

# Novel Wide Bandgap Crystals: Low Temperature Growth of 2H-SiC and $\beta$ -Gallium Oxide

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Darren Thomson

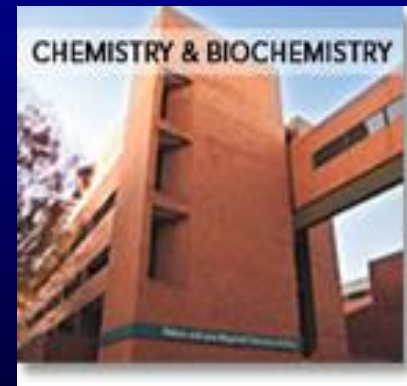
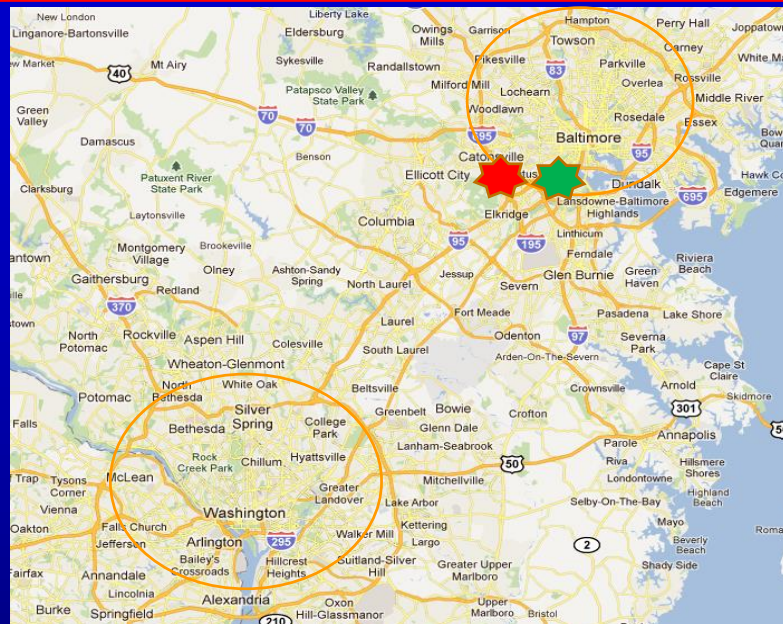
Tahira Raja, Vishnu Razdan, Emily Schultheis

**Materials are chosen to meet the strategic needs of DoD, DOE and DHS**

# Low temperature approach of reactive flux growth (LTRFG) of $\beta$ -gallium oxide

## Outlines

- Background and Objectives
  - Need for lattice/chemistry matched substrates for GaN
  - Problems of SiC were motivation for  $\beta$ -Ga<sub>2</sub>O<sub>3</sub>
- $\beta$ -Ga<sub>2</sub>O<sub>3</sub> Comparable to 2H-SiC
  - Low temperature reactive flux growth
  - Characteristics of 2H-SiC
- $\beta$ -Ga<sub>2</sub>O<sub>3</sub> Crystals
  - Growth approach: Co-crystallization and growth from melt
  - Solution growth
  - Characterization of grown  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> Crystals



**We have an excellent team and facilities to execute semiconductor, EOIR and Laser materials and device programs at UMBC**

# Low temperature growth of $\beta$ -Ga<sub>2</sub>O<sub>3</sub>

## Objective:

The objective is to investigate scientific parameters for growing cm size of an exciting novel large bandgap material  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> crystals by low temperature innovative approach of reactive solution growth (LTRSG) method.

## Benefits:

- The bandgap of  $\beta$ - Ga<sub>2</sub>O<sub>3</sub> is >4.5 eV, which corresponds to the second largest bandgap after that of diamond among semiconductors, good thermal conductivity and mobility and excellent substrate for GaN.
- This will enable to develop optical devices,  $\beta$ - Ga<sub>2</sub>O<sub>3</sub> deep ultraviolet photo detectors and high power amplifiers for variety of commercial and DOD applications.

*We had used a similar approach of reactive growth for 2H-SiC ( $\beta$ -Ga<sub>2</sub>O<sub>3</sub>)*

# We have excellent growth and characterization facilities at UMBC



Solution crystallizer



High temperature annealing furnace



Blue M furnace for high temp synthesis



Flux growth furnace



Custom CZ growth furnace

Bulk synthesis and crystal growth



Bridgman growth gold furnace



Three zone CVD growth furnace



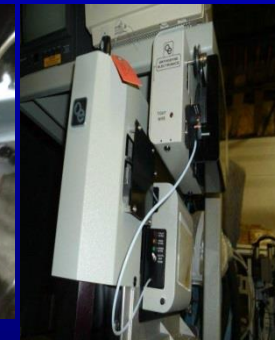
Low temperature Bridgman furnace



Multizone vertical furnace



Wire Saw



Wire Bonder

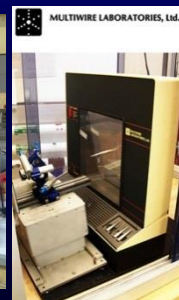
Thin Film, nano crystal growth and fabrication



Nanoline



Optical and nanoSEM microscopes



Realtime X-ray Laue



Probe facilities for materials and device characterization

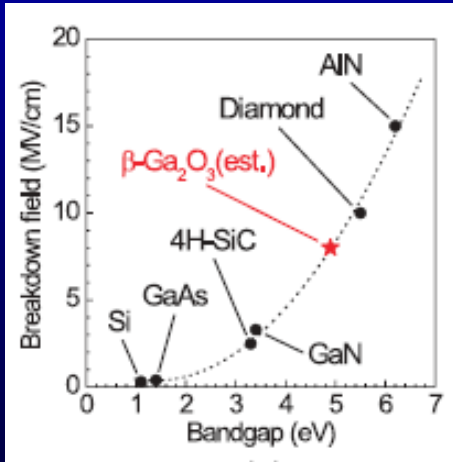


Crystal and device characterization

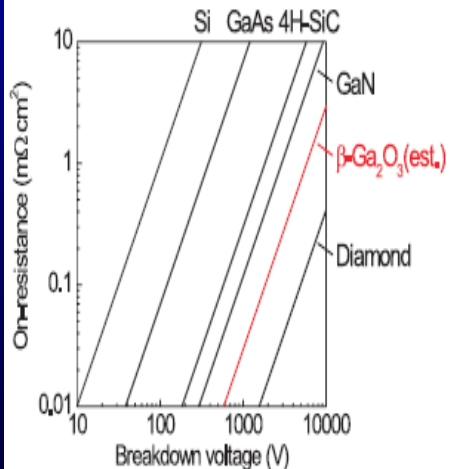
# $\beta\text{-Ga}_2\text{O}_3$ : A comparison of the merit of several semiconductors

**Figure of merit for various semiconductor materials**

	Si	4H-SiC	GaN	Diamond	$\beta\text{-Ga}_2\text{O}_3$
Bandgap (eV)	1.1	3.18	3.3	5.5	<b>4.85</b>
Electron Mobility (cm <sup>2</sup> /Vs.)	1400	700	1200	2000	<b>400</b>
Breakdown Field E <sub>b</sub> (MV/cm)	0.3	2.5	3.3	10	<b>8</b>
Dielectric Constant	11.8	9.7	9.0	5.5	<b>9.5</b>
Baliga's Figure of Merit	1	340	870	24664	<b>3444</b>



**Bandgap and breakdown field for semiconductor materials**



**Theoretical limit of on-resistance as the function of breakdown voltage**

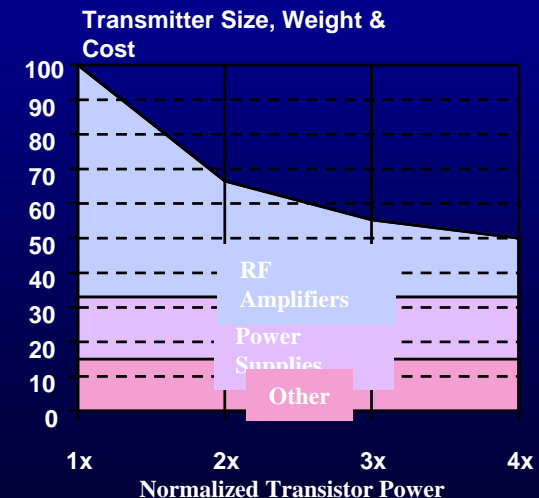
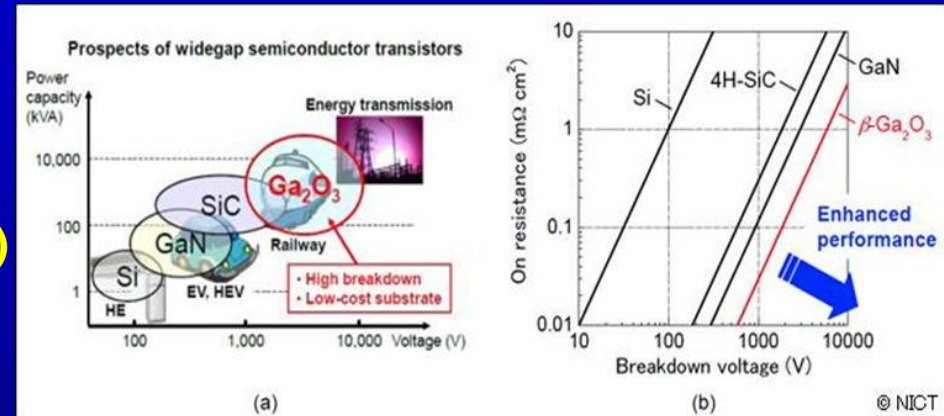
**Bandgap and breakdown voltage of  $\beta\text{-Ga}_2\text{O}_3$  of semiconductor materials**

**$\beta\text{-Ga}_2\text{O}_3$  has very good properties compared to developed wide bandgap materials**

# System Advantages of $\beta$ -Ga<sub>2</sub>O<sub>3</sub> Material

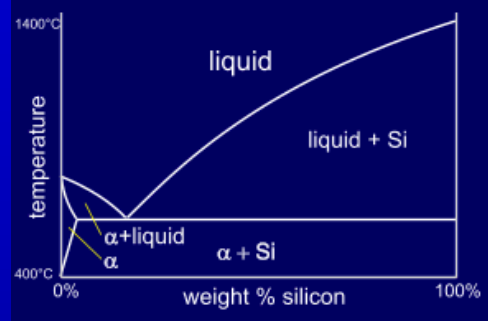
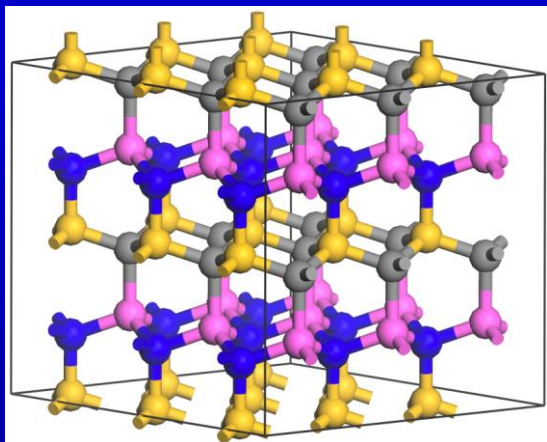
**Higher Power Transistors Can Lead to 50% Reduction in Transmitter Size, Weight, Cost**

- **Higher Power Transistors**
  - Fewer Circuits
  - Smaller, Lower Cost Systems
- **Improved Thermal Conductivity (>GaN) & Higher Temperature Capability with GaN**
  - More Cooling Options
- **Higher Voltage Operation**
  - Higher Device Impedance
- **Reduced Parasitics per Watt**
  - Higher Efficiency & Wider Bandwidth
- **Pure Heterostructure Possible (GaON)**
  - Reduced thermal and mismatch problems

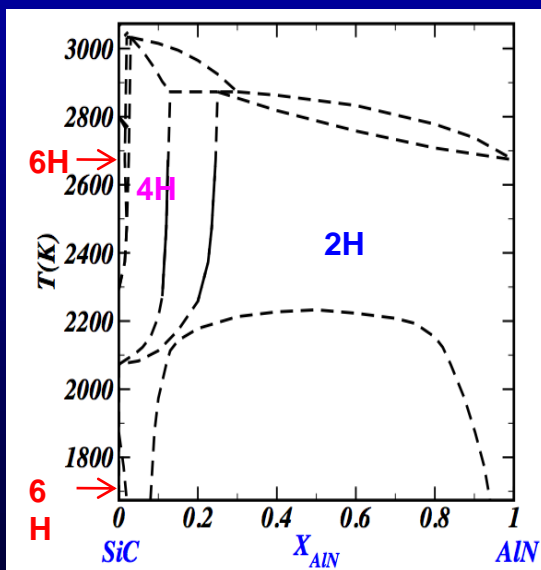


**Potentially 10X lower surface traps as compared to AlGaIn**

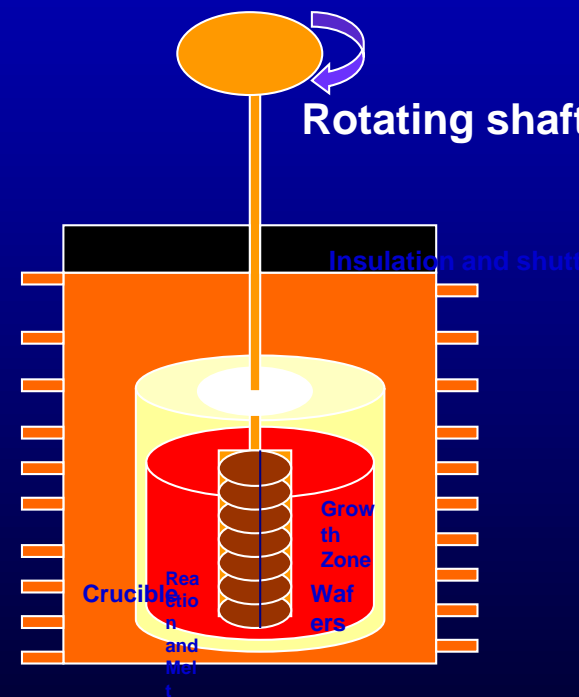
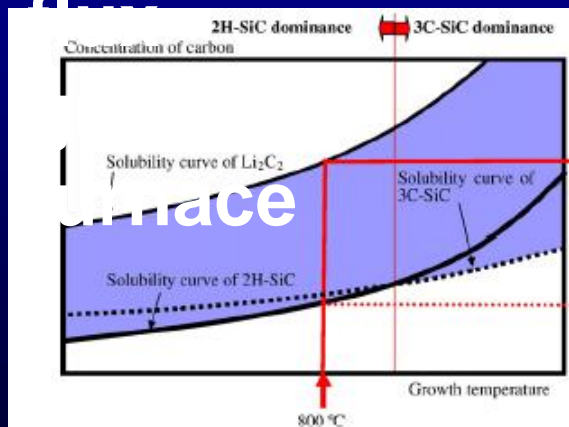
# 2H-SiC: Synthesis by Reactive Liquid-Solid Incorporation (RLSI) Growth is a novel concept to produce large volume



