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Directional solidification and mechanical properties of NiAl-NiAlTa alloys

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Directional solidification of eutectic alloys is a promising technique for producing *in-situ* composite materials exhibiting a balance of properties. Consequently, the microstructure, creep strength and fracture toughness of directionally solidified NiAl-NiAlTa alloys were investigated. Directional solidification was performed by containerless processing techniques to minimize alloy contamination. The eutectic composition was found to be NiAl-15.5 at% Ta and well-aligned microstructures were produced at this composition. A near-eutectic alloy of NiAl-14.5Ta was also investigated. Directional solidification of the near-eutectic composition resulted in microstructures consisting of NiAl dendrites surrounded by aligned eutectic regions. The off-eutectic alloy exhibited promising compressive creep strengths compared to other NiAl-based intermetallics, while preliminary testing indicated that the eutectic alloy was competitive with Ni-base single crystal superalloys. The room temperature toughness of these two-phase alloys was similar to that of polycrystalline NiAl even with the presence of the brittle Laves phase NiAlTa.

Key words: in-situ composites, NiAl-NiAlTa, directional solidification, creep resistance, fracture toughness.

1 INTRODUCTION

Alloys based on the ordered intermetallic compound NiAl have the potential to be used in high temperature structural applications due to their superior oxidation resistance, high melting point, and high thermal conductivity.^{1,2} However, improvements in both the room temperature fracture toughness and elevated temperature strength are necessary before NiAl alloys can be used in load-bearing applications. Considerable research has been aimed at improving the room temperature ductility and fracture resistance of NiAl¹⁻⁴ with only a few promising results. For example, a tensile ductility of five percent has been measured for both high purity⁵ and iron-doped single crystals of stoichiometric NiAl.⁶ In addition, room temperature fracture toughness values of the order of 24

MPa√m have been measured for NiAl-(Cr,Mo) eutectics that were directionally solidified to produce *in-situ* NiAl-based composites containing a refractory metal phase.⁷

Improving the elevated temperature strength of NiAl-based alloys may not be considered as challenging as improving low temperature fracture properties. Nevertheless, very few attempts to strengthen NiAl have actually produced materials that are competitive with the nickel-based superalloys.² Conventional strengthening techniques such as precipitation, solid solution, and dispersion strengthening have all been applied with varying success,^{2,3} with strengthening by precipitates of the Heusler phases Ni₂AlTi and Ni₂AlHf exhibiting the most promise.^{1,3,8,9} For example, Polvani *et al.*¹⁰ have demonstrated that the creep resistance of a two-phase NiAl+Ni₂AlTi alloy is significantly greater than that of the individual phases and was comparable to polycrystalline Ni-based superalloys. While impressive yield strength values have been

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measured for NiAl containing Heusler precipitates,^{2,9} the improved strengths of polycrystalline alloys, due to solid solution and precipitation strengthening, are diminished at low strain rates such as those in the creep regime.^{2,11}

Another approach to strengthening NiAl consists of developing composite materials containing phases much stronger than the Heusler phase. For example, Sauthoff has suggested the use of C14 Laves compounds such as NiAlNb and NiAlTa for strengthening NiAl.^{12,13} These phases are very brittle and have the hexagonal MgZn₂ type crystal structure. However, compressive yield strengths surpassing those of advanced nickel-based superalloys have been measured for NiAl-Laves alloys.¹³ In general, the improvements in strength of these two-phase alloys are proportional to the volume fraction of Laves phase present.^{12, 14}

In addition, eutectic microstructures are possible within the NiAl–NiAlNb and NiAl–NiAlTa systems allowing for *in-situ* composite studies. Whittenberger *et al.* have found that the creep strength of the NiAl–NiAlNb eutectic is extremely sensitive to microstructure and processing conditions.¹⁴ By directional solidification of this eutectic, an order of magnitude increase in creep resistance was measured compared to materials processed using a casting and extrusion procedure. Unfortunately, the creep strength of the NiAl–NiAlNb alloys is still less than that of most nickel-based superalloys.¹⁴

Phase equilibria in the NiAl–NiAlTa system are not very precisely defined.^{15,16} Nevertheless, based

on Sauthoff's work^{12,13} with polycrystalline materials, directionally solidified alloys from this system may display better strengths than NiAl–NiAlNb alloys. Therefore, the purpose of this study was to characterize the effects of containerless processing on the microstructure, the elevated temperature strength, and the room temperature fracture toughness of *in-situ* composites based on the NiAl–NiAlTa system.

2 PROCESSING AND MATERIALS

2.1 Arc-melted ingots

A series of alloys containing high purity Ni, Al, and Ta were arc-melted using a non-consumable tungsten electrode in order to identify promising microstructures for subsequent *in-situ* composite studies. Each button, weighing approximately 12 g, was melted at least five times and flipped over between each melting to promote homogeneity. The arc-melted ingots were then metallographically examined. From this survey study, a near-eutectic alloy of NiAl–14.5Ta (in atomic percent) was chosen for further investigation.

2.2 Directional solidification

Precursor ingots for subsequent directional solidification were produced by induction melting of elemental Ni and Al, and a Ni–Ta master alloy.

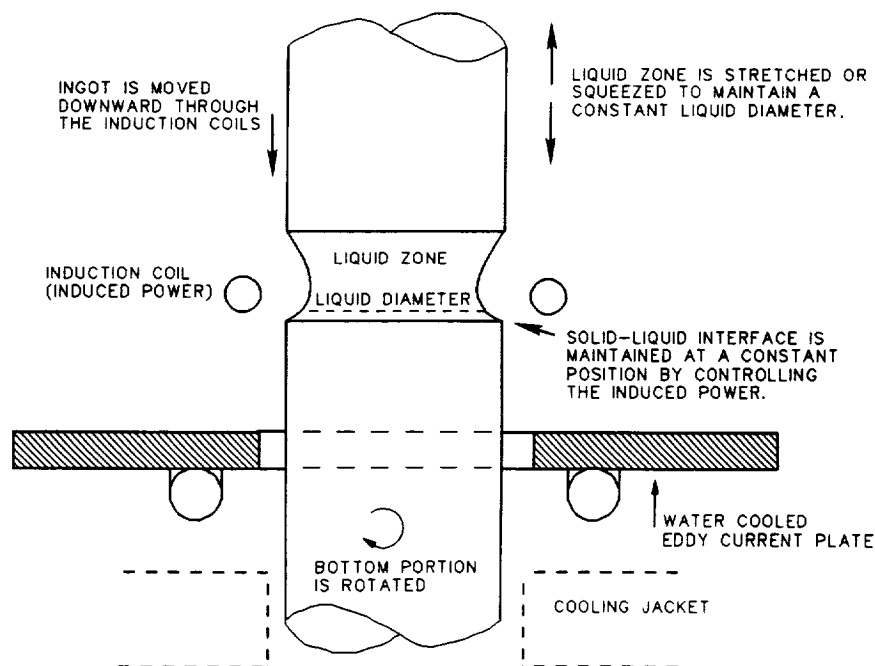


Fig. 1. Schematic of molten zone during directional solidification.

