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BOILING HEAT TRANSFER TO LIQUID HYDROGEN

AND NITROGEN IN FORCED FLOW

By James P. Lewis, Jack H. Goodykoontz, and John F. Kline

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Page 19, equation (All): The symbol R should be R_0 .

Page 52, figure 17: The abscissa scale label should be Wall superheat, $t_{w,i}$ - t_{sat} , ^OR instead of Water superheat, $t_{w,i}$ - t_{sat} , ^OR.

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SUMMARY

Boiling heat transfer to liquid hydrogen and nitrogen was investigated experimentally. Results are presented from a study of bulk boiling inside a cylindrical tube under vertically upward forced-flow conditions. A 0.555-inch-inside-diameter and $16\frac{1}{8}$ -inch-long electrically heated stainless-steel tube was used. The range of variables studied for hydrogen were mass velocity of 2850 to 17,000 pounds per hour per square foot, local heat flux of 3600 to 40,000 Btu per hour per square foot, inlet pressure of 30 to 74 pounds per square inch absolute, and inlet subcooling of 0° to 9° R. Nitrogen test conditions were mass velocity of 15,000 to 56,000 pounds per hour per square foot, local heat flux of 2300 to 40,000 Btu per hour per square foot, inlet pressure of 47 to 56 pounds per square inch absolute, and inlet subcooling of 1° to 6° R.

The axial distribution of the tube-wall temperatures is presented. A transition in the type of boiling heat transfer was obtained. The critical heat flux corresponding to this transition was determined over a range of flow and heating rates and local qualities. At specific combinations of flow and transition location, a range of critical-heat-flux values was obtained and maximum values were determined. The maximum critical heat flux increased with increasing fluid-flow rate and decreased with increasing length of tube before transition. Similar variations of the maximum critical heat flux have been reported for water. The tube-inner-wall temperatures upstream of transition were essentially uniform and were only slightly greater than the fluid saturation temper-The wall-temperature profiles downstream of transition generally ature. resembled those obtained in film-boiling studies and appeared to be strongly dependent upon local quality at the point of transition. Maximum wall temperatures of 900° and 1800° R were obtained with hydrogen and nitrogen, respectively. Fluctuations of pressure, flow rate, and temperature occurred during some of the boiling tests. Under some conditions, maximum critical-heat-flux values were attained during steady-state operation with fluctuations. In other cases the fluctuations became uncontrolled, and critical-flux values less than the maximum values were obtained upon restabilization of the test conditions. No measurable pressure drop across the test section was obtained at any condition.

INTRODUCTION

Liquid hydrogen has been proposed for use in several advanced propulsion systems. In these systems, hydrogen may be used both as a propellant and as a coolant. The low boiling point of hydrogen and the desirability of storing it in the liquid state in addition to the requirements of some systems for gaseous hydrogen necessitate a knowledge of two-phase flow and heat transfer for hydrogen. Information is especially desired for boiling heat transfer of hydrogen under forced-flow, confinedgeometry conditions. In addition, the wide variance of the physical properties of hydrogen from those of more conventional fluids make it attractive as a test fluid in research directed towards a more complete understanding of the general problem of boiling heat transfer.

Information in the literature concerning boiling heat transfer, primarily for the case of pool (or pot) boiling and usually for conventional fluids, such as water and alcohols, is extensive. Present thinking with respect to pool boiling and related investigations with hydrogen are summarized in reference 1. Pool boiling is characterized by three distinct modes of boiling, namely, nucleate, transition, and film boil-Analytical and empirical relations between the heat flux (or heating. transfer coefficient) and the wall- to fluid-temperature difference have been obtained for pool boiling. Pool boiling also exhibits a distinctive value of heat flux obtained at the boundary between the nucleate and transition boiling regions that has been variously termed maximum nucleate flux, departure from nucleate boiling (DNB), or burnout heat flux. Analytical and experimental correlations of the maximum nucleate flux with fluid properties and test operating variables have been made with varying degrees of success (ref. 1).

For the case of forced flow in confined geometries, the current understanding of boiling heat transfer is much more limited, especially for the case of net vapor generation. Again, considerable data have been obtained and several correlations have been proposed (refs. 2 to 6). Much of the available data are incompletely presented or contradictory, and the correlations, which successfully relate the results of a single study. have not been successful when applied to other tests or fluids of widely differing properties. The experimental data of several investigations (refs. 2 to 4) have indicated the existence of a critical heat flux, which somewhat resembles the maximum nucleate flux obtained in pool boiling, in that a well defined reduction in the heat-transfer coefficient is obtained. Some data of this type that resemble the usual results obtained for pool boiling were obtained in limited tests of boiling hydrogen (ref. 7). Data were obtained in reference 8 for hydrogen for the region that might be termed film boiling in tubes with forced flow. For the investigations of boiling heat transfer with forced flow in confined geometries, there is a wide variation in the assumptions regarding the physical nature of the heat-transfer process and in the definition of the critical heat flux.

Because of the aforementioned limitations of present knowledge, a program was initiated at the Lewis Research Center to investigate boiling heat transfer to liquid hydrogen under conditions of forced flow inside a vertical tube. The principal objective of the investigation was to determine in engineering terms the effect of the operating variables (flow rate, pressure, liquid inlet subcooling, heating rate, and tube geometry) on the mode of heat transfer and their relation to the value of the critical heat flux. The program was directed towards conditions resulting in net vapor generation. In addition, information on flow instability and its effect on heat transfer were desired.

The test apparatus consisted of a pressure-fed, once-through system with an electrically heated vertical tube of 0.555-inch inside diameter and $16\frac{1}{8}$ -inch length. Both subcooled liquid para-hydrogen and liquid nitrogen flowed through the tube in vertical upflow. The range of variables investigated was limited to the following:

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	Hydrogen	Nitrogen
Mass velocity, lb/(hr)(sq ft)	2850 to 17,000	15,000 to 56,000
Local heat flux, Btu/(hr)(sq ft)	3600 to 40,000	2300 to 40,000
Liquid inlet subcooling, ^{OR}	0 to 9	1 to 6
Inlet pressure, lb/sq in. abs	30 to 74	47 to 56

Tube-exit qualities ranged from essentially 0 to 1.0 (with superheat), and transition from a relatively high to a lower value of heat-transfer coefficient occurred over a range of axial locations from tube entrance to exit. A few tests were made with cold hydrogen gas flowing through a heated tube. The results obtained from the investigation are presented in tabular and graphical form.

APPARATUS

General Arrangement

The test equipment included a liquid-supply Dewar, a controlled source of pressurizing gas, a flash cooler to subcool inlet test liquid, the test section and electric power supply, inlet and exit control valves, a vaporizer, an orifice-type flowmeter, and vent, pressure relief, and purge systems (fig. 1). In practically every case, the test liquids (para-hydrogen and nitrogen) were pressurized by their own gases. Gaseous helium was used for system purging and inerting. The vaporizer was used to ensure that only a fully vaporized product would pass through the flow orifice. All fluid lines from the supply Dewar to a point past the end of the test section were insulated with a vacuum jacket.

Test Section

The test-section assembly consisted of the electrically heated tube, inlet and outlet chambers, a vacuum jacket, and test instrumentation (fig. 2). The tube was made of type 304 stainless steel with a 0.555inch inside diameter and a 0.035-inch-thick wall. As indicated in figure 2, the effective heated length of the tube was $16\frac{1}{2}$ inches, measured between the inner faces of the end flanges. All distances along the tube from the tube inlet were measured from the downstream side of the inlet flange. The actual inlet end of the tube extended 1/2 inch up-The inlet chamber consisted of a $1\frac{1}{2}$ -inch-insidestream of this point. diameter stainless-steel cylinder attached to the inlet flange. The inlet chamber, which was lined with 1/8-inch-thick thermal insulation, was designed to provide a low-velocity plenum at the test-section entrance and to minimize heat leakage from the heated test section to the incoming Two copper bus bars, diametrically opposite, were connected beliquid. tween the test-section inlet flange and the bottom flange of the vacuum jacket, which also served as the ground side of the electrical circuit. These bus bars were 4 inches long with a cross section of 1/8 by 1 inch. The outlet chamber was designed to minimize heat losses from the tube, to provide an electrically insulated, low-electrical-resistance connection to the tube, and to provide a thermally insulated mixing chamber, in which the test fluid could come into thermal and phase equilibrium. The outlet chamber contained an inner liner consisting of stainless steel and Teflon. This liner, which was not attached directly to the tube,

was designed to allow cool gas to accumulate between it and the outer shell and thus to act as thermal insulation. The outlet section was connected to a 1-inch-outside-diameter copper tube that passed through the vacuum jacket and was electrically isolated from it. Two conically shaped mixing screens were placed in the outlet section. The vacuum jacket around the test-section assembly consisted of stainless-steel

flanges with O-ring seals, a $3\frac{1}{2}$ -inch-diameter Lucite tube, and an aluminum-foil radiation shield.

Flash Cooler

The flash cooler shown in figure 1 was provided to supply subcooled liquid to the test section. The cooler consisted basically of three concentric tubes. The flow of the liquid to the test section was brought through the small innermost tube. Some liquid was allowed to pass into the annular space around the inner tube through bleed holes at various points along the length of the subcooler. The pressure in this annulus was maintained intermediate between atmospheric pressure and the supply Dewar pressure by a throttle valve. The liquid entering the annular space vaporized because of the drop in pressure and thus cooled the inner supply tube. The inner tube had a 0.38-inch outside diameter with a 0.032-inch wall. Stainless-steel rods, 1/4 inch in diameter, were inserted into the inner tube to promote cooling of the supply liquid. The outer annular space provided a vacuum jacket for thermal insulation.

