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Effect of chemical form, heating, and oxidation products of linoleic acid on rumen bacterial population and activities of biohydrogenating enzymes

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ABSTRACT

Heating polyunsaturated fatty acids (PUFA) produces oxidation products, such as hydroperoxides, aldehydes, and oxypolymers, which could be responsible at least in part for modification of PUFA rumen biohydrogenation (BH). Three in vitro experiments were conducted to investigate the effects of linoleic acid (*cis*-9,*cis*-12-C18:2) oxidation products on BH. In the first experiment, we studied the effects of free linoleic acid (FLA), heated FLA (HFLA, at 150°C for 6 h), triacylglycerols of linoleic acid (TGLA), heated TGLA (HTGLA, at 150°C for 6 h), 13-hydroperoxide (13HPOD), *trans*-2-decenal (T2D), and hexanal (HEX) on BH in vitro after 6 and 24 h of incubation. In the second experiment, aldehydes differing in chain length and degree of unsaturation [pentanal, HEX, heptanal, nonanal, T2D, *trans*-2,*trans*-4-decadienal (T2T4D)] were incubated in vitro for 5 h in rumen fluid. In the third experiment, 9-hydroperoxide (9HPOD), 13HPOD, HEX, or T2T4D were incubated for 1 h in rumen fluid inactivated with chloramphenicol to investigate their effects on enzyme activity. In experiment 1, heat treatment of TGLA generated TGLA oxypolymers, did not affect *cis*-9,*cis*-12-C18:2 disappearance, but did decrease BH intermediates, especially *trans*-11 isomers. Heating FLA decreased *cis*-9,*cis*-12-C18:2 disappearance and *cis*-9,*trans*-11-CLA and *trans*-11-C18:1 production. Treatment with HEX and T2D did not affect *cis*-9,*cis*-12-C18:2 disappearance and barely affected production of BH intermediates. The bacterial community was affected by 13HPOD compared with FLA and HFLA, in parallel with an increase in *trans*-10 isomer production after a 6-h incubation. After 24 h of incubation, 13HPOD decreased *trans*-11 isomer production, but to a lesser extent than HFLA. In experiment 2, some weak but significant effects were observed on BH, unrelated to chain length or degree of

unsaturation of aldehydes; the bacterial community was not affected. In experiment 3, 9HPOD inhibited Δ^9 -isomerization, and both 9HPOD and 13HPOD inhibited Δ^{12} -isomerization. We concluded that oxypolymers did not affect *cis*-9,*cis*-12-C18:2 disappearance. Heating both esterified and free *cis*-9,*cis*-12-C18:2 greatly altered Δ^{12} -isomerization. Aldehydes had few effects. Hydroperoxides are responsible, at least in part, for the effects of fat heating: 13HPOD increased *trans*-10 isomer production (probably by affecting the bacterial community) and decreased *trans*-11 isomer production by inhibiting Δ^{12} -isomerase activity, whereas 9HPOD inhibited both isomerases.

Key words: rumen biohydrogenation, lipid oxidation, linoleic acid, *trans* fatty acids

INTRODUCTION

Trans FA, including various *trans*-octadecenoic acid (*trans*-C18:1) isomers and most conjugated linoleic acid (CLA) isomers, which are positional and geometric isomers of linoleic acid (*cis*-9,*cis*-12-C18:2) with conjugated double bonds, have been reported to potentially affect the risk of cancer and cardiovascular diseases in humans (Gebauer et al., 2011). Among CLA isomers, *cis*-9,*trans*-11-CLA has beneficial properties for human health, whereas *trans*-10,*cis*-12-CLA may be harmful (Troegeler-Meynadier and Enjalbert, 2005). Many human foods contain *trans* FA, with dairy products having the highest concentrations. In dairy cows, milk *trans* FA have 2 origins. The first is microbial ruminal biohydrogenation (BH) of *cis*-9,*cis*-12-C18:2, which begins with isomerization, mainly producing *cis*-9,*trans*-11-CLA and *trans*-10,*cis*-12-CLA (Enjalbert and Troegeler-Meynadier, 2009). Thereafter, initial reduction produces different *trans*-C18:1, in particular *trans*-11-C18:1 (vaccenic acid) and *trans*-10-C18:1. Finally, a subsequent reduction produces stearic acid (C18:0). *Butyrivibrio fibrisolvens* is believed to be responsible for the production of *cis*-9,*trans*-11-CLA and *trans*-11-C18:1 (Enjalbert and Troegeler-Meynadier,

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2009), whereas *Megasphera elsdenii* (Kim et al., 2002) and *Propionibacterium acnes* (Wallace et al., 2007) have been reported to form *trans*-10,*cis*-12-CLA in vitro. A more significant origin of milk CLA is mammary desaturation of *trans*-11-C18:1 into *cis*-9,*trans*-11-CLA (Grinari et al., 2000). *trans*-11-C18:1 is an intermediate of ruminal BH of both *cis*-9,*cis*-12-C18:2 and α -linolenic acid (*cis*-9,*cis*-12,*cis*-15-C18:3). Thus, milk CLA content could be influenced both by factors affecting *cis*-9,*cis*-12-C18:2 and *cis*-9,*cis*-12,*cis*-15-C18:3 BH and by factors affecting desaturation in the mammary gland.

One way to efficiently increase CLA milk concentration is the addition of a fat supplement to the dairy cow diet, particularly a fat source rich in *cis*-9,*cis*-12-C18:2 or *cis*-9,*cis*-12,*cis*-15-C18:3. In dairy cow diets, fat is usually provided by oilseeds, which are often heat treated; for example, by roasting or extrusion. These processes have been shown to have variable effects on BH in different studies, which need to be described and explained. Heating oilseeds often induces a decrease in *cis*-9,*cis*-12-C18:2 and *cis*-9,*cis*-12,*cis*-15-C18:3 BH in vivo (Gonthier et al., 2005), in situ (Troegeler-Meynadier et al., 2006b), and in vitro (Reddy et al., 1994; Kaleem et al., 2013). Furthermore, heat treatment of oilseeds results in increased milk proportions of BH intermediates, particularly *cis*-9,*trans*-11-CLA and *trans*-11-C18:1 (Chouinard et al., 1997a,b; Chilliard et al., 2009).

A previous study in our laboratory (Privé et al., 2010) showed that increasing the heating temperature of sunflower oil led to partial protection of PUFA against BH in vitro, but contrary to previous reports with heated oilseeds, heated oil increased *trans*-10 isomers and decreased *trans*-11 isomers. These effects were linked to the peroxide value of oil and associated with a modification of the bacterial community. More recently, a study with heated oilseeds (Kaleem et al., 2013) showed that aldehydes, in particular hexanal (**HEX**), were linked to a decrease in PUFA disappearance in ruminal cultures.

These changes in BH caused by heated fat could be mediated by potentially active molecules generated during heat treatment: the lipid oxidation products (mostly FA hydroperoxides and aldehydes). Oxidation of *cis*-9,*cis*-12-C18:2 can be divided into 3 steps (Frankel, 2005). During initiation, *cis*-9,*cis*-12-C18:2 is converted to an alkyl radical. During propagation, oxygen binds with this radical forming a peroxy radical, which takes a hydrogen radical from another *cis*-9,*cis*-12-C18:2, forming a hydroperoxide [13OOH-*cis*-9,*trans*-11-linoleate (**13HPOD**) and 9OOH-*trans*-10,*cis*-12-linoleate (**9HPOD**) for *cis*-9,*cis*-12-C18:2] and spreading the oxidation reaction to other FA. The third step is termination, which may produce hydroxy-

acids, aldehydes, ketones, or triacylglycerol (**TAG**) oxypolymers.

A hypothesis for the action of oxidative products on BH is that hydroperoxides could interfere with BH reactions through competitive inhibition of enzymes because of their structural similarity to PUFA and mainly CLA. Moreover, hydroperoxides could be toxic to bacteria because they have been shown to inhibit the activity of rumen anaerobic microorganisms (Brioukhanov and Netrusov, 2004). Aldehydes have also been reported to have toxic effects on rumen bacteria, depending on their chain length (Kubo et al., 1995) and degree of unsaturation (Deng et al., 1993), and some have been shown to promote BH, in particular *trans*-2-decenal (**T2D**; Lee et al., 2007). They also could protect PUFA from BH, in particular **HEX** (Kaleem et al., 2013). Moreover, TAG oxypolymers could be less sensitive to lipases because of their steric hindrance, resulting in a partial protection of PUFA from BH (as only free FA can be hydrogenated by rumen bacteria).

