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GAS COOLED FUEL CELL SYSTEMS TECHNOLOGY DEVELOPMENT

FINAL REPORT FOR THE SECOND LOGICAL UNIT OF WORK CONTRACT PERIOD: MAY 1983 – MAY 1984

J. M. Feret Westinghouse Electric Corporation Advanced Energy Systems Division Pittsburgh, PA 15236-0864

AUGUST 1986

Prepared for NATIONAL AERONAUTICS AND SPACE ADMINISTRATION Lewis Research Center Under Contract DEN 3-290

For

U.S. DEPARTMENT OF ENERGY Morgantown Energy Technology Center

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The following personnel contributed to this Final Report

M. S. Barrett

D. E. Dickey

G. G. Elia

J. M. Feret

L. L. France

N. Haines

R. R. Holman

T. K. Houghtaling

J. L. Kelly

H. D. Kulikowski

A. K. Kush

K. F. Lampe

J. R. Lance

M. T. Le

B. L. Pierce

J. F. Pierre

R. Rosey

E. F. Saladna

F. R. Spurrier

S. Venkatesh

D. A. Wiseman

B. M. Woodle

M. K. Wright

J. F. Zippay

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LIST OF ACRONYMS

AICM	Acid Inventory Control Member
BOU	Beginning of Use
COE	Cost of Electricity
ĊT	Combustion Turbine
DHE	Dynamic Hydrogen Electrode
DSC	Differential Scanning Calorimetry
EOU	End of Use
EPRI	Electric Power Research Institute
ERC	Energy Research Corporation
ETU	Electrolyte Take-Up
FCS	Fuel Cell System
FEP	Fluorinated Ethylene-Propylene
FHS	Fuel Handling System
FMEA	Failure Mode Control System
FPS	Fuel Processing System
ICD	Interface Control Drawing
ICS	Instrumentation and Control System
LVDT	Linear Variable Displacement Transducer
0&M	Operating and Maintenance
000	Upen Circuit Voltage
PAFC	Phosphoric Acid Fuel Cell
PUS	Power Conditioning System
PDS	Purchasing Department Specifications
PES	Polyethersultone
PFA .	Perfluoroalkoxyethylene
KES ·	Rotating Equipment System
KHE	Reversible Hydrogen Electrode
565	Steam Generation System
	Simulated Reformer Gas
	Technical Assessment Guide
	lo be Determined Thormoonsyimotric Analysis
	Inerniograv Inecric Analysis
	Utility Equipment System
Urr W	Westinghouse
	Hork Broakdown Structure
H D S WNDF	Nork Dreakdown Structure Westinghouse Distributed Drocoss Family
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1.0 SUMMARY

This report addresses the work performed during the Second Logical Unit of Work of a multi-year program designed to develop a phosphoric acid fuel cell (PAFC) for electric utility power plant application. The Second Logical Unit of Work, which covers the period May 14, 1983 through May 13, 1984, was funded by the U.S. Department of Energy, Office of Fossil Energy, Morgantown Energy Technology Center, and managed by the NASA Lewis Research Center. The PAFC Technology Development Program efforts were performed by a team comprised primarily of personnel from the Westinghouse Advanced Energy Systems Division and the Energy Research Corporation (a division of St. Joe Minerals Corporation), in parallel and integrated with the Utility Power Plant Program which was sponsored by Westinghouse, a host utility, the Electric Power Research Institute, the Empire State Electric Energy Research Corporation, and other participating organizations.

1.1 Scope of Work

The major elements of work to be performed in the multi-year program have been integrated into a Work Breakdown Structure (WBS) as shown in Figure 1-1. This WBS establishes the basic framework for all contractually required effort and thereby provides a uniform structure for planning, assessing progress achieved, controlling costs, etc.

The Second Logical Unit of Work comprised twenty WBS tasks (see Table 1-1) for which specific objectives, schedular and cost plans, and deliverables have been established.

1.2 Technical Objectives

The primary objectives established to be achieved in the Second Logical Unit of the program were:



1-2

TABLE 1-1

PHOSPHORIC ACID FUEL CELL TECHNOLOGY DEVELOPMENT PROGRAM SECOND LOGICAL UNIT TASKS

WBS TASK NO.	DESCRIPTION
1101	Program Management
1102-01-100	System Analyses
1102-01-200	Fuel Cell, Fuel Processing, Power Conditioning, Rotating Equipment, and Instrumentation and Control Systems Design Requirements
1102-01-300	Fuel Cell Requirements
1102-02-200	100 kW Full Stack Fuel Cell Test Specification
1102-02-300	25 kW Short Stack Fuel Cell Test Specification
1102-02-400	Fuel Cell Hardware Test Specifications
1102-02-500	Fuel Cell Manufacturing Processes Specifications
1102-03-100	Fuel Cell System Design
1102-03-300	Systems Interface Requirements and Control
1103-01	Subscale Fuel Cell Development
1103-02	9-Cell Stack Development
1103-03	10 kW Fuel Cell Stack Development
1103-04	Fuel Cell Materials Characterization and Testing
1103-05	25 kW Short Stack Development
1103-06	100 kW Full Stack Development
1103-07	375 kW Module Development
1103-08	Advanced Fuel Cell Development
1104-01	Subscale 2 x 2 Inch Cell and 9-Cell Testing Facilities Development
1108-01	Technology Proof-of-Concept Module Fuel Cell System

- Establish the final system-level requirements for the Fuel Cell, Fuel Processing, Power Conditioning, Rotating Equipment, and Instrumentation and Control Systems.
- Develop the Fuel Cell System preliminary design.
- Reduce the total fuel cell resistance to less than 0.2 m Ω .
- Increase the average cell beginning of use performance at pressure on the order of 50 mV, and improve the long term performance of the fuel cell to a 2 mV loss per 1000 hours of operation.
- Develop an optimum stack startup procedure from ambient conditions to the nominal operating conditions: 480 kPa (70 psia), 190°C (374°F), and 325 mA/cm³ (300 A/ft²) operating on reformate hydrogen (83 percent utilization) and air (50 percent utilization), and changes above/below these nominal conditions.
- Develop performance repeatable and cost effective manufacturing processes for the electrodes, matrices, and cooling and bipolar plates.

Considerable progress was achieved relative to each of these objectives and is summarized below:

System level design requirements were established for the Fuel Cell, Fuel Processing, Power Conditioning, Rotating Equipment, and Instrumentation and Control Systems. This was accomplished by the performance of numerous studies and analyses which addressed the areas of system trade-offs, performance, operation, and maintenance. The specifications prepared for these systems include functional, operational, interface, and quality assurance requirements based on the selected plant design parameters: 1) 7.5 MW gross electrical output (7.2 MW net), 2) a heat rate of less than 9500 kJ/kWh

1-4

(9000 BTU/kWh), 3) full power operation with naphtha as the backup fuel, and 4) automatic control during all power operations.

A number of modifications were incorporated into the PAFC power plant design which resulted in improved performance, or a simpler configuration. These requirements, studies, and modifications are discussed in detail in Section 2.1.

- The Fuel Cell System preliminary design was fully developed. A complete design description, system maintenance requirements, and structural analysis are presented in Sections 2.2 and 2.3. The system interface requirements were identified and are documented in Section 2.3.2.
- Technology in the fuel cell development program has progressed significantly. Subscale cell testing verified stack design and manufacturing processes, as well as provided a base for stack scaling. The various test plans and results are reported in Section 3.1.
- The nine cell stack design has been further developed and refined. The primary areas of improvement were electrode edge sealing, acid transport within the cells, and manifold sealing. Additionally, new manufacturing, startup, and operating procedures resulted a substantial 48 mV increase in cell performance, while a further increase of 30 mV is projected due to thinner matrices. Complete details and testing results are given in Section 3.2.
- Based on the achievements in the nine cell stack area, the 10 kW, 25 kW, and 100 kW stacks were re-designed, incorporating a new inverted cell configuration. This configuration provides improved acid transport. The development status and design description of each of these stacks are presented in Sections 3.3 through 3.5.

- The 375 kW module design, necessary to the utility plant application, was updated to incorporate advancements achieved by the fuel cell technology program. The major changes included addition of a continuous acid supply system, addition of electric heaters to maintain higher internal temperatures, and the inverted cell stack design. A complete design description, including maintenance and replacement schedules, of the 375 kW module is given in Section 3.6.
- The characterization of various raw materials utilized in the manufacture of repeating cell components was addressed. Specifically, plate resistivity, porosimetry, and corrosion, seal materials, electrode materials, and other stack materials were evaluated. The comprehensive results of this effort are reported in Section 3.7.
- The advanced fuel cell development effort achieved considerable success by defining performance goals and identifying problem areas. A Performance Improvement Work Plan was formulated and implemented to improve cell performance and reduce variance. The areas primarily addressed were alternate catalyst evaluation and support, manufacturing, cell resistance, plate coatings and corrosion, acid management, and impurity effects. A complete summary of this effort is presented in Section 3.8.
- A new presurized test facility, SE-1, was constructed and checked out at ERC. This facility, which is capable of accommodating a full sized (12x17 in.) nine cell stack, improved upon past ERC facilities by incorporating advanced methods of measurement and control. This facility is described in Section 4.0.

The aggregate of this effort resulted in the revision of manufacturing and stack assembly procedures, stack design, and system specifications. These, along with the associated drawings, were updated and released.

2.0 SYSTEMS ENGINEERING (WBS 1102)

The objectives of this task were to develop the final system-level design requirements for the Fuel Cell, Fuel Processor, Power Conditioning, Rotating Equipment, and the Instrumentation and Control Systems (FCS, FPS, PCS, RES, and ICS, respectively). The following sections summarize the results of analyses and design studies performed.

2.1 Development System Requirements (WBS 1102-01)

This section summarizes system analysis, technology assessments, and the resulting design requirements for the FCS, FPS, PCS, RES, and ICS.

2.1.1 System Analyses (WBS 1102-01-100)

The primary objectives of the system analysis effort were to perform system trade-off studies, performance analyses, and operational studies to establish preliminary system level design requirements.

These analyses identified the functional and conceptual interface requirements in conjunction with the Task 1102-04-100 Plant Systems Analyses effort. System level trade-off studies that considered heat rate, power level, cost of electricity (COE), operating characteristics, and development risk were performed to select system level design requirements for each of these systems for a prototype power plant that produces 7.5 MW_e DC with a heat rate goal of less than 9500 kJ/kWh (9000 Btu/kWh). The effect upon plant performance and operating characteristics for startup, steady-state, shutdown, and malfunction conditions associated with these systems was determined by utilizing appropriate steady-state, transient, and controllability analysis models.

A number of design modifications were incorporated into the PAFC power plant. These modifications include:

- Provide addition of 100 cells per module.
- Incorporate the functions of the FPS turbocompressor into the RES.
- Make plant operation at a lower pressure level at part power compatible with RES recommended speed.
- Provide part of the steam required for reforming in the FPS.
- Design for full power beginning of use with naphtha fuel.

The above items resulted in improved performance or a simpler configuration.

2.1.1.1 Performance and Operational Studies

Four normal power plant operating points were established in the First Logical Unit. These are: (1) full power beginning of use (BOU), (2) full power end of use (EOU), (3) part power BOU, and (4) part power (EOU). The full power BOU point is identified as the baseline prototype power plant design. The EOU condition is defined at a cell voltage degradation of 80 mV. Though the remaining fuel cell operating parameters (temperature, pressure, current density, etc.) remain constant at BOU and EOU for full and part power modes respectively, there is a difference in the operating parameters between full and part power. The full power BOU operating conditions for the fuel cell are 190°C (374°F), 480 kPa (70 psia), and 300 mA/cm^2 (280 A/ft²) producing a cell voltage of 0.690 V (0.610 V at EOU). The corresponding part power conditions are 190°C (374°F), 207 kPa (30 psia), and 75 mA/cm² (70 A/ft²), which result in a BOU cell voltage of 0.752 V (0.672 V at EOU). The part power operating pressure represents a pressure level compatible with the RES speed. Part power operation at constant temperature maintains the pressure level of the steam generators. The performance summary for these operating points is given in Table 2-1, where the heat rates have been reduced from one to four percent below the First Logical Unit values.

TABLE 2-1 PERFORMANCE SUMMARY

	FULL POWER		PART POWER	
	BOU	EOU	BOU	EOU
Cell Voltage (mV)	690	610	752	672
Pressure, kPa (psia)	480 (70)	480 (70)	207 (30)	207 (30)
Temperature, °C (°F)	190 (374)	190 (374)	190 (374)	190 (374)
Current Density, mA/cm ² (A/ft ²)	300 (280)	300 (280)	75 (70)	75 (70)
Gross DC, kW	7500	6625	2042	1825
Gross AC, kW	7200	6360	1878	1679
Net RES, kW	98	103	-233	-203
Parasitic Losses, kW	202	202	64	66
Net AC, kW	7096	6243	1581	1410
Thermal Input, GJ/h (10 ⁶ Btu/h)	62.6 (59.39)	62.6 (59.39)	16.1 (15.23)	16.1 (15.23)
Plant Heat Rate,	8875 (8370)	10,028 (9510)	10,155 (9630)	11,388 (10,800)

kJ/kWh (Btu/kWh)

2.1.2 Fuel Cell, Fuel Processing, Power Conditioning, Rotating Equipment, and Instrumentation and Control Systems Design Requirements (WBS 1102-01-200)

Updated system design requirements were prepared for the FCS, FPS, PCS, RES, and ICS. These specifications include functional, operation, interface, and quality assurance requirements.

2.1.2.1 Fuel Cell System

The primary function of the FCS is to produce DC electric power and thermal energy using air-cooled 375 kW fuel cell modules. The fuel cell modules transform the chemical energy from the fuel and oxidant reaction (hydrogen-rich feed gas supplied by the FPS and oxygen in the form of pressurized air supplied by the RES) into electrical energy.

Design requirements established for the FCS include:

- Provide a gross electrical output of 7.5 MW DC at full power operating conditions at BOU.
- Control electrical output based on current demand. The design basis (BOU full power) fuel cell module current output is 325 amps.
- Provide hydrogen utilization consistent with fuel cell performance characteristics and FPS performance requirements. The design hydrogen utilization is 83 percent.
- Equalize fuel and process air flows to the individual fuel cell modules and provide for on-line checking for cross leakage between the anode and the cathode.
- Design for stable operation at power operation conditions and for response to the transient operating conditions within the appropriate time frames.

2-4

- Maintain fuel cell module temperatures greater than 107°C (225°F) during STANDBY.
- Maintain the fuel cell temperature between 38°C (100°F) and 54°C (130°F) during shipping, handling, maintenance, installation and non-power operating conditions to prevent solidification of the installed phosphoric acid.
- Maintain the air pressure level between 205 and 480 kPa (30 70 psia). The system design shall include the capability to isolate any fuel cell module or modules without causing significant pressure differentials across individual cells.
- Maintain the fuel cells in a dry inert atmosphere having a dew point of -23°C (10°F) or less during shipping, handling, installation, and non-power operating conditions to limit phosphoric acid water-content.
- Design for operation on the fuel cell anode supply composition defined in Table 2-2.
- Design to allow replenishment of fuel cell phosphoric acid during operation and shutdown.

2.1.2.2 Fuel Processing System

The primary function of the FPS is to convert steam and hydrocarbon fuel (either natural gas or naphtha) to a hydrogen-rich gas for use in the fuel cells. Fuel is provided by the Fuel Handling System (FHS) as natural gas or as liquid naphtha. Fuel cell anode exhaust is used to provide thermal energy for the reforming process. Steam and feedwater are supplied by the Steam Generation System (SGS). Pressurized air for combustion is provided by the RES.

Secondary functions of the FPS are to reduce impurities (hydrogen sulfide, etc.) from the fuel stream which are detrimental to fuel cell performance, to

TABLE 2-2

FUEL CELL ANODE SUPPLY COMPOSITION REQUIREMENTS

	Composition (vol	Composition (volume fraction)		
<u>Component</u>	Design Basis	Acceptable Limit		
Hydrogen	.742	> 0.50		
Carbon Dioxide	.182	diluent		
Carbon Monoxide	.008	< 0.01		
Water	.044	< 0.06		
Methane	.014	< 0.05		
Nitrogen	.010	diluent		

Impurities

Composition (ppm by volume)

H ₂ Š		< 100
cos	1	< 100
Olefins		< 300
Higher Hydrocarbons (C_2^+)		< 1000
NH ₃		< 10
ด้า		< 1
Metal Ions		< l (by weight)
Tar/Oils		< 0.05 (by weight)
Particulatés		< 0.05 (by weight)

remove water from the process gas stream, and to provide process heat for steam generation.

The FPS "top level" design requirements are summarized in Table 2-3.

2.1.2.3 Power Conditioning System

The primary functions of the PCS are to control the magnitude of the direct currents generated by the FCS, to combine these currents, and to convert them to alternating currents suitable for application to a distribution system and to an electric utility transmission network through the Utility Equipment System (UES).

Secondary functions are to control fuel cell module electrical loading, provide appropriate fault protection, harmonic reduction, power factor correction capability, and means of integrating system control and monitoring into the overall power plant.

The PCS design will be designed to meet the performance and equipment requirements identified in Tables 2-4 and 2-7 based on the input and output conditions as specified in Tables 2-5 and 2-6.

2.1.2.4 Rotating Equipment System

The primary functions of the RES are to provide and circulate a supply of clean pressurized air to the FCS and to provide clean pressurized air to the FPS. The air provides a source of oxygen for both the FPS combustion and the cathode side of the fuel cells and also removes thermal energy produced by the exothermic reaction within the FCS. The RES uses steam generated in the SGS, cathode exhaust gas from the FCS and combustion exhaust gas from the FPS to develop mechanical shaft power to help drive circulating and compressing equipment, thus helping to maximize the overall power plant efficiency. Key operating statepoints are summarized in Table 2-8.

2-7

TABLE 2-3				3	
	FPS	тор	LEVEL	DESIGN	REQUIREMENTS

PERFORMANCE

- H_2 Production Rate, normal m³/s (SCFD)
 - Gross
 - Net (Hydrogen Utilized) $1.3 (4.1 \times 10^6)$

Natural Gas Feedrate,* kg/s (1b/h)

OPERATIONAL

Cold Start to Standby, hours< 4</td>Standby to Min. Power, hours< 1/2</td>Min. to Max. Power, Percent Full Power per min7.7Operating Range, Percent production rate23-OperationAutomatic

Start/Stop Cycles (Over Lifetime)
- Warm Startup/Warm Shutdown

- Cold Startup/Cold Shutdown

GENERAL

Fuel Capability Availability, Percent Design Life, Years Maximum Plot Plan Area, m² (ft²) Maximum Height of Equipment Above Grade, m (ft) Noise Level (db) Transportability 7.7 23-100 Automatic Dispatch from Standby to Full Power

 $1.5 (4.9 \times 10^6)$

0.34 (2,720)

10,000 150

Natural Gas/Naphtha 95 25 230 (2500) 9.1 (30) 55 Common Carrier Truck (For all subsystems except reformer)

* Includes raw fuel used as reformer furnace fuel

TABLE 2-4 PCS PERFORMANCE REQUIREMENTS

Power Conversion

Rated Full Power Input Rated Full Power Output Rated Part Power Input Rated Part Power Output Minimum Operating Input

Efficiency

Full Power Part Power

Power Factor

Normal Excess VARs Available

۰.,

Fault Protection

DC Power AC Power

Response Time

Controlled Power Change Controlled MVAR Change Fault Clearing Trip

Harmonic Distortion

7.5 MWDC 7.2 MWAC 1.875 MWDC 1.725 MWAC .75 MWDC

96 percent 92 percent

Unity - Part to Full Load 1.4 MVAR - Full Load, Rated Voltage 5.4 MVAR - Part Load, Rated Voltage +6.8 MVAR or - 3.4 MVAR - Zero Load

Electronic circuit interruption Circuit breaker

.5 seconds over operating range.1 seconds over operating range.05 seconds

Maximum total harmonic distortion of 3 percent RMS of fundamental voltage and 1 percent for any one individual component when operating into a system with 250 MVA short circuit capacity.

TABLE 2-4

PCS PERFORMANCE REQUIREMENTS (CONT'D)

Manual Control

Automatic Control

Power Dispatch

Reactive Power Dispatch

To bring system to STANDBY condition during plant startup and for system checkout.

For all normal power operation via plant computer.

In response to changes in currents from fuel cell modules.

In response to command signals from plant computer independent of real power dispatch.

Power Level Control Accuracy

Electromagnetic Interference

Black Start Capability

Low Power Bleed Loading

<u>+</u> 150 kW

Shall not result in misoperation of local communications equipment or any sensitive loads.

None

Provide 50 ± 2.5 ohm load for each group of 5 modules - to load modules before PCS is connected.

TABLE 2-5 PCS INPUT FROM FUEL CELL SYSTEM

20

Number of Fuel Cell Modules

Nominal Input Per Module

	375 kW Module
Full Power - Beginning of Use	
Current, amps	325
Voltage, volts DC	1156
Power, kWDC	375
Full Power - End of Use	:
Current, amps	325
Voltage, volts DC	1025
Power, kWDC	333
Part Power - Beginning of Use	
Current, amps	80
Voltage, volts DC	1260
Power, kWDC	100
Part Power - End of Use	
Current, amps	80
Voltage, volts DC	1130
Power, kWDC	90
Module Voltage Maximum Limit	1300 V (s

Module Current Maximum Limit

Module Minimum Controlled Current

Module Low Power Bleed Current

Module Ripple Current Limit

Module Effective Resistance

Module Effective Capacitance

1300 V (sustained more than 1 second)
450A (sustained more than 1 second)
50A

5 to 10A during shutdown operation 100A peak to peak, 180 or 360 Hertz .95 ohm during power operation Variable with magnitude of maximum current reached during a load change

current reached during a load change 24 MFD for 100A maximum 120 MFD for 350A maximum

TABLE 2-6PCS OUTPUT TO UTILITY EQUIPMENT SYSTEM

Output Voltage

voltage variations)

13,800 VAC, 3 phase, 60 Hz

<u>System Performance Degradation</u> Performance (Resulting from utility AC grid Bus Voltage

System

Impact

Variation

2-12

+5 Percent >+5 Percent to +10 Percent (continuous) <-5 Percent to -10 Percent (continuous) <-10 Percent to -20 Percent >+10 Percent or -20 Percent Rated Power 95 Percent of Rated Power 95 Percent of Rated Power 85 Percent of Rated Power Unpredictable

Utility System Unbalance

Frequency

Rated Power Output with up to 3 Percent unbalance and 2 Percent negative sequence voltage

Rated Power Output - 57 - 61 Hertz (Trips off line outside this range due to transformer core limitations)

TABLE 2-7 PCS EQUIPMENT DESIGN REQUIREMENTS

Design Life

Availability

Reliability

Structural Welding

Structural Design

Footprint

Environment

Governing Electrical Code

Shipment

25 years 10000 Operations (mechanical)

99 Percent

99.5 Percent

AWS D1.1

ANSI A58.1 (Seismic Zone 4)

 43.5 m^2 (1500 ft²) maximum

Temperature	-29°C to 49°C (-20°F to 120°F)
Humidity	100 percent
Pressure	85-110 kPa (12.5-16 Psia)
Altitude	1525 m (5000 ft)

National Electrical Code (NFPA No. 70 - Latest Revision)

Truck Transportable (maximum size modules 3.5 m (11'-6") high x 2.25 m (7'-6") wide x 9.5 m (31'-6") long, 18600 kg (41,000 lbs) - shock loads of up to 3 g during shipment)

WAESD=TR=85=0030

TABLE 2=7 PCS EQUIPMENT DESIGN REQUIREMENTS (CONT'D)

<u>Maintenance_Requirements</u>

<u>Controlled Mechanical Interfaces</u>

<u>Controlled Electrical Interfaces</u>

Clearance for removal of major component and assemblies Replaceable air filters Self checking diagnostic circuits Conservative thermal margins -24 hours/year scheduled maintenance downtime

Equipment foundations Equipment fencing Fire protection alarms

(+) or (-) connection to each FCS module

13.8 KVAC oùtput connections to a Utility Équipment System

480 VAC power feeder from Power Distribution System

Ground connections from Power Distribution System

Control interface with Westinghouse WDPF Data Highway

TABLE 2-8 RES OPERATING STATEPOINTS

		Full Power	Part Power	
	`	EOU	BOU	
•	Air Circulator			
	Flow, kg/h (lb/h)	642, 200 (1,416,000)	102,490 (225,950)	
	Inlet Pressure, kPa (psia)	482 (70)	207 (30)	
	Inlet Temperature, °C (°F)	146 (294)	117 (243)	
	Outlet Pressure, kPa (psia)	489 (71).	207 (30)	
	Outlet Temperature, °C (°F)	147 (297)	141 (286)	
	Fluid	Air	Air	
•	Centrifugal Compressor			
	(Two Stages)			
	Flow, kg/h (lb/h)	33,790 (74,490)	17,919 (39,480)	
	Inlet Pressure, kPa (psia)	101.3 (14.7)	101.3 (14.7)	
	Inlet Temperature, °C (°F)	27 (80)	27 (80)	
	Outlet Pressure, kPa (psia)	482 (70)	207 (30)	
	Outlet Temperature, °C (°F)	146 (294)	117 (243)	
	Fluid	Air	Air	
•	Steam Turbine			
	Flow, kg/h (lb/h)	8,193 (18,060)	1,126 (2,476)	
	Inlet Pressure, kPa (psia)	365 (53)	365 (53)	
	Inlet Temperature, °C (°F)	140 (284)	140 (284)	
	Outlet Pressure, kPa (psia)	21 (3)	21 (3)	
	Outlet Temperature, °C (°F)	61 (141)	61 (141)	
	Fluid	Steam	Steam	
•	Cathode Gas Expander			
	Flow, kg/h (lb/h)	28,230 (62,230)	11,730 (25,860)	
	Inlet Pressure, kPa (psia)	475 (69)	200 (29)	
	Inlet Temperature, °C (°F)	193 (379)	172 (341)	
		-		
	Т	ABLE 2-8		
-----	-----------	-------------	----------	
RES	OPERATING	STATEPOINTS	(CONT'D)	

		Full Power	Part Power
		EOU	BOU
Outlet	Pressure, kPa (psia)	110 (16)	110 (16)
Outlet	Tempgrature, °C (°F)	27 (146)	114 (238)
Fluid	· · · · · · ·	Air (O ₂ depleted)	Air (O ₂ depleted)
) Combust	ion Gas Expander		

Flow, kg/h (lb/h) Inlet Pressure, kPa (psia) Inlet Temperature, °C (°F) Outlet Pressure, kPa (psia) Outlet Temperature, °C (°F) Fluid

9,288 (20,480)	2,457 (5,416)
352 (51)	138 (20)
440 (824)	434 (813)
110 (16 <u>)</u>	110 (16)
295 (563)	408 (767)
Combustion Products	$w_{1} = 1$

Table 2-9 gives the power summary for the RES design for full and part power at BOU and EOU conditions. The BOU part power case requires the most power from the electric motor, 231 kW (310 HP). Table 2-10 lists the RES design requirements. A compact arrangement of the RES is illustrated in Figure 2-1.

2.1.2.5 Instrumentation and Control System

The primary functions of the ICS are to provide real time monitoring of all plant parameters and provide warnings of deviations from normal operating conditions or plant configuration, to control all startup, power and shutdown operations of plant systems, and to implement automatic corrective action in the event of faulted or emergency conditions. Secondary functions are to provide data recordings of plant parameters, plant configuration status, and alarm event sequences for overall plant performance evaluation use and event documentation.

The ICS will be designed to meet the performance requirements identified in Table 2-11, based on the numbers of inputs and outputs specified in Table 2-12.

Table 2-13 lists equipment design requirements that will be met, and Figure 2-2 shows the basic system configuration in block diagram form.

2.1.3 Fuel Cell Requirements (WBS 1102-01-300)

The fuel cell baseline technology requirements development effort needed to meet the PAFC system design and operation goals imposed by FCS performance requirements is discussed in this section. Current fuel cell technology and the anticipated improvements resulting from the development effort are identified and compared to the prototype plant requirements and goals.

2.1.3.1 Approach and Assumptions

The overall objectives of the fuel cell requirements effort were to establish the baseline fuel cell technology requirements, define goals for design and

	TABLE	2-9
RES	POWER	SUMMAR Y

• •	Full Power		Part Power	
· ·	BOU	EOU	BOÙ	EOU
<u></u>	<u>:</u>	<u></u>		<u> </u>
Steam Turbine (kW)	561	672	46	77
Combustion Gas Expander (KW)	458	458	22	22
Cathode Gas Expander (kW)	1175	1175	207	207
Air Compressor (kW) (Two Stages)	-1842	-1842	-492	-492
Air Circulator (kW)	-255	-360	-14	- 16
Motor Power Required (kW)	97	103	-231	-202

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TABLE 2-10 RES DESIGN REQUIREMENTS

Performance

See Operating Statepoints

Operational

Warm Startup/Warm Shutdown Cycles Cold Startup/Cold Shutdown Cycles Operating Range, Percent of Rated Speed Operation

(Table 2-8)

10,000
150
70 - 100
Automatic Dispatch
from Standby to Full
Power

General

·	.
Availability Percent	94
Design Life, years	25
Maximum Plot Plan Area, m^2 (ft ²)	50 (525)
Maximum Height of Equipment Above Grade, m (ft)	9.1 (30)
Noise Level (db) at plant parameter	55
Transportability	Common Carrier Truck



TABLE 2-11

ICS TOP LEVEL DESIGN REQUIREMENTS

<u>System Type</u> Westinghouse WDPF distributed control system using local programmable controllers, data highway, and central control room.

<u>Manual Control</u> Between COLD STOP and STANDBY from local control stations of individual systems or central control room.

<u>Automatic Control</u> During POWER operations from local control stations of individual systems or central control room or remote utility dispatch room.

Alarms

Lock-in type warnings of deviations from normal operating condition or plant configuration and identification of fault and emergency conditions and sequence of occurence.

Monitoring Real time indications of plant alignment and operating parameters.

<u>Redundancy</u> As required to achieve desired reliability and accomplish safe shutdown.

UninterruptibleAs required to monitor plant status after loss of plantPower Supplypower.

<u>Software</u> User friendly - no previous computer programming experience required.

Diagnostics

Self checking and failure diagnostic programming built in.

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WAESD-TR-85-0030

TABLE 2-12 ICS INPUT/OUTPUT REQUIREMENTS

Type of 1/0	<u>No. Required*</u>	Electrical, <u>Characteristic</u>	Function
Digital Input	768	384-TTL,	Level, temp., and pressure
		192-25VDC,	switches, limit switches,
And States and Antonia and		192-115VAC	events, handshakes, binary
			data

÷	Digițal Output		384-TTL,	Digital controls, contac-
	A Constant of the second second		192-115VAC,	tors, inhibits, enables,
		· · · · · · · · · ·	192-Relay	handshakes, binary data

Analog Input		1	5
• Thermocouple	288	Туре К	Temperature sensors
• High level	288	0-5 VDC	from level, pressure,
	<i>*</i> .	, , , , , , , , , , , , , , , , , , ,	flow, voltage and current
galler og vir som en som		u in an trut i	sensors and strain gauges

Analog Output

• High level	72	O-5 VDC	To meters, trend
• PID Controller	72	4-20 ma	recorders, analog devices
and the second second second	с. С. с.		
-	· · ·	· · · · ·	t.

TOTAL 2,256 (720 Analog, 1536 Digital)

· · *Current estimate

1.1

TABLE 2-13 ICS EQUIPMENT DESIGN REQUIREMENTS

Response Times

Controller computational frequency	TBD
Operator display update	TBD
Multiplexer scan rate	TBD
Time to change display format	TBD
Alarm scan rate	TBD
Alarm timing resolution	TBD

Accuracy

<u>Type of Measurement</u> TBD Control Accuracy

Design Life

25 years 10,000 operations (mechanical)

TBD (percent of full scale)

+ 2 percent Power

Availability

Reliability

Structural Welding

Structural Design

Configuration

99 percent

Error

99.8 percent

AWS D1.1

ANSI A58.1 (Seismic Zone 4)

93 m² (1000 ft²) maximum footprint. Skid mounted, prewired assemblies with weather/explosion proof enclosures as required. See Figures 2-8 and 2-9.

TABLE 2-13 ICS EQUIPMENT DESIGN REQUIREMENTS (CONT'D)

Environment

Temperature -29°C to 49°C (-20°F to 120°F)

15°C to 32°C ((60°F to 90°F)

inside enclosure)

Humidity 100 percent Pressure 85-110 kPa (12.5-16 psia)

<u>Governing Electrical Code</u> National Electrical Code (NFPA No. 70, latest revision)

Shipment

Maintenance Requirements

Controlled Mechanical Interfaces

Controlled Electrical Interfaces

Truck transportable (maximum size modules 3.5 m (11'-6") high x 2.25 m (7'-6") wide x 9.5 m (31'-6") long, 18600 kg (41,000 lbs) - shock loads of up to 3 g during shipment)

Clearance for removal of major components and assemblies Conservative thermal design margins On-line diagnostic and verification capability and software

Equipment foundations Fire protection alarms

All wiring to other plant systems





development of the fuel cells, and determine further development effort required. Specific efforts during this reporting period were to:

- determine fuel cell testing performance and technology baseline requirements to meet system development goals.
- establish and update the PAFC technology database needed for system development requirements assessments.
- define the fuel cell design and operating requirements for the continuing development testing to ensure meeting system requirements.
- provide projected fuel cell requirements for the PAFC module design specification and update as required.

The key baseline fuel cell technology existing at the beginning of this reporting period and from which the fuel cell development requirements effort progressed included design and fabricating processes for:

- nominal 30.5 cm x 43 cm (12 in. x 17 in.) "Z" channel process gas routing bipolar plates and double branched (tree) cooling plates.
- rolled electrode catalyst layers dry bonded to Teflon wet-proofed Stackpole backing paper.
- _____five cells between cooler plates
- a matrix utilizing .25 mm (10 mil) thickness MAT-1 and nominal .20 mm (8 mil) thickness SiC layer.
- performance development testing to be conducted with simulated reformer gas (SRG) fuel and oxidized with air.

- Viton seals selected as the first baseline (more detail is presented in Section 3.2.1.1 Stack Elastomer Edge Seal Development).
- nominal 48 cm³ (12.7 x 10^{-3} gal) of phosphoric acid applied at 97 percent concentration for each cell.

During this development period variations in fuel cell assemblies of alternate electrodes, processes, seal materials, seal configuration and acid loadings were investigated. However, the above baseline configuration was the basis for comparison of the technology development.

2.1.3.2 Technology Assessment Updates

During the First Logical Unit, the reference PAFC design technology baseline was not fully established for assessing development efforts for meeting prototype power plant applications. Effort was directed during this earlier period toward upgrading technology to improve performance. This is noted in Table 2-14, where the parameters developed in the First Logical Unit effort were updated to concur with the fuel cell prototype performance goals determined during this reporting period. A review of the prototype goals resulted in the definition of potentially more compatible fuel cell and plant (systems) operating conditions for the specified plant requirements (Refer to Section 2.1.2). PAFC development goals and test performance experienced during this reporting period are presented in Table 2-15.

Although most of the fuel cell operating parameters that were needed for design and concept assessments had some level of definition, the impacts of the selected operating conditions on cell life, overall stack performance, economics, and risk were not definable since corresponding test data were not available. Work continues to identify the parameter trade-offs and sensitivity of the fuel cell to its operating requirements in plant system conditions.