Electric Power Supply

The tube was heated by alternating current supplied through a $2\frac{1}{2}$ -

kilowatt, 60-cycle transformer with a maximum current rating of 500 amperes. The power to the test section was controlled by a variable autotransformer in the primary circuit. The current to the test section was measured by a laboratory-quality ammeter connected to a current transformer with a ratio of 100. Voltage drops across the tube at various locations were measured with a Ballantine vacuum-tube voltmeter.

Instrumentation

Instrumentation was provided to measure the inlet and the exit fluid bulk temperatures, the tube-wall temperatures, the inlet-fluid pressure, and the test-fluid flow rate.

Fluid temperatures. - The fluid bulk temperatures were measured by carbon resistors in the inlet and the exit of the test section (see fig. 2). The carbon resistors at the test-section outlet were located both above and below the mixing screens. The carbon resistors were hermetically sealed in a protective sheath about 0.1 inch in diameter by 0.2 inch long. The carbon resistors acted as one arm of a bridge circuit, the output of which was recorded on a self-balancing potentiometer. The slope of the temperature-resistance curve was obtained in a laboratory calibration and was essentially invariant. Shifts of the curve occurred, however, that required daily adjustment with a trimming resistance at a known temperature condition. The fluid temperature at the orifice flowmeter was measured with a copper-constantan thermocouple. The overall accuracy of the fluid bulk temperatures is estimated at approximately $\pm 0.5^{\circ}$ R.

<u>Wall temperatures.</u> - The temperatures of the tube wall and of adjacent sections were obtained with copper-constantan thermocouples. The thermocouples were soldered to the outside of the tube wall and the leads were wrapped around the tube several times and were finally wrapped with glass-fiber tape. The tube-wall thermocouples were positioned in one longitudinal plane and their axial locations are given in table I. The positions of the thermocouples that were installed on the inlet section, the outlet section, and the ground bus are also given in table I. These thermocouples were used to monitor the flow of heat to and from the test section. The constantan wire from each thermocouple junction was led without interruption to individual reference junctions located in a liquidnitrogen bath at atmospheric pressure. Copper leads led from the bath to a manual selector switch. The thermocouple voltage was bucked by a 1millivolt voltage to obtain positive values, and the resultant signal was recorded on a self-balancing potentiometer. The calibration of the thermocouples was determined from the National Bureau of Standards calibration (ref. 9) and laboratory calibration checks. The calibration indicated a very low sensitivity for the copper-constantan thermocouples near liquid-hydrogen temperatures. Wall temperatures, however, were obtained from approximately 40° to 1800° R. Above 1200° R, an extrapolation of the curve of reference 9 was used. The sensitivity and accuracy of the thermocouple readings are indicated by the following table:

	Liquid-hydrogen temperature (45° R)	Liquid-nitrogen temperature (160 ⁰ R)	Room tempera- ture (530º R)
Sensitivity, mv/ ^O R Chart reading limit, mv Chart reading limit, ^O R	0.004 0.01 to 0.02 2.5 to 5	0.01 0.01 to 0.02 1 to 2	0.022 0.01 to 0.02 0.5 to 1

The thermocouple calibration points also showed a scatter of approximately $\pm 3^{\circ}$ R at liquid-hydrogen temperatures and $\pm 1^{\circ}$ R at liquid-nitrogen temperatures. Approximately the same scatter was obtained from the actual tube-wall thermocouples during no-heat runs. The tube-wall thermocouples were attached to the outside of the tube wall. The temperature of interest, however, is that of the inner surface. An analysis and computation of the temperature drop through the tube wall is given in appendix A for the case of negligible axial temperature gradients. This analysis indicates wall drops of up to 20° R for liquid-hydrogen conditions and up to 7° R for liquid-nitrogen conditions over the range of the test heat fluxes.

<u>Pressure</u>. - The fluid pressures were sensed with strain-gage-type transducers and were continuously recorded on a high-speed recording potentiometer (0.3-sec full-scale travel). Pressure was sensed at the test-section inlet (see fig. 2) by a transducer having a range of 0 to 100 pounds per square inch absolute and an overall accuracy of ±0.5 percent of full scale. Initially a differential pressure transducer was installed to measure the pressure drop across the test section. Since no measurable pressure drop was obtained at the largest flow and vaporization conditions, the downstream pressure tap was removed to aid in eliminating flow and pressure oscillations. The fluid pressure far downstream of the tube exit (upstream of the exit control valve) was monitored on a visual gage but showed no significant drop from the test-section inlet pressure. The pressure in the flash cooler was also sensed by a strain-gage transducer.

Flow rate. - The test-fluid flow rate was measured by a sharp-edge orifice downstream of the vaporizer. The vaporizer ensured that all the fluid was in the gaseous phase and at a temperature at which fluid properties are well known. The discharge coefficient was determined with water for a range of Reynolds numbers. The orifice pressure and pressure drop were measured with strain-gage-type transducers and recorded on a self-balancing potentiometer. The flow-rate measurements are estimated to have an accuracy of ±2 percent.

PROCEDURE

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Establishment of Test Conditions

Obtaining information on boiling heat transfer for various heating rates at several pressure levels and over a range of flow rates in a systematic way was desired in order that the effects of each variable could be determined. It was also desired to have the test liquid enter the test section slightly subcooled and to study heat transfer and twophase flow in forced flow over as great a range of fluid quality as possible and to obtain a critical heat flux at arbitrary locations along the tube axis. (The critical heat flux is defined as the flux immediately before the transition from the high upstream heat-transfer coefficient to a lower value.) Completely systematic operation was not always possible because of limitations of the test equipment and of the boiling process itself. In addition, operation at a precise preselected condition was difficult to attain because of fluctuations of flow and pressure that occurred in the system.

The general operating procedure consisted of setting conditions of flow rate, pressure, and inlet subcooling without heat addition and then gradually increasing the heat to the test section in small increments until the desired condition was obtained. As heat was added to the system, the flow and pressure conditions changed and had to be continually readjusted. The most consistent and repeatable results were obtained by always increasing the heat control setting and/or decreasing the flow rate. The range of test variables for the investigation of boiling heat transfer were test-section pressure, 30 to 74 pounds per square inch absolute; mass velocity, 2850 to 17,000 pounds per hour per square foot for hydrogen and 15,000 to 56,000 pounds per hour per square foot for nitrogen: inlet subcooling of 0° to 9° R for hydrogen and 1° to 6° R The heated-tube-wall temperatures varied from 36° to for nitrogen. 1800⁰ R. The point of transition from a high to a lower heat-transfer coefficient was obtained at various locations along the length of the tube. Occasional unheated runs were made before and after a heated For an unheated run made after a heated condition, the flow and run. pressure controls were left unchanged in order that the effect of boiling on the flow conditions might be studied. A few runs were made in which cool hydrogen gas flowed through the tube at nominal pressures of 50 and 70 pounds per square inch absolute, inlet temperatures of 46°

to 82° R, and mass velocities of 7800 to 13,000 pounds per hour per square foot.

Data Reduction and Computations

All wall temperatures were obtained from the thermocouple chart readings and the aforementioned copper-constantan thermocouple calibration. Inner-tube-wall temperatures for a negligible axial temperature gradient were obtained from the calculations of appendix A. All pressures were read directly from the recorder charts. The flow rate was computed by the standard ASME orifice equations. Fluid properties were taken primarily from National Bureau of Standards sources (refs. 9 and 10).

The local heat flux was computed from the measured current and tube-outer-wall temperature by equation (All).

The local vapor quality was obtained from a heat balance by the relation

$$x = \frac{Q - wc_p(t_{sat} - t_{in})}{wh_{fg}}$$
(1)

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(All symbols are defined in appendix B.) For the special case of negligible axial temperature gradient (hence, constant heat flux), the quality is given by

$$x = \frac{4\left(\frac{q}{G}\right)\left(\frac{L}{D}\right) - c_{p}(t_{sat} - t_{in})}{h_{fg}}$$
(2)

Heat balances were computed for a few cases by comparing the enthalpy rise of the fluid through the test section with the amount of electric heat supplied to the test section. The heat balance could be computed only for cases with subcooled inlet liquid and superheated exit vapor (except for the hydrogen-gas runs) because there was no independent means of measuring quality. The heat balances agreed in most cases within ±10 percent. Usually the heat input was greater than the measured increase in fluid enthalpy, which indicated a heat loss. The main sources of heat loss (or gain) are the vacuum jacket, the copper ground bus, and the inlet and the exit sections. Calculations indicated that heat transfer across the vacuum jacket to the test section was negligible. Conduction through the copper ground bus was into the inlet section and was less than 5 percent of the heat generated in the tube. The heat loss from the tube through the walls of the inlet section was less than 2 percent of the heat generated in the tube. Evaluation of the heat loss at the exit of the test section was impossible. An additional source of error arose from the possible nonequilibrium of temperature and the phase of the test

fluid at the points of measurement in the inlet and the exit sections. The heat balance, however, was satisfactory and within the accuracy expected from the individual measurements.

RESULTS AND DISCUSSION

Tabulation of Data

The data obtained in 160 separate runs are tabulated in table II. Included in the table are data for runs both with liquid hydrogen and with liquid nitrogen, both heated and unheated, and also heated and unheated gaseous-hydrogen runs. The original data consisted primarily of the tube-outer-wall temperatures and the fluid exit bulk temperature obtained for various tests conditions of pressure, inlet fluid temperature, flow rate, and heating rate. For cases in which significant fluctuations of tube-wall temperatures occurred, the magnitudes of such fluctuations are also tabulated. The runs are numbered in chronological order. An omission in the run-number sequence indicates an aborted run or a significant change in the testing program. Also presented in table II are the temperatures measured on the electrical ground bus and in The table also contains the calcuthe inlet and the outlet sections. lated values of the fluid inlet subcooling, the fluid exit superheat, the critical heat flux (or the uniform flux on the tube if it is below the critical value). the position of the point of transition in heat transfer (termed the critical-boiling-length-to-diameter ratio), and the local quality at the point of transition (termed the critical quality). The remarks tabulated for each run are based on observations made during the test and also on subsequent study of the data.

