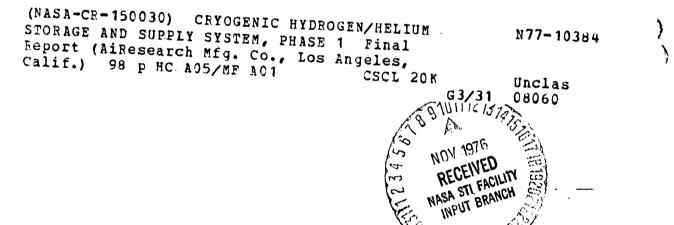
Phase I Final Report

🚭 https://ntrs.nasa.gov/search.jsp?R=19770003442 2020-03-22T12:09:35+00:00Z

# CRYOGENIC HYDROGEN/HELIUM STORAGE AND SUPPLY SYSTEM

75-11607

February 3, 1976



Prepared for

National Aeronautics and Space Administration George C. Marshall Space Flight Center Marshall Space Flight Center, Alabama 35812



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

# FOREWORD

The Cryogenic Hydrogen/Helium Storage and Supply System Program, Phase I, was completed under NASA Contract NAS8-30574, by the AiResearch Manufacturing Company of California, a division of The Garrett Corporation. This work was sponsored by the George C. Marshall Space Flight Center, MSFC, Alabama, and administrated under direction of Mr. Eric Haschal Hyde, EP43.

The Phase I program performance period dates from 15 May 1974 to 21 November 1975. This final report was prepared by Mr. R. H. Norman and Mr. R. D. Raynor, who were also responsible for the analytical and development efforts on the program.



75-11607 Page 111

# ABSTRACT

The Cryogenic Hydrogen/Hellum Storage and Supply System Program, Phase 1, involved refurbishing an existing cryogenic tank, installing microspheres in the tank annulus, and testing the thermal performance of the unit. The performance data was compared with NRC-2 multilayer insulation and low-emittance aluminized surfaces installed in tanks of the same basic design. Phase 11 (pending authorization to proceed) is concerned with developing a He-II (superfluid) Joule-Thomson cryostat to provide fractional watt cooling at 2 K, making use of the present tank for storage and supply of supercritical helium.

The cryogenic tank modified during the program was originally designed for the Manned Orbiting Laboratory (MOL) Program, and subsequently modified by vacuum-depositing aluminum on all annulus surfaces and leaving out the NRC-2 multilayer insulation.

A comparison of performance data from all three tanks shows the heat leak of the low-emittance-annulus tank to be almost 40 percent higher than the multilayer-insulated tank, while the aluminized-microsphere-insulated tank heat leak is higher by a factor of 4.8. That same microsphere-insulated tank measured heat leak is 2.5 times higher than is predicted using some of the published data and current theory, while the measured VCS temperature is 69 R lower than predicted. Other recent data agrees quite well with measured tank performance.

It is concluded that the application of aluminized-microsphere insulation is not yet very predictable for tank design purposes, especially at  $LH_2$ temperature and in the presence of a vapor-cooled shield. It is recommended that this tank design be further evaluated and tested with microsphere insulation to determine the reasons for the poor thermal performance, and to provide better understanding of microsphere thermal behavior and application.



75-11607 Page lv CONTENTS

Section		Page
1	INTRODUCTION AND SUMMARY	1-1
	Introduction	1-1
	Program Overview	1-1
	Summary	1-2
2	EQUIPMENT DESCRIPTION	2-1
	Introduction	2-1
	General Description	2-1
	Detailed Description	2-1
3	PROGRAM TASK NARRATIVE	3-1
	Introduction	3-1
	Program Review	3-1
	Evaluation and Repair	3-2
	Instrumentation Installation	3-6
	Tank Modification	3-6
	Microsphere Insulation	3-11
	Annulus Evacuation and Pressure Test	3-14
	Vented Weight Loss Test	3-14
4	PERFORMANCE AND ANALYSIS	4-1
	Introduction	4-1
	Tank No. 1 (Multilayer Insulation)	4-1
	Tank No. 2A (Low-Emittance Surfaces)	4-3
	Tank No. 2B (Microsphere Insulation)	4-10
	Performance Summary	4-23

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

• ;

75-11607 Page v

. . . . . . . . . .

# CONTENTS (Continued)

Section				Page
5	CONCLUSIONS	AND	RECOMMENDATIONS	5-1
Appendix				
Α	TEST DATA			A-1



75-11607 Page vi

# SECTION 1

# INTRODUCTION AND SUMMARY

# INTRODUCTION

This report describes Phase I of a two-phase program concerned with development of a cryogenic hydrogen/helium storage and supply system for the NASA Marshall Space Flight Center (MSFC). Phase II is to develop a Joule-Thomson cryostat for cooling to 2 K. The Phase I work was conducted for NASA MSFC under NASA Contract NAS8-30574 by the AiResearch Manufacturing Company of California, over the period between 15 May 1974 and 21 November 1975. The issuance of this report concludes Phase I of the program.

The information presented in this report is summarized below.

Section 1, <u>Introduction and Summary</u>, briefly overviews the program activity and summarizes the results of the work accomplished during Phase 1.

Section 2, Equipment Description, describes the physical characteristics of the cryogenic hydrogen/helium tank assembly modified and tested during Phase 1.

Section 3, <u>Program Task Narrative</u>, discusses the six individual work tasks of Phase I and describes the work performed.

Section 4, <u>Performance and Analysis</u>, presents the test data analysis and performance comparison of three different insulation systems in cryogenic tanks of the same basic (MOL) design.

Section 5, <u>Conclusions and Recommendations</u>, gives conclusions based on the performance and analysis of Section 4, and makes recommendations for future work.

# PROGRAM OVERVIEW

The planned work consisted of a six-task effort briefly described below.

# Task 1--Evaluation and Repair

An evaluation of the second MOL tank built under Air Force contract F04611-71-C-0020 was conducted to determine the location of annulus leakage and the repair method that would render the tank useful for the NASA MSFC program. The tank was disassembled, the leak located, repaired, and leak checked. The leak check showed the tank to be satisfactory for use.

> AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

# Task II--Instrumentation Installation

Three sensors were installed in the tank to monitor various temperatures: (1) in the pressure vessel fill port to measure stored liquid temperature; (2) at the pressure vessel vent port to measure the ullage temperature; and (3) on the vapor-cooled shield (VCS) surface near the exit of the vent line.

# Task III--Tank Modification

During Task III, a preliminary analysis of helium delivery to the cryostat (for Phase II) was performed. This analysis included: literature search for helium-II data; performance predictions of the microsphere-insulated tank; analysis of the J-T process for He-II; and test setup considerations. This work led to the design of a separate supply line and new outer tank boss (cryostat adapter). This modification should allow the tank performance required to deliver cold helium to the Phase II J-T cryostat.

#### Task IV--Microsphere Insulation

The required amount of microspheres were ordered from The 3M Company, and inspected for quality. Inspection revealed negligible sphere-breakage, therefore no processing was performed to remove broken spheres. Following a tareweight determination, the microspheres were installed in the tank annulus, and the tank with microspheres installed was weighed.

# Task V--Annulus Evacuation and Pressure Test

During Task V, the tank assembly was processed for 35 days at elevated temperature (210 F) and reduced pressure (1 x  $10^{-4}$  torr) to remove entrapped gases. The ultimate pressure reached during the last 6 days of the bake-out period was 3.5 x  $10^{-4}$  torr.

# Task VI--Vented Weight Loss Test

A thermal performance test (vented weight loss) with liquid hydrogen was conducted at the AiResearch Mint Canyon Remote Test Facility. During testing, thermal acoustic oscillations were observed giving rise to unanticipated experimentation and procedure changes. The test was significantly extended in order to compile enough data from which to select suitable periods for analysis (presented in Section 4).

#### SUMMARY

The thermal performance of the NASA MSFC tank assembly was compared with the thermal performance of the first MOL design configuration (employing multilayer insulation) and the second (employing low-emittance aluminized annulus surfaces, without multilayer insulation), both of which were reported under Air Force contract F04611-71-C-0020.



The original multilayer insulation design exhibited the lowest heat leak; the second configuration exhibited a 40 percent higher heat leak than the first; and the NASA MSFC tank assembly (the third configuration, with aluminized microspheres) exhibited much greater heat leak than the first two, by about a factor of 4.

The predicted performance of the third tank configuration based on recently published theory of microsphere insulation and calorimeter data, did not agree with the test data. Some of the most recent data for aluminized microspheres tended to agree more closely. It was concluded that, at the present time, aluminized-microsphere insulation is not sufficiently predictable for VCS tank design purposes, (especially LH<sub>2</sub> and LHe applications), although more technical information is becoming available.

AiResearch recommends that testing be continued with the NASA MSFC test article to aid in developing improved predictability of the applied thermal performance of aluminized-microsphere insulation.



# SECTION 2

# EQUIPMENT DESCRIPTION

# INTRODUCTION

Equipment modified and tested during Phase I of this contract consisted of a cryogenic hydrogen/helium tank (AiResearch P/N 851240-2-1, modified), mounted in a cryogenic tank handling dolly (AiResearch P/N 956112-1-1). Tank modifications were not formally documented by engineering drawings; engineering sketches were used by manufacturing to perform tank modifications and other work operations.

The cryogenic hydrogen/helium tank assembly was originally designed to store and maintain hydrogen in a supercritical state. In the original design, internal instrumentation consisted of fluid quantity gauging and temperature measurement, with associated electronic signal conditioners (mounted externally). For this application, the instrumentation system was modified to remove the electronic signal conditioning packages, disable the quantity gauging system, and install an additional temperature sensor to monitor the VCS surface temperature. A cross sectional schematic of the tank is shown in Figure 2-1, showing the approximate orientation of the fluid lines and bosses in the normal vent test configuration. The physical characteristics of the tank assembly are given in Table 2-1.

# GENERAL DESCRIPTION

The tank assembly (Figure 2-2) is constructed of two concentric, leaktight spheres. The inner sphere is a pressure vessel which stores approximately 87 lbs of liquid hydrogen, or 150 lbs of liquid helium, in a volume of 20 cu ft. The outer shell provides a vacuum environment for the annular volume which contains a vapor-cooled shield and insulating material. Instrumentation is provided for monitoring temperatures at selected locations.

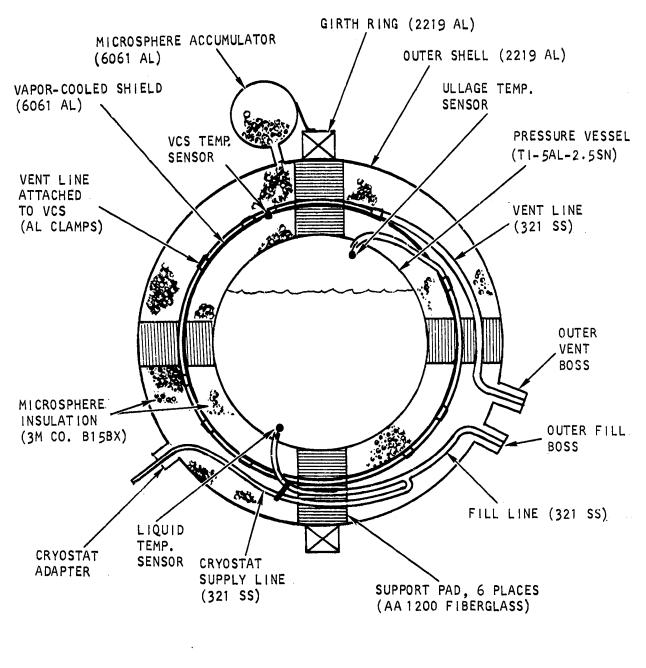
The insulation system consists of an evacuated annulus, a vapor-cooled shield (VCS), and an insulating material consisting of aluminized glass microspheres. The annulus surfaces of the inner and outer shells and the VCS are covered with vacuum-deposited aluminum, over a resin undercoat. Glass microspheres in the evacuated annulus reduce heat conduction and radiation from the outer shell to inner tank wall. During normal venting, fluid exits from the tank through tubing physically and thermally attached to the aluminum VCS. By this method of fluid delivery the VCS is cooled, removing heat from the annulus and improving thermal performance. The inner pressure vessel and the VCS are supported within the outer vessel by compressed fiberglass support pads that have a low thermal conductivity.

# DETAILED DESCRIPTION

#### Inner Pressure Vessel

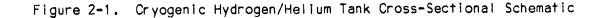
The inner pressure vessel is fabricated from forged titanium 5A1-2.5Sn ELL hemispheres which are finished, machined, and welded together. The wall of

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA



Normal Vent Orientation

**S-**98020





AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

TABL	E	2-1
------	---	-----

# TANK PHYSICAL CHARACTERISTICS

ltem	Design Value
Materials	
Pressure vessel	Titanium 5A1-2.5Sn
Vapor-cooled shield	Aluminum 6061
Outer shell and girth ring	Aluminum 2219
Insulation	Microspheres (~100µ)
Support pads	Fiberglass AA-1200
Dimensions (nominal), in.	
Pressure vessel ID	40.445
Wall thickness	0.045
Vapor-cooled shield ID	41.50
Wall thickness	0.060
Outer shell OD	43.26
Wall thickness	0.055
Volumetric Capacity (nominal), cu. ft.	
At room temperature	20.05
At LH <sub>2</sub> normal boiling point	19.93
Weights (nominal), Ib	
Dry weight	160
Liquid hydrogen	87
Liquid helium	150

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

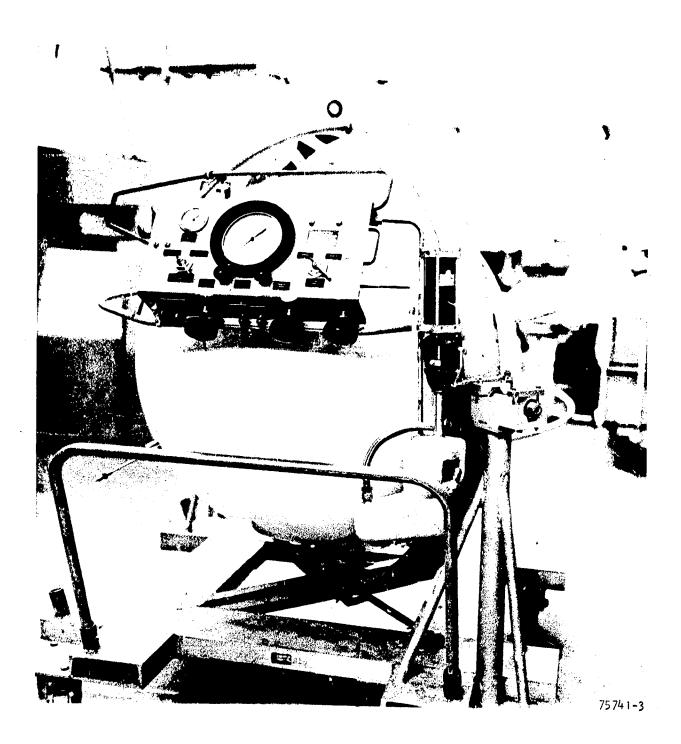


Figure 2-2. Cryogenic Hydrogen/Helium Tank Assembly (View 1) -----



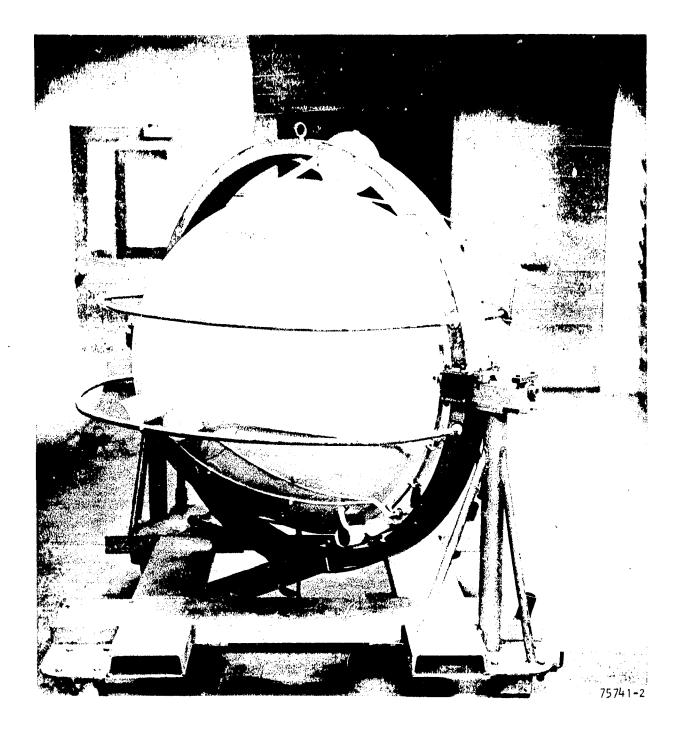


Figure 2-2. Cryogenic Hydrogen/Helium Tank Assembly (View 2)



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

this vessel is designed to withstand both the internal pressure loads and local buckling loads imposed by the compressive support pads. Nominally, the shell is 0.045 in. thick; this wall thickness is designed for the burst pressure of 640 psi (2 times the maximum operating pressure of 320 psi) at 200 R. The wall thickness of the region under the support pads is increased to 0.080 in. to withstand the loads of the supports. The diameter of the thicker wall region is 9.00 in. which is 3.0 in. larger than the diameter of the support pads.

# Access Lines for Fill, Vent, Fluid Delivery, and Instrumentation

The access lines (both fluid and instrumentation) to the interior of the inner vessel are routed through the annular space between the inner and outer shells in such a manner that the heat conducted through these lines is minimized. All of the fluid and instrumentation paths are combined into three composite lines: the fill line, the vent (VCS) line and the fluid delivery line to the cryostat. The 3/8 In. fill line (approximately 42 in. long) is attached to the bottom of the inner vessel and contains leads for the stored liquid temperature sensor. The 3/8 in. vent line is attached to the top of the inner vessel, and inside this line are leads for two temperature sensors. The vent line (approximately 16 ft long) is used as the delivery line and is attached to the vapor-cooled shield over most of its length. The outermost 2 feet, or so, of this line is enlarged to 7/16 in. to accommodate the temperature sensor attached to the VCS. This sensor is inserted into a separate 5/32 in. sheath extending about 8 in. from a tee joining the 7/16 in. and 3/8 in. tubing. The ullage temperature sensor sheath extends along the full length of the VCS/vent line terminating inside the pressure vessel at the inner vent boss. The 3/16 in. diameter (0.142 in. ID) cryostat delivery line tees off the fill line about 16 in. from the outer fill boss. This section of line, about 48 in. long, its tee and outer boss (cryostat adapter), represent the major design modification (other than addition of microspheres) to the previously existing MOL tank (P/N 851240-2-1). The reasons for the modifications are discussed in Section 3.

All the annular lines are made of 300 series stainless steel tubing. The dissimilar material joints near the inner vessel and the outer shell are made with sections of coextruded tubes (except for the fluid delivery line to the cryostat): titanium-to-stainless steel coextrusions at the inner vessel, and aluminum-to-stainless steel coextrusions at the outer shell.

#### Cryostat Adapter

The cryostat adapter is a stainless steel boss on the outer shell of the tank assembly which provides for fluid delivery to the cryostat (Phase 11). The boss is located on the outer shell approximately 155 degrees from the existing fill and vent ports. The adapter connects to the inner tank through a 48-in. segment of 3/16 in. tubing which tees into the fill line. The adapter provides for fluid delivery to the cryostat with minimum heat leak.

The 3/16 in. delivery tube connects to a new tee, designed to be assembled around and welded onto the existing fill tube without separating the fill tube from the inner vessel. The completed assembly forms a tee about 16 in. from the outer shell fill boss and about 26 in. from the inner fill boss.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

# Insulation and Intershell Support System

A 6061 aluminum alloy vapor-cooled shield is located in the 1.3 in. annular insulation space, about 0.5 in. from the inner vessel. The delivery line (vent line) is attached to this shield over most of its length with 29 aluminum U-shaped clamps, 1-3/8 in. wide, each with 6 rivets. The fluid leaving the tank cools this shield and most of the heat being transferred into the insulation from the outer shell will be intercepted before it reaches the inner vessel, which contains the cryogen. Hemispherically-aluminized, hollow, glass microspheres, type B15BX, manufactured by The 3M Company, fill the annular space between the inner and outer shells and serve as annulus insulation.

Six equally-spaced compressed fiberglass pads are used to support the inner vessel and the vapor-cooled shield within the outer shell. The pads are approximately 6.0 in. In diameter and provide a total support area of 170 sq. in. A 6.64-in. diameter hole is cut in the VCS for each pad stack, which contains a contact disc (of 7075 aluminum) located at the same distance from the inner vessel as the VCS when the pads are finally compressed. The contact discs serve to provide a near isothermal extension of the VCS through the fiberglass pads thus intercepting most of the prd heat leak from the outer shell.

# Outer Shell

The outer shell of the tank subassembly is made of formed 2219 aluminum hemispheres. The design of the shell is based on resisting the buckling load exerted by the atmospheric pressure on the evacuated outer shell. The tank assembly is supported by a forged and machined girth ring which forms an integral part of the outer shell. Three mounting points are provided on the girth ring to support the tank assembly in the handling dolly. The attachment points are located 90 degees apart, and with the tank in the normal position (the plane of the girth ring vertical), two mounting points are in a horizontal plane and one at the uppermost point of the ring. The girth ring also serves as a mounting base for the external manifold assembly and microsphere accumulator.

# Microsphere Accumulator

The microsphere accumulator is a small vessel mounted on the outer shell and opening directly into the tank annulus; it functions to compensate for an increase in annular volume when the inner vessel is filled with liquid helium. The accumulator is a spherical vessel, 8 in. in diameter (256 cu. in. capacity), and fabricated of aluminum 6061-T6. The sphere is welded directly to the outer shell and supported with two struts welded to the tank assembly girth ring (Figure 2-2, Views 1 and 2).

# External Manifold Assembly

The external manifold assembly (Figure 2-2, View 2, bottom of tank) consists of a filter subassembly, a vacuum ion pump, and a manual shutoff valve, and is mounted on the girth ring. The manifold assembly functions to enable maintenance and monitoring of the vacuum level in the tank annulus.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

The manual shutoff value allows for connection of a vacuum source directly to the annulus so that any gas within the annulus can be removed. The filter assembly prevents aluminized microspheres from being drawn from the annulus by the pump, or the vacuum ion pump. The vacuum ionization pump is discussed below.

# Vacuum Ionization Pump

The vacuum ionization pump is included with the cryogenic tank as part of the external manifold assembly to maintain tank annulus vacuum at low pressure, to assist in maintaining thermal integrity of the tank assembly. Pump operation can be monitored to assess the quality of annulus vacuum. The pump current is proportional to pressure, and by monitoring the current, an adequate determination can be made of the annulus vacuum level.

The actual operating life of the vacuum ionization pump is dependent on the average pressure of the annulus vacuum, with life expectancy directly proportional to the operating pressure of the pump. Using this vacuum ionization pump, the annulus will attain steady-state pressures of approximately  $1 \times 10^{-7}$ torr (if sufficient outgassing has occurred). At this pressure the expected life of the pump is greater than 100,000 hours.

# Barostat Regulator

A government-furnished National Bureau of Standards (NBS) Barostat Regulator, P/N D-4828, was used during the thermal performance testing to regulate the tank ullage pressure to very near atmospheric pressure.

# Tank Handling Dolly

The tank handling dolly (Figure 2-2, Views 1 and 2) is used to hold the cryogenic hydrogen/helium tank assembly during testing and while being transported.

