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NASA TM-78887

(NASA-TM-78887) DIMINIODE THERMIONIC ENERGY  
CONVERSION WITH LANTHANUM-HEXABORIDE  
ELECTRODES (NASA) 18 p HC A02/MF A01

N78-24617

CSCL 10A

Unclass  
16804

G3/44

NASA TM-78887

**DIMINIODE THERMIONIC ENERGY CONVERSION  
WITH LANTHANUM-HEXABORIDE ELECTRODES**

by Erich W. Kroeger, Virginia L. Bair, and James F. Morris  
Lewis Research Center  
Cleveland, Ohio 44135

TECHNICAL PAPER to be presented at the  
International Conference on Plasma Science  
sponsored by the Institute of Electrical and Electronics Engineers  
Monterey, California, May 15-18, 1978



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National Aeronautics and Space Administration  
Lewis Research Center  
Cleveland, Ohio 44135

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ABSTRACT

This paper presents thermionic-conversion data obtained from a variable-gap cesium diminiode with a hot-pressed, sintered lanthanum-hexaboride emitter and an arc-melted lanthanum-hexaboride collector. Performance curves cover a range of temperatures: emitter 1500 to 1700 K, collector 750 to 1000 K, and cesium reservoir 370 to 510 K. Calculated values of emitter and collector work functions and barrier index are also given.

INTRODUCTION

TEC with metal-hexaboride electrodes (ref. 1) was part of a diode-screening project at the Lewis Research Center (LeRC) in 1970. The work depended on economical "mass production" of the diminiode, shown in figure 1 (refs. 2 to 5). This cesium diode had been designed with guarded electrodes small enough to accommodate obtainable 0.64-cm-diameter single crystals of refractory materials like tungsten, rhenium, iridium (ref. 6), and metal hexaborides. In fact procurement initiated in February 1971 for 0.8-cm-diameter  $\text{LaB}_6$  single crystals to be made by R. W. Johnson (ref. 7). But the in-core nuclear thermionic program incorporated the diminiode capability for a statistical study of reactor-compatible electrode materials (refs. 8 to 10). During the 1973 termination of space nuclear activities (including TEC) an attempt to evaluate a diminiode with  $\text{LaB}_6$  electrodes fell short. Subsequently presentation at the 1974 IEEE International Conference on Plasma Science reviewed TEC electrodes of  $\text{LaB}_6$  and other metallides (ref. 1) in conjunction with the diminiode (ref. 11). Then the current work began with LeRC's technical and fiscal management of NASA's applied-research-and-technology (ART) program for TEC in July 1975 (refs. 12 to 17).

NASA's TEC-ART Program prompted important new information on metal hexaborides and  $\text{LaB}_6$  in particular (refs. 18 to 26). Additional emphasis came

with the July 1977 revelation of 1.7-V barrier indices for cesium diodes with  $\text{LaB}_6$  collectors at the Sukhumi Institute of Physics and Technology (ref. 27). Shortly thereafter programmatic retrenchment terminated TEC-ART activities at LeRC. And NASA transferred fiscal and technical management of its TEC-ART program to the Jet Propulsion Laboratory (JPL), where NEP-prototype emphasis (refs. 28 to 34) has led to indicated discontinuations of research on metal hexaborides.

In the interim LeRC overcame impurity and attachment problems for  $\text{LaB}_6$  electrodes in a diode with a hot-press-sintered emitter and an arc-melted collector. And after a long source search 0.8-cm-diameter single crystals are on order for  $\text{LaB}_6$  and are available for the even more promising cerium hexaboride ( $\text{CeB}_6$ ). Diode evaluations of the better-performing metal-hexaboride crystal faces would allow optimum-surface selections. Then parametrically controlled vapor deposition should enable an approach to such desirable TEC-electrode surfaces in practical cylindrical configurations. But diode personnel and facilities will be dispersed before the monocrystalline metal hexaborides arrive.

