

Phage ϕ 29 Protein p6 Is in a Monomer–Dimer Equilibrium That Shifts to Higher Association States at the Millimolar Concentrations Found *in Vivo*[†]

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ABSTRACT: Protein p6 from *Bacillus subtilis* phage ϕ 29 ($M_r = 11\,800$) binds *in vitro* to DNA forming a large nucleoprotein complex in which the DNA wraps a multimeric protein core. The high intracellular abundance of protein p6 together with its ability to bind the whole ϕ 29 DNA *in vitro* strongly suggests that it plays a role in viral genome organization. We have determined by sedimentation equilibrium analysis that protein p6 (1–100 μ M range), in the absence of DNA, is in a monomer–dimer equilibrium, with an association constant (K_2) of $\sim 2 \times 10^5 \text{ M}^{-1}$. The intracellular concentration of protein p6 ($\sim 1 \text{ mM}$) was estimated measuring the number of copies per cell (7×10^5) and the cell volume ($1 \times 10^{-15} \text{ L}$). At concentrations around 1 mM, protein p6 associates into oligomers. This self-association behavior is compatible with a dimer–hexamer model ($K_{2,6} = 3.2 \times 10^8 \text{ M}^{-2}$) or with an isodesmic association of the dimer ($K = 950 \text{ M}^{-1}$), because the apparent weight-average molecular mass ($M_{w,a}$) does not reach saturation at the highest protein concentrations. The sedimentation coefficients of protein p6 monomer and dimer were 1.4 and 2.0, respectively, compatible with translational frictional ratios (f/f_0) of 1.15 and 1.30, which slightly deviate from the hydrodynamics of a rigid globular protein. Taking together these results and considering the structure of the nucleoprotein complex, we speculate that the observed oligomers of protein p6 could mimic a scaffold on which DNA folds to form the nucleoprotein complex *in vivo*.

The genomes of prokaryotic organisms are organized in higher order nucleoprotein complexes that, besides a packaging role, mediate fundamental processes of DNA such as replication, transcription, recombination, and transposition (Echols, 1990). Some of these complexes are assembled by architectural elements that by analogy to eukaryotic systems are referred to as histone-like proteins. In *Escherichia coli*, the major constituents of bacterial nucleoid are the HU heterodimer [reviewed in Drlica and Rouviere-Yaniv (1987)] and H-NS [reviewed in Ussery et al. (1994) and Atlung and Ingmer (1997)]. These proteins are small, very abundant, and bind DNA with little or no sequence specificity, usually through the minor groove. The pleiotropic nature of mutations in their corresponding genes indicates that these proteins have multiple functions. HU is reported to be involved in replication, transposition, and transcriptional control, while H-NS has been suggested to be a modulator of regulated gene expression.

Protein p6 from *Bacillus subtilis* phage ϕ 29 has some features resembling those expected for a histone-like protein. Protein p6 is the most abundant protein in ϕ 29-infected cells and binds *in vitro* to the viral DNA forming multiple complexes spread virtually throughout the entire genome,

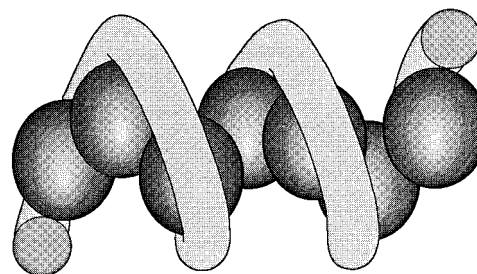


FIGURE 1: Model of protein p6–DNA complex. The path of the DNA in the complex is described in Serrano et al. (1993): one superhelical turn has 63 bp and every protein p6 dimer, represented by ellipsoids, is bound to 24 bp. The DNA in the complex is strongly bent (66° every 12 bp), undertwisted (11.5 bp per turn) and highly compacted (4.2-fold).

of sizes ranging from ~ 100 base pairs (bp) up to ~ 2 kilobases (kb) (Gutiérrez et al., 1994). Protein p6 binds to DNA through the minor groove, and it does not recognize a specific sequence, but rather a DNA structural feature such as bendability (Serrano et al., 1989). In the nucleoprotein complex, the DNA adopts a right-handed toroidal conformation winding a multimeric protein core (Serrano et al., 1993) (see Figure 1). Therefore, complexed DNA has a high degree of compaction and it is strongly distorted. The formation of nucleoprotein complexes at the origins of replication activates the initiation of ϕ 29 DNA replication *in vitro* (Serrano et al., 1989), and indeed protein p6 is absolutely required for viral DNA synthesis *in vivo* (Carcascosa et al., 1976). In addition, protein p6 also represses transcription from ϕ 29 early promoter C2 both *in vivo* and *in vitro* (Whiteley et al., 1986; Barthelemy et al., 1989).

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Footprinting studies of the protein p6–DNA complex showed a repeating binding pattern of protein dimers (Serrano et al., 1990), and in fact, protein p6 formed dimers in solution (Pastrana et al., 1985). Fluorescence studies have shown that protein p6 binding to DNA is highly cooperative (A. M. Abril, unpublished results); thus, presumably, the complex is propagated by dimer–dimer interaction. In the present study, we have characterized the state of association of protein p6 by sedimentation equilibrium, as a function of temperature, ionic strength, and protein concentration. We have also addressed the question whether the *in vivo* amount of protein p6 suffices to complex all the DNA of the viral progeny by quantification of the intracellular amounts of protein p6 and Ø29 DNA, and studied the self-association behavior of protein p6 at concentrations approaching the *in vivo* conditions. Furthermore, global hydrodynamic properties of the protein were obtained at different states of association by means of sedimentation velocity, from which a gross shape of the protein p6 species can be derived.