The tank handling dolly is comprised of two major components: the dolly frame and the yoke. The dolly frame is an open structure consisting of two braced upright supporting members welded to an open square frame. Fork lift provisions are built in as part of the frame. Four swiveling casters and a handle provide for mobility. Two casters have separate foot-operated brakes. The yoke is designed to support a specific cryogenic tank. The yoke consists of two bolt-on channels, a lifting eyebolt on the top channel, a position securing eyebolt on the bottom channel, and a control panel with valves, pressure gages, regulators, and pneumatic fittings.

Additional tubing (Figure 2-2, View 2) was welded to the yoke of the tank handling dolly, to serve as additional protection for the tank during handling.

The overall dimensions of the hydrogen handling dolly (i.e., frame and yoke) are: 59 in. long by 55 in. wide by 69 in. high.

CARRETT

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

# SECTION 3

# PROGRAM TASK NARRATIVE

# INTRODUCTION

Phase I of the cryogenic hydrogen/helium storage and supply system program was conducted during the period between May 15, 1974 and November 21, 1975. The objectives of Phase I consisted of refurbishing an existing cryogenic tank, installing microspheres in the tank annulus to augment the insulation system, and testing the completed unit. The activities involved with accomplishing these objectives are listed below as Phase I work tasks:

Task I--Evaluation and Repair

Task 11--Instrumentation Installation

Task III--Tank Modification

Task IV--Microsphere Insulation

Task V--Annulus Evacuation and Pressure Test

Task VI--Vented Weight Loss Test

Following a Program Review, the work activities are described in chronological order of occurrence.

PROGRAM REVIEW

## Program Scope

The NASA MSFC program, conducted under Contract NAS8-30574, is concerned with developing a system for cooling a fractional watt load to a temperature in the region of 2 K. When complete, the cooling system will consist of a cryogenic storage and supply vessel, a Joule-Thomson expansion device, and a counterflow heat exchanger. The storage and supply vessel will hold cryogenic supercritical helium for delivery to the heat exchanger. The Joule-Thomson expansion device/heat exchanger will provide cooling to a specific load in the Helium-II temperature range.

Development of the system will be conducted in two separate and distinct phases. Phase I (the subject of this report) was concerned with preparing the cryogenic fluid storage and supply system. Phase II, if approved by MSFC, will involve development of the Joule-Thomson cryostat.

# Historical Background Information

The origin of the cryogenic tank hardware used for this program is of interest and will be discussed below. The discussion traces the design from



75-11607 Page 3-1 \_ the original Air Force Manned Orbiting Laboratory (MOL) Program, through the Air Force Rocket Propulsion Laboratory programs (Edwards Air Force Base), and finally to the NASA MSFC program.

# 1. Air Force Manned Orbiting Laboratory Program

AiResearch, under contract to Douglas Aircraft Corp., was developing cryogenic tankage for use in the Manned Orbiting Laboratory atmosphere and reactants supply subsystem. Although the program was terminated for the convenience of the government, the cryogenic tankage design was complete, and materials had been ordered. As a result of this program, sufficient materials were available to construct several sets of cryogenic tanks.

# 2. Air Force Rocket Propulsion Laboratory Evaluation Programs

Two cryogenic hydrogen tankage systems, utilizing materials from the original MOL program, were manufactured by AiResearch for the Air Force Rocket Propulsion Laboratory, Edwards Air Force Base, California, under contract F04611-71-C-0020. These were intended for ground test evaluation of thermal insulation systems. The first test article was nearly identical to the hard-ware configuration planned for the MOL program, with an insulation system of multilayer aluminized Mylar (NRC-2) and a vapor-cooled shield. Certain structural changes were made to reduce cost and make the unit more suitable for ground handling. This first unit exhibited high thermal performance, verifying original design predictions, prior to delivery.

The second unit was produced with a different thermal insulation system. In this unit, the NRC-2 multilayer insulation was eliminated, and all annulus surfaces were covered with vacuum-deposited aluminum over a resin undercoat. This coating was intended to reduce heat losses caused by radiated heat entering the fluid storage vessel. During final testing two failures occurred: (1). a leak into the tank annulus degraded tank performance, and (2) a short circuit was observed in the quantity gauging circuits. Due to lack of funding, the AFRPL program was terminated at this point.

# 3. NASA MSFC Cryogenic Hydrogen/Helium Storage and Supply System

With the second Air Force MOL tank as government-furnished equipment, NASA MSFC contracted with AiResearch to repair the unit to use for storage and supply of cryogenic hydrogen and helium. The cryogenic fluid was to be stored for delivery to a Joule-Thomson cryostat designed to provide cooling for a fractional watt load at 2 K. The contract became effective 15 May 1974.

# EVALUATION AND REPAIR

Task I involved the evaluation of the cause of leakage, and its method of repair, with consideration given to incorporating design features to permit installation of microspheres in the tank annulus. The cryogenic tank and its associated drawings were reviewed by engineering and manufacturing personnel for the purpose of establishing manufacturing operations and tooling (MOT) instructions. MOT's provide the detailed step-by-step manufacturing uses to perform

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

work. In this case the MOT's detailed the exact procedures to be followed in disassembling the cryogenic tank for repair.

# Tank Disassembly - Outer Shells Removed

Prior to installing the tank assembly in the disassembly fixture, all external components and associated tubing were removed and stored for future use. After the tank was placed in the disassembly fixture, the outer shells were separated at the girth weld and removed, leaving the inner tank and VCS within the fixture (Figure 3-1). Comparison of the condition of the vacuumdeposited aluminum with photographs showing its condition just prior to assembly in mid-1972, indicated no surface degradation. Visual inspection of the exposed inner tank revealed no obvious defective areas that would have contributed the vacuum degradation.

The tank assembly was attached to a helium mass spectrometer leak detector and evacuated to a pressure of  $10^{-6}$  torr. The leak was located at the entrance of the vent tube into the pressure vessel (Figure 3-2). A fluorescent-penetrant test revealed a series of small cracks in a weld area that had been repaired during fabrication of the pressure vessel.

# Rework Activity

After the defective area was reviewed by engineering and manufacturing personnel, rework instructions were prepared. The rework consisted of the following activities:

- (a) The old weld material in the crack area was routed out to the parent boss material, but not into the boss.
- (b) A fluorescent-penetrant test was performed to verify removal of the cracks.
- (c) The area was cleaned with water and a solvent compatible with titanium.
- (d) During welding, the area to be reworked was bagged so that an Argon purge provided an inert atmosphere to ensure a weld free of contamination.

After completing the rework, a helium mass spectrometer check was performed on the complete inner tank assembly. The unit was surrounded with a plastic bag and the bag pressurized with helium. Leakage was monitored for a period of 60 minutes with no indication of leakage observed. Thus the leak rate was less than  $2 \times 10^{-10}$  scc per sec of gaseous helium, the lowest measurement limit of the helium mass spectrometer.

After completing the leak test, the repaired area was subjected to another fluorescent-penetrant test. This test verified that the weld repair was free of cracks.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

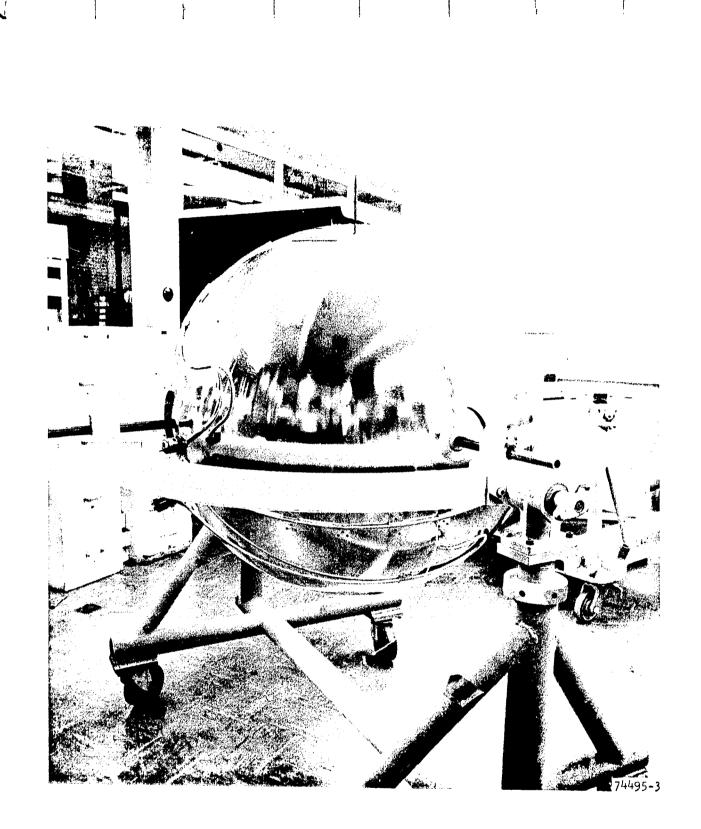
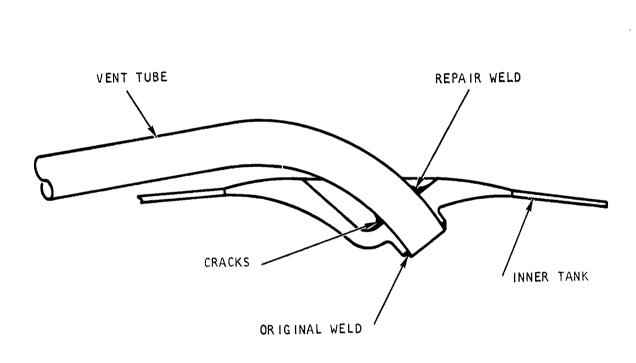


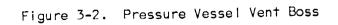
Figure 3-1. Inner Fark and VCP After Removal of Outer Shell

# ORIGINAL PAGE IS OF POOR QUALITY





s-88325



į

7

# Temperature Shock and Proof Pressure Tests

Following helium leak test, the inner tank assembly was transported to the AiResearch Mint Canyon Remote Test Facility for the temperature shock and proof pressure tests.\* At the test facility, the tank assembly was repositioned in the fixture such that the inner vent boss was located at the bottom of the tank. Thermocouples were attached to the tank in four locations. Liquid nitrogen was then fed into the tank through the vent port. Due to the difficulty of insulating the tank for this test, the minimum temperature obtainable at the inner vent boss was -150 F. The tank assembly was then warmed to 60 F and a repeat of the above test was performed. Following the temperature shock test, the tank was pressurized with precooled nitrogen gas at a rate of less the 30 psig per minute until a proof pressure of 300 psig was attained, and average temperature of the thermocouples stabilized at 40 F. Figure 3-3 shows the equivalent proof pressure levels at various combinations of pressure and temperature for this tank. Visual examination of the tank assembly revealed no evidence of deformation or permanent damage. Following the proof pressure test, the inner tank was again subjected to a helium leak test. As there was no indication of leakage, this completed the activities of Task 1.

# INSTRUMENTATION INSTALLATION

Three temperature sensors consisting of a platinum resistance element wound on an aluminum oxide mandrel were installed in the tank. The leads of the elements are insulated and encased in a flexible sheath, then inserted into dry wells inside the vessel through tubes located in the fill and vent lines. Two of the sensors are located at the inner vessel, one at the fill port to measure the liquid temperature, and the other is at the vent port to monitor the ullage temperature. The third sensor is located on the vaporcooled shield (VCS) near the exit of the vent line.

Temperature sensor calibration data from the instrumentation vendor, Rosemount Engineering Company, is given in Table 3-1.

# TANK MODIFICATION

#### Phase || Préliminary Analysis

During this portion of the program the proposed method of helium delivery to a J-T cryostat was evaluated. Investigation of this problem at this time instead of in Phase II of the program, appeared desirable in that modifications to the tank could be planned to facilitate installation of the cryostat during Phase II. The analytical work consisted of the following four items.

\*Test data is given in Appendix A.

OF CALIFORNIA

AIRESEARCH MANUFACTURING COMPANY

**-280** s-89262 -240 -200 -160 -120 0 –40 –80 TEMPERATURE, <sup>O</sup>F 0 +20 <del>1</del>60 +100 +140 270 +180 300 400 350 450 PROOF PRESSURE, PS1

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA 75-11607 Page 3-7

Figure 3-3.

Proof Pressure Versus Temperature - MOL Hydrogen Inner Tank Assembly

TABLE	3-1
-------	-----

TEMPERATURE SENSOR CALIBRATION DATA

LOCATED ON VAPOR- COOLED SHIELD	ROSEMOUNT ENGINEERING COMPANY TEST REPORT         OUALITY CONTROL APPROVED       IPTS 1948         MODEL 1460P86         SERIAL 6383         DATE 07 25 73         ACTUAL CALIBRATION POINTS         TEMP         RESISTANCE OHMS         500.36000         -452.0992         .57860         -320.7470         94.98100         FSC IS         19.58415600         0RM IS
LOCATED AT INNER FILL BOSS	ROSEMOUNT. ENGINEERING COMPANY TEST REPORT         QUALITY_CONTROL APPROVED       IPTS 1948         MODEL 1460P86         SERIAL 6382         DATE 07 25 73         ACTUAL CALIBRATION POINTS         TEMP         RESISTANCE OHMS         500.60000         -452.0992         -320.7470         FSC IS       19.59408500         DRM IS
LOCATED AT INNER VENT BOSS	ROSEMOUNT ENGINEERING COMPANY TEST REPORT QUALITY CONTROL APPEDOVED       IPTS 1948         MODEL 146DP86 SERIAL 6384 DATE 07 25 73



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

ORIGINAL PAGE IS OF POOR QUALITY

#### Literature Search for Helium-II Data 1.

To examine the Joule-Thomson (J-T) process in sufficient deta. to define the tank requirements necessary to produce superfluid helium (He-II), it was necessary to conduct a literature search for pressure-enthalpy data which included the two-phase region below the lower lambda point (coexisting He-I vapor and liquid He-II). No such pressure-enthalpy diagram could be found in the Helium literature. The analysis was performed by using the latest NBS publication of helium thermophysical properties (McCarty, NBS-TN-622, 1972) above 1 psia and older published data on heat capacity, latent heat, and vapor pressure of liquid He-II, (Cook, Argon, Helium, and The Rare Gases, Vol. I, Interscience Publishers, New York, 1961).

#### 2. Performance Prediction of Microsphere-Insulated Helium Tank

To determine the best method of helium delivery to the cryostat, it was nécessary to establish the baseline tank performance (to be derived later from Phase I test results) for supercritical helium operation of the present tank with the addition of microsphere insulation. Using thermal conductivity data for hemispherically-aluminized, 15 to 135 micron diameter glass spheres. (from Adv. Cryo. Engrg, Vol. 18, 1973, p. 103) in conjunction with the AiResearch Cryogenic Tank Design and Performance Prediction computer program, \_\_\_\_ the isobaric delivery rate of helium at 50 and 100 psia was determined as a function of helium storage temperature, for two modes of operation: (1) all flow through the vapor-cooled shield (VCS) vent circuit, and (2) all flow through the fill line, not shorted to the VCS which becomes a passive radiation shield. Over the probable temperature range of interest, 5 to 20 R, the vent rate through the VCS averaged about 0.5 to 0.6 lb per hr, while for flow through the fill line (no VCS cooling) about 2.5 to 4 lb per hr resulted. These vent rates were computed for steady-state, constant-pressure operation with the outer shell temperature at 70 F.

# 3. Analysis of the J-T Process for He-11

OF CALIFORNIA

The J-T process for liquefying helium below the lambda point (0.73 psia, 3.92 R) was analyzed to determine the worst-case tank conditions (maximum temperature at minimum pressure) for which expansion into the He-II region could be achieved. Several important assumptions had to be made in order to begin the study, such as dewar (liquid collector) operating point, heat exchanger high-pressure-side exit temperature, and He-II cooling load (taken to be 1 watt, for reference). Tank pressures (inlet to J-T heat exchanger) of 50 and 100 psia were studied, together with a 1.0 R approach at the cold-end of the heat exchanger (that is, J-T expander inlet temperature of 4.0 R, with He-II dewar operating at 3.0 R, 0.142 psia). The resulting flow rates were on the order of 0.5 lb per hr. The maximum inlet temperature to the J-T heat exchanger (prior to counterflow cooling to the expansion inlet condition, 4 R) was found to vary from 10.5 to 15.0 R, for 50 to 100 psia tank pressure. Somewhat lower temperatures would be required in the tank to allow for cryostat supply line heat leak. A very rough estimation of J-T heat exchanger size was made for the 50 psia, 10.5 R inlet condition. The low-pressure side characteristics (minimal allowable pressure drop and poor heat transfer coefficient) require

ARESEARCH MANUFACTURING COMPANY

significantly larger finned tubing than that, for example, in current use for  $LN_2$  cryostats built at AiResearch. Detailed sizing and design will be performed in Phase II.

# 4. <u>Test Setup and Tank Operating Considerations</u>

The most critical parameter in the operation of a He-II cryostat from the supercritical helium tank is the stored helium temperature, which increases as the tank is depleted. Since the J-T heat exchanger inlet must be very cold to minimize HX size and make liquefaction possible, the transfer losses (heat leak) between the tank and cryostat must be very small. From the J-T process analyzed above and the predicted tank thermal performance, a reasonable operating time of 10 to 20 hrs requires that the temperature rise from tank to HX inlet be held to less than 0.3 R. With a flow of about 3 lb per hr from the tank, this corresponds to a transfer loss of 2 Btu per hr. This level of performance can be achieved if all the tank flow is delivered through a special boss, after which the cryostat flow is split off, and the remaining flow is routed over the boss to provide vapor cooling. In this configuration the tank can deliver about 3 or 4 lb per hr of helium and the pressure will remain constant at 100 or 50 psia due to heat leak, provided that no flow is allowed over the VCS. This special boss and a small line connecting it to the present fill line, near the inner vessel, was recommended as the simplest design modification to the existing tank which could ensure delivery of cold helium to the cryostat.

For Phase I testing with liquid hydrogen (vented-weight-loss test), the effect of this extra line and the removal of the thermal short from the fill line to the VCS, was estimated to be less than a 0.5 percent increase in vent rate.

# Cryostat Adapter Design

Using results obtained from the analytical work above, a new tank boss and a "tee" were designed and fabricated. The external boss which is the cryostat adapter, was located on the outer shell approximately 155 degrees from the existing fill and vent ports. The adapter allowed a 3/16 in. diameter supply tube to exit from the fill line, minimizing the heat-leak to the pressure vessel.

A special tee was designed to be assembled around and welded onto the existing fill tube without separating the tube from the pressure vessel. A small hole was machined in the fill tube, about 16 in. from the existing outer fill boss and the tee then welded to the tube. Approximately 48 in. of the 3/16 in. OD tubing (measured ID of 0.142 in.) was required to connect the tee to the external adapter.

Modification of the outer shell was required for the addition of the microsphere fill port and the cryostat supply tube. In addition, a cylindrical 10-micron filter was installed between the annulus evacuation port and the external manifold consisting of an ion-pump and a manual shutoff valve.



# MICROSPHERE INSULATION

# Purchase and inspection

Thirty pounds of hemispherically-aluminized, hollow, glass microspheres, type B15BX, were purchased from the 3M Company Special Enterprises Department. Random sampling inspection indicated that the nominal size was 100 microns, with approximately 30 percent of the spheres without aluminum coating (Figure 3-4). A density check of the spheres was performed to confirm the amount required. The packed density of the spheres was 0.063 g/cc.

The original plan of processing the microspheres to remove broken spheres by alcohol floatation was revised.\* Inspection of random samples revealed negligible breakage. Therefore, no attempt was made to remove broken spheres.

Transfer of the microspheres to the processing containers was accomplished and the containers were then placed in a vacuum oven. The microspheres were evacuated to a pressure of  $1 \times 10^{-4}$  torr at 300 F for 72 hours. At the end of that time the containers were back-filled with 5 psig, filtered, dry, nitrogen gas and stored for future use.

# Microsphere Installation

Prior to installation of the microsphères, the tank assembly was baked out for a period of approximately one week at a temperature of 210 F, while the tank annulus pressure was less than 1 x  $10^{-4}$  torr.

After tank annulus preconditioning was completed, the system and its holding fixture were mounted on a set of scales and positioned so that the microsphere accumulator fill port was located at the top. This accumulator is part of the tank outer shell and sized to correspond to the equivalent increase in tank annulus volume when the tank is filled with liquid helium. At this point, a system tare weight was recorded.

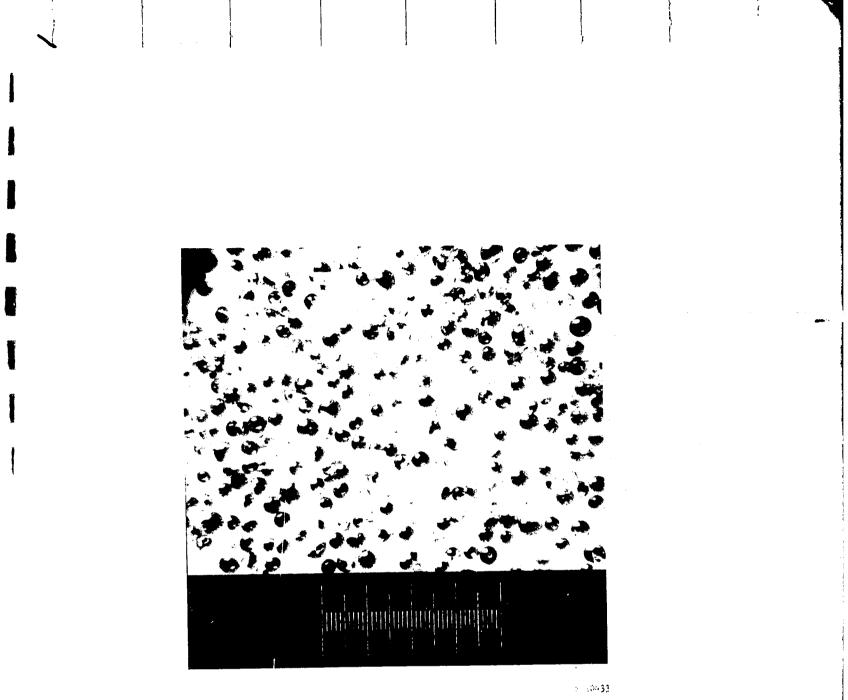
A cannister containing the preconditioned microspheres was suspended above the accumulator fill port and connected to the port with a clear plastic tube (Figure 3-5). By alternately pressurizing the microsphere cannister and evacuating the tank annulus, the microspheres were transferred from the cannister to the tank annulus. The tank holding fixture was vibrated during transfer to assure proper packing of microspheres in the annulus. Numerous holes exist in the VCS (where the lines pass through, and in the vicinity of the inner and outer shell bosses) to enable the microsphere's to fill the annulus under the VCS.

When the predetermined microsphere fill weight was approached, and visual inspection indicated that the annulus was full, the tank assembly was rotated 180 degrees. The tank was vibrated in the inverted position for a short period. After vibration, the tank was rotated back to the upright position and an additional quantity of microspheres installed in the annulus.

\*Discussion with Dr. Peter Howell of 3M Co. indicated that the number of broken spheres is usually negligible.