The objectives of this study were to investigate whether TAG oxypolymers, hydroperoxides, or aldehydes were responsible, at least in part, for the BH alterations observed with heated fat, focusing particularly on the effects of chain length and degree of unsaturation of aldehydes. Moreover, as a first mechanistic approach, we studied the effects of oxidation products on the bacterial community, using an overall quantitative and qualitative approach, and the effects on enzymatic activities using chloramphenicol to prevent synthesis of enzymes during ruminal incubation.

MATERIALS AND METHODS

Three experiments were conducted in vitro: the first explored the effect on BH and rumen bacteria of TAG oxypolymers, obtained by heating TAG, and the effects of heating pure *cis*-9,*cis*-12-C18:2 or adding pure 13HPOD or 2 pure aldehydes (**HEX** and **T2D**); the second investigated the effect of aldehyde chain length and degree of unsaturation on rumen BH and bacteria. The third investigated the effects of 13HPOD and 9HPOD and 2 aldehydes (**HEX** and *trans*-2,*trans*-4-decadienal, **T2T4D**) on BH enzyme activities.

In Vitro Incubations

Experiment 1. As there is no available commercial preparation of TAG oxypolymer, we produced these compounds by heating pure TAG of *cis*-9,*cis*-12-C18:2 (**TGLA**, purity $\geq 99\%$, Sigma Co., St Louis, MO) at 150°C for 6 h, resulting in heated TGLA (**HTGLA**). Pure *cis*-9,*cis*-12-C18:2 (free linoleic acid, **FLA**, purity $\geq 99\%$, Sigma Co.) was heated separately at 150°C for

6 h, resulting in heated FLA (**HFLA**). Once heated, TGLA and FLA were left at ambient temperature for 30 min before storage at -80°C until utilization. For hydroperoxides, 13HPOD was synthesized in the laboratory from pure *cis-9,cis-12-C18:2* using the methods of Fauconnier et al. (1997) and López-Nicolás et al. (1999). Briefly, *cis-9,cis-12-C18:2* was incubated in a pH 9 borate buffer (0.2 M) at 0°C for 30 min with soybean lipoxygenase as the catalyst, under constant aeration. Then, the pH of the buffer solution was adjusted to 3 using 3 N HCl and the solution was purified on a solid-phase cartridge (Strata C18-E, 500 mg, Phenomenex, Le Pecq, France). The 13HPOD was eluted with ethanol, which was then evaporated under vacuum at 40°C . Once the solvent was completely eliminated, the samples were placed in small bottles filled with nitrogen and stored at -80°C until further use. Before the experiment, 13HPOD in this preparation was quantified as described later: purity was 78%. The 2 aldehydes, HEX and T2D, used in the experiment were those already shown to modify BH (Kaleem et al., 2013 and Lee et al., 2007, respectively) and were of commercial origin (purity $\geq 97\%$ and $\geq 95\%$, respectively, Sigma Co.). Ruminal cultures were performed in a water bath rotary shaker (Aquatron; Infors AG, Bottmingen, Germany). Rumen contents were collected from a lactating dairy cow fed a corn silage-based diet (29.6 kg of corn silage, 2 kg of soybean meal, and 0.2 kg of a mineral and vitamin supplement, as fed) in 2 equal meals. The cow was adapted to this diet for 18 d before the beginning of the experiment. One liter of ruminal fluid was taken from the cow with a vacuum pump 30 min before the morning meal and strained through a metal sieve (1.6-mm mesh). Rumen fluid was quickly transferred to the laboratory in anaerobic conditions at 39°C .

Erlenmeyer flasks, containing 52 mg of TGLA or HTGLA, 50 mg of FLA or HFLA, or 50 mg of FLA together with 5 mg of 13HPOD, HEX, or T2D, and a fermentative substrate (1 g of 2-mm-ground meadow hay, 0.2 g of 1-mm-ground soybean meal, and 0.25 g of 1-mm-ground corn grain), were filled with 40 mL of rumen fluid and 40 mL of bicarbonate buffer gassed with CO_2 and heated to 39°C (pH = 7; $\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$ 19.5 g/L, NaHCO_3 9.24 g/L, NaCl 0.705 g/L, KCl 0.675 g/L, $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ 0.108 g/L, and $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ 0.180 g/L). The filled flasks were gassed with CO_2 and placed in the water bath at 39°C . The flasks were then closed with a rubber bung through which a plastic tube led into the water to vent fermentation gas without ingress of oxygen. Flasks were stirred at 130 rpm and kept out of the light for 6 or 24 h. At the end of incubation, fermentations were stopped by placing the flasks in iced water. One replicate for each treatment and incubation

duration was incubated during each of the 6 d of culture, regrouped in 2 wk (i.e., a total of 6 replicates per treatment for 6-h and for 24-h incubations). Each day, 2 blanks, without fat or any oxidative product and not incubated, were prepared to represent the initial state of the media and placed immediately in iced water. In incubated flasks and blanks, pH was measured and 2×1 mL aliquots were sampled in Eppendorf tubes for bacterial community analysis. The contents of the flasks and Eppendorf tubes were then immediately frozen. Samples were freeze-dried (Virtis Freezemobile 25; Virtis, Gardiner, NY) and weighed. Flask contents were ground and homogenized in a ball mill (Dangoumau, Prolabo, Nogent-sur-Marne, France) and kept at -18°C for later analysis of FA. Eppendorf tubes were kept at -80°C until analysis.

Experiment 2. Ruminal cultures were prepared as for experiment 1, with some differences. Rumen fluid was collected and mixed from 2 lactating dairy cows fed a corn silage-based diet (20 kg of corn silage, 3 kg of meadow hay, 1.6 kg of soybean meal, and 0.2 kg of mineral vitamin mixture, as fed, per day). The cows were adapted to this diet for 16 d before the beginning of the experiment. Six different aldehydes were chosen as the oxidation products, taking into account their chain length and degree of unsaturation: pentanal (**PEN**, purity $\geq 97\%$), HEX (purity $\geq 97\%$), heptanal (**HEP**, purity $\geq 95\%$), nonanal (**NON**, purity $\geq 95\%$), T2D (purity $\geq 95\%$), and T2T4D (purity $\geq 85\%$), all of commercial origin (Sigma Co.). Ten milligrams of 1 aldehyde was added to the cultures along with 80 mg of pure *cis-9,trans-11-C18:2*, and a fermentative substrate (2 g of 2-mm-ground meadow hay, 0.4 g of 1-mm-ground soybean meal, and 0.5 g of 1-mm-ground corn grain). Then, 80 mL of rumen fluid and 80 mL of bicarbonate buffer were added to each flask. Two replicates of each treatment were incubated for 5 h, to give priority to CLA concentrations, during each of the 3 d of culture, regrouped in 1 wk (i.e., a total of 6 replicates per treatment). Each day, 2 blanks, without fat and not incubated, were prepared to represent the initial state of the media. Because the main objective of this experiment was to investigate the effects of chain length and degree of unsaturation of aldehydes, no control without added aldehydes was incubated. At the end of the incubation, fermentations were stopped by placing the flasks into iced water, pH was measured, and samples were prepared and kept as for experiment 1 for bacterial and FA analysis.

Experiment 3. To explore the effect of lipid oxidative compounds on the 3 reactions of *cis-9,cis-12-C18:2* BH, 4 substrates were used: pure *cis-9,cis-12-C18:2* for studying isomerization, *cis-9,trans-11-CLA* and *trans-10,cis-12-CLA* (for both, purity $\geq 98\%$, Cayman Chem-

ical, Ann Arbor, MI) for studying the first reduction and *trans*-11-C18:1 (purity $\geq 99\%$, Sigma Co.) for the last reduction. No incubation with *trans*-10-C18:1 was performed because no commercial standard is available for *trans*-10-C18:1. Oxidative products tested were the hydroperoxides 13HPOD and 9HPOD (for both, purity $\geq 98\%$, Cayman Chemical) and the 2 main aldehydes of *cis*-9,*cis*-12-C18:2 oxidation, HEX and T2T4D. Their effects were assessed through the quantity of reaction products after a 1-h incubation with inactivated rumen fluid.

The lactating dairy cow from which the rumen fluid was taken was fed a corn silage-based diet (36 kg of corn silage, 2 kg of hay, 10 kg of a mixture of concentrates, and 0.2 kg of mineral vitamin mixture, as fed, per day). The adaptation period was 4 wk. The procedure for rumen fluid collection and transportation was the same as in the previous 2 experiments. In the laboratory, 200 mg of chloramphenicol was immediately added to 200 mL of rumen fluid in a flask and the flask was gassed with CO₂, closed, and stirred. After 5 min of continuous stirring, the flask was put in an oven at 39°C for 5 h to completely stop the bacterial activity (Allison et al., 1962), resulting in an enzymatic solution.

