

1 Mapping the Global Flow of Tungsten to Identify Key
2 Material Efficiency and Supply Security Opportunities

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16 **ABSTRACT**

17 Tungsten is an economically important metal with diverse applications ranging from wear
18 resistant cutting tools to its use in specialized steels and alloys. Concerns about its supply security
19 have been raised by various studies in literature, mostly due to trade disputes arising from supply
20 concentration and exports restrictions in China and its lack of viable substitutes. Although tungsten
21 material flows have been analysed for specific regions, a global mass flow analysis of tungsten is
22 still missing in literature and its global supply chain remains opaque for industry outsiders. The
23 objective of this paper is to create a map of global tungsten flows to highlight and discuss key
24 material efficiency (i.e. using less of a material to make a product or supply a service, or reducing
25 the material entering production but ending up in waste) and supply security opportunities along
26 tungsten's supply chain that could be incorporated into the planning and prioritization of future
27 supply security strategies. The results indicate the existence of various intervention alternatives that
28 could help to broaden the supply base and improve the overall material efficiency of the system. In
29 particular, future policy and research and development (R&D) efforts to improve tungsten's
30 material efficiency should focus on minimizing tungsten losses as fine particles during beneficiation
31 and extraction (current global losses estimated at 10–40%), as well as on evaluating alternatives to
32 improve recycling collection systems and technologies, which could lead to 17–45% more tungsten
33 discards being recycled into new products.

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40 **KEYWORDS:** tungsten, mass flow analysis, material efficiency, mineral supply security.

41 **1. Introduction and background**

42 The high rate of technological evolution experienced in the world during the last three decades
43 has resulted in the development of increasingly complex products that employ intricate material
44 mixes. Combined with population and economic growth across the world, this has generated a rapid
45 growth in demand for many mineral commodities that were previously not produced in large
46 amounts. For example, 713 million smartphones were shipped globally in 2012, an increase of
47 44.1% over 2011 (IDC, 2013). This situation has raised concerns from governments, industries and
48 academics about whether the non-fuel mineral resources needed to satisfy the growing economic
49 demand will become scarce or difficult to obtain in the future. One such material is tungsten, as
50 evidenced by its inclusion in the European Union's (EU) raw material supply criticality list
51 (European Commission, 2014), which was motivated by its high economic importance stemming
52 from its wide range of applications, its lack of viable substitutes, the EU's dependence on imports
53 and trade concerns arising from China's dominant market position. Similarly, the British Geological
54 Survey's (BGS) risk list (BGS, 2012a) ranked tungsten as number two in a supply criticality index
55 list containing forty one elements, mainly due to alleged political instability in supplying regions
56 and its limited number of substitutes.

57

58 Tungsten's unique properties include the highest melting point, the lowest coefficient of thermal
59 expansion and the lowest vapour pressure of any non-alloyed metal (BGS, 2011; Lassner &
60 Schubert, 1999). In addition, tungsten is among the heaviest metals with a density similar to that of
61 gold and presents a high modulus of compression, high wear resistance, high tensile strength and
62 high thermal and electrical conductivity (International Tungsten Industry Association (ITIA), 2009;
63 BGS, 2011). These properties make it extremely important for a large variety of products. In
64 particular, tungsten's use in cemented carbide represents its most important application (ITIA,
65 2011b). Tungsten carbide is widely employed in the mining, petroleum, construction and metal-

66 working industries in drill bits and in machine tools for shaping metals, wood, composites, plastics
67 and ceramics (e.g. punches, stamping dies, bushes, rollers, milling inserts and tile and glass cutters
68 among others) (BGS, 2011). In addition, tungsten is commonly alloyed with steel, especially in
69 high speed steels (HSS) that allow high productivity levels in metal cutting and in superalloys with
70 applications in the aerospace, industrial gas turbine and marine turbine industries due to high
71 resistance to corrosion and wear (Lassner & Schubert, 1999). Other tungsten alloys find important
72 applications in electronics, power engineering and medical devices. Pure tungsten mill products are
73 used as light bulb filaments, vacuum tubes and heating elements. Additional applications include an
74 extensive range of chemical uses including catalysts, colouring agents for porcelain and paint
75 pigments, among many others (BGS, 2011).

76

77 Numerous security of supply strategies are discussed in literature (as exemplified by the summary
78 presented in Table S1 of the supplementary information); some of the most common being mineral
79 resource exploration incentives for supply diversification, material substitution, recycling systems
80 and technological improvement, material re-use and waste reduction. However, the authors believe
81 that the potential development and application of such approaches is usually hindered by the lack of
82 transparency and data availability that exists across the supply chain of these materials, which limits
83 the analysis of each strategy's potential material benefits and overall economic and technical
84 feasibility. This is also a difficulty for tungsten, as evidenced by a recent study of data needs for
85 mass flow analysis (MFA) relating to 21 raw materials (RPA, 2012), which identified tungsten
86 amongst the five elements that have the least data available. This type of analysis (also referred to
87 as material flow analysis or substance flow analysis) is an analytical method of mapping
88 quantitative data about material flows and their relationships and transformations through the entire
89 production system. Such analyses have been performed at a global level for base metals such as
90 steel and aluminium (Cullen and Allwood, 2013; Cullen et al., 2012), as well as for materials such

91 as rare earths (Du and Graedel, 2011a; Du and Graedel, 2011b), cobalt (Harper et al., 2012), indium
92 (Yoshimura et al., 2011) and a joint-study for neodymium, cobalt and platinum (Nansai et al.,
93 2014). Although tungsten flows have been analysed for the United States of America (Harper and
94 Graedel, 2008; Harper, 2008), a global mass flow analysis of tungsten is still missing in literature
95 and its global supply chain remains opaque for industry outsiders.