Tube-Wall Axial-Temperature Profiles

The tube-outer-wall temperature profiles along the length of the tube for liquid-hydrogen tests at a pressure of approximately 50 pounds per square inch absolute and an average inlet subcooling of 2° R are presented in figure 3 for various heating rates at two different nominal mass velocities. Also included in the figure are the temperatures of the inlet and the outlet sections and a schematic diagram of the testsection geometry, including thermocouple locations. All the profiles of figure 3 have the same general shape but show trends with respect to heating rate and mass velocity. Starting at the tube inlet, the wall temperatures are essentially constant until a sudden temperature rise is obtained at various downstream locations. Following the initial sharp rise, the slope of the temperature profile decreases and in some cases the curves appear to approach a constant temperature. Listed in figure 3 are the values of the local heat flux existing immediately upstream of the point of temperature rise. This heat flux, arbitrarily termed the critical heat flux, is indicative of a boiling heat-transfer condition at which, for a given flow and pressure, a transition occurs from a relatively large heat-transfer coefficient to a smaller coefficient at a specified position along the tube axis. (The flux downstream of the transition point is larger than the critical flux because of the increase in tube electrical resistance, but the proportionate increase in flux is much less than the increase in wall temperature.) The critical flux may not correspond to the maximum nucleate or burnout flux obtained in pool boiling; for the terms to be synonymous, evidence would be required that the critical heat flux results from surface ebullition.

All the data of figure 3 show an increase in the critical heat flux as the location of transition moves upstream. This inverse relation was also found in tests with water for transition occurring at the exits of tubes of various lengths (ref. 4). The temperature-rise curves for the nominal mass velocity of 12,000 pounds per hour per square foot (fig. 3(a)) show a more pronounced change in slope and tend to approach a constant value of temperature sooner than those for the lower flow rate (fig. 3(b)). These effects are probably related to the lower qualities at the critical point obtained at the higher flow rate; however, a difference in the two-phase flow pattern (void-fraction distribution) is felt to be the controlling factor.

All the wall temperatures of figure 3 upstream of transition are uniform along the tube within the limits of the instrumentation and show a small and nonsystematic variation between runs. The inner-wall temperatures can be obtained by subtracting the wall-temperature drop (given in appendix A) from the outer-wall temperatures of figure 3. The innerwall temperatures are approximately 12° and 3° R above the inlet fluid saturation temperature for the conditions of figures 3(a) and (b), respectively. These small temperature differences are not considered accurate enough for further analysis because of the inherent inaccuracy and lack of sensitivity of the temperature measurements at these low temperatures (40° to 70° R).

Since the highest wall temperatures obtained with hydrogen never exceeded 900° R and in most cases were less than 600° R, safe operation over a considerable range of conditions in a region equivalent to film boiling with forced flow seemed possible. Similar magnitudes of wall temperature were obtained in reference 8.

The tube-wall temperature profiles obtained with liquid nitrogen as the test fluid were generally similar to the results with hydrogen. (All nitrogen data are given in table II(b).) The rise in wall temperature following transition was much steeper for nitrogen than for hydrogen and the high temperatures (up to 1800° R) obtained finally caused failure of the test apparatus. During operation with liquid nitrogen,

attempts to obtain transition upstream of the tube exit generally resulted in unstable conditions with extreme fluctuations of flow, pressure, and wall temperature.

A few tests were made in which cool hydrogen gas flowed through the heated tube. These tests were made to obtain tube-wall axial temperature profiles for conditions of gas convective heat transfer for comparison with the temperature profiles obtained with boiling heat transfer. The profiles shown in figure 4 are for turbulent convective heat transfer and are considerably different from those obtained with boiling heat transfer (fig. 3). For the convective heat transfer, the tube-wall rise always started close to the tube inlet and the wall-temperature rise was generally more gradual than for the boiling heat-transfer tests. The shape of the convective wall-temperature profiles results primarily from entrance effects and variations in the tube-wall- to fluid-bulk-temperature ratio. The shape of the wall-temperature profiles for the boiling case (fig. 3), however, reflects changes in phase and in the boiling heat-transfer mechanism.

Temperature profiles are presented in figure 5 for boiling hydrogen at two constant values of transition location for several values of mass velocity and critical heat flux. An increase in mass velocity tends to skew the temperature-rise curves by increasing the slope at first and then by decreasing it at downstream locations. This effect of mass velocity on the temperature-rise curves was previously shown by the data of figure 3.

Critical Heat Flux

The critical heat flux for the conditions of this investigation corresponds to the local heat flux just upstream of the location of a sudden rise in wall temperature. For the critical heat flux at the end of the tube, transition was defined as the point corresponding to the conditions existing just previous to the increase in flux that first caused the thermocouple located 1/2 inch from the tube exit to rise. The critical heat flux obtained for boiling liquid hydrogen at a pressure of approximately 50 pounds per square inch absolute is presented in figure 6 as a function of mass velocity for four nominal values of the criticalboiling-length-to-diameter ratio L/D. All these curves show a significant increase in the critical flux with increasing mass velocity, but the slope of the curves generally decreases with increasing mass veloc-Generally the critical flux increases as the L/D decreases for ity. constant mass velocity. Similar relations were found for water in reference 4. in which transition occurred at the exit of tubes of various lengths and diameters. In the present investigation, the length variation was obtained by causing transition to occur at various locations along a tube of constant length and diameter. The data of figure 6 show

an increased scatter with reduction in the critical L/D. The points of greatest heat flux in figure 6(d) also had the greatest fluctuations of wall temperature, flow rate, and pressure but were essentially steadystate conditions. The points along the lower envelope of critical flux in figure 6(d) did not show any fluctuation. Some of these lower points were obtained by deliberately overheating and then decreasing power and/or by increasing the flow rate until a stable condition was obtained. The rest of the lower points were obtained by a similar, but uncontrolled, process that could occur independently following a perturbation and that would eventually result in a stable condition. The highest flux values obtained at a given operating condition are arbitrarily termed the maximum critical heat fluxes. Throughout the investigation the maximum critical heat fluxes were generally associated with fluctuations of wall temperature, flow rate, and pressure, while the lower values of critical flux normally occurred without fluctuations. The scatter of the criticalflux values increased as the transition point moved upstream for the entire investigation with both liquid hydrogen and liquid nitrogen. The temperature profiles presented in figures 3 and 5 are from tests in which the maximum critical flux was obtained.

The temperature profile for a maximum-critical-heat-flux case is compared with the temperature profile obtained at a lower value of critical flux in figure 7. All other conditions of flow rate, pressure, and inlet subcooling are essentially the same. The main difference in the two profiles is a higher temperature level for the maximum critical flux case, which reflects the increased heating rate. The lower critical flux was obtained by deliberately overheating and then by cooling.

Tube-wall-temperature profiles for tests with critical heat fluxes less than the maximum are presented in figure 8 for an essentially constant mass velocity and various transition locations. For transition occurring at a value of L/D of less than 8 (axial distance L of about 4), the profiles each have a definite peak and a minimum as contrasted with the profiles for larger values of L/D and the profiles of figures 3 and 5.

Some runs were made with the wall-temperature rise occurring at or near the tube inlet. The resulting profiles are shown in figure 9 for two pressures and various mass velocities. Many of these curves have peaks and minimum points and in this respect are similar both to the profiles of figure 8 for values of L/D of less than 8 and to the profiles reported in reference 8. The shape of the curves of reference 8 was explained on the basis of an inlet end effect, a two-phase annularflow model with the associated momentum pressure drop along the tube, and the attainment of "dry-wall" or "vapor-binding" conditions. In the tests presented herein, no measurable pressure drop across the test section was obtained, but dry-wall conditions could be attained. The data of figure 9 do not seem to indicate any significant effect on the critical heat flux of the increase in pressure from 50 to 70 pounds per square inch absolute. Whether the critical-heat-flux values for the data of figure 9 should be classified as maximum or submaximum values is not known. Additional data for small values of transition length are necessary to resolve this question.

The critical-heat-flux data for hydrogen that are considered to be maximums are shown in figure 10 in logarithmic coordinates as functions of mass velocity for several critical L/D's. Lines of constant quality of 1.00 and 0.50 computed from a heat balance are also indicated in figure 10. The general trend of the data is similar to that obtained for the boiling of water at low pressure (ref. 4). The water data indicated a change of slope or a knee in the curve of flux against mass velocity with the knee at a quality of approximately 0.50. The hydrogen data, however, do not exhibit any marked change in slope, particularly in the region of a quality of 0.50. The hydrogen data are fairly limited compared with the water data of reference 4. The hydrogen data can be extrapolated to higher and lower qualities in a manner which would show that a knee occurs in the quality range of 0.60 to 0.70. These same data are cross plotted against the critical-length-to-diameter ratio in figure 11, which shows the inverse relation between the critical heat flux and the critical L/D. This effect is greatest for high qualities. Extrapolating the curves to small critical values of L/D would indicate a small effect of L/D on the maximum critical flux. This condition makes it difficult to determine if the data shown in figure 9 represent maximum-critical-flux values.

The variation of the critical heat flux with mass velocity for boiling liquid nitrogen is presented in figure 12. The results are given for transition at the end of the tube only (L/D = 29). For smaller values of critical L/D, the critical heat fluxes that were obtained were less than those of figure 12 at corresponding operating conditions. For this reason, the critical fluxes obtained upstream in the tube with nitrogen are not regarded as maximum critical fluxes as defined herein. Attainment of such maximum critical fluxes at critical values of L/Dof less than 29 would be difficult and would require an improved apparatus with respect to stability control and material temperature limits. The general trend of the data of figure 12 agrees with that for hydrogen at a similar critical value of L/D (fig. 10) but with the critical flux at a larger value of mass velocity at approximately the same quality. This result reflects the lower latent heat of vaporization of nitrogen compared with hydrogen.

