1	Major and trace element geochemistry of El Chichón volcano-hydrothermal system
2	(Chiapas, Mexico) in 2006-2007: implications for future geochemical monitoring
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13	RESUMEN
14	Se presenta un estudio detallado de la composición isotópica y química (elementos
15	mayores y traza) del lago cratérico, Soap Pool y de los manantiales termales en el Volcán
16	Chichón para el período noviembre 2006-octubre 2007. Después de dos décadas de
17	estudios del lago cratérico, se confirma la relación compleja entre la distribución anual de
18	la precipitación y el volumen y la química del lago cratérico: durante tres años (2001,
19	2004 and 2007) un volumen grande del lago rico en Cl se relaciona con la reactivación de
20	la descarga alta (>10 kg/s) hacia el lago de aguas salinas con un pH cerca de la
21	neutralidad de los manantiales hirvientes Soap Pool, únicamente algunos meses (~enero)
22	después del final de la temporada de lluvias (junio-octubre). El volumen más grande
23	jamás observado occurrió en marzo 2007 (~6x10 ⁵ m ³).
24	No obstante el hecho que los manantiales termales de Agua Tibia 2 descargan al pie del

domo SO, su química indica un regimen de temperaturas más bajas, una interacción aguaroca y una contribución del basamento (evaporitas y carbonatos) más avanzada y una

27 lixiviación de anhidrita de los depósitos piroclásticos de 1982; el domo SO no demuestra 28 actividad. Existen nuevas evidencias sobre una posible infiltración del lago cratérico hacia el manantial de Agua Caliente. Los modelos existentes del sistema "lago cratérico-29 30 Soap Pool" y el sistema hidrotermal más profundo se justifican y se detallan. Creemos 31 que los cambios químicos en el acuífero geothermal profundo alimentando los 32 manantiales termales, podrían antecipar el crecimiento futuro de un domo. El monitoreo 33 volcánico futuro se tendría que enfocar en los cambios en la química de los manantiales, 34 aparte del monitoreo del lago cratérico.

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PALABRAS CLAVE: El Chichón, geoquímica de fluidos, monitoreo volcánico.

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38 ABSTRACT

39 We report a detailed study of isotopic, major and trace element composition in the 40 crater lake, Soap Pool and thermal springs at El Chichón volcano for the period 41 November 2006-October 2007. After two decades of studying the crater lake, it is 42 possible to confirm the complex relationship between the annual rainfall distribution and the crater lake volume and chemistry: during three years (2001, 2004 and 2007) a large 43 volume high-Cl lake can be related to the reactivation of high discharge (>10 kg/s) of 44 45 saline near-neutral water from the Soap Pool boiling springs towards the lake, only a few months (~January) after the end of the rainy season (June-October). The highest lake 46 volume ever observed occurred in March 2007 ($\sim 6x10^5 \text{ m}^3$). 47

Despite the fact that the Agua Tibia 2 thermal springs discharge at the foot of the SW dome, their chemistry indicates a lower temperature regime, an enhanced water-rock interaction and basement contribution (evaporites and carbonates), and anhydrite leaching from the 1982 pyroclastic deposits, rather than dome activity. New suggestions on crater lake seepage are evidenced by the Agua Caliente thermal springs. 53 54

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Existing models on the "crater lake-Soap Pool spring" and the deep hydrothermal system are justified and detailed. We believe that chemical changes in the deep geothermal aquifer feeding the thermal springs will anticipate dome rise. Future volcanic surveillance should focus on the changes in spring chemistry, besides crater lake monitoring.

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KEY WORDS: El Chichón, fluid geochemistry, volcanic surveillance.

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1. Introduction

El Chichón volcano is located in northwest Chiapas, southern Mexico, in between 61 62 two major volcanic regions, the Transmexican Volcanic Belt and the Central American 63 Volcanic Arc (Fig. 1(a)). Dome destroying eruptions are frequent at El Chichón: each 100-600 years during the past 8,000 years (Espíndola et al., 2000). In fact, El Chichón is 64 65 a dome complex volcano: two "fresh looking" domes (NW and SW) are the main morphological features of the volcanic edifice formed by the 2 km wide Somma crater 66 67 (~0.2 Ma, Damon and Montesinos, 1978; Duffield et al., 1984; Layer et al., this volume) 68 (Fig. 1(b)); moreover, the two centrally nested domes were blasted away in 1982 (Luhr et 69 al., 1984; Varekamp et al., 1984).

70 During the last two decades, the El Chichón volcano-hydrothermal system has 71 been extensively studied (Taran et al., 1998; Tassi et al., 2003; Capaccioni et al., 2004; 72 Rouwet et al., 2004; Rouwet, 2006; Taran and Rouwet, 2007; Rouwet et al., 2008; Taran et al., 2008). As for other crater lake bearing volcanoes (Giggenbach, 1974; Takano, 73 74 1987; Ohba et al., 1994; Christenson, 2000), crater lake dynamics and chemistry seems to 75 be an adequate monitoring tool in predicting volcanic crisis at El Chichón. Surprisingly, 76 the 1982 Plinian eruptions were not followed by dome growth in the 1 km wide-200 m 77 deep explosion crater. Although changes in chemistry of thermal springs at active 78 volcanoes have scarcely proven to be efficient precursors before magmatic crisis, we believe that the El Chichón spring network is adapted to start up volcanic surveillance.
Before this, one needs to know the baseline behaviour of the springs during quiescent
degassing periods; the main goal of this study.

82 Taran et al. (2008) reported for the first time the chemistry of the Agua Tibia 2 springs (AT2 thereafter), discharging from beneath the SW dome. Since April 1998, the AT2 83 84 springs have been revisited only twice: in November 2006 and March 2007. This study 85 could be the missing link to shed light on (1) the relationship of the AT2 springs with the state of activity of the volcano, and (2) the connection between the crater manifestations 86 87 and the thermal springs. Crater lake seepage, is a common feature at crater lake hosting 88 volcanoes, justified by similarities in major element contents, Cl/SO₄ ratios and pH (e.g. 89 Poás, Copahue, Kawah Ijen, Patuha, Keli Mutu, Rincón de la Vieja, Ruapehu; Pasternack 90 and Varekamp, 1994; Rowe et al., 1995; Sanford et al., 1995; Deely and Sheppard, 1996; 91 Sriwana et al., 1998; Delmelle and Bernard, 2000, Kempter and Rowe, 2000, Varekamp 92 et al., 2001, Löhr et al., 2005). This process was suggested for the highly saline Agua 93 Salada acidic springs discharging at the NW dome (Taran et al., 2008), but these springs 94 are not necessarily the only direct output of crater lake water at El Chichón.

