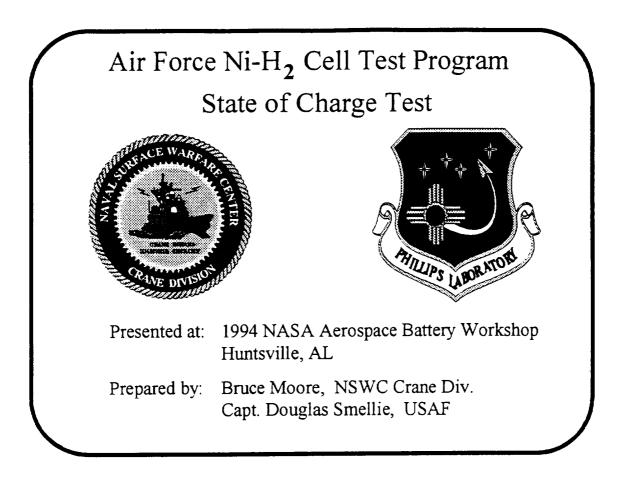
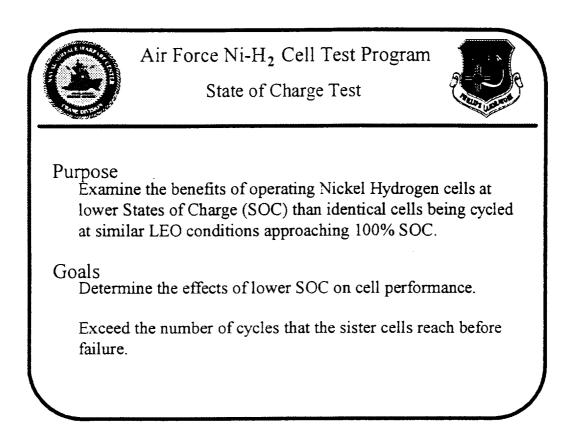
1995-120369

N95-26789



Nickel Hydrogen cells are being cycled under a LEO test regime to examine the benefits of operating the cells at lower States of Charge (SOC) than typically used. A group of four cells are being cycled using a voltage limiting charge regime that limits the State of Charge that the cells are allowed to reach. The test cells are being compared to identical cells being cycled at or near 100% State of Charge using a constant current charge regime.



Four 50 AmpHr 3.5" diameter cells manufactured by Eagle Picher in Joplin Mo. are being used for the test, part numbers RNH50-43 and RNH50-53. RNH50-43 uses a back-to-back stack design with a 26% KOH electrolyte concentration. RNH50-53 uses an alternating stack with an electrolyte concentration of 31% KOH.

Each of the two designs were originally split up into two packs of ten cells each, 3314E and 3214E. They are running an identical constant current test regimes with a C/D ratio of 1.03 to 1.04 at 40% Depth of Discharge and 10 degrees C. Approximately one year or 5000 cycles later the four cell SOC test pack, 3001C, was started. Two cells from each design were combined into one pack. The charge and discharge for the SOC test pack are identical to the original packs with a voltage limit placed on the charge cycle that will cause the current to taper towards the end of the charge.

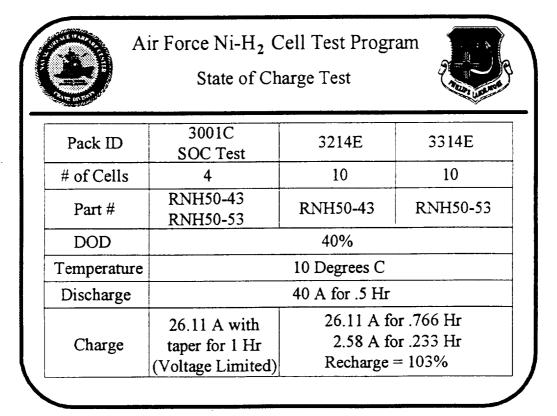


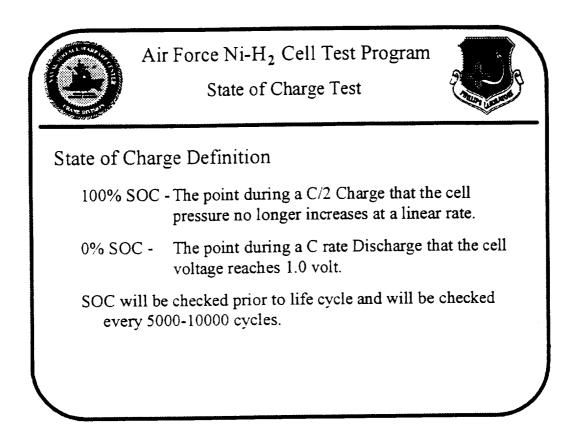
Air Force Ni-H₂ Cell Test Program



State of Charge Test

Manufacturer	Eagle Piche	r, Joplin
Capacity	50 Am	pHr
Size	3 1/2	
Separator	Asbest	OS
Part # · j	RNH 50-43	RNH 50-53
Stack Configuration	Back to Back	Alternating
KOH Concentration	26%	31%

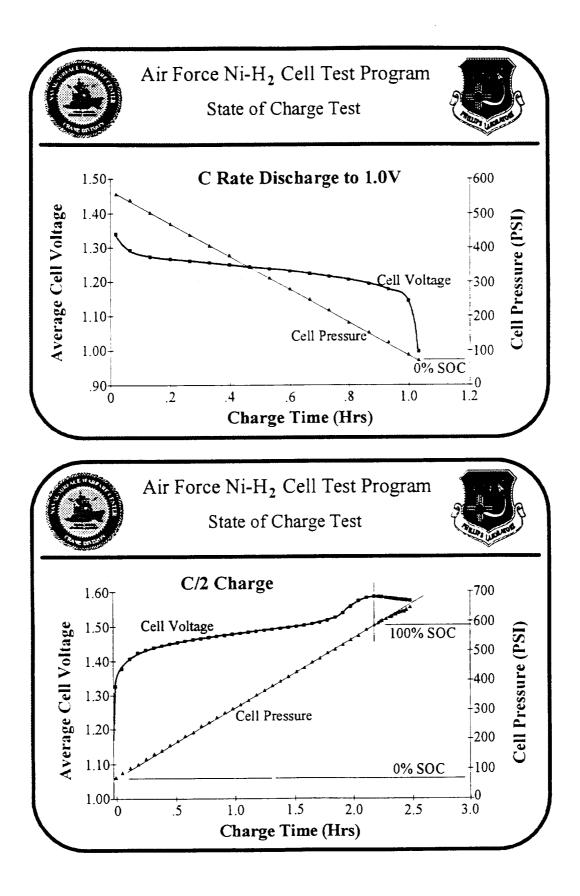


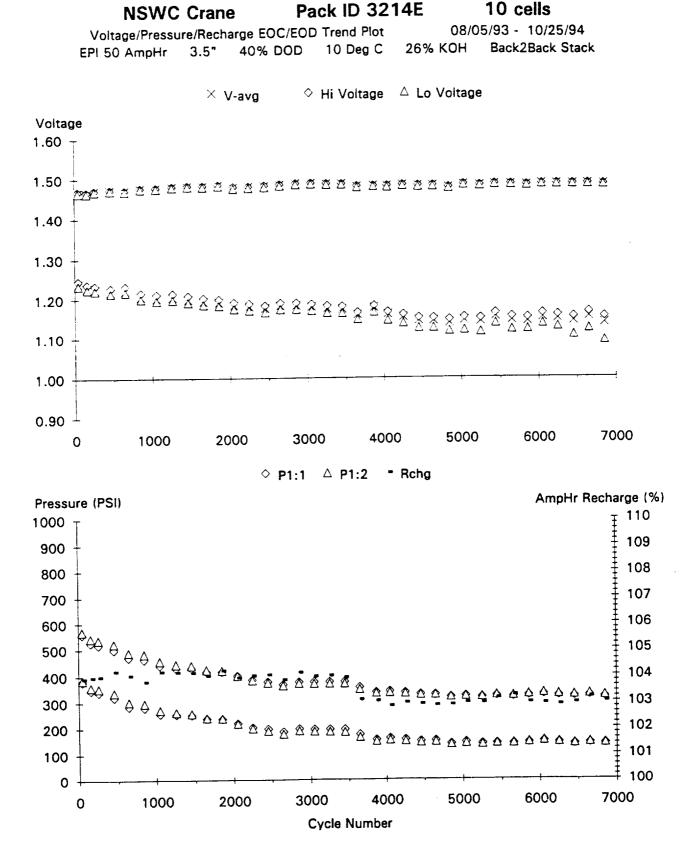


For the purposes of this test, 100% state of charge is defined as the point during a C/2 charge that the cell pressure no longer increases at a linear rate. 0% state of charge is the point during a C rate discharge that the cell voltage reaches 1.0 volts. Prior to starting life cycle, the four cells destined for the SOC test were cycled to find the zero and one hundred percent SOC points. According to the results the pressures related to those points are 80 and 590 psi respectively. Although this data is probably accurate for the cell in its current state it is not useful information for the purposes of life cycle testing.

An examination of the Trend Plot for 3314E shows that at the beginning of life, the End of Charge (EOC) pressures were at the same 590 psi for the SOC test cells. After only 1000 cycles the EOC pressures were reduced to approximately 425 psi. It appears that changes occur very quickly during the first 1000 cycles of a life cycle regime. Since the SOC test cells seem to follow the characteristics of their sister cells, the 425 psi EOC pressure value was assumed for 100% SOC.

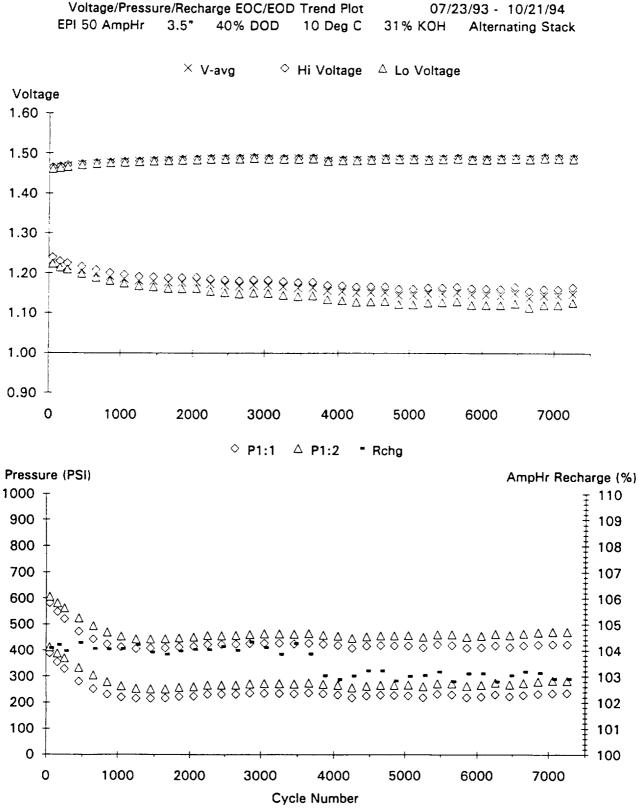
The target SOC for the test cells is 60 to 70%. This value was chosen to keep the cells at significantly below 100% SOC and to provide for a reserve capacity at the end of discharge, in this case 20 to 30% of rated capacity. Assuming the previously stated values of 425 psi for 100% SOC and 80 psi for 0% SOC, the pressure should be maintained at 304 psi for 65% SOC.

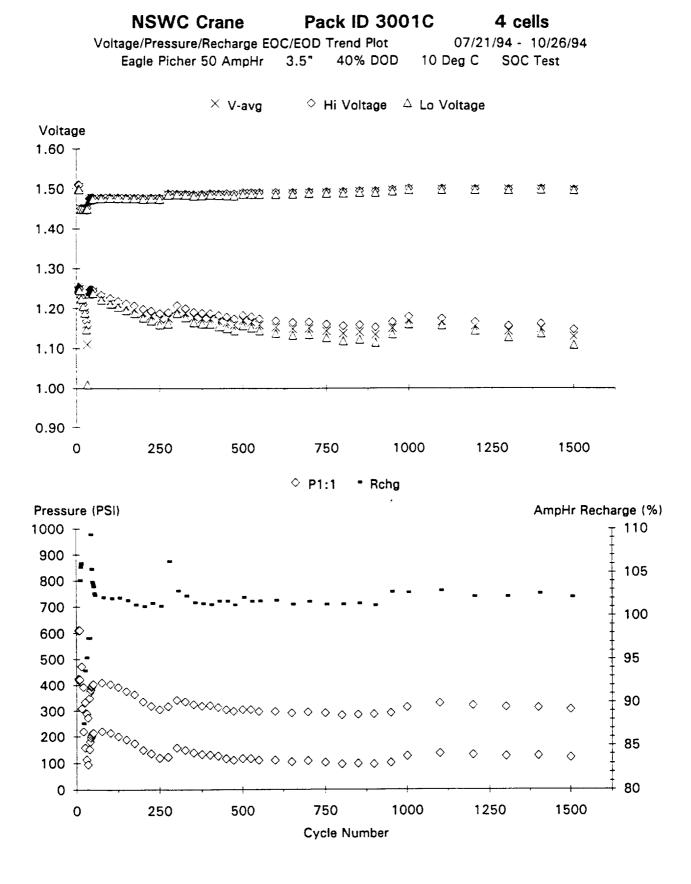




1994 NASA Aerospace Battery Workshop

Nickel-Hydrogen Design Session





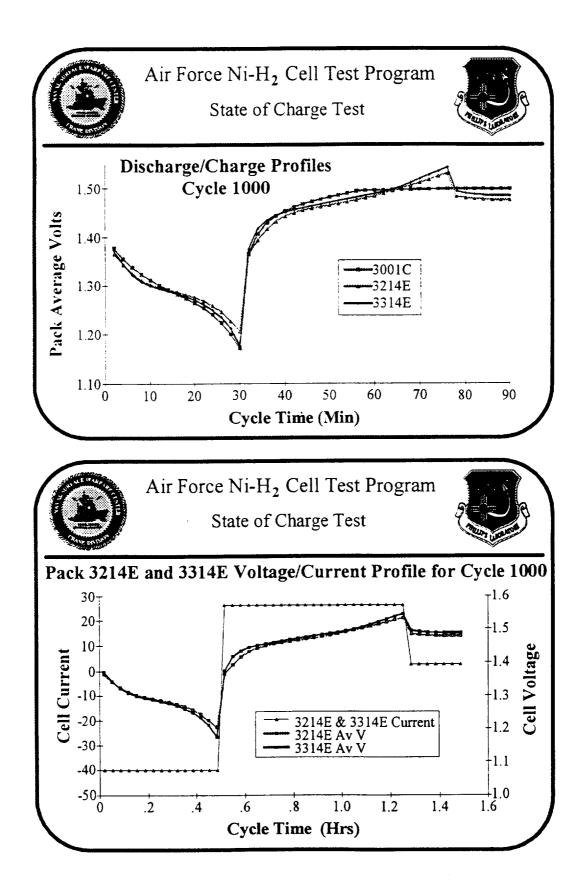
1994 NASA Aerospace Battery Workshop

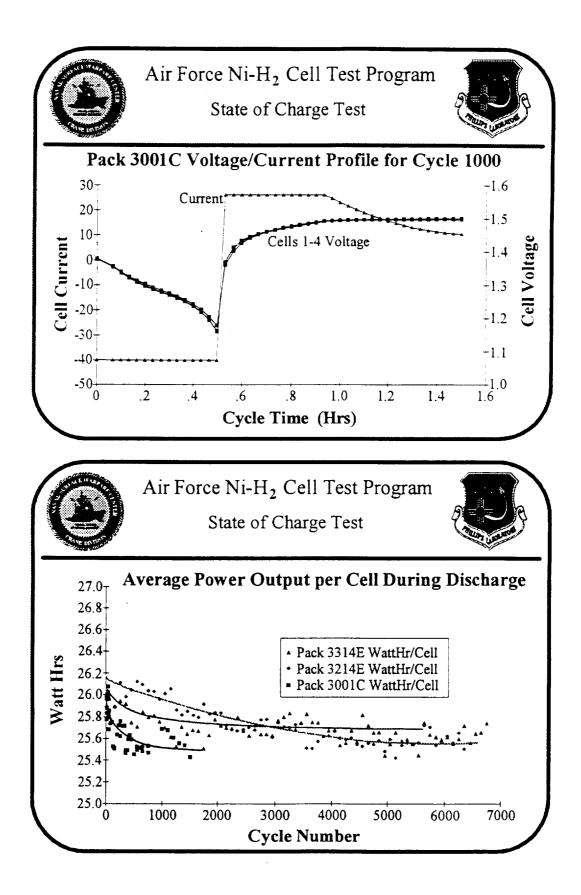
Nickel-Hydrogen Design Session

DULT	State	of Charge T	est	A COLORED ST
		Cycle Cell Ca Rate to 1.0 Vo	-	
	RNH:	50-43	RNH	[50-53
Temp	3314E	3001C	3214E	3001C
-5 °C	62.94	60.84	65.69	65.23
0 °C	68.64	67.33	74.14	75.60
10 °C	58.10	55.68	64.98	62.41
20 °C	51.58	52.30	59.15	58.78
30 °C	49.37	49.83	53.56	55.11

The SOC test cells will be cycled with a C/2 charge and C rate discharge again at 5000 cycles to determine the pressure values for 0 and 100% SOC. The SOC/pressure relationship will also be checked again every 5000 to 10000 cycles.