96

97 The objective of this paper is to create a global mass flow analysis of tungsten to discuss key
98 supply security opportunities where intervention could be most effective in broadening the supply
99 base and improving the material efficiency of the system. Such a map could work as reference
100 material for the planning and prioritization of future supply security strategies for tungsten based on
101 criteria such as prospective material gains, investment requirements and economic
102 certainty/motivation, existing technological readiness, geological knowledge and understanding of
103 potential new deposits, research and development capacity and sustainability performance. This
104 study is also expected to contribute to tungsten's supply chain transparency by gathering the scarce
105 public information that exists on this material and complementing it with new unpublished insights
106 obtained by the authors through a stakeholder consultation process. The assumptions that underlie
107 this analysis are discussed further in the next section.

108

109 **2. Methodology and data considerations.**

110 This section describes the tool employed to carry out the global mass flow analysis of tungsten
111 (Section 2.1) and the methods, assumptions and data sources used to build such analysis, including
112 a short account of data availability issues (Section 2.2).

113

114 **2.1. Description of the global tungsten Sankey diagram**

115 The Sankey diagram has been adopted as the visualisation tool employed to present the mass
116 flows of tungsten in this paper. Sankey diagrams applied to mass flows help to highlight
117 inefficiencies and potential savings in connection with material use by illustrating quantitative
118 information about flows, their relationships and their transformations, as suggested by Schmidt
119 (2008). Since their development over 100 years ago, Sankey diagrams have been used to represent
120 the energy and material balances of complex systems and have been widely used in industrial
121 ecology to depict industrial metabolisms (Schmidt, 2008).

122

123 The mass flow analysis presented in this paper displays the allocation of tungsten across its
124 supply chain by following the mining–manufacturing–use route, in addition to recycling and re-use
125 flows and the points where material losses occur. The Sankey diagram shows the total amount of
126 materials that were extracted, processed and used in 2010, but does not indicate the accumulated
127 natural and anthropogenic material stocks available for human exploitation. The thickness of the
128 flows are proportional to the amount of mass in each of them (i.e. the thickness of each link
129 represents the magnitude of flux) and the mass balance is maintained along the diagram. Therefore
130 all tungsten entering and leaving the system is accounted for and any mass balance irregularities
131 due to losses or inefficiencies are intuitively displayed (Schmidt, 2008).

132

133 Tungsten rarely exists in a pure state along the system, therefore, vertical divisions (slices) along
134 the flows indicate where important transformative processes occur. They are accompanied by an
135 indication of the resulting material forms and the amount of energy (including both electricity and
136 fuel converted to kWh units) that is consumed during each transformation per unit mass, to provide
137 an insight into their environmental cost. Additional resources and emissions involved during these
138 material transformation processes (e.g. water, chemicals or gas emissions) have not been included
139 due to lack of suitable data. Colour is used to distinguish the different tungsten grades contained in

140 each flow (i.e. to describe the typical tungsten concentration within the carrier materials in each
141 flow).

142

143 **2.2. Data availability and sources**

144 The tungsten Sankey diagram presented in this paper was populated using data from a variety of
145 industrial and academic sources. In some cases the data had to be inferred, estimated or back-
146 calculated if the direct values were not available. In order to overcome the problem of public data
147 scarcity, a stakeholder consultation was performed through the organisation of a workshop named
148 “Understanding the tungsten lifecycle in Europe” (BGS, 2012b). This workshop gathered experts
149 from across all levels of the supply chain, from mining to final manufacturing, in addition to
150 academia and consultancies. The lead author also visited the Mittersill tungsten mine in Austria,
151 operated by Wolfram Bergbau und Hütten (WBH, 2013), where tungsten mining experts were
152 consulted.

153

154 Table S2 in the Supplementary Information provides additional detailed information about the
155 methods, data and assumptions applied to the mass flow analysis to support the explanations
156 presented in this section and to help the reader to see overall characteristics of the estimation at a
157 glance. The mass flow estimations can be divided into five categories, as follows:

158

159 **i. Mining and extraction**

160 a. Global mine production figures (given in metric tonnes of tungsten content), following
161 ore beneficiation, are the starting point for the mass flow analysis building process.
162 Global mine production data per country (67 kt for China, 9.9 kt for the rest of the
163 world), flows to stock (5.9 kt) and data on total scrap input for 2010 (24 kt) were obtained

164 from the International Tungsten Industry Association's (ITIA) website (ITIA, 2011a;
165 ITIA, 2011b).