We propose new ideas on the southern parts of the El Chichón volcano-hydrothermal system, by adding chemical and isotopic data on Agua Caliente (AC thereafter) and the AT2 thermal waters. The existing model for the "lake-Soap Pool spring" system in the crater (Rouwet *et al.*, 2004; Taran and Rouwet, 2007; Rouwet *et al.*, 2008) is tested by introducing recent data on crater lake volume, and lake and Soap Pool geyser-like boiling spring chemistry.

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2. Hydrothermal manifestations at the El Chichón dome complex

106 El Chichón is an actively degassing volcano, as manifested by boiling temperature fumaroles, abundant bubbling degassing through the lake bottom, and diffuse degassing 107 108 inside the 1982 crater. An acidic (pH = 2.2-2.7), warm (T $\sim 30^{\circ}$ C) and shallow crater lake 109 (1 to 3.3 m deep) covers the largest part of the crater floor at ~850 m a.s.l (Fig. 2(a)). This 110 lake changes dynamically in chemistry and volume, due to its direct connection with a 111 group of geyser-like boiling springs on the northern shore of the lake, called "Soap Pools" 112 by Taran et al. (1998) (SP thereafter). The SP alternates periods of high discharge of near 113 neutral, saline and Cl-rich waters towards to lake with periods of exhalation of only 114 vapour. Based on a chemical and isotope balance approach, the total heat output from the crater is estimated to be 35-60 MW, and the diffuse CO₂ flux from the crater is not lower 115 than 140 g/m²day (Taran and Rouwet, 2007). This last value coincides with the minimum 116 117 values obtained by direct CO₂ flux measurements from the crater lake surface (March 118 2007; Mazot and Taran, this volume).

119 Besides the crater manifestations, El Chichón is known for its extremely high discharge 120 of thermal waters (>300 kg/s), through numerous springs near domes outside the 1982 121 crater, at the contact between the permeable volcanic edifice and sedimentary basement. All thermal springs finish into Río Magdalena, finally draining towards the Gulf of 122 123 Mexico (Fig. 1(b)). The total thermal heat output through the springs is estimated to be 124 ~100 MW (Rouwet, 2006). Despite the fact that pre-1982 reports (Molina Berbeyer, 1974; Templos et al., 1981) already mention thermal springs at El Chichón, up until now, 125 126 the springs have not been targeted during volcanic surveillance, mainly because they are 127 located in remote and densely vegetated canyons. The "fresh looking" domes (Fig. 2(a-128 b)), outside the 1982 crater, may still be rooted by a cooling magma body, possibly 129 remaining the principal heat source at the present hydrothermal system.

130 The AC springs discharge in the southeastern horseshoe-shaped canyon (Guyabal Tuff 131 cone, <10 ka; Layer *et al.*, this volume), testifying Holocene explosive activity (Fig. 1(b) and 2(c)-(d)). Based on the Cl content of Río Tuspac, Taran et al. (1998) estimated the 132 133 total outflow rate of the AC springs to be >100 kg/s. The temperature of the slightly acidic to near-neutral AC waters (pH 5.7-7.6) remained stable at ~71°C since before 1982 134 135 (Molina Berbeyer, 1974; Templos et al., 1981). The conductivity varies from 4.3 to 5.9 136 mS/cm (2004-2007), representing similar salinities than the present crater lake (Table 1). 137 The AC springs form numerous cascades and hot water spurts discharging into densely vegetated swampy pools (Fig. 2(d)). Weak bubbling degassing can be observed at AC, 138 139 and coinciding with negative Eh values, this indicates a H₂S input (Taran *et al.*, 1998). The AC springs discharge at less than 1.5 km distance from the 1982 crater with a vertical 140 141 difference of only 200 m.

142 Some 2.5 km west of AC, the AT2 springs discharge east of the SW dome (Fig. 1(b) 143 and 2(b)). This dome shows surface alteration, the result of past fumarolic activity (dome 144 age 217 ± 10 ka; Layer *et al.*, this volume); active fumaroles cannot be distinguished. A 145 total discharge of 80 kg/s of the AT2 thermal springs is estimated (Taran et al., 2008). In 146 April 1998, discharge temperatures of AT2 waters of 51-49°C were detected, similar to 147 the pre-1982 temperature reported by Templos et al. (1981). We measured lower 148 discharge temperatures for various AT2 springs in November 2006 and March 2007 (35.4 149 to 46.5°C, Table 1). The AT2 spring mainly discharges from beneath rocks in the river bed, forming numerous pools downstream of 10-20 meters of translucent whitish-150 151 turquoise water (Fig. 2(e)). Abundant amorphous milky-white precipitates can be 152 observed at the bottom of the pools. More downstream, Fe-oxy-hydroxides colour the 153 river bed orange, such as at AC. The horizontal distance of the AT2 springs from the 154 crater lake is ~1.6 km, with a height difference of ~150 m. Besides temperature, the conductivity is also lower than for AC waters, varying from 1.8 to 4.2 mS/cm (Table 1). 155

Eh values are negative, which might indicate an H_2S input, although no strong evidence on bubbling degassing exists. The pH of AT2 waters is slightly lower than at AC: 5.15 to 5.85 (2006-2007).

The most acidic and saline waters of the entire volcano-hydrothermal system discharge at the northwestern ends of El Chichón (Agua Salada, pH ~2.2-5.6, 79°C), north of the younger NW dome (age 90 \pm 18 ka; Layer *et al.*, this volume) (Fig. 1(b)). The Agua Tibia 1 springs (68°C) are similar to AC waters (Taran *et al.*, 2008) (Fig. 1(b)). The Agua Salada and Agua Tibia 1 springs were not revisited since 2005. Nevertheless, the head of the Agua Salada canyon was visited in March 2007. The Agua Suerte cold spring discharges here at <1km distance from the crater at an elevation of ~800 m (Fig. 1(b)).