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA



------

MAGNIFIED APPROXIMATELY 50 TIMES (EACH MICROMETER DIVISION = 25.4 MICRONS)

Figure 3-4. Microscopic Inspection of Microspheres (Typical)



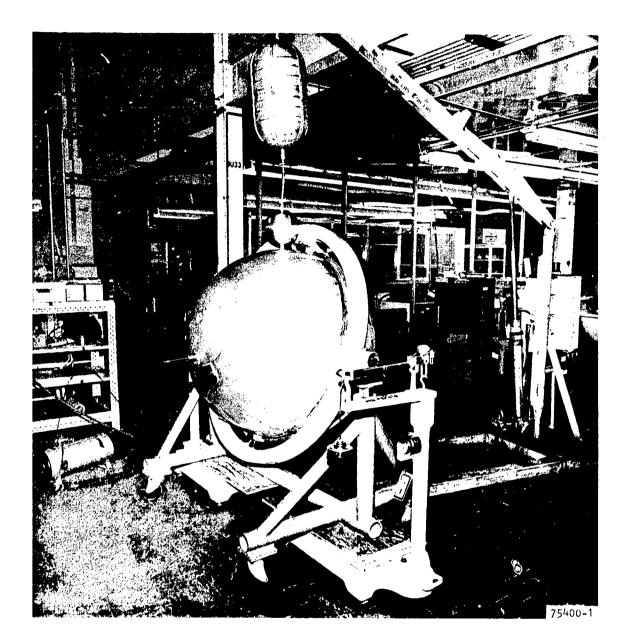


Figure 3-5. Tank Prepared for Microsphere Installation

ORIGINAL PAGE IS OF POOR QUALITY



These operations were repeated until visual inspection indicated that the annulus fill was complete. At the completion of fill, the microsphere weight was 16.88 lb,\* an installed bulk density of 0.068 g/cc.

# ANNULUS EVACUATION AND PRESSURE TEST

When microsphere installation was complete, the fill port was welded closed and a helium leak check performed. This test revealed no leaks.

The tank was installed in an oven and setup for annulus evacuation. It was planned that the annulus be evacuated at a temperature of 210 F until annulus pressure reached a level of less than 1 x  $10^{-4}$  torr. After an evacuation-bakeout period of 10 days, a static annulus pressure test was performed to aid in predicting the length of the evacuation period\*. The test indicated that at least 10 additional days of evacuation at 210 F were required. The evacuation-bakeout period was continued.

At the end of the second 10-day evacuation-bakeout period, a static annulus pressure test was again conducted. This test indicated that there was no appreciable improvement from the continued evacuation bakeout. Investigation of available test data (from previous tests conducted using microspheres as an insulating material) indicated that either a dynamic vacuum approach was utilized, or the annular volume was back filled with CO<sub>2</sub> and allowed to cryopump during the vented weight loss test. As a result, it was proposed that the 30-day static annulus pressure test be run until the pressure builds beyond the upper limit capability of the ion pump (approximately 5 x 10-4 torr). At that point, the pump would be turned on to reduce pressure to  $1 \times 10^{-4}$  torr, or less, and then the test would be continued with the ion pump off.

The evacuation bake-out procedure was concluded after a total of 35 days. During the last 6 days of the test period there was no appreciable improvement noted in the evacuation-system pressure,  $(3.5 \times 10^{-4} \text{ torr})$ , indicating that the ultimate pressure had been reached. A static annulus test substantiated that additional bake-out/evacuation would not improve the condition of the tank annulus.

After stabilizing the tank assembly at ambient temperature and a constant dynamic annulus pressure of  $5 \times 10^{-7}$  torr, the tank was isolated from the pump and the annulus pressure was monitored hourly until a static pressure of  $3 \times 10^{-4}$  torr was reached. The pressure was then monitored twice a day for the remaining 30 days. It was not required to turn on the ion pump, as the annulus pressure did not increase above  $4.6 \times 10^{-4}$  torr during the 30 day test.

# VENTED WEIGHT LOSS TEST

The completed system (Figure 3-6) was transported to the AiResearch Mint Canyon Remote Test Facility and prepared for the liquid hydrogen vented-weightloss (VWL) test. The tank assembly was purged and cooled prior to filling with

\*Test data given in Appendix A.

GARRETT

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

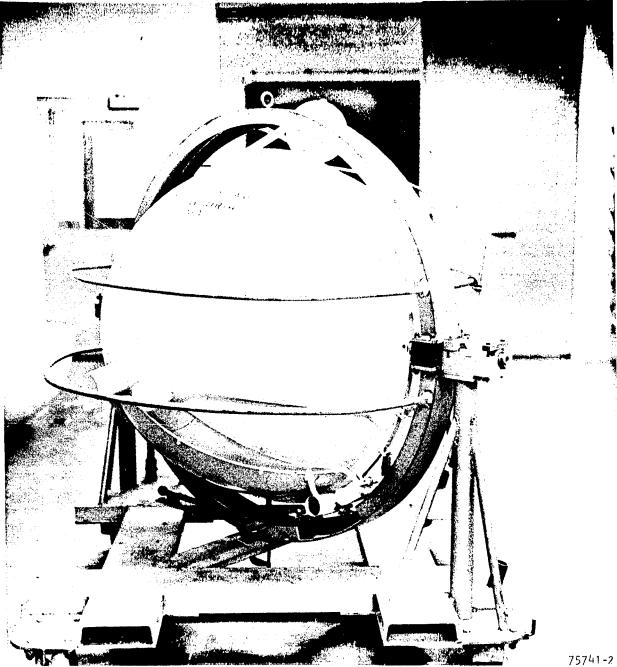


Figure 3-6. Completed Cryogenic Hydrogen/Helium Storage and Supply System

ORIGINAL PAGE IS OF POOR QUALITY



87 ib of liquid hydrogen. After an 18-hour stabilization period the vent port was connected to a wet test meter, through a water saturator. Five days later the NBS barostat and a Baratron pressure readout were installed (Figure 3-7). Two days later Pace dynamic pressure transducers were installed on the fill and vent ports, and an oscillograph was attached. During the WL test the tank annulus pressure, the fill and vent port pressures, the 3 temperature sensors, the scale and wet test meter were monitored and recorded. (Refer to Figure 3-8 for test schematic.) Oscillograph traces were obtained over.

Pressure oscillations were observed on both the fill and vent pressure gages (scale ranges of 0-30 psig and 0-30 in. Hg, respectively) from the start of testing. Prior to installing the barostat, both the fill and vent port oscillation amplitudes (peak-to-peak) averaged about 0.02 atm; no oscillation frequency was established at this time. With the barostat in the vent circuit, the amplitude of the vent pressure oscillation approximately doubled, with no change in the fill pressure oscillation (gage is less sensitive, however). Suspecting thermal acoustic oscillations (TAO) in the dead-ended cryostat supply line which tees off the fill line, the fill and cryostat support ports were externally connected through an orifice (micrometer valve). The gages showed no change in oscillation amplitude. An oscillograph was then attached to a pair of fast response (time constant of 12 milliseconds) Pace dynamic pressure transducers, one in the fill line and the other in the vent line. The external shunt was then reconnected between the fill and vent ports. Examination of the resulting pressure traces 3 différent micrometer valve settings) revealed a steady, constant-amplitute vent pressure oscillation of about 0.006 atm peak-to-peak amplitude, with trequency of 8.5 Hz. The fill line pressure was oscillating at a variable amplitude between 0.017 atm and zero, with a frequency of 12 Hz (the period between peak amplitudes was also variable). This reduction in amplitude is not very significant, as it did not significantly reduce the vented weight loss. A calculation of the TAO effect on tank heat leak, based on recently published data, is included in Section 4.

The second attempt to reduce the TAO involved rotating the tank assembly 90° from its normal vent orientation (referring to Figure 2-1, the tank was rotated 90° counterclockwise), such that both inner bosses were out of the liquid phase, and the outer fill and vent bosses were facing upward. In this orientation the fill pressure gage showed no detectable oscillation, while the vent pressure continued to oscillate at about 0.03 atm. At this point in time, the tank contained about 38 percent liquid (by volume) and the ullage gas temperature was about 100 R. The vent line oscillation was finally eliminated by removing the wet test meter saturator, indicating that the vent line oscillation was not thermally induced (no TAO).

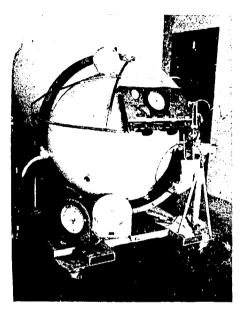
Measuring the tank vent rate in the 90° rotated configuration with a low liquid level was undesirable because of the high degree of ullage stratification. Another attempt to eliminate the TAO in the normal vent configuration was made by inserting a ball within the cryostat supply tube, thus blocking off fluid at the tee on the fill line. With the ball in the supply tube, the tank was refilled and subjected to another vented weight loss test. No change was detected in the oscillations, as compared with original testing. (The ball



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

AIRESEARCH MANUFACTURING COMPANY	ENVIRONMENTAL TEST - PHOTOGRAPH	IC RECORD
851240-1 SERIA	NÔ 1 VE	NTED WEIGHT LOSS
EWO 3400-250338-01-0601	DATE 6-2-75	

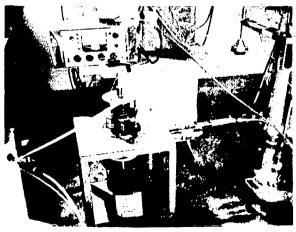




NEW TARE WEIGHT CONFIGURATION

BAROSTAT INSTALLATION

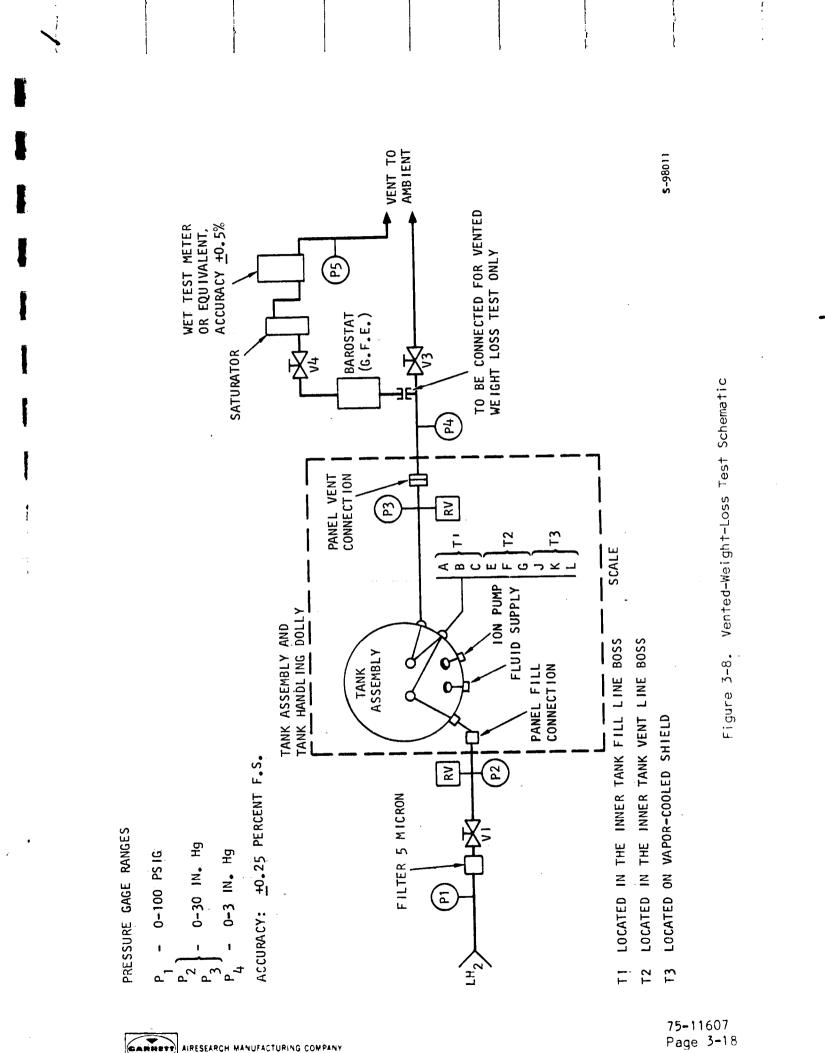
ORIGINAL PAGE IS OF POOR QUALITY



F-21818

Figure 3-7. Vented-Weight-Loss Test Let ge





AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

GARHETT

could not be made to completely block the opening, it was estimated to be about 85 percent plugged, at best.) About half-way through the test the ball was removed and no change in the WL was detected.

For the final WL determination, the tank assembly was inverted with fluid venting out the cryostat supply line, thus bypassing flow around the vaporccoled shield. Referring to Figure 2-1, the tank was rotated approximately 170° clockwise from the position shown. In this orientation the inner fill boss is almost exactly at the top of the pressure vessel. A transducer on the vent line showed a pressure oscillation amplitude of about 0.0020 atm during the first 48 hr. From that time on, the oscillation amplitude dropped to 0.0007 atm, for some unknown reason. However, neither of these magnitudes are very significant, and perhaps are not even due to TAO, since the vent line is shorted to the VCS over most of its 16 ft length.

A suitable amount of data was collected to determine the effectiveness of the annulus insulation both with and without flowing through the VCS.



# SECTION 4

# PERFORMANCE AND ANALYSIS

# INTRODUCTION

This section presents a review of the performance data obtained from the two previously tested tanks, a summary of the present test data analysis, and a performance comparison of all three tanks. The following tank designations will be adopted for brevity in referring to them henceforth:

- Tank No. 1 The original MOL hydrogen tank containing 69 layers of crinkled, single aluminized Mylar multilayer insulation (referred to as NRC-2) and a vapor-cooled shield.
- Tank No. 2A The second tank of the MOL design assembled with all surfaces within the annulus coated with 1500-2000A of 99.99 percent pure aluminum, vacuum-deposited over a resin undercoat, with finish better than 8 RMS; no multilayer insulation
- <u>Tank No. 2B</u> The same tank as No. 2A with the addition of a mixture of uncoated and hemispherically-aluminized microspheres (3M Co.) in the entire annulus.

# TANK NO. 1. (MULTILAYER INSULATION)

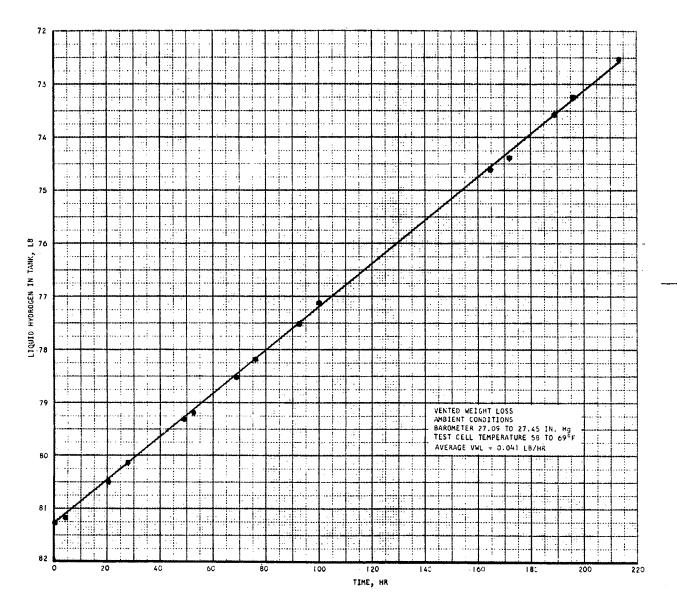
## Measured Performance

The MLI-insulated tank, as designed for the MOL Program, was fabricated and tested under contract to the Air Force Rocket Propulsion Laboratory (Edwards Air Force Base, California), and documented by AiResearch Report No. 72-8359, April 14, 1972. The vented weight loss (WL) test data, using liquid hydrogen, is reproduced in Figure 4-1 (as previously reported on Page 3-17 of Report 72-8359), showing a 213-hour average vent rate of 0.041 lb per hr. At the tank vent pressure of about 13.5-psia, this corresponds to a heat leak to the stored fluid of about 8.0\_Btu per hr.

# Analytical Performance Prediction

The AiResearch Cryogenic Tank Design and Performance Prediction computer program is briefly described as follows. The function of the program is to determine the steady-state flow rate from a specific tank operating at constant pressure in a constant temperature (outer shell) environment. A heat balance is performed between the various constituents (heat paths) of the insulated annulus, the fluid cooling rate of the vapor-cooled shield (VCS) and the isobaric specific heat input of the stored fluid. The two unknowns, flow rate and VCS temperature, are obtained by simultaneous iteration between the thermodynamics of the stored fluid and the heat transfer rates across each annulus (including the VCS cooling rate). The insulation between either shell and the VCS can be vacuum (radiation only), for which an emittance-versustemperature table must be provided, or any insulation (evacuated multilayer,





5-728te

Figure 4-1. LH<sub>2</sub> Vented Weight Loss Test - Tank No. 1

AIRESEARCH MANUFACTURING COMPANY

fiberglas batt, etc.) for which a thermal-conductivity-integral-versustemperature table is provided. Several multilayer insulations are characterized by equations dependent on temperature, uncompressed layer density, and number of layers. The annulus can contain support pads or tension support struts, or both, and provision is made for stainless steel lines and copper electrical leads. The program also calculates shell thicknesses and weights of the tank's major constituents.

The computer printout of the predicted VWL performance was presented on Page 3-25 of Report 72-8359. The insulation thermal conductivity curve used for NRC-2 was that derived from Apollo/Lunar Module (LM) supercritical helium tank test data, representing the thermal performance attained by more than half the tanks in a sample of 15 tanks of identical design. The predicted vent flow rate was 0.0399 lb per hr with a tank pressure of 13.5 psia and outer shell temperature of 70 F. The corresponding heat leak to the stored fluid is about 7.8 Btu per hr. Thus, the VWL-test-measured thermal performance was within 3 percent of the analytically predicted performance.

TANK NO. 2A (LOW-EMITTANCE SURFACES)

#### Measured Performance

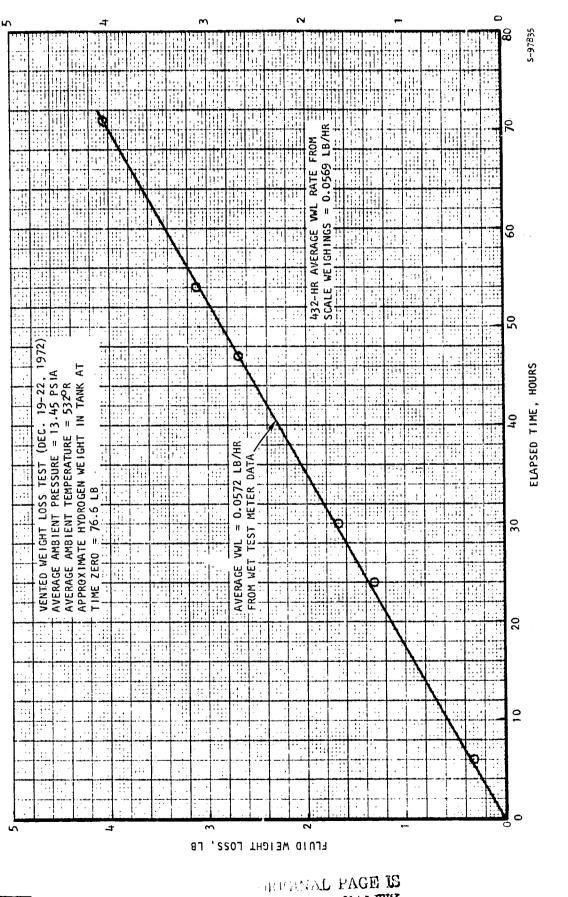
A summary of the vacuum-aluminized, low-emmittance-surface-insulated tank fabrication and testing program was documented by AiResearch Report No. 73-9418, July 25, 1973, under contract to the Air Force Rocket Propulsion Laboratory. The thermal performance test (VWL) data sheets were presented on Pages 3A-12 through 3A-18 of that report. The wet test meter flow rate data are presented here in Figure 4-2, and the various temperature measurements are plotted in Figure 4-3.

The flow rate data, for the 71-hour test period, show an average vent rate of 0.0572 lb per hr. Unfortunately, no scale weight data were taken during this period (due to the removal of the pressure control panel which was leaking) as a check on the flow rate. The Test Log reveals, however, that scale weighings were recorded 96.83 hours before and 264.25 hours after the 71-hour test period, over which time 24.59 lb of hydrögen boiled off. This data yields an average vent rate of  $24.59/432.08 \pm 0.0569$  lb per hr, an excellent corroboration of the wet test meter data. Therefore, the measured vent rate is taken to be 0.0572 lb per hr, corresponding to a heat leak of 11.18 Btu per hr. -

The temperature data appear to be somewhat anomolous, especially the inner fill boss readings. The average fill boss temperature of about 48 R is 12 R above the saturated liquid temperature at the tank pressure of approximately 13.5 psia. This discrepancy is greater than should be attributed to instrument inaccuracy. It is possible that the fill boss temperature sensor tube is not properly positioned at the inner fill boss temperature of 55 R, or about 19 R above saturation, could be due to thermal stratification in the ullage. This is not unreasonable to expect, considering the very low thermal conductivity of the thin titanium-alloy vessel wall, and the presence of at least one, possibly two, fiberglass support pads conducting heat to the hydrogen vapor.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

ARRETT



1 POOR QUALITY

Figure 4-2. LH2 Vented Weight Loss Test - Tank No. 2A

75-11607 Page 4-4

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

CARRETT

310 305 535 530 525 30 33 60 55 45 0<del>1</del> 50 s-97832 φ φ φ Π 20 TEST (DEC 19-22, 1972) 60 Temperature Data From VWL Test - Tank No. 2A Φ Φ WL Ś ¢ ወ ELAPSED TIME, HOURS WET TEST METER **1**<sup>4</sup>0 1 VAPOR-COOLED SHIELD BOSS TEMP. 13.5 PSIA INNER FILL BOSS 30 INNER VENT ወ đ Figure 4-3. ٠ 20 SATIN TEST CELL  $^{\rm H}_{\rm 2}$ 10 0 35 50 45 100 60 55 535 530 525 310 Я<sup>0</sup> ARESEARCH MANUFACTURING COMPANY

The average VCS temperature from Figure 4-3 is taken to be 304 R, and the average ambient (outer shell) temperature 532 R.

### Surface Emittance Determination

An analytical prediction of Tank No. 2A thermal performance cannet be made with a high degree of confidence because the total normal emittance of the aluminized vacuum annulus is not known. The best that can be done ... "a priori" is to assume an emittance versus temperature characterization based on what could be similar surfaces. The difficulty here is that a very wide range of data exists for aluminum surfaces because of the wide range of surface conditions possible. At room temperature, an RMS rating of 33 microinches on aluminum 2024-T4 (92.7 percent Al) can exhibit an emittance of 0.09 (and 0.018 at 10 R)\*, whereas the aluminized side of crinkled, 1/4-mil, aluminized Mylar can be as low as 0.03. Since radiation is the predominant mode of heat transfer in this tank, a variation like that can give a wide range of heat leak (vent rate) results. The problem is further complicated by the range of possible emittance values to be assigned to the VCS, whose temperature is also an unknown (in the computer analysis). Therefore, it is suggested that the most meaningful analysis to be performed here is to back out an "apparent" emittance characteristic from the test data (both vent rate and VCS temperature) and compare it with data from other aluminum surfaces of known emittance.