In 200-mL tubes, 0.1 mg of one oxidative product (except in control tubes) and 1 mg of one of the FA were added, investigating all possible combinations: no oxidative product or 1 of 4 oxidative products \times 4 FA. To precisely dose the products at such low quantities, HEX, T2T4D, and FA were first dissolved in hexane and HPOD was dissolved in ethanol. Once the appropriate volume was introduced in tube, the solvent was eliminated at 50°C under nitrogen. Thereafter, 1 mL of enzymatic solution and 1 mL of bicarbonate buffer (pH = 7; Na₂HPO₄·12H₂O 19.5 g/L and NaHCO₃ 9.24 g/L) was added to each tube. The tubes were filled with CO₂ and a magnetic stirrer was added to each tube before closing them firmly. Thereafter, these tubes were incubated and stirred for 1 h in a water bath at 39°C. Moreover, for each of the 4 FA used in this study, one tube was frozen without incubation to measure the initial status of the medium with enzymatic solution. After incubation, the tubes were frozen at -20°C and freeze-dried.

Chemical Analyses

FA Analysis. The FA of feed and in vitro substrate ingredients, TGLA, HTGLA, FLA, HFLA, and 13HPOD, control flasks, and incubated media were extracted and methylated in situ using the procedure of Park and Goins (1994), except that the solution of 14% boron trifluoride in methanol was replaced by a solution of methanol-acetylchloride (10:1). Nonadeca-

noic acid (C19:0) was used as the internal standard at a dose of 0.8 mg. The FA methyl esters were then quantified by gas chromatography (Agilent 6890N, Network GC System, equipped with a model 7683 auto injector, Agilent Technologies, Palo Alto, CA) using a fused-silica capillary column (100 m \times 0.25 mm i.d., 0.20- μ m film thickness, CPSil88, Varian, Middelburg, the Netherlands), as described in Zened et al. (2011). Peaks were identified and quantified by comparison with commercial standards (Sigma Co.), except C18:1 (other than *trans*-9-C18:1, *trans*-11-C18:1, and *cis*-9-C18:1), which were identified by order of elution. Chromatograms were integrated using the Peak Simple software (Peak Simple Data System, version 3.29, SRI, Torrance, CA).

Lipid Oxidation Product Analysis. The analyses were performed as described by Kaleem et al. (2013). For the TAG polymer assay, the NF T60-268 official method was used (AFNOR, 2000), except for the detector. The analysis was performed by HPLC with an P680 HPLC pump (Dionex, Voisins le Bretonneux, France) coupled with an evaporative light-scattering detector (PL-ELS 2100, Polymer Laboratories, Church Stretton, UK) equipped with a size-exclusion PLgel column (particle size 5 μ m, pore size 300 Å, 300 \times 7.5 mm i.d., Polymer Laboratories). The mobile phase was tetrahydrofuran. The injected volume was 20 μ L, the column oven temperature was set at 30°C, and the mobile phase flow rate was 1 mL/min. The conditions for evaporative light scattering detection were 1 mL/min for nitrogen flow rate, 50°C for nebulization temperature, and 75°C for evaporation temperature.

Analysis of hydroperoxides and hydroxyacids was performed according to the HPLC method of Browne and Armstrong (2000), after saponification of lipids for TGLA and HTGLA: 100 mg of sample plus 2 mL of 1 M potassium hydroxide (prepared in 95% ethanol) were kept for 16 h under continuous stirring at 20°C. Then, 5 mL of ultrapure water and 10 mL of hexane were added and homogenized by vigorous manual stirring. The upper organic phase was eliminated as it contained the nonsaponified molecules, and the alcoholic phase was acidified using 5 mL of 6 N HCl. Afterward, this phase was washed 3 times with 10 mL of hexane and filtered with a phase separator silicone-treated filter paper (1PS, Whatman Inc., Buckinghamshire, UK). Hexane was evaporated under vacuum at 35 to 40°C. The residue was weighed and diluted with ethanol to obtain 10 mg of residue per mL of ethanol; FLA, HFLA, and 13HPOD were diluted directly in ethanol in the same proportions. The analysis was performed on an automated HPLC-UV system using a P680 HPLC pump and diode array detection at 320 nm (Agilent Technologies). The column used for separation was an

Table 1. Oligonucleotide sequences used for PCR capillary electrophoresis single-strand conformation polymorphism (CE-SSCP) and real-time PCR

Group	Direction	Position	Oligonucleotide sequence (5'-3')	Reference
PCR CE-SSCP Bacterial fingerprint	Forward	329-348	ACGGTCCAGACTCCTACGGG	Delbès et al. (2000)
	Reverse	515-532	FAM-TTACCGCGGCTGCTGGCAG	Zumstein et al. (2000)
Real-time PCR All Bacteria	Forward	338-357	ACTCCTACGGGAGGCAGCAG	Fierer et al. (2005)
	Reverse	518-534	ATTACCGCGGCTGCTGG	Fierer et al. (2005)

Uptisphere C18 ODB (particle size 3 µm, 15 × 0.46 cm i.d., Interchim, Montluçon, France). The mobile phase was a mixture of 1 g/L acetic acid, acetonitrile, and tetrahydrofuran (52:30:18, vol/vol/vol). The injected volume was 20 µL of the ethanol solution. The oven temperature of the column was set at 40°C. The mobile phase flow rate was 1 mL/min. Peaks were identified and quantified by comparison with commercial standards (Interchim), focusing on products from *cis-9,cis-12-C18:2* oxidation, 13HPOD, and 9HPOD, and their respective hydroxyacids (**13HOA** and **9HOA**).

Aldehyde analysis was performed according to the method of Li et al. (2005): 75 mg of extracted oil was mixed with 10 mL of hexane, and then 1 µL was injected into a gas chromatograph (Agilent 7890A, GC System, equipped with a model 7683 auto injector, Agilent Technologies) using a capillary column (30 m × 0.53 mm i.d., 0.5-µm film thickness, DB-5, Agilent Technologies). For analysis, the flame-ionization detector temperature was maintained at 280°C and the injector at 280°C; the split ratio was 1:5. Nitrogen was used as the carrier gas with a constant flow of 4 mL/min. The initial oven temperature was 60°C, and was increased by 2°C/min to 70°C, then increased by 10°C/min to 250°C and held for 1 min. Peaks were identified and quantified by comparison with commercial standards (Sigma Co.).

Chromatograms for TAG polymers, hydroperoxides or hydroxyacids, and aldehydes were integrated using the Chromeleon chromatography data system software (version 6.60, Dionex).

Microbiological Analyses

DNA Extraction. Total DNA from 1 mL of freeze-dried ruminal sample was extracted and purified using the QIAamps DNA Stool Mini kit (Qiagen Ltd., West Sussex, UK) according to the manufacturer's documentation with a previous bead-beating step in a FastPrep Instrument (MP Biomedicals, Illkirch, France).

Total Bacterial Community Abundance. Total numbers of bacteria in the cultures was estimated by quantitative PCR, as described in Combes et al. (2011). Assays were performed using the ABI Prism 7900HT sequence detection system (Applied Biosystems, Carls-

bad, CA) in optical-grade 384-well plates in a final volume of 10 µL. The SYBR Green assay reaction mixture contained template DNA, a specific primer set at 100 nM (Table 1), and 1× Power SYBR Green PCR Master Mix (Applied Biosystems). A dissociation curve was added to SYBR Green assays to check the specificity of the amplification. Then, the results were compared with a standard curve to obtain the number of target copies in the sample. The standard DNA curves were generated by amplification of the serial 10-fold dilutions of a reference plasmid containing the target 16S rRNA gene (accession no. EF445235 *Prevotella bryantii*).

Structure and Diversity of the Bacterial Community. Bacterial community analysis was performed as described by Michelland et al. (2009). The V3 region of the 16S rRNA genes of bacterial species, corresponding to a 205-bp fragment in *Escherichia coli*, was used as a diversity marker by performing PCR (Table 1). Bacterial communities were spread out in about 1,700 scans. Capillary electrophoresis single-strand conformation polymorphism (**CE-SSCP**) profiles were aligned and normalized and the diversity index was estimated using StatFingerprints program version 1.3 (Michelland et al., 2009). The Simpson index, which is an estimate of bacterial population diversity, was calculated according to Michelland et al. (2009) and the formula $-\log_{10}\sum(ai)^2$, where ai is the relative area under the i th peak detected in the CE-SSCP profile (Rosenzweig, 1995).