Al-Si liquid melt is suitable for incorporation of carbon and SiC both



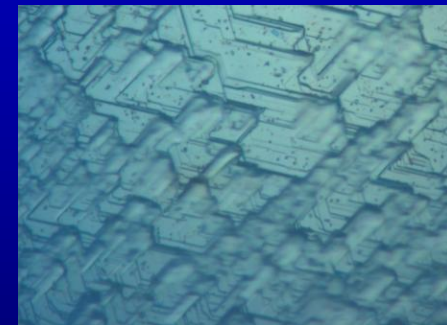
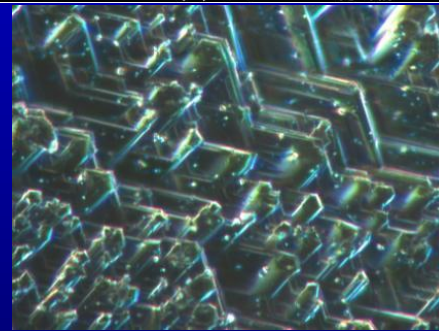
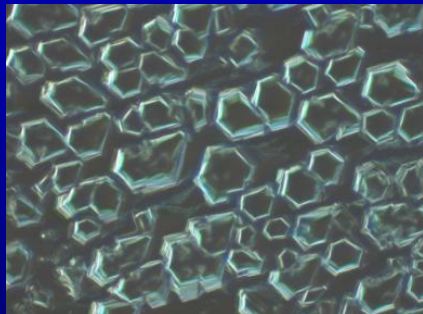
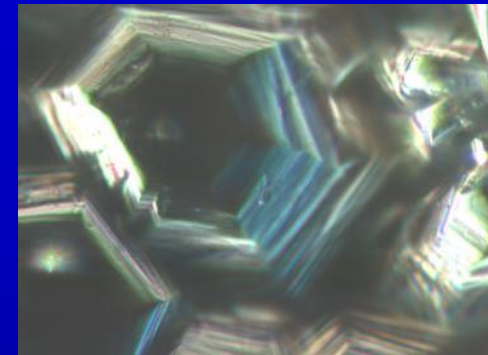
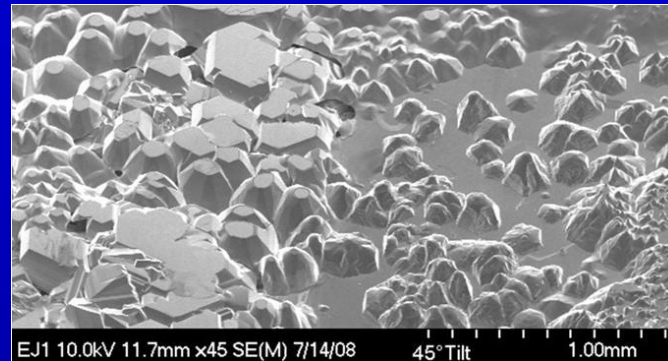
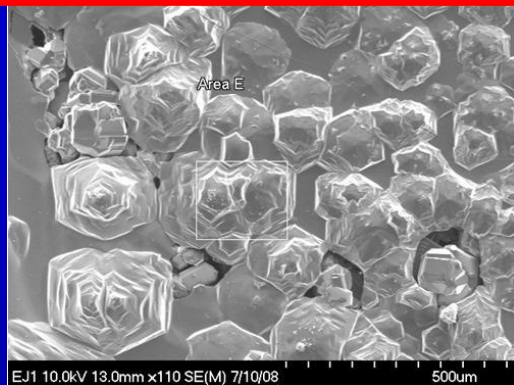
Custom  
growth



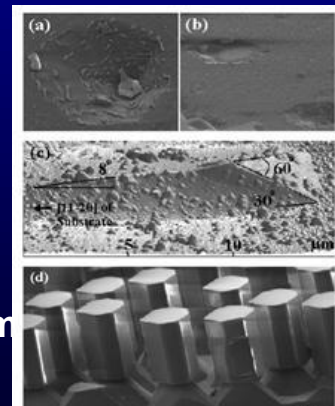
Addition of AlN stabilizes SiC into hexagonal 2H-SiC polytype

*J. Cryst. Growth Des.*, 2010, 10 (8), pp 3508–3514

# Material nucleated as a disc, developed into hexagons and then in film



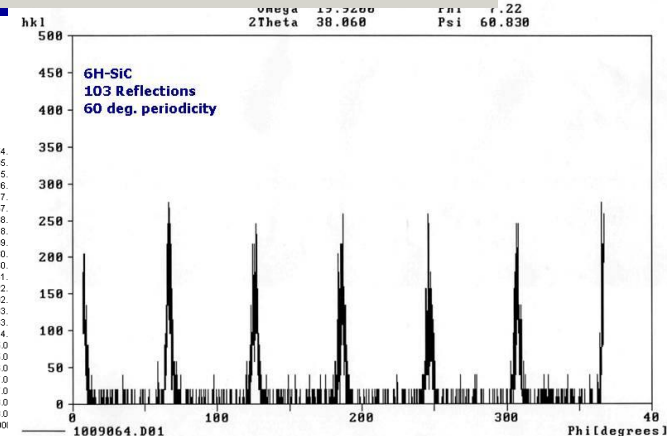
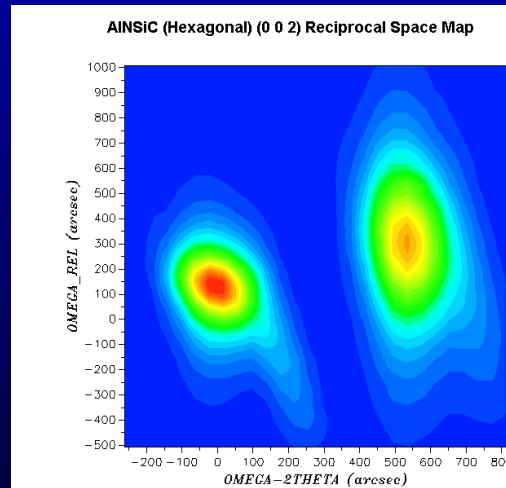
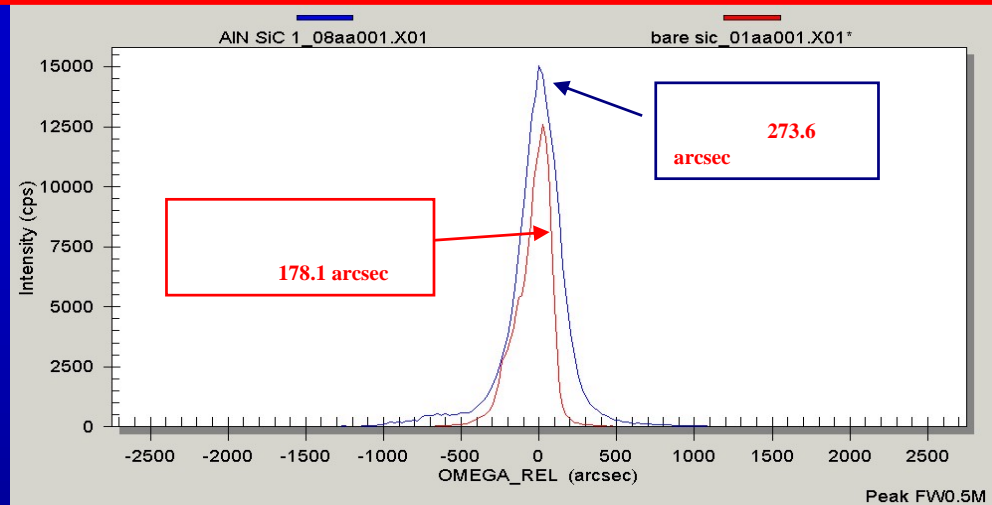
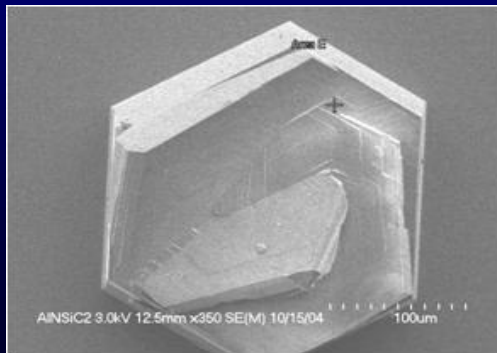
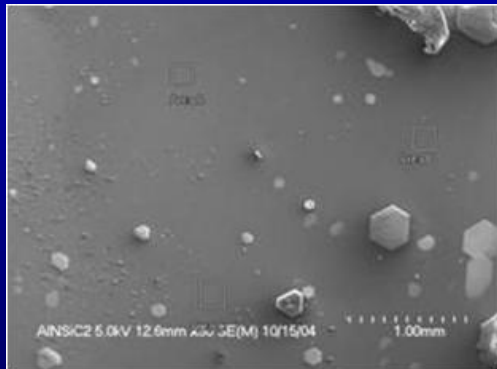
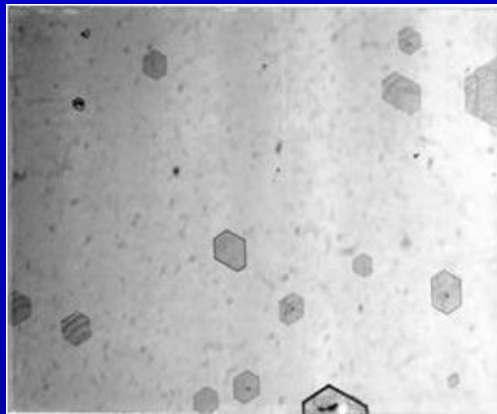
- The SiC deposits begin as discs and grow to become pyramids, before finally coalescing to become a continuous layer.
- 2H is the natural crystal orientation of AlN. AlN rods can be grown on 2H-SiC. The rods coalesce and form a continuous, low-stress layer.
- The rods are typically 12  $\mu\text{m}$  in diameter and 17  $\mu\text{m}$  in length, and exhibit flat (0001) surfaces.



- Nucleation occurred like a disc.
- SiC grew by layer-by-layer growth mechanism.
- We did not observe micropipes in the grains (Yakimova, Yazdi and Syväjärvi – Linköping University, Sweden)



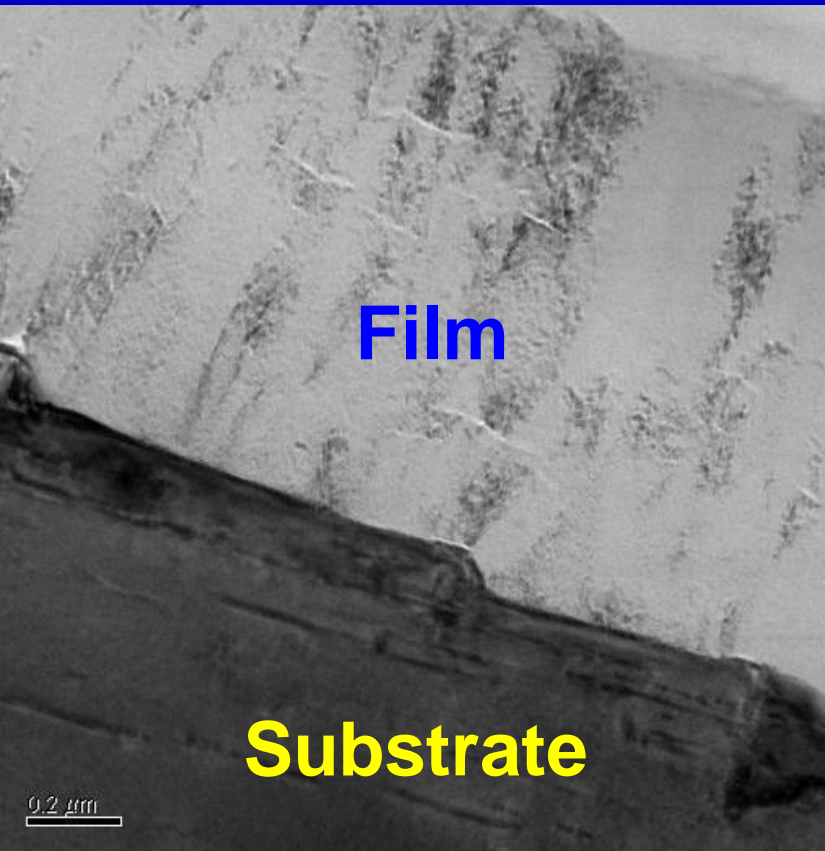
# AlN-SiC alloys have structure of AlN (2H) which indicates that SiC changes to Wurtzite