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3. Sampling and analytical methods

168 The crater lake was sampled in November 2006, and January, March, September 169 and October 2007. The SP water was collected during the March 2007 campaign. In 170 November 2006 and September-October 2007, the SP geyser only emitted vapour, so no 171 water sample could be collected. The AC springs were sampled in November 2006 (Fig. 172 2(d)); the AT2 springs in November 2006 and March 2007. Temperature, pH, 173 conductivity and Eh were directly measured at the lake surface or spring outlet. Waters 174 were stored in polyethylene bottles after passing through 0.45 µm filters. Samples for 175 cation analyses were acidified in the field with a 60% HNO₃ solution. Anion analyses 176 were elaborated from non-acidified filtered samples. Water samples for minor and trace element analyses were stored in ultra-pure HDPE Nalgene flasks and acidified by a 60% 177 178 HNO₃ and 60% HCl solution in 4/1 proportions.

Major element contents (Na, K, Ca, Mg and F, Cl, SO₄) were obtained by Liquid Chromatography (Dionex) with an accuracy of 3%. HCO₃ concentrations were detected by means of titration with a 0.01N HCl solution of 10 ml of a non-filtered non-acidified

sample aliquot. SiO₂ concentrations were measured by colorimetric photospectrometric 182 183 methods on diluted samples. Trace element concentrations were determined by ICP-MS (Agilent 7500 CE). All determinations were performed with the external standard 184 185 calibration method, using Re and In as internal standards. The accuracy of the results (±5%) was obtained by analyzing certified reference materials (NRCSLR-4, SPS-SW1 186 187 and NIST-1643e). The water samples were analyzed for their oxygen and hydrogen 188 isotopic composition, using Analytical Precision AP 2003 and Finnigan MAT Delta Plus 189 spectrometers, respectively. The isotope ratios are expressed as the deviation per mil (δ %) from the reference V-SMOW. The uncertainties are $\pm 0.1\%$ for δ^{18} O and $\pm 1\%$ for δ D (one 190 191 standard deviation).

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4. Results and discussion

194 4.1 Stable isotopes and major ion species

195 4.1.1 Isotopic composition of thermal waters: δD and $\delta^{I8}O$

The stable isotopic composition (δD and $\delta^{18}O$) of thermal and cold spring waters 196 at El Chichón are presented in Table 1, and plotted in Fig. 3. Considering δD and $\delta^{18}O$ of 197 198 crater lake and SP waters, the same tendency as observed in earlier years is repeated: (1) 199 in November 2006 the SP did not discharge water towards the crater lake, and consequently, δD and $\delta^{18}O$ of the crater lake water clearly follow the evaporation trend 200 201 originating from the local meteoric water, and (2) in March 2007, the SP water discharge towards the lake was high, and the δD and $\delta^{18}O$ of lake and SP waters plot near each 202 203 other in Fig. 3, although the evaporation effect for the lake water is not excluded. The δD and δ^{18} O for the March 2007 SP water plot near the range established in Rouwet *et al.* 204 205 (2008) of -8 ± 2 ‰ and $+1.5\pm0.7$ ‰, respectively. River and cold spring waters (i.e. Agua 206 Suerte) determine the correct isotopic composition of local meteoric water, slightly

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shifted to the left off the meteoric water line (MWL in Fig. 3). The isotopic composition 208 of AC and AT2 thermal spring waters plot near the values for the local meteoric waters 209 (Fig. 3).

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4.1.2 Major ion species

212 The major element composition of the 2006-2007 thermal waters at El Chichón 213 are shown in Table 1. All thermal waters at El Chichón, flank springs, as well as crater 214 fluids, are of a Na-Ca-Cl-SO₄ type. As Cl behaves as a conservative element, the cation 215 vs Cl plots (mixing plots) serve to distinguish main ion sources. Cationic species roughly 216 have a common source for all waters (Fig. 4(a), i.e. the trachy-andesitic rock, see section 217 4.2.1), nevertheless, the data of AC and AT2 spring waters sometimes scatter along the 218 general mixing trends between the meteoric and hydrothermal end-member (Fig. 4). The 219 AT2 springs have significantly lower discharge temperatures compared to AC. Lower 220 temperatures result in lower relative K contents (Fig. 4(b)). El Chichón thermal waters are 221 less "immature" than conventional solute geochemistry suggests (Giggenbach, 1988; 222 Henley et al., 1984); a partial equilibrium between waters and an Al-silicate alteration 223 mineral assemblage is attained, and the equilibrium temperatures coincide well with the 224 chalcedony equilibrium temperature (Taran *et al.*, 2008). The AT2 springs are fed by a 225 colder water source than the neighbouring AC springs, the chalcedony geothermometers 226 result in (Henley et al., 1984): 100-130°C for AT2 springs against 160-180°C for AC springs. The latter temperatures are even higher than for SP (~115°C) and lake waters 227 228 (115-130°C). AT2 waters are enriched in Ca and are less rich in K, with respect to AC 229 waters (Fig. 4(b)-(c)), so AT2 waters cannot be considered just diluted AC waters, but 230 indicate re-equilibration processes to lower temperatures of a same source as AC. 231 We show evidence on a contribution of the evaporite basement for the thermal waters at

El Chichón (Fig. 4(d)). The AT2 and AC springs seem to be affected by a non-volcanic 232

233 halogen source. The AC and AT2 waters plot towards the line of Cl/Br~289, the ratio 234 representative for "seawater" (≈evaporite) (Böhlke and Irwin, 1992). The January 2007 crater lake water plots more towards the "volcanic Cl/Br line" (Cl/Br >1,450; Taran et al., 235 1995; Fig. 4(d)). During this period, the SP geyser had returned to its high water 236 discharge. The March 2007 SP and lake water plot along the same mixing trend, 237 238 confirming the strong influence of SP waters on the crater lake chemistry. This mixing trend has a Cl/Br concentration ratio of ~500, meaning that Cl and Br at El Chichón in 239 240 any way can be considered a mixture of "seawater" and "volcanic" originating fluids (Fig. 241 4(d)). In any case, all observed Cl/Br ratios for El Chichón fluids are common in volcanic 242 waters produced in the magmatic-hydrothermal environment.

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4.1.3 The crater lake-Soap Pool system in 2006-2007

Within the scope of continuing the monitoring of the current activity of the crater lake-Soap Pool volcano-hydrothermal system, we present the updates for 2007, on: (1) the crater lake volume (Table 2, Fig. 5), and (2) the Cl content in the crater lake and SP spring waters (Table 1, Fig. 5).

