

Evaluation of Meso-NH and WRF/CHEM simulated gas and aerosol chemistry over Europe based on hourly observations

A. Berger, C. Barbet, M. Leriche, L. Deguillaume, C. Mari, N. Chaumerliac, Nelson Bègue, Pierre Tulet, D. Gazen, J. Escobar

▶ To cite this version:

A. Berger, C. Barbet, M. Leriche, L. Deguillaume, C. Mari, et al.. Evaluation of Meso-NH and WRF/CHEM simulated gas and aerosol chemistry over Europe based on hourly observations. Atmospheric Research, Elsevier, 2016, 176-177, pp.43-63. http://www.sciencedirect.com/science/article/pii/S0169809516300230>. <10.1016/j.atmosres.2016.02.006>. http://www.sciencedirect.com/science/article/pii/S0169809516300230>.

HAL Id: hal-01307985 https://hal.archives-ouvertes.fr/hal-01307985

Submitted on 27 Apr 2016 $\,$

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Evaluation of Meso-NH and WRF/CHEM simulated gas and

2 aerosol chemistry over Europe based on hourly observations

A. Berger^{1*}, C. Barbet^{2*}, M. Leriche^{1*}, L. Deguillaume², C. Mari¹, N. Chaumerliac², N. Bègue³, P. Tulet³, D. Gazen¹, J.
Escobar¹.

- ⁵ ¹Laboratoire d'Aérologie, UMR 5560 (CNRS, Université Paul Sabatier), Toulouse, France
- 6 ²Laboratoire de Météorologie Physique, UMR 6016 (CNRS, Université Blaise Pascal), Aubière, France
- 7 ³Laboratoire de l'atmosphère et des Cyclones, UMR 8105 (CNRS, Météo-France, Université de La Réunion), Saint-
- 8 Denis de La Réunion, France.
- 9 *Correspondence to:
- 10 M. Leriche (Maud.Leriche@aero.obs-mip.fr),
- 11 C. Barbet (C.Barbet@opgc.univ-bpclermont.fr).
- 12
- 13

14 Abstract

15 Gas and aerosol chemistry of 10 km -resolution mesoscale models Meso-NH and WRF/CHEM were evaluated on three 16 cases over Europe. These one-day duration cases were selected from Freney et al. (2011) and occurred on contrasted 17 meteorological conditions and at different seasons: a cyclonic circulation with a well-marked frontal zone on winter, an 18 anti-cyclonic situation with local storm precipitations on summer and a cold front in the northwest of Europe associated 19 to a convergence of air masses over eastern Europe and conflicting air masses over Spain and France on autumn. To 20 assess the performance of the two models, surface hourly databases from observation stations over Europe were used, 21 together with airborne measurements. For both models, the meteorological fields were in good agreement with the 22 measurements for the three days. Winds presented the largest normalized mean bias integrated over all European 23 stations for both models. Daily gas chemistry was reproduced with normalized mean biases between -14 and 11 %, a 24 level of accuracy that is acceptable for policy support. The two models performances were degraded during night-time quite likely due to the constant primary species emissions. The PM2.5 bulk mass concentration was overestimated by 25 26 Meso-NH over Europe and slightly underestimated by WRF/CHEM. The absence of wet deposition in the models partly 27 explains the local discrepancies with the observations. More locally, the systematic low mixing ratio of volatile organic 28 compounds in the gas phase simulated by WRF/CHEM at three stations was correlated with the underestimation of OM 29 (Organic Matter) mass in the aerosol phase. Moreover, this mass of OM was mainly composed of anthropogenic POA 30 (Primary Organic Aerosols) in WRF/CHEM, suggesting a missing source for SOA (Secondary Organic Aerosols) mass 31 in WRF/CHEM aerosol parameterization. The contribution of OM was well simulated by Meso-NH, with a higher 32 contribution for the summer case. For Meso-NH, SOA made the major contribution to the OM mass. The simulation of the mass of SO_4^{2-} in particles by both models was often overestimated and correlated with an underestimation of the 33 SO₂ mixing ratio. The simulated masses of NO₃⁻ and NH₄⁺ in particles were always higher for Meso-NH than for 34 35 WRF/CHEM, which was linked to a difference in NO_X mixing ratio between the models. Finally, computations of 36 model performance criterion and model performance goals show that both models can be considered acceptable for 37 standard modelling applications. In particular, Meso-NH model, using a gaseous chemical mechanism designed to 38 compute the organic precursors of aerosols, shows comparable simulated amounts of SOA to observations at local sites. 39

1 1 Introduction

Air pollutants have an effect on human health (Pope et al., 2004; Rueckerl et al., 2011), ecosystems and regional climate
(Monks et al., 2009). Greenhouse gases impact the climate primarily through shortwave and longwave radiation (Myhre
et al., 2013), while aerosols, in addition, affect the climate through cloud-aerosol interactions (O'Donnell et al., 2011;

5 Rap et al., 2013; Boucher et al., 2013).

Aerosols are composed of solid and liquid particles of varying chemical complexity, size, and phase. New particles are 6 7 added into the atmosphere by direct emissions and nucleation (secondary particles). Primary particles originate from 8 anthropogenic sources such as fossil fuel combustion and natural sources (fires, desert dust, sea salt, etc). Secondary 9 particles are formed through nucleation and condensation of the gas phase or by in-cloud processes (Ervens et al., 10 2011). Organic aerosols are a key issue for models as their formation processes and evolution are poorly known. 11 Depending on meteorological conditions and aerosol properties, aerosol particles act as cloud condensation nuclei with 12 the potential to impact the precipitation pattern (Poschl et al., 2005; Duseck et al., 2006; Rosenfeld et al., 2008). The 13 coupling between aerosols and cloud remains an important but poorly understood issue. Unlike well-mixed greenhouse 14 gases, short-lived aerosols exhibit a strong regionality in climate forcing and air quality impacts (Monks et al., 2009). 15 Chemistry-transport models (CTMs) are essential to capture the regional forcing and impacts of aerosols.

16 For the last two decades, numerical CTMs have experienced significant improvements thanks to: the increase in high 17 performance computing resources (Colette et al., 2014), the "online" coupling between meteorological and chemical 18 fields (Zhang et al., 2008; Zhang et al., 2013; Kukkonen et al., 2012; Baklanov et al., 2014) and the improved 19 knowledge of atmospheric processing. While the progress in simulating air quality are notable, many challenges remain 20 for CTMs. the AQMEII (Air Quality Model Evaluation International Initiative) is a joint effort between North America 21 and Europe to establish common methodologies for model evaluation with a focus on ozone and aerosols (Rao et al., 22 2011). Vautard et al. (2007) concluded that the majority of CTMs used for AQMEII captured the observed gas phase 23 mean values and daily variability fairly well, except for city centres. The skill of aerosol simulations, however, is generally lower. Nopmongcol et al. (2012) highlighted the role of emissions and dilution in the performance of their air 24 25 quality model. In their comprehensive evaluation of the on-line coupled CTM COSMO-ART, Knote et al. (2011) noted 26 that ozone and NO_x were well reproduced; $PM_{2.5}$ and PM_{10} were, on average, underestimated. Several processes needed 27 to be improved in the model, such as wet scavenging, SOA formation, distribution and concentrations in primary 28 emissions of aerosol particles. Several authors have also mentioned lateral boundary conditions for aerosols as a source 29 of uncertainties. Aksoyoglu et al. (2011) concluded that the offline-coupled CAMx (Comprehensive Air quality model 30 with extensions) model reproduced the relative composition of aerosols very well over Switzerland but underestimated 31 the absolute concentration by 20%. Tuccella et al. (2012) validated the online-coupled WRF/CHEM (Weather Research 32 and Forecast - Chemistry) model against ground-based measurements over Europe. The model reproduced daily PM2.5 33 aerosol mass with a slight negative bias but underpredicted particulate sulphate by a factor of 2 and overpredicted 34 ammonium and nitrate by about a factor of 2. Missing processes in the aqueous-phase could explain the differences 35 (Ervens et al., 2011). Zhang et al. (2013) compared the offline-coupled WRF/Polyphemus with the online-coupled model WRF/CHEM-MADRID over Western Europe. No model was shown to be superior in terms of aerosol 36 37 representation. Although the online WRF/CHEM-MADRID accounted for interactions between the meteorology and 38 the chemistry, the model comparison showed that the simulation of atmospheric pollutant was mainly sensitive to the 39 vertical structure, emissions and parameterizations for dry/wet depositions. Online biogenic emissions significantly 40 improved the simulated temporal variations and magnitudes for most variables and for both models. Meteorological 41 conditions also contribute to aerosol composition. Tulet et al. (2005) used the online-coupled Meso-NH (Mesoscale 42 Non-Hydrostatic model) model to simulate a coastal summer pollution episode during the ESCOMPTE ("Expérience

sur Site pour Contraindre les Modèles de Pollution Atmosphérique et de Transport d'Emissions") campaign over 1 2 southern Europe. Results showed good agreement between observed and simulated aerosol compounds. However, nitrate and ammonium were underestimated, probably due to an underestimation of relative humidity. Pollution levels 3 4 were also controlled by continental advection of aerosols. Aouizerats et al. (2011) used Meso-NH to simulate a two-day 5 period in the context of the CAPITOUL ("Canopy and Aerosol Particle Interactions in the Toulouse Urban Layer") field experiment in order to reproduce the spatial distribution of specific particle pollutants produced at regional and local 6 7 scale. Their simulation using three nested domains (10 km, 2.5 km and 500 m horizontal resolution) showed that urban 8 meteorology could locally affect the pollutant concentrations by up to a factor of 5. Bègue et al. (2012) studied the 9 evolution of dust optical properties during a major dust event, originating from northern Africa and advected over 10 northwestern Europe. The impact was found to be large over the Netherlands, with a maximum of aerosol optical 11 thickness close to 1.

This work fits in with the current effort to make a careful benchmarking of numerical CTMs against observational data and/or between models. Two online-coupled models, Meso-NH and WRF/CHEM, are evaluated over Europe during three one-day episodes here. The objective is to assess the capacity of the models to reproduce the magnitude of intraday gas and aerosol fluctuations over Europe for specific episodes. These episodes were chosen from the seminal study of Freney et al. (2011) and occurred during contrasted meteorological conditions over Europe. Models are compared with daily and hourly ground-based observations of meteorological data, gases and aerosols and to vertical profiles of meteorological data and gaseous pollutants.

A first section describes the Meso-NH and WRF/CHEM models (section 2). A second section concerns the measurement databases and the simulation set-up (section 3). The evaluation of the meteorology, gases and $PM_{2.5}$ aerosol at the regional scale is discussed in section 4. The last section is dedicated to model performance in terms of gases and aerosols at three particular chosen sites where detailed observations of aerosol chemical composition are available (section 5).

24

25 2 Model descriptions

Meso-NH (Lafore et al., 1998; Tulet et al., 2003) and WRF/CHEM (Grell et al., 2005) are two non-hydrostatic, and 26 27 "on-line" mesoscale atmospheric models (Baklanov et al., 2014). Meso-NH is anelastic and WRF/CHEM is fully 28 compressible. Both models simulate atmospheric phenomena with horizontal resolutions from a few metres (LES) to a 29 few kilometres (synoptic-scale). Meso-NH is developed by the Laboratoire d'Aérologie and Météo-France. In the 30 present study, version 4.9.3 of Meso-NH is implemented. It uses terrain-following z coordinates. WRF/CHEM is 31 developed among the community and the code is controlled by NOAA/ESRL (National Oceanic and Atmospheric 32 Administration/Earth System Research Laboratory) scientists. WRF/CHEM uses terrain-following Eta-coordinates and, 33 in this study, the version 3.4.1 is used.

Several parameterizations have been integrated in these models for convection, cloud microphysics, turbulence, surface processes, gas chemistry, and aerosol composition (<u>http://mesonh.aero.obs-mip.fr/mesonh410</u> and <u>www.wrf-model.org</u>).

36 The parameterizations used by the two models are reported in Table 1 and dynamical and physical parameterizations are

37 detailed in Appendix. A. Chemical options used here are detailed in the two next sections.

38 2.1 Gas-phase mechanism

The gas phase chemistry in Meso-NH was described by Suhre et al., (1998) and Tulet et al., (2003). The ReLACS2 scheme (Regional Lumped Atmospheric Chemical Scheme 2) used in this study is based upon a reduction of the original CACM mechanism (Caltech Atmospheric Chemistry Mechanism; Griffin et al., 2002; Griffin et al., 2005). ReLACS2 is derived from a reduction by reactivity weighting towards the hydroxyl radical OH developed by Crassier 1 et al., (2000). It includes 82 prognostic gaseous chemical species and 363 reactions enabling the formation of SOA

precursors to be addressed (Tulet et al., 2006), compared with 189 prognostic species and 361 reactions in CACM.
The gas phase chemical mechanism used in WRF/CHEM in the present study is RACM (Regional Atmospheric

The gas phase chemical mechanism used in WRF/CHEM in the present study is RACM (Regional Atmospheric
Chemistry Mechanism; Stockwell et al., 1997; Geinger et al., 2003). This mechanism includes 84 species and 252

5 reactions. It includes 16 aggregated anthropogenic species (alkanes, alkenes, toluene, xylene, cresol) and 3 aggregated

6 biogenic species (isoprene, monoterpenes and sesquiterpenes) representing VOCs.

Both ReLACS2 and RACM allow O₃ chemistry to be simulated for a range of conditions found in remote areas to polluted urban sites, from the surface to the upper troposphere. In addition, ReLACS2 was designed to simulate the formation of SOA precursors. For both models, the photolytic rates are calculated using the TUV (Tropospheric Ultraviolet and Visible; https://www2.acd.ucar.edu/modeling/tropospheric-ultraviolet-and-visible-tuv-radiation-model) radiation model. The version 5.0 (Madronich and Flocke; 1999) is used for Meso-NH and the version 4.1 (Madronich et al., 1998; McKenzie et al., 2007) for WRF/CHEM.

13 2.2 Aerosol scheme

14 Emissions, nucleation, coagulation, condensation, dry deposition, sedimentation, and advective and diffusive transport 15 are considered as processes modifying the aerosol population in both models. The evolutions of aerosol size distribution by aerosol microphysics (nucleation, coagulation, sedimentation and dry deposition) and of aerosol chemical 16 composition are predicted respectively by the ORILAM-SOA (Organic Inorganic Lognormal Aerosols Model; Tulet et 17 18 al., 2005; Tulet et al., 2006) model in Meso-NH and by the MADE-VBS (Modal Aerosol Dynamics Model for Europe; 19 Ackermann et al., 1998, Volatility Basis Set; Donahue et al., 2006) model in WRF/CHEM. Assuming that aerosols are 20 internally mixed (each size class corresponds to a single chemical composition), the particle size distribution of the 21 aerosol is represented by three overlapping lognormal modes (Aïtken, accumulation and coarse modes) in these two 22 schemes. 23 The gas to particle conversion for the inorganic species is handled by the EQSAM model (Equilibrium Simplified

Aerosols Model; Metzger et al., 2002) in Meso-NH and by MADE, a modified version of MARS (Binkowski and Shankar, 2005), in WRF/CHEM, both including sulphuric and nitric acid, ammonia and water vapour.