Prior to life cycle testing the four SOC test cells were stored discharge, open circuit at 5 degrees centigrade for one year. Comparison of the sister cells that started life cycle testing one year before the SOC test cells show very little difference in the capacities. The reported capacity values for the SOC test cells in pack 3001C are after the one year stand. The capacity was not checked when they were received.







Air Force Ni-H₂ Cell Test Program



State of Charge Test

Concerns

Sample is small. Only four cells are being used for the comparison of the SOC charge regime.

Possibility of voltage divergence because of the different cell designs.

Possible effect of capacity checks on life of cell.

MARTIN MARIETTA ASTRO SPACE



bу

D.J. Keys, G.M. Rao, H.E. Wannemacher, R.B. Wingard NASA/GSFC

C.W. Bennett, W.M. Gibbs, E.W. Grob and A.F. Mucciacciaro MARTIN MARIETTA ASTRO SPACE

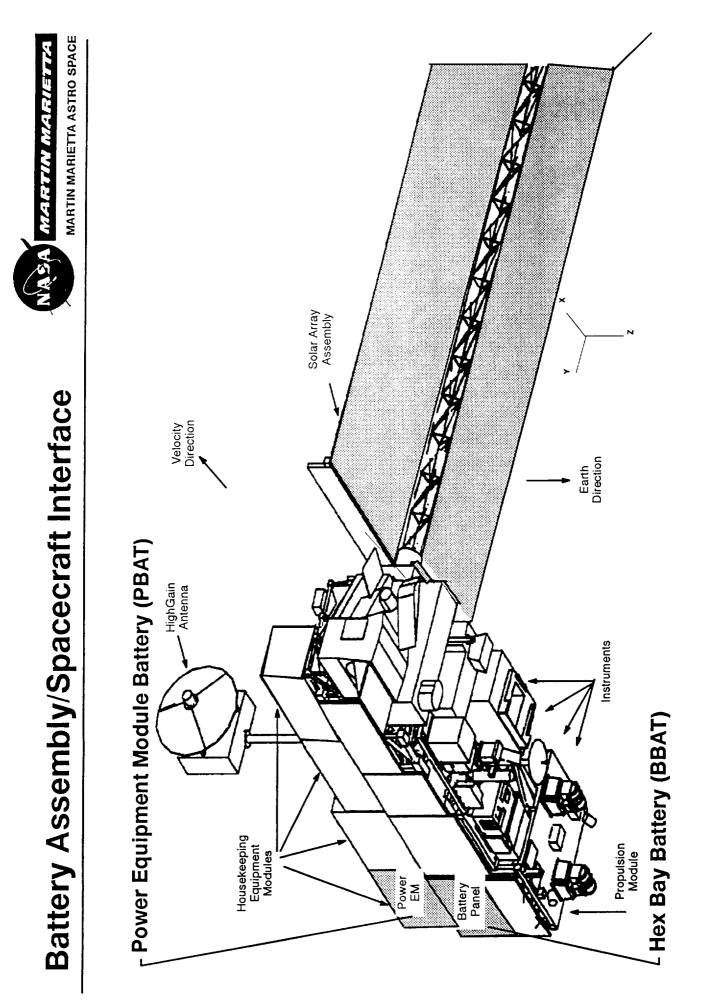
(Performed under NASA Contract NAS5–32500)





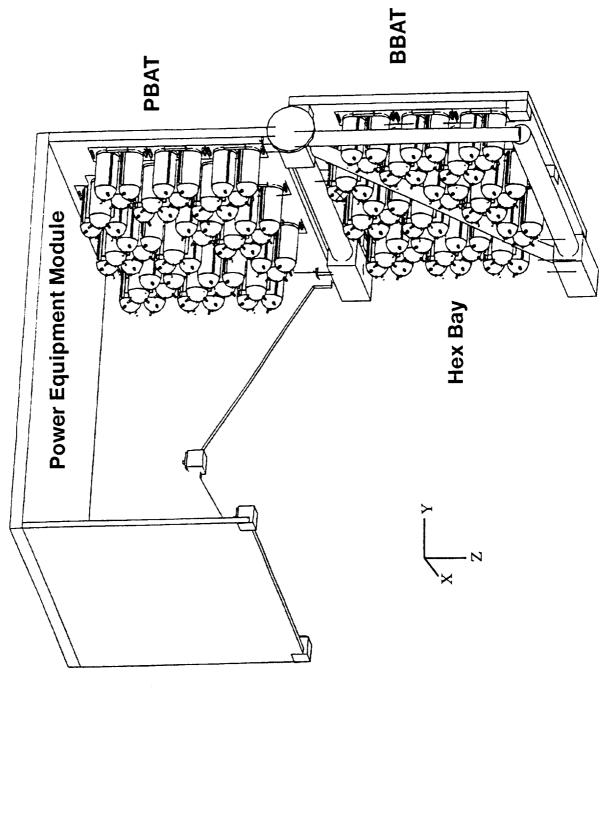


- Spacecraft Interfaces
- **Top Level Requirements** •
 - Design Summary Electrical •
 - - Thermal
- Mechanical
- Battery Assembly
- **Design Drivers**





Integrated Battery Assemblies



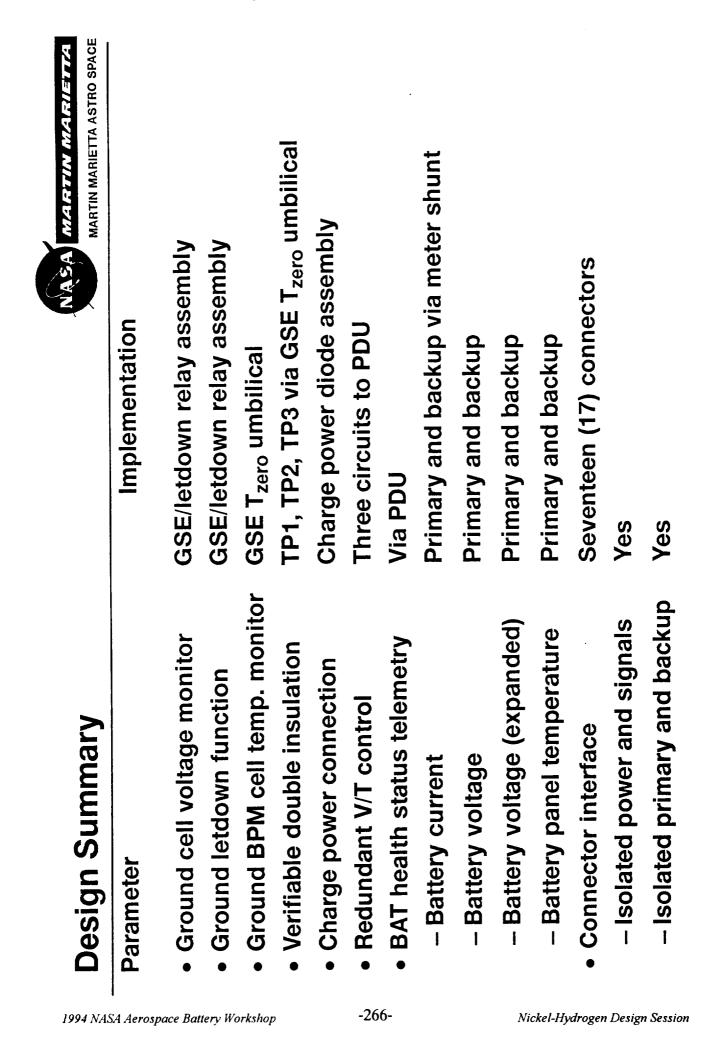
Top Level Requirements	NASA MARTIN MARIETTA MARTIN MARIETTA ASTRO SPACE
 Electrical Requirement 	Actual or Prediction
 Capacity (Amp Hours Minimum) 	
58.8 at 0°C	65.0
57.0 at 10°C	64.5
– – 50.0 at 20°C	54.5
42.0 at 30°C	51.5
– Voltage	
54.0 to 89.1	70.2 to 80.4 (Max Science)
54.0 to 89.1	71.8 to 80.9 (Survival and Safe)
 Depth of Discharge 	
– – 30% Max	19.1% (Max Science, 54 cells)
– – 35% Max	19.6% (Max Science, 53 cells)
– – 30% Max	10.0% (Survival and Safe)
– Current	
– – 30 Amps Max Discharge	20.7 Amps (Max Science, 53 cells)
– – 23 Amps Max Charge	12.2 Amps (Max Science, 53 cells)

Top Level Requirements	MARTIN MARIETTA MARTIN MARIETTA ASTRO SPACE
 Mechanical Requirement 	Actual or Prediction
– Mass (Maximum)	
310.1 (PBAT)	299.7 (10.4 lbs Contingency)
–– 315.9 (BBAT)	304.7 (11.2 lbs Contingency)
 Structural Load Cases 	
 – Launch acceleration environment 	ent
 – Qualification level acoustic loading 	iding
 Thermal Requirement 	Actual or Prediction
 Operating Temperature Limits 	
5°C to 10°C	2.6°C to 4.7°C (Min Science)
5°C to 10°C	1.2°C to 3.9°C (Max Science)
Thermal Gradients	
Cell to Cell 3°C max	2.4°C (EOL Cold)
Stack to Dome 7°C max	2.1°C (EOL Cold)
 – Dome to Dome 10°C max 	2.3°C (EOL Cold)
 – PBAT to BBAT 3°C max 	1.5°C (EOL Cold)

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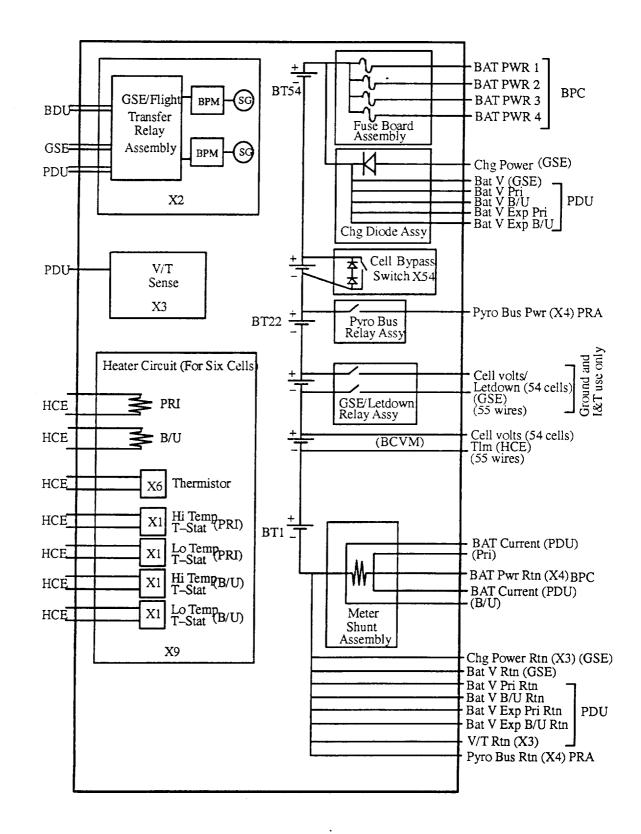
Design Summary	MARTIN MARTIN MARTIN MARTIN
Parameter	Implementation
 Four fused outputs 	Fuse Board
 Isolated pyro bus 	Pyro bus relay assembly
 Conductive thermal design 	Cell/sleeve assembly & baseplate
 Open cell protection 	Bypass switch assembly (one per cell)
 Cell pressure telemetry 	BPM (4 per BAT)
 – GSE/Flight isolation 	GSE/Flight transfer relay assembly
 Heater control 	Separate primary and backup circuits
 multiple cells per circuit 	Six cells per circuit
- Nine primary circuits	Through HCE5A
 – Nine backup circuits 	Through HCE5A
 Thermistor control 	one thermistor per cell (controls Pri. & B/U)
 Over and under temp. protectio 	Over and under temp. protection High and low temp. T-stats/circuit
 Flight cell voltage telemetry 	Through HCE5A (for each cell)
 Flight cell temp. telemetry 	Through HCE5A (for each cell)

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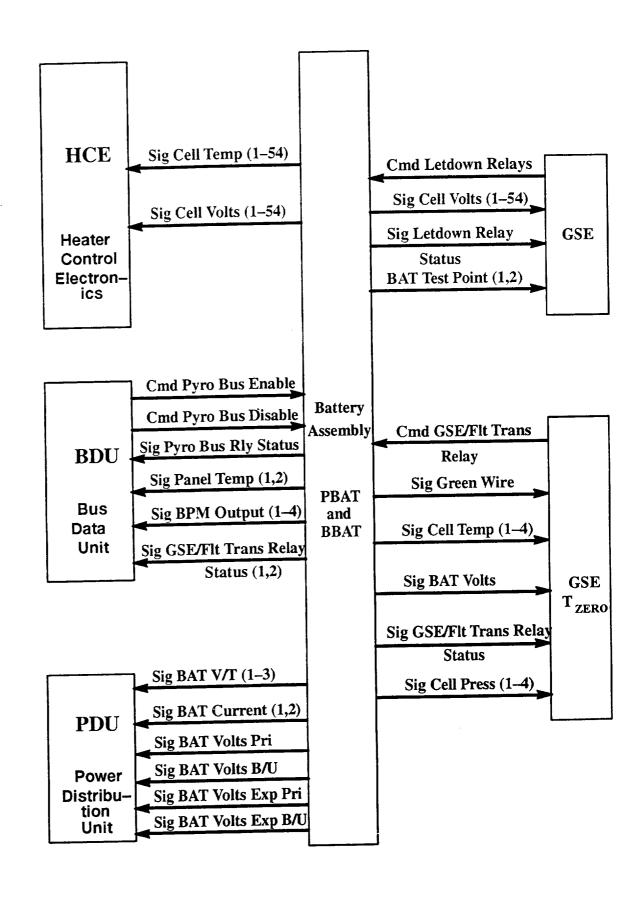