The data of figure 12 are also plotted in figure 13 together with the critical-flux data obtained with critical values of L/D of less than 29. The dashed line in figure 13 represents a quality of 1.00 for L/D of 29. With the exception of one point, all the critical-flux data for the short L/D tests fall below that for L/D of 29. In addition, the critical flux obtained upstream of the tube exit appears to be independent of the location of transition over a considerable range of L/D. The resemblance between figure 13 of this report and figure 4 of reference 4 should be noted. Figure 4 of the reference for water (ref. 4) showed that the presence of compressible volumes limited the stability of the system and caused low values of critical flux. A similar, though unknown, limitation of system stability apparently existed for the nitrogen tests with transition upstream of the tube exit.

Normalization of Critical-Heat-Flux Data

Previous investigators (refs. 2 to 4) have tried various means of correlating and normalizing critical-heat-flux results. These efforts have been primarily empirical approaches. In reference 4, a large amount of data was normalized for forced-flow boiling of low-pressure water by using parameters including tube diameter, tube length, and mass velocity. The maximum-critical-heat-flux data of the present investigation are presented in terms of the parameters of reference 4 in figure 14 and compared with the water data of reference 4. The cryogenic data appear to be successfully normalized into single curves for each fluid with an acceptable degree of scatter. The normalized curves for the three fluids have the same general trends and are separated in the order of their respective latent heats of vaporization. A similar normalization of the data is shown in figure 15 but with slightly different powers of the length and the diameter terms. The normalization of the data in figure 15 appears to be equally as good as that in figure 14. Selection of the correct correlating parameters seems difficult without a realistic model of the two-phase flow and heat transfer, particularly for the cases of gualities approaching 0 and 1.00.

Acceptance of the normalization of the critical-heat-flux data in the form of figures 14 and 15 would imply an effect of the critical boiling length on the critical heat flux in addition to that required by a heat balance. If the length term is assumed to have no other effect than that required by a heat balance, the critical-flux data should be normalized by a plot of the critical flux against the mass velocity divided by the critical-length-to-diameter ratio L/D; that is, the critical heat flux is a unique function of the local critical quality for a given fluid. The hydrogen maximum-critical-heat-flux data is shown in this way in figure 16. The dashed line represents a quality of 1.00 for all length values. The scatter of the data in figure 16 is only slightly worse than in figures 14 and 15. The actual data scatter appears to be unsystematic with the possible exception of the smallest L/D conditions, for which the heat-flux values fall lower than the rest of the data. Similar trends were obtained with the water data of reference 4; that is, the fluxes for small L/D data were low. The failure of the small L/Ddata to correlate with the rest of the results in a graph such as

figure 16 may be attributed to several factors, in addition to questions concerning the validity of the choice of correlating parameters. These (1) The data at small critical values of L/D were the most difare: ficult to obtain and had the greatest tendency towards instability; (2) at short lengths, heat transfer and two-phase flow equilibrium may not have been achieved; and (3) the data at small values of L/D may be reflecting entrance effects. The relation between the maximum critical heat flux and the critical length has therefore apparently not been completely determined. An additional complicating factor is involved in the selection of the correct critical length. The water tests of reference 4 had considerable inlet subcooling and would be expected to have an appreciable length of subcooled boiling, whereas, in the present investigation, the subcooling was negligible and bulk boiling occurred over nearly the entire length. Whether the effective length should be measured from the tube inlet or from the location at which the fluid bulk reaches the local saturation temperature is unknown. This problem is treated in reference 3 in a discussion of the use of quality as a correlating parameter for the critical heat flux.

Wall Superheat

A conventional method of presenting boiling heat-transfer data (especially for pool or pot boiling) is a graph of the heat flux against the wall superheat (wall temperature minus the fluid saturation temperature). Data for nitrogen are presented in this form in figure 17. These data include both the maximum-critical-heat-flux conditions and conditions below critical (no transition). Most of the data appear to fall on a single curve with no significant effect of mass velocity. Included in figure 17 are the predictions of reference 11 for nitrogen and of reference 5 for nitrogen and water. It is claimed in reference 5 that the method presented therein of predicting boiling heat fluxes applies to flowing systems as well as to nucleate pool boiling. The results shown in figure 17 should not be interpreted as supporting the analytical predictions or their application to flowing systems. The agreement may be fortuitous, especially because of the limited extent and accuracy of the nitrogen data. Similar graphs for the hydrogen data are not presented because of the poor sensitivity of the copper-constantan thermocouples at hydrogen temperatures. In fact, the sensitivity of the thermocouples for the nitrogen conditions is considered marginal. Analytical predictions indicate a wall superheat of 1° to 3° R for the range of the hydrogen test conditions. The experimental data show a wall superheat of the order of 10° R or greater. Attributing this lack of agreement entirely to limitations of the thermocouples appears difficult. The data of reference 7 for hydrogen do not fully correlate with the analytical predictions of references 5 and 11.

SUMMARY OF RESULTS

The results of the investigation of boiling heat transfer to liquid hydrogen and nitrogen in forced flow may be summarized as follows:

1. Boiling heat-transfer data (wall temperatures and heat fluxes) were obtained for bulk boiling of liquid hydrogen and nitrogen under forced flow upward inside an electrically heated tube. Data were obtained over ranges of flow and heating rates and pressures for small amounts of inlet subcooling. A limited amount of data was obtained with flowing cool hydrogen gas.

2. A transition in the type of boiling heat transfer was obtained. The critical heat flux corresponding to this transition was determined over a range of flow and heating rates and qualities. At specific combinations of flow and transition location, a range of critical-flux values was obtained and maximum values were determined. The maximum critical boiling heat flux increased with increasing fluid-flow rate and decreased with increasing length of tube before transition. The variation of the maximum critical flux with flow rate and critical boiling length was similar to that previously obtained with water.

3. Tube-inner-wall temperatures upstream of transition were essentially uniform and were only slightly greater (less than 20° R) than the fluid saturation temperature. Wall temperatures downstream of transition were considerably greater and the wall-temperature profiles generally resembled those obtained in film-boiling studies. The form of the wall-temperature rise downstream of transition appeared to be strongly dependent on the fluid quality at the point of transition. Maximum wall temperatures of 900° and 1800° R were obtained with hydrogen and nitrogen, respectively.

4. Fluctuations of pressure, flow rate, and temperature occurred during some of the boiling tests. Under some conditions, maximum criticalheat-flux values were attained during stable operation with fluctuations. In other cases the fluctuations became uncontrolled, and restabilization of the test condition resulted in critical-flux values less than the maximum values.

5. No measurable pressure drop across the test section was obtained at any condition.

Lewis Research Center National Aeronautics and Space Administration Cleveland, Ohio, May 29, 1962

APPENDIX A

COMPUTATIONS AND DATA REDUCTION

Temperature Drop Across Tube Wall

An analysis of the thermal and electric flow in an electrically heated tube is given in references 12 and 13. The basic assumptions of this analysis are negligible radial-voltage gradient and negligible axial-temperature gradient. For the case of a perfectly insulated outer wall, in which the thermal and electrical conductivities of the wall are linear functions of temperature, the equation for the temperature drop across a tube wall may be written as

$$t_{w,o} - t_{w,i} = \frac{JI^2 r_o^2 R_o}{k_o A_c^2} \left(\frac{R_{av}}{R_o}\right)^2 \frac{F}{1 + \sqrt{1 - AF}}$$
(A1)

where

$$F = \ln\left(\frac{1}{1-S}\right) - \left(S - \frac{S^2}{2}\right) \tag{A2}$$

$$S = 1 - \frac{r_i}{r_0}$$
(A3)

$$\frac{R_{av}}{R_{o}} = \frac{S - \frac{S^{2}}{2}}{S - \frac{S^{2}}{2} + \frac{BS^{3}}{6}}$$
(A4)

$$A = \frac{JI^2 r_o^2 R_o}{A_c} \left(\frac{R_{av}}{R_o}\right)^2 \frac{\alpha_o}{k_o}$$
(A5)

$$B = \frac{JI^2 r_o^2 R_o}{A_c^2} \left(\frac{R_{av}}{R_o} \right)^2 \frac{\beta_o}{k_o}$$
(A6)

K

(All symbols are defined in appendix B.) For the conditions of this investigation,

$$\left(\frac{R_{av}}{R_{o}}\right)^{2} \approx 1$$

Substituting the proper constants and dimensions in equation (Al) gives the tube-wall-temperature drop as

$$t_{W,0} - t_{W,1} = 2465I^2 \frac{R_0}{k_0} \left(\frac{0.0131}{1 + \sqrt{1 - 0.0131 A}} \right)$$
 (A7)

and

$$A = 2465I^2 \frac{R_0 \alpha_0}{k_0}$$
(A8)

 $(R_{o} = (ohms)(sq ft)/ft; k_{o} = lb force/(sec)(^{o}R))$. The heat flux at the tube inner wall is given by

$$q = \frac{JI^2 R_o}{2\pi r_1 A_c} \left(\frac{R_{av}}{R_o} \right)$$
(A9)

which for this investigation becomes

$$q = 5.22 \times 10^4 R_0 I^2$$
 (A10)

 $(R_{o} = (ohms)(sq ft)/ft).$

The variation of the thermal conductivity of 303, 304, and 347 stainless steel with temperature is given in figure 18. The variation of the tube electrical resistance with temperature is given in figure 19. The computed tube-wall-temperature drop is given in figure 20 as a function of the heat flux and the tube-outer-wall temperature. For the conditions of the investigation, the wall-temperature drop ranges up to 20° R for the hydrogen conditions and up to 7° R for the nitrogen test conditions. These computed wall-temperature drops should be applied only for readings of thermocouples located in a region of negligible axialtemperature gradient.