249 Since a large lake has been observed for the first time in January-April 2001, and a 250 second time in March-April 2004, the lake has entered a large volume situation for the 251 third time by March 2007. Since January 2007, water has been discharging from the SP 252 site (no sample), and the lake volume has consequently been increasing (January 2007, 253 pers. comm. Protección Civil Chapultenango staff). In March 2007, El Chichón crater lake reached its largest volume ever observed ($\sim 6x10^5$ m³, Table 2). The SP discharge 254 was estimated in the field, resulting in ~ 10 kg/s. Assuming we did not miss any 255 256 significant volume changes of the lake with our observation frequency, for the first time, 257 three real trends can be observed in the alternation between a small and a large volume lake: (1) a large volume lake occurs each three years (2001, 2004 and 2007) starting a few 258

259 months (~January) after the rainy season (June-October), (2) the maximum volume 260 slightly increased from 2001 to 2007 (white dotted arrow in Fig. 5) and (3) intermediate 261 volume lakes were observed in 2002 and 2006, probably coinciding with shorter periods 262 of SP water discharges (black arrows, Fig. 5).

The Cl content in the lake increased from 135 mg/l in November 2006 to 1,256 mg/l in 263 September 2007. In March 2007 the Cl content was 1,149 mg/l. In November 2006 and 264 September 2007 the SP geyser emitted only vapour, while in March 2007 SP water 265 266 discharge was high (~10 kg/s). This sequence probably indicates that in the summer of 2007, the Cl content in the lake, and probably the lake volume, was even higher than 267 268 detected in March 2007, and that the relatively high Cl content in September 2007 is the remnant Cl shortly after the ceasing of SP water discharge towards the lake. Because the 269 estimated seepage flux (>17 kgm⁻¹d⁻¹; Rouwet *et al.*, 2004) is higher than the yearly 270 average precipitation flux (<12 kgm⁻²d⁻¹; Rouwet et al., 2004), the lack of SP water 271 272 feeding the lake results in a rapid decrease in lake volume, although no decrease in Cl 273 concentration. Lake water residence times were calculated to be extremely short (~2 274 months; Taran and Rouwet, 2007). The El Chichón crater lake is probably the best 275 example in the world to demonstrate the delicate balance between input and output of 276 heat, water and chemical species, to actually make a crater lake exist.

Rouwet *et al.* (2008) deduced an empirical equation to quantify the linear trend of
diminishing Cl content in the SP waters for the period 1983-2005:

 $Cl_L = -79.221x t + 25,628$ (1)

280 Where Cl_L is the Cl content in the SP water and t the number of months since January 281 1983 (t = 1). Applying this equation, the SP springs and crater lake should theoretically 282 be Cl-free by 2009±1; instead, for March 2007 a Cl content in SP waters of ~1,600 mg/l 283 should be reached. Despite this trend, the Cl content in the March 2007 SP waters is as 284 high as 3,028 mg/l. For the first time since 1995, a non-linear though diminishing Cl trend 285 for SP water is shown (Fig. 5). These higher than expected Cl contents in SP waters could 286 imply the presence (or appearance) of an additional Cl-source feeding the shallow boiling aquifer beneath the El Chichón crater. At present it is difficult to rule out the origin of this 287 288 Cl-enrichment, though different scenarios can be proposed. The first probable source of 289 Cl-rich waters feeding the shallow aquifer is the crater lake itself, due to an efficient 290 recycling of lake water through infiltration at the lake bottom. Consecutive boiling in the 291 shallow aquifer, enriches the liquid phase (SP geyser water) in Cl (Ohba et al., 2000). If 292 so, steam separation has become more efficient recently. Secondly, the SP geyser and the 293 shallow aquifer can be fed by the main hydrothermal aquifer, probably the same as the 294 one feeding the flank thermal springs (Cl ~2,000 mg/l), through upward fluid migration. Until now, a linear dilution of the remains of the 1982 ultra-acidic brine-like 295 296 hydrothermal fluid (24,000 mg/l of Cl, Casadevall et al., 1984; Rouwet et al., 2004) could 297 clearly be observed. In the near future (2009±1) it will become clear if all original Cl will 298 be flushed out off the crater hydrothermal system (zero-Cl in SP and lake water) or if Cl 299 content in the SP and lake water remains constant. If not, the regime of fluid flow inside 300 the volcanic edifice should be more complex, and existing models should be revised.

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- 4.2 Trace element geochemistry

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4.2.1 Relative mobility of trace metals

The trace element compositions of thermal waters at El Chichón are presented in Table 3. Figure 6 is a scatter plot of the concentration of a large amount of metals in the crater lake and AC and AT2 spring waters, with respect to the 1982 trachy-andesitic rock (Luhr *et al.*, 1984). Besides the major rock forming elements (thereafter RFEs), Cl and S are plotted as well, because both can originate from the basement rock. The 1982 deposits are renowned for their exceptionally high content of microphenocrystic anhydrite (2.6 wt% as SO₃, Varekamp *et al.*, 1984). Thus, S can be an abundant leaching product of the 311 1982 deposits, while Cl can enter thermal waters by water-basement interaction 312 (evaporites). The correlation between these two variables is generally rather good in all 313 graphs (Fig. 6(a)-(d)), stating that the rock composition plays a major role in the 314 distribution of RFE in thermal waters. Considering the main RFE, the alkalis and alkaline 315 earths (Na, K, Ca, Mg, Sr, Rb), it can be noticed that these have a strong tendency to 316 concentrate in solution. The water-rock ratios, marked by full diagonal lines in Fig. 6, demonstrate that 100 to 1,000 l of water is needed to leach out 1 kg of the main RFEs 317 318 from the deposits. Especially Ca and S are more concentrated in the thermal springs, 319 supporting the hypothesis of anhydrite leaching from the 1982 deposits. Also Sr, Mg, and 320 Rb are slightly more mobile in the spring waters with respect to the lake water. This can be due to the high state of alteration of sediments at the lake bottom by the acidic crater 321 322 lake water: the most mobile major elements are already exhausted in the lake sediments. 323 Additionally, Sr, Mg and Rb are also more abundant in carbonate rocks, present in the 324 basement. Sulphur, and especially Cl, show lower water-rock ratios (100 to <10) than the 325 main RFEs, indicating that the rock (trachy-andesite and its available anhydrite) is 326 obviously not the only source. Chlorine and sulphur might enter by magmatic degassing, 327 or even magmas and its resulted eruptive products are initially influenced by the Cl-SO₄rich basement rocks. All elements but S are more mobile in the March 2007 crater lake 328 329 with respect to November 2006. This indicates that besides direct leaching by crater lake 330 water of lake sediments, the SP springs are an additional source of metals: in November 331 2006 the crater lake was not fed by SP waters. Aluminium behaves differently if 332 comparing the crater lake waters (Fig. 6(a)-(b)) with the AC (Fig. 6(c)) and AT2 waters 333 (Fig. 6(d)). Aluminium generally has a strong tendency to concentrate in weathering 334 minerals such as oxides and clays (e.g. Aiuppa et al., 2000). Under acidic conditions clay 335 minerals are not stable, thus Al remains in solution (crater lake with pH < 2.7). In the less

acidic spring waters (pH 5.15-5.85) Al is lost by secondary mineral precipitation (Fig.
6(c)-(d)).