Calculations. The FA data were expressed as quantities of FA produced or disappeared, to directly assess BH capacity of bacteria. For each FA, production or disappearance was calculated as amount in incubated flask – (amount in blank + amount from added fat).

The rates (**v**) and efficiencies (**E**) of the *cis9,cis12-C18:2* isomerization (v1, E1), CLA reduction (v2, E2), and *trans-C18:1* reduction (v3, E3) were calculated as previously described by Troegeler-Meynadier et al. (2006a):

$$v1 = ([C18:2]b - [C18:2]e)/t,$$

$$E1 = ([C18:2]b - [C18:2]e)/[C18:2]b,$$

where [C18:2]b and [C18:2]e represent the concentrations (mg/L) of *cis-9,cis-12-C18:2* at the beginning

and at the end of incubation, respectively, and t is the incubation time;

$$v_2 = ([C18:2]_b - [C18:2]_e + [CLA]_b - [CLA]_e)/t,$$

$$E_2 = ([C18:2]_b - [C18:2]_e + [CLA]_b - [CLA]_e) / ([C18:2]_b - [C18:2]_e + [CLA]_b),$$

where $[CLA]_b$ and $[CLA]_e$ represent the concentration (mg/L) of total CLA isomers measured at the beginning and at the end of the incubation, respectively; and

$$v_3 = ([C18:2]_b - [C18:2]_e + [CLA]_b - [CLA]_e + [trans-C18:1]_b - [trans-C18:1]_e)/t,$$

$$E_3 = ([C18:2]_b - [C18:2]_e + [CLA]_b - [CLA]_e + [trans-C18:1]_b - [trans-C18:1]_e) / ([C18:2]_b - [C18:2]_e + [CLA]_b - [CLA]_e + [trans-C18:1]_b),$$

where $[trans-C18:1]_b$ and $[trans-C18:1]_e$ represent the concentration (mg/L) of total *trans*-C18:1 isomers measured at the beginning and at the end of the incubation.

Data Analysis. Fatty acid production or disappearance, rates and efficiencies of the BH reactions, and quantitative PCR and diversity indices of the bacterial community were analyzed using the general linear model of SYSTAT (version 9, SPSS Inc., Chicago, IL) and are reported as mean values with standard errors of the mean. Several models were used.

Experiment 1 had 2 objectives: testing the effects of oxypolymers on BH of *cis*-9,*cis*-12-C18:2 as TAG, and testing the effects of oxidative products on BH of free *cis*-9,*cis*-12-C18:2, so we used different statistical analyses:

Model 1 analyzed the effects of TAG polymers on BH in experiment 1:

$$\begin{aligned} \text{variable} = & \text{mean} + \text{“day of incubation” effect} \\ & + \text{“duration of incubation” effect} + \text{“heating” effect} \\ & + \text{interaction “heating} \times \text{incubation duration”} + \epsilon, \quad [1] \end{aligned}$$

where ϵ is the residual error.

Model 2 compared the effects of FLA, HFLA, 13HPOD, HEX, and T2D on BH in experiment 1, analyzing 6- and 24-h incubations separately:

$$\begin{aligned} \text{variable} = & \text{mean} + \text{“day of incubation” effect} \\ & + \text{“treatment” effect} + \epsilon. \quad [2] \end{aligned}$$

Model 3 compared the effects of TGLA, HTGLA, FLA, HFLA, 13HPOD, HEX, and T2D on bacterial number and diversity in experiment 1:

$$\begin{aligned} \text{variable} = & \text{mean} + \text{“day of incubation” effect} \\ & + \text{“treatment” effect} + \text{“duration of incubation”} \\ & \text{effect} + \text{interaction “treatment} \\ & \times \text{incubation duration”} + \epsilon. \quad [3] \end{aligned}$$

In experiment 2, the effects of the nature of aldehyde on BH and on bacterial number and diversity were analyzed with the model

$$\begin{aligned} \text{variable} = & \text{mean} + \text{“day of incubation” effect} \\ & + \text{“nature of aldehyde” effect} + \epsilon. \end{aligned}$$

In experiment 3, the effect of the hydroperoxides (13HPOD and 9HPOD) and aldehydes (HEX and T2T4D) on enzymatic BH reactions was analyzed with the model

$$\begin{aligned} \text{variable} = & \text{mean} + \text{“day of incubation” effect} \\ & + \text{“oxidative products or none” effect} + \epsilon. \end{aligned}$$

When the treatment effects were significant, a Tukey test was used for pairwise comparisons.

In experiment 2, linear correlations between individual FA balances and chain length and degree of unsaturation of aldehydes were computed.

To compare the structure of the bacterial communities, the pairwise Euclidean distances of the CE-SSCP profiles were used. Analysis of similarity (ANOSIM) was performed on the distance matrix using 10,000 Monte Carlo permutations. Global ANOSIM was performed to test the fixed effects of treatment (all data except TGLA and HTGLA, experiment 1) or of aldehydes (experiment 2). The ANOSIM R-value indicates the extent to which the groups differed and lies between 0 (no separation) and 1 (well-separated groups). When a fixed effect was significant, an iterative Mann-Whitney test was carried out on the 1,700 scans of the CE-SSCP profiles to estimate the percentage of the population that differed between 2 factors. Significance was set at $P \leq 0.05$.

RESULTS

For experiments 1 and 2, pH was not affected by addition of oxidative products. In experiment 1, initial pH was 7.0, and final pH was 6.8 and 6.6 after 6- and

Table 2. Quantities of *cis-9,cis-12-C18:2* and oxidative products added to the different cultures (per flask, experiment 1)

Item	Culture ¹						
	TGLA	HTGLA	FLA	HFLA	13HPOD	HEX	T2D
<i>cis-9,cis-12-C18:2</i> (mg)	49	48	48	45	50	50	50
Triacylglycerol polymers (mg)	ND ²	13	NA ²	NA	NA	NA	NA
Hydroperoxides and hydroxyacids (μg)							
13HPOD	ND	ND	ND	805	3,904	NA	NA
9HPOD ³	ND	ND	ND	345	346	NA	NA
13HOA ⁴	26	515	ND	105	19	NA	NA
9HOA ⁵	10	224	ND	55	6	NA	NA
Aldehydes (μg)							
Hexanal	2	24	3	6	NA	5,000	—
<i>trans-2-decenal</i>	ND	0.3	ND	ND	NA	NA	5,000
<i>trans-2,trans-4-decadienal</i>	8	73	5	52	NA	NA	NA

¹TGLA = pure triacylglycerol of *cis-9,cis-12-C18:2*; HTGLA = heated TGLA (150°C for 6 h); FLA = pure *cis-9,cis-12-C18:2*; HFLA = heated FLA (150°C for 6 h); 13HPOD = 13-OOH *cis-9,trans-11-C18:2*; HEX = pure hexanal; T2D = pure *trans-2-decenal*.

²ND = not detected; NA = not analyzed.

³9HPOD = 9-OOH *trans-10,cis-12-C18:2*. ⁴13HOA = 13-OH *cis-9,trans-11-C18:2*.

⁵9HOA = 9-OH *trans-10,cis-12-C18:2*.

24-h incubations, respectively; in experiment 2, pH was 7.0 and 6.8 after 6- and 24-h incubations, respectively.

Oxidative Products of Heated Fat Sources

The thermal treatment of TGLA was sufficient to generate TAG polymers, which represented 27% of added fat in HTGLA incubations (Table 2). It produced small amounts of aldehydes and large amounts of hydroxyacids, but hydroperoxides were not detectable. Conversely, heating FLA resulted in a high production of hydroperoxides and a low production of hydroxyacids and aldehydes. Among hydroperoxides and hydroxyacids, 13OOH or 13OH isomers were the predominant forms. The main aldehyde obtained by heating either TGLA or FLA was T2T4D.

Effect of TAG on BH

Heating TAG had no effect on *cis-9,cis-12-C18:2* disappearance regardless of incubation duration; therefore, the 2 incubation durations were combined (Figure 1). Nevertheless, some differences were observed on BH intermediates: heating of TAG significantly ($P < 0.01$) decreased the production of CLA, *trans-C18:1*, and C18:0 (Figure 2). This decrease was less important for *trans-10* isomers (−41%) than for *trans-11* isomers (−86%).