26 To simulate SOA formation, two different approaches are used in Meso-NH and WRF/CHEM. In Meso-NH, the 27 partitioning of the low volatility organic species between the gas and aerosol phases is based on the thermodynamic 28 equilibrium scheme MPMPO (Model to Predict the Multiphase Partitioning of Organics; Griffin et al., 2003; Griffin et 29 al., 2005). It considers the aerosol phase divided into a liquid organic phase and an aqueous phase. Hydrophobic 30 compounds condense on the organic liquid phase through Raoult's law and hydrophilic compounds condense mainly on 31 the aqueous phase according to the Henry's law amount of liquid water and the pH computed initially by EQSAM (Pun 32 et al., 2002). Finally, 10 groups of SOA (8 from anthropogenic sources and 2 from biogenic sources) are considered in 33 the MPMPO model.

34 To simulate SOA formation in WRF/CHEM, a new parameterization is used: the VBS set (Volatility Basis Set; Donahue 35 et al., 2006), recently coupled by Ahmadov et al. (2012) to the aerosol module MADE and to the gas phase chemical mechanism RACM in the WRF/CHEM model. In this parameterization, the oxidation of the VOCs by the hydroxyl 36 37 radicals OH, O_3 or nitrate radicals leads to the production of organic species, which partition between the gas and the 38 particle phases according to their volatility. The volatility spectrum is divided into four bins with saturation vapour concentrations ranging from 1 to 1000 µg m⁻³. For each bin and each SOA precursor, an SOA yield based on smog 39 40 chamber studies is defined. Yields are different for two regimes, high and low NO_x conditions. Moreover, the 41 photochemical aging of the first generation VOC oxidation products is taken into account by shifting mass from high

42 volatility bins to lower ones when further oxidation processes occur.

1 2.3 Dry deposition

In Meso-NH, the dry deposition of chemical species is treated according to the resistance concept of Wesely (1989) in 2 3 the SURFEX model, which treats all surface processes (Masson et al., 2013). The deposition velocity is computed by determining the net flux of the species and its concentration at the surface. It depends strongly on surface conditions 4 5 from the ISBA (Interaction Sol-Biosphère-Atmosphère) scheme for nature surface (Noilhan and Planton, 1989), from the TEB (Town Energy Budget) scheme for town surface (Masson, 2000) and from the sea surface scheme based on the 6 7 Charnock (1955) formula. The deposition depends on the turbulence in the boundary layer and on the molecular 8 diffusion, which lead gases to effective surface deposition. The surface resistance is based on solubility and equilibrium 9 with vegetation-specific processes such as stomatal activities of leaves for nature surface. The mesophyll resistance and 10 the external leaf uptake are considered as a sink for some soluble gases (inorganic and organic) and are determined by their effective Henry's law constants. Dry deposition of aerosol particles is also based upon Wesely's (1989) 11 12 parameterization according to the surface type. 13 In WRF/CHEM, the dry deposition of gas phase species is parameterized according to Erisman et al. (1994) also based 14 on Wesely's (1989) scheme. The deposition velocity is proportional to the sum of the aerodynamic resistance between a 15 specified height and the surface, the sub-layer resistance (dependant upon the molecular diffusion) and the surface 16 resistance. The impact of the nature, the town and the sea surfaces on the exchanges between the soil, the surface and 17 the atmosphere is taken into account by the unified Noah Land Surface Model (Chen and Dudhia, 2001), which

considers 24 land use categories (including urban and built-up lands, water bodies, snow and ice) and 16 soil categories according to the United States Geological Survey (USGS). However unlike in Meso-NH, the mesophyll resistances for all gases are assumed to be zero, thus increasing the dry deposition velocity. For organic species, the dry deposition velocity is parameterized as proportional to the dry deposition velocity of nitric acid. Since no data are available to constrain the model, the parameter that determines the fraction of nitric acid is set to 0.25 as defined by Ahmadov et al.

23 (2012). For aerosol particles, the Wesely approach is used (Wesely and Hicks, 2000).

- 24 In both models, no wet deposition scheme is activated.
- 25

26 3 Observation databases and simulation set-up

27 3.1 Databases

Surface meteorological fields were compared to the ISD (Integrated Surface Database) database of NOAA's NCDC (National Climatic Data Center) (<u>http://cdo.ncdc.noaa.gov/pls/plclimprod/cdomain.abbrev2id</u>), which provides public access to hourly averages of meteorological surface parameters over Europe. Wind direction (WD), wind speed (WS), temperature and dewpoint were used for this study. Only stations below 700 m elevation were selected for the model evaluation in order to avoid bias due to terrain smoothing by the models (Jimenez and Dudhia, 2012).

33 The MOZAIC (Measurements of OZone, water vapour, carbon monoxide and nitrogen oxides by In-service Airbus 34 airCraft) database provides measurements of meteorological and chemical fields made during a large number of 35 commercial flights all over the world since 1993 (http://www.iagos.fr/web/; Marenco et al., 1998). These 20 years of 36 measurements enable climatology and air quality studies to be made for the troposphere. The database includes data collected from the ground to the upper troposphere during take-offs and landings, with high temporal and 3D-spatial 37 38 resolution. These vertical profiles are used here to evaluate meteorology and chemistry in the vertical direction. During 39 landings and take-offs, data is collected every 4s, which corresponds approximately to 50-100 m in the vertical direction 40 (Solazzo et al., 2013).

Surface chemical measurements were provided by AIRBASE (European AIR quality dataBASE) 1 2 (http://www.eea.europa.eu/data-and-maps/data/airbase-the-european-air-quality-database-8). AIRBASE provides hourly mean concentrations of O_3 , SO_2 (sulphur dioxide), NO_x (nitrogen dioxide) and $PM_{2.5}$ bulk mass concentration for a large 3 4 number of European stations, each characterized by its own kind of area (urban, rural, suburban) and its type of station 5 (background, traffic, industrial). Only a subset of stations is used for the model evaluation, with the following criteria:

stations must be below 700 m elevation (Nopmongcol et al., 2012) and must be classified as background rural stations. 6

The study excludes urban or suburban stations and stations for which pollution is determined predominantly by traffic 8 or industrial sources. Because of the low horizontal resolution and the static emission profiles used by the two models,

9 local anthropogenic (point or mobile) sources, for which time-varying daily profiles are crucial, are not represented in

10 the models.

7

- 11 The chemical composition of aerosol particles at ground level is compared with the AMS (Aerodyne Aerosol Mass 12 Spectrometer) global database (https://sites.google.com/site/amsglobaldatabase/). This database provides hourly 13 measurements of submicron non-refractory aerosol mass concentrations (NR-PM₁) for sulphate ($SO_4^{2^{-}}$), ammonium (NH4⁺), nitrate (NO3⁻) and OM (Organic Matter) taken throughout Europe and broken down into several kinds of 14 15 environment (urban, urban downwind, rural/remote, aircraft). AMS measurements were available at 3 stations 16 distributed over Europe for the three simulated case studies: the puy de Dôme station, the Melpitz station and the K-17 Puszta station.
- 18 The puy de Dôme (pdD) station is located in the centre of France (45°77N; 2°96E) at 1465 m a.s.l. This site is 19 surrounded by agricultural land and forest. As discussed in Sect. 4.1, air masses characterized by specific chemical 20 compositions reach the pdD station according to the season and the origin of the air masses (Freney et al., 2011).
- Melpitz (51°54 N; 12°93 E, 86 m a.s.l) is a German station located 50 km east of Leipzig. This site is representative of a 21
- 22 larger rural area in Saxony with almost no local anthropogenic sources. Two main wind directions are observed at 23 Melpitz station. When the wind blows from the southwest, the air mass reaching Melpitz is mostly a modified maritime
- 24 air mass and is less polluted than when the wind direction is from the east, bringing Melpitz a dry continental air mass composed of anthropogenic pollutants (Spindler et al., 2010). 25
- 26 K-Puszta (46°96N; 19°58E, 136 m a.s.l) station is located in Hungary; 80 km southeast of Budapest; in a rural, 27 continental environment. This station is surrounded by forest and is a good site for studying the formation of SOA from 28 biogenic VOCs (BVOCs). Maenhaut et al., (2007) determined the chemical aerosol characteristics for the period 24 29 May-29 June 2006, period in which two different flows occurred (an oceanic air mass from the north west and a static
- 30 continental air mass). Results showed that the mass concentration of aerosols was higher for the warm period than for
- 31 the cold one and that OM made the greatest contribution by far to the $PM_{2.5}$ and PM_{10} mass concentration.

32 3.2 Simulation set-up

- 33 An intensive campaign (Freney et al., 2011) took place at the pdD station (45°77N, 2°96E) in autumn 2008, winter 2009 34 and summer 2010, and documented chemical and microphysical properties of atmospheric aerosol particles. From these 35 measurements, Freney et al., (2011) showed, firstly, that for each season, the aerosol was characterized by its own 36 chemical composition and, secondly, that the origin of the air mass influenced the mass concentration of aerosols. For 37 this reason, it was interesting to simulate three cases extracted from this database, which cover three situations, one for
- 38 3 March 2009, one for 26 June 2010 and one for 18 September 2008. The air masses reaching the pdD station for these
- 39 days are representative of an oceanic, a continental and a Mediterranean environment respectively.
- 40 The computational domain extended over Europe. It has 360 x 360 grid points with a horizontal resolution of 10 km.
- 41 The vertical grid was common to both models and had 50 levels, stretched to allow better definition in the boundary

1 layer. The first level was at 30 m and the last one was at about 19 km. Each simulation lasted for 96 h, of which 72 h

2 concerned spin-up. The time step was 50 s.

3 The initial and boundary conditions for meteorology were provided by the ECMWF (European Centre for Medium-

4 Range Weather Forecasts) analysis with a horizontal resolution of 0.5° and were forced every 6 h. The initial and

5 boundary conditions for gaseous chemical species including inorganic nitrogen species, CO, SO₂, NH₃, NMVOCs

6 (Non-methane VOCs), primary (BC, POA) and secondary (inorganics, SOA) aerosol species were taken from

7 MOZART-4 (Model for Ozone and Related chemical Tracers, version 4) (Emmons et al., 2010) driven by meteorology

8 from NCEP centres (National Centers for Environment Prediction). Boundary chemical fields were forced every 6h.

9 For Meso-NH, surface emissions of atmospheric compounds were taken from the ECCAD database (Emissions of atmospheric Compounds & Compilation of Ancillary Data, <u>http://eccad.sedoo.fr/eccad_extract_interface/JSF/</u> <u>page_login_jsf</u>), which compiles the main emissions datasets available for the community. The MACCity emissions dataset (MACC/CityZEN EU projects) (Van der Werf et al., 2006; Lamarque et al., 2010; Granier et al. 2011; Diehl et

13 al. 2012) provided "off-line" monthly anthropogenic emissions for CO, NO_x, SO₂, NH₃, NMVOCs such as alkanes,

14 alkenes, alcohols, aldehydes, ketones and aromatics, lumped into 21 species, and for primary aerosol species, with a

15 grid resolution of 0.5°. The MEGANv2 model (Model of Emissions of Gases and Aerosols from Nature, version 2)

16 (Guenther et al., 2006) gave "off-line" monthly net emission of gases and aerosols from vegetation into the atmosphere

17 at 0.5° resolution (for NOx and VOCs). The monthly GFED3 (Global Fire Emissions Database, version 3; Van der Werf

et al., 2010) database was used to represent biomass burning emissions (CO, NMVOCs, BC and POA) with a spatial
 resolution of 0.5°.

Emissions in WRF/CHEM also came from MACCity for anthropogenic species. Biogenic emissions were different: the MEGAN model was used in "on-line" configuration because this option was already available in WRF/CHEM. In MEGAN, emissions were estimated using a 1km x 1 km resolution map of vegetation defined for the year 2003. For biomass burning emissions, the daily FINN inventory (Fire INventory from NCAR) (Wiedinmyer et al., 2011) model

24 was used in WRF/CHEM with a 1 km^2 spatial resolution.

For Meso-NH and WRF/CHEM, NO_x emissions were assumed to be 70% NO and 30% NO₂. In the chemical 25 26 mechanisms CACM and RACM, in order to limit the number of model species, some similar organic compounds were 27 grouped together based on the principle of reactivity weighting. Aggregation factors, computed by Middleton et al. 28 (1990), were used for VOCs. For aerosol species, in WRF/CHEM, a conversion factor of 1.6 proposed by Bessagnet et 29 al. (2008) was used to convert the emissions of organic carbon (OC) into particulate organic matter (POA). Then, 20% 30 of BC and POA emissions were allocated to the Aïtken mode of the aerosol distribution and 80% to the accumulation 31 mode according to Ackermann et al. (1998). In Meso-NH, 5% of OC and BC were allocated to the Aitken mode and 32 95% to the accumulation mode, as, according to the actual knowledge, organic aerosol mass is assumed to be 33 distributed mainly in the accumulation mode (Kanakidou et al., 2005). For both models, parameters for the three lognormal modes (total particle number concentrations, mean radius and standard deviation) are initialized from Asmi 34 35 et al. (2011).

36

37 4 Evaluation of models at regional scale

38 4.1 Three contrasted meteorological situations over Europe

In general, both models showed similar meteorological patterns (Fig. 1). On 3 March 2009, a cyclonic circulation with a well-marked frontal zone was simulated west of 0°E associated with heavy precipitation behind the cold front (10-20 mm over United Kingdom). Clear-sky conditions and strong southerly winds developed ahead of the cold front with

wind speeds exceeding 20 m.s⁻¹ over France. A second south-north rain band was simulated with weaker precipitation 1 except over eastern Spain (up to 30 mm). On 26 June 2010, an anticyclone (55°N; 10°E) prevents precipitation over 2 northern Europe with strong northerly flow over northeastern Europe. Continental southern Europe experienced local 3 4 storm precipitations (up to 20 mm) over mountain ranges. A second anticyclone centred on Russia induced a southerly 5 flow over the extreme east of Europe and a second rain band with strong precipitation (up to 30 mm). On 18 September 2008, three precipitation zones were simulated. The first was associated with a cold front in the northwest of the 6 7 domain. The second was induced by convergent air masses over eastern Europe, and the third resulted from conflicting 8 air masses over Spain and France. Southern France, northern Spain and northern Italy experienced strong precipitation. 9 Both models simulated the contrasting meteorology and the precipitation over Europe for the three episodes described

10 in Freney et al. (2011).