166 b. The beneficiation recovery rate for the Mittersill tungsten mine in Austria has been
167 estimated at 75–85% (WBH, 2013), whereas that of the Los Santos project in Spain has
168 been reported at 57–65% (Almonty, 2012) and that of the Cantung mine in Canada is
169 around 75–79% (NATC, 2013). These numbers agree with estimates from Lassner and
170 Schubert (1999) and Smith (1994), who have suggested that tungsten recovery rates
171 normally range between 60–90%. A recovery rate of 75% was assumed in Figure 1. Mine
172 production data was back-calculated considering this recovery rate to infer the amount of
173 tungsten contained in ore prior to beneficiation. A 75% recovery rate means that out of
174 103 kt of tungsten mined as ore, 26 kt are lost during beneficiation, while the rest
175 becomes the official mine production figure.

176

177 **ii. Recycling routes**

178 a. Based on data from ITIA (2011b), a total of 24 kt of tungsten were incorporated into the
179 supply chain through the recycling of scrap from end-of-life products in 2010. As shown
180 in Figure 1, two major recycling processes exist: the zinc process and chemical recycling.
181 Records showing the exact amounts of tungsten scrap that were processed through each
182 of these two methods could not be found in literature nor in industrial reports. Similarly,
183 none of the experts consulted during the stakeholder meeting (BGS, 2012b) were able to
184 provide information to clarify this point. Given that industrial recycling of tungsten
185 carbide is a well-established procedure (BGS, 2012b; WBH, 2013; Weiss, 1985) and that
186 carbide products account for at least 50% of end-products (BGS, 2011), it has been
187 assumed that 50% of tungsten is recycled through the zinc process (12 kt), which is the
188 preferred carbide recycling route due to its lower energy consumption and lower cost

189 compared to chemical recycling (WBH, 2013; Weiss, 1985). The remaining 12 kt of
190 tungsten are assumed to be recycled through the chemical route and transformed into
191 ammonium paratungstate (APT).
192

193 **iii. Fabrication of intermediate products and finished sectors**

194 The construction of this section of the Sankey diagram involved three main steps: defining the
195 structure and connections between its flow routes (part ‘a’), identifying relevant data and
196 assumptions regarding the likely values for each flow (part ‘b’) and connecting these two pieces
197 of information to back-calculate and estimate the final mass flows shown in this intermediate
198 section of the Sankey diagram (part ‘c’).
199

200 a. The flow structure and links depicted in the intermediate section (“Fabrication of
201 intermediate products”) have been based on previous work by Smith (1994) and a
202 subsequent adaptation of the same work by Harper and Graedel (2008), who mapped
203 tungsten flows for the United States. Based on this, a fraction of tungsten concentrate
204 flows directly towards the tungsten carbide production step, while another concentrate
205 fraction is used directly in the manufacturing of steel and alloys. A larger portion of
206 tungsten concentrate is chemically converted to APT, which is mainly an intermediate
207 compound used in the production of tungsten chemicals and tungsten metal powder. The
208 latter is commonly employed to manufacture tungsten forms, ferrotungsten and tungsten
209 carbides.

210 b. Mass allocation for the “Fabrication of intermediate products” section was based on the
211 following evidence:
212

- 213 i. The global distribution of finished sectors was obtained from BGS (2011) (i.e. the
214 percentage of tungsten used to produce chemicals [6%], mill products [13%], steels
215 and alloys [27%] and carbide products [54%]). In addition, the detailed distribution
216 of carbide products was obtained from ITIA (2010) (i.e. metal cutting [22%], wear
217 applications [17%], stoneworking [26%], wood and plastic working [26%] and
218 chipless forming [9%]). Moreover, information on chemical product categories was
219 extracted from ITIA (2011c).
- 220 ii. Information about processing losses during each manufacturing step was initially
221 obtained from the work of Smith (1994) and Weiss (1985) and later on corroborated
222 through conversations with industry representatives (BGS, 2012b; WBH, 2013):
223 4% losses during APT production (2.6 kt), 1% losses during metal powder
224 manufacturing (0.6 kt), 1% losses from transforming metal powder into tungsten
225 carbide plus 4% losses from converting tungsten concentrate into carbide (0.5 kt in
226 total), 4% losses during the production of tungsten chemicals (0.2 kt) and 4–5%
227 losses from the use of tungsten concentrate in steels and alloys (0.8 kt). All these
228 numbers combined produced an overall 5% mass loss, equivalent to 4.7 kt.
- 229 iii. Two key assumptions have been made, based on the work from Lassner and
230 Schubert (1999): roughly 70–80% of tungsten is used in powder metallurgy and
231 approximately 70–80% of tungsten powder is used to produce tungsten carbides.
232 For the purpose of the Sankey diagram, these numbers were fixed at 73% and 70%
233 respectively to ensure the system mass balance. These assumptions resulted in a
234 total of 65 kt of tungsten (out of 88.9 kt) converted to APT and 47.9 kt of tungsten
235 metal powder (out of 68.7 kt) allocated to carbide production.

236

237 c. Connecting the information given in part ‘b’ above while following the structure
238 described in part ‘a’ allowed the back-calculation and estimation of all the flows that
239 form the “Fabrication of intermediate products” section:

240 i. Considering a total consumption of 95 kt of tungsten in 2010 (76.9 kt virgin
241 tungsten + 24 kt scrap input – 5.9 kt flow to stocks) and 5% overall losses during
242 manufacturing (4.7 kt), it was possible to allocate the appropriate shares to each
243 finished sector (90.3 kt distributed across four categories). In this way, tungsten
244 chemicals flowing from APT production were back-calculated to 5.6 kt (accounting
245 for losses), tungsten forms reaching mill products were estimated at 11.7 kt,
246 tungsten carbide going to carbide products was 48.7 kt and a total of 24.4 kt of
247 tungsten went to steel and alloys production coming from three different sources:
248 tungsten concentrate, ferrotungsten and tungsten forms.

