

ISSN 1420-3049 © 2007 by MDPI www.mdpi.org/molecules

Full Paper

# **Evaluation of the Activity of the Sponge Metabolites Avarol and Avarone and their Synthetic Derivatives Against Fouling Micro- and Macroorganisms**

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Received: 23 April 2007; in revised form: 13 May 2007 / Accepted: 14 May 2007 / Published: 15 May 2007

Abstract: The sesquiterpene hydroquinone avarol (1) was isolated from the marine sponge *Dysidea avara*, whereas the corresponding quinone, avarone (2), was obtained by oxidation of avarol, and the significantly more lipophilic compounds [3'-(p-chloro-phenyl)avarone (3), 3',4'-ethylenedithioavarone (4), 4'-isopropylthioavarone (5), 4'-tert-butylthioavarone (6), 4'-propylthioavarone (7), 4'-octylthioavarone (8)] were obtained by nucleophilic addition of thiols or *p*-chloroaniline to avarone. All these compounds were

tested, at concentrations ranging from 0.5 to 50 µg/mL, for their effect on the settlement of the cyprid stage of *Balanus amphitrite*, for toxicity to both nauplii and cyprids and for their growth inhibitory activity on marine bacteria (*Cobetia marina*, *Marinobacterium stanieri*, *Vibrio fischeri* and *Pseudoalteromonas haloplanktis*) and marine fungi (*Halosphaeriopsis mediosetigera*, *Asteromyces cruciatus*, *Lulworthia uniseptata* and *Monodictys pelagica*).

**Keywords:** Antifouling activity, settlement inhibition, antimicrobial activity, avarol, avarone.

#### Introduction

One of the most serious problems that marine technology is currently facing is the control of biofouling on man-made structures [1, 2]. Efficient antifouling paints are based on copper compounds and booster biocides that when submerged, release toxic compounds causing adverse environmental effects [3-8]. Organotin-based paints have been linked to pollution of food webs and are of particular concern to human consumers [9, 10]. Environmental protection is now a major concern in the search for new active compounds. New environmental regulations generate further restrictions on the use of biocides in industrial formulations [11] for the protection of rivers, lakes, estuaries, coastal waters and groundwater from further deterioration and for protection of biodiversity. There is clearly a need to develop new non-toxic or environmentally benign antifouling alternatives that would be efficient against the most severe fouling organisms such as barnacles, blue mussels, bryozoans and algae [12]. On one hand, there are solutions based on interference with biological adhesion to surfaces; the socalled foul-release coatings. On the other hand, there is an increasing interest in exploring the antifouling potential of natural compounds. In the marine environment, where all surfaces are constantly exposed to the threat of surface colonization, many sessile organisms remain relatively clean and control epibiont growth using effective antifouling mechanisms. In addition to physical defense mechanisms comprising structural elements made of lignin, CaCO<sub>3</sub>, silica, etc., sessile organisms such as sponges, soft corals, and seaweeds are known to elaborate chemical defense mechanisms against predation and epibiont growth. Many sponges have been shown to synthesize toxic metabolites to prevent predation, and because of this, frequently other organisms attach sponges to themselves for their protection [13].

In continuation of our studies towards the discovery of natural products that are effective against marine biofouling [14, 15], we investigated barnacle settlement inhibition and marine bacterial growth inhibition with the sponge metabolites avarol (1), avarone (2) and the synthetic derivatives **3-8**. Barnacles were selected as test organisms because they are one of the most significant forms of animal fouling [14]. We also assayed the same set of compounds on fouling bacteria since it is widely accepted that biofilm formation is problematic on even the most effective antifouling and foul-release coatings [16].

A great number of marine quinones and hydroquinones are of considerable interest with regard to their diverse biological activities. Furthermore, sesquiterpenes in which a decaline type unit and a quinoid moiety are structurally associated, often exhibit pronouced cytotoxicity [17-20]. Another reason that led us to evaluate the barnacle settlement inhibition of *D. avara* metabolites is that a significant number of other sponge metabolites have exhibited promising activities in an array of antifouling assays [21-24].