Heat Flux

The local heat fluxes tabulated in table II were computed by

 $q = 282I^2 R$ (All)

which is the same as equation (AlO) with a change in the constant resulting from using the resistance in ohms per inch of tube. The heat fluxes tabulated in table II include not only the critical heat flux but also the heat flux at the end of the tube, which was essentially constant over the entire tube length for the subcritical flux conditions (no transition).

APPENDIX B

SYMBOLS

A	factor defined in eq. (A5), dimensionless
Ac	tube-wall cross-sectional area, 4.5×10-4 sq ft
В	factor defined in eq. (A6), dimensionless
c _p	specific heat of liquid at constant pressure, $Btu/(lb mass)(^{O}R)$
D	tube inside diameter, 0.04625 ft (0.555 in.)
F	factor defined in eq. (A2), dimensionless
G	test fluid mass velocity, 1b mass/(hr)(sq ft)
^h fg	heat of vaporization, Btu/1b mass
I	heating current, amp
J	mechanical equivalent of heat, 778.3 ft-lb/Btu or 0.7376 lb force/(w)(sec)
k	thermal conductivity, $Btu/(hr)(sq ft)(^{O}R/ft)$ or 1b force/(sec)(^{O}R)
L	distance along tube axis measured from inlet station, in. (total length of tube, $16\frac{1}{8}$ in.)
p	pressure, lb/sq in. abs
Q	rate of heat flow, Btu/hr
đ	heat flux, Btu/(hr)(sq ft)
R	tube electrical resistance, (ohms)(sq ft)/ft or ohms/in. of tube
r	radius measured from tube centerline, ft
S	factor defined in eq. (A3), dimensionless
t	temperature, ^O R
W	fluid mass-flow rate, 1b mass/hr
x	fluid quality or mass fraction of vapor defined in eq. (1), dimensionless

α	coefficient of thermal conductivity as function of temperat l/OR	ure,
β	coefficient of electrical resistivity as function of temper $1/{}^{O}\!R$	ature,
Subscr	ripts:	
av	arithmetical average	
cr	critical (conditions at point of sudden rise of tube-wall t ature)	,emper-
ex	exit of tube	
i	inside surface of tube	
in	inlet of tube	
0	outside surface of tube	
sat	saturation condition	
W	wall of tube	

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Description	Sta- tion	Distance from inlet (measured positively downstream), in., for -						
		Runs 100 to 220	Runs 220 to 327					
Tube outer wall	1 2 3 4 5	0.5 1.5 2.5 3.5 4.5	0.52 1.5 2.5 3.41 4.5					
	6 7 8 9 10	5.5 6.5 7.5 8.5 9.5	5.41 6.45 7.53 8.53 9.53					
	11 12 13 14 15	10.5 11.5 15.6	10.5 11.48 14.06 14.56 15.61					
Copper bus Far	16	-1 <u>7</u>	-1.81					
Near	17	-7/8	88					
Inlet section Near Far	18 19	-1/2 -1 <u>1</u> 2	-0.5 -1.44					
Outlet section	20	17	17.22					

TABLE I. - THERMOCOUPLE LOCATIONS

TABLE II. -

(a) Hydrogen

Run	Pressure,	Satura-	Fluid-	Fluid-	Inlet	Exit	Mass veloc-	Heater	Critical	Critical-	Critical		Tu	Tube-outer-			ı
	sq in. abs	temper- ature, Op	tem- pera- ture,	tem- pera- ture,	cool- ing,	heat,	lb mass (hr)(sq ft)	amp	Btu (hr)(sq ft)	length- to- diameter	qua⊥lty	1	2	3	· 4	5	6
			°R .	°R	· n	°sat, °R				ratio							
100 101 103 105 106	47.8 51.0 53.0 49.0 50.2	45.0 45.5 45.9 45.2 45.4	138.0 69.0 82.0 52.0 54.0	138.0 69.0 117.0 52.0 92.0		93.0 23.5 71.1 6.8 46.6	6,480 7,725 7,780 13,100 12,800	254 326				157 90 174 69 148	151 82 209 69 200	146 81 225 69 223	145 81 238 69 261	145 79 245 65 280	145 81 255 69 306
107 108 109 114 115	49.0 52.5 70.5 49.5 52.9	45.2 45.8 48.3 45.3 45.8	46.0 55.0 57.0 40.2 41.7	98.0 121.0 107.0 45.9 46.1	 5.1 4.1	52.8 75.2 58.7 .6 .3	12,550 7,780 12,500 6,000 12,500	332 346 348 323 462	9,400 19,350	16.5 19.2	0.52	160 61 189 56 60	226 248 272 56 60	265 354 326 56 60	305 442 385 54 58	336 492 438 54 58	375 533 498 56 58
117 118 123 124 125	50.6 74.0 51.0 51.0 78.0	45.5 49.1 45.5 45.5 49.5	45.9 43.7 43.7 43.9 44.7	64.0 48.6 169.0 240.0 223.0	5.4 1.8 1.6 4.8	18.5 5 123.5 194.5 173.5	7,850 9,050 4,340 2,850 3,090	409 376 391 375 377	15,100 12,750 13,800 12,700 12,800	14.7 19.0 4.5 4.5 3.0	.65 .55 .31 .44 .23	56 56 58 63 60	58 60 63 64	60 56 60 70 290	56 56 323 350 361	56 54 393 410 408	56 56 441 365 455
126 127 128 129 130	76.0 30.0 50.0 50.3 48.5	49.2 41.3 45.4 45.4 45.1	43.2 39.6 44.2 43.9 42.3	222.0 218.0 45.5 45.3 45.7	6.0 1.7 1.2 1.5 2.8	172.8 176.7 .1 1 .6	3,090 2,970 7,725 12,800 7,500	379 369 399 417 412	13,000 12,250 14,300 15,700 15,300	4.2 3.0 6.5 11.7 15.7	.32 .25 .26 .30 .69	60 56 58 58 58	64 60 60 60	170 316 60 58 60	331 437 60 58 60	375 486 324 58 60	422 533 361 58 60
131 132 133 134 135	50.0 50.0 48.0 50.0 50.0	$\begin{array}{c} 45.4\\ 45.4\\ 45.0\\ 45.4\\ 45.4\\ 45.4\end{array}$	44.1 44.1 44.4 44.3 44.1	45.4 45.7 70.0 45.5 103.0	1.3 1.3 .6 1.1 1.3	.3 25.0 .1 57.6	13,400 7,970 5,530 5,650 5,700	408 378 360 339 388	15,000 12,900 11,700 10,400 13,600	10.5 21.0 17.0 20.0 10.3	. 25 . 76 . 82 . 83 . 55	60 60 56 58	60 60 58 58 58	60 60 58 58 58	60 60 58 58 58	60 60 56 58 58	60 60 58 58 58
136 137 138 139 140	50.0 50.0 50.0 50.0 50.0	45.4 45.4 45.4 45.4 45.4	$\begin{array}{r} 44.3 \\ 44.3 \\ 44.4 \\ 44.1 \\ 44.2 \end{array}$	45.3 99.0 149.0 70.0 83.0	1.1 1.1 1.0 1.3 1.2	1 3.6 103.6 24.6 37.6	5,830 4,040 4,040 4,220 4,100	359 338 371 303 320	11,600 10,300 12,400 8,300 9,300	15.5 12.0 8.5 17.5 15.8	.59 .69 .58 .77 .81	58 56 56 56 54	58 56 56 56 54	58 56 56 56 54	58 56 56 56 54	58 56 56 56 54	58 56 315 56 54
144 145 146 147 148	15.0 69.0 51.0 73.0 73.0	36.6 48.3 45.5 48.8 48.8	36.8 46.5 43.3 46.7 46.5	36.8 46.7 50.6 88.0 130.0	1.8 2.2 2.1 2.3	.2 -1.6 5.1 39.2 81.2	10,900 12,800 9,690	455 498 488	18,700 22,400 21,400	 	 	45 54 284 227 266	47 56 484 366 465	47 56 512 456 518	47 56 504 494 525	45 56 484 488 513	47 56 476 485 513
149 150 158 159 160	71.0 71.0 51.0 52.0 54.0	48.6 48.6 45.5 45.7 46.0	47.7 48.4 42.4 42.9 43.4	90.0 88.0 140.0 67.0 166.0	.9 .2 3.1 2.6	41.4 39.4 94.5 21.3 120.0	10,600 10,750 8,090 13,100 5,230	527 519 484 495 419	25,100 24,300 21,100 22,100 15,800			250 241 332 242 200	422 381 577 464 414	514 464 605 537 453	539 500 598 536 473	519 493 579 505 482	512 386 573 488 504
161 162 163 164 165	51.0 30.0 31.0 52.0 52.0	45.5 41.3 41.5 45.7 45.7	39.3 39.0 39.1 42.7 42.8	153.0 41.5 90.4 168.0 132.0	6.2 2.3 2.4 3.0 2.9	107.5 .2 48.9 122.3 86.3	6,420 13,500 13,500 6,060 8,680	442 463 523 448 483	17,600 19,300 24,600 18,100 21,000			351 356 492 267 303	587 605 826 490 575	625 598 771 531 604	625 546 664 545 591	608 500 589 549 567	603 472 560 562 560
166 167 168 169 170	50.0 52.0 71.0 71.0 69.0	45.4 45.7 48.4 48.4 48.3	39.1 39.0 39.0 39.0 44.1	137.0 166.0 178.0 150.0 74.2	6.3 6.7 9.4 9.4 4.2	91.6 120.3 129.6 101.6 25.9	8,390 5,770 7,380 8,150 15,200	483 447 512 508 513	21,000 18,100 23,700 23,300 23,700			260 192 171 182 122	537 446 453 447 381	597 503 570 542 490	597 525 599 568 537	574 530 593 560 518	572 545 604 561 505
171 172 202 203 204	65.0 70.0 49.8 49.0 51.0	47.7 48.4 45.4 45.3 45.6	45.4 46.0 43.2 45.0 45.6	158.0 141.0 43.0 45.0 45.0	2.3 2.4 2.2 .3 0	110.3 92.6 -2.4 3 6	7,020 8,440 13,600 13,875 14,000	468 490 363 473	18,700 21,700 ⁸ 11,900 20,200	a294 20.2	a.57 .67	199 260 43 47 60	497 534 45 50 59	536 570 45 42 54	543 564 45 39 50	539 548 45 36 45	547 552 45 39 42
205 206 208 209 210	50.1 50.0 49.0 49.5 48.5	45:4 45.4 45.3 45.3 45.1	41.4 45.4 42.4 42.7 42.4	44.9 44.8 44.7 44.8 44.8	4 2.9 2.6 2.7	5 7 6 5 3	12,900 11,325 9,535 10,830 11,690	457 390 458 477 518	18,850 13,750 19,000 20,600 24,350	19.5 20.2 8.0 8.0 8.5	.59 .57 .32 .31 .36	59 52 50 50 52	56 54 54 54 58	54 50 52 54 58	48 50 54 56 58	45 45 74 70 58	40 45 368 388 405
211	48.0	45.1	42.6	44.9	2.5	2	8,350	467	19,750	8.7	.43	52	54	54	54	54	329
212 213 214	49.2 51.1 50.2	45.3 45.6 45.4	43.5 43.7 42.9	43.5 43.7 44.8	1.8 1.9 2.5	-1.8 -1.9 6	9,940 12,525 9,240	200	a3,560	a29+	a.21	44 43 48	45 45 50	45 45 50	45 45 50	44 44 50	44 45 50
215 216 217 218 237	49.0 50.5 52.7 52.8 50.2	45.3 45.5 45.9 45.9 45.4	42.1 44.3 44.3 42.8	44.6 45.3 45.5 44.9	3.2 1.6 1.6 2.6	6 4 5	11,180 7,815 10,450 13,235 11,430	300 448 448 485 397	⁸ 8,150 18,200 18,250 21,350 14,400	^a 29+ 8.0 14.0 14.0 26.2	a,43 .54 .50 .72	50 50 54 58 48	52 56 66 64	52 54 60 65	526 562 666 56	52 216 62 66	54 382 65 70 64
238 239	46.5 50.4	44.8 45.5	42.3 43.4	42.3 44.0	2.5 2.1	0 -1.5	14,375 11,500	396	14,300	27.5	.75	50 48	60 64	54 58	52 70	54 68	52 62
240 241	50.6 50.6	45.5 45.5	42.1 40.6	42.8 41.0	3.4 4.9	-2.7 -4.5	11,650 11,550	405 415	14,900 15,600	25.2 24.5	.69 .68	50 48	68 66	62 60	62 68	68 68	62 66
242	50.0	45.4	43.1	43.1	2.3	-2.3	13,900					50	58	52	5Ó	52	50
243 244	50.4 45.7	45.5 44,6	43.0 42.3	43.4 42.3	2.5 2.3	-2.1 -2.3	11,690 15,860	452	18,500	20.7	.72	45 50	62 56	60 52	50	70 54	68 52
248 250 251 252	39.3 50.5 50.8 51.0	43.4 45.5 45.5 45.6	41.6 41.7 41.7	43.1 45.5 42.4	3.9 3.8 3.9	-2.3 0 -3.2	4,100 4,380 4,410 4,820	261 255 	6,200	29.0	.83	42 36 36 36	80 56 58 58	52 64 50 42	52 66 45 45	56 66 54 45	54 64 50 42