11 4.2 Surface Meteorological fields

12 NOAA surface data, coming from European meteorological centres, were compared with results from both models.

For each selected surface meteorological station, the daily bias between observations and the two models was computed for WD10 (10m-Horizontal Wind Direction) (°/N), WS10 (10m-Horizontal Wind Speed) (m.s⁻¹), T2m (Air temperature at 2 m) (K) and Td2m (dewpoint at 2 m) (K) for the three chosen dates of 18 September 2008, 3 March 2009 and 26

16 June 2010 (Fig. 2 for 18 September 2008; Fig. S1 and Fig. S2 in the supplement for the other two days).

The corresponding daily NMB (Normalized Mean Bias), expressed as a percentage, is reported in Table 2 and definedas:

$$NMB = 100 \frac{\sum_{i=1}^{N} (Mod_i - Obs_i)}{\sum_{i=1}^{N} Obs_i}$$
(1)

20 Results in Table 2 show that there were no significant differences between the models for the simulated WD10, as shown by the weak NMB (-3 to 1%). For 3 March 2009, the well-marked frontal zone (Fig. 1) was correlated with a 21 22 zone of weak biases for WD10 (between -20°/N and 20°/N) from northern France and the United Kingdom to Scandinavia (Fig. S1). The NMB of WS10 for Meso-NH was low (-2 to 2%) while, for WRF/CHEM, it showed an 23 24 overestimation (15 to 20%) (Table 2). This behaviour is visible on Fig. 2, Fig. S1 and Fig. S2 showing a higher bias on 25 WS10 for WRF/CHEM than for Meso-NH. In general, except for coastal zone, the Meso-NH bias for WS10 is about +/-2 m.s⁻¹, which is consistent with Aouizerats et al. (2011). For WRF/CHEM, previous studies had already pointed out a 26 27 systematic overestimation of surface wind speed (Zhang et al., 2010; Tuccella et al., 2012; Wyszogrodzki et al., 2013). The NMB of T2m and Td2m was close to 0% for both models (Table 2). The large underestimation of T2m for 28 29 WRF/CHEM for the 26 June 2010 (Fig. S2) is consistent with results from Tuccella et al. (2012) showing a cold bias

30 during the spring-summer period over Europe with WRF/CHEM.

31 Determining the origin of these biases in the two models would require an extensive study of the current uncertainties in 32 surface-atmosphere coupling, which is beyond the scope of this paper. Simulated near-surface variables depend on 33 surface conditions, including soil moisture and temperature (Sutton et al. 2006), land surface characteristics (land use, 34 land cover, vegetation), and the coupling between these surface parameters within the land-surface model (LSM) and 35 boundary layer parameterizations (Liu et al. 2006; Trier et al., 2008; Misenis et al., 2010; Noilhan et al., 2011). The parameterization of cloud microphysics and radiation may represent additional sources of biases for temperature. In a 36 37 recent study, Zhang et al. (2013a) emphasized the role of the radiative impact of aerosols on shortwave radiation; they 38 improved the simulation of surface temperature in WRF/CHEM. A recent study from Jimenez and Dudhia (2012) 39 proposed a new parameterization to account for a subgrid-scale orography effect on surface winds for models at low 40 resolution.

1 4.3 Vertical distribution of meteorological variables

2 The simulated vertical distribution of meteorological variables was compared with the high resolution vertical profiles

3 of T (temperature), Td (dewpoint), WD (wind direction) and WS (wind speed) measured during take-off and landing of

- 4 MOZAIC-IAGOS aircraft above Frankfurt (Fig. 3 for 18 September 2008 and Fig. S3 for 3 March 2009). There was no
- 5 MOZAIC flight from Frankfurt on 26 June 2010. As Meso-NH and WRF/CHEM grids are horizontally and vertically
- 6 close, the comparison between models was not spatially biased.

7 Fig. 3 shows that both models capture the vertical distribution of T and Td between the surface and 300 hPa on 18 8 September 2008. Below 850 hPa, both models tend to overestimate T and Td. Models simulate stratiform clouds 9 between 500 hPa and 300 hPa as observed during flights where T was equal to Td. In the planetary boundary layer for 10 18 September 2008, there is a positive bias of T but the observed temperature inversion is well reproduced by the 11 models early in the morning and in the evening. Measurements show a sudden increase in T at 850 hPa, which is less 12 visible for both models. Observed Td decreases strongly at 850 hPa early in the morning, meaning a dry air mass limit. 13 On 3 March 2009 (Fig. S3), the temperature vertical profile is well reproduced except at the surface where there is an 14 underestimation at 10:00 UTC. Larger discrepancies are obtained with both models for Td. Fig. 1 shows that Frankfurt 15 was at the edge of a rain band on 3 March 2009, which could explain the difference between local measurements and 16 low-resolution model outputs. The models reproduce the northeasterly-easterly flow at the surface on 18 September 17 2008 and the southwesterly flow on 3 March 2009.

18 Overall, these four meteorological parameters were relatively well represented over Europe by Meso-NH and by

WRF/CHEM. This constitutes a good basis for the following part of the study, concerning reactive gases and aerosolparticles over Europe.

21 **4.4 Gaseous pollutants at the surface**

Simulated results for O_3 , NO_x and SO_2 are compared with the AIRBASE surface data. The number of rural and background stations for statistics varied between 111 and 259. As emissions have no diurnal variation, the bias of O_3 , NO_x and SO₂ was calculated for daytime only, between 08:00 and 16:00 UTC (Fig. 4, Fig. S4 and Fig. S5 for 18 September 2008, 3 March 2009 and 26 June 2010 respectively, and Table 2). Fig. 5 shows the diurnal cycle of the median bias for these three chemical species.

27 Meso-NH tended to overestimate O₃ for the three days (Fig. 4, Fig. 5, Fig. S4 and Fig. S5) with a highest NMB (+11%)

28 on 18 September 2008 (Table 2). Bias of NO_x was slightly negative (-3% <NMB< -1%, Table 2). Looking at the diurnal

29 cycle of the bias for NO_x, Meso-NH showed poorer performance during night-time, with mean differences up to 6 µg.m⁻

 30^{-3} for 50% of stations (Fig. 5). During daytime, the largest underestimations for NO_x were found in industrial areas of the

31 Po Valley (Italy), the Ruhr, the Netherlands and Austria, especially on 18 September 2008 (Fig. 4). In the Po Valley, for

32 the three study cases (Fig. 4, Fig. S4 and Fig. S5) a positive bias of up to $20 \ \mu g.m^{-3}$ of O₃ was associated with a strong

33 negative bias of NO_x (-5 µg.m⁻³ for 50% of stations).

- WRF/CHEM underestimated O_3 on 3 March 2009 and 26 June 2010 (-14% <NMB< -3%, Table 2) but gave good predictions of ozone during the third episode (NMB =1%, Table 2). Daytime NO_x concentrations were reproduced to within -1 to 2% (Table 2), the worst prediction being made during night-time, with a systematically strong overestimation of up to 10 µg.m⁻³ for 50% of stations (Fig. 5). Fig. 4, Fig. S4 and Fig S5 show patterns of maximum biases for ozone and NO_x over the Po Valley and northern Europe for the WRF/CHEM model that are similar to those
- 39 of Meso-NH.

Daytime values of NMB ranged between -2% and 0% for SO₂ in both models (Table 2). As shown in Fig. 5, 1

WRF/CHEM tends to underestimate SO₂ for the three cases, with maximum bias during daytime (up to -1.5 μ g.m⁻³ for 2

3 50% of stations).

4 Except in the Po Valley, during daytime, the level of accuracy of both models for the prediction of ozone, NO_x and SO_2 5 at the surface is acceptable for regulatory application (AQD, 2008). The maximum bias for both models over the Po Valley is suspected to be due to the absence of diurnal variation of anthropogenic emissions (underestimation for NO_x 6 7 and overestimation for ozone). In the same way, both models showed a positive bias for NOx during night-time due to 8 this absence of diurnal variation. This effect was also visible on the diurnal cycle of SO₂ bias for both models. However, 9 for ozone bias, the link with the diurnal cycle of anthropogenic emissions is indirect due to the non-linearity of ozone 10 chemistry. For instance, Tao et al. (2004) changed the temporal emission profiles (uniform vs. time-varying) of 11 anthropogenic species and found no impact on daytime ozone concentrations. In the same idea, Menut et al. (2012) 12 increased the NO₂ concentrations by 10-20% in their model using new hourly fluxes representative of traffic emissions 13 with a moderate impact on ozone (0-7% decrease) and the most important changes occurred during night-time. Among 14 potential sources of errors in the models, the chemical lateral boundary conditions are essential to capture the variability 15 of pollution inflow and provide background concentrations, to which local production is added (Tang et al., 2007, 16 Pfister et al., 2011). Global models provide chemical top and lateral boundary conditions for regional models but 17 introduce uncertainties that are inherent in the model or due to the coupling process. This point is discussed in the next 18 part. Obviously, differences between the two models could be related to differences in chemical reaction schemes 19 coupling to differences between thermodynamical modules for gas to particles conversion, dynamics of the planetary 20 boundary layer (venting of pollutants) and dry deposition parameterizations.

21 4.5 Vertical distribution of ozone and carbon monoxide

22 The simulated vertical distributions of O₃ and CO were assessed using measurements from the MOZAIC database (Fig. 6 for 18 September 2008 and Fig. S6 for 3 March 2009; no data for 26 June 2010).

23

24 CO is a significant trace gas in the troposphere, which strongly influences the concentrations of oxidants such as 25 hydroxyl radical (OH.) and O₃. CO is mainly emitted by anthropogenic activities and has a lifetime of a few months in 26 the boundary layer (BL) and the free troposphere (FT) (Seinfeld and Pandis, 1998).

27 Above 9-10 km altitude, O_3 and CO mixing ratios and their daily variability are controlled by stratosphere-troposphere exchanges, venting by convective clouds, chemical reactions and long-range transport. The vertical profiles in Fig. 6 28 29 show that both models fail to reproduce a well-marked stratospheric dry ozone-enriched intrusion. Both models 30 underestimate O₃ with negative biases of up to 35 ppbv on 18 September 2008 (Fig. 6 at 0700 UTC) and 200 ppbv on 3 31 March 2009 (Fig. S6). Stratospheric air corresponds to poor CO mixing ratios. CO is overestimated at these altitudes with a large positive bias of 50 ppbv on 18 September 2008 (Fig. 6 at 0700 and 1000 UTC) and 3 March 2009 (Fig. S6 32

- 33 at 0400 UTC).
- 34 Below 500 m, the two models reproduced CO mixing ratios poorly, with large biases between -90 and 90 ppby. These 35 large discrepancies can possibly be attributed to two principal processes:
- 36 The influence of local sources near the airport, the intensity and temporal variations of which are not captured by the models. 37
- 38 The BL dynamics and the associated vertical venting of the pollutants. Interestingly, the worse biases are _ 39 obtained during night-time, when the surface BL is expected to be low, favouring the accumulation of 40 pollutants. The models performed better during daytime (1000 UTC on Fig. 6 and Fig. S6) as the depth of the 41 BL increased.

1 Solazzo et al. (2013), which studied the performance of regional-scale air quality models in reproducing the vertical

2 distribution of pollutants over the North American and European continents for 2006, also found such strong biases for

- 3 CO in the BL.
- 4 O₃ in the BL was generally well simulated with biases between -10 and 10 ppbv. The best scores were obtained for 3
 5 March 2009 (Fig. S6).

Above the BL and below 9 km of altitude, biases for O_3 and CO were variable depending on the altitude and on the time 6 of day. Both models exhibited similar vertical distributions of the biases. The vertical distribution of biases between O₃ 7 8 and CO observations and the MOZART model interpolated on the Meso-NH vertical and horizontal grid at 0600 and 9 1200 UTC (Fig. 6) clearly shows that the vertical profiles of CO and O₃ for both models are strongly influenced by the 10 MOZART fields. The difference between CO biases for MOZART and the two models is the smallest (< 10 ppbv) in the entire vertical column above the BL. Below 500 m, simulated CO departs from the MOZART fields. The largest 11 12 differences between the regional models and MOZART are found below 500 m for CO. Below 500 m, the MOZART 13 model gives a strong overestimation of CO compared to the measurements (30-100 ppbv). The boundary conditions 14 impact the simulated surface concentrations of CO but the biases are reduced in the regional models in comparison with 15 MOZART bias because of local meteorology and chemical processes. The difference of CO bias between Meso-NH and 16 WRF/CHEM at the surface and up to 3 km is certainly due to the difference in WD and in WS up to this altitude (Fig.

17 3).

18 The vertical profiles of O_3 biases for the two models also show the impact of MOZART fields. The bias of ozone 19 MOZART fields is strongest in the upper troposphere and in the BL; this bias is reduced for Meso-NH and for 20 WRF/CHEM. The difference between ozone biases for MOZART and both models is the weakest (5 ppby) in the 21 middle troposphere (2500-5000 m). This difference is stronger (20 ppbv) below 1000-1500 m. As for CO, the biases are 22 reduced in the regional models in comparison with MOZART bias because of local meteorology and chemical 23 processes. This is consistent with Solazzo et al., (2013) who found that model performance for ozone in the BL was 24 generally good and mainly influenced by lateral boundary conditions and surface processes such as emissions, transport and photochemistry. Appel et al. (2012) underlined the influence of the forcing model for surface ozone over North 25 26 America by improving the CMAQ model simulations using the global GEOS-Chem model instead of GEMS (Global and regional Earth-system Monitoring using Satellite and in-situ data) data (Schere et al., 2012) for the boundary 27 28 conditions However, if the surface ozone is mainly influenced by local photochemistry instead of the pollution inflow, 29 the surface ozone mixing ratio can have low sensitivity large-scale chemical conditions (Szopa et al., 2009). For vertical 30 profiles, Tang et al. (2007) found that the mean O3 vertical profile below 3 km was insensitive to changes in chemical 31 boundary conditions, whereas CO concentrations were sensitive throughout the troposphere.

32 4.6 PM_{2.5} bulk aerosol at the surface

Simulated daily mean $PM_{2.5}$ bulk aerosol mass concentrations over Europe are compared with AIRBASE stations measurements for the three chosen dates in Fig. 7. For both models, $PM_{2.5}$ mass concentration was computed by integrating Aitken and accumulation lognormal modes up to 2.5 micrometres in diameter. It included primary aerosol mass (BC and POA for organic), secondary inorganic (NO₃⁻, SO₄²⁻ and NH₄⁺) and organic (SOA) aerosol components.