Two basic assumptions are made in this method of analysis: 1) the heat transfer through either annular space can be separated into radiation and conduction components, where the conduction contributions are known (that is, they can be calculated independently), and 2) the emittance of all aluminized surfaces is a linear function of temperature, or a constant (an overall average, or effective, value). The first assumption is necessary to do any tank performance analysis, the total heat transfer rate being the sum of the individual members' heat leaks. The second assumption is not necessary, but is very convenient because there is an infinite number of possible combinations of individual surface emittances that yield the same emittance interchange factor for a given annulus. In this program, the radiation heat transfer rate and emittance interchange factor are defined as follows:

> $Q_{R} = \sigma \bar{e} A_{1} \left( T_{2}^{4} - T_{1}^{4} \right), \text{ Btu per hr}$ (1)  $\bar{e} = \left[ \frac{1}{e_{1}} + \frac{A_{1}}{A_{2}} \left( \frac{1}{e_{2}} - 1 \right) \right]^{-1}$ (2)

\*K. H. Hawks and W. B. Cottingham, "Total Normal Emittances of Some Real Surfaces at Cryogenic Temperatures," <u>Adv. Cryo. Eng.</u>, Vol. 16, 1971, page 473.



 $\sigma = 1.714 \times 10^{-9} \text{ Btu/hr-ft}^2 - \text{R4}$ 

T = surface temperature, R

A = surface area, f+2

e = surface emittance (relative to black body = 1.0)

and subscripts "1" and "2" refer to the inner and outer surfaces, respectively, bounding a given annulus.

Using the average test data, as presented in Figures 4-2 and 4-3, for ambient temperature (532 R), VCS temperature (304 R), and vent rate (0.0572 Ib per hr); and assuming the actual inner vessel wall temperature to be 36 R, and the VCS effectiveness to be 1.0, the breakdown of heat transfer rates was calculated. The results are presented in Table 4-1. The emittance interchange factors from Equation (1) are then

#### TABLE 4-1

#### Heat Transfer Rate (Btu/Hr) Radiation Fill line Total Supports Annulus Space 3.35 0.56 7.27 11.18 Below VCS 0.22 55.99 59.63 3.42 Above VCS

HEAT LEAK BREAKDOWN CALCULATED FOR TANK NO. 2A FROM TEST DATA

0.0143 and .0125, below and above the VCS, respectively. These emittance results are contrary to the well-known behavior of metallic surfaces, namely, that "emissivity" increases with increase in temperature\*. If a straight-line equation were fit to the above interchange factors, the zero temperature value would be 0.0303 and the 530 R value would be only 0.0222; the equation has a negative-slope temperature dependence:

(3)

The above data analysis depends on an accurate measurement of the VCS temperature. To estimate the effect of VCS temperature on determination of emittance, a computer study was performed to calculate vent rate and VCS temperature as a function of a constant apparent emittance on all annulus surfaces. The results are presented in Figure 4-4, where the sensitivity to VCS temperature is shown to be quite dramatic. A VCS temperature increase of only 6 R raises the vent rate by over 26 percent, and the apparent emittance by almost 32 percent. The measurement of vent rate is considered to be more

\*W. H. McAdams, Heat Transmission, 3rd Ed., McGraw-Hill, 1954, page 61.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

75-11607 Page 4-7

# $e = 0.0303 - 1.53 \times 10^{-5}(T)$

where

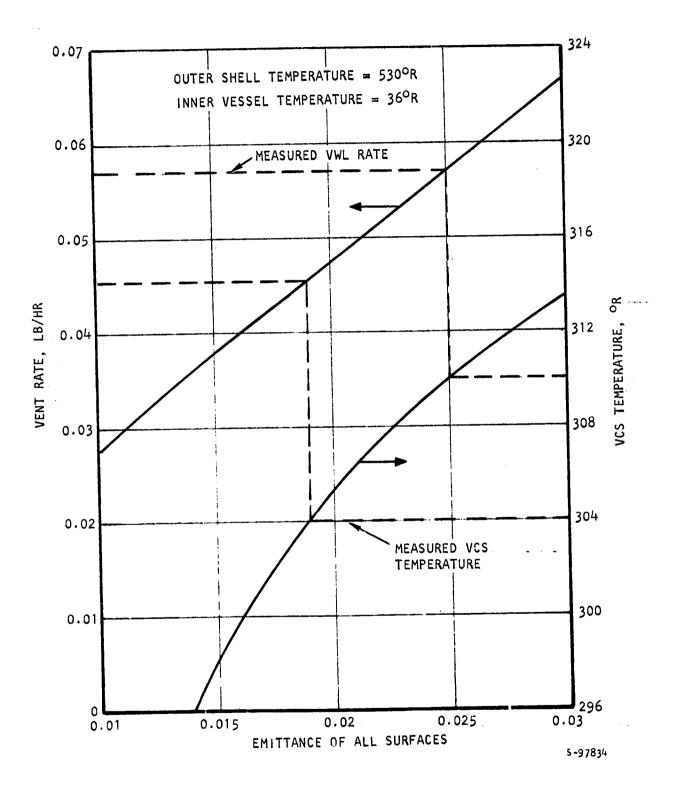


Figure 4-4. Determination of Constant Emittance Required to Simulate Tank No. 2A VWL Test



accurate, since it is obtained by two different methods. The VCS temperature is less accurately known (only one sensor), and, from Figure 4-3, was found to vary by more than 2 R during the 71-hr test period. Another uncertainty in using the VCS temperature is the assumption that it is a measure of the temperature over the entire shield. Further, the computer study is based on 100 percent effectiveness (i.e., the fluid leaves the VCS at the VCS temperature). This study points toward an apparent emittance of approximately 0.025, if it were independent of temperature. The value of 0.019, which would be required if the VCS were 304 R, is lower than any data in the literature for aluminized surfaces at temperatures above 300 R, and results in a vent rate of only 0.0455 lb per hr.

A possible explanation for the unrealistic emittance temperature dependence of Equation (3), assuming the VCS temperature measurement is correct, may be in the uncertainty in calculating the "known" heat leaks. Table 4-1 shows that the conduction heat leaks are more important below the VCS (35 percent of total) than above it (only 6 percent of the total). If the conduction heat leaks shown in Table 4-1 are too low the radiation heat transfer below the VCS will be less, and by a larger percentage than above the VCS. Hence, it follows that a lower emittance interchange factor could result below the VCS than above it, allowing the possibility of a positive-slope (more reasonable) emittance characteristic.

Another way to evaluate the apparent emittance of the annulus is an extension of the above study, using an emittance characteristic which increases with increasing temperature. This was actually done before the WL testing in order to estimate the expected vent rate. A linear emittance characteristic proportional to that measured by Cunnington, et.al.\* for the aluminum side of single-aluminized Mylar multilayer insulation was used. The best values reported were 0.013 at 113 R and 0.029 at 560 R. A series of computer runs were made with increasing factors applied to the emittance curve. Two print-outs were presented in AiResearch Report No. 73-9418, Pages 3-19 and 3-20, with vent rate results bracketing the measured value. A curve 20 percent higher than the base emittance curve gave a vent rate of 0.058 lb per hr (emittance range from 0.012 at 36 R to 0.0336 at 530 R). The predicted VCS temperature, however, was 335 R, or 32 R higher than indicated by the test results.

A summary of the analytical determination of emittance in an attempt to match the vent rate and shield temperature test data for Tank No. 2A is presented in Table 4-2, together with judgments on the validity of the emittance-temperature dependence obtained. In conclusion, it appears that vacuum-deposited, 2000. A aluminized annulus surfaces exhibit apparent emittances more nearly equal to those of 400 Å aluminized polyester film than polished aluminum plate, especially at room temperature. It also appears that the VCS temperature should be determined more accurately in order to evaluate the annulus emittance, and that for a given emittance characteristic (if available from independent sources) the least accurately predicted paraeter would be VCS temperature.

\*G. R. Cunnington, C. W. Keller, and G. A. Bell, "The Thermal Performance of Multilayer Insulations," Interim Report, <u>NASA-CR 72605</u>, 20 April 1971, page 4-22.

AIRESEARCH MANUFACTURING COMPANY

#### TABLE 4-2

SUMMARY OF EMITTANCE ANALYSES FOR TANK NO. 2A

Vent Rate (Lb/Hr)	VCS Temp. ( R)	Apparent 36 R	Emittance 530 R	Comments on Temperature Dependence
0.0572*	304*	0.0297	0.0222	Unrealistic - negative slope
0.0455	304*	0.019	0.019	Constant - too low for 530 R
0.0572*	310	0.025	0.025	Constant - reasonable average
0.050	331	0.010	0.028	Reasonable - positive slope,
0.058	336	0.012	0.0336	<pre>dervied from Al-Mylar data</pre>

\*Test values

TANK NO. 2B (MICROSPHERE INSULATION)

#### Measured Performance

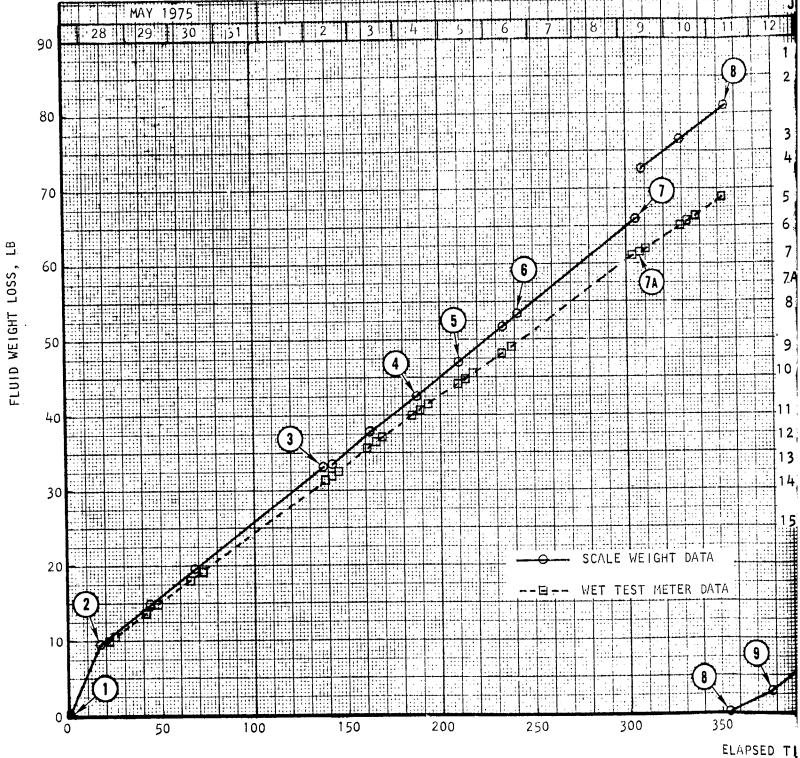
The tank insulated with partially-aluminized microspheres exhibited such unusual behavior (as compared with analytical predictions, to be discussed later), that the boiloff test was increased beyond the time usually required to establish a typical vessel steady-state vent rate. Figure 4-5 presents the vented weight loss data\*\* for the entire 31-day test series, and denotes the points at which significant events or procedure changes occurred. Both scale weight data and wet test meter (WTM) integrated data are plotted. In general, over long periods of time, one would tend to place more confidence in the scale weighings, although the wet test meter appears to have responded to some short term events, where the scale did not. The discrepancy between the two is in the right direction (assuming correct calibrations) since plumbing leaks would escape being recorded by the WTM. Vent rates calculated by test segment showed the minimum difference to be 1.6 percent (WTM rate higher than scale rate), and the maximum\_difference to be 19.8 percent.(WTM rate lower than scale rate).

The presence of thermal acoustic oscillations (TAO) with the tank in the normal vent configuration prompted numerous procedure changes and experimentation. Among them were rotating the tank 90 degrees and 170 degrees from the normal, installing a plug (only partial plugging) in the cryostat supply line, and removing the water saturator (upstream of the WTM). Some of these schemes altered the intensity of the oscillations, but since they were of relatively small magnitude,\*\*\* examining the entire data set reveals very little change in average vent rate from one time segment to the next, with two exceptions. The obvious one is when the tank is inverted 170 degrees and fluid is vented out

\*\*Raw test data from which this curve was constructed is given in Appendix A. \*\*\*A quantitative estimate of the TAO effect on tank heat leak is given later

in this section.





FOLDOUT FRAME /

GANRETT

LAPSED

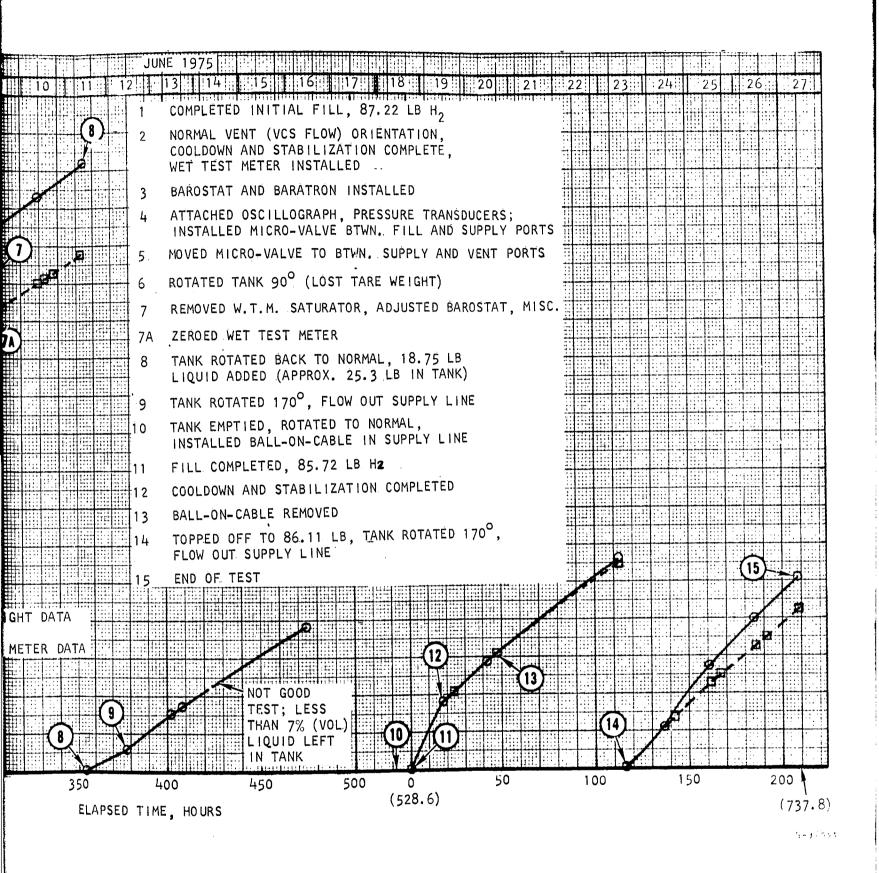


Figure 4-5. LH<sub>2</sub> Vented weight cost Test - Tank No. 28

> 75--176--) Forgan - **4--**17

FOLDONT FRAME 2

the fill/supply line, thus bypassing the vapor-cooled shield (VCS). This vent rate is expected to be higher, as was evident from the test data. The other exception is more subtle, and could possibly be due to experimental error: that is, the wet test meter recorded a drop in vent rate of approximately 15 percent when a micro-valve was installed between the fill and supply ports. Since the scale weight data gave the same vent rate as before installing the micro-valve (the tare weight was adjusted for the plumbing change), it is suspected that a small leak may have bypassed the WTM, or there was some error in recording the data.

Two portions of the test sequence were selected for analysis: 1) the last series in the normal vent configuration, where the agreement between scale and WTM data agreed to within 3.5 percent (scale = 0.199 lb per hr; WTM = 0.192 lb per hr); and 2) the last test segment, where the tank was inverted 170 degrees and the flow bypassed the VCS. Vent rate data from the WTM was 17.5 percent lower than from the scale (scale = 0.280 lb per hr; WTM = 0.231 lb per hr). The earlier test data for the inverted tank was not used because the tank was nearly empty and excessive stratification was evident--the vented fluid temperature was 60 -70 R above the liquid temperature.

The temperature data for the selected test periods is presented in Figure 4-6 and Figure 4-7. The various times denoted by circled numbers coincide with the events described on the weight loss curve, Figure 4-5.

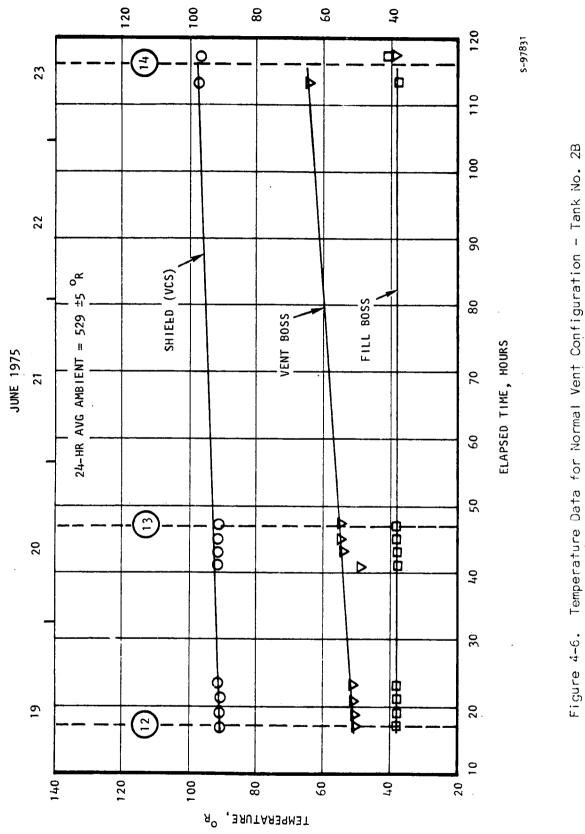
The fact that the fill boss temperature now reads within 2 R of 13.5 psia saturated liquid hydrogen temperature does not preclude the possibility that the temperature sensor is actually in the tube, as was considered in observing the temperature data from Tank No. 2A. The clamp (an unintended thermal short) supporting the supply tube at about 8-10 in. from the inner fill boss could result in the presence of liquid several inches into the fill tube.

The vent boss temperature in Figure 4-6 and the fill boss temperature in Figure 4-7 show a gradual divergence from the corresponding data for Tank No. 2A, Figure 4-3 (inner vent boss temperature nearly constant at 55 R). This is probably due to the much higher heat leak (factor of 3.5 to 4.5 above Tank No. 2A) causing more thermal stratification in the ullage. The slope of the curves could probably be related to the increased rate of liquid level recession in the present tank (No. 2B).

Although the VCS temperature varied gradually over the test period, the following averages are taken for comparison with analysis: 94 R in the normal vent configuration, and 1.2 R with the tank inverted and no flow through the VCS line.

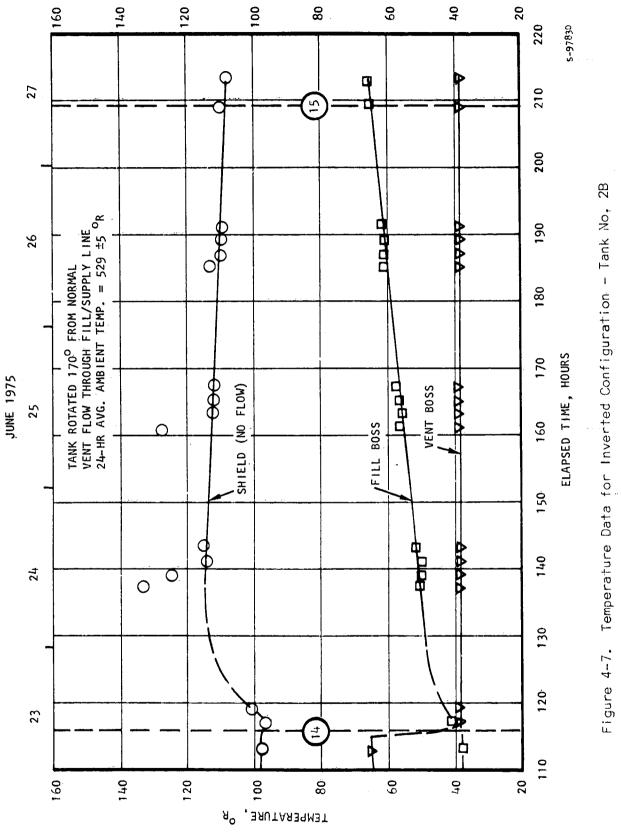
Table 4-3 summarizes the key parameters derived from the selected VWL test periods described above. The uncertainty value expressed for the vent rate is simply the deviation from the mean value between scale and WTM data. The uncertainty expressed for the VCS temperature is the deviation from the average value estimated over the time period. The 24-hr average ambient temperature was estimated to be  $529 \pm 5$  R.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

DARRETT



GARRETT AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

75-11607 Page 4-14

Ţ

TAB	LE	4-3
-----	----	-----

#### 'MEASURED PERFORMANCE OF TANK NO. 2B

	Tank Vent C	onfiguration
ltem	Normal	Inverted
Average Vent Rate, Ib per hr	0.1.96. <u>+</u> 0.003	0.256 <u>+</u> 0.024
Heat Leak, Btu per hr	38.3	<b>50.0</b>
VCS Temperature, R	94 +4	112 <u>+</u> 3

#### Estimated Effect of TAO on Tank Heat Leak

A recently published investigation of thermal acoustic oscillations by Spradley, et. al.\* provides an excellent basis for estimating the effect of TAO on tank heat leak. Although this reference work is confined to liquid helium systems and an idealized gas-filled tube geometry, the range of appropriate parameters studied was broad enough to merit applying here.

Our tube configuration is somewhat complicated in that it is a compound line with two different diameters (one line containing 3 tubes within it) with an unknown temperature at the tee joining them. Referring to Figure 2-1, the worst conceivable situation for TAO is suspected to be the path from the cryostat adapter, down the supply line, and into the liquid via the fill line. Using a hydraulic diameter of 0.4456 in. for the fill line gives an aspect ratio of L/D = 58; the cryostat supply line aspect ratio is approximately L/D = 338. Spradley's work does not suggest how to treat a compound line, so assuming aspect ratios are additive, for the total path we obtain an aspect ratio of approximately L/D = 400. The temperature ratio from ambient to LHz is  $T_h/T_c = 15$ . Using these two parameters the following parameters are obtained from the Spradley model (page No. of referenced work is also given):

Pressure amplitude,  $P_A = 0.04$  atm (page 43)

Oscillation frequency, f = 23 Hz (page 85)

Intensity, I = (0.04) (23) = 0.92 atm - Hz

Tube heat leak ratio,  $Q/Q_t = 9$  (page 89).

OF CALIFORNIA

 $Q/Q_{t}$  is defined as the ratio of total tube heat leak (with TAO) to normal conduction heat leak.

\*L.W. Spradley, W.H. Sims and C. Fan, "Thermal Acoustic Oscillations," Volume II, Final Report, LMSC-HREC TR D390690-II, March 1975.



The predicted Tank No. 2B heat leak breakdown is presented later in the thermal performance summary, Table 4-6, showing a fill line heat leak (normal conduction) of 0.27 Btu/hr, and a total tank heat leak of 15.1 Btu/hr. Applying the above TAO results gives a line heat leak of 2.43 Btu/hr; resulting in a total tank heat leak of 17.3 Btu/hr, an increase of only 15 percent above the "no-TAO" predicted value.

If the maximum measured oscillation amplitude of 0.02 atm and frequency of 12 Hz is used in place of the Spradley prediction based L/D and  $T_h/T_c$ , the resulting  $Q/Q_t$  is approximately 2.5 (from an experimental point on page 89, which grossly exceeds the theoretical value). In this case the total line heat leak would be 0.68 Btu/hr; resulting in a tank heat leak of 15.5 Btu/hr, an increase of only 3 percent over the "no-TAO" predicted value.

With a measured tank heat leak of 38.3 Btu/hr, it is concluded that the fill line TAO effect is negligible, since the line conduction heat leak is such a small fraction of the total apparent heat leak to the vessel.