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Functional Block Diagram

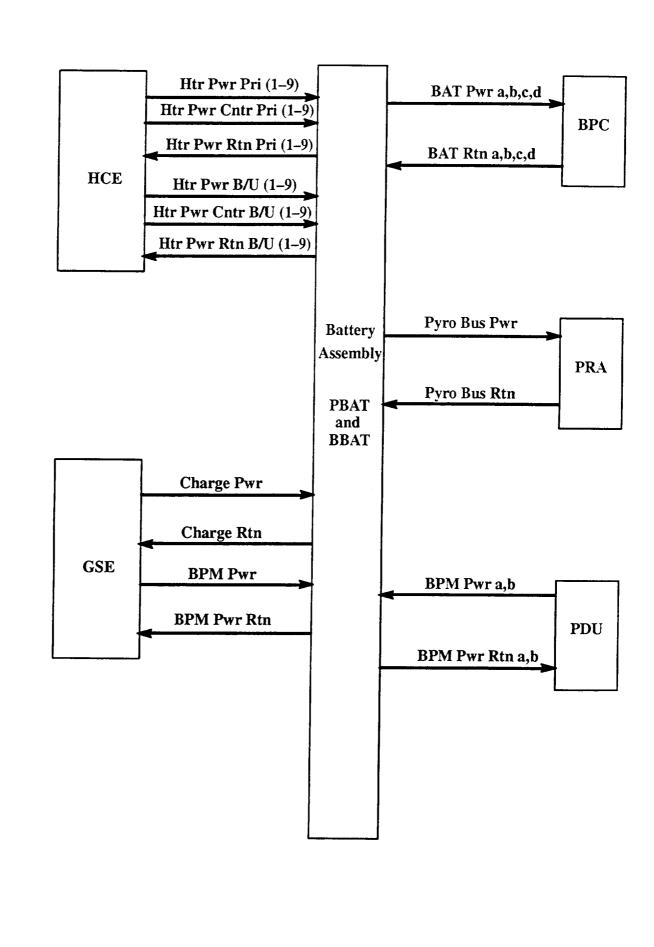


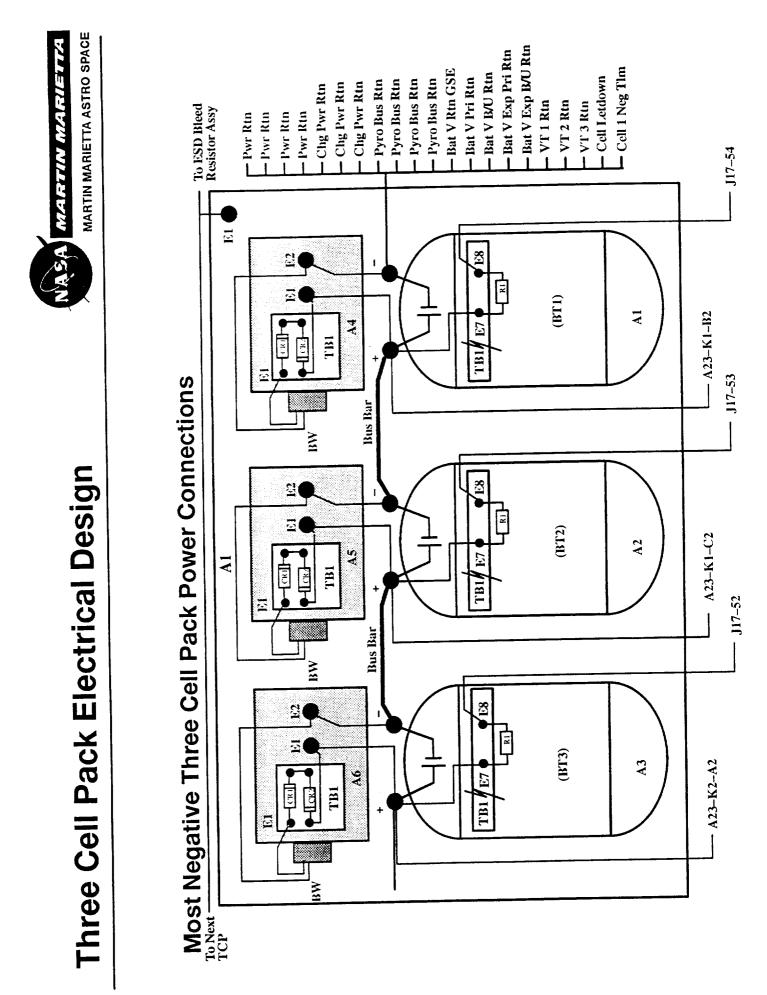
MARTIN MARIETTA ASTRO SPACE Signal, Command and Telemetry Interfaces 1

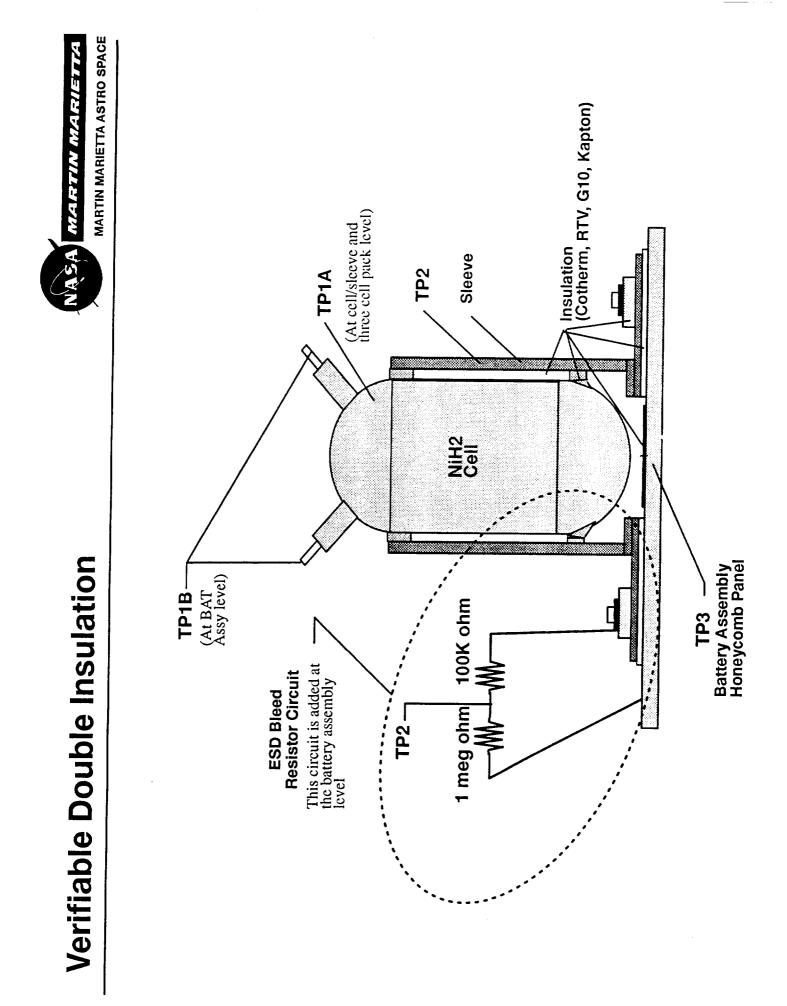


Signal, Command and Telemetry Interfaces⁴

MARTIN MARIETTA ASTRO SPACE

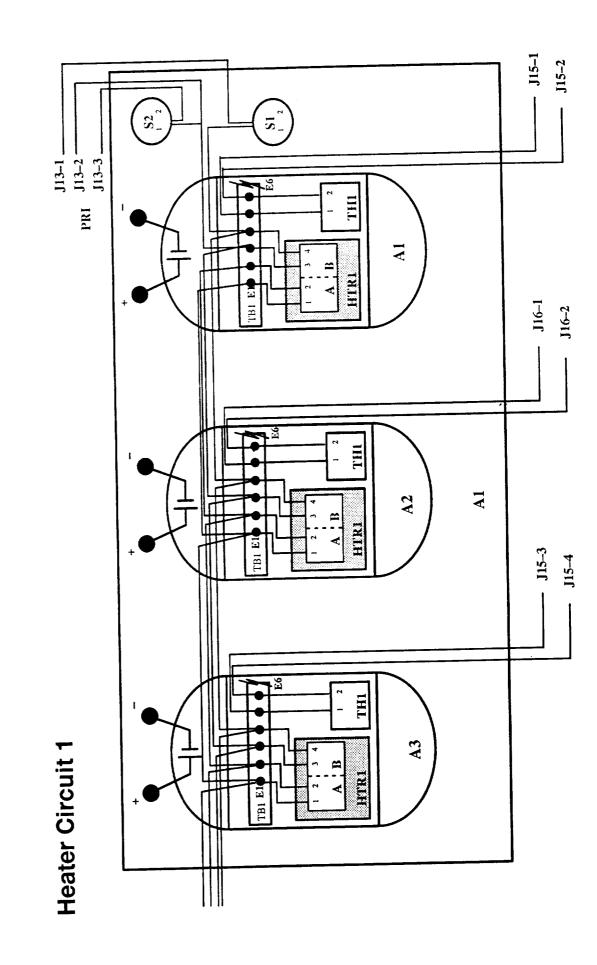






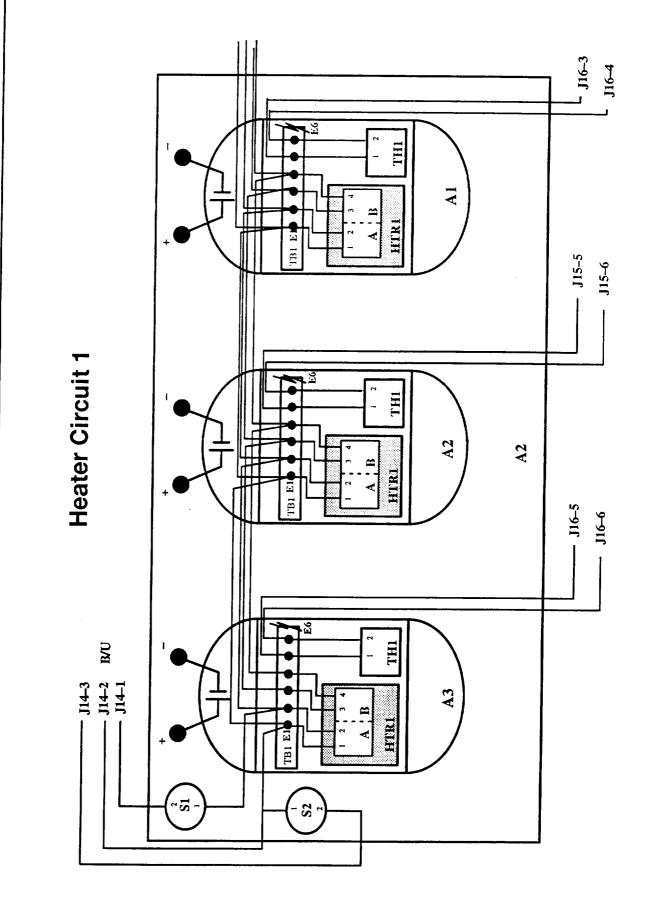


Three Cell Pack Heater Circuits





Three Cell Pack Heater Circuits

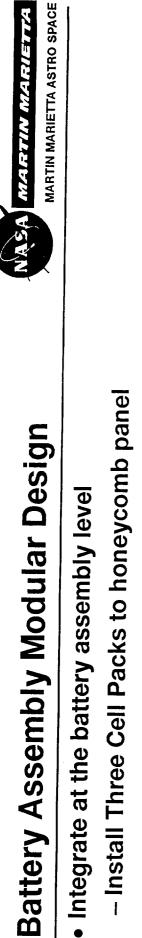


Battery Assembly Modular Design	NASA MARTIN MARIETTA MARTIN MARIETTA ASTRO SPACE
 54 Cell/Sleeve assemblies 	
 – 54 Bypass switch assemblies 	
 18 Three cell pack assemblies 	
 Electrical subassemblies 	
 4 Battery Pressure Monitor Assembly (BPM) 	
 – 1 Meter Shunt Assembly 	
 – 2 GSE/Flight Transfer Assembly 	
 – 1 GSE/Letdown Relay Assembly 	
 – 1 Pyro Bus Relay Assembly 	
– 1 Charge Power Diode Assembly	
 1 ESD Bleed Resistor Circuit Assembly 	
 Cover/Connector Box Assembly (with 17 connectors) 	
 – 1 Fuse Board Assembly 	
 Built and test all subassemblies 	

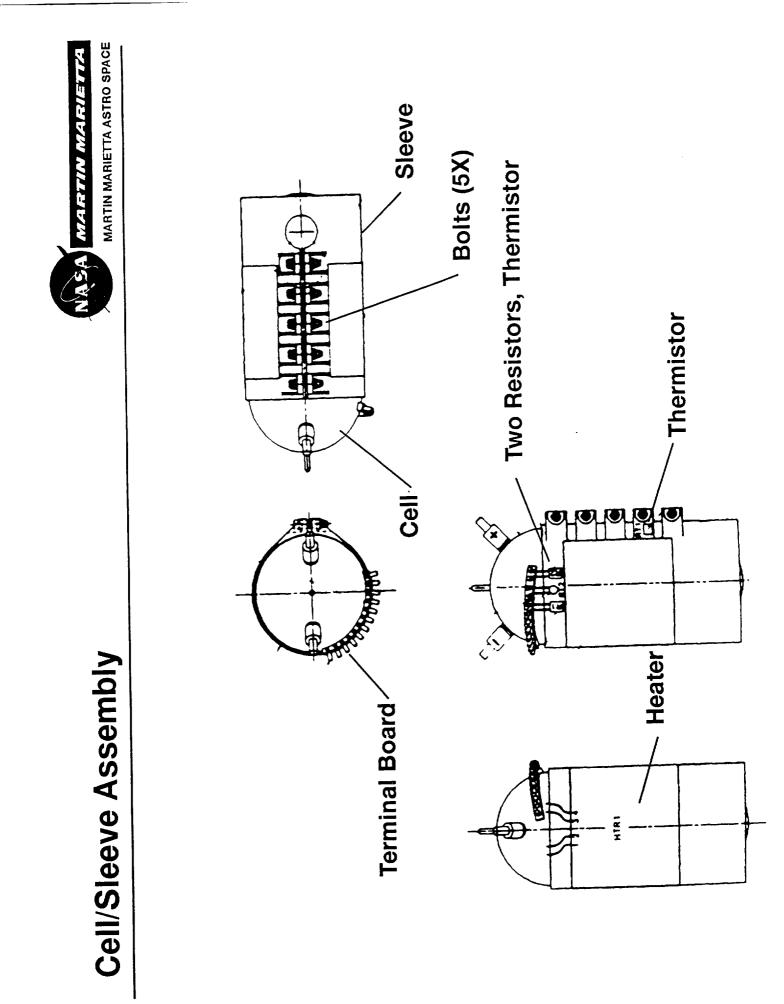
 Three cell pack assemblies - Electrical subassemblies