- 37 The corresponding daily NMB (in %) was computed (Table 2) but it should be borne in mind that $PM_{2.5}$ measurements
- 38 were available for only a small number of rural and background stations (6 and 12 respectively).
- 39 Table 2 and Fig. 7 indicate that Meso-NH overestimates the $PM_{2.5}$ mass concentration with positive NMB between 6
- 40 and 14% for the three dates. WRF/CHEM slightly underestimates the $PM_{2.5}$ mass concentration with a NMB between -3
- 41 and 0% (Fig. 7 and Table 2). The same computation considering all stations (urban and suburban stations, traffic and

- industrial sources), which represented more than 50 stations of AIRBASE (not shown here), changed the results 1
- 2 considerably, with a large underestimation for WRF/CHEM (NMB between -17% and -1%) and a weaker positive
- NMB for Meso-NH (-5% to 8%). 3
- The high observed values for the Po Valley in Italy (range from 20 to 40 µg.m⁻³) were well captured by both models for 4
- the three cases despite an overestimation (ranging from 30 to 60 µg.m⁻³) by Meso-NH and an underestimation (ranging 5
- from 5 to 25 μ g.m⁻³) by WRF/CHEM (Fig. 7). Over stations in the Netherlands, PM_{2.5} mass concentrations were 6
- relatively low. This was well reproduced by WRF/CHEM whereas Meso-NH overestimated the observations by 15 to 7
- 20 µg.m⁻³ (Fig. 7). 8
- 9 Several recent studies have shown that air quality models tend to underestimate PM_{2.5} mass concentrations over Europe.
- 10 Appel et al. (2012) studied the performance of the CMAQ model over Europe and North America for the entire year of
- 2006. They found a global underestimation of the PM_{2.5} mass concentration, mostly in winter (NMB=-55% and MB=-11
- 12.9 µg.m⁻³), with better performance over France, Spain and Portugal. However, in their study, all the stations of the 12
- 13 AIRBASE database were taken into account, which certainly translated, for WRF/CHEM, into a lack of anthropogenic
- 14 sources or a too strong dry deposition. Tagaris et al., (2013) found an underestimation of PM_{2.5} mass concentration over
- 15 most European countries during winter and summer 2006. Their paper discusses the chemical aerosol speciation, which
- 16 indicates an insufficient contribution of OC to PM2.5 mass concentration in their model. By simulating gas and aerosol
- 17 with WRF/CHEM over Europe, Tuccella et al (2012) showed an underestimation with a mean bias of PM_{2.5} mass (-
- 18 7.3%) linked to a strong underestimation of the carbonaceous fraction by RADM2 mechanism.
- 19 The Meso-NH model overestimated the observed PM_{2.5} mass concentration. This could be due to a missing sink, to an 20 underestimation of dry deposition, or to an overestimation of aerosol sources, as well as a too low BL height. The
- 21 following section provides a detailed analysis of the chemical composition of the simulated aerosols and discusses the 22 aerosol sources.
- 23

5 24

Evaluation and discussion of sites

25 Both models present significant biases on daily PM_{2.5} mass concentration whatever the situation. To explain these 26 biases, it is necessary to look at the detail of the aerosol chemical composition by comparison between simulated and 27 observed hourly mean mass concentration for each aerosol constituent. AMS measurements are available at three local 28 sites: puy de Dôme, Melpitz and K-Puszta for the three selected dates. In this section, the local mixing ratios of aerosol 29 gaseous precursors are studied before analysing the aerosol hourly mean mass composition for the three local sites. For 30 the pdD station, in order to avoid errors due to topography smoothing, all simulated results are taken at the real altitude 31 of the station: 1465 meters.

5.1 32 Gas precursors of aerosols

33 Fig. 8 suggests that simulated NH_3 , NO_x and VOC mixing ratios have a behaviour partly linked to the season with 34 higher values for 3 March 2009 and lower values for 26 June 2010 during daytime at the surface stations (Melpitz and 35 K-Puszta). At these stations, VOC, NH₃, NO_x and SO₂ show diurnal variations with higher mixing ratios at night and 36 minimum values during the day, although this is visible to a lesser extent for 3 March 2009. These diurnal evolutions 37 are due to the combination of the smallest BL height during the night and the constant temporal profiles of 38 anthropogenic emissions in both models. The chemical species emitted accumulate during the night, when the BL 39 height is small and no photochemistry takes place. The association of rain and cold temperatures simulated for 3 March 40 2009 limits photochemistry, which partly explains the contrast with the two summer cases where the weather was 41 warmer (Fig. 1). Another reason could be connected with the fact that the BL height is smaller in winter than in summer,

implying a weaker contrast between night-time and daytime mixing ratios. For K-Puszta station, some data are available 1 2 on the EMEP database (Tørseth et al., 2012) for nitrogen dioxide, ammonia and sulphur dioxide as measurements on filter packs corresponding to 24 hours averaged values between 07:00 on the day of interested to 07:00 on the next day. 3 4 However, because of the error already discussed on nigh-time simulated mixing ratios, the following discussion focuses 5 on simulated daytime values. For 3 March 2009, observed ammonia mixing ratio was 2.5 ppbv showing an overestimation by both models, and especially by WRF/CHEM. However, both models simulated mixing ratio for 6 7 ammonia close to observations for 18 September 2008 and for 26 June 2010 (observed mixing ratio: 2.5 ppbv and 1.8 8 ppbv respectively). For NO_x , the comparison was done with NO_2 as no NO observations are available but, as NO_2 9 dominates NO_x mixing ratio, mixing ratio of NO₂ can be seen as a lower limit for NO_x. Except for WRF/CHEM for the 10 3 March 2009, simulated NO_x mixing ratio for both models are of the same order of magnitude than observed NO_2 mixing ratio (observed mixing ratios: 5.4 ppbv for 3 March 2009, 2.6 ppbv for 26 June 2010 and 3.5 ppbv for 18 11 12 September 2008). For SO₂, Meso-NH overestimated the observed values for the three episodes whereas WRF/CHEM 13 tended to slightly underestimate it (observed mixing ratios: 0.8 ppbv for 3 March 2009, 0.7 ppbv for 26 June 2010 and

14 0.25 ppbv for 18 September 2008).

15 At the mountainous pdD station, night-time observations, with mixing ratios lower than or similar to those found during 16 the day, are representative of FT air masses (Freney et al., 2011). During the day, depending on the season and the 17 meteorological conditions, the station lies in the FT (winter cases) or in the BL (summer) or in a transition area between 18 BL and FT (spring and autumn) (Freney et al., 2011). Results for the three cases show the same trend, with the station 19 probably located in a transition area between BL and FT during the day for both summer cases, which show no diurnal 20 cycle. Some EMEP data are available for pdD on the EMEP database for sulphur dioxide as hourly observations. As 21 these data are few dispersed, they were not added on Fig. 8 for clarity. For 3 March 2009, the observed mixing ratio was 22 0.11 ± 0.03 ppby; for 26 June 2010, it was 0.12 ± 0.04 ppby and for 18 September 2008, it was 0.17 ± 0.06 ppby. In 23 comparison to these values, Meso-NH overestimated observed SO2 mixing ratio whereas WRF/CHEM slightly 24 underestimated it. This trend is similar to K-Puszta.

The NH₃ mixing ratio typically ranges between 0.1 and 10 ppb over continents (Seinfeld and Pandis, 1998). NH₃ 25 26 mixing ratios higher than 10 ppbv are reported close to urban centres, or in recent plumes impacted by burning biomass 27 (Yokelson et al., 2003) or an industrial accident (Nowak et al., 2010). The main sinks of NH₃ are dry and wet deposition 28 and its transfer in the particulate phase as NH₄⁺. NH₃ simulated by Meso-NH at the Melpitz station reached 55 ppbv at 29 0700 UTC on 18 September 2008 and WRF/CHEM simulated more than 30 ppbv for 3 March 2009 at 0800 and 1800 30 UTC (Fig. 8). These high mixing ratios of simulated NH₃ can probably be attributed to the combination of high 31 emission rates and the absence of a sink by precipitation, as both models simulate precipitation for 3 March 2009 and 32 for 18 September 2008.

33 The largest differences in NH₃ mixing ratio between the two models are simulated on 3 March 2009 at all stations, with 34 higher values simulated by WRF/CHEM. Simulated NO_x mixing ratios are close except for 3 March 2009, when values 35 simulated by WRF/CHEM are higher, as already shown on Fig. 5. VOCs simulated by Meso-NH are almost twice those 36 given by WRF/CHEM at the three stations and for the three dates. The maximum of VOCs is simulated for the Melpitz 37 (26 ppbv) and K-Puszta (24 ppbv) stations on 18 September 2008 and for Melpitz (30 ppbv) on 26 June 2010 (Fig. 8). 38 SO₂ simulated by Meso-NH is three or four times higher than simulated by WRF/CHEM most of the time. The behaviour of NO_x and SO_2 is consistent with the conclusions drawn over Europe as seen above (cf. Sect. 4.4; Fig. 5). 39 40 The Comparison of the diurnal cycles of boundary layer height at the tree sites for the three dates show quite similar results for both models. This means that the observed differences between NO_x and SO₂ cannot be explained by 41

1 thermodynamical modules for gas to particles conversion. Indeed, discrepancies between NO_x and SO₂ are highest

2 when boundary layer height is small (during night-time at plain sites and all along the day at K-Puszta for the 3 March

3 2009), so, when chemical species are more concentrated and chemical reactivity is higher.

4 5.2 Chemical composition of aerosol particles

The model performance in term of aerosol composition is evaluated with the AMS global database. This database 5 provides the mass concentration of NR-PM₁ for NO₃⁻, SO₄²⁺, NH₄⁺ and OM, mentioned as ORG on Figs. 9 and 10. OM 6 includes POM and SOM (for Primary and Secondary OM respectively). Here, OM measured by AMS is compared to 7 8 the sum of anthropogenic primary organic aerosol (APOA) and secondary organic aerosol (SOA) simulated by Meso-9 NH and WRF/CHEM. Simulated and observed mass concentrations are compared as hourly mean to highlight how 10 models are able to represent the detailed processes driving the aerosol composition and its temporal evolution. For each 11 station, this temporal evolution of the aerosol mass composition on 18 September 2008 and on 3 March 2009 is presented in Fig. 9 and Fig. 10 respectively (cf. supplementary materials for 26 June 2010, Fig. S7). The daily mean 12 mass concentration for each aerosol component is also indicated as well as their daily mean mass fraction. 13

14 In order to assess the ability of the models to reproduce aerosol chemical composition, statistical parameters such as

- 15 Root Mean Square Error (RMSE), mean fractional bias (MFB) and mean fractional error (MFE), as discussed by
- 16 Boylan and Russell (2006), are computed for each chemical component of aerosols and for the three case study cases
- 17 (Tables 3, 4 and 5). When the performance criterion (|MFB| < 60% and MFE < 75%) is satisfied, the level of accuracy is
- 18 considered to be acceptable for standard modelling applications. When the performance goals (|MFB| < 30% and MFE
- 19 < 50%) are met, the level of accuracy is considered to be close to the best a model can be expected to achieve.
- 20 Models provide aerosol mass compositions for Aitken and accumulation modes for an aerodynamic diameter less than
- 21 2.5 µm or PM_{2.5}. As AMS-observed NR-PM₁ mass concentration is compared to the simulated PM_{2.5} in the following,
- 22 the models are expected to give an overestimation of aerosol mass concentration. The difference between PM₁ and
- 23 PM_{2.5} mass concentrations can be high (previous studies report PM₁/PM_{2.5} mass ratios between 55 and 75%) and can
- vary according to the meteorological conditions and the emission rates of chemical compounds (Aksoyoglu et al, 2011;
- 25 Spindler et al., 2013). However, the proportions of the predominant chemical components in NR-PM1 and simulated
- 26 PM_{2.5} are expected to be of the same order.

27 5.2.1 At puy de Dôme

- 28 Observations (Fig. 9 and Fig. S7) suggest that, during the summer cases (26 June 2010 and 18 September 2008), the
- 29 total aerosol mass was mostly composed of OM (15.69 μg.m⁻³ (53.05%) and 2.39 μg.m⁻³ (35.09%) respectively). Meso-
- 30 NH shows consistent results, with the highest contribution of simulated OM on 26 June 2010 (11.56 µg.m⁻³ (66.49%))
- and a significant contribution on 18 September 2008 (5.13 μ g.m⁻³ (43.77%)). SOA mass makes the highest contribution
- 32 to OM for the three study cases. WRF/CHEM simulates the highest masses of OM during the two summer cases (0.69
- μ g.m⁻³ for 18 September 2008 and 1.57 μ g.m⁻³ for 26 June 2010) associated with the highest SOA mass fraction in OM.
- 34 At the pdD station, the mass of OM simulated by WRF/CHEM is weaker by a factor of 10 than those simulated by
- 35 Meso-NH.
- 36 On 3 March 2009 (Fig. 10), both models and measurements show a minimum of the mass concentration of NH_4^+ (1.36
- 37 μg.m⁻³, 0.93 μg.m⁻³ and 1.28 μg.m⁻³ for Meso-NH, WRF/CHEM and observations, respectively). This day corresponds
- to the presence of an oceanic air mass (Fig. 1). Both models simulate a minimum of the mass concentration of NO_3^- (0.6
- 39 μg.m⁻³ and 0.57 μg.m⁻³ respectively for Meso-NH and WRF/CHEM) for 18 September 2008, during the presence of a
- 40 Mediterranean air mass. The observed mass concentration of NO_3^- is the weakest for this situation (0.93 µg.m⁻³) but still
- 41 about twice that simulated by the models. Whatever the case, WRF/CHEM systematically underestimates the mass

- 1 concentration of NH_4^+ and NO_3^- at pdD by a factor of 2-4. Meso-NH underestimates the mass concentration of NH_4^+
- 2 and NO₃⁻.on 18 September 2008 and 26 June 2010 (Fig. 9 and Fig. S7). Meso-NH overestimates slightly NH₄⁺ and NO₃⁻
- 3 on 3 March 2009.
- 4 Concerning SO₄²⁻, Meso-NH and WRF/CHEM simulate a comparable mass concentration at pdD for the three days.
- 5 The highest mass concentrations and the highest difference between models for SO_4^{2-} are simulated for 18 September
- 6 2008 (4.3 µg.m⁻³ and 3.03 µg.m⁻³ respectively for Meso-NH and WRF/CHEM). The observed mass concentration is
- 7 about 2, roughly 4 times weaker than the simulated ones for 18 September 2008 and 3 March 2009. In contrast, for 26
- 8 June 2010, the observed mass concentration of SO_4^{2-} is 2-3 higher than the simulated ones.
- 9 The performance criterion, computed for Meso-NH, is satisfied for NH_4^+ on 18 September 2008 (Table 4), for NH_4^+ and
- 10 OM on 26 June 2010 (Table 5), and for NH₄⁺ on 3 March 2009 (Table 3). Concerning WRF/CHEM, the performance
- 11 goal is only attained for NH_4^+ on 3 March 2009.

12 5.2.2 At Melpitz

- 13 Fig. 9 and Fig. 10 indicate that Meso-NH (respectively WRF/CHEM) systematically overestimates (respectively 14 underestimates) the observed OM at Melpitz. Measurements (Fig. 9, Fig. 10) show a more significant contribution of 15 OM to the aerosol mass concentration for 18 September 2008 (3.42 µg.m⁻³ (47.71%)) than for 3 March 2009 (1.83 µg.m⁻³ (13.58%)). Meso-NH reproduces the observed predominant contribution of OM to the aerosol mass 16 concentration for 18 September 2008 (5.74 µg.m⁻³ (53.32%)) and a lower value for 3 March 2009 (3.34 µg.m⁻³ 17 18 (16.23%)). The speciation of OM provided by Meso-NH indicates a significant mass contribution of SOA for both 18 September (88%) and 3 March 2009 (80%). Unlike the observations and Meso-NH, WRF/CHEM does not show a 19 weaker mass of OM on 3 March 2009 than on 18 September 2008 (1.09 µg.m⁻³ (10.76%) versus 1.03 µg.m⁻³ (20.56%) 20
- 21 respectively). For WRF/CHEM, speciation for 3 March 2009 shows a higher contribution of POA to the OM mass
- concentration (89%) than on 18 September 2008 (72%).
- For 3 March 2009, measurements show NH_4^+ (6.60 µg.m⁻³ (49.02%)) and SO_4^{2-} (2.94 µg.m⁻³ (21.86%)) to be the main components of the aerosol. For this date, Meso-NH gives more than half of the total simulated aerosol mass concentration in the form of NO_3^- (10.94 µg.m⁻³ (53.14%)). Like the observations, WRF/CHEM shows a strong contribution of SO_4^{2-} (4.72 µg.m⁻³ (46.27%)) and NH_4^+ (2.36 µg.m⁻³ (23.15%)).
- 27 On 18 September, the proportion of each inorganic aerosol component is quite similar (~1 µg.m⁻³) for the observations
- and both models. While NH_4^+ (44%) dominates the inorganic fraction in observations, SO_4^{2-} dominates the inorganic fraction simulated by WRF/CHEM (45%) and Meso-NH (38%).
- 30 On 3 March 2009, the two models meet the performance criterion for SO4²⁻ and OM (Table 3). It is also met by Meso-
- 31 NH for NH_4^+ on 3 March 2009 (Table 3). On 18 September 2008, the performance criterion is only fulfilled by
- 32 WRF/CHEM for NO₃⁻ and Meso-NH for OM. (Table 4). For this date, the two models overestimate the observed SO_4^{2-}
- 33 (Fig. 9).

