

**NASA TECHNICAL  
MEMORANDUM**



**NASA TM X-3564**

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**COMPUTER PROGRAMS FOR THE INTERPRETATION  
OF LOW RESOLUTION MASS SPECTRA:  
PROGRAM FOR CALCULATION OF MOLECULAR  
ISOTOPIC DISTRIBUTION AND PROGRAM  
FOR ASSIGNMENT OF MOLECULAR FORMULAS**

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1. Report No. NASA TM X- 3564	2. Government Accession No.	3. Recipient's Catalog No.	
4. Title and Subtitle <b>COMPUTER PROGRAMS FOR THE INTERPRETATION OF LOW RESOLUTION MASS SPECTRA: PROGRAM FOR CALCULATION OF MOLECULAR ISOTOPIC DISTRIBUTION AND PROGRAM FOR ASSIGNMENT OF MOLECULAR FORMULAS</b>		5. Report Date July 1977	6. Performing Organization Code
		8. Performing Organization Report No. E-9133	10. Work Unit No. 506-16
7. Author(s) by Robert A. Miller and Fred J. Kohl		11. Contract or Grant No.	
9. Performing Organization Name and Address National Aeronautics and Space Administration Lewis Research Center Cleveland, Ohio 44135		13. Type of Report and Period Covered Technical Memorandum	
		14. Sponsoring Agency Code	
12. Sponsoring Agency Name and Address National Aeronautics and Space Administration Washington, D.C. 20546		15. Supplementary Notes	
16. Abstract Two FORTRAN computer programs for the interpretation of low resolution mass spectra have been prepared and tested. One is for the calculation of the molecular isotopic distribution of any species from stored elemental distributions. The program requires only the input of the molecular formula and has been designed for compatibility with any computer system. The other program is for the determination of all possible combinations of atoms (and radicals) which may form an ion having a particular integer mass. It also uses a simplified input scheme and has been designed for compatibility with any system.			
17. Key Words (Suggested by Author(s)) Mass spectrometry Computer program Isotopic distribution		18. Distribution Statement Unclassified - unlimited STAR Category 61	
19. Security Classif. (of this report) Unclassified	20. Security Classif. (of this page) Unclassified	21. No. of Pages 17	22. Price* A02

COMPUTER PROGRAMS FOR THE INTERPRETATION OF LOW RESOLUTION  
MASS SPECTRA: PROGRAM FOR CALCULATION OF MOLECULAR  
ISOTOPIC DISTRIBUTION AND PROGRAM FOR ASSIGNMENT  
OF MOLECULAR FORMULAS

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SUMMARY

Two FORTRAN computer programs for the interpretation of low resolution mass spectra have been prepared and tested. One is for the calculation of the molecular isotopic distribution of any species from stored elemental distributions. The program requires only the input of the molecular formula and has been designed to be compatible with any computer system. The other program is for the determination of all possible combinations of atoms (and radicals) which may form an ion having a particular integer mass. It also uses a simplified input scheme and has been designed to be compatible with any system.

INTRODUCTION

The mass spectroscopist is faced with several laborious tasks when he is attempting to identify an ion from its mass and isotopic distribution. One task is to select all ions which may give rise to a particular mass. Another is to calculate the relative molecular isotopic abundances of these ions for comparison with the experimental distribution. Two new computer programs which may be used for these tasks are described here. These programs are easier to handle and more encompassing than previous methods and computer programs (refs. 1 and 2). The ease of use of our pro-

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grams arises from having all necessary atomic data contained within the program. Moreover, the programs are easily adaptable to most computer systems.

## CALCULATION OF MOLECULAR ISOTOPIC DISTRIBUTIONS

Grimley (ref. 3) has presented an algorithm which may be used to obtain molecular isotopic distributions from known elemental distributions. With this method, two elemental distributions are first combined. The resulting diatomic distribution is then combined with another elemental distribution to construct a triatomic distribution. This process is repeated until the distribution for the desired ionic species is obtained. This method, although tractable for hand-calculations of small ions, becomes very tedious for larger species.

For multiatomic species high-speed computer techniques are most convenient. For this reason we have prepared a computer program which may be used to calculate these distributions. In order to encourage the use of this program over hand calculations, thereby avoiding the associated error and tedium, a simplified input scheme is used. The only input required is the molecular formula; all other information is stored permanently within the program. For example, the distribution for the molybdenum trioxide trimer molecule  $(\text{MoO}_3)_3$ , would be obtained after entering MO 3 O 9.

