Carbonate system dynamics in the Gulf of Trieste (North Adriatic sea)

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<u>Why studying carbonate system in the</u> <u>Gulf of Trieste ?</u>

EC. N

Bottom cyclonic circulation
Shallow (max depth 25 m)
Strong cold winds (Bora)
Cold water in winter (7-8°C)

Potential for high CO₂ absorption and acidification

Dense water formation

Basin scale influence





River loads

 High susceptibility to meteorological forcings

 Strong seasonal variability in biogeochemical parameters

How do they influence carbonate system and pH?

Are they buffering or do they accelerate acidification?

PALOMA Dynamic Pilon

Approach

PALOMA station

Start Jan 2008 - monthly

Temperature, salinity, fluorescence, DO.
Dissolve Oxygen (winkler)
Nutrients: NO₃, NH₄, NO₂, PO₄, SiO₂

-pH: spectrophotometric det. ± 0.003 (Byrne & Breland 1989, DOE 2007)

Start Aug 2008 - monthly

•Alkalinity: potentiometric titration ± 3 µmol/kg_{sw} (adapted Hernandez-Ayon 1999)

AT / S correlation used to estrapolate AT values from Jan to July 2008

 pH_T situ, pCO₂, other parameters: calculated with "CO₂sys" (Lewis&Wallace 98)





Meteorological data: •Air temperature •Solar radiation •Wind speed and direction •Precipitations

Sea water temperature













<u>Comparison with other</u> <u>Mediterranean coastal areas</u>

Bay of Angels

Gulf of Trieste



Comparison of Surface data, Monthly 2 years averages

Borges, Schiettecatte, Abril, Delille, Gazeau, 2006, Est.Coats. Shelf Sci. Carbon dioxide in European coastal waters



pCO₂ @ 15°C

More pronounced seasonal cycle •Higher winter values \Leftrightarrow lower SST •Decrease in summer, increase autumn \Leftrightarrow trophic status pCO_2 situ Inversion of seasonal cycle in both Med. sites \Leftrightarrow effect of high seas. Δ T on H₂CO₃ equilibria

 $H_2CO_3 \Leftrightarrow H^+ + HCO_3^- \Leftrightarrow 2 H^+ + CO_3^=$



Future scenarios and in situ variabilty

Selected months were atmospheric CO₂ solubilization was the main driver of sea pCO₂ : Jan, Feb, March 2008-09; 0-14 m depth
 Same S, T, Alkalinity, nutrients concentrations
 Same pCO₂ atm – pCO₂ sea surface difference

[CO₂]atm 385 ppm Today [CO₂]atm 750 ppm By 100 years [IPCC, Business as Usual scen.]

pH_T in situ





Some first answers

River loads

 High susceptibility to meteorological forcings

 Strong seasonal variability in biogeochemical parameters In winter CO_2 undersaturation \longrightarrow High potential for CO_2 absorption and acidification

Spring riverine loads trigger phytoplankton blooms but they loads more acidic waters that mask a potential pH increase due to biological CO₂ uptake

How do they influence carbonate system and pH? Are they buffering or do they accelerate acidification? Strong variations in carbonate system can be expected by the end of the century in cold winter waters.

But.... High alkalinity values keep the Gulf of Trieste saturated for Ca and Ar even 750 ppm scenario

And ... what will happen in summer?

Thank you for your attention !

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