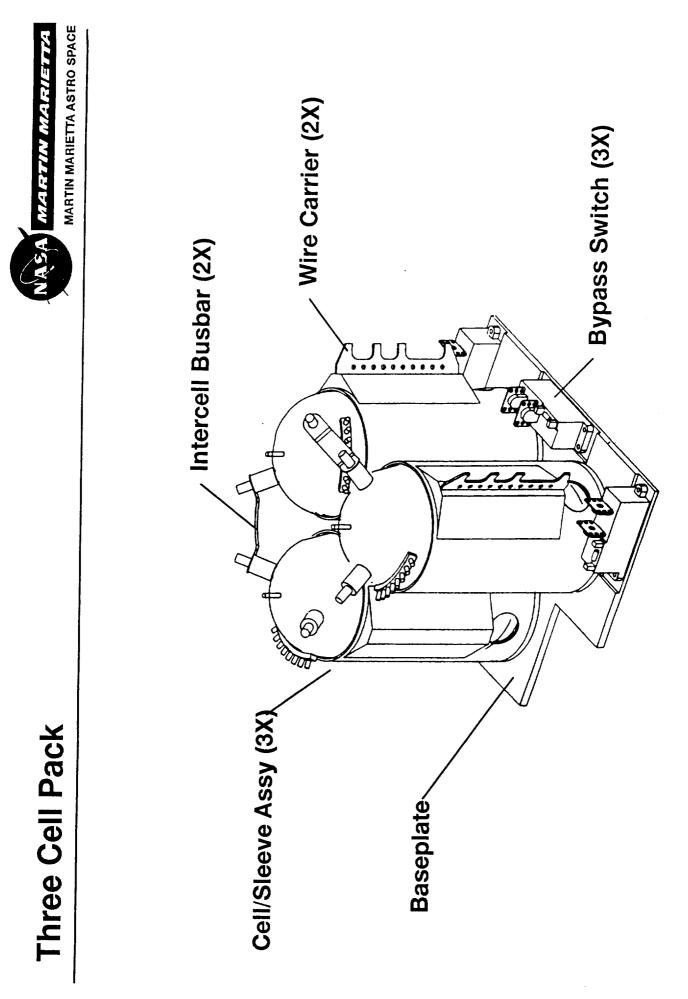
Cell/sleeve assemblies

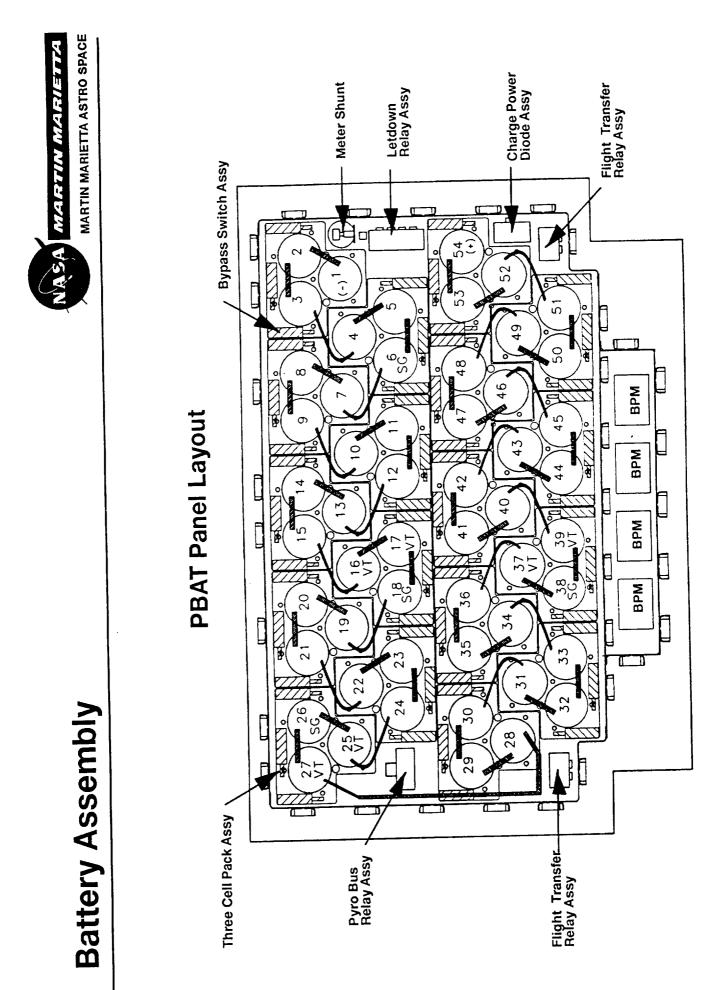
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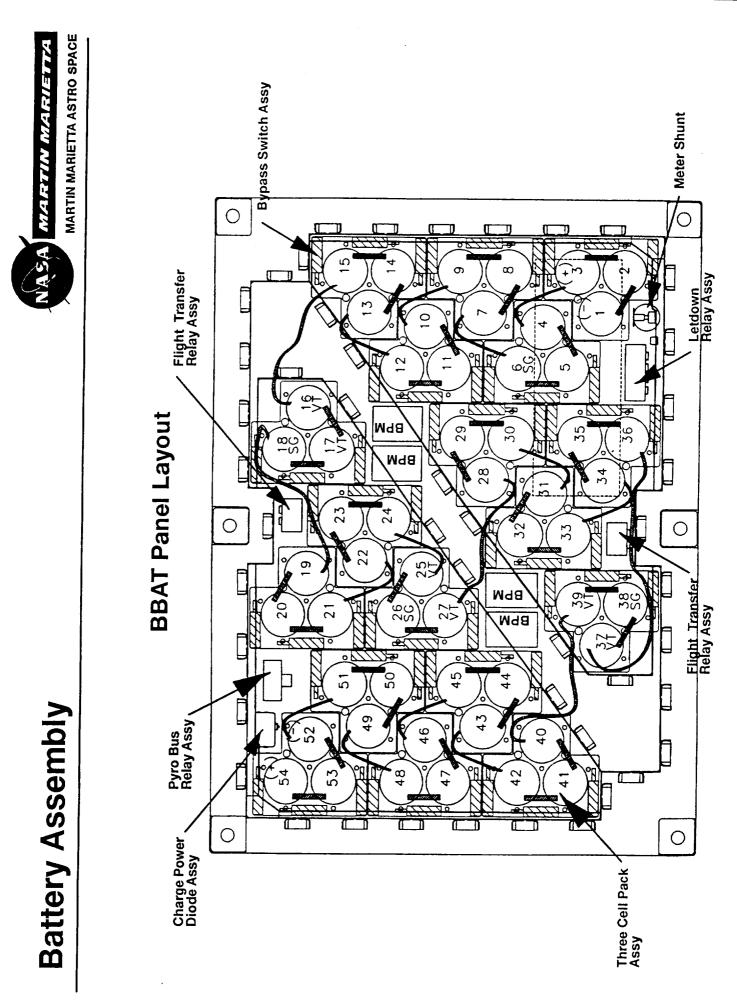


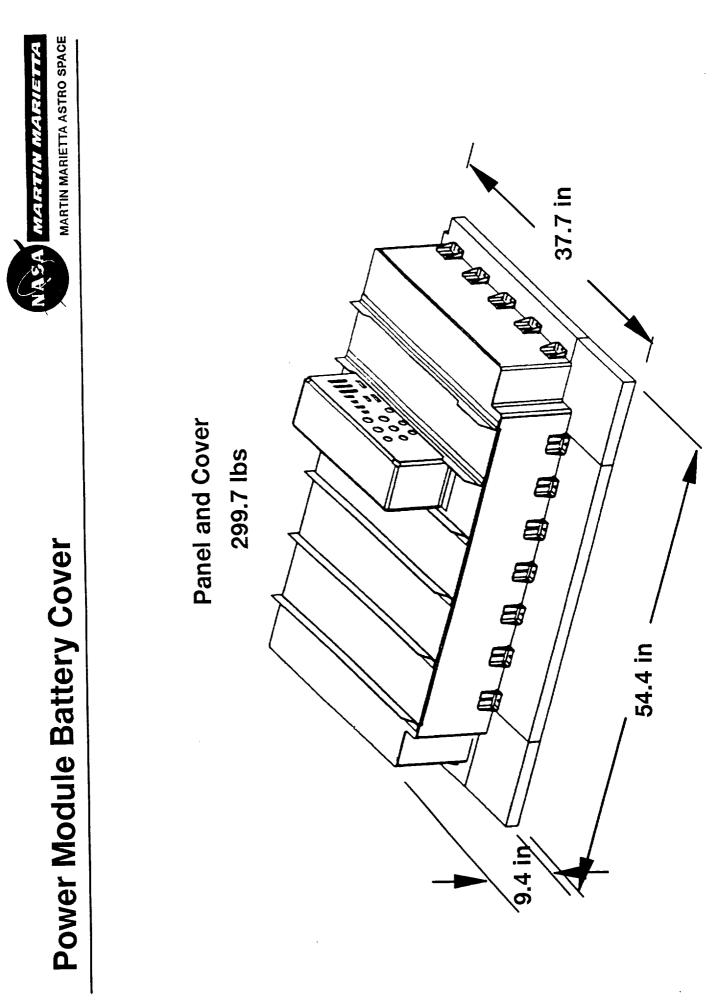
- Install electrical subassemblies to honeycomb panel
- Complete final point to point wiring
- Route wires to connectors
- Install battery cover
- Tape all cover seams with copper tape for EMC requirement





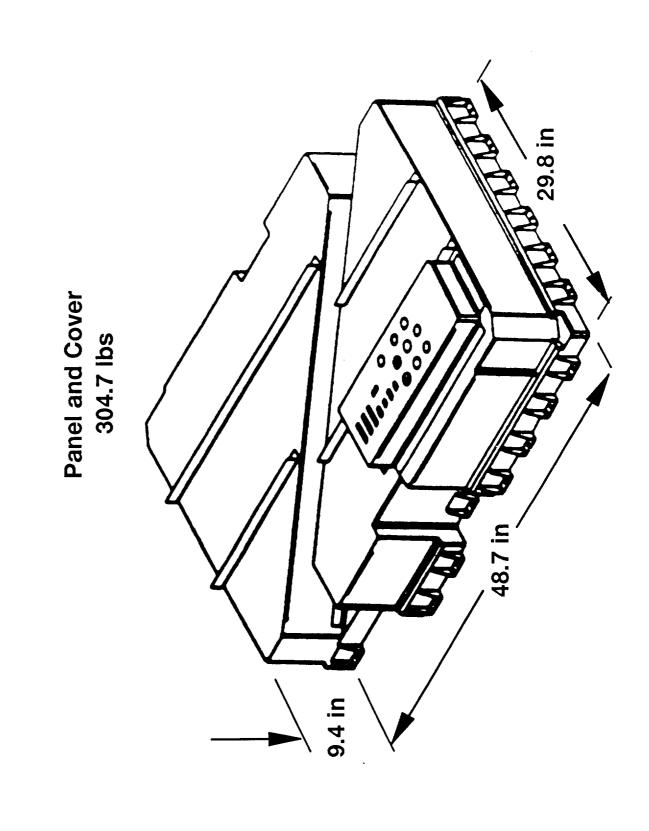


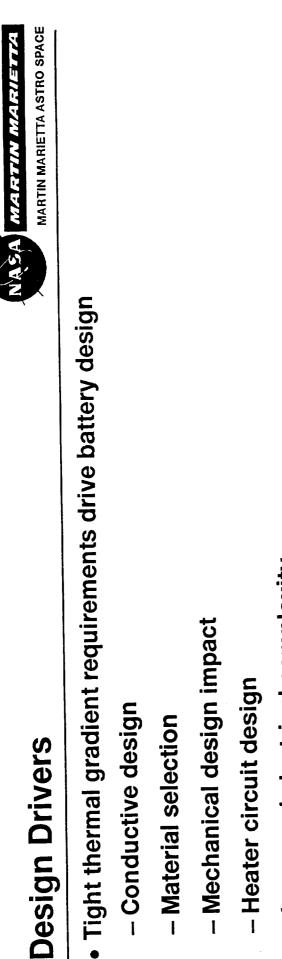










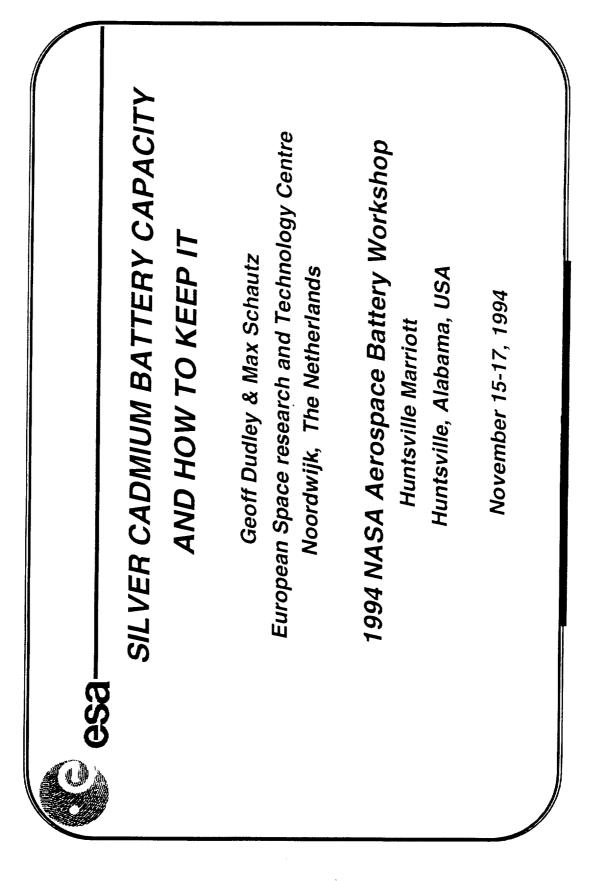


- Increased electrical complexity
- Electrical Complexity
- modular design approach
- individual cell voltage and temperature telemetry
- 17 connectors with 428 wires for spacecraft interface
- open cell protection device
- EMC
- .063" thick cover required
- large weight impact
- gold plated sintered washers
- copper tape at seams

Advanced Technologies Session

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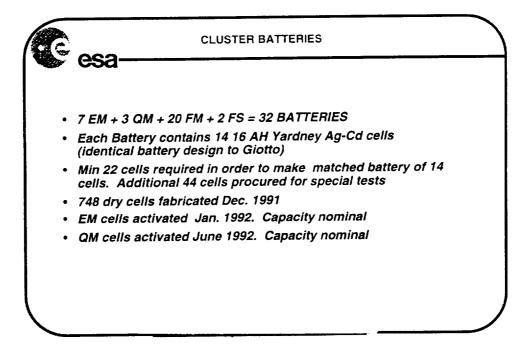
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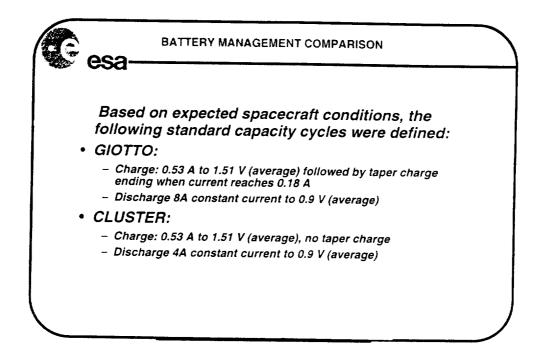
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esa		
	entific spacecrafi	used on a considerable where high levels of ry:
SPACECRAFT	LAUNCH DATE	Number of batteries
HEOS -1	Dec. 1968	1 (5 Ah)
HEOS - 2	Jan 1972	1 (5 Ah)
GEOS 1	April 1977	1 (16 Ah)
ISEE -B	Oct 1977	1 (10 Ah)
GEOS 2	July 1978	1 (16 Ah)
GIOTTO	July 1985	4 (16 Ah)
CLUSTER (x4 S/C) Dec. 1995	4 x 5 (16 Ah)

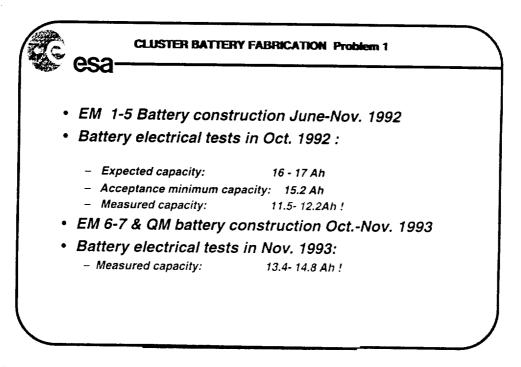
Yardney silver cadmium cells have been used on all the above spacecraft. Geos, Giotto and Cluster use the YS16(S)-4 16 Ah cell. The battery design for Cluster is identical to Giotto. The 4 Cluster spacecraft will be in elliptical 66 hour polar orbits. During the 2.5 year mission the batteries will see a total of about 35 charge-discharge cycles with a maximum depth of discharge of 65%. 32 of these will take place during 4 short eclipse seasons which occur roughly every 6 months.



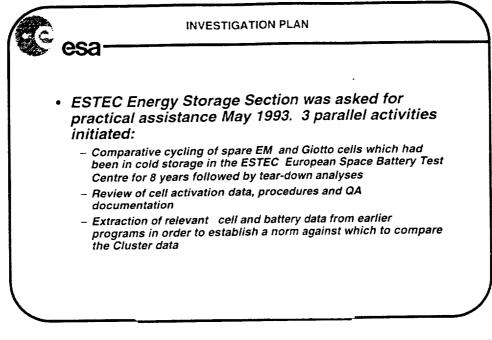
Cell activation entails filling with electrolyte, 3.5 formation cycles, cell sealing, last formation discharge, 5 stabilization cycles, and 2 acceptance cycles. This is carried out in groups of 22 cells. Matching cycles are then performed and the 16 best matched cells shipped for battery manufacture (14 cells + 2 spares).



In contrast to the Giotto program, taper charging is not implemented for Cluster. Both programs employ a trickle charge when the battery voltage falls below 19.43 (1.39 V/cell).



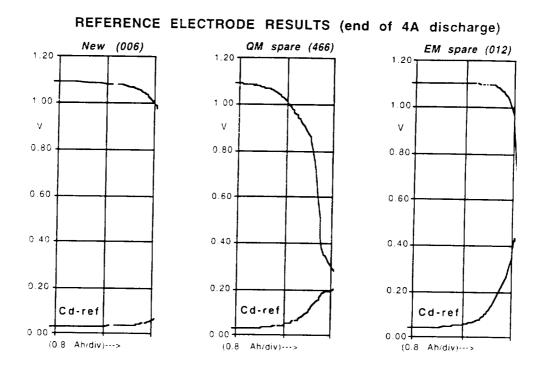
Cells were not cycled by the battery manufacturer until after battery construction, which involves potting the cells into the aluminum structure. The low capacity was found both before and after battery vibration and thermal vacuum tests.



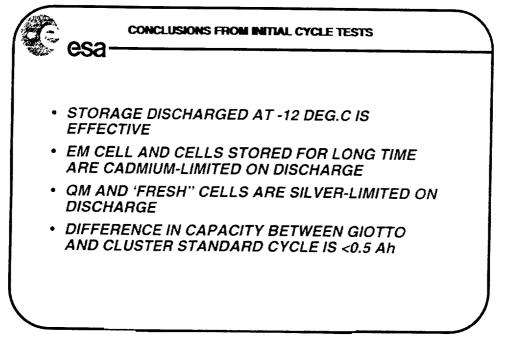
Cells were immersed in a water bath at 20 deg.C for all electrical tests. The reference electrodes were pieces of cadmium wire, mechanically cleaned and washed in hydrochloric acid, water and potassium hydroxide immediately before insertion into the cell. Inside an oxygen - filled glove box, cell tops were pierced with a 1 mm drill and a 0.4 ml electrolyte sample taken with a syringe. The reference electrode, sheathed in PTFE tubing except for the last 1 cm, was inserted and the cell re-sealed with epoxy cement.