Generally, As, Zn, and Cu enter as highly volatile compounds in high-temperature 338 339 magmatic gases (Symonds et al., 1987; Taran et al., 1995). No strong enrichment in 340 waters of any of these metals can be detected, not surprisingly indicating the absence of a near-surface high temperature degassing magma batch. Comparing these metal 341 342 abundances in the lake water with respect to the spring waters, the crater lake water is 343 enriched by one order of magnitude with respect to the springs, probably due to the higher 344 acidity enhanced leaching capacity. Cu and Zn in AC waters are less mobile than in AT2 345 waters by two orders of magnitude of water-rock ratio (Fig. 6(c)-(d)).

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4.2.2 log-log plots and crater lake seepage

We already detected some differences in the major element chemistry between the AT2 and AC springs. Nevertheless, we would like to rule out whether AT2 is fed by an independent aquifer near the SW dome, or if it is part of a large S-SW deep geothermal aquifer. If plotting the trace element content of the November 2006 AT2 waters (46.5°C) against this of AC, the good correlation ($\mathbb{R}^2 \sim 0.95$) infers that the AT2 springs are fed by the same large aquifer as AC (Fig. 7).

354 Going beyond the application proposed by Taran et al. (2008), the log-log plots could be 355 used to detect possible lake seepage of crater lake water towards the deep geothermal aquifer, if plotting data sets of earlier samples of the crater lake against more recent data 356 sets for spring waters. An extremely good correlation exists for the November 2006 AC 357 and June 2004 crater lake waters ($R^2 \sim 0.98$; Fig. 8(a)). Correlation coefficients for later 358 359 dates (lake in March 2005 and November 2006) are near 0.90. If crater lake seepage 360 towards the flank springs takes place, the changes in chemistry in the lake water thus 361 seem to be transmitted into the deep geothermal aquifer. With these observations, it

362	would take \sim 30 months -the period between June 2004 and November 2006- for the lake
363	water to reach the deep geothermal aquifer and come out through the AC springs. The
364	same trend can be noticed for the AT2 spring (46.5°C, November 2006), although
365	correlation coefficients are lower (Fig. 8(b)). At Poás volcano a tritium-based residence
366	time of seeping crater lake water of 3 to 17 years was deduced (Rowe et al., 1995). Our
367	estimate of 30 months seems reasonable considering that the Poás fluids need to travel a
368	larger horizontal distance (3.25 km vs 1.5 km at El Chichón). The only tritium value for
369	the Agua Caliente spring at El Chichón (2.4 T.U. in 1998) corresponds well with the
370	values for Central America in the 1970's (2.6 T.U., as reported in Rowe et al., 1995),
371	suggesting that in 1998 the aquifer feeding the AC spring had an age of 20-25 years in
372	1998. Crater lake water in 1998 showed lower tritium isotopic values (1.3 T.U.)
373	corresponding with meteoric waters (1.1 T.U. for Agua Roja cold spring water).

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4.2.3 Chemical distribution of rock forming elements

Incongruent dissolution of volcanic rocks is the major, but not the only source of
RFEs in acidic volcano-hydrothermal systems and crater lakes. Fig. 9 shows the chemical
distribution of RFEs with respect to the 1982 trachy-andesitic rock (Luhr *et al.*, 1984,
Varekamp *et al.*, 1984), defined as follows:

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381	(RFE/Mg) _w /(RFE/Mg) _r	(2)
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With RFE each rock forming element (including S and Cl); subscript w stands for water, r for rock. We choose Mg to normalize because it is well-reserved in acidic sulphate-rich solutions (Giggenbach, 1974; Delmelle and Bernard, 1994; Pasternack and Varekamp, 1994; Takano *et al.*, 2004). The crater lake, AC and AT2 waters roughly follow the same element distribution (Fig. 9). Sulphur is relatively enriched with respect to the other 388 RFEs, confirming anhydrite leaching. The acidic crater lake water is more efficient in
389 leaching Cu, Co, and V from rocks or sediments.

A remarkable difference exists in the distribution of Cu and Zn between the AC and AT2 390 391 spring waters (Fig. 9): Cu and Zn are depleted in AC waters, with respect to AT2. At AC bubbling degassing takes place, accompanied by an H₂S smell, features absent at AT2. 392 393 The entrance of H_2S creates reduced environments adapted to sink Cu and Zn as sulfides. 394 The turquoise colour at the head of the AT2 spring might be due to a Cu enrichment. At 395 the AC, and more downstream at the AT2 stream, massive Fe-oxy-hydroxides form, 396 indicating strongly oxidized conditions, responsible for the Fe depletion. This effect is 397 stronger at AT2.

It is highlighted in Fig. 9 that As is depleted in the AC spring with respect to the crater lake water. Under acidic oxidized conditions (\approx crater lake) As is stable as H₃AsO_{4(a)}, and thus remains in solution. On the other hand, under near-neutral reduced conditions (\approx AC spring outlet) As will precipitate as As₃S₃ (Aiuppa *et al.*, 2000). Moreover, As coprecipitates with or adsorbs on Fe-oxy-hydroxides (Fig. 9) (Ballantyne and Moore, 1988; Aiuppa *et al.*, 2003).

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405 5. Conceptual models: present and future state of the El Chichón volcano406 hydrothermal system

407 5.1 The crater lake-Soap Pool system

For the first time since the observations started in 1995 (Taran *et al.*, 1998), a relationship with the annual rainfall distribution is noticed. In January-April 2001, March-June 2004 and March 2007 a large volume lake was observed. It is a fact that since 2001, every three years, a few months after the rainy season (June-October), the SP spring enters in a high water discharge activity, responsible for the lake growth and consecutive increase in Cl content in the lake water. Considering an average rainfall flux of 1.34x10⁻⁴ 414 kgm⁻²s⁻¹ (Rouwet *et al.*, 2004), the total volume of rainwater accumulated during a time 415 span of three years beneath the crater floor ($\sim 8x10^5 \text{ m}^2$) is in the order of 10^7 m^3 . This can 416 be considered the maximum volume of the shallow boiling aquifer beneath the 1982 417 crater floor, because part of the infiltrating rainwater dissipates in the volcanic edifice.