#### **Results and Discussion**

The redox couple avarol (1) and avarone (2), isolated for the first time from the sponge *Dysidea avara* by Minale and coworkers [25], has shown strong activity against mouse lymphoma cells [26-28], antiviral activity [29-31], antiinflammatory activity [32], antipsoriatic properties [33], as well as moderate antifungal and antibacterial activity against Gram-positive strains [34, 35]. Intrigued by the wide range of biological activities of these metabolites and the fact that avarol (1) is biosynthesized in large amounts by the sponge, implying an important ecological role [36], we decided to investigate their antifouling activity and prepare a number of lipophilic derivatives that would reduce their solubility in sea water and improve their incorporation in a potential antifouling preparation.

Figure 1. Structures of assayed compounds 1-8.



# Antimicrobial activity against Cobetia marina, Marinobacterium stanieri, Vibrio fischeri and Pseudoalteromonas haloplanktis

Compounds 1-8 were tested against the marine bacteria *C. marina*, *M. stanieri*, *V. fischeri* and *P. haloplanktis* (Table 1). Among the eight compounds tested, only compound 4 did not exhibit antimicrobial activity with a minimum inhibitory concentration (MIC) above  $50\mu g/mL$ . High levels of inhibition against *C. marina* were observed with compounds 5 and 7 (MIC = 0.5  $\mu g/mL$ ), as well as with avarone (2) and compounds 3, 8 (MIC =1  $\mu g/mL$ ). Avarol (1) and 6 were noticeably less active with MIC of  $2.5\mu g/mL$  and  $5\mu g/mL$ , respectively.

*M. stanieri* growth was strongly inhibited in the presence of compounds **5** and **7** (MIC = 0.5  $\mu$ g/mL), and avarone (2) (MIC = 1  $\mu$ g/mL). Compound **8** gave a MIC of 2.5 $\mu$ g/mL. MIC values of 5 $\mu$ g/mL were recorded when bacteria were incubated with avarol (1) and compounds **3** and **8**.

Compounds 5 and 7 were the most efficient growth inhibitors of *V. fischeri* (MIC =  $1.0 \ \mu g/mL$ ). Moderate levels of activity with MIC ranging between 2.5 and 5  $\mu g/mL$  were found for avarone (2), compounds 3, 8 and 6. The less active compound was found to be avarol (1) with a MIC =  $10 \ \mu g/mL$ .

Growth of *P. haloplanktis* was strongly inhibited by compounds **5**, **6** and **7**, with MICs =  $1\mu g/mL$ . Moderate levels of activity with MIC ranging between 2.5 and 5  $\mu g/mL$  were found for avarone (2), **8** and **3**. Finally, avarol (1) showed weak activity (MIC =  $10 \mu g/mL$ ).

No	Compound	Cobetia marina	Marinobacterium stanieri	Vibrio fischeri	Pseudoalteromonas haloplanktis
1	Avarol	2.5	5.0	10.0	10.0
2	Avarone	1.0	1.0	2.5	2.5
3	3'-(p-Chlorophenyl)avarone	1.0	5.0	2.5	5.0
4	3',4'-Ethylenedithioavarone	>50.0	>50.0	>50.0	>50.0
5	4'-Isopropylthioavarone	0.5	0.5	1.0	1.0
6	4'-tert-Butylthioavarone	5.0	5.0	5.0	1.0
7	4'-Propylthioavarone	0.5	0.5	1.0	1.0
8	4'-Octylthioavarone	1.0	2.5	2.5	2.5

**Table 1.** Evaluation of antimicrobial activity (MIC, μg/mL).

# Antifungal activity against Halosphaeriopsis mediosetigera, Asteromyces cruciatus, Lulworthia uniseptata and Monodictys pelagica

The inhibition of marine fungal growth by compounds **1-8** is presented in Table 2. Compounds **4** and **6** were inactive towards the four strains of marine fungi tested.

