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# SILICON IONS BELOW 100 KM: A CASE FOR SiO<sub>2</sub><sup>+</sup>

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AUGUST 1974



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Richard A. Goldberg  
Upper Atmosphere Branch

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### ABSTRACT

Silicon ions are normally detected at altitudes above 100 km and within sporadic E layers. Traces have rarely been observed within the more permanent metallic layer near 93 km. This is surprising since silicon is an important constituent of chondritic meteorites, which ablate material in this region to provide a primary source of the metallic species observed there. In this work, we present evidence that  $\text{Si}^+$  ions form  $\text{SiO}_2^+$  at the lower altitudes, and exist in this ionic state prior to recombination. A rocket launched from El Arenosillo, Spain on 3 July 1972, at 0743 LMT ( $\lambda = 57.1^\circ$ ), during the predicted period of the  $\beta$  Taurids meteor shower, passed through a continuous belt of metallic ions that began near 85 km, ended near 115 km, and exhibited an order of magnitude increase in the form of a layer near 114 km.  $\text{Si}^+$  was measured in and below the ledge down to 103 km. It showed a rapid decrease below this height. A comparison of the isotopic ratio for nickel,  $[\text{}^{60}\text{Ni}]/[\text{}^{58}\text{Ni}]$ , with the measured ions shows too large a ratio for the ions below 100 km, suggesting contribution to  $60^+$  from other sources. A prime candidate is  $\text{SiO}_2^+$ , since  $\text{Si}^+$  is observed to decrease in the region where the ratio becomes too large. No evidence for  $\text{SiO}^+$  ( $44^+$ ) is present in the data, indicating that  $\text{Si}^+$  forms  $\text{SiO}_2^+$  either directly or through  $\text{SiO}^+$  in a short lived state. Radiative association is offered as a primary mechanism for  $\text{SiO}_2^+$  production and yields a calculated reaction rate of  $2 \times 10^{-16} \text{ cm}^3 \text{ sec}^{-1}$ .

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## SILICON IONS BELOW 100 KM: A CASE FOR $\text{SiO}_2^+$

by Richard A. Goldberg

Numerous rocket-borne mass spectrometric measurements of ion composition in the upper atmosphere have defined a metallic ion belt between 90 and 95 kilometers as a reasonably permanent feature independent of location and time (e. g. Young et al, 1967; Narcisi 1967; Johannessen and Krankowsky, 1972; Aikin and Goldberg, 1973). Figure 1 illustrates this layer as observed at the daytime equator by Aikin and Goldberg (1973). The data have shown the dominant species within the layer to be  $\text{Fe}^+$  and  $\text{Mg}^+$  with reduced quantities of  $\text{Na}^+$ ,  $\text{Al}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^+$ ,  $\text{Ni}^+$ , and other metal ions. These constituents are believed to be principally of extraterrestrial origin, deposited at these altitudes by meteoric ablation. The remainder must be terrestrial, precipitated by natural and artificial processes. A thorough knowledge of the species' abundances can thereby provide information concerning structure and evolution of certain extraterrestrial bodies such as comets, and simultaneously assist study of the transport processes responsible for bringing trace constituents of terrestrial origin into the upper atmosphere.

Ion-neutral chemistry plays a critical role in analyzing this problem, especially since most of the rocket measurements to date have only provided ion information. Silicon is a semi-metallic constituent that must be important because of its high abundance in meteoric materials (e. g. Mason, 1971) yet as an ion it is distributed in a manner unexplainable in terms of its known chemistry. Above 100 km and primarily within the confines of mid and high latitude sporadic E(Es) layers, metal ions are observed to dominate the ion structure (Istomin, 1963; Young et al, 1967; Narcisi, 1967; Goldberg & Aikin, 1973) and here  $\text{Si}^+$  is usually observed to be an important, if not dominant atomic ion. However, within the permanent metallic ion layer below 95 km,  $\text{Si}^+$  has rarely been observed. The ion-neutral chemistry of silicon must therefore be unique, to permit its existence in a free ionic state above but not below 100 km. This work is concerned with an investigation of this problem, made possible by the study of an unusual metal ion structure recently observed during a meteor shower event.