The 1-kg charges were chill cast into a copper mold. After removal of the hot-top, the precursor ingots were nominally 25 mm in diameter and 300 mm in length. The ingots were then directionally solidified in the containerless mode by the electro-magnetically-levitated zone process in ultra-pure helium atmospheres.

Containerless processing offers the advantage of enhanced purity by eliminating alloy contamination from crucible materials while simultaneously zone-purifying the initial ingot material. The crucible is eliminated by using induction power to heat, levitate, and constrain the liquid zone. Directional solidification is then accomplished by moving the liquid zone through the length of the ingot by physically traversing the ingot through the induction coil. A schematic of the levitated molten zone is shown in Fig. 1. Without crucible containment, precise dimensional control of the freezing ingot requires that the shape and position of the molten zone within the induction coils be controlled. The control variables are the position of the solid-liquid interface and the liquid zone diameter (melt diameter). Current practice allows ingots to be processed under full computer control independent of any temperature measurement.^{17,18}

Computer control is accomplished by using a digitizing camera to capture an image of the molten zone. The location of the solid-liquid interface is then found and held to a target position by controlling the induced power. The shape of the zone is controlled by maintaining a constant melt diameter by a stretch-squeeze action on the liquid, (Fig. 1). Both the interface position and the melt diameter are held to their target values by proportional, integral control loops. While not a control variable, the temperature near the solid-liquid interface is measured using an infrared pyrometer and recorded as a function of run time for all melts.

An as-processed NiAl-14.5Ta ingot and the corresponding melt record are shown in Fig. 2. The data in Fig. 2 represents the first attempt to process this material. Computer control was initiated at about 3000 s. From Fig. 2, a constant melt diameter was maintained by a stretch-squeeze action on the liquid zone. Similarly, the position of the solid-liquid interface was controlled by varying the induced power. However, an oscillation in the interface position (Fig. 2(c)) created an unusually rough surface for the processed ingot, Fig. 2(d)). The oscillation of the interface position was caused by the target value for the melt diameter being too large. This created an unstable situation where the diameter of the

freezing ingot was larger than that of the unprocessed section.

To allow for volume expansion upon melting, a gap is left between the upper and lower portions of the ingot during initial heating. If the spacing of the gap is too small, a portion of the molten zone may spill over the edge of the solid-liquid interface before a stable zone can be established creating too large a base. The tendency for the liquid zone to spill during the initial heating was a common problem for all the NiAl-NiAlTa alloys processed. However, good dimensional control was obtained for all the other directionally solidified NiAl-NiAlTa ingots by decreasing the freezing diameter manually before initiating computer control.

3 EXPERIMENTAL PROCEDURES

3.1 Compression testing

Cylindrical compression specimens were electrical discharge machined from selected ingots. The samples were 5 mm in diameter by 10 mm in length, with the compression axis parallel to the growth direction of the crystal. Mechanical properties were generated under both constant velocity conditions in a screw driven universal machine and under constant load conditions in lever-arm creep machines at temperatures between 1200 and 1400 K. Constant velocity experiments were used to determine the behavior at fast strain rates ($> 10^7 \text{ s}^{-1}$) while constant load testing was employed for slower rates. Overlapping steady state stress-strain rate data from the two techniques indicated excellent correlation between constant load and constant rate tests. All testing was performed in air as a secondary check for environmental resistance under load.

3.2 Flexure testing

Room temperature fracture toughness was determined by performing four-point bend tests on notched samples. The specimen size and test geometry are shown in Fig. 3. Bend specimens were electrical discharge machined from the directionally solidified ingots and notched perpendicular to the growth direction using a slow speed diamond impregnated saw. A fatigue pre-crack was not initiated at the notch tip prior to testing. Bend tests were performed on a servo-hydraulic test frame using a displacement rate of $7.6 \times 10^{-4} \text{ mm/s}$. Fracture toughness values were calculated using the *K*-calibration for pure bending.¹⁹

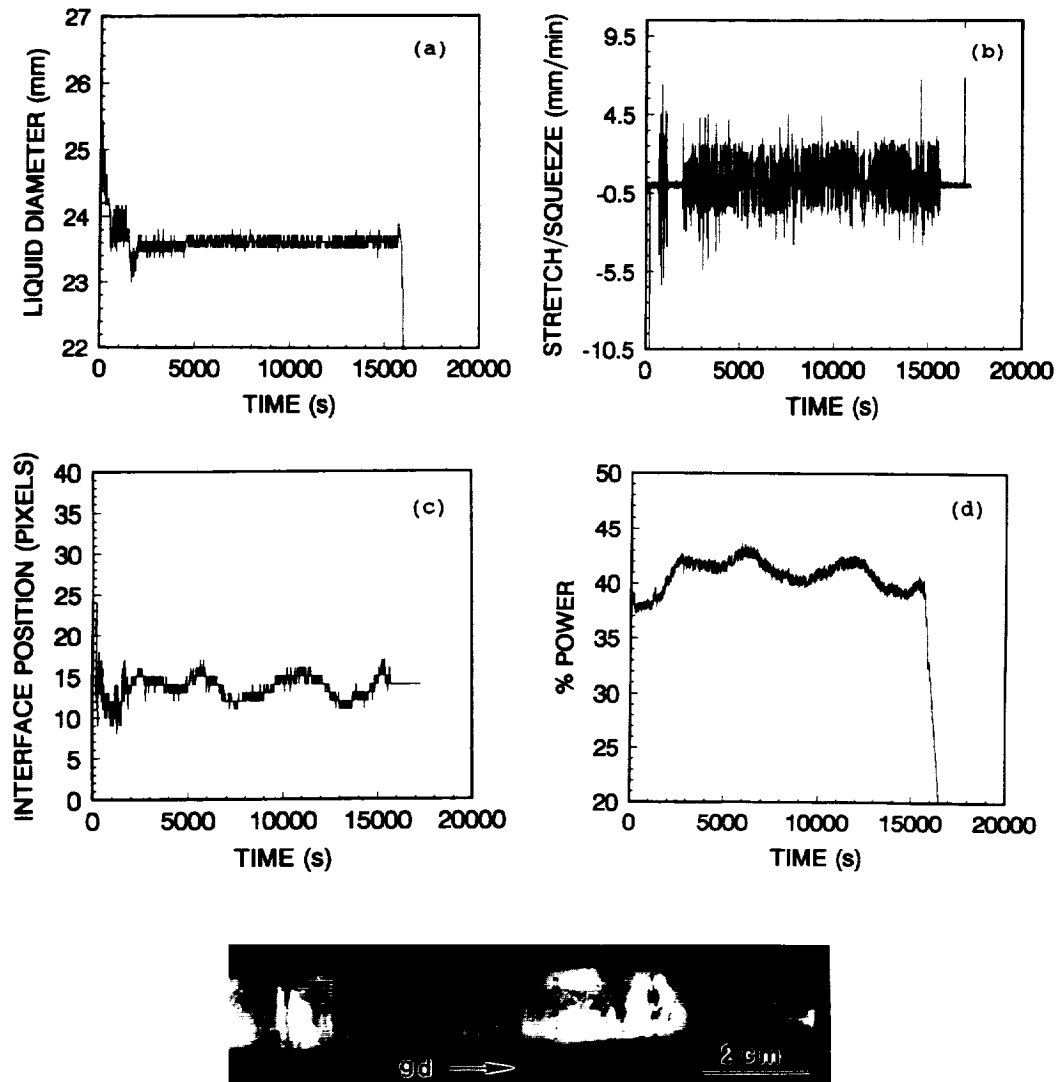
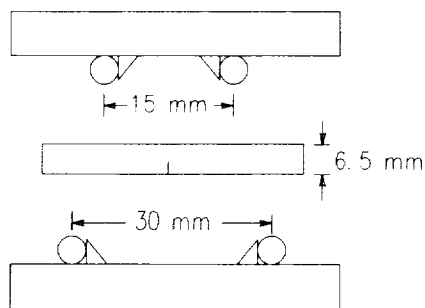


