# The effect of P<sub>CO2</sub> on the high temperature behavior of dolomite: insights from deformation experiments

**Claudio Delle Piane, Luigi Burlini. Geological Institute ETH Zurich Switzerland Bernard Grobety. Department of Geosciences, University of Fribourg, Switzerland** 

200 um



## <u>Goals</u>

Infer the influence of decarbonation reaction on:

- mechanical response of deforming rocks
- fluid transport properties
- micro-seismicity

## <u>Methods</u>

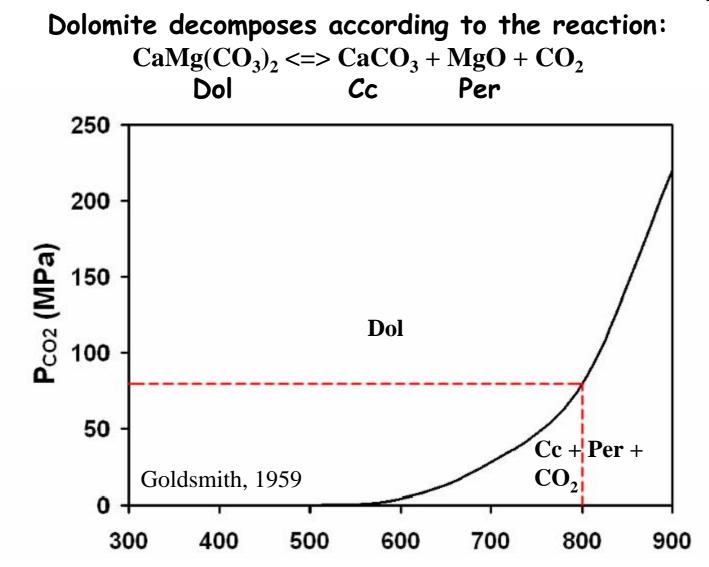
- Large strain experiments on a reacting system
- Microstructural characterization of the deformed samples
- $\boldsymbol{\cdot}$  Acoustic emission measurment during static HT/HP experiments

# Dolomite: starting material **250** μm

X-Ray Fluorescence and single-point microprobe analyses show that the dolomitic marble is very pure with almost stochiometric composition