The output of this program is shown in figure 1(a). It consists of the molecular formula, the individual elemental distributions, and the molecular distribution. This molecular distribution is displayed both numerically and graphically; the latter display is especially useful for rapid comparisons of experimental and calculated distributions. The run time for this calculation was 1.2 seconds on the IBM 360. A second example, for dibromochlorotrifluorobutane,  $\text{C}_4\text{H}_4\text{Br}_2\text{ClF}_3$ , is given in figure 1(b).

The few other programs available for this type of calculation (refs. 1 and 2) all have limitations. Of the three programs given in these references, one is limited to a few atoms; another requires the input of atomic distributions for each atom in the molecule; and the remaining program requires the input of a complete table of distributions with each run. Furthermore, all require the input of additional data such as the atomic numbers and the number of atoms.

A listing of our program is given in appendix A. This program has run successfully on the UNIVAC 1110 system at Lewis Research Center and, with a few modifications, on the IBM 360 system. This program is called DISMIS for distribution of molecular isotopes. It is coded in standard FORTRAN IV and can be readily adapted to both time-sharing and batch-processing systems. Moreover, its output can be easily changed from the 129-column form to a 79-column form for teletypes. The required modifications are given in comment cards located within program listing. The ele-

mental isotopic distributions of all stable elements are contained in a subroutine called DISEIS. These data were taken from a recent tabulation by Roboz (ref. 4).

The program is set up to accept molecules having up to 10 different atomic species. The molecular formula, which is the only input required, is entered in 10(A2, I2) format. The common symbols for the elements are used, and those symbols and subscripts having a single character must be preceded by a blank space. Sufficient memory has been set aside within the program to calculate the distribution for virtually any molecule which can be detected mass spectrometrically.

### ASSIGNMENT OF MOLECULAR FORMULAS

The technique described here can be used to prepare a list of those species, possibly present in a system, which have a specified mass number. For example, in figure 2(a) all possible combinations of K, Cl, Cr, and O atoms which have a mass number of 294 have been determined. From this list one may choose those species which are intuitively reasonable, and then further analyze them using the isotopic distribution program. For the example in figure 2(a) the best candidate was the potassium dichromate ion,  $K_2Cr_2O_7^+$ . The run time required for this calculation was 1.2 seconds on the IBM 360. A second example is shown in figure 2(b), where the most reasonable choice for the mass to charge ratio  $m/e$  118 was  $CrO_2(OH)_2^+$ .

There have been a few tabulations of molecular formulas as a function of mass (refs. 5 and 6), but these are limited to four and 16 elements, respectively. There are also a few computer programs available (e. g., refs. 7 and 8). However, of the three programs listed in these references, two are designed for high resolution, generally organic, spectra, and the third is written in BASIC. All are limited to a few atoms.

A listing of our program, including pertinent COMMENT cards for the conversion from batch processing to time sharing computer systems, is given in appendix B. This version has run on the IBM 360. The mass number of the most abundant isotope for 57 elements have been included in the program. These represent all of the stable elements for which this isotope accounts for at least 60 percent of the total. Thus, elements such as platinum have been excluded. In addition to the elements contained in this program, the radicals OH,  $H_2O$ , and  $CH_3$  have also been included. Up to 12 more radicals may be added to the array ATOM by replacing two of the present zeroes with a two-digit symbol and the mass number, respectively. More radicals may be added if the dimension is increased.

The input to the program consists of the symbols of up to nine of the stored atoms or radicals entered in 9A2 format. This is followed by the value of  $m/e$  under consideration. It is entered in I4 format and must be right justified; for example, a two-

digit number must be preceded by two blank spaces. One then enters the maximum number of each atom to be considered. This is mandatory with the batch processing program; however, with the time sharing program, upper limits for these values will be calculated and then lower values may be added if desired.

As seen before, the output of the program consists of a listing of all possible combinations of the entered atoms which have a specified mass. Because these molecular masses are based on the masses of the most abundant atomic isotope, errors can occur if those atomic isotopes do not account for most of the intensity. This is especially true when several of the same atoms are being considered. For example, for the ion  $K_4Cl_3^+$ , the atoms  $^{39}K$  and  $^{35}Cl$  account for about 93 and 75 percent of the total, respectively. For this ion the most abundant mass is not 261 atomic mass units (amu) as one may have expected but instead 263 amu. Thus, one must always consider the full distribution, as calculated using the program DISMIS, before assigning an ion to a mass number.