418 Geothermometry suggests boiling processes at shallow depth beneath the crater floor 419 (115-130°C). Fluid migration takes place inside the heterogeneous shallow aquifer, 420 testified by rumbling and low frequency noise. Such rumbling has been observed at other 421 active volcano-hydrothermal systems such as e.g. Kusatsu-Shirane volcano, the host of 422 the Yugama crater lake, where steam-driven explosive activity was common in the 1970's 423 and 80's (Ohba et al., 2008). Thus, these features at El Chichón can be interpreted as small steam-driven phreatic explosions inside the shallow aquifer. Considering that the 424 lake volume strongly depends on the SP water input, the slight increase of the maximum 425 426 lake volume observed through 2001, 2004 and 2007 (Fig. 5) could demonstrate an 427 increase in volume of the shallow aquifer feeding the SP geyser.

428

429 5.2 The deep geothermal aquifer

430 We found arguments in favour of a small degree of crater lake seepage into the S-SE deep geothermal aquifer (~200 m) reaching the AC and, in probably lower proportion, the AT2 431 432 springs. Crater lake seepage towards the S of the volcano might be enhanced by the 433 morphology of the southern rim of the Somma crater, opening into a big "barranca" ending into the AT2 canyon. The horse-shoe shaped explosion crater at AC could be 434 435 another morphological weakness, enhancing lateral and downward fluid flow. The seeped crater lake water is thought to reach the AC springs in ~30 months, a reasonable residence 436 437 time to migrate along a distance of ~1.5 km for a height difference of ~200 m. This effect 438 has been seen for the June 2004 crater lake infiltration, when lake volume was high (~ $5x10^5$ m³, Table 2). When the crater lake level is high, more permeable parts of the 439

440 crater floor are flooded than at low crater lake level: the constantly covered lake bottom is 441 composed of sealing clays, while near the coast of the lake, the lake bottom floor is composed of pumiceous permeable sediments (Fig. 11). High level lakes thus tend to 442 443 infiltrate preferentially (Fig. 11). If this mechanism is a fact, the March 2007 crater lake will have reached the AC springs by the autumn of 2009. Trace element contents in a 444 large volume lake are generally higher than in a small volume lake. A monitoring with 445 446 time of the RFE patterns between the lake and spring waters will contradict or affirm the 447 here proposed lake seepage. The Cl/SO₄ ratio in both lake and AC waters has shown to be 448 variable with time and tracing this parameter will be less efficient to detect possible lake 449 seepage. Moreover, estimates of important parameters of the physical properties (i.e. 450 hydraulic conductivity, active porosity, permeability, etc.) of the volcanic edifice still 451 lack, and are necessary to establish a hydrogeochemical modelling of the entire El 452 Chichón volcano-hydrothermal system. The AT2 springs, at the foot of the SW dome, are 453 fed by the same deep geothermal aquifer. This aquifer is formed as the result of the 454 permeability decrease at the contact between the volcanic edifice and the basement rocks. 455 The main heat source beneath El Chichón is probably long-lived and stable, as spring 456 temperatures have not changed since the late 1970's. The remains of the 1982 magma is 457 an additional heat source. No chemical evidence exists on the location of any heat source 458 or a high temperature degassing magma body. Moreover, the normal faults could serve as 459 upward migration pathways for deep Cl-rich fluids, frequently present at hydrothermal systems (e.g. Pinatubo volcano; Stimac et al., 2004). The Cl from the deep geothermal 460 461 aquifer probably originates partly from the same evaporites (~Cl/Br ratios). The current 462 uncertainty on an additional Cl input at the crater system still points to a possible masking 463 effect of the deep geothermal aquifer to absorb all rising (magmatic?) Cl before reaching 464 the crater area. That is the reason why we believe dome intrusion will probably be

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anticipated by chemical changes in the thermal springs, even before modifying crater lake dynamics and chemistry of gas emissions.

467

468 **6.** Conclusions

In this study, the results of the geochemical survey obtained during the period November 2006-October 2007 are integrated in the monitoring program of the crater lake-Soap Pool spring system at El Chichón volcano. Moreover, we propose new ideas for future volcanic surveillance of magmatic activity using the AT2 and AC thermal springs. New data on stable isotopes (δD and $\delta^{18}O$), major, minor and trace elements verify existing hydro-geochemical models (Taran and Rouwet, 2007; Rouwet *et al.*, 2008; Taran *et al.*, 2008).

476 In March 2007 the crater lake reached its largest volume ever observed ($\sim 6x10^5 \text{ m}^3$). For 477 the first time a trend between the lake volume and the annual rainfall distribution could be 478 noticed: during three years (2001, 2004 and 2007), high water discharges from the Soap 479 Pool spring result in a large lake, starting some months after the rainy season (June-480 October). The diminishing-Cl trend with time for Soap Pool and crater lake waters shows 481 a non-linear behaviour for the first time; current Cl contents are higher than expected, 482 though few evidence on a renewed Cl input exist.

483 Earlier suggestions on a strong basement-volcano interaction are confirmed. Cl has a nonpure volcanic origin for thermal spring waters. The AT2 springs testify an enhanced 484 485 water-rock interaction, anhydrite leaching, evaporite contribution, absence of degassing 486 and lower temperature conditions. Thus, the SW dome does not seem to be active. A 487 small portion of the deep geothermal aquifer seems to originate from the direct seepage of 488 crater lake water. Strongest evidence is found for the AC springs. We suggest further 489 research on the spring systems to ascertain the response of the thermal springs during 490 increased volcanic activity (e.g. future dome growth).

491

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Table 1

Chemical and isotopic composition of thermal waters at El Chichón volcano. Concentrations are expressed in mg/l, isotopic composition in ‰ vs V-SMOW, Conductivity in mS/cm, Eh in mV. nm: not measured.