#### Analytical Performance Prediction

Microsphere insulation performance data have been published by Cunnington and Tian\*, where hemispherically-aluminized spheres (among others) were tested in a flat-plate calorimeter with cold boundary maintained at 78 K. The data for a 5.84-mm thick layer (average sphere diameter of 84 microns) was curvefitted to the following heat flux equation:

$$q = 0.0315 (T_1 - T_2) + 0.0166 \sigma (T_1^4 - T_2^4)$$
(4)

where

 $q = heat flux, watt/m^2$ 

 $T_1$ ,  $T_2$  = température of the warm and cold boundaries, respectively, K  $\sigma = 5.667 \times 10^{-8} \text{ watt/m}^2 - \text{K4}$ 

The conduction term in Equation (4) assumes that the aluminum film thermal conductivity is constant to temperatures as low as 78 K. This is approximately correct, but as the cold boundary temperature approaches 36 R (20 K), liquid hydrogen normal boiling temperature, puré aluminum conductivity rises rapidly, as shown in Figure 4-8. The constant value implied in Equation (4) is probably 137 Btu/hr-ft- R, as quoted in a paper by Chan and Tien\*\*, which agrees with the reference used for Figure 4-8. The performance prediction computer program described earlier was modified to include the characterization of microsphere

\*G. R. Cunnington and C. L. Tien, "Heat Transfer in Microsphere Cryogenic

Insulation," Adv. Cryo. Eng., Vol. 18, 1973, pp 103-111. \*\*C. K. Chan and C. L. Tien, "Conductance of Packed Spheres in Vacuum," Journal of Heat Transfer, Vol. 95, 1973, pp. 302-308.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

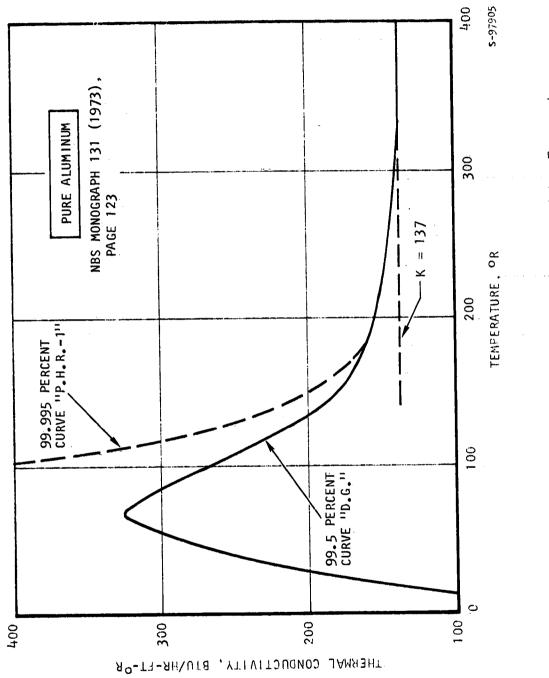


Figure 4-8. Thermal Conductivity of Pure Aluminum at Low Temperature



insulation. The program equation for heat flux has a correction for the thermal conductivity of 99.5-percent-pure aluminum, by making use of a built-in integral curve:

$$\int_{T}^{0.00} k \, dT \text{ versus T}$$

where T is the cold boundary temperature, R. Another correction applied to Equation (4) is an insulation thickness factor, which converts the flux equation into an apparent thermal conductivity equation. According to the theory advanced by Cunnington and Tien, as pointed out in the discussion following their 1973 paper, the conduction heat flux would be proportional to t-2/3. (In a more recent paper, by Nayak and Tien, this theory has been slightly revised; it will be examined further below.) In addition, the entire equation is multiplied by  $t_0$ , the thickness which applies to the experimental heat flux equation. The heat flux, in general, would then take the following form:

$$q = \frac{q_0^{\dagger} o}{t} = \left[ C_1 \left( T_1 - T_2 \right) \left( \frac{k_{A1}}{k_0} \right) \left( \frac{t_0}{t} \right)^2 + C_2 \left( T_1^4 - T_2^4 \right) \right]_{t}^{\frac{1}{2}}$$
(5)

where

 $k_{A1} = \int_{T_2}^{T_1} \frac{k_d T}{(T_1 - T_2)}, Btu/hr-ft-R$ t = insulation thickness, in.

This is the microsphere insulation characterization built into the computer program. The constants are determined as follows ( $T_1$  and  $T_2$  remain in degrees Kelvin):

 $C_1 = 0.0315 [3.41443/(3.28084)^2] = 0.00999215$   $C_2 = 0.0166 [3.41443/(3.28084)^2] (5.667 \times 10^{-8}) = 2.98408 \times 10^{-10}$   $k_0 = 137 \text{ Btu/hr-ft- R}$  $t_0 = 5.84/25.4 = 0.22992 \text{ in.}$ 

A more recent paper, by Nayak and Tien\*, explores further the influence on microsphere insulation apparent thermal conductivity of the compressive pressure resulting from self-loading (microsphere weight). This effect is expressed in terms of an insulation thickness L. Introducing the concept of lattice conductivity (or conductive component, the radiative conductivity is determined separately) for packed beds of hollow microspheres, experimental data from a spherical calorimeter indicated a relationship of the form:

k<sub>c</sub> ~ L<sup>0.66</sup>

(6)

\*A. L. Nayak and C. L. Tien, "Thermal Conductivity of Microsphere Cryogenic Insulation," Paper K-4, 1975 Cryogenic Engineering Conference, Kingston, Ontario.

AIRESEARCH MANUFACTURING COMPANY

The revised theory, however, based on the deformation of thin spherical shells under light loads and density stratification, implies an exponent of 0.6. At a mean temperature of 157 K (283 R), the Nayak and Tien data for a 2.3-cm thickness of fully-aluminized microspheres (average diameter of 26 microns) gives  $k_{\tilde{c}} \approx 1.7 \times 10^{-3}$  watts/m- K (9.83 x 10<sup>-4</sup> Btu/hr-ft- R). This reference point would give the following relationship:

$$k_{z} = 5.41 \times 10-3 + 0.66$$

(7)

where

k<sub>c</sub> = lattiče thermal conductivity, Btu/hr-ft- R

L = insulation thickness, ft

For the MOL tank nominal insulation thickness of 1.3 in., and mean temperature of 283 R, the lattice conductivity from Equation (7) would be  $1.25 \times 10^{-3}$  -Btu/hr-ft- R. Using this insulation thickness implies that the VCS is not present. The test data, Table 4-3, for the inverted tank yields an apparent thermal conductivity (which includes the radiation component) of 2.64 x 10-4 Btu/hr-ft- R, which is lower than the Equation (7) value by a factor of 4.7. Therefore, applying this data as if no VCS were present would predict performance to be much poorer than observed. This concept will be pursued in more detail later.

The use of Equation (5) in the computer program gives the tank performance presented in Figure 4-9. Two input variables were examined in the attempt to correlate the test results: 1) VCS effectiveness, and 2) innerannulus microsphere degradation factor. The effectiveness is a measure of how efficiently the VCS acts as a heat exchanger, that is, how close the fluid exit temperature approaches the average shield temperature (isothermal VCS assumed). The degradation factor is a multiplier (a value of 1.0 means no effect) applied to the heat flux equation for the inner annulus only. This would account for any increased compressive loading on the microspheres, or any other phenomenon which may increase the apparent conductivity of the microspheres at lower temperatures (exclusive of the aluminum conductivity effect and thickness proportionality), or a thermal short between the VCS and inner vessel.

As noted in Figure 4-9, a zero effectiveness shield simulates the inverted tank condition, when the flow by-passes the VCS. The trend toward infinite degradation factor simulates a VCS completely shorted to the inner vessel, thus rendering the inner annulus useless as a heat transfer barrier (VCS temperature and temperature of the fluid in the tank become identical). The tests results for the inverted tank, however, indicate a VCS temperature 205 R lower than predicted (the predicted value is 317 R, off the scale of Figure 4-9). This could be explained by a partially-shorted VCS (shorted to the inner vessel), except that the measured vent rate (from scale data) is even larger than the predicted vent rate if the VCS were completely shorted to the inner vessel (infinite degradation factor). Thus, even if the VCS were shorted, that alone is not enough to explain the disagreement between the micro-sphere analytical model and the test data for the inverted tank.

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

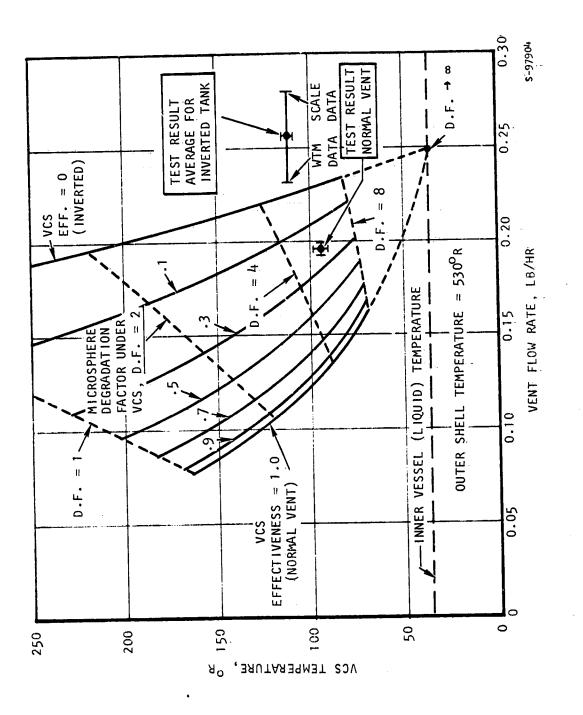


Figure 4-9. Analyticul Study of Microsphere-Insulated Tank Performance



The normal vent test result could be explained by a grossly inadequate shield efficiency (about 20 percent effectiveness) and a degradation factor of about 6. A degradation factor of this magnitude would not be unusual to apply to idealized equations for multilayer insulation systems, where anisotropic effects can be significant. However, for a near-isotropic media like microspheres, this factor seems to be exceedingly high.

An alternate approach would be to calculate an apparent thermal conductivity from the WL test results, and compare them with the latest lattice conductivity data of Nayak and Tien, similar to the previous calculation for VCS not present. Using the WL results, the apparent thermal conductivity is calculated separately for above and below the VCS, assuming that the support pad heat leak is predictable and the VCS effectiveness is 0.9. Then Nayak and Tien's theoretical prediction ( $k_c \sim L^{0.6}$ ) is applied to the tank annulus dimensions and mean temperature. The comparison is presented in Table 4-4.

#### TABLE 4-4

#### APPARENT THERMAL CONDUCTIVITY COMPARISON WITH NAYAK AND TIEN THEORY

	Insulation Ap	parent Thermal	Conductivity (B	tu/hr-ft- R)
Annulus	Normal Vent	Configuration	Invérted Con	figuration
Region	From Test	Theory	From Test	Theory
Above VCS	2.29 x 10-4	7.68 x 10-4	1.80 × 10-4	7.79 x 10-4
Below VCS	7.35 x 10-4	1.82 x.10-4	7.37 x 10-4	2.06 x 10-4

It is interesting to note that the theory predicts conductivities 3.35 and 4.33 times greater than the test results for the annulus region above the VCS, while the test results are 4.04 and 3.58 times greater than the theory below the VCS. The results are even more conflicting than shown, since the "theory" values do not include radiation (it is negligible below the VCS, only). The aluminum conductivity factor not included in the theory could account for some of the difference below the VCS, but the aluminum factors,  $(k_{A1}/k_{O})$  in Equation (5), would be only 2.17 and 2.10, for the normal vent and the inverted configurations, respectively. It is apparent that using the new Nayak and. Tien  $(k_{C}-L0.6)$  in the computer program would only result in further divergence between predicted and measured VCS temperature and vent flow rate; that is, for the theory to give higher vent rates (closer to test data), the VCS temperature must be hotter (further away from test data).

CARRETT

ARESEARCH MANUFACTURING COMPANY OF CALIFORNIA

More recent data published by Reinker, et. al.\* is in much closer agreement with the WL test results, although it uncovers a potentially serious problem with using hemispherically-aluminized microspheres. It was found that the thermal conductivity was highly dependent on sample history; that is, thermal cycling to  $LN_2$  temperature increased the conductivity significantly. Table 4-5 presents a comparison of the apparent thermal conductivity previously calculated (for the "above VCS" case, an average of the normal vent configuration and inverted configuration from Table 4-4 is used) with data from Reinker at the same mean annulus temperature.

#### TABLE 4-5

	Apparent Thermal Condu	ctivity (Btu/hr-ft- R)
ltem	Above VCS T <sub>m</sub> = 175 K	Overall Annulus T <sub>m</sub> = 157 K
VWL Test	2.05 × 10-4	$2.64 \times 10^{-4}$
Reinker Data:		
Uncycled (extrapolated)	$1.4 \times 10^{-4}$	1.1 × 10-4
Cycled Once to LN.2	$2.5 \times 10^{-4}$	$2.0 \times 10^{-4}$
Cycled Twice to LN2	3.8 × 10 <sup>-4</sup>	3.1 × 10-4

#### APPARENT THERMAL CONDUCTIVITY COMPARISON WITH REINKER DATA

This comparison shows much better agreement than is obtained by using Equation (5), or the Nayak and Tien theory. Note that the WL test value at 175 K is only 18 percent lower than Reinker's data after one cycle. In the VWL test the insulation was not fully cycled, although some of the microspheres under the VCS were warmed to 260 R (144 K), at 480 hours into the test (see Appendix, page A-14), prior to being rechilled down to LH<sub>2</sub> temperature. This history-dependent behavior of microsphere insulation was reported for the aluminized spheres only; the test data for uncoated microspheres was repeatable after cycling. The conductivity data for the uncoated microspheres fell between the half-aluminized data for one and two cycles.

From the above comparisons it can only be concluded that present theories and experimental (calorimeter) data are inadequate to explain the phenomena observed in the tank WL tests.

<sup>\*</sup>R. P. Reinker, K. D. Timmerhaus, and R. H. Kropschot, "Thermal Conductivity and Diffusivity of Selected Porous Insulations Between 4 and 300K," <u>Adv. Cryo</u>. Eng., Vol. 20, 1975, pp. 343-354.



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

#### PERFORMANCE SUMMARY

The MOL Tank Program and its follow-on programs offer a unique opportunity to compare the performance of three different insulation systems in a production tank of the same basic design, with an evacuated annulus, a vapor-cooled shield, and fiberglass support pads. Table 4-6 presents a performance summary of the three tank configurations tested, and includes the calculated performance based on the best available material data. The reason for the good agreement between test data and calculations (without assumptions) with Tank No. 1 is that AiResearch has more experience with and more technical data for NRC-2 multilayer insulation than the other types.

Figure 4-10 illustrates graphically the relative magnitude of the tank heat leaks tabulated in Table 4-6.



#### TABLE 4-6

#### THERMAL PERFORMANCE SUMMARY OF THREE INSULATION SYSTEMS

Tank No.	1	2۸	28	
Insulation . System	NRC-2 Multilayer	Low-Emittance Surfaces	Hemisphèrica Aluminizéd M	ally- Aicrospherės
Test Configuration	Normal Vent	Normal Vent	Normal Vent	Inverted. (No VCS Flow)
Test Performance				
Vent Rate (Lb/Hr) Heat Leak (Stu/Hr) VCS Temp ( R)	0.0410 8.02	0.0572 11.18 304	0.196 38.3 94	0.256 50.0 112
Calculated Performanc	( :e			
Vent Rate (Lb/Hr) Heat Leak (3tu/Hr) VCS Temp ( R)	0.0399 7.80 232	0.0572* 11.18* 310*	0.077** 15.1** 163**	0.162 31.7 317
Heat Leak Breakdown Calculated (Btu/Hr)				
Insulation*** Support Pads*** Lines*** VCS Cooling Rate Total from Ambient (outer shell)	5.25 2.10 0.45 21.7 29.5	7.12 3.48 0.58 49.6 60.8	13.69** 1.09** 0.27 25.9** 41.0**	27.92 3.62 0.12  31.7

\*Assumed emittance of 0.025 on all surfaces.

\*\*Assumed VCS effectiveness of 100 percent.

\*\*\*Heat leak to pressure vessel.

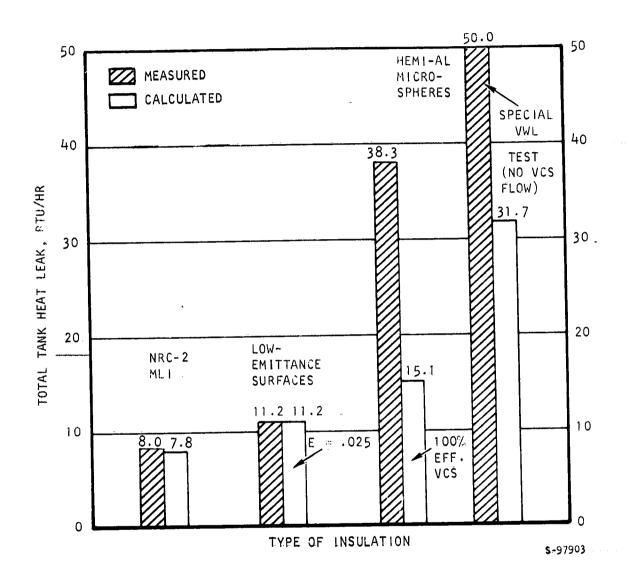


Figure 4-10. Comparison of Three Types of Insulation by Tank Heat Leak

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

#### SECTION 5

#### CONCLUSIONS AND RECOMMENDATIONS

Analytical studies and thermal performance testing of three similar tank configurations with different insulation systems resulted in the following conclusions.

- 1. Tank No. 1 (Multilayer Insulation) exhibited the lowest heat leak, and its thermal performance was most predictable, due to the extensive experience and tank-installed performance data generated at AiResearch.
- 2. Tank No. 2A (Low-Emittance Surfaces) demonstrated a heat leak almost 40 percent higher than Tank No. 1.
- 3. Tank No. 2A aluminized surfaces exhibited (by calculation) apparent emittances very close to aluminized polyester film values, which are about as low as can be achieved.
- 4. Tank No. 2A data analysis revealed the importance of a very accurate vapor-cooled shield (VCS) temperature determination; a 6 R error in VCS temperature was shown to produce errors of 26 and 32 percent in calculated vent rate and apparent emittance, respectively.
- Tank No. 2B (Microsphere Insulation) exhibited a much higher heat leak than the other two tanks, by factors of 4.78 (over Tank No. 1) and 3.43 (over Tank No. 2A).
- 6. Tank No. 2B thermal acoustic oscillations (of low intensity, however) prompted additional experimentation in the vented-weight-loss (VWL) testing, culminating in rotating the tank approximately 170° and venting out the fill/supply line. In the inverted position, the average heat leak increased by about 30 percent.
- 7. Tank No. 2B VCS and inner vent boss temperatures were observed to slowly drift upward over long periods of time as the tank fluid quantity was depleted (the vent boss was always warmer than the liquid, as in Tank No. 2A); and a gradual decrease in vent rate was detected. These observations indicated ullage stratification; hence, the test periods selected for analysis followed filling or topping-off.
- 8. The predicted performance of Tank No. 2B, based on published theory and experiments with microsphere insulation (both flat-plate and spherical calorimeters), was in poor agreement with the test data. The measured VCS temperatures were much lower than predicted, while the measured vent rates were much higher than predicted. The best predicted heat leaks were a factor of 2.54 and 1.58 lower than the measured normal and inverted tank values, respectively. The corresponding predicted VCS temperatures were high by 69 R and 205 R, respectively. A comparison of test-data-derived apparent thermal conductivities of the

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

insulation above and below the VCS with theoretical lattice (conductive only) thermal conductivities of Nayak and Tien between the same temperatures showed contradiction; results varied by factors of 3.4 to 4.3, and in opposite directions above and below the VCS.

- 9. A low-temperature correction factor for pure aluminum thermal conductivity in the aluminum-coated microsphere theory may account for some of the discrepancy below the VCS; most of the literature data and curvefitting have been limited to temperatures above 78 K (LN<sub>2</sub> boiling point), where the constant-aluminum-conductivity assumption is valid.
- 10. A recent paper by Reinker, Timmerhaus, and Kropschot (Adv. Cryo. Eng., Vol. 20, 1975) found that the thermal conductivity of half-aluminized microspheres increased each time the sample was cycled to LN<sub>2</sub> temperature. The apparent thermal conductivity calculated from Tank No. 2B test data, for the insulation above the VCS, was only 18 percent below Reinker's data after one cycle.
- 11. The Tank No. 2B test results, considered in conjunction with the three previously published calorimeter studies, indicate that perhaps the thermal performance of microspheres (especially the aluminized type) is not as predictable as expected when first proposed as a cryogenic insulation. In addition, for very low temperature cryogens (LH<sub>2</sub> and LHe), its thermal performance may not be competitive with multilayer insulation or low-emittance surfaces, where VCS are employed.
- 12. A microsphere-insulated tank testing program is recommended to obtain additional performance data on the practical application of microsphere insulation in high performance cryogenic tankage. A suggested tank evaluation and test program would include the following:
  - Radiographic (X-Ray) examination of the tank annulus to determine the microspheres location (possibly density stratification), and the uniformity of VCS spacing.
  - Gas analysis of the annulus to investigate the possibility of excessive SO<sub>2</sub> and O<sub>2</sub> (gases inside the microspheres) contaminating the vacuum ion pump leading to potentially erroneous annulus vacuum readings.
  - Vented weight loss testing and cycling with liquid nitrogen to determine if aluminized-microsphere insulation conductivity increases with cycle time.
  - Thermal performance testing with supercritical helium, with the addition of temperature measurement and insulation around the cryostat supply adapter, to verify the feasibility of satisfying the Phase II J-T cryostat supply requirements.



#### APPENDIX A

#### TEST DATA

This appendix contains test data sheets compiled during testing of the NASA MSFC cryogenic hydrogen/helium storage and supply system. Test data given in this appendix is indexed below.