^aHeat flux, length-to-diameter ratio, and quality are subcritical values and are taken as of tube-exit conditions.

EXPERIMENTAL DATA

(a) Hydrogen

temperature, OR at station -								0	. hua	т	ot	Ev.1 F	Amr	itude	Remarks
tempe	eratur	10	ι, a	t sta 12	13	1 - 14	15	tempe:	rature, B	plenum wall temperature,		plenum wall	of to temp	be wall erature	
								Far	Near	o Near	R Far	temper- ature, ^O R	fluct ±°R	At sta- tion ~	
145 14 82 8 263 27 69 6 330 35	15 149 2 8 71 27 57 6 59 37	5 145 1 81 5 283 5 65 9 410	146 81 288 64 439	147 81 291 64 466			152 82 300 65 532	367 344 345 318 359	339 312 310 283 316	207 157 158 128 135	182 128 131 91 95	221 123 162 98 194			<pre>No-heat run No-heat run </pre>
420 46 560 57 555 60 58 5	39 51 77 58 00 61 58 5	0 554 5 595 8 631 8 160 8 60	590 598 623 279 60	609 599 619 319 244			574 583 546 395	379 377 388 356 411	335 330 344 305 352	139 109 145 87 92	102 59 108 49 49	211 233 216 99 185			Cool-gas run
58 6 58 5 479 51 521 57	62 26 58 5 18 54 76 61	3 400 8 60 1 569 5 659	473 66 593 697	517 248 611 729			 799	384 388	332 338	89 95 96	49 56 58	297 365			Transition; maximum-critical-flux value Transition Transition; maximum-critical-flux value
498 54 463 51 574 61 382 39	45 57 16 55 16 64 99 41	9 620 3 593 3 676 4 427	644 635 700 434	687 645 720 440			770 750 737 463	388							> Transition
64 27 60 6 271 31 60 6	77 31 50 6 18 32 50 6	2 332 0 305 7 343 0 60	347 382 351 60	430 359 66 369			595 525 377 416 497							 	Transition; maximum-critical-flux value Transition } Transition; maximum-critical-flux value
58 58 301 37 58 5	58 5 79 42 58 6	8 58 0 458 9 256	58 486 330	241 509 373			411 556 470 530								J. Transition } Transition; maximum-critical-flux value
390 45 56 5 54 5	50 48 56 5 54 5 47 4	5 521 6 56 4 249 5 45	548 238 324 45	577 294 366 45			631 402 459 47	 338	290	 92	 47		 		Transition] Transition; maximum-critical-flux value] No-heat run
56 5 471 4 479 48 519 53	58 5 73 46 88 48 36 53	6 56 9 473 8 500 4 547	56 468 500 556	56 472 509 562			56 477 536 599	347 448 411 408	297 385 352 355	86 104 111 109	56 58 56 56	64 193 220 246			
498 - 480 4 567 5 476 4 520 5	95 48 72 57 76 47 40 55	3 495 0 583 1 476 3 570	495 583 472 586	504 588 476 600			534 603 488 619	416 384 402 387	362 332 348 336	111 95 95 98	56 57 57 57 57	191 185 211 260			
596 5 456 4 537 5 568 5 559 5	96 57 47 44 35 53 81 59 61 55	7 595 2 442 1 536 1 610 9 56	592 435 531 618 560	2 593 437 5 534 5 626 5 573			588 435 540 646 585	390 398 414 392 400	338 344 360 339 346	95 94 97 98 100	51 53 52 57 55	285 216 189 294 293			Transition at inlet; oritical length indeterminate
567 5 553 5 615 6 568 5 495 4	70 56 70 57 32 64 79 56 93 49	9 57 5 590 6 604 5 503	570 600 675 61- 50-	5 582 5 606 5 697 4 629 4 509			596 631 750 659 527	500 393 400 394 391	347 341 348 344 339	95 100 96 90 94	52 55 57 55 58	274 287 320 234 176			
554 5 539 5 45 39 42	73 57 68 57 45 4 29 3 40 4	7 59 2 58 7 4 33 2 0 30	60 59 4 2	4 616 2 604 5 48 5 29 9 227			641 635 47 36 446	402 400 280 319 343	349 349 328 369 398	105 105 74 91 102	57 57 47 45 54	260 260 52 18 173	 6 2	 15 12	No-heat run No transition Transition
45 48 405 4 422 4 467 4	42 3 45 3 33 44 47 45 96 50	56 40 56 30 47 46 58 47 58 52	0 40 5 46 2 48 7 53	0 276 2 20] 9 476 1 484 4 54]	5 5 1		401 377 492 499 556	342 323 341 347 353	389 371 395 402 409	94 101 85 86 91	53 52 42 42 45	144 126 243 239 239	17 2 2 2	11 15 9 12	<pre>Transition; critical flux less than maximum Transition; long period required for stabilization; transition point moved slowly upstream Transition; with increase of heat, temperature at station 5 did not rise but downstream tempera- tures increased</pre>
413 4 45 47	56 47 45 4 47 4	79 50 44 4 5 4	52 4	3 534 4 44 5 45	1	. .	593 45 46	338 285 284 294	392 334 333	88 72 72 79	42 45 47 47	264 52 52 60	2	15	Transition; similar to run 210. Maximum-critical- flux value No-heat run
52 56 423 4 68 72	52 58 43 43 70 29 80 30	6 5 3 46 34 38 31 43	3 5 5 48 5 43 1 46	6 62 1 488 4 464 5 489			62 511 539 546	307 338 340 355	359 390 394 409	83 100 101 112 68	48 45 48 50	65 257 244 235 66	2 14 1	 15 1 15 15	No transition Transition; maximum-critical-flux value
62 52 64 64	52 5 62 6 64 6	52 10 52 50 52 18		4 50 2 60 2 60			50 97 246 307	272 294 296 296	315 339 341 341	50 68 70 70	50 66 68 68	52 66 86 103	102 17 26 11	10 15 13 4	No-heat run Transition; maximum-oritical-flux value
50 66 52	50 66 52	50 48 68	5 5 7 5 5	4 5 9 9 4 5	5 499 5 499 5 50	50 50 50 50	50 50 482 50	268 307 268	312 356 311	50 350 50	50 409 50	52 219 52	 1	 14 	No-heat run; control valves in same position as in run 241 Transition; maximum-critical-flux value No-heat run; control valves in same position as in run 245
54 64 50 42	54 62 50 39	54 5 62 49 50 8 39 49	2 6 9 4 3 5 0 4	0 5 7 3 6 5 7 4	6 56 9 39 2 85 2 39	5 75 5 54 5 39	186 39 54 39	272 258 269 260	314 301 314 304	54 39 52 39	54 39 52 39	56 42 52 42	15 30	14 1- 13 	Transition; maximum-oritical-flux value No-heat run Transition; maximum-oritical-flux value No-heat run; control valves in same position as in run 251