The user of this program may also wish to limit both the number of different atoms to be considered and the maximum number of each. Such restrictions, when made carefully, can often result in substantial savings of computer time without missing any reasonable species. However, the user must severely restrict the number of hydrogen atoms to be considered. If this is not done, the number of species having the molecular mass of interest would be prohibitive. Thus, the program is especially useful for inorganic species where the number of hydrogen atoms in a molecule is generally less than for organics. Indeed, we have used the program to interpret the low resolution spectra of inorganic systems and have found it to be an invaluable aid for recognizing all possible species which may be in a system.

The program is called FINDME which stands for: find m/e. As with the previously described program it has been designed for ease of use, compatibility with other FORTRAN IV compilers, use with batch processing or time sharing systems, and use with any output device having, in this case, at least 52 columns.

Lewis Research Center,  
National Aeronautics and Space Administration,  
Cleveland, Ohio, April 28, 1977,  
506-16.

## APPENDIX A

### LISTING OF COMPUTER PROGRAM FOR THE CALCULATION OF MOLECULAR ISOTOPIC DISTRIBUTIONS

```

DIMENSION AT(10),NAT(10),ATM(100)
C   FOR ITY CHANGE TO PLOT(50)
DIMENSION A1(500),A2(10),FRC(500),PLOT(100)
DATA DASH,BLANK,PLUS/'-',' ','+'/'
C   FOR TIME SHARING SYSTEMS REMOVE THE FOLLOWING 'C'
C   WRITE(6,500)
500  FORMAT('OENTER MOLECULE IN 10(A2,I2) FORMAT; RIGHT JUSTIFY')
50  READ(5,100,END=51) (AT(I),NAT(I),I=1,10)
100  FORMAT(10(A2,I2))
    NIS=10
    NA=0
    DO 1 I=1,NIS
      J=NAT(I)
      IF(J.EQ.0) GOTO 1
      ITOP=I
      DO 2 II=1,J
        NA=NA+1
        ATM(NA)=AT(I)
2     CONTINUE
1     CONTINUE
    WRITE(6,101) (AT(I),NAT(I),I=1,ITOP)
101  FORMAT(///'1 ISOTOPIC DISTRIBUTION FOR      '10(A2,I2))
    WRITE(6,102)
102  FORMAT('0 ELEMENTAL DISTRIBUTIONS')
    CALL DISEIS(ATM(1),L1,A1,SUM,NSI)
    LH=L1+NSI-1
    LE=LH
    IF(NSI.GT.7) LE=L1+6
    WRITE(6,103) ATM(1),(M,M=L1,LE)
    NISO=NSI
    IF(NSI.GT.7) NISO=7
    WRITE(6,107) (A1(I),I=1,NISO)
    IF(NSI.LE.7) GOTO 20
    LE=LE+1
    WRITE(6,108) (M,M=LE,LH)
    WRITE(6,107) (A1(I),I=8,NSI)
20   IF((SUM.GT.1.0005).OR.(SUM.LT..9995)) WRITE(6,109) SUM
103  FORMAT(A7,I3,6I11)
107  FORMAT(7X,7F11.6)
108  FORMAT(7X,I3,6I11)
109  FORMAT(9X,' NOTE SUM =*F9.6)
    NSI=NIS
    ZERO=1.E-8
    DO 3 IA=2,NA
      CALL DISEIS(ATM(IA),L2,A2,SUM,NIS)
      IF(ATM(IA).EQ.ATM(1)) GOTO 21
      LH=L2+NIS-1