No	Compound	Halosphaeriopsis mediosetigera	Asteromyces cruciatus	Lulworthia uniseptata	Monodictys pelagica
1	Avarol	10.0	10.0	25.0	25.0
2	Avarone	1.0	2.5	1.0	1.0
3	3'-(p-Chlorophenyl)avarone	25.0	25.0	25.0	50.0
4	3',4'-Ethylenedithioavarone	>50.0	>50.0	>50.0	>50.0
5	4'-Isopropylthioavarone	5.0	10.0	10.0	5.0
6	4'-tert-Butylthioavarone	>50.0	>50.0	>50.0	>50.0
7	4'-Propylthioavarone	1.0	1.0	2.5	1.0
8	4'-Octylthioavarone	10.0	10.0	5.0	5.0

Table 2. Evaluation of antifungal activity (MIC, µg/mL).

The two most active compounds were avarone (2) and compound 7, which exhibited a MIC < 2.5  $\mu$ g/mL against the four strains. Moderate levels of inhibition (MIC < 10.0  $\mu$ g/mL) were obtained against all strains with compounds 5 and 8. Low levels of inhibition (MIC < 25.0  $\mu$ g/mL) were recorded for avarol (1) and compound 3.

# Cyprid settlement and mortality

The effects of compounds **1-8** on cyprid settlement and mortality are presented in Table 3. Only avarol (**1**) and avarone (**2**) were toxic at high concentrations. Avarol was the most toxic, with a concentration inducing 50% mortality in comparison to the control (LC<sub>50</sub>) of 13.28 µg/mL, while avarone had an LC<sub>50</sub> of 27.12 µg/mL. All compounds had concentrations inhibiting settlement by 50% in comparison to the control (EC<sub>50</sub>) lower than the corresponding LC<sub>50</sub>. The three most active compounds were avarol (**1**), **3** and **7** with EC<sub>50</sub> < 1 µg/mL. Avarone (**2**), **5**, **6** and **8** had EC<sub>50</sub> between 1 and 5 µg/mL. Compound **4** was the least active, with EC<sub>50</sub> = 26.22 µg/mL.

# Toxicity of compounds 1-8 to nauplii

The toxicity test results are presented in Table 3. Six compounds (3–8) showed no toxicity at the tested concentrations (LC<sub>50</sub> >50  $\mu$ g/mL). Mortality was noted only for avarol (1) and avarone (2) which had LC<sub>50</sub> 1.58  $\mu$ g/mL and 25.12  $\mu$ g/mL respectively.

	Compound	Balanus amphitrite				
No		$\mathbf{EC}$ (up/mL)	Cyprid	Nauplii		
		$EC_{50}$ (µg/mL)	LC <sub>50</sub> (µg/mL)	LC <sub>50</sub> (µg/mL)		
1	Avarol	$0.65\pm0.03$	$13.28\pm0.70$	$1.58 \pm 0.05$		
2	Avarone	$3.41\pm0.12$	$27.12 \pm 1.51$	$25.12\pm0.93$		
3	3'-(p-Chlorophenyl)avarone	$0.65\pm0.02$	>50.0	>50.0		
4	3',4'-Ethylenedithioavarone	$26.22\pm0.27$	>50.0	>50.0		
5	4'-Isopropylthioavarone	$1.33\pm0.06$	>50.0	>50.0		
6	4'-tert-Butylthioavarone	$4.23\pm0.12$	>50.0	>50.0		
7	4'-Propylthioavarone	$0.45\pm0.02$	>50.0	>50.0		
8	4'-Octylthioavarone	$1.46\pm0.05$	>50.0	>50.0		

Table 3. Evaluation of anti-settlement and toxicity of compounds 1-8 on *B. amphitrite*.