CELL	Storage time	Capacity (Ah)	Chglimiting electrode	Dischlimiting electrode	K2CO3 (wt%)
OLD CELL 022	10.5 y	11.8	Ag	Cd	13.7
GIOTTO 451	8 y	12.4	Ag	Cd	7.6
CLUSTER EM 012	20 m	13.8	Ag	Cd	10.0
CLUSTER QM 466	14 m	15.8	Ag	Ag	9.8
FRESH CELL 006	1 m	18.5	Ag	Ag	9.8

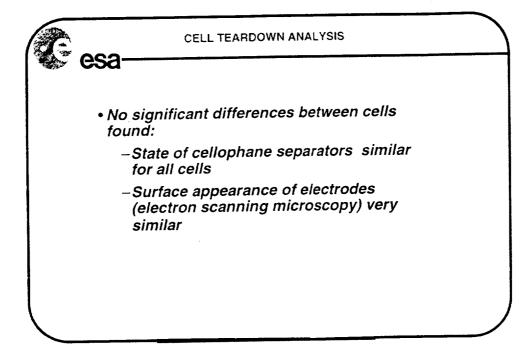
It is noteworthy that the lower capacity cells are all cadmium - limited on discharge. As cells age, carbonate ions build up in the electrolyte as a by-product of oxidation of the cellophane separators. The level of carbonate present was considered an important parameter both as an indicator of the extent of the attack and because high levels are known to impede operation of the cadmium electrodes. The concentrations measured in the test cells were unexpectedly high, but the activation of cell 006 at ESTEC revealed that the levels already reach about 6 wt.% at the time cells are first sealed. This is formed presumably during the deliberate overcharge that takes place during the first formation charge.



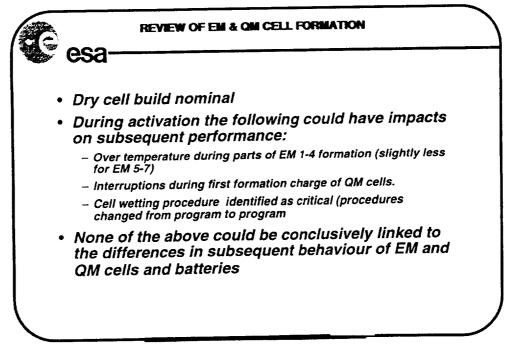
Cycles with reference electrodes were extended to 1.56V on charge and 0.2V on discharge (except for cell 006 which only went to 0.9 V on discharge). In all cases charge was limited by the silver electrode since the cadmium reference - cadmium electrode potential remained within +/- 20 mV right up to the end of charge. On discharge (above), both electrodes can be seen eventually to polarize with respect to the cadmium reference. On the basis of which electrode polarized first, one could conclude that the discharge capacities of the two electrodes were within 0.5 Ah of each other in all cases including the 'fresh' cell 006 (the curves for which can be superimposed upon those of QM 466). This is not enough to explain the differences in capacity. However, fresh cells should have a much larger excess cadmium capacity than the result for cell 006 suggestes. so it is probably not valide to estimate quantitative electrode capacity differences from this type of data.



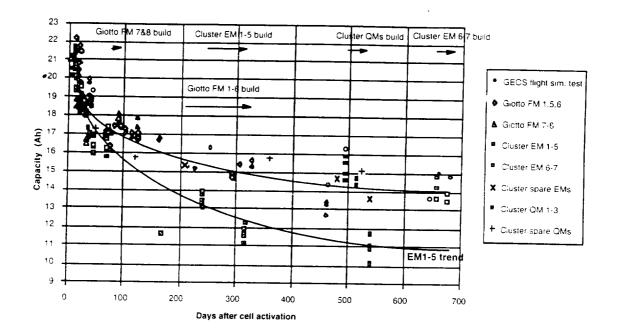
The question was raised as to the best storage temperature and state of charge for silver cadmium cells. The remarkably high capacities still available from the cells stored for many years at ESTEC confirm that -12 deg. C is a suitable temperature. Although most probably discharged when put into storage, we cannot confirm this with certainty.



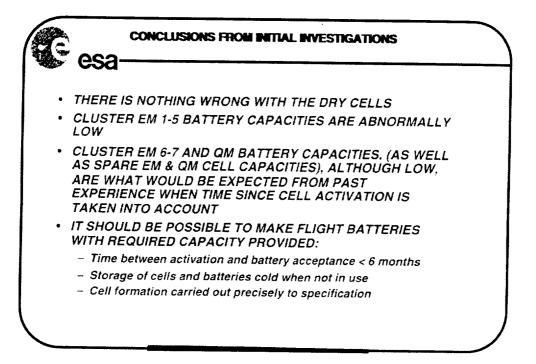
Cells were dismantled in the oxygen-filled glove box to avoid further carbonation. A sample of the electrolyte was recovered for analysis and the components were washed in distilled water and allowed to dry. The extent of silver penetration of the cellophane separator was estimated from its appearance and later by atomic absorption analysis. Electrode surfaces were examined under a scanning electron microscope. The results did not show any abnormalities in any of the cells studied. Cell 006, which had not been stored long and had seen only moderate cycling showed somewhat less silver penetration of the cellophane but the electrode surfaces were indistinguishable from the stored cells.



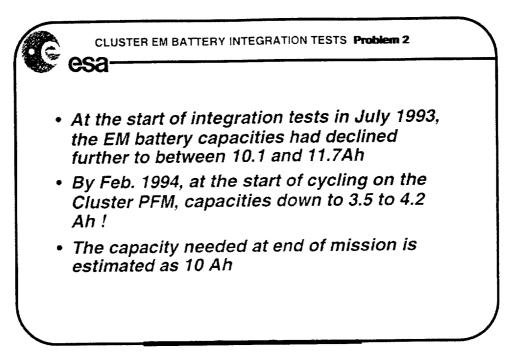
The factors above could all play a role in subsequent performance. In particular, the first attempt to activate dry Cluster cells at ESTEC confirmed how sensitive the obtained capacity is to the thoroughness of the electrolyte filling step.



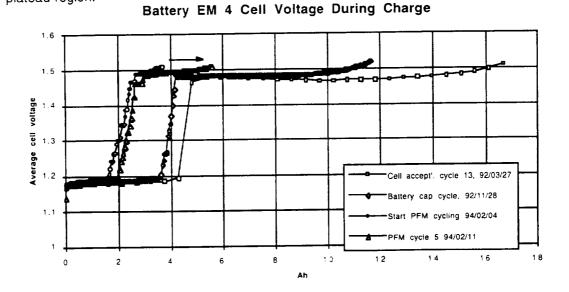
It can be seen that the capacities of Cluster EM batteries 1-5 fall significantly below the trend for all other cells and batteries. Spare EM cells and Cluster QM cells and batteries. on the other hand, are not anomalous. Cluster EM batteries 6.7 which also gave capacities below the acceptance level (15.2 Ah), nevertheless show capacities which are nominal when the (nearly 2 years) interval since activation is taken into account. Since the cells in EM 6,7 were activated in parallel with battery EM 5, this suggests that the difference may have more to do with the storage conditions since activation than with the cell formation. Whilst batteries were generally stored discharged at ambient temperature, detailed records of time batteries spent at different temperatures and states of charge and are not available (and were not required).



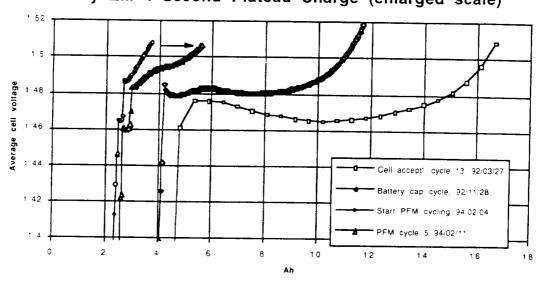
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Since their construction, the EM batteries had been stored discharged at ambient temperature for 4 months before they were needed for integration testing. Before the start of these tests, a further standard capacity check was carried out and revealed a further drop in capacity of all 5 batteries. By February 1994, at the start of cycling test on the Cluster proto-flight model, capacities had fallen dramatically. Comparing the shapes of the battery voltage curves during cycling at various times reveals significant changes, particularly during charge in the second plateau region.

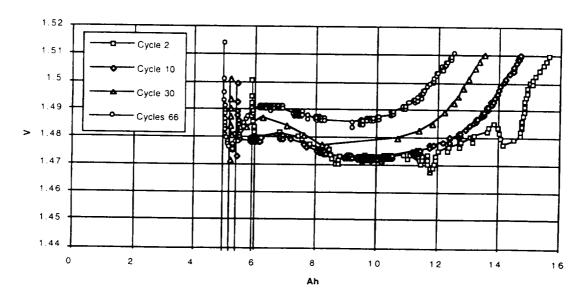


This compares the average cell voltage charge curves of the cells used to make battery EM 4 with the curves measured at battery level before the start of integration testing. The curves are dominated by the transition from the first plateau (Ag ----> Ag2O) to the second (Ag2O----> AgO). Although these over-simplified reactions would suggest a ratio of the capacity of the second to the first plateau as 1;1, in practice a cell a fresh cell gives a ratio of about 2.5:1 as was the case for the EM 4 cells immediately after activation. It can be seen that this ratio has fallen to about 0.5:1. While the capacity of the first plateau has also fallen, the majority of the capacity reduction is associated with the second plateau, charge being terminated when the average cell voltage reaches 1.51 V.



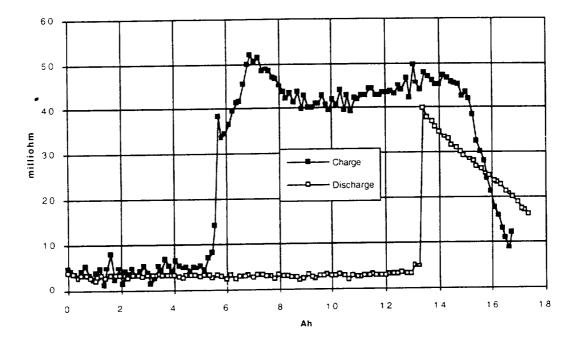
Battery EM 4 Second Plateau Charge (enlarged scale)

Here the same second plateau charge data is shown on an expanded scale. The change in shape of the charge voltage curves can now be seen clearly. From the reference electrode data from single cell tests we believe that these changes are due only to the silver electrode. (EM batteries are not yet available for reference electrode tests). The last curve shows that it is possible to regain some lost capacity by repeat cycles.

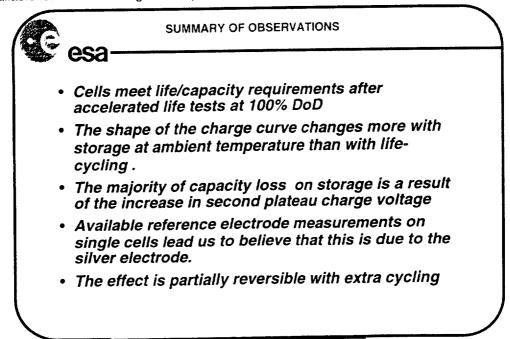


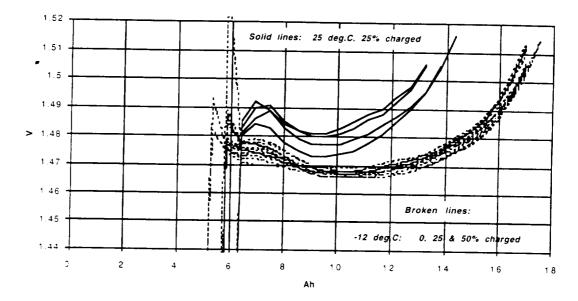
Second Plateau Charge During Life Cycle Test at 20 deg.C

Second plateau average cell voltage charge curves are shown from a 100% DoD life cycle test on two cells at ESTEC. (Fluctuations during cycle 2 were due to instabilities in the temperature of the water bath). Although the cell voltage increases with number of cycles, the increase is rather uniform and the fall in capacity of the second plateau is moderate. The capacity of the first plateau falls by less than 1 Ah after 66 cycles. These results demonstrate the cell's capability to meet the capacity requirements at the end of the mission. The observation that storage could cause more capacity loss than continuous cycling had, however, not been anticipated and therefore needed further investigation.



The change in charge voltage curve with storage was considered most likely to be the result of silver electrode kinetic limitations caused, for example by morphological changes or surface contamination. Internal ohmic resistance measurements should provide useful information. The 'fresh' cell (006) was subjected to current-interruption internal resistance measurements during one complete charge - discharge cycle. Results are shown for charge (solid squares), going from left to right and discharge (open squares), going from right to left. Results are based upon voltage measurements 2 mS before and after the current was reduced to zero by an electronic switch. Reference electrode measurements in the same cell confirmed that the large resistance changes are associated with the silver electrode. This is a known feature of the couple, but it explains how sensitive the second plateau voltage could be to small changes in the silver electrode surface. (The fall in resistance towards the end of charge is probably associated with the onset of oxygen evolution. Comparative resistance data are not yet available for cells exhibiting second plateau capacity loss.

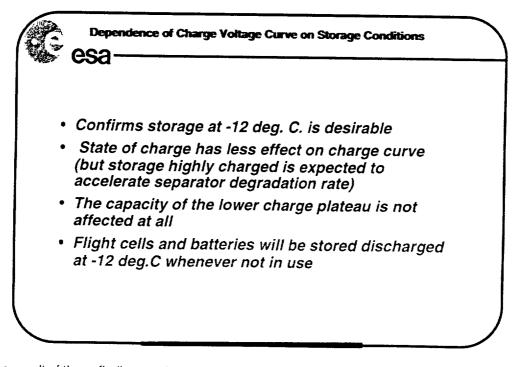




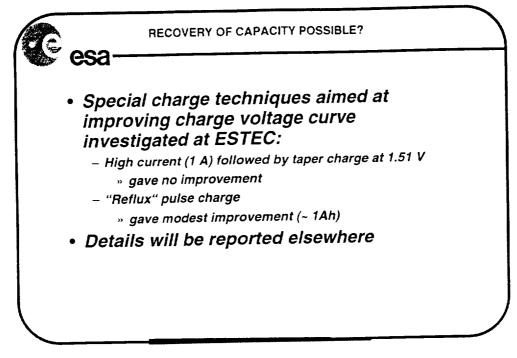
Cell Second Plateau Charge after storage for 6 months

To settle remaining doubts over the best storage conditions. Yardney stored 5 groups of 4 cells for 6 months at 0%, 25%, 50% and 100% charged at -12 deg.C.

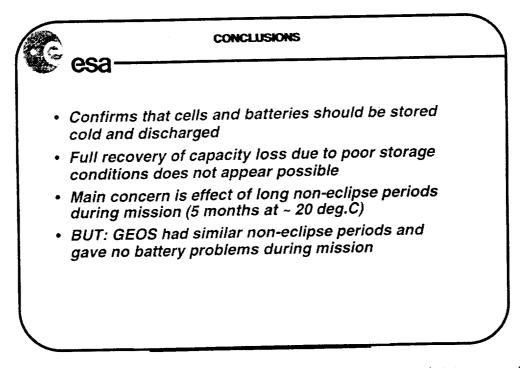
Standard capacity cycles at 20 deg. C, after removal from storage were performed, and the second plateau charge region is shown above. It can be seen that all cells stored at -12 deg.C irrespective of state of charge, showed little change whereas the group stored at 25 deg. C show a loss of 3 to 4 Ah from the second plateau. The charge curve was rather similar to that of a cell subjected to 66 cycles (see vu-graph 17). (The variability in voltage near to the plateau transition is due to the low number of measurements (one per hour)).



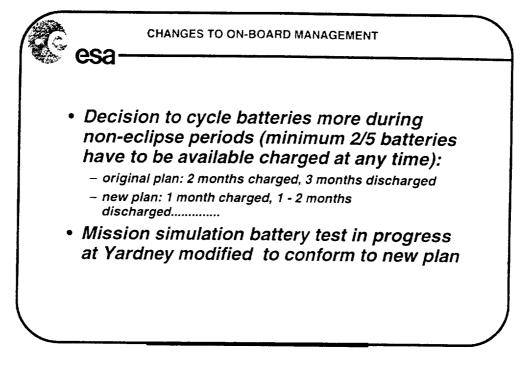
As a result of these findings, strict rules for storage of flight cells and batteries are in preparation.



