

ded by NASA Technical Reports Server Frank J. Gruntmaner's abstract is presented mere. his paper and visual material were not presented for publications.

CHEMICAL STRUCTURE OF INTERFACES

Frank J. Grunthaner

Jet Propulsion Laboratory California Institute of Technology Pasadena, California

The interfacial structure of silicon/dielectric and silicon/metal systems is particularly amenable to analysis using a combination of surface spectroscopies together with a variety of chemical structures of Si/SiO2. Si/SiO₂Si₃N₄, Si/Si₂N₂O, Si/SiO₂/A1, and Si/Native Oxide interfaces using high resolution (0.350 eV FWHM) X-ray photoelectron spectroscopy. The general structure of these dielectric interfaces entails a monolayer chemical transition layer at the Si/vielectric boundary, which consists of specific distributions of the intermediate oxidation states of silicon (Si⁺¹, Si⁺², and Si⁺³), which appear to be related to local morphology of the silicon substrate. Amorphous Si substrates show a wide variety of hydrogenated Si and Si(OH), states that are not observed in thermal oxidation of single crystal material. Extended SiO₂ layers greater than 8 Å in thickness are shown to be stoichiometric SiO₂, but to exhibit a wide variety of local network structures. In the nitrogen-containing systems, we see an approach to stoichiometric oxynitride compounds with interesting impurity- and electron-trapping properties. In native oxides, we find substantial topographical non-uniformity in oxide thickness and composition (e.g., a nominal 25Å oxide can be shown to have 4 different areal distributions of SiO₂ ranging in thickness from 12 to 165 Å). Analysis of metal/oxide interfacial layers is accomplished by analytical removal of the Si substrate by UHV XeF2 dry etching methods. The general question of the importance of the chemical state of the starting silicon surface to the final dielectric/silicon interface composition will be addressed in a series of experiments using covalent silicon surface substitution. Finally, the modification of silicon surface bonding and hybridization in the presence of sub-monolayer levels of transition metals will be examined from the perspective of a modification in interfacial reactivity.

PRECEDING PAGE BLANK NOT FILMED