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Water Activities in the Kerala Khondalite Belt

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The determination of $a(H_2O)$ and $a(H_2O)$ gradients in granulite terrains can provide important constraints on their petrogenesis[1,2,3]. In this study, we calculate $a(H_2O)$ in various rock types of the Kerala Khondalite Belt (KKB) and evaluate the granulite-facies metamorphism of the region in light of this information.

Two of the major rock types of the KKB contain mineral assemblages that permit the characterization of $a(H_2O)$. The charnockites contain bt + $qtz + opx + kfs + (grt-ilm \pm gr)$ and the khondalites contain bt + qtz + sil + grt+ $kfs \pm (crd-spl-ilm-gr)$. These two assemblages define the equilibria: (R1) 2 Phl + 6 Qtz = 3 En + 2 Kfs + 2 H_2O and (R2) Phl + 2 Qtz + Sil = Prp+ Kfs + H_2O, respectively, and can be used to calculate $a(H_2O)$, provided that pressure and temperature are known and the relevant thermodynamic data and activity-composition models are available.

Geothermobarometric studies [4] indicate that the entire KKB was metamorphosed at relatively uniform conditions of 5.5 kb and 750°C. Therefore, all calculations were made at this pressure and temperature. The position of the Mg-end-member reactions were calculated using thermodynamic data from the internally consistent data set of Holland and Powell [5]. It is possible to make the calculated positon of R1 agree with the experimental bracket of Bohlen et al. [6] at 5 kb and $X(H_2O)$ = 0.35 by adjusting the thermodynamic values of Phl [7]. Because recent calorimetric measurements [8] suggest that the $\Delta H^{f}_{Phl,298}$ listed in Holland and Powell [5] is correct, we have chosen to increment the S°_{Phl,298} until the calculated position of R1 agrees with the experimental bracket. The shallower slope of R1 calculated with this larger S°_{Phl,298} is in better agreement with the slope of R1 (at X[H₂O] = 1) determined by Wood [9].

 a_{phl}^{Mice} and a_{en}^{Opx} were calculated using ideal on-sites mixing

models [10,11]. The a_{Kfs}^{Afs} was assumed to equal X_{Kfs}^{Afs} , where X_{Kfs}^{Afs} was determined from the composition of coexisting plagioclase at 750°C according to the model of Stormer [12]. The a_{Prp}^{Ort} was calculated using the model of Newton et al.[13].

The results of the calculations are shown in map-form in Fig. 1. The charnockites give an average $a(H_2O)=0.27\pm0.05$ (1 σ) and the khondalites an $a(H_2O)=0.26\pm0.06$ (1 σ). The striking feature is the uniformly low $a(H_2O)$ recorded over a large region. This uniformity is in marked contrast to a number of other granulite terrains where significant gradients in $a(H_2O)$ have been documented over a kilometer or even a meter scale [2,3].

Two lines evidence suggest the uniform $a(H_2O)$ of the KKB rocks were not caused by the extraction of a partial melt. First, within each rock type, $a(H_2O)$ shows no obvious correlation to bulk compositional variables such as silica content, Fe/Mg ratio or $a(TiO_2)$. Thus, "restite-rich" assemblages record approximately the same $a(H_2O)$ as more leucocratic assemblages. Second, there is a remarkably good agreement between $a(H_2O)$ in the charnockites and khondalites. If this agreement is correct then it would seem highly fortuitous that two contrasting rock types, which encountered different melting reactions, partially melted to yield identical $a(H_2O)$.

The simplest interpretation of the $a(H_2O)$ data is that the rocks of the KKB equilibrated with a low $a(H_2O)$ fluid that had a roughly constant composition throughout the region. The patchy replacement of garnet-biotite gneiss by coarse-grained charnockite along deformation zones and foliation planes provides field evidence for this fluid-present metamorphism [14,15]. It is the opinion of the authors that the low $a(H_2O)$, presumably CO_2 -rich, fluids were introduced from deeper levels. However, a model invoking internally-derived fluids, such as those generated by the reaction bt + qtz +gr = opx + kfs + v [16], possibly under conditions of P_{fluid} $P_{ilthostatic}$ [14], would also be consistent with the $a(H_2O)$ data, provided that these fluids were sufficiently water-poor.

References

(1) Phillips, G.N. (1980). Contr. Mineral. Petrol., 75:377-386. (2) Valley, J.W., McLelland, J., Essene, E.J. and Lamb, W. (1983). Nature, 301:226-228.

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(3) Waters, D.J. and Whales, C.J. (1984). Contr. Mineral. Petrol., 88:269-275. (4) Chacko, T., Ravindra Kumar, G.R. and Newton, R.C. (1987). J. Geol., 95:343-358. (5) Holland, T.J.B. and Powell, R. (1985). J. Metam. Geol., 3:343-370. (6) Bohlen, S.R., Boettcher, A.L., Wall, V.J. and Clemens, J.D. (1983). Contr. Miner. Petrol., 83:270-277. (7) Bhattacharya, A. and Sen, S.K. (1986). J. Petrol., 27:1119-1141. (8) Clemens, J.D., Navrotsky, A., Circone, S., McMillan, P., Smyth, B.K. and Wall, V.J. (1987). EOS, 68:460. (9) Wood, B.J. (1976). N.E.R.C., 11:17-19. (10) Bohlen, S.R., Peacor, D.R. and Essene, E.J. (1980). Am. Miner., 65:55-62. (11) Wood, B.J. and Banno, S. (1973). Contr. Miner. Petrol., 42:109-124. (12) Stormer, J.C., Jr. (1975). Am. Miner., 60:667-674. (13) Newton, R.C., Geiger, C.A. and Kleppa, O.J. (1987). In Korzhinski volume, Springer-Verlag. (14) Srikantappa, C., Raith, M. and Speiring, B. (1985). J. Geol. Soc. India. 26:849-872. (15) Ravindra Kumar, G.R. and Chacko, T. (1986). J. Geol. Soc. India. 28:277-288. (16) Hansen, E.C., Janardhan, A.S., Newton, R.C., Prame, W.K.B.N. and Ravindra Kumar, G.R. (1987). Contr. Miner. Petrol., 96:225-244.





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