2-28	Parameter Identity Maximum Voltage (Cell) Minimum Voltage (Cell) Current Density (Min) Current Density (Min) Current Density (Max) Temperature (Nom Max) Temperature (Max) Temperature (Max) Pressure (Max) Pressure Difference Process Gases (Max) Coolant Temperature (Min)	KEY PAFC Time Continuous (TBD) h Continuous Continuous Continuous Continuous Continuous Continuous Continuous	DESTGN PARAMETE <u>Level</u> <800 mV None 0-75 mA/cm ² (0-70 A/ft ²) > 75-100 ² mA/cm ² (70-93 A/ft ²) > 75-100 ² mA/cm ² (70-93 A/ft ²) 220°C (426°F) (70-93 A/ft ²) 220°C (426°F) (70-93 A/ft ²) 220°C (426°F) (700 psia) (100 psia) 34.5 kPa (100 psia) 34.5 kPa (5 psia) 110-150°C (230-302°F)	<pre>RS. VALUES, AND CONSTRAINTS¹ Origin/Comments Corrosion Constraint to < 800 mV from life performance Functional Constraint - minimum design voltage required Corrosion Constraint - Voltage dependent - requires further development data to set time limit at temperature Corrosion Constraint² - function of temperature and pressure and not yet characterized Performance Constraint - low efficiency and cell perfor- mance (requires trade-offs - efficiency and cell perfor- mance (requires trade-offs - efficiency versus cost) Performance Constraint - electrode life requirements from impact of electrolyte loss Performance Constraint - temperature control requires AT variance and maximum local temperature in cell is constraint Performance Constraint - temperature control for acid conc., performance Variance & CO poisoning Performance Constraint - temperature control for acid conc., performance Variance & CO poisoning Performance Constraint - temperature control for acid conc., performance Constraint - temperature control for acid conc., performance Constraint - temperature control for acid performance Constraint - temperature control for acid conc., performance Versus pressure level needed functional Constraint - temperature control for acid perating cost versus pressure level needed functional Constraint - temperature control of cell acid performance Constraint - temperature control of cell acid</pre>
	Coolant Temperature (Max)	ч (TBD) h	200°C (398°F)	Performance Constraint - life requirement impact from excessive cell temperature conditions

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TABLE 2-14

TABLE 2-14 Key PAFC DESIGN PARAMETERS, VALUES AND CONSTRAINTS¹ (CONT'D)

ar	ameter Identity	Time	Level	Origin/Comments
Cel Pre	l Coolant Diff. ssure (Max)	Instantaneous	(TBD)	Functional Constraint - design limit for separation requirement and cell assembly and sealing
Rea Pre	ctant Diff. ssure (Max)	Continuous	(180)	Functional Constraint - design limit for separation requirement and cell assembly and sealing
Cel (Ma	l Stack Temp Diff. x to Min)	(TBD) h	20-50°C (36-90°F)	Performance Constraint - life requirement impact
Cel (Ma	l Stack Temp Diff. x to Min)	Continuous	℃ 20°C (36°F)	Performance Constraint - life requirement impact
Tra Pow	nsient Rates - er	Startup Shutdown	(TBD) (TBD)	Performance Constraint - life requirement impact
Sta	ck Mechanics:		·	
1	Contact Resistances	Continuous	<20ເເດີ-cm ² (8ແນ-in ²)	Performance Constraint for practical losses (<2 percent output)
ı	Clamping Łoad (Initial)	Continuous	70-210 kPa (20-60 psi)	
1	Clamping Load (Min)	Continuous	(TBD)~6.9 kPa (1 psi)	Performance Constraint - leakage and electrical losses high contact resistance
	Creep Rates/ Strengths	;	(TBD)	Performance Constraint - life requirements impacts
ł	Maximum Stack Height (Module Unit	1	3.3 m (10 ft)	Imposed Constraint - truckability
1	Maximum Width (Module Unit)	1		Imposed Constraint - truckability

1

TABLE 2-14

KEY PAFC DESIGN PARAMETERS, VALUES AND CONSTRAINTS^E (CONT'D))

Para	meter Identity	Time	Level	<u>Origin/Comments</u>
E V	faximum Weight (Module Unit)	1		Imposed Constraint - truckability
<u>></u> >	4aximum Output /oltage (Stack)	Instantaneous	< 400V DC	Functional Constraint - design limit is fixed as functior of number cells in stack at 0.C. voltage
∑≥ ï	Aaximum Output /oltage (Stack)	Cont i nuous	< 300V DC	Nominal design condition
<u>ک</u> >	dinimum Output (oltage (Unit)	Continuous	> 2000V DC	Functional Constraint - nominal design condition Performance Constraint - conversion efficiency

Condensed list of key constraints remaining after having assumed fixed process controls and fixed performance parameters for baselined reference design. These parameters are needed for further performance parameters for baselined reference design. design and operating requirements definition. Ê

(2) Corrosion impact - not yet determined.

2-30

(3) Amount of H3PO4 vaporized; currently perceived upper limit.

TABLE 2-15 PAFC DEVELOPMENT GOALS

Cell Performance:(1)

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Pri	nary	<u>Goal</u>	Demonstration
•	Voltage: Cell Potential - ayg. mV/cell 2 x 2 @ Design ⁽²⁾ 12 x 17 @ Design ⁽²⁾	705 680	678 + 12 645 <u>+</u> 12
•	Endurance Voltage Loss - mV/1000 h Corrosion Loss - cm/yr (mil/yr)	2 2 (2.54 x 10 ⁻³)	10-15 NA
Sec.	<u>ondary</u> Catalyst Anode Loading - mg/cm ² (1b/ft ²) Cathode Loading - mg/cm ² (1b/ft ²)	0.25 (.5 x 10 ⁻³) 0.50 (1.0 x 10 ⁻³)	0.30 (.5 x 10 ⁻³) 0.50 (1.0 x 10 ⁻³)
•	Oxidant Utilization with Air: Oxygen ⁽³⁾ - percent	50	50 _.
•	Fuel Utilization with SRG: Hydrogen ⁽³⁾ - percent	83	83
•	Operating Temp. of Cell: Cell Avg. Temp °C (°F)	190 (374)	190 (374)
(1)	Fixed Parameters: Natural gas feedstock ERC-Westinghouse Air Cooled System Co Mark II "zee"-"tree" Plates MAT-1 matrix 1982 Process Specs/Fab Controls	onfiguration	
(2)	 2) Design Conditions (Beginning of Use): 480 kPa (70 psia) Pressure Level 190°C (374°F) Average Cell Temperature 325 mA/cm² (300 A/ft²) Average Cell Current Density Loading Acid Control Equilibrium 80 percent H₂ Utilization (SRG) 50 percent Air utilization 		
(3)	At 485 kPa (4.8 Atm)	·	•

2.1.3.3 PAFC Module and 100 kW Stack Fuel Cell Requirements

•

The performance and design requirements for the 100 kW stack and module design defined in the First Logical Unit report were assessed and updated during this reporting period. Revised requirements are presented in Table 2-16 for the key PAFC operating parameters.

2.2 Fuel Cell System Requirements (WBS 1102-02)

The primary objective of this task is to develop test specifications in accordance with the FCS requirements discussed in the previous section for subscale (2x2 in.) fuel cells, nine and ten cell stacks, 10 kW stack and the 25 kW stacks. The test specifications include requirements for steady state and transient performance, endurance, key statepoint data, gas analyses and chemistry, and structural considerations. The following sections summarize the results of these efforts.

2.2.1 25 kW Stack Fuel Cell Test Specification (WBS 1102-02-300)

Although a preliminary 25 kW stack specification has not been issued, system performance requirements and design requirements were identified and documented during the First Logical Unit of Work. Updated 25 kW stack performance requirements are given in Table 2-17.

2.2.2 Fuel Cell Hardware Test Specifications (WBS 1102-02-400)

A number of subscale cell and nine cell stack test specifications were issued to perform tests to obtain the technology characterization for fuel cell and stack design and performance prediction. The 10 kW stack test and measurement requirements and testing conditions were defined and the test specification initiated.

Subscale cell test specifications were issued for a group of technology characterization tests and quality verification tests. A summary of the

TABLE 2-16

100 kW STACK PERFORMANCE AND DESIGN REQUIREMENTS

PARAMETER	UNITS	NOMINAL	DESIGN RANGE
POWER	^{k₩} dc	94	0 < 120
TEMPERATURE	°C(°F)		
Oxidant Inlet*		182 (360)	Amb. to 204 (400)
Coolant Inlet		143 (290)	Amb. to 160 (320)
Fuel Inlet		190 (374)	Amb. to 204 (400)
Plate Avg.		190 (374)	Amb. to 204 (400)
*Same as coolant outlet.			
PRESSURE	kPa (psia)	485 (70)	101-107 (15-85)
FLOW	g/s (lb/hr)		
Fuel 83% H ₂ util.	•	12.6 (100)	15 (< 120)
Oxidant (air) 50% O ₂ Util.		97.6 (775)	113 (< 900)
Coolant (air)		1934 (15,350)	2268 (< 18,000)
CELL VOLTAGE	mV	·	
Open Circuit		920	0 - 1000
Operating Limit		800	
Operating Design Point		680	600 - 690
CELL CURRENT DENSITY	mA/cm ² (A/ft	²)	
Minimum Operating		50 (47)	0 < 400 (372)
Design Point	•	325 (300)	250-400 (235-372)
FUEL STACK VOLTAGE	Volts		<i>,</i> .
Open Circuit		360	< 385
Operating Limit		315	< 335
Operating Point		270	210-335
FULL STACK CURRENT	Amps	350	0 < 430

TABLE 2-17

25. kW STACK PERFORMANCE REQUIREMENTS

The performance requirements for the 25 kW stack design are listed below:

PARAMETERS	<u>UNITS</u>	NOMINAL	DESIGN RANGE
POWER	^{KW} dc	25	< 30
TEMPERATURE	°C (°F)		
Oxidant Inlet		(184) 363	(149-204) 300-400
Coolant Inlet		(136) 277	(110-160) 230-320
Fuel Inlet		(190) 375	(149-204) 300-400
Plate Avg.		(190) 375	(149-204) 300-400
PRESSURE	kPa (Atm)	485 (4.8)	101-606 (1 - 6)
FLOW	g/s (lb/hr)		
Fuel		1.1 (8.75)	1.1 (8.75)
Oxidant (air)		26.3 (209)	<52.7 (<418 <u>)</u>
Coolant (air)		419 (3325)	365-548 (2900-4350)
CELL VOLTAGE	mV		
Open Circuit		920	< 1100
Operating Limit		800	
Operating Point		680	500-800
CELL CURRENT DENSITY	mA/cm ² (A/ft ²)	325 (300)	< 400 (372)
· .			·· ·

subscale test specifications issued is given in Table 2-18. The purpose of the first group of tests was to: (1) verify process and fabrication reproducibility and the process control requirements, (2) establish comparative capabilities of electrodes with other vendor catalysts, (3) to determine the impact of some process variables and material sources, and (4) to meet the requirements stipulated in the technology development program.

Test specifications were defined and issued for nine cell stacks W009-08, -09, -10 and -11. Test article data and the test operations are listed in Tables 2-19 and 2-20. Revisions were issued for stacks W009-08, -09 and -11 to accommodate changes in the test plan. Test results and conclusions derived from these test plans from the experiments that had been conducted are presented in Section 3.0.

2.2.3 Fuel Cell Manufacturing Process Specifications

The Nine Cell Stack Subassembly Procedure (PAFC-006) and Final Assembly Procedure (PAFC-007) were modified for each stack fabricated during the report period. The modifications to the procedures were made to accommodate the design changes peculiar to each stack.

Disassembly procedures were prepared and utilized for stacks W-009-04 and W-009-10.

2.3 Systems Integration (WBS 1102-03)

2.3.1 Fuel Cell System Design (WBS 1102-03-100)

FUEL CELL SYSTEM PRELIMINARY DESIGN DESCRIPTION

The FCS (Figure 2-3) consists of groups of ten fuel cell modules supported in two rows of five together with their piping on an elevated platform. Two such ten-module groups are employed in a 7.5 MW plant. The fuel cell module design is described in Section 3.6. Each module has an overall height of approximately 3.5 m (11 ft 6 in.) and diameters of 1.4 m (4 ft 6 in.) over the

TABLE 2-18 SUBSCALE TEST SPECIFICATIONS ISSUED

Test Specification	Test Objective
TS 2 X 2-001	- Reproducibility (Fab, Ass'y & Test)
TS 2 X 2-002	- Five Manufacturing Variable Effects
TS 2 X 2-004	- Initial Baseline Electrode Characterization at Pressure
TS 2 X 2-005	- Initial Baseline Electrode Characterization
TS 2 X 2-006	- Screening of Alternate Vendor Catalyst
TS 2 X 2-007	- Alternate Backing Paper/Process Evaluation
TS 2 X 2-008	- Electrode Bonding and Sintering Assessment
TS 2 X 2-010	- 10 kW Stack Electrode Performance
TS 2 X 2-011	- Alternate Acid Transport Evaluation
TS 2 X 2-012	- Manufacturing Variables Evaluation, Catalyst and Backing Paper Wet Proofing
TS 2 X 2-013	- 10 kW Stack Electrode Quality Verification
TS 2 X 2-014	- Pressurized Endurance Demonstration
TS 2 X 2-015	- Screening of Manufacturing Variables
TS 2 X 2-016	 Non-Electrode Catalyst Poison/Contamination Assessment

	30	20	00	00		
M009	90-	10-	-08	60-	-10	-
s 1 & 2	Basel ine	Dry Laminated (900° HT Catalyst)	Baseline	o006)	Laminated - HT Catalyst)
s 3-7	Baseline	Baseline	Baseline	Dry (900°	Laminated - HT Catalyst)
s 8 & 9	Baseline	Dry Laminated (900° HT Catalyst)	Baseline	Dry (900°	Laminated - HT Catalyst)
l Matrix	Yes	Yes	Yes	No	Yes	Yes
lar Plates	Machined, 9	1H °000	Molded	A99 Regrind	Molded	
·	Teflon		Viton	Viton	Viton	Viton
	97% 50 ml	93% 46.5 ml	93% 42 ml*	98% 27 m]*	98% 45 ml*	98% 45 ml
Instrument	No	Yes	No	No	No	Yes
tional Tests	Start/Stop	Load Var.	Start/Stop	Start/Stop	Start/Stop	Load Var.

TABLE 2-19 NINE CELL STACK TEST ARTICLE DATA *Stack to be acid wicked at 130°F, dry room conditions for 3 to 7 days prior to testing.

· .		NINE CEI	TABLE 2-20 LL STACK TEST PLAN SEQUE	.NCE(1)	
PHASE I		• .			PHASE II
STEP 1 ATMOSPHERIC TESTS		STEP 2 PRESSURIZE	STEP 3 STEADY STATE PRESSURIZED	STEP 4 MAPPING PRESSURIZED	ADDITIONAL OPTIONAL TEST(4)
< 50 Hrs		4 Hrs	500 Hrs	< 100 Hrs	500 Hrs
 Repeatability(2) Open Circuit Voltage Conditioning 	• • •	Startup to part power Up to full rated power condition Initiate SRG	Polarization ⁽³⁾ Conditions 375°F 70 psia 40% air, 83% H ₂ 325 mA/cm ²	l6 points Polarization @ end of mapping	 Shutdown Polarization Polarization Polarization Open Circuit Voltage Operational Transients Endurance Additional Mapping Additional Mapping Start/Stop Sequence Cycling Other Development Options
Noted: (1) All full pre	nssa	ırized tests run wı	ith SRG except H ₂ gain		
<pre>(2) Verify measu required.</pre>	nen	ents with stack ir	ıstalled and stack contr	ol repeatabilit	y within uncertainty
(3) Minimum of 1 levels each	four poi	<pre>points - H2 util nt from 25% to 603</pre>	ization constant, oxidan 6.	it flow rate hel	d at four utilization
<pre>(4) Continued of development</pre>	pera may	ition for up to 500 / extend these opti	0 hours on SRG and TBD o ions).	ptions, transie	nts (continuing
				· · · ·	· · · · · · · · · · · · · · · · · · ·



pressure vessel cylinder and 1.7 m (5 ft 6 in.) over its lower flange and support region. The weight of each module is approximately 5500 kg (6 tons).

Each of the fuel cell modules is designed to be truck transportable by common carrier. The module dimensions and weight are within the envelope permissible for truck transportation. The structural design of the modules is such that they can withstand the loads imposed by this type of transportation, as well as expected seismic loads. The module pressure vessels will be designed in accordance with Section VIII (Division 1) of the ASME Pressure Vessel Code.

Cooling and process air is supplied through a 1.2 m (4 ft) diameter pipe which traverses the length of the system immediately below the platform and midway between the two rows of modules. Air is extracted from this pipe and is supplied to each module through a short 40 cm (16 in.) diameter branch pipe. The cooling air, after traversing the module internals, is returned through another 40 cm (16 in.) line to a 1.2 m (4 ft) diameter air return line, supported immediately below the air supply line, midway between the rows of modules. A portion of the spent cooling air is extracted for use as process air internally to each module.

The 15 cm (6 in.) diameter fuel inlet and outlet lines are supported in two rows immediately below the outermost edges of the mounting platform. The fuel lines and the 20 cm (8 in.) diameter process air return lines, which are similarly supported, are connected to penetrations at the module pressure vessel lower head through lengths of small diameter piping. This piping is 5 cm (2 in.) diameter in the case of the fuel supply and return, and 10 cm (4 in.) diameter in the case of the process air return.

The symmetry of the piping arrangement provides for uniform distribution of fuel, cooling and process air between the modules comprising the FCS. The pipe sizes are selected so that the pressure drops through the fuel cell stacks dominate the flow resistances in the system, thus promoting equal flow distribution between modules as well as between the fuel cells within a module.

Shutoff valves are provided in all the fuel and process air connections to the modules to enable any of the modules to be shut down independently of the system. Valves are not provided in the 40 cm (16 in.) cooling air connecting lines to the modules so that continued operation of the fuel cell system with a module or modules shut down will therefore result in the continued circulation of cooling air through the shut down module or modules. The modules and their piping system and valves are completely encased in thermal insulation to reduce heat losses to acceptable levels.

Cable trays, located below the platform, support the electrical power leads and instrumentation lines between the modules and the PCS and ICS.

The overall height of the FCS is 8 m (25 ft). The overall length and width of the ten module assembly are approximately 10 m (33 ft) and 7 m (23 ft), respectively. The module support platform is approximately 4 m (14 ft) above foundation level. Foundation level must be located approximately 1.5 m (5 ft) below ground to provide for module handling using a permanent overhead crane and meet the 9.1 m (30 ft) height limitation of the crane.

The physical location of the FCS ten module groups in relation to the interfacing plant systems is suitably arranged so that the thermal expansion effects in the large 1.2 m (4 ft) diameter cooling air pipes can be accommodated by hinged bellows without overstressing the pipes in bending or exceeding limiting axial loads and movements at the FCS interfaces.

The smaller process gas piping is less constrained in this respect due to the substantially smaller diameters involved. However, some modification of the process gas piping geometry within the FCS is required to increase the bending flexibility of the vertical legs to accommodate thermal expansion of the horizontal piping.

2.3.1.1 Fuel Cell System Maintenance

Access is provided to the fuel cell modules from below for attention to electrical connections and acid replenishment systems. Access to the piping

and shutoff valves below the ten module group is also available. Access is also provided from the sides of each ten module group and via an aisle between the two five module rows within each ten module group to facilitate removal of the pressure vessel flange bolts when fuel cell cartridges must be replaced.

Detailed maintenance requirements for all of the components of the FCS have not yet been defined. However, an initial assessment of the module maintenance requirements including acid replenishment and cartridge replacement, is included in Sections 3.6.4 and 3.6.5. As the FCS design matures and specific items of equipment are selected, such as valves, pipe hangers, instrumentation, etc., additional maintenance requirements may be identified.

2.3.1.2 Structural Analysis

A structural analysis of the entire fuel cell system piping and support structure was initiated using the finite element computer program ANSYS. Figure 2-4 illustrates a three-dimensional view of the entire FCS piping and support structure model containing over 1100 elements. The purpose of this analysis is to demonstrate that the stresses in the piping and support structure are within allowable limits or to recommend design changes to insure structural adequacy of all components. The allowable design stresses are based upon ANSI B31.1 (1980 edition) Power Piping Code and the ASME B&PV code, Section VIII (1983 edition), Division 1, Appendix G.

The finite element model includes all large and small pipes, support structure, and the module vessel baseplates. The weights of the vessels are included as mass elements acting at the center of the baseplates. The work platform and handrails were not included as structures; however, their weights were included. The bolted connection between each vessel base and the support beams was included. Specific details of this connection will be considered in the thermal stress analysis. Flexibility factors for the pipe elbows and stress intensification factors at the pipe connections are included based on the ANSI B31.1 Power Piping Code.



Figure 2-4. 3-Dimensional ANSYS Finite Element Model

The loading conditions considered in the analysis included dead weight, earthquake loads (1/2 g), thermal expansion (assuming a -29°C (-20°F) day based on EPRI TAG guidelines), and internal piping pressure. The frame columns are assumed to be rigidly connected to the concrete base.

This analysis effort is now in progress; however, no results or conclusions are available at this time.

2.3.2 Systems Interface Requirements and Control (WBS 1102-03-300)

This section presents the drawings which will be used to define and control the developmental systems interfaces. The Interface Control Drawings (ICDs) illustrate each system in simplified form and show the interfaces with other plant systems. These drawings were developed as part of a complete ICD package for each plant system which also contains information regarding each interface, foundations and supports requirements, electrical interface requirements, and measurements requirements. The ICD packages provide the means to identify and maintain control of all interfaces as needed throughout each system's design development. The developmental systems ICDs which are shown in Figures 2-5 through 2-9 are listed below.

<u>System</u>	Figure No.
FCS	2-5
FPS	2-6
RES	2-7
PCS	2-8
ICS	2-9



Electrical control interfaces with ICS 3

.

Figure 2-5. Fuel Cell System Interface Control Drawing



NOTES: 1. Equipment foundations and supports by AE 2. Electrical power interfaces with PDS

Figure 2-6. Fuel Processing System Interface Control Drawing



Figure 2-7. Rotating Equipment System Interface Control Drawing



ES

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LEGEND

PCS., PDS., etc. - System Acronymus:

AE. -- Architect. Engineer

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Power Conditioning System Interface Control Drawing

Figure 2-8.



Figure 2-9. Instrumentation and Control System Interface Control Drawing

3.0 FUEL CELL DEVELOPMENT AND TEST (WBS 1103)

3.1 Subscale Fuel Cell Development (WBS 1103-01)

3.1.1 Design and Assembly Status

The principal subscale design and assembly procedure utilized was that developed during the First Logical Unit of Work. Gasket requirements were revised to accommodate cells assembled without MAT-1 matrix material as well as cells with alternate matrix material.

A metal end plate design was evaluated to provide suitable end plates for pressurized endurance testing of up to 5,000 hours. Gold plating of tantalum was considered but since a reliable plating process was not readily available, it was concluded that gold plated nickel 200 would be evaluated. To select the gold plating process for the nickel 200 design, four simplified end plates with various types of gold plating were corrosion tested by immersion in 97 w/o H_3PO_4 at 190°C (374°F), 485 kPa (4.8 atms.), with no applied potential for over 100 hours. Following visual examination, three types of coating were subjected to a second pressurized, 485 kPa (4.8 atm.), corrosion test for an additional 500 hours. The gold plating selected was a combination of 5 microns (.2 mils) of gold applied by electroplating. Five cells were assembled utilizing gold plated hardware and tested, four at pressure and one unpressurized. The gold plating failed during cell operation.

Work was initiated on heat treating sections of subscale 900°C (1650°F) heat treated carbon end plates to 2700°C (4950°F). Based on the success of this work on high temperature heat treatment of carbon end plates and the relatively poor performance of cells built with gold plated nickel 200 plates, it was concluded that the effort would be concentrated on procuring 2700°C (4950°F) heat treated carbon end plates for pressurized endurance evaluation of baseline and alternate catalysts.

A design revision was completed and drawings prepared for a subscale cell using heat treated carbon end plates that can accommodate a variety of matrices. The matrix size was increased to the cathode size and is now larger than the anode, making the subscale configuration similar to the stack configuration. Anode and cathode gasket widths were modified to provide a 0.25 cm (0.10 in.) overlap between the anode gasket and the matrix or between the anode gasket and the cathode if a matrix is not used. This overlap should decrease the possibility of cell cross leakage in this area. Teflon gaskets (of different thicknesses) are included to facilitate disassembly of the cell after testing and to provide capability to adjust for variation in electrode and gasket thickness. The design provides for a nominal 12 percent electrode compression coupled with a

15 - 20 percent elastomer seal compressive strain range.

3.1.2 Subscale Testing

. 9 TO 1 F

A total of 160 cells, unpressurized and pressurized, were assembled and tested at Westinghouse. With the exception of five cells, all the testing was based on test plans, and the status of these test plans and cells built is provided in Table 3-1.

Entry Processing of the Second Second

An additional sixty-nine subscale fuel cells were assembled and tested during this period. Testing of eleven subscale cells assembled during the first logical unit of work was also conducted. These cells were tested to address various areas of work (Table 3-2) as identified in the performance improvement work plan developed by Westinghouse and ERC to achieve the power plant performance goals. Table 3-3 summarizes the test data. Specific test objectives and results for these cells are discussed under relevant subsections of Task 1103-08.

The overall objectives of the various test plans are:

 To establish the baseline technology with existing ERC/<u>W</u> design/processes.
WAESD-TR-85-0030

TABLE 3-1

CELLS TESTED UNDER VARIOUS TEST PLANS

Test Specification TS 2X2-002, Rev. 0 TS 2X2-002, Revs. 1 and 2 TS 2X2-004, Rev. 2 and 4 TS 2X2-004, Rev. 3 'TS 2X2-005, Revs. 2 and 3 TS 2X2-006, Rev. 0 TS 2X2-006, Rev. 1 TS 2X2-007, Revs. 0 and 1 TS 2X2-008, Rev. 0 TS 2X2-010, Rev. 0 'TS 2X2-013, Revs. 0, 1, 2, 3 and 4 TS 2X2-014, Rev. 0 and 1

TS 2X2-016

Cells Built and Tested	
CAS-001 through -016, -001-S1, 005-S1	
CAS-061 through -072, CAS-112 through -115	
CPS-017 through -024, CPS-018-S1 -020-S1, -021-S1, -023-S1, -023-S2, CPS-045 through -047	,
CPS-026, CPS-028, -029, CPS-031	
CAS-017, -023, -090, -091, -096 -097	
CAS-026 through -039	
CAS-047 through -049, CAS-051 through -060, CAS-100 through -106, CAS-108	
CAS-040 through -045, CAS-073 through -076	
DAS-001 through -010	
QAS-001, -002, -015, -016, -019 through -024, DAS-047, -048	3
QAS-006 through -009, QAS-009-S1 QAS-016 through -020, QAS-019-S1 QAS-019-S2, QAS-025 through -027 QAS-009 through -020	2
CPS-033 through -036	
CAS-116 through -121	

Test Purpose	Number of Cells Tested	Task 1103-08 Reference Section
Alternate Catalyst	19	1.1
Alternate Catalyst Support:	. 7	1.2
Electrode Manufacturing	14	1.3
Cell Resistance	10	1.4
Plate Corrosion	9	1.5
Quality Control	6	1.8
Miscellaneous	15	-

TABLE 3-2 SUMMARY OF SUBSCALE CELLS TESTED

Total

80

TABLE 3-3 SUMMARY OF SUBSCALE TEST DATA

			IEST TECHINALICH DALE	14/02/101	04/14/83 04/19/83 08/22/83 08/22/83	01/10/84 02/01/84 09/13/83	11/15/03 11/15/03 02/03/04 12/09/05	01/01/01 01/01/01 03/20/04		13/41/10 18/61/10 18/61/10	02/01/04 02/01/04 02/02/04 01/02/04 01/02/04 01/02/04 04/04	11.007.05 11.007.05 11.14.01.01 11.14.01 11.14.01 11.14.01 11.14.01 11.14.01 11.14.0	05/04/81 02/24/84 03/06/44	03/27/#4 02/24/84 02/24/84
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		• .	1651 SPEC								8000 8000 8000 8000 8000 8000 8000 800	100-100-100-100-100-100-100-100-100-100	~ ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	600 600 600 600 600 600 600 600 600 600
			CELL	MI-I MI-I SIC	51C 51C 51C	1-141 1-141 1-141 1-141	20020	SIC SIC SIC SIC SIC	NAT-1 NAT-1	RAT-1 SIC SIC				
· .			Stnr.	A A A A A A A A A A A A A A A A A A A	818 818 818 818	ala Ala Ala Ala			22	an a			4 23	393
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TABLE 3-3 SUMMARY OF SUBSCALE TEST DATA (CONT'D)

- To provide a base for stack scaling, and
- To determine the effects of principal parameters in order to develop predictive correlations for performance scaling.

Table 3-4 shows the objectives and status of the various test plans.

3.1.2.1 Objectives and Results of Various Test Plans

### TS 2X2-002, Rev. 0, 1, 2

The goal of this test plan is to determine the effect of the following parameters on electrode/cell performance.

- Two different shipments/lots of catalyst.
- Catalyst heat treatment/nonheat treatment.
- Two different heat treatment lots.
- Catalyst loading on cathode.
- Wet and dry bonding of catalyst layer to the substrate.
- Two different SiC lots.
- Thin, < 125 microns (< 5 mils), and thick SiC layers.</li>
- SiC material Polyox 1 and Polyox 2.

Table 3-5 shows the performance and status of the cells built in this test plan with the cell voltages (terminal) reported at 200 mA/cm² (186 A/ft²), 101 kPa (1 atm), 190°C (374°F), 80 percent hydrogen utilization and 20 percent air utilization.

Initial baseline electrode characterization at pressure Extension to screen a production wetproofing process Manufacturing variables evaluation, catalyst and Reproducibility (fabrication, assembly and test) Rev. O in new format and 10 kW stack electrodes Additional tests required by facility problems Initial baseline electrode characterization Addition of cells for Kureha 604 alternate Alternate backing paper/process evaluation Electrode bonding and sintering assessment Addition of other catalyst and elecrodes Addition of other catalyst and electrodes Addition of other catalyst and electrodes Addition of cells for 4-1/2% Teflon-MAT-1 Screening of alternate vendor catalyst Update to use new baseline electrodes Update to use new baseline electrodes Alternate acid transport evaluation Five manufacturing variable effects Replacement of inappropriate cells Test Objective 10 kW stack electrode performance Addition of four variables backing paper wetproofing Rev. 0 reformattted SUBSCALE TESTING STATUS TABLE 3-4 2 2 erminated (updated) Reformatted as Rev. Reformatted as Rev. Cancelled (updated) Test Results Retest needed Retest needed Ongoing Scheduled Ongoing Scheduled Completed Completed Completed Completed Cancelled Scheduled Scheduled Complete Complete Ongoing Ongoing Ongoing Ongoing Rev. 0 0 4 3 2 0 Rev. 0 Test Specification  $\circ$  $\sim$ O 0 Rev. ( Rev. ] Rev. TS-2X2-006 **TS 2X2-001** TS 2X2-002 **TS 2X2-005** TS 2X2-012 **TS 2X2-004** TS 2X2-008 TS 2X2-010 TS 2X2-007 TS 2X2-011

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TABLE 3-4 SUBSCALE TESTING STATUS (CONT'D)

Test Specif	ication	Test Results	Test Objective
TS 2X2-013	Rev. 0 Rev. 1 Rev. 2 Rev. 3 Rev. 4	Completed Ongoing Ongoing Ongoing Scheduled	10 kW stack electrode quality verification Additional batch quality verification Additional batch quality verification Additional batch quality verification Additional tests to qualify pressurized endurance cells
TS 2X2-014 TS 2X2-015	Rev. 0	Hold	Pressurized endurance demonstration
TS 2X2-016	Rev. 0	ocreau rea Ongoi ng	Surgenting of manufacturing variables Non-electrode catalyst poison/contamination assessment
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# TEST 2x2-002 REV. 0 ASSESSMENT OF FIVE ELECTRODE VARIABLES

(Status as of 5/31/84)

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	Test Condi	tions: 1	Atm, 80% H ₂	Util., 20%	Air Uti	1., - 2	:00 mA/c	cm2 @ ]	90°C						
-	Initial	200 Hour	s 600 Hours	1,000 Hour	s '	PRESENT	TEST S	STATUS	CATI	HODE		ANC	DE	•	
ID	kesistance (ma)	E Cell (mV)	E cel-I (mV)	E cell (mV)	E cell (mV)	Total C Hours 1	)ff Test (	Cat.	Bond. S	int. (	Cat. E	sond. Si	nt. Ma	- t-l Assembly and	est Comments
CAS-001	6.8	623	405	ſ	405	715	7/83 5	Std	Wet A1	. <u>.</u>	Std v	let Ai	- ×	Low Pt, Thin	iC. Polvox #1
CAS-001-S1	8.8	623	626	616	566	1749 1	0/83 5	Std I	Wet Ai	<u>ب</u>	itd W	et Ai	r R	Low Pt, Thin S	iC. Polyox #1
CAS-002	6.6	646	646	618	630	2160	9/83 h	- F	Wet Ai	L L	ہ ب	let Ai	ч К	Low Pt, Thin	iC, Polyox #2
CAS-003	3.1	626	600	528	476	1293	6/83 5	Std	Wet Ai	Ŀ,	Std W	et Ai	۲ ۲	High Pt, Thin	SiC, Polyox #2
CAS-004	5.8	627	625	611	594	1992	9/83 F	÷	Wet Ai	<u>ب</u>	Ť.	let Ai	r R	High Pt, Thin	SiC, Polyox #1
CAS-005	•	610	ı	ı	490	480	7/83 S	std	Dry Ai	<u>י</u>	itd D	ry Ai	r	Low Pt, Thin S	iC, Polyox #2
CAS-005-S1	6.7	632	628	624	617	1221 1	0/83 5	Std	Dry Ai		Std C	'ry Ai	r	Low Pt, Thin	iC, Polyox #2
CAS-006	6.9	657	655	655	563	1824	9/83 H	-	Dry Ai	<b>.</b>	Ц	ry Ai	r	Low Pt, Thin S	iC, Polyox #1
CAS-007	7.8	615	610	611	594	2472	9/83 5	std	Dry Ai	<u>ب</u>	Std C	'ry Ai	r	High Pt, Thin	SiC, Polyox #1
CAS-008	8.0	. 650	589	609	541	2160	9/83 H	ť	Dry Ai	۲.	1	ry Ai	۲	High Pt, Thin	SiC, Polyox #2
C.AS-009	6.9	607	580	. 009	592	2176	9/83 S	itd	Wet Ai	<u>ب</u>	td W	et Ai	r R	Low Pt, Thick	SiC, Polyox #2
CAS-010	6.8	641	642	639	634	1486	7/83 H	IT L	vet Ai	r T	M 11	et Ai	r R	Low Pt, Thick	SiC, Polyox #1
CAS-011	7.6	644	641	. 632	607	1733 1	0/83 S	std 1	Wet Ai	<u>د</u>	Std W	let ⊂ Ai	r R	High Pt, Thick	SiC, Polyox #1
CAS-012	2.9	644	643	630	584	1845 1	0/83 H	1	vet Ai	<u>۲</u>	H H	et Ai	r	High Pt, Thick	SiC, Polyox #2
CAS-013	6.7	635	632	625	523	2376	8/83 5	itd (	Dry Ai	r.	std D	ry Ai	r R	Low Pt, Thick	SIC, Polyox #1
CAS-014	7.4	639	637	594	296	1648	9/83 H	ц т	Dry Ai	۲ ۲	1	ry Ai	2	Low Pt, Thick	SiC, Polyox #2
CAS-015	0.6	621	605	605	561	1652	9/83 S	itd .	Dry Ai	r.	std D	ry Ai	r R	High Pt, Thick	SiC, Polyox #2
CAS-016	9.4	635	634	630	472	2520	9/83 H	Ŧ	Dry Ai	r	Ľ	ry Ai	r R	High Pt, Thick	SiC, Polyox #1
		HT - 9(	00°C Heat Tr	eated Pt.			÷.	· ·	R - Smal	11 Batc	ch Ross	Mixed	•		-
	·	(-51) Ret	built identi	cal sister (	cells		•	•		•	• •	-		•	
•				• •	•		4								
Lot #3 Cati	ilyst was us	ed in all	cells		•	۰.									

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TEST 2x2-002 REV. 1 and REV. 2 ASSESSMENT OF FIVE ELECTRODE VARIABLES (CONT'D)

(Status as of 5/31/84)

Test Conditions: 1 Atm, 80% H₂ Util., 20% Air Util., - 200 mA/cm² 0 190°C

200 Hours 600 Hours 1,000 Hours PRESENT TEST STATUS

Initial

ANODE

CATHODE

CAS-DEE	4	667	656	6 E.A	620	0010		τn	200	Ann	140	200	Ann		
CAS-064	6.2	650	652	654 .	645	1341	3/84 4/84	HT H	ory Ury	Arg Arg	Std	ory	Arg	n n	
CAS-065	6.8	657	656	654	628	2488		Ħ i	Dry	Arg	Std	Dry	Arg	·ν	
CAS-066	6 <u>.</u> 3	647	653	653	645	2488		ΗT	Dry	Arg	Std	Dry	Arg	S	
CAS-067	7.6	656	661	653	652	1509	4/84	HT	Dry	Arg	Std	Dry	Arg	S	
CAS-068	10.1	627	630	633	632	1341	4/84	Ħ	Dry	Arg	Std	Dry	Arg	S	
CAS-069	8.1	642	647	645	635	1744	5/84	H	Dry	Arg	Std	Dry	Arg	S	
CAS-070	7.8	630	636	637	641	1509	4/84	Ħ	Dry	Arg	Std	Dry	Arg	S	Compares to CAS-014
CAS-071	0.6	. 643	ı	ı	647	378	3/84	HT	Dry	Arg	Std	Dry	Arg	s	
CAS-072	.9	652	658	. 658	635	2848		HT	Dry	Arg	Std	Dry	Arg	S	• .
CAS-112	6.0	. 630	598	608	609	640 -		HT	Dry	Arg	Std	Dry	Arg	S	
CAS-113	7.6	638	632	626	633	640		HT	Dry	Arg	Std	Dry	Arg	S	Compares to CAS-008
CAS-114	8.3	634	626	615	624	640		Ħ	Dry	Arg	Std	Dry	Arg	S	
CAS-115	7.7	654	655	652	655	640		HT	Dry	Arg	Std	Dry	Arg	S	Compares to CAS-016

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S - Large Batch Schold Mixed

HT - 900°C Heat Treated

The results of this plan can be summarized as follows:

- The present baseline cell performance (mean of 12 cells) is 648 + 6 mV.
- The effect of different catalyst lots (lots 2 and 3) was within the expected + 10 mV deviation.
- Catalyst heat treatment has 10 25 mV advantage over non-heat treated catalyst.
- The two furnace heat treatment runs of catalyst were within the  $\pm$  10 mV expected deviation.
- The effect of catalyst loading, SiC coating thickness and manufacturing process were interactive with + 21 to -26 mV variability over mean.
- There was no significant effect of dry bonding over wet bonding on cell voltage (was within the <u>+</u> 10 mV expected deviation); however, dry bonding was preferred over wet bonding due to ease of manufacturing.