MATERIALS AND METHODS

Determination of the in Vivo Concentration of Protein p6. *B. subtilis* 110NA (*try⁻ spo A⁻ su⁻*) (Moreno et al., 1974) cells were grown at 30 °C in LB medium supplemented with 5 mM MgSO₄ to 0.45 OD₅₅₀ and then infected with Ø29 *sus* 14(1242) mutant phage (delayed lysis phenotype) at a multiplicity of infection of 4. Aliquots were taken at different times after 5 min of adsorption for determination of the number of cells and their volume, phage development and the amount of Ø29 DNA and protein p6. The number and volume of cells was measured in a Coulter Counter model ZM. In addition, *B. subtilis* cells, previously fixed in 0.1% (v/v) formaldehyde, were observed with an Olympus System Microscope model BHS and photographed with an Olympus Photomicrographic System model PM-10AD. The volume of cells harvested 90 min after infection was also calculated directly from the photographs (about 600 measurements), taking into account the magnification factor and assuming an ellipsoidal shape. Intracellular phage development was followed by plating on *B. subtilis* MO-101-P (*thr⁻ spo A⁻ su⁺⁴⁴*) (Mellado et al., 1976) after treatment with lysozyme (500 µg/mL). The amount of Ø29 DNA was determined by electrophoresis as reported by Bravo et al. (1994), and the corresponding bands were quantified by densitometry in a Molecular Dynamic 300A densitometer using purified Ø29 DNA as standard. The amount of protein p6 was determined by two-dimensional gel electrophoresis, performed as previously described in Santarén et al. (1993). Cultures were concentrated 20-fold in lysis buffer (9.8 M urea, 2% ampholytes, pH 7–9, 4% NP-40 and 100 mM dithiothreitol), loaded on gels that, after electrophoresis, were stained with Coomassie blue. Protein p6 was quantified by densitometry as above, using as standard purified protein p6.

Sedimentation Equilibrium. The experiments were performed in a Beckman Optima XL-A analytical ultracentrifuge equipped with absorbance optics, using an An60Ti rotor. Protein p6 was equilibrated in buffer A (50 mM Tris-HCl, pH 7.5, and 10 mM MgCl₂) with 50 mM NaCl unless otherwise stated. Short column experiments (60–80 µL of protein p6, loading concentration ranging between 2 µM and 0.7 mM) were done at different speeds (15, 20, 25, and 30

krpm) by taking absorbance scans (0.001 cm step size, 4–10 averages) at the appropriate wavelength (230, 250, 280, or 290 nm) at sedimentation equilibrium. We used standard (12 mm optical path) double-sector or six-channel centerpieces of charcoal-filled Epon, with the exception of the experiments at the highest loading protein concentrations, where 4 mm double-sector centerpieces were employed, at scanning wavelengths between 280 and 300 nm. The equilibrium temperature was 20 °C, except otherwise indicated. In all the cases, high-speed sedimentation (42 krpm) was conducted afterward for baseline correction.

Whole-cell apparent weight-average molecular masses ($M_{w,a}^c$) were determined by fitting a sedimentation equilibrium model for a single sedimenting solute to individual datasets with the programs XLAEQ and EQASSOC [supplied by Beckman; see Minton (1994)]. The partial specific volume of protein p6 was 0.728 mL/g, calculated from the amino acid composition of the protein deduced from the gene 6 sequence (Murray & Rabinovitz, 1982) and, when necessary, corrected for temperature according to Laue et al. (1992). Several procedures were used to determine the self-association behavior of protein p6. (i) A monomer–dimer association model at sedimentation equilibrium was globally fitted to multiple experimental data using either the Micro-Cal-Origin version of NONLIN (Johnson et al., 1981) or the conservation of signal algorithm [MULTEQ1B and MULTEQ3B programs, see Minton (1994)]. This procedure was used at protein p6 concentrations below 0.1 mM. The monomer relative molecular mass was taken as 11 800 and a value of 8380 M⁻¹ cm⁻¹ was used for the extinction coefficient of protein p6 at 280 nm (Perkins, 1986). (ii) When the whole set of experimental data was considered, models for self-association (Chatelier & Minton, 1987; Muramatsu & Minton, 1989) were fitted to secondary data (apparent weight-average molecular mass, $M_{w,a}$, versus protein concentration) using a nonlinear least-squares method.

Sedimentation Velocity. The experiments were carried out at 60 krpm and 20 °C in the same XL-A instrument. Protein p6 (loading concentrations 3, 85, and 600 µM) was equilibrated in buffer A with 50 mM NaCl, unless otherwise specified. The sedimentation velocity data were analyzed with the programs XLAVEL (Beckman) and SVEDBERG (Philo, 1997). The former calculates the apparent sedimentation coefficients from the rate of movement of the solute boundary. The latter uses Faxen's approximation of the Lamm equation to fit the sedimentation profiles up to four sedimenting species. As a third method of analysis, the distribution of the apparent sedimentation coefficients, $g(s^*)$, was computed with the program DCDT (Stafford, 1994). The experimental sedimentation coefficients were corrected to standard conditions [water, 20 °C; see van Holde (1986)] to get the corresponding $s_{20,w}$ coefficients. A gross estimation of the shape of protein p6 (monomer, dimer) was determined as follows. The translational frictional coefficient (f) of protein p6 was calculated from the molecular mass and sedimentation coefficient of p6. The frictional coefficient of the equivalent hydrated sphere (f_0) was estimated using an hydration coefficient (∂_w) of 0.3 g of H₂O/g of protein (Pessen & Kumosinski, 1985). From these coefficients, the translational frictional ratio (f/f_0) of protein p6 at different states of association was determined, which allows the generation of a family of ellipsoids of revolution compatible