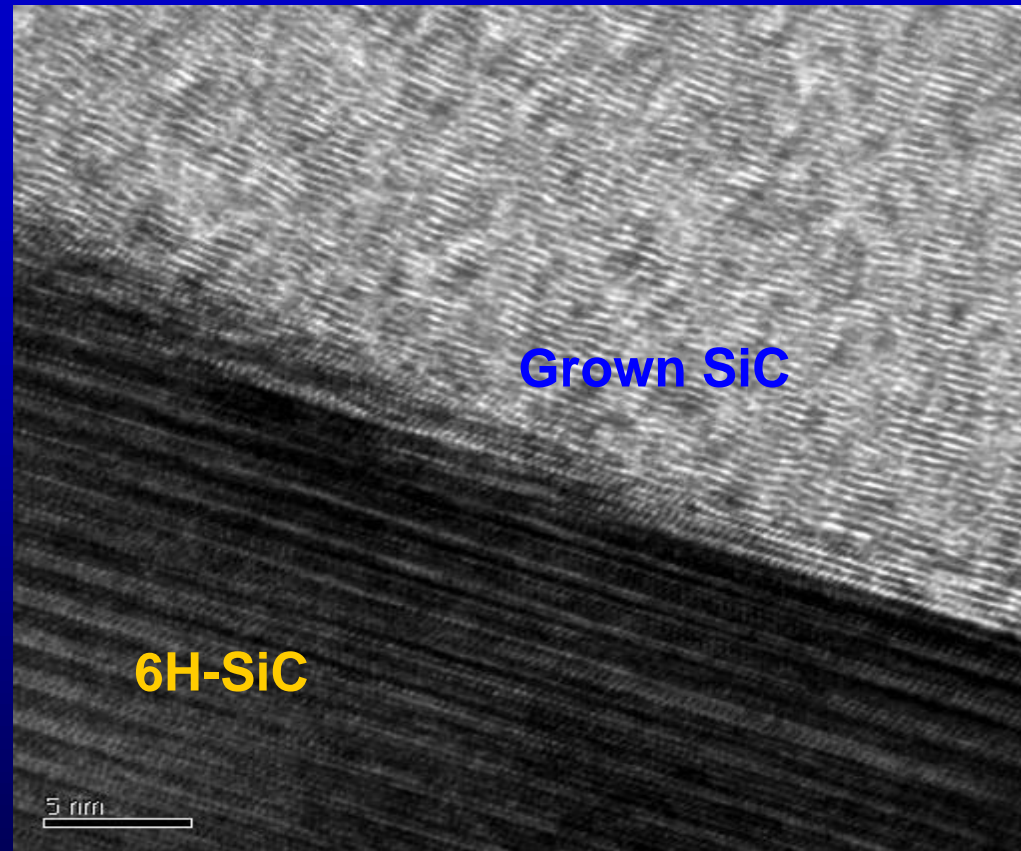
ALNSIC13 Bede Scientific May-23-05

Data shows both substrate and alloy peaks  
Hexagonal morphology was observed

# HRTEM clearly shows formation of Wurtzite structure on 6H-SiC substrate



TEM image showing film and substrate. Film/substrate interface is not flat and film morphology is columnar.

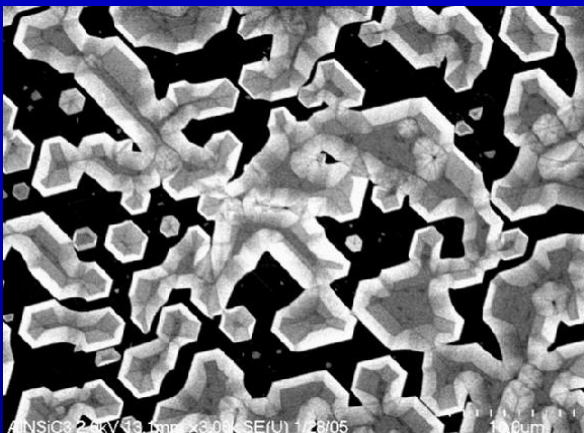


High resolution TEM image showing part of the film and 6H-SiC substrate

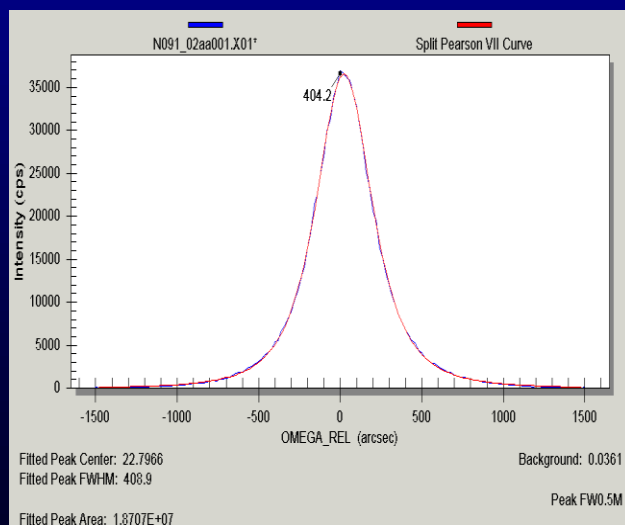
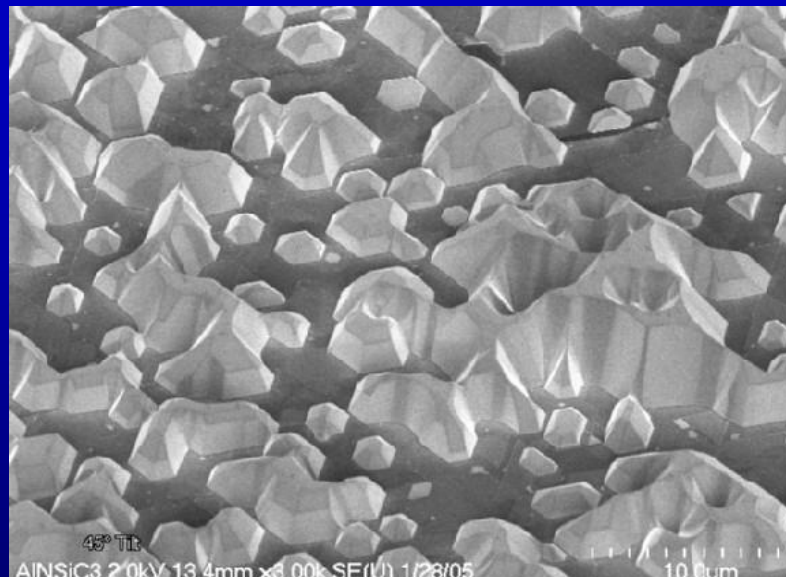
TEM morphology at the interface is very different for the substrate and new material

# SiC grown on thick film of AlN was used for GaN growth.

- Growth morphology of GaN and formation of small angle boundaries
- One can design experiments to reduce the defect density



Growth morphology of GaN



- FWHM is 404.2 arcsec
- GaN peak is located at expected location for epitaxial (0001) GaN
- Important to note that the orientation was maintained from substrate, to alloy material, to GaN, thus the GaN is in fact epitaxial

# Summary for 2H-SiC

- 2H-SiC with pure hexagonal symmetry is stable in presence of AlN
- 2H-SiC has higher electron velocity and larger bandgap which translates into better devices.
- SiC can be grown at temperature below 1200C and hence Si wafers with AlN seed can be used to nucleate SiC.
- Low temperature reactive growth is good process for SiC growth
- Low temperature SiC growth opens possibility of large area SiC growth at low cost
- Low cost large area SiC with superior properties and low defects will be possible

**Low temperature process provides a pathway for growth of large crystals**

# Low temperature growth of $\beta$ -Ga<sub>2</sub>O<sub>3</sub>

## Objective:

The objective is to investigate scientific parameters for growing cm size of an exciting novel large bandgap material  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> crystals by low temperature innovative approach of reactive solution growth (LTRSG) method.

## Background

- Commercially available Ga<sub>2</sub>O<sub>3</sub> often consists of a mixture of the  $\alpha$ - and  $\beta$ -phases.
- The  $\beta$ -phase is the thermodynamically stable modification, with a formation Gibbs energy that surpasses that of corundum.
- Additionally the  $\gamma$ -,  $\delta$ -, and  $\varepsilon$ -Ga<sub>2</sub>O<sub>3</sub> phases are known in literature.

*We used a similar approach of reactive growth process for  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> crystal*

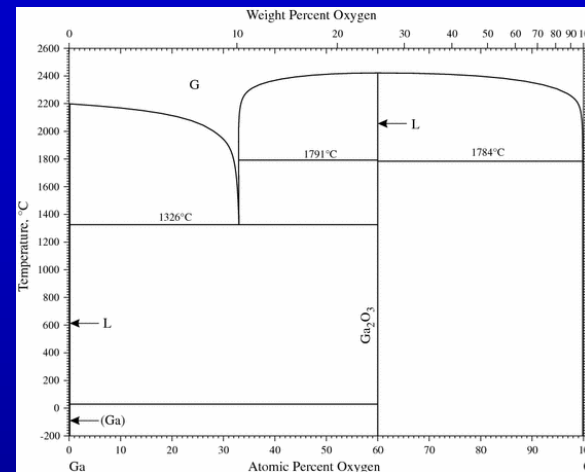
# Chemistry and Solid-liquid equilibrium data

## Background on Crystal Chemistry

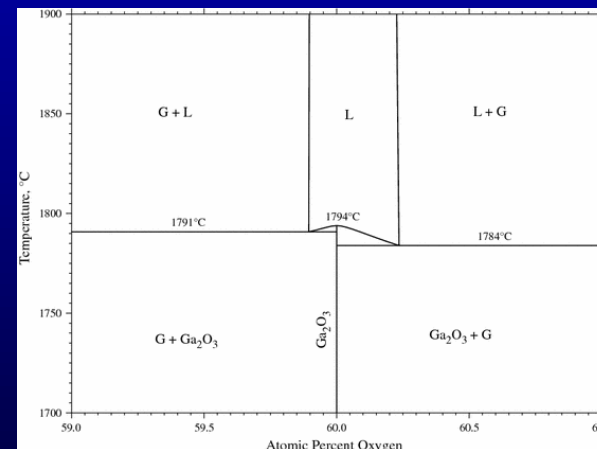
Gallium oxide is precipitated in hydrated form upon neutralization of acidic or basic solution of gallium salt. It can occur in five different modifications,  $\alpha, \beta, \delta, \gamma$  and  $\epsilon$ .

$\beta$ - $\text{Ga}_2\text{O}_3$  is the most stable form

- $\beta$ - $\text{Ga}_2\text{O}_3$  can be prepared by heating nitrate, acetate, oxalate or other organic derivatives above  $1000^\circ\text{C}$ .
- $\alpha$ - $\text{Ga}_2\text{O}_3$  can be obtained by heating  $\beta$ - $\text{Ga}_2\text{O}_3$  at 65kbars and  $1100^\circ\text{C}$  for 1 hour giving a crystalline structure. The hydrated form can be prepared by decomposing precipitated and "aged" gallium hydroxide at  $500^\circ\text{C}$ .
- $\gamma$ - $\text{Ga}_2\text{O}_3$  is prepared by rapidly heating the hydroxide gel at  $400^\circ\text{C}$ - $500^\circ\text{C}$ .
- $\delta$ - $\text{Ga}_2\text{O}_3$  is obtained by heating  $\text{Ga}(\text{NO}_3)_3$  at  $250^\circ\text{C}$ .
- $\epsilon$ - $\text{Ga}_2\text{O}_3$  is prepared by briefly heating  $\delta$ - $\text{Ga}_2\text{O}_3$  at  $550^\circ\text{C}$  for 30 minutes



H. Okamoto, J Phase Equilibrium and Diffusion, December 2008, Volume 29,6, pp 550-551



M. Zinkevich and F. Aldinger, Thermodynamic Assessment of the Gallium-Oxygen System. *J. Am. Ceram. Soc.*, 2004, 87(4), p 683-691

***A very tight control of composition and temperature and high gradient will be required to grow from melt***

# Approach for the low temperature growth

There are two approaches:

- (α)β-Ga<sub>2</sub>O<sub>3</sub> single crystals using tin and tin-gallium solution method (problem of Sn)
- (b) GaCl<sub>3</sub> nitration followed by treatment with hydroxide
- (c) Growth by hydrolysis of gallium(III)-isopropoxide and from aqueous GaCl<sub>3</sub> solution by addition of aqueous tetramethylammonium hydroxide (TMAH)

- Controlling the growth rate: We will control the growth rate and hence the quality by cooling rate
- Nucleation rate: We propose to use nucleus grown by urea method

• Characterization:

- Bandgap
- Lattice parameters:
- Quality by X-ray rocking curve
- Morphology by SEM
- Resistivity and Mobility
- Impurities by PL

Phase β-Ga <sub>2</sub> O <sub>3</sub>	
Crystal system	Monoclinic
Space group	C2/m
a [° A]	12.214 (3)
b [° A]	3.0371 (9)
c [° A]	5.7981 (9)
Beta	103.83 (2)
Cell volume[ ° A <sup>3</sup> ]	208.8

- Fabrication by cutting and polishing (AFM)

*This low cost approach has potential for producing large and high quality crystals*