249 ii. After considering all the data and assumptions explained until this point, four mass
250 flows remain undefined: the exact amount of tungsten concentrate flowing to
251 carbide production and steel and alloys production, as well as the amount of
252 ferrotungsten and tungsten forms flowing to steel and alloys production. The
253 values shown for these flows in Figure 1 have been allocated by considering the
254 expert opinions from consulted stakeholders (BGS, 2012b; WBH, 2013) and
255 applying the conservation of mass principle. In this way, it was assumed that
256 roughly 70% of the input for steels and alloys production came directly from
257 tungsten concentrate (16.7 kt), while the amount of ferrotungsten used in steels and
258 alloys manufacturing was defined as nearly double than that of tungsten forms (5.8
259 and 2.7 kt respectively). By mass conservation, the remaining tungsten concentrate
260 (1.3 kt) was allocated to the production of tungsten carbide.

261

262 **iv. Tungsten grades**

263 a. The purity of each flow specified in Figure 1 has been obtained from Lassner (1995) for
264 ore deposits, ore concentrate and tungsten scrap grades; GTP (2015) for APT; THPP
265 (2014) for tungsten metal powder; USGS (2011) for carbide metal powder and carbide
266 products grades; ITIA (2011d) for ferrotungsten grade; ITIA (2011e) for high speed
267 steels, tool steels, cast steels and heavy metal alloys; Haynes (2013) for superalloys and
268 ITIA (2011f) for heavy metal alloys.

269

270 **v. Processing energy consumption**

271 a. The energy consumption of distinct processing steps has been obtained from USDOE
272 (2007) and consultation with industry specialists (WBH, 2013) for mining, extraction,
273 handling and beneficiation; Krishna Rao (1996) for general beneficiation figures; De
274 Wang et al. (1995) for APT production from concentrate; Hairunnisha et al. (2007) for
275 APT production from scrap; Acharyulu and Rama Rao (1996) and Suchkov et al. (1971)
276 for tungsten powder production from concentrate; and Acharyulu and Rama Rao (1996)
277 and Gürmen and Friedrich (2004) for powder production from tungsten carbide scrap.
278 Additional information on energy estimates is presented in Table 2. Data availability on
279 tungsten's mining and processing energy intensities is low and therefore the data
280 presented in Table 2 does not necessarily represent all existing technologies or best
281 practices across the industry, but simply shows the data that is available in literature.

282

283 **3. Results**

284 Figure 1 presents the global mass flow of tungsten through its entire supply chain in 2010, as well
285 as the energy requirements of key transformation processes and the material grades of main flows.
286 The total mine production of tungsten in 2010 was 76.9 kt (ITIA, 2011a), in addition to the

287 consumption of around 24 kt of scrap from end-of-life products (ITIA, 2011b). The tungsten
288 lifecycle starts with the mining of tungsten ore minerals, chiefly scheelite and wolframite, which
289 contain about 80.6% and 76.5% tungsten trioxide (WO_3) respectively (BGS, 2011). The ore grade
290 of tungsten deposits varies between 0.08–1.5% of WO_3 (i.e. 0.06-1.2% tungsten metal content), as
291 indicated in Figure 1. Mining is an energy intensive activity, requiring 20.3 kWh per metric tonne
292 of processed ore (including both electricity and fuel converted to kWh units). The tungsten-
293 containing minerals are extracted from ore through traditional beneficiation techniques such as
294 crushing, grinding, magnetic, gravity and flotation separation, to form market-grade concentrates
295 with WO_3 contents between 15–75% WO_3 (i.e. 12–60% tungsten metal content) (BGS, 2011). The
296 beneficiation process is slightly less energy intensive than mining, requiring 13–15 kWh per tonne
297 of processed ore.