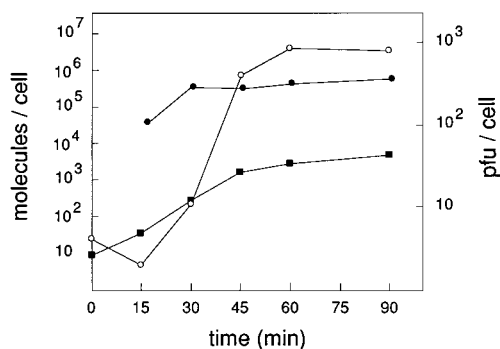


FIGURE 2: Time course accumulation of Ø29 DNA and protein p6. *B. subtilis* cells were infected with Ø29 *sus14* (1242) mutant phage, and phage development (○), amount of viral DNA (■), and protein p6 (●) per cell was determined as described in the Materials and Methods.

with the hydrodynamic properties of the protein (Stafford & Szent-Györgyi, 1978; Waxman et al., 1993).

RESULTS AND DISCUSSION

Determination of the *in Vivo* Concentration of Protein p6.

A previous rough estimation of the amount of protein p6 *in vivo*, using monodimensional gel electrophoresis, was about 3×10^6 copies per cell (Serrano et al., 1994). According to this value, assuming a cell volume of 2.6×10^{-15} L (Luria, 1960), the protein concentration was preliminarily assessed to be in the millimolar range (Gutiérrez et al., 1994). Since protein p6 was not resolved from host proteins and the cell volume is dependent of growing conditions and phage infection, we have now determined more accurately the intracellular protein p6 concentration by calculating the amount of protein by high-resolution two-dimensional gel electrophoresis and directly measuring the cell volume under our experimental conditions.

The high amount of protein p6 synthesized in phage Ø29 infected *B. subtilis* cells allowed us to detect the protein by Coomassie staining and to determine its time course accumulation after phage infection, as well as the intracellular concentration, taking into account the number of cells and their volume. The number of cells at different times after infection was measured in a Coulter. After 90 min of infection, the number of cells per milliliter [$(2.75 \pm 0.03) \times 10^8$] was very similar to that of uninfected cells [$(2.48 \pm 0.02) \times 10^8$]. The cell volume was also determined by the Coulter [$(0.78 \pm 0.02) \times 10^{-15}$ L after 90 min of infection]; however, since the cells have not a spherical shape, we used in the calculations a more accurate value, $(1.00 \pm 0.17) \times 10^{-15}$ L, obtained at the same time by optical microscopy assuming the cells have an ellipsoidal shape. Figure 2 shows the time course accumulation of the intracellular protein p6 and viral DNA, along the phage development. Under the experimental conditions, about 10^3 viable phages per infected cell was produced. Accumulation of protein p6 levels off 30 min after infection, shortly after the onset of intracellular viral production. The number of molecules of protein p6 per cell reached a maximum average of $(0.66 \pm 0.09) \times 10^6$, corresponding to a concentration of 1.09 ± 0.15 mM. The estimated amount of viral DNA after 30 min of infection was about 300 molecules per cell. Since the stoichiometry of protein p6 binding to DNA is one dimer per 24 bp (Serrano et al., 1990) and taking into account Ø29 DNA has

19 285 bp (Vlcek & Paces, 1986), 1607 molecules of protein would be required to complex entirely a single DNA molecule. Thus, at 30 min postinfection, the amount of intracellular protein p6 is about 1.4 times that required to saturate all the *in vivo* Ø29 DNA molecules. The number of Ø29 DNA molecules steadily increases up to 45 min after infection, when the available protein p6 could bind only about one third of the DNA. This figure could be underestimated, since as the viral production indicates, many DNA molecules are already encapsidated, precluding even a rough estimation of DNA molecules available for protein p6 binding.

The figures obtained for the amount of protein p6 are larger than those reported for bacterial histone-like proteins. In the case of *E. coli* HU, it has been estimated to be about 6×10^4 HU copies per cell allowing only 16% of the genome packaged (Drlica & Rouviere-Yaniv, 1987). In the case of H-NS, the amount reported is even lower, $\sim 2 \times 10^4$ molecules per cell (Spassky et al., 1984).

Characterization of the Protein p6 Monomer–Dimer Equilibrium. Earlier analytical ultracentrifugation work (Pastrana et al., 1985) showed that protein p6 was a dimer in solution (30, 60, and 120 μ M), and protein p6 dimers were detected after glutaraldehyde (Freire et al., 1994) or GGH-Ni(II) cross-linking (A. M. Abril, unpublished results). The first question to answer, the strength of the interaction between protein p6 monomers to form dimers, was undertaken by performing sedimentation equilibrium over a broad range of protein concentrations. Figure 3 (Panels A and B) shows that the average molecular mass ($M_{w,a}^c$) of protein p6 increases with protein concentration from a value corresponding to the theoretical molecular mass of the monomer (11 800) (2 μ M, panel A) to that of a dimer (95 μ M, panel B), indicative of a monomer–dimer equilibrium. In Figure 3C, the degree of association was plotted as a function of mid-channel protein concentration. As shown, the half-saturation of dimerization is reached at a protein p6 concentration of approximately 10^{-5} M (estimated $K_2 \approx 10^5$ M⁻¹). Figure 4 shows an example of further analysis of the dimerization equilibria of protein p6, performed by global fitting of sedimentation equilibrium data taken at different loading protein concentrations and speeds. Again, a monomer–dimer equilibrium model fit adequately the experimental data, and the best fit parameter value for the dimerization constant (K_2), using the Origin-NONLIN algorithm, was 2.0×10^5 M⁻¹ (1.5×10^5 , 2.8×10^5 , 95% confidence limits), in agreement with the previous estimation of K_2 . A similar result, $(3.5 \pm 1.5) \times 10^5$ M⁻¹, was obtained when the data were analyzed with the conservation of signal algorithm (Minton, 1994).