# There are pathway to move ahead where we can avoid use of Sn

- **Cocrystallization in catalyzed environment**: In this biomimetic approach, the hydrolytic enzyme motif from the silicateins, consisting of nucleophilic hydroxyl of serine and amine of histidine, catalyzes the hydrolysis of the gallium precursors and results in both GaOOH and  $\text{-Ga}_2\text{O}_3$  nanocrystal formation with a wide size range.
  - Urea melt
- **Growth from Melt**
- **Ga-Sn Experiment**: Experiments were performed using several Ga:Sn ratio in the range (1:1 to 3:1 ratio)
- **Effect of temperature**: High temperature required (1100 to 1300C)
  - Sn-Ga Eutectic melt
- **Solution growth experiments with and without urea**: Dissolve 5 gram urea and 1-2 gm gallium salt in  $\text{HNO}_3 + \text{H}_2\text{O}$  (10%  $\text{HNO}_3$ ) in a test tube
  - Dissolved and cooled for recrystallization
  - Got gallium oxide co-crystallize with urea

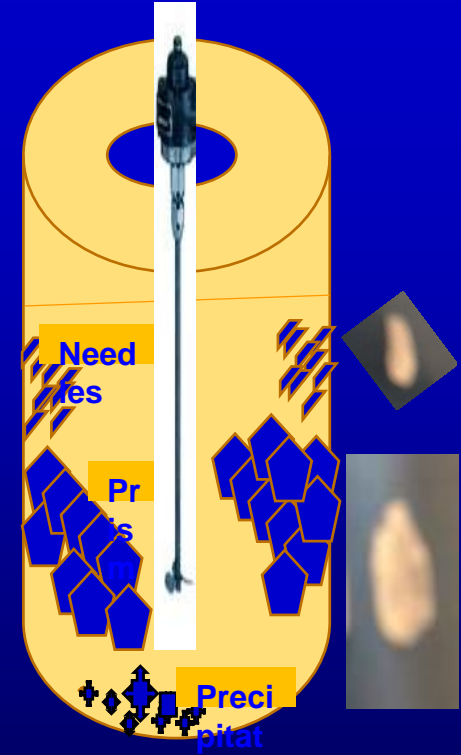
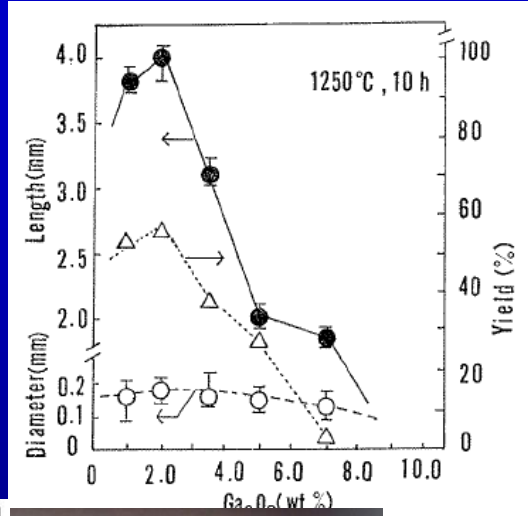
***Co-crystallization and flux growth are two promising approach***



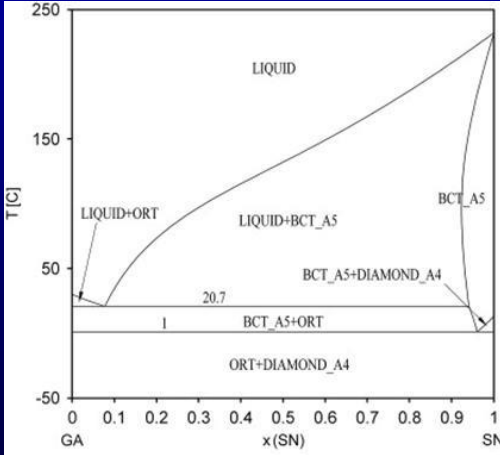
# Synthesis by Reactive Liquid-Solid Incorporation (RLSI)

## Growth is a novel concept to produce large volume

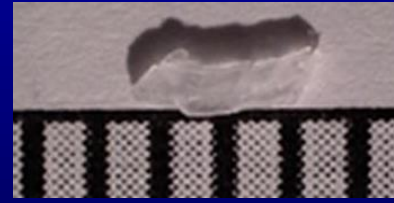
RLSI Furnace, Alumina crucible and substrate



Different morphologies were observed in Alumina crucible



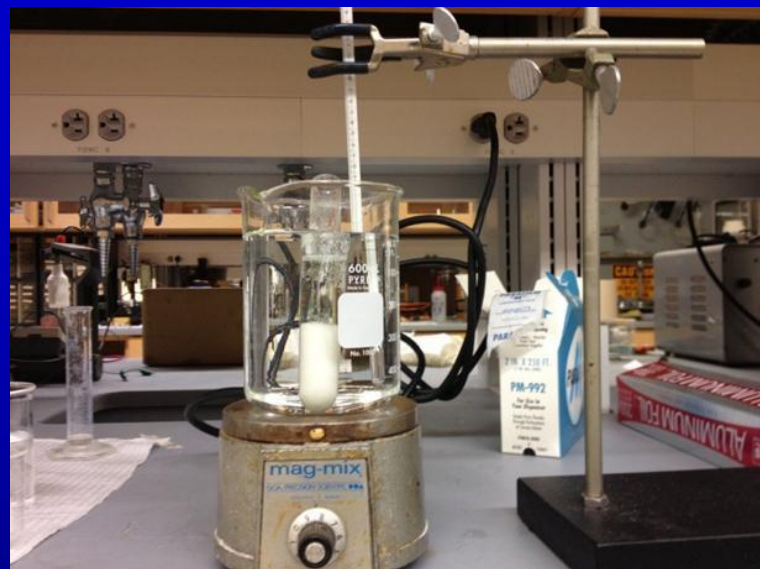
Ga-Tin Eutectic melt was used



Effect of composition of Ga/Tin has been reported in Japan

**Ratio of flux and position controls morphology. However, by stirring one can alter the morphology and hence the crystal size**

# Synthesis from gallium chloride and hydroxide material



There traces of gallium chloride  
in presence of traces of HCl.  
Excess  $\text{HNO}_3$  is added to avoid  
chloride

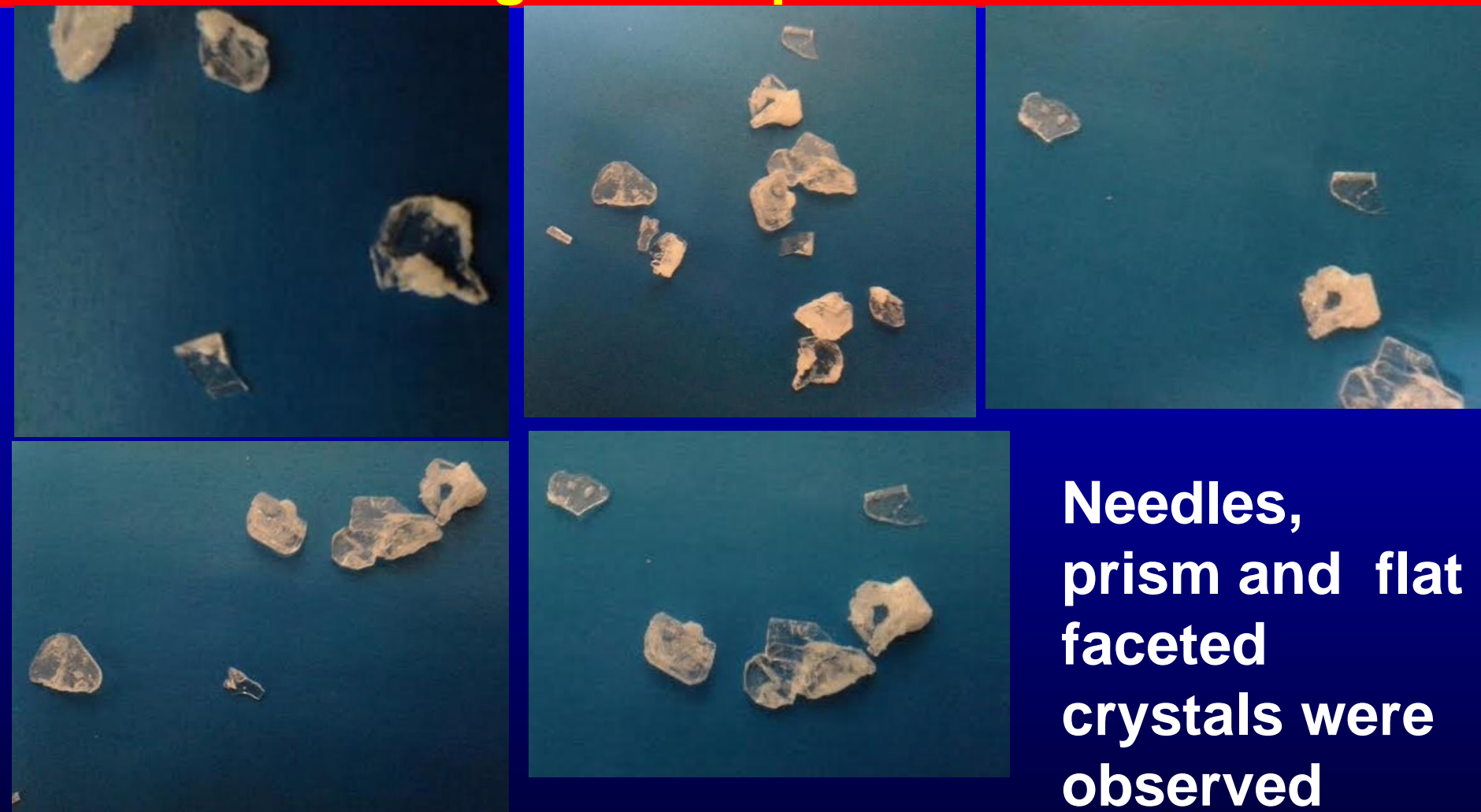
***Dissolution and nitration and neutralization has to be optimized***

# Morphology of crystals



***Uncontrolled crystallization was slow and produced mm to cm size crystals***

# Morphology of crystals harvested from the second growth experiment



Needles,  
prism and flat  
faceted  
crystals were  
observed

***Harvested crystals had different morphologies***

# Four major morphologies of crystals were observed in the second growth experiment



- Different morphology of as grown 2mm to 7mm size virgin crystals
- Very important clues about effect of
  - pH
  - Concentration and gradient
  - Nucleation
  - Stirring
  - Cooling rate

***Crystals had four morphologies indicating effect of pH, concentration and rotationstirring***

# Preliminary characterization of grown crystals

- Characterization:
  - Bandgap
    - Measured optically: 4.7eV
  - Lattice parameters:
    - $a = 5.797^{\circ}\text{A}$
    - $b = 3.04$
    - $c = 12.198$
    - Angle  $\beta = 103.68$
  - Morphology by SEM
  - Quality by X-ray rocking curve
  - Resistivity and Mobility
  - Impurities by PL
  - Fabrication by cutting and polishing (AFM)

Phase $\beta\text{-Ga}_2\text{O}_3$	
Crystal system	Monoclinic
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Beta	103.83 (2)
Cell volume[° A <sup>3</sup> ]	208.8

*X-ray produced the lattice parameters very closed to literature value*

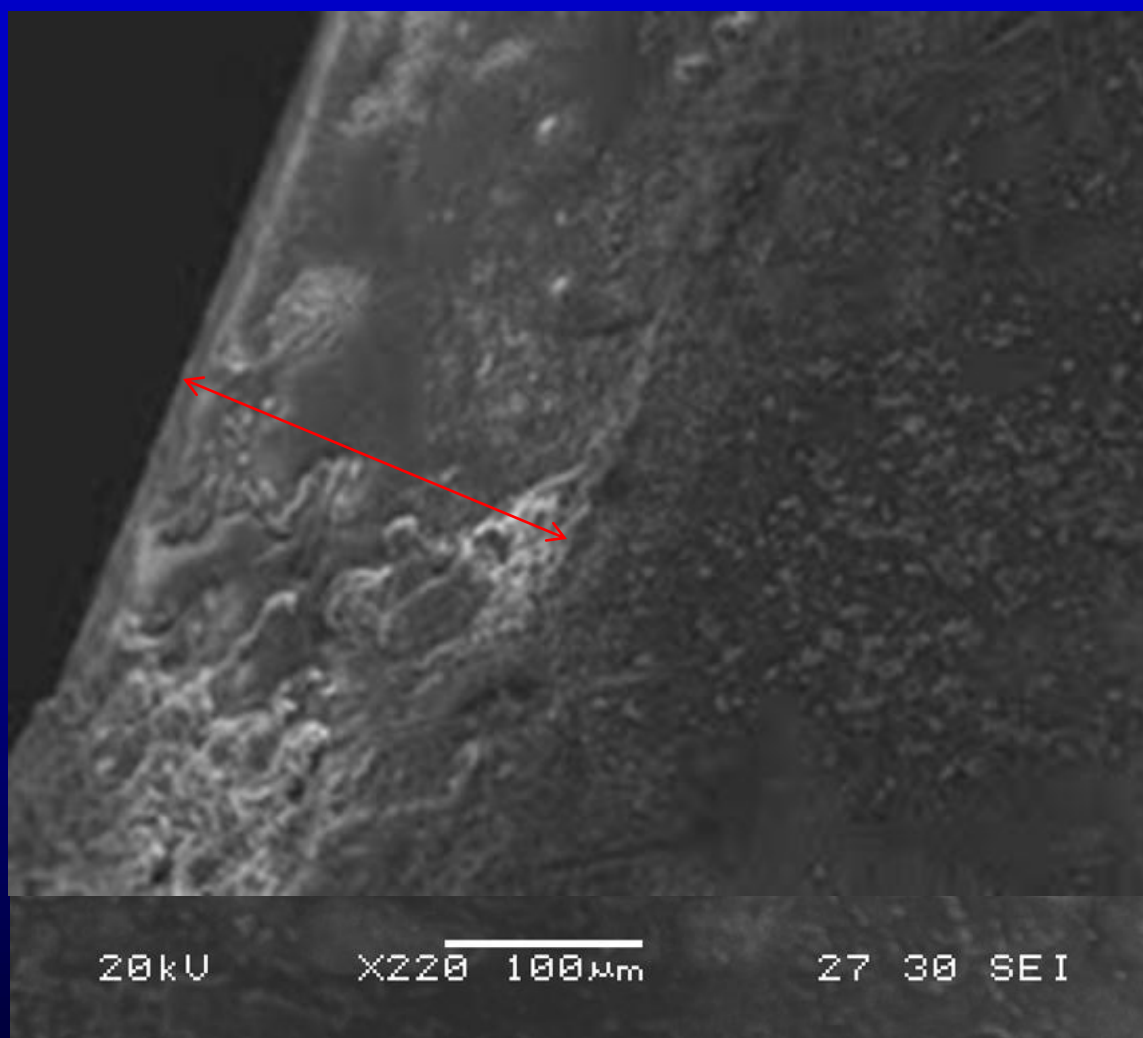
# JEOL SEM system was used for morphology