# Conclusions

Biofouling is one of the most serious problems the maritime domain currently faces. It has been estimated that the growth of marine fouling organisms costs the shipping and other marine industries over \$6.5 billion per year [37]. Biofouling is considered to have four distinct stages, the first one starting from the moment a man-made object is immersed in water. The surfaces of these objects quickly accumulate dissolved organic matter and molecules, such as polysaccharides and protein

fragments. Gradually, bacteria and single-cell diatoms sense the surface and start settling on it, forming a microbial film [38]. Subsequently, the adhesive substances and rough irregular microbial colonies trap more particles and organisms. Spores of algae, e.g. species of *Enteromorpha intestinalis*, *Ulothrix zonata*, marine fungi and ciliate protozoa soon appear on the film [39]. In the final stage, other marine organisms, such as barnacles, tunicates, mussels, bryozoans, polychaetes and tubeworms, settle on the submerged surfaces [40].

Many marine sponges, as well as other benthic organisms, are relatively free of settlement by fouling organisms [41] due to the production of biogenic compounds that possess antibacterial, antialgal, antifungal, antiprotozoan and antimacrofouling properties. Therefore, the isolation and production of these natural products from marine organisms could be used for the prevention of biofouling.

To gain a better understanding of chemical antifouling mechanisms of marine organisms, it is necessary: a) to identify the settling preferences of common fouling species, b) to determine the concentrations of secondary metabolites that settlers would experience in the field and c) to develop assay methodologies that deploy compounds in ecologically realistic ways [42].

For marine antifouling research, bioactive substances of particular interest should be ones that show deterrence properties and can be used for the development of antifouling coatings. Natural products rarely are available in sufficient quantity to be commercially harvested from marine macroorganisms. Moreover, most potent natural product compounds are structurally too complex to be commercially synthesized.

The observation that the sponge *D. avara* maintains an epibiont-free surface, in conjunction to the fact that the main metabolite avarol (1), which as already mentioned exhibits a wide range of biological activities, can be relatively easily synthesized in eight steps with a 28% overall yield [43], prompted us to evaluate its antifouling properties.

Avarol (1) showed promising antisettlement activity, but showed significant toxicity towards barnacle larvae. Therefore, unless rapidly biodegradable, it seemed unlikely to be a good candidate for new antifouling agents. For that reason, we proceeded with the preparation of simple synthetic derivatives aiming at an analogue that would maintain the levels of activity but without toxicity.

Avarone (2) and 3 showed significant antisettlement activity, but were almost equally toxic to barnacle larvae as avarol (1). Conversely, compound 4 possessed insufficient antimicrobial and antisettlement activity to be a promising antifoulant. The most active antisettlement compounds were 1, 3 and 7 with  $EC_{50}$  values < 1 µg/mL.

With the exception of compound **4**, all tested compounds showed significant antibacterial activity. Compounds **5** and **7** were found to be the most active with MIC < 1  $\mu$ g/mL. A comparison of activity in relation to structure suggests that steric hindrance exerted by bulky substituents on the quinone ring significantly reduced antimicrobial activity. Blockage of both positions 3' and 4' also deprived the compounds of their antimicrobial properties, as was the case for compound **4**. The low activity of compounds **1** and **2** compared to some derivatives (**3**–**8**) can be explained by their higher polarity and therefore lower membrane permeability. A summary of all results obtained in this study is shown in Figure 2.





The therapeutic ratio,  $LC_{50}/EC_{50}$  is a way of expressing the effectiveness of the compound in relation to its toxicity. From the perspective of potency for use in an antifouling coating, the desired target ratio should be much greater than 1.0 [44].

Compounds 5 and 7 are of particular interest because they showed good antisettlement and antimicrobial activity at concentrations that were not acutely toxic to barnacle larvae, with an  $LC_{50}/EC_{50} > 40$ , thus satisfying the above mentioned criterion. In the future, field assays are planned to determine how specific or broadly active these bioactive compounds are.