Effect of Temperature on the Dimerization of Protein p6.

The thermodynamic parameters associated with the dimerization of protein p6 were calculated from the temperature dependence of K_2 by means of sedimentation equilibrium. The K_2 value decreases with temperature, from 4 to 37 °C. The K_2 value at 4 °C was the same before and after incubation of protein p6 at 37 °C (data not shown). Figure 5A summarizes this analysis, in the form of a van't Hoff plot, where the natural logarithm of the association constant is plotted against the reciprocal of the absolute temperature. The standard enthalpy change ($\Delta H_{\text{obs}}^\circ$) on dimer formation

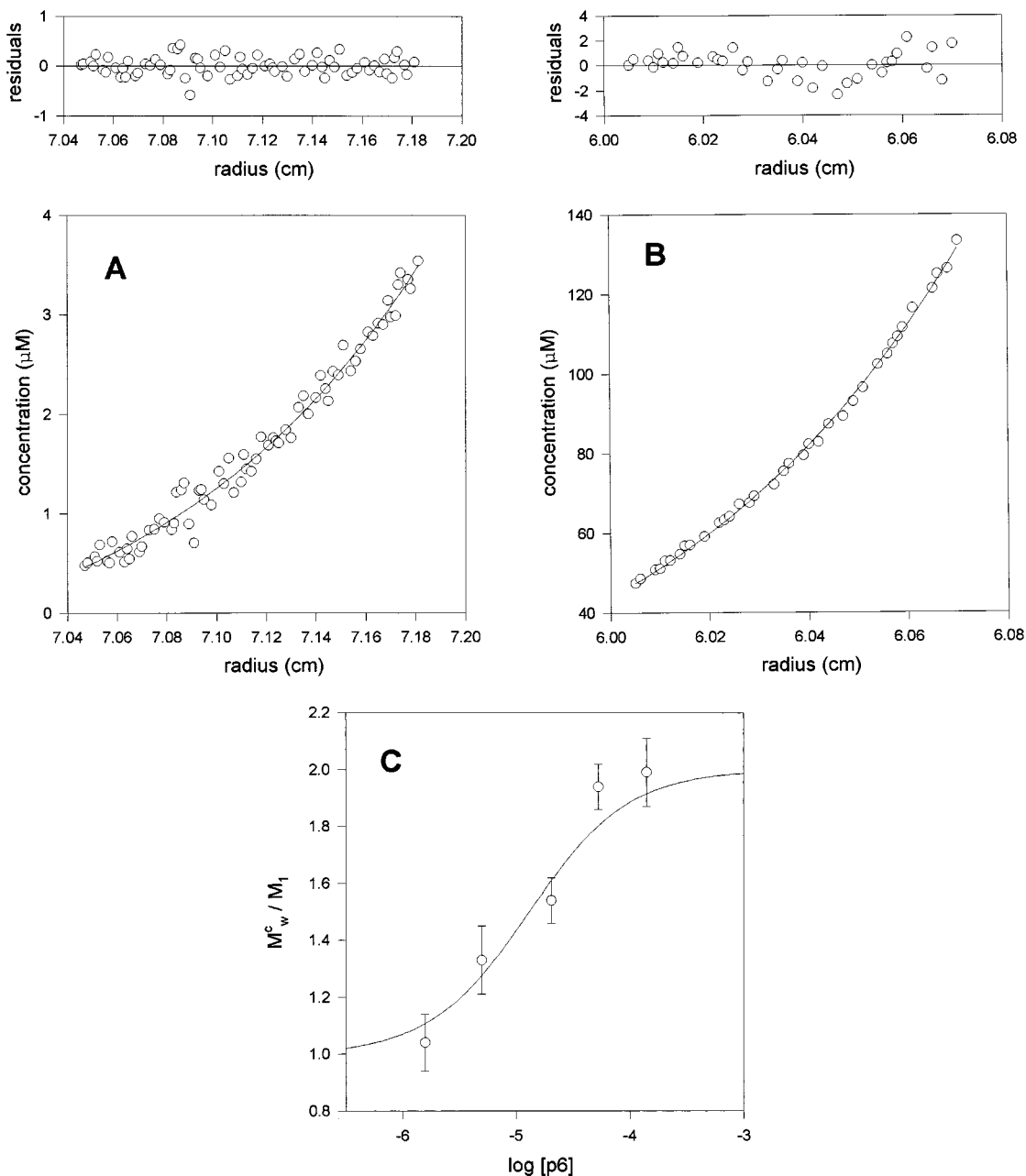


FIGURE 3: Sedimentation equilibrium analysis of protein p6. (Panel A) Sedimentation equilibrium profile of 2 μM protein p6 taken in buffer A with 50 mM NaCl, at 30 krpm, 20 $^{\circ}\text{C}$, as described in the Materials and Methods. The symbols represent the experimental data. The solid lines show the best fit functions for a single solute, at sedimentation equilibrium. The average molecular weight ($M_{w,a}^c$) was $11\,500 \pm 1000$ at this protein p6 concentration. (Panel B) The same as panel A with 95 μM protein p6. In this case, $M_{w,a}^c$ was $23\,800 \pm 400$. (Panel C) Dependence of the degree of association (M_w^c/M_1) of protein p6 on protein concentration. The solid line represents the best fit to a monomer–dimer equilibrium model (see text and Figure 4 for details). Error bars indicate ± 2 SD.

