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# ,, NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS WARTIME REPORT ORIGINALLY ISSUED October 1943 as Advance Restricted Report 3J11 EFFECT OF PH ON STRENGTH OF RESIN BONDS By R. C. Rinker, F. W. Reinhart, and G. M. Kline National Bureau of Standards



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#### NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS

#### ADVANCE RESTRICTED REPORT

#### EFFECT OF DH ON STRENGTH OF RESIN BONDS

By R. C. Rinker, F. W. Reinhart, and G. M. Kline

#### **INTRODUCTION**

The increased use of resin-bonded plywood for structural parts of aircraft has made it necessary to determine the effect of various chemical properties of the resins on the strength properties of the resin bonds. Information of this nature is needed to utilize the materials properly in building satisfactory aircraft and to evaluate the causes of failures. Determination of the effect . of acid on the strength and aging properties of various types of resin bonds is one important phase of this work. This report presents the results of an investigation which was made to determine these relationships.

The degree of acidity or hydrogen ion concentration can conveniently bo reported as a pH value which is the logarithm of the reciprocal of the gram ionic hydrogen equivalents per liter; that is,  $p\overline{H} = \log 1/H^+$  per liter.<br>Water has a concentration of  $H^+$  ion of  $10^{-7}$  and of  $QH^$ ion of 10<sup>-7</sup> moles por liter or a pH value of 7, and is said to be neutral in reaction. The presence of an acid in a water solution increases the concentration of hydrogon ions. Honce the concentration of hydrogen ions in an acid solution becomes  $10^{-6}$ ,  $10^{-5}$ , or greator, and the pH value is less than 7. The presence of an alkali in a water solution increases the concentration of hydroxyl ions and decreases that of the hydrogen ions. Hence the concentration of hydrogon ions in an alkaline solution becomes 10<sup>-8</sup>, 10<sup>-9</sup>, or less, and the pH value is greater than 7. The product of the hydrogen ion concentration and the hydroxyl ion concentration is always equal to 10-<sup>14</sup> in aqueous medium at 25° C. The pH value has been used throughout this report to indicate the degree of acidity of the various specimens.

The two most commonly used types of bouding agents in the manufacture of resin-bonded plywood are the phenolformaldehyde and the urea-formaldehyde resins. Both types are cured oither by the "hot-sot" or the "cold-set" method.

Since the demarcation between "cold-set" and "hot-set" bonding resins has not been definitely established in the industry, the resins used in this project were classified according to the temperature required to cure the resin in a commercially practical period of time, as followat

- Class R. These resins do not require a higher degree of heat for curing than that available at ordinary room or factory conditions.
- Olass M. These resins require a degree of heat greater than that available at room or factory conditions, but not over 160° F  $(71^{\circ}$   $\circ)$ .
- These resins require a temperature Class H. greater than  $160^{\circ}$  F (71<sup>c</sup> C).

To obtain a satisfactory degree of cure of class R and some class M resins, it is necessary with most of the commercial resins to use very active catalysts. One of the most active catalysts for curing these types of resins is the hydrogen ion which is usually expressed in terms of pH units when the concentration is less than one molar.

It is an established fact that weed deteriorates rapidly in acidic media. It is also known that ureaformaldehyde resins are not as resistant to acid conditions as are phenolic resins. The werk reported herein was designed to determine the effect of the pH of the resin bond on the strength properties of the resin-wood composite since the failures may be in the resin, in the wood, or in both resin and wood. It should be noted. however, that the acid conditions in the resin-bonded birch panels tested are attributable to the ingredients in the resin-glue mixtures and not to the wood or any extraneous source.

This investigation, conducted at the National Bureau of Standards, was sponsered by, and conducted with financial assistance from, the National Advisory Committee for Aerenautics.

The authors wish to acknowledge the assistance given by Mr. B. M. Axilrod and Miss M. C. Fordney in supplying the data on strength properties herein reported.

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#### MATERIALS

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> A froup of commercial resins which are being used to a great extent in the manufacture of resin-bondod plywood nircraft was selected for this work. The commercial designations and the manufacturers of the resins, and the classification of the various resins and resin-oatalyst mixtures on the basis of the temperature required for curing, are given in table I.

The test panels were made with sliced birch veneers. having an average thickness of 0.01 inch. The thin ve-<br>neers were used to obtain a higher resin content than that normally used in aircraft plywood. Since the acidic conditions result from the resin, a high resin content would be expected to magnify the effect of the pH on the strength properties of the composite.