	Pressure	Satura	Fluid	Fluid	Tplef	Ex1+	Mage voloo	Noctor	074.84	0	0.411	r					
1000	1b abs	tion	inlet	exit	sub-	super	ity,	current,	heat flux,	boiling-	quality	<u> </u>	1	Tub	e-out	er-wal	
		ature,	pera- ture,	pera- ture,	ing, Op	tex -	(hr)(sq ft)	anp	(hr)(sq ft)	to diameter		1	ſ -		1*	2	ľ
			°R	°R		°R				ratio			ļ				
253	51.9	45.7	42.3	45.4	3.4	-0.2	4,520	263	6,250	25.7	0.77	39	73	50	50	54	52
254	51.0	45.6	42.4	45.4	3.2	2	4,500	268	6,500	26.0	.82	42	77	50	29	54	52
255	49.8	45.4	42.4	45.1	3.0	3	4,380	272	6,700	24.5	.81	42	77	50	42	54	52
257	46.0	45.6 44.7	43.1 43.5	45.3	2.5	3 4	4,450 4,300	313	8,850 	20.5	. 90	42 39	80 70	52 45	45 45	58 45	54 42
253 259	14.7 50.0	36.4 45.4	36.4 43.2	36.4 45.5	0 2.2	0 .1	0 4,100	291	7,700	21.5	.89	33 45	70 82	36 54	36 45	36 58	33 54
260 261	49.5 50.0	45.3 45.4	42.5	45.7	2.8	.4	4,170	312	8,800	20.5	.95	36	74	50	45	54	52
262	50.2	45.4	43.4	45.5	2.0	.1	4,030	316	9.050	19.0	. 95	42	80	47 52	45 50	47	45
265	43.5	45.2	43.5	80.5	1.7	35.3	4,470	351	11,150	15.2	.85	45	86	54	66	60	58
297	47.0	44.9	43.5	44.6	1.4	3	6,770	317	9,100	25.0	.75	50	86	58	58	62	58
298	47.7	45.0	43.6	44.9	1.4	1	6,520	316	9,050	25.2	.78	50	88	58	68	64	60
300	40.4 46.1	90.1 44.7	43.7 43.7	45.0 44.8	1.4 1.0	1	6,780 8,720	312 363	8,800 11,950	26.2 26.2	.76 .81	50 50	88 90	58 58	58 	62 68	60 62
301	52.3	45.8	43.0	45.5	2.8	3	4,440	263	6,300	26.2	.82	48	80	54	47	60	56
302 303 304	52.0 50.5 50.4	45.7 45.5 45.5	43.1 43.1 43.4	45.6 45.4 45.4	2.6 2.4 2.1	1 1 1	10,630 12,620 16,960	387 416 458	13,600 15,700 19,150	26.2 26.2 26.2	.73 .71 .65	48 48 48	85 86 86	60 60 64	58 52 50	68 68 74	64 64 70
305 306	50.0 48.0	45.4 45.1	43.6		1.8		7,990 7,990	433 424	17,150 16,400	12.0 7.2	.56 .30	48 70	83 79	66 72	64 72	75 385	74 443
308 309	49.8 49.8 50.6	40.4 45.4 45.5	43.6 43.3 43.2	45 4 45.9 46.2	1.8	0 .5	5,740 5,510 6.060	373 382 389	12,600 13,200	18.2 15.7	.89	48 48	83 82	56 58	42 58	66 66	62 64
310	49.0	45.2	43.3	45.9	1.9	.7	8,330	440	17,650	13.5	.63	50 50	85	- 58 - 62	62	68 74	66 72
311 312 313	51.6 53.0 50.9	45.7 45.9 45.6	42.3 44.9 41.2	45.7 46.0 45.5	3.4 1.0 4 4	°.1	13,380 12,660	443 470 415	17,850	7.5	.17	45 47	79 93	62	54	269 74	390 72
314	50.5	45.5	41.7	45.5	3.8	0.1	16,000	415	16,300	10.0	.04 .17	45 47	75 75	60 60	327 54	389 66	393 100
315 316 317	50.8 51.5 52.1	45.6 45.7 45.8	41.7 42.9 43.5	45.5	3.9 2.8	1	16,660	425 444	16,350 17,900	12.5 13.5	.22 .45	47 47	73 88	60 58	54 58	66 66	60 60
318 319	50.8 51.8	45.6 45.7	43.8 43.6	45.6 46.0	2.3 1.8 2.1	08.2	6,190 9,310	5/8 397 419	12,950 14,300 15,900	8.7 8.5 8.5	.64 .42 .30	42 42 39	75 77 75	52 52	52 52	58 58	277 281 299
320	51.8	45.7	44.6	45.8	1.1	.1	10,920	430	16,750	8.5	.25	39	80	52	36	64	276
325	48.8 51.5	45.2 45.7	44.6 45.6	45.4 	.6 .1	.2	11,850 8,700	447 448	18,150 18,150	8.5 2.5	.29 .12	36 45	79 97	54 268	47 305	70 374	292 459
323 324	50.0 52.2	45.4 45.8	45.5 44.2	45.5 45.7	1 1.6	.1 1	10,690 14,670	476	20,500	15.2	.46	45 42	80 75	50 58	47 52	47 66	45 64
							(۵)) Nitroger	1								
264 265	55.3 54.1	163.2 162.8	162.0 158.0	165.0 164.0	1.2 4.8	1.8 1.2	17,030 24,030	297 304	8,870 9,250	21.5 25.0	0.57	165 163	198 195	170 169	175	173 172	171
266 267	54.6 50.2	163.0	160.0	165.0 161.0	3.0	2.0	24,300 38,000	322 345	10,400 11,900	23.7 25.0	.50	164 162	197 195	170 169	179 167	174	172 171
269	40.5 54.3	163.0	160.0	163.0	3.2	2	30,400	323 313	10,400	23.5	.39	160	194	167	171	170	170
270	56.0 50.2	163.5 161.4	161.5 157.0	163.5 162.0	2.0	ŏ .6	42,300 31,000	355 325	12,600 10,600	24.2	.36	165	197 193	170 167	165 171	175 171	173 170
272	49.8 50.1	161.2	157.0 157.0	162.0 160.5	4.2 4.3	.8 8	30,800 15,700	326 309	10,650 89,550	29+ ^a 29+	a.49 87	161 163	193 195	168 167	173 169	171 170	170 169
274 275	49.8 55.3	161.2 163.2	158.0	160.5 163.0	3.2 5.7	7	15,100 19,750	314 301	a9,850 9,050	⁸ 29+ 22.0	a.95 .48	161	194 194	165 167	168 163	169	168
276	50.4 52.5	161.4	158.0	160.5	3.4	9	16,250	316 302	a10,000 9,150	a29+ 22.5	a 89 .44	163 164	195 196	167 169	165 167	171	170 171
279	52.0	161.3	157.0	161.0	3.0 4.3	0 3	27,200	316 352	10,000 ^B 12,350	24.0 a ₂₉₊	.43 a.70	164 : 159 :	197 194	168 165	175 171	171 170	170 169
280 281	50.5	161.4	157.0	162.0	4.3 4.4	.7 4	24,400 24,100	367 361	13,300 813,000	29.0 ^a 29+	.78 a.77	159	194	165 166	176	170	169 170
282	50.6	161.5	157.5	161.0	4.0	5	23,600	371	13,800	29.0	.84	160	195	165	165	171	170
283 284 285	50.3 51.0 54.4	161.4	158.0 157.5	162.0	3.4	2	24,500 31,100	308 408	9,500 ^a 16,700	a23.5	a 44	160	191	165	167	168	167 171
286 287	55.0 55.0	163.2 163.2	159.0 159.0	163.0 163.0	4.2	2	31,700 31,100	406 413	16,500 a17,100	829+0 829+	a.75 a.79	161	200	170 170	178 166	174 173	173 172
289	50.2	161.4	157.0	160.0	4.4	4	41,900	453	a20,550	a29+	a.70	160	201	170	176	175	172
291 292	49.2	160.8	157.0	161.0	3.8	5	41,700	400 346 489	11,950 124,000	29.0 24.0 a29+	. /2 .40 8.61	161	202	167 171	168 171 170	171 171	173 170 176
293	49.8	161.2	157.0	161.0	4.2	2	32,400	154	ā2,350	829+	A.08	159	187	162	167	163	162
294 295 296	49.5 49.7 49.6	161.1	157.0	161.0	4.1	-:1	31,400 31,200	208 255	^a 4,300 ^a 6,500 ^a 10,400	*29+ a29+ a20.	a.28 a.28	160	188	163	167	164	163
325 326	47.5	160.2	158.5	159.7 161.0	1.7	- 5	31,900	325	10,550	17.0	.31	160	184	159	163 166	168 160 171	160 167
327	51.3	161.8	159.7	162.0	2.1	.2	24,400	384	14,700	3.0	.08	164	294	833	1132	1320	455
272b 276a	50.0	161.2	158.2	161+	3.0		37,500	459 315	21,300	29.0	.80						
292a 325a	51.0	161.7	158.0	162.0	3.7	0.3	53,800	503 367	25,800 13,500	29.0	.69 .82						
0208	47.5	101.0	19910	101.0	2,0	Ů	50,400	390	15,300	29.0	.73						

(a) Concluded. Hydrogen

^aHeat flux, length-to-diameter ratio, and quality are subcritical values and are taken as of tube-exit conditions.