can be calculated from the slope of the van't Hoff plot (Record et al., 1991):

$$\left[\frac{\partial \ln K_2}{\partial (1/T)}\right]_p = -\Delta H_{\text{obs}}^{\circ}/R \quad (1)$$

In the case of protein p6, the van't Hoff plot is nonlinear, which would correlate with a considerable standard heat capacity change ($\Delta C_{p,\text{obs}}^{\circ}$) upon complex formation, since

$$\Delta C_{p,\text{obs}}^{\circ} = (\partial \Delta H_{\text{obs}}^{\circ} / \partial T)_p \quad (2)$$

The data were analyzed according to the nonlinear integrated van't Hoff equation (Andreu et al., 1983; Naghibi

et al., 1995):

$$\ln(K_2/K_2^{\circ}) = [(\Delta H_{\text{obs}}^{\circ} - T^{\circ} \Delta C_{p,\text{obs}}^{\circ})/R](1/T^{\circ} - 1/T) + (\Delta C_{p,\text{obs}}^{\circ}/R) \ln(T/T^{\circ}) \quad (3)$$

where the superscripts indicate the corresponding thermodynamic values at a reference temperature (T°). This analysis resulted in a constant negative heat capacity change $\Delta C_{p,\text{obs}}^{\circ} = -0.92 \pm 0.14$ kcal mol $^{-1}$ and variable enthalpy values (at 20 $^{\circ}\text{C}$, $\Delta H_{\text{obs}}^{\circ} = -11.3 \pm 1.9$ kcal mol $^{-1}$). The free energy change at each temperature was calculated directly as $\Delta G_{\text{obs}}^{\circ} = -RT \ln K_2$ (-7.23 ± 0.37 kcal mol $^{-1}$ at 20 $^{\circ}\text{C}$) and the entropic contribution was calculated by difference

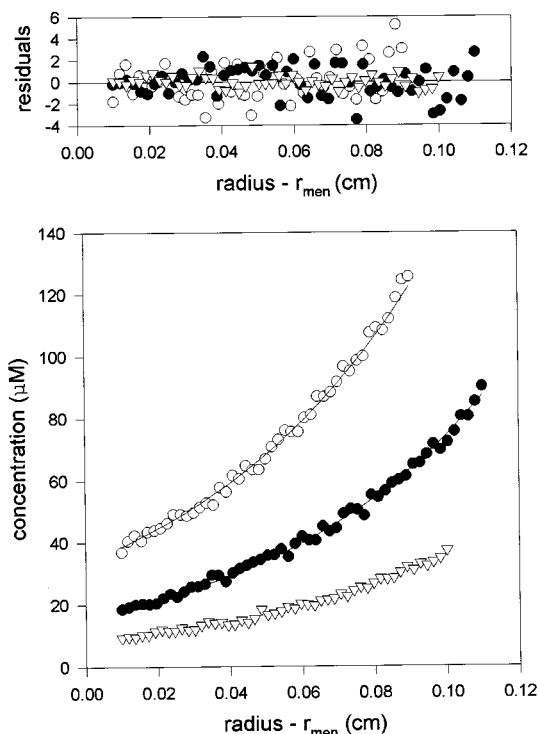


FIGURE 4: Monomer-dimer equilibrium of protein p6. The symbols show the experimental radial distribution of protein p6 at sedimentation equilibrium at 23 μM (∇), 67 μM (\bullet) and 95 μM (\circ). The solid lines represent the best-fit curves of the global analysis of multiple sedimentation equilibrium data to the monomer-dimer equilibrium model (buffer A with 50 mM NaCl at 20 $^{\circ}\text{C}$), described in the text (M_1 was constrained to 11 800; $K_2 = 2 \times 10^5 \text{ M}^{-1}$). From six fitted data sets (the three previously indicated protein concentrations at 25 krpm and 30 krpm) only three (30 krpm) are shown. r_{men} (95 μM) = 6.01 cm; r_{men} (67 μM) = 6.53 cm; r_{men} (23 μM) = 7.02 cm.

as $T\Delta S^{\circ}_{\text{obs}} = \Delta H^{\circ}_{\text{obs}} - \Delta G^{\circ}_{\text{obs}}$ ($-4.1 \pm 0.7 \text{ kcal mol}^{-1}$ at 20 $^{\circ}\text{C}$).

Figure 5B summarizes the temperature dependence of the observed thermodynamic functions for the dimerization of protein p6. The variation of the standard free energy change is relatively small in the temperature range investigated. However, as the temperature increases, there is a large decrease of both the enthalpic and entropic values contributing to this free energy change. This indicates the existence of a shift with temperature of the thermodynamic forces involved in protein p6 dimer formation: at low temperature it is an entropically driven process, which becomes enthalpically driven as temperature raises.

The negative values obtained for $\Delta H^{\circ}_{\text{obs}}$ and $\Delta S^{\circ}_{\text{obs}}$, that are common in protein associations, could be compatible with van der Waals contacts or hydrogen bonding occurring upon dimer formation (Sturtevant, 1977; Ross & Subramanian, 1981). The negative $\Delta C^{\circ}_{\text{p,obs}}$ could result from the reduction in water-accessible nonpolar surface area on the protein upon complex formation (Livingstone et al., 1991).