The effect of new SiC lot was within the ± 10 mV expected deviation.

 Thinner SiC coating (from 250 microns (10 mil) to 125 microns (5 mil)) provided 12 + 6 mV benefit.

### TS 2X2-004, Rev. 2, 3 and 4

The objective of this test plan was to characterize pressurized 2 x 2 subscale cells; i.e., to quantify the map of acceptable fuel cell operation and to provide data base for assessment of stack scaling effects. The approach was to assemble and test pressurized 2 x 2 cells with the following variables: catalyst heat treatment, current density, pressure, and temperature. The results and the status of these tests are given in Table 3-6.

TEST 2x2-004 REV. 2 AND REV. 4 PRESSURIZED ELECTRODE CHARACTERIZATION

(Status as of 5/31/84)

Test Conditions: 4.8 Atm., Util. 80% H₂ (SRG), 20% Air Util. - 200 mA/cm²0 190°C

**Cells terminated due to low performance. 20 hr voltages at rated conditions before hardware failures were 715 mV, 737 mV, 685 mV, and 736 mV, respectively. (-S1 or -S2) Rebuilt identical sister cells

***Cell went erratic after shift from H to SRG (680 mV performance before acid addition shutdown at 700 hrs)

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Only preliminary results were obtained from this test plan, since there were difficulties associated with the functioning of the facility and some premature cell failures due to unknown causes.

### Test Plan - TS 2X2-005, Rev. 2 and 3

The aim of this plan was to characterize the baseline performance of subscale cells by mutual exchange of cells between ERC and  $\underline{W}$ . This type of "Round Robin" test also enables a comparison of the two facilities.

The approach was to fabricate and assemble eight 2 x 2 cells with the pertinent variables being the assembly facility, test facility and test cells with and without MAT-1. The test plans are shown in Table 3-7 with some preliminary results.

These tests are in progress both at ERC and  $\underline{W}$ . From the preliminary data, it can be stated that  $\underline{W}$  built and tested cells averaged 659 mV at 200 mA/cm² (186 A/ft²) and 190°C (374°F) with 80 percent hydrogen utilization and 20 percent air utilization, whereas the ERC cells under similar conditions averaged 660 mV. Further testing, in terms of rebuilding and exchange, is in progress.

### Test Plan TS 2X2-006, Rev. 0 and 1

This test plan was designed to evaluate alternate catalyst/electrode performance and also to perform an assessment of vendor manufacturing processes. The test plan for this task is shown in Table 3-8 and all the cells were evaluated under standard atmospheric conditions at 190°C ( $374^{\circ}F$ ) at 200 mA/cm² (186 A/ft²) and 80 percent hydrogen utilization and 20 percent air utilization.

It can also be observed that Prototech (DCO1) could be considered as an alternate supplier of catalyst. Giner fabricated cathodes (G82-5-10 and 7) had good initial performance which was better than the baseline, but the electrodes

	·······			WAESD-TR-85-0030
Z	l Assembly and Test Comments	Cancelled due to wet sinter low performance. Test units redefined below as CAS-090 thru CAS-097	<u>M</u> Fab. <u>M</u> Test - Sent to ERC ERC Test <u>M</u> Fab. <u>M</u> Test - Sent to ERC ERC Test <u>M</u> Fab. <u>M</u> Test <u>M</u> Fab. <u>M</u> Test	red v
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LITY	Bond	Dry Dry Dry	Dry Dry Dry	arge
DFAGI	E . Cat.	HT-W HT-W HT-W HT-W	Std Std Std	
E ANI	Sint	Wet Wet Wet Wet	Arg Arg Arg	
MANC  /84)	Bond	Dry Dry Dry	Dry Dry Dry	
3-7 PERFOR F 5/31	r STATUS Cat.	HT-W HT-W H-TH HT-W	HT - W HT - W HT - W HT - W	
TABLE S BASELINE F atus as of atus - 200 mA/c	PRESENT TESI Total Off Hours Test	1124 10/83 Not Built 1124 10/83 Not Built	1144 761 1144 761 1144 1144	Fabricated
. 3 (St	E cell (mV)	609 - 610	655 628 632 584 606 649	house ells.
. 2 & REV	1,000 Hours E cell   (mV)	596 - 605	656 635 608 650	ted, Westing cal sister c
x+005 REV 8.8 24 2 20 1. 80% H ₂ ⁻⁰	s 600 Hours E cell (mV)	588 598 -	658 628 642 592 618 652	°C Heat Trea Juilt identi
TEST x2	200 Hour: 2 E cell (mV)	584 -021, -022 600 -	662 630 664 613 637 667	HT-4 900 (-S1) Rel
Test Conditio	Initial Resistance (m2)	8.8 -019, -020, 7.6 -	6.3 8.0 7.0 10.8 9.1 7.0	
	Cell ID	CAS-017 CAS-018, CAS-023 CAS-024	CAS-090 CAS-091 CAS-096 CAS-097	

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	ssembly and Test Comments				ancelled	•			-	endor Fabricated	endor Fabricated	endor Fabricated, <u>W</u> Backing	endor Fabricated, <u>W</u> Backing		erminated; poor performance
	Mat-1 A:	S	s	s	ů		, s	s	s	S	S Ve	s S	s ve	s	S Te
ł	int. 1	Air	Air	Air		Air	Air	Air	Air	Air	Air	Air	Air	Air	Air
ANODE	Sond. S	Wet	Wet	Wet		Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet
	Cat. E	Std	Std	Std		Std	Std	Std	Std	Std	Std	Std	Std	Std	Std
	Sint.	Air	Air	Air		Air	Air	Air	Air	NA	NA	NA	NA	Air	Air
THODE	Bond.	Wet	Wet	Wet		Wet	Wet	Wet	Wet	NA	NA	NA	NA	Wet	Wet
5	Cat.	DC05	DC05	DC04	DC04	DC06	0006	DC07	0007	0006	0007	0006	DC07	0001	DC02
ŝ	0ff Test	12/83	2/83	1/84	ı	2/84		1/84		1/83	2/83	11/83	1/84	1/84	0/83
	Total Hours	2235 1	2235 1	2176	١	3877	6184 [,]	1966	5008	1298 1	1250 1	1316 1	0771	1772	139 1
PKESENI	E cell (mV)	612	631	632	•	604	592	597	629	441	251	636	630	623	537
1,000 HOURS	<pre>E cell (mV)</pre>	632	649	638	: <b>I</b>	663	681	623	672	538	390	649	654	627	•
DUU HOURS	E cell (mV)	631	646	638	J	668	. 682	623 ·	670	612	545	658	660	627	•
ZUU HOURS	E cell (mV)	644	654	· 642	•	673	680	625	672	. 644	627	664	675	633	537
Initial	Resistance (ma)	7.4	6.4	6.3		7.6	6.6	13.2	8.0	8.1	8.2	8.1	5.4	6.9	5.2
	[0	AS-026	:AS-027	CAS-028	:AS-029	CAS-030	AS-031	CAS-032	AS-033	AS-034	AS-035	AS-036	AS-037	:AS-038	AS-039

____ TABLE 3-8

TEST 2x2-006 REV. 0 ALTERNATE CATALYST COMPARISON AND ELECTRODE EVALUATION

(Status as of 5/31/84)

WAESD-TR-85-0030

TEST 2x2+006 REV. 1 ALTERNATE CATALYST COMPARISON AND ELECTRODE EVALUATION (CONT'D)

(Status as of 5/31/84)

<b>--</b>	·.· .				belli																•	WA	ESD	-TR-	·85
•	Assembly and Test Comments			Anode load 0.45 mg/cm ²	Anode load 0.43 mg/cm ² -Cance					Cancelled		•									•				
	Mat-1	s S	s	Ś	S	Ś	ŝ	S	s	s	S	S	s	S	s	s	No	No	S	S	Ś	s	N,		
	Sint.	Air	Air	Air	Air	Air	Air	Air	Air	Air	Air	Air	Air	Air	Air	Arg	Arg	Arg	Arg	Air	Air	Arg	Arg	lixed	
ANODE	Bond.	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Dry	Dry	Dry	Dry	Wet	Wet	Dry	Dry	hold	
	Cat.	DC03	DC03	DCO1	DC02	DC03	DC03	0003	DC03	Std	Std	DC03	DC03	Std	· Std	Std	Std	Std	Std	0003	0003	Std	Std	atch So	
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VTHODE	Bond.	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Wet	Dry	Dry	Dry	Dry	Wet	Wet	Dry	Dry	Dry	Dry	Wet	Wet	Dry	Dry	S - L	
3	Cat.	DC05	0005	DCO1	DC02	DC06	DC06	DC07	DC07	0010	DC10	DC10	0010	0010	0010	DC06	0007	DC06	0007	DC07	9000	0011	0C12		
ATUS -	Off  Test	1/84	1/84	3/84	ı					•	4/84	4/84	4/84	4/84	4/84					5/84	ı	5/84	5/84		
TEST ST	Total Hours	2134	2182	622	ł	5026	5170	3784	4192	ł	1336	1336	1240	1240	1168	2056	1840	2226	2344	2104	2344	232	232		
PRESENT	E Cell (mV)	553	626	594	•	441	587	516	596	·	585	547	522	559	625	640	585	603	616	535	633	43	54		
1,000 Hours	t cell (mV)	641	645	J	•	680	672	660	661	ı	591	595	555	605	612	673	672	699	663	649	672	·	<b>a</b>		
600 Hours	t cell (mV)	639	651	594	ı	683	119	662	661	•	608	596	555	613	616	672	683	675	665	654	677	ŀ	ı		
200 Hours	د دور ۱ (mV)	649	656	601	ŀ	686	675	667	662	•	612	602	553	623	622	677	693	619	670	655	680 .	53	129		
Initial	kesistance (mu)	5.5	6.7	9.5	•	6.2	7.7	7.2	8.9	ı	7.5	7.7	.0.9	7.6	8.3	7.6	6.8	7.7	7.9	9.6	7.0	7.2	22.2		
11-0	10	CAS-047	CAS-048	CAS-049	CAS-050	.CAS-051	CAS-052	CAS-053	CAS-054	CAS-055	CAS-056	CAS-057	CAS-058	CAS-059	CAS-060	CAS-100	CAS-101	CAS-102	CAS-103	CAS-104	CAS-105	CAS-106	CAS-108		

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TEST 2x2-007 REV. 0 THRU REV. 2 ALTERNATE BACKING PAPER COMPARISON AND PROCESS EVALUATION

(Status as of 5/31/84)

Stackpole PC206, 39% IFE, Std. Pročess Stackpole PC206, 39% TFE, Paste Proc. Silk Screen Production Paste Process Silk Screen Production Paste Process Kureha E715, 42% TFE, Paste Process Kureha E715, 33% TFE, Paste Process Kureha E715, 42% TFE, Std. Process Kureha E715, 33% TFE, Std. Process Cat. Bond. Sint. Mat-1 Assembly and Test Comments ۶ Ñ ۶ S S Arg Arg Arg Arg Air Air Air Air Air Air ANODE Dry Wet Dry Pry Wet Wet Wet Dry Wet Wet Std Bond. Sint. Arg Arg Arg Air Arg Air Air Air Air Air [est Conditions: 1 Atm., 80% H2 Util., 20% Air Utilization, - 200 mA/cm² @ 190°C CATHODE 5 Dry Dry Wet Wet Wet Dry Wet Wet Wet Cat. Std Std Std Std Std Std H 보보 r TEST STATUS Total Off Hours Test 5/84 5/84 5/84 1/84 3/84 2/84 2/84 5/84 1/84 3/84 2018 1600 1123 1134 1132 1600 1600 1600 1990 1081 PRESENT E cell (≣V) 614 613 535 555 575 633 645 653 572 584 ,000 Hours E cell 568 645 654 622 639 585 577 587 636 587 200 Hours 600 Hours 1 E cell E cell (mV) (mV) 575 598 644 665 578 602 633 662 677 577 675 647 618 665 669 580 583 592 604 631 Resistance (መበ) [nitia] 6.8 5.9 7.5 7.7 6.7 18.0 12.2 14.7 11.1 9.۱ CAS-043 CAS-044 CAS-045 CAS-074 CAS-076 CAS-041 CAS-073 CAS-075 CAS-040 CAS-042 Cell 10

WAESD-TR-85-0030

S - Large Batch Schold Mixed

HT - 900°C Heat Treated

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had high decay rates. The performance of this cell is shown in Fig. 3-1. Vendor cathodes DC05 and 10 provided acceptable initial performance and DC04 also appears to be good but these electrodes need further endurance testing. Alternate catalyst materials DC-11 and DC-12 initially appear to be unacceptable. A number of these tests are being continued for up to 5000 hours and a detailed analysis of their performance will be delayed until the tests are completed. However, based on the available data, it is apparent that reliable predictions regarding the performance degradation of certain catalysts cannot be made based on test duration of less than 3000 hours.

### Test Plan TS 2X2-007, Rev. 0 and 1

The objective of this plan was to assess the performance of alternate backing paper, alternate wet proofing methods and effect of changing Teflon levels in the backing paper on 2x2 subscale cells. The performance histories for various cells in this test plan are shown in Table 3-9.

One series of cells, CAS-040 through -043, was fabricated from electrodes supported by a possible alternate support layer, namely, Kureha E715 carbon paper. The backing or support layer was wet proofed to two levels of Teflon using two processes, the standard dipping process and a paste spreading process. The terminal cell voltage of these cells was 600 mV or lower and the cell resistance was high, ll m $\Omega$  or greater (since the backing paper resistance was 50 percent higher than Stackpole paper). The poor cell performance can tentatively be attributed to the higher resistance of the cells, compared to baseline cells. Additionally, the rapid degradation of the cells, approximately 12 mV in 800 hours, indicates possible flooding of the backing layer at both Teflon levels. This data indicates that the E715 paper is not an acceptable alternative or substitute for the material presently being used for electrode layer backing.

The use of the alternate "paste" wetproofing process, or "silkscreen" process, was found to be comparable to the double dip process, and average cell voltages from the two processes were similar. This indicates that the silkscreen wetproofing process is an acceptable alternate process for further study.



Fuel Cell Performance.

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### Test Plan 2X2-008, Rev. 0

A total of ten cells were built under this test plan to evaluate the effects of (1) dry and wet bonding, (2) alternate sintering environment, viz., steam, air and argon. This plan also was aimed at providing an acceptance screening of Schold Mix MAT-1 compared to Ross Mix MAT-1. The performance of the various cells is shown in Table 3-10.

The results from wet/dry lamination comparisons conclusively show that dry bonding produces an acceptable result, which was also shown earlier in TS 2x2-002. Sintering of electrodes in argon is also an acceptable process (also confirmed by ERC test plan TSE 2x2-006 results). There was no significant difference in subscale cell voltages whether Ross Mixed MAT-1 or Schold Mixed MAT-1 is used, and this clearly shows that Schold Mixed MAT-1 matrix is acceptable.

### Test Plan 2X2-010, Rev. 0

The main aim of this test plan was to verify the quality of electrodes for a 10 kW stack. Seven cells were assembled and tested to assess the variation in the performance of electrodes within a batch and batch-to-batch. During the initial phase of this plan, the baseline processing procedure was revised due to the change of electrode sintering gas to argon; hence, this test plan was cancelled.

### TS 2X2-011, Rev. 0, 1 and 2

The purpose of this test plan was to study the effect of alternate acid transport medium, especially zirconium pyrophosphate insulation material, Teflon content in MAT-1 matrix, and the thickness of SiC coating. A series of tests was designed and cells were tested as described in Table 3-11.

TEST 2x2-008 REV. 0 ELECTRODE BONDING AND SINTERING PROCESS DEVELOPMENT TESTS

(Status as of 5/31/84)

Same Electrode Batch as CAS 17 & 23 Same Electrode Batch as CAS 17 & 23 Bond. Sint.Mat-lAssembly and Test Comments Reproducibility of Standard electrode and matrix Wet Air Arg Air Net Wet Wet Arg Arg ANODE Dry Wet Wet Pry Pry Pry Pry Wet Wet Dry Bond. Sint. Cat. M--1H M-1H Std Std Std Std g Std Std Std S - Large Batch Schold Mixed R - Small Batch Ross Mixed Wet Net Arg Air Wet Arg Wet Arg Arg Air CATHODE Wet Wet Wet Dry Dr.Y Pry Dry ΡY Dry Wet Test Conditions: 1 Atm. 80% H₂ Util., 20% Air Util., - 200 mA/cm² @ 190°C Cat. HT-W HT-W HT-W M-TH Std Std Std Std Std Std 11/83 4/84 12/83 12/83 11/83 2/84 12/83 2/84 12/83 1/84 Hours Test Total Off 200 Hours 600 Hours 1,000 Hours PRESENT TEST STATUS **635** 1635 1778 1586 1586 586 3194 2662 4266 2900 E cell () M 616 602 643 614 635 646 642 633 581 630 HT-W Heat Treated Westinghouse Fabricated E cell ) M 639 611 625 626 643 657 641 653 665 663 E cell (Mm) 655 636 623 660 646 664 623 641 664 627 Resistance E cell ( m ) 624 657 649 657 634 626 623 640 661 664 (шу) Initial 7.0 6.9 6.6 5.8 6.9 6.9 6.7 6.4 6.3 ... DAS-002 DAS-006 **DAS-009 DAS-003 DAS-004 DAS-005** DAS-007 **DAS-008** DAS-010 DAS-001 Cell . 1

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WAESD-TR-85-0030

Low       Ander       Ander       Ander         10       (iff.1)       (iff.1) </th <th></th> <th></th> <th>5 00 0</th> <th>200 1.1411</th> <th>1 5 te 1 te 1 te 1</th> <th>v- 000</th> <th></th> <th>7 9001</th> <th></th> <th></th> <th></th> <th></th> <th>•</th> <th></th> <th></th>			5 00 0	200 1.1411	1 5 te 1 te 1 te 1	v- 000		7 9001					•		
Initial conditions         Condition         Condition         Anole         Anole           10         (ad)         (ad) <td< th=""><th>ובאר רסטת</th><th>161005. 1 AC</th><th></th><th>ULII., 20%</th><th></th><th></th><th></th><th>ואת ר</th><th></th><th></th><th></th><th></th><th></th><th>i</th><th>J</th></td<>	ובאר רסטת	161005. 1 AC		ULII., 20%				ואת ר						i	J
	Cell	Initial Resistance	200 Hour: E cell	s 600 Hours E cell	1,000 Hour: F cell	s <u>PRESEN</u> F' cell	IT TEST	STATUS	- Dff	CAI	HODE	1	ANO	ы	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	01	( WU )	( mV )	( mV )	( m)	(mV)	• 표	ours	Test: C	at.B	ond. S	int. Ca	t. Bor	nd. Si	nt.Mat-lAssembly and Test Comments
D5-011       8.3       643       640       1900       Z/84       HT       Dry       Arg       Std       Met       Air       No.       Mominal SIC         D5-013       -       -       -       -       -       -       -       HT       Dry       Arg       Std       Wet       Air       No.       Mominal SIC         D5-013       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -															
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	DAS-011	8.3	636	644	645 '	640	0061	2/84	HT D	hry A	ي ق	td We	t Air	יא י	Std 3% Teflon MAI-1
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	DAS-012	7.5	658	663	653	642	1994	2/84	HT D	ry A	rg S	td. We	t Air	r No	Nominal SiC
MS-014       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       -       - <td>DAS-013</td> <td>•</td> <td>•</td> <td>•</td> <td>ľ</td> <td>,</td> <td></td> <td></td> <td>HT D</td> <td>hry A</td> <td>rg S</td> <td>td Dr</td> <td>y Arg</td> <td>oN E</td> <td>Kureha E715 - Cancelled</td>	DAS-013	•	•	•	ľ	,			HT D	hry A	rg S	td Dr	y Arg	oN E	Kureha E715 - Cancelled
$ \begin{array}{rcccccccccccccccccccccccccccccccccccc$	DAS-014	,	• <b>1</b>	•	ı	ı	ı	•	HT D	Iry A	rg S	td Dr	y Arg	oN 6	Stackpole PC 206 - Cancelled
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	<b>DAS-015</b>	6.1	666	671	669	628 1	0554		HT	Jry A	rrg S	td Dr	y Arg	g S	Std 3% Teflon MAT-l.
AS-019 8.1 639 648 646 653 2116 3/84 HT Dry Arg 5td Dry Arg 5 5td 3X Tefton MT-1 AS-0201 7.9 666 649 2140 3/84 HT Dry Arg 5td Dry Arg 5 5td 3X Tefton MT-1 AS-021 7.1 665 661 660 624 4168 HT Dry Arg 5td Dry Arg 5 2r P ₂ O ₇ Coated Cathode AS-021 1.4 613 614 617 619 1408 5/84 HT Dry Arg 5td Dry Arg 10 Nominal 51C AS-023 11.4 613 614 617 619 1408 5/84 HT Dry Arg 5td Dry Arg 10 Zr P ₂ O ₇ Coated Cathode AS-024 11.6 623 603 1408 5/84 HT Dry Arg 5td Dry Arg 10 Zr P ₂ O ₇ Coated Cathode AS-024 8.2 628 632 140 5/84 HT Dry Arg 5td Dry Arg 10 Zr P ₂ O ₇ Coated Cathode AS-024 8.2 628 630 522 HT Dry Arg 5td Dry Arg 5 4 1/2X Tefton Mat-1, Thin 51C AS-024 8.2 628 232 HT Dry Arg 5td Dry Arg 5 4 1/2X Tefton Mat-1, Thin 51C AS-047 8.2 628 624 232 HT Dry Arg 5td Dry Arg 5 4 1/2X Tefton Mat-1, Thin 51C AS-047 8.2 628 53 575 560 1408 5/84 HT Dry Arg 5td Dry Arg 5 4 1/2X Tefton Mat-1, Thin 51C AS-047 8.2 628 53 575 560 1408 5/84 HT Dry Arg 5td Dry Arg 5 4 1/2X Tefton Mat-1, Thin 51C AS-047 8.2 628 53 575 560 1408 5/84 HT Dry Arg 5td Dry Arg 5 4 1/2X Tefton Mat-1, Nom 51C AS-047 8.2 628 53 575 560 1408 5/84 HT Dry Arg 5td Dry Arg 5 5 4 1/2X Tefton Mat-1, Nom 51C AS-048 8.8 624 57604 Mat-1 Prested 5-1 Large Batch Schold Mixed	0AS-016	6.1	669	675	673	625 ]	0554	-	HT D	)ry A	rg S	td Dr	y Arg	о <mark>У</mark> Б	Thin SiC
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	DAS-019	8.1	639	648	646	625	2116	3/84	HT	)ry A	Irg S	td Dr	y Arg	9 N	Std 3% Teflon MAT-1
$\begin{array}{llllllllllllllllllllllllllllllllllll$	DAS-020	7.6	.999	670	666	649	2140	3/84	HŢ	lry A	rg S	td Dr	y Arg	ov - E	Thick SiC
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	DAS-021	7.9	655	658	660	623	4168		HT D	hry A	lrg S	td Dr	y Arç	g S	Std 3% Teflon MAT-1
DAS-023 11.4 613 614 617 619 1408 5/84 HT Dry Arg Std Dry Arg S $Zr P_2O$ , Coated Cathode DAS-024 11.6 623 603 575 560 1408 5/84 HT Dry Arg Std Dry Arg No $Zr P_2O$ , Coated Cathode DAS-047 8.2 628 603 575 560 1408 5/84 HT Dry Arg Std Dry Arg S 4 1/2% Terflon Mat-1, Thin SiC DAS-048 8.8 624 624 232 HT Dry Arg Std Dry Arg S 4 1/2% Terflon Mat-1, Nom SiC HT - 900°C Heat Treated $S - Large Batch Schold Mixed$	DAS-022	1.1	667	661	667	624	4168	-	HT D	lry A	rg S	td Dr	y Arg	oN 6	Nominal SiC
DAS-024 11.6 623 603 575 560 1408 5/84 HT Dry Arg Std Dry Arg No Zr P ₂ 07 Coated Cathode DAS-047 8.2 628 232 HT Dry Arg Std Dry Arg S 4 1/2% Teflon Mat-1, Thin SiC DAS-048 8.8 624 232 HT Dry Arg Std Dry Arg S 4 1/2% Teflon Mat-1, Nom SiC HT - 900°C Heat Treated S - Large Batch Schold Mixed	DAS-023	11.4	613	614	617	619	1408	5/84	HT	hy A	lrg S	td Dr	y Arg	9 S	Zr P ₂ 0 ₇ Coated Cathode
AS-047       8.2       628       232       HT       Dry       Arg       5 4 1/2% Teflon Mat-l, Thin SiC         DAS-048       8.8       624       232       HT       Dry       Arg       5 4 1/2% Teflon Mat-l, Thin SiC         DAS-048       8.8       624       232       HT       Dry       Arg       5 4 1/2% Teflon Mat-l, Nom SiC         HT - 900°C Heat Treated       5 - Large Batch Schold Mixed	0AS-024	11.6	623	603	575.	560	1408	5/84	HT D	hry A	rg S	td Dr	y Arg	oN 6	Zr $P_2O_7$ Coated Cathode
AS-047       8.2       628       232       HT       Dry       Arg       S       4       1/2%       Teflon       Mat-1, Thin SiC         DAS-048       8.8       624       232       HT       Dry       Arg       S       4       1/2%       Teflon       Mat-1, Nom SiC         DAS-048       8.8       624       232       HT       Dry       Arg       S       4       1/2%       Teflon       Mat-1, Nom SiC         DAS-048       8.8       624       232       HT       Dry       Arg       S       4       1/2%       Teflon       Mat-1, Nom SiC         HT       900°C Heat Treated       S - Large       Batch Schold Mixed       S       2       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2%       1/2% </td <td></td> <td></td> <td>•</td> <td>:</td> <td></td> <td></td> <td>·<b>.</b></td> <td>,</td> <td></td> <td>.*</td> <td></td> <td></td> <td></td> <td></td> <td></td>			•	:			· <b>.</b>	,		.*					
DAS-048 8.8 624 624 232 HT Dry Arg 5td Dry Arg 5 4 1/2% Tefion Mat-1, Nom SiC HT = 900°C Heat Treated 5 - Large Batch Schold Mixed	04S-047	8.2	628			628	232	-	HT	lry A	rg	td Dr	y Arç	S S	4 1/2% Teflon Mat-1, Thin SiC
HT - 900°C Heat Treated S- Large Batch Schold Mixed	DAS-048	8.8	624			624	232	•	HT C	)ry β	lrg S	td Dr	y Arg	С	4 1/2% Teflon Mat-1, Nom SiC
			HT - 900	°C Heat Trea	ited		S	- Lar	ge Batch	i Schol	d Mixed				
							•	-							
												,	•	•	-
				- - -		•	-	, ,	· .			•			•

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One of the important conclusions from this test plan is that the insulation coating from zirconium pyrophosphate is not acceptable since the cells had lower cell voltages and also 60 percent (approximately) higher resistance than silicon carbide. It can be seen once again (previously shown in TS 2X2-002) that thin SiC coating is highly desirable (also confirmed by ERC Test Plan TSE 2X2-003 results) and the measured resistance of SiC coating is about .02-.03 m\Omega/micron (0.5-0.7 m\Omega/mil).

The effect of MAT-1 carbon layer on cell internal resistance was assessed using subscale cells DAS-Oll and DAS-Ol2. MAT-1 was assembled in DAS-Ol1, but omitted in DAS-Ol2. After continuous testing at 200 mA/cm² (186 A/ft²) for  $\sim$ 200 hours, the polarization data and internal resistances were measured at 190°C (374°F) and ambient pressure. At the current densities of this study, no major differences were observed in the internal resistances of the two cells. Since the presence of MAT-1 increases the interelectrode distance, one would expect a higher resistance in DAS-Ol1 than that measured in DAS-Ol2. The additional resistance was not detected, presumably due to the electronic conductivity of the MAT-1.

Figure 3-2 shows the IR-free cell voltage vs. current density relationships for the two cells. At very low current densities, both cells exhibited similar voltages. With increased operating current density, however, the voltage difference between these cells became significant. If one plots the voltage difference as a function of current density, a linear curve is obtained, revealing that the MAT-1 in DAS-O11 contributes approximately an additional 3.1 m to the overall cell internal resistance. At 190°C ( $374^{\circ}F$ ), the measured electrical conductivity of 100 w/o  $H_3PO_4$  is  $\sim 0.5$ /ohm-cm. Assuming that the tortuosity factor for a loose random pore structure is  $\sqrt{3}$ , the calculated electrical resistance through a typical MAT-1 of thickness 0.229 mm ( $\sim 0.009$  in.) is  $\sim 3.09$  m $\Omega$ . This estimated value is in good agreement with the additional internal resistance of 3.1 m $\Omega$  obtained experimentally. In conclusion, the inclusion of MAT-1 increases the cell internal resistance by  $\sim 3$  m $\Omega$  and the inferred resistance is about .012-.02 m $\Omega$ /micron (0.3-0.5 m $\Omega$ /mil). As a result, the terminal voltage of





a subscale cell without MAT-1 at 200 mA/cm² (186 A/ft²) should be about 16 mV higher than the cell with MAT-1. However, additional data are required to completely define the resistance contribution of MAT-1.

### TS 2X2-013, Rev. 0, 1, 2, 3 and 4

The major objective of this test plan is to verify (using subscale cells) the quality of the electrodes and cell components that are incorporated in the baseline Stacks W-009-10, W-009-11 and W-009-12. A series of 27 cells were built and testing is in progress as shown in Table 3-12.

From the tables, it can be observed that the QAS-006-009 (S1) cell voltages ranged from 635 - 652 mV with an average of 645 mV (based on 200 hour data) at the specified test conditions. The same electrodes in cells QPS-009 through -012 performed at an average voltage of 640 mV (range 627-647 mV) with SRG at 485 kPa (4.8 atm.), 80 percent H₂ (SRG), 20 percent air utilization,  $325 \text{ mA/cm}^2$  ( $300 \text{ A/ft}^2$ ) at 190°C ( $374^\circ$ F). These are the electrodes that were used in assembling in Stack W-009-10.

The electrodes that were obtained from heat treatment run No. 4 were incorporated in Stacks W-009-11 and W-044-1 and these were tested in subscale cells QAS-016 through -019, -019-S1 and -019-S2 under atmospheric pressure and in QPS-017 through -020 under pressurized conditions. The atmospheric test cells showed an average of 648 mV with a range from 624 mV to 680 mV (based on 200-hour data). The same electrodes in pressurized cell conditions with SRG exhibited 659 mV (range 647-669 for 3 cells) indicating that the effect of SRG was about -25 mV as expected with the range from -15 mV to -30 mV. These electrodes appear to be acceptable for stacks (with a minimum performance > 630 mV) and will presumably aid in understanding the data from the stack development program.

### TS 2X2-014, Rev. 0 and 1

This test plan was designed to evaluate alternate catalysts in terms of endurance under pressurized conditions. During this reporting period, tests

TEST x2x-013 REV. 0 THRU REV. 5 ELECTRODE QUALITY VERIFICATION TEST

(Status as of 5/31/84)

Test Conditions: 1 Atm., 80% H₂ Util., 20% Air Utilization, - 200 mA/cm² @ 190°C

•	est Comments	ardware failure)		·	
	Assembly and T	High nominal Low nominal (h	High nominal Low nominal	Low nominal High nominal	
	Mat-1			o o o N N N	
	Sint.	Arg Arg Arg Arg	Arg Arg Arg Arg Arg	Arg Arg Arg	
ANODE	Bond.	00000 0100 0100	777777 7777777777777777777777777777777	222	xed.
	Cat.	std std std	std std std std std std std std std std	Std Std Std	M plot
	Sint.	Arg Arg Arg Arg	Arg Arg Arg Arg Arg	Arg Arg Arg	tch Sci
VTHODE	Bond.	77 77 77 77 77 77 77	777777 7777777777777777777777777777777	0 7 7 7 7 7 7 7	irge Ba
5	Cat.	HT2 HT2 HT2 HT2 HT2	НТ4 НТ4 НТ4 НТ4 НТ4 НТ4 НТ4	нта нта нта	S - La
ATUS	0ff Test	2/84 2/84 2/84 1/84 3/84	5/84 5/84 5/84 4/84 5/84		
TEST ST	Total Hours	1034 1056 1005 286 1000	1312 1312 150 160 240 649 313	304 304 304	
PRESENT	E cell (mV)	645 638 600 610 632	648 557 598 597 645 645	617 641 653	
1,000 Hours	E cell (mV)	643 642 600 - 632	651 633 574 -		reated er cell
600 Hours	E cell (mV)	644 651 638 638 638	662 643 625 - 623 623		00°C Heat T itical sist
200 Hours	E cell (mV)	648 652 646 609 635	667 680 641 598 607 624 646	650 649	HT - 9( ebuilt ider
Initial	lesistance (ເໝີ)	8.3 8.4 9.0 11.8 8.8		10.5 8.8 8.2	or -52) Re
•	Cell F	QAS-006 QAS-007 QAS-008 QAS-009 QAS-009 QAS-009-51	04S-016 04S-017 04S-018 04S-019 04S-019-51 04S-019-52 04S-020	QAS-025 QAS-026 QAS-027	15-)

WAESD-TR-85-0030

TEST 2x2-013, REV. 0 THRU REV. 5, ELECTRODE QUALITY VERIFICATION TEST (CONT'D)

(Status as of 5/31/84)

Test Conditions: 4.8 Atm., 80%  $H_2$  (SRG), 20% Air - 325 mA/cm² </code> 190°C

C - Z

50%) H ₂ utilization	<u>د</u> ۲	° *		.ed.	viM blo	ch Schi	ge Bat	S - Lar		×	urnace Run	<b>Freated Pt, F</b>	DO°C Heat ⁻	HTX - 9(	
	ν	Arg	Dry	Std	Arg	Dry	HT4		256	561(H2)*			522	10.6	QPS-020
tow nominal	S	Arg	Dry	Std	Arg	Dry	HT4		256	679(H2)*	•		667	5.9	QPS-019
	S	Arg	Dry	Std	Arg	Dry	HT4		256	673(H2)*.	•		638	6.9	QPS-018
High nominal	S	Arg	Dry	Std	Arg	Dry	HT4	·	256	679(H2)*	•		629	. <b>[</b> °9	QPS-017
4	S	Arg	Οry	Std	Arg	Dry	HT2	5/84	794	602(H2)	.1	621(H2)	623(H2)	10.0	QPS-016
3 (location) dependence te	S	Arg	Dry	Std	Arg	Dry	HT2	5/84	794	589(H2)	•	629(H2)	660(H2)	8.0	QPS-015
<ul> <li>2 Pressurized loop-station</li> </ul>	S	Arg	Dry	Std	Arg	Dry	HT2 ·	5/84	794	635(H2)	•	639(H2)	640(H2)	8.0	QPS-014
[	S	Arg	Dry	Std	Arg	Dry	HT2	5/84	794	606(H2)	•	609(H2)	608(H2)	·9.7	QPS-013
	No	Arg	Dry	Std	Arg	Dry	HT2	3/84	857	0	ı	0	627	4.5	. QPS-012
Low nominal	Ŷ	Arg	Dry	Std	Arg	Dry V	HT2	3/84	857	0	•	0	647	4.6	QPS-011
	No	Arg	Dry	Std	Arg	Dry	HT2	3/84	857	450(H2)	ı	671(H2)	642	4.9	QPS-010
High nominal	N N	Arg	Dr.y	Std	Arg	Dry	HT2	3/84	857	0		670(H2)	645	5.5	<u>qPS-009</u>
t-lAssembly and Test Comments	Mat	. Sint	Bond	Cat.	Sint.	Bond.	Cat.	Test	Hours	r cell (mV)	E Cell	(mV)	100 L Cel (mV)	Kesistar (m3)	10
		ANODE			ATHODE	3	STATUS	T TEST	PRESEN	urs	1,000 Ho	urs 600 Hours	1 200 Hoi	Initia	

st

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were initiated with an alternate catalyst, DC-06, in two cells, along with two Westinghouse baseline cells (CPS-033 through -036) as shown in Table 3-13. Further tests are planned in this area with various types of catalysts.

The preliminary tests in this test plan were performed with gold-plated hardware (plated bipolar plates). The initial results showed a poor performance with an average of 580 mV under pressurized conditions which indicates that gold plated hardware may not be acceptable.

### Test Plan TS 2X2-016

The objective of this test plan was to evaluate the effects of various nonelectrode related poisons such as by-product materials dissolution at various temperatures. Two sealing materials, viz., Viton and Teflon were assessed at 190° and 200°C (374° and 392°F) temperatures with and without MAT-1. An attempt was also made to evaluate the effect of alternate end plates. Testing of the cells built in this plan is in progress and preliminary results are shown in Table 3-14.

Although the initial performance data on cells CAS-116 through -121 exhibit a scatter, it is safe to conclude that Teflon is an acceptable sealant. Further re-building and re-testing is required to obtain any endurance data and/or other conclusions.

EFFECT OF VARIATIONS IN KEY OPERATIONAL PARAMETERS ON UNPRESSURIZED CELL PERFORMANCE

To normalize the experimental data for unpressurized subscale tests, the effects of minor variations in test temperature and current density on cell performance were investigated using cells DAS-019 and -021. Both cells were assembled using dry bonded argon sintered cathodes made using Lot No. 3 heat treated catalyst, dry bonded argon sintered anodes made using as-received catalyst and carbon layer (MAT-1). Tests were performed at 80 percent hydrogen utilization, 25 percent air utilization at 200 mA/cm² (186 A/ft²) (5.13 amps).

