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BURDEN FOR MANUFACTURE OF CERAMIC MATERIAL

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Translation of "Shikhta dlya izgotovleniya keramicheskogo materiala", USSR Patent No, 655691, Opisaniye Izobreteniya k Avtorskomu Svidetel'stvu, (Description of an Invention with Author's Certificate), Published in Bulletin Izobreteniy, No. 13, May 5, 1979, 2 pp.

(NASA-TM-77255) BURDEN FOR MANUFACTURE OF CERAMIC MATERIAL (National Aeronautics and Space Administration) 5 p HC A02/MF A01  
 CSCL 11B

N83-30655

G3/27 Unclas 13178

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STANDARD TITLE PAGE

1. Report No. NASA TM-77255	2. Government Accession No.	3. Recipient's Catalog No.	
4. Title and Subtitle BURDEN FOR MANUFACTURE OF CERAMIC MATERIAL		5. Report Date May 1983	
		6. Performing Organization Code	
7. Author(s) T. V. Chechenya and Ye. A. Osipova		8. Performing Organization Report No.	
		10. Work Unit No.	
9. Performing Organization Name and Address SCITRAN Box 5456 Santa Barbara, CA 93108		11. Contract or Grant No. NASu- 3542	
		12. Type of Report and Period Covered Translation	
12. Sponsoring Agency Name and Address National Aeronautics and Space Administration Washington, D.C. 20546			
15. Supplementary Notes  Translation of "Shikhta dlya izgotovleniya keramicheskogo materiala", USSR Patent No. 655691, Opisanije Izobreteniya k Avtorskomu Svidetel'stvu, (Description of an Invention with Author's Certificate), Published in Bulletin Izobreteniy, No. 13, May 5, 1979, 2 pp.			
16. Abstract  The invention refers to ceramic materials which can be obtained by methods of powder metallurgy and can be used in high temperature technology.			
17. Key Words (Selected by Author(s))		18. Distribution Statement  Unclassified - Unlimited	
19. Security Classif. (of this report) Unclassified	20. Security Classif. (of this page) Unclassified	21. No. of Pages 5	22. Price

Description of Invention for Certificate of Authorship 655691.  
Applied for 01 October 1976, Application No. 2406926/29-33, Published  
05 April 1979, Bulletin No. 13, Date of publication 08 April 1979.  
M. K1<sup>2</sup> C 04 B 35/56, C 04 B 35/58. UDC 066.798.2(088.8).

## BURDEN FOR MANUFACTURE OF CERAMIC MATERIAL

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The invention refers to ceramic materials which can be obtained by methods of powder metallurgy and can be used in high temperature technology. /1\*

There are widely known ceramic materials, refractory masses which are used in high temperature technology [1].

The closest to the invention is burden for the manufacture of ceramic material which contains silicon carbide, chrome aluminophosphate binding agent, magnesium oxide and chromite. It contains the indicated ingredients in the following quantities, percent by weight: silicon carbide 55 - 63, chrome aluminophosphate binding agent 13 - 17; chromite of fraction less than 0.1 mm 20 - 25; magnesium oxide 3 - 5 [2].

The material has the following physical characteristics: volumetric weight 2.4 g/cm<sup>3</sup>; ultimate compression strength (roasting temperature 1600°C) 6.00 - 7.00 kg/mm<sup>2</sup>; gas permeability 0.6 - 0.8 l.m./m<sup>2</sup> hour x wat.col.

The shortcomings of this material are high temperature of sintering and low mechanical strength. This excludes it from being used as structural material (for example, material for heat exchanger frames in gas-turbine engines).

The purpose of the invention is to improve the mechanical strength of the material.

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\* Numbers in margin indicate pagination in original text.

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In order to achieve this goal, the burden for manufacture of the ceramic material which includes silicon carbide, magnesium oxide and aluminum chrome-phosphate binding agent, also contains silicon nitride with the following ratios of components, percent by weight:

Silicon carbide	32.9 - 47.3
Magnesium oxide	1.3 - 1.5
Aluminum chrome-phosphate binding agent	16.4 - 19.7
Silicon nitride	31.5 - 49.4

The raw material used was: green silicon carbide, silicon nitride, roasted magnesium, chrome aluminum phosphate binding agent. 12

The method of making the material consists of preparing a mass consisting of powder of silicon carbide and chrome aluminum phosphate binding agent to which silicon nitride powder and roasted magnesium are then added. All are carefully mixed and directly before molding, the mass is prepared.

It is molded in steel molds which are chrome-plated on the working surfaces with specific molding pressure of 300 - 1000 kg/cm<sup>2</sup>.

The intermediate products are sintered according to the following technology:

Drying from 20° to 150°C for 5 hours in air.

Sintering from 20 to 700°C for 6 hours and holding at 700°C for one-two hours in air.

The effect of the percentage content of silicon nitride on the strength of the samples is indicated in the table.

A	GPB	SiC	MgO	Si <sub>3</sub> N <sub>4</sub> %	Density, g/cm <sup>3</sup>	$\sigma_{\text{max}}$ kg/mm <sup>2</sup>	$\sigma_{\text{C}}$ kg/mm <sup>2</sup>
1.1	1.3	1.3	31.5	2.35	4.45	8.24	
1.2	1.5	1.4	39.5	2.35	7.10	8.95	
1.3	1.9	1.5	49.4	2.4	8.10	13.26	

As is apparent, the magnitude of strength exceeds the same magnitudes of the known ceramic material ( 6 - 7 kg/mm<sup>2</sup> -  $\sigma_c$ ). The items made of the proposed material, in addition, have low sintering temperature (700°C instead of 1600°C for the known ceramic material). The technology for manufacturing the items is comparatively simple and the employed materials are inexpensive. The volumetric weight of the material is about 2.4 g/cm<sup>3</sup>, thermal stability is 1000 - 20°C of 20 air thermal cycling without destruction.

The proposed ceramic material can be used as structural high-temperature material, for example for heat exchanger frames of a gas-turbine engine.

#### Formula of the Invention

The burden for making the ceramic material which includes silicon carbide, magnesium oxide and aluminum chrome-phosphate binding agent is distinguished by the fact that in order to improve mechanical strength, it additionally contains silicon nitride with the following component ratio, percent by weight:

Silicon carbide	32.9 - 47.3
Magnesium oxide	1.3 - 1.5
Aluminum chrome-phosphate binding agent	16.4 - 19.7
Silicon nitride	31.5 - 49.4

Sources of information considered in the expert evaluation.

1. USSR Certificate of Authorship No. 348634, k. C 22 c 29/00, 1971.
2. USSR Certificate of Authorship No. 408935, k1. C 04 v 35/56, 1972.