- System used : JEOL 560
- Voltage: 20KV
- No sputtering
- We studied morphology to determine layering
- Composition
- Growth steps
- Surface structures
- Major point defects
  - Precipitates
  - Voids
  - Cracks
  - bubbles

**Morphology and composition (EDS) was studied**

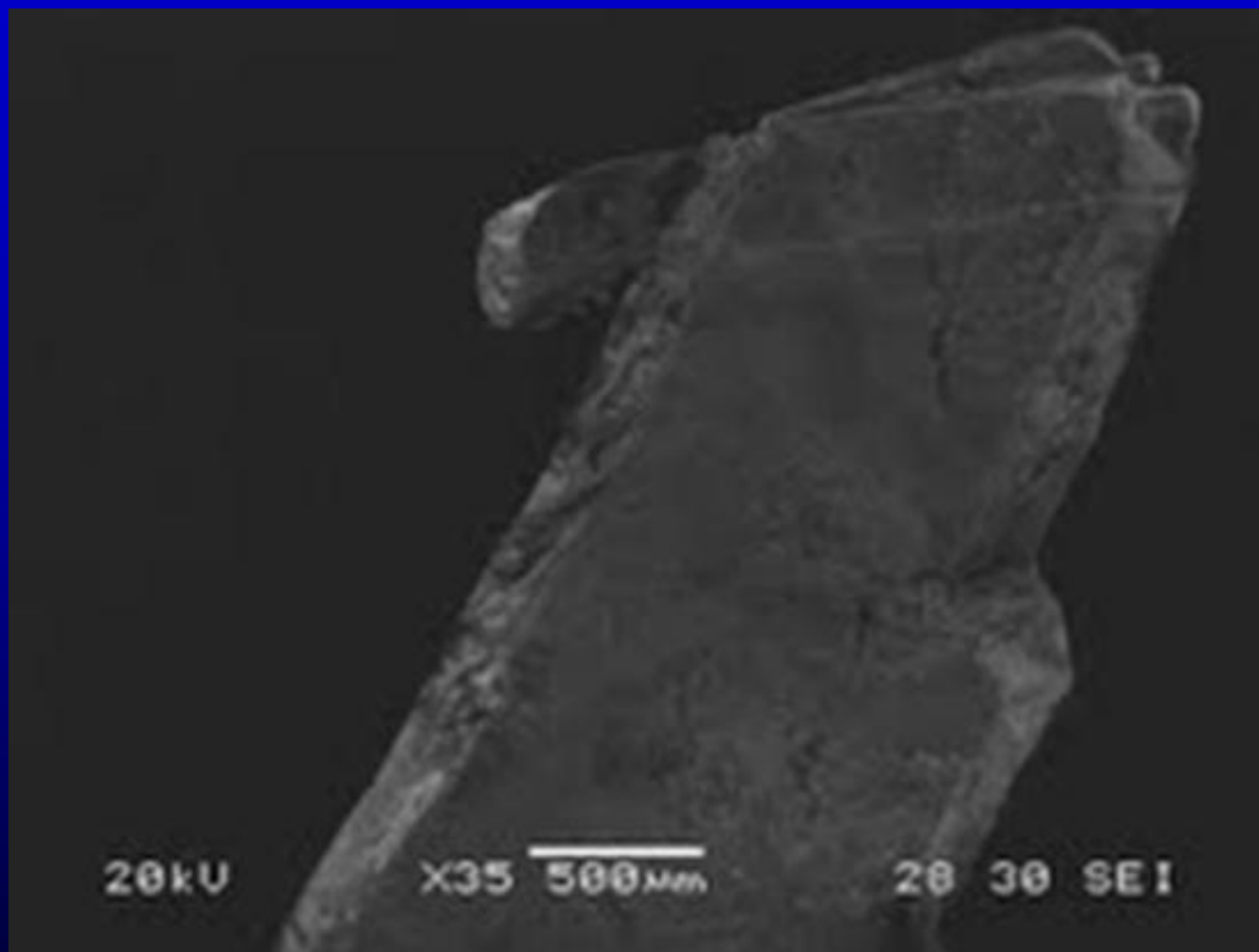
# As grown needles shape crystals



***Large needles (100  $\mu\text{m}$  diameter) of  $\beta\text{-Ga}_2\text{O}_3$  were observed***

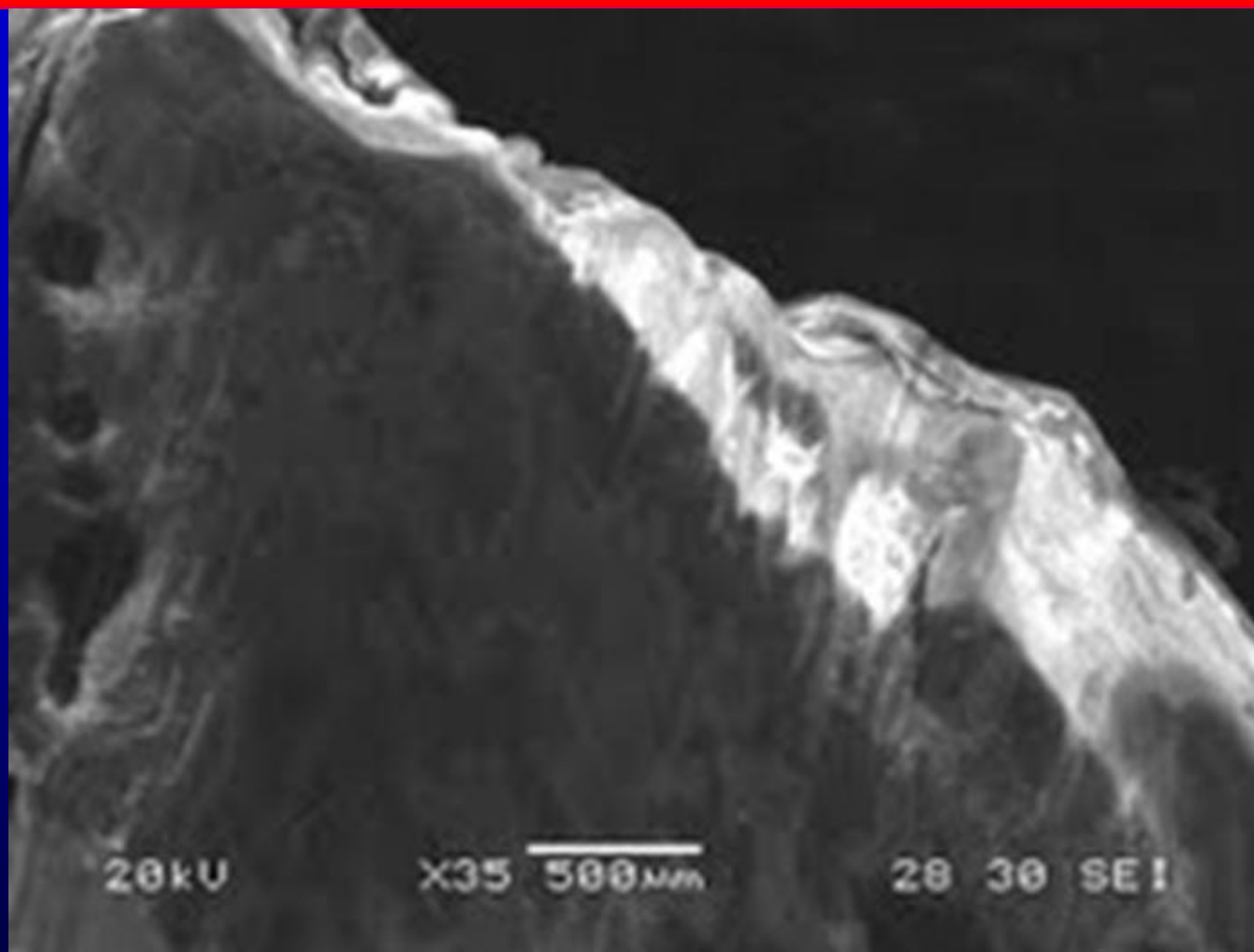


# As grown prism shape crystals



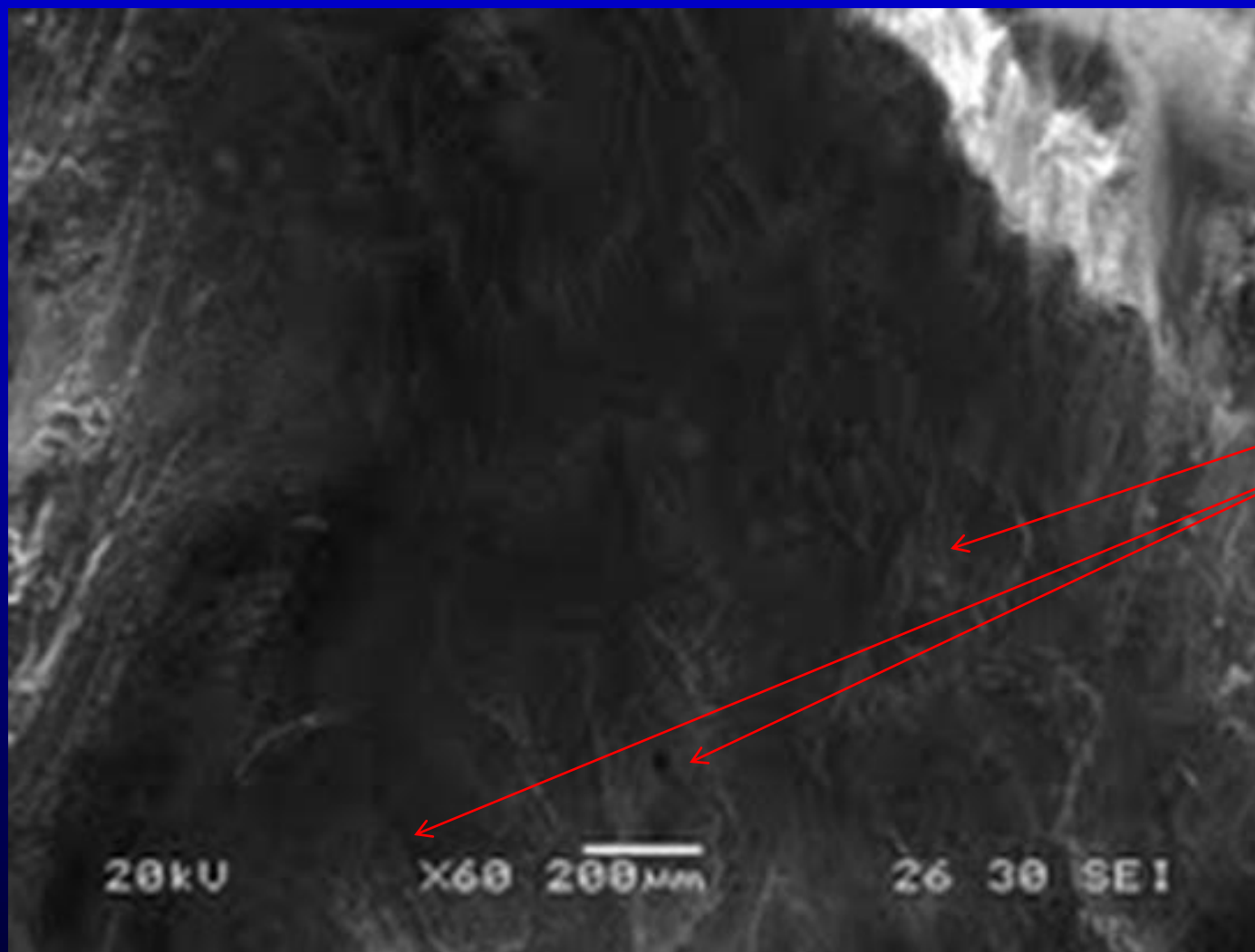
***Large prism shape crystals of  $\beta\text{-Ga}_2\text{O}_3$  were observed in the crucible***