The region between the permanent metallic ion layer near 95 km and the Es layers above 100 km does not normally exhibit significant metallic ion structure except during specific events such as a meteor shower. An example is that of Goldberg and Aikin (1973), illustrated in Figure 2. These results were obtained during the  $\beta$  Taurids meteor shower on July 3, 1972 at 0743 LMT ( $\chi = 57.1^\circ$ ) with a rocket-borne quadrupole ion mass spectrometer launched from El Arenosillo, Spain. Absolute ion magnitudes were obtained by normalization procedures, using the measured total electron and ion densities determined by

Faraday rotation radio wave propagation and Gerdien probe experiments aboard the same flight.

These data exhibit an Es peak near 114 km. The metal ions dominating this peak are  $\text{Fe}^+$  ( $56^+$ ),  $\text{Si}^+$  ( $28^+$ ), and  $\text{Mg}^+$  ( $24^+$ ). Identification of  $28^+$  as  $\text{N}_2^+$  is not feasible because of the difficulty in explaining the presence of a short lived molecular constituent within the peaked distribution. Here, the only constituents present should be metal atomic ions, since only these possess sufficiently long lifetimes to permit transport mechanisms to redistribute them into the narrow ledge observed. Above the ledge,  $28^+$  takes on the chemical equilibrium distribution predicted for  $\text{N}_2^+$ .

Under the metal ion ledge, metallic ions are observed in abundance to altitudes below 93 km, the region where meteoric ablation is maximum.  $\text{Si}^+$  is observed to rapidly decay below 102 km, and at 93 km is a very minor atomic ion. The high abundance of metallic ions between 95 and 105 km has also been observed by Narcisi et al (1972) during a polar cap absorption (PCA) event and by Zbinden et al (1973) during the Geminid meteor shower. Both results display a similar profile for  $28^+$ .

Figure 3 represents the relative number abundances of the metal ion species when normalized to  $28^+$  as silicon. The results within the layer are compared to those obtained near 101 km. These results are further compared with the neutral relative number abundances measured for the same constituents in chondritic and crustal materials (converted from the weight abundances given in Ahrens, 1965). The agreements obtained demonstrate the validity for identifying  $28^+$  as  $\text{Si}^+$  within and below the metallic ledge, and suggest most species observed to be of extraterrestrial origin.

The rapid depletion of  $\text{Si}^+$  below 100 km and its minor contribution within the primary metallic layer shows the  $\text{Si}^+$  chemistry to be more complicated than that of the more common metallic constituents, e. g.  $\text{Fe}^+$  and  $\text{Mg}^+$ . As noted by Brown (1973) in his recent review on metallic ion chemistry in the upper atmosphere, production processes for  $\text{Si}^+$  are not significantly different than for other metal ions, unless silicon ablates from meteorites in a unique form currently unspecified. If we turn to loss processes for an explanation of the unique  $\text{Si}^+$  distribution, we must consider oxides (e. g.  $\text{SiO}^+$ ,  $\text{SiO}_2^+$ ) since dissociative recombination of these constituents is far more rapid than radiative recombination of  $\text{Si}^+$ .

Information concerning  $\text{SiO}_2^+$  is available in the data of 14.483.  $\text{SiO}_2^+$  ( $60^+$ ) has the same mass as the nickel isotope  $^{60}\text{Ni}^+$ . By comparing the known isotopic ratio of  $^{60}\text{Ni}/[^{58}\text{Ni}]$  to the measured ratio of  $[60^+]/[58^+]$ , as shown in

Figure 4, we find a trend leading to values well in excess of the nickel ratio below 105 km, thereby suggesting a second contribution to  $60^+$  (e. g.  $\text{SiO}_2^+$ ). This is of interest because the  $60^+$  enhancement occurs at the very location where  $\text{Si}^+$  begins to deplete.

The reader is cautioned to note that this result has been extracted from data obtained near the sensitivity limit of the instrument. This also accounts for the absence of data points below 100 km, where  $60^+$  was not clearly discernable. Figure 5 offers a comparative plot for the measured ratios of  $([25^+] + [26^+])/[24^+]$  as referenced to the known relative abundance of the magnesium isotopes. In this case the scatter of measured data is tightly bound to the theoretical line throughout the entire altitude range where measurement was possible. This result is offered to demonstrate the scatter expected, and lend credence to the  $[60^+]/[58^+]$  curve.