TEST 2x2-014 REV. 0 AND REV. 1 PRESSURIZED ELECTRODE ENDURANCE VERIFICATION

(Status as of 5/31/84)

Mat-1 Assembly and Test Comments Preliminary tests with Gold Plated Hardware ကကုက Bond. Sint. Arg Arg Arg ANODE Test Conditions: 4.8 Atm., Util. 80%  $H_2$  (SRG), 20% Air Util. -  $32^{\circ}$  mA/cm² @ 190°C S -- Large Batch Schold Mixed. Cat. Std Std Std Bond. Sint. Arg Arg Arg CATHODE 2222 . : 200 Hours 600 Hours 1,000 Hours PRESENT TEST STATUS CA E cell E cell E cell E cell 10tal 0ff (mV) (mV) Hours Test Cat. DC06 Std Std . 4/84 4/84 4/84 4/84 · ... -14 5 0000 516(H₂) 612(H₂) 570(H₂) 573(H₂) 1.1 HT -- 900°C Heat Treated ĻĻ ţ 1 . Initial 2 Resistance (m2) . . . . CPS-033 CPS-034 CPS-035 CPS-035 CPS-036 Cell 10

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<u>3-3]</u>

## TEST 2x2-016 NON-ELECTRODE RELATED CATALYST POISONS (Status as of 5/31/84)

•
0°190°C
mA/cm ²
200
Util
Air
20%
Н2,
80%
Ut11.
Atm.,
-
Conditions:
Test

:	ts .						ites	ates	
	Commer						end pla	end pl	
	d Test				•		Alt.	Alt.	
	olv and	,	seal	seal	1 seal	i seal	seal,	ı seal,	
	Assent		Viton	Viton	Teflor	Teflor	Viton	Teflor	
	Mat-1		ŝ	No	No	د	S	S	
	Sint.		Arg	Arg	Arg	Arg	Arg	Arg	
ANODE	Bond.		Dry	Dry	Dry	Dry	Dry	Dry	
	Cat.		Std	Std	Std	Std	Std	Std	
	Sint.		Arg	Arg	Arg	Arg	Arg	Arg	
ATHODE	Bond.		Dry	Dry	Dry	Dry	Dry	Dry	
U	Cat.		TH .	HT	H	ΗT	ΗT	HT	
ATUS	<u>Off</u> Test								
TEST SI	Total Hours		592	736	592	328	736	736	
PRESENT	E cell (mV)		667	642	676.	635	669	647	
1,000 Hours	E cell (mV)								
600 Hours	E cell (mV)			643	·		668	649	
200 Hours	E cell (mV)		664	644	677	589	666	657	
Initial	Resistance (m2)		6.7	10.5	7.0	11.6	6.8	7.6	
	Cell IO		CAS-116	CAS-117	CAS-118_	CAS-119	CAS-120	CAS-121	

HT - 900°C Heat Treated

S - Large Batch Schold Mixed.

Figures 3-3 and 3-4 show the effect of varying cell current between 4.5 and 5.5 amperes on cell voltage at 185° and 190°C (365° and 374°F). Both cells varied approximately 2.9 mV per 0.1 amps at 190°C (374°F) and between 2.7 and 3.0 mV per 0.1 amp at 185°C (365°F).

Cell voltage as a function of temperature is plotted in Figure 3-5. Over the range of temperatures examined, the cells vary 1.2 mV/°C (.7 mV/°F) (DAS-019) and 1.1 mV/°C (.6 mV/°F) (DAS-021).

Similar data was previously determined for wet bonded, non-heat treated catalyst electrodes. The agreement between the sets of data indicates very little effect of process changes. The new data was utilized in the data acquisition programming to perform the normalization of actual test values.

3.1.2.2 Performance Improvement From Pressurization

The effect of operating pressure on performance of subscale fuel cells continued to be studied. This work was initiated in the First Logical Unit. Test cells were wet assembled using 93 w/o  $H_3PO_4$  and measurements made at 190°C (374°F) over the pressure range of 101-687 kPa (1 - 6.8 atms). The cells were operated for over 300 hours at pressure to stabilize the electrode performance and improve the reproducibility of the results prior to taking the measurements for pressurization effects. Polarization data was obtained at each pressure and the cells were equilibrated at each current level for approximately ten minutes before data was taken. Cell internal resistance was measured using an <u>in-situ</u> technique with an AC milliohmeter (Hewlett-Packard Model 4328A).

Typical cathode potential-current density relationships at pressures of 101, 240, and 480 kPa (1.0, 2.4, and 4.8 atms) are given in Figure 3-6. Linear Tafel regions were observed at current densities below 150 mA/cm² (140  $A/ft^2$ ), the measured slopes ranging from 106 to 109 mV/decade and independent of operating pressure. Increased pressure did not affect the mechanism of the oxygen reduction reaction. Based on the assumption that pressure effects on

WAESD-TR-85-0030 مارومة الإخارة وأثرياته 4 -S. <u>4</u>. 1 . . : ; , -3-13  $\frac{1}{2}$ ÷., : 1.1  $\hat{\phi}$ · · . . · :• . . . : 3 · . . . Charle the Carl 111-1-1 • ; Ó ഗ 2.7 mV/.1 Rap 3.8 av/1 85° ີທ໌ ທ at a Function of Current Amps **S** ហ I Voltage as AIR: 25% @ 200 mA/cm2 H2: 80% 6 208 mA/cm2 PRESSURE: 1 Ata **BIG-SH** Cell **BS-82** 015 3-3. Figure 620 650 630 680 660 670 640 0 • . G . : ; • . Vm ۰:, 13 17 llaci i÷ .

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900A19 ဖ ·2.9 mV/.1 Amps 2.9 mV/.1 Rmp: CURRENT vs VOLTS 5.5 Amps · TEMP = 198 C ហ 1 AIR: 25% 8 200 mA/cm2 H2: 88% & 200 mA/cm2 PRESSURE: 1 Atm. 4. N DAS-819 DAS-821 618 680 640 670 660 650 630 620 ٧w [[ab] · **—** 

Figure 3-4. Cell Voltage as a Function of Current at 190°C



Figure 3-5. Cell Voltage as a Function of Temperature

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[1]

[2]

the activation and concentration overpotentials of the hydrogen electrode are small and can be neglected, the mathematical equation for predicting voltage increase resulting from pressurization was re-examined. At low current densities, where the concentration overpotentials of air electrodes are small, the expected voltage gain is given by:

$$\Delta Ep = [2.3 \text{ RT} (2 + 4/\alpha)/4F] \cdot \log P$$

where P is the operating pressure in atmospheres,  $\alpha_{c}$  denotes the apparent transfer coefficient for oxygen reduction, and F is Faraday's constant. In practice,  $\alpha_{c}$  is obtained from the measured Tafel slope b using the expression:

 $\alpha_c = 2.3 \text{ RT/(b_cF)}$ 

Using an average Tafel slope of 108 mV/decade, the cell voltage gains at various pressures were calculated using equation [1] and the solid line in Figure 3-7 shows this gain. The experimental voltage gains at current densities of 100, 200 and  $325 \text{ mA/cm}^2$  are presented in Figure 3-8. At both 100 and 200 mA/cm² (93 and 186 A/ft²), the observed voltage gains are in excellent agreement with those predicted by equation [1]. Thus indicating that, in the operating current density range where cell voltage losses are mainly due to the activation overpotentials at the cathode, Eq. [1] predicts the gains from pressurization.

As shown in Figure 3-8, the voltage gains at  $325 \text{ mA/cm}^2$  ( $300 \text{ A/ft}^2$ ) are slightly higher than predicted by Eq. [1]. Pressurization results in increased oxygen solubility which, in turn, reduces the concentration overpotential. Thus, the additional voltage gains at higher current densities are presumably attributed to the reduction of concentration overpotentials at the cathode.

The influence of operating pressure on cell internal resistance is given in Figure 3-9. Increasing pressure from 101 kPa to 240 kPa (1 atm to 2.4 atms)



Figure 3-6. Tafel Plots for Oxygen Reduction on a Fuel Cell Cathode at 1.0, 2.4, and 4.8 atm






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Figure 3-8. Effect of Pressure on Tafel Slopes



Figure 3-9. Voltage Gain at Various Pressures

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results in a relatively significant decrease in cell resistance. At  $100 \text{ mA/cm}^2$  (93 A/ft²), for example, the resistance drops from about 10.5 mΩ to 9.0 mΩ, while above 303 kPa (3 atms), the measured resistance is relatively independent of pressure. This behavior is related to the acid concentration decreasing with increased pressure, essentially due to increased partial pressure of water vapor at higher pressures. As has been well documented, the electrical conductivity of  $H_3PO_4$  increases with decreasing concentration in the range of 86-102 w/o. Under pressurization, the decreased cell internal resistance is primarily associated with decreased acid concentration.

At higher current densities from  $200 - 400 \text{ mA/cm}^2$  (186-372 A/ft²) and pressures above 240 kPa (2.4 atm), an empirical equation for total cell voltage gain from pressurization was derived by taking into consideration the combined effect of pressure on the Nernst potential, cell internal resistance, and both activation and concentration overpotentials at the cathode, namely:

 $\Delta E_{cell} = (2.3 \text{ RT/4F}) \cdot (2 + 4 \alpha_c) \log P + 1.22 \times 10^{-4} \text{ i}$   $+ 4.8 \times 10^{-3} P - 1.93 \times 10^{-5} \text{ iP} - 1.95 \times 10^{-2}$ [3]

where i is the current density in  $mA/cm^2$  and P is the operating pressure in atms.

For the apparent average transfer coefficient of 0.85, Table 3-15 compares the measured voltage gains with the cell voltages predicted by both Eqs. [1] and [3]. At 485 kPa (4.8 atm), the observed voltage gains are higher than predicted by Eq. [1] which accounts for voltage gains in reversible potential and activation overpotentials of the cathode. As shown in Table 3-16, the total cell voltage gains given by Eq. [3] are in excellent agreement with the measured values, the gains predicted by Eq. [3] being within  $\pm$  2 percent of the measured gains.

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Pressure ( (atm)	Current Density (mA/cm ² )	Measured Cell Voltage gain (mV)	ΔEr + IΔnc,act ^{1,-,} by Eq. [1] (mV)	Predicted Voltage Gain by Eq. [3] (mV)	Deviation ⁽³ (%)
4.8	100	111	105	111	0.0
4.8	200	115	105	, 114	-0.9
4.8	325	120	105	118	-1.7
2.4	200	67	58	67	0.0
3.7	200	98	87	96	-2.0
5.8	200	127	117	128	+0.8
6.8	200	137	128	139	+1.5

**TABLE 3-15** 

(2)  $\Delta E_r + 1\Delta \eta_{c,act}$  = the cell voltage gains arising from improvements in Nernst potential and activation overpotentials at the cathode.

(3) Deviation = (predicted voltage gain - measured voltage gain)/(measured voltage gain)

## 3.1.2.3 Correlation of Performance Decay with Cell Voltage

Electrochemical investigations were completed on the correlation between performance deterioration and cell voltage at ambient pressure, 190°C (374°F), 20 percent air utilization and 80 percent hydrogen utilization. After reproducibility testing at 200 mA/cm² (186 A/ft²) for approximately 1600 hours, three subscale cells SC-037, SC-038 and SC-039 were operated at 80, 100, and 200 mA/cm² (75, 93, and 186 A/ft²), respectively, so that their initial IR-free cell voltages were in the range of 700-800 mV, above 800 mV and below 700 mV. In the course of testing these cells, 100 w/o H₃PO₄ was added twice to the cell acid reservoirs. Electrode flooding was found in SC-039 after  $\sim$ 1850 hours of continuous operation. The IR-free cell voltages and internal resistances are depicted as a function of time (up to  $\sim$ 2400 hours) in Figures 3-10 and 3-11, respectively.

With an initial IR-free cell voltage above 800 mV, subscale cell SC-037 exhibited a performance deterioration rate as high as 6.0 mV/1000 hours (see Figure 3-10). At 80 mA/cm² (75 A/ft²), the IR-free cell voltage of SC-038 decreased with time at  $\sim$ 2.1 mV/1000 hour. Prior to electrode flooding, however, the performance deterioration rate of SC-039 operating at 200 mA/cm² (186 A/ft²) was only  $\sim$ 1.6 mV/1000 hour.

As shown in Figure 3-11, the in-test addition of phosphoric acid resulted in no beneficial effect on the internal resistances of SC-037 and SC-038. After refilling the acid reservoirs with  $H_3PO_4$ , the internal resistance of SC-039 decreased  $\sim 0.5 \ m\Omega$ , indicating that electrolyte dryout may have occurred in this cell. As seen in Figure 3-11, however, the measured internal resistance of SC-069 continued to increase with time.

3.1.2.4 Dependence of Cell Performance on Hydrogen Utilization

The effect of hydrogen utilization on cell performance was re-examined in subscale cells ERC-3036 and CPS-023. [Note that cell ERC-3036 was previously tested at Energy Research Corporation for about 700 hours and was transferred

1



IR-FREE CELL VOLTAGE, mV

TIME, HOURS





TIME, HOURS



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to WAESD for testing.] At a constant current density of 200 mA/cm² (186  $A/ft^2$ ) and air flow rate of 32.8 cm³/min (2 in³/min) (25 percent utilization), experiments were performed using hydrogen and SRG (simulated reformer gas containing 74 percent hydrogen, 25 percent  $CO_2$  and 1 percent CO). With hydrogen as fuel, the cell voltage remained constant with increasing utilization up to 90 percent, but it started to drop significantly for a utilization of 92 percent. In the case of SRG fuel, the cell voltage decreased linearly at approximately 1 mV per each 10 percent increase in hydrogen utilization, cell voltage decreased more rapidly with increasing utilization, presumably due to the CO blocking and/or poisoning effects on the platinum catalyst.

# 3.2 Nine Cell Stack Development (WBS 1103-02)

3.2.1 Design Developments

In accordance with the performance goals established for the prototype powerplant, the primary objectives of the nine cell stack developmental effort were to:

- establish startup and operating procedures for pressurized stack testing
- establish baseline performance, reproducibility and endurance at rated power operating conditions
- increase stack average cell performance to 680 mV at operating conditions of 480 kPa (70 psia), 190°C (374°F), 325 mA/cm² (300 A/ft²), 83 percent fuel (SRG) and 50 percent air utilizations with a maximum degradation rate of 2 mV/1000 hours of operation.

The nine cell stack design, developed during the First Logical Unit of Work, was refined and developed further. The refinements were primarily in the

areas of electrode edge seal improvements, acid transport improvements within the cells, and manifold sealing. Additionally, some minor changes were incorporated into the design to address performance problems primarily affecting end cells of numerous stacks.

3.2.1.1 Nine Cell Stack Elastomer Edge Seal Development

The initial nine cell stack design incorporated elastomer cell edge seals of a single thickness at each edge of the electrode pair. The seal thickness exceeded the thickness of the electrode pair to provide a compressed edge seal. The seals provided edge sealing only and did not prevent internal crossleaks within the seal boundary at the edge of the electrode pair. Edge crossleaks were to be prevented by application of a sealant material along the edges of each electrode. Problems with obtaining an effective edge sealant application caused this design to be abandoned after testing only one stack during the First Logical Unit of Work.

The elastomer cell edge seal design was revised to address the problem of internal crossleak at the edge of the electrode pair. The single thickness edge seal was replaced by a double layer seal. The thickness of each layer was sized to the thickness of the corresponding cell component (electrode or electrode and MAT-1). One seal was made wider than the other for the seal to overlap the edge of the electrode to block crossleak at the edge of the electrode. Figure 3-12C illustrates a typical section through a cell with the overlapped double Viton seal as compared to Figure 3-12A, which illustrates the standard Teflon and MAT-1 seal arrangement, and Figure 3-12B, which illustrates the single thickness Viton seal.

A 10 cm by 10 cm (4 in. by 4 in.) single cell was built to assess the sealability of the overlap design against crossleak before the nine cell stack design was revised. For this test, the proposed overlap condition was duplicated on all four sides of the electrode pair, which was assembled with acid-wetted MAT-1. Pressurized helium gas was then introduced to one side of



Figure 3-12. Nine Cell Stack Cell Edge Seal Configurations

the cell with the other side connected to a manometer. The helium source and manometer were then interchanged and the test repeated. For both tests, the cell and seal withstood a maximum 6.9 kPa (10 psi) differential with no observable leakage. Tests were performed at room temperature only.

Following the success of the single cell test, a five cell engineering stack Was built to develop the manufacturing techniques and to test a multi-cell stack which had acid applied with the standard syringe technique. This technique was considered to provide less uniform acid wetting than that used for the 10 cm by 10 cm (4 in. by 4 in.) test assembly in which the MAT-1 was floated on acid for uniform wetting. Helium leak testing was performed on the five cell stack using the same methods as applied to all previous nine-cell stacks. At helium pressures up to 356 mm (14 in.) of water, crossleaks of less than 20 cm³/min (72 ft³/h) were measured. This was considered quite satisfactory in comparison to crossleak results obtained with previous stacks built with the standard Teflon and MAT-1 edge seal design. The nine cell stack drawings were revised to reflect the double layer Viton seal. The first application of the design to an operating nine cell stack was W009-08.

A second major change to the basic cell design is illustrated in Figures 3-12, 3-13, and 3-14. The cell assembly (anode, MAT-1, cathode) between the bipolar plates was inverted and the cathode was extended below the acid channel in the upper plate. Thus, the acid supplied through the acid channel directly contacts the silicon carbide layer of the cathode. The silicon carbide layer serves as the primary acid transport member for the cell and addition of acid to an assembled cell is significantly improved by this change. As shown in Figure 3-12, all previous cell designs suffered from poor lateral acid transport because the acid had to first move laterally through a length of MAT-1 until contact with the silicon carbide layer was made. Independent tests to assess lateral acid transport rates associated with both MAT-1 and silicon carbide have shown that the MAT-1 has a slower rate. As shown in Figure 3-14, the feature of MAT-1 extending beneath the seal inboard of the acid channel has been retained in the revised design. The MAT-1 serves as the





main barrier to anode/cathode crossleak at the joint between anode and seal, and it also cushions and prevents damage to the silicon carbide layer of the cathode caused by seal edge shearing effects.

The inverted cell design has been used in three stacks built to date; Stacks W009-09, -10, and -11. However, only Stacks W009-10 and -11 employed a MAT-1 layer as shown in Figure 3-14. Stack W009-09 was assembled without MAT-1, but with a thicker silicon carbide layer, to test the feasibility of stack operation without MAT-1. Acid additions to these stacks after assembly have been successfully performed within much shorter time periods, whereas for prior stacks, acid additions were difficult to perform in a reasonable time period. The acid addition data for individual stacks is provided in Table 3-16.

## 3.2.1.2 Nine Cell Stack Manifold Seal

Limited process gas manifold seal tests were performed in the dry room on Stack W009-09 after initial performance tests. The purpose of these tests was to begin building seal leak test experience on a relatively inexpensive test bed and to identify seal configurations for further development on the 10 kW "Metal Stack" (Section 3.3.1). Seals of Viton elastomer (both 0-rings and flat gaskets) and expanded Teflon ("Gore-Tex" joint sealant) were tested. To test O-rings, a special gasket/frame (Figure 3-15) was made with standard O-ring grooves on both sides to seal between the stack to frame and frame to manifold. Prior to starting the tests, the frame to flat aluminum manifold joint was proven to be leak tight. The "Gore-Tex" and flat gaskets were applied directly to the manifold. Tests were performed at 150°C to insure that the standard gaskets (uncured Viton sheets) on the end not being tested would remain leak tight. The test results are summarized on Table 3-17. These tests show that flat, cured Viton 0.32 cm (.125 in.) thick by 0.64 cm (.25 in.) wide produces a good seal, two layers of 0.48 cm (.189 in.)"Gore-Tex" shows some promise, and 75 durometer O-rings with standard compression (25 percent) is not an acceptable seal. These test results were further verified by the "Metal Stack" tests (Section 3.3.1).

TABLE 3-16

STACK ASSEMBLY COMPARISON



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TABLE 3-16

STACK ASSEMBLY COMPARISON (CONT'D)

	UNITS	10-600M	W009-02	M009-03	W009-04	M009-05	90-600M	M009-07	M009-08	M009-09	N009-10	11-600M
CATHODES											,	
Catalyst Catalyst Heat Treatment Barkinn Paner	ິ	10% Pt o None Stacknol	n XC72 e PC206						4	10% Pt on 900 Starknole	xC72	
Wet Proofing Method Lamination Method/Pressure Sinter Atmosphere & Temp 5101 Trithore Dance	°C Con	FEP Doub Wet @ 30 Air @ 36	le Dip	30,23	16/86	06780	667 UE	12/80	06776	FEP Doubl Dry @ 30 Argon @ 3	e Dip	067.06
SiC Thickness Range	mils	5/6	76/06	cc /nc	10/07	nc /07	JC INC	10/07	a S <b>↑</b>	+c/1c	cc /oc 8	5/6 5/6
<u>MAT-1</u> Manufacturing Type	1	Ross Ble	inder	Expmt'l.	Schold Mix	ker			<b>∳</b>	, <b>•</b>	Schold –	<b>^</b>
letion Percent Final Thickness Range	a mils	3 9/12	1	11/01	11/01	11/01	11/01	01/6	<b>∳</b> =		11/01	01/6
ACID LOADING												
Anode Application	ເພິ່	0									<b>A</b>	
Cathode (SiC) Application Matrix (MAT-1) Application	<u> </u>	15 24.5	13 35	13 33	10 32	10 35	13 35	10 35	10 32	27 -	5000	1
Acid Groove Application Total Cell Application	<b>J</b>	2 41.5	20. 20.	2 48	0 42	1.5 46.5	2 50	1.5	0 42	0 27	45	
Acid Concentration Acid Supplier	wt. X	94 Fisher –	97	54	93 Baker	93	97.3	92.7	92.5	98.4	98	97.8
Dry Rm Humidity Range ⁽¹⁾	æ	5/75			<b>↑</b>	13/175	5/75	13/175	13/175	3.5/6.5	5.5/6.0	5.0/6.5
Testing	cm ³	Acid Tra	nsport Ine	ffective -				¢	0(3)	с.	27	30
STACK CONDITIONING												
Stack Temperature ⁽¹⁾ Conditioning Period Drv Rm Humidity Ranoe ⁽¹⁾	۴ hrs	180 19 5/75 —	150 18	1805 ~18	150	128 21 13/175	162 18 5/75	130 16 13/175	130 20 13/175	130 21 6.0/12	120 20 5.5/6.0	130 18 5.0/6.0
STACK COMPRESSION	l	•					•					
Avg. Compression Applied	ISd	20	60	55	50	45	55	55	50	30	40	=.
stack Temperature	۹F ۲	176	143	- 178	108	127	168	130	130	130	130	130
Compressed Stack Resist. Pre-Test Stack Resist.	e e	9.5	18.5 6.5		9.9 6.8	8.2 6.5	7.8 6.5	8.6 7.0	6.5 5.5	24.0 16.0	10.0 8.2	13.2 9.2
<ol> <li>S = Specification requir</li> <li>S = Specification requir</li> <li>S = Specification second (2)</li> <li>S = Stack W009-08 experised 5</li> </ol>	ement, no   from 15 0 cm ³ aci	t actual m to 30 psi, id after as	easurement. and 3rd fi semblv. Av	rom 30 to 5 cid added a	50 psi. and retaine	50 50 50 50 50 50 50 50 50 50 50 50 50 5	un ³ from O	net additi	5			

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# TABLE 3-17

NINE CELL STACK MANIFOLD SEAL TEST SUMMARY

			•	External Leakage
1)	GOR	E-TEX 0.32 cm (1/8 in.)		∿< 400 cm ³ /min
	0	Final Manifold to Stack Gap	$\sim 0.013$ cm (.005 in)	
	0	Average Seal Width	∿0.51 cm (.20 in)	
2)	0-R	INGS (75 Durometer, 0.53 cm [3/1	6 in])	∿60 cm ³ /min
	0	Standard Compression	$\sim 28$ percent	
3)	GOR	E-TEX (2 Layers of 0.48 cm (3/16	"))	∿9 cm ³ /min
	0	Final Manifold to Stack Gap	∿ 0.11 cm (.045 in)	
	0	Average Seal Width	∿0.64 cm (.25 in)	
4)	FLA (2 Thi	T CURED VITON Layers at 0.16 cm (1/16 in.) ck by 0.64 cm (1/4 in.), ∿50 Du	rometer	∿0 cm ³ /min
	0	Compressed	$\sim$ 40 percent	· · ·

NOTE: 1) Extruded edge seal gaskets cut flush with razor blade for all tests. 2) All data at 150°C (300°F).

## 3.2.1.3 Plate Cracking

Other refinements were made to the nine cell stack design to address performance problems primarily affecting end cells. Cracks were observed in both end plates and cooler plates on a majority of the stacks that had been disassembled following testing. In most cases the cracks were located at the process ends of the plates in the nonfunctional grooves that are there to reduce material thickness for heat treatment purposes. The design of these grooves was modified to reduce the depth and gussets added. Both of these changes are designed to improve the bending capability of the plates near the edges, which are subject to bending moments resulting from gasket loads. The revised design (Figure 3-16) was incorporated into the cooler dies and molded plates will be tested in Stack W009-12 (Third Logical Unit of Work).

In parallel with the above plate design modification, a thicker insulator plate was incorporated in the stack design between each end of the stack and the stack heaters. The thickness was increased as much as practical within the height limits, which are controlled by existing manifolds. The insulation plates were made 3/8 in. thicker at each end. The purpose of this change was to provide more uniform load distribution to the repeating stack components by providing a stiffer member between the stack and the flexible heaters, which do not react well to loading. Also, a single layer of carbon paper was placed between the copper current collector plate and the carbon end plate at each end of the stack. The compressibility of the carbon paper compensates for variations associated with thickness matchup between terminal post foot and the pocket stamped into the collector plate.

3.2.1.4 Stack Assembly Status

During this period of time, five additional nine cell stacks were assembled and tested. Table 3-16 provides a summary of the characteristics for each new stack and also provides a summary for previous stacks to allow comparison.



## 3.2.2 Acid Transport Test Fixture

The Acid Transport Test Fixture was developed to provide a technique for measuring wicking rates and wicking capabilities of various candidate fuel cell matrices. The initial design of the Acid Transport Test Fixture is shown in Figure 3-17.

The test fixture consists, basically, of a single cell combined with the nine cell assembly non-repeating hardware. The anode and cathode end plates were modified to include nine resistance measuring contacts recessed in each of the "A" and "B" faces of the ends plates. The Viton insulations located in the recesses of the plates provide a small spring force to maintain contact between the resistance measuring contacts and the electrodes and insulate the contacts from the end plate. The resistance measuring terminal wires are cemented within the end plate grooves and routed out to a rotary switch. The rotary switch is used to connect the individual pairs of contacts to a milliohmmeter.

An acid level indicator tube is provided at the anode end plate near the acid inlet to monitor the inlet acid head. Three acid level indicator tubes are provided at the cathode end plate in a location opposite to the acid groove to monitor the acid that moves across the cell from the acid groove. The stack assembly heaters are included in the fixture so the wicking process can be observed with the cell at the desired temperature (presently 54°C (130°F)).

The first test was conducted to determine the sensitivity of the technique used to measure the wicking rate and to establish improved techniques for future tests. The test results indicated that the wicking rate measurement technique could be improved by insulating the contacts in the anode end plate from the anode backing paper. This can be achieved by providing an anode with nine 1.3 cm (0.50 in.) diameter cut-outs at the anode resistance measuring contact locations. The cut-outs could be replaced with blotter paper or silicon carbide. Since the cathode backing paper and cathode end plate are both good conductors and make contact with each other, the only resistance measuring connection needed on the cathode side of the fuel cell is to the cathode end plate.



The test results indicated that in a single cell with no MAT-1 the cell resistance after adding acid reduced from  $85 \text{ m}\Omega$  to  $10 \text{ m}\Omega$  in 24 hours. The test results also indicated that with an acid inlet head up to 10 cm (4 in.) and the acid outlet closed off no acid could be forced horizontally out of the opposite edge of the cathode in a 24 hour period. The post test examination after disassembly of the fixture revealed that the acid wicking in the silicon carbide layer of the cathode progressed approximately 2/3 of the distance across the cell.

## 3.2.3 Acid Delivery System

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*1*г с An Acid Delivery System (Figure 3-18) was designed for bench-top testing in order to obtain design and operational information. The bench-top design was modified to include in-vessel testing in conjunction with a phosphoric acid fuel cell stack and provide startup and operational information for the final design of the acid pumping system.

The pump selected for the bench-top testing is a pneumatically driven commercial metering pump made of PVC material. Although the pump has an adjustable stroke, the volume of fluid delivered (with the shortest stroke possible) is much greater than required for addition to an operating stack. An overflow line from the acid reservoir on top of the stack returns the excess acid to the pump sump outside of the pressure vessel. A metering valve was added between the acid reservoir and the inlet connection on the stack to restrict the flow of acid and allow sufficient time for the excess to flow through the overflow line.

## 3.2.4 Nine Cell Stack Testing

The overall performance results in the nine cell stack testing activity at Westinghouse are summarized in Table 3-18 and Figure 3-19. Note that Table 3-18 is cumulative from the start of the stack testing program. Details regarding the performance of each nine cell stack tested are given in the sections which follow.



# Figure 3-18. Acid Delivery System - Nine Cell System

**TABLE 3-18**.

SUMMARY OF NINE CELL STACK PERFORMANCE

N/A 0.136 0.203 N/A 0.3 0.28 0.28 Cell'Resistance, mΩ @ Full Power 0.19 N/A 0.62 N/A 0.31 N/A 0.7* N/A 0.27 End 6 hrs 0.59* N/A 0.145 0.143 0.19 N/A 0.22 0.17 0.20 0.17 0.20 0.21 N/A 0.3 0.28 N/A Stack Resistance, mn @ Full Power 2.3 7.0 1.85 3.2 4.5 7.1* End N/A 3.3 N/A 3.0 N/A 3.5 2.9 3.4 2.95/2.7 6 hrs 2.23 N/A 1.91 N/A 2.8 N/A 1.77 2.17 3.70 6.3* N/A N/A 3.5 3.0 Total Hours Oper. 48: 570 830 140 60 101 57 178 128 90 186 1007 100 80 778 Test Time (Hrs)0atm2atm (SRG) < 1hr 33* 23* 376 ₂(2) 470 30¥ 21* 35* 63* 153 23 25***** 254 *H2 Fueled **Cells 1 & 2 were excluded from the selection because of voltage sharing problem. . 118 32 33 33 37 133. 37 133. (H₂) 901 S 67 60 4.8 atm (SRG), mV? Avg. of 5 center cells 634(2): 587* 382* 634 654 650 650 654* 667(5) 558* 620 631* 627* 614 632* 632* 530 Performance Range @ 4.8 atm 486/633 (542/652)* (518/647)* (598/633)(6) (579/652)* (606/655)* (303/557)* 369/657 303/657 59/664 (512/642)* Stack 88/706(2) (49/660)* 66/681(5) 403/655 (SRG), mV 542/601* 82/586 Did not Test Range @ l'atm Perfőrmance 745/774 515/684(3) 667/699 459/711⁽⁴⁾ (H₂), mV 669/684 730/743 500/664 508/733 625/699 484/691 664/706 625/669 664/694 645/696 t67/655* 642/708 III (12/14/83) IV (1/3/84) W009-06 W009-07 W009-08 I (9/22/83) II (11/21/83) -3rd -2nd 1009-05-1st / (1/24/84) Stack No. M009-02 W009-03 M009-04 W009-09 W009-10 10-600M 11-6001

Notes (1) All testing @ 1 atm @ 47 mA/cm² unless otherwise indicated; all testing @ 4.7 atm @ 325 mA/cm², SRG/Air unless otherwise indicated 200 Amps on SRG Cell #3 not included

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90 Amps on H₂ Cell 7 not included

5 center cells range



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Additional stack testing at ERC established an initial performance baseline using pure hydrogen fuel in Stack E-009-02. The performance level of Cells 1 thru 5 was 624 mV at the following operating conditions: 480 kPa (70 psia), 190°C (374°F), 325 mA/cm² (300 A/ft²), 80-83 percent fuel utilization, 40-50 percent oxidant utilization. A new performance baseline was established in Cells 1 thru 5 of Stack E-009-04. This new baseline is 672 mV at the same operating conditions mentioned previously. The 48 mV increase is attributed mainly to the increased inlet gas temperatures of the new test facility, new startup procedures and new manufacturing procedures may have contributed to this increase. A quantitative separation of the individual contributions cannot be made from the existing data. Factors which will provide additional improvements in performance were demonstrated by comparing the resistance of cells with 175 microns (7 mil) and 75 microns (3 mil) silicon carbide layers (.12 m $\Omega$  and .09 m $\Omega$ , respectively), and then comparing performance of cells with the standard 250 microns (10 mil) carbon plus 175 microns (7 mil) silicon carbide with the thin, 150 microns (6 mil) carbon plus 75 microns (3 mil) silicon carbide (624 mV and 654 mV, respectively). These thinner matrices are projected to result in a 30 mV improvement over the present 672 mV baseline and thus meet the goal of 700 mV. Future stacks which incorporate both matrix improvements and operating procedure improvements are required to confirm this expectation. Reproducibility will also be demonstrated in future stacks.

The goal of 500 hours of continuous operation at 480 kPa (70 psia) was achieved in Stack E-009-04, but there was significant performance decay due to a resistance increase. The longest stable performance was achieved in Stack E-009-02. Over a period of 359 hours (between the peak and final performance) there was an average loss of 8 mV/cell. A number of factors have limited the achievement of long term endurance. These factors include facility upsets in  $\Delta p$ , power failure to controls, cell assembly defects, cross leaks and acid management problems. Procedures have been identified to improve each of these factors in future stacks.

Though the factors mentioned above were present, some limited data was obtained using SRG containing 75 percent hydrogen, 24 percent  $CO_2$ , and 1 percent CO

from cylinders of premixed gases. The expected performance loss on SRG is 20 mV. Stack E-009-04, which was tested after revising the operating procedures, showed a 20-30 mV SRG loss. Higher losses were observed in those cells which were diagnosed to have cross leaks and poor temperature distribution. Higher than expected SRG losses are suspected to be caused by the following factors:

- cross leaks
- fuel maldistribution
- inaccurate flow measurement
- poor cell temperature distribution caused by low coolant and process gas inlet temperatures.

Steps have been identified to improve test facility and cell design to correct the problems mentioned above.

#### Stack W-009-06

Test and disassembly data were developed for a pressurized nine cell stack (W-009-06) tested at Westinghouse and ERC. The stack was operated for over 500 hours, of which 325 were at 480 kPa (70 psia). Average stack performance for the center five cell group at 140 mA/cm² (130 A/ft²) was approximately 718 mV/cell (190°C (374°F), 480 kPa (70 psia), 70 percent hydrogen utilization, 40 percent air utilization).

The following is a summary of important observations/conclusions:

- The final average stack performance of the center five cell group at 480 kPa (70 psia) at Westinghouse was within approximately 6 mV/cell of the initial average stack performance at 480 kPa (70 psia) at ERC.
- Initial OCV testing at ERC indicated possible problems with Cells 1 and
   9. Throughout the testing, Cells 1 and 9 were substantially below the stack average. Eliminating these two cells from the average showed

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this stack performance approximately equaled Stack E-009-01 performance, 725 mV/cell and 721 mV/cell respectively at 140 mA/cm² (130 A/ft²). These two cells (1 and 9) were found to be corroded upon disassembly of the stack.

- Cathode plate and backing paper corrosion was found in Cells 1, 3, 6 and 9.
  - Cell 1 oxidant inlet area and acid channel area
  - Cell 3 fuel outlet area of the cathode plate
  - Cell 6 fuel outlet area of the cathode plate
  - Cell 9 oxidant inlet area
- Low performance of Cells 1 and 9 made it necessary to test the stack at a reduced current.

# Stack W-009-07

Stack W-009-07 underwent testing in two phases of operation which totaled 778 hours.

Major results during each phase of testing are given in Tables 3-19 and 3-20 and noted below:

### Phase I

Total Phase I test time was 385 hours which includes 262 hours at atmospheric conditions and 123 hours at pressure. Testing included OCV tests, polarization tests at 143kPa (20.7 psia) and 480 kPa (70 psia), and process gas utilization tests for the anode and cathode. The desired full power hold was not achieved due to problems with the cathode air flow.