34 5.2.3 At K-Puszta

- Observations at the K-Puszta station are available only for 18 September 2008 (Fig. 9). As for the other two stations of pdD and Melpitz, the two models overestimate the observed SO_4^{2-} .
- 37 As found at the other two sites, observations show a high contribution of OM to the aerosol mass $(3.09 \ \mu g.m^{-3}$
- 38 (35.51%)), comparable to the other sites. Meso-NH simulates the same tendency with a significant contribution of OM
- $39 \quad (7.86 \,\mu\text{g.m}^{-3}(60.42\%))$, stronger than at the pdD and at Melpitz stations, for 18 September 2008. SOA makes the main
- 40 contribution to OM (79%). For WRF/CHEM, OM mass also makes a higher contribution than the other two sites.

- 1 Unlike Meso-NH, WRF/CHEM simulates a high mass fraction (81%) of POA in OM. For this particular day, Meso-NH
- 2 overestimates the relative contribution of OM while WRF-CHEM underestimates it.
- 3 In the inorganic part of aerosols, WRF/CHEM and Meso-NH simulate a dominant contribution of SO_4^{2-} while 4 measurements show a dominant contribution of NO_3^- (2.36 µg.m⁻³). Moreover, the observed mass concentration of NO_3^- 5 found at K-Puszta is higher than at the other two sites on the same date. Both models underestimate NO_3^- (Fig. 9).
- 6 The model performance criterion is met for all species and for both models only at K-Puszta for 18 September 2008
- 7 (Table 4). On this day, the Meso-NH performance goal is met for NH_4^+ and NO_3^- at this station. Concerning
- 8 WRF/CHEM, the performance goal is reached for NH_4^+ and for OM.

9 5.3 Discussion

- Whereas the chemical sinks and sources are specific to each chemical compound, physical processes applied to the total aerosol mass conserve the mass fraction of each aerosol compound.
- 12 The sudden decrease in measured aerosol mass concentration observed at pdD at 10:00 and 18:00 UTC on 18 September 2008 (Fig. 9) and at 11:00 on 26 June 2010 (Fig. S7) is certainly linked to wet scavenging of aerosol 13 14 particles by rain. Both models show rainy weather at pdD on 18 September 2008 and stormy weather with convective 15 cells is visible over the Massif Central Mountains on 26 June 2010 (Fig. 1 and Sect. 4.1). The effect of wet scavenging on measured aerosol mass concentration is also observed at 10:00 UTC on 3 March 2009 (Fig. 10) and at 12:00 UTC on 16 17 18 September 2008 (Fig. 9). The weather simulated by both models is rainy at Melpitz on 3 March 2009 and on 18 September 2008. As wet scavenging is not activated in either model, the effect of this process is not visible. For Meso-18 19 NH, as the mass fraction of each compound in the aerosol is generally well represented, this missing sink process 20 probably explains the overestimation of aerosol mass concentration. For instance, Chung and Seinfeld (2002) showed 21 that about 80% of SOA could be scavenged by cloud water and rainwater.
- 22 Variations of the total aerosol mass concentration at pdD can be also associated with the transition between FT and BL 23 and with the long distance transport of polluted air mass up to the pdD station. For instance, on 3 March 2009, both 24 observed and simulated mass concentrations of all compounds are weak and comparable (Fig. 10), indicating that the 25 pdD station is located inside the FT as expected for a winter air mass (Freney et al., 2011). In contrast, on 26 June 2010, 26 the observed mass concentrations are high for all compounds (Fig. S7), indicating that the pdD station was in the BL as 27 is typical in summer (Freney et al., 2011). Even though simulated local meteorological conditions show that the altitude 28 of the pdD station in the models is within the BL, simulated mass concentrations for this day show an underestimation 29 for all compounds in both models (Fig. S7). This can be attributed to, for example, a more pronounced influence of 30 Clermont-Ferrand city pollution in observations than in simulations. It can be noted that Meso-NH simulates a large 31 mass contribution of SOA on 26 June 2010, in agreement with the high mass contribution of OM observed (Fig. S7).
- 32 Observations (Fig. 9,, Fig. S7) suggest that, in the summer cases (26 June 2010 and 18 September 2008), the aerosol is 33 mostly composed of OM at all sites. Freney et al. (2011) attributed the organic aerosol mass at the pdD station to 34 biogenic sources by analysing the correlation of low volatility oxygenated organic aerosol particles with fragments of 35 mass spectral markers of wood burning. For all cases, Meso-NH simulates a significant contribution of SOA to OM mass concentration (>80%) whereas, for WRF/CHEM, OM is predominantly composed of POA (>70% for 3 March 36 37 2009 and for 18 September 2008 at Melpitz and K-Puszta). However, for WRF/CHEM, at the pdD station for summer 38 cases, SOA dominates the OM mass concentration (57% for 18 September 2008 and 79% for 26 June 2010). These 39 results stress that the pdD station is farther from polluted sources than Melpitz and K-Puszta, which are located in plains 40 and that, as expected, the SOA production is highest for summer cases in both models. OM mass concentration
- 41 simulated by WRF/CHEM is systematically underestimated by a factor varying between 2 and 10 in comparison with

Meso-NH and observations. This systematic underestimation of OM by WRF/CHEM is associated with VOC mixing 1 2 ratios that are systematically lower than those in Meso-NH (Fig. 8). This underestimation of OM is the strongest for 26 June 2010 at pdD, where the difference of VOC mixing ratios between WRF/CHEM and Meso-NH during daytime is 3 4 the highest. As both models simulate close values of POA at all stations and in all cases, the lowest OM mass 5 concentration simulated by WRF/CHEM comes from its lower ability to convert VOCs into SOA across the aging constant of organic condensable vapours (OCVs) according to the oxidizing with the hydroxyl radicals used into the 6 7 VBS parameterization (cf. Sect. 2.2). Moreover, the low VOC mixing ratios in WRF/CHEM can be explained by a too 8 high dry deposition of VOCs due to a failure to consider mesophyll resistances and the use of a factor of 0.25 (cf. Sect. 9 2.3). An important difference between both models is also their chemical mechanism: RACM was designed to compute 10 the classical regional atmospheric chemistry whereas CACM treated in addition to this classical chemistry the formation 11 of gaseous organic precursors of SOA. This difference likely contributed to the low VOC mixing ratios associated to 12 low amounts of SOA simulated by WRF/CHEM. While anthropogenic emissions came from the same inventory for 13 both models, the difference of VOC mixing ratios is strengthened by the use of different biogenic and biomass burning 14 emissions. In particular, Meso-NH used, for biogenic emissions, a static inventory computed from the MEGAN model 15 while WRF/CHEM computed biogenic emissions from MEGAN model coupled online with its surface module.

16 For inorganic aerosol compounds, the differences between the two models and between models and observations are 17 very contrasted. For the pdD station, when the contribution of inorganic species mass to the total inorganic mass is 18 considered (not shown), observations and simulated inorganic mass contributions are very close on 26 June 2010. On 18 19 September 2008 and 3 March 2009, both models overestimate the inorganic mass contribution of sulphate. For 18 20 September, the inorganic mass contributions simulated by both models are in agreement whereas, for 3 March 2009, the 21 nitrate inorganic mass contribution and the nitrate mass concentration simulated by Meso-NH is higher than those 22 simulated by WRF/CHEM. On 3 March 2009 during the day, local weather conditions show a temperature up to 4 K 23 higher simulated by Meso-NH than by WRF/CHEM (not shown). In addition, NH₃ and NO_x simulated by WRF/CHEM are higher than simulated by Meso-NH on 3 March 2009 at all stations (Fig. 8). These differences in temperature and 24 gas-phase aerosol precursors on 3 March 2009 at pdD are likely reasons for the higher nitrate inorganic mass 25 26 contribution simulated by Meso-NH than by WRF/CHEM.

At the two plain stations (Melpitz and K-Puszta), both models underestimate NH₄⁺ mass concentrations. Simulated 27 NO₃⁻ and NH₄⁺ mass concentrations are higher for Meso-NH than for WRF/CHEM (Fig. 9 and Fig. 10) except on 18 28 September 2008 at K-Puszta, where the NH₄⁺ mass concentration simulated by WRF/CHEM is slightly higher than that 29 30 simulated by Meso-NH. This general behaviour of NO3⁻ and NH4⁺ mass concentrations is linked to a systematic 31 underestimation of simulated NO_x by Meso-NH compared to WRF/CHEM (Fig. 5, Fig. 8), which is maximum on 3 March 2009. This underestimation of simulated NO_x by Meso-NH may come from a more efficient transfer of oxidation 32 products of NO_x as nitric acid from the gas to the aerosol phase. Both models overestimate the SO_4^{2-} mass 33 concentration, except for Meso-NH at Melpitz on 3 March 2009 - linked to its high simulated NO₃⁻ mass concentration. 34 SO₂ simulated by WRF/CHEM is underestimated in comparison with observations (cf. Sect. 4.4 and Sect. 5.1). In both 35 models, SO_4^{2-} is formed by oxidation of SO_2 in the gas phase. Therefore, the underestimation of SO_2 in the gas phase by 36 WRF/CHEM, which is linked to its overestimation of aerosol SO_4^2 , seems to indicate a faster oxidation in WRF/CHEM 37 than in Meso-NH. On 18 September 2008 at K-Puszta and on 3 March 2009 at Melpitz, the SO₄² mass concentration 38 simulated by WRF/CHEM is higher by a factor of about two than that simulated by Meso-NH. However, as noted 39 40 before, the relative behaviour of NH_4^+ mass concentration for these two cases is opposite: Meso-NH simulates more NH₄⁺ than WRF/CHEM at Melpitz on 3 March 2009 and less at K-Puszta on 18 September 2008. This behaviour can be 41 42 explained by the very high NO3⁻ mass concentration simulated by Meso-NH at Melpitz on 3 March 2009, leading to

additional transfer of NH_3 to the particulate phase. After ammonium sulphate was formed, the formation of NH_4NO_3 and subsequently of NH_4^+ and NO_3^- ions in the aerosol phase is favoured by high relative humidity and low temperature, which was the case on 3 March 2009 at Melpitz as shown by the local simulated meteorological condition.

4

5 6 Summary and Conclusion

6 In this paper, our goal was to evaluate the ability of the Meso-NH and WRF/CHEM models to simulate three one-day 7 cases over Europe. Meteorological fields (wind direction, wind speed, temperature, dewpoint and precipitation), 8 gaseous species concentrations (O_3 , NO_x , SO_2) and aerosol particle compositions (inorganic and OM) have been 9 compared model to model as well as to available measurements. These one-day duration cases came from Freney et al. 10 (2011) and were chosen in order to simulate different seasons and air mass characteristics.

Simulated surface fields were compared to several surface hourly databases from stations (NOAA's NCDC for meteorology, AIRBASE for gaseous species and AMS global database for aerosol composition) over Europe. Simulated vertical profiles were evaluated above Frankfurt against airborne measurements of meteorological parameters and gaseous species from the MOZAIC database. The differences in simulated and observed CO and ozone at Frankfurt are

15 mostly due to the initial fields provided by the MOZART CTM.

16 The two models reproduced the contrasted meteorological conditions for the three selected days. The comparison with

17 vertical profiles of meteorological variables at Frankfurt and with observations at the surface was satisfactory.

18 The results concerning gaseous species at the surface are in good agreement with observations, mostly during the 19 daytime for NO_x and SO₂. Globally, O₃ presents the same diurnal evolution of bias between observed and simulated 20 concentrations for both models with a quasi-constant gap of between 10 and 20 μ g.m⁻³. In the vertical direction, O₃ and

21 CO are well represented in the FT. The maximum bias appears near the surface for CO, with local emissions not

22 correctly included in the models, and near the tropopause for CO and O_3 , due to stratospheric intrusion not being well

simulated by either model. The results suggest that the photochemistry is comparable for both models. Differences are

- 24 probably due to simulated dynamics.
- 25 The simulated aerosol chemical composition is encouraging, with several model performance criteria met.

At the three local stations (pdD, Melpitz and K-Puszta), the systematic underestimation of simulated VOCs by WRF/CHEM is correlated with an underestimation of OM mass concentration in the aerosol phase. Moreover, OM simulated by WRF/CHEM presents a major contribution of POA, suggesting a missing source for SOA in WRF/CHEM

29 parameterization. The contribution of OM is well simulated by Meso-NH in both proportion and quantity, with a higher

30 contribution for summer cases. For Meso-NH, SOA make the major contribution to OM. Mass concentration of SO_4^{2-}

simulated by both models is often overestimated, a fact that is certainly associated with biases of SO_2 mixing ratio. Simulated NO_3^- and NH_4^+ mass concentrations are almost always higher for Meso-NH than for WRF/CHEM, in

connection with differences in NO_X mixing ratio between the models. Finally, computations of model performance criterion (met if (|MFB| < 60% and MFE < 75%) and model performance goals (met if (|MFB| < 30% and MFE < 50%)

35 show that both models can be considered acceptable for standard modelling applications.

It should be kept in mind that the use of different biogenic and biomass burning emission inventories associated with the use of different gaseous chemical mechanisms in Meso-NH and WRF/CHEM lead to differences between the simulated amounts of gas phase precursors of aerosols, and thus aerosol amounts. In particular, Meso-NH model, using a gaseous chemical mechanism designed to compute the organic precursors of aerosols, shows comparable simulated amounts of SOA to observations at local sites.