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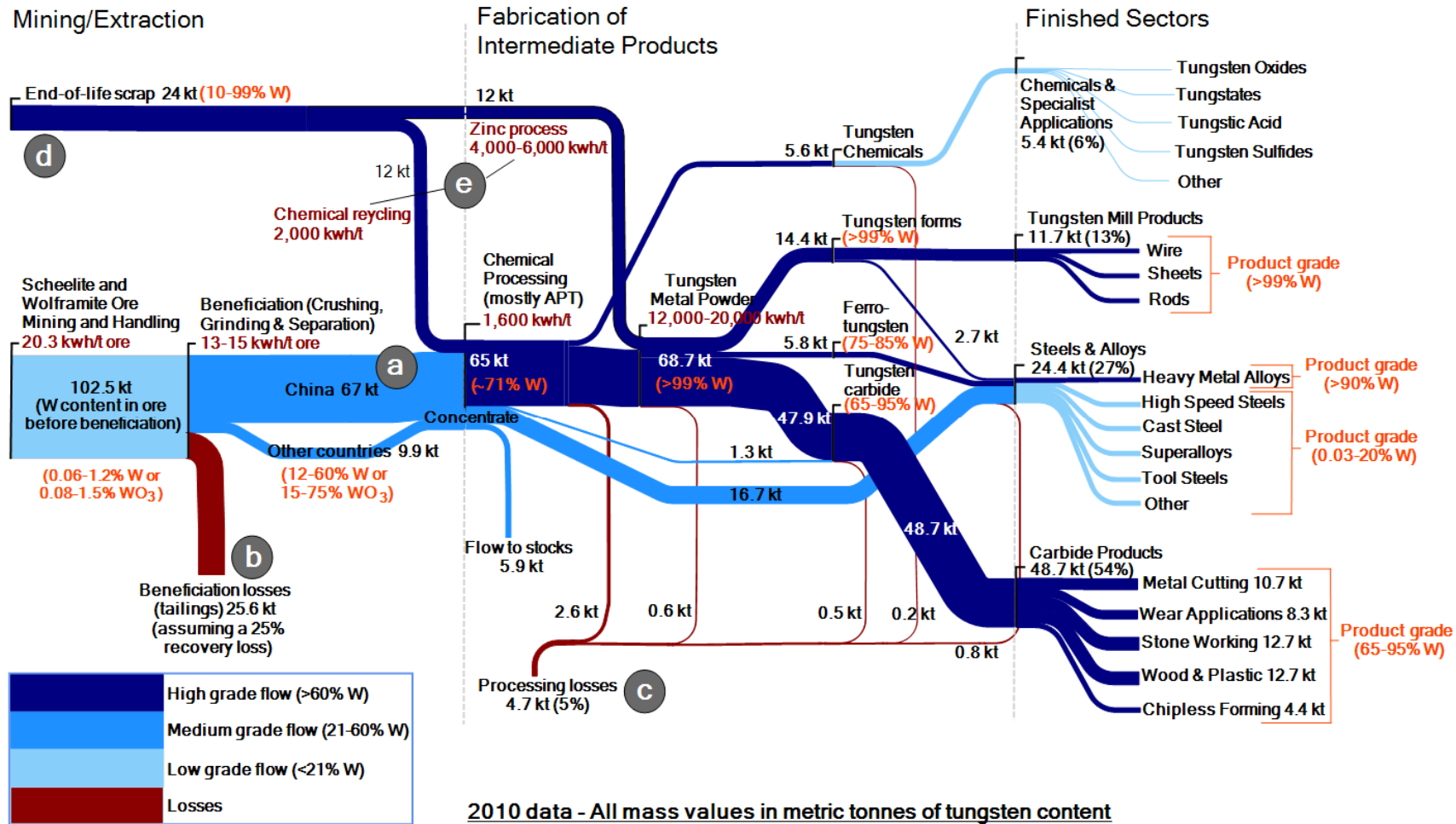
299 The resulting tungsten concentrate can either be used directly as an alloying element in steel or
300 converted to intermediate tungsten compounds through hydrometallurgy (mostly to APT with a
301 typical WO_3 content of 89.5%, or 71.6% tungsten content (GTP, 2015)). This process requires
302 1,600 kWh per tonne of APT produced. Intermediate compounds can be further refined through
303 pyrometallurgy, leading to the production of tungsten metal powder containing >99% tungsten
304 (THPP, 2014) and typically requiring 12,000–20,000 kWh per tonne of powder produced. Tungsten
305 metal powder is then converted into final products, mostly in the form of tungsten carbide for
306 cutting tools (65–95% tungsten content) (USGS, 2011), ferrotungsten (75–85% tungsten content)
307 (ITIA, 2011d) for steels and alloys (with final tungsten contents ranging between 0.03 and more
308 than 90% (ITIA, 2011e; ITIA, 2011f; Haynes, 2013)) and tungsten metal for mill products such as
309 wires, rods and sheets containing more than 99% tungsten (BGS, 2011). An additional application is
310 the production of chemicals with low tungsten content such as tungsten oxides, tungstates, tungstic
311 acid and tungsten sulphides.

312 Secondary supply of end-of-life scrap is also incorporated into the supply chain presented in
313 Figure 1, mainly through chemical recycling (hydrometallurgy) and the zinc process, which result
314 in the production of APT and tungsten metal powder, respectively. Figure 1 indicates that the zinc
315 process has a higher energy consumption of around 4,000–6,000 kWh per tonne of tungsten
316 processed, compared to 2,000 kWh per tonne for chemical processing. Both represent viable
317 recycling routes for this material.

318

319 The supply chain presented in Figure 1 has an overall mass efficiency (measured as the ratio of
320 output of tungsten mass to input of tungsten mass) of about 71% (which varies between 55% and
321 83% for worst- and best-case scenarios based on the beneficiation and intermediate processing
322 losses shown in Figure 1). Key aspects of the global tungsten map are discussed in the following
323 section.