Fortunately some preliminary results for the diode with 99.8 percent-pure  $\text{LaB}_6$  electrodes are now available. These data represent a 1500 to 1700 K sintered  $\text{LaB}_6$  emitter, a 750 to 1000 K arc-melted  $\text{LaB}_6$  collector, a 370 to 510 K cesium reservoir, and 0.25 mm interelectrode spacing. The present paper discusses performance values obtained from this thermionic energy converter with emitter and collector surfaces of  $\text{LaB}_6$ .

W. E. Frey and R. D. Schaal developed and performed special procedures and conducted research tests necessary to fabricate this diode. References 4, 5, and 10 provide detailed descriptions of all equipment and procedures used in the experiments treated in this paper.

#### ELECTRODE MATERIAL PROCESSING

Many types of lanthanum hexaboride ( $\text{LaB}_6$ ) were purchased or formed in-house in an attempt to fabricate thermionic diode electrodes. These forms of  $\text{LaB}_6$  powder were tried: cold-pressed; hot-pressed; electron-beam melted; arc-melted; and hot-pressed, sintered as well as single crystal. Some of the problems encountered were high impurity levels, low material strengths, high porosities, voids, cracks, small sample sizes, various reactivities, and thermal expansion mismatches. The best form for a diode would be of course a 0.7-cm-diameter, high-purity, oriented, uniform, single crystal of  $\text{LaB}_6$ . Since this form was not available the following polycrystalline  $\text{LaB}_6$  material was used.

The collector electrode started with hot pressed 85 percent dense and 99.8 percent pure  $\text{LaB}_6$ . This hot-pressed material was arc melted into a high-density button. The button was shaped by an electrical-discharge machine (EDM) into a 0.780-cm-diameter by 0.185-cm-thick disk for the collector. A Zr, 22.8 w/o-Ru braze (1523 K melt point) was used to attach the collector to a diminiode niobium, alumina-cerniet sub-assembly. The final preparation of the collector face and guard ring separation was done by EDMing. A heat-cycle test was performed on this collector sub-assembly in an RF vacuum furnace. The sub-assembly was heated 10 times from room temperature to 1073 K at  $<10^{-6}$  torr.

The biggest problem with the  $\text{LaB}_6$  emitter electrode was finding a method to attach it to the diminiode tantalum-top-hat structure. Many brazing or adhesive materials were tried to adhere  $\text{LaB}_6$  to Ta for high-temperature emitter applications. References 1 and 39 mention most of the materials tried. Failure was more common than not for several reasons. The major cause of failure was a high reactivity of  $\text{LaB}_6$  at temperatures above 1500 K with practically all materials tried. Another reason was the brittleness of  $\text{LaB}_6$  which caused cracking under the stress, generally as a result of thermal cycling and thermal expansion mismatch. Additionally, some materials, such as Pt and  $\text{MoSi}_2$ , wicked into the porous  $\text{LaB}_6$  and contaminated the electrode surface.

Special low vapor pressure ( $<10^{-10}$  torr at 1500 K) braze alloys were prepared and used with  $\text{LaB}_6$  (ref. 40). The alloy Zr, 22.8 w/o-Ru which melts at about 1523 K was a fair braze but applicable only for the low temperature collector electrode. A Zr, 31.1 w/o-Mo braze was fair as an emitter braze but limited to about 1700 K maximum due to reactions and remelting problems.

Attempts were made to melt  $\text{LaB}_6$  on W, Ta, and Mo pedestals. Generally, a reaction occurs well before the melting of  $\text{LaB}_6$ , thereby raising doubts as to the chemical integrity of an electrode joined by this method. Further proof of chemical diffusion was the depression of the  $\text{LaB}_6$  melting point by 200 to 400 K below literature values.

Clean metal-to-metal diffusion bonding also had the problem of  $\text{LaB}_6$  reactions. As a result, TaC coated Ta was tried as the base pedestal. Limited success occurred in that reactions were prevented. The bonding, in general, did not hold up after thermal cycling to about 1700 K. Subsequently, private communications with E. K. Storms (LASL) indicated 150 cycles to over 1772 K without failure for a  $\text{TaCo}_2$  braze of  $\text{LaB}_6$ .