Comparative Effects of *cis-9,cis-12-C18:2* Heating and Addition of Hydroperoxide or Aldehydes on BH

In experiment 1, compared with FLA, HFLA led to a decrease of *cis-9,cis-12-C18:2* disappearance and of CLA and *trans-C18:1* production, regardless of incuba-

tion duration, and a decrease in C18:0 production for 24-h incubations (Tables 3 and 4). The BH intermediate isomers whose production was significantly decreased by heating were *trans-11-C18:1* and *cis-9,trans-11-CLA*. Conversely, 13HPOD, HEX, and T2D treatments did not affect *cis-9,cis-12-C18:2* disappearance or production of total *trans-C18:1* or CLA. Nevertheless, some differences were observed in distribution of BH intermediate isomers. In particular, 13HPOD resulted in the highest *trans-10-C18:1* production for 6-h incubations, and, as observed with HFLA, decreased *trans-11* isomer production during 24-h incubations. Both aldehydes, HEX and T2D, had little effect on the production of BH products, but T2D resulted in significantly higher

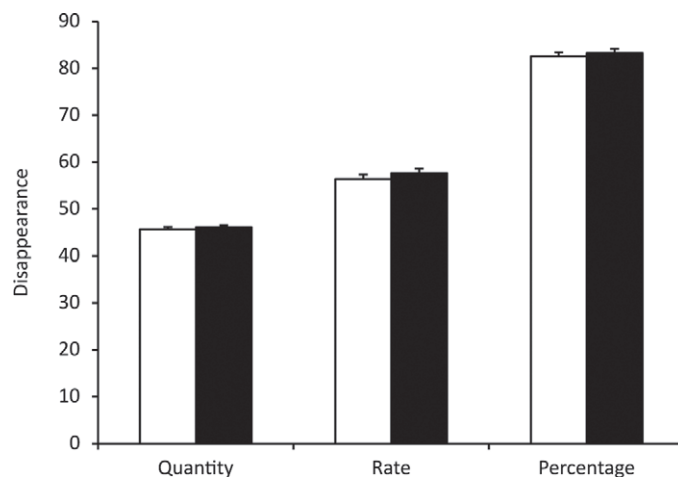


Figure 1. Quantity (mg/flask), rate (mg/L per hour), and percentage (%) of *cis-9,cis-12-C18:2* disappearance when incubated with rumen fluid as heated (■) or nonheated (□) triacylglycerols, combining values from 6- and 24-h incubations. Error bars = SEM.

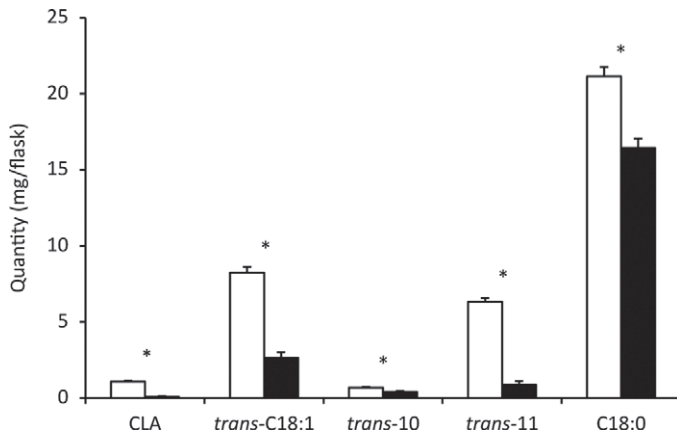


Figure 2. Quantities (mg/flask) of total conjugated linoleic acid (CLA), total *trans*-C18:1, *trans*-10 isomers (*trans*-10,*cis*-12-CLA + *trans*-10-C18:1), *trans*-11 isomers (*cis*-9,*trans*-11-CLA + *trans*-9,*trans*-11-CLA + *trans*-11-C18:1), and C18:0 produced from *cis*-9,*cis*-12-C18:2 when incubated with rumen fluid as heated (■) or nonheated (□) triacylglycerols, combining values from 6- and 24-h incubations. Error bars = SEM; **P* < 0.01.

trans-10 isomer production than did HEX for 6-h incubations. Moreover, 13HPOD increased *cis*-12-C18:1 production in 6- and 24-h incubations and T2D only in 6-h incubations compared with FLA, with HFLA and HEX resulting in intermediate values.

Compared with FLA, HFLA decreased the rate and efficiency of isomerization, regardless of incubation duration (Tables 3 and 4). Rates of both reductions were also significantly decreased by heating, but their efficiencies were not affected. Other treatments (13HPOD, HEX, and T2D) did not modify BH kinetic parameters compared with FLA.

Effects of Aldehyde Chain Length and Degree of Unsaturation on BH

In experiment 2, the different assayed aldehydes resulted in similar disappearances of *cis*-9,*cis*-12-C18:2 (Table 5). No effect was noticed on *trans*-C18:1 produc-

Table 3. Comparison of heating linoleic acid or adding oxidation products on production (positive values), disappearance (negative values) of FA (mg/flask), rates (v, mg/L per hour), and efficiencies (E, %) of the 3 reactions of *cis*-9,*cis*-12-C18:2 biohydrogenation after a 6-h incubation with ruminal fluid, focusing on its biohydrogenation products (experiment 1)

Item	Culture ¹					SEM	<i>P</i> -value
	FLA	HFLA	13HPOD	HEX	T2D		
C18:0	13.62	11.57	12.47	12.80	12.65	0.514	0.13
<i>cis</i> 12-C18:1	-0.04 ^b	0.14 ^{ab}	0.48 ^a	0.14 ^{ab}	0.41 ^a	0.101	<0.01
<i>trans</i> -C18:1 ²	5.54 ^a	3.63 ^b	5.44 ^a	4.92 ^{ab}	6.15 ^a	0.320	<0.01
<i>trans</i> -4-C18:1	0.02	0.01	0.04	0.03	0.04	0.011	0.24
<i>trans</i> -5-C18:1	0.03	0.02	0.02	0.03	0.03	0.008	0.92
<i>trans</i> -6+7+8-C18:1	0.29 ^{ab}	0.23 ^b	0.33 ^a	0.25 ^{ab}	0.32 ^a	0.020	<0.01
<i>trans</i> -9-C18:1	0.13 ^{ab}	0.13 ^{ab}	0.18 ^a	0.12 ^b	0.16 ^{ab}	0.012	0.01
<i>trans</i> -10-C18:1	0.65 ^{bc}	0.57 ^{bc}	0.84 ^a	0.55 ^c	0.73 ^b	0.039	<0.01
<i>trans</i> -11-C18:1	3.23 ^a	1.55 ^b	2.28 ^a	2.87 ^a	3.20 ^a	0.302	<0.01
<i>trans</i> -12-C18:1	0.33 ^b	0.34 ^{ab}	0.47 ^a	0.29 ^b	0.46 ^a	0.032	<0.01
<i>trans</i> -13+14-C18:1	0.49 ^b	0.53 ^{ab}	0.77 ^a	0.46 ^b	0.74 ^a	0.059	<0.01
<i>trans</i> -15-C18:1	0.14 ^b	0.18 ^{ab}	0.25 ^a	0.14 ^b	0.23 ^a	0.020	<0.01
<i>trans</i> -16-C18:1	0.21	0.07	0.26	0.19	0.23	0.055	0.19
<i>cis</i> -9, <i>cis</i> -12-C18:2	-48.81 ^b	-44.16 ^a	-48.73 ^b	-48.72 ^b	-48.50 ^b	0.239	<0.01
CLA ²	0.40 ^a	0.18 ^b	0.30 ^{ab}	0.38 ^a	0.35 ^{ab}	0.045	0.02
<i>trans</i> -10, <i>cis</i> -12-CLA	0.03	0.02	0.04	0.03	0.05	0.006	0.03
<i>cis</i> -9, <i>cis</i> -11-CLA	0.00 ^{ab}	0.02 ^a	0.00 ^b	0.00 ^b	0.00 ^{ab}	0.004	0.01
<i>cis</i> -9, <i>trans</i> -11-CLA	0.24	0.08	0.16	0.24	0.21	0.040	0.04
<i>trans</i> -9, <i>trans</i> -11-CLA	0.13 ^a	0.07 ^b	0.09 ^{ab}	0.10 ^{ab}	0.09 ^{ab}	0.012	0.05
<i>trans</i> -10 isomers ²	0.68 ^{bc}	0.59 ^c	0.89 ^a	0.58 ^c	0.77 ^{ab}	0.040	<0.01
<i>trans</i> -11 isomers ²	3.60 ^a	1.69 ^b	2.54 ^{ab}	3.22 ^a	3.50 ^a	0.336	<0.01
v1 ³	101.69 ^a	92.01 ^b	101.52 ^a	101.51 ^a	101.05 ^a	0.498	<0.01
E1 ³	0.89 ^a	0.85 ^b	0.89 ^a	0.89 ^a	0.88 ^a	0.004	<0.01
v2 ³	100.84 ^a	91.63 ^b	100.89 ^a	100.72 ^a	100.32 ^a	0.499	<0.01
E2 ³	0.99	0.99	0.99	0.99	0.99	0.001	0.08
v3 ³	89.31 ^a	84.07 ^b	89.55 ^a	90.47 ^a	87.51 ^{ab}	0.889	<0.01
E3 ³	0.78 ^{ab}	0.80 ^a	0.78 ^{ab}	0.79 ^{ab}	0.77 ^b	0.006	0.02

^{a-c}Means in the same row with different superscripts differ significantly (*P* < 0.05).