```

```

LE=LH
IF(NIS.GT.7) LE=L2+6
WRITE(6,103) ATM(IA),(M,M=L2,LE)
NIS0=NIS
IF(NIS.GT.7) NIS0=7
WRITE(6,107) (A2(I),I=1,NIS0)
IF(NIS.LE.7) GOTO 22
LE=LE+1
WRITE(6,108) (M,M=LE,LH)
WRITE(6,107) (A2(I),I=8,NIS)
22 IF((SUM.GT.1.0001).OR.(SUM.LT..9999)) WRITE(6,109) SUM
21 CONTINUE
ATM(1)=ATM(IA)
L1=L1+L2
NI=NSI+NIS-1
DO 4 NMI=1,NI
N=NMI+1
NMT=NMI
IF(NMT.GT.NSI) NMT=NSI
NS=N-NIS
IF(NS.LE.0) NS=1
FRAC=0.
DO 5 I1=NS,NMT
I2=N-I1
FRAC=A1(I1)*A2(I2)+FRAC
5 CONTINUE
FRC(NMI)=FRAC
IF(FRAC.LT.ZERO) FRC(NMI)=0.
IF(FRAC.GT.ZERO) NSS=NMI
4 CONTINUE
NSI=NSS
DO 6 II=1,NSI
A1(II)=FRC(II)
6 CONTINUE
3 CONTINUE
NIMIN=0
NIMAX=NSI
SUMM=0.
FRCMAX=0.
DO 7 I=1,NSI
F=FRC(I)
IF(F.GT.FRCMAX) FRCMAX=F
IF(F.GT.ZERO) NIMAX=I
SUMM=F+SUMM
IF(NIMIN.GT.0) GOTO 7
IF(F.GT.ZERO) NIMIN=I
7 CONTINUE
C FOR TTY CHANGE TO NPL0T=50
NPL0T=100
IPLT=NPL0T/10
WRITE(6,104) (I,I=1,10)
104 FORMAT('0 MOLECULAR DISTRIBUTION',/,-
C FOR TTY CHANGE TO 1015
& ' M/E ABSOLUTE RELATIVE ',10I10)
DO 8 I=NIMIN,NIMAX
F=FRC(I)
MASS=L1+I-1

```



```

RELFRC=F/FRCMAX
IF(RELFRC.LT..005) GOTO 19
DO 9 J=1,NPLOT
9  PLOT(J)=BLANK
N=RELFRC*NPLOT+.5
DO 10 J=1,N
10 PLOT(J)=DASH
CONTINUE
DO 11 J=IPLT,N,TPLT
IF(J.LE.N) PLOT(J)=PLUS
11 CONTINUE
WRITE(6,105) MASS,F,RELFRC,PLOT
GOTO 8
19 WRITE(6,105) MASS,F,RELFRC
8 CONTINUE
C FOR TTY CHANGE TO 50A1
105 FORMAT(I6,F11.7,F11.7,2X,100A1)
WRITE(6,106) SUMM
106 FORMAT('D TOTAL PROBABILITY =',F9.6,///)
GOTO 50
51 STOP
END

```

```

SUBROUTINE DISEIS(AT,L,A,SUM,NIST)
DIMENSION A(500),ATOM(15,85),ATAM(15,19),ATEM(15,19)
DIMENSION ATIM(15,15),ATUM(15,18),ATYM(15,14)
DATA ATAM/
& 'H',1.,.999855,.000145,11*0.,
& 'HE',3.,.00000137,.99999863,11*0.,
& 'LI',6.,.0750,.9250,11*0.,
& 'BE',9.,1.,12*0.,
& 'B',10.,.1978,.8022,11*0.,
& 'C',12.,.98888,.01112,11*0.,
& 'N',14.,.99633,.00367,11*0.,
& 'O',16.,.99759,.000374,.002039,10*0.,
& 'F',19.,1.,12*0.,
& 'NE',20.,.9092,.00257,.0882,10*0.,
& 'NA',23.,1.,12*0.,
& 'MG',24.,.7870,.1013,.1117,10*0.,
& 'AL',27.,1.,12*0.,
& 'SI',28.,.9221,.0470,.0309,10*0.,
& 'P',31.,1.,12*0.,
& 'S',32.,.95018,.00760,.04215,0.,.00014,8*0.,
& 'CL',35.,.7553,0,.2447,10*0.,
& 'AR',36.,.00337,0,.00063,0.,.99600,8*0.,
& 'K',39.,.9310,.000118,.0688,10*0./
DATA ATEM/
& 'CA',40.,.9697,0.,.0064,.00145,.0206,0.,.00003,0.,.00185,4*0.,
& 'SC',45.,1.,12*0.,
& 'TI',46.,.0793,.0728,.7394,.0551,.0534,8*0.,
& 'V',50.,.0024,.9976,11*0.,
& 'CR',50.,.0431,0.,.8376,.0955,.0238,8*0.,