#### PREPARATION OF TEST PANELS

The resin glues were prepared according to directions received from the manufacturers and were applied to the birch veneers by means of rollers. This method produced resin films of uniform thickness on both sides of the ve-The veneers coated with the class H resins were neers. suspended from a drying rack and allowed to dry about 20 hours before assembling and pressing. The veneers coated with the class R and class M resins were assembled and pressed immediately after coating. Each panel consisted of 8 birch veneers arranged with the grain of plies 1, 3, 6, and 8 parallel to one another and with the grain of plies 2,  $\frac{1}{4}$ , 5, and 7 perpendicular to the face plies. All the test panols were pressed at approximately 100 pounds por square inch.

The birch venocrs used in each panel were conditioned nt 77<sup>0</sup> F (25<sup>0</sup> 0) and 50 percent relative humidity, and were weighed before the resin coating was applied. The completed test panel was also conditioned and weighed. The resin content of the test panel was then calculated by menns of the following equation:

Resin content, percent =  $Wt$ . of test panel -  $Wt$ . of conditioned voneors  $\times$  100 wt. of test panol

Three panels **were** prepared with each resin or reslncatalyst mixture. The panels made with class H resins were 12 by 12 inches in area; the panels made with class R and class M resins were 9 by 9 inches. The thicknessof the test panels were approximately 0.08 inch. The conditions used to cure the panels, the average densities, and the average resin contents are given in table I. The densities were determinod by weighing and measuring machined specimens.

#### TESTING PROCEDURE

#### Agiag

*Eaoh* test panel was out into quarters and treated as followst

- 1. One quarter section was not subjected to any aging treatment.
- 2. One quarter section was heated in a forcod-draft oven at  $176^\circ$  F (80° C) for 40 hours.
- 3. One querter section was subjected to a continuous<br>oven-fog cyclic accelerated seing test. The oven-fog cyclic accelerated aging test. cycle in this test consisted of the following:



The sections were exposed for a totnl of 200 hours in the oven aad 40 hours in the fog cabinet.

4. One.quarter section was exposed on tho roof of the In&ustrial Builaing on racks at an angle of 45° facing south. This exposure test is still in progress.

#### Determination of pH

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A thin film of the resins of class R and class H was cast on glass and allowed to dry for 15 hours at a tempora-<br>ture of 70° to 73° F (21° to 26° C). The resin film was then removed from the glass and ground to a finencss of 40 mosh. Two grams of the powdered resin were suspended in 10 millilitors of distilled water and the pH of the suspension was measured by means of a glass electrode aftor 15 minutos, and after 24, 48, 72, and 96 hours. Tho pH values were constant after 72 hours.

Films were prepared from the class H resins by casting thom upon a glass plato, using a knife blade to remove excess resin and make the thickness of the coating 0.02 inch or loss. The cast films were placed in a circulatingair oven at 150° F (65° C) until examination showed that most of the solvent had evaporated; this process required about 4 hours except in the case of Plaskon 107, which was cured after 3 hours at 150º F (65º C) and was not subjected to any further heating. This drying was followed by a cure in the even at 3000 F (148º C) until the films were hard and brittle, the latter operation requiring about 30 minutes. The hard, brittle films were pulverized in a sunll rock-crushing mortar and passed through a 40-mesh screen. The pH valuss of the powdered films were measured in the same manner as those of the class R and the class M  $f11$  $B.$ 

The acidity of the tast panols was determined by grinding a portion of the panel to 40 mesh in a Wiloy mill and suspending 1 gram of the powder in 5 milliliters of distilled water. Hoasurements of pH were made after 24, 48, and 72 hours. The pH values of the water susponsions wore constant after 48 hours.

The pH of the distilled water used in making the resin suspensions was 5.3. A fow of the resin films and powdered panels were also suspended in dilute hydrochleric acid solution of pH 4.5. The pH values of the acid susponsions are reported in table II and did not differ appreciably from those of the water suspensions. All the pH monsuromonts were made at a temporature of 77° F (25°C) with a glass electrode. The measurements reported are accurato to ±0.05 pR unit.

#### Strength Properties

The test specimens for determining the strength properties were cut from the quarter eootlons after the aging treatments. The specimens were aaohined and then conditioned at 77° F (25° C) and 50 percent relative humidity prior to testing. All the tests were made at 770 F (25° C) and 50 percent relative humidity.