If, in fact, the data does provide a measurement of  $\text{SiO}_2^+$ , then we arrive at a magnitude of  $2 \text{ cm}^{-3}$  near 100 km. Swider (1969) has suggested  $\text{SiO}_2^+$  formation by the process  $\text{Si}^+ + \text{O}_2 + \text{N}_2 \rightarrow \text{SiO}_2^+ + \text{N}_2$  with a rate coefficient  $k_1 = 10^{-30} \text{ cm}^6 \text{ sec}^{-1}$ . This would then be followed by dissociative recombination  $\text{SiO}_2^+ + e \rightarrow \text{Si} + \text{O}_2$  with a rate  $k_2 = 3 \times 10^{-7} \text{ cm}^3 \text{ sec}^{-1}$ . An equilibrium value for  $\text{SiO}_2^+$  of  $0.1 \text{ cm}^{-3}$  is calculated for 100 km, suggesting need of a larger value for  $k_1/k_2$  than that used. On the other hand, assuming Swider's numbers to be correct, this low value for  $\text{SiO}_2^+$  implies that either  $60^+$  contains an additional unknown component, or that another  $\text{SiO}_2^+$  source is present. If we postulate this source to be the product of radiative association, i. e.  $\text{Si}^+ + \text{O}_2 \rightarrow \text{SiO}_2^+ + h\nu$ , then we calculate a rate  $k_3$  of  $2 \times 10^{-16} \text{ cm}^3 \text{ sec}^{-1}$  for this process.