Test Dats Description	Page(s)
Shock and Proof Pressure Test	A-2 through A-5
Microsphere Eilling	A-6
Vacuum Decay	A-7 through A-10
Tank Performance Dàta	A-11 through A-19



$\frac{1}{1242} = \frac{1}{1242} = 1$
E.W.O/CHGE. NO. 3400-350332-0/-0100 SUPP. 1.D. 1.D. 1.D. TECHNICIAN <u>ELGLICH</u> <u>SALA</u> <u>SUPP. DATA</u> <u>LOG</u> <u>I</u> <u>TIME</u> <u>ENY</u> <u>PUBGE</u> <u>OAN BORG</u> <u>SUPP. OUADARGENERS</u> <u>SHEET</u> <u>I</u> <u>TIME</u> <u>ENY</u> <u>PUBGE</u> <u>OAN BORG</u> <u>SUPP. OUADARGENERS</u> <u>SUPP. COMMERCES</u> <u>OF</u> <u>CANA</u> <u><u>E</u><u>HEELICH</u> <u>ENAN</u> <u>ENCLIPONCE</u> <u>SUPP. OUADARGENERS</u> <u>ELECTON</u> <u>ENAN</u> <u>ENCLIPONCE</u> <u>SUPP. OUADARGENERS</u> <u>CANA</u> <u><u>E</u><u>HEELICH</u> <u>ENAN</u> <u>ENCLIPONCE</u> <u>SUPP. OUADARGENERS</u> <u>ENCLIPONCE</u> <u>ENCLIPONCE</u> <u>ECTIPE</u> <u>ENCLIPONCE</u> <u>ECTIP</u> <u>IIIIG</u> <u>TC</u><u>HI</u> <u>400</u> <u>ELCON</u>. <u>ENCELIPONCE</u> <u>ECTIPE</u> <u>ECTIP</u> <u>ENCELIPONCE</u> <u>ECTIPE</u> <u>ENCIP</u> <u>ENCELIPONCE</u> <u>ECTIP</u> <u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u> <u>IIIIG</u> <u>ECON</u>. <u>ENCELIPONCE</u> <u>ECTIP</u> <u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u> <u>ENCELIPONCE</u> <u>ECTIP</u> <u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u> <u>ENCELIPONCE</u> <u>ECTIP</u> <u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u> <u>ENCELIPONCE</u> <u>ECTIP</u> <u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u>E</u><u></u></u></u></u></u></u></u></u></u></u></u>
PIN
THE GRY PORCE ON BAG SOURCONDING TORK PAR FULLIPORT SHUPLES OF CASE FREQUEN GAS; 3CT OP PER FLUXO, SHUPLES OK GIVE STARTER COCL DOWN TO #1 IS ON TOP OF TANK AND TO # 2 IS ON THE BOTTON OF TANK AND TO # 2 IS ON THE BOTTON INCO TO #1, 62°F TO #2, 49°F INCO TO #1, 62°F TO #2, -12°F INCO TO # 1, 62°F TO #2, -12°F INCO TO # 1, 46°F TO 2, -55 INCO TO # 1, 46°F TO 7, -50°F INCO TO # 1, 43°F TO # 2, -40°F IZOO TO # 1, 46°F TO # 2, -40°F
PAR FULLIPORT SAMPLES OF CASE FREEDOM GAS, 357 OP PER FULLO. SHUPLES OR GIVE STARTED COCL DOWN TO # 15 ON TOP OF TANK AND TO # 2 15 ON THE BOTTON OF TANK AND TO # 2 15 ON THE BOTTON IO TO TO #1, 62°F TO #2, -12°F 10:20 TO #1, 62°F TO #2, -12°F 10:20 TO #1, 46°F TO 2, -55 11:10 TO #1, 46°F TO 2, -55 11:40 TO #1, 40°F TO #2, -60°F AFTOR LOWER BOTICE HAD WORMADD UP TO OF LOST COULD FLOW. 1245 LIGNID FLOWING AGAIN! TO #1, 43°F TO #2, -40°F
$\frac{G_{HS}}{STARTEP} \xrightarrow{SET} UP PEN FEMOR SHUPLES AR STARTEP COCL DOWN TO # 15 ON TUP OF THNK AND TO # 2 15 ON THE BOTTON OF THNK AND TO # 2 15 ON THE BOTTON 10:00 TO #1, 02 F TO # 2, 4/3 °F 10:20 TO #1, 62 F TO #2, -12 °F 10:40 TO # 1, 62 F TO #2, -12 °F 11:40 TO # 1, 4/6 °F TO 2, -55 11:40 TO #1, 4/6 °F TO 2, -55 11:40 TO #1, 4/6 °F TO 2, -55 0°F LOST LIQUID FLOW. 1245 LIQUID FLOWING NEARLY 1245 LIQUID FLOWING NEARLY 1200 TO #1, 4/3 °F TO #2, -40°F$
$\frac{51'44}{5TARTEP} \frac{CORL}{DORL} \frac{DORN}{TC} \frac{TC}{T} \frac{T}{15} \frac{DN}{TCP} \frac{TCP}{OF} \frac{TCP}{15} \frac{TCP}{15} \frac{TCP}{2} \frac{15}{15} \frac{DN}{THE} \frac{TCP}{BOTTON}$ $\frac{DCRC}{TC} \frac{TCP}{1} \frac{TCP}{1} \frac{TCP}{TC} \frac{TCP}{2} \frac{15}{2} \frac{CP}{F} \frac{16'2}{F} \frac{16'2}{F} \frac{12'F}{15} \frac{16'2}{F} \frac{16'2}{F} \frac{12'F}{15} \frac{16'2}{F} \frac{16'F}{15} \frac{TCP}{2} \frac{-55}{F} \frac{11'40}{15} \frac{TCP}{1} \frac{46'F}{F} \frac{TCP}{2} \frac{-55}{F} \frac{11'40}{F} \frac{16'F}{F} \frac{TCP}{2} \frac{-55}{F} \frac{11'40}{F} \frac{16'F}{F} 16$
$\frac{OF}{THNK} AND TC#2 IS ON THE BOTTON NOICO TC#1, 2/9F TC#2, 4/9F 10:20 TC#1, 62F TC#2, 30F 10:20 TC#1, 62F TC#2, -12F 10:10 TC#1, 46F TC2, -55 11:10 TC#1, 46F TC2, -55 11:40 TC#1, 40°F TC#2, -60°F AFTER LUNGH BETILE HAD WANNED UP TO OF LOST LIGUID FLOW. 1245 LIGUID FLOWING AGAIN 1245 LIGUID FLOWING AGAIN 1245 LIGUID FLOWING AGAIN 1245 LIGUID FLOWING AGAIN 1245 LIGUID FLOWING AGAIN$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$
$\frac{10!20}{10!40} = \frac{10!7}{10!40} = \frac{10!7}{10!7} = 1$
$\frac{10!20}{10!40} = \frac{10!7}{10!40} = \frac{10!7}{10!7} = 1$
11:10 $TC^{\#}$ , 46° $F$ $TC^{2}$ , -55 11:40 $TC^{\#}$ , 40° $F$ $TC^{\#}$ , -50° $F$ AFTER LUNCH BETTLE HAD WHATTED WAS TO OF LOST LAUD FLOW. 1245 LIGUID FLOWING AGAIN TC^{\#}, 43° $F$ $TC^{\#}$ , -15° $F$ 1200 $TC^{\#}$ , 46° $F$ $TC$ $TF$ 2, -40° $F$
$\frac{11:40}{1240} = \frac{140}{1245} = \frac{15}{1245} = \frac{15}{1245} = \frac{15}{1245} = \frac{15}{1245} = \frac{15}{1245} = \frac{1240}{1245} = \frac{140}{1245} = \frac{15}{1245} = \frac{1240}{1245} = \frac{140}{1245} = $
AFTER LUNCH BETILE HAD WHATTED UP FO OF LOST LAQUID FLOW. 1245 LIQUID FLOWING AGAIN TCH, 43°F TCH2 -15°F 1200 THI 46°F TCH2, -40°F
AFTER LUNCH BETILE HAD WHATTED UP TO OF LOST LIQUID FLOW. 1245 LIQUID FLOWING AGAIN TCHI, 43°F TCH2 -15°F 1200 THI 46°F TCH2, -40°F
1245 LIGNID FLOWING NGALLI TOM, 43°F TOM 2, -15°F 1200 TOTI 46°F TOTTO, -40°F
TCH, 43°F TCH2 -15°F
1200 + #1 46 °F TC # 2 - 40°F
$\frac{1300}{12.15} + \frac{46}{12.15} + \frac{46}{12.15} + \frac{10.17}{12.15} + \frac{46}{12.15} + \frac{10.17}{12.15} + \frac{46}{12.15} + \frac{10.17}{12.15} + 10.$
12 + 7c + 1 + 48 + 7c + 2 - 1901
1330 TC#1, 477 TC#2, -701-
1540 STARTED. TO PRESS WITH HE TO TO UE
1343 TANK PPESS 301516 TO TH 46-F
1343 TANK PRESS 30 MIG TO #1, 46°F 1346 TANK PRESS, 45 Price TO #1, 48 °F
170 ) CHAINA - B KAT K ISA MCAS
1355 TANK PRIEDS 140 1516 TO 1 43°F

PAGE TIME: \_\_ ----TOTAL TIME:

Ĩ

DARHETT AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

ORIGINAL PAGE IS OF POOR QUALITY



#### LABORATORY TEST LOG

ARTICLE ON TEST HOLHE SUNDLY THNK SIN 01 DATE 8-9-24 E.W.O/CHGE, NO. 3400-250 336-01-0/00 SUPP. I.D. TECHNICIAN TULLISH DATA SHEET DATA LOG 2 P/N TIME CHANGED 'K' BOTTLE CONTINED TO PRESS. 1400 TANK PRESS 160 FS16 TC#1 44 0F TC # 2, -16°F TANK PRESS ROO NSIG, TO#1, 44 8F TO#2, -20F 1405 THING PRESS DRST TOTH, 48°F 1410 CHARACES K' BOTHE 1412 CONTINUED TO PRESS, THINK PRESS 3901516 1413 TC#1, 46 7 TC#2, - # FF START 4 MILD HOLD THING PRESS 294 PSIG 1415 TC#1, 42°F TC#2, -16 1417 TANK PRETSS 295 PSIG TOT2, -1695 TC 77 \$ 416 COF 1419 TALVE PRESS 296 PEILE TOTAL 416 OF TOTAL -12°FOF, END 4 FILM HOLD UTKITING AND START WARREN OP 1520 TELLA 45 OF ALLE STOPED AND BETTLE SHIP TO LIA,

PAGE TIME:

TOTAL TIME:

ORIGINAL PAGE IS OF POOR QUALITY

75-11607 Page A-3

ARRETT AIF

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

			FLUID ANALYSIS TEST DATA SHEET	VALYSIS A SHEET		
Part Name <u>/</u>	la/He Su	Part Name Lia/He Supply TANK (FUNCE)	מנגדן אנא	્રા (ઝરુ	1	•
Procedure No.				Para No.	No.	
hs put tid	S.SIDIO	ME O	T			
Date 2-0	HC-5-	Test	Engineer	Test Engineer R. RAYADOR	2101	Technician EUGUSAL
Inspect ion	kiResearch	<u> KiResearch Quality Control</u>	lo			DCAS-QAR
Fluid:	ELIUM GAS	Gus				
Para	aranețer	Units	Req	Actual		Remarks
Particle	0ver 500	Microns	0	Ś	•	
ŝ2 - S	350 to 500		хып 2	¢		
	200 to 350		IO max	M		

Figure 2. Fluid Analysis Test Data Sheet

AIRESEARCH MANUFACTURING COMPANY 70213 AT851240

ORIGINAL PAGE IS OF POOR QUALITY



AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

1 <sup>1</sup>	6.532	10-10	ĊŨDATE	<u>6-</u>		TEST PURPOSE		SHOC	K AN	PRC	C F L	SHOCK AND PROF PRESSURE
	0			( (	1711							
р. т			BAROI TEMP	× 4	BARON . 0 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	TEST PERS.		ENGLISH	HSH			
114	. V . 1		11	<u> </u>	TANK							
	- iL # 8				Key							
2 1 0 2	m	4	5	\$	7 8	6	2	11	5	7	2	REMARKS
13:40	ю т				•	517	RTED	10 0	PRESS, EHANK	HANK		STREET IS L'ASSO
13:413	10				30				, ! i			PETER STRAINS
13. 416	л Т				(بري	C H	CHANGED.	, k	EXTT2	1		$\frac{1}{2}\sqrt{2t_{i}} = \frac{1}{2}\sqrt{2t_{i}} = \frac{1}{2}\sqrt{2t_{i}}$
13:55	43				140			: • •		+		CALLS MARTIN
14:00	H-H-		- 119		1601				, , , ,			H.Mour. Cont. Down
5-11	-1-1		-20		200							
++; ] G	E C				285	Ū	CHANGED	011	ENT	4. 5.		
14113	46		3/-		200		· · - •	·	•			
111.15	ω,		-16		1. [H	12:	STPACT -	NILL H	biold (			
1-1-1	46		7/-	•••••••	295	: 			•	4		
51:11	74		21-13				END 11	1/11/1 +101d.		Z VID T	557	DUMPPKE
	·	• • • • • • • • • • • • • • • • • • •						! 				
+	•			•   	1	 			,	•		
			•				               					
	· · · · ·											
							·····					
• • • •	•	• •	• ••	• • • • • • • • • •							1	
		••   						• • •	•	· · · ·	• · • •	
•	•	•••		•			···· •	; • •••	;			
			•••	-				-				

75-11607 Page A-5

••

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

ORIGINAL PAGE IS OF POOR QUALITY

; . à **1**0 - - - i,  $\frac{1}{1}$ ĥ Pege | у П К FILLING į h ij ł . 15 ÷ MICROSPHERE 16.88 LoSi КЧ TO THL 2 - LAB DATA SHEET 24.22.19.64 4.58 Less A 34.36 19.10 5.26 24.60 22,98 1.62 Ë, SUPAY TARE EMPTY WT. 24.18 18.76 5.42 WT. 13 K. Ξ MICRU No: TEST PURPOSE. 7 2 ł TEST PERS. ٥ æ 1. 12 400-250338-01-040 HATE 3-4-75 i BAROW TEMP ŝ SPUEREd 16.88 MICRO 1 ズン 0F 4 ģ MOL POL POL 502,50519.38 Ы Ч ო 626043 DATE WT. 2 FORM #1339A i, 3-4-22 2 | 3 | 21 G 1 ---• 2 7 2 1 18 5 24 Ξ. 2 12 ž



	3400-250335-01-0500 DATE -X-9-75	543		MUANA	יב דעיייני ינ סא	TEST PURPOSE						
				TEMP	lp	TEST PERS.	s		- ~			
	1	1	CVEN	CWEN FURBL	Prin sys	T MANYI TOND	UNVH					
•	TIN'	11115	11:MP. 11	P. TI-MP.	LAUNT	1 YACULY	UAL VI					
C Z		2	4	+ د	6 7		10	12	13	14	15	REMARKS
-	- 44 - 44	0805	1.1	74	1.5.110-3		CPIEN			1		
7	5-6-	C340	.47	14	-1.6110-	5 35110-7	CHIN					
 	:	C 545	- 74	74	1.3 110-1	1.8410-5	Ciesi D				1	ICH PUND ON RANKE
4	· .	5.62.2	77	74	1.5.1/0	1410-5	~~					C1:1-
 	"	× 2 2	-1.7%	74	1.3410	5 11/0-5-	~~					
	"	0.160	7.1	74	1.3.110	S-1X 10-5	->-					
~		16:0	7.0-	75-	1.3 110	5 11 X10-5	~					
80		1410	. 25	25	1.3 11	1.1 × 105	; ; ; ;				4	10N PUMD CITI
•	=	1315	75	75	1.3 1/0	× 71/2 3	~	• : : :				
	;	1620	2	7: 1	1 110	Y 3.1 1/1 -5-	•	4 - -			_	
=		1125	2-	75	1.51.10	F 91/10 - 5-	~					
12	+ ; ;	1: 1/	1, i 15.,	2		1.9.5×1/6 -5						
	<b>-</b>	5.5	121	73,	1/21/-	1.1.1 - 21		+				
; • •		1166	5	751	114 2. 17	1.2.1 10 -4	~	+			¥	START I BEAR TRADING
+		1200	12	75	112410	Y. 1.7. XV: -1		i	+	<b>1</b>		
۱۴	··· ·	2.	2	1,	1.24	1. X.Y Y				 : ;		
17	~	1400	·)/	76	1.3410	3. 2.6X15-4		•	+	 		
81		1.1.		7.	1.3.11	1. 4. 5. 4. 5 - 4.	· · · · ·	 !		:		
 £	 	1706	11	76	1. 1/6	1.218.5.2	~	<b>-</b> -	•	:	1	
20 ≶.	16 J 1	5765	. 75	7.4	·	31.1.5-4			ı			
~ ~	1-1-	こうらい	1.1.	1.1.		3.1X 10-1						

AIRESEARCH MANUFACTURING COMPANY OF CALIFORNIA

## ORIGINAL PAGE IS

: **X** 1 the state of the s i 5 REV ARKS i P oge ļ i ł 15 1 ... LAB DATA SHEET 13 Ú 2 CLOSL'D 1111/1 11 : 01 QWVH 'n, 3 2 ~ ~ 3.1 × 10 -4 3.2 × 10 -4 2.1 × 10 -4 2.1 × 10 -4 7.1 A 15-4 3.1 1/104 3,7×10 4 1.2181 11.15-4 41×10-1 4.1.4 15 A 3.1 1.12-4 21 1/5-4 TEST PERS. 7 8 6 8 7 - 3/1/5 - 4/ 3.1 X 16-4 3× 10-11 3.24154 3.24104 3, 2, 8, 10 4 3.3.4.64 Ling port System What we We are worked 1. 3460 250358-01-0500 BATH 4.9-7. ~ 11 ł BAROM .... 74 74 76 75 76 75 76 76 76 76 71 76 TEMP Jun 11 du 11 1200 76 76 76 16 76 76 76 1.2 202 7. 75 1700 75 15 19 1260 76 1160 76 1360 76 1360 76 1360 76 G700 75 4 1.700 0.700 1100 I'INI'I 0.250 m 51-256 44 •• FORM #1333A V-11-1 11 Tiut 21-4-1 1 4 × . : 2 ~ ÷ ٠ ្ខ័ o ŝ ŝ Ð

AIRESEARCH MANUFACTURING COMPANY

ORIGINAL PAGE IS OF POOR QUALITY

IN SYSTEM VACUUM Page \_\_\_\_\_\_ Y REPARKO 104 TUBE ..... R R R R 11111 ĥ 12 1 ; 2 - LAB DATA SHEET ENO 3400 250 358-01-0500 CATE 4-9-75 TEST PURPOSE . 1/2 C & VIJ. D. C'AL **2** 2 74 74 6.1125 4.2.16-4 Custo 75 75 6.1126 5 4.2.10 4 11 75 75 6.5216 5 4.2.10 4 11 11 01 01 VALIN UN HH . : 1 ~ : 2 2 2 2 2 2 2 ~ ~ 2 TEST PERS. 1406177 1:5-X15-1 hr. of Kletr 4.4X15"4 4-1415-4 MANIELD 4.2×10-4 4. 4X12 4 3.5 1. 2 1 5X16 4 4.6×10-4 76 76 22115 + 452.00 4.2. 4 10-4 4.3. XI - 1 1. 2315 4 1.5X 15 1 4-218-2 1 4.4X1.4 S.1410 8 4 2X10-8 211: 3 6.6416 8 1.44.5 76 76 3.6210 - 4 4. NK 2.7 1.4.10.4 21 1º - 5 12.72.17W ONTH PURCE SYSTEM 1.4410-1 TITAR TEMP. VILOUNA 3.2115 - F 6 7 5 71 ę, EAROM TEMP ۰ ۳ ۳ 75 73 76 75 75 75. 75 , ' ' 74 74 76 76 1.1 77 74 76 12 1. . . . . . . . ~~~~ 5 12 Ň 74 70 ¿700 0.700 1500 1266 455 1:135 c7c c 1430 17.1 1700 1700 15-1:0 1230 1766 5.200 02020 1300 17:00 71145 •• 11 95214 3 ~ 1: 4-13-75 FORM +15714 1.15 2 11 4.18 18 1.74. :1. Int-k 2 4-17-15 7-12-6 c2-11-1 115-7 DAN 1 - 12-1 " 1; *и* : е: 2 1 *!!* . 1 ~ \* 11 6 ٠ z |. ∞ |. ្លុំខ្ល a t :, -•7 ÷ <u>...</u> œ . #\*\*

ORIGINAL PAGE IS

75-11607 Pare 4-9

AIRESEARCH MANUFACTURING COMPANY

OF CALIFORNIA

Poge \_\_\_\_\_\_ 0' 4 <- CONTINUE ON FURNAGE - OPEN HAND VALVE RECORD SHEET 4 RENARKS ŀ THE REAL PROPERTY IN 1 ñ APO-250335-01-0500 DATE 4-9-75 TEST PURPOSE .. VA " UVA" DE UAY H L -- LAB DATA SHEET <u>۳</u> Sach 2 711 **=** .. 01631-0 OPEN CIARH 6 2 2 : 1 ~ 2 : 11 2 -3 4,5×1,6-4 0 TEST PERS. 4.49.10 " 4:3 ×10-4 1.8710-6 4.5 11: 1 1. 2187.1 1.11.5 4 4-314 412 + 11×11+ 4.3 \$15-4 4.3415 4 3.5415-4 4.07 VA 14 10 4 4. 11. 11. 24 CUTN FURDE SYSTEM MANIFLAD 4.3×104 2.3 4 16 4 42 110-4. 1. 8 X10-1 TEMP TEMP EDUNA LACENTY б .. З 2,2710-5 2 2410-5-41 8-0182 2.2810 8 4 1 8-01×+19. 1. 9. 4. - 8-1-014 11 3.34/2-1 2115 ٢ 1 . . . 1506 75 75 2416 1506 77 77 2.2416 TEMP 74 BAROM 1200 74 74 75 1200 75 75 75 1205 75 75 75 77 77 75 75 : 74 74 75 75 73 73 75-75 77 77 75-75 73 7-1 74 72 72 7.77 11.11 76 76 23 1700 . 100 1700 1530 0.100 0700 SOEI 1700 1.761 ċ,Ζ.,΄ 15.30 Juic 0.700 0:700 TIMI 11 95-6043 5-2-2 5/-1-5 52-5-5 5.2-75 5-5-2 5-2-2 56-6-5 11 1. 12-4 11 5-2 1 Propert #1353A . . JUNIL 11 : ;; 2 11 : -= 1 . 21 • : -2 ιc. **..** ...

URING COMPANY OF CALIFORNIA

AIRESEARCH MANUFACT

1---111

# HYUROGEN TANK 851240-3400-250338-01-0601

			COLUMN TRADE OF A DESIGNATION OF	FILL T	EMP.		A COMPANY OF A COMPANY	SHIELT	TEM	P.	1
DATE	TIME	ELAPSED	T, A-B	F. B.C	T. AB-BL	T, Temp °F	Tz EF	Ti Si	TI LF-FG	Tz TEMP	13
MAY 27 75		-	543,98	1.534	542.446	+ 70.678			543.495	+71.640	541
	15:00	1	UN	T No	7 5	TABLE		•		SISTANC	
28	08:05	17.1	3,672	,806	2.866	-421	24.00	334	27.3.14	-414	7.
	11:11	20,2	3,750	. 803	2,1947		34,91	12.89	34.621	-373	8.
4	13:11	•	3:760	1802	2.958	-422.5	29.40	. 2.86	29.114	-319	6.
	15:11		3.754	.802	2.952	- 422.5	29,40	-283	29.117	- 379	2.
29		41.2	3,757	. 798	2.959	- 42.2	3480	·285	3-1.515	-313	8.
	9:11	42.2	3.760	.199	2.961	- 422_	36.30	1285	36.015	-312	8.
	10:11		3.160	(199	2.961	- 1122	38,00	12.85	37.115	- 370	9.2
1	11:11		3.747	, 799	2.948	- 421	37.798	1283	37,515	- 370	9.2
	12:11	45.2	3,760	1800	2.960	- 422	38-100	•283	37.812	-320	9.2
	13:11	146.2	3.750		2.450	. 422	38.300	1285	38,015	-320	9.5
	14:11	47.2	3.750	,803	2.947	· 47.2	37,800	- 255	37.512	-310	9.
		48.2	3.750	,804		- 422	34.700	12.85	34,412	-373	9.
30	08:11	49.2	3.760	,805		- 422	35.200	1288	34.912	-373	9.4
	9:11	65. % 66.2	3.120	6771		-422_	141,500	1283	41.217.		11.
1	10:11	67.2	3.730	. 223	2.957	- 42.2	116,400	: 28-44	-10.116	-369.	11:
	11:11	68.2	3,730 3,727	·777 ·782	2.953	-422	39.100	, 2.85	38.815	-368	11 0
	12:11	69.2	3,732	,182		-422	37,700	1285	37.415	- 370	110
- - -	14:10	71.2	3.729	, 185.	2.447	-422	31.870	1238	3753%	-370	110
	16:11	23.2	3,740	,190		-422	38.200	1239			11.
JUNE 2		137.2	3.720		2.950	-422	33.400	1290		-369	11 1
		137.5	5.720	1186	2.940	-422	40,1179	, 2.94	40.18.5	- 368	16 .
	13:00	142.0	3.515	• 781	2.734	-422	29 . 11 2		360 10 -		
	14:15	1	3,726	. 283	2.943	-4122	39.243	61.96	38 947	-369	16.
1		145.25	3.498	- 783	2.715	- 421	44.736- 39.994	. 277	44,459	- 364	19.
	08:15	161.25	3.698	. 77/	2,927	- 47.2	59.794 59.398	·291	39.694	- 368	16.
1 1	10:15	163.25	3.710	1776	2.934	- 47.2	53 858	12.99 2.98	59,099	- 352	2.4.
		165.25	3.709	1777	2,932	- 422	49.798	. 278 1998	53600 149.50C	-357 -359	24.
		167.25	3.712	.777	2 935	.422	119,499	1299	149.700	- 359	22.
.24	1.15	169.25	3.713	•778	2.935	1	149.26 4	,300	48.964	-360	22,
4 1	08:15	185.2.5	3. 17.5	1777		-422	54,800	1302	54.498	- 355	22.
	10:15	187.25	3.706	,779	2.927		51.100	,303	50.797	- 357	26.0
1		189.2.5	3.694	.770	2,924	. 422	50.900	,302	50.598	- 358	25.0
		191.7.5	3.696	1771	2,925	-422	30.100	,303	49.797	-359	241
· · · · · ·	16:15	193.25	3-687	,765		422	50,100	,303		- 359	24.6
;										,.,	24.8

ORIGINAL PAGE IS OF POOR QUALITY

.