At this point it was concluded that a more readily attainable method to hold the  $\text{LaB}_6$  would be mostly mechanical. A tantalum carbide coating was used to prevent  $\text{LaB}_6$  from reacting with a tantalum cup into which it was hot-press

sintered. Several different configurations were tried. The one that worked best is described below.

A carbonized tantalum cup as shown in figure 2 was machined. The carbonizing was accomplished by packing carbon powder into the tantalum cup and RF heating in a vacuum for 1 hour at 2100 K. The excess carbon powder was removed from the cup and 320-mesh, high-purity  $\text{LaB}_6$  powder was cold pressed into the cup. The tantalum cup and  $\text{LaB}_6$  were RF heated to 2100 K for 2 hours in a vacuum with a tantalum rod pushing on the  $\text{LaB}_6$  with a constant  $210 \text{ N/cm}^2$  (305 psi) pressure to sinter the  $\text{LaB}_6$ .

Since the  $\text{LaB}_6$  was below the tantalum cup surface, the excess tantalum was removed using high speed tools. No EDMing was done on this electrode because of past experience which showed a loosening of the sintered  $\text{LaB}_6$  material in the tantalum cup after EDMing. The surface of the  $\text{LaB}_6$  emitter was prepared using fine emery paper. A Zr, 18 w/o-Re braze alloy (melting point 1870 K) was used to bond the Ta,  $\text{LaB}_6$  cup to the tantalum top hat of the diminiode. This sub-assembly was heated at  $<10^{-6}$  torr for 10 cycles from room temperature to 1700 K. The bond between the  $\text{LaB}_6$  and tantalum remained good as was observed from the constant temperature difference at a given temperature between the  $\text{LaB}_6$  and tantalum.

One additional property measured on the sintered  $\text{LaB}_6$  was its thermal expansion. As can be seen in figure 3, the thermal expansion of sintered  $\text{LaB}_6$  matches that of annealed niobium (ref. 36). This means that the diminiode collector sub-assembly has a good thermal expansion match. The emitter sub-assembly has the  $\text{LaB}_6$  in compression which should make a good mechanical bond.

After any parts of the diminiode were subjected to cutting fluids or contamination of any kind, they were ultrasonically cleaned in trichloroethane (NA 500) before vacuum bakeout, welding, brazing, or assembly.

At this point the two basic electrode sub-assemblies were ready for final assembly. For a detailed description of the diminiode assembly, see reference 11 (note that the cesium fill procedure, given below, has been modified so that capsules are no longer used).

#### DIMINIODE PROCESSING

A new processing chamber was used for the  $\text{LaB}_6$  diminiode. This chamber shown in figure 4 allows diminiode bake-out, emitter-top-hat-temperature calibration, electrode-spacing calibrations, cesium filling and a copper-braze

closure. Both turbomolecular and vac-ion pumping are used on this water jacketed chamber to maintain  $10^{-8}$  torr (diminiode cold) and  $10^{-7}$  torr (diminiode hot).

In this case the  $\text{LaB}_6$  diminiode was electron-bombardment (EB) heated to 1673 K emitter and a 933 K collector temperature for bakeout. A temperature calibration of the black-body hole in the  $\text{LaB}_6$  (sighting down the open cesium-reservoir tube) versus the black-body hole in the top hat was done. As indicated in references 37 and 38, this calibration is acceptable because the diminiode holds at zero current before and after the data-taking cycle. These low-transport conditions and thermal inertia assure good sensing of the emitter temperature. The electrode gap was also checked at this time with the diminiode hot.

Cesium is introduced into the chamber from a high-purity cesium bottle through a heated stainless-steel tube as shown in figure 5. The flow is controlled with a metal seal stainless-steel valve. The cesium is dropped from the end of the tube onto a small, heated stainless steel tray. After visually observing the cesium, the tray can be tipped to allow the cesium to run into the diminiode or into a catch can. Following cesium addition a tantalum ball is then dropped into the end of the cesium reservoir and copper brazed in place to seal the diminiode.