2 Acknowledgements

3 We thank ECCAD for the archiving and distribution of the emission data, including MACCity, MEGAN and GFED.

- 4 For initial and boundary conditions, we would like to acknowledge Louisa Emmons for providing MOZART outputs
- 5 and for answering our questions, and also the European Centre for Medium-Range Weather Forecasts (ECMWF) for the
- 6 meteorological data. We also thank NOAA's National Climatic Data Center (NCDC), the MOZAIC Data Centre, the
- 7 European Air quality dataBase (AIRBASE) and the Aerosols Mass Spectrometry (AMS) Global Database for their
- 8 meteorological fields, airborne gaseous data, surface gaseous data and aerosol composition data, respectively, over
- 9 Europe. The authors are very grateful to the Agence Nationale de la Recherche (ANR) for its financial support through
- 10 the CUMULUS project (ANR-10-BLAN-0617). Computer resources were provided by CINES (Centre Informatique
- 11 National de l'Enseignement Supérieur). We acknowledge everyone at the GAW puy de Dôme station. We also would 12 like to thank the Meso-NH team for its technical assistance.
- 13

1

14 **References**

- Ackermann, I.J., Hass, H., Memmesheimer, M., Ebel, A., Binkowski, F.S., Shankar, U., 1998. Modal aerosols dynamics
 model for Europe: Development and first applications. Atmos. Environ. 32, 2981–2999.
- 17 Ahmadov, R., McKeen, S.A., Robinson, A.L., Bahreini, R., Middlebrook, A.M., de Gouw, J.A., Meagher, J., Hsie, E.-
- 18 Y., Edgerton, E., Shaw, S., et al., 2012. A volatility basis set model for summertime secondary organic aerosols over the
- 19 eastern United States in 2006. J. Geophys. Res. 117, D06301.
- 20 Aksoyoglu, S., Keller, J., Barmpadimos, I., Oderbolz, D., Lanz, V.A., Prevot, A.S.H., Baltensperger, U., 2011. Aerosols
- modelling in Europe with a focus on Switzerland during summer and winter episodes. Atmos. Chem. Phys. 11, 7355–
 7373.
- Aouizerats, B., Tulet, P., Pigeon, G., Masson, V., Gomes, L., 2011. High resolution modelling of aerosol dispersion
 regimes during the CAPITOUL field experiment: from regional to local scale interactions. Atmos. Chem. Phys. 11,
- 25 7547–7560.
- 26 Appel, K.W., Chemel, C., Roselle, S.J., Francis, X.V., Hu, R.-M., Sokhi, R.S., Rao, S.T., Galmarini, S., 2012.
- Examination of the Community Multiscale Air Quality (CMAQ) model performance over the North American and
 European domains. Atmos. Environ. 53, 142–155.
- 29 AQD, 2008. Directive 2008/50/EC of the European Parliament and of the Council of 21 May 2008 on ambient air
- 30 quality and cleaner air for Europe. Official Journal No. 152.
- 31 Asmi, A., Wiedensohler, A., Laj, P., Fjaeraa, A.-M., Sellegri, K., Birmili, W., Weingartner, E., Baltensperger, U.,
- 32 Zdimal, V., Zikova, N., Putaud, J.-P., Marinoni, A., Tunved, P., Hansson, H.-C., Fiebig, M., Kivekäs, N., Lihavainen,
- 33 H., Asmi, E., Ulevicius, V., Aalto, P.P., Swietlicki, E., Kristensson, A., Mihalopoulos, N., Kalivitis, N., Kalapov, I.,
- 34 Kiss, G., de Leeuw, G., Henzing, B., Harrison, R.M., Beddows, D., O'Dowd, C., Jennings, S.G., Flentje, H., Weinhold,
- 35 K., Meinhardt, F., Ries, L., Kulmala, M., 2011. Number size distributions and seasonality of submicron particles in
- 36 Europe 2008–2009. Atmos. Chem. Phys. 11, 5505–5538. doi:10.5194/acp-11-5505-2011.
- 37 Baklanov, A., Schluenzen, K., Suppan, P., Baldasano, J., Brunner, D., Aksoyoglu, S., Carmichael, G., Douros, J.,
- 38 Flemming, J., Forkel, R., et al., 2014. Online coupled regional meteorology chemistry models in Europe: current status
- and prospects. Atmos. Chem. Phys. 14, 317–398.
- 40 Bechtold, P., Bazile, E., Guichard, F., Mascart, P., Richard, E., 2001. A mass-flow convection scheme for regional and
- 41 global models. Q. J. R. Meteorol. Soc. 127, 869–886.

- 1 Bègue, N., Tulet, P., Chaboureau, J.-P., Roberts, G., Gomes, L., Mallet, M., 2012. Long-range transport of Saharan dust
- 2 over northwestern Europe during EUCAARI 2008 campaign: Evolution of dust optical properties by scavenging. J.
- 3 Geophys. Res.-Atmos. 117, D17201.
- 4 Bessagnet, B., Menut, L., Curci, G., Hodzic, A., Guillaume, B., Liousse, C., Moukhtar, S., Pun, B., Seigneur, C.,
- 5 Schulz, M., 2008. Regional modeling of carbonaceous aerosols over Europe focus on secondary organic aerosols.
- 6 Journal of Atmospheric Chemistry 61, 175–202.
- Binkowski, F.S., Shankar, U., 1995. The Regional Particulate Matter Model: 1. Model description and preliminary
 results. J. Geophys. Res.-Atmos. 100, 26191–26209.
- 9 Boucher, O., Randall, D., Artaxo, P., Bretherton, C., Feingold, G., Forster, P., Kerminen, V.-M., Kondo, Y., Liao, H.,
- 10 Lohmann, U., Rasch, P., Satheesh, S.K., Sherwood, S., Stevens B., Zhang, X.Y., 2013. Clouds and Aerosols. In: Climate
- 11 Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the
- 12 Intergovernmental Panel on Climate Change [Stocker, T.F., D. Qin, G.-K. Plattner, M. Tignor, S.K. Allen, J. Boschung,
- 13 A. Nauels, Y. Xia, V. Bex and P.M. Midgley (eds.)]. Cambridge University Press, Cambridge, United Kingdom and
- 14 New York, NY, USA.
- 15 Bougeault, P., Lacarrere, P., 1989. Parameterization of Orography-Induced Turbulence in a Mesobeta-Scale Model.
- 16 Mon. Weather Rev. 117, 1872–1890.
- 17 Boylan, J.W., Russell, A.G., 2006. PM and light extinction model performance metrics, goals, and criteria for three-
- 18 dimensional air quality models. Atmos. Environ. 40, 4946–4959.
- 19 Carlson, T.N., Boland, F.E., 1978. Analyses of urban-rural canopy using a surface heat flux/temperature model. J. Appl.
- 20 Meteor. 17, 998–1013.
- 21 Charnock, H., 1955. Wind stress on a water surface, Q. J. R. Meteorol.Soc. 81, 639–640, 1955.
- 22 Chen, F., Dudhia, J., 2001. Coupling an advanced land surface-hydrology model with the Penn State-NCAR MM5
- 23 modeling system. Part I: Model implementation and sensitivity. Mon. Weather Rev. 129, 569–585.
- Chou, M.D., Suarez, M.J., Ho, C.H., Yan, M.M.H., Lee, K.T., 1998. Parameterizations for cloud overlapping and shortwave single-scattering properties for use in general circulation and cloud ensemble models. J. Clim. 11, 202–214.
- Chung, S.H., Seinfeld, J.H., 2002. Global distribution and climate forcing of carbonaceous aerosols. J. Geophys. Res. Atmos. 107, 4407.
- Cohard, J.M., Pinty, J.P., Bedos, C., 1998. Extending Twomey's analytical estimate of nucleated cloud droplet
 concentrations from CCN spectra. J. Atmos. Sci. 55, 3348–3357.
- Colette, A., Bessagnet, B., Meleux, F., Terrenoire, E., Rouil, L., 2014. Frontiers in air quality modelling. Geosci. Model
 Dev. 7, 203–210.
- Crassier, V., Suhre, K., Tulet, P., Rosset, R., 2000. Development of a reduced chemical scheme for use in mesoscale
 meteorological models. Atmos. Environ. 34, 2633–2644.
- Cuxart, J., Bougeault, P., Redelsperger, J.L., 2000. A turbulence scheme allowing for mesoscale and large-eddy
 simulations. Q. J. R. Meteorol. Soc. 126, 1–30.
- 36 Diehl, T., Heil, A., Chin, M., Pan, X., Streets, D., Schultz, M., Kinne, S., 2012. Anthropogenic, biomass burning, and
- 37 volcanic emissions of black carbon, organic carbon, and SO2 from 1980 to 2010 for hindcast model experiments.
- 38 Atmos. Chem. Phys. Discuss. 12, 24895–24954.
- 39 Donahue, N.M., Robinson, A.L., Stanier, C.O., Pandis, S.N., 2006. Coupled partitioning, dilution, and chemical aging
- 40 of semivolatile organics. Environ. Sci. Technol. 40, 2635–2643.

- 1 Dusek, U., Frank, G.P., Hildebrandt, L., Curtius, J., Schneider, J., Walter, S., Chand, D., Drewnick, F., Hings, S., Jung,
- 2 D., et al., 2006. Size matters more than chemistry for cloud-nucleating ability of aerosol particles. Science 312, 1375-
- 3 1378.
- 4 Emmons, L.K., Walters, S., Hess, P.G., Lamarque, J.-F., Pfister, G.G., Fillmore, D., Granier, C., Guenther, A., Kinnison,
- 5 D., Laepple, T., et al., 2010. Description and evaluation of the Model for Ozone and Related chemical Tracers, version 4
- 6 (MOZART-4). Geosci. Model Dev. 3, 43–67.
- 7 Erisman, J., Vanpul, A., Wyers, P., 1994. Parametrization of surface-resistance for the quantification of atmospheric
 8 deposition of acidifying pollutants and ozone. Atmos. Environ. 28, 2595–2607.
- 9 Ervens, B., Turpin, B.J., Weber, R.J., 2011. Secondary organic aerosol formation in cloud droplets and aqueous particles
- 10 (aqSOA): a review of laboratory, field and model studies. Atmos. Chem. Phys. 11, 11069–11102.
- 11 Fouquart, Y., Bonnel, B., 1980. Computations of solar heating of the earth's atmosphere: A new parametrization. Beitr.
- 12 Phys. Atmosph. 53, 35-62.
- 13 Freney, E.J., Sellegri, K., Canonaco, F., Boulon, J., Hervo, M., Weigel, R., Pichon, J.M., Colomb, A., Prevot, A.S.H.,
- 14 Laj, P., 2011. Seasonal variations in aerosols particle composition at the puy-de-Dome research station in France.
- 15 Atmos. Chem. Phys. 11, 13047–13059.
- 16 Geiger, H., Barnes, I., Bejan, I., Benter, T., Spittler, M., 2003. The tropospheric degradation of isoprene: an updated
- 17 module for the regional atmospheric chemistry mechanism. Atmos. Environ. 37, 1503–1519.
- 18 Granier, C., Bessagnet, B., Bond, T., D'Angiola, A., Gon, H.D. van der, Frost, G.J., Heil, A., Kaiser, J.W., Kinne, S.,
- 19 Klimont, Z., et al., 2011. Evolution of anthropogenic and biomass burning emissions of air pollutants at global and 20 regional scales during the 1980–2010 period. Climatic Change 109, 163–190.
- Grell, G.A., Devenyi, D., 2002. A generalized approach to parameterizing convection combining ensemble and data
 assimilation techniques. Geophys. Res. Lett. 29, 1693.
- Grell, G. A., Peckham, S. E., Schmitz, R., McKeen, S. A., Frost, G., Skamarock, W. C. Eder, B., 2005. Fully coupled
 "online" chemistry within the WRF model. Atmos. Environ. 39,, 6957–6975, doi:10.1016/j.atmosenv.2005.04.027.
- Griffin, R.J., Dabdub, D., Seinfeld, J.H., 2002. Secondary organic aerosol: 1. Atmospheric chemical mechanism for production of molecular constituents. J. Geophys. Res. 107(D17), 4332, doi:10.1029/2001JD000541.
- Griffin, R.J., Nguyen, K., Dabdub, D., Seinfeld, J.H., 2003. A coupled hydrophobic-hydrophilic model for predicting
 secondary organic aerosols formation. J. Atmos. Chem. 44, 171–190.
- 29 Griffin, R.J., Dabdub, D., Seinfeld, J.H., 2005. Development and initial evaluation of a dynamic species-resolved model
- 30 for gas phase chemistry and size-resolved gas/particle partitioning associated with secondary organic aerosols
- 31 formation. J. Geophys. Res.-Atmos. 110, D05304.
- 32 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P.I., Geron, C., 2006. Estimates of global terrestrial isoprene
- emissions using MEGAN (Model of Emissions of Gases and Aerosols from Nature). Atmos. Chem. Phys. 6, 3181–
 3210.
- 35 Jimenez, P.A., Dudhia, J., 2012. Improving the representation of resolved and unresolved topographic effects on surface
- 36 wind in the WRF model. J. Appl. Meteorol. Climatol. 51, 300–316.
- 37 Hong, S.-Y., Noh, Y., Dudhia, J., 2006. A new vertical diffusion package with an explicit treatment of entrainment
- 38 processes. Monthly Weather Review 134, 2318–2341.
- Kain, J., Fritsch, J., 1990. A one-dimensional entraining detraining plume model and its application. J. Atmos. Sci. 47,
 2784–2802.
- 41 Kanakidou, M., Seinfeld, J. H., Pandis, S. N., Barnes, I., Dentener, F. J., Facchini, M. C., Van Dingenen, R., Ervens, B.,
- 42 Nenes, A., Nielsen, C. J., Swietlicki, E., Putaud, J. P., Balkanski, Y., Fuzzi, S., Horth, J., Moortgat, G. K., Winterhalter,