The flexural modulus of elasticity was measured on an Olsen Stiffness Tester, Tour-Marshall design. Specimens 5 inches long and 0.5 inch wide were cut from the panels. Two measurements were mado on each cpecimen, one on each end, The test span was 2 inches long; the total bending moment applied to the specimen wae 3 lnchpounds . The angular deflections were plotted against the bonding moments and the deflection at a stress of 2500 pounds per squaro inch was determined from the curve. The modulus of elasticity ia flexuro then was calculated from the approximate expression

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\mathbf{J} = \frac{239.2 \text{ Pl}^3}{\text{D a h}^3}
$$

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where

- E modulus of elasticity in flexure
- P load
- 1 length of beam
- D deflection, degrees
- a width of beam
- and

h thickness of beam.

This expression was darived from the formula for the deflection of a cantilever beam with a concentrated load at one end.

The flexural strength was measured on specimens 1,0 inch long and 0.75 inch wide cut from the panels. The specimon was supported on two parallel supports with a

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span of 5/8 inch. The load was applied at the center of the span by a pressure piece similar to the supports. The odges of the support pieces and of the pressure piece wore rounded to 1/8-inch radius. The tests were made on a hydraulic testing machine with a head speed of 0.05 inch per minute. The machine was accurate to 2 percent of the lowest applied load.

The impact tests were made on an Izod impact machine of 2 foot-pounds capacity. Specimens 2.5 inches long and 0.5 inch wide wore cut from the panels.

The tonsile tests were made according to the method described in soction B-1 of Federal Specification L-P-406 for plastics. Type I specimens were used; the width of the reduced section was 0.5 inch. The tests were made on a hydraulic tosting machine with self-alining Templin The rate of head speed was 0.05 inch per minute. grips.

Shear specimens 4 inches long and 0.75 inch wide wore cut from the panols. A groove 1/8 inch wide and extonding through approximately  $4\frac{1}{2}$  vencers was milled on one face of the panel parallel to the 0.75-inch dimension. A similar groove was milled on the opposite face. The. grooves on the specimens used in the proliminary tests were 1/2 inch apart, but, since many tensile failures were obtained, the distance between the grooves was reduced to 1/4 inch on the lator specimons. The specimons wore broken on a hydraulic testing machine at a rate of loading of 200 pounds per square inch per minute.

#### Delamination

One strip 0.5 inch wide cut from each quarter section of each test panel was subjected to a delamination test. The strips were placed in individual 3-by 20-centimeter test tubes which contained distilled water previously heated to the boiling point by immersion of the tubes in a water bath. The tubes containing the test strips were loft in the bath of boiling water for 1 hour. On removal from the tost tubes the specimens were immersed in water at 77° F (25° C) for 15 minutes and then dried at 140° F  $(60^{\circ}$  C) in a forced-draft oven for 22 hours. This procodure constituted one cycle of the test. At the end of each crole the test specimens were bent over a mandrel of 8-inch radius. After five cycles the specimens were bent over a 4-inch mandrel. Observations regarding dolamination were made.

#### RESULTS OF TESTS

A preliminary investigation was made to obtain data for use in selecting the strength properties to be measured on all the test panels. Six panels were prepared with Tego film and six with Uformite 430 catalyzed with 10 percent ammonium chloride. These two materials vere selected to determine the effects of high and low pH conditions, respectively. One-half of each panel was tested unaged and the other half was subjected to an aging test prior to measurement of the strength properties. The strength properties measured in these preliminary tests were flexural modulus of elasticity, and flexural, impact, tensile, and shear strengths. The results are given in table II.

On the basis of the results obtained in these preliminary tests, the size of the tost specimens required. and an analysis of the stresses in the various tests, it was decided to employ the flexural, impact, and shear strengths for detecting the deterioration of the resinbonded birch plywoods. The critical pH values in the acid range - that is, the pH value below which serious degradation of the plywood may occur because of free acid, observed for the urea-formaldehyde and phenol-formaldehyde resin-bonded panels in the various physical tests, are summarized for convenionce in table III. Detailed results of the tests are presented in tables IV, V, VI, and VII. The results of the shear tests were not available at the time this report was propared and will be presented in a subsequent report.