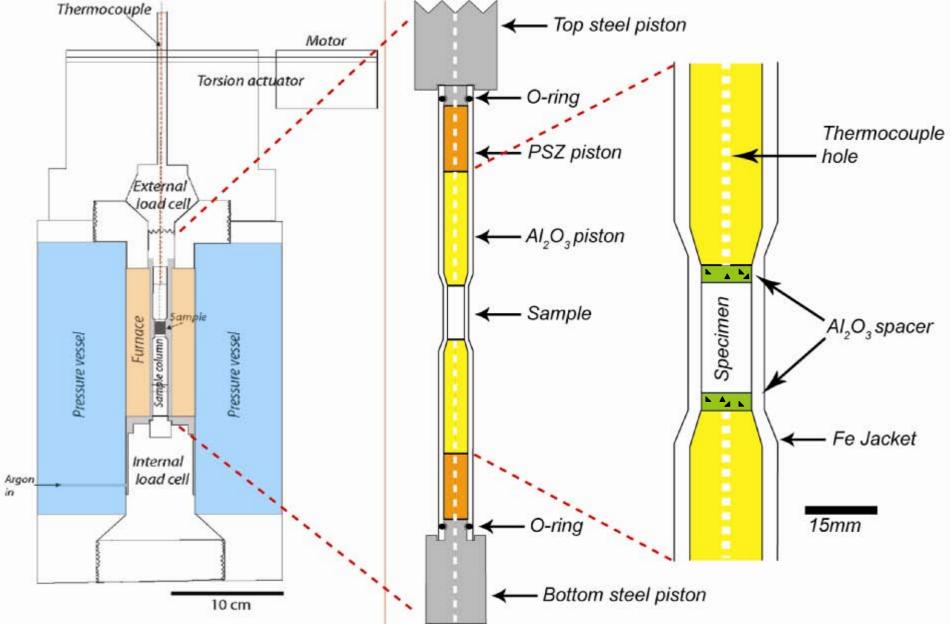
#### Dolomite equilibrium

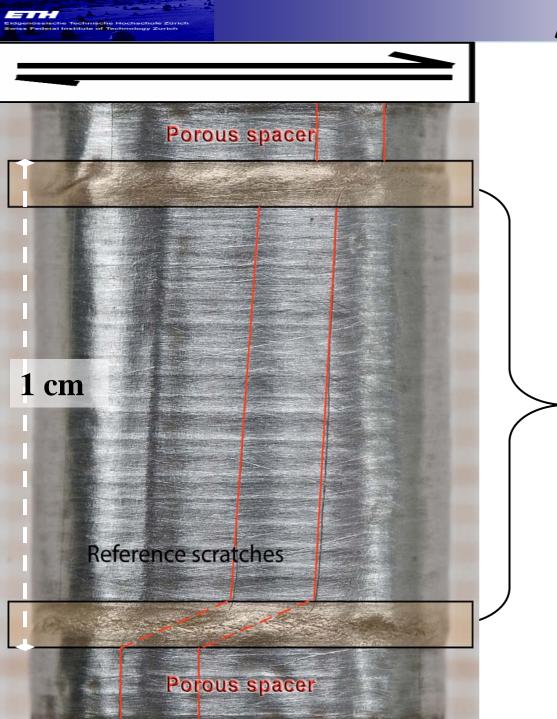


The kinetic of the reaction is controlled by the partial pressure of  $CO_2$ 



#### Experimental set-up





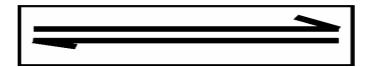
Dolomite: mechanical behavior

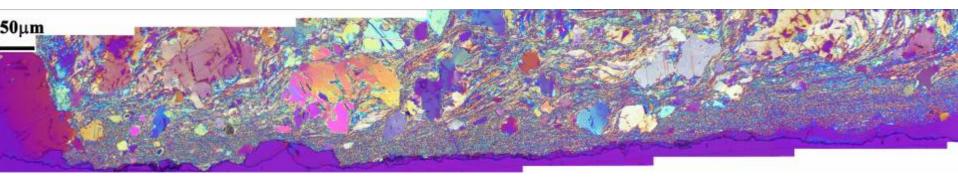
After the experiments deformation is higly nonhogeneous along the sample