Fig. 2. Processing record for a NiAl 14-5Ta ingot: (a) diameter of the molten zone and (b) the corresponding stretch-squeeze control, (c) position of the solid-liquid interface and (d) the corresponding power variation, and (e) the as-processed NiAl 14-5Ta ingot (gd = growth direction).

To examine the effect of thermal exposure and cooling rate on fracture resistance, a number of bend specimens were heat treated prior to testing.



SAMPLE SIZE: 4.5 mm x 6.5 mm x 40 mm
NOTCH: 2.00 mm x 0.35 mm

Fig. 3. Geometry of the four-point flexure testing fixture and bend sample.

The samples were heat treated in air at 1000 K for 7.2 ks (2 h) and either air cooled, oil quenched, or water quenched. After heat treatment, the samples were notched and the room temperature fracture toughness was determined.

3.3 Materials characterization

Light optical microscopy was used to characterize the general morphology of the NiAl-NiAlTa alloys. Sections taken from directionally solidified and arc-melted ingots were metallographically prepared and etched with a solution of 5%HF-5%HNO₃-90% H₂O by volume.

Scanning electron microscopy (SEM) was also used to characterize the material. Backscattered electron imaging was used to help identify the phases present in the arc-melted and directionally

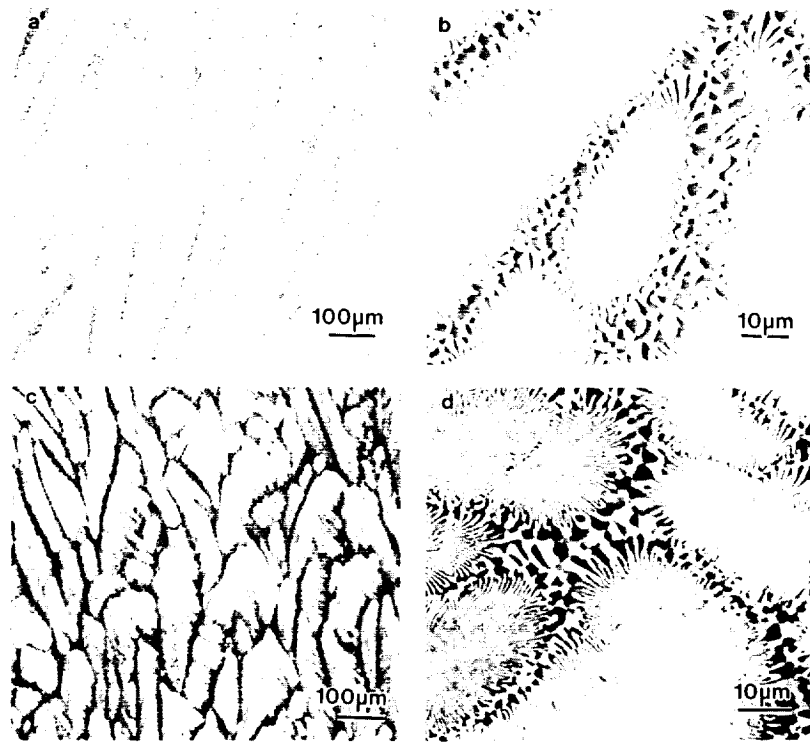


Fig. 4. SEM photomicrographs of arc-melted NiAl-NiAlTa alloys having a composition of (a),(b) NiAl-14.5Ta, and (c),(d) NiAl-15.5Ta.

solidified ingots. Quantitative X-ray analysis of the different phases was performed on a scanning electron microscope equipped with an energy dispersive spectrometer (EDS) detector. The 'ZAF' method, which attempts to correct the data for Z(atomic number), A(absorption) and F(fluorescence) effects, was used to determine the phase compositions. Finally, the fracture surfaces of the bend specimens were examined using secondary electron imaging.

Transmission electron microscopy (TEM) was used to further characterize the microstructure of the directionally solidified NiAl-NiAlTa eutectic. Thin slices were taken from broken bend specimens using a low speed diamond saw. Thinning was performed by grinding, followed by twinjet electropolishing in a solution of 5 vol% perchloric acid-95 vol% acetic acid at 40 volts and 300 K.

4 MICROSTRUCTURES

Fully coupled microstructures consisting of NiAl and NiAlTa occurred in arc-melted ingots of NiAl containing 14-16 at% tantalum. These microstructures consisted entirely of the eutectic NiAl-NiAlTa and contained no single phase dendrites, Fig. 4. Since alloy compositions that contain lower percentages of

tantalum are expected to have better oxidation resistance, a NiAl-14.5Ta (at%) alloy was initially chosen for further study. Large induction melted and drop cast ingots of this composition were directionally solidified for mechanical property evaluation.

