

Self-Cleaning Boudouard Reactor for Full Oxygen Recovery from Carbon Dioxide

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Outline

- Importance of oxygen recovery from carbon dioxide
- Self-cleaning reactor designs at KSC
- Results
- Future Work



O₂ Recovery from CO₂

- Only 50% of O₂ can recovered from respiratory CO₂ on the ISS
- Sabatier reactor makes CH₄ and H₂O
- CH₄ is vented, losing H₂
- H₂O from cargo limits H₂ availability to 50% recovery
- RFP seeks at least 75% recovery
- Deep space missions (Moon, Mars moons, Mars surface, asteroids, etc.) need closer to 100% recovery



Bosch Reaction

- Bosch Reaction: $CO_2 + H_2 \rightarrow C_{(s)} + 2 H_2O (\rightarrow 2 H_2 + O_2)$
- RWGS: $CO_2 + H_2 \rightarrow CO + H_2O (\rightarrow H_2 + \frac{1}{2}O_2)$
- Boudouard: 2 CO \rightarrow C_(s) + CO₂ (Fe catalyst, H₂ enhancer)
- Need a method to remove C from catalyst as it forms





Design Concepts

- Criteria: Expected Durability, catalyst surface area, mechanical interface, ease of use/fabrication, ability to evaluate design variations in same reactor.
- Did not seek to choose the best catalyst
- Most concepts centered around a catalyst that was either a brush or springs
- Others included planetary gears (like a pencil sharpener), ball bearings



Brush Design Concepts

- Catalytic brush with mechanism for carbon removal
- Variations included the number of brushes and method of carbon removal







Spring Design Concepts

 Catalytic springs with different mechanisms that compress/release springs to remove carbon





Initial design

- Spinning carbon steel spiral brush with brass rods
- Stainless steel reactor body





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Initial Design

- Tested steel wool reactor for comparison
- Tested 1" and 2" ID reactors
- Collected carbon in HEPA^{Reactor} Flow filter bag as it was generated (1 of 3)

Wrapped Controller









Methods

- CO, H₂, N₂ fed into reactor
- Reactor temperature 500-600 °C
- Carbon collected and weighed

Parameters for Each Reactor

	1" REACTOR	2" REACTOR
REACTOR VOLUME, ML	76	300
CATALYST MASS, G	1.31	11.82
H ₂ FLOW, SCCM	232	909
CO FLOW, SCCM	232	909
N ₂ FLOW, SCCM	52	202





Methods

- Product Gas quantified with GC
- A total carbon balance was used with the GC data to calculate CO₂ yield

 $yield = \frac{mol \ CO_2 \ produced}{0.5 \times mol \ CO \ in}$





Results

- 1" reactor ran for 12 h
 - Reached 51% CO₂ yield, collected 27% of C in filter bag (5.5g in filter bag, 20.5 g total)
 - Found to be damaged upon disassembly

- 2" reactor run for 37 h before failure
 - Reached 73% conversion, collected 25% of C in filter bag
 - Equivalent to 1 crew $CO_2 \rightarrow O_2/day$
 - Multiple modules + RWGS can recover
 ALL the O₂ on ISS





Results: 1 inch reactor

- After 12 hours of test time, the reactor jammed
- Brush bristles had become knotted and brush was starting to fall apart
- Some carbon still in reactor







Results: 2 inch reactor

- Pressure inside the reactor began to increase after 27 hours, and reactor was stopped after 37 hours due to the pressure increase
- Reactor x-rayed to determine cause



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Carbon Analysis

- Carbon analysis with SEM/EDA indicated iron was present
- Source is likely the brush



Secondary electron, left, and backscatter electron, right, images of carbon collected from the two inch reactor. The bright spots in the right image are iron

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Future Work

- New design: Catalytic wall with non-catalytic scraper
- Using pipe inserts as catalyst so it will protect the reactor wall
- Different inserts could be made of different catalysts





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