```

```

E 'MN',55.,1.,12*0.,
E 'FE',54.,.0582,0.,.9166,.0219,.0033,8*0.,
E 'CO',59.,1.,12*0.,
E 'NI',58.,.6788,0.,.2623,.0119,.0366,0.,.0108,6*0.,
E 'CU',63.,.6909,0.,.3091,10*0.,
E 'ZN',64.,.4889,0.,.2781,.0411,.1857,0.,.0062,6*0.,
E 'GA',69.,.604,0.,.396,10*0.,
E 'GE',70.,.2052,0.,.2743,.0776,.3654,0.,.0776,6*0.,
E 'AS',75.,1.,12*0.,
E 'SE',74.,.0087,0.,.0902,.0758,.2352,0.,.4982,0.,.0919,4*0.,
E 'BR',79.,.50537,0.,.49463,10*0.,
E 'KR',78.,.0035,0.,.0227,0.,.1156,.1155,.5690,0.,.1737,4*0.,
E 'RB',85.,.7215,0.,.2785,10*0.,
E 'SR',84.,.0056,0.,.0986,.0702,.8256,8*0./
  DATA ATIM/
E 'Y',89.,1.,12*0.,
E 'ZR',90.,.5146,.1123,.1711,0.,.1740,0.,.0280,6*0.,
E 'NB',93.,1.,12*0.,
E 'MO',92.,.1584,0.,.0904,.1572,.1653,.0946,.2378,0.,.0963,4*0.,
E 'RU',96.,.0551,0.,.0187,.1272,.1262,.1707,.3163,0.,.1858,4*0.,
E 'RH',103.,1.,12*0.,
E 'PD',102.,.0096,0.,.1097,.2223,.2733,0.,.2671,0.,.1191,4*0.,
E 'AG',107.,.51817,0.,.48183,10*0.,
E 'CD',106.,.0122,0.,.0088,0.,.1239,.1275,.2407,.1226,.2886,0.,
E   .0758,0.,0.,
E 'IN',113.,.0428,0.,.9572,10*0.,
E 'SN',112.,.0096,0.,.0066,.0035,.1430,.0761,.2403,.0858,.3285,0.,
E   .0472,0.,.0594,
E 'Sb',121.,.5725,0.,.4275,10*0.,
E 'TE',120.,.00089,0.,.0246,.0087,.0461,.0699,.1871,0.,.3179,0.,
E   .3448,0.,0.,
E 'I',127.,1.,12*0.,
E 'XE',124.,.00096,0.,.00090,0.,.01919,.2644,.0408,.2118,.2689,
E   0.,.1044,0.,.0887/
  DATA ATUM/
E 'CS',133.,1.,12*0.,
E 'BA',130.,.00101,0.,.00097,0.,.0242,.0659,.0781,.1132,.7166,4*0.,
E 'LA',138.,.00089,.99911,11*0.,
E 'CE',136.,.00193,0.,.00250,0.,.8848,0.,.1107,6*0.,
E 'PR',141.,1.,12*0.,
E 'ND',142.,.2711,.1217,.2385,.0830,.1722,0.,.0573,0.,.0562,4*0.,
E 'SM',144.,.0309,0.,0.,.1497,.1124,.1383,.0744,0.,.2672,0.,
E   .2271,0.,0.,
E 'EU',151.,.4782,0.,.5218,10*0.,
E 'GD',152.,.0020,0.,.0215,.1473,.2047,.1568,.2487,0.,.2190,4*0.,
E 'TB',159.,1.,12*0.,
E 'DY',156.,.00052,0.,.00090,0.,.0229,.1888,.2553,.2497,.2818,4*0.,
E 'HO',165.,1.,12*0.,
E 'ER',162.,.00136,0.,.0156,0.,.3341,.2294,.2707,0.,.1488,4*0.,
E 'TM',169.,1.,12*0.,
E 'YB',168.,.00135,0.,.0303,.1431,.2182,.1613,.3184,0.,.1273,4*0.,
E 'LU',175.,.9741,.0259,11*0.,
E 'HF',174.,.0018,0.,.0520,.1850,.2714,.1375,.3524,6*0.,
E 'TA',180.,.00012,.99988,11*0./
  DATA ATYM/
E 'W',180.,.0014,0.,.2641,.1440,.3064,0.,.2841,6*0.,
E 'RC',185.,.3707,0.,.6293,10*0.,

