0.14 by Day 10 of being harvested (Figure 5B). Preharvest application of GB lowered the decline of F_v/F_m in chilled fruits, as by Day 5, F_v/F_m remained as high as in control fruits. By Day 6, F_v/F_m began to fall, but this group always showed intermediate F_v/F_m values between chilled and control groups from Day 6 to Day 10 after being harvested. This might indicate that the application of GB reduced the damage to membranes caused by chilling, perhaps by showing a protective effect on the structure and function of the PSII complex. Previous reports exist of GB's ability to protect the photosynthetic oxygen-evolving complex [16,17].

Although it is argued that a decreased F_v/F_m in fruits is more associated to the process of fruit senescence [57], resulting from chlorophyll degradation during ripening [58,59], in banana fruits, it has been argued that F_v/F_m declines when chlorophyll content decreases as reported by Sanxter et al. [60] or by membrane degradation during ripening as suggested by Marangoni et al. [61]. F_v/F_m has been mentioned as a good indicator for membrane damage caused by chilling [62,63], and other authors also found that the exogenous application of GB allowed chilled tomato plants to maintain higher F_v/F_m values than non-GB-treated chilled fruits [55].

In conclusion, exposure of banana fruits to chilling stress caused fruit browning, an increase in the activity of the PPO when compared to control fruits and consequently a reduction in the concentration of phenols. Thus, the browning of the chilled fruits was not related to the content of phenols. Chilling also caused uneven ripening, an increased degradation of proteins, a reduction in the carotenoid content, and an increase in the content of chlorophylls, membrane damage (increased electrolyte leakage), and reduced efficiency of photosystem II (as shown by low F_v/F_m values). Chilled fruits also showed an acceleration of ethylene production and a reduction in fruit respiration.

On the contrary, the preharvest foliar application of GB (100 mM) to banana plants reduced the biochemical and physiological alterations caused by chilling injury on their harvested fruits. When fruits from GB-treated plants were exposed to 10°C for 6 h, the sub-epidermal browning was attenuated and the accumulation of total phenols was delayed. Similarly, the electron efficiency of photosystem II (F_v/F_m) as well as the CMS remained higher (24% and 27%, respectively) than in chilled fruits from non-GB-treated plants. It might be possible that the reduction of chilled-induced fruit browning by GB could be associated to a protection of the membrane integrity and protection of photosystem II. GB also seemed to protect proteins from chilling damage, as GB-treated fruits not only did not decrease protein content but even had increased protein content than both the chilled fruits or even the control fruits. GB also increased respiration and prevented the accelerated ripening associated with a premature appearance of an ethylene peak in non-GB-treated chilled fruits. However, it did not have effect on the chilling-related changes on the concentration of phenols or chlorophyll content, but it promoted the synthesis of carotenoids.

Further research is needed to elucidate the cellular and molecular mechanisms that may explain how the preharvest foliar application of GB induced a subsequent protection of fruits to chilling injury. The protection was related to processes of membrane stabilization, electrolyte leakage prevention, and protection to photosystem II, maintaining the photosynthetic efficiency of this protein complex.

In other species such as tomato, it has been suggested that, in addition to protecting macromolecules and membranes directly, GB may enhance chilling tolerance by inducing H_2O_2 -mediated antioxidant mechanisms, e.g., enhanced catalase expression and catalase activity [55]. These results suggest that GB not only can protect temperate plants from chilling injury including *A. thaliana* (see review by Ashraf and Foolad [27]), but also has a protective effect in tropical plants such as banana. The results of the present work suggest that foliar application of GB might be used in banana plantations to reduce the fruit damage caused by their postharvest storage at low temperature. Our results also might serve as the basis to further evaluate whether the foliar application of GB could also reduce the damage caused by frost-associated chilling injury on banana fruits in the field.

Finally, this study also might give support to the possibility to improve chilling tolerance in this monocot tropical fruit by engineering its capacity to over accumulate GB, in a similar way as it has been achieved in maize [64], rice [28,31], and more recently in sweet potato [65].

Competing interests

The authors declare that they have no competing interests.

Author's contribution

Luis Carlos Rodríguez-Zapata provided financial support of the experiments from a research grant. Francisco Espadas y Gil assisted in the conduction of field experiments. Susana Cruz Martínez conducted field experiments and most of lab determinations. Carlos Talavera May and Fernando Contreras-Marin assisted in the conduction of field experiments. Gabriela Fuentes contributed to the experimental design and in the manuscript writing. Enrique Sauri-Duch offered expert advice during the conduction of the enzymatic activity assays. Jorge M. Santamaría provided general conception and coordination of the experiments, interpretation of results and manuscript writing. All authors read and approved the final manuscript.

Author details

¹Unidad de Biotecnología, Centro de Investigación Científica de Yucatán, Mérida 97200, México. ²Instituto Tecnológico de Mérida, Mérida 97200, México.

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