 Zones of localized deformation

#### **Dolomite:** deformation microstructure

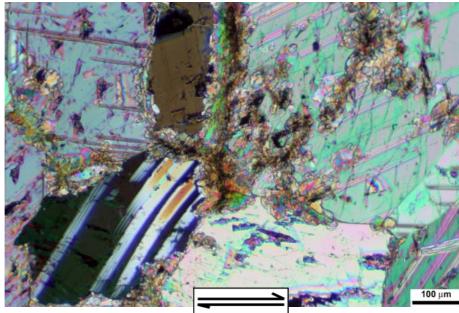


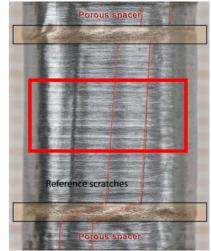


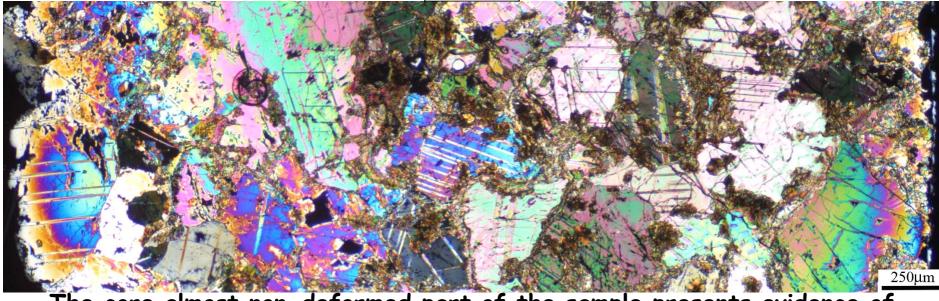


Localised shear bands occur in zones of grain size refinement

#### Dolomite: deformation microstructure

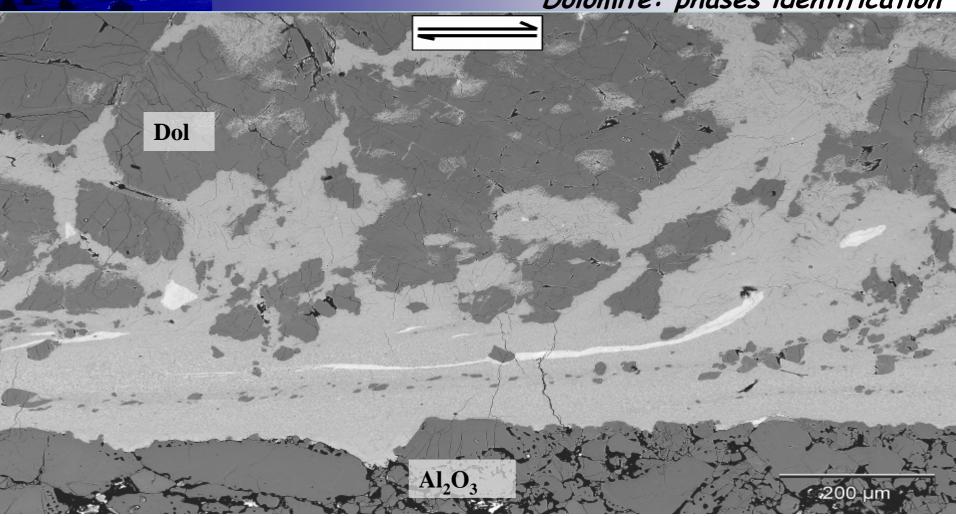


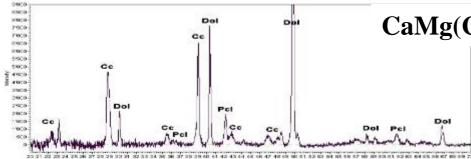




The core almost non-deformed part of the sample presents evidence of intracrtysatalline deformation and microcracks