The diminiode is now ready for performance testing after it is removed from the processing chamber and mounted into the test station.

#### DIMINIODE RESULTS

Figure 6 shows the current-density, voltage (I, V) envelopes for constant emitter temperatures of 1500, 1550, 1600, 1650, and 1700 K with the collector varied from 750 to 1000 K and the cesium reservoir varied over a temperature range of 510 to 370 K. The power density (P, V) envelopes calculated from the I, V results are shown in figure 7. These two figures are summarized in the following table to show where maximum power occurred.

Emitter temperature, $T_E$ , K	Approximate collector temperature, $T_C$ , K	Approximate cesium temperature,	Maximum power output, $P_{max}$ , W/cm <sup>2</sup>	Voltage at maximum power, V
1500	850	440	1.86	0.22
1550	850	440	2.50	.28
1600	850	440	3.50	.38
1650	850	450	4.20	.40
1700	900	460	5.29	.45

As can be seen from the table, the cesium reservoir temperatures are much lower than the expected 550 K. Cesium reservoir thermocouples and readout instrumentation were checked with no apparent discrepancies found. A procedure check revealed that more cesium was added to this diminiode than was usual. Steps taken to ascertain possible cesium temperature problems are described below.

The diminiode was carefully removed and X-rays were taken to see if any cesium extended beyond the cesium-temperature control zone. The reservoir tube axis is in a 3° below horizontal position during operation. No apparent cesium extension could be seen.

The diminiode was then positioned with the cesium-reservoir pointing down. The reservoir was cooled and the rest of the diminiode heated with an air heat gun to make sure all the cesium was in the reservoir. Another set of X-rays was taken. No cesium could be seen above the cesium-reservoir-temperature control zone.

The diminiode was then placed in its normal operating position. Hot water was used to heat the cesium reservoir with the rest of the diminiode at room temperature. After heating in this manner for 5 to 10 minutes the reservoir was cooled and another series of X-rays taken. This time the X-rays revealed cesium in the unheated portion of the cesium reservoir tube. At this point the decision was made to remount the diminiode in its test station with the cesium reservoir pointing down to prevent any possible cesium extension beyond the control zone.

When trying to test the diminiode in this new position it was discovered that the collector was no longer operative. Apparently a large segment of the collector had dislodged locally, but not enough to short out another element. Arc-melted samples had demonstrated this separating tendency. A series of



runs was made using the guard ring as the collector to see if there were any changes in the cesium reservoir temperature for maximum power. Maximum power occurred at a cesium reservoir temperature of 450 to 460 K at 1500 K emitter and 850 K collector temperatures. Thus the diode apparently did operate at the cesium-reservoir temperatures indicated with a possible error of  $\pm 25^\circ$ .

A comparison of the I, V and P, V envelopes of the  $\text{LaB}_6$  diode with a TECO W, O, Cs diode, and a Re, Nb diode all with 1700 K emitters is shown in figure 8. As can be seen in this figure the  $\text{LaB}_6$  diode produces a maximum power that is almost equal to the best diode (Re, Nb) but at almost twice the voltage, 0.45 volt versus 0.23 volt.

Indeed the most efficient operation of the  $\text{LaB}_6$  diode occurs at 0.75 volt and  $5 \text{ A/cm}^2$  ( $3.8 \text{ W/cm}^2$ ). There the work function of the 1700 K emitter is less than 2.64 eV, the barrier index is about 1.9 volts, and the calculated efficiency for optimum leads is approximately 1.3 percent. Unfortunately back emission data to enable determination of the work function for the 853 K collector was unavailable.

In fact because of the complete lack of experience with cesium diodes having  $\text{LaB}_6$  emitters and collectors, these initial tests were made based on accumulated results for refractory-metal electrodes. As such these findings are very nonoptimum. And a search of the data collection revealed no suitable back-emission numbers. Correcting this deficiency was a goal of the next set of tests before changing the interelectrode spacing. But the collector malfunction intervened.