FOLDOUT FRAME /



GEN TANK 851240-1 S/N 1 150338-01-0601

. .-

5	HIELD	TEM	P.	L V	ENT	TEMP	>	1	k-16	20112	ωειζη	7-	
	Ti A	Tin	TE TEMP	L3 L	T3 KL	T3 _1K-KL	T3 TEMP	SCALE H	SCALE #Z	GROSS	TARË	LIQUID	X
		543 495				542.090	+70.351	+	WT. Di Ber	THE B	WT FEACIN'	WT Ibs.	
00	6 H	27.2.16	=5157ANC. -381	1	FADING						494.75	87.22	
	314	6.777	5 <u>41</u>	7.353		6.979			278:10			77.77	
1	1289	34.621	-373	8.720	. 390		-407		-	-	-		
0	. 2.86	29.114	-319	6.33	.375	5,955	-412.5						
2	-283	-	- 379	7.651	1393	7.278							
2	1285	34.515	-313	8.62	.376	8,224							
) 	1285	36.015	-312	8.82	.378	8.442	- 407	ĺ					
0 8	12.85	37.215	- 370	9,2.70.	,380	889	- 407	290.18	2.76.30	567.08	1194.75	2233-	
	1283	37,515	- 370	9.268	•378	8.39	-407						
0	•283 •285	37.812	-3200	9.260	•378		-407						
v v	-255	33,015	-320	9,500	-382	9.118	-406						
	12.85		-310	9.710	1383	9.32.7	-406						
0	12.05	34,412	-373	9.360-		••	-406						
0	1283	34.912		9.490	,350	9,110	- 406						
0	. 28-4	41.217	_	11.400		11,014	-403						
0	12.85	•	-367	11:220	,336								
0	285	38:815 37:415	-368	11.080	. 385	10.695	- 403	296.96	277.36	574,32	506.87	67,45	
0	1238	37.58%	- 370	11.050.	.384	10.666	- 403						
	1239	37.911		11,172	1385	10, 184	- 1103						
0			-370	11.520	,390								
	1290	38.610	-369	11:550		-	-402-						
	1.14	40.18.5	- 368	16.439	1427	16,012	- 396-						
	• 296	38.947 -		-				288.41	272.60	561.01	506.87	54.14	
	- 241	44,459		16,220	• 423	15,2.97		295,27	2.73.2.0	568,47	514,50	53.97	
	•211 •291		- 36.4	19.219		-18.793							
	12.99	59.099	-	16.410	e 428	15.982	-395						
	. 2.98	53.600		24,651	• 458	24.193							
S	\$ 2.48	149.500		24.178	,449	23.729	-385	292.60	271.30	563,96	514.50	49.40	
			-359			22.189							
		48.96.4	- 359 - 360	22,461	144.1	-							
		40.764 54.498		22.473		22.031	-						
		56.997	-355	-	,460								
ł		50.598				24,578		296.20	264.20	559.40	914.50	144.90	
		~		-		24,247							
		49.797		24.695		24.244							
			- 527	24.820	1435	2.4.365	- 384						
							I						

TOTAL OF FRANK

\_ · · · · 

#### PAGE D

HYDROGEN TANK 8512

3400-250338-01=0601

				i			WTM				
· · · · · ·	DATE	TIME	ELAPSED	∆ Timé	WTM	AWTM	BARO.	BARO.	ToF	T	K
	MAY 28	09:11	0	~	0.0		27.02	<b></b>	70°		
		11:11	2.0	2.0	36.56	36.54	27.01	27.015		20.0	. 815
		13:11	4.0	2.0	120.00	83 4.4	27.00	27.00	69 "	69.5	18171
•		15:11	6.0	2.0	205.09	85.09	26.97	26 19		690	.516
	29	08:11	23.0	17.0	937.63	732,54	27.06-	27.01	68	68,5	.818
	•	9:11	24.0	110	918.50	110.87	22.07	27.06	694	69°	. 819
		10:11	25.0	1.0	1020.60	42,10	: 27,08	27.07	68'	68.5	. 8214
• •		11:11	26.0	1.0	1064.45	43.55	27.08	27,08	45	65.20	1821
		12:11	27.0	1:0	1108.22	43.7.7		7.7.07	6-80	68	.821
•		13:11	28.0	1.0	1.151 .34	43.11		27.05	680	68	,8216
		14:10	29.0	1.0	1193,17	÷		27.05	· 68°	6.50	,821
•		15:11	30.0	110	1232.58	39.41		27.04	68'	68	. 320
		16: 11	31.0	1.0	1272.74	40.16		27.03	69"		, 818
	30	08:11	47:0	16.0		643.65		27.06	6.2"	63.	.828
		9:11	1 48.0	1.0	1	43.79		27.11	620	620	,836
		10:11	49.0	1.0		41.67		27.10	63	63	, 833
	ļ	1 11:11	50. 0	1.0	2042.31	40.46	27,10		65	6.4	18-311
		12:11	51,0	1.0	2081.70	39.39	2709			k k	,826
		14:11	53 0	2.0		38.96	27.07			6.7	, \$21
		16:11	55.0	2.0	2239.82	40,16	27.06	2706	70	69	. 819
	JUNE 2	08:11	119.0	64,0	4761.58	2.527.26	27.04	27 05	6 l	65	. * 27
		TOOK .	SCALE	WTS	REMA	TED G	ROF C	n' Vien	T ,	INSTALL	-U BA
			UNIT			• •					
		1130	RECE	FIUED Y	ASTALL	ED BAI	ATAYON F.	RESSUR.			
		1300	0		0.0	<b></b>	2763	22.03			
		14:15	1.25	1,25	18.0	148.01	27,02	27.00			182
		16:15	3.25	210	12.3.5	7 75.56	27,00	27.01			,82
	3	08:15		16.0		7 714.10					181
			2/125			9 76.7.7	26.93			59	1831
		12:15	23.25	2.0	1016.72	5 78.34	24 . 4.3		~	60	. 53 ;
		14/15	25.25	2.0	1033.4	6 77.17				62	, 831
	l ,	12:15	27.2	2.0	1	- 14.5c			le et	ر. ج	. 22
	; 4	+ c3'15				15:11		5 62 1			22
AME /	:		45.2.5		1894.						, 83
		12:15	17.2	5 2.0		1 18 18 1			; ;		. 52
•		14:15	1 49.25	5 2.0	204/11	6 13.58	26.96		65		52
		16 1-	5 51.2	> 2.10	2/13.4	12 26	26.93		67		, <sup>3</sup> <b>7</b>

FOLDOUT FRAME /

.

ORIGINAL PAGE IS OF POOR QUALITY

1 <b>993 (</b> -1993) 1			ŵ				<u>STAR</u> BARO.	Tor	K I	۔ ت		l
;+-	Ŧ	K										
<u>c</u> , [	20.0	. 8151	,0836	READING					1 1 A	1.1	7°	
Ü	69.5	18171	,1912	, NO.			20	F77		1100-2		
C.	690	15-16-2-	,1948	•	-				7-0-1			
	16 rel	,8185	1973	1.								
C	69°	. 8190	1878									
, e 	68,5	1.8216	1941		HRS	0,0		-				
	68.20	18219	, 2.022		6	. 83,44 -	19.00	690	,8171	1912		
6	68		,2017		24	168.53	26.995	690		, 1979		
/ C	68	,8210	,1986		21	•		690	18174	, 1967		
С. с	68° 68°	,8210 ,8207	1926		22	941.94		690	18177	1964		
, c 7 <sup>c2</sup>	69	, 8180			23	984.04	27:03	69°	18180	,1963		
2."	65	,8281			24	1027 89	27.04	69 6.	,8184	1966		
2 <sup>i</sup>	62°		. 2055		25	1071.66	27:04	690	18184	,1968		
3	63		,1950		24	1114.78	22,04	690	,8184	1968		
5	6.4		,1583		27		27.84	690	, 8184			
4	66		.1827	7	28	1196.02	27.04	69°	18184	,1962		
ç	67	.8242	1796	, ,	29		27.104	67	18154	,19-24		
'C'	69		1837		45		27.105	68°		1926		
1	65	, 8275	.1823	2	40	· · · ·	27105	67°	8233	1923		
					4		07.062	670	,8236	. 1931		
- 14	NSTALLE	O BAK	CSTAT	<i></i> .	4	8 2005 ~~ 9 2045 14	27:04	670	.8236	1928		
	_	1			5		\$ 27.06	6.70	,8236	:1924		
540	CUT.			NEW SET	-un r	3. 2.2.03, 2.4		67°	18236			E RUA CH
50				Take TH J-16 2		7 4731.03			187.36			Van Ser OP
		Ball	6.172		1.3	10			18246			
6	•	1027	0. 174	46	3.		27.01	66		.1738		•
7- 8	62.	10 47	9 ,203	8	19,	25 837.67				,2027		
6 (1	59		3 ,211			25 997.89		. £		.2039		
· (	60		9,183			2.5 1006.23		r (		,2020		
:3	62	1	4 , 179			25 108314				12002		
64 	-	1 ·	3 ,17%			. 2.5 1159.90			13316	1 , 1986 . 1962		
62	43		5 , 192			,25 1819.03			18514	. 1953		
6 )	43	1 831	7,175	8.	45	251 394.4	0 26 45 2 21 95	6.3	15319	1437		
65	<b>~</b>		53 , 16			125 204/.1		<u>(</u>	1329	1 . 1929	34 3	
65			0 .170		1 119	こえち ムレアハイ	12 76 95	ر کما		1 . 1418	~	

1

ORIGINAL PAGE S

25-11602 Hage A-11

HYDRUGE 3400-

TANK	PR
------	----

			1	GAU	GLS	BARATIZON	LON	HUMP	
DATE	TIME	ELAPSED	IBAKO. "HYAN	FILL PRESS PSIG	VENT PILLS. "Hy	VILINT PRESS "HIG	1	MICRO MAP	TICC 65 TURR
MAY 28:25	08:05	12.1	27,02	212.5			11000	14	15.10
	11:11	2012	27.01				3900	13	2410 1
	13:11	22.2	27.00	-	.grsic		3900	10.5	1 × 10 -
	15:11	24.2	26.97	-	190516		, 3930	10.	1 510
29	08:11	41.2	27,06	-	19		3950	12.	2110-
-	<b>6</b> :11	42.12			-		3930		1 110-
İ	10:11	43.2	27.08	-	19		3900	10	1.10
	11:11	44.2	27.08	-	• 9		13900	818	9 × 10
	12:11	45.2-	27.06		09	-	3950	7.	7×10-
	13:11	46.2	27.05	-	09	- 1	3900	8.	8×10-
	14:11	47.2	27.05	-	- 9		3950	6.	6×10
	15:11	48.2	27.04	<u> </u>	, 9psis	. —	3900	11.	1 × 10 -
	16:11	49:2	27.03	, 8.5	115 kg	· ·	3950	12.	2. × 10
30		65.2	27,10		1,48	~ ]	3900	H.S.	318×10-
	51:11	66.2	27,11	i	1.48		3900	4.5	4×10-
	10:11	67.2	27,10		1,45	-	37100	,1,5	4110
	11:11	68.2	27.10	•	1141		3950	6.	GX10-
	12;11	69.2	27,09	,78	1.40	-	3 <b>90Q</b>	3.5	5×10-
1	14:11	7:12	27,07	,78	1,40		3900	16	1410-
1	16:11	73.2	27.06	,78	1,40		395C	9	9×10:
JUNE 2	8:11	137.2	27.05	172	1.2.8	- '	3830	Ø	SX 10 -
1	13:15	1+2.25	27,03	,74	1,45	1.3830	3830	8-	5.416 -
ļ	14:15	143.25	27.02	,74	1,40	1,3900.			• ••
۱ ب	16:15	145.2.5	27,00	175	1.40	114090	3900	2	8.5 ×16
3	8:15	161.25	26.94	,77	1.50	1,4600	3850	6.8	5.5×10
	10:15	163.25	24.93		1.50	1.4630	3850	65	55410
		165.25	24.93		1.50	1.4610	3400	5,	5 + 10 -
ł	14:15	167.25	26.92		1.50	114740	3900	5 5	5.5×10
1		169.25		•	1.60	114940	3900	5:5	5.5110
4	8:15	185.25		176	1.45	1.4372	3850	5.5	3×10-
		187.25	24.96	17.5	1.40	1.4162	3850	3.5	3210-
		189.25		17.5	1.40	1,4110	3850	5.5	5 5×10
		191.25		,	1.45	114200	3850	6.5	6 8 18
	16:15	193 25	24.95	176	1. 45	1.4361	3850	6.	6-× 10-
5		209.25	F	172	1.40	1,3625	3.830	5.	1 810
	10:15	211.25	27.03	.71	1.38	1.3570	3900	3.5	5 210
	12:15	2.13.2.5	27,03	171	1.35	1.3520	39,00	14.5	3.5 + 10
1	14:15	215.25	27:03	171	1.35	1,3520	1858	5	1 + 10 -

FOLDOUT FRAME

1

ORIGINAL PAGE IS OF POOR QUALITY

#### HYDROGEN TANK 851240=1 5/N1 3400-250338-01-0601

ION	PUMP			Anna a Anna an Anna an An	<b>TILLER TO AND AND THE</b>	ali kur sili kur sani asul kur di kur di kur si su da da kur	an ann an	1994 - 1899 - 1995 - 1202 (1874 - 1875) 1995 - 1997 -		<b>RE-ENDINGINAL PROPERTY OF</b>
Vocis	MICRO-MAP	TO ICR		4		1	 		i	
4000	14	15.10 6					1	1	]	ł
3900	/3	3110-6		÷	1					
3900	10.5	1×10-6								
3950	10.	1 × 10 - 6								
3950	. 12.	: 110-6								
3950	11	1×10-6								
3.900	10	1×10-6								
3900	818	910-								
3950	7.	2×10-7			•					
3900	8,	8×10-7								
3950	· 6.	6×10	1							
3900	11.	1 × 10 - 40								
3950	. 12.	2-×10-6								
39 00	4,8	318× 10-1								
3900	4,5	4×10-1								
37100	i 11,5	4×10 ?								
3950	Ŀ.	GX10- (								
398Q	5.5	5×10-7								
3900	10	1×10- 0	-							
3950	9	9×10-7	1						-	
3830	Ø	SX10 - 7								
3850	8-	5×10-7								
-	-	·								
3900		8.5 ×10 - 7	,							
3850	6.8	5.5 × 10 - 1								
3850	65	5 5710-1								
3400	5.	5×10-7								
3900	55	5.5x10-1								
3900	5.5	5.5110-1	2							
3850	3.5	5x10								
3850		3×10-7								
3850	5.5	5.5×10-								
3850		6×10-1								
3850	6.	6×10-7								
3850	3,	2×10-7			1					
3900	3.5	2 2 10 - 1								
3900	4.5	3.5×10-7 -5×10-7								
3850	5. 5 i	5×10-7								
3400	/		I							FOTT

FOLDOUT PROJECT

Parte 7 3

> 75-11607 Page A-13

HYDROGEN TAN

			[	FILL	TEMO	)	1	diliri	N 771	110
DATE		LAPSED TIME		FILL FBC	TEMP		\$2.4. WE VE 1821 194.4	SHIFI.		
DATE	TIME	1	1		1	1	To h			12-
JUNE 5	08:15	209.25	3.699	. 17.3	2.926	- 42.2	59.400	.307	54.043	- 35
	09:15	210.25		. ·		· ••••				
	10:15	211.25		.760	1	-42.2	55,300	1305	54.995	- 35
		213.25	3:565		2.951		54,100	.307	53.793	- 35
		215.25				- 4.22		1308	53.892	- 35
-	16:15	E	3.548		1	- 1122			34.2.90	
-	8:15	1	3.551	• 6.4 *	J. J. C. 14 1	-422	louiste co	n 1141	61.286.65	
	9:15	235.25		1.40	1.5 (2.7)					•
		236.25			2.457	-4122.5	51 700		51.58C	
			07ATE		ANK.	700	5 7. Car	a )/(,	58.484	24
	14:45				21,567	10 - 25 k	100055	2.0	117.912	
	8:45	1			48.348		48.7.80	1308		- 30
1	11:30		48,452		48,123	- 36 /	701700	·334	20.366	-39
	13:30				1 48,113		61,600	, 328		-3. ~3
	15:30				11-8,389		61.900	1318	01.372	ر 3-3
10		329.5	59.800		58.691		60,970	1328 1343	60,642. 21,747	ر د
<i>,</i> -	10:30	•	56.509		55,395		66.470	1.340	66,130	- 3
		333.5	55.500	10117	54.383	-3.56	66.580	1340	66.245	-3
		335.5 -	5,5,517	112.0	54.397	~ 355		340	46.260	 3
	16:30		56.200	1.121	55.079	- 355	68.300	341		- 3
1			86.221	1.057	85.164	- 32.7	97.201	.377	96.824	-3
		_			Norm				ipiet 1.	
	13:30	355.5	3,739	1809		- 422	11.9.989	- 311	149.626	- 3
	15.30		3.189	,850	2.939	-42.2		.319		. 3.
12	8:30	377.5	3.803		2.935	-422		131.8	16.012	
	10:30	379 5	3.810	, 82 8	2.942	- 4222		.324	63.371	- 3
	1	ROTA TZ	D UNI		PROX.			THRU	Suppr	
	12:45		44.368	16.38	43.730	-36.5	53.055	, 587	52.301	-3:
	14:45	383.7.5	116.200	.637	45,43	-363	54.013	. 623	53, 390	- 3
13		401.75	69,339	16.59	68.618	-34.2	81.172	16-53	20 314	3
	10:45	403.75	61.159	165k	60,503		65,700	.646	65.054	- 3
	12:45	405.25	57.416		56.966		66,075	, 648	15.427	- 34
		467.75	52.300	1644	56.651	1.4.5	al Per	, ( ~ <sup>1</sup>		· .
16	8:4.5	473.75	110.70	,82	169.85	-254	214.30	132	213.48	2
	10:45		191 32		19046		2.46.75	- 1 G -	246-1011	- /
		41917.25		170	2.11.16	I	288.60	193	2.57.6.7	1.
i	14:45.	479.75	231.70	.76	230 .94		330.0.3	.97	327.04	- 11

## ORIGINAL EACH AND OF POOR QUALITY

FOLDOUT FRAME /

DROG	REN.	TANK	( 8.	5124	0-1	- 57N	1				ËAGE	1
SHIEL	DTEN	1P.		VFN	T TEI	MP.		LIGU	ID W	EIGH	T	(4)
The St	FA FC	Teme Teme	73 A 1	13 n=	JK-XL 73	76114! T3 9	SCALE SCALE	SCAL F 02 WT	GROSS	TAR.	1.19010 1.19010 1.19010	an a shippe buyer, some bala per -
.307	59.093	-352	30.600	.1182					• • • • • • • • • • • • • • • • • • •		- brit - 6664	ann an tha tha an an an an an a
~	1						288,00	269.40	557.40	517.14	40.26	
1305	54.495	- 355	28.902	. 470	28,432	-380						
.307	53.293	- 356	28.820	.472	28.348							
1308	53,892	-356	28:990	1423	28,517							
1310	34.290	-356			2.8.843							
· 314	61.806	- 350	35.170	• 5/Cs	35.260	-312						
		<b></b>				، المسرُّ	284.15	266.60	552.75	517.14	35.61	
	54.380				33025		<b>1</b>					
0316	58.484	-353	33.510	0499	33-01	-375				TUNE		
							272 34	256,60	527,94	TAKE UNITU.		
	117.972	- 362	9.580	c 4 31	9.149	F .	270,80				33.0	
1334	20.366	-340	36.100		35.521	a			514 46		*20.5	
, 328	61.272	-350	31.992	1542	34,450		265 5%	2.47.51	507.55		\$ 14.0	
1328	01.572	-349	31.520	1539	30,981	-37.7	•					
,328	60,642	-352	31.539	1538	31.001	-327	<b></b>				ц:	
1343	71.747	- 330	42,990	.612		- 3/45	256,83	247.10	503.93		* 10,0	
340	66.130	- 345	119197	1591	391606	- 368	-					
1340	66.240	-345	39.350	15 PC	38.794	-370						
340	66.260.	-345	39,600	1589	39.011	-36.9						
1341	• •	- 343	140,300	, 594	39.706	-368					¥.	
, 377	96.824	-318	61.501	,119	60.722	-350	251.50	248.14	499.64		* 6 <b>.9</b> 5	and an
		9010 160	1							REFFR		SEFORE FILL
,311	119.6.26	- 359			25,217	-38.3	2.5085	7.31.61	482.46		18.75 A	CCFC
.319		356	32,589	1506	32.053	- 375					.*	
	66.012	-345	43 140	.557	42.583	-364			479.71		* 20.0	
.329	-	- 347		1555	41.595	-3665	245.30	249.19	497.99	1-1111 1	per lier	FRUCKIE.
	•	y Port		_								
. 587	52.501				21931							
+ 623		- 356	1		21913							
1653	30. 519	-332			2.910			245.95	493.50		19 6	
.646	65:054	-346	3.884			-422	1				<b>1</b> -	
	65.427	-346			21884			2.45 6.0	4 12 35		** 59 s*	
	12 512	- :46		1287	2595	- 422	1'				*	
182	213.48	- 2.15	4.64	2.07	2.1.59	- 2.59	2.40.82	241.00	481,82		* 3.0	
	246.07	,		2.34	2,59	- 259	1					
	2.87.67		3.29	2.6.5	2.61		PACE T.	MAL Fil	1. ICAT	•		
.97	327.04	-113	5.63-	3 00	2.63	-263	l					
			l				l					T. C. (19)

•

.