Sample	Date	T(°C)	pН	Cond.	Eh	δD	δ ¹⁸ Ο	SiO ₂	Na	K	Mg	Ca	F	Cl	Br	SO ₄	HCO ₃
LE	19/11/2006	26	2.48	2.4	nm	-17	-2.5	108	79	14	8	45	-	135	-	327	-
LE	23/01/2007	nm	2.70	3.4	nm	-	-	-	267	47	17	134	-	600	0.80	381	-
LE	10/03/2007	29.2	2.57	5.0	525	-4	0.0	138	488	68	21	216	0.95	1149	2.40	332	-
LE	12/09/2007	nm	nm	nm	nm	-	-	-	610	-	34	20	0.30	1256	-	619	-
LE	19/10/2007	nm	nm	nm	nm	-	-	-	265	-	12	10	-	508	-	431	-
SP	10/03/2007	98	7.07	9.9	-319	-12	0.5	105	1401	215	64	365	0.00	3028	6.39	264	43
AC1	23/11/2006	70.6	5.77	5.2	9	-23	-3.7	251	708	93	49	462	0.57	1459	3.20	733	293
AC2	23/11/2006	58.8	5.85	4.3	-27	-21	-4.2	205	553	72	51	436	0.38	1141	3.20	624	271
AT2.1	27/11/2006	36.4	5.75	1.8	-131	-22	-4.7	83	137	15	17	469	1.14	292	0.80	984	52
AT2.2	27/11/2006	46.5	5.85	3.4	-67	-21	-4.4	133	334	30	27	427	1.14	803	1.60	738	122
AT2.TOP	27/11/2006	36	5.35	2.6	3	-19	-4.5	99	246	21	22	369	0.38	583	1.60	618	76
AT2.TOP	28/03/2007	35.4	5.15	4.2	-120	-21	-4.1	93	349	25	26	477	0.30	911	1.52	797	49
Agua Suerte	27/03/2007	22.5	6.28	nm	16	-18	-4.0	54	3	2	1	7	0.08	2	-	17	12
river water AC	23/11/2006	nm	nm	nm	nm	-24	-5.4	-	-	-	-	-	-	-	-	-	-
river water AC	23/11/2006	nm	nm	nm	nm	-22	-4.6	-	-	-	-	-	-	-	-	-	-
river water AT2	27/11/2006	nm	nm	nm	nm	-23	-5.0	-	-	-	-	-	-	-	-	-	-

The changes in crater lake surface area, depth and volume. The estimates were obtained applying the methods evolved by Rouwet *et al.* (2008). The lake depth was calculated by the equation: d = -0.913 + 0.0097.S^{1/2} (Rouwet *et al.*, 2004; 2008), the lake volume by the multiplication S.d. References: ¹Rouwet *et al.* (2008), *: direct measurement of the lake depth during bathymetric surveys.

Date	$S(x10^4 m^2)$	d (m)	$V(x10^4 m^3)$	Ref.
May-92	9.7	2.1	20.5	1
Jan-93	4.6	1.2	5.4	1
Mar-93	4.3	1.1	4.7	1
16/05/95	6.0	1.5	8.8	-
24/03/96	4.2	1.1	4.5	1
16/01/97	5.7	1.4	8.0	1
07/04/98	4.0	1.3*	5.1	1
20/11/98	4.8	2.1*	10.1	1
01/03/99	3.7	1.0	3.5	1
10/08/99	4.8	1.2	5.8	1
24/04/00	4.8	1.2	5.8	-
22/01/01	14.3	3.3*	47.3	1
19/04/01	14.8	2.8	41.9	1
05/07/01	10.4	2.2	23.0	1
15/01/02	4.6	1.2	5.4	1
29/04/02	8.2	2.4*	19.6	1
03/03/03	5.8	1.4	8.2	1
03/04/03	5.8	1.4	8.2	1
01/05/03	5.3	1.3	7.1	1
10/11/03	6.3	1.5	9.5	1
01/03/04	12.4	2.5*	31.1	1
10/04/04	15.4	2.9	44.7	1
15/05/04	17.1	3.1	52.7	1
25/06/04	15.9	3.0	47.2	1
31/10/04	9.2	2.0	18.7	-
10/03/05	6.2	1.5	9.2	1
04/06/05	3.9	1.0	3.9	1
18/06/05	4.3	1.1	4.7	1
12/09/05	4.6	1.2	5.4	1
06/12/05	6.1	1.5	9.0	1
05/03/06	5.7	1.4	8.0	-
19/11/06	3.9	1.0	3.9	-
09/03/07	18.0	3.2	57.6	-
25/03/07	18.1	3.2	58.2	-
12/09/07	8.5	1.9	16.3	-
19/10/07	9.1	2.0	18.3	-

 Table 3

 Trace element content in thermal and cold spring waters at El Chichón for the period 2006-2007. Concentrations are expressed in µg/l.

Sample	LE	LE	AC1	AC2	AT2.1	AT2.2	AT2.TOP	AT2.TOP	Agua Suerte
Date	19/11/06	10/03/07	23/11/06	23/11/06	27/11/06	27/11/06	27/11/06	28/03/07	27/03/07
Li	196	815	1505	1107	211	596	392	516	1
В	4421	16410	18370	14000	3521	8895	6728	8943	210
Al	3250	6167	10.1	5.0	374	7.4	28.1	116	17.6
V	11.8	6.5	0.90	0.99	0.59	1.3	4.8	3.5	3.6
Cr	0.34	< 0.1	0.16	0.20	0.07	0.53	0.08	< 0.1	< 0.1
Mn	1217	3654	2030	2223	3580	4754	3518	6224	4.6
Fe	6738	8931	5233	1223	202	5486	7.4	31.6	4.7
Со	0.95	16.7	1.4	1.9	6.9	8.1	1.9	6.3	< 0.1
Ni	0.41	6.2	1.0	4.2	14.4	4.6	6.9	12.0	< 0.1
Cu	5.9	163.0	0.82	5.1	61.4	12.8	60.3	119	0.68
Zn	11.3	162	3.7	5.9	116	41.9	68.5	120	0.61
As	59.8	36.8	4.5	2.6	0.81	2.1	1.3	1.1	0.41
Se	0.19	0.32	-	-	0.25	-	-	-	0.14
Rb	99	452	666	501	97.9	243	163	175	12.8
Sr	346	1828	4394	3602	3289	3907	3138	3536	83.5
Mo	0.64	0.23	2.6	2.0	0.38	1.5	0.56	0.33	0.41
Cd	< 0.1	1.7	< 0.1	< 0.1	0.49	< 0.1	0.40	0.84	< 0.1
Sb	0.38	0.23	< 0.1	0.17	< 0.1	< 0.1	0.15	< 0.1	< 0.1
Cs	24.6	96.1	236	158	26.9	99.0	57.9	63.3	0.3
Ba	44.6	159	88.2	92.0	34.2	45.9	75.0	98.7	6.0
Hg	< 0.05	0.08	0.06	< 0.05	< 0.05	< 0.05	< 0.05	0.06	0.07
Pb	< 0.1	< 0.1	0.24	0.18	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1
Th	-	< 0.05	-	-	-	-	-	< 0.05	< 0.05
U	-	0.13	-	-	-	-	-	0.55	< 0.05
Be	-	0.13	-	-	-	-	-	-	-