Typical microstructures of the directionally solidified NiAl-14.5Ta alloy are shown in Figs 5(a) and (b). The microstructure of this alloy consists of NiAl dendrites surrounded by aligned eutectic regions. Since the directionally solidified ingot was processed at near-equilibrium conditions, the dendritic microstructure represents an off-eutectic composition. To better determine the eutectic composition, another NiAl-14.5Ta ingot was directionally solidified with a small amount of tantalum added to the initial molten zone. The added tantalum resulted in a well-aligned microstructure in the beginning section of the processed ingot (Figs 5(c) and (d)). A bulk eutectic composition of NiAl-15.5Ta (at%) was determined for this region by inductively coupled plasma atomic emission spectroscopy. Another NiAl-14.5Ta ingot processed in a similar manner (with additional tantalum) also resulted in a well-aligned, fully eutectic microstructure. This ingot was used for elevated temperature testing.

The NiAl-NiAlTa eutectic has a lamellar microstructure as shown by the SEM photomicrographs

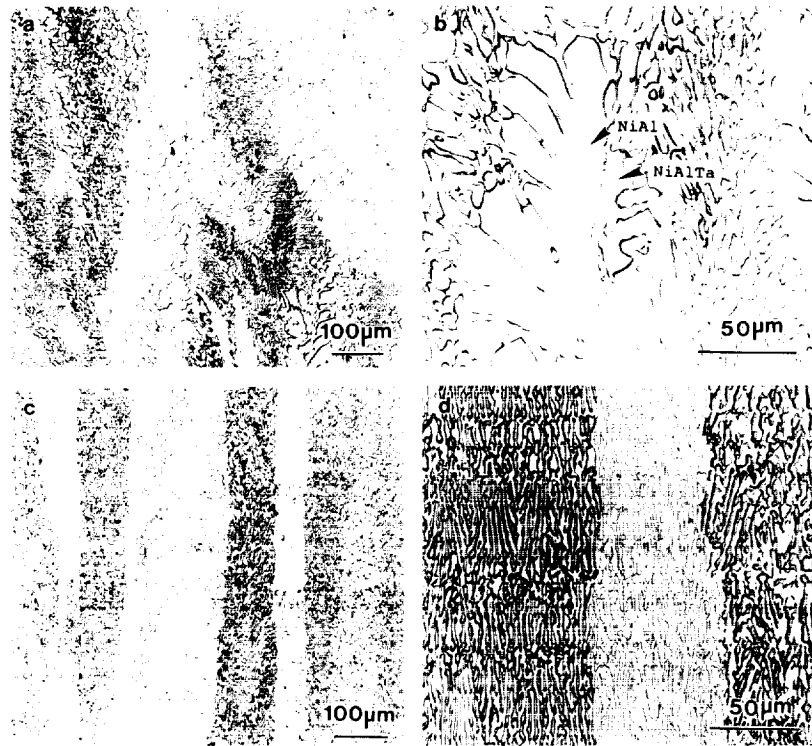


Fig. 5. Light optical photomicrographs showing (a),(b) a directionally solidified NiAl-4.5Ta ingot (19mm/h) and (c),(d) a directionally solidified NiAl-15.5Ta ingot (15 mm/h).

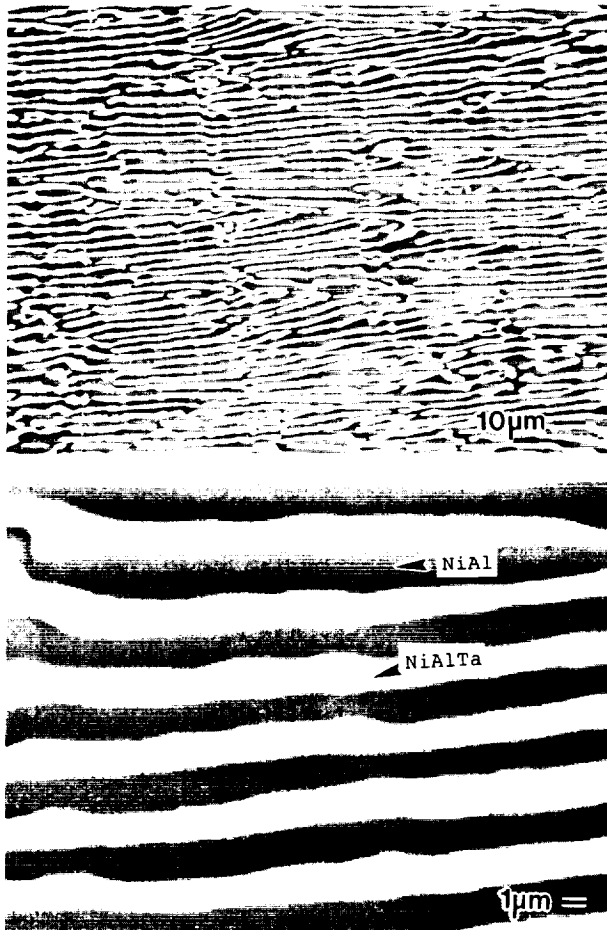


Fig. 6. SEM photomicrographs of the directionally solidified NiAl-NiAlTa eutectic (NiAl-15.5Ta).

in Fig. 6. The eutectic temperature was found to be near 1820 K by differential thermal analysis (DTA). The volume fractions of the two phases were estimated at 53% NiAl and 47% NiAlTa from SEM photomicrographs at 4000 \times . The composition of the individual eutectic constituent for the slowly cooled directionally, solidified NiAl-15.5Ta ingot was 50.5 Al, 48.5Ni and 1.0Ta for the NiAl phase and 34.0Al, 36.0Ni, 30.0Ta for the NiAlTa Laves phase as determined by quantitative X-ray analysis. A $\langle 100 \rangle$ growth direction was found for the NiAl phase by selected area diffraction patterns. No definite crystallographic relationship could be determined for the NiAlTa phase.

The arc-melted versus directionally solidified microstructures were different for the NiAl-14.5Ta alloy as shown by Figs 4 and 5. The microstructure for the arc-melted ingot was essentially eutectic while the directionally solidified ingot contained dendrites of NiAl. However, higher solidification rates and undercoolings are produced by the water cooled hearth during arc-melting compared to containerless solidification. Consequently, the difference in the arc-melted and directionally solidified microstructures suggests, that for rapid solidification rates, the coupled growth region is skewed towards lower tantalum contents at moderate undercoolings.