#### DISCUSSION OF RESULTS

Use of the various commercial resins with their catalysts selected for this investigation resulted in pH values for birch plywood ranging from 1.7 to 8.4. (See table I.) The pH valuos for the test panels made from the urea-formaldehyde materials renged from 1.9 to 5.7; the pH valuos for the test panels made from the phenolic materials ranged from 1.7 to 8.4. Test panels were made also with a now resin, Laminac, which is neither a phenolic nor a uren-formaldehyde type. The pH values for the test panels made from Laminac ranged from 3.7 to 4.0.

The pH values of the birch plywood were not affected

by moderate baking or by exposure to cycles of heat and fog. This indicated that the acidic compounds determining the pH of the composite *did* not escape readily from the structure or did not react with the birch or its decomposition produots in such a way that they lost their . chemical identity. It would seem reasonable, therefore, to assume that the deterioration caused by pH would continue until failure occurred.

The results of the preliminary tests reported in table II indicated that the weakening of the resin-bonded birch plywoods was first evident in the flexural and impact strengths. Tentatively, the shear strength also appeared to be reduced early in aging treatments of the composite. In ordor to alarlfy thie latter *point, more* data on shear strength *are* being obtained. The flexural modulus of elasticity and tensile strength did "not appear to he markedly affected in short-time aging tests.

The floxural strength sf the urea-formaldehyde restnbonded birch plywood depended markedly on the pH of the composite. This is shown by the data *in* table IV and graphically in figures 1 and 2. These is apparently a critioal pH value between 3.El and 4.6 for birch plywood banded with urea-formaldehyde resins, below which optimum flexural strengths are not obtained even on unaged material. The oven-aged specimens with pH values of 3.6 and less underwent a greater proportionate loss of strength than those with pH values of 3.8 and more. The oven-fogagod specimens with pH values of J.g and less underwent, with one ezception, *n* greater proportionate loss of strength than those with pH values of  $4.6$  and mor.a. This Indicates that if optimum strengths are desired nnd if these strengths aro to ho retained on aging, the pH of tho composite should be greater than 3.8.

The flexural strength of the phenolic resin-bonded panels dld not show an exnet correlation with pH, but an examination of the values in tables IV and VII shows that the prosenoe of aoid aatalyst auusoa a decrease in this property in the unaged panels in every case. This decrease was noticed espealally with the pahels prepared with the oatabond resins 590 mnd 2000Z, wherein oonoontrated hydrochlorfo aoid catalysts were used. It is well known that hydrochloric acid has a decidedly deleterious effect on most woods. There is apparently a critical pH value for the initial flexural strength of plywood bonded with phenolic resin between 3.1 and 3.6. Exposure to

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both oven and oven-fog-aging treatments caused marked docreases in floxural strengths when the pH values wore 3.1 or loss. The critical limit of pH with respect to loss in strength on aging appears to be between 3.1 and 3.6 for the phenolic-resin-bonded birch panels.

The impact strength of the plywoods bonded with urea-formaldehyde resins was also found to be dependent on the pH. This is shown by the data in table V and graphically in figure 3. The critical value of the pH for the unaged specimens appears to be between 3.6 and 3.8. For the oven-agod and the oven-fog-aged specimens there appears to be a definitely greater less in strength below a pH of 3.6, but the specimens of 3.6 and higher pH also showed appreciable loss in strength on aging. This indicates that the loss in impact strength on aging can be attributed to both deterioration of the wood at low pH and deterioration of the resin over the whole pH rango investigatod. It should be noted that Plaskon 700-2 is a modified urea-formaldohyde resin having some of the charactoristics of a phonolic resin, which probably accounts for its greator initial impact resistance.

The impact strongth of the plywood panels bonded with phenolic resins shows the same goneral relationship to pH as the floxural strength values. In each case, the presence of acid catalyst caused a decrease in strength. The lowest values were again obtained with the catabond resins catalyzed with hydrochloric acid. The critical pH value is apparently in the range between  $3.1$ The critical pH for the oven-aged specimons is and 3.6. between 2.7 and 3.1. The even-fog-aged specimens with pH values of 3.1 and less underwont a greater proportionate loss in impact strongth than those with pH values of 3.6. and more. Since the oven-fog-aging treatment is considered to simulato natural aging moro closoly than the continuous dry heat, the critical pH for the phenolic resin. compositos should be considered as in the range of 3.1 to 3.6 on the basis of the impact tests.