Effect of Ionic Strength on the Self-Association of Protein p6. Preliminary data from protein p6 cross-linking experiments indicated that formation of oligomers higher than dimers was favored at concentrations of NaCl above 0.2 M (R. Freire and A. M. Abril, unpublished results). The dependence of protein p6 self-association on ionic strength was tested by sedimentation equilibrium at 0.2 and 1.0 M NaCl, in addition to the previous one (50 mM). Figure 6

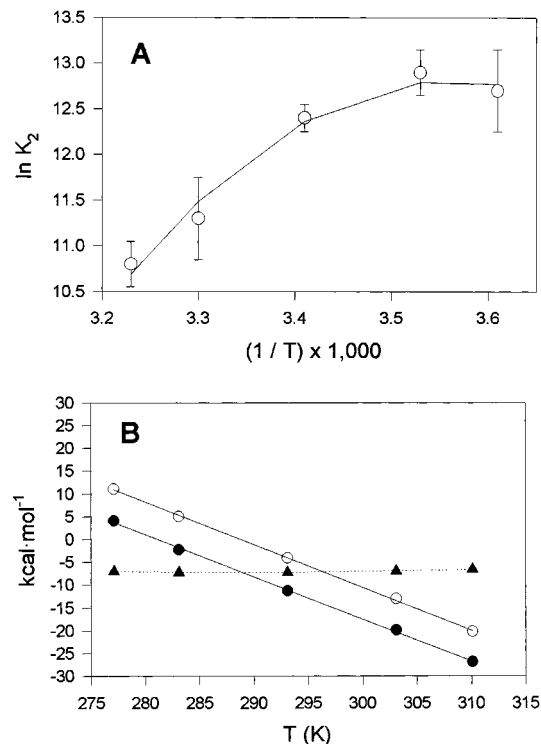


FIGURE 5: Effect of temperature on the self-association of protein p6. (Panel A) van't Hoff plot for the dimerization of protein p6 over a temperature range 4–37 $^{\circ}\text{C}$. The open circles represent the dimerization constant determined by a global fit to a monomer-dimer association model of three sedimentation equilibrium data (loading concentrations 5, 20, and 45 μM) taken at 30 krpm. The solid line is the best fit to the nonlinear integrated van't Hoff equation (see ref 3). Error bars are ± 2 SD. (Panel B) Effect of temperature on the thermodynamic functions for the dimerization of protein p6. ΔH° (\bullet), $T\Delta S^{\circ}$ (\circ) and ΔG° (\blacktriangle).

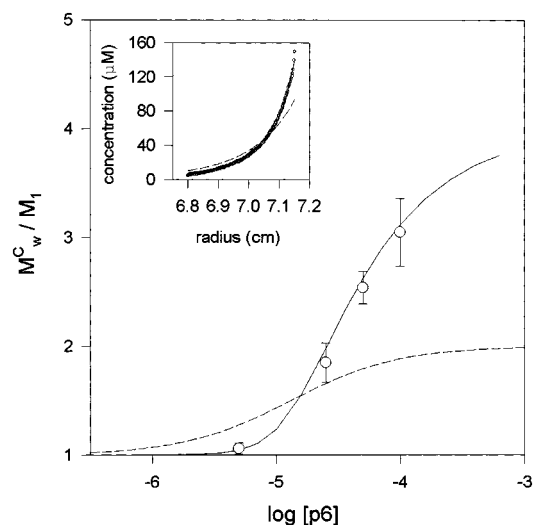


FIGURE 6: Effect of ionic strength on the self-association of protein p6. Dependence of the degree of association (M_w^c/M_1) of protein p6 on NaCl concentration. Open circles correspond to 1 M NaCl data. The solid line shows the best fit for a monomer-dimer-tetramer association scheme ($K_2 = 1.8 \times 10^4 \text{ M}^{-1}$; $K_4 = 3.1 \times 10^{14} \text{ M}^{-3}$). Error bars indicate ± 2 SD. The dotted line is the monomer-dimer equilibrium model at 50 mM NaCl, previously described in Figure 3C. (Inset) Sedimentation equilibrium profile of protein p6 (50 μM loading concentration, 25 krpm at 20 $^{\circ}\text{C}$). Open circles are the experimental data and lines are the best fits to the association models described in the figure.

summarizes the results obtained, where the degree of oligomerization is plotted against approximate mid-channel

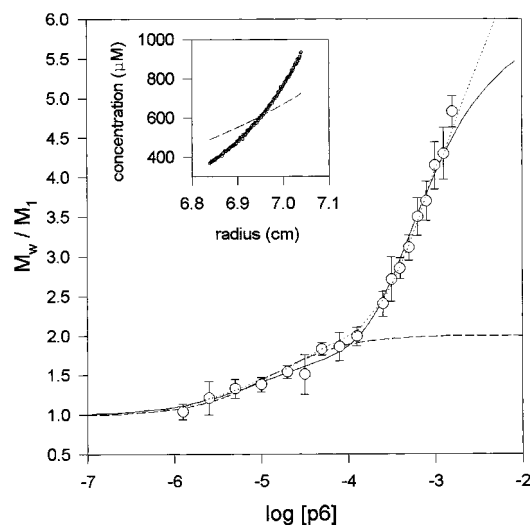


FIGURE 7: Self-association of protein p6 at the *in vivo* protein concentration. Summary of the dependence of the apparent weight-average molecular weight on protein p6 concentration. Open circles represent experimental values of $M_{w,a}$ preaveraged as described in the text; error bars indicate ± 2 SD. The dashed line is the theoretical monomer-dimer equilibrium model previously described. The solid line is the best fit to a monomer-dimer-hexamer model ($K_2 = 1.1 \times 10^5 \text{ M}^{-1}$; $K_{2,6} = 3.2 \times 10^8 \text{ M}^{-2}$). The dotted line is the best fit to an isodesmic dimer association model ($K = 950 \text{ M}^{-1}$). (Inset) Radial distribution of protein p6 at sedimentation equilibrium (0.7 mM loading concentration, 15 krpm at 20 °C). Open circles are the experimental data and lines are the best fits to the association models described in the figure.