In an attempt to produce a more aligned microstructure in the NiAl-14.5Ta alloy, a cooling jacket

Table 1. Compositions and processing conditions for NiAl–NiAlTa alloys

Nominal compositions (at%)	Growth velocity (mm/h)	Rotational velocity (rpm)	Number of passes
NiAl 14.5Ta	19	75	2
NiAl 14.5Ta*	19	75	2
NiAl 15.5Ta**	15	75	1

*Cooling jacket used for increased thermal gradient.

**NiAl 14.5Ta ingot with extra Ta added to the molten zone before directional solidification.

was mounted below the water cooled eddy current plate (see Fig. 1) during one processing run. This was an attempt to increase the thermal gradient through the solid-liquid interface. A similar setup was previously used to produce successfully aligned microstructures from alloys in the NiAl–(Cr,Mo) eutectic system.⁷ However, there was no change in microstructure for the NiAl–14.5Ta alloy when processed with the cooling jacket. This indicates that a much greater thermal gradient may be needed for coupled growth at this composition. The composition and processing conditions of all the directionally solidified NiAl–NiAlTa alloys evaluated in this study are listed in Table 1.

5 ELEVATED TEMPERATURE STRENGTH

Most of the compression testing was performed on an off-eutectic NiAl–14.5Ta directionally solidified ingot processed at 19 mm/h with a microstructure typical of that shown in Figs 5(a) and (b). However, a limited number of tests were performed on an alloy with a fully eutectic microstructure (NiAl–15.5Ta), similar to that shown in Figs 5(c) and (d). The flow stress, σ , and strain rate, $\dot{\epsilon}$, data for these alloys were fitted to a temperature compensated-power law equation:

$$\dot{\epsilon} = A\sigma^n \exp(-Q/RT)$$

where A is a constant, σ is the steady state creep strength (MPa), Q is the activation energy for deformation (kJ/mol), T is the absolute temperature, R is the gas constant, and n is the stress exponent. The creep characteristics for the NiAl–NiAlTa alloys are compared to a NiAl–NiAlNb Laves alloy and single crystal NiAl in Table 2.

The elevated temperature strength of the off-eutectic NiAl–14.5Ta alloy over the temperature range of 1200–1400 K is shown in Fig. 7. Except for the fastest strain rate at 1200 K, where power-law breakdown behavior has occurred, the NiAl–

Table 2. Representative creep behavior for NiAl–Laves alloys compared to binary NiAl

Nominal composition (at%)	Representative creep behavior
NiAl[001] (Ni–50Al)	1110 1300K ²⁰ $\dot{\epsilon} = (1.48 \times 10^3)\sigma^{6.3}\exp(-439/RT)$
NiAl–NiAlNb (NiAl–16.5Nb)	200 1300 K: ¹⁴ $\dot{\epsilon} = (40.0)\sigma^{4.2}\exp(-415/RT)$
NiAl–NiAlTa (NiAl–14.5Ta)	200 1400 K: $\dot{\epsilon} = (4.55 \times 10^3)\sigma^{4.9}\exp(-522/RT)$
NiAl–NiAlTa (NiAl–15.5Ta)	1300 K: $\dot{\epsilon} = (3.06 \times 10^{21})\sigma^{6.0}$

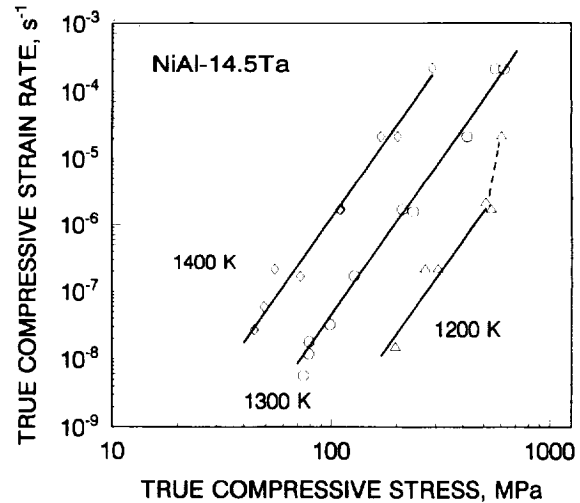


Fig. 7. 1300 K compressive flow stress–strain rate behavior for the directionally solidified NiAl–14.5Ta alloy as a function of temperature.

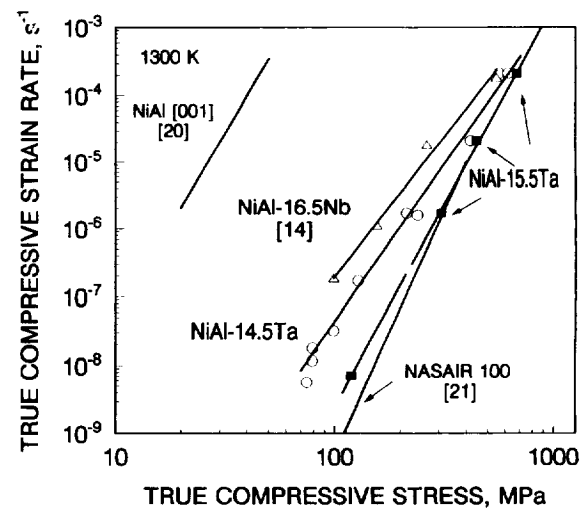


Fig. 8. 1300 K compressive flow stress–strain rate behavior for directionally solidified NiAl–Laves alloys compared to NiAl and a single crystal superalloy.

14.5Ta alloy exhibits a very consistent power law behavior over the strain rates and temperatures investigated.

In Fig. 8, the 1300 K compressive creep behavior of the NiAl–Laves alloys are compared to single

crystal binary NiAl²⁰ and a nickel-based single crystal superalloy.²¹ The NiAl-Laves alloys have significantly higher strengths compared to binary NiAl. Moreover, the strengths of the NiAl-NiAlTa alloys are greater than that of the NiAl-NiAlNb eutectic. However, the NiAl-14.5Ta alloy is still marginally weaker than a single crystal nickel-based superalloy.

Conversely, the 1300 K compressive strength of the NiAl-15.5Ta eutectic alloy is comparable to that of the nickel-based superalloy as shown in Fig. 8. While the data for the NiAl-15.5Ta alloy is limited to only 1300 K tests, the data does represent a modest improvement in creep resistance due to a modification of the microstructure. The improvement in strength is likely due to a refinement in microstructure for the fully eutectic alloy and the elimination of the large NiAl dendrites. The lamellar spacing of the eutectic microstructure was measured at 2–3 μm from SEM and TEM photomicrographs. This is much finer than the phase distribution found in the NiAl-14.5Ta alloy (Fig. 5(b)).