- R., Myhre, C. E. L., Tsigaridis, K., Vignati, E., Stephanou, E. G., Wilson, J., 2005. Organic aerosol and global climate
 modelling: a review. Atmos. Chem. Phys. 5, 1053-1123, doi:10.5194/acp-5-1053-2005.
- 3 Knote, C., Brunner, D., Vogel, H., Allan, J., Asmi, A., Äijälä, M., Carbone, S., van der Gon, H.D., Jimenez, J.L.,
- 4 Kiendler-Scharr, A., et al., 2011. Towards an online-coupled chemistry-climate model: evaluation of trace gases and
- 5 aerosols in COSMO-ART. Geosci. Model Dev. 4, 1077–1102.
- 6 Kukkonen, J., Olsson, T., Schultz, D.M., Baklanov, A., Klein, T., Miranda, A.I., Monteiro, A., Hirtl, M., Tarvainen, V.,
- 7 Boy, M., et al., 2012. A review of operational, regional-scale, chemical weather forecasting models in Europe. Atmos.
- 8 Chem. Phys. 12, 1–87.
- 9 Lafore, J.P., Stein, J., Asencio, N., Bougeault, P., Ducrocq, V., Duron, J., Fischer, C., Hereil, P., Mascart, P., Masson, V.,
- 10 et al., 1998. The Meso-NH atmospheric simulation system. Part I: adiabatic formulation and control simulations. Ann.
- 11 Geophys.-Atmos. Hydrospheres Space Sci. 16, 90–109.
- 12 Lamarque, J.-F., Bond, T.C., Eyring, V., Granier, C., Heil, A., Klimont, Z., Lee, D., Liousse, C., Mieville, A., Owen, B.,
- 13 et al., 2010. Historical (1850-2000) gridded anthropogenic and biomass burning emissions of reactive gases and
- 14 aerosols: methodology and application. Atmos. Chem. Phys. 10, 7017–7039.
- 15 Lee-Taylor, J., Madronich, S., Aumont, B., Baker, A., Camredon, M., Hodzic, A., Tyndall, G.S., Apel, E., Zaveri, R.A.,
- 16 2011. Explicit modeling of organic chemistry and secondary organic aerosol partitioning for Mexico City and its
- 17 outflow plume. Atmos. Chem. Phys. 11, 13219–13241.
- Lin, S.J., Rood, R.B., 1996. Multidimensional flux-form semi-Lagrangian transport schemes. Mon. Weather Rev. 124,
 2046–2070.
- 20 Liu, Y., Li, F.L., 2006. A preliminary approach on the land surface temperature (LST) lapse rate of mountain area using
- 21 MODIS data art. no. 619907. In Remote Sensing and Space Technology for Multidisciplinary Research
- and Applications, Q. Tong, X. Chen, A. Huang, and W. Gao, eds. (Bellingham: Spie-Int Soc Optical Engineering), pp.
 19907–19907.
- Madronich, S., McKenzie, R.L., Bjorn, L.O., Caldwell, M.M., 1998. Changes in biologically active ultraviolet radiation reaching the Earth's surface. J. Photochem. Photobiol. B-Biol. 46, 5–19.
- 26 Madronich, S. Flocke, S., 1999. The role of solar radiation in atmospheric chemistry, in: Handbook of Environmental
- 27 Chemistry, edited by: Boule, P., Springer-Verlag, Heidelberg, 1–26.
- 28 Maenhaut, W., Raes, N., Chi, X., Cafmeyer, J., Wang, W., 2008. Chemical composition and mass closure for PM2.5 and
- 29 PM10 aerosols at K-puszta, Hungary, in summer 2006. X-Ray Spectrom. 37, 193–197.
- 30 Marenco, A., Thouret, V., Nedelec, P., Smit, H., Helten, M., Kley, D., Karcher, F., Simon, P., Law, K., Pyle, J., et al.,
- 31 1998. Measurement of ozone and water vapor by Airbus in-service aircraft: The MOZAIC airborne program, An
- 32 overview. J. Geophys. Res.-Atmos. 103, 25631–25642.
- Masson V., 2000. A physically-based scheme for the urban energy budget in atmospheric models. Bound. Layer Meteor.
 1994, 357-397.
- Masson, V., Champeaux, J.L., Chauvin, F., Meriguet, C., Lacaze, R., 2003. A global database of land surface parameters
 at 1-km resolution in meteorological and climate models. J. Clim. 16, 1261–1282.
- 37 Masson, V., Le Moigne, P., Martin, E., Faroux, S., Alias, A., Alkama, R., Belamari, S., Barbu, A., Boone, A., Bouyssel,
- 38 F., Brousseau, P., Brun, E., Calvet, J.-C., Carrer, D., Decharme, B., Delire, C., Donier, S., Essaouini, K., Gibelin, A.-L.,
- 39 Giordani, H., Habets, F., Jidane, M., Kerdraon, G., Kourzeneva, E., Lafaysse, M., Lafont, S., Lebeaupin Brossier, C.,
- 40 Lemonsu, A., Mahfouf, J.-F., Marguinaud, P., Mokhtari, M., Morin, S., Pigeon, G., Salgado, R., Seity, Y., Taillefer, F.,
- 41 Tanguy, G., Tulet, P., Vincendon, B., Vionnet, V., Voldoire, A., 2013. the SURFEXv7.2 land and ocean surface platform
- 42 for coupled or offline simulation of earth surface variables and fluxes. Geosci. Model Dev. 6, 929-960.

- 1 McKenzie, R.L., Bjorn, L.O., Bais, A., Ilyasd, M., 2003. Changes in biologically active ultraviolet radiation reaching
- 2 the Earth's surface. Photochem. Photobiol. Sci. 2, 5–15.
- 3 Metzger, S., Dentener, F., Pandis, S., Lelieveld, J., 2002. Gas/aerosols partitioning: 1. A computationally efficient
- 4 model. J. Geophys. Res. 107, ACH 16–1.
- 5 Menut, L., Goussebaile, A., Bessagnet, B., Khvorostiyanov, D., Ung, A., 2012. Impact of realistic hourly emissions
- 6 profiles on air pollutants concentrations modelled with CHIMERE. Atmos. Environ. 49, 233–244.
- 7 Metzger, S., Dentener, F., Pandis, S., Lelieveld, J., 2002. Gas/aerosol partitioning: 1. A computationally efficient model.
- 8 J. Geophys. Res. 107, ACH 16–1.
- 9 Middleton, P., Stockwell, W., Carter, W., 1990. Aggregation and analysis of volatile organic-compound emissions for
- 10 regional modeling. Atmos. Environ. Part a-General Topics 24, 1107–1133.
- 11 Misenis, C., Zhang, Y., 2010. An examination of sensitivity of WRF/Chem predictions to physical parameterizations,
- 12 horizontal grid spacing, and nesting options. Atmos. Res. 97, 315–334.
- 13 Mlawer, E.J., Taubman, S.J., Brown, P.D., Iacono, M.J., Clough, S.A., 1997. Radiative transfer for inhomogeneous
- 14 atmospheres: RRTM, a validated correlated-k model for the longwave. J. Geophys. Res.-Atmos. 102, 16663–16682.
- 15 Monin, A. S., Obukhov, A. M., 1954. Basic laws of turbulent mixing in the atmosphere near the ground. Tr. Inst. Teor.
- 16 Geofiz. Akad. Nauk SSSR 24, 1963–1987.
- 17 Monks, P.S., Granier, C., Fuzzi, S., Stohl, A., Williams, M.L., Akimoto, H., Amann, M., Baklanov, A., Baltensperger,
- U., Bey, I., et al., 2009. Atmospheric composition change global and regional air quality. Atmos. Environ. 43, 5268–
 5350.
- 20 Morrison, H., Thompson, G., Tatarskii, V., 2009. Impact of cloud microphysics on the development of trailing stratiform
- precipitation in a simulated squall line: Comparison of one- and two-moment schemes. Mon. Weather Rev. 137, 991–
 1007.
- Myhre, G., Shindell, D., Breon, F.-M., Collins, W., Fuglestvedt, J., Huang, J., Koch, D., Lamarque, J.-F., Lee, D.,
 Mendoza, B., Nakajima, T., Robock, A., Stephens, G., Takemura, T., Zhang, H., 2013. Anthropogenic and Natural
- 25 Radiative Forcing. In: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth
- 26 Assessment Report of the Intergovernmental Panel on Climate Change [Stocker, T.F., D. Qin, G.-K. Plattner, M. Tignor,
- 27 S.K. Allen, J. Boschung, A. Nauels, Y. Xia, V. Bex and P.M. Midgley (eds.)]. Cambridge University Press, Cambridge,
- 28 United Kingdom and New York, NY, USA.
- Noilhan, J., Planton, S., 1989. A Simple Parameterization of Land Surface Processes for Meteorological Models. Mon.
 Weather Rev. 117, 536–549.
- Noilhan, J., Donier, S., Lacarrere, P., Sarrat, C., Le Moigne, P., 2011. Regional-scale evaluation of a land surface
 scheme from atmospheric boundary layer observations. J. Geophys. Res.-Atmos. 116, D01104.
- 33 Nopmongcol, U., Koo, B., Tai, E., Jung, J., Piyachaturawat, P., Emery, C., Yarwood, G., Pirovano, G., Mitsakou, C.,
- 34 Kallos, G., 2012. Modeling Europe with CAMx for the Air Quality Model Evaluation International Initiative
- 35 (AQMEII). Atmos. Environ. 53, 177–185.
- 36 Nowak, J.B., Neuman, J.A., Bahreini, R., Brock, C.A., Middlebrook, A.M., Wollny, A.G., Holloway, J.S., Peischl, J.,
- 37 Ryerson, T.B., Fehsenfeld, F.C., 2010. Airborne observations of ammonia and ammonium nitrate formation over
- 38 Houston, Texas. J. Geophys. Res.-Atmos. 115, D22304.
- 39 O'Donnell, D., Tsigaridis, K., Feichter, J., 2011. Estimating the direct and indirect effects of secondary organic aerosols
- 40 using ECHAM5-HAM. Atmos. Chem. Phys. 11, 8635–8659.
- 41 Pergaud, J., Masson, V. Malardel, S., Couvreux, F., 2009. A parameterization of dry thermals and shallow cumuli for
- 42 mesoscale numerical weather prediction. Bound.-Layer. Meteor. 132, 83-106.

- 1 Pfister, G.G., Parrish, D.D., Worden, H., Emmons, L.K., Edwards, D.P., Wiedinmyer, C., Diskin, G.S., Huey, G.,
- 2 Oltmans, S.J., Thouret, V., et al., 2011. Characterizing summertime chemical boundary conditions for airmasses
- 3 entering the US West Coast. Atmos. Chem. Phys. 11, 1769–1790.
- 4 Pinty, J.-P., Jabouille, P., 1998. A mixed-phase cloud parameterization for use in mesoscale non-hydrostatic model:
- 5 simulations of a squall line and of orographic precipitations. Proc. Conf. of Cloud Physics, Everett, WA, USA, Amer.
- 6 Meteor. soc., Aug. 1999, 217 220.
- 7 Pope, C.A., Hansen, M.L., Long, R.W., Nielsen, K.R., Eatough, N.L., Wilson, W.E., Eatough, D.J., 2004. Ambient
- 8 particulate air pollution, heart rate variability, and blood markers of inflammation in a panel of elderly subjects. Environ
- 9 Health Perspect 112, 339–345.
- Poschl, U., 2005. Atmospheric aerosols: Composition, transformation, climate and health effects. Angew. Chem.-Int.
 Edit. 44, 7520–7540.
- 12 Pun, B.K., Griffin, R.J., Seigneur, C., Seinfeld, J.H., 2002. Secondary organic aerosols 2. Thermodynamic model for
- 13 gas/particle partitioning of molecular constituents. J. Geophys. Res.-Atmos. 107, AAC 4–1–AAC 4–15.
- 14 Rao, S.T., Galmarini, S., Puckett, K., 2011. Air Quality Model Evaluation International Initiative (AQMEII) Advancing
- the State of the Science in Regional Photochemical Modeling and Its Applications. Bull. Amer. Meteorol. Soc. 92, 23–
 30.
- Rap, A., Scott, C.E., Spracklen, D.V., Bellouin, N., Forster, P.M., Carslaw, K.S., Schmidt, A., Mann, G., 2013. Natural
 aerosols direct and indirect radiative effects. Geophys. Res. Lett. 40, 3297–3301.
- 19 Rosenfeld, D., Lohmann, U., Raga, G.B., O'Dowd, C.D., Kulmala, M., Fuzzi, S., Reissell, A., Andreae, M.O., 2008.
- 20 Flood or drought: How do aerosols affect precipitation? Science 321, 1309–1313.
- Rueckerl, R., Schneider, A., Breitner, S., Cyrys, J., Peters, A., 2011. Health effects of particulate air pollution: A review
 of epidemiological evidence. Inhal. Toxicol. 23, 555–592.
- Seinfeld, J.H., Pandis, S.N., 1998. Atmospheric chemistry and Physics: From Air Pollution to Climate Change. J. Wiley,
 New York.
- 25 Schere, K., Flemming, J., Vautard, R., Chemel, C., Colette, A., Hogrefe, C., Bessagnet, B., Meleux, F., Mathur, R.,
- 26 Roselle, S., et al., 2012. Trace gas/aerosol boundary concentrations and their impacts on continental-scale AQMEII
- 27 modeling domains. Atmos. Environ. 53, 38–50.
- 28 Skamarock, W.C., Klemp, J.B., Dudhia, J., Gill, D.O., Barker, D.M., Duda, M.G., Huang, X.-Y., Wang, W., Powers,
- 29 J.G., 2008. A description of the advanced research WRF version 3. NCAR Tech. Note.
- 30 Solazzo, E., Bianconi, R., Pirovano, G., Moran, M.D., Vautard, R., Hogrefe, C., Appel, K.W., Matthias, V., Grossi, P.,
- Bessagnet, B., et al., 2013. Evaluating the capability of regional-scale air quality models to capture the vertical distribution of pollutants. Geosci. Model Dev. 6, 791–818.
- 33 Spindler, G., Brüggemann, E., Gnauk, T., Grüner, A., Müller, K., Herrmann, H., 2010. A four-year size-segregated
- characterization study of particles PM10, PM2.5 and PM1 depending on air mass origin at Melpitz. Atmos. Environ. 44,
 164–173.
- 36 Spindler, G., Gruener, A., Mueller, K., Schlimper, S., Herrmann, H., 2013. Long-term size-segregated particle (PM10,
- PM2.5, PM1) characterization study at Melpitz influence of air mass inflow, weather conditions and season. J. Atmos.
 Chem. 70, 165–195.
- 39 Stockwell, W.R., Kirchner, F., Kuhn, M., Seefeld, S., 1997. A new mechanism for regional atmospheric chemistry
- 40 modeling. J. Geophys. Res. 102, 25847–25879.
- 41 Suhre, K., C. Mari, T. Bates, J. Johnson, R. Rosset, Q. Wang, A. Bandy, D. Blake, S. Businger, F. Eisels, B. Huebert, G.
- 42 Kok, R. I. Mauldin, A. Prévôt, R. Schillawski, D. Tanner, D. Thornton, 1998, Physico chemical modeling of the first

- aerosol characterization experiment (ace 1) lagrangian b, 1. a moving column approach, J. Geophys. Res., 103, 16,433-1
- 16,455. 2
- 3 Sun, Y., Wang, Z., Dong, H., Yang, T., Li, J., Pan, X., Chen, P., Jayne, J.T., 2012. Characterization of summer organic
- 4 and inorganic aerosols in Beijing, China with an Aerosols Chemical Speciation Monitor. Atmos. Environ. 51, 250-259.
- 5 Sutton, C., Hamill, T.M., Warner, T.T., 2006. Will perturbing soil moisture improve warm-season ensemble forecasts? A
- proof of concept. Mon. Weather Rev. 134, 3174-3189. 6
- Szopa, S., Foret, G., Menut, L., and Cozic, A., 2009. Impact of large scale circulation on European summer surface 7 8 ozone and consequences for modelling forecast. Atmos. Environ. 43, 1189-1195.
- 9
- Tagaris, E., Sotiropoulou, R.E.P., Gounaris, N., Andronopoulos, S., Vlachogiannis, D., 2013. Air quality over Europe: 10 modelling gaseous and particulate pollutants. Atmos. Chem. Phys. 13, 9661–9673.
- Tang, Y., Carmichael, G.R., Thongboonchoo, N., Chai, T., Horowitz, L.W., Pierce, R.B., Al-Saadi, J.A., Pfister, G., 11
- 12 Vukovich, J.M., Avery, M.A., et al., 2007. Influence of lateral and top boundary conditions on regional air quality
- 13 prediction: A multiscale study coupling regional and global chemical transport models. J. Geophys, Res. 112, D10S18.
- 14 Tao, Z.N., Larson, S.M., Williams, A., Caughey, M., Wuebbles, D.J., 2004. Sensitivity of regional ozone concentrations
- 15 to temporal distribution of emissions. Atmos. Environ. 38, 6279-6285.
- 16 Tørseth, K., Aas, W., Breivik, K., Fjæraa, A. M., Fiebig, M., Hjellbrekke, A. G., Lund Myhre, C., Solberg, S., Yttri, K.
- 17 E., 2012. Introduction to the European Monitoring and Evaluation Programme (EMEP) and observed atmospheric
- 18 composition change during 1972-2009. Atmos. Chem. Phys. 12, 5447-5481, doi:10.5194/acp-12-5447-2012.
- 19 Trier, S.B., Chen, F., Manning, K.W., LeMone, M.A., Davis, C.A., 2008. Sensitivity of the PBL and precipitation in 12-
- 20 day simulations of warm-season convection using different land surface models and soil wetness conditions. Mon.
- 21 Weather Rev. 136, 2321-2343.
- 22 Tuccella, P., Curci, G., Visconti, G., Bessagnet, B., Menut, L., Park, R.J., 2012. Modeling of gas and aerosols with 23 WRF/Chem over Europe: Evaluation and sensitivity study. J. Geophys. Res. 117.
- Tulet, P., Crassier, V., Solmon, F., Guedalia, D., Rosset, R., 2003. Description of the MESOscale NonHydrostatic 24
- 25 Chemistry model and application to a transboundary pollution episode between northern France and southern France. J.
- 26 Geophys. Res. 108(D1), 4021.
- 27 Tulet, P., Crassier, V., Cousin, F., Suhre, K., Rosset, R., 2005. ORILAM, a three-moment lognormal aerosol scheme for
- 28 mesoscale atmospheric model: Online coupling into the Meso-NH-C model and validation on the Escompte campaign.
- 29 J. Geophys. Res. 110, D18201.
- 30 Tulet, P., Grini, A., Griffin, R.J., Petitcol, S., 2006. ORILAM-SOA: A computationally efficient model for predicting 31 secondary organic aerosols in three-dimensional atmospheric models. J. Geophys. Res.111, D23208.
- 32 Twomey, D.S., 1959. The nuclei of natural cloud formation part II: The supersaturation in natural clouds and the
- 33 variation of cloud droplet concentration. Geofisica Pura E Applicata 43, 243-249.
- Van der Werf, G.R., Randerson, J.T., Giglio, L., Collatz, G.J., Kasibhatla, P.S., Arellano Jr., A.F., 2006. Interannual 34 35 variability in global biomass burning emissions from 1997 to 2004. Atmos. Chem. Phys. 6, 3423-3441.
- Van der Werf, G.R., Randerson, J.T., Giglio, L., Collatz, G.J., Mu, M., Kasibhatla, P.S., Morton, D.C., DeFries, R.S., 36
- Jin, Y., van Leeuwen, T.T., 2010. Global fire emissions and the contribution of deforestation, savanna, forest, 37 38 agricultural, and peat fires (1997-2009). Atmos. Chem. Phys. 10, 11707-11735.
- Vautard, R., Builtjes, P.H.J., Thunis, P., Cuvelier, C., Bedogni, M., Bessagnet, B., Honore, C., Moussiopoulos, N., 39
- 40 Pirovano, G., Schaap, M., et al., 2007. Evaluation and intercomparison of Ozone and PM10 simulations by several
- 41 chemistry transport models over four European cities within the CityDelta project. Atmos. Environ. 41, 173–188.