The failure of the urea-formaldehyde resin-bonded materials in the delamination test is also affocted by the pH of the plywood. The critical pH value in this test appears to be botween 3.8 and 4.6 for both the unagod and the agod spocimens.

No failuro of the phonolic resin-bonded composites occurred in the delamination test. The unaged and aged

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spocimens with pH values of 3.1 or less were brittle in the final flexibility test on the 4-inch mandrel. With one exception, those with pH values of 3.6 or more wore floxible throughout this test.

The critical pH for the phonolic resins is not dofinitely established by the data in this report because phenolic resin films in the range of pH between 2.0 and 4.0 were not included in the tests. Further work is under way to establish closer limits on this critical pH. It is evident from the pH values obtained on the test panels that the pH of the resin-catalyst mixture will have to be such that the pH at the bond will be above 3.1. It appoars, in general that birch wood is subject to scrious deterioration when the pH at the bond is below about 3.5.

It is to be expected that there also will be a critical pH in the alkaling range above which resin-bonded birch would be subject to deterioration in strength. This point has not yet been established in the experimental work on this project.

#### CONCLUSIONS

1. The pH values of the birch plywoods are not markedly affected by moderate baking or by exposure to cycles of heat and fog.

2. The flexural and impact strengths, both initially and after aging, of birch plywoods bonded with urea-<br>formaldehyde resins are definitely affected by the pH. The critical pH value, below which optimum strengths are not obtained and deterioration upon aging becomes appreciable, lies between 3.8 and 4.6.

3. The flexural and impact strengths, both initially and after aging, of birch plywoods bonded with phenolic resins are definitely affected by the pH. The critical pH value, below which optimum strengths are not obtained and deterioration upon aging becomes appreciable, lies between 3.1 and 5.6. Hydrochloric acid has a decidedly deleterious effect on plywoods made with phenolic resine.

4. The delamination of birch plywoods made with urea-formaldehyde resins is affected by the pH. The

lower the pH, the fewer cycles required for delamination to oocur.

5. The delamination of birch plywoods made with phonolic resins Is not affected by the PH. When the pH is 3.1 or less, the materials are not as flexible as those with pH values of 3.6 or more.

National Bureau of Standards, Washington, D. C. August 6, 1943.

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#### TABLE I.- DESCRIPTION OF RESINS AND RESIN-PONDED BIRCH PANELS

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a. The resins are classified according to the temperature required to cure the resin. Class R includes those which<br>cure quickly at room temperature. Class M includes those which require a temperature above room temperature

#### TABLE II.- RESULTS OF PRELIMINARY TESTS MADE ON RESIN-BONDED BIRCH PLYWOODS





e. The pH of the water used in these tests wes 6.3; that of the dilute hydrochloric scid solution was 4.5. The pH of the water and dilute hydrochloric acid solution extracts of the birch veneers used in these tests was 4.

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# TABLE III.- SUMMARY OF CRITICAL pH VALUES FOR BESIM-BONDED BIRCH PLYWOOD<sup>6</sup>



Critical value of pH in the acid range for resin-bonded birch plywood

<sup>a</sup>The critical pH is the pH belov which serious degradation of the plywood may occur because of free acid.

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#### TABLE IV. - EFFECT OF pH ON FLEXURAL STRENGTH OF RESIN-BONDED BIRCH PLYWOOD



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#### TABLE V.- EFFECT OF pH ON IMPACT STRENGTH OF RESIN-BONDED BIRCH PLYWOOD

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#### TABLE VI.- EFFECT OF pH ON DELAMINATION OF RESIN-BONDED BIRCH PLYWOOD

are specimens were subjected to 5 cycles of immersion in boiling water and drying, described on page 8. Figure in parenthesis refers to cycle in which observation was made. Abbreviations are as follows:

- $D =$  delaminated<br>
SD = slightly delaminated<br>
ND = no delamination<br>
B = brittle<br>
F = flexible
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#### TABLEVI -- EFFECT OF CATALYST ON FLEXURAL AND IMPACT STRENGTHS OF RESIN-BONDED BIRCH PLYWOOD

a Decrease in strength for the unaged, oven-aged, and oven-fog-aged panels, respectively, is calculated on the besis of the strength of the unaged, oven-aged, and oven-fog-aged panels, respectively, made without catalyst.

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# Urea-formaldehyde

Figure 1.

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Figure 2.





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Fig. 3