·			CELL VOLTA	ge and resi	STANCES AT RAT	ed power (\$	SRG FLOW)		
		<u>с</u> ,	hase I				hase II	~	
	Beg	inning	End		Begin	ning	(H ₂ Fueled)	End	
Date, Time	7/3	1, 0:27:51	8/9,	5:00:02	10/11	, 8:20:15	10/18, 8:55:5	51 10/25	. 10:11:13
Current, A	351		350.	5	351.2		351	351	
Temp, °C	189	•6 (373°F)•	195.	7 (384°F)	193.8	(381°F)	191 (376°F)	191.2	(376°F)
Pressure, kPa	1 480	(70 psia)	480	(70 psia)	, 480 (	70 psia)	480 (70 psia	) 480 (	70 psia)
<b>%</b> Н ₂	83		83		70		70	70	
% 02 2	40		50		51		50	51	
Cell # Volt	age, mV	Res, m	Voltage, mV	Res, m ^Q	Voltage, mV	Res, mM	Voltage, mV	Voltage, mV	Res, mΩ
-	574	0.285	540	0.34	567	0.27	576	535	0.32
2	576	0.24	552	0.28	601	0.19	608	579	0.20
m	479	0.335	489	0.35	542	0.29	545	520	0.285
4	47.1	0.435	354	0.70	432	0.52	427	266	0.85
ى ک	557	0.280	440	0.49	511	0.29	520	459	0.40
6	503	0.370	410	0.56	496	0.36	506	467	0.40
7	508	0.355	467	0.40	606	0.17	608	586	0.16
8	569	0.240	293	0.71	537	0.31	550	513	0.295
6	493	0.390	230	0.895	547	0.29	557	515	0.30
AVERAGE	525		419		538		544	493	
STACK	4.514V	3.6 mΩ	3.56V	5.45 m	4.77V	2.69 mΩ	<b>4.</b> 83V	<b>4</b> .38V	3.8 mΩ

TABLE 3-19

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		8/1, 23:1 (after de
	OWER POINT	8/1, 23:07:53
TABLE 3-20	PERFORMANCE AT QUARTER F	7/29, 16:17:29
	STEADY STATE	4:01:14

Date, Ti	me [`] 7/2!	9/83, 14:01:	.14 7/29	<b>,</b> 16:17:29	8/1,	23:07:53	8/1, (after due to	23:15:38 r depressurization o loss of power)
Fuel	Н ₂		SRG		Н		SRG	
Current,	A 94.5	~	92.3	•	93.5		93.6	•
Temp, °C	190	(374°F)	190	(374°F)	) 161	376°F)	) 161	376°F)
Pressure	, kPa 276	(40 psia)	276	(40 psia)	276 (	40 psia)	276 (4	40 psia)
<b>%</b> Н ²	70-8	35	90.5		86		87	
يم 0 م	40	• •	42		42		42	
Cell #	Voltage, mV	Res, m	Voltage, mV	Res, mR	Voltage, mV	Res, mΩ	Voltage, mV	Res, mM
	723	0.45	713	0.49	728	0.51	725	0.54
∾ .70	716	0.48	701	0.52	708	0.63	706	0.67
с	669	0.57	657	0.60	686	0.54	686	0.55
4	674	0.69	. 662	0.75	660	0.85	657	0.90
ۍ ۲	703	0.45	691.	0.49	691	0.56	689	0.58
9	679	0.73	662	0.83	667	0.72	662	0.77
7	686	0.55	679	0.59	686	0.59	684	0.61
8	733	0.38	718	0.40	728	0.46	725	0.46
б	669	0.66	681	0.70	694	0.67	689	0.70
AVERAGE	698		685		694		691	
STACK	6.20V	0°9	6.1V .	6.3 mΩ	6.18V		6.145V	6.6 mΩ

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TABLE 3-20 STEADY STATE PERFORMANCE AT QUARTER POWER POINT (CONT'D)

8/2,	19:28:31	8/2,	19:43:20	8/6,	0:20:26	8/6,	0:33:55
		SRG		H ₂		SRG	
		92		90.4		88.9	
(374	°F)	190	(374°F)	) 161	376°F)	189.9	(374°F)
(70	psia)	480	(70 psia)	276 (	40 psia)	276 (	40 psia)
		86		93		73	
		. 42	•	40		40	
Res.	C III	Voltage, mV	Res, mn	Voltage, mV	Res, mn	Voltage, mV	Res, mΩ
0.5	[	716	0.54	718	0.58	וול	0.615
0.6	4	696	0.68	686	0.84	672	0.925
0	54	681	0.56	686	0.63	677	0.66
0 <b>.</b> 0	36	647	16-0	623	1.22	608	1.32
0	58	679	0.61	657	0.83	645	06.0
0	76	650	0.84	623	1.0	623	1.10
0	58	674	0.61	689.	0.62	672	0.645
0		וול	0.52	169	0.82	677	0.84
0	66	674	0.70	664	0.96	650	1.05
		681		671		649	
<b>6.</b> 6	Cum (	6.057V	6.9 mV	5.964V	8.4 mS	5.867V	8.95 mS

	(CONT'D)
	POINT
	POWER
3-20	QUARTER
ABLE	AT
17	PERFORMANCE
	STATE
	STEADY

Date, Time	8/9, 10:37:	05	8/9, 10:54:0	2	10/13, 2:04:24
Fuel	SRG		H ₃		H,
Current, A	89.6		90.9		5. 89.6
Temp, C	191.5 (377°	F) .	191.7 (377°F	(	189.5 (373°F)
Pressure, kPa	276 (40 psi	a )	276 (40 psia	·	480 (70 psia)
<b>ж</b> Н ₂	70		84		50
% 02 2	45		47.5		30
Cell #	Voltage, mV	Res, m	Voltage, mV	Res, mû	Voltage, mV
-	101	0.57	708	0.56	767
2	674	0.80	681	0.78	772
m	664	0.65	674	0.64	757
4	567	1.59	572	1.52	730
2	611	1.50	618	1.00	733
6	584	1.36	594	1.32	733
7	660	0.60	669	0.59	779
8	520	1.85	528	1.90	750
. 6	576	1.50	596	1.42	750
AVERAGE	617		627		752
STACK	5.495V	10.9 m	5.569V	10 <b>.</b> 8 mΩ	6.741V

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## Phase II

Acid additions in the dry room between testing phases recovered cell voltage degradation. Average cell voltage increased 119 mV from the end of Phase I to the start of Phase II. Total test time was 393 hours with 99 hours under atmospheric conditions and 294 hours at pressure. The stack was operated 125 hours at the full power condition:  $325 \text{ mA/cm}^2$  ( $300 \text{ A/ft}^2$ ),  $190^{\circ}$ C ( $374^{\circ}$ F), 50 percent cathode air utilization, 70 percent hydrogen utilization.

### Stack W-009-08

Stack W-009-08 was tested a total of 567 hours which included 103 hours under atmospheric conditions and 464 hours at pressurized conditions. During pressurized operation, hydrogen fuel testing at full power conditions ran for 173 hours while SRG fuel testing lasted 177 hours.

Testing was accomplished in five different phases of operation. Open circuit voltage tests were conducted. Hydrogen utilization tests occurred during Phase V. Table 3-21 lists the various conditions and test times for the five phases of testing. Performance of this stack is indicated in Table 3-22 for each phase. These data indicated no voltage penalty from power or pressure cycling for cells with stable performance. Acid addition helped recover performance and enhance performance of several cells.

### Stack W-009-09

Test operations lasted a total of 157 hours, which included 60 hours at atmospheric conditions and 97 hours at pressure. This stack was the first of the inverted design with anode on top and cathode on the bottom. No MAT-1 matrix was used and the silicon carbide layer on the cathode was increased nominally by 50 microns (2 mils) to improve acid transport. Dry laminated electrodes were also used for the first time in a full size stack.

Phase	Total Operating Time	Atmospheric Operation	Pressurized Operation	Full Power Operation with H ₂	Full Power Operation with SRG
	110 hrs	32 hrs	78 hrs	63 hrs	0 hrs
11	54	30	24	Ċ	
ĨĨĨ	181	œ	173	2	153
ĬV	130	10	120	80	23
>	92	23	69	25	0
Totals:	567	103	464	173	177

TABLE 3-21 TEST TIMES FOR STACK W-009-08

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:
	() () ()				بر سین در ه		<u></u>				-WAESD-TR-85-0030
	<pre>@ 130°F Resistances in Dry Room Before ransferred to FCTF (</pre>	0.52 0.57	0.66 0.64	0.67	0.63	0.62	0.76	0.74	7.7	n ³ , retained 56 cm ³ . d (98%)	HALSD TH CO
	Electrodes T						C160-6 A163-6	C161-2 A163-8		. Wicked in 150 cr ick in 190 cm ³ acid	
<b>09-08 DATA</b>	ance @ FP H2, 40% 02	(0.25) (0.28)	(0.43) (0.65)	(0.53)	(0.85)	(0.65)	(0.50)	(0.86)	(6.4)	uring assembly cid, 6% RH. W	
TABLE 3-22 OF STACK WOO	Perform F w/H2, 50%	557 523	471 386	4/1 386 315 315 369 418 303 3.45 3.45 3.45 3.45 t 28 wt% ac	³ (total) d cs 98 wt% a	ressure.					
MMARY 0	Room to FCTI								×	s 50 cm	rs at pi
SU	rodes (Before Transferred A163-6 0.43 A162-7 0.46	0.45	0.44	0.50	0.51	0.71	0.46	Sk = 5.6 mΩ	<pre>% RH) spec/cell - loss min - No cross leak. n, cells 8 &amp; 9 was ac se I = 100 hrs. 76 h</pre>	se I = 100 hrs, 76 hr	
		A163-6 A162-7	A162-5 A164-1	A162-6	A164-5	A164-4	A163-2	A162-8		% and 16% د 30 cm ³ /r gperatio	irs of Pha
	Electr	C159-4 C161-6	C159-1 C160-5	C161-5	C160-1	C160-2	C159-7	C160-3		1 acid (93 nal leak e phase 2 ined 120 c	l test hou
	Cell No.	- 2	ω4	ک	9	7	8	6 3.	-75	Note: (1) Addec (2) Exter (3) Befor retai	(4) Tota

Cell No.	Phase II (18:30, 11/23) Performance W/SRG, 50% H ₂	Phase III (8:30, 12/16) e @ FP 2, 33% 02	Phase IV (22:06, 1/4)	Electrodes	Resistance in DR Before Transferred to FCTF (mΩ)	Phase (13:00, 1/25) Performance 50% H ₂ ,	v (11:00, 1/29 @ FP w/H2 40% 02	~
	Beginning	Peak	End (3)			Beginning	End	
-	547	603 (0.2)	59 (0.19)	C201-8 A207-8	2.72	271	230	
2	598	672 (0.16)	721 (0.15)	C204-6 A207-1	0.64	623	657	
n	608	652 (0.14)	667 (0.14)		0.70	6.35	660	
4	657	657 (0.12)	660 (0.12)		0.70	655	664	
2	642	655 (0.13)	657 (0.14)		0.74		. 664	
9	65 <u>0</u>	650 (0.11)	655 (0.12)		0.71	. 655	655	
	608	613 (0.13)	616 (0.14)		0.67	650	642	
φ	410	594 (0.13)	596 (0.16)	•	0.80	. 642	635	
6	591	635 (0.14)	650 (0.13)		0.74	650	647	
	5.20 V 5	5.25 V (2.2 mQ)	5.10 (2.3)		.9.40	5.37 V	4.91 V	
		acid adde	p		· · ·	shutdow of cell	n because 1 perf.	
Note:			• •	•				
(1) Tota (2) Tota (3) Shut	ll test time of Ph l test time of Ph down, replaced Cé	hase 2 + 3 + 4 = nase V = 90 hrs, ells 1 and 2. T	363 hrs, at pre at press = 69 h otal acid added	essure = 315 hrs, irs. - 150 cm ³ ; total	w/SRG at press = 182 h acid retained - 88 cm ³	S.	WAESD-TR-85-003	

TABLE 3-22

Testing was conducted in two phases. During Phase I, the stack ran for one hour at atmospheric conditions when an emergency shutdown occurred due to extremely high temperatures. Phase II testing lasted 156 hours. Atmospheric pressure tests ran for 59 hours. The remaining test time included 36 hours at full power conditions with hydrogen fuel and nine hours at full power using SRG. OCV and utilization tests were conducted.

The stack exhibited an improved ability to wick acid due to the design change which provided partial surface exposure of the acid transport member (SiC) to the acid groove. Stack resistance was higher than noted in previous stacks, partly due to the thicker silicon carbide layer. However, Cell 1 was the single largest contributor to the increased resistance.

After the high temperature excursion in Phase I, the stack was returned to the dry room and examined. No visible damage was apparent but cross leaks were much higher than before testing. Additional tests showed that the stack retained acceptable bubble pressure.

Significant accomplishments achieved with this stack included:

- Recovery of stack performance after an over-temperature excursion of nearly 56°C (100°F) was experienced.
- Completion of a full startup on SRG.
- Demonstration of an emergency shutdown from rated full power.
- Mapping of 100°C (212°F) shutdown from two atmospheres with SRG.
- Performance of numerous startup and pressurizing cycles that verified the results obtained from stack W-009-08 (indicating no performance penalty resulting from pressure cycling).

 Acid addition in the test loop to recover cell performance after apparent electrolyte loss.

This stack provided a number of diagnostic data including effects of:

- Oxygen gain of 80-85 mV at atmospheric test conditions.
- Negative voltage condition startup.
- No apparent enhancement from the reversal of process gases to electrodes (exchange of anode for cathode, and vice versa).
- Current step and flow interruption cell voltage and response characteristics.

#### Stack W009-10

The stack accumulated 1007 hours of test operation with 117 hours at atmospheric conditions and 890 hours at pressure. During pressure testing the stack ran 290 hours on SRG with 254 hours at rated conditions.

The stack design is similar to Stack W009-09 except that MAT-1 was included in the matrix. Minor design changes involved increased thickness of the insulator plates between the end heaters and current collectors to more uniformly redistribute the load transferred through the flexible heater elements to the end plates. A sheet of carbon paper was also placed between the current collector and end plate at each end of the stack to cushion any high load points caused by the interface of the terminal point base in the collector plate and the end plate.

#### Test Results

Table 3-23 summarizes steady state performance at full power condition with SRG. At peak performance, which occurred after 45 hours in the SRG endurance run, the center four cell average was 630 mV.

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### TABLE 3-23

STACK W-009-10 STEADY STATE PERFORMANCE AT FULL POWER POINT WITH SRG

	Phase 1b		Phase 1c	
	lst FP data pt	lst FP Data	Peak	Last FP Data
Nate Time	3/19 9.24	4/4 3.00	4/6 10.00	4/6 16.00
Time on Test	129 hours	597 hours	470, 10.00	900 hours
	420 10015	597 Hours	052 110415	090 HOURS
(at pressure)	2004	2204	2264	2474
Current	322A	338A	336A	- 34 I A
Temperature	192°C	190°C	193°C	192°C
Pressure	70 psia	70 psia	70 psia	70 psia
% H ₂ Util.	70%	70%	70%	70%
% 02 Util.	31%	40%	33%	33%
Cell Voltage				
Cell l, mV (Res, mΩ)	547 (0.23)	616 (0.17)	618 (0.18)	611 (0.19)
Cell 2, mV (Res, m $\Omega$ )	442 (0.19)	611 (0.16-0.18)	620 (0.17)	572 (0.08-0.12)
Cell 3, mV (Res, mΩ)	486 (0.24)	515 (0.23)	508 (0.26)	447 (0.33)
Cell 4, mV (Res, m $\Omega$ )	611 (0.21)	620 (0.19)	628 (0.19)	623 (0.21)
Cell 5, mV (Res, mΩ)	635 (0.16)	647 (0.16)	655 (0.16)	647 (0.18)
Cell 6, mV (Res, mΩ)	591 (0.20)	594 (0.18)	608 (0.18)	576 (0.21)
Cell 7, mV (Res, mΩ)	589 (0.21)	608 (0.15)	630 (0.17)	596 (0.22)
Cell 8, mV (Res, mΩ)	349 (0.15)	544 (0.15)	403 (0.165)	550 (0.20)
Cell 9, mV (Res, m $\Omega$ )	620 (0.16)	623 (0.15)	667 (0.15)	620 (0.16)
Stack Voltage, V	4.704	5.158	5.124	4.922
(Stack Resistance, $m\Omega$ )	(2.60)	(2.50)	(2.50)	(2.90)

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Effects of humidification were studied during Phase 1b operation. The design levels of moisture were chosen for the anode and cathode stream, which were 4.4 percent and 1 percent by mole, respectively, for the anode and cathode. Table 3-24 presents the results of the tests. For the anode fuel, moisture was added to hydrogen first and then to hydrogen with  $CO_2$  later. About 5 mV gain per cell was observed in both cases with humidified anodes. For humidified cathodes, an average of 5 mV increase in cell voltage was also observed. However, an endurance run with the humidified hydrogen with  $CO_2$  fuel was conducted but cell 1 dropped 47 mV from 611 mV to 564 mV. The stack was then returned to dry fuel.

Polarization test data are shown in Figures 3-20 thru 3-24. The first figure is for the polarization test with hydrogen and the following four with SRG. The SRG tests were conducted at the beginning and end of Phase 1c. The hydrogen polarization curve shows an average 20 mV higher than that of the first SRG curve.

Before the stack was depressurized, 13 mapping points were conducted. Table 3-25 shows the results and cases used in the tabulation. In summary, the good cells of stack 10 showed higher voltage gain, due to increase in pressure than the theoretical value. The temperature gain, effect 1.5 mV/1°C (.8 mV/1°F), was considered to be high at 166A and 325 kPa (47 psia). The current effect varied with pressure operating level. A high pressure, 690 kPa (100 psia), the current effect rate was -0.57 mV per ampere increase.

#### Stack W-009-11

Stack W-009-11 was of the same cell arrangement as stack W009-10. At assembly, the stack was lightly loaded at an average stack pressure of 55 kPa (8 psi) and an average cell strain of 6 percent based on 190°C ( $374^{\circ}F$ ). The objectives of the test were: (1) to determine the minimum initial cell compressive load and strain for acceptable cell performance, (2) to determine stack and cell operating characteristics over limited operating map including startup and shutdown

**TABLE 3-24** 

STACK W-009-10 EFFECT OF HUMIDIFICATION ON CELL PERFORMANCE (PHASE 1b)

Stack Operating Conditions

Date, Time	3/12, 18:00	3/12, 20:00	3/13, 12:00	3/13, 16:00	3/13, 1:00	3/13, 3:00
Current, A	339A	340A	339A	339A	340A	341A
Temp, °C	190°C	190°C	192°C	192°C	.189°C	191°C
Pressure, psia	70 psia	70 psia	70 psia	70 psia	70 psia	70 psia
Anode Content	H ₂ only	н ₂ & н ₂ 0	$H_2$ and $CO_2$	$H_2 + CO_2$	H ₂ only	H ₂ only
		4.4% by mole		& H ₂ 0	I	I
	·			4.4% by mole		
Cathode Content	Air Only	Air Only	Air Only	Air Only	Air Only	Air & H ₂ 0
			·			1% by mole
% H ₂ Util.	51%	51%	51%	49%	84%	84%
% 0 ₂ Util.	40%	40%	40%	40%	33%	33%
Cell 1 mV	625	620	603	611	628	630
Cell 2 mV	635	640	623	633	640	642
Cell 3 mV	630	630	613	620	630	633
Cell 4 mV	623	628	620	623	628	633
Cell 5 mV	652	650	645	647	655	657
Cell 6 mV	613	618	608	613	620	620
Cell 7 mV	601	606	598	611	611	613
Cell 8 mV	606	608	545	547	59 <u>8</u>	603
Cell 9 mV	635	640	647	642	625	625
Stack Voltage, Volts	5.412	5.437	5.310	5.339	5.432	5.451

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CELL VOLTAGE.

,Vm







Cases Used	Variable Effect	Range of Variable	Experimental Value	Theoretical Value
8 and 9	Current @ 30 psia	80-160A	0.90 mV/amp	
2 and 11	Current @ 40 psia	160-340A	0.85 mV/amp	
3 and 4a	Current @ 100 psia	160-420A	0.57 mV/amp	
6 and 7	Pressure @ 165A	30-47 psia	38 mV	26 mV
3 and 5a	Pressure @ 165A	40-100 psia	.77 mV	54 mV
7 and 10	Temperature @ 166A, 47 psia	170-190°C	1.5 mV/1°C	w.2 to 1.3
				mv/1°C*
8 and 12	Temperature @ 166A, 47 psia	170-190°C	1.5 mV/1°C	4.2 to 1.3
			·	mV/1°C*
		·		*2 x 2 data
	•	•		
	-			

TABLE 3-25

procedures, and (3) to continue verifying stack and cell performance with components to be used in the 10 kW stack and with the inverted cell stack configuration.

Only the first of the three testing phases (Phase 1a) totalling 159 hours was completed by the end of this report period. Table 3-26 summarizes the results.

3.3 10 kW Fuel Cell Stack Development (WBS 1103-03)

3.3.1 Development Status

Design modifications were made to account for the larger shrinkage (5.2 percent) incurred with the A4421 regrind plate material. Molded, heat treated plates were inspected and statistically analyzed to establish the appropriate dimensions. All drawings affected by plate shrinkage were revised and released. The instrumentation drawing was also completed and released. Based on nine cell work, the electrode configuration was re-evaluated and re-designed in an inverted configuration with the cathode on the bottom to provide better acid transport. Assembly, process gas piping, and instrumentation drawings were revised as necessary to incorporate the "inverted" cell configuration.

Plate/electrode subassembly and final stack assembly procedures were also revised for the inverted cell design. Fabrication of all non-repeating hardware was completed and plate subassembly initiated.

Samples of polyethersulfone (PES) process gas manifold material were tested in hot acid for compatibility. Some surface attack was observed (primarily in the outer glass fibers) but does not warrant a special surface coating for short term testing. This subject will be reconsidered following completion of this test. Flatness was difficult to maintain when machining the manifolds from extruded slab material. This was overcome by making light cuts and reclamping frequently without bending against the machine table.

## TABLE 3-26

STACK W-009-11 STEADY STATE PERFORMANCE AT FULL POWER POINT WITH H2

Operating Conditions		Phase la	
	Beginning	Peak	End
Date, Time	5/14, 12:00	5/14, 18:00	5/17, 10:00
Time on Test, hrs	78 hrs	84 hrs	148 hrs
Current, A	345 A	345 A	336 A
Temperature, °C	190°C	188°C	193°C
Pressure, psia	70 psia	70 psia	70 psia
ΔLoad, psi	0	0	0
%H ₂ Utilization	70%	72%	70%
%0 ₂ Utilization	33%	33%	33%
Cell 1, mV (mΩ)	169 (0.25)	181 (0.28)	100 (0.65)
Cell 2, mV (mΩ)	594 (0.25)	601 (0.29)	594 (0.26)
Cell 3, mV (mΩ)	554 (0.25)	557 (0.25)	557 (0.25)
Cell 4, mV (mΩ)	591 (0.19)	601 (0.20)	596 (0.20)
Cell 5, mV (mΩ)	574 (0.23)	586 (0.23)	576 (0.24)
Cell 6, mV (mΩ)	586 (0.19)	594 (0.18)	589 (0.20)
Cell 7, mV (mΩ)	542 (0.24)	550 (0.24)	550 (0.25)
Cell 8, mV (mΩ)	562 (0.20)	574 (0.20)	569 (0.22)
Cell 9, mV (mΩ)	581 (0.22)	591 (0.22)	586 (0.24)
Stack Volt V (mΩ)	4.53 (3.0)	4.62 (3.0)	4.49 (3.4)

A prototype metal stack was designed and assembled according to test procedures using aluminum plates in place of graphite plates and prototypic edge seal gaskets. The primary purpose of this prototype was to develop a manifold seal system. It also provided experience in building a larger stack and an opportunity to evaluate the stacking assembly fixture as shown in Figure 3-25. Based on manifold seal tests on a nine cell stack, five different seal configurations were tested on this metal stack with plates aligned as well as possible. Since the aluminum plates had no flow channels, only one manifold was required and only one end of the stack was involved in the leak test. Two different manifolds were used for these tests. A PES manifold with O-ring grooves was used to test 0-ring seals 0.48 cm (0.19 in.) dia. of 50 durometer and 75 durometer Viton. Based on nine cell stack tests, the grooves were reduced in depth to allow greater squeeze on the O-ring. An aluminum manifold with a flat seal surface was used to test flat cured Viton 0.32 cm (0.13 in.) by 0.64 cm (0.25 in.) of 50 durometer and "Gore-Tex" joint sealant 0.48 cm (0.19 in.) size. Table 3-27 summarizes these tests. The leakage data is reported for both chambers (anode and cathode) in the manifold and for two differential pressures 1.7 kPa and 34.5 kPa (0.25 psi and 5.0 psi). These tests show that the flat 0.32 cm Viton gasket is the best, followed by one layer of "Gore-Tex" on flat Viton, then the 50 durometer O-ring or two layers of "Gore-Tex", and finally the 75 durometer O-ring. Future tests will evaluate the best of these on a misaligned stack.

#### 3.3.2 10 kW Stack Design Description

The 10 kW stack mounted in the test loop pressure vessel is shown in Figure 3-26. The stack contains 44 cells in eight groups of five with two cells at each end. Each group of five cells is separated by cooling plates. The remaining cell components in the five cell group are separated by bipolar plates. The nine cell stack discussion relative to the "zee" and "tree" pattern bipolar and cooling plates, respectively, is applicable for this stack and is, therefore, not repeated herein. Additional design details were previously described in the "Final Report for the First Logical Unit of Work."



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## TABLE 3-27 10 kW METAL STACK MANIFOLD SEAL TEST RESULTS (ALIGNED STACK)

		Leaka cm²/	.ge ⁽²⁾ min
	Compression (Percent)	0.172 kPa (0.25 psi)	3.45 kPa (5 psi)
<u>Aluminum Manifold Tests(1)</u>			
<ul> <li>Flat Cured Viton 0.32 cm (1/8 in.) Thick by 0.64 cm (1/4 in.) Wide</li> </ul>	32	0/0	0/0
• Gore-Tex 2 Layers of 0.48 cm (3/16 in.)	(0.045 gap)	0/5	19/53
<ul> <li>Gore-Tex 0.48 cm (3/16 in.) on Top of Flat Viton 0.32 cm (1/8 in.)</li> </ul>	(0.10 gap)	0/0	6/12
Polyethersulfone (PES) Manifold Test(3)			
<ul> <li>Viton O-Rings (0.53 cm (0.210 in.), 75 Durometer)</li> </ul>	43	19/24	High
<ul> <li>Viton O-Rings (0.48 cm (0.187 in.), 50 Durometer</li> </ul>	41	0/5	11/82

(1) Flat face sealing flange.

(2) Leakage data given for each half of one manifold.

(3) Reduced depth O-ring groove 0.25 cm (0.100 in.) vs. 0.38 cm (0.150 in.).

(4) All tests at room temperature.



Figure 3-26. PAFC 10 kW Stack Test Assembly (Non-Inverted Arrangement Shown)

#### 3.4 25 kW Short Stack Development (WBS 1103-05)

#### 3.4.1 Development Status

Consistent with the 10 kW stack, this stack was re-designed for an inverted cell with the cathode on the bottom and the anode on the top. This provides better acid transport since the acid must now run over the cathode in the acid groove area which insures good contact with the acid supply. Detail and assembly drawings were revised as required.

#### 3.4.2 25 kW Stack Design Description

The 25 kW stack mounted in the test facility pressure vessel is illustrated in Figure 3-27, which has not been revised for the inverted cell design. The major difference would be to move the process gas connections from left to right and vice versa. The stack assembly is supported from a carbon steel lower support plate which is mounted on the cooling air outlet pipe and two support columns. Many of the 10 kW stack design features described in Section 3.3.2 are applicable to the 25 kW stack and are not repeated here.

The 25 kW stack contains 104 cells in 20 groups of five with two cells at each end of the stack. Each group of five cells is separated by cooling channels with the design identical to that used for 10 kW stacks.

3.5 100 kW Full Stack Development (WBS 1103-06)

3.5.1 Development Status

The 100 kW Stack Preliminary Design was updated to include the most current stack design features. The major revisions included in the update were as follows:



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Figure 3-27. 25 kW Stack Design (Non-Inverted)

- Revision of the stack and process gas supply lines to reflect the inverted cell arrangement, i.e., the inversion of the anode and cathode. Gas supply line connections to the vessel and stack manifolds were repositioned.
- Stack related provisions for a continuous acid pumping system were added. This included an acid distribution box at the top of the cooling air plenum and designated acid and vent line routings with appropriate vessel penetrations.
- The process gas manifolds were modified from a two piece design to a molded four piece design. The shorter sections should be more suitable for injection molding than the half-height components.
- Other minor changes were incorporated. In addition, the 100 kW Stack Preliminary Design Description was revised to reflect the changes made to the 100 kW Stack Design.

#### 3.5.2 100 kW Stack Design Description

The 100 kW Stack Preliminary Design (inverted) was developed in detail for the stack mounted in the Westinghouse-provided test facility Loop 1 pressure vessel and the stack in the prototypical 375 kW module pressure vessel. Except for test assembly differences, the basic stack design is essentially identical for these two test configurations.

The facility vessel test assembly is very similar in concept to the 25 kW stack design and employs identical support structure and piping components where possible. The main differences are the longer stack, cooling air outlet plenum box, electrical power take-off wires, acid lines, and pressure vessel. In addition, process air in the 100 kW stack is piped directly from the cooling air outlet plenum box (as in the 375 kW module design) instead of via a separate vessel penetration provided in the 25 kW stack.

The module vessel test assembly incorporates maximum component commonality with the Loop 1 test assembly configuration and the 375 kW module design (see Section 3.6). In this arrangement, the 100 kW stack and its cooling air outlet plenum box interface with the module designed internal support structure. Process air is supplied from the cooling air outlet plenum box. A departure was made from the prototypical module piping manifold arrangement on the vessel lower head, to avoid the need for blanking off and, possibly, bleeding the piping legs for the three missing stacks.

The 100 kW stack preliminary design contains 419 cells in 83 groups of five with two cells at each end of the stack. Each group of five cells is separated by cooling channels and is of a design configuration identical to that described for the 10 kW stack. Many of the 10 and 25 kW stack design features as described in Sections 3.3 and 3.4 are applicable to the 100 kW stack, and thus are not reported herein. Additional design details were previously described in the "Final Report for the First Logical Unit of Work."

#### 3.6 375 kW Module Development (WBS 1103-07)

3.6.1 Development Status

The fuel cell module preliminary design was completed and is illustrated in Figure 3-28 and described in Section 3.6.2. The nominal design parameters of the module preliminary design are presented in Table 3-28 at beginning of use of the plant. The major changes made were:

- Inversion of cell components (anode and cathode) and relocation of process piping to accommodate the anode-over-cathode configuration.
- Addition of electric resistance heaters into the cooling air outlet plenum to maintain module internals temperature above 38°C (100°F) to prevent crystallization of the phosphoric acid.

3. Addition of a continuous acid supply system.



Figure 3-28. Fuel Cell Module

# TABLE 3-28 FUEL CELL MODULE DESIGN PARAMETERS

PARAMETER	UNITS	NOMINAL
POWER	ĸ₩ _{DC}	375
TEMPERATURE	°C(°F)	
Oxidant Inlet*		186 (366)
Coolant Inlet		147 (297)
Fue) Inlet		191 (376)
Plate Avg.		191 (376)
*Same as coolant outlet.		
PRESSURE	kPa (psia)	483 (70)
FLOW	kg/h (lb/h)	
Fuel		181 (398)
Oxidant (air)		1400 (3086)
Coolant (air)		27,842 (61,380)
	•	
CELL VOLTAGE	mV	
Open Circuit		920
Operating Limit		800
Operating Point	· · · · ·	690
CELL CURRENT DENSITY	mA/cm ² (A/ft ² )	300 (280)
MODULE VOLTAGE	Volts	
Open Circuit		1540
Operating Limit		1340
Operating Point		1156
MODULE CURRENT	Amps	324

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A stress analysis of the module preliminary design was completed and documented. The analysis indicates that the module structural design satisfies ASME Code requirements and is adequate to withstand the loads expected during transportation, operation, and maintenance.

A preliminary maintenance assessment of the module installed in the Fuel Cell System was performed. The most important maintenance considerations were determined to be the monitoring and replenishment of the phosphoric acid and the periodic replacement of the fuel cell cartridge at approximately 5 year intervals during the 30 year life of the plant. (The fuel cell "cartridge" is the assembly of four 100 kW stacks and associated hardware that fit inside the pressure dome.) These maintenance considerations are discussed in more detail in Sections 3.6.3 and 3.6.4.

3.6.2 Module Preliminary Design Description

The fuel cell module is illustrated pictorially in Figure 3-28. The module consists of four stacks of fuel cells supported inside a containment vessel with their cooling air outlet passages discharging into the space formed between the stacks. Cooling air is admitted to the pressure vessel cavity, surrounding the stack assembly, and flows through the stack cooling air passages to the space between the stacks. Corner seals, provided between the adjacent stack inner corners, minimize the leakage of coolant air into the exhaust plenum. The cooling air enters and leaves the vessel through the lower head, allowing the vessel cylinder and upper head to be easily removed for rapid replacement of the fuel cell stacks without disturbing the external piping. Process air is piped to the stacks from the cooling air outlet cavity between the stacks through holes in the top plate which separates the inlet and outlet cooling air cavities. Fuel is piped to the stacks from a stainless steel piping assembly in the lower region of the pressure vessel which is fed, in turn, from a penetration in the lower vessel head. The fuel and process air return lines from the stacks are led to similar piping assemblies and penetrations at the lower vessel head. Figure 3-29 illustrates all of the above flow paths in a schematic illustration of the module.

90 V. 1





Electric resistance heaters are installed inside the cartridge cooling air outlet plenum for use in preventing crystallization of the phosphoric acid during transportation, storage, and plant shutdown. The heaters are energized from an external source of power and are controlled to maintain a minimum temperature of 38°C (100°F) at the fuel cells. Explosion-proof temperature switches mounted outside the module pressure vessel, (or cartridge shipping container) monitor the cell temperature via a sensor bulb in the cartridge support plate and capillary tubing. The electrical leads from the switches to the heaters and the capillary tubing penetrate the lower vessel head via sealing bushings.

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The electrical connections at the stack consist of copper connectors clamped to the copper collector plates with bolts and supported from the insulator plates. 4/0 nickel-coated TFE insulated cable is swaged to the copper connectors and led to the appropriate connectors on the adjacent stacks and a sealing bushing at the pressure vessel penetration. The cables are supported at intervals along their lengths to prevent whipping under short circuit fault conditions.

Two 6 in. diameter penetrations are provided in the vessel lower head to accommodate instrumentation. Each penetration has the capability of accommodating several multiple lead compression fittings. It is expected that the need for instrumentation will progressively reduce with time as the full stack and module test program proceeds.

Support for the fuel cell stack cartridge against seismic loads in the horizontal direction is provided by four adjustable support screws on a 90° spacing around the upper circumference of the pressure vessel. The supports are individually adjustable to suit the particular dimensions of initial or replacement cartridges following installation of the pressure vessel. Four guide bars supported between the stack compression plates protect the cartridge from damage during installation of the vessel.

The major physical characteristics of the module are listed in Table 3-29. The module external interfaces are listed in Table 3-30.

### TABLE 3-29 MODULE PHYSICAL CHARACTERISTICS

Vessel Height

Vessel Diameter

Base Plate Diameter

Weight

Module Contains 4 Stacks of 419 Cells

Cell Plate Dimensions (machined)

Stack Height

Stack Weight

Vessel Cylinder Weight

Automatic Disconnection of Process Fluid lines during cartridge replacement

Pressure Vessel - Carbon Steel to ASME Section VIII Division 1 for 586 kPa (85 psia) and 204°C (400°F) capability

3-103

3.5 m (11 ft, 6 in) 1.37 m (4 ft, 6 in)

1.68 m (5 ft, 6 in)

5400 kg (6 tons)

83 5-cell groups 2 2-cell groups

0.419 m x 0.3 m (16.5 x 11.75 in)

2.44 m
(8 ft) (approximately)

621 kg (1370 lb)

1088 kg (2400 lb)

		Press	ure	Temper	rature	Flow	Rate
•	Pipe Size	(psia)	(kPa)	(s)	( )。)	(1b/hr)	<u>(kg/hr)</u>
Cooling Air - In	16" Flanged (150 lb)	70	(483)	297	(147)	61,380	(27,842)
Cooling Air - Out	16" Flanged (150 lb)	69	(476)	366	(186)	58,294	(26,442)
Process Air - Out	4" Flanged (150 lb)	69	(476)	379	(193)	3,110	(1,410)
Fuel - In	2" Flanged (150 lb)	70	(483)	376	(161)	398	(181)
Fuel - Out	2" Flanged (150 lb)	69	(476)	385	(196)	373	(691)
Acid - In	TBD	N	, A	Z	/A .		
Electric Connections	o Vessel feedthrough o Wire size, 4/0 avg	, two wire stranded n	sealing bu ickel-coat	ishing (0. ed copper	-Z Gedney T r conductor	ype CSBG) (TFE insulate	
Module Voltage	<pre>1156 (Operating) 1340 (Operating Limit 1540 (Open Circuit)</pre>	<b>•</b>					
Module Current	324 amps						
Module Power	375 kW		×				
Module Heat Rate	7340 BTU/kWh						

TABLE 3-30 MODULE EXTERNAL INTERFACES

3-104

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#### 3.6.3 Acid Supply System

A continuous acid supply system has been incorporated. Phosphoric acid is supplied to the stacks through Teflon tubes clamped to connectors in the stack. The supply lines to the stacks receive acid from a distribution box attached to the cover plate of the cooling air outlet plenum.