6 ROOM TEMPERATURE TOUGHNESS

The NiAl-NiAlTa alloys exhibit brittle behavior at room temperature with a fracture toughness of approximately 5 $\text{MPa}\sqrt{\text{m}}$. However, this is not considerably different from typical toughness values reported for single crystal or polycrystalline NiAl materials.² A total of nine fracture toughness specimens were tested. Three fracture specimens were tested in the as-processed condition. The remaining specimens were heat treated at 1000 K for 7.2 ks (2 h) and then either air cooled, oil quenched, or water quenched. The results of all these tests are summarized in Table 3.

Table 3. Room temperature fracture toughness of directionally solidified NiAl-NiAlTa alloys (NiAl-14.5Ta) after various heat treatments

Material condition	K_{Q} ($\text{MPa}\sqrt{\text{m}}$)
As-processed	4.1
	5.4
	4.1
Heat treated at 1000K for 7.2 ks, Air cooled:	5.0
	4.9
Oil quenched:	5.7
	6.3
Water quenched:	5.1
	6.0

Table 4. Room temperature fracture toughness of alloys containing NiAl and Laves phases

Material	Morphology	K_{Q} ($\text{MPa}\sqrt{\text{m}}$)
NiAl	Polycrystalline (HIP)	6 ²³
NiAl [001]	Single crystal (DS)	11 ⁵
NiAlNb	Single phase (HIP)	2 ²⁸
NiAl-NiAlNb	Eutectic (cast)	4 ²⁸
NiAl-14.5Ta	Near-Eutectic (DS)	5.1 \pm 0.8

HIP = hot isostatically pressed powder metallurgy
DS = directionally solidified

Hack *et al.*²² have shown improvements in the fracture toughness of NiAl when rapidly cooled through the temperature range of 673–300 K. Their results suggest that NiAl may be susceptible to strain-aging embrittlement. However, the post-processing heat treatments used in this study provided no improvement in toughness of the NiAl-NiAlTa alloys and a fracture toughness of approximately 5 $\text{MPa}\sqrt{\text{m}}$ was measured from all samples.

In Table 4, the fracture toughness of the NiAl-14.5Ta alloy is compared to NiAl and the NiAl-NiAlNb eutectic. The fracture toughness of the NiAl-NiAlTa alloy is comparable to that of polycrystalline NiAl.²³ Hence, the large increase in creep strength is not gained at the expense of fracture toughness. However, the fracture toughness of the NiAl-NiAlTa eutectic is less than that of single crystal NiAl having a [001] notch plane. Assuming that the fracture toughness of the NiAlTa phase is the same as that of NiAlNb, i.e. 2 $\text{MPa}\sqrt{\text{m}}$, then the fracture toughness of the NiAl-NiAlTa eutectic is very close to that predicted by the rule of mixtures, 6.5 $\text{MPa}\sqrt{\text{m}}$ predicted versus 5.1 \pm 0.8 $\text{MPa}\sqrt{\text{m}}$ measured.

Therefore, materials containing a large volume fraction of the extremely brittle Laves phase can be reliably produced, handled, and tested by processing alloys near the eutectic composition. Ironically, NiAl is the toughening component in these two-phase alloys. The fracture surface of a NiAl-NiAlTa bend specimen is shown in Fig. 9. The eutectic microstructure is notably visible from the fracture surface as a result of partial debonding between phases during fracture. Further evidence of this fracture behavior is shown in Fig. 10. A section of the directionally solidified NiAl-14.5Ta ingot was polished and then broken with the polished surface in tension. The resulting fracture profile reveals a series of microcracks in the Laves phase with the NiAl phase bridging the crack path. In addition, cracking is also visible along the NiAl/NiAlTa phase boundaries.

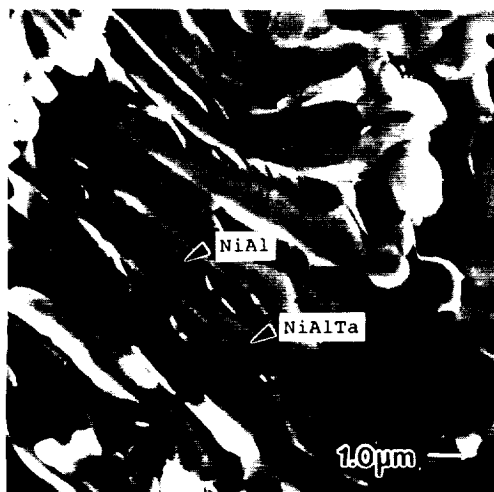
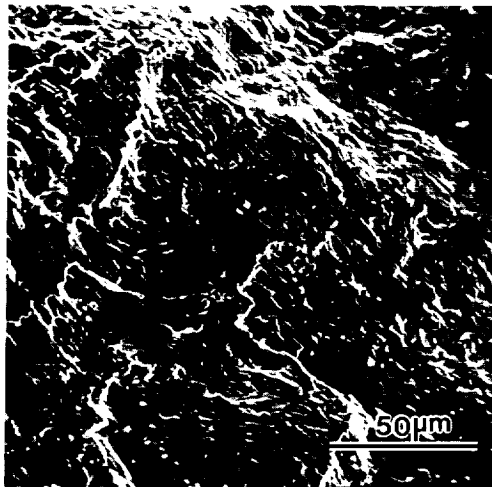


Fig. 9. SEM photomicrographs of fracture surfaces from a NiAl-14.5Ta bend specimen.

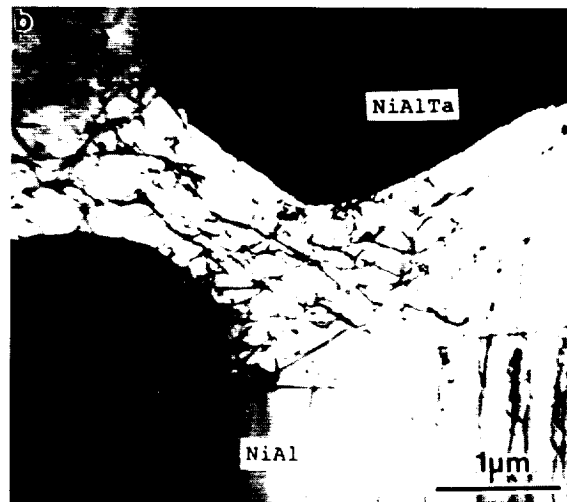
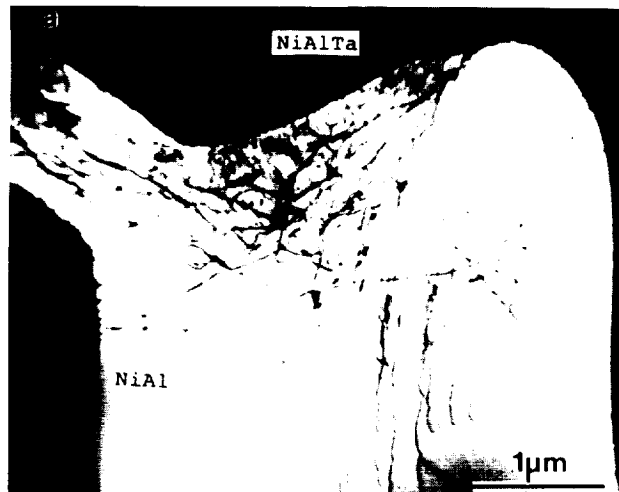


Fig. 11. TEM photomicrographs of samples taken near the fracture surface of a NiAl-14.5Ta bend specimen.