Additional I, V curves did reveal barrier indices less than 1.95 volts for both 1653 and 1596 K emitters: 1653 K emitter, 801 K collector, 0.62 V,  $4.8 \text{ A/cm}^2$ ,  $3.0 \text{ W/cm}^2$ ,  $<2.57 \text{ eV}$  emitter work function,  $\sim 1.95 \text{ V}$  barrier index and 1596 K emitter, 853 K collector,  $5.3 \text{ A/cm}^2$ , 0.51 V,  $2.7 \text{ W/cm}^2$ ,  $<2.46 \text{ eV}$  emitter work function,  $\sim 1.95 \text{ V}$  barrier index.

#### CONCLUDING REMARKS

As can be seen in the diode results, the performance of the  $\text{LaB}_6$  is equal to some of the best electrode combinations at this time. It is felt that the  $\text{LaB}_6$  material and electrode spacing used in this test are far from the optimum. Work functions for single crystal  $\text{LaB}_6$  are  $\varphi_{\text{LaB}_6(100)} = 2.52 \text{ eV}$ ,  $\varphi_{\text{LaB}_6(110)} = 2.60 \text{ eV}$ , and  $\varphi_{\text{LaB}_6(346)} = 2.41 \text{ eV}$ .

Field emission work by L. Swanson indicates even lower work functions are available. Swanson also obtained nonoptimized cosiated  $\text{LaB}_6(100)$  work functions of 1.3 eV (refs. 17 to 21) without oxygenation. With collector work functions approaching this low value and interelectrode losses diminished by reduced cesium pressures perhaps coupled with some enhancement, barrier indices significantly lower than the measured 1.9 eV value seem quite probable. Furthermore the work function for  $\text{CeB}_6(100)$  is less than  $\text{LaB}_6(100)$ , which is even more promising.

Incidentally reference 41 indicates vaporization rates between  $4 \times 10^{-3}$  ( $\text{LaB}_6_{014}$ ) and  $4 \times 10^{-4}$  ( $\text{LaB}_6_{111}$ ) cm/yr (2 to 0.2 mils/yr) for congruently vaporizing  $\text{LaB}_6$  at 1700 K. So TEC with 1700 K congruently vaporizing  $\text{LaB}_6$  emitter surfaces and congruently depositing  $\text{LaB}_6$  collector surfaces appear practical both in performance and lifetime. The interelectrode gap would merely shift between 0.04 and 0.004 cm (16 and 2 mils) in 10 years of service.

Work with promising metallides like  $\text{LaB}_6$  and  $\text{CeB}_6$  for TEC electrodes should include tests of selected faces (refs. 17 to 21) of high-purity single crystals of specified stoichiometries (refs. 17 to 23). If such systematic research is neglected, a far more complex situation could result than occurred for refractory-metal TEC electrodes in the early 1960's. Understanding and control of TEC variables had eluded researchers for many years. Finally investigations of pure, oriented monocrystalline electrodes in exceedingly clean converters lifted the veil to reveal a systematic technology where confusion formerly prevailed. Having defined and described the datum base for refractory-metal electrodes, TEC research workers could then invoke crystal-face and additive effects predictably - rather than haphazardly.

Metallide TEC electrodes begin with more chemical complexity, hence greater permutability than the relatively simple single elements of their refractory-metal counterparts. Thus systematic research to point the way to controlled, productive development is even more important for success with metallide TEC electrodes than it ultimately proved to be for the refractory-metal predecessors.