The acid pump consists of a piston, cylinder, and check valve assembly. The piston rod of the pump extends through a sealing gland into a double-acting air cylinder. All of the pump parts exposed to acid are made of Teflon. In operation, compressed air is admitted to the piston rod side of the air cylinder causing the acid pump to withdraw acid from the reservoir. The compressed air supply is then valved to the opposite side of the air piston and the piston rod side is allowed to exhaust, causing the acid pump to discharge the swept volume of acid into the supply line to the acid distribution box at the upper end of the fuel cell cartridge.

The acid pump is mounted from the acid tank, which is suspended below the module vessel lower head, clear of the vessel thermal insulation. Under normal operating conditions the acid tank and pump are maintained at a temperature below 65°C (150°F). Heat conducted into the tank from the module pressure vessel is removed from the tank by free convection. Under low temperature ambient air conditions heater coils around the tank and pump are energized to maintain a minimum temperature of 38°C (100°F). Explosion-proof temperature switches mounted on the acid tank monitor the tank temperature and control the electrical supply to the heaters. Provision is made for monitoring the acid level in the tank and for manually replenishing the acid supply when required. Monitoring may be visual, by means of a translucent Teflon sight gauge affixed to the tank. Final details of the selected approach will be verified in testing prior to being applied to the module. Provision for filtering the acid supply from the pump and for minimizing the recirculation of corrosion products and debris may also be incorporated if testing experience shows such provisions to be necessary.

#### 3.6.4 Module Maintenance

The only consumable, apart from the fuel and air supplies, requiring periodic attention is the phosphoric acid supply. Nominal module acid consumption is expected to be in the region of  $1.2 \text{ cm}^3$  per hour (approximately 0.25 gallon per month). However, consumption conservatively could be four times this value, or  $4.8 \text{ cm}^3$  per hour (1 gallon per month).

The acid reservoir suspended below the module pressure vessel has a capacity of approximately 15 liters (4 gallons) with, perhaps, 11 liters (3 gallons) effectively available to the pump. This volume is sufficient for one year of operation under expected nominal consumption conditions or, conservatively, under maximum conceivable consumption conditions, three months. Therefore, acid replenishment is not expected to be necessary more frequently than every three months. Acid level monitoring and logging of acid tank levels (possibly once a month) will be required in the early operational phases. These may be relaxed later as consumption rates are more accurately characterized. It is also conceivable, depending on the rate of contamination of the acid experienced in operation, that periodic removal of the tank for complete draining, cleaning, and replacement of the acid may be required. This would be performed during a scheduled shutdown period during the module life. Possibly, operational experience may show this to be unnecessary during the five year life of the module, in which case tank removal and cleaning would be performed only during the cartridge replacement shutdown period.

#### 3.6.5 Cartridge Replacement

Cartridge replacement is expected to be required at five year intervals. The cartridge and pressure vessel dome must be removed together to satisfy lifting height limitations and protect the cartridge from the elements. The cartridge removal and replacement procedure outlined below satisfies these requirements and avoids disturbing the external piping connections to the module lower pressure vessel head. After removal, cartridges will be transported to an assembly area for packaging in a shipping container prior to being returned to the factory for refurbishment.

Replacement cartridges will be shipped from the factory in special shipping containers equipped with heaters and designed to maintain the requisite fuel cell environment. Assembly of a replacement cartridge into the power plant will follow the reverse of the procedure described below.

Replacement of a fuel cell module cartridge requires, first of all, that the power plant be shutdown, depressurized, and cooled. Following a sufficient period of cooling, the insulation segments at the pressure vessel bolting flange and the bolts can be removed. The pressure vessel dome can then be lifted, using a crane, until its lower flange is level with the cartridge lower support plate, a distance of 44.5 cm (17.5 inches).

While continuing to support the vessel dome from the crane, temporary bracing is deployed to ensure safe working conditions while lifting fixtures attaching the vessel lower flange to the cartridge lower support plate are bolted into position. After the vessel dome is securely bolted to the cartridge, the temporary bracing can be removed and work started to separate the cartridge from its connections to the module lower pressure vessel head. This requires the separation of electrical power takeoff leads, acid feed, drain and vent lines, heater power supply leads, temperature monitors, and cartridge holddown bolts. Any diagnostic or control instrumentation provided in the module must also be disconnected. The cartridge and vessel dome assembly can then be removed and transferred to an assembly/disassembly room for preparation for shipment to the factory. During transfer from the plant to the assembly/ disassembly room, cartridge heating can be maintained, if required, by the use of a transportable power source with temporary heater power and temperature monitor connections.

In the assembly/disassembly room, preparation for shipment to the factory will normally involve removing the pressure vessel dome from the cartridge and transferring it, together with its lifting fixtures, to the replacement cartridge. The spent cartridge is then placed in a shipping container for transportation by truck. (The replacement cartridge will have been delivered to the plant site in such a shipping container.) Transfer of the replacement

cartridge and pressure vessel dome from the assembly/disassembly room to the plant follows the reverse of the procedure described above.

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The shipping container envisaged for use in shipping cartridges to and from the power plant has not yet been designed. The design of the shipping container must allow the cartridge to be heated and maintained in a suitably "inert" environment during shipment. In addition, the shipping container must protect the cartridge from handling damage during shipment and must be designed to minimize the volume and weight of the shipment.

#### 3.7 Fuel Cell Materials and Component Characterization and Testing (WBS 1103-04)

This section addresses the characterization of the various raw materials utilized in the manufacture of repeating cell components, as well as the characterization of repeating cell components, including chemical, physical, mechanical, electrical and corrosion behavior.

#### 3.7.1 Raw Material Specifications

The Purchasing Department Specifications (PDS) previously prepared were utilized for the procurement of raw materials for repeating component manufacture. Additional data were accumulated and will be utilized in future revisions of the PDSs as required.

#### 3.7.2 Plate Materials

This section summarizes the characterization effort devoted to bipolar and cooling plates and the raw materials used in their formulation.

#### 3.7.2.1 Plate Evaluation

Emphasis was placed on the evaluation of "Zee-Zee" bipolar plates produced primarily from graphite or carbon powders exhibiting approximately equiaxed particle shapes, namely Asbury Regrind A99, Asbury Regrind 4421 and Stackpole

;

MF958 powder. These three materials produce plates which are generally flat to within less than 0.8 mm (0.030 inch).

Plates were heat treated to 900°C (1650°F) and those evaluated are identified in Table 3-31.

The in-plane electrical resistivity and mercury porosimetry for a number of these plates are given in Tables 3-32 and 3-33.

Based on the corrosion data obtained by ERC, it was concluded that additional heat treatment to higher temperatures would be required for adequate pressurized cell performance. Sections of larger plates previously heat treated to 900°C (1650°F) were heat treated to both 1500°C and 2700°C (2732°F and 4900°F) for evaluation of selected mechanical and physical properties and for autoclave corrosion studies. Typical mercury porosimetry data for the two regrind type materials and Stackpole MF958 powder plate sections are given in Table 3-34. It is to be noted that the porosity of the three materials increased from 3.6 to 5.0 percent after 900°C (1650°F) heat treatment to between 11.0 and 14.7 percent after 2700°C (4900°F) heat treatment.

Weight loss data for ZFA64, TZA27 and TZA30 resulting from the second heat treatment cycle to 2700°C (4900°F) were determined. The data indicate a weight loss ranging from 1.2 to 3.2 w/o.

Flexural strengths at room temperature were determined for the three materials after heat treatment at 900, 1500, and 2700°C (1650, 2732, and 4900°F). The average data are plotted in Figure 3-30, together with reference data on A99 (33 w/o resin) material. After the 900°C (1650°F) heat treatment, the regrind A99 material had the lowest flexural strength while the Stackpole MF958 (33 w/o resin, dry mix) plate material had the highest. Heat treatment at 1500°C (2732°F) resulted in a small strength increase in the three materials while 2700°C (4900°F) heat treatment resulted in a strength decrease, particularly for the MF958 powder plate material.

# TABLE 3-31 DEVELOPMENTAL PLATE COMPOSITIONS

Plate <u>No.</u>	Type Graphite	Mix	Comments
ZZ293	Poco PXB-5Q-325	Drý	
ZZ299	Airco Speer Texaco Wilmington	Dry	
ZZ300	Stackpole MF958	Dry	·
ZZ302	A99 Regrind (-40)	Dry	
ZZ330	A99 Regrind	Dry	Regrind prepared from cold pressed material.
ZZ332	A99	Dry	
ZZ341	Stackpole MF958	Dry	
ZZ346	A99/MF958	Dry	50 w/o of each powder.
ZZ367	A99 Regrind (-40)	Dry	
ZZ404	Stackpole MF1001	Dry	
ZZ405	Desulco 9026	Ũry	
ZZ411	A99	Dry	
ZZ414	A99	Dry	
ZZ419	Asbury 4421	Dry	
ZZ464	Asbury 4421 Regrind	Dry	5.7 day heat treatment.
ZZ465	Asbury 4421 Regrind	Dry	10.7 day heat treatment.
ZFA64	Asbury 4421 Regrind	Dry	
TZA27	Stackpole MF958 (Lot 1)	Dry	First powder lot.
TZA30	A99 Regrind	Dry	
ŻZ539	Stackpole MF958	Dry	Second powder lot.
22597	Stackpole MF958	Dry	Second powder lot.
ZZ608	Stackpole MF958	Dry	Second powder lot.
ZZ612	Stackpole MF958	Dry	Second powder lot.
ZZ642	Stackpole MF958	Wet	Second powder lot.
ZZ643	Stackpole MF958	Wet	Second powder lot.
ZZ650	Regrind A99	Dry	-40 mesh.
ZZ660	Regrind A99	Ďry	-40 mesh.
ZZ660 [°]	Regrind A99	Dry	-40 mesh.
## TABLE 3-32

# IN-PLANE ELECTRICAL RESISTIVITY OF DEVELOPMENT PLATES AFTER 900°C HEAT TREATMENT

Plate No.(1)	Resistivity ρ Plane (mΩ-cm)	Resistivity(2) ρ Thickness (mΩ-cm)	Ratio(2) Pp/Pt
ZZ293	2.98	·-	-
ZZ299	2.01	-	-
ZZ300	2.12	-	-
ZZ302	3.42	-	-
ZZ330	3.07	-	-
ZZ332	1.53	-	-
ZZ341	2.00	-	
ZZ346	1.78	-	-
ZZ367	3.26	-	-
ZZ404	1.69	-	<b>-</b> '
ZZ405	2.37	-	-

(1)See Table 3-31 for compositions, etc. (2)Data to be collected.

## TABLE 3-33

MERCURY POROSIMETRY DATA FOR DEVELOPMENT PLATES HEAT TREATED TO 900°C

Plate No.(T)	Skeletal Density (g/cm ³ )	Bulk Densiţy (g/cm ³ )	Specific Pore Yolume (cm ³ /g)	Specific Pore Surface (m²/g)	Porosity (Percent)
ZZ293	2.15	2.11	0.010	4.7	2.1
ZZ299	2.18	2.11	0.014	7.1	2.9
ZZ300	2.18	2.10	0.016	8.9	3.4
ZZ302	2.10	1.94	0.039	17.6	7.5
ZZ330	2.08	1.91	0.042	21.6	8.0
ZZ332	2.24	2.11	0.027	14.4	5.7
ZZ341	2.23	2.13	0.021	10.9	4.4
ZZ346	2.23	2.11	0.026	14.1	5.5
ZZ367	2.08	1.98	0.025	12.7	5.0
ZZ404	2.25	2.06	0.040	20.6	8.3
ZZ405	2.10	1.78	0.084	23.5	15.0
ZZ411	1.95	1.87	0.024	11.4	4.5
ZZ414	1.97	1.86	0.030	14.3	5.6
ZZ419	1.97	1.91	0.017	7.8	3.2
ZZ464	1.83	1.76	0.022	11.9	4.1
ZZ465	1.83	1.77	0.020	10.8	3.6

(1)See Table 3-31 for compositions, etc.

# TABLE 3-34

# MERCURY POROSIMETRY DATA FOR SELECTED DEVELOPMENTAL PLATES HEAT TREATED TO 2700°C

Plate) No. Percent	Skeletal Densiţy (g/cm³)	Bulk Densiţy (g/cm³)	Specific Pore Yolume (cm/g)	Specific Pore Surface (m²/g)	Porosity	
ZZ341	2.10	1.87	0.059	25.3	11.0	•
ZZ367	2.09	1.83	0.069	24.4	12.6	
ZZ465	2.07	1.76	0.083	33.3	14.7	
Subscale End Plate (A99 Graphite + 33% Resin)	2.14	1.73	0.109	27.3	18.9	

(1)See Table 3-31 for composition, etc.

3600 (42% TOTAL RESIN) (42% TOTAL RESIN 3300 STACKPOLE MF958 + 33% RESIN 2 A99 + 30% RESIN (REF. (REF. 3000 Figure 3-30. Flexural Strength Vs. Heat Treatment Temperature A99 + 30% RESIN 2700 REGRIND 4421 REGRIND A99 HEAT TREATMENT TEMP., DEG. C 2400 ll łI lİ łt H 2100 ۰ × C 1800 1500 1200 ピ X

FLEXURAL STRENGTH . MPd

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The in-plane electrical resistivity of the three materials was determined and is provided in Figure 3-31. As would be expected, the resistivity decreased with increased heat treatment temperature.

Shrinkage data were obtained on small samples of Stackpole MF958 plate material heat treated to  $2700^{\circ}C$  (4900°F) and ranged from 0.5 to 0.8 percent. The accuracy of these measurements is limited and more reliable data will be obtained from full size  $2700^{\circ}C$  (4900°F) heat treated plates.

Helium permeability testing performed on small (4.5 cm (1.8 in.) diameter) samples of Zee-Zee plates heat treated to both 900°C and 2700°C (1650°F and 4900°F) indicated a very low permeability of 8 x  $10^{-8}$  cm³/sec (0.48 x  $10^{-8}$  in.³/sec) or less for Stackpole MF958 material, while the regrind A99 material gave values of about 6 x  $10^{-6}$  cm³/sec (0.36 x  $10^{-6}$  in.³/sec).

Based on the screening tests performed to date, the most promising graphite powder for production of large flat as-pressed plates is Stackpole MF958. Additional samples of this material were heat treated to 900°C and 2700°C (1650°F and 4900°F) for flexural and electrical resistivity measurements. The data obtained are given in Tables 3-35 and 3-36 and are in relatively good agreement with the screening data presented earlier.

The backup to this material will most likely be the Asbury Regrind A99 plate material. Several additional plate samples were evaluated after 900°C (1650°F) heat treatment (ZZ650, 660 and 666). The average flexural strength was 53.2 MPa (7,700 psi) and ranged from 48.5 to 55.6 MPa (7,000 to 8,100 psi), while the in-plane electrical resistance for the same samples ranged from 2.99 to 3.17 mΩ; the average was 3.08 mΩ.

#### 3.7.2.2 Plate Corrosion

Autoclave corrosion testing of the three principal candidate plate materials, Stackpole MF958 (33 w/o resin), Asbury Regrind A99 ( 49  w/o resin) and Asbury Regrind 4421 ( 49  w/o resin) was performed in 97 w/o  $H_3PO_A$  at



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TABLE 3-35 FLEXURAL STRENGTH OF BIPOLAR PLATE MATERIAL

laterial	<u>Heat</u> Treatment	Ave Fle	rage xural ength	Rar Str	nge of exural rength
Plate Nos.)	( <u>)</u> )	MPa	Psi	MPa	Ps i
1F958 (new powder lot) 13 w/o Resin Drv Mived	006	77.2	11,200	75.8 to 78.9	11,000 to 11,450
(ZZ 589, 597)	2700	47.5	6,900	36.5 to 54.8	5,300 to 7,950
1F958 (new powder lot) 16 w/o Pasin Drv Mivad	006	74.4	10,800	69.6 to 83.4	10,100 to 12,100
(ZZ 608, 612)	2700	52.4	7,600	49.3 to 55.1	7,150 to 8,000
1F958 (new powder lot) 13 w/o Basin Wat Mived	006	68.9	10,000	59.9 to 74.4	8,700 to 10,800
(ZZ 642, 643)	2700	45.1	6,550	39.9 to 52.0	5,800 to 7,550

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TABLE 3-36 IN-PLANE ELECTRICAL RESISTIVITY OF BIPOLAR PLATE MATERIALS

Material (Plate Nos.)	Heat Treatment (°C)	Average Resistivity (mûrcm)
MF958 (new powder lot) 33 w/o Resin Dry Mixed (77 580 597)	900 2700	1.90
MF958 (new powder lot) 36 w/o Resin, Dry Mixed (ZZ 608, 612)	900 2700	2.01
MF958 (new powder lot) 33 w/o Resin, Wet Mixed (22 642, 643)	900 2700	1.98 1.47

190°C (374°F) and 485 kPa (4.8 atms) after heat treatment at 900, 1500, and 2700°C (1650, 2732, and 4900°F). The dynamic hydrogen electrode (DHE), used as a reference electrode, was polarized at 0.1 mA/cm² (0.09 A/ft²) and exhibited a stable potential of approximately 22 mV below the reversible hydrogen electrode (RHE). After each corrosion test, the density of the acid was determined and showed a change of less than 0.5 w/o.

The corrosion potential-current density relationships (using one side of specimen geometric surface area) for the 900°C and 2700°C (1650°F and 4900°F) heat treated samples are shown in Figure 3-32. Relatively linear Tafel slope regions over two to three decades were observed for the samples tested. The 2700°C (4900°F) heat treated samples exhibited slopes of from 57 to 85 mV/decade which were somewhat lower than those of the 900°C (1650°F) material which gave slopes of from 116 to 145 mV/decade.

Table 3-37 summarizes the corrosion data at 0.8 and 0.9 volts versus RHE. At similar test conditions, the 900°C (1650°F) heat treated plate material made from Stackpole MF958 powder (33 w/o resin) gave a slightly lower corrosion current density than the two regrind materials ( 42  w/o resin), presumably due to the lower resin content of the former. At 0.8 V versus RHE, the 2700°C (4900°F) heat treated materials exhibited corrosion currents of 0.0017 mA/cm² (0.0015 A/ft²) or lower while for the 900°C (1650°F) material, corrosion currents ranged from 0.17 to 0.77 mA/cm² (0.7 A/ft²). Increasing the heat treatment temperature from 900°C to 2700°C (1650°F to 4900°F) thus reduced the corrosion current density at 0.8 V versus RHE by at least two orders of magnitude while the 1500°C (2732°F) heat treated material showed only about a factor of two reduction.

After corrosion testing, samples were gently washed in hot water to remove acid residue, vacuum dried, mounted and polished for optical microscopic examination. Microstructures of the corrosion surface cross section are shown in Figure 3-33 for the Stackpole MF958 plate material. It is noted that the carbon phase resulting from pyrolysis of the resin binder was selectively and severely corroded in the 900°C (1650°F) heat treated plate resulting in a



Figure 3-32. Corrosion Current Vs. Corrosion Potential for Heat Treated Plate Materials

TABLE 3-37

CORROSION RATE OF CANDIDATE BIPOLAR PLATE MATERIALS

(Experimental Conditions: 190°C, 4.8 atm, 97 w/o H₃PO₄)

Raw Materials	Sample ID	Heat Treatment Temperature, °C	Corrosion Current at 0.8 V, mÄ/cm ²	Corrosion Current at 0.9 V, mA/cm ²
A-99 Regrind	ZZ367R	006	0.77	3.6
A-99 Regrind	TZA-30	1500	0.34	1.06
A-99 Regrind	ZZ367	2700	0.0010*	0.057
Stackpole MF958	ZZ341R	006	0.17	1.4
Stackpole MF958	TZA-27	1500	. 0.12	0.66
Stackpole MF958	ZZ341	2700	0.0017	0.064
Asbury 4421 Regrind	ZZ465R	006	0.70	2.2
Asbury 4421 Regrind	ZFA-64	1500	0.30	1.26
Asbury 4421 Regrind	ZZ465	2700	0.0015*	0.020
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Value obtained by extrapolation of

*No anodic corrosion current was observed at 0.8 V. higher voltage data to 0.8 V.

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## BIPOLAR PLATE MATERIALS AFTER CORROSION TESTING



Stackpole MF958 Graphite, 900°C Heat Treatment

200X



Stackpole MF958 Graphite, 2700°C Heat Treatment

200X



surface layer of relatively large loose particles similar to the Stackpole MF958 graphite powder used in preparation of these plates. The 2700°C (4900°F) heat treated sample shows relatively uniform surface attack with little or no selective attack of the carbon binder phase apparent. Metallography of the Asbury Regrind A99 and Asbury Regrind 4421 samples revealed similar corrosion behavior. The appearance of the 900°C (1650°F) samples agrees with the observations made previously on samples of plate corrosion from pressure-tested stacks and subscale tests - namely, that the resin carbon phase of the 900°C (1650°F) heat treated bipolar plates seems to be selectively corroded while the graphite powder particle shows very little attack.

Limited autoclave corrosion data were obtained on bipolar plate materials in 93 w/o phosphoric acid at 190°C ( $374^{\circ}F$ ) and 690 kPa (100 psia). Plate samples were heat treated for times up to 100 hours at 900, 1100, and 1200°C (1650, 2000, and 2200°F). These tests indicate that no improvement in corrosion behavior was obtained from longer heat treatment times at 900°C ( $1650^{\circ}F$ ) (4 versus 100 hours) and that only limited improvement is achieved at temperatures of 1100°C and 1200°C ( $2000^{\circ}F$  and  $2200^{\circ}F$ ) for 100 hours (see Figure 3-34). These tests confirm the conclusion that heat treatment times at temperature has the major impact on corrosion behavior, while longer times at temperature seamined.

#### 3.7.2.3 Plate Raw Materials

Particle size distribution for a number of graphite powders utilized in plate production, including as-received samples of A99, 4421, PXB-5Q and MF958 (two lots) as well as reground A99 (-40 mesh) and reground 4421, was determined.

Samples of Varcum 29-703 were heat treated to 900, 1500, and 2700°C (1650, 2732, and 4900°F) to provide samples of the binder resin carbon for X-ray diffraction analysis. X-ray diffraction profiles show a contraction in the C direction (002) from 3.885 to 3.472 Å coupled with an increase of about



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2 times in the LOO2 crystallite size as a result of the  $2700^{\circ}C$  (4900°F) heat treatment, see Table 3-38. Typical diffraction profiles at 900, 1500, and 2700°C (1650, 2237, and 4900°F) are shown in Figure 3-35. Selected samples were used for determination of resin carbon weight loss resulting from additional heat treatments to temperatures above 900°C (1650°F); at 1600°C (2900°C) the loss was 1.3 w/o while at 2700°C (4900°F) it was 1.6 w/o.

#### 3.7.3 Seal Materials

The stack design relies on effective edge cell-to-cell seals as well as manifold-to-stack seals. Seal tests were conducted to simulate the seal configuration planned for the 10 and 25 kW stacks. Selection of the materials tested was based on several specific parameters such as gas sealability, structural stiffness, acid corrosion resistance, compression set, and stability at 200°C (392°F). The commercial availability of stock sheet forms is an important consideration; therefore, the thickness of samples tested are not necessarily the thickness for the stack application. Generally, gas sealability, structural stiffness, short-term compression set, and operating temperature stability parameters are exhibited via a sealing test at ambient temperature and 200°C (392°F) . The general seal test configuration is shown on Figure 3-36. A flowmeter capable of detecting flow rates as low as 0.003  $cm^3/min$  (0.011 in.³/h) air is placed in series with the gas inlet to provide a more quantitative measure of seal leakage. The testing was done on a stack of five or six picture frame seals attached to heat treated  $10 \times 10$  cm (4 x 4 in.) graphite/resin plates on one surface with Viton cement. A linear variable displacement transducer (LVDT) is used to define the seal displacement as a function of time. The load and displacement values are recorded continuously on a strip recorder to evaluate the structural characteristics. Dry nitrogen gas was used to internally pressurize the seal stack to either 108 kPa (1 psig) or 135.5 kPa (5 psig). Usually, two stacks were used to evaluate the ambient temperature and operating temperature, 200°C (392°F), conditions.

The perfluoroelastomer Kalrez was tested at both ambient and operating temperatures. This material was tested in the standard dry seal configuration

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H.T. Temperature °C	d ₀₀₂ /Å	a ₁₀₀ /Å	L ₀₀₂ /Å	L ₁₀₀ /Å
900	3.885	2.06	9	18
1500	3.667	2.08	11	27
1800	3.645	2.08	13	28
2200	3.535	2.08	15	34
2700	3.472	2.10	20	48
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TABLE 3-38 CRYSTALLITE SIZES AND LATTICE SPACINGS OF HEAT TREATED VARCUM 29-703 RESIN

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using no acid. The Kalrez tested, Compound 1-58, was 60 durometer Shore hardness. The Kalrez seal stack was cycled four times at ambient temperature to compressive strains of 10.3 to 28.9 percent producing tangent elastic moduli values of 9.85 MPa (1428.6 psi) to 16.01 MPa (2321.4 psi). During the first cycle, the seal stack was leak tested at 10.5 percent strain. The result was corner leakage at a pressure differential of 108 kPa (1.0 psig) of nitrogen gas. On succeeding cycles, the seal stack was compressed to progressively higher strains and the corner leak became nonexistent even at an internal pressurization of 446 kPa (50 psig) of nitrogen gas.

Ambient temperature stress-vs-strain cycle data indicate a good strain range, 0-22 percent, with a maximum seal compression pressure of 1.67 MPa (240 psi) with relatively linear behavior beyond the 3 percent strain value. The modulus of elasticity values for Kalrez is similar to Viton with tangent values equivalent to 10.4 MPa (1500 psi). A summary of these modulus of elasticity values is shown on Table 3-39 for the various loading cycles. This table also provides a comparison between the Kalrez and the Viton values for similar conditions.

The operating temperature test cycles are shown on Figure 3-37. These cycles indicate a consistent tangent modulus of elasticity of 6.894 MPa (1000 psi) over the seal strain range of 4 to 20 percent. This also indicates that the material becomes softer at temperature. Again the modulus of elasticity values are linear and repeatable.

No evidence of seal leakage could be detected for an internal nitrogen pressurization of 135.5 kPa (5.0 psig) at both test temperatures. The minimum seal contact pressure during these leak tests was 1.554 MPa (225.4 psi) at ambient temperature and 815.6 kPa (118.3 psi) at operating temperature. Kalrez seals also demonstrated the ability to withstand internal pressurizations of up to 446 kPa (50 psig) at both ambient and operating temperature with no leakage.

**TABLE 3-39** 

DEFORMATION COMPRESSION TEST DATA SUMMARY OF LOAD VS.

÷.: Chord Modulus of Elasticity MPa (psi)x10³ 8.00 7.10 (1.03) 7.79 9.86 (1.43) 11.79 9.79 (1.42) 5.72 (0.83) 9.17 (1.33) 9.17 (1.33) 14.89 (2.16) 7.52 (1.09) 6.89 (1.00) E=v/€ E = V c Tangent Modulus of Elasticity MPa (psi)x10³ 9.86 (1.43) 5.31 (0.77) 7.86 (1.14) 7.03 7.03 8.27 (1.20) 16.00 (2.32) 10.96 6.83 (0.99) 13.10 4.83 10.41 Temp. of Test °C (°F) 201.7 190 (374) 190⁻ (374) 190 (374) 190 (374) 190 (374) 22.2 22.2 22.2 (72) 22.22(72) 22.2 22.2 ٩ Total Strain (cm/cm ( in/in) খা " " .2049 .1516 .1025 . 1516 .2016 .2020 .2020 .2898 .2131 .2004 .195 20 .0370) .0940[.] (.0370) .1250 (.0492) .1252 (.0493) .1252 (.0493) in cm (in) .1242 (.0489) AL Change .127 (.050) .183 (.072) .0635 .188 (.074) .1796 .1321 L⁽¹⁾ Height of Test Article .620 (.244) .620 .620 (.244) .620 .620 (.244) .620 .620 (.244) .620 (.244) .620 (.244) .370) .940 (.370) .620 (.244) <u>ب</u> ت Specimens per Test Article z 823.37 (119.42) 892.60 (129.46) 1800.63 (261.16) 3044.17 (441.52) 1085.03 (157.37) 1048.07 (152.01) 1046.56 (151.79) 3047.21 (441.96) Maximum Stress Applied v = <u>Load</u> Load Årea Newtons KPa (1b) (psi) 1662.12 (241.07) 801.79 (116.29) 1474.1 (213.8) 1379.0 1189.9 (267.5) 2668.9 (600.0) 2602.2 (585.0) 1290.0 1158.8 (260.5) 1568.8 (352.5) 1514.6 (340.5) 1512.4 (340.0) 4403.7 (990.0) 3647.5 (820.0) 4399.3 (989.0) 2402.2 (540.0) Area Nominal cm² (in²) 24.80 (3.844) 19.35 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) 14.45 (2.24) Test Contact Description Surface Flat Stack 2 Cycle 3 Stack 2 Cycle 4 Stack 2 Cycle 5 Stack 2 Cycle 6 (2) Cycle 1 (2,3) Cycle 4 Cycle 1 Cycle 2 Cycle 3 Stack 2 Cycle 1 20 Cycle / Stack Cycle Stack <u>۵</u> Seal Material 2 Kalrez Viton

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obtained from previously conducted tests and documented in Reference 3. had been previously cycled according to the following sequence:

preload applied

ambient temperature ur. soak at 201.7°C (395°F)

(1)Measured with a 5 lb. (2.23 psi) prelo. (2)This data was obtained from previously (3)Viton stack 3 had been previously cycli Cycle 1 - compression cycle at ambient Cycle 2 - (thermal cycle) - 5-hr. soak Cycle 3 - compression cycle at ambient

ambient temperature.



Compressive Seal Stress Vs. Seal Strain for Kalrez Seal Configuration at Operating Temperature (190°C) Figure 3-37.

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Another key aspect of the mechanical testing is that of permanent set at both ambient and operating temperatures. The results of this study indicated after one cycle at ambient temperature Kalrez exhibited a 2.75 percent permanent compression set, and after four cycles permanent compression set of 4.8 percent occurred. Kalrez demonstrated larger compression sets of 5.8 percent and 7.8 percent after one and four cycles at operating temperature. These values are consistent with other elastomers which have been tested.

As part of this study, a comparison of material characteristics was made between Kalrez and Viton. A comparison of modulus of elasticity values at ambient temperature, operating temperature, and finally at ambient temperature after being at 193°C (380°F) for 'a period of time is shown on Figure 3-38. It was found that the Kalrez seals were slightly stiffer than Viton seals at both ambient and operating temperatures where the elastic moduli of Viton are 8270 kPa (1200 psi) and 4830 kPa (700 psi), respectively. Kalrez seals incurred a smaller permanent compression set after four cycles at ambient temperature than did Viton seals. Kalrez demonstrated a permanent compression set of only 16.5 percent of the maximum compressive strain developed during testing; Viton exhibited a permanent set of 51.3 percent of the maximum compressive strain developed during its testing. Both materials demonstrated good sealing capabilities under compressive loads normally encountered and with internal pressurization of 135.5 kPa (5.0 psig).

A short compression/seal leakage test stack made of Armalon seal material was assembled and tested. The Teflon coated fiber material, which indicated acceptable acid resistance, was tested at ambient temperature conditions. The dry stack was compressed and leak tested at three compression levels of 10, 15, and 20 percent strain. At all three loading conditions, the seals showed leakage at 135.5 kPa (5 psig) internal nitrogen pressurization. This leakage decreased at the higher compression strains. The exact source of this leakage could not be located due to a limitation of the size of the test unit and the test assembly. The material exhibited large permanent strains and very nonlinear behavior at ambient temperature conditions. Even though the tangent modulus values are on the order of 13.79 MPa (2000 psi), the large permanent ORIGINAL PAGE 15 OF POOR QUALITY



PERCENT STRAIN

Figure 3-38.

Compressive Stress Vs. Seal Strain for Viton/Viton Cement and Kalrez/Viton Cement Seal Comparative Tests strains (95 percent of maximum strain obtained) indicate a lack of compliance. This lack of compliance of the material has precluded consideration of Armalon as a viable candidate seal material.

Mechanical testing of candidate seal material Kel-F was completed for both ambient temperature and operating temperature of 198°C (390°F). This seal material has similar elastomer characteristics to the Viton compound. Ambient temperature tests were conducted using four compression cycles for seal strain values ranging from 15 to 30 percent. The Kel-F exhibited tangent modulus of elasticity values on the order of 8.67 MPa to 10.45 MPa (1258 psi to 1515 psi). The original hardness of this seal was measured as 61 Shore durometer. The seal material exhibited a relatively small amount of permanent strain (3 percent) after two cycles to a maximum of 20 percent strain. The test unit indicated good sealing characteristics over the range of 108.6 kPa to 135.5 kPa (1.1 psig to 5 psig). A 10 percent strain working range is available for sealing once the seal has been loaded to 30 percent strain. At ambient temperature, no evidence of seal leakage could be detected for compressive stresses greater than 853 kPa (109 psi) under an internal pressure of 135.5 kPa (5 psig). The Kel-F seal material also demonstrated the ability to withstand internal pressures of up to 445 kPa (50 psig) under a normal compressive stress of 2.5 MPa (352 psi). Since the Kel-F material exhibited excellent seal characteristics at ambient temperatures, no leak tests were performed at operating temperatures.

A second Kel-F mechanical test performed at 198°C (390°F) was completed and the material exhibited relatively linear characteristics over the range of 4 percent to 20 percent strain. The Kel-F material did not soften during the 198°C (390°F) test phase. The increase in modulus of elasticity values for first cycle loading was on the order of 33 percent. The modulus of elasticity value changed from 8.674 MPa (1258 psi) to 11.69 MPa (1696 psi) for the ambient to operating temperature conditions. It should be noted for both ambient and operating temperature conditions, the amount of permanent strain developed was on the order of 4 percent after the seals had been compressed to 15 percent and 20 percent strain values. This is considerably less permanent strain than any Viton compound previously tested.

The seal stiffening effect after exposure to high temperature was also evaluated during this test phase. The evaluation indicated the Kel-F material increased in stiffness from 11.610 MPa (1684 psi) at 198°C (390°F) to 19.305 MPa (2800 psi) at ambient temperature after exposure at 198°C (390°F) for ^40 hours. This stiffness increase is consistent with Viton type elastomers.

Compression creep tests of the Viton edge seal material were initiated. Four units are currently being tested at the following conditions:

Test <u>Condition</u>	Test Unit <u>No. 1</u>	Test Unit <u>No. 2</u>	Test Unit <u>No. 3</u>	Test Unit <u>No. 4</u>
Temperature	149°C/300°F	198°C/390°F	198°C/390°F	54°C/130°F
Compressive Pressure (using Constant Load)	1.72 MPa/ 250 psi	0.689 MPa/ 100 psi	1.72 MPa/ 250 psi	2.76 MPa/ 400 psi
Test Time as of 6/1/84	400 hours	1000 hours	1600 hours	400 hours

Each test unit consists of five layers of Viton seal material separated by graphite disks, Figure 3-36. The specimens were loaded, and deflection versus time data are presently being recorded. Each specimen will be tested until a sustained creep limit has been reached. Documentation of the test results will be presented upon completion of the entire test.

Assessment of acid exposure of various fluoroelastomer material being considered for cell to cell and manifold to stack seals was initiated. Samples, as received, were leached in 94 w/o acid for 150 hours at 185°C (365°F), and ion chromatography (Dionex Model 14 Ion Chromatograph) was performed on the resultant acid after filtering through 8000 Å opening Millipore filter and dilution to 47 w/o  $H_3PO_4$ . An  $H_3PO_4$  sample was run in the Teflon beaker used for leaching to establish the blank. In addition, weight gain/loss tests were performed on the various samples. Test results, Table 3-40, indicate rather substantial removal of F⁻ and  $SO_4^{2-}$  from the samples as a result of leaching. Selected samples were re-leached and relatively minor TABLE 3-40

LEACHING EFFECT ON SEVERAL AS-RECEIVED FLUORDELASTOMER MATERIALS

(Leached in 94% H₃PO4 at 185°C after 150 hours. Ion chromatography used for analysis. Samples washed to neutral and dried in vacuo before weighing.)

		Wt. Change	ANALYSI	S. PPM	BY WEIG	TH	
Sample	Description	(%)	и. 1	c1 -	N0 ⁷	504	Comments
VMOAC-1	Viton 10125 Cured 164°C, 3 hrs	+ 6.1	890	2	8	212	Electrolyte black
VMDAC-2	Viton 10149 Cured 164°C, 3 hrs	+ 11.4	006	ъ	5	112	Electrolyte black
KALAR-1	Kalrez K00506	+ 0.08	712	9	4	001	Electrolyte straw colored
KLFAR-1	KEL-F 06-3655 As received	+ 0.4	512	m	ω	112	Electrolyte amber
VPAAR-1	Pelmor	+ 7.3	460	e	2	252	Light straw
VPRAR-1	Precision Rubber No. 42679	- 0.5	172	60	` ى	112	Acid dark brown
V MÚAR – 1	MOSITE	+ 33.1	752	10	9	106	Electrolyte black
GRAY VITON	Pelmor, PLV-10059	+ 4.2	316	9	5	00 L	Electrolyte black
BLACK VITON	Precision Rubber No. 19356	+ 3.5	418	Ξ	8	144	Electrolyte black
BLANK	H ₃ PO4 in Teflon			ю	5	4	No color change

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changes in the anionic species levels were noted, Table 3-41. A third group of samples heated at 190°C (374°F) in air for 100 hours prior to leaching show a large increase for several samples in the F⁻ due to prior heat treatment, Table 3-42. These data indicate that all of the fluoroelastomer materials examined as possible seal materials have the potential of being a source of  $F^-$  and  $SO_A^{2-}$  contamination or poison to the cell.

