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TROPOSPHERIC HO2 DETERMINATION BY FAGE R.J. O'Brien, T.M. Hard, C.Y. Chan, and A.A. Mehrabzadeh Portland State University Portland, OR

In contained-flow HO analyzers, ambient  $\mathrm{HO}_2$  can be detected by its conversion to HO via

 $HO_2 + NO --> HO + NO_2$ 

The detection efficiency is greatest at low pressures, where the subsequent removal of the HO product by the NO reagent (via HO + NO + M  $\rightarrow$  HONO + M) is relatively slow. Moreover, nozzle expansion of the air from ambient to low pressures produces a turbulent zone that assists in mixing the reagent with the sample. If the HO product is observed by laser-excited fluorescence, then the other advantages of low-pressure detection by FAGE (Fluorescence Assay with Gas Expansion) also apply.

The method and some of its potential interferences are set forth in Reference 1. Regarding RO<sub>2</sub>, we incorrectly assumed that the intermediate reaction RO + O<sub>2</sub> -=> R'CHO + HO<sub>2</sub> is instantaneous in our pressure range; correction reduces this interference considerably. When NO is the modulating reagent, any contaminant that fluoresced (or gave photolytic HO) would produce a false positive HO<sub>2</sub> signal. This can be prevented by using a steady NO flow, modulating the resulting ambient HO<sub>2</sub> + HO signal with an HO-removing reagent, and performing a parallel determination of ambient HO in another channel.

We have calibrated the FAGE instrumental response to external HO<sub>2</sub> by observing NO decay in the photolysis of HO-CH<sub>2</sub>O mixtures and by choosing conditions in which HO<sub>2</sub> + NO is the only significant NO destruction path. We have determined HO<sub>2</sub> in urban air.

Reference

T.M. Hard, R.J. O'Brien, C.Y. Chan and A.A. Mehrabzadeh, Environmental Science and Technology 18 768-777 (1984).

## Comments

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Using this chemical conversion/LIF detection of OH method, this group estimates that  $HO_2$  can be detected at levels of 60 to  $120 \times 10^{\circ}/\text{cm}^3$  in a 6-minute average. They claim it is specific to  $HO_2$  in that other peroxyl radicals, such as methyl peroxyl, can be prevented from yielding OH at low pressure due to kinetic freezing out. Current calibrations are by NO decay due to  $HO_2$  + NO in the absence of  $O_3$ , and  $HO_2$  decay due to  $HO_2$  +  $NO_2$  in the absence of NO,  $NO_2$ , and  $O_3$ . These two methods give differing results by a factor of 2, and it will be necessary to resolve such discrepancies. The workshop participants felt that the method appeared sound from a spectroscopic/laser basis, but that there remained questions concerning its chemical aspects.