¹FLA = pure *cis*-9,*cis*-12-C18:2; HFLA = heated FLA (150°C for 6 h); 13HPOD = 13-OOH *cis*-9,*trans*-11-C18:2; HEX = pure hexanal; T2D = pure *trans*-2-decenal.

²*trans*-C18:1 and conjugated linoleic acid (CLA) are the sums of the respective assayed isomers; *trans*-10 isomers = *trans*-10,*cis*-12-CLA + *trans*-10-C18:1; *trans*-11 isomers = *cis*-9,*trans*-11-CLA + *trans*-9,*trans*-11-CLA + *trans*-11-C18:1.

³v1 and E1 refer to the isomerization of *cis*-9,*cis*-12-C18:2 to CLA; v2 and E2 refer to the reduction of CLA to *trans*-C18:1 isomers; v3 and E3 refer to the reduction of *trans*-C18:1 to C18:0.

Table 4. Comparison of heating linoleic acid or adding oxidation products on production (positive values), disappearance (negative values) of FA (mg/flask), rates (v, mg/L per hour), and efficiencies (E, %) of the 3 reactions of *cis-9,cis-12-C18:2* biohydrogenation after a 24-h incubation with ruminal fluid, focusing on its biohydrogenation products (experiment 1)

Item	Culture ¹					SEM	P-value
	FLA	HFLA	13HPOD	HEX	T2D		
C18:0	25.49 ^a	19.28 ^b	26.68 ^a	27.30 ^a	25.11 ^a	0.985	<0.01
<i>cis-12-C18:1</i>	0.09 ^b	0.15 ^{ab}	0.32 ^a	0.12 ^b	0.13 ^b	0.044	<0.01
<i>trans-C18:1</i> ²	6.08 ^a	3.60 ^b	5.08 ^{ab}	5.16 ^{ab}	5.13 ^{ab}	0.536	0.04
<i>trans-4-C18:1</i>	0.03 ^{ab}	0.03 ^b	0.05 ^a	0.03 ^b	0.03 ^b	0.004	<0.01
<i>trans-5-C18:1</i>	0.02	0.02	0.03	0.01	0.02	0.007	0.43
<i>trans-6+7+8-C18:1</i>	0.25 ^{ab}	0.19 ^b	0.27 ^a	0.21 ^{ab}	0.22 ^{ab}	0.018	0.03
<i>trans-9-C18:1</i>	0.13	0.12	0.15	0.10	0.13	0.013	0.13
<i>trans-10-C18:1</i>	0.66	0.53	0.72	0.54	0.53	0.068	0.11
<i>trans-11-C18:1</i>	3.77 ^a	1.45 ^b	2.2 ^{9b}	3.09 ^a	2.84 ^{ab}	0.374	<0.01
<i>trans-12-C18:1</i>	0.31	0.31	0.39	0.28	0.34	0.031	0.11
<i>trans-13+14-C18:1</i>	0.52 ^{ab}	0.56 ^{ab}	0.71 ^a	0.51 ^b	0.59 ^{ab}	0.051	0.04
<i>trans-15-C18:1</i>	0.18	0.20	0.24	0.17	0.21	0.024	0.21
<i>trans-16-C18:1</i>	0.21	0.21	0.24	0.22	0.24	0.019	0.63
<i>cis-9,cis-12-C18:2</i>	-51.43 ^b	-46.71 ^a	-51.05 ^b	-51.39 ^b	-51.55 ^b	0.371	<0.01
CLA ²	0.77 ^a	0.05 ^b	0.35 ^{ab}	0.58 ^{ab}	0.36 ^{ab}	0.160	0.02
<i>trans-10,cis-12-CLA</i>	0.04	0.01	0.03	0.04	0.04	0.009	0.08
<i>cis-9,cis-11-CLA</i>	0.03	0.00	0.00	0.01	0.00	0.008	0.09
<i>cis-9,trans-11-CLA</i>	0.41 ^a	-0.03 ^b	0.17 ^{ab}	0.36 ^a	0.20 ^{ab}	0.100	0.02
<i>trans-9,trans-11-CLA</i>	0.30	0.07	0.15	0.19	0.12	0.067	0.15
<i>trans-10 isomers</i> ²	0.70	0.53	0.75	0.58	0.56	0.071	0.10
<i>trans-11 isomers</i> ²	4.47 ^a	1.49 ^d	2.61 ^{bcd}	3.64 ^{ab}	3.17 ^{abc}	0.420	<0.01
v1 ³	26.79 ^a	24.33 ^b	26.59 ^a	26.76 ^a	26.85 ^a	0.193	<0.01
E1 ³	0.94 ^b	0.90 ^a	0.93 ^a	0.93 ^a	0.94 ^a	0.006	0.00
v2 ³	26.39 ^a	24.30 ^b	26.40 ^a	26.46 ^a	26.66 ^a	0.209	<0.01
E2 ³	0.98 ^b	0.99 ^a	0.99 ^{ab}	0.98 ^{ab}	0.99 ^{ab}	0.003	0.03
v3 ³	23.22 ^{ab}	22.42 ^b	23.76 ^{ab}	23.77 ^{ab}	23.99 ^a	0.330	0.02
E3 ³	0.78	0.81	0.80	0.80	0.80	0.009	0.22

^{a-d}Means in the same row with different superscripts differ significantly ($P < 0.05$).

¹FLA = pure *cis-9,cis-12-C18:2*; HFLA = heated FLA (150°C for 6 h); 13HPOD = 13-OOH *cis-9,trans-11-C18:2*; HEX = pure hexanal; T2D = pure *trans-2-decenal*.

²*trans-C18:1* and conjugated linoleic acid (CLA) are the sums of the respective assayed isomers; *trans-10 isomers* = *trans-10,cis-12-CLA* + *trans-10-C18:1*; *trans-11 isomers* = *cis-9,trans-11-CLA* + *trans-9,trans-11-CLA* + *trans-11-C18:1*.

³v1 and E1 refer to the isomerization of *cis-9,cis-12-C18:2* to CLA; v2 and E2 refer to the reduction of CLA to *trans-C18:1* isomers; v3 and E3 refer to the reduction of *trans-C18:1* to C18:0.

tion, but CLA production was significantly decreased by NON. The highest value for *trans-10,cis-12-CLA* and *trans-10-C18:1* production was obtained with HEP, and the highest value for *cis-9,trans-11-CLA* and *trans-11-C18:1* production was obtained with T2D and T2T4D. However, differences were small and not always significant. We observed no significant relationship between individual FA balances and chain length or degree of unsaturation of aldehydes.

Effects of Fat Heating and Addition of Aldehydes or Hydroperoxides on Ruminal Bacterial Population

In experiments 1 and 2, fat heating and addition of oxidative products had no effect on ruminal bacterial density or diversity, resulting in values similar to those of blanks (Table 6). The bacterial community assessed by CE-SSCP was affected by 13HPOD addition to cultures in experiment 1 (Table 7), compared with FLA and HLA, with HEX and T2D being intermediate. It-

erative tests showed that differences were large for 6-h incubations but not for 24-h incubations; for example, the percentage difference was 27% between a bacterial community obtained from ruminal cultures with FLA and 13HPOD ($P < 0.05$) for 6-h incubations and only 2% ($P < 0.05$) for 24-h incubations. In experiment 2, ruminal bacterial density and diversity did not differ between aldehydes, and values obtained with aldehydes were similar to those obtained with blanks (Table 6). The bacterial community assessed by CE-SSCP did not differ according to aldehydes ($P > 0.05$; results not shown).