# Experimental

#### Isolation of avarol (1) and avarone (2)

*Dysidea avara* specimens (5.1 kg wet tissue) were collected from the Gulf of Kotor, Monte Negro, Yugoslavia, at 5-15 m by SCUBA diving. The freshly collected sponge was initially freeze dried and then exhaustively extracted at room temperature with  $CH_2Cl_2/MeOH$  (2/1, v/v) mixtures. The residues were subjected to vacuum chromatography using silica gel and a step-wised gradient solvent system ranging from 100% cyclohexane to 100% ethyl acetate. The medium polarity fractions, containing avarone (2) and avarol (1) (identified by TLC comparison with authentic material) were further purified either by vacuum/column chromatography and HPLC until pure metabolites 1 (0.15% wet weight) and 2 (0.02% wet weight) were isolated. Structural elucidation of the natural products was based on their spectral data (NMR, MS, IR, UV) and comparison with literature values [25] (Figure 1).

#### Preparation of compounds 3-8

Alkylthio and *p*-chlorophenylamino derivatives of avarone **3-8** were synthesized by nucleophilic addition of thiols and *p*-chloroaniline, respectively, to avarone, as previously described [45]. In a typical experiment, the nucleophile (1.6 mmol) was added to a solution of avarone ( $\mathbf{2}$ , 1.6 mmol) in

ethanol-water (1:1, 50 mL). To improve the reactivity of aliphatic thiols, the solution was made weakly alkaline with NaHCO<sub>3</sub> and the reaction was carried out under an inert nitrogen atmosphere to prevent polymerization of the quinone moiety. Following purification by chromatographic methods, the structure of the obtained quinones was determined using <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectroscopy (Figure 1).

#### Antibacterial assays

Compounds were tested for inhibitory activity against four strains of marine bacteria: *Cobetia marina* (ATTC 25374), *Marinobacterium stanieri* (ATCC 27130), *Vibrio fischeri* (ATCC 7744) and *Pseudoalteromonas haloplanktis* (ATCC 14393). Experiments were performed as previously described [46]. Compounds (at concentrations of 0.5, 1, 2.5, 5, 10, 25 and  $50\mu g/mL$ ) were incubated with the bacteria (2x10<sup>8</sup> cells/mL) in 96-well plates (MERCK) in MHB medium (Mueller Hinton Broth, SIGMA), supplemented with NaCl (15g/L), at 25°C for 24 h. Each treatment and the seawater control were replicated six times. MICs compared to the seawater control, were determined by the microtitre broth dilution method [47].

# Antifungal assays

Compounds were tested for inhibitory activity against four strains of marine fungi obtained from the culture collection of the School of Biological Sciences, University of Portsmouth, UK: *Halosphaeriopsis mediosetigera, Asteromyces cruciatus, Lulworthia uniseptata* and *Monodictys pelagica.* Experiments were performed as previously described [48]. Compounds (at final concentrations of 0.5, 1, 2.5, 5, 10, 25 and 50µg/mL) were incorporated into maize agar 12% (SIGMA) into Pyrex Petri dishes (15mm x 30mm, Fisher, UK). Plates were inoculated aseptically at their centre with an 8 mm diameter agar plug of mycelia. After incubation for 4 weeks at 25°C, MIC were determined. Assays were performed in duplicate.

# Balanus amphitrite adult broodstock maintenance

Adult barnacles, *Balanus amphitrite* Darwin, were supplied courtesy of Prof. Dan Rittschof (Duke University Marine Laboratory, Beaufort, North Carolina). After removal of epibionts, adults were maintained in 20-L plastic aquaria containing 10 µm filtered and UV-irradiated seawater. Tanks were aerated and kept at 22°C on a 14:10 L:D cycle. The adults were fed daily with *Artemia* sp. nauplii and the water was changed every other day, at which time the adults were cleaned again as previously described [49].