```

```

& 'OS',184.,.0002,0.,.0159,.0164,.133,.161,.264,0.,.410,4*0.,
& 'IR',191.,.373,0.,.627,10*0.,
& 'PT',190.,.00013,0.,.0078,0.,.329,.338,.253,0.,.0721,4*0.,
& 'AU',197.,1.,12*0.,
& 'HG',196.,.00146,0.,.1032,.1684,.2313,.1322,.2980,0.,.0685,4*0.,
& 'TL',203.,.2950,0.,.7050,10*0.,
& 'PB',204.,.0148,0.,.236,.226,.523,8*0.,
& 'BI',209.,1.,12*0.,
& 'TH',232.,1.,12*0.,
& 'U',234.,.000057,.0072,0.,0.,.9927,8*0.,
& 30*0./

```

```

EQUIVALENCE(ATOM(1),ATAM(1)),(ATOM(286),ATEM(1)),(ATOM(571),
& ATIM(1)),(ATOM(796),ATUM(1)),(ATOM(1066),ATYM(1))

```

```
SUM=0.
```

```
ATM=AT
```

```
DO 1 I=1,85
```

```
J=I
```

```
IF(ATM.EQ.ATOM(1,I)) GOTO 2
```

```
1 CONTINUE
```

```
2 L=IFIX(ATOM(2,J))
```

```
DO 3 I=1,13
```

```
A(I)=ATOM(I+2,J)
```

```
IF(A(I).LE.0.) GOTO 3
```

```
SUM=SUM+A(I)
```

```
NIST=I
```

```
3 CONTINUE
```

```
RETURN
```

```
END
```

## APPENDIX B

### LISTING OF COMPUTER PROGRAM FOR THE ASSIGNMENT

#### OF MOLECULAR FORMULAS

```

INTEGER ATOM(140),AMASS(9),ATM(9),L(9),J(9)
DATA ATOM/' H',1,'LI',7,'BE',9,' B',11,' C',12,' N',14,'ME',15,
& ' O',16,'OH',17,'WA',18,' F',19,'NE',20,'NA',23,'MG',24,'AL',27,
& 'SI',28,' P',31,' S',32,'CL',35,'AR',40,' K',39,'CA',40,'SC',45,
& 'TI',48,' V',51,'CR',52,'MN',55,'FE',56,'CO',59,'NI',58,'CU',63,
& 'GA',69,'AS',75,'RB',72,'SR',88,' Y',89,'NB',93,'RH',103,
& 'IN',96,' I',100,'CS',133,'BA',138,'LA',139,'CE',140,'PR',141,
& 'TB',159,'HO',165,'TM',169,'LU',175,'TA',181,'RE',187,'IR',193,
& 'AU',197,'TL',205,'BI',209,'TH',232,' U',238,
& 24*0,
& ' ',0/
DATA YES/'Y'/
C   FOR TIME SHARING SYSTEMS REMOVE 'C' FROM I/O AND IF STATEMENTS
C   WRITE(6,200)
200 FORMAT('CENTER ATOMS AS 9A2')
501 READ(5,100,END=500) ATM
100 FORMAT(9A2)
C   WRITE(6,201)
201 FORMAT('CENTER M/E AS I4')
READ(5,101) MOLMAS
101 FORMAT(I4)
DO 1 I=1,9
DO 2 K=1,99,2
IF(ATM(I).NE.ATOM(K)) GOTO 2
AMASS(I)=ATOM(K+1)
GOTO 1
2   CONTINUE
1   CONTINUE
102 FORMAT('OMASS ',9A3)
DO 5 I=1,9
NB=I
IF(ATM(I).EQ.ATOM(99)) GOTO 6
5   CONTINUE
NB=NB+1
6   CONTINUE
17  NC=NB-1
NCMB=1
DO 16 II=1,NC
L(II)=MOLMAS/AMASS(II)
NCMB=(L(II)+1)*NCMB
16  CONTINUE
WRITE(6,202) NCMB,(L(I),I=1,NC)
202 FORMAT(' THERE ARE',I10,' COMBINATIONS',/
& ' MAX NO. OF EACH ATOM IS',9I3)
C   WRITE(6,109)
109 FORMAT(' DO YOU WISH TO RESET MAXIMUMS ')
C   READ(5,203) HUH