Effects of Aldehydes and Hydroperoxides on Enzymatic BH Reactions

In experiment 3, BH was almost complete as shown by C18:0 being the major BH product. Nevertheless, some effects of added products could be noticed on isomerization (Table 8): Δ^9 -isomerization (producing

Table 5. Comparison of effects of different aldehydes on production (positive values) or disappearance (negative values) of FA (mg/flask) after a 5-h incubation with ruminal fluid, focusing on *cis*-9,*cis*-12-C18:2 biohydrogenation products (experiment 2)

Item	Culture ¹						SEM	<i>P</i> -value
	PEN	HEX	HEP	NON	T2D	T2T4D		
C18:0	37.07	36.13	36.54	35.15	35.62	34.41	0.792	0.15
<i>cis</i> -12-C18:1	0.81	0.81	0.85	0.77	0.83	0.78	0.026	0.16
<i>trans</i> -C18:1 ²	12.31	12.12	11.98	11.73	12.54	12.35	0.320	0.47
<i>trans</i> -4-C18:1	0.04	0.04	0.04	0.04	0.04	0.04	0.002	1.00
<i>trans</i> -5-C18:1	0.03 ^b	0.03 ^{ab}	0.03 ^{ab}	0.03 ^{ab}	0.04 ^{ab}	0.04 ^a	0.003	0.02
<i>trans</i> -6+7+8-C18:1	0.37	0.37	0.38	0.36	0.37	0.37	0.016	0.85
<i>trans</i> -9-C18:1	0.22	0.22	0.22	0.19	0.22	0.20	0.019	0.68
<i>trans</i> -10-C18:1	0.68 ^{ab}	0.70 ^a	0.72 ^a	0.61 ^{ab}	0.64 ^{ab}	0.57 ^b	0.033	0.01
<i>trans</i> -11-C18:1	5.49	5.39	5.08	5.14	5.64	5.62	0.158	0.04
<i>trans</i> -12-C18:1	0.96	0.99	0.99	0.95	0.97	1.02	0.054	0.93
<i>trans</i> -13+14-C18:1	1.31	1.31	1.34	1.37	1.37	1.42	0.037	0.26
<i>trans</i> -15-C18:1	2.72	2.66	2.68	2.55	2.71	2.54	0.064	0.17
<i>trans</i> -16-C18:1	0.49	0.41	0.50	0.51	0.54	0.53	0.034	0.11
<i>cis</i> -9, <i>cis</i> -12-C18:2	-7.66	-7.86	-7.66	-7.34	-7.04	-6.76	0.308	0.07
CLA ²	1.56 ^a	1.55 ^a	1.35 ^{ab}	1.13 ^b	1.56 ^a	1.57 ^a	0.098	<0.01
<i>trans</i> -10, <i>cis</i> -12-CLA	0.08 ^{ab}	0.09 ^{ab}	0.09 ^a	0.05 ^c	0.06 ^{bc}	0.05 ^c	0.007	<0.01
<i>cis</i> -9, <i>cis</i> -11-CLA	0.03	0.03	0.03	0.02	0.02	0.03	0.005	0.35
<i>cis</i> -9, <i>trans</i> -11-CLA	1.30 ^{ab}	1.30 ^a	1.10 ^{ab}	0.95 ^b	1.36 ^a	1.37 ^a	0.090	<0.01
<i>trans</i> -9, <i>trans</i> -11-CLA	0.15	0.13	0.13	0.11	0.12	0.12	0.013	0.28
<i>trans</i> -10 isomers ²	0.76 ^{ab}	0.78 ^{ab}	0.81 ^a	0.66 ^{bc}	0.70 ^{abc}	0.62 ^c	0.035	<0.01
<i>trans</i> -11 isomers ²	6.93 ^{ab}	6.82 ^{ab}	6.31 ^{ab}	6.20 ^b	7.12 ^a	7.11 ^a	0.209	<0.01

^{a-c}Means in the same row with different superscripts differ significantly ($P < 0.05$).

¹PEN = pure pentanal; HEX = pure hexanal; HEP = pure heptanal; NON = pure nonanal; T2D = pure *trans*-2-decenal; T2T4D = pure *trans*-2,*trans*-4-decadienal.

²*trans*-C18:1 and conjugated linoleic acid (CLA) are the sums of the respective assayed isomers; *trans*-10 isomers = *trans*-10,*cis*-12-CLA + *trans*-10-C18:1; *trans*-11 isomers = *cis*-9,*trans*-11-CLA + *trans*-9,*trans*-11-CLA + *trans*-11-C18:1.

trans-10 isomers) was inhibited by 9HPOD, and Δ^{12} -isomerization (producing *trans*-11 isomers) was inhibited by both 9HPOD and 13HPOD. The first reduction was significantly but only slightly altered by oxidative products. The second reduction was not affected.

DISCUSSION

Polymers of TAG did not affect *cis*-9,*cis*-12-C18:2 disappearance, which suggests that TAG polymers

have no effect on either lipolysis or *cis*-9,*cis*-12-C18:2 isomerization. This is consistent with the results of Tarvainen et al. (2010), who observed that oxidized TAG are hydrolyzed by the porcine lipase as efficiently as unoxidized molecules. Nevertheless, in HTGLA compared with TGLA incubations, a significant decrease in all BH products (CLA, *trans*-C18:1, and C18:0) was observed, suggesting that the increase of oxidative products (Table 2) could have shifted *cis*-9,*cis*-12-C18:2

Table 6. Effects of heating fat and adding oxidative products on bacterial density (total bacteria quantitative PCR, no. of copies per mL of culture) and diversity (Simpson index) of bacterial community in experiments 1 and 2

Item	Culture ¹							SEM	<i>P</i> -value	
	Nonincubated blank	TGLA	HTGLA	FLA	HFLA	13HPOD	HEX			
Experiment 1										
Log (no. of copies/mL of culture)	4.16	4.06	4.09	4.14	4.06	4.12	4.13	4.15	0.03	0.20
Simpson index	7.39	8.22	7.75	7.92	7.71	7.68	8.10	8.05	0.22	0.40
Experiment 2										
Log (no. of copies/mL of culture)	10.54	10.44	10.62	10.46	10.65	10.49	10.53		0.11	0.69
Simpson index	8.09	8.87	8.73	8.81	8.98	8.83	8.46		0.20	0.44

¹Nonincubated blanks reflect initial status and were not included in statistical analysis; TGLA = pure triacylglycerol of *cis*-9,*cis*-12-C18:2; HTGLA = heated TGLA (150°C for 6 h); FLA = pure *cis*-9,*cis*-12-C18:2; HFLA = heated FLA (150°C for 6 h); 13HPOD = 13-OOH *cis*-9,*trans*-11-C18:2; HEX = pure hexanal; T2D = pure *trans*-2-decenal; PEN = pure pentanal; HEP = pure heptanal; NON = pure nonanal; T2T4D = pure *trans*-2,*trans*-4-decadienal.

Table 7. Capillary electrophoresis single-strand conformation polymorphism similarity values (R-ANOSIM) between the different treatments used in experiment 1, combining 6- and 24-h incubations¹

	FLA	HFLA	13HPOD	HEX
HFLA	-0.138	—	—	—
13HPOD	0.615	0.888	—	—
HEX	-0.138	0.052	0.325	—
T2D	0.037	0.415	0.278	0.251

¹FLA = pure *cis-9,cis-12-C18:2*; HFLA = heated FLA (150°C for 6 h); 13HPOD = 13-OOH *cis-9,trans-11-C18:2*; HEX = pure hexanal; T2D = pure *trans-2*-decenal.

rumen metabolism toward another pathway of *cis-9,cis-12-C18:2*, such as hydration, which can be performed by *Streptococcus bovis* (Hudson et al., 1998). With TGLA, the decrease in *trans-11* isomers was greater than that of *trans-10* isomers, showing that the balance between the 2 main BH pathways was affected.

With HFLA, despite a slightly lower initial quantity of *cis-9,cis-12-C18:2* compared with other media, inhibition of *cis-9,cis-12-C18:2* disappearance was observed because isomerization efficiency was decreased ($P < 0.01$). This inhibition was observed only on Δ^{12} -isomerization because only *trans-11* isomers were decreased ($P < 0.01$). This great alteration of the *trans-11* BH pathway with heating has been reported previously (Vázquez-Añón et al., 2008; Privé et al., 2010). This inhibition of Δ^{12} -isomerization with HFLA could be linked to the presence of aldehydes or hydroperoxides.