FIGURE CAPTIONS

6

7

Fig. 1

8 (a) Location map of El Chichón volcano. TR: Tehuantepec Ridge, MAT: Middle 9 America Trench. Triangles are volcanoes. Po: Popocatépetl, I: Iztaccíhuatl, Pi: Pico 10 de Orizaba, SM: San Martín. Black triangles indicate volcanoes of the Central 11 American Volcanic Arc, white triangles are Mexican volcanoes (Transmexican 12 Volcanic Belt, besides El Chichón and San Martín). Cities, DF: Distrito Federal, V: 13 Villahermosa, TG: Tuxtla Gutiérrez. (b) Location map of the hydrothermal 14 manifestations at El Chichón. Thermal manifestations, AC: Agua Caliente, AT1: Agua Tibia 1, AT2: Agua Tibia 2, AS: Agua Salada, SP: Soap Pool, L: crater lake. 15 16 17 Fig. 2 18 (a) Panoramic view of the 1982 crater from the NE rim, showing the largest lake 19 ever observed (27/03/2007). The top part of the SW dome peeps over the crater rim. 20 Soap Pool (SP) site is also indicated. (b) Panoramic view from the southern Somma 21 crater rim, showing the SW dome and the canyon where AT2 springs discharge. To 22 the right of the picture, the outside rim of the 1982 crater is shown. (c) The AC 23 thermal springs and Río Tuspac seen from the Somma crater rim, towards the SE 24 (27/11/2006). (d) Sampling at the main AC thermal spring (23/11/2006). (e) The 25 AT2 thermal spring (27/11/2006) discharging at the foot of the SW dome. (Pictures 26 by M. Jutzeler). 27 28 Fig. 3.

29 δ^{18} O vs δ D plot for El Chichón thermal spring, cold spring, crater lake and Soap 30 Pool waters for the period November 2006-October 2007. MWL: Meteoric water

31	line, M: meteoric water at El Chichón. SP: range for Soap Pool spring waters taken
32	from Rouwet <i>et al.</i> (2008).
33	
34	Fig. 4.
35	Mixing plots for El Chichón thermal waters for the period November 2006-October
36	2007. (a) Cl vs Na, (b) Cl vs K, (c) Cl vs Ca, (d) Cl vs Br.
37	
38	Fig. 5.
39	Changes with time for the Cl content in the crater lake and Soap Pool waters (right-
40	hand scale), and the crater lake volume (left-hand scale). Pre-November 2006 data
41	are compiled from Rouwet et al. (2008).
42	
43	Fig. 6.
44	Ion concentrations in thermal waters at El Chichón vs the average concentrations in
45	the 1982 trachy-andesitic rocks (taken from Luhr et al., 1984; and Varekamp et al.,
46	1984), for (a) the November 2006 crater lake, (b) the March 2007 crater lake, (c)
47	the November 2006 Agua Caliente springs, and (d) the November 2006 Agua Tibia
48	2 springs.
49	
50	Fig. 7.
51	Log-log scatter plots: metal concentrations in AT2 (November 2006, T= 46.5° C)
52	thermal spring water vs metal concentrations in November 2006 Agua Caliente
53	spring water. R ² represents the correlation coefficient.
54	
55	
56	

57 Fig. 8.

58	Log-log scatter plots for ion contents in thermal waters, showing changes of R^2 with
59	time. (a) November 2006 Agua Caliente spring waters vs the June 2004, March
60	2005 and November 2006 crater lake waters, (b) November 2006 (T= 46.5° C) AT2
61	spring waters vs the June 2004, March 2005 and November 2006 crater lake waters.
62	Pre-2006 data are taken from Rouwet et al. (2008).
63	
64	Fig. 9.
65	Relative abundances of elements normalized against Mg in the crater lake waters
66	(November 2006 and March 2007), AC (November 2006), AT2 (March 2007) and
67	that in the average 1982 trachy-andesitic rocks (taken from Luhr et al., 1984; and
68	Varekamp et al., 1984).
69	
70	Fig. 10.
70 71	Fig. 10. Current dynamics of the "crater lake-Soap Pool spring" system. Phreatic explosions
70 71 72	Fig. 10.Current dynamics of the "crater lake-Soap Pool spring" system. Phreatic explosions (stars) inside the shallow boiling aquifer breach through secondary sealing minerals,
70 71 72 73	Fig. 10.Current dynamics of the "crater lake-Soap Pool spring" system. Phreatic explosions (stars) inside the shallow boiling aquifer breach through secondary sealing minerals, increasing on its turn the aquifer volume. The boiling aquifer is continuously filled
 70 71 72 73 74 	Fig. 10. Current dynamics of the "crater lake-Soap Pool spring" system. Phreatic explosions (stars) inside the shallow boiling aquifer breach through secondary sealing minerals, increasing on its turn the aquifer volume. The boiling aquifer is continuously filled by infiltrating rain and crater lake water.
 70 71 72 73 74 75 	Fig. 10. Current dynamics of the "crater lake-Soap Pool spring" system. Phreatic explosions (stars) inside the shallow boiling aquifer breach through secondary sealing minerals, increasing on its turn the aquifer volume. The boiling aquifer is continuously filled by infiltrating rain and crater lake water.
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 70 71 72 73 74 75 76 77 	 Fig. 10. Current dynamics of the "crater lake-Soap Pool spring" system. Phreatic explosions (stars) inside the shallow boiling aquifer breach through secondary sealing minerals, increasing on its turn the aquifer volume. The boiling aquifer is continuously filled by infiltrating rain and crater lake water. Fig. 11. Model to explain enhanced crater lake seepage during high volume lake situations
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Figure 1. Rouwet et al.





Fig. 3. Rouwet et al.



Fig. 4. Rouwet et al.



- SP Cl content
- Crater lake volume

Fig. 5. Rouwet et al.



Fig. 6. Rouwet et al.



Fig. 7. Rouwet et al.



Fig. 8. Rouwet et al.



- -•- Crater lake March 2007

Fig. 9. Rouwet et al.



Fig. 10. Rouwet et al.



Fig. 11. Rouwet et al.