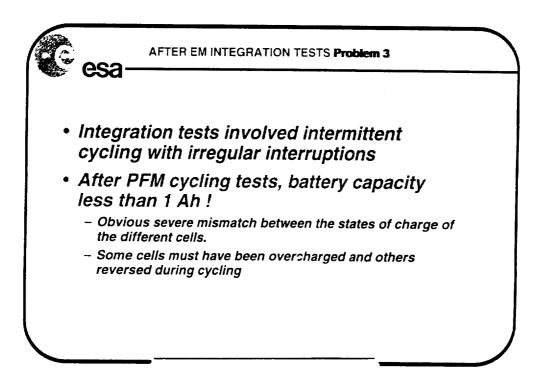
A natural question is whether the capacity loss is recoverable or permanent. So far, only slight recovery has been possible, so it is essential to avoid such losses in the first place.



Whereas long periods of non-use at ambient temperature and a charged state can easily be avoided on the ground, they are unavoidable during the mission, where the temperature during eclipse-free periods is expected to be in the region of 20 deg.C.

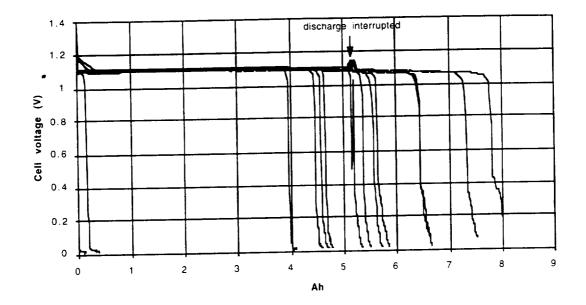


The change in plan reduces the maximum time any battery will be left charged and un-cycled from 2 months to 1 month.

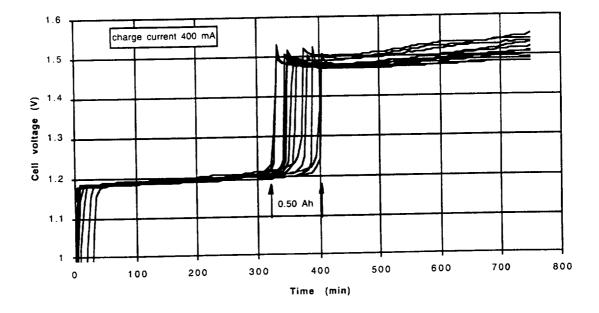


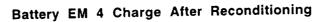
During integration tests, battery cycling was started and stopped according to the needs of the equipment under test. Consequently batteries sometimes remained for prolonged periods at intermediate states of charge. As a result individual cell's state of charge began to diverge. This in turn led the most charged cells tending to be overcharged and the least charged cells reversed in subsequent cycles, because maximum and minimum voltages are defined only at battery level.

Discharge of Battery EM 4 with Individual Cell resistors

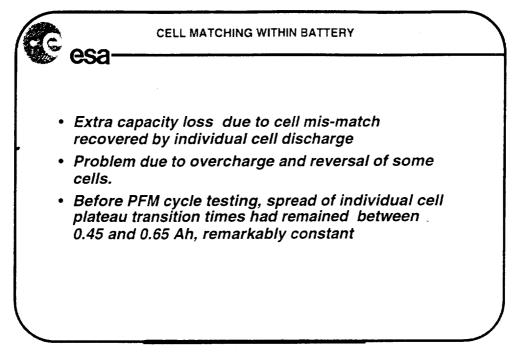


Battery EM 4 was discharged connecting 2.8 ohm resistors across each cell. The voltage curves show an enormous dispersion of 8 Ah between the extreme cells.

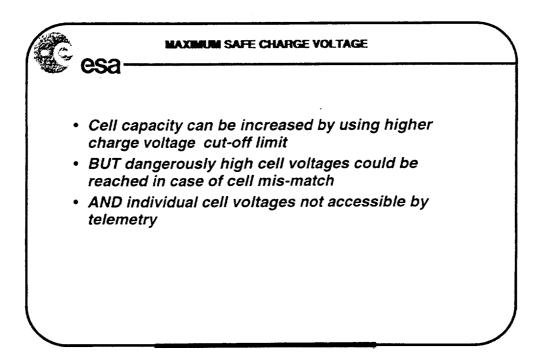




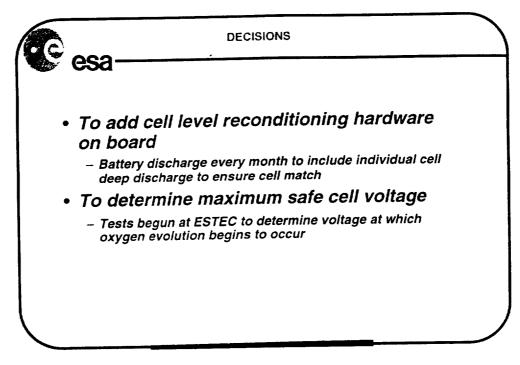
Following the above 'reconditioning' discharge, the next charge was normal again in the sense that the dispersion in plateau transition times between cells in the battery was 0.50 Ah, very close to that observed immediately after battery manufacture (0.53 Ah) and even to that during acceptance cycling of the cells that were made into the battery. This is quite remarkable considering the abuse some cells had suffered.



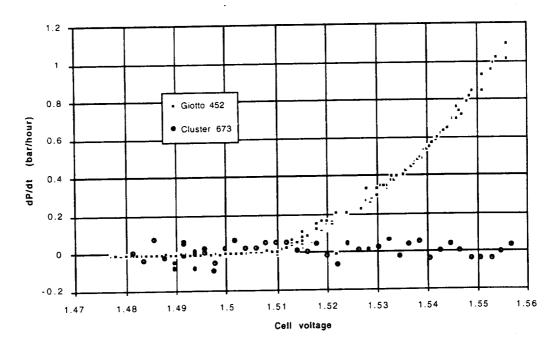
Cells in a battery maintain their relative state of charge during normal cycling and storage. Prolonged periods at intermediate states of charge will eventually lead to mis-match, but the overcharge and reversal some cells evidently suffered during the PFM cycle test were probably the main cause. Reconditioning has restored cell's relative states of charge but not the capacity lost during storage and due to cell overcharge and reversal.



Tests on battery EM 3, in which the end of charge voltage limit had been slightly raised, showed that the capacity could be increased by several ampere-hours because it was then possible to get past the 'hump' in the second plateau charge curve. It is nevertheless essential to avoid any cell in a battery being charged into the region where oxygen is evolved from the silver electrode, because the recombination reaction in such a 'flooded' cell is too slow to prevent the build up of dangerously high pressure in the cell.

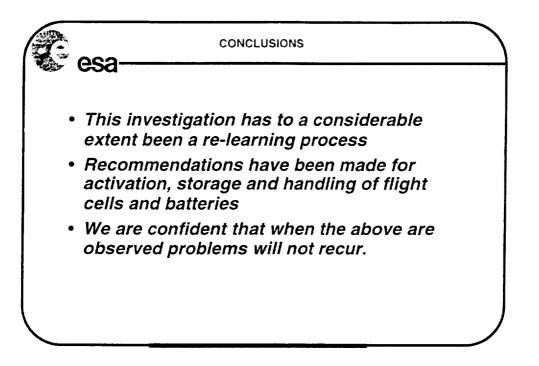


Because of the experience with cell mis-match and the unavailability of individual cell voltage data during the mission, it was decided to implement individual cell reconditioning on board the Cluster spacecraft. In addition it was decided to determine at what charge voltage (at normal charge current) oxygen evolution begins to occur.

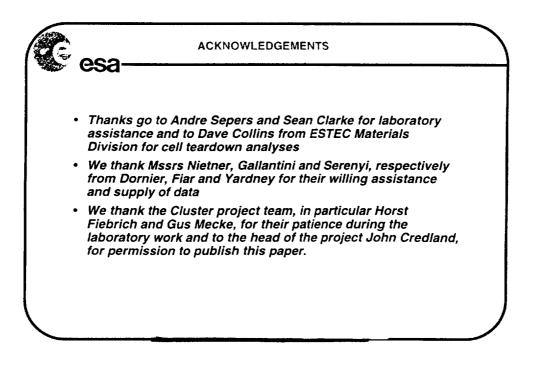


CELL PRESSURE TEST RESULTS

Cells were opened in an oxygen filled glove box and a pressure transducer fitted through a hole drilled in the fill tube and sealed with epoxy cement. Since the cells were clamped across their large faces and the free space in the cell plus pressure transducer remained practically constant, the rate of generation of oxygen is roughly proportional to the rate of pressure increase. There is a clear difference in the behavior of the old Giotto cell, which begins oxygen evolution at 1.51 V at normal charge rates, and the Cluster cell. Since it appears that this voltage decreases with aging, it will be necessary to carry out further measurements on Cluster cells in an "end of life" condition.



Because of the infrequent use of silver cadmium batteries, continuity in knowledge of how to handle them has been hard to maintain and this exercise has been somewhat of a re-learning process. Whilst we believe we know how to avoid these problems during preparation of the flight batteries, it is intended to continue these investigations with the aim of better understanding the underlying processes responsible for them.



FNC

Sal Di Stefano & D. Perrone

Jet Propolsion Laboratories

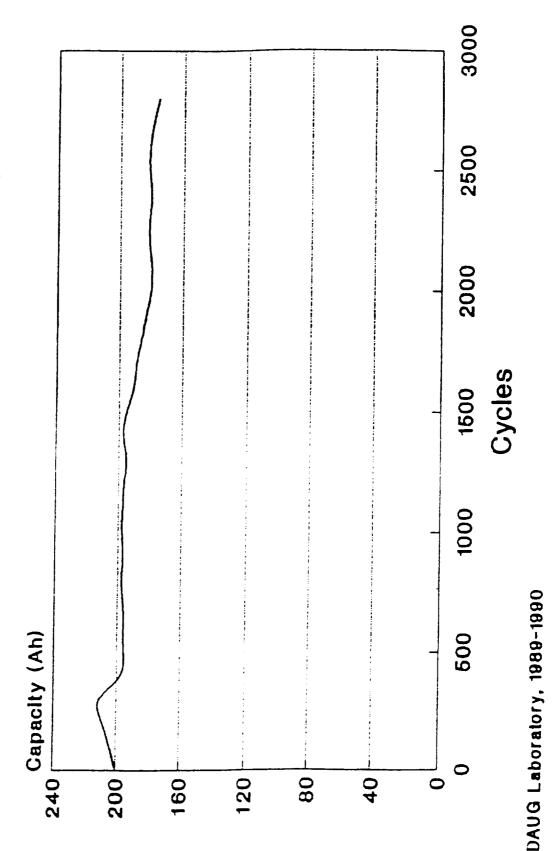
Menahem Anderman Acme Electric Corporation Aerospace Division

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Design and Performance Data Fiber Nickel-Cadmium (FNC) Cells for Sealed



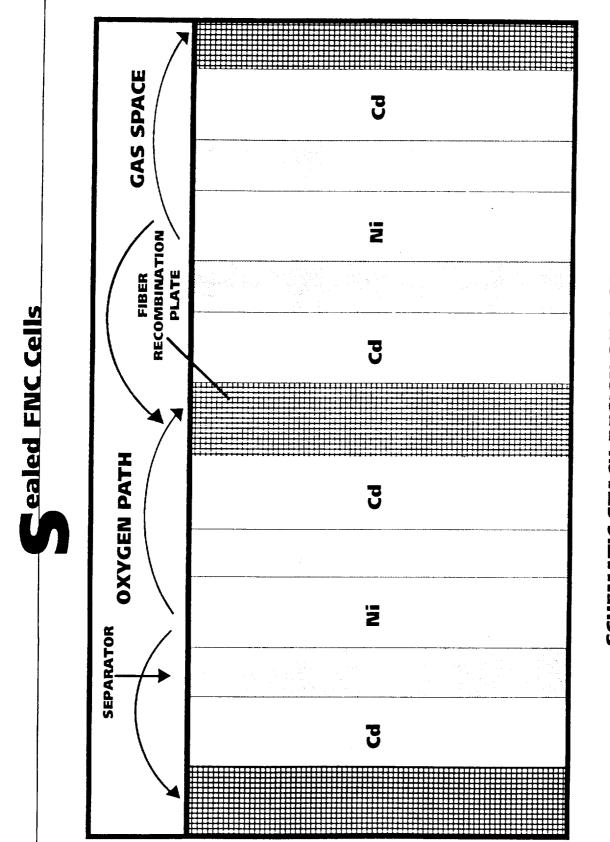
Cycle Life Data 200 Ah FNC Traction Cell (KFMP 200) 100 % DOD, Discharge Load 100 Amp



FNC Main Features

- Use of Fiber Plate (negative and positive)
- Use of Recombination Plates
- Use of Fully Wet Separator
- Larger Amount of Electrolyte per Ahr.
- Negative Internal Pressure
- Hydrogen Removal Catalyst

FZC



SCHEMATIC STACK DESIGN OF A SEALED FNC RECOMBINATION CELL FIN

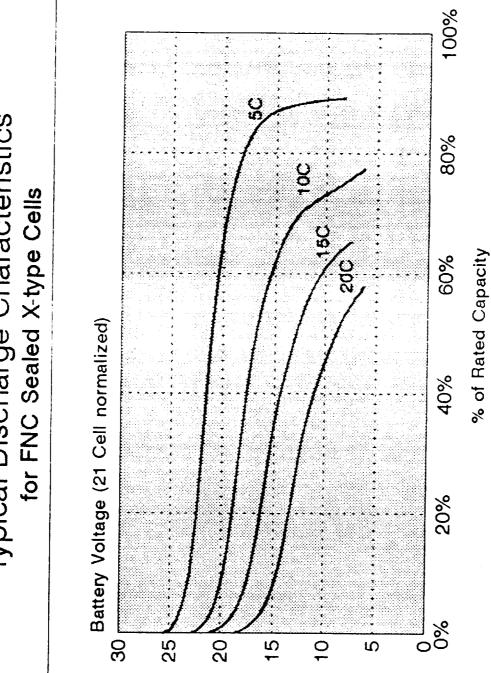
ison	Acme ENC	Nickel-plated Felt	ical Mechanical	1.5g/cc Void	nylon 2-5 micron pore size PP or 20-40 micron pore size nylon	LNC
Ni-Cd Cells Component Comparison	NASA <u>Standard</u>	Sintered Nickel	Chemical/Electrochemical	2g/cc Void	30-80 micron pore size nylon	
		Plaque	Impregnation	Positive Loading	Separator	

Sta	Space Ni-Cd Cells Stack Comparison	
	NASA <u>Standard</u>	Acme <u>ENC</u>
Recombination Plates	NON	YES
Overcharge Pressure	> 30 PSIA	< 5 PSIA
Hydrogen Removal Capability	NO	YES
Inner Electrodes Distance	8-10 mil	11-13 mil
Electrolyte ml/Ah	2.5 - 3.5	4 - 4.5
Measured Negative/Positive Capacity	1.5 - 2	2 - 2.4

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FIC

Typical Design Parameters SPFNC Cells	S
Electrodes Active Area	55 cm ² /Ah
Positive Electrode Loading	1.5g/cc Void
Positive Electrode Charge Density	0.15 Ah/g
Negative Electrodes Theoretical Capacity	3.3 Rated Capacity
Negative Electrodes Flooded Capacity	2.5 Rated Capacity
Separator Material	PP or Nylon
Inner Electrode Distance	11 - 13 Mil
Electrolyte Concentration	30 - 32% KOH 2 - 3% LiOH
Electrolyte Volume	4.3 ml/Ah
Cell Impedance (mohm Ah)	30
Specific Energy (Cell Level)	34 - 43 Wh/kg
	FNC