```

```

203  FORMAT(A1)
C    IF(HUH.NE.YES) GOTO 24
C    WRITE(6,204)
204  FORMAT('ENTER NO. ATOMS AS 9I2')
      READ(5,205)(L(I),I=1,NC)
205  FORMAT(9I2)
24   IMASS=0
      L1=L(1)+1
      L2=L(2)+1
      L3=L(3)+1
      L4=L(4)+1
      L5=L(5)+1
      L6=L(6)+1
      L7=L(7)+1
      L8=L(8)+1
      L9=L(9)+1
      WRITE(6,102)(ATM(I),I=1,NC)
      DO 30 I1=1,L1
      J(1)=I1-1
      M1=J(1)*AMASS(1)
      DO 30 I2=1,L2
      J(2)=I2-1
      M2=J(2)*AMASS(2)+M1
      DO 30 I3=1,L3
      J(3)=I3-1
      M3=J(3)*AMASS(3)+M2
      DO 30 I4=1,L4
      J(4)=I4-1
      M4=J(4)*AMASS(4)+M3
      DO 31 I5=1,L5
      J(5)=I5-1
      M5=J(5)*AMASS(5)+M4
      IF(M5.GT.MOLMAS) GOTO 30
      DO 32 I6=1,L6
      J(6)=I6-1
      M6=J(6)*AMASS(6)+M5
      IF(M6.GT.MOLMAS) GOTO 31
      DO 33 I7=1,L7
      J(7)=I7-1
      M7=J(7)*AMASS(7)+M6
      IF(M7.GT.MOLMAS) GOTO 32
      DO 34 I8=1,L8
      J(8)=I8-1
      M8=J(8)*AMASS(8)+M7
      IF(M8.GT.MOLMAS) GOTO 33
      DO 35 I9=1,L9
      J(9)=I9-1
      MSS=J(9)*AMASS(9)+M8
      IF(MOLMAS-MSS) 34,21,35
21   WRITE(6,108) MSS,(J(I),I=1,NC)
108  FORMAT(I5,9I3)
35   CONTINUE
34   CONTINUE
33   CONTINUE
32   CONTINUE
31   CONTINUE
30   CONTINUE

```

```
C      PRINT OUT OF MAX. NO. OF EACH ATOM (LEAVE OUT OF T/S VERSION)
      WRITE(6,104) (L(I),I=1,NC)
104    FORMAT(/5X,9I3,/)
      GOTO 501
500    STOP
      END
```

## REFERENCES

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2. Dombek, B. D.; Lowther, J.; and Carberry, E.: A Computer Program for the Prediction of Mass Spectrum Isotope Peaks. *J. Chem. Educ.*, vol. 48, no. 11, Nov. 1971, p. 729.
3. Grimley, Robert T.: *Mass Spectrometry. The Characterization of High Temperature Vapors.* John L. Margrave, ed., John Wiley & Sons, Inc., 1967, pp. 195-243.
4. Roboz, John: *Introduction to Mass Spectrometry.* Interscience Publications, 1968, appendix I.
5. Beynon, John H.; and Williams, A. E.: *Mass and Abundance Tables for Use in Mass Spectrometry.* Elsevier Publishing Co., 1963.
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8. Quantum Chemistry Program Exchange. Programs No. 296 and 301, Indiana Univ.

```

run dismis
ENTER MOLECULE IN 10(A2,I2) FORMAT; RIGHT JUSTIFY
MO 3 0 9
    
```

ISOTOPIC DISTRIBUTION FOR MO 3 0 9

ELEMENTAL DISTRIBUTIONS		94	95	96	97	98
10	92	0.158400	0.000000	0.157200	0.004600	0.237800
	99	0.000000	0.096300	0.165300	0.004600	0.237800
0	16	0.997500	0.00374	0.002030		
	17					
	18					