protein concentration for 50 mM and 1.0 M NaCl. The equilibration of protein p6 with 1.0 M NaCl favors the formation of higher order states of association, in agreement with the cross-linking data (0.2 M NaCl gave an incipient effect; not shown). Global analysis of multiple sedimentation equilibrium experiments at 1 M NaCl does not support a monomer-dimer association scheme, rather a monomer-dimer-tetramer model, with a K_2 value of $1.8 \times 10^4 \text{ M}^{-1}$ (1.0×10^4 , 2.9×10^4 ; 95% confidence limits) and K_4 equal to $3.1 \times 10^{14} \text{ M}^{-3}$ (0.4×10^{14} , 6.4×10^{14} ; 95% confidence limits). The inset in Figure 6 illustrates the effect of salt concentration in a single sedimentation equilibrium gradient of protein p6.

The dependence of a self-association reaction on ionic strength gives qualitative information on the role of specific amino acid groups upon protein associations (Cole & Ralston, 1992; Kim et al., 1977; Record et al., 1978). In the case of protein p6, the effect of salt indicates that charged residues do not seem to play a critical role in the formation of higher order states of association. A similar behavior with salt was found on the monomer-dimer equilibrium of α -chymotrypsin (Aune et al., 1971). More recently, Koblan and Ackers (1991) showed an increase in λ repressor dimer stability with KCl concentration.

Protein p6 Forms Higher Order Oligomers at the *in Vivo* Protein Concentrations. Since the intracellular concentration of protein p6 was estimated to be in the millimolar range, we have further studied by sedimentation equilibrium the self-association behavior of protein p6 at that concentration. Figure 7 summarizes the results of sedimentation equilibrium (in buffer A with 50 mM NaCl at 20 °C) as a plot of degree of association *versus* protein p6 concentration, over the whole range of protein concentrations employed in the present study. Each symbol represents the mean of all the data

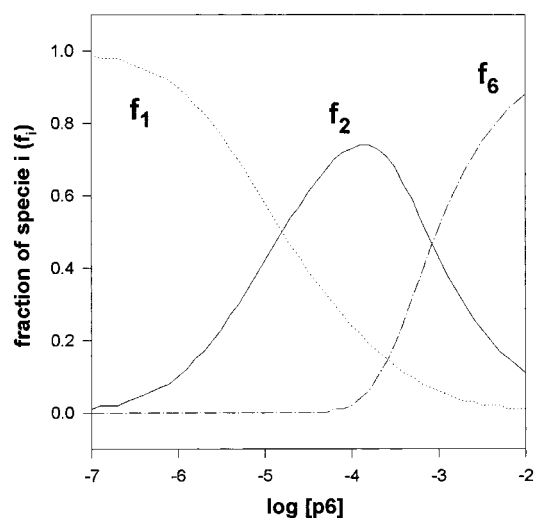


FIGURE 8: Predicted fractional distribution of protein p6 species, calculated from the best fit parameter values of the monomer-dimer-hexamer model described in the text (see also Figure 7).

averaged over ± 0.1 log unit on the concentration axis. Also plotted in the same figure are curves representing the dependence of the degree of association upon protein concentration calculated using three different models of association. The first one (dashed line) is based on the previously described monomer-dimer equilibrium model, and it is shown only for illustrative purposes, to indicate the disagreement between the experimental data and the calculated function above ~ 0.1 mM protein p6 concentration. The second calculated curve (solid line) is in reasonably good agreement with the data and was generated from the best fit parameter values of a monomer-dimer-hexamer association scheme, with a dimerization constant in the same range as the one previously found for the monomer-dimer equilibrium at low protein concentration [$K_2 = (1.1 \pm 0.5) \times 10^5 \text{ M}^{-1}$, $K_{2,6} = (3.2 \pm 1.1) \times 10^8 \text{ M}^{-2}$]. Because the $M_{w,a}$ does not reach saturation at the highest protein p6 concentrations, the data do not exclude the formation of oligomeric structures capable of further growth, or even an isodesmic type of association (Adams & Lewis, 1968) from the protein p6 dimer, in which the equilibrium constants for incorporation of protein dimers to the oligomer are identical. In fact, the third model (dotted line) is an isodesmic association of the dimer ($K = 950 \pm 60 \text{ M}^{-1}$). Experimental limitations preclude to discriminate between second and third models since we cannot go higher in protein concentration, due to the optical capabilities of the analytical ultracentrifuge. Figure 8 shows the fractional distribution of protein p6 species plotted as a function of protein concentration to illustrate the self-association scheme of protein p6. We found no evidence of nonideal behavior throughout this work, according to the residual distribution of the best fit models for the sedimentation equilibrium gradients. Moreover, including a reasonably large nonideality term in the M_w *versus* concentration data [based upon Chatelier and Minton (1987)] did not change the stoichiometry of the best-fit model or even significantly alter the best fit values of the association constant with respect to those obtained ignoring nonideality.