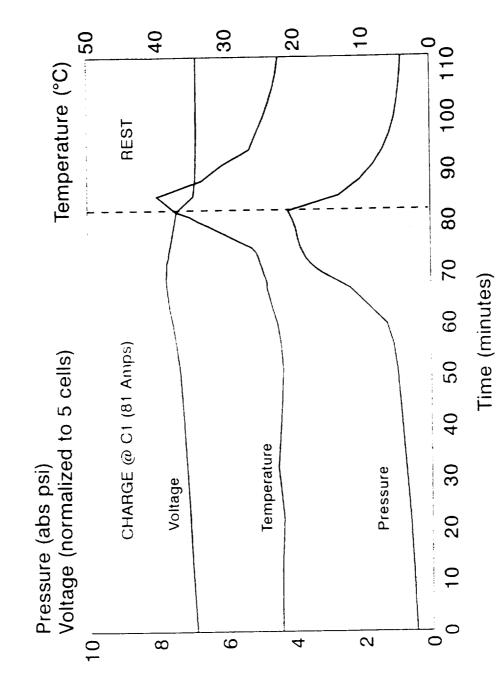


Typical Discharge Characteristics for FNC Sealed X-type Cells

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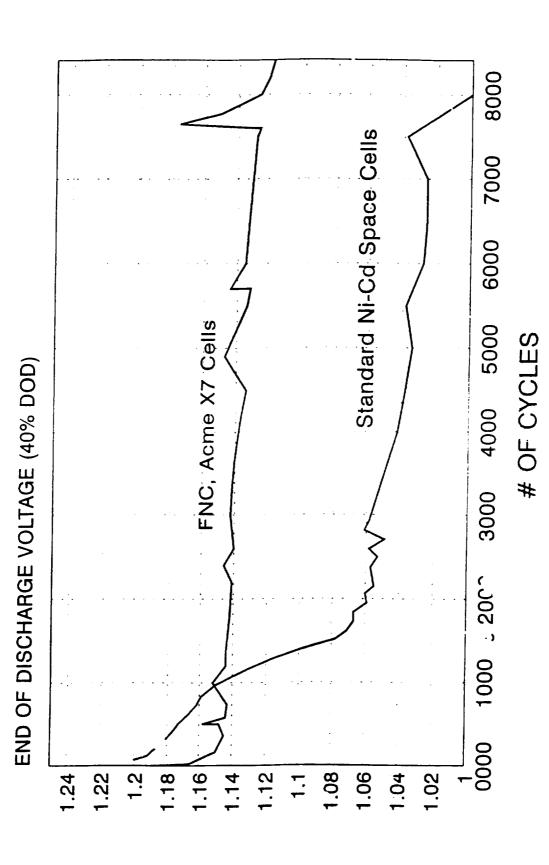
FINC

Charging Characteristics X81 Cell, 24°C

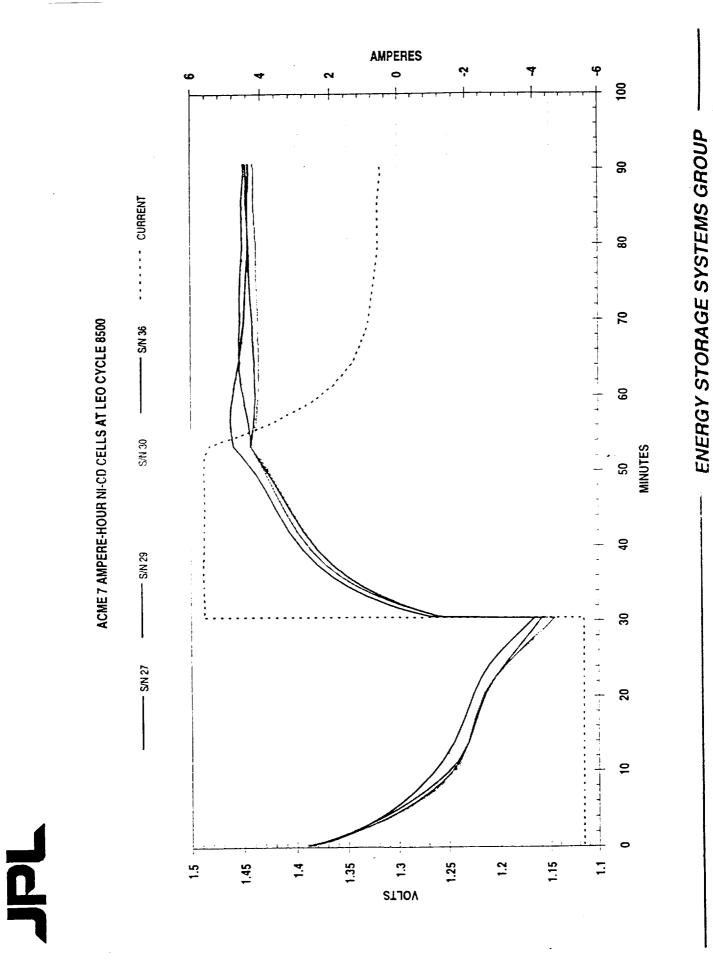


Accelerated LEO Stress Test 20°C / 40% DOD

FNC

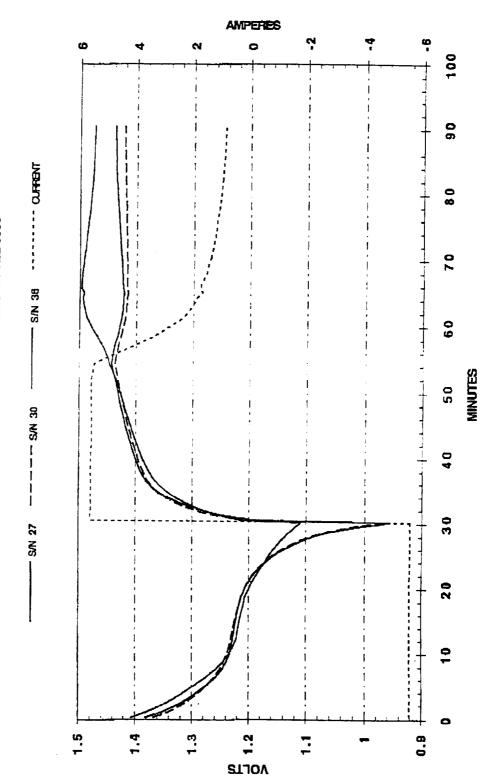


JPL Data 1991 - 1994



1994 NASA Aerospace Battery Workshop

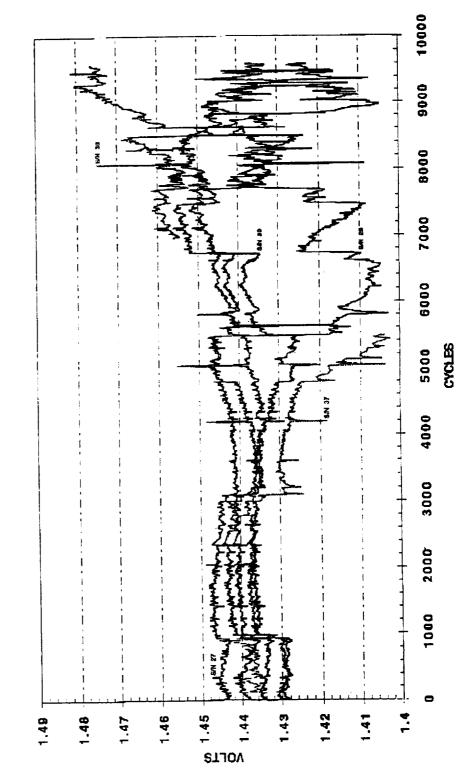
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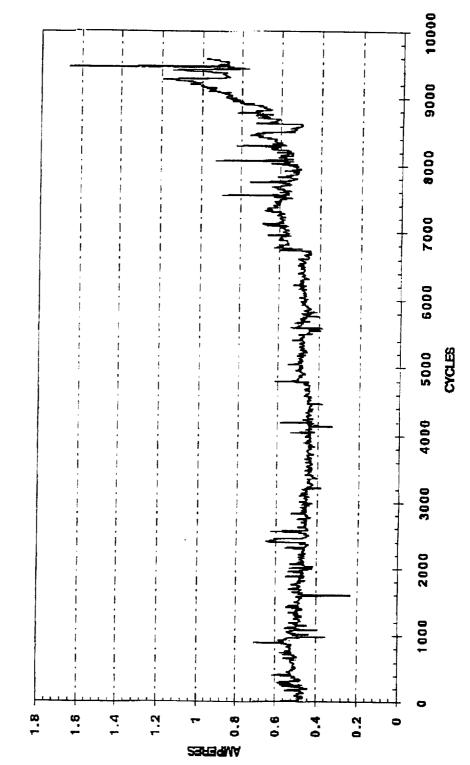
ACME 7 AMPERE-HOUR NHCD CELLS AT LEO CYCLE 9550

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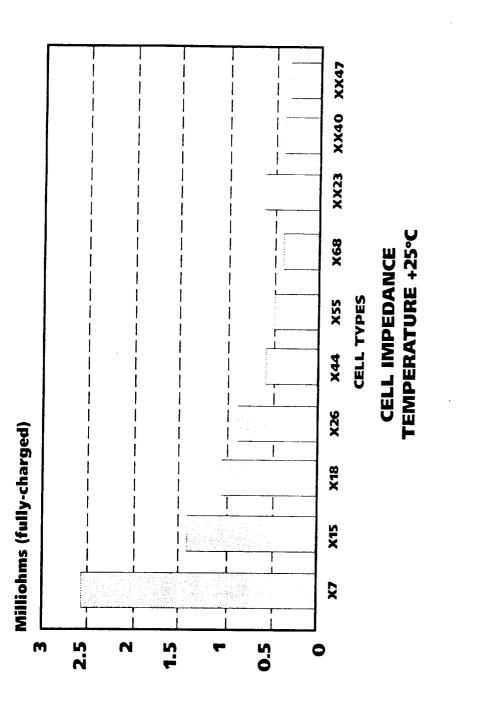




Sealed FNC Cells

(mm)	105	178	147	168	168	166	166	224	168	168	166
TOTAL HEIGHT (inches) (mm)	4.12	7.03	5.79	6.61	6.61	6.52	6.52	8.82	6.61	6.61	6.52
(mm)	24	29	36	25	41	54	59	54	25	41	54
LENGTH (inches) (mm)	<u>.94</u>	1.14	1.41	66'	1.62	2.13	2.33	2.13	66 [.]	1.62	2.13
(mm) T	57	61	67	115	115	115	115	115	115	115	115
WIDTH (inches) (mm)	2.24	2.41	2.66	4.53	4.53	4.53	4.53	4.53	4.53	4.53	4.53
ЗНТ (kg)	0.28	0.57	0.73	1.13	1.72	2.05	2.59	3.13	1.13	1.72	2.05
WEIGHT (Ibs.) (kg	0.63	1.25	1.62	2.50	3.80	4.52	5.71	6.90	2.50	3.79	4.52
RATED CAPACITY (Ah/1h to1.0Vpc)	6.5	15	17	26	44	55	68	81	23	40	47
CELL	X7	X15	X18	X26	X44	X55	X68	X81	XX23	XX40	XX47

FNC



FNC

FNC

FNC Peripheral Advantages

- More tolerable to manufacturing variations.
- Potential for shorter development and qualification time for new cells. 0
- Lower cost.
- Capacity up to 150 Ah possible.

FINC

SUMMARY	 Improved fiber plates are utilized in Ni-Cd cells for a variety of applications. 	 Sealed cell design uses oxygen recombination plates that allow the use of a dendrite-resistive fully wet separator. 	 Conservative design: low positive mass loading, high negative to positive capacity ratio, high electrolyte per Ah. 	 Negative cell pressure at all times. 	 Robust against manufacturing variations. 	 Cycle life testing ongoing. 	 Potential for increased reliability with reduced cost compared to standard cells. 	 Ready for full space qualification programs. 	
							-	-	

NASA BATTERY TESTBED CAPABILITIES AND RESULTS	μ	Frank Deligiannis, Sal DiStefano, and Dave Perrone

1994 NASA Aerospace Battery Workshop Huntsville Marriott Huntsville, Alabama November 15-17, 1994

A COUPLE OF NASA SATELLITES (UARS & CGRO) EXPERIENCED ANOMALIES WITH THEIR Ni-Cd BATTERIES - EARLY 1992	BATTERIES DEVELOPED LARGE HALF-BATTERY VOLTAGE DIFFERENTIALS (>100 mV) EARLY IN LIFE (<1 YR) A BATTERY ON CGRO WAS REMOVED OFF LINE DUE TO THE HALF-BATTERY VOLTAGE DIFFERENTIAL EXCEEDING 750 mV	UPCOMING LAUNCHES OF EUVE & TOPEX WITH SIMILAR DESIGN BATTERIES - MID 1992	BATTERY MANAGEMENT WAS INITIATED ON MOST MISSIONS -MANY NON-TRADITIONAL METHODS OF OPERATION APPEARED	NASA BATTERY TESTBED EFFORT WAS INITIATED	ENERGY STORAGE SYSTEMS GROUP
Igga NASA Aerospace Battery Workshop		•	- MANYER	•	



NASA TESTBED OBJECTIVE

OPERATIONAL STRATEGIES PRIOR TO IMPLEMENTATION DETERMINE ON THE GROUND THE IMPACT OF VARIOUS **ON THE SPACECRAFT** .

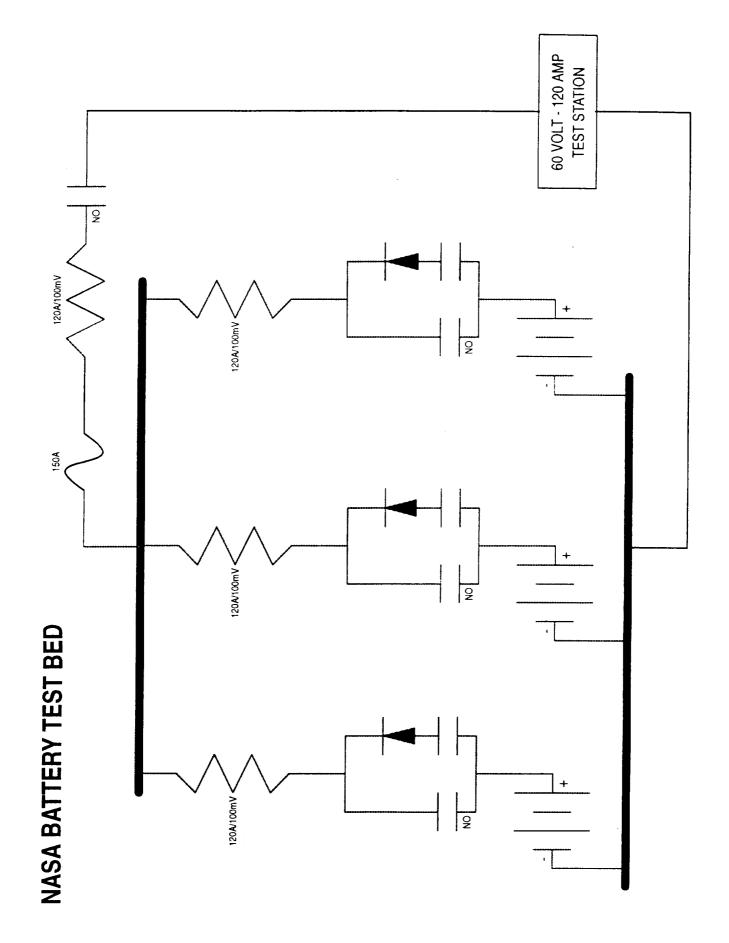
	APPROACH	 DEVELOP COMPUTER CONTROLLED TESTING WHICH WILL DEVELOP COMPUTER CONFIGURATION TO MODEL VARIOUS BATTERY ENABLE FAST RECONFIGURATION TO MODEL VARIOUS BATTERY SYSTEMS: CGRO, UARS, EUVE TOPEX AND OTHER FUTURE NASA MISSIONS 	 OBTAIN & OPERATE IMBALANCED BATTERIES WITH HIGH HALF-BATTERY VOLTAGE DIFFERENTIALS 	 IMPLEMENT VARIOUS OPERATIONAL STRATEGIES SUCH AS 	 DEEP DISCHARGES DURING FULL SUN PERIODS CONSTANT CURRENT MODES OF CHARGING ETC. 	ENERGY STORAGE SYSTEMS GROUP
1994 I	vASA Aerospace	Ballery Worksnop	-520-		Advanced Technol	ogies Session