- 1 Wesely, M., 1989. Parameterization of surface resistances to gaseous dry deposition in regional-scale numerical-models.
- 2 Atmos. Environ. 23, 1293–1304.
- 3 Wesely, M.L., Hicks, B.B., 2000. A review of the current status of knowledge on dry deposition. Atmos. Environ.
- 4 34(12–14), 2261–2282.
- 5 Wicker, L. J., Skamarock, W. C., 2002. Time-splitting methods for elastic models using forward time schemes. Mon.
- 6 Weather Rev., 130, 2088–2097.
- 7 Wiedinmyer, C., Akagi, S.K., Yokelson, R.J., Emmons, L.K., Al-Saadi, J.A., Orlando, J.J., Soja, A.J., 2011. The Fire
- 8 INventory from NCAR (FINN): a high resolution global model to estimate the emissions from open burning. Geosci.
- 9 Model Dev. 4, 625–641.
- 10 Wyszogrodzki, A.A., Liu, Y., Jacobs, N., Childs, P., Zhang, Y., Roux, G., Warner, T.T., 2013. Analysis of the surface
- 11 temperature and wind forecast errors of the NCAR-AirDat operational CONUS 4-km WRF forecasting system.
- 12 Meteorol. Atmos. Phys. 122, 125–143.
- 13 Yokelson, R.J., Christian, T.J., Bertschi, I.T., Hao, W.M., 2003. Evaluation of adsorption effects on measurements of
- 14 ammonia, acetic acid, and methanol. J. Geophys. Res. 108, 4649.
- 15 Yokelson, R.J., Bertschi, I.T., Christian, T.J., Hobbs, P.V., Ward, D.E., Hao, W.M., 2003. Trace gas measurements in
- 16 nascent, aged, and cloud-processed smoke from African savanna fires by airborne Fourier transform infrared
- 17 spectroscopy (AFTIR). J. Geophys. Res. 108, 8478.
- 18 Zhang, Y., Pan, Y., Wang, K., Fast, J.D., Grell, G.A., 2010. WRF/Chem-MADRID: Incorporation of an aerosols module
- 19 into WRF/Chem and its initial application to the TexAQS2000 episode. J. Geophys. Res. 115, D18202.
- 20 Zhang, Y., Sartelet, K., Wu, S.-Y., Seigneur, C., 2013. Application of WRF/Chem-MADRID and WRF/Polyphemus in
- 21 Europe Part 1: Model description, evaluation of meteorological predictions, and aerosols-meteorology interactions.
- 22 Atmos. Chem. Phys. 13, 6807–6843.
- 23 Zhang, Y., Sartelet, K., Zhu, S., Wang, W., Wu, S.-Y., Zhang, X., Wang, K., Tran, P., Seigneur, C., Wang, Z.-F., 2013.
- 24 Application of WRF/Chem-MADRID and WRF/Polyphemus in Europe Part 2: Evaluation of chemical concentrations
- and sensitivity simulations. Atmos. Chem. Phys. 13, 6845–6875.
- 26

1 Figure captions

precipitation.

2

Fig. 1: Daily averaged accumulated precipitation (in mm) over Europe simulated by Meso-NH (top) and WRF/CHEM (bottom) right to left for 3 March 2009, for 26 June 2010 and for 18 September 2008. Wind direction is represented by black vector in m.s⁻¹ at 850 hPa at 12:00 UTC. Accumulation precipitation (in mm) is daily averaged, represented by colours and is the sum of accumulated explicit precipitation (rain, snow and graupel) and convective accumulated

8

7

9 Fig. 2: Simulated daily biases for 18 September 2008 between models and observations at NOAA stations. Wind
10 direction (°/N) and wind speed (m.s⁻¹) at 10 m, and temperature (°C) and dewpoint (°C) at 2 m are represented for
11 Meso-NH and WRF/CHEM respectively top to bottom and right to left.

12

Fig. 3: Simulated (blue and red lines, for Meso-NH and WRF/CHEM respectively) and observed (black line) skew-T
plots for temperature (solid lines), dewpoint (dashed lines), wind speed and wind direction (staffs and attached barbs).
Observed vertical profiles were collected during take-off and landing of MOZAIC-IAGOS aircraft at Frankfurt airport
on 18 September 2008 at four different times of day.

17

Fig. 4: Simulated daytime biases of mixing ratio (in μ g.m⁻³) for 18 September 2008 between models and observations at "background" and "rural" AIRBASE stations. O₃, NO_X and SO₂ are represented for Meso-NH and WRF/CHEM respectively left to right and top to bottom.

21

Fig. 5: Diurnal cycle of median bias of mixing ratio of O_3 , NO_X and SO_2 between simulated and observed values on the Airbase stations with mention "Background" and "rural". The left column is for 3 March 2009, the middle one is for 26 June 2010 and the right one is for 18 September 2008. On each plot, the blue and red colours are for Meso-NH and WRF/CHEM respectively. Solid lines represent median, and dashed lines 25^{th} and 75^{th} percentiles. Black solid line separates negative from positive values.

27

Fig. 6: Vertical profiles of CO and ozone mixing ratio biases (in ppbv) between models and observations from take-off and landing of IAGOS-MOZAIC aircraft at Frankfurt airport on 18 September 2008 at four different times of day: Meso-NH (CO dark grey, O₃ gold), WRF-Chem (CO black, O₃ red). CO and O₃ mixing ratio biases between MOZART model and observations at 06:00 and 12:00 UTC are superimposed on the 07:00 UTC and 10:00 UTC plots respectively (light grey for CO, light gold for O₃).

33

Fig. 7: Maps of daily mean $PM_{2.5}$ mass concentration (in $\mu g.m^{-3}$) simulated by Meso-NH (top) and WRF/CHEM (bottom). Daily means from observations are represented by circles at "background" and "rural" AIRBASE stations for

36 3 March 2009 (left), for 26 June 2010 (middle) and for 18 September 2008 (right).

Fig. 8: Temporal evolution of mixing ratio of gas phase aerosol precursors simulated by Meso-NH (solid lines) and
 WRF/CHEM (dashed lines). Top to bottom for 3 March 2009, 26 June 2010 and 18 September 2008. Left to right at puy
 de Dôme, Melpitz and K-Puszta stations. Precursors are represented with different colours: NH₃ (grey), SO₂ (red), NO_x
 (blue) (left Y axis) and VOCs (green) (right Y axis).

5

Fig. 9: Diurnal cycle of aerosol chemical composition in mass concentration (μ g.m⁻³) of NR-PM₁ on the left of each plot (top: measured by AMS, middle: simulated by Meso-NH, bottom: simulated by WRF/CHEM) for 18 September 2008 at puy de Dôme, Melpitz and K-Puszta stations. On the right of each plot, the corresponding daily averaged fractional mass composition is also indicated by pie charts. Percentages in parentheses correspond to the mass fraction of the total aerosol mass for each compound. NR-PM₁ include NH₄⁺ (orange), SO₄²⁻ (red), NO₃⁻ (blue) and ORG (green). Meso-NH and WRF/CHEM differentiate APOA (dark green) and SOA (light green) in OM.

12

Fig. 10: Diurnal cycle of aerosol chemical composition in mass concentration (μ g.m⁻³) of NR-PM₁ on the left of each plot (top: measured by AMS, middle: simulated by Meso-NH, bottom: simulated by WRF/CHEM) for 3 March 2009 at puy-de-Dome, Melpitz and K-Puszta stations. On the right of each plot, the corresponding daily averaged fractional mass composition is also indicated by pie charts. Percentages in parentheses correspond to the mass fraction of the total aerosol mass for each compound. NR-PM₁ include NO₃⁻ (blue), SO₄²⁻ (red), NH₄⁺ (orange) and ORG (green). Meso-NH

18 and WRF/CHEM differentiate APOA (dark green) and SOA (light green) in OM.

| Scheme | Meso-NH | WRF/CHEM | | | | |
|--|---|--|--|--|--|--|
| Surface layer Land and ocean-surface schemes | SURFEX (Masson et al., 2013) ISBA for nature surface (Noilhan and Planton, 1989) TEB for town surface (Masson, 200) | NOAH LSM (Chen and Dudhia, 2001) | | | | |
| Turbulence | Boundary layer with diagnostic ML (Bougeault and Lacarrere, 1989) Prognostic TKE (Cuxart et al., 2000) | | | | | |
| Deep convection Shallow convection | KAFR (Kain and Fritsch, 1990; Bechtold et al., 2001) EDKF (Pergaud et al., 2009) | Grell-Devenyi (Grell and Devenyi, 2002) | | | | |
| Shortwave radiation | ECMWF (Fouquart and Bonnel, 1980) | Goddard (Chou et al., 1998) | | | | |
| Longwave radiation | RRTM (Mlawer et al., 1997) | RRTMG (Mlawer et al.,1997) | | | | |
| Microphysics | ICE3 (Pinty and Jabouille, 1998) | Morrison double-moment scheme (Morrison et al., 2009) | | | | |
| Advection scheme | 4 th order scheme centred on space and time for horizontal and vertical velocities PPM for meteorological and scalar variables (Lin and Rood, 1996) Leapfrog scheme for time integration | Horizontal advection for momentum and scalar is 5 th order Vertical advection for momentum and scalar is 3 rd order Moisture, scalars, TKE, chemical variables: monotonic transport Third-order Runge-Kutta (RK3) time integration scheme (Wicker and Skamarock, 2002) | | | | |
| Gas-phase mechanism | ReLACS2 (Tulet et al., 2006) | RACM (Stockwell et al., 1997) | | | | |
| Aerosols model | ORILAM-SOA (Tulet et al., 2006) | MADE (Ackermann et al., 1998) – VBS for SOA (Ahmadet al., 2012) | | | | |

1 Table 1: Meso-NH and WRF/CHEM schemes

1 Table 2: Normalized Mean Bias (%) of meteorology, gaseous pollutants and PM_{2.5} bulk mass concentration from

2 Airbase and NOAA datasets for the three study cases on a daily basis or for daytime. For meteorology, the number of

3 stations used for the statistics is between 1239 and 1397 depending on the parameter and on the day. For gaseous 4 pollutants and for $PM_{2.5}$ mass concentration, the number of stations used is presented, respectively on Fig. 5 and Fig. 7.

| | 03/0 |)3/09 | 26/ | 06/10 | 18/09/08 | | | |
|-------------------|---------------------|----------|---------------|----------------|----------|----------|--|--|
| | Meso-NH | WRF/CHEM | Meso-NH | WRF/CHEM | Meso-NH | WRF/CHEM | | |
| | Meteorology (daily) | | | | | | | |
| WD10 | 1 | -1 | -2 | -3 | -2 | -3 | | |
| WS10 | -2 | 15 | -1 | 16 | 2 | 20 | | |
| T2m | 0 | 0 | 0 | 0 | 0 | 0 | | |
| Td2m | 0 | 0 | 0 | 0 | 0 | 0 | | |
| | | | Gaseous pollu | utants daytime | | | | |
| O ₃ | 7 | -3 | 4 | -14 | 11 | 1 | | |
| NO _X | -3 | 2 | -1 | -1 | -3 | -2 | | |
| SO_2 | 0 | -1 | -1 | -1 | -1 | -2 | | |
| | | | Bulk ma | ss (daily) | | | | |
| PM _{2.5} | 12 | 0 | 14 | -1 | 12 | -3 | | |

- Table 3: Root Mean Square Error (RMSE), Mean fractional bias (MFB) and error (MFE) for aerosol components for 3
- March 2009 at puy de Dôme (FR) and Melpitz (DE) stations. In bold when the model performance criteria are met (|MFB| < 60% and MFE<75%).

| <u></u> | puy de Dôme | | | | | | Melpitz | | | | | |
|------------------------------|-------------|------|-----|------|---------|-----|---------|------|---------|-----|---------|-----|
| | MNH | WRF | MNH | WRF | MNH | WRF | MNH | WRF | MNH | WRF | MNH | WRF |
| $\mu g.m^{-3}$ | RMSE | | MFB | (%) | MFE (%) | | RMSE | | MFB (%) | | MFE (%) | |
| $\mathrm{NH_4}^+$ | 0.51 | 0.59 | 3 | -31 | 37 | 49 | 3.42 | 4.87 | -44 | -90 | 46 | 90 |
| $\mathrm{SO_4}^{2}$ | 1.19 | 1.48 | 101 | 112 | 101 | 112 | 1.06 | 1.92 | -21 | 49 | 30 | 49 |
| NO ₃ ⁻ | 1.37 | 1.16 | 20 | -74 | 72 | 84 | 9.00 | 0.39 | 138 | 1 | 138 | 16 |
| ORG | 0.93 | 0.91 | 33 | -118 | 58 | 118 | 1.70 | 0.83 | 60 | -49 | 60 | 49 |