# Morphology of Gallium Oxide



***We did not observe soft layering and hence no sign of cleavage was observed***

# Morphology of $\beta$ -Ga<sub>2</sub>O<sub>3</sub> crystal



Growth steps

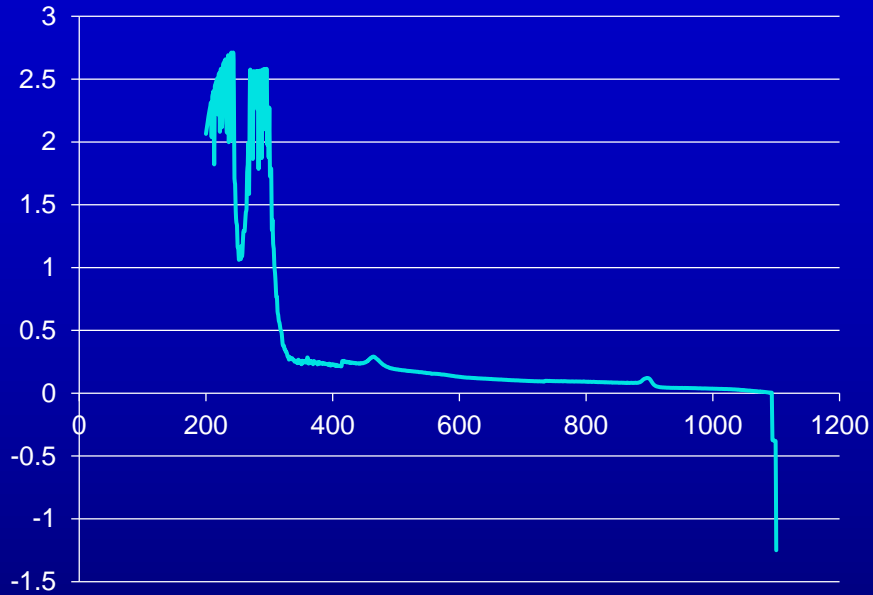
***Growth steps (needles and prism) were observed in virgin crystals***

# Optical absorbance and transparency was measured for the $\beta$ -Ga<sub>2</sub>O<sub>3</sub> as grown crystal



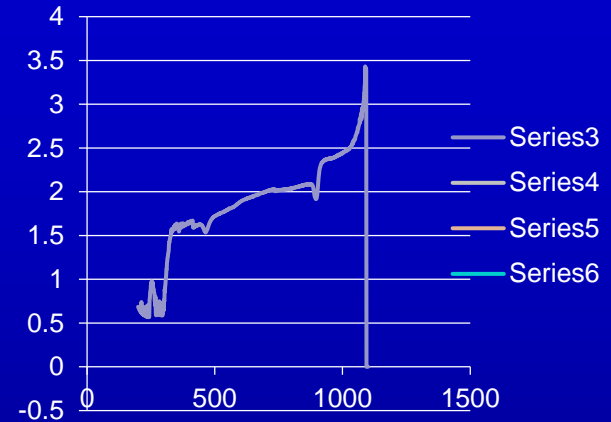
***Optical method showed bandgap >4.7eV for as grown crystal***

# Optical characteristics of the material

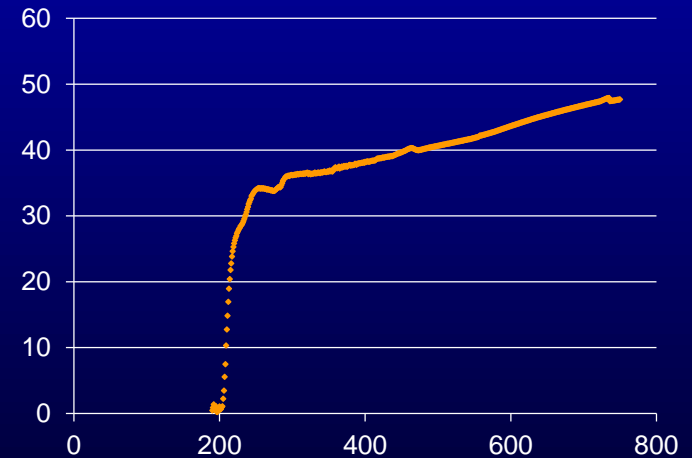


Absorbance with as grown crystal

Transmission of as grown new crystal crystal

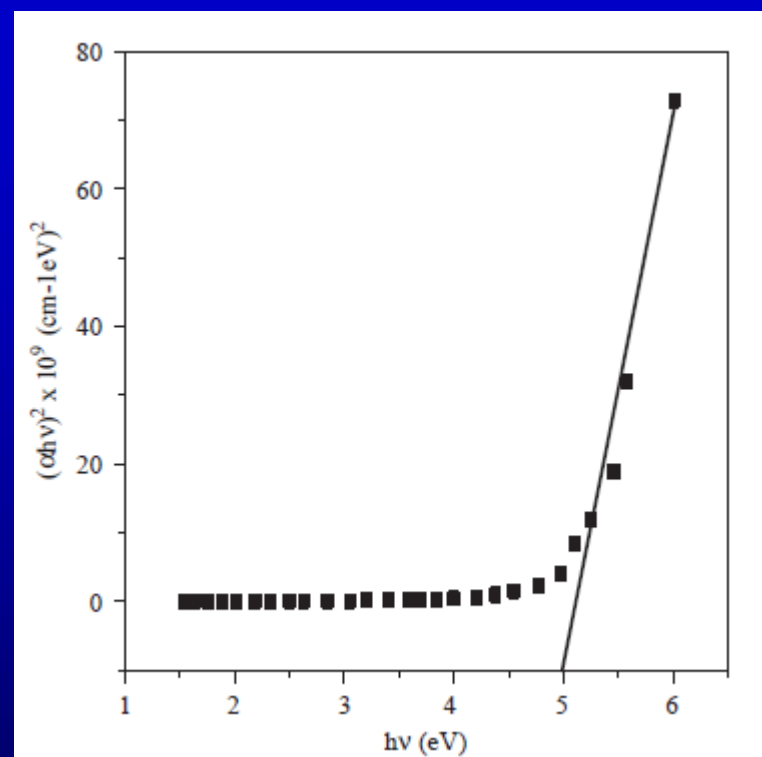
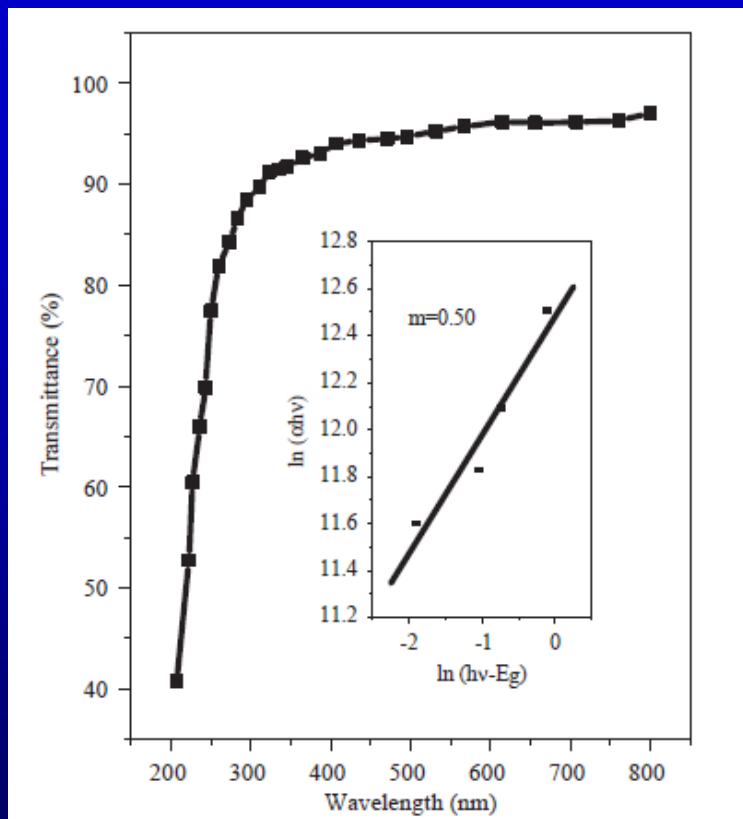


Cutoff wavelength for crystal in tin flux



**Both absorbance and transmission indicated the bandgap of 4.7eV  
New crystals indicated bandgap of 5.90eV**

# Optical characteristics of film grown by dip coating (JCG 276 (2005) 204-207)

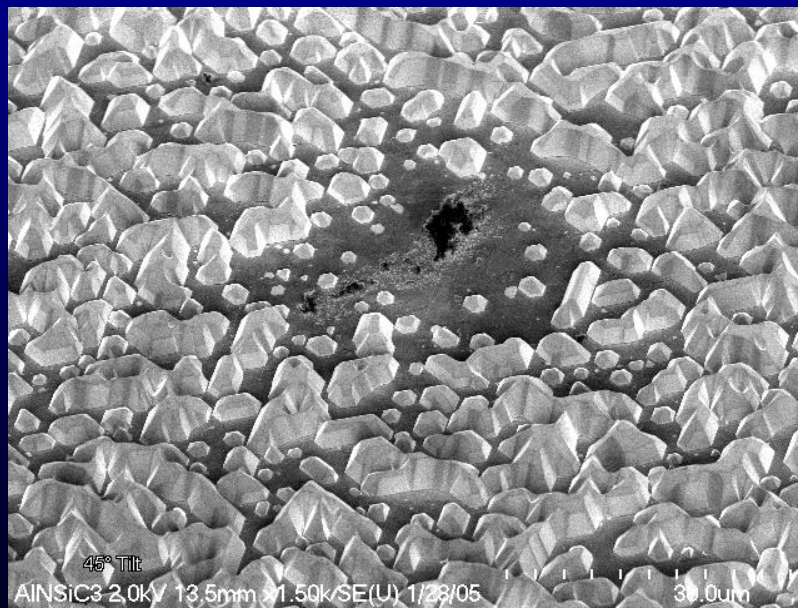
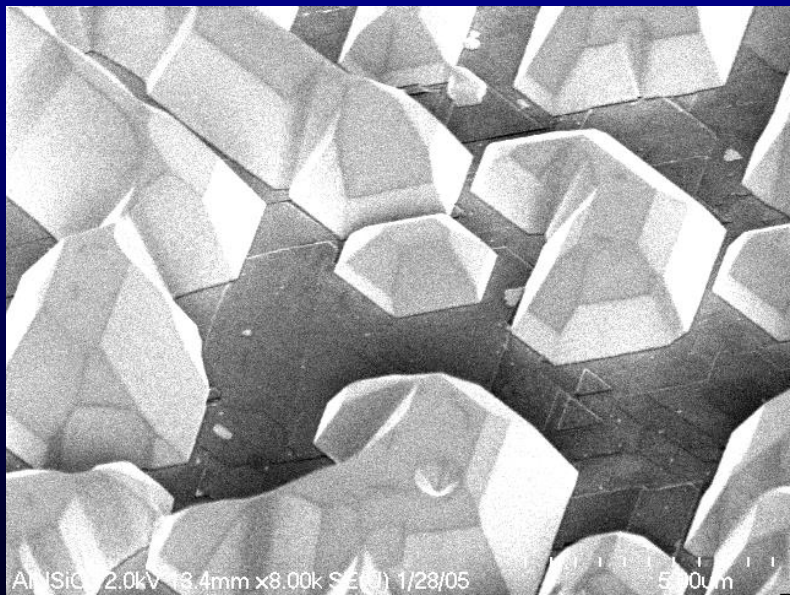
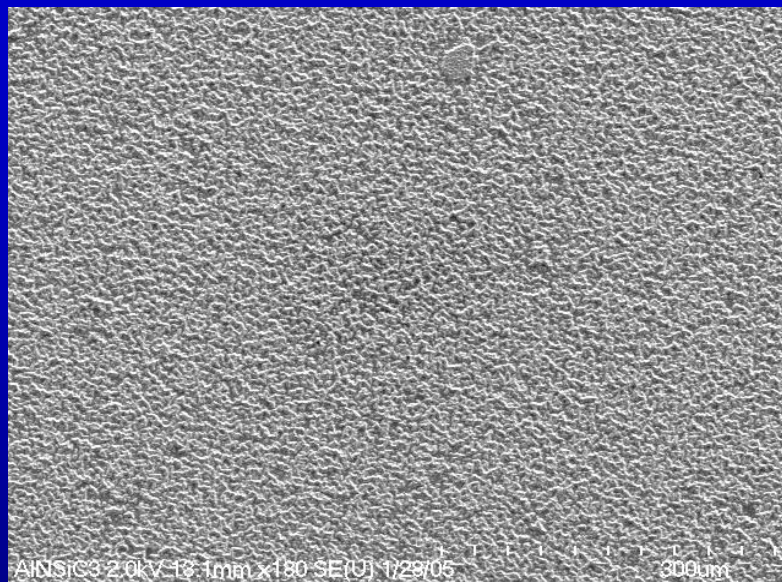
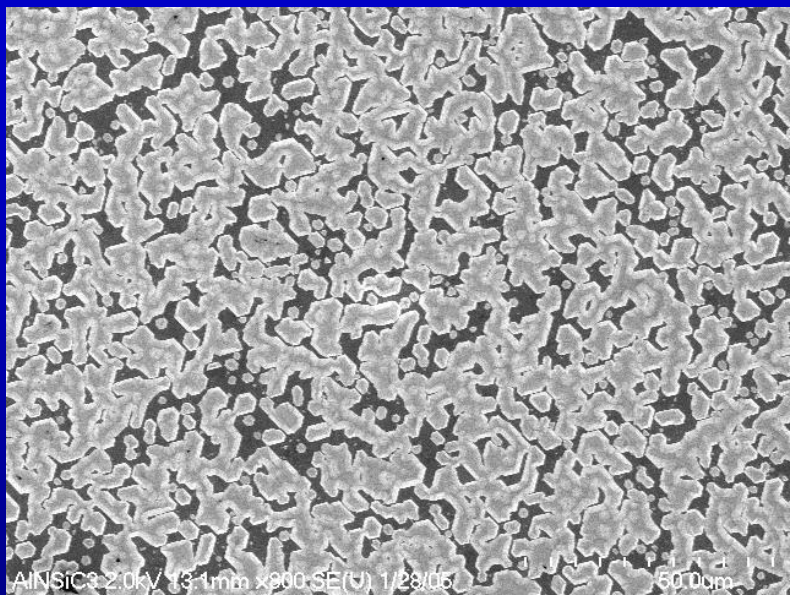


Transmittance Vs wavelength trace for pure  $\alpha\text{-Ga}_2\text{O}_3$  thin film. Inset shows  $\ln(\alpha h\nu)$  vs.  $\ln(h\nu - E_g)$  plot.