#### 3.7.4 Electrode Materials

Characterization of finished electrodes by half cell measurements was initiated and other evaluations were continued.

#### 3.7.4.1 Design and Optimization of Half Cell Fixture

The design and optimization of a half cell assembly for electrochemical evaluation of individual electrodes (both cathodes and anodes) was accomplished. Several generations of cell designs were evaluated in the optimization process. As shown in Figure 3-39, the final half cell fixture machined from Teflon utilizes standard existing subscale test hardware (compression plate, carbon end plate, current collector, heating element and insulation). A reference DHE was fabricated and calibrated in the half cell over the temperature range of  $18 - 193^{\circ}C$  ( $64 - 380^{\circ}F$ ). At 0.1 mA/cm² ( $0.09 \text{ A/ft}^2$ ), the potential of this reference electrode was measured with respect to a RHE in the same electrolyte. Preliminary tests were run in this cell to identify optimum testing conditions, establish electrode pre-treatment conditions, and to select the electrochemical method for polarization measurements.

## 3.7.4.2 Effect of Sintering Temperature on Cathode Performances: Half Cell Studies

The baseline manufacturing practice requires the fuel cell electrodes be sintered to enhance the Teflon binding strength. The sintering temperature is thought to be one of the key parameters affecting the wetting characteristics

**TABLE 3-42** 

LEACHING EFFECT ON HEAT TREATED FLUOROELASTOMER MATERIALS*

(Leached in 94% H3PO4 on several polymer materials at 190°C for 150 hours. Samples were washed to neutral in hot deionized water and dried in vacuo before weighing.)

		Wt. Change	ANALY	YSIS, F	PM BY WE	IGHT	
Sample	Description	(%)	۲ ۴	c1_	NO ³	50 <mark>4</mark>	Comments
VPR HT-1	Precision Rubber No. 42679	6.1 -	130	Ŕ	Ŕ	52	Electrolyte amber
VPA HT-1	Pelmor	- 21.1	610	Ŕ	₽	156	Electrolyte light amber
VMDAC HT-2	Cured Viton No. 10149	- 7.0	29,500	Ş	Ŕ	320	Sample broken
VMDAC HT-1	Cured Viton No. 10125	ı	5,284	ۍ	Ş	256 small	Sample fractured into fragments
VMO HT-1	MOSITE	+ 17.0	7,300	Ą	Ą	80	
KLF HT-1	KEL-F 06-3655 Compound	1	1,300	\$ S	Ŕ	184 small	Sample fractured into fragments
KAL-HT-1	Kalrez Keo Sol Compound	+ 1.2	80	Ą	€	112	Electrolyte clear

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*HEAT TREATED FOR 100 HOURS AT 190°C IN AIR PRIOR TO LEACHING



of the Teflon-bonded catalyst layer. Electrochemical behaviors of three "oversintered" and two baseline cathode samples were studied in the half cell at 190°C (374°F) and 101 kPa (1 atm). The "oversintered" cathodes were heated an additional 6°C and 12°C (11°F and 22°F).

Throughout the current density region of this study, the polarization data were highly reproducible on the two baseline cathodes. The increased sintering temperature produced a substantial reduction in the achievable electrode potential. At  $200 \text{ mA/cm}^2$  (186 A/ft²), for example, the high temperature sintered cathode exhibited a potential of only 650 mV versus RHE, about 60 mV less than the baseline samples. Furthermore, the measured Tafel slopes also increased slightly with increased sintering temperature.

After polarization at 200 mA/cm² ( $186 \text{ A/ft}^2$ ) for 40 hours or longer, only slight improvements (less than 5 mV) were observed on the two baseline cathodes. The performance of "oversintered" cathodes was significantly improved, but was still 20-30 mV below the baseline level. This result suggests that increasing the sintering temperature improves the hydrophobicity of the catalyst layer and thus retards the penetration of electrolyte into the porous structure. It also raises a question as to whether reducing the sintering temperature below the 360°C ( $680^{\circ}$ F) baseline level would result in improved cathode performance.

#### 3.7.4.3 Half Cell Evaluation on Alternate Electrocatalysts

The electrochemical behaviors of several alternate catalysts were investigated in the half cell apparatus at 190°C (374°F) and atmospheric pressure. Shown in Figure 3-40 is the Tafel plot for oxygen reduction on DC-06 catalyst as compared to two present baseline cathode samples, C234-4 and C249-8. To optimize the catalyst utilization, each electrode was pretested at 200 mA/cm² (186 A/ft²) for approximately 90 hours. The polarization curve of DC-06 catalyst was nearly parallel to those on the baseline cathodes. However, the observed electrode potentials on DC-06 were  $\sim$ 30 mV higher than the baseline cathodes in the current density range of 5-600 mA/cm² (4.7-560 A/ft²). At



Figure 3-40. Tafel Plots for Oxygen Reduction on an Alternate Catalyst and Two Baseline Cathodes

200 mA/cm² (186 A/ft²), for instance, the measured cathode potential on DC-06 was  $\sqrt{740}$  mV versus RHE, as compared to the baseline performance of  $\sqrt{710}$  mV.

Tafel plots for oxygen reduction on alternate catalyst DC-10 and DC-05 are illustrated in Figure 3-41. Electrochemical behavior of DC-10 was quite similar to the two baseline cathodes, showing a Tafel slope of 96 mV/decade. However, DC-05 exhibited slightly better performance than the baseline cathodes. The measured electrode potential was 726 mV versus RHE at 200 mA/cm² (186 A/ft²).

Figure 3-42 shows the polarization curves for hydrogen oxidation on a baseline anode, A212-2, and an alternate anode catalyst, DC-03. The anodic overpotentials increased significantly with increasing current density; consequently, no linear Tafel region was obtainable on either electrode. The alternate anode, DC-03, exhibited lower overpotentials than the baseline anode. At  $200 \text{ mA/cm}^2$  (186 A/ft²), the observed hydrogen overpotential was  $\sim$ 15 mV at the anode A212-2,  $\sim$ 50 percent higher than the literature reference value. This suggests that the platinum catalyst at this baseline anode may have been contaminated by impurities from the Vulcan XC-72 carbon support and/or the wetproofed backing paper.

## 3.7.4.4 Effect of Electrolyte Contamination by Viton Gasket Corrosion Products on the Anode Performance

Half cell tests were performed to investigate the impact of electrolyte contamination from Viton gasket-acid reactions on the anode performance for hydrogen oxidation. The contaminated electrolyte, prepared by immersing and soaking Viton gaskets in 97 w/o H₃PO₄ at 190°C (374°F) and 485 kPa (4.8 atm) for N170 hours, contained 320 ppm F⁻ and 68 ppm SO₄²⁻ as determined by ion chromatography. This contaminated solution was then mixed with various proportions of fresh electrolyte to provide a series of test electrolytes containing a range of contaminant concentrations.



Figure 3-42. Polarization Curves for Hydrogen Oxidation on an Alternate Catalyst and a Baseline Anode
Using pure hydrogen, the equilibrium potential of a typical WAESD-produced anode, A212-2, decreased with increasing concentration of impurities in the electrolyte. At the highest impurity level used (i.e., 320 ppm F⁻ and 68 ppm  $SO_4^{2-}$ ), the equilibrium potential decreased by  $\sim$ 20 mV. In the practical current density region of 200-400 mA/cm² (186-372 A/ft²), however, the contaminated electrolyte resulted in about 5 mV higher polarization potential than that measured for the uncontaminated electrolyte.

### 3.7.4.5 Baseline Catalyst Characterization

Characterization of Lot No. 3 catalyst was performed in both the as-received condition and after heat treatment at 900°C (1650°F). The data is in relatively good agreement with both the vendor data and the PDS requirements. In the case of Lot No. 3, the as-received metal surface area is somewhat larger, 16.3 m²/g (79.6 x  $10^4$  ft²/lb), than that of Lot No. 2, 13.6 m²/g  $(66.4 \times 10^4 \text{ ft}^2/\text{lb})$ , or the PDS value  $(11 + 2 \text{ m}^2/\text{g})$ . Heat treatment of the catalyst to 900°C (1650°F) results primarily in changes in the platinum. The metal surface area decreases from about 190 to 40 - 45  $m^2/g$  (9.28 x  $10^5$ to  $1.95 \times 10^5$  - 2.20 x  $10^5$  ft²/lb) while the surface area of the carbon support remains unchanged. The particle size as determined by both transmission electron microscopy and X-ray line broadening increases from 15 to 20 A to 39 to 44 A as a result of heat treatment. Increases in Fe, Ni and Cr have been noted in the heat treated catalyst; this may be related to either non-uniform distribution of foreign particles in the as-received catalyst or to possible contaminant pickup from the metal trays used to contain the catalyst during heat treatment.

### 3.7.4.6 Compression Behavior of Electrode Materials

Compressibility tests were performed on Stack W-009-09 anode-cathode components and Stack W-009-11 anode-matrix-cathode components. Both sets of components were fabricated using the following procedure: Cathode (900°C (1650°F) heat treated catalyst, dry laminated, argon sintered) and anode (as-received catalyst, dry laminated, argon sintered). However, the Stack 09 components incorporated a thicker SiC layer and higher platinum loadings than the Stack 11 components.

Three assemblies were tested. The first assembly consisted of five layers of Stack 09 anode-cathode components wetted with 1.1 cm³ (0.07 in³) of 98 percent w/o phosphoric acid ( $H_3PO_4$ ) per cell layer. This specimen was subjected to five cycles at 56°C (133°F). The second unit consisted of five layers of dry Stack 09 anode-cathode components and was subjected to four cycles at 56°C (133°F) and two additional cycles at 198°C (390°F). The third specimen consisted of five layers of dry Stack 11 components and was subjected to three cycles at 55°C (130°F), two cycles at 191°C (375°F) and then one cycle at 55°C (130°F). A typical compression cycle was as follows:

- load to 1000 N (225 pounds)
- 30 minute load relaxation
- unload to 8.89 N (2 pounds)

Although each compression cycle followed this general pattern, there was some variation in maximum load and relaxation time. A LVDT and a load cell were used to record displacement and load as a function of time. The data were reduced and evaluated to determine stress-strain relationships, permanent set characteristics and load relaxation effects for use in the stack analyses.

At the operating temperature of 56°C (133°F), the wet test unit with Stack 09 anode-cathode components exhibited a very nonlinear stress-strain curve. In the 0 to 15 percent strain range, the material was extremely soft as indicated by a low modulus of elasticity. The 15 to 19 percent strain range is a transition region which was strongly nonlinear. From 19 percent to the maximum strain range, it is characterized by relatively stiff, linear behavior. After the third compression cycle, the stress-strain curves demonstrated good repeatability. In the range from 19 to 22 percent strain, the average modulus of elasticity (based on nominal area) for cycles 3, 4 and 5 was 5.49 MPa (796 psi). After five cycles, the wet specimen incurred a 10 percent permanent

set. This permanent set represents 43 percent of the maximum strain developed during the compression loading. During the 30 minute relaxation periods, this specimen showed a load relaxation of 10 to 20 percent of the initial load. Although this relaxation did not reach a steady state value, the amount of additional relaxation for longer time periods would be relatively small.

The structural properties of the dry W-009-09 anode-cathode components at 56°C (133°F) exhibited a strong correlation to the wet structural properties. Although the permanent set values were slightly lower, the modulus of elasticity and load relaxation effects were nearly the same for both test assemblies. Overall, the phosphoric acid had a negligible effect on the structural properties of the W-009-09 anode-cathode components.

At 190°C (374°F) the Stack 09 components appeared significantly softer than at 56°C (133°F). Additional tests are being considered to fully determine the structural properties of the Stack 09 components at this elevated temperature.

At the operating temperature of 55°C (130°F), the Stack W-009-11 anode-matrixcathode components also exhibited a very nonlinear stress-strain curve. For example, the cycle 2 test data indicate a modulus of elasticity (based on nominal area) of 117 kPa (17.0 psi) in the 0 to 10.6 percent strain range and a modulus of elasticity of 4.09 MPa (593 psi) in the 15.0 to 18.0 percent strain range. At 191°C (376°F) the Stack 11 components exhibited a fairly linear and repeatable stress-strain curve. The average tangent modulus of elasticity at this temperature was 6.45 MPa (936 psi). After three cycles, Stack 11 specimen accumulated a constant permanent set of 11.5 percent strain. No additional permanent set is expected for a higher number of cycles.

Figure 3-43 compares the stress-strain behavior of the Stack 09 components, the Stack 11 components and the standard anode-matrix-cathode components. (Standard components refer to the electrodes used in Stacks W-009-01 through W-009-08). Note the Stack 09 components and Stack 11 components were significantly softer than the standard components. This difference was primarily due to the soft behavior of the Stack 09 and Stack 11 components in



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Strain Comparison Between Standard and Stacks W009-09 and -11 Cell Components

Figure 3-43.

the low strain range. The increased stiffness of the Stack 11 components as compared to the Stack 09 electrodes is the result of the thinner SiC layer and reduced loadings on the Stack 11 cell components.

The primary design impact of the softer Stack 09 and Stack 11 electrodes occurs with the stress distribution in the cell. Due to the soft electrode material, the stresses will tend to redistribute around the outside seal regions. This redistribution is being investigated in the nine cell stack finite element models.

3.7.4.7 Silicon Carbide Layers and Powder

Helium bubble pressures were determined for various thickness layers of silicon carbide (all acid-filled for 15 minutes with 98.4 w/o  $H_3PO_4$ ) from Cathode C204-4. The ambient temperature data obtained are tabulated in Table 3-43. There is no apparent relationship of bubble pressure with coating thickness. The bubble pressure varies from 89.7 kPa (13 psi) to 165.6 kPa (24 psi). These data are in agreement with previously reported bubble pressures for acid filled silicon carbide layers.

An all-Teflon Gelman in-line filter holder was modified to perform elevated temperature bubble pressures up to 230°C (446°F). Data were obtained over the range from 190°C to 228°C (374°F to 442°F) and are given in Table 3-44. The bubble pressure is somewhat lower than the ambient temperature average data, but not below 65 kPa (9.5 psi).

A second lot of SiC used for cathode insulation layer was evaluated. Comparison of the particle size distribution of this lot of material with that of the previous lot indicates the two lots of powder have relatively similar particle distribution.

Bubble pressure was determined at ambient temperature for cathode C214-2 fabricated with this new lot of SiC. The values ranged from 82.8 to 117.2 kPa (12 to 17 psi), which are similar to values obtained with the previous lot of SiC.

		TAI	BLE;	3-4	3	
HELIUM	BUBBLE	PRESS	URE	ÔF	VARIOUS	THICKNESS
	SicL	AYERS	(CA	тноі	DE C204-4	4)

Sample	Sic T	hickness	<u>Bubble</u> P	ressure
Ident.	<u>mm</u>	<u>in</u>	<u>kPa</u>	<u>psi</u>
1	0.10	0.004	165.6	24
2	0.10-0.14	0.004-0.0055	144.9	21
3	0.14	0.0055	103.5	15
4	0.11-0.14	0.0045-0.0055	103.5	15
5	0.11	0.0045	124.2	18
6	0.22	0.0085	96.6	. 14
7	0.22-0.27	0.0085-0.0105	124.2	18
8	0.27	0.0105	96.6	14
9.	0.18-0.27	0.007-0.0105	110.4	16
10	0.18	0.007	89.7	13

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### TABLE 3-44 ELEVATED TEMPERATURE BUBBLE PRESSURE OF TYPICAL CATHODE (C204-4)

	Test Temperature	BUBB	LE PRESSURE
Sample No.	(°C)	kPa	psi
l	190	65.46	9.5
2	193	78.55	11.4
3	208	77.17	11.2
4	228	65.46	9.5

Initial evaluation of zirconium pyrophosphate applied as an alternate to SiC was performed on cathode C221-5. Bubble pressure for a 0.13 mm (0.005 in.) layer at ambient temperature ranged from 131 to 172 kPa (19 to 25 psi), and was slightly higher and more uniform than data on SiC layers.

### 3.7.4.8 Backing Paper

A limited evaluation of Kureha E715 carbon paper was performed. This paper was evaluated as a possible alternative to the Stackpole PC-206 paper currently in use. Comparison data are given in Table 3-45. The thickness variation of the E715 sample measured was larger than that of the PC-206 and the average in-plane electrical resistivity of the E715 was about double that of the PC-206. Based on these data and subscale cell resistances, the decision was made that E715 was not an acceptable alternative to the PC-206.

X-ray diffraction of as-received samples of PC-206 and E715 are shown in Figure 3-44. These data indicate that the PC-206 is somewhat more "graphitic" than the E715, having both a lower  $d_{002}$  and higher  $L_{002}$ .

Evaluation of the structural properties of the Kureha paper E715 was initiated. The Kureha paper is an alternative replacement candidate for the electrode layer support as well as a possible acid transport member. Compression tests of E715 at both ambient and operating temperatures were completed. Two test units, each 20 pieces thick, were compressed in a dry (no acid) flat loading configuration. A typical compression cycle consists of loading to a specified strain and then unloading to a 22 N (5 lb) reference load. Figure 3-45 indicates the compressive stress versus strain curve for three cycles at ambient temperature. According to Figure 3-45, a compressive stress of 480 kPa (70 psi) would produce a 12 percent strain in the Kureha paper. A similar test of MAT-1 (carbon layer) indicates the same 480 kPa (70 psi) would induce only a 4.1 percent strain in the MAT-1. This verifies the Kureha paper (E715) is much more compliant than the MAT-1 (carbon layer) at ambient temperature.

TABLE 3-45 BACKING PAPER COMPARISON SUMMARY

0.406 (16.0) 0.560 (26.1) 0.477 (18.8) Kureha E715 0.021-0.022 0.318 Stackpole PC-206⁽¹⁾ 0.009-0.018(3) 0.406 (16.0) 0.368 (14.5) 0.457 (18.0) 0.034 (4) 0.004 0.307 0.375 0.037 samples) and P. O. 60816 (4 samples) 6.0 62 0.9 V vs. RHE, mA/cm² 1.0 V vs. RHE, mA/cm² 1.1 V vs. RHE, mA/cm² 74351 unless noted entified percent H₃PO₄, l atm Average Thickness, mm (mils) Wetproofed, Paste Process Min. Thickness, mm (mils) Max. Thickness, mm (mils) Dimensional (3 sheets of each) Wetproofed, Dip Process metry technique Calculated Density, g/cc Measured Density, g/cc Sulfur Analysis, w/o Porosity, percent⁽²⁾ Resistivity, ohm-cm Baseline P. Corrosion(5)



≅ + 8 + 8 + AMBIENT TEMP 3 18 KUREHA PAPER % STRAIN IN (8.842 CH) (25.8 SO CH) H H 12 Ē Ē Ê KUREHA PAPER ANG. LAYER THICONCSS-0.8124 (17.77) (18.25 | (18.77 COMPRESSIVE RREP-4.0 50 IN CYALE 2 ET-2577 PSI CTOLE 3 ET-2722 PSI CYOLE I ET=1486 PSI 10 28 LANERS 140 120 88 40 20 6 100 60 COMPRESSIVE STRESS (ISd)

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Figure 3-45. Compressive Stress-Strain Data for Kureha E715
 Paper at Ambient Temperature

The second Kureha paper stack was tested at operating temperatures. Figures 3-46 and 3-47 present the stress-strain curves at temperature. Note that cycle 1, cycle 2 (loading only) and cycle 5 are at 55°C (130°F), whereas, cycle 2 (unloading), cycle 3 and cycle 4 are at 198°C (390°F). In addition, the cycle 4 data (Figure 3-47) indicate a yield stress at approximately 500 kPa (80 psi). The plastic strain associated with this yielding accounts for the offset of cycle 5.

3.7.4.9 Differential Scanning Calorimetry

Characterization of Teflon 6C powder as-received, in a non-sintered electrode layer (C209-1) and in a sintered electrode layer (C233-8) was performed using differential scanning calorimetry (DSC). The thermogram of the Teflon 6C is shown in Figure 3-48, data from a computer-controlled DuPont 1090 analyzer. The glass transitions around 20 and 30°C (68 and 86°F) are in agreement with the values reported for polytetrafluoroethylene (PTFE). ⁽¹⁾ The virgin material has a melting point peaking at 339°C (642°F). Upon cooling, only a portion of the virgin crystalline phase resolidifies at about 310°C (590°F). The recrystallized portion of Teflon melts at a lower temperature than the virgin material, at about 326°C (619°F), as reported. ⁽²⁾ No further changes, both in the temperature and heat of melting, are observed upon continued cycling between 200 and 400°C (392 and 752°F). The thermal data are summarized in Table 3-46.

Figure 3-49 is a thermogram of a catalyst layer (C209-1) that had only been heated to 93°C (200°F) to drive off the  $NH_4HCO_3$ . On first heating, a broad exothermic transition peaking occurs at about 180°C (356°F). This transition was not observed in as-received TFE 6C. It was also absent on a second heating scan. This is attributed to the exothermic effect with the stress relaxation of the cold-worked, fibrillated Teflon. The enthalpy of the solid-state reaction was found to be different in different samples, 25.2, 27.7, and 8.6 J/g (10.8, 11.9, and 3.7 Btu/1b).^(*) These data may be an

(*)Based on the portion of Teflon in the catalyst layer (40 w/o).



Figure 3-46. Kureha Paper Stress-Strain Data at 55°C (130°F)

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Figure 3-47. Kureha Paper Stress-Strain Data at 199°C (390°F)

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Figure 3-48. DSC Thermogram of TFE 6C

Transition	Onset Temperature (°C)	Peak Temperature (°C)	Enthalpy Change J/g
l Melting	329	340	+ 40.5
Freezing	314	311	- 23.2
2 Melting	320	325	+ 23.1

### TABLE 3-46MELTING POINT DATA FOR TFE 6C



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important indicator of the state of fibrillation. Further work in this area is suggested, including thermomechanical and dilatometric studies on catalyst layers.

Unsintered catalyst layers (including the one shown in Figure 3-49) have a melting transition peaking at around  $335^{\circ}C$  ( $635^{\circ}F$ ) which is reasonably close to the peak melting point of virgin TFE 6C; however, the heat of melting is only about half that of as-received Teflon (20 J/g).

After melting, no further transition is observed, either on cooling or reheating. This suggests that the Teflon, once melted in the Pt/C matrix, is totally amorphous. It is thought that Teflon penetrates into the carbon pores and forms thin films around carbon/platinum aggregates, thereby disallowing any opportunity for crystallinity. For the same reason, no success has been experienced in obtaining electron diffraction patterns of Teflon in electrodes.

Catalyst layers that have been sintered at 360°C (680°F) on Stackpole supports show no transition at all.

3.7.4.10 Ammonium Bicarbonate

Emphasis was placed on determining a satisfactory technique to determine the particle size distribution of ground ammonium bicarbonate. Utilization of the Microtrac particle size analyzer with either isopropyl alcohol, toluene or Shell Sol as the fluid proved satisfactory. A number of wet grinding results were analyzed to assess the effect of the ball jar size on particle distribution. This work was in support of manufacturing and had as its goal the preparation of layer grinding batches of bicarbonate. Initial dry grinding experiments were analyzed for particle size distribution as well.

### 3.7.5 Other Stack Materials

### 3.7.5.1 Manifold Material

Evaluation of polyethersulfone (PES) was initiated for possible stack manifold application. Coefficient of thermal expansion tests for this material confirmed that it is anisotropic. The material expands differently in all three directions and some shrinkage occurs between cycles. The critical coefficient of thermal expansion in the "X" direction of measurement was found to be  $2.02 \times 10^{-5}$ /°C ( $1.12 \times 10^{-5}$ /°F). The coefficients of thermal expansion in the "Y" and "Z" directions were  $2.55 \times 10^{-5}$ /°C ( $1.41 \times 10^{-5}$ /°F) and  $2.17 \times 10^{-5}$ /°C ( $1.20 \times 10^{-5}$ /°F), respectively. These values are based on an average of two tests per direction. Compressive creep tests were initiated for the Victrex (PES) manifold material. Two specimens are being tested for 1000 hours at 199°C ( $390^{\circ}$ F) and 20.75 MPa (3010 psi). The results of the test will help predict orthotropic creep rates for the Victrex material.

### 3.7.5.2 Stack End Insulation Material

The thermal expansion of the insulating block material (H-18408) was evaluated in three directions up to a maximum temperature of 232.2°C (450°F). Preliminary results indicate that the material has nonlinear expansion characteristics and is anisotropic. The largest variation occurs in the "Z" direction. The "X" and "Y" directions are relatively consistent with values of  $1.99 \times 10^{-5}$ /°C (1.11 x  $10^{-5}$ /°F) and  $1.95 \times 10^{-5}$ /°C (1.08 x  $10^{-5}$ /°F), respectively. The "Z" direction expansion is greater by a factor of 3.74, for a value of  $13.2 \times 10^{-5}$ /°C (7.37 x  $10^{-5}$ /°F). This value affects the amount of compressive force developed in the stack during thermal excursions. The second cycles indicate that some shrinkage occurs during the first cycle. Therefore, it may be necessary to thermally cycle the finished parts made from this material at least twice to minimize the shrinkage effect.

### 3.7.5.3 Phosphoric Acid

Ion chromatography was performed on three samples of  $H_3PO_4$ , Table 3-47. The data indicate that all three are essentially the same with regard to F⁻,  $Cl^-$ ,  $NO_3^-$  and  $SO_4^{-2-}$ .

### 3.8 Advanced Fuel Cell Development

To achieve the plant performance goals, the fuel cell technology needs to be developed to obtain an average start of life performance of 680 mV/cell at  $325 \text{ mA/cm}^2$  (300 A/ft²), 190°C (374°F), 480 kPa (70 psia), 83 percent fuel utilization and 50 percent oxidant utilization. A performance degradation rate of 2 mV/1000 hours was considered desirable. Although significant progress has been achieved in the development effort to obtain the above goals, further improvements are required. Problem areas and potential solutions were identified and a Performance Improvement Work Plan was 'developed to achieve these goals.

Performance improvements are being sought by:

- improving catalysts.
- reducing the cell resistance.
- better understanding of operational parameters.

Efforts are also underway to reduce cell performance variance by:

- reducing the matrix thickness variations.
- improving quality control techniques.
- improving fabrication and assembly procedures.

Performance degradation rates are expected to be lowered by the use of a more corrosion resistant catalyst support, improved corrosion resistance of the plates, and acid management improvements. The advanced development effort is described in the following areas:

		Analysis, p	pm by weight	
Acid	 F	C1 ⁻	NO3	s04 ²⁻
Baker Analyzed (0260-3)	2	4	١	6
Fisher Certified ACS (A242)	3	4	2.5	13
MCB, Reagent Grade	3	5	4	11

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TABLE 3-47ION CHROMATOGRAPHY OF VARIOUS SAMPLES OF PHOSPHORIC ACID

- alternate catalyst evaluation.
- alternate catalyst support.
- electrode manufacturing.
- cell resistance.
- acid management.
- plate corrosion.
- plate coatings.
- impurity effects.

### 3.8.1 Alternative Catalysts

Heat treated platinum is being used as the cathode catalyst in the present electrodes. Improved catalyst activity with alloy catalysts has been reported in the literature by several investigators. Evaluation of developmental catalysts DC-05 and DC-06 was, therefore, initiated according to test specification TSE 2x2-008A. A heat treated platinum cathode was also evaluated to provide a baseline for comparison.

Initial performance of the developmental catalysts was 25 to 30 mV better than the heat treated platinum/Vulcan. Catalyst DC-06 showed the best performance. A total of 1,760 hours of pressurized, 480 kPa (4.76 atm), testing showed that the IR-free performance degradation rate of the developmental catalyst is 15 to 25 mV/1000 hours. This is much higher than observed for the baseline heat treated platinum/Vulcan catalyst. Cell performance data are shown in Table 3-48 and Figure 3-50.

The performance level of the alloy catalysts decreased to that of the heat treated platinum/Vulcan in 1,760 hours of testing. Similar results were previously obtained with alloy catalysts tested at ERC. An increase in cell resistance during testing was noted in all cells. This is presently thought to be due largely to the corrosion of the 900°C (1650°F) heat treated plates used in these cells. Cells containing 2700°C (4900°F) heat treated plates will be tested to verify this hypothesis. Testing of all cells will continue in order to obtain long term degradation rates.

TABLE 3-48

# ENDURANCE TESTING OF DEVELOPMENTAL CATALYST CELLS ECPS-05 THRU -08

(TEST SPEC. TSE 2X2-008A)

utilizations.	Cell , mV	Most k Recent	750) 652 (710)	761) 650 (731)	765) 657 (738)	731) 650 (734)
(air)	erminal Voltage	Pea	693 (	697 (	) 969	) 699
d 50% oxidant	F	Initial	693 (740)*	(192) 269	696 (765)	667 (729)
83% fuel (H ₂ ) an	Hours Tested (As of 5/25/84)		1759	1759	1759	1759
325 mA/cm ² ,	Cell Matrix		MAT-1	MAT-1	MAT-1	MAT-1
70 psia, 190°C,	Anode Catalyst		DC-03	DC-03	DC-03	P <b>t</b>
Conditions:	Cathode Catalyst		DC-05	DC-06	DC-06	HT Pt
<b>Uperating</b>	Cell No.		ECPS-05	ECPS-06	ECPS-07	ECPS-08

*Numbers in parentheses are IR-free voltages

70 psia, 190° C, 325 mA/cm ² DC-05 DC-03 C ECP8-05 83% H ₂ Utilization DC-06 DC-03 $\triangle$ ECP8-06 50% Air Utilization DC-06 DC-03 $\triangle$ ECP8-07 H.T. Pt Pt Pt O ECP8-08 150% Air Utilization Cole DC-03 $\triangle$ ECP8-06 150% Air Utilization DC-06 DC-03 $\triangle$ ECP8-06 150% Air Utilization DC-03 $\triangle$ ECP8-07 150% Air Utilization DC-03 $\triangle$ ECP8-08 150% Air Utilization DC-03 150% Air Utilization DC-03 150% Air Utilizat		ö	beratir	10 Col	nditio	US	<b>.</b>		ÿ	athod	le Cat	talyst	Anod	le Cata	lyst		Ce Ce	I No.	
83% H ₂ Utilization     DC-06     DC-03     Δ ECPS-06       50% Air Utilization     DC-06     DC-03     Δ ECPS-07       60% Air Utilization     DC-06     DC-03     □ ECPS-07       11     11     Pressure Cycle     Pressure Cycle     0 ECPS-08       12     0     0     0     0       13     Pressure Cycle     2nd Pressure Cycle     0 ECPS-08       14     Pressure Cycle     2nd Pressure Cycle     0 ECPS-08       13     0     0     0     0       14     0     0     0     0       15     0     0     0     0       16     0     0     0     0       17     1     1     1       1     1     1     1		10	) psia,	, 190	C, 3	25 m	A/cm	2	ļ	ă	0-05			)C-03		U		PS-05	• •
50% Air Utilization     DC-06     DC-03     □ ECPS-07       -     -     H.T. Pt     Pt     Pt     O ECPS-08       -     -     -     -     -     O ECPS-08       -     -		8	3% H ₂	Utiliza	ation					ă	C-06			C-03		7		90-Sc	
H.T. Pi H.T. Pi H.T. Pi H.T. Pi H.T. Pi H.T. Pi A.C. CPRS-08 H.T. Pi A.C. CPRS-08 H.T. Pi A.C. CPRS-08 A.C. CPRS-08		50	0% Air	Utilla	ation	_				ă	C-06			C-03		J		20-Sc	
1 1 Pressure Cycle 1 1 Pressure Cycle 1 1 Pressure Cycle 1 2 1 Pressure Cycle 1 2 1 000 2 000 2 000 3 000 1 1 000 1 1 1 Pressure Cycle 2 1 1 1 Pressure Cycle 3 000 3 000 3 000 1 1 1 Pressure Cycle 3 000 3 000 1 1 1 1 Pressure Cycle 3 000 1 1 1 1 Pressure Cycle 1 2 1 1 Pressure Cycle 3 000 1 00 1 0 1		,	• -							Ϊ	T. Pt		ē.	<b>*</b>		J	O ECI	80-Sc	·
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1st Pressure Cycle       2nd Pressure Cycle         1st Pressure Cycle       2nd Pressure Cycle         0       0         0       0         0       0         0       0         0       0         0       0         1,000       2,000         1,000       2,000         1,000       2,000					\$	·			<u> </u>							· .		• •	
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Six additional cells (ST-01 thru ST-06) were tested to compare the performance stability of developmental catalyst cathodes with the 900°C (1650°F) heat treated platinum/Vulcan cathodes under atmospheric stress test conditions (870 mV terminal voltage) and nominal atmospheric operating conditions, per Test Specification TSE 2x2-008. The cell performance at 200 mA/cm² ( $186 A/ft^2$ ) is presented in Table 3-49. The IR-free performance of developmental catalyst cells degraded faster than that of the cells with heat treated platinum catalyst, both under stress conditions and at nominal atmospheric operating conditions. Of the developmental catalysts, performance of cathodes with DC-06 catalyst degraded much slower (36 mV/1000 hours) than cathodes with DC-05 catalyst (77 mV/1000 hours). Performance of the cells with the heat treated platinum cathode increased during the test period. This may be due to continued wetting of the catalyst layer.

Based on the data available, it appears that the performance of cathodes containing developmental catalysts DC-05 and DC-06 is less stable under stress test conditions than under nominal atmospheric operating conditions, and degrades faster than the heat treated platinum catalyst.

Cell 3009 containing a 900°C (1650°F) heat treated platinum/Vulcan cathode has been tested at 101 kPa (1 atm) and 200 mA/cm² (186 A/ft²) for 11,000 hours. A lifegraph of Cell 3009 is shown in Figure 3-51. During this period its terminal voltage dropped from 643 mV (peak) to 619 mV. This represents an overall degradation rate of  $\sim$  2 mV/1000 hours. The performance degradation appears mainly to be related to a decrease in catalyst activity. The cell resistance appears to be quite stable.

3.8.2 Alternate Catalyst Support

Analysis of stack and subscale cell data obtained under pressurized test conditions during the First Logical Unit of Work indicated that the performance degradation rates of cells containing heat treated platinum/Vulcan electrodes could be as high as 25 mV per 1,000 hours. Currently, performance

**TABLE 3-49**:

## STRESS TESTING OF DEVELOPMENTAL CATALYST CELLS ST-01 THRU -06

(TEST SPEC. TSE 2X2-008)

Operating Conditions: 15 psia, 190°C, 100 mA/cm², 83% fuel (H₂) and 25% oxidant (air) util.

Cathode Catalyst U	Test Jbjective	Cell No.	Anode Catalyst	Matrix	stress testing	of stress testing	of stress testing	mV M
0C-05	Stress	ST-01	Pt	SiC	681 (708)	644 (674)	, 600 (631).	-77
DC-05	Control	ST-02	Pt	Sic	(101) [70]	663 (698)	654 (691)	-10
DC-06	Stress	ST-03	Pt	SiC	679 (715)	659 (707)	620 (679)	-36
DC-06	Control	ST-04	Pt	Sic	697 (726)	697 (729)	697 (726)	0
HT Pt	Stress	ST-05	Pt	SiC	628 (669)	634 [、] (688)	(129) 809	+ 2
HT Pt	Control	ST-06	Pt	SiC	623 (663)	636 (679)	635 (681)	+18

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Figure 3-51. Lifegraph of Cells 3009 and 3038

CELL TERMINEL VOLTAGE ۸ ۴

degradation rates of similar cells, based on short-term test results, is 8 to 12 mV per 1,000 hours. It is important to note that the 2 mV/1,000 hours degradation rate was demonstrated in Stack 431 operated at atmospheric pressure for more than 16,000 hours (Table 3-50) and Stack 560 (Figure 3-52). Recently, this rate was confirmed in subscale Cell 3009 which has been running for over 11,000 hours (Figure 3-51).

During the Second Logical Unit of Work, Gulf Corporation's Shawinigan Acetylene Black and 2700°C (4900°F) heat treated Vulcan XC-72 were selected as alternative catalyst support materials. A suitable catalyst support material should have a low corrosion rate, high surface area, and be electrochemically clean. Shawinigan Black carbon has a low sulfur content (0.01 percent). As-received Shawinigan Black has a surface area, 90 m²/g (4.40 x  $10^5$ ft²/lb), similar to that of 2700°C (4900°F) heat treated Vulcan, 80 m²/g (3.90 x  $10^5$  ft/lb). The corrosion rate of Shawinigan carbon at 0.8 volt is at least an order of magnitude lower than that of Vulcan XC-72. These desirable properties formed the basis for its selection.

Platinum/Shawinigan electrodes with 25 percent and 30 percent polytetrafluoroethylene were fabricated. Fabrication of Shawinigan electrodes was more difficult than Vulcan electrodes. Platinum/Shawinigan electrodes were tested for over 7,500 hours at 101 kPa (1 atm). Performance data for the cells tested are summarized in Table 3-51. The electrode performance is  $\sim$ 20 mV lower than that of heat treated Platinum/Vulcan at 101 kPa (1 atm). The performance decay rate (2 mV/1,000 hours), however, is similar to that of heat treated platinum/Vulcan electrodes.