Aldehydes are not important modulators of free *cis-9,cis-12-C18:2* BH or the ruminal bacterial community. Among them, T2D has been reported to increase BH (Lee et al., 2007), but the culture conditions in that experiment were different from ours; in particular, those

authors did not add *cis-9,cis-12-C18:2* to their cultures, so that their effects could be due to modifications of either lipolysis or BH, and if these effects were on BH reactions, they could have been less obvious in our experiment using larger amounts of *cis-9,cis-12-C18:2* as substrate. Lee et al. (2007) also observed a decrease in *trans-10* isomers and *cis-12-C18:1* and an increase in *trans-11* isomers with T2D compared with other oxidative products, in particular other aldehydes, which was the case in our experiment 2 but not in experiment 1. In our experiment 2, T2T4D also slightly increased *trans-11* isomers ($P < 0.01$). In fact, in our experiments, aldehydes had few and variable effects on BH so that no definite conclusions about their effects can be drawn. In a previous study in our laboratory (Kaleem et al., 2013), heating oilseeds decreased *cis-9,cis-12-C18:2* disappearance, and we proposed 2 possible modes of action: an antimicrobial effect of aldehydes on ruminal bacteria or the formation of Maillard complexes during heating. In the present study, no effect of aldehydes on the bacterial community was observed.

In the present study, the major BH modulators were hydroperoxides. In particular, in enzymatic solutions, 13HPOD and 9HPOD led to an inhibition of *trans-11* isomer production, and 9HPOD to an inhibition of production of both *trans-11* and *trans-10* isomers (Table 8). Moreover, 13HPOD in ruminal cultures led to an increase in *trans-10* isomer production ($P < 0.01$; Table 3). Such a modification of *trans* isomer production was reported by Privé et al. (2010), linked to increased peroxide values (PV). As in the present study, Privé et al. (2010) found large effects on isomerization. Hence, hydroperoxides could explain, at least in part, the ef-

Table 8. Effects of hydroperoxides and aldehydes on the production ($\mu\text{g}/\text{tube}$) of *cis-9,cis-12-C18:2* biohydrogenation intermediates and final products in chloramphenicol-treated ruminal fluid initially containing 1 mg of free *cis-9,cis-12-C18:2*, *cis-9,trans-11*-conjugated linoleic acid (CLA), *trans-10,cis-12-CLA*, or *trans-11-C18:1* (experiment 3)

Production	Culture ¹					SEM	P-value
	No oxidative product	9HPOD	13HPOD	HEX	T2T4D		
Incubation of <i>cis-9,cis-12-C18:2</i>							
<i>trans-10</i> isomers	69 ^{ab}	44 ^c	52 ^{bc}	80 ^a	62 ^{ab}	4	<0.01
<i>trans-11</i> isomers	84 ^a	57 ^b	55 ^b	92 ^a	85 ^a	3	<0.01
C18:0	401	407	402	425	403	16	0.79
Incubation of <i>cis-9,trans-11-CLA</i>							
<i>trans-11-C18:1</i>	60 ^{ab}	53 ^b	53 ^b	73 ^a	73 ^a	4	0.01
C18:0	396	408	411	423	410	12	0.65
Incubation of <i>trans-10,cis-12-CLA</i>							
<i>trans-10-C18:1</i>	59 ^a	50 ^b	50 ^b	53 ^b	52 ^b	1	<0.01
C18:0	433	426	427	413	427	8	0.53
Incubation of <i>trans-11-C18:1</i>							
C18:0	420	418	407	443	425	11	0.27

^{a-c}Means in the same row with different superscripts differ significantly ($P < 0.05$).

¹9HPOD = pure 9-OOH *trans-10,cis-12-C18:2*; 13HPOD = pure 13-OOH *cis-9,trans-11-C18:2*; HEX = pure hexanal; T2T4D = pure *trans-2,trans-4*-decadienal.

fects of fat heating. In HFLA incubations, with large amounts of both 9HPOD and 13HPOD, *trans*-11 isomer production was greatly inhibited compared with 13HPOD alone (Tables 3 and 4). The 2 hydroperoxide isomers, as prooxidants, probably directly interfered with Δ^{12} -isomerization because this reaction, like lipid oxidation, involves radicals (Wallace et al., 2007), so it could involve some intermediates chemically similar to 13HPOD. An effect of hydroperoxides on the Δ^{12} -isomerase would also be in agreement with the appearance and persistence of effects on *trans*-11 isomer production in 24-h incubations, because this enzyme is not recyclable (Kim et al., 2000). Indeed, as it cannot be reused, it becomes less and less available in the media. Moreover, HFLA did not affect the structure of bacterial community (Table 7) or its diversity and density (Table 6). However, an action via the bacterial population cannot be excluded because hydroperoxides have bacteriostatic effects that could not be assayed by our study, because inactivated bacterial DNA is detected as efficiently as DNA from active bacteria. Indeed, some peroxides (e.g., benzoyl peroxide) are used in medicine as potent oxidizing agents to treat skin infections, for example (Fanta et al., 1979). Furthermore, the CE-SSCP technique could have failed to reveal changes in the bacterial community affecting scarce bacteria; *Butyrivibrio fibrisolvens* and *Butyrivibrio proteoclasticus*, which are responsible for *trans*-11 isomerization, have low relative abundances in the rumen (Zened et al., 2013). Thus, hydroperoxides could partly explain the effects of highly oxidized fat on *trans*-11 isomers through inhibition of Δ^{12} -isomerization or a bacteriostatic effect on *trans*-11 isomer-producing bacteria. Moreover, as 13HPOD was not as efficient as HFLA in decreasing *trans*-11 isomer production, some other compounds might be involved, such as 9HPOD.

In our experiment 1, the increase in *trans*-10 isomer production was observed only in 13HPOD cultures with short incubations ($P < 0.01$; Table 3). Privé et al. (2010) also reported this increase of *trans*-10 isomers production when incubating rumen fluid for 6 h with a high PV oil (PV = 238 mEq of O₂/kg of oil corresponding to 75 mg of hydroperoxides with 18 carbons and 2 double bonds, which is similar to the 85 mg of total hydroperoxides per kg of added fat in our 13HPOD cultures). In HFLA incubations in the present experiment, there was probably insufficient 13HPOD to result in a measurable increase of *trans*-10 production (i.e., 25 mg of total hydroperoxides per kg of added fat). This evolution of *trans*-10 isomers was concomitant with a modification of the bacterial community by 13HPOD, in particular compared with FLA (R-ANOSIM = 0.62), whereas FLA and HFLA did not result in different communities

(R-ANOSIM = -0.14; Table 7). Similarly, Privé et al. (2010) observed a modification of the bacterial community without any change in its diversity when adding heated oil to ruminal cultures. Thus, 13HPOD could be partly responsible for the increase of *trans*-10 isomers observed with highly oxidized fat, shifting the bacterial community toward *trans*-10-producing bacteria. Such an effect did not persist in 24-h incubations because hydroperoxides are not stable molecules (Frankel, 2005), especially under the reducing conditions observed in the rumen, so that they were probably rapidly broken down in our incubations, showing a strong resilience of the rumen microbiota after a disturbance caused by oxidative products. On the other hand, 13HPOD did not significantly affect Δ^9 -isomerization, whereas 9HPOD decreased it (Table 8). Indeed, this reaction is quite different from Δ^{12} -isomerization and would have a carbocation chemically similar to *trans*-10,*cis*-12-CLA as an intermediate (Liavonchanka et al., 2006). This could explain why only 9HPOD could inhibit this reaction.

Hence, in HFLA, there was probably insufficient 13HPOD to modify the ruminal population and increase *trans*-10 isomer production, and an insufficient amount of 9HPOD to significantly decrease *trans*-10 production, but 13HPOD and 9HPOD together inhibited production of *trans*-11 isomers. In HTGLA cultures, BH was likely affected in a same way.

CONCLUSIONS

Polymers of TAG did not affect *cis*-9,*cis*-12-C18:2 disappearance. Heating FLA and TGLA greatly inhibited Δ^{12} -isomerization, leading to decreased *cis*-9,*cis*-12-C18:2 disappearance. This may have occurred because of the production of hydroperoxides and aldehydes during heat treatment. Neither bacterial community nor disappearance of *cis*-9,*cis*-12-C18:2 was affected by chain length or degree of unsaturation of aldehydes, and aldehydes only slightly affected BH intermediates. Among hydroperoxides, 9HPOD inhibited Δ^9 -isomerization and Δ^{12} -isomerization, whereas 13HPOD only inhibited Δ^{12} -isomerization but it also affected the structure of the bacterial community, resulting in increased production of *trans*-10 isomers. Thus, hydroperoxides could explain (1) the increased *trans*-10 isomer production observed with highly oxidized fat associated with a modification of the bacterial population due to 13HPOD; and (2) the decrease in *trans*-11 isomer production with heated oil, caused by the inhibiting effect of 13HPOD and 9HPOD on Δ^{12} -isomerization. Nevertheless, some other oxidative products might also be involved.

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