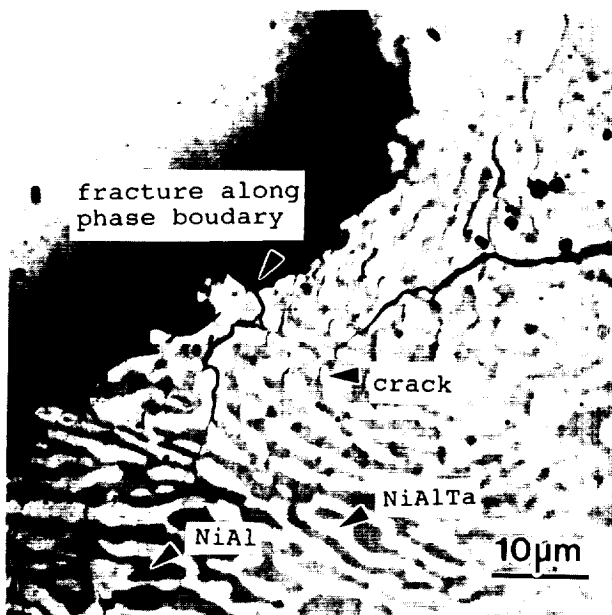


Fig. 10. Light optical photomicrograph of the fracture region from a directionally solidified NiAl-14.5Ta alloy.

Figure 11 is a TEM micrograph of the NiAl phase near the fracture surface of a broken bend specimen. A high dislocation density was found in the NiAl phase indicating possible plasticity of the material prior to fracture. The microstructure of NiAl consists of coarse dislocation tangles with the dislocation density greatest in the sections where the NiAl lamellae are the thinnest. Material taken away from the fracture surface contained a much lower dislocation density than that shown in Fig. 11. These observations suggest that the NiAl phase provides most of the fracture toughness in these brittle composite alloys as suggested by the rule of mixtures analysis. No precipitates were found in the NiAl phase.

7 DISCUSSION

The directionally solidified NiAl-NiAlTa alloys exhibit good creep resistance. In fact the 1300 K

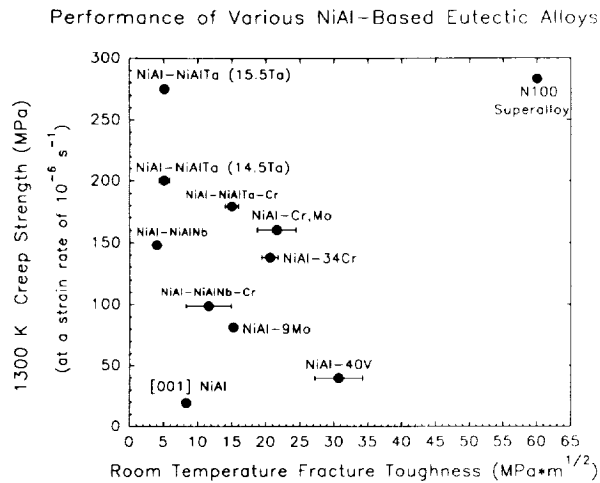


Fig. 12. Performance of NiAl-based directionally solidified eutectics compared to binary single crystal NiAl and a single crystal superalloy.

compressive strength of the fully eutectic NiAl-15.5Ta alloy approaches that of a single crystal nickel-based superalloy (Fig. 12). The modest improvement in strength of the NiAl-NiAlTa eutectic alloy compared to the NiAl-14.5Ta alloy is probably due to a refinement of the microstructure. Sauthoff has shown that the strength of NiAl-Laves alloys follows a simple rule of mixtures behavior except for very fine phase distributions.^{13,24} Additional strengthening is provided once the lamellar spacing becomes less than a critical value corresponding to the mean free dislocation path.²⁴ Klöwer and Sauthoff have studied the effects of lamellar spacing on the creep behavior of Ni-Al-Fe alloys and have calculated a critical lamellar spacing of 8 μm to benefit from additional contributions in strength due to phase boundary strengthening.²⁵ The preliminary results for the creep resistance of the NiAl-NiAlTa eutectic with a lamellar spacing of 2-3 μm are consistent with their argument.

However, the directionally solidified NiAl-15.5Ta alloy has a coarse eutectic spacing when compared to NiAl-refractory metal eutectics such as a NiAl-Cr and NiAl-(Cr,Mo), which exhibit eutectic spacings of less than 1 μm .⁷ As the eutectic spacing was relatively coarse even for the quickly cooled, arc-melted NiAl-NiAlTa alloy (Fig. 4), lamellar spacing is probably not a strong function of processing conditions. Hence further significant strengthening of the NiAl-NiAlTa eutectic alloy is not expected at greater solidification rates.