The above equilibrium system seems feasible since Fehsenfeld (1969) has found  $\text{SiO}^+ + \text{O} \rightarrow \text{Si}^+ + \text{O}_2$  to be exothermic, implying that  $\text{SiO}^+$  production through the reverse process is not likely. The absence of  $\text{SiO}^+$  ( $44^+$ ) within the metallic ion region observed on this rocket flight, or for that matter, in any other metallic ion data obtained by this observer, appears to concur with the laboratory result.

In conclusion, we have interpreted an excess abundance of  $60^+$  ions measured near 100 km to be caused by the presence of  $\text{SiO}_2^+$ . The results, although somewhat marginal due to the sensitivity restrictions of the detector, appear valid. Furthermore, the unusual metallic ion distribution present on the day of the measurement provided a unique opportunity for the silicon problem to be studied. The results suggest that below 100 km  $\text{Si}^+$  is rapidly depleted by two and three body reactions with  $\text{O}_2$ , forming  $\text{SiO}_2^+$  which then recombines. Further measurements under similar conditions should be conducted to determine the validity of the results suggested here, especially to establish the importance of radiative association, which has a reaction rate too small for laboratory verification.

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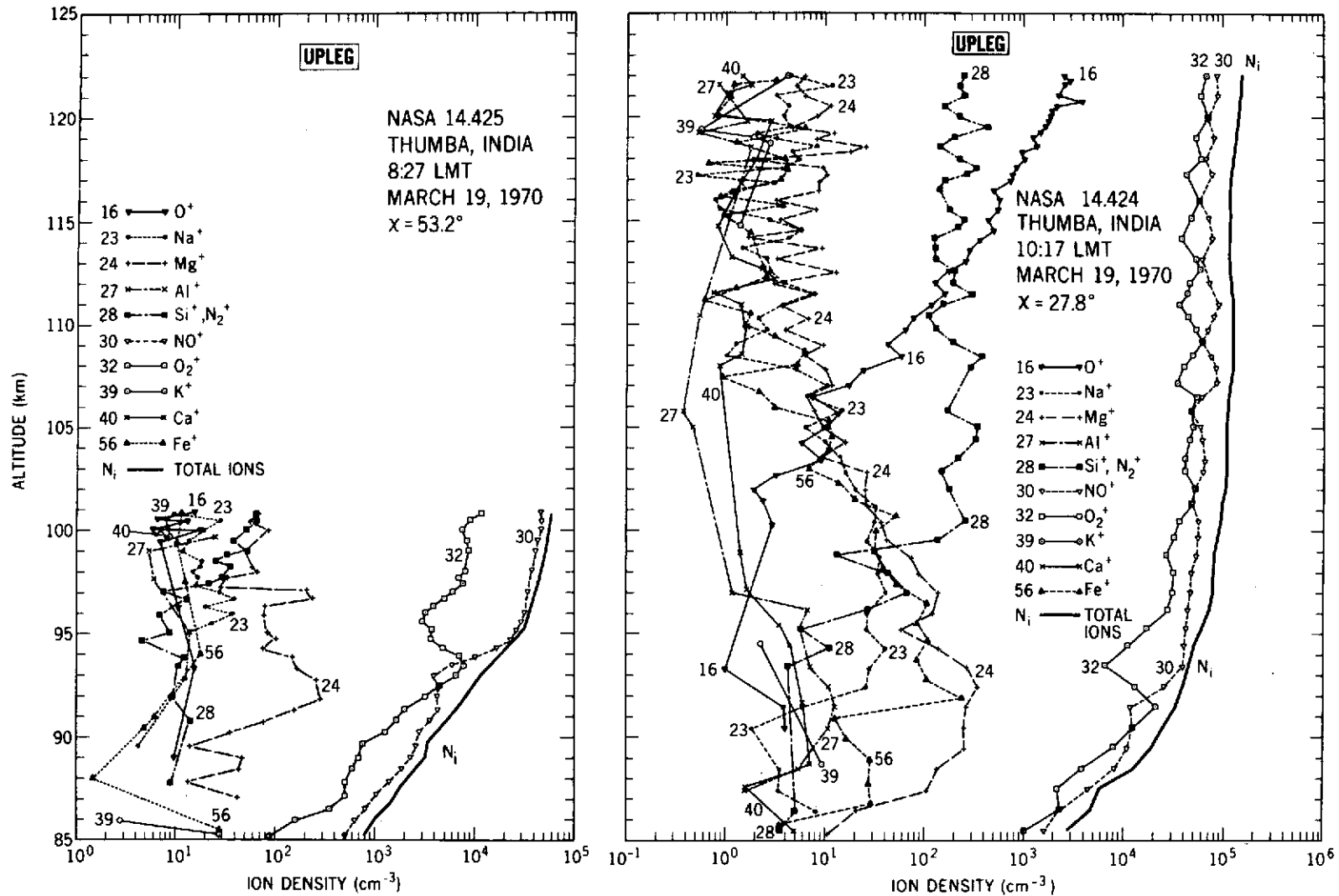