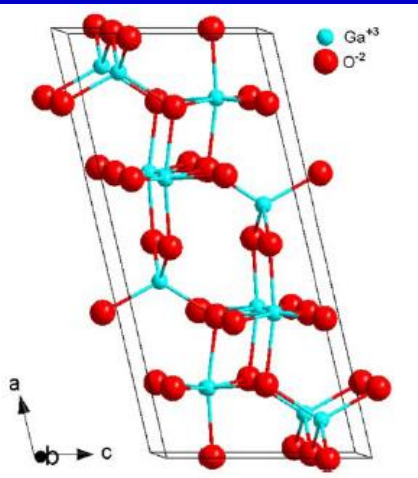
$(\alpha h\nu)^2$  vs.  $h\nu$  trace for pure  $\alpha\text{-Ga}_2\text{O}_3$  thin film indicates  $E_g = 4.98 \text{ eV}$ .

**Both absorbance and transmission indicated the bandgap  $> 4.7\text{eV}$**

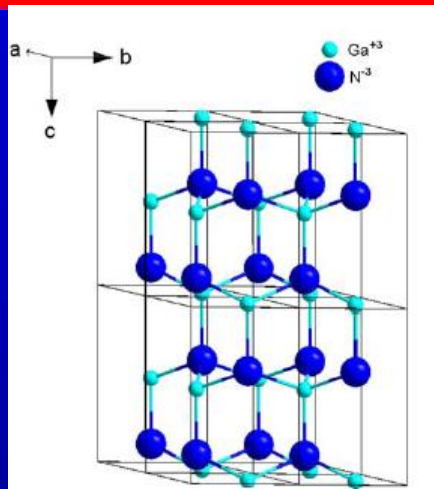
# Morphology of GaN shows excellent morphology and Ostwald ripening



# Suitability $\beta$ -Ga<sub>2</sub>O<sub>3</sub> for GaN Devices

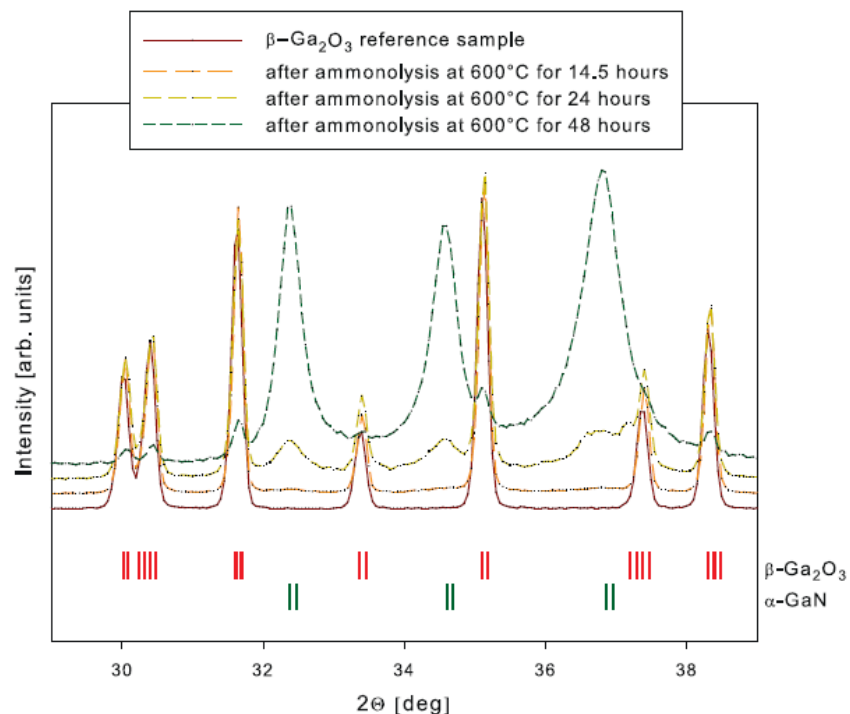


Crystal structure of  $\beta$ -Ga<sub>2</sub>O<sub>3</sub>



Hexagonal structure of GaN

X-Ray Diffraction pattern for amonolyzed at 600C for extended time duration



**Literature indicates that a buffer layer can be grown for deposition of GaN growth.**



# Summary

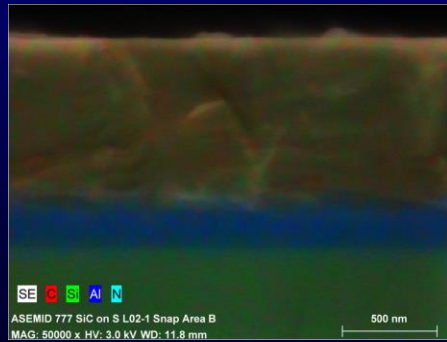
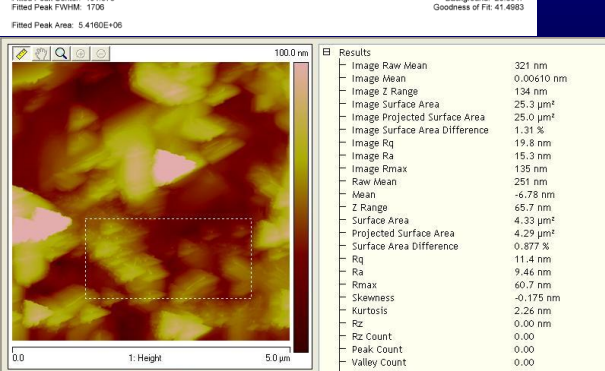
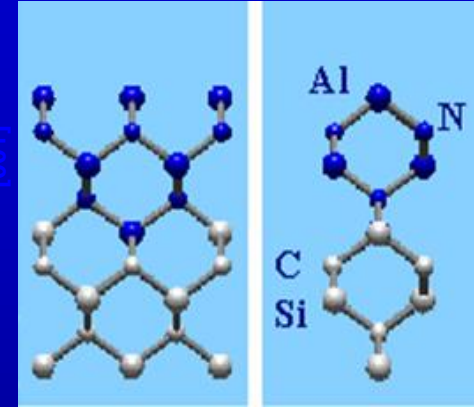
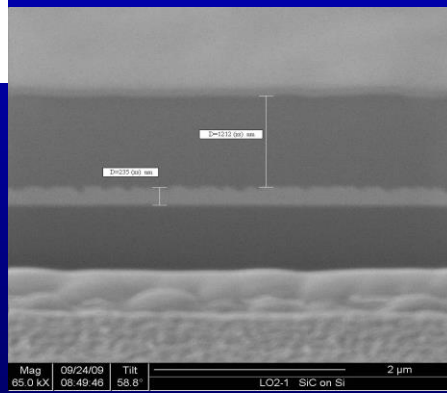
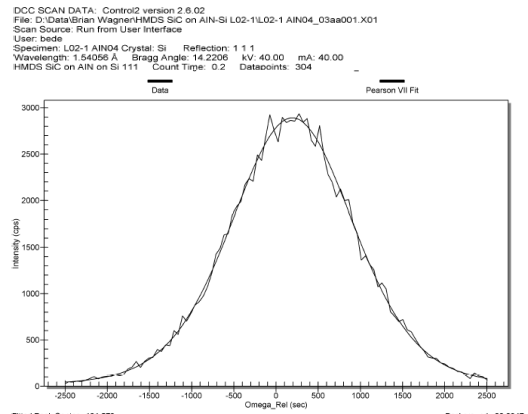
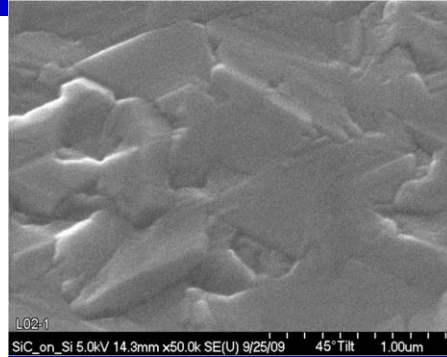
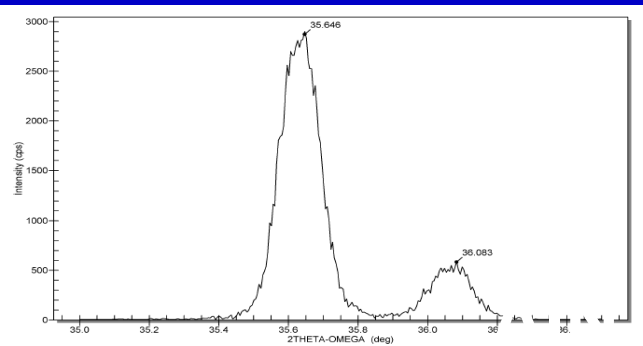
- A low temperature growth method for the growth of a novel wide bandgap materials  $\beta\text{-Ga}_2\text{O}_3$
- The approach involves low temperature reactive flux growth of crystals
- Preliminary studies indicated that co-crystallization and Sn-based flux can be used for the growth of large crystals
- Small crystals grown in Sn showed a bandgap of 4.7eV, while crystals with nitrate showed as high as bandgap of 5.90 eV.
- The proposed method is suitable for scaling the growth without large equipment investments
- Lattice parameters, bandgap, thermal conductivity and mobility are main parameters for investigation
- The  $\beta\text{-Ga}_2\text{O}_3$  is an excellent substrate for GaN growth

***Preliminary results based on small crystals are very exciting***

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***Thank you very much***

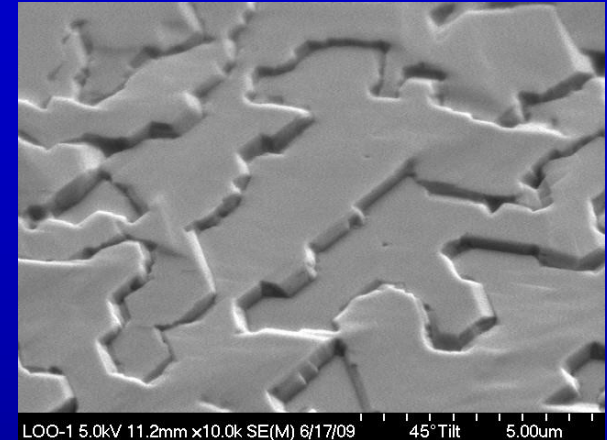
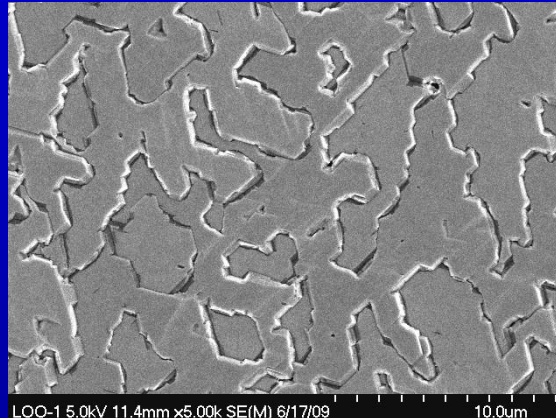
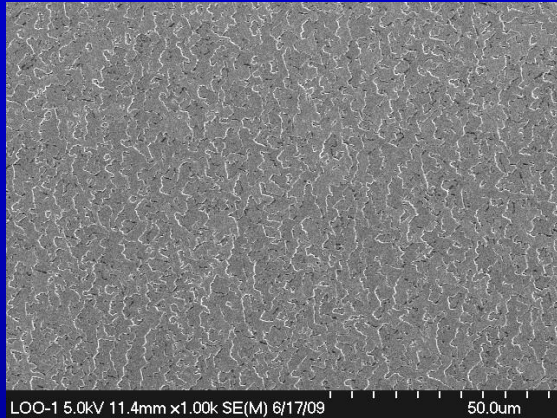
# XRD Survey Indicates Crystalline SiC Deposited on AlN/Si <111> Substrate.



- For coverages <1/2 Monolayer (ML), nitrated Si-pairs are formed
- At 1ML zincblend sites are metastable: Hydrazine like Si<sub>4</sub>N<sub>2</sub> complexes are formed coating the surfaces
- Nitridation is favored, and growth is inhibited
- Interface mixing occurs as a stable configuration and reconstruction occurs by deposition of Si<sup>4+</sup> ions.
- There is a possibility of amorphisation of the interface during above process.