324



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Figure 1: Global mass flows of tungsten in 2010. The grade of different flows and the energy consumption of selected processes are indicated with orange and red text respectively. Note: the letters are referred to in section 4

328 **4. Discussion**

329 The following key messages have been highlighted in Figure 1 (each item in the list indicated in
330 Figure 1 under the same letter):

331

- 332 a) In 2010, China accounted for about 87% of all mine production.
- 333 b) Tungsten's beneficiation process leads to considerable losses (10–40% of the tungsten
334 content of the ore may be lost).
- 335 c) Processing losses during the fabrication of intermediate products are comparatively lower
336 (<5%). In addition, nearly three quarters of all tungsten are processed through powder
337 metallurgy and about half of total tungsten is used to manufacture carbide products.
- 338 d) Roughly 25% of total tungsten supply came from end-of-life scrap in 2010.
- 339 e) Based on their processing energies, tungsten recycling is less energy intensive than virgin
340 production. Figure 1 presents the latter in terms of energy per tonne of processed ore
341 given that tungsten content in ore can vary greatly. At a typical cut-off ore grade of 0.2%
342 [WBH, 2013], virgin production results in roughly 10,000 kWh per tonne of tungsten, in
343 comparison to <6,000 kWh per tonne of tungsten for recycling (More details in Section
344 5.5).

345

346 The analysis of tungsten flows shown in Figure 1 leads to important questions, as discussed in
347 Sections 4.1-4.5.

348

349 **4.1 Can we develop alternative supply chains in the rest of the world?**

350 Historical evidence suggests that China has not always been the dominant actor in the tungsten
351 market. Figure 2 (Brown, 2012) shows that although global tungsten mine production barely

352 changed between 1980 and 1990, totalling little more than 50 kt, China's share of the total changed
353 from only 29% in 1980 to 62% in 1990, increasing even further in the following years.

354

355 As shown in Figure 3, tungsten prices rose significantly in the early 1970s, reaching US\$164 per
356 metric tonne in 1977 (USGS, 2013). Brown (2012) and the USGS (2013) suggest that a supply
357 shortfall in 1978 had to be compensated by the release of US Government stockpiles and increased
358 exports from China, with prices falling as a result. Reduced demand from Western Europe caused
359 by the 1981 global recession, coupled with an increase in the supply of Chinese tungsten
360 concentrates and intermediate products at cheaper prices than those from Western sources,
361 contributed to a downward trend of tungsten prices until the mid-1980s, reaching a low point in
362 1986 (Brown, 2012; USGS, 2013) (Figure 3). Despite strong demand between 1986 and 1990,
363 prices continued to be relatively low as a result of continuous oversupply from China at cheaper
364 prices (USGS, 2013). This trend was exacerbated in the early 1990s, when the oversupply from
365 China coincided with a period of reduced demand due to another global recession and a reduction in
366 imports to the former Soviet countries following the 1991 breakup of the Soviet Union (Brown,
367 2012; USGS, 2013). Low prices during such an extended period of time led to the closure of a
368 significant number of mines in the Western world, leaving China as the main player in the market.
369 Once demand and prices started to recover, Chinese producers were able to react more quickly and,
370 as a consequence, China's output grew rapidly compared to other nations (Brown, 2012).

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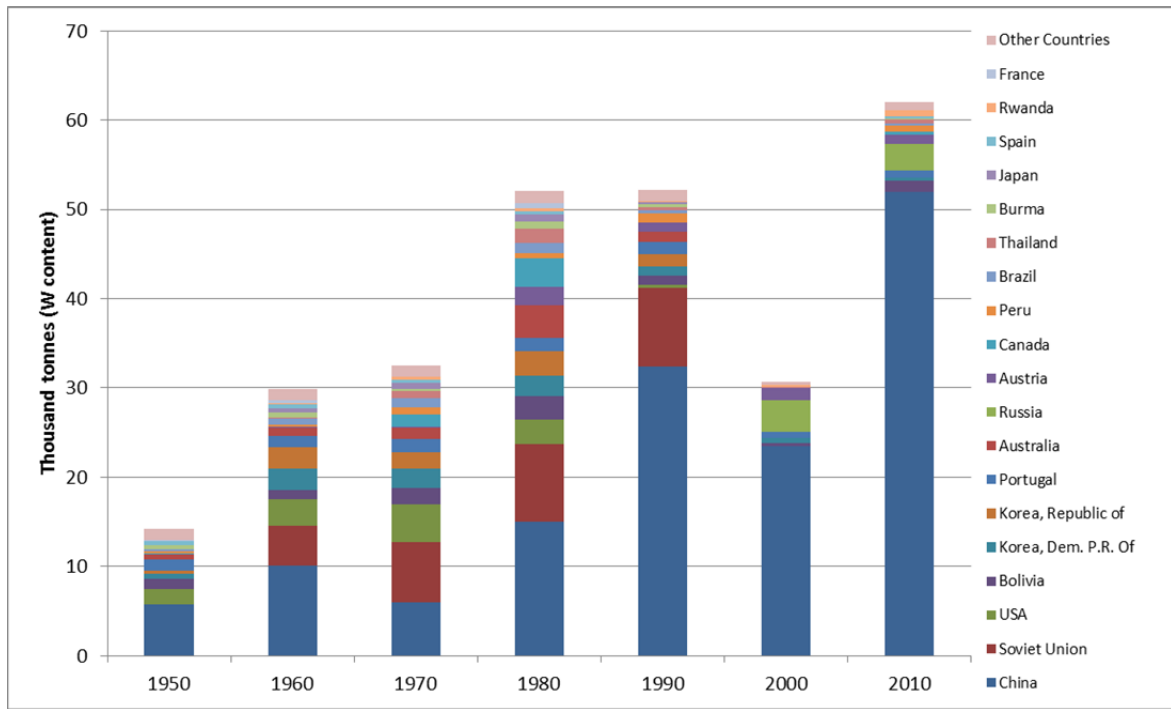


Figure 2: Evolution of tungsten production concentration 1950–2010 (Data from Brown (2012)).