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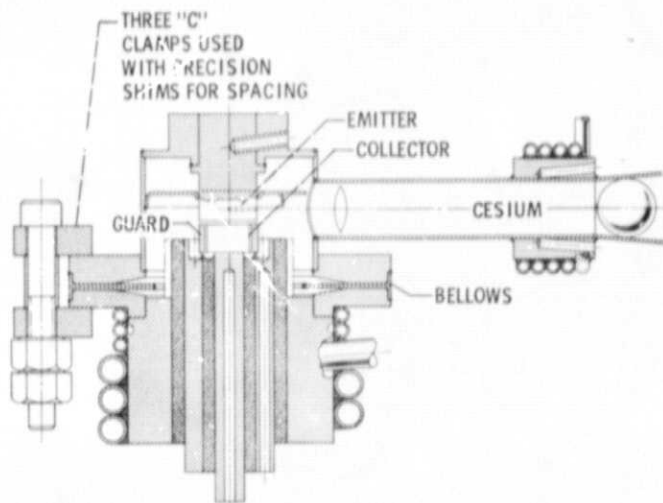
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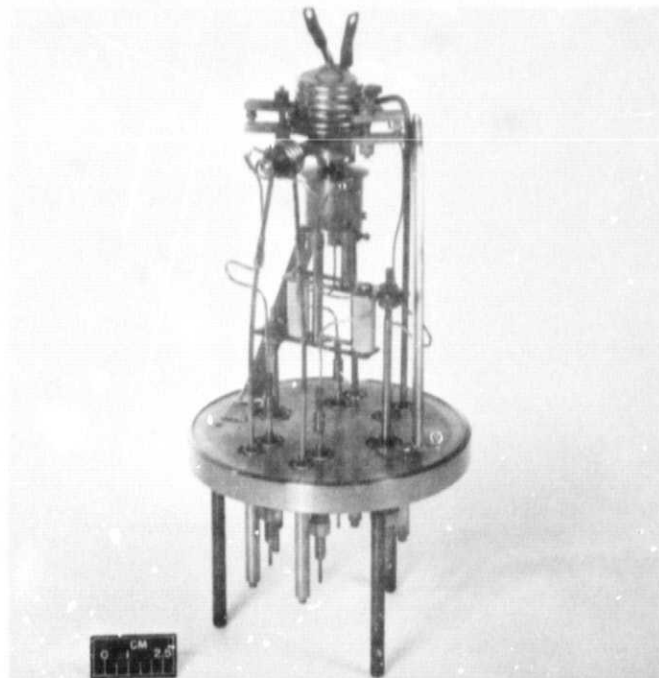
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(a) DIMINIDE CROSS SECTION.



(b) DIMINIDE MOUNTED ON FLANGE.

Figure 1. - Diminide.

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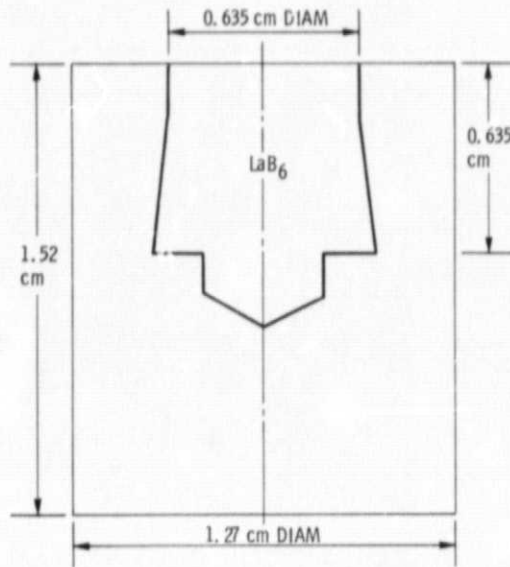


Figure 2. - Tantalum cup for sintering LaB<sub>6</sub> emitter.