Figure 1. Two observed distributions of positive ion species in the daytime lower thermosphere over Thumba, India on 19 March 1970, illustrating the metallic layer near 93 km.

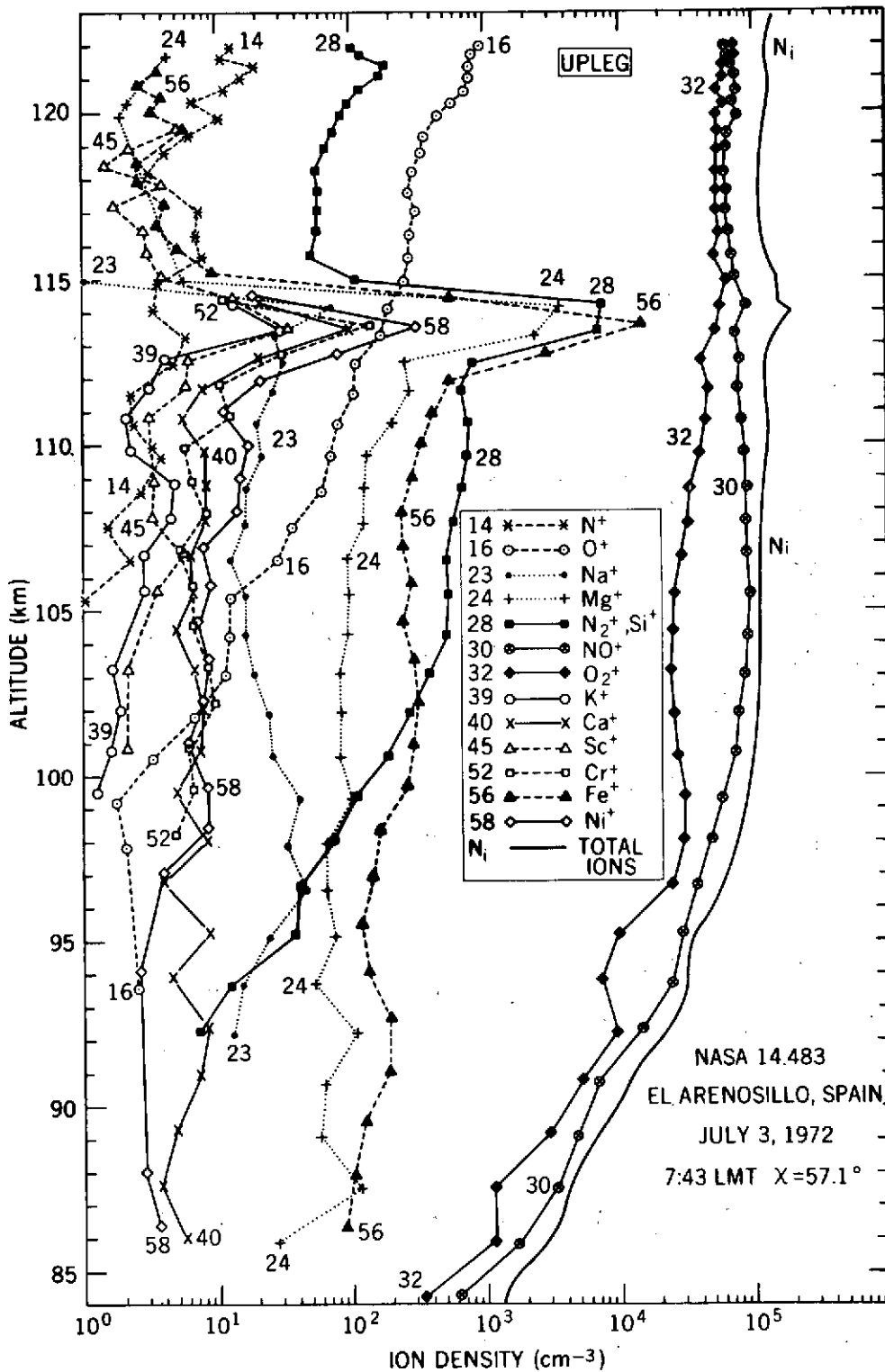


Figure 2. Observed distribution of positive ion species in the lower thermosphere over El Arenosillo, Spain during the  $\beta$  Taurids meteor shower on 3 July 1972.

ELEMENT	ION ABUNDANCES		NEUTRAL ABUNDANCES	
	NASA	14.483	(AHRENS, 1965)	
	101 km	114 km	CHONDRITES	CRUST
Na	12	9.1	4.6	10.4
Mg	49	57	93.5	8.0
Si	100	100	100	100
K	0.81	0.38	0.37	5.3
Ca	3.6	1.2	5.5	10.4
Sc	1.0	0.35	0.003	0.005
Cr	3.4	1.77	0.92	0.02
Fe	149	158	70.9	10.0
Ni	4.2	4.7	3.6	0.13

Figure 3. Relative number abundances of ions (NASA 14.483 rocket data) and neutral elements (chondritic and crustal data) normalized to Si<sup>+</sup>, Si = 100.