One method to further improve the strength of the NiAl-NiAlTa alloys may simply be to improve the strength of the NiAl phase. Promising

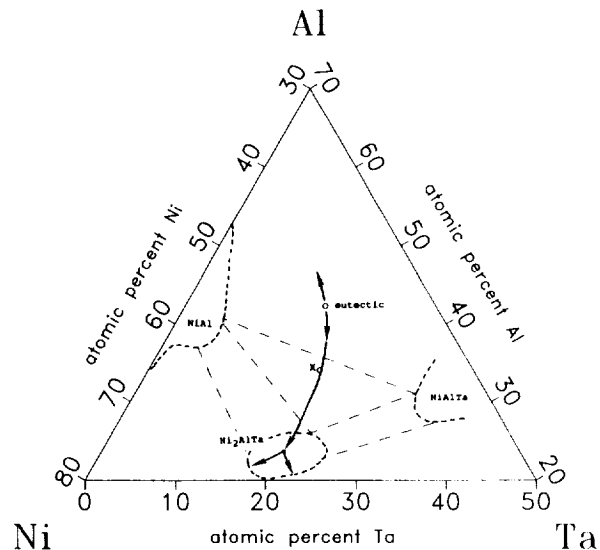


Fig. 13. A 1753 K isotherm from the Ni-Al-Ta system¹⁶ combined with a schematic of the liquidus surface.²⁹

strengths have been found for two phase material consisting of NiAl and the Heusler precipitate Ni_2AlTa .⁹ Hence, alloys heat treated in the three phase region shown in Fig. 13 may show improved strengths. For example, an arc-melted ingot with the composition near the NiAl-NiAlTa liquidus trough (indicated by the *x* in Fig. 13) lies well within the NiAl- Ni_2AlTa -NiAlTa region on the 1753 K isotherm. The cast microstructure of this alloy (Fig. 14) consists mainly of NiAl-NiAlTa dendrites surrounded by an interdendritic Heusler phase, Ni_2AlTa . Post-processing heat treatments of this alloy should produce precipitates of the Heusler phase within the NiAl phase. Alloys with compositions closer to the NiAl-NiAlTa eutectic should contain smaller percentages of the interdendritic Heusler phase. Hence, it may be possible to produce aligned NiAl-NiAlTa microstructures with NiAl strengthened by the precipitates of the Ni_2AlTa phase.

While the NiAl-NiAlTa alloys show promising creep strengths, the room temperature fracture toughness of these alloys is poor. One scheme for improving the toughness is to include a metallic phase within the NiAl-NiAlTa microstructure. From Fig. 12, it is evident that the NiAl-Cr and the NiAl-(Cr,Mo) alloys have a much higher fracture toughness than the NiAl-Laves phase alloys.⁷ Furthermore, a ternary eutectic exists between the NiAl, NiAlTa, and Cr phases,²⁶ as shown in Fig. 15. Fracture toughness and elevated temperature strength data for this ternary eutectic fall somewhere in between those of the separate binary eutectics.²⁷ Since NiAl, NiAlTa, and Cr are all

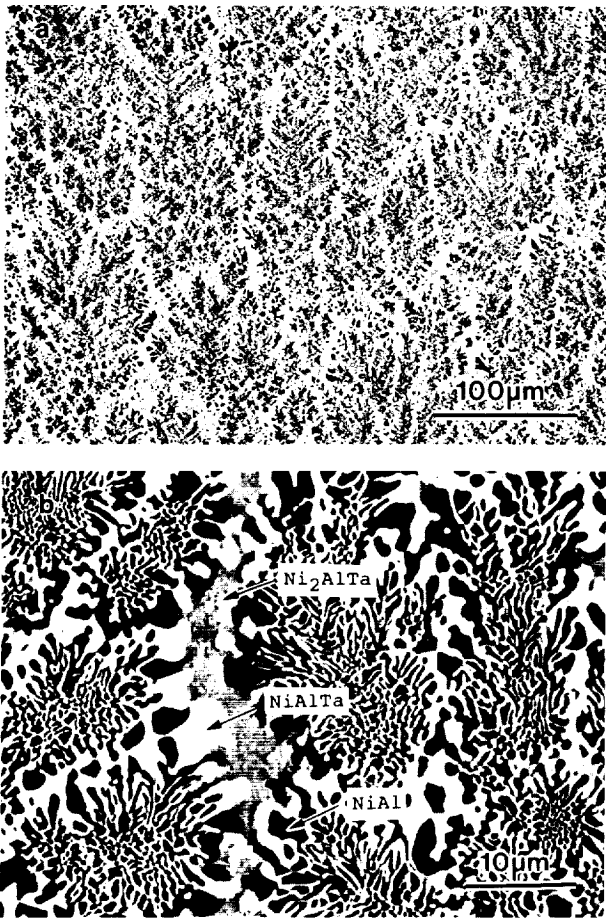


Fig. 14. SEM backscattered electron photomicrographs of an arc-melted Ni-34Al-19Ta alloy in the as-cast condition.

thermodynamically compatible, composite materials formed by combining the NiAl-NiAlTa and the NiAl-Cr eutectics may be possible.

8 CONCLUSIONS

Promising creep strengths approaching that of a single crystal nickel-based superalloy were obtained in a directionally solidified NiAl-NiAlTa eutectic alloy. While the Laves phase NiAlTa is extremely brittle, materials that could be reliably handled and tested were produced by directional solidification of the NiAl-NiAlTa eutectic. Through an empirical alloy design approach it may be possible to produce materials with even better creep resistance and fracture toughness by the generation of multi-phase alloys.

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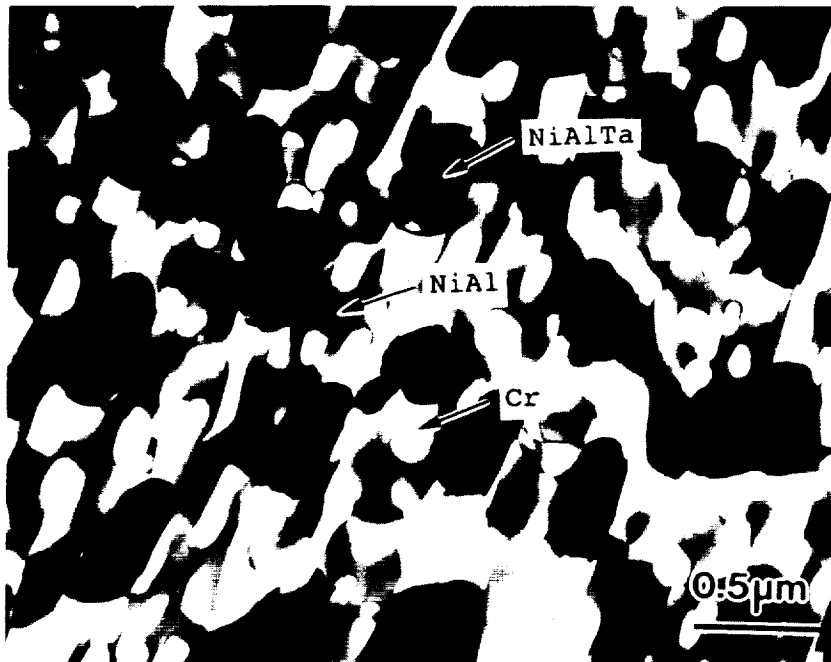


Fig. 15. TEM photomicrograph of the NiAl-NiAlTa-Cr ternary eutectic in an as-cast Ni-30.5Al-6Ta-33Cr ingot.

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