References also support that AlN forces SiC to grow to Wurtzite (2H-) symmetry

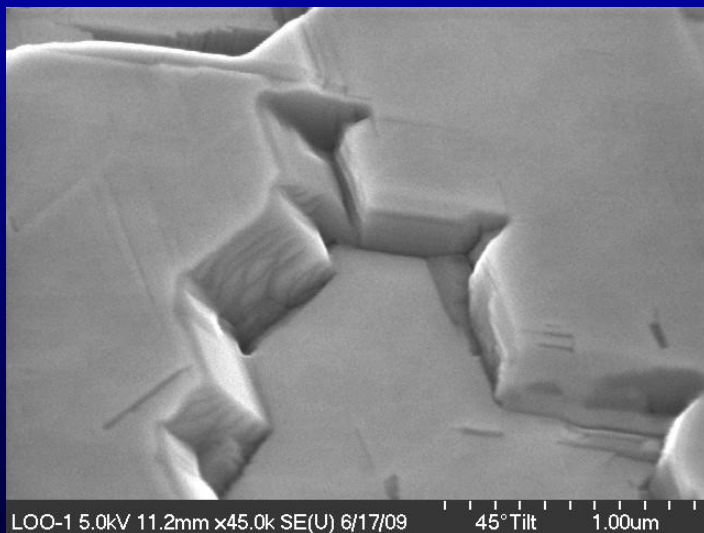
# We developed low temperature growth of SiC . There was no sign of screw dislocation and micropipe



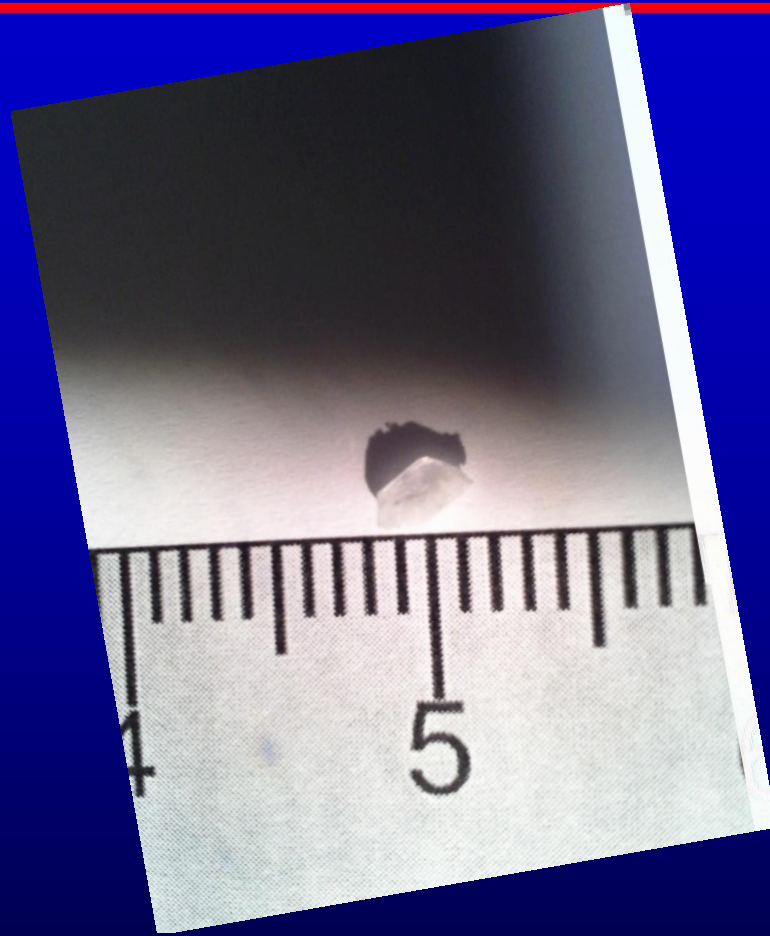
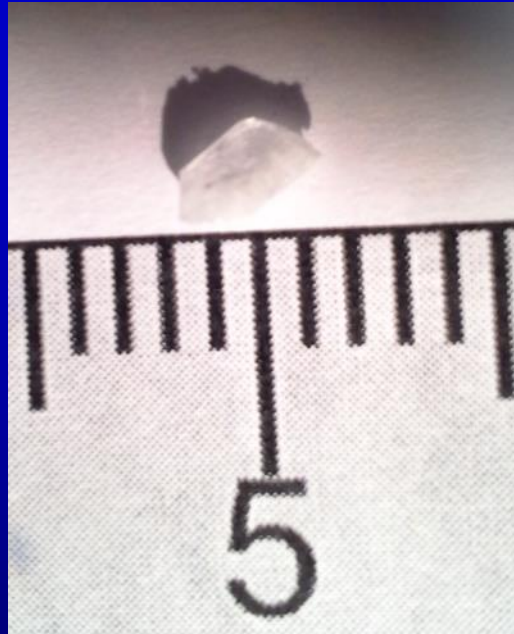
Growth using HMDS shows layer by layer growth

References support 2H- structure in presence of AlN for SiC

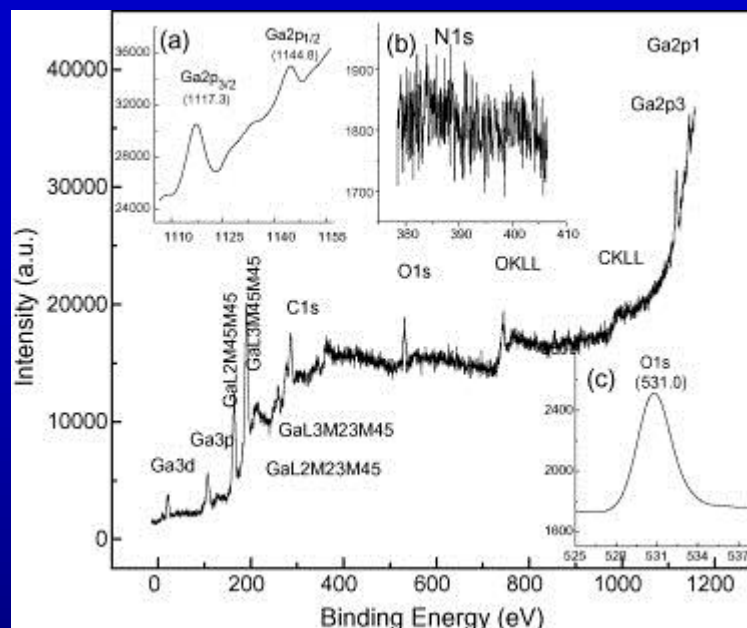
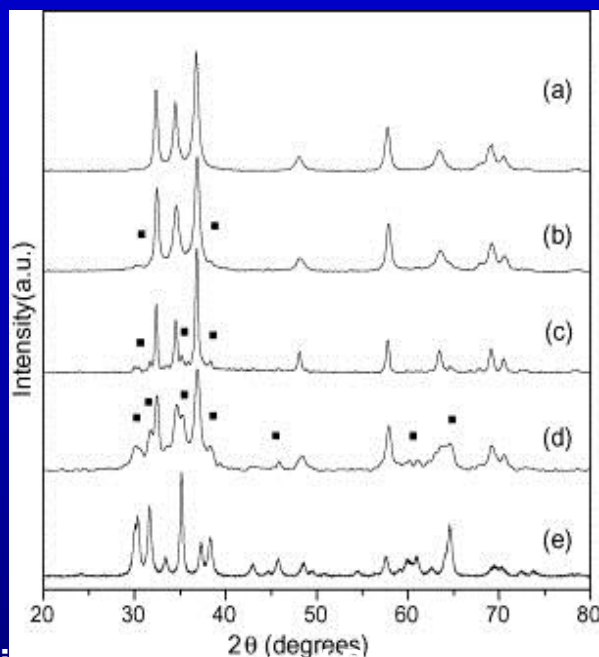
- J-F, Li and R. Watanabe, J. Materials Science. Vol. 26 (1991) 4813.
- Y. Xu, Y; A. Zangvil, A; M. Landon, and F. Thevenot, J. American Ceramic Society, 75 (1992) 325.
- G. E. Hilmas and Tseng-Ying Tien, J. Materials Science, 34 (1999) 5613..
- Miura, M; Yogo, T; S-I, Hirano, J. Ceramic Society of Japan, 101 (1993) 1281.
- Huang, J-L; Jih, J-M, J. American Ceramic Society, 79 (1996), 1262



Low temperature growth occurs as layer by layer mechanism  
and does not show micropipe in the bulk



# Synthesis and structural properties of beta-gallium oxide particles from gallium nitride powder

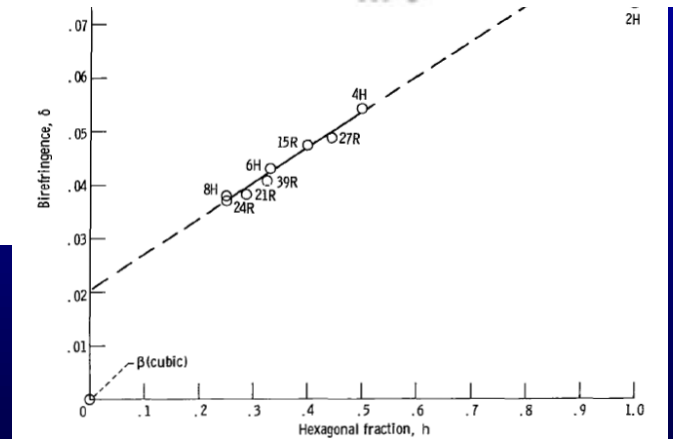
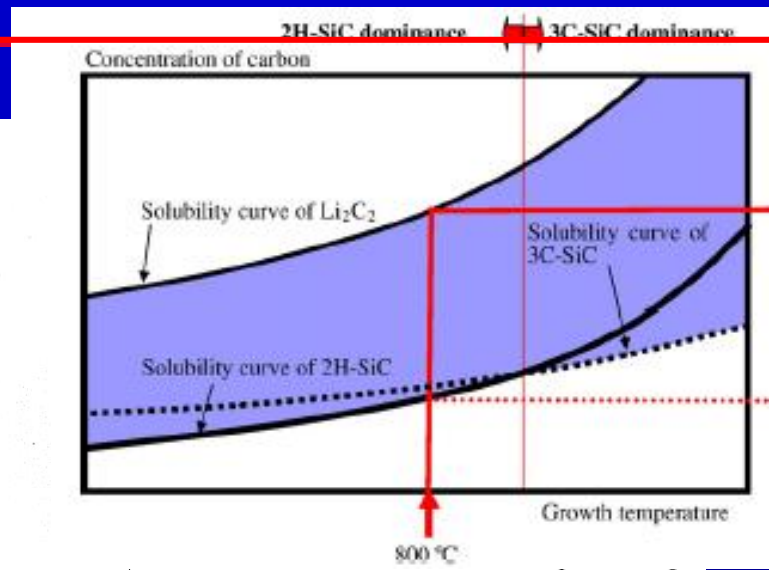
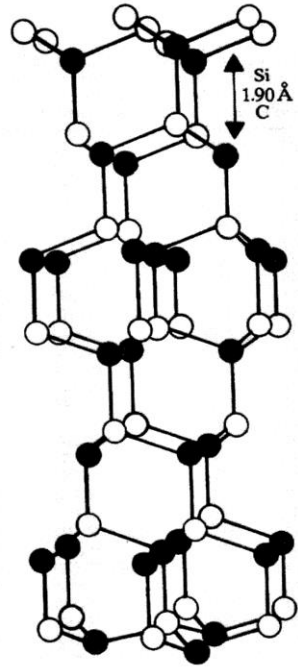
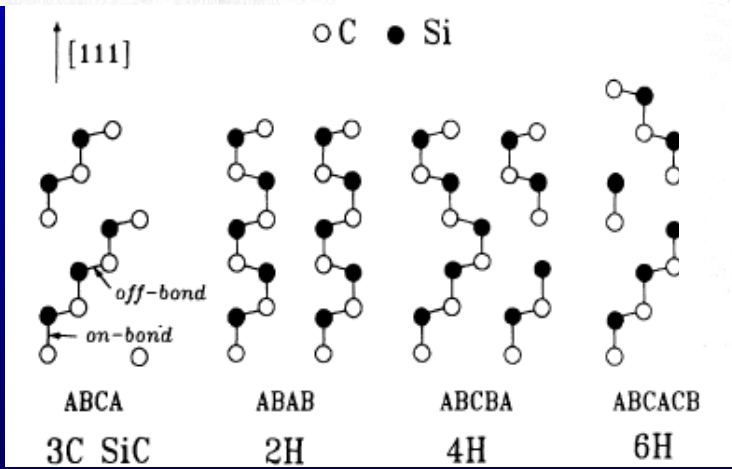
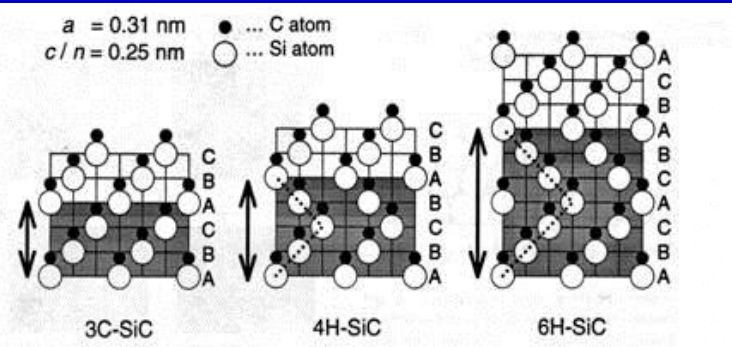


Materials Chemistry and Physics, 101, 1, 15 January 2007, Pages 99–102

Beta-gallium oxide ( $\beta$ -Ga<sub>2</sub>O<sub>3</sub>) powders have been synthesized through simple thermal annealing gallium nitride (GaN) powders in the opening air at 900 °C. The observations revealed that Ga<sub>2</sub>O<sub>3</sub> on the surface of GaN particles has been formed below 500 °C, the rate of Ga<sub>2</sub>O<sub>3</sub>'s formation under air is slow in the temperature range from 500 to 800 °C and is fast in the temperature range of 800–900 °C. The as-obtained products at 900 °C are pure, single-crystalline monoclinic Ga<sub>2</sub>O<sub>3</sub> particles, and the size of  $\beta$ -Ga<sub>2</sub>O<sub>3</sub> is about 50–300 nm

***Oxidation process is very difficult to produce gallium oxide***

# What is 2H-SiC: It is 100% Hexagonal



Low temperature reactive growth facilitates 2H-SiC