# Larval culture

Egg hatching was synchronized by drying adult barnacles overnight followed by re-immersing them in fresh seawater. Nauplii were attracted to a point source of light over a 1.5 h period and collected using a Pasteur pipette. Larvae were then transferred to a 2-L beaker containing filtered seawater. They were then reared to the cyprid stage by conventional methodology [40] in aerated 0.7  $\mu$ m filtered seawater laced with antibiotics (streptomycin at 36.5 mg/mL and penicillin at 21.9 mg/mL) at 28°C on a 14:10 L:D cycle. Larvae were fed daily on a diet of *Skeletonema costatum* (1 L of 2x10<sup>5</sup> cells/mL) [50]. When the majority of larvae had metamorphosed into cyprids (4 days), the culture was filtered through a series of plankton mesh filters (300  $\mu$ m, 250  $\mu$ m and 160  $\mu$ m). Cyprids were retained on the 250 $\mu$ m filter and subsequently transferred to fresh 0.45  $\mu$ m filtered seawater [51] and stored at 6°C [52].

# Algal culture

Seed cultures of *S. costatum* were purchased from Seasalter Shellfish Ltd (Whitstable, U.K.) and cultured in f/2 medium [53] at 18 °C under constant illumination. Cultures were enriched by gassing with  $CO_2$  (20 mL/min for 30 sec day<sup>-1</sup>).

#### Balanus amphitrite cyprid settlement assay

Settlement assays were conducted by adding 10-15 cyprids (3 days old) to the wells of a 24-well microplate (Iwaki) containing 2 mL of the compounds in seawater. The compounds were tested at concentrations of 0.5, 1, 2.5, 5, 10, 25 and 50  $\mu$ g/mL. Test plates were incubated at 28 °C in the dark and results were recorded after 24 hours incubation. Each larva was examined under a dissecting microscope and its condition recorded. Cyprids that did not move, had extended appendages and did not respond after a light touch by a metal probe were regarded as dead [54]. Both permanently attached and metamorphosed individuals were counted as settled, the remaining living cyprids were counted as swimmers. The EC<sub>50</sub> and LC<sub>50</sub> values of each compound were calculated using Sigma Plot (SYSTAT Software Inc.). All bioassays were conducted with six replicates per treatment.

#### Toxicity tests on Balanus amphitrite nauplii

Toxicity tests on nauplii of *B. amphitrite* were conducted as described [55]. Only positively phototactic stage II nauplii were used. Compounds were tested at concentrations of 0.5, 1, 2.5, 5, 10, 25 and 50  $\mu$ g/mL with 6 replicates of each concentration and the seawater control. Tests were conducted by adding 10-15 nauplii to wells of a 24-well plate containing 2 mL of solution. The number of swimming nauplii was recorded after a 24 h exposure to the compounds. Non-swimming larvae were regarded as dead. LC<sub>50</sub> values of each compound were calculated using Sigma Plot. All bioassays were conducted with six replicates per treatment.

#### Acknowledgements

The Greek research team for this project was co-funded by the European Social Fund and National Resources – (EPEAEK II) PYTHAGORAS II, the General Secretariat for the Research and Technology and the University of Athens. The British team was supported by awards from the US Office of Naval Research (N00014-02-1-0311) and from the Natural Environment Research Council

(NER/B/S/ 2003/00273). Our special thanks to Ms S. Henry (School of Marine Science and Technology, Newcastle University, UK) for assistance with cyprid cultures and to Prof. D. Rittschof (Duke University Marine Laboratory, Beaufort, North Carolina, USA) for providing *B. amphitrite* adults. The Serbian team was supported by a bilateral research program of the Yugoslavian Federal Ministry for Development, Science and Environment.

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Sample Availability: Samples of the compounds 1-8 are available from the authors.

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