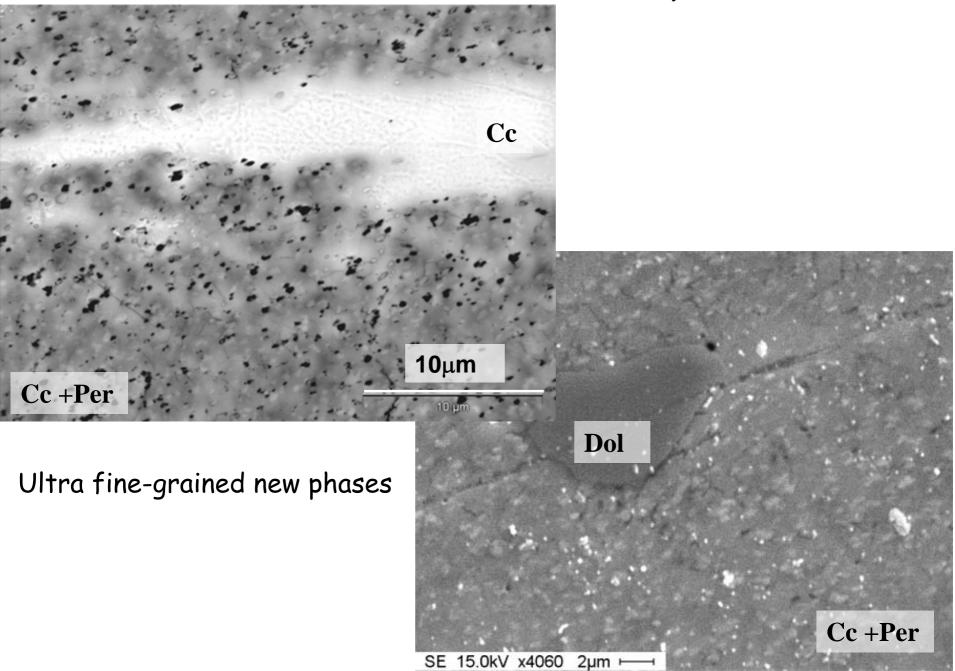


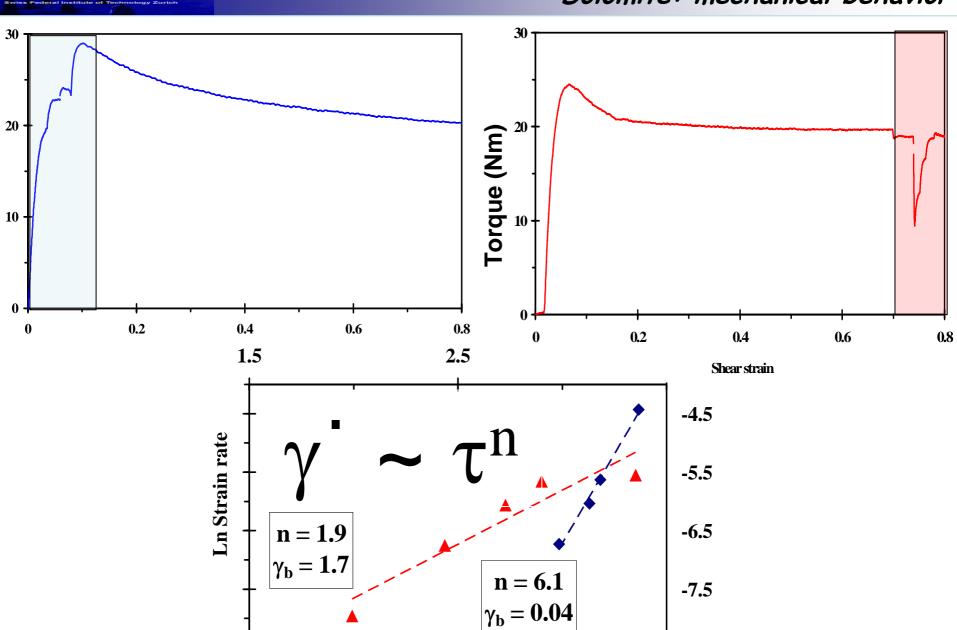
#### $CaMg(CO_3)_2 \ll CaCO_3 + MgO + CO_2$