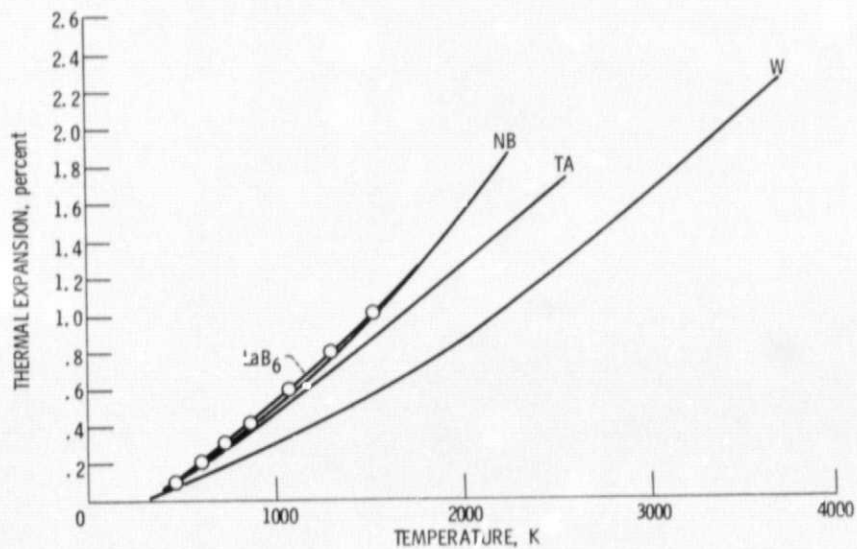


Figure 3. - Thermal expansion (ref. 35).

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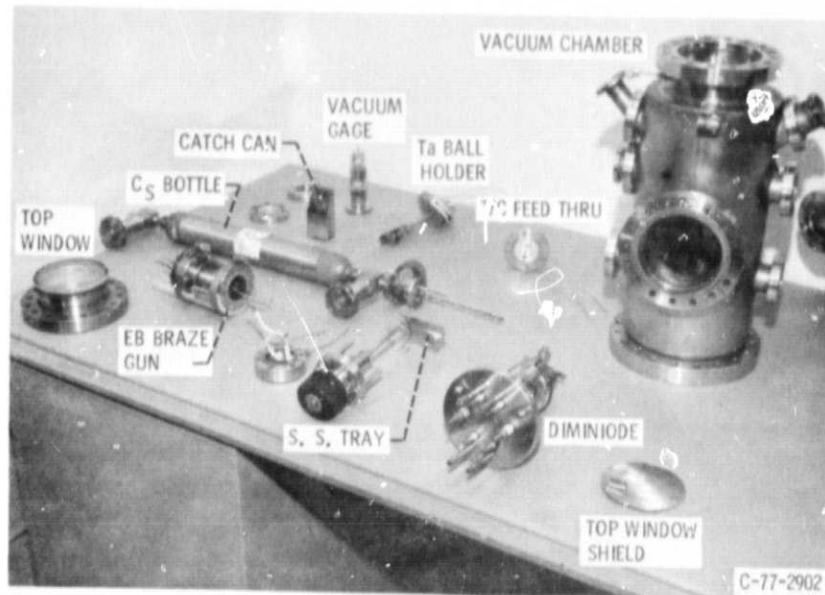


Figure 4. - Processing chamber parts.

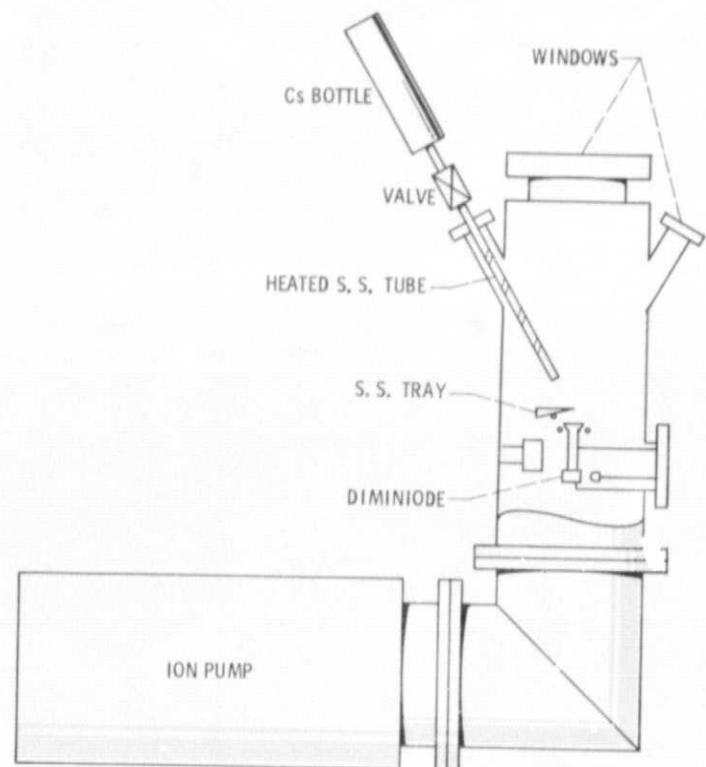


Figure 5. - Processing chamber schematic.