Performance of 2500°C (4550°F) heat treated platinum/heat treated Vulcan electrodes was very low. Platinum/heat treated Vulcan electrodes tested in other ERC programs showed better results than those obtained here. The performance in this program was still 20 to 30 mV lower than that of heat treated platinum/Vulcan electrodes. The poor performance of heat treated platinum/heat treated Vulcan electrodes may be due to a large platinum crystallite size. This hypothesis, however, needs to be investigated further.

CELL NO.	INITIAL PERFORMANCE	PERFORMANCE AT 16,382 HRS.	DEGRADATION RATE, mV/1,000 HOURS
1	670	630	2.4
2	660	630	1.8
3	670	620	3.1
4	670	630	2.4
· 5	650	630	1.2
AVG. CELL	660	630	1.8

### TABLE 3-50

### TERMINAL PERFORMANCE STABILITY OF STACK 431, mV

**OPERATING CONDITIONS:** 

PRESSURE:	l atm
TEMPERATURE:	180-185°C
REACTANTS:	H ₂ /AIR
CURRENT DENSITY:	$100 \text{ mA/cm}^2$
CATALYST:	Pt/Vulcan
PLATES:	900°C HEAT TREATED MK-I
MATRIX:	MAT-1
CONTRACT:	DEN3-205



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TABLE 3-51

### PERFORMANCE OF ALTERNATE CATALYST SUPPORT CELLS, mV (TEST SPEC. TSE 2X2-002)

		CATHODE					HOURS	TERMINAL	(IR-FREE) V	<b>ULTAGE</b>
SUPPORT	TEFLON CONTENT %	LAMINATION PRESSURE	AMOUNT OF FILLER USED	CATALYST	CATALYST LOADING, mg/cm ²	CELL NU.	TESTEU AS 0F 5/25/84	INITIAL	PEAK	MUST RECENT
Shawinigan. 1200°C HT	25	0.13 x Std	Std	HT Pt	0.6	AS-02	2,088	636 (677)	654 (690)	647 (683)
Shawinigan, 1200°C HT	30	0.13 x Std	None	HT Pt	0.6	AS-05	528	595 (63 <b>5</b> )	623 (659)	614 (651)
Shawinigan, as-received	30	0.13 x Std	Std	HT Pt	0.6	AS-06	2,088	622 (668)	637 (680)	634 (678)
Shawinigan, 1200°C HT	30	Std	Std	HT Pt	0.5	AS-08	1,152	644 (680)	648 (680)	579 (618)
Vulcan, 2500°C HT	25	0.13 x Std	Std	HT Pt	0.6	AS-09	360	303 (380)		280 (359)
Vulcan. 2500°C HT	30	Std	Std	HT Pt	0.6	AS-12	360	435 (497)		423 (489)
Shawinigan, as-received	30	Std	Std	Pt	0.5	3037	2,040	622 (656)	641 (672)	576 (607)
Shawinigan, as-received	30	Std	Std	Pt	0.5	3038	7,584	608 (640)	643 (668)	630 (654)

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H2/Air 200 mA/cm²

| atm 190°C

Pressure: Temperature: Reactants: Current Density:

**Uperating Conditions** 

### 3.8.3. Electrode Manufacturing

Some of the performance variations observed in subscale cells and stacks may be attributed to the variations in electrode manufacturing processes. Occasionally, partial burning of the electrodes has been observed during the air sintering process. Subscale cell and out-of-cell tests were, therefore, conducted to study the effect on cell performance of the following variables:

- sintering gas environment.
- electrode lamination pressure.
- amount of filler ammonium bicarbonate used in the electrode fabrication process.

Cathodes sintered in various gas environments were tested in subscale Cells MF-Ol thru -O4 at atmospheric conditions. Although no significant difference in cell performance was observed in the limited number of tests conducted, it was decided to use an argon gas environment to prevent the possibility of electrode burning.

Thermogravimetric analysis (TGA) of the unsintered electrode layers in air had shown that burning occurred around  $160^{\circ}C$  ( $320^{\circ}F$ ). It is known, however, that heat treatment of the catalyst powder is carried out at  $900^{\circ}C$  ( $1650^{\circ}F$ ) in a nitrogen environment without any noticeable burning. Experiments indicated that burning can be limited by reducing the amount of air available at the catalyst surface.

Effects of electrode lamination pressure and the amount of filler used in electrode fabrication on cell performance was studied in Cells MF-05 thru MF-08 at atmospheric pressure. No difference in the cell performance associated with these manufacturing variables was observed. However, very low lamination pressure sometimes resulted in poor lamination of the electrode. A more detailed study is necessary to optimize the required lamination pressure. Elimination of the filler from the manufacturing process can result in a simplified process with significant cost savings. Further testing in stacks is required to confirm the above results.

### 3.8.4 Cell Resistance

Subscale cell tests indicated that a 31 mV performance improvement could be achieved by reducing the present matrix thickness, 0.043 cm (0.017 in.), by 50 percent, 0.023 cm (0.009 in.). These results were further verified in nine cell stacks. It was recognized that manufacturing processes would need to be refined to fabricate thin and defect-free matrices. A brief account of the cell resistance model and experimental results are presented in this section.

Previous effort under DOE/NASA contract DEN3-67 identified the potential resistance contribution of various individual cell components. Significant reduction in cell resistance was achieved in that program by heat treating the graphite-resin plates. Initial estimates indicated that the electrolyte matrix accounted for a large part of the measured cell resistance.

The present MAT-1 matrix consists of a carbon layer, 0.025 cm (0.010 in.), and a SiC layer, .018 cm (0.007 in.) thick. It has become apparent that the resistance of the carbon layer is not directly measured by the AC milliohmeter normally used to measure resistance. Two subscale cells were tested to obtain a preliminary estimate of the resistance of the carbon layer. Although the thickness of the matrix in Cell 3028 was 60 percent greater than that in Cell 3029, the measured resistances of both cells were the same. The peak IR-free performance of Cell 3029 was  $\sim$  20 mV higher than that of Cell 3028. This suggested that the resistance of the carbon layer is not being measured by the present resistance measurement technique. A similar gain in performance for cells without carbon layers in matrices had also been observed in Stack E-009-02.

Understanding of the effective cell resistance network was, therefore, desired. The previous resistance model did not include the interaction of electronic and ionic resistance of the electrode catalyst layers and the carbon layer in the cell matrix. The model was modified to include this interaction. Figure 3-53 presents the revised model. When the cell resistance is measured using a milliohmmeter, current from the meter passes through all the resistances. The path of the current generated in the cell is different than in the milliohmmeter case. Thus, the actual cell resistance is different that the measured value. Resistance equations for the new model were derived and individual component resistances estimated using these equations. The estimates also confirmed that the matrix accounts for most of the cell resistance.

Resistance of the cell matrix can be reduced by reducing the thickness of the SiC and carbon layers in it. However, these thicknesses may only be reduced to the minimum required for these layers to perform their functions. The main function of the SiC layer is to serve as an electrical insulator between anode and cathode, while that of the carbon layer is to provide a bubble pressure barrier to prevent gas crossover between the electrodes. Very thin layers of SiC and carbon are sufficient to perform these functions.

As a first attempt, the thickness of the SiC layer was reduced from 0.018 to 0.008 cm (0.007 to 0.003 in.) and that of the carbon layer from 0.025 to 0.015 cm (0.010 to 0.006 in.). These thin matrix layers were evaluated in subscale cells (TH-01 thru TH-06) at atmospheric pressure. Performance and resistance data are presented in Table 3-52. Individual resistances of the SiC layer, carbon layer, and hardware (electrodes, plates, and contact) in standard cells (2 x 2 in. and 12 x 17 in.) were estimated using the test data. It is again apparent from these estimates that the matrix layers offer most of the resistance. A resistance reduction of  $\sim 2 \text{ m}\Omega$  resulted from 0.01 cm (0.004 in.) reduction in the SiC layer thickness, which represents a performance gain of  $\sim 10 \text{ mV}$  at 200 mA/cm² (186 A/ft²). A carbon layer thickness reduction of 0.01 cm (0.004 in.) resulted in a performance gain of  $\sim 9$  mV at 200  $mA/cm^2$  (186 A/ft²). The total observed gain at 200 mA/cm² (186 A/ft²) was 19 mV. At power plant operating conditions,  $325 \text{ mA/cm}^2$  (300 A/ft²), a performance gain of  $\sim$  31 mV can be expected. This was verified in nine cell Stack E-009-01 and is a significant improvement in cell performance.



Figure 3-53. Modified Resistance Model Compared with Old Model

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. · ·		•	ßm	Last	as <b>4</b> °a] a s	<b>`3</b> .5	4.0	6 <b>.</b> ]	4.8	6.1	-	e -	•
		air) util.	ssistance,	At Peak Voltage	3.3	3.5	4.0	6.1	4.7	6.0		•	
	STANCE	% oxidant (	Cell Re	Initial	3.5	3.4	3.8	6.0	4.7	6.1			
	NCE AND RESI	(H ₂ ) and 259	.ee*)	Last	673 (694)	677 (695)	684 (705)	681 (712)	670 (695)	658 (689)			·
	3-52 ELL PERFORMA	1 <mark>2,</mark> 83% fuel	minal (IR-Fr oltage, mV	Peak	687 (704)	677 (695)	686 (706)	681 (712)	674 (698)	(069) 629		• • •	
	ICKNESS ON C	'С, 200 mA/ст	Cell Ter Vo	Initial	679 (697)	668 (686)	675 (694)	670 (701)	668 (692)	642 (673)	cluded.		
	MATRIX TH	l atm, 190°		Hours Tests	740	740	740	740	740	740	yer not ind		
	FFECT OF	L :SNOI		Cell No.	TH-05	TH-01	TH-02	TH-06	TH-03	TH-04	arbon la		
·		PERATING CONDIN	l Matrix 1ess, mil	Carbon Layer	0	. <b>9</b> .	10	0	9	10	tion for the co		
	 	10	Cell Thickr	SiC Layer	3	ε	e	7	7	7	*IR correct		
				3-1	82								

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The 0.008 cm (0.003 in.) thick SiC layer did, however, cause an electrical short resulting in low OCV and cell performance. The manufacturing of thin SiC layer should be improved to yield a stronger, uniformly thin layer.

The 0.008 cm (0.003 in.) thick SiC layer was also evaluated in a nine cell stack (E-009-04) at pressure. Performance and average cell resistance data are presented in Table 3-53. A performance gain of 14 mV was expected based on the measured resistance reduction. The observed gain at  $325 \text{ mA/cm}^2$  (300 A/ft²) was 18 mV.

3.8.5 Bipolar Plate Corrosion

Corrosion of the cathode side of bipolar plates has been observed in subscale cells and stacks after pressurized operation. The following factors are suspected to play a significant role in plate corrosion:

- acid volume expansion.
- high water partial pressure in oxidant exhaust stream.
- defective wetproofing of backing paper.
- fuel starvation in a cell driven by other cells.
- oxidant starvation in a cell driven by other cells.

The objectives of this task were to identify the cause(s) and to gain an understanding of the mechanism of bipolar plate corrosion. Conditions suspected of causing bipolar plate corrosion were investigated by introducing these conditions into subscale cells and evaluating their effects on plate corrosion. The ultimate goal is to find a means to prevent or protect against bipolar plate corrosion.

Subscale Cells PC-O1 thru PC-O8 were tested for  $\sim 600$  hours at atmospheric pressure to evaluate the suspected contributors to corrosion. Acid expansion and higher water partial pressure in the cathode exhaust stream were obtained by running these cells with pure oxygen at greater than 60 percent utilization

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#### TABLE 3-53

## EFFECT OF MATRIX THICKNESS ON PERFORMANCE AND RESISTANCE OF STACK E-009-04 AT PRESSURE

AND RESISTANCE OF STACK E-009-04 AT PRESS

(1)	DECREASE IN M	EASURED RESISTANCE	
· .	Matrix Thi	ckness, mil	Average Cell (12 x 17 in.) Resistance, $m\Omega$
•	SiC Layer	Carbon Layer	At 325 mA/cm ² , 70 psia, 190°C
	7	10	0.12
	3	10	0.08
Differenc	ce 4	0	0.04 mΩ

Average Resistance of SiC layer = 0.01 m $\Omega$ /mil

(2) MEASURED VOLTAGE GAIN

Cell No.	Matrix TI SiC Layer	nickness, mil Carbon Layer	Cell Terminal Voltage, mV At 70 psia, 325 mA/cm ² , 190°C
1	7	10	662
2	7	10	654
3	. 7	10	.676 Avg. 669.8
4	7	10	673
55	7	10	684
6	3	10	641*
7	3	10	678 Avg. 676.5
8	3	10	708
9	3	10	680

*Cells 6, 7, 8, 9 were diagnosed to have electronic shorts

## (3) CONCLUSIONS

Based on resistance reduction:

predicted gain	=	14 mV
observed gain, Cells 6-9	=	7 mV
observed gain, Cells 7, 8 and 9	=	18 mV

and then running them on air. This operation is referred to as an oxygen cycle. Fuel and oxidant starvation due to gas maldistribution were simulated by restricting the gas flow and fixing the current density via an external power supply.

Maximum acid expansion was expected in Cell PC-O1. No change in performance, however, was observed after two oxygen cycles. Cell PC-O2 could not be operated at 84 percent oxygen utilization, (partial pressure of water corresponding to plant conditions) hence it was operated at 78 percent oxygen utilization. Performance of the cell decayed by 25 mV after two oxygen cycles. Upon disassembly of Cell PC-O2, a platinum deposit was found on the SiC matrix and on the anode graphite plate in the acid reservoir area. There was no apparent reason for this rather strange occurence. No significant decay was observed in the other two cell (PC-O3 and PC-O4). During oxygen cycles, acid was seen to have expanded in the acid fill tubes for all four cells. Upon disassembly, no visible softness was found on any of the cathode plates.

Cell PC-05 was operated in an oxidant starved mode and driven by an external power supply. A decay of 31 mV was observed after two oxygen cycles. No softness of the cathode plate was evident upon disassembly.

Cell PC-O6 was run in a fuel starved mode during the oxygen cycle. This cell required nearly 2V from the external power supply. After one hour of operation in this mode, a significant increase in cell resistance was noted (from  $\sim 4 \ m\Omega$  to 60 m $\Omega$ ). Testing was terminated when acid in the fill tube turned black. Significant corrosion was observed in the acid channel area on the anode plate.

The results of these experiments may be summarized as follows:

• Up to 12 percent acid expansion and corresponding high water partial pressure did not cause any apparent softness of the cathode plate or

the backing paper. This may be due to the lateral transport of electrolyte in the 2 x 2 cells allowing the acid to escape from the edges. The above experiments also suggest that similar acid expansion experiments in  $12 \times 17$  stacks would be helpful in isolating the effects of acid expansion from any other effects of high pressure operation.

 Oxidant and fuel starvation caused significant performance loss in fuel cells in a driven mode. Fuel starvation is more detrimental than oxidant starvation although fuel starvation causes anode plate corrosion; no apparent <u>cathode</u> plate corrosion was found in either of these cells.

Cell PC-07 was tested to evaluate defective wetproofing of the backing paper on plate corrosion. The cathode backing paper in this cell had a non-wetproofed area ( $\sim$  1.3 cm (0.5 in.) diameter) in the center. This cell was first conditioned and tested at 200 mA/cm² (186 A/ft²), 101 kPa (1 atm) pressure, 25 percent oxidant (air) and 80 percent fuel (hydrogen) utilizations. It was then tested at 325 mA/cm² (300 A/ft²), 101 kPa (1 atm) pressure, 84 percent oxidant (oxygen) utilization and 80 percent fuel (hydrogen) utilization to simulate acid expansion and high water partial pressure in the oxidant stream. After two cycles of acid expansion, no significant decay in cell performance was observed. After  $\sim$  600 hours of testing, the cell was disassembled, and no corrosion was visually apparent underneath the non-wetproofed area of the backing paper. A small area of the cathode plate near the acid reservoir was found to be corroded.

Cell PC-08 was assembled with a specially fabricated cathode. The backing paper of this cathode had a 1.3 cm (1/2 in.) wide SiC strip in the middle in place of standard backing paper. This provided an acid path from the catalyst layer to the cathode plate. Cell PC-08 was first conditioned and tested at  $200 \text{ mA/cm}^2$  (186 A/ft²), 190°C (374°F), 101 kPa (1 atm), 25 percent oxidant (air) utilization, and 80 percent fuel (hydrogen) utilization. It was then

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tested at 325 mA/cm² (300 A/ft²), 190°C (374°F), 101 kPa (1 atm), 84 percent oxidant (oxygen) utilization and 80 percent fuel (hydrogen) utilization to provide a 8 percent acid expansion and high water partial pressure in the oxidant stream. After two cycles of acid expansion, a steady decline in cell performance was observed. Testing of this cell was terminated after 580 hours.

Careful examination of the cell components during post-test disassembly revealed no apparent corrosion of the plates. The anode backing paper and the lower half of the anode plate were found to be wet and may have been the cause of the poor cell performance.

Potentiostatic corrosion testing of as-received and wetproofed Stackpole backing paper was also conducted. Test results are presented in Figure 3-54. Corrosion behavior of this material was very similar to that of pyrolitic graphite. The corrosion current at 1.05 V for the as-received backing paper was  $\sqrt{7} \times 10^{-3} \text{ mA/cm}^2$  (6.5  $\times 10^{-3} \text{ A/ft}^2$ ) (in comparison to  $\sim 4 \text{ mA/cm}^2$ (3.7 A/ft²) for the 900°C (1650°F) heat treated A-99 graphite/resin plate material). The corrosion current for a wetproofed backing paper at 1.05V was  $\sim 2.5 \times 10^{-2} \text{ mA/cm}^2$  (2.3  $\times 10^{-2} \text{ A/ft}^2$ ). This is higher than the value obtained for the plain backing paper. This may be due to some residual dispersing agent from the flourinated ethylene-propylene (FEP) emulsion. These results suggest that the backing paper is graphitic and any further heat treatment of the backing paper may not be helpful.

Corrosion testing of Asbury A-99, Asbury 4221 and Superior 9026 molded plates was completed. The Asbury 4421 plates showed the best corrosion resistance in these tests. At 0.85V, wet-mixed Asbury 4421 plates showed a three-fold reduction in corrosion current as compared with standard Asbury A-99 dry-mixed plates. The dry-mixed Superior 9026 plate showed no improvement over the standard Asbury A-99 dry-mixed plates. The improved corrosion resistance of the Asbury 4421 plate may be attributed to the higher purity of the graphite and to the wet mixing process.

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STEP TAFEL PLOT

CORROSION POTENTIAL, VOLT

### 3.8.6 Bipolar Plate Coatings

A possible solution to prevent or minimize plate corrosion is to apply a Teflon (e.g. FEP) coating on the cathode side of plate (except the rib area) and thereby reduce the plate surface area in contact with phosphoric acid. Another solution may be to heat treat the plates to a high temperature (e.g. 2700°C (4900°F)) to form a structure which is more corrosion resistant. Because Teflon (FEP, PFA) is a poor conductor of electrons, the bulk of the coating should be sanded off the rib area to minimize the electrical resistance added by the coating. Another option which minimizes the electrical resistance of Teflon coating is to use carbon or graphite filling.

A test plan was written to evaluate plate coatings which includes the following items:

- Evaluate the corrosion resistance of FEP, FEP/Graphite A-99, PFA/Vulcan XC-72 coated plates, and 2700°C (4900°F) heat treated plates.
- Determine the stability of each coating.
- Determine the effect of coating thickness on measured cell resistance, coating stability, and corrosion resistance.
- Demonstrate subscale cell performance.

The objective of plate coating is to form a protective layer against acid penetration into the graphite/resin plates and thus prevent plate corrosion. The development of a protective coating with a minimum increase in fuel cell electrical resistance is being pursued through vendors and at ERC. The current status of the development is presented below. Two subscale cathode plates were coated by a vendor with FEP to yield 0.0013 cm and 0.0025 cm (0.0005 in. and 0.001 in.) thick coatings. Preliminary tests indicated that these coatings were electrical insulators. By shaving most of the coating off the ribs, the increase in electrical resistance due to coating was reduced; however, the cell resistance was still high.

To lower the electrical resistance of Teflon coatings, development of a conductive graphite/Teflon composite coating was initiated. Materials and plans for a composite coating were submitted to a vendor for formulation and supplying coated subscale plates for future testing.

DuPont's "Topcoat Finish Black" (dispersion of Teflon and graphite) was received and evaluated. The electrical resistance of a coating less than 0.0013 cm (0.0005 in.) thick was not acceptable. Discussion with DuPont also revealed that an unacceptable anionic wetting agent was used in "Topcoat Finish Black". Further work with DuPont is planned to establish materials suitable for fuel cell use.

Presently, bipolar plates are heat treated to  $900^{\circ}C$  ( $1650^{\circ}F$ ). Higher heat treatment temperatures are expected to increase the corrosion resistance of the bipolar plates. Ten sets of Westinghouse subscale cell graphite/resin plates were successfully heat treated to  $1150^{\circ}C$  ( $2100^{\circ}F$ ). Five sets of these plates were heat treated to  $1500^{\circ}C$  ( $2732^{\circ}F$ ) by a vendor. Heat treatment cycles for these trials have been developed by ERC under in-house programs. The other five sets of subscale plates were successfully heat treated to  $2700^{\circ}C$  ( $4900^{\circ}F$ ) by a vendor using ERC procedures. Visual inspection of these plates indicated no delamination or blistering but the plates were bowed. Careful sanding of the plate reduced bowing satisfactorily. Dimensional changes due to heat treatment above  $900^{\circ}C$  ( $1650^{\circ}F$ ) were less than 1 percent and current MK-II stack hardware should accommodate these changes. As expected, the porosity measured by mercury porosimetry (pore diameters between 50 microns (2 mils) to .006 microns ( $2.4 \times 10^{-4} \text{ mils}$ )) increased for materials heat treated at  $2700^{\circ}C$  ( $4900^{\circ}F$ ).

#### 3.8.7 Acid Management

A comprehensive acid management system study was submitted during the First Logical Unit of Work. Some experimental effort on different acid management-related issues was also initiated during that period. The effort during the Second Logical Unit of Work focused on four different areas as identified in the Performance Improvement Work Plan document:

- Improve lateral transfer of electrolyte during non-steady state operation and acid addition by improving matrix and transport and an alternate seal design.
- Test acid volume control methods.
- Estimate acid requirements based on individual component characteristics.
- Establish criteria and procedures for acid replenishment during stack operation.

Three nine cell stacks were tested in support of this activity. Stack E-009-02 was assembled with a seal design which extended the silicon carbide coated cathode and the carbon layer over the acid replenishment channel. A diagram depicting this design is shown in Figure 3-55. The design improves the wicking of acid into cells. The effectiveness of this design was demonstrated by acid addition to Stack E-009-02 after its first pressure cycle. The cell OCV increased significantly as shown in Table 3-54. This stack had five standard cells and four cells which contained no carbon layer in the matrix. Cells without a carbon layer had very low OCV before the acid addition, but increased to the level of the standard cells after acid addition. Apparently, acid was effectively transported to the cells. This data supports the recommendation that this seal design be used in future cells.

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b) ALTERNATE SEAL DESIGN FOR IMPROVED ACID TRANSPORT

Figure 3-55. Cell Seal Design

## TABLE 3-54

# STACK E-009-02 OPEN CIRCUIT VOLTAGES, mV

# Operating Conditions: 1 atm, 125°C, H₂/Air

CELL NO.	MAT - 1	BEFORE ACID ADDITION	AFTER ACID ADDITION
1	Yes	888	896
2	Yes	878	895
3	Yes	923	937
· 4	Yes	913	943
5	Yes	886	988
6	No	71	772
7	No	9	1003
8	No	46	895
9	No	137	903
Avg. MAT-1 Matr	iv Cells	898	032
Avg. SiC Maturi		66	902
Avg. Sil Matrix	cerrs	00	893

Total Acid Addition

∿ 65 cc 97% H₃PO₄

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Stack E-009-03 was assembled with Cells 1 - 5 containing Kureha 604 paper instead of Vulcan carbon layer. This material was tested to see if it improved the lateral transport of acid during operational changes that result in acid volume expansion. The stack was tested for a total of  $\sim$  160 hours, including  $\sim$  90 hours at 480 kPa (70 psia). An average performance of 653 mV/cell in Cells 1-5 containing Kureha paper in cell matrices and 661 mV/cell in Cells 6-9 containing MAT-1 matrices was observed at 480 kPa (70 psia), 190°C (374°F), 325 mA/cm² (300 A/ft²), 83 percent fuel (hydrogen) and 40 percent oxidant (air) utilizations. A reversal of polarity in Cell 3 caused stack testing to be terminated. Kureha paper in the cell matrices resulted in a slightly lower average cell performance ( $\sim$  10 mV/cell) as compared to cells containing MAT-1 matrices. Several other differences were observed between Cells 1-5 containing the Kureha paper matrices and Cells 6-9 containing MAT-1 matrices:

- 1.5
- Open circuit voltages of Cells 1-5 were lower (∿ 50 mV/cell) than those in Cells 6-9. This may have been due to internal short circuits or slight gas cross leaks.
- Due to the more porous structure (and thus higher electronic resistance) of Kureha paper compared to the standard carbon layer, the measured resistances of Cells 1-5 were  $\sim$  50 percent higher than those of Cells 6-9. This higher electronic resistance was not expected to adversely affect cell performance since ionic resistance of the matrix has a much more significant effect on performance than does electronic.
- SRG loss in Cells 1-5 was 82 mV/cell at the design point compared to a loss of 34 mV/cell for the cells containing MAT-1. Overheating also occurred in several cells containing Kureha paper in the matrices which was attributed to reactant crossover.

Post-test disassembly showed degradation of the Cell 1 cathode plate near the oxidant inlet and both the Cell 3 anode plate and anode backing paper near the fuel inlet.

Based on the limited experience thus far with Kureha paper in cell matrices, Kureha paper is not satisfactory as a barrier to crossleaks. Post-test disassembly also showed that it does not prevent plate wetting by transporting the electrolyte laterally.

During upset conditions it is possible to get relatively large changes in acid volume. There are two methods for dealing with this potential problem:

- allow the acid to expand laterally and collect in the replenishment channel (modified matrix tested in Stack E-009-03)
- allow acid expansion perpendicular to the electrode (tested in Stack E-009-01).

An acid inventory control member (AICM) was initially developed and tested on Contract DEN3-205. The best component design from that effort was incorporated into the anode of Cells 6-9 in Stack E-009-01. The basic design of this member is shown in Figure 3-56.

The stack was operated for  $\sim$  530 hours, of which 100 hours were at 480 kPa (70 psia). As shown in Figure 3-57, Cell 6-9 containing AICM anodes had performance equivalent to the other cells. Disassembly inspection showed that one cell with the AICM and one cell without an AICM had cathode plate corrosion. Apparently, the AICM did not prevent this type of corrosion.

The amount of acid required to fill standard or modified components is obtained from analysis of ERC electrolyte take-up (ETU) data and from Westinghouse calculations. Since components can vary in thickness and density, the amount of acid to completely fill them during the assembly procedure cannot be ascertained with any significant accuracy. This variability, however, can be accommodated by applying a constant amount during the initial assembly and then adding acid through the acid replenishment channel in each cell after assembly. The optimum conditions for post-assembly acid addition remain to be determined.



Figure 3-57. Initial Performance Data for Stack E-009-01

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TERMINAL POTENTIAL, mV

Analysis of acid replenishment requirements during stack operation has been limited by the relatively short periods of operation for each stack and the variations of the cell components built into the stacks. Each stack was found to accept acid when it was added. A summary of the acid additions is presented in Table 3-55. Apparently the cells had not received enough acid before operation. Improved procedures for acid addition prior to operation are required for the future.

#### 3.8.8 Impurity Effects

An effort to catalog and understand the level of impurities in various starting materials was initiated during the First Logical Unit of Work. Spectrographic analyses showed that as-received Vulcan XC-72 (standard catalyst support) contained  $\sim$  1 percent sulfur. Because high levels of sulfur may cause catalyst poisoning and create problems during plant startup, an effort to remove these impurities was undertaken. Various pretreatments including leaching in phosphoric acid and heat treatments at 900°C, 1800°C, and 2500°C (1650°F, 3300°F, and 4550°F) were examined as shown in Table 3-56. Both free sulfur and total sulfur were measured. Acid leaching removed only a small amount of sulfur. Similarly, heat treatment to 900°C (1650°F) ("standard" heat treatment temperature after catalyzation) did not change the sulfur content. Thus, as expected, sulfur appears to be strongly bound within the carbon structure. Further heat treatment to 1800°C (3300°F) and 2500°C (4550°F) did remove  $\sim$  95 percent of the sulfur. It is possible that an intermediate temperature between 900°C (1650°F) and 1800°C (3300°F) may also provide a similar reduction in sulfur levels. However, such heat treatments do change the chemical and physical properties of carbon and may also be expensive.

The sulfur content of Shawinigan Black, a purer carbon support, was also analyzed. As shown in Table 3-56, the sulfur level is  $\sim$  160 ppm in the as-received material. Heat treatment to 1200°C (2200°F) and 1800°C (3300°F) reduced this level slightly.

	Stack Number				
Addition No.	E-009-01	E-009-02	E-009-04	E-009-06	
]	25	12	10	28	
2		40	43	-	
<b>3</b> .	· _	20	54	-	
Total	25	72	107	28	

TABLE 3-55 SUMMARY OF ACID ADDITIONS TO NINE CELL STACKS,  $\mbox{cm}^3$ 

## TABLE 3-56

TOTAL AND FREE SULFUR CONTENTS OF CARBON SUPPORTS

<u>Material</u>	Sulfur Con <u>Total</u>	ntent % <u>Free</u>
As-received Vulcan XC-72	0.93	0.53
Hot Phosphoric Acid (190°C) Leached Vulcan XC-72	0.88	0.27
900°C heat treated Vulcan XC-72 (in N ₂ atmosphere)	0.89	0.51
1800°C heat treated Vulcan XC-72 (in N ₂ atmosphere)	0.025	Not detectable
2500°C heat treated Vulcan XC-72 (in N ₂ atmosphere)	0 <b>.</b> 024	Not detectable
As-received Shawinigan	0.016	Not detectable
1200°C heat treated (in N ₂ atmosphere) Shawinigan Acetylene Black	0.013	Not detectable
1800°C heat treated (in N2 atmosphere) Shawinigan Acetylene Black	0.012	Not detectable

*Less than 40 ppm.

## 4.0 ERC NINE CELL STACK TEST FACILITIES DEVELOPMENT

A pressurized test facility to accommodate a  $12 \times 17$  in. nine cell stack was completed. This facility improved upon the past ERC facilities in the following areas:

- capable of testing up to a nine cell stack.
- measurement of fuel, oxidant, and cooling flow pressure drops (bringing the total number of  $\Delta P$  measurements to six).
- cooling inlet temperature control.
- redundant flow monitoring.

Current Density Range:

- fine control of electrical load.
- automatic failure mode control system.
- load transients with greater than 3 milliseconds switching time able to be studied.

A schematic flow diagram for the improved facility is presented in Figure 4-1. This facility was designed for the following range of operating conditions:

Pressure Range: 0 to 1037 kPa (0 to 135 psig), pressurization rate of approximately 13.8 kPa (2 psi) per minute

Standard Gases: Hydrogen fuel, air oxidant, auxiliary inputs for special gases

50 to 460 mA/cm² (47 to 430 A/ft²)



Fuel FLow Range:

Up to 80 percent utilization at 50 mA/cm² (47 A/ft²) 50 to  $\sim$  100 percent utilization at 460 mA/cm² (430 A/ft²)

Oxidant Flow Range:

Up to 50 percent utilization at 50 mA/cm² (47 A/ft²) 20 to  $\sim$  100 percent utilization at 460 mA/cm² (430 A/ft²)

Temperature Range:

30 to 250°C (86 to 480°F), independently controlled end plate temperatures

Cooling:

100 scfm air, Ambient to 130°C (265°F)*

Unattended operation of the facility is made possible by the use of a Failure Mode Control System (FMCS). This system continuously monitors the facility for conditions that could have a harmful effect on either the facility or the stack. The FMCS monitors several stack voltages, several temperatures, line power, and hydrogen concentration in the vessel cover gas. The specifications for the FMCS are presented in Table 4-1.

Construction and checkout of the first nine cell stack pressurized test facility, SE-1, was completed. The checkout of facility SE-1 was conducted by testing Stack E-009-06 and when completed, the facility was put into service. Stacks have been tested in this facility up to 825 kPa (120 psia) and  $400 \text{ mA/cm}^2$  (372 A/ft²) using pure hydrogen as fuel with a 55°C (100°F) coolant  $\Delta T$ . The verified operating condition ranges are presented in Table 4-2. The facility has accumulated  $\sim$  3,500 hours of testing.

Several recommended modifications to the original design were identified during the testing of facility SE-1. These changes are outlined below:

*At blower element

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TABLE 4-1

SPECIFICATIONS OF THE FMCS FOR 12 X 17 STACK PRESSURIZED TEST FACILITY

Problem	Trip Point (Adjustable)	Action
Stack Undervoltage (group of 2 or 3 cells)	0.5 - 3.0 V	Remove Load
Stack Over Temperature	150 - 250°C	Remove Load
Vessel Over Temperature	100 - 175°C	Remove Load Stop Process Flows Depressurize System
H ₂ in Vessel Gas	1% or 4%	Remove Load Stop Process Flows Depressurize System
Power Failure	Longer than 1 min.	Remove Load
	Longer than 30* min.	Remove Load Stop Process Flows Depressurize System

*Due to loss of backup power and supply air.

### TABLE 4-2

## VERIFIED OPERATING CONDITIONS (USING NINE CELL STACK IN PRESSURIZED TEST FACILITY SE-1)

#### OPERATING CONDITIONS

Pressure Range:

14.7 to 120 psia (Pressurization Rate of  $\sim 0.5$  psi per min.)

Standard Gases:

H₂ fuel, air oxidant auxiliary inputs for other gases

Current Range:

At 14.7 psia-up to 150 mA/cm²

At 70 psia-86 to 400  $mA/cm^2$ 

Fuel Flow Range:

Up to 83% utilization at 86 mA/cm², 50 to 85% utilization at 325 mA/cm²

Uxidant Flow Range: Up to 50% utilization at 86 mA/cm², 40 to 65% utilization at 325 mA/cm²

Stack Temp Range:

Ambient to 250°C (independently controlled for each end plate)

Cooling  $\Delta T$ :

∿ 55°C (100°F) at 325 mA/cm² with 120°C (250°F) inlet manifold temperature

Coolant  $\Delta T$  higher than the design value of 100°F above 400 mA/cm². Modification for increased coolant flow capacity submitted to Westinghouse to reduce coolant  $\Delta T$  from 100 to 68°F.

Pressure above 120 psia not tested. Pressurized at 83 mA/cm² load, 50% H₂ utilization 40% oxidant utilization.

COMMENTS

Modification for long term SRG planned

 Addition of a flow control/gas blending system for extended operation on SRG. This system should allow incremental addition of the various component gases to investigate their effect on stack performance. Minor adjustments in gas compositions should also be possible.

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Addition of a temperature controller in the fuel inlet line to achieve a more uniform thermal profile and to simulate the temperature of the fuel gas entering the stack from a reformer.

- Increase in cooling loop capacity. The program specification of 55°C (100°F) temperature rise in the cooling air system from inlet manifold to outlet manifold was revised after the design of test facility SE-1. The new specification of 38°C (68°F) rise and increased pressure drop in the cooling plate, along with other factors, will require modification of the design to single pass cooling and installation of a dedicated compressed air system.
- Single pass cooling design to eliminate the blowers presently in use. The blowers are unreliable for long term continuous operation. By replacing them with a dedicated compressor system, overall system reliability will increase.
- Oil-free compressors are recommended because oil decomposition products generated at elevated temperatures in the compression process may poison fuel cell catalysts. Although filters can be used to remove oil particles, they do not remove vapors and gases. Periodic compressor shutdowns required to replenish the oil also reduce the compressor availability and may require unwarranted stack shutdowns. The air in these compressors is thoroughly mixed with the oil and a filter malfunction can introduce oil in the compressed air lines, requiring a long shutdown to decontaminate.

A comparison of the design for recirculation and single pass cooling is presented in Figure 4-2.



Work on modification of the second nine cell stack pressurized test facility SE-2 was initiated.

Test facility SE-3 is planned for construction in the next logical unit of work. After it is completed, facility SE-1 will be modified to the single pass cooling scheme.

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15. ABSTRACT-- PURPOSE, SCOPE, APPROACH, RESULTS, CONCLUSIONS, SIGNIFICANCE: (MAXIMUM: 200 WORDS) This report summarizes the work performed by Westinghouse Electric Corp. and Energy Research Corp. during the second phase of a planned multiphase program to develop a Phosphoric Acid Fuel Cell (PAFC) for Electric Utility <u>Prototype Power Plant Application. The results of this effort include:</u> 1) establishment of the final system level requirements for the Fuel Cell, <u>Fuel Processing, Power Conditioning, Rotating Equipment, and Instrumentation</u> and Control Systems, 2) advancement of fuel cell technology through innovative improvements in the areas of acid management, catalyst selection, electrode materials, and quality assurance programs, and 3) demonstration of improved fuel cell stack performance.

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