XRD pattern indicating presence of calcite, periclase and dolomite



#### Dolomite: phases identification



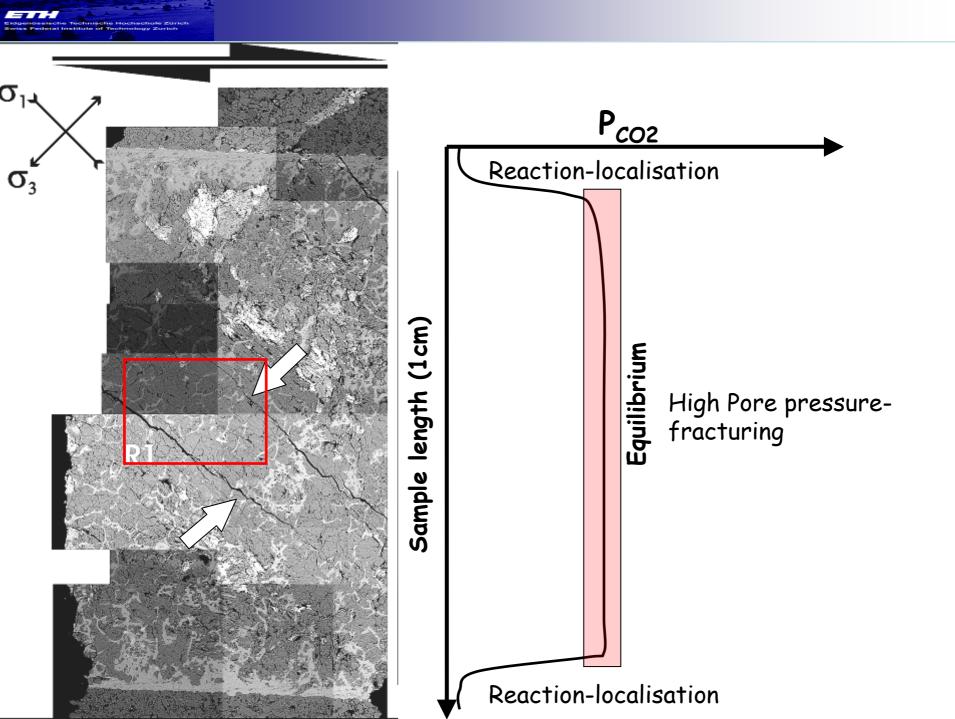


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#### Ln Torque

#### Dolomite: mechanical behavior

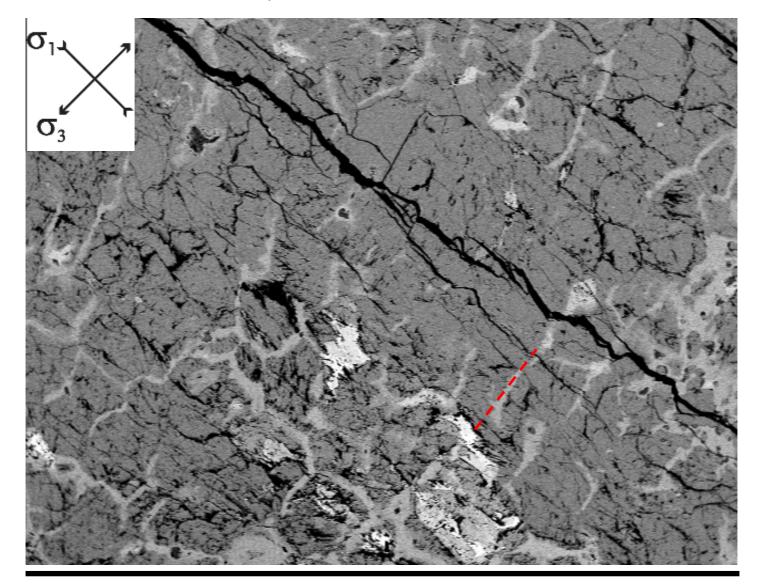
-8.5





Reaction induced compaction bands

Solid volume decreases by ca.25% due to the decarbonation reaction

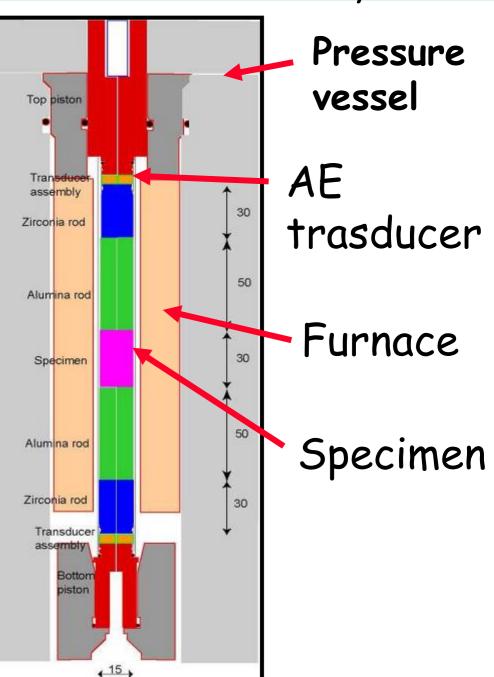


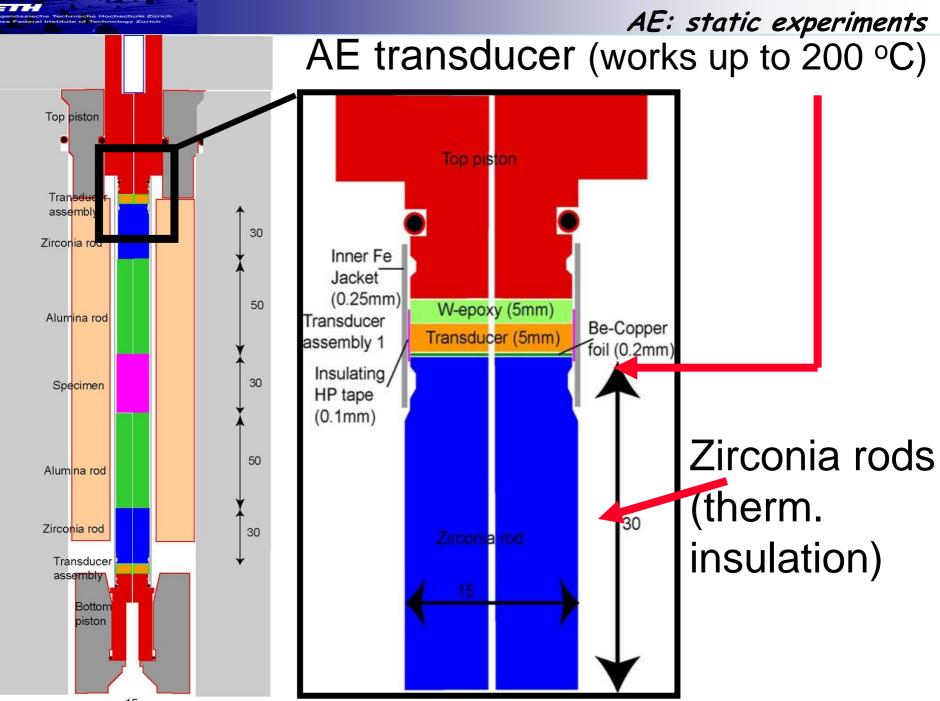
**250µm** 

#### AE: static experiments

# Sample assembly /

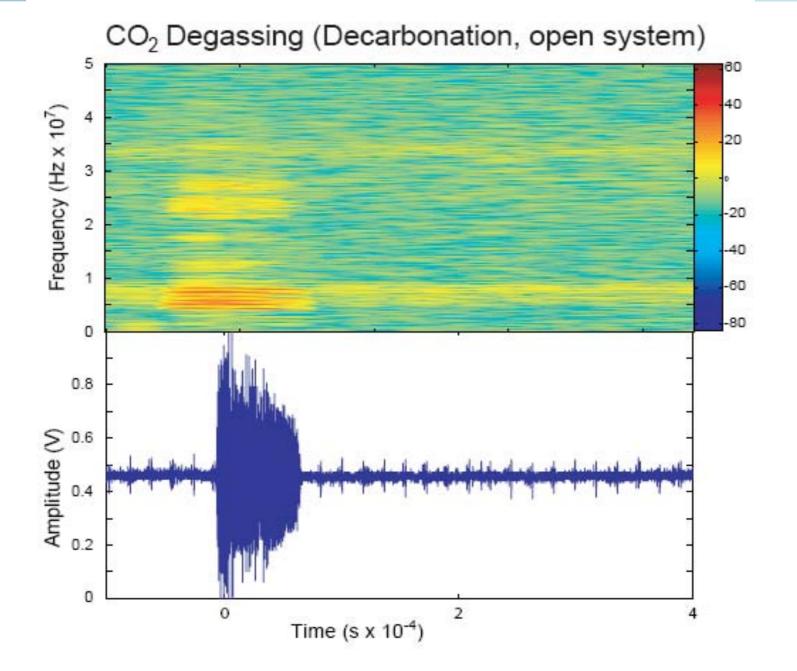


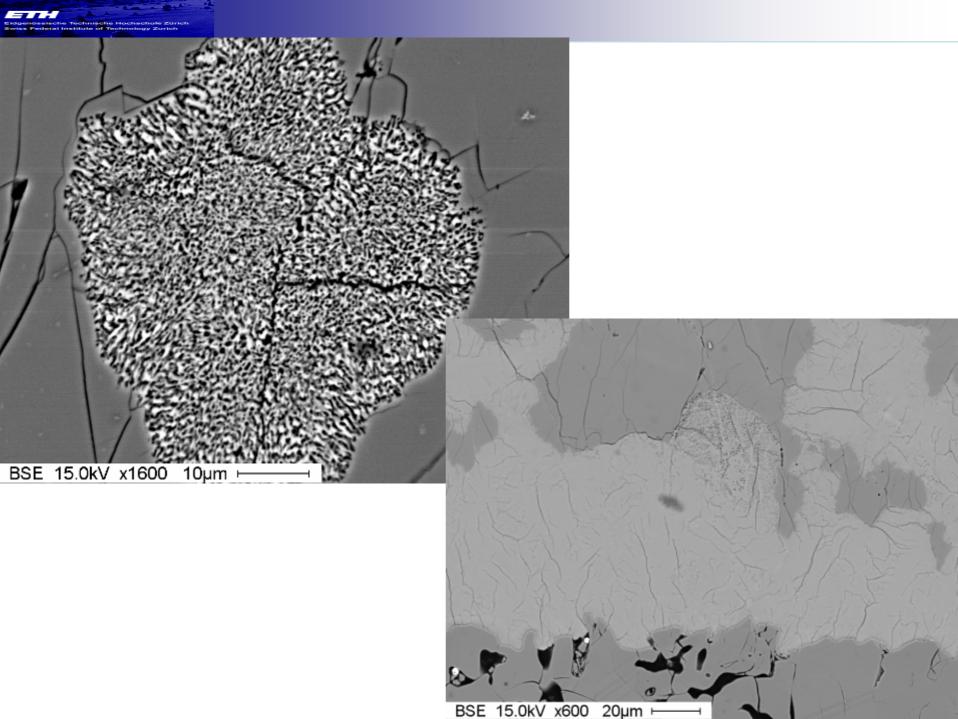




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#### Decarbonation 850°C







- Natural dolomitic marble was deformed drained conditions during decarbonation reaction. The reaction progress is controlled by the CO<sub>2</sub> pressure gradient along the sample
- Sub-micron size products are found at the interfaces between the sample and the porous Al<sub>2</sub>O<sub>3</sub> spacer
- Strong strain partitioning occurs between the reacted and non-reacted portions of the deformed samples.
- Stress exponent evolves with strain towards values close to 1, suggessting an increasing contribution of grain size sensitive mechanisms in accommodating deformation as reaction progresses.
- Localisation is driven by weakening due to grain size refinement associated to the decarbonation reaction
- Solid volume reduction is responsible for compaction bands formation
- AE associated with pore collapse and fluid migrationin under hydrostatic conditions