.

.

11日1月月1日1日1日1月月1日

ħ

11

1.4

75-11607 Dage A-14

HYDROGEN TANK 85124 3400-250338-01-0601

Dere	T.	ELANSED	en meneralisaria 1 a		I miner	BARC	WrM			1
DATE	TIME	TIME	and the second sec		AWTO	1 11/8 1	BARO	IT?	Ŧ	1
JUNE 5	08:15	1			552.8	2- 27.0	3 26.99			
	10:15	1 - 1 - 1		2740 168	74.44	\$ 27.03		64°		
		71,25		2.810.87				65		
		73.25		2880.90				61		
	-	175.25		2950.67	r 69.7	1 27.03		68		
		91.25		0.11.0			2.1.07	64	66	
	10:15		2.0	36-15.94	1 66.19	27.10		46		
	11 ; 15	1	1.0		33.92		1	67		
	13:45			are Ro	TATE	0 90	W	TAI	KI-AMIN	
	14:45		1.0		41.33		•	20		
	8:45	Colic.	lile C	2833.39			2.7.08	65	67	
	10:30		1.75	liene	KED S	ATURAT	W - CHAI	GED LUN	IGHTS CA	12
	13:30		1.0	\$2.85	52.53	21,05		6-14		+
	15:30 :		2.0		64.39	100		6.4		,
	8:30		2.0	164.19				6- 14		,
1	10:30			763.36	\$99.17	2.7.05	27.03	6-1	63	1.
	•	94.75	2.0	876.35				6.4		
1		96.75	2.0	889,57				6-60		
1		98.75	2.0	953,42		-		67		
		1.4.75	X.C	1018-27		27.04		67		
	THN	to Kaiz	10.0	1613.51:	375.24	27.10	27.07	60	64	- 1
	11:30	0.0		CIC N	A Chip	AL fost	TION.	hEMAL.	KINE LIG	61
		2.0	2.0	75.35	116 21				-	
1		4.0		146.168				47		1 (
12	8:30 1			761.94	(0100	27.05		67		1
1	0:30		2.0	826.55	CA 91	27.08	-7.08	63	6.5	÷
	;	ROTATE		T APV	22-6 1 1			65		•
1	0;45	010	0.0	OIC			000 1200	1 Mil	5. M	4
1	2:45	20	2.0	68.66	68.66	27,10		/ · 3		
1	1:45 .	\$10		131.07	67.41	27.09		67		(
13	3145.	12.0 ,		-	624.45	2.1.1.3	27,11	67 61		1
/	6:45	24.0		825.68	10 36	27.14	r ( ( ) )	6.64	64	, 8
	14.5			890 81	65.13	27.1.5				•
	-			958.24	67.45	27.14		66		• 1
14 8		4.0		39.38.70 ;		17.06-	27.10	6.6		- 5
				4 028.91	90.91	21.04-	A110	62	6.4	κ ۶
12	: 4.5 9		2.0	4117.14	35.23	17 11		61		, i
14	:45 10	0.0	2.0 4	+201.46	\$4.32	27.03		li 1 6 1		·

ORIGINAL PAGE IS OF POOR QUALITY

FOLDOUT



Ĵ

]

7ANK 851240-1. S/N 1 250338-01-0601

ŀ

ł

PACE 2 W'= Kenne AWTM

		r	$w^{\circ} = \frac{K e_{\text{FID}} A W T M T}{A T T M F} $ (5)	
	Ŧ	K ŵ	A TIME A WIM BARO TOF K. J	
	6.50	18259 1601	67,25 2666,24 26 96 63 18295 1845	
	-	,8294 .1732	69.25 2740,60 26.96 43 ,82.95 ,1542	
		18272 1629	71.25 2.810.87 26.97 63 ,82.95 ,1837	
		,82.2.6 ,1616	73 2.5 2880.90 26.97 64 18225 1826	
		18204 11606	75.25 2950,67 24.98 64 18279 1821	
	66	182.62 11736	91.25 3549.75 26.98 64 18279 1807	
		,8271 ,1534 SHE LOU	· 93,25 3415.94 24.99 64 ,8282 ,1802	
		182-51 11570	94.25 3649.86 27.00 64 18285 1800	
R	=APING		· · · · · · · · · · · · · · · · · · ·	
		18176 12825		
	67	18242= 1936		
161	ts cri	BAROSTAT. SET WITH TO	500	
		18301 ,152.9		
		18294 ,1498		
		, 32.88 . 1557		
	63	18317 1644		
		18301 11467		
		, 82.58 , 1465		
		, 82-24 . 1473		
		. 87.7.9 . 1497		
	6.4	, 8307 , 1733		
c.,	ve Lig	VID HUDED. W.TA	ZEROED.	
		18248 , 17,43		
		18242,1650		
	65	18287 ,1699	WITM REMOVED FOR 10 MIN, ADDED 5.94 TOTOTAL FLOW FOR ALFORDE.	
		, 82.87 , 1509		
1.	Surr	4 KINE WTOTA	EROFO.	
		·		
		18248 ,1589		
		182.45 1444		
	44	18319 1/618		
		. 8329 , 1649		
		.8387 .1514		
		.8284 .1568		
	6.4	(8316 , 1840		
		8370 12119		
		18376 1	FLOW THRU FILL FORT.	
		18361 , 1995	)	PULI

75-11607 Page A-15 I.

-

} :

•				•	_			0338-01-06	0/	
					TANK				0	
			CIAPCED	BARD	GALGE	VENT	BARATRON		Pum	- P7
DAT	F	TIME	ELAPSED TIME	IHSA.	PRESS PSIG	PRESS 'Hg	PRETS. 114 2		MILRO AMP	1 1
JUNE	6	08:15	233.25	27.10	,68	1.30	1.2910	3900	3.5	2.3
			235.25	27,10	,	1,53	1.5250	3900	11.5	32
			236.25	27.11		11.50	115250	3900	4115	31
			23.9.75	27.0%	67	1.30	1.3120	3900	50,	4X
			240.75	-				3706 -	=	2.1
	9		305.75.		1. 69		1.3310	3900	35,5	2.51
			30815		703/100	1	0.5405	3900	19.	91
			310.5		7.3 H20	1-	05515	3850	115	OX
		•	312.5	27.01		1.55		: 3900	23.	1.5
	10				7.4	.50	0,5210	3900	38.5	2.7
	1		331.5	27.05		1.50	0.5210	3850	19.5	.9
		12:30	333.5	27.06		,50	0,5210	3900	17.5	16
	1	14:30	335.5	27.03			0.5485		17.	.6 1
		16:30	337.5	2.7.04	-	52	0.5400	3900	15.	.51
	1	8:30	353.5	27.10		•47	0.4800	3500	Net STA	
			1				To NCA	MAL POSITIO	л т Г. 16.	
		13:30	358,5	27,10		· •	0.26.20	3910		CX.
		15:30			5.06	2.6 "	0.2700	3900	//	OX.
		8130	377.5	27.08	4.7.7 "	2.4 "	0.2550	3850	14.	51
		10:30	379.5	27.08	4.8	2.4 "	0.2610	3700	13	38
			TA.	rK No	TA TEO A	SUPPLY	170 1-1	OUL THRU SU.	PPXY FO	CR I
			381.75	2.7.10			0.24.50	3800	90	91 •9
			383.75	1	4.86		0,2510	F	19	
	13	8:4.5	1 -	27.13			0.2500	3700	2.2.0	141
			403.75	27.14			0.2.2.50	3900	32	2.8
		2	405.75	27,15				3850	2.2-	1.
			407.7.5	27.14	4.6. 4		0 2250	3400	25-	.5
	14	8:45	473.25	27.06			0.3140	1900	2.5 MA	
							03080	1550	29 NA 31 MA	
			477.75				w C.303C.	1400	31 MA	
		14:45	#19.75	27.03	, 30	· 6.10 '	· C. 3:60	1 300	51 11 4	' <b>í</b>
						1				
4						•				
						:				
t				1						
j			;			i		4 1 1		
(CD)		I	,		1	ł				
GF, IS				1				1		
LITY FRAME /										

HYDROGEN TANK 851240-1 5/W 3400-250338-01-0601

ORIGINAL PAGE IS OF POOR QUALITY FOLDOUT FRAME

.

GARHETT AIRESEANCH MANUFACTURING COMPANY CF CALIFURNA

:

240-1 -01-0601	5/w 1.		PAGE X (6)	
<u>VOLTS MILRO AM</u> 3900 3.5 3900 4.5 8400 2115	PRESS.	- SFE LOG PAGE #30 TO EXILAIN TANK I	RESSURE RISE	
3900 50, 3700 345 3700 35,5 3700 19, 3850 115	4×10-6 2×10-6 25×10-6 9×10-6 0×10-6	TANK RETATED.		
3900 23, 3900 38,5 3850 19,5 3900 17,5 3900 17,	1.5×10-6 2.7×10-6 .9×10-6 .16×10-6 .6×10-6	•		
	.5×10=6 ВБЕ-122Е0 схю=6 охис=6 5хю=6	WITH LIPAID NE HAD LEFT. 1.	SLBS REMOVED FROM BAROS TAT	
3700 13 0 SUPPXF 1 3800 90 3900 19 3700 220	3×10-6	FORT IERS 5.52/120 5.68 " 6.3 "		
3900 32 3850 24 3400 25 1900 25MA	2. ×10 - 6 1×10 - 6 .5×10 - 6 7. 1.5×10 - 4	5.3 " 5.3 " 5.35 <sup>[]</sup> 9.30 <sup>[]</sup>		
1400 31 MA 1400 31 MA 1300 31 MA		9.50" 5.23m FLOW THRU FILL FORT. 5.36 "		
				on <u>er e</u> rvet

75-11607 Page A-16

	7	IME		(			$\hat{\mathcal{O}}$	•	dur		
•		and the second		TI A-B	FILL	TEM		- IF	SHIEL	DE	M
•	DATE	TIME			TBC	TI AB-BC	T. Tem. "			Ta 25-	2
	679-75		0.0	3.664	.775	2.389		39.500	7	31417	
		10:45	20	3.673	;	2.908	- 422	39.400	1280	29.12.0	1
		12:45	4.0	3.693	,778	2.915	-422		,251	34,119	1
		14:45	6.0	3.686		2.909	-422	40.066		39.785	1
	20	8:45	24.0	3.682	•	1	1-422				7
		10:45	26.0	3.700			-422			39.714	]
		12:45	30.0	3.697			-422	39-710 40.490		391422	1
	23	8:45	960	3.691	1	1	-422			110,202	
		10:03	UNIT	TOPP	•		LIQUIC			Ker 170	0
		12:45	0.0	4.000	•		-4/95			45.647	
		14:45	2.0	3.727	•		- 421	49.407		48.724	
	24	8145	20.0	2.500		7,052	- 409,5	82.000	,712	\$5 2.88	1
		10:45	22.0	7.530	1460	7.070	- 410	77,870	1707	77.163	4
	1	12:45	24.0	7.640	1467	7.173	- 409.5	18.425	691	67.728	
	1	14:45	26.0	2.930	1467	7,463	- 408 . 5	68,420	169.5	67.12.5	•
	1	8:45	44.0	10.8.40	•	12.34	-405.5	SC.30?		2. 5.4	
		10:45		10.520	•	10.058	- 404	63.400		62.710	
		12:45	48.0	11.120		10.715	- 403	63 605	• -	02.916	
	i	111:45	30.0	11.326		10.913	- 402.5	61.454		65 765	٦
	26		168.0	12.203		12.748		61.160		1 163	4
		10:45		13 252		12.842		58.805			1
		12;45		13.774		15 343	-3.19	2		1.1.1	
	27	14:45 8:45	92.0	16.365		1 3 640	- 344.8			61 407	
		9:45	93.0				- 3763			57 045	
		12 45	96 0				-3145			58,200	-
		1- 12	•••				-0740			Q + m	
	[					1		ł			
						•					
						٠					
	1										
			!								
	,			t				İ			
FOLDOUT FRAM	E /										
- TAL PA	GED										
ORIGINAL IN OF POOR QU	ALITY										1
(vi 🗢 i )											
GARNETT ARESEARCH MANUFACTUR	<ul> <li>NG COMPANY</li> <li>OF CALIFORN #</li> </ul>										1
											4

			•			,					PAGE (	7)
IEL	DTE	M.P.		VENT	T TE	MP.		LIQUI	D WZ	ELGH	I	
ŝ	T2 FF-FG	Tatemp. "F-	73 JK	3 2	T3 JK-KL	To Teme "F	SCALE EI WI	SCALE #2_WT.	GROSS WIL	-TARF LUT	1.19010 NUT. 180 1.45.12	AUTIB
83	37.217	1	2,401	.378	7,023	-4095	279.76	265,60	545.36	465.90	76.46	9.26
80 '	29.120	-369	7.413	.374	7.039	.409.5			1		i.	
81	39.119	-369	7,503	1375	7.128							
81	39.785	-368.5	7.666	1376		- 409			<b>.</b>			
82	40.028	-368	9:061	,329		- 4115	277,10	263,30	540,40	468.90	71.50	4.96
186	39.7.14	- 36.8.5	9.282	,382		-406						
284	39,422	-368,5	9.432	-383	9,049	- 405						
88	110,202	-368.3	9,480	o 384	9,094		BALL hemo					
297	. 16 234	-362	16.031	1422	. 15.6.09	. 395	269.20	256.53			57.35	1417
AVI	Kox 170	· · ·		<del>.</del>	-		4	269.70			86.11	
53	45647	- 363	3. 549			- 421	2.75.09	えぶれて	553.71	468.90	84.81	1.30
\$83	48.72.4	-360.3	3 941							NEN THEC	01	tica
12	85,288	- 327	4.125	1,081			2.73.29	2.76.03	549,32	470,10	79 22.	5.59
707	77.16.3	- 325	4,070	1.030	-	- 421,5						
91	67.728	-345	4.025	,975								
9.5	67,125	- 345	4,02.0		3.054	- 421,5		<b>^</b> -			7	0.0
P05	2: 5%	- 332	1.102		. 3.061	- 4-23.5	2.6.7.44	271,50	541.24	470.16	1 1 1 4	8.08
690	62.710	-348	3.96.3	. 148		-421.5						
689	6-2.916	-348	3.95.5	1942	•	-421.5						
639	60.265	- 34 8	3:955	1941	3.014				and a start		1 4 13	
672		- 347	3.960	19419		-4215	2.66.2.0	265.15	334,55	14.20,11	6	6-36
692.	5811.3	-350	3 9 4.9	, <i>Ŷ3</i> 4		- 42125						
644		-350	2.933			-42.6.5						
623		-35C	3. 4. 11	. 949		- 421.5						
695		-350	3.982	157	· 3,045	-421.5		01-10	rail.		590d	574
91	57.045		3.934			5 - 421.5 []101.2	-12.5.54	265.00	56 1.14	410.10	59.04	3.77
91	58,200	-352	3.942	.945	2.997	-4121.5						
l I												•_+ ) ;
			I									

Ì

the star and the star and a

}

----

-----

1 2

1

75-11607 Pate A-17

					$ \begin{array}{cccccccccccccccccccccccccccccccccccc$					
DATE	TIME	ELAPSED TIME	ATIME	WTM	AWTM	BARC	BARO	Tenr. °F	Trop T.	- K
6-19-25	8:45	0.0	-	1.50	-	27.05		640		-
	10:45	2.0	2.0	24.94	73.44	27.08	-	60	·	,837 <b>9</b>
	1.2:45	410	210	156.60	81.66	27.08	1	58		1:443
	14:45	6.0	2.0	237,45	80.85	27.08	-	57	-	,8465
20	8:45	24.0	18.0	1080.02	842.57	27.06	27.07	54	56	18483
	10:45	26.0	2:0	1154.80	24.78	27.16	27.11	56	55	1:515
	12:45	28.0	2.0	1225,77	13.97	27.17			BAH PEALSO	. 3493
	14:45	30.0	2.0	1303,89	•				14.20	, 84-71
23	8:45	96.0	66.0		4					1:42-1
			UN.	1	5-15-12		1		CUN Y	lic T
	12:45	0.0	0.0	0.0			1	•		-
	14:45	2.0	2.0	90.59	í '					. 3300
	8:45	70.0	18.0				1		64	,8338
	10:45	22.0	2.0	1						· 8599
	12:45		2.0	1292.67	1.	-				- 8 <b>553</b>
	14:45	26.0	2.0		i					1 852 <b>5</b>
4	8:45	44.0	18.0	1					55	13556
•	10:45	46.0	2.0	1	i		i			. 8584
	12:45	48.0	2.0	2544.75						, 3534
1	14:45	50.0	2.0	1	1					1.1.81
1	8:45	6.8.0			1		21.11		- د	. 471
1	10:45	70.0	2.0	1	1					1920
	12:45	12.0	2.0				I			- /7
	8:45	92.0	2.0				27 10		EI	. 8283
	9 45		12.0	111010		· /./ C	21,10	(U )	• /	• .) ~ • •
1		96.0	4.0	#573.88	81.72	27.09		66		.8275
			I		1					
										-
				1	ł					
		ł.	I		i					
	•	•								
		1		1	:					1
l			1							
				•						

#### WTM

### FOLDOUT FRAME /

RENAL PORT GARHETT A ST TAREN MANCHAUTOR (CONSAN) CAREETT AST TAREN MANCHAUTORING (CONSAN) CE CALIFORNA



PA6E (8)

	4				D	ATA FR	om STA.	RTCI	TES	T.		
=	TENP "F	K	ŵ			AWTM			K	Contraction of the state of the state	TUINI ACIS 18	<b>HECKER FARMER</b>
				ZERO TIME	0.0	-	_			~		
- ¦		,8399	1730		2.0	73.44	27.08	60°	.8399	11730	i	
	_	18443	, 1934		4.0	15510	27.07	61°		11822	·	
		, 5465			6.0	235.95	27.07	60 6	18396	1853	1	
:	56	1	2228		14.0	1078.52		58	18440	.2128	5.11	
	55	.8515			24.0	1153 30	27.05	58		,7.102		
•		. 8493	. 1763		28.0	1227,2.7	27.10	58	.8449	,2078	,	
:	BALL XEAL- 20 14:20	. 8471	, 1785		30.0	1302.39	27.10	.58	18449	.2.058		
•	6 Č	1542-1	. 18.55		96.0		•	5E	16452	. 197-3		
Ģ	UID Y	Roth	+ 7150	" ~ 11	ROX	170 0	·	-			•	
		a 117			-			-				
		, 8300		_	2.0	90.59			8300			
	64	,8338			20.0	•		64		. 2538	5.08	
		18599		•		1193.95		61	18412			
		18553	•			129267		.59	.8459			
		1 8525				1393,50		59		12544	1.00	
	55	1 8556				2352.26		58		,2545		
		, 8584			146.0		•		, 8506			
		, 8534			48.0				18509			
	55	5715-1			30.0	- 2	-	57 57	15509	17.4 0 60	571	
	77	1 52471			65.0	3512.00	· ·	58		12387		
•			1932		20.0			55	2475°			
		1 87.71			74.0			-39		.2362		
	61	. 8283			92.0			61		.2321		
		1 000.5	•••					·			5.57	
		.8275	1895		960	457384	27.09	66	8-481	. 22.67		
			.,,			10.00		•••		• •		
•												
					ļ							
		1										
		1			1							

\*

and a state of the

	•			}		1	_		0	
	7	IME	ELAPSE D			RESSU		2 CN'	Pu 2:1	
	DATE	TIME	TIME	FILL PRES			BARATRON		MICKGINGS	
	6-19-75	08145	0.0	,23 416	3.5		0.2740	3850	3.4. 21	~ X1
	1	10:45	2.0	6.85 HE.O	3.7.5 Has	l	0.3000	3300	20	3.1%
		12:45	4.0	6.80	3.75	•	0.2900	1900	24	1.7×1
		14:45	6.0	6.50.	3.15	•	6.2900	2130		.Sx1
	20	8:45	24.0	6.10	3.13		0.2430	3805	· ~	19X
		10:45	26 0	5.85	3.04	•	6-2336	57752	53	2 2.XA
		12,45	28.0	5.74	2.42	+ -	0.2270	BALL PERCIAC	23	3210
		14:45	30.0	5.67	2.80	•	C 2225	14:20 3722	27.9	2.9×1
	23		96.0	6-21	5.70		0.2830	2020	·	2.34
				HED V	ite h	1 1		1		
		12,45	0,0	CAGENG. PSIG.		3.37	0.2620	3400	2 U	1.2.1
		14:45	2.0	PS16.		4.0	0.3050	3530	۷ ،	1.51
	24		20.0	,34		. 4.5	0,3350		يك 12 مر. مورجو رافعا ومر	1 × 10
	!	10:45	22.0	127		3.8	0.2460	37.50	1	- 10 1
		12:45	24.0	122	-	3.25	0.2500	5900	25	1.28
		14:45	24.0	,20	-	322	0.2500	33.2 -	•	1.6X
	1 25	8:45	44.0	127		3.56	0.2740	3785	= 7	( Tri
		10:45	46.0	,17		2.70	0.2110	3900	; C	2.1.21
		12.45	48.0	.14	_	2.69	0.2090	1.157		E.C.XI
		14:45	50.0	116	-	2.46	0.2250	0,200	1	1.4 <b>X</b> I
	26	8:45	68.0	.18		3.50	6-2230		1 (f. 17) 1	
	•	10:45	76.0	,17	-	4.0	0.2710	2400		2414
	I.	12:45	72.0	1185	-	3.25	0.9875			. 10
	27	845	74.0 92.0	·20 .21	· _	5.92	- 103-00 - 0,310	5850	ه : سم <del>س</del>	1 ×10 3.81
	27	9:45	930	.19			0,310	2401		2.24
		12:45	96.0	.10			0 120	7830	t s a	4.54
	•	12113	10.0	0,0		<b>A</b>	0 1 / 0			7.51
	1									
	I									
	;									1
				Í.						
	1									
	,									
	1									
	1			i				! •		
			-	-				•		
- MF	)									
FOLDOUT FRAME										
R. V.										

1

Í

٣

#

.

•

CANNETT A RELEAR OF WING ST. T. S.Y. WEANN

ION 1	o <sub>um</sub> r		785007 <b>-00</b> 7-007-00-00-00-00-00-00-00-00-00-00-00-0		a nyangarik ku sharat 1 managariy	and a state of the	the sub-latence records	PAGE	9)	
YOUTS_	Muikt fam'	PATESS. TCRP				:	:			
3850	30	7.411 -6.	t	I			i			
3800	50	3.1×10-6 1.2×10-4								
3100	24	5510 6								
3830 3800	+? 32	1.9 X 10 - 60								
	53	2 2×10-6								
3700 3550 3750	27 43	3.710 - 4								
3720	24.9	2.9 ×10 - 4								
3850	34	2.3×10 - 4								
MRox 1		1								
3900	- 4	12. ×12 - 6								
3550	6.6	1.5×10-6								
3500	225	1×10-6								
39.50	10	- 10 - 6								
3960	25	1.2×10-6								
3800	u C	14.6×10-4								
3830	2.7	1.7.10 6								
3900	30	2.1×10-6-								
3150	5:5	2.6×11-6								
3700	27.	1.4X F - 6								
3500 3700	7. 8 33	3. C. A. M. C.								
3760 - 366	13	1810 " K								
		4. e								
3850	50	38×10-6-								
3900	32	7×10-6								
3850	60	4.5×10-6								
		r -								
									**	

· ·

.

n = t t gal Rage A = ta

ł