MOLECULAR DISTRIBUTION		1	2	3	4	5	6	7	8	9	10
420	0.0038860	0.0371095									
421	0.0000131	0.0001255									
422	0.0067299	0.0643744									
423	0.0116012	0.1109701									
424	0.0161372	0.1543500									
425	0.0204509	0.1956210									
426	0.0439892	0.4207762									
427	0.0564112	0.5482886									
428	0.0650703	0.6284203									
429	0.0700911	0.6700098									
430	0.0830748	0.7946470									
431	0.0941674	0.8950083									
432	0.1043430	1.0000000									
433	0.0783243	0.7492050									
434	0.0980610	0.931934									
435	0.0700300	0.6698731									
436	0.0683002	0.6533218									
437	0.0471772	0.4312710									
438	0.0417658	0.3955670									
439	0.0173891	0.1720758									
440	0.0212988	0.2037324									
441	0.0290664	0.283749									
442	0.0068604	0.0656801									
443	0.0000750	0.0030080									
444	0.0009062	0.0095205									
445	0.0000037	0.0000355									
446	0.0000171	0.0001033									
447	0.0000000	0.0000002									
448	0.0000001	0.0000012									

TOTAL PROBABILITY = 1.000017

(a) Molybdenum trioxide trimer.

Figure 1. - Isotopic distributions. Output taken from time-sharing version of program DISMIS.



run d1sm1s  
 ENTER MOLECULE IN 10(A2,12) FORMAT; RIGHT JUSTIFY  
 C 4 h 4br 2cl 1 f 3

ISOTOPIC DISTRIBUTION FOR C 4 H 4BR 2CL 1 F 3

ELEMENTAL DISTRIBUTIONS

Element	Isotope	Relative Abundance	Abundance
C	12	0.988880	0.011120
	13		
H	1	0.999855	0.000145
	2		
BR	79	0.505370	0.000000
	80		0.494630
CL	35	0.755300	0.000000
	37		0.244700
F	19	1.000000	

MOLECULAR DISTRIBUTION

m/e	Absolute	Relative
302	0.1843574	0.4381624
303	0.0083994	0.0199628
304	0.4207312	1.0000000
305	0.0191641	0.0455473
306	0.2938514	0.6933971
307	0.0133755	0.0317895
308	0.0574463	0.1365327
309	0.0020380	0.0061998
310	0.0000449	0.0001007
311	0.0000004	0.0000008

TOTAL PROBABILITY = 0.999999

(b) Dibromochlorotrifluorobutane.

Figure 1. - Concluded

```

run findme
ENTER ATOMS AS 9A2
kclcr o

ENTER M/E AS 14
294
THERE ARE      8208 COMBINATIONS
MAX NO. OF EACH ATOM IS 7 8 5 18
DO YOU WISH TO RESET MAXIMUMS?
no

MASS  K  CL  CR  O
294  0  2  0  14
294  0  2  4  1
294  0  6  1  2
294  1  1  3  4
294  1  5  0  5
294  2  0  2  7
294  4  2  1  1
294  5  1  0  4

```

(a) Potassium, chlorine, chromium, and oxygen for mass of 294 atomic mass units.

```

run findme
ENTER ATOMS AS 9A2
cr onaclohalsifent

ENTER M/E AS 14
118
THERE ARE      907200 COMBINATIONS
MAX NO. OF EACH ATOM IS 2 7 5 3 6 4 4 2 2
DO YOU WISH TO RESET MAXIMUMS?
yes

ENTER NO. ATOMS AS 912
2 7 4 3 4 2 2 1 1

```

MASS	CR	O	NA	CL	OH	AL	SI	FE	NI
118	0	0	0	0	2	0	1	1	0
118	0	0	0	1	0	1	0	1	0
118	0	0	0	1	0	1	2	0	0
118	0	0	1	0	4	1	0	0	0
118	0	0	2	0	1	1	1	0	0
118	0	1	0	0	1	1	0	0	1
118	0	1	1	0	3	0	1	0	0
118	0	1	1	1	1	1	0	0	0
118	0	1	2	0	0	0	0	1	0
118	0	1	2	0	0	0	2	0	0
118	0	2	0	0	0	0	1	0	1
118	0	2	0	1	3	0	0	0	0
118	0	2	1	1	0	0	1	0	0
118	0	2	3	0	1	0	0	0	0
118	0	3	0	2	0	0	0	0	0
118	0	4	0	0	0	2	0	0	0
118	1	1	1	0	0	1	0	0	0
118	1	2	0	0	2	0	0	0	0

(b) Chromium, oxygen, sodium, chlorine, hydroxyl, aluminum, silicon, iron, and nickel for mass of 118 atomic mass units.

Figure 2. - Combinations of elements or radicals. Output taken from time-sharing version of program FINDME.



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—NATIONAL AERONAUTICS AND SPACE ACT OF 1958

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