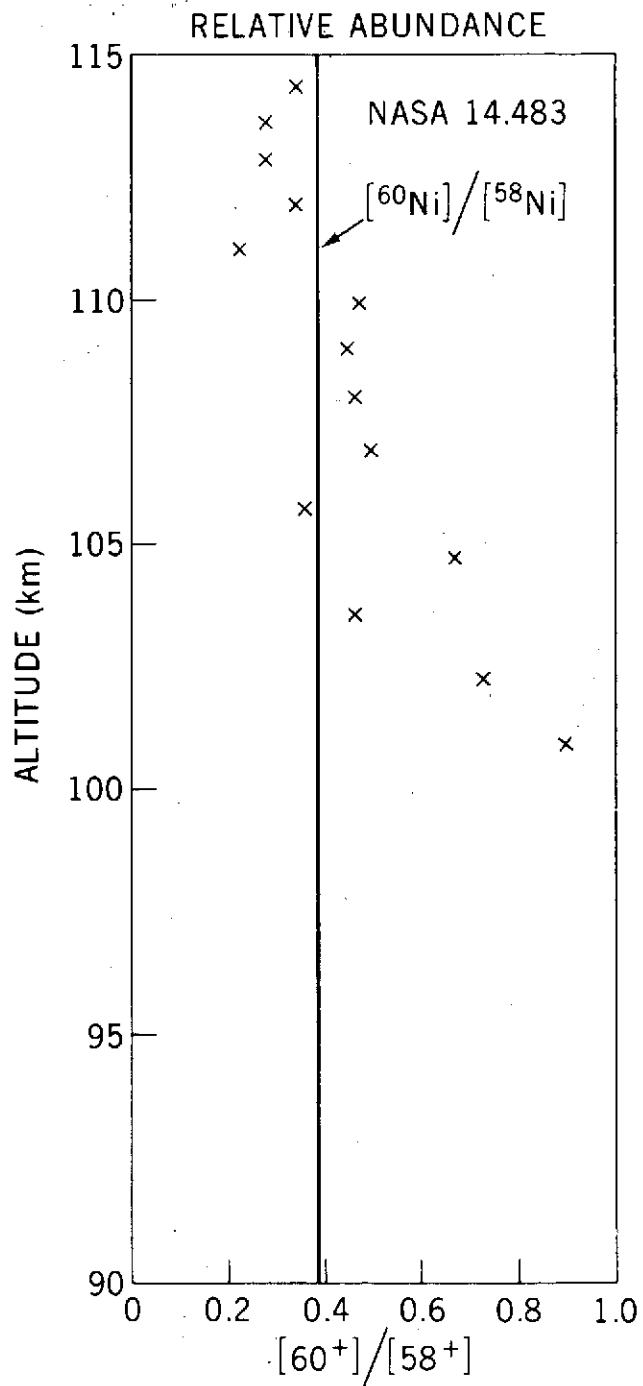


Figure 4. Comparison of measured ratio  $[60^+]/[58^+]$  to the known ratio  $[^{60}\text{Ni}]/[^{58}\text{Ni}]$  for nickel isotopes.

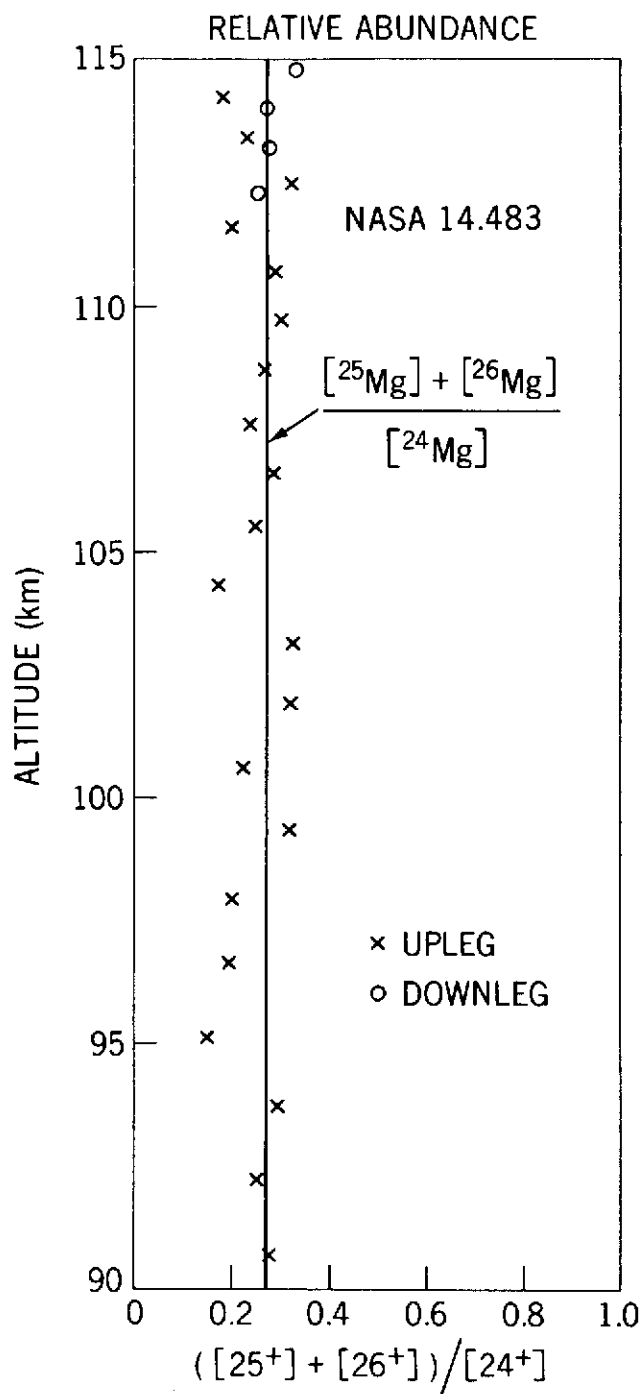


Figure 5. Comparison of the measured ratio  $([^{25+}] + [^{26+}])/[^{24+}]$  to the known ratio  $([^{25}\text{Mg}] + [^{26}\text{Mg}])/[^{24}\text{Mg}]$  for magnesium isotopes.