EXPERIMENTAL DATA

(a) Concluded. Hydrogen

te	mpera	ture,	° _R ,	at st	ation	-			Copp	er bus	Inl	et wall	Exit	Ampl of tu	itude be wall	Remarks
7	8	9	10 :	11	12	13	14	15	Far	^o R Near	temper o	ature, R	wall temper- ature,	temp fluct	erature uations	
											Near	Far	°R	±°R	At sta- tion -	
50	50	50	94	58	52	85	100	171	272	316	52	52	54	30 36	13 14	
52	52	52	90	58	52	.62	80	122	274	318	54	54	54	20	13 14	<pre>Transition; maximum-critical-flux value</pre>
52	52	52	80 91	58 66	379	379	392	375	280	325	294	242	183	14	12	
42	42	42	504	47	42	39	39	42	263	307	42	42	45	·		No-heat run; control valves in same position as in run 256
33 56	33 54	33 54	505 90	36 64	33 58	29 328	29 341	33 329	262 281	306 326	- 33 - 249	192	36 145	2	14	Transition; maximum-critical-flux value
52	52 45	52 45	108	62 50	172	402 42	413 42	397 42	281 268	326 311	315 45	268 45	285 45	4	12	No-heat run; control valves in same position as
54 56	56 56	56 109	566 549	103 389	260 443	428 592	437 605	421 564	287 293	333 338	346 489	306 455	211 272	23 21	11 9	in run 260 Transition; maximum-critical-flux value; Transition; maximum-critical-flux value; difficult Transition; maximum-critical-flux value; difficult
	=					104	075	188	287	330	62	60	60	49	13	to stabilize because of flow oscillations
58	58	58	62	68	62	91	167	188	287	331	62	60	60	17 26	14 13	
58	58	58	62	70	62	62	66	155	287	332	62	60	60	12 6	14	
• 60	60	60	62	72	64	64	70	230	294	338	66	54 58	88		14	> Transition; maximum-critical-flux value
55	62	54 62	64	72	- 58 - 66	68	72	105	203	343	68	66	64	38 34	15 15	·
64 68	62 68	64 66	· 66 70	72	68 72	70	72	236	301 307	347 354	70 74	68 72	105 100	6	15 15	J
75	361	423	461	492	509	568	571	538	303	350	529	519	270			} Transition; conditions impossible to stabilize;
451 60	457 60	466 60	470 62	480 260	489 344	511	512 522	489	302 296	349	490	488	241) semperatures kept enanging
64	62 64	305	375	420	426 451	524	554	521	298	346	491	467	266		==	Transition; maximum-critical-flux value
72 424	132 416	387 406	439 405	471 409	492 410	556 424	564 429	535 409	307 309	354 359	525 411	514 409	269 207		· ==	Transition
68 373	68 361	239 357	375 351	413	434	504 347	511 352	487	314	361	465	450	231			Transition; maximum-critical-riux value
365	380	3/5	355	361	360	361	362	349	300	346	357	357	143			Transition
56	104	352	387 487	411	426	469	472 665	450 635	302 294	349 340	445 609	439 583	222 329	8	13	Transition; maximum-critical-flux value
359 377	388 392	415 404	430 409	451 418	463 421	517 444	520 447	493 424	295 299	342 346	487 427	425	260 212			Transition
365	393	406	412	418	420	441	444	418	300	348	423	416 430	204	13	4	
553	618	434 626	433 596	575	561	573	577	540	319	364	555	561	270	4	12	Transition; took 3 hr to obtain equilibrium; at first had increasing temperature profile;
45	. 45	45	45	50	45	42	42	45	276	321	47	47	45		·	No-heat run to check wall thermocouples
54	50	101	473	420	441	4/9	482	455	309		402	(b)	Nitroger	<u></u>	L	
171	173	172	173	178	175	865	890	650	346	384	549	178	175			
171	172	171	172	175	173	451	618	524 635	348	380	175	174	174	30	15	
171	172	171	171	173	172	805	850	660	345	377	173	171	171	4	13	Transition; on all these runs either power was decreased or flow rate increased as desired
172	174	173 173	175 174	178 177	176	805 675	830 800	626 860	340 346	376 383	175	179 175	175		15	vent overheating and instability
170 170	171 171	170	171 171	173 173	171	725 173	805 172	639 172	337 336	373	173	173	172			No transition, close to maximum-critical-flux value
169	170	170	170	171	171	172	172	172	343	380	170	170	168			No transition
170	171	170	171	174	172	760	795	590 172	340 345	377 382	174	173	172			Transition; close to maximum-critical-flux value No transition; close to maximum-critical-flux value
171 170	171 171	171	171 171	175 173	172	721 660	764 761	601 619	343 346	380 383	175	173	172	2	15	} Transition
-168	169	168	169	171	170	171	171	171	342	379	171	171	170		13	No transition Transition: maximum-critical-flux value
168	169	168	170	171	170	171	171	171	344	381	171	171	170	11	14	No transition
169	170	169	170	171	170	172	172	172	346	383	171	171	170			Transition; maximum-critical-flux value
167 170	168	167	168	170	169	173	736	172	351	390	172	172	171			No transition; close to maximum-critical-flux value
172	173	172	173	175	174	175	176	175	352	391 392	174	174	171	3		No transition; close to maximum-critical-flux value
171	172	171	172	178	173	175	176	174	361	392	174	174	171			No transition
172	173 170	172	173	176	174	175	177	175	362	404 381	174	174	172	4	15	Transition Transition No transition: close to maximum-critical-flux value
1/4	163	163	163	164	163	164	164	165	325	361	163	163	166			
164 165	164 166	163 165	164 167	165 168	165 167	167 168	166 168	167 168	328 333	365 370	166	165 167	167			> No transition
167 160	168	167 160	168	170	169 160	171	171	170	342	356	170	160	160		10	No-heat run; thermocouple check
168	169	166	386	1760	900	175 1745	1770	1600	470	494	1801	1815	765	5	6	Transition; difficult to control
1030																All these data taken while attempting to obtain maximum critical flux for transition at end
																before final power or flow adjustment, which caused instability and cave transition unstream
																at heat-flux values considered to be less than maximum critical flux







Figure 2. - Details of test section, inlet, and exit.



(a) Mass velocity, approximately 12,000 pounds per hour per square foot.

Figure 3. - Tube-outer-wall temperature profiles for constant mass velocity and varying heat flux and location of transition. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2[°] R.

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Figure 3. - Concluded. Tube-outer-wall temperature profiles for constant mass velocity and varying heat flux and location of transition. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2° R.





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Figure 5. - Tube-outer-wall temperature profiles for constant transition location and varying heat flux and mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2[°] R.



Figure 5. - Concluded. Tube-outer-wall temperature profiles for constant transition location and vary-ing heat flux and mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2° R.

Critical heat flux, q_{cr} , Btu/(hr)(sq ft)



(a) Critical-boiling-length-to-diameter ratio, approximately 26.

Figure 6. - Variation of critical heat flux with mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2° R.

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Critical heat flux, q_{cr} , Btu/(hr)(sq ft)



(b) Critical-boiling-length-to-diameter ratio, approximately 20.

Figure 6. - Continued. Variation of critical heat flux with mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet sub-cooling, 2° R.



(c) Critical-boiling-length-to-diameter ratio, approximately 15.

Figure 6. - Continued. Variation of critical heat flux with mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet sub-cooling, 2° R.





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(d) Critical-boiling-length-to-diameter ratio, approximately 8.5.

Figure 6. - Concluded. Variation of critical heat flux with mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet sub-cooling, 2° R.



Figure 7. - Comparison of tube-outer-wall temperature profiles for maximum and submaximum criticalheat-flux conditions. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2^o R.



Figure 8. - Comparison of tube-outer-wall temperature profiles for submaximum critical-heat-flux conditions at various heat fluxes and constant mass velocity. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2^o R.

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(a) Test-section pressure, approximately 50 pounds per square inch absolute.





(b) Test-section pressure, approximately 70 pounds per square inch absolute.

Figure 9. - Concluded. Tube-outer-wall temperature profiles for liquid hydrogen with wall temperature rise at tube inlet.



Figure 10. - Variation of maximum critical heat flux with mass velocity at various critical-boiling-length-to-diameter ratios. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2° R.



Figure 11. - Variation of maximum critical heat flux with critical-boiling-length-todiameter ratio at various mass velocities. Liquid hydrogen; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2° R.



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Figure 14. - Comparison of maximum critical heat flux for cryogenic liquids with water correlation of reference 4.





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Figure 16. - Maximum critical heat flux as function of flow and length parameter. Boiling liquid hydrogen; tube inside diameter, 0.555 inch; test-section pressure, approximately 50 pounds per square inch absolute; average inlet subcooling, 2^o R.

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Figure 17. - Nucleate heat flux as function of wall superheat. Liquid nitrogen; testsection pressure, 48 to 56 pounds per square inch absolute; inlet subcooling, 3° to 5° R; length-to-diameter ratio, 29. Thermal conductivity, k, Btu/(hr)(sq~ft)(OR/ft)

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Figure 18. - Variation of thermal conductivity of 303, 304, and 347 stainless steel with temperature.

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Temperature drop across tube wall,

(a) Hydrogen test conditions.

Figure 20. - Temperature drop across tube wall as function of heat flux and tubeouter-wall temperature. Negligible axial temperature gradient; tube of 304 stainless steel; inside diameter, 0.555 inch; wall thickness, 0.035 inch.



(b) Nitrogen test conditions.



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