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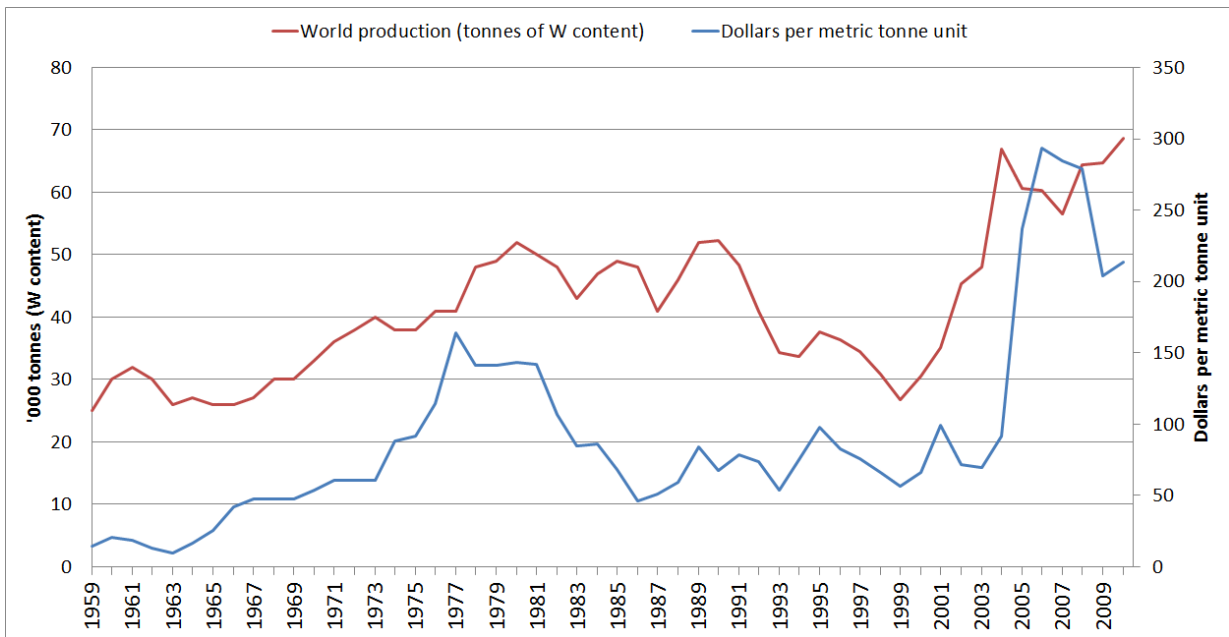


Figure 3: Historical tungsten production and unit value 1959–2010. Production data in tonnes of W content (Data from BGS (2013)). Prices in dollars per metric tonne unit (Data from USGS (2013), converted from short ton to metric tonne).

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380 Several tungsten mining projects are under development worldwide, both for newly discovered
381 deposits and for the reopening of dormant mines (Table 1), indicating that supply diversification is

382 possible. These may contribute to global supply in the near future, given favourable market
 383 conditions, and help to reduce China's dominant position as a result.
 384

Name	Country	Current Status (as at Feb 2015)	Possible Production	Resources (tonnes contained tungsten)
Hemerdon	United Kingdom	Feasibility study completed May 2011, mine construction started early 2014	2015	>420,000
Mount Carbine	Australia	Tailings retreatment commenced in 2012, reopening of hard rock mine scheduled for 2016	2016	>50,000
Watershed	Australia	Feasibility study completed Sept 2014, permitting completed Dec 2013, raising funds	2016 or 2017	>55,000
Barruecopardo	Spain	Feasibility study completed Feb 2012, mine permit granted Nov 2014, raising funds	2016 or 2017	>55,000
King Island	Australia	Feasibility study completed Feb 2012, all permits in place, updated resources and reserves statement Sept 2014, progressing with raising funds	2016 or 2017	>195,000
Sisson Brook	Canada	Feasibility study completed early 2013, permitting expected in 2015, raising funds	2017?	>270,000
Sangdong	South Korea	Feasibility study completed April 2012, updated January 2015	unknown	>280,000
Mactung	Canada	Feasibility study completed 2009, environmental permitting completed in 2014	Unknown	>370,000
Northern Dancer	Canada	Preliminary economic assessment completed in 2011, development currently suspended	Unknown	>390,000
O'Callaghans	Australia	Prefeasibility work continuing	Unknown	>200,000

385 Table 1: Selected major developing tungsten deposits and those where production is expected in the near future.
 386 Note: Resources are from all categories and in some cases include reserves (Data compiled from individual
 387 company reports and websites).
 388

389 Known tungsten deposits occur in many countries of the world, as illustrated by Brown & Pitfield
 390 (2014). Detailed, up-to-date figures for global resources of tungsten are difficult to obtain but Hinde

391 (2008) estimated the total to be approximately 7 million tonnes of contained tungsten. In addition to
392 the locations mentioned in Table 1, significant deposits are known to exist in Kazakhstan, Russia
393 and the United States as well as China (Brown & Pitfield, 2014). The United States Geological
394 Survey estimates worldwide reserves of tungsten to be approximately 3.3 million tonnes, with 42%
395 of those reserves being located outside of China (Shedd, 2015).