Histone-like proteins seem to have a similar behavior. Using cross-linking reagents, HU- α monomers are predominant at protein concentrations about 50 nM, while the predominant form is dimeric at 50 μM , and from 100 μM (the estimated *in vivo* concentration), the amount of dimers

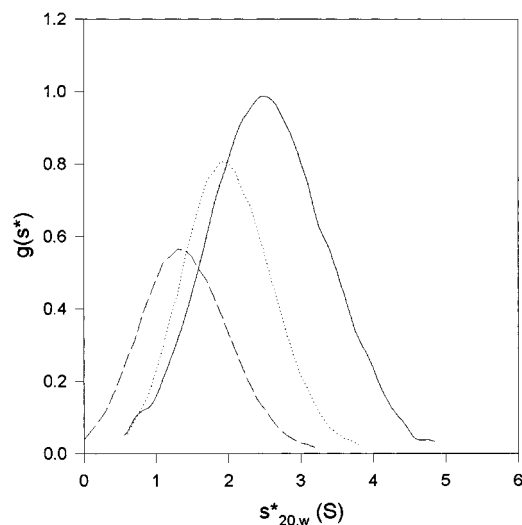


FIGURE 9: Distribution of the apparent sedimentation coefficients taken in buffer A with 50 mM NaCl at 60 krpm and 20 °C at three different protein p6 concentrations: 3 μ M (dashed line), 85 μ M (dotted line) and 600 μ M (solid line).

Table 1: Average Molecular Weights ($M_{w,a}^c$), Sedimentation Coefficients ($s_{20,w}$), and Frictional Coefficient Ratios (f/f_0^a) of Protein p6

protein p6	$M_{w,a}^c$	$s_{20,w}$	f/f_0^a
monomer	11 500 \pm 900	1.4 \pm 0.1	1.15 \pm 0.08
dimer	23 800 \pm 400	2.0 \pm 0.1	1.30 \pm 0.07
tetramer ^b	46 200 \pm 1400	2.6 \pm 0.2	

^a Assuming ∂w 0.3 g of H₂O/g of protein. ^b Average value.

decreases as the amount of tetramers and larger aggregates increases (Losso et al., 1986). A similar result was obtained by gel-filtration analysis of H-NS, indicating that this protein forms a dimer as a structural unit in solution, and undergoes oligomerization, depending on its concentration; for instance, tetramers (and hexamers in some mutants) were predominant at 10 μ M, but neither trimers nor odd-numbered oligomers could be detected (Ueguchi et al., 1996). More recently, H-NS mutants have been described with impaired capacity to oligomerize both *in vitro* and *in vivo* (Spurio et al., 1997).

Sedimentation Velocity. Sedimentation velocity profiles of protein p6 at two protein concentrations which essentially correspond to monomer and dimer protein p6 (3 and 85 μ M, respectively; see figure 8) revealed an apparent single boundary in each case, with a higher s value as protein p6 concentration increases (Figure 9). In both cases, the data are well fitted by means of the program SVEDBERG with a single sedimenting species (not shown), with $s_{20,w}$ of 1.4 and 2.0 S for monomer and dimer protein p6, respectively (Table 1). These values agree, within experimental error, with those obtained with the program XLAVEL, and correspond to the peak position in the $g(s)$ profiles determined with the program DCDT (Figure 9). In the 85 μ M sample (predominantly protein p6 dimer), there was no significant improvement in the best fit parameter obtained with SVEDBERG if a second sedimenting species is introduced. These sedimentation coefficient values are compatible with a translational frictional ratios (f/f_0) of 1.15 and 1.30, respectively (Table 1), consistent with axial ratios for a prolate ellipsoid of 5.8 and 10.8, respectively. Therefore, the global hydrodynamic behavior of monomer and dimer protein p6 slightly deviates from the one corresponding

to a rigid spherical particle (Teller, 1973; Waxman et al., 1993). A third protein concentration, corresponding to protein p6 oligomers (0.6 mM) was also analyzed. This sample has a whole-cell average molecular weight of 46 200 ($M_1 = 11\ 800$) and sedimenting species model. The best fit parameters corresponded to a 1:1 mixture of two particles with $s_{20,w}$ values of 1.9 and 3.3, respectively; the first can be identified with the dimer of protein p6 the latter has an apparent s value intermediate between those corresponding to tetramer and hexamer protein p6 (not shown). The complexity of this equilibrium mixture, together with the uncertainties in the shape of the oligomers species, makes difficult the discrimination between models of higher order self-association of protein p6 and precludes a more rigorous hydrodynamic analysis.

CONCLUDING REMARKS

The present study demonstrates that protein p6, in the absence of DNA, self-associates in solution. Furthermore, at physiological concentrations, protein p6 is capable of oligomerization from a preformed dimer. The oligomer formation of protein p6 should be enhanced *in vivo* due to excluded volume effects in a crowded media (Zimmerman & Minton, 1993). The monomer-dimer-oligomer association would suggest the existence of two protein-protein interaction domains, one for monomers and the other for dimers. In agreement with this, cross-linking studies have shown that a deletion mutant of protein p6 that forms dimers is impaired in oligomerization (A. M. Abril, unpublished results). Taking into account the model proposed for the structure of protein p6-DNA complexes (Figure 1, Serrano et al., 1993), the biological problem posed is how to assemble the nucleoprotein structure to maintain this configuration. From the results obtained in this study, we speculate that protein p6 could behave as a scaffolding protein on which DNA folds.

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