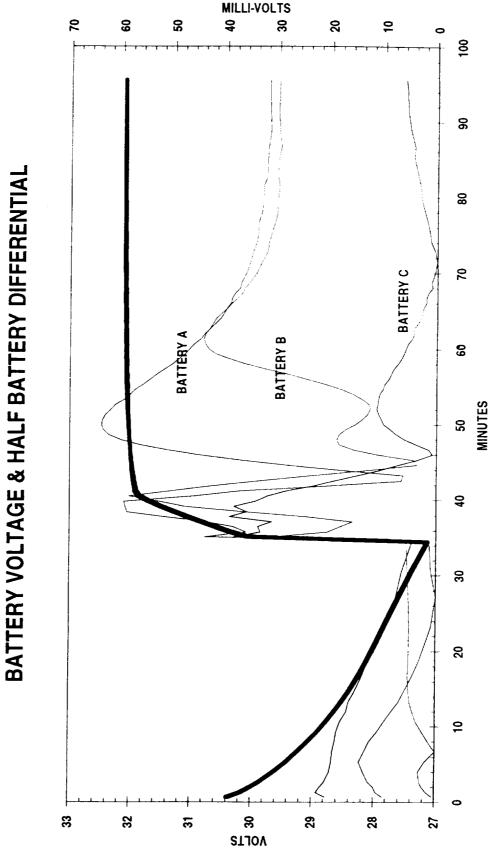
TESTBED CAPABILITIES	CURRENTLY CONFIGURED TO HANDLE THREE 50 Ah NiCd BATTERIES IN PARALLEL	 SIMULATION THROUGH COMPUTER HARDWARE & SOFTWARE 	 VARIOUS CHARGE/DISCHARGE MODES CAN BE IMPLEMENTED (CONSTANT CURRENT, CONSTANT POWER, CONSTANT VOLTAGE etc) 	 SIMULATION OF ORBIT PROFILES WITH VARYING OCCULTATION PERIODS 	TEMPERATURE, VOLTAGE, CURRENT LIMITS CAN BE SET	 24 HOUR AUTOMATED OPERATION 	ENERGY STORAGE SYSTEMS GROUP
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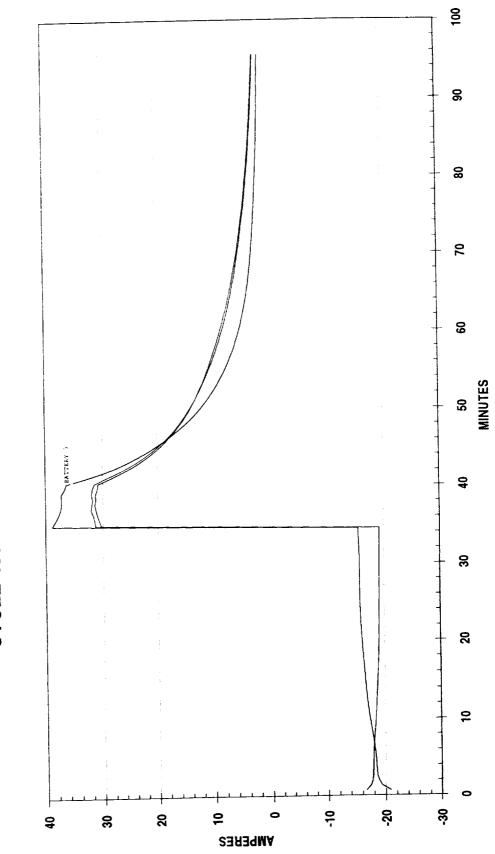
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TESTBED CAPABILITIES (cont.)	
 POWER & ORBIT PROFILES EASILY CHANGED BY COMPUTER COMMAND 	
 MONITORING & DATA COLLECTION OF INDIVIDUAL CELL VOLTAGES, BATTERY CURRENTS, TEMPERATURES & VOLTAGES INCLUDING PARAMETERS SUCH AS PEAK CHARGE CURRENT, TAPER CURRENT, C/D RATIO, NET OVERCHARGE etc. 	
 THERMAL ENVIRONMENT CONTROLLED BY AN ENVIRONMENTAL CHAMBER 	
 SYSTEM HAS MAX 40 A PER BATTERY CURRENT CAPABILITY 	
 SYSTEM HAS MAX 60 V PER BATTERY VOLTAGE CAPABILITY 	
ENERGY STORAGE SYSTEMS GROUP	1
ENERGY STORAGE SYSTEMS GROUP	1

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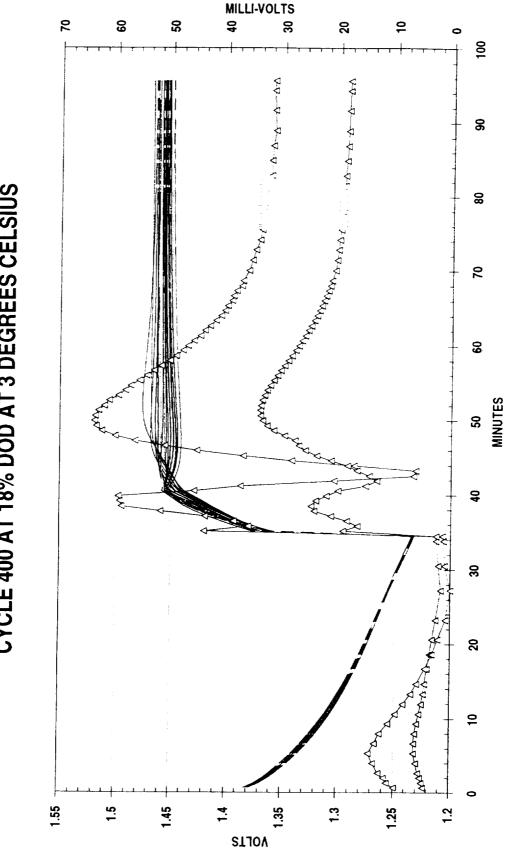




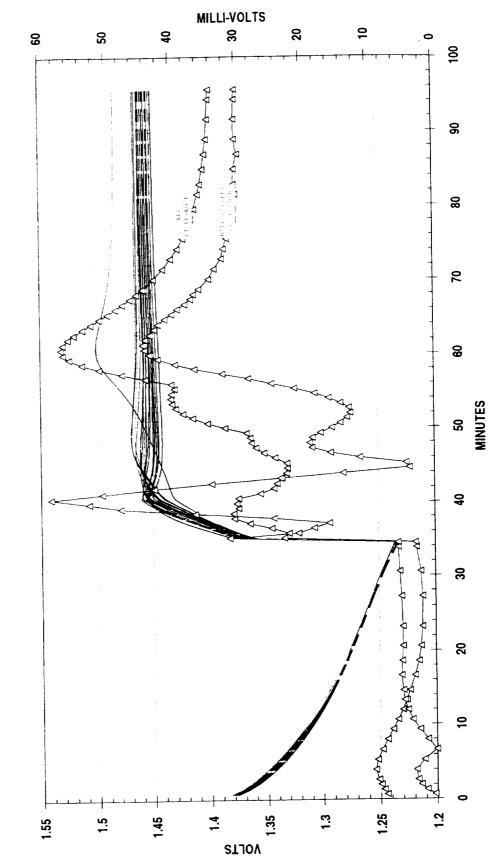
NASA BATTERY TEST BED -- BATTERY A, B & C CYCLE 400 AT 18% DOD AT 3 DEGREES CELSIUS BATTERY VOLTAGE & HALF BATTERY DIFFERENTIA



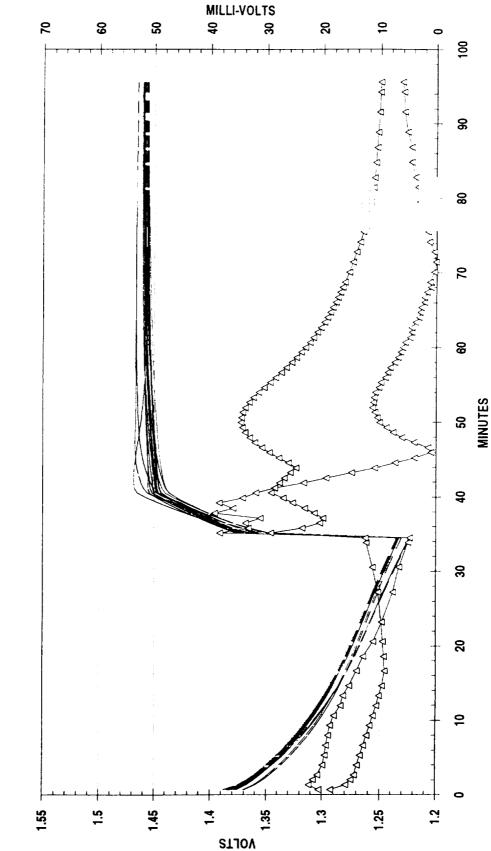
NASA BATTERY TEST BED CYCLE 400 AT 18% DOD AT 3 DEGREES CELSIUS







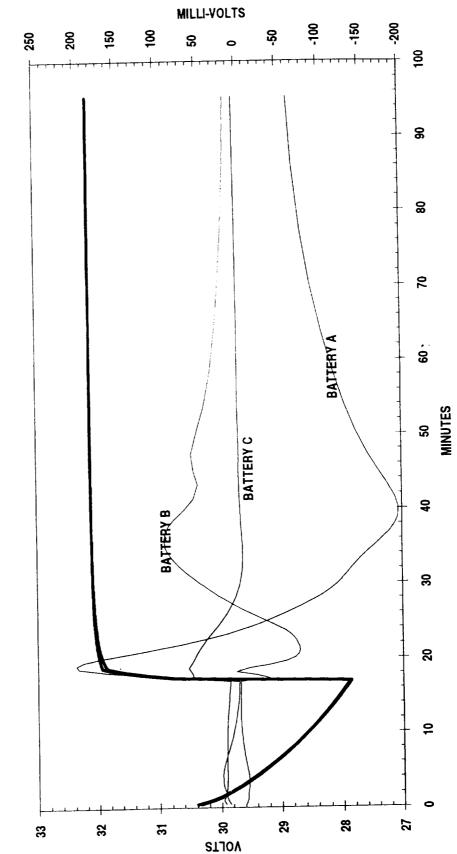
NASA BATTERY TEST BED -- BATTERY B CYCLE 400 AT 18% DOD AT 3 DEGREES CELSIUS



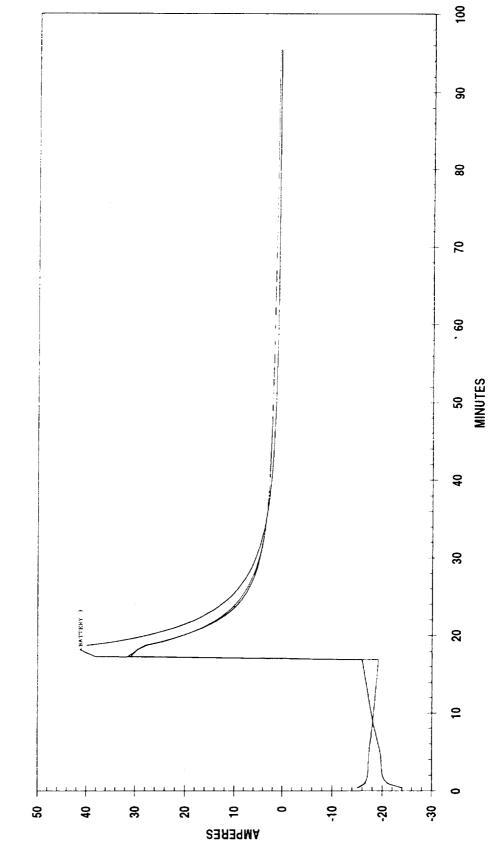
NASA BATTERY TEST BED -- BATTERY C CYCLE 400 AT 18% DOD AT 3 DEGREES CELSIUS

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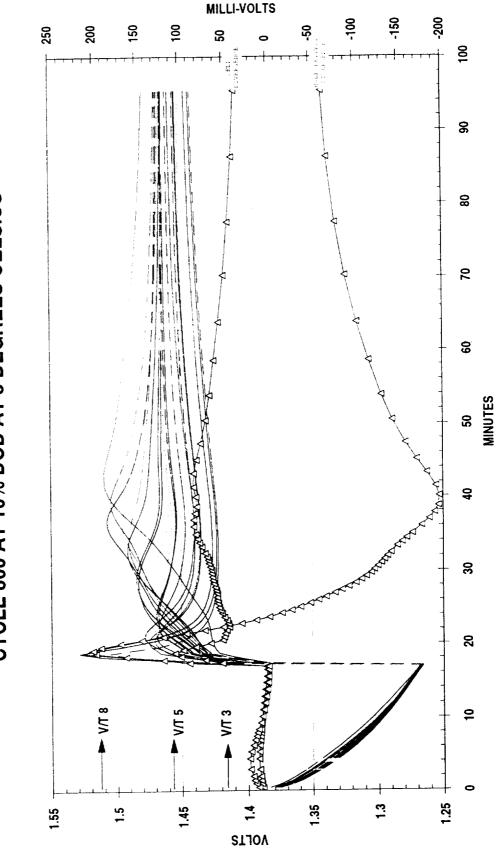
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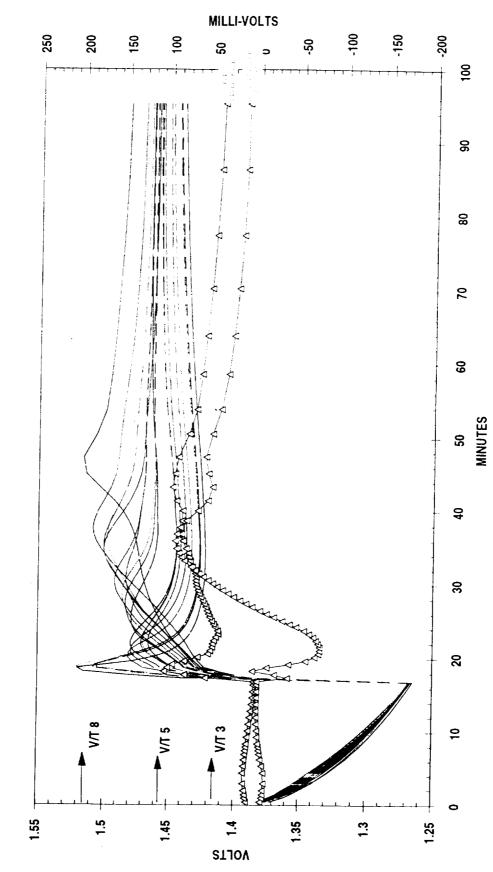






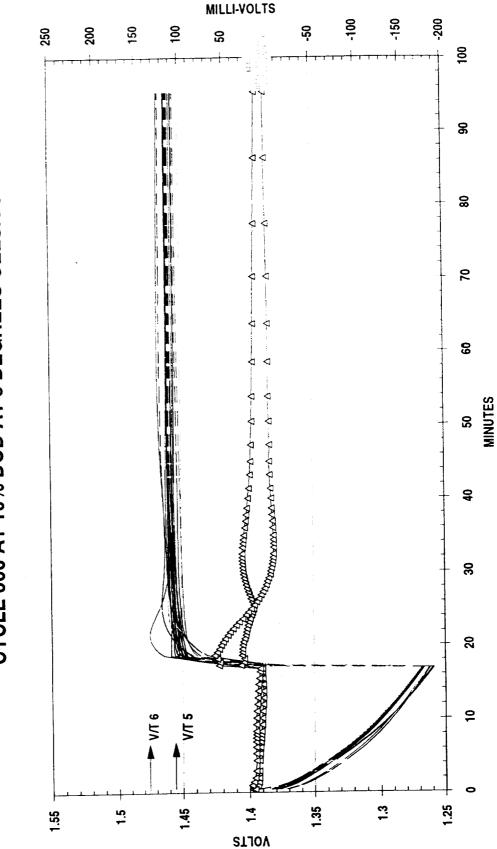


NASA BATTERY TEST BED -- BATTERY A CYCLE 800 AT 10% DOD AT 3 DEGREES CELSIUS





Advanced Technologies Session



NASA BATTERY TEST BED -- BATTERY C CYCLE 800 AT 10% DOD AT 3 DEGREES CELSIUS

SUMMARY	 COMPLETED THE COMPUTER SOFTWARE & HARDWARE 	 OBTAINED THREE BATTERIES FOR TESTING 	 ESTABLISHED A PERFORMANCE DATABASE OF THE THREE BATTERIES UNDER THE UARS PROFILE 	 CREATED A IMBALANCED BATTERY SYSTEM 	 FUTURE PLANS ARE TO IMPLEMENT VARIOUS OPERATIONAL STRATEGIES TO CORRECT THE IMBALANCE 	ENERGY STORAGE SYSTEMS GROUP
	- COMPLETE	• OBTAINED	 ESTABLISH BATTERIES 	 CREATED A 	 FUTURE PL STRATEGIE 	

ACKNOWLEDGMENT

- This work was performed at The Jet Aeronautic and Space Administration. The Propulsion Laboratory, California Institute of work was sponsored by the Office of Safety Technology under contract with the National Reliability and Maintainbility, Code Q, of NASA.
 - Test Batteries and useful discussions were provided by Dr. Gopalakrishna M. Rao and Mr. Mark Toft of the Power Branch at Goddard Space Flight Center.