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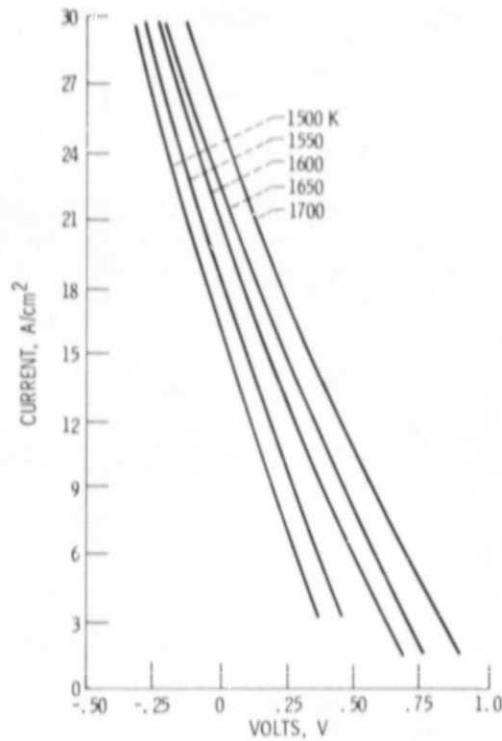


Figure 6. - LaB<sub>6</sub>, LaB<sub>6</sub> constant emitter temperature envelopes, optimum collector and reservoir. 0.25 mm gap.

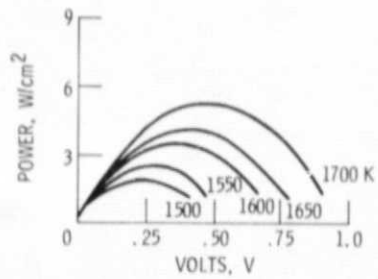


Figure 7. - LaB<sub>6</sub>, LaB<sub>6</sub> constant emitter temperature envelopes, optimum collector and reservoirs.

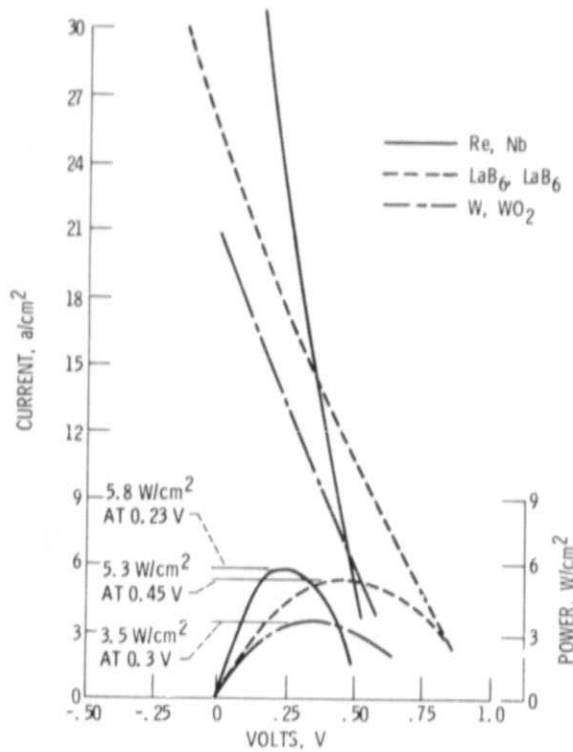


Figure 8. - Constant emitter temperature envelopes  
 1700 K emitter, optimum collector and reservoir.  
 0.25 mm gap.

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