396

397 The evidence described in this section suggests that it is possible to develop alternative supply
398 chains; however these are subject to financing being available to open projects outside China, which
399 depends on the perceptions of investors with regards to risks. Detailed discussions of the many
400 factors that affect supply diversification are beyond the scope of this paper. Reducing the
401 dominance of China in the supply of tungsten will require both time and appropriate policy efforts.

402

403 **4.2 What variables determine beneficiation losses and how can they be reduced?**

404 Beneficiation losses are mostly attributed to the friable nature of tungsten minerals (WBH, 2013;
405 Weiss, 1985), which leads to the excessive generation of fine particles (<25 µm) during ore
406 grinding and crushing to liberate tungsten minerals from the rest of the gangue material. As
407 suggested by Wills (1988), a mineral deposit will be economic to work if its contained value per
408 tonne is higher than the sum of total processing costs (including mining and subsequent separation
409 steps) plus losses and other costs per tonne. In other words:

410

$$411 \text{ Contained value / t} > (\text{total processing cost} + \text{losses} + \text{other costs}) / \text{t}$$

412

413 In the case of tungsten, mining represents a major cost. This is because the mining methods
414 required to exploit underground vein-type deposits such as tungsten deposits are among the most
415 expensive, as suggested by Wills (1988). Therefore, a balance is required between beneficiation

416 costs and material losses if the economic viability of the entire operation is to be preserved. This
417 means that sometimes losses have to be tolerated in exchange for less efficient but more cost-
418 effective processing methods. Even if more efficient beneficiation methods exists, these need to be
419 economical enough to guarantee that total costs do not exceed the contained value of the deposit.
420 There are two traditional approaches for solving this trade-off between cost and efficiency: either by
421 creating economic methods to avoid the creation of tungsten fines in the first place or by developing
422 economic processes for extracting these tungsten particles from waste slimes and tailings.

423

424 Fine tungsten particles are hard to capture by the most widely used separation methods and are
425 commonly lost in slimes and tailings instead, from where it is even harder to recover the tungsten
426 (Weiss, 1985; WBH, 2013; Krishna Rao, 1996). Tungsten minerals are friable and tend to be
427 ground preferentially during crushing and grinding. Also, due to their high density, tungsten
428 particles can be misclassified by cyclones or hydraulic classifiers, often being sent to the over-size
429 fraction and getting recycled to the grinding mill, resulting in over-grinding (Krishna Rao, 1996).
430 This argument has been supported by Clemente et al. (1993), who provide an applied example from
431 a wolframite mine at Minas da Panasqueira in Portugal. In this mine, tungsten ore is crushed to a
432 coarse average size of 2.25 mm to liberate the tungsten minerals from the gangue. The higher
433 friability of wolframite compared to the rest of the gangue minerals in the ore means that
434 wolframite tends to end up disseminated in fine particle form, leading to a fines feed with almost
435 double the tungsten content than the rest of the plant feeds. Most of these fine wolframite particles
436 (below 25 μm) end up in tailings, where they are mixed with a wide range of other minerals,
437 complicating their recovery through normal separation methods. The same problem has been
438 reported with scheelite at the Mittersill tungsten mine in Austria (WBH, 2013). Scheelite losses are
439 exacerbated when extracted from low grade ores, as this type of rock requires even finer grinding to

440 liberate scheelite from gangue material, leading to higher losses of tungsten as fine particles in
441 slimes (Marinakakis and Kelsall, 1987).

442

443 Multi-stage crushing and grinding has been suggested by Krishna Rao (1996) as an effective
444 technique to reduce the excessive generation of tungsten fine particles. In this process, multi-stage
445 sizing of the ore takes place, attempting to recover as much tungsten as possible from each size at
446 each stage. Selective disintegration has also been suggested by Chanturiya (2008) who advocates
447 substituting the traditional processes of crushing and grinding in jaw and cone crushers and ball
448 mills by processes that cause disintegration across the boundaries of mineral grains and thus
449 promote mineral liberation with reduced fines production. Other approaches to avoid losses by
450 over-grinding include coarse narrow-range grinding of wolframite by rod milling, as proposed by
451 Jakhu and Ray (1996) who have reported eighty per cent liberation of wolframite by this process.

452 The extraction of tungsten fine particles from tailings has been investigated by Clemente et al.
453 (1993), who mention high-efficiency slimes gravity separators and new flotation reagents as
454 examples of processing technologies capable of extracting tungsten from slimes. A three-stage
455 gravity separation process developed and tested by Clemente et al. (1993) was capable of producing
456 a 50–55% WO_3 concentrate from tungsten slimes at Minas da Panasqueira in Portugal, achieving a
457 68–73% recovery of tungsten particles in the 10–125 μm range and about 50 to 54% of all tungsten
458 contained in the tailings of that mine. In summary, it is technically feasible to minimise the
459 dissipation of tungsten during beneficiation by reducing the production of fine tungsten particles
460 through optimisation of the comminution stages and/or recovering tungsten from tailings when
461 economic conditions allow it. However, the technical viability of these approaches has to be
462 accompanied with economic viability for these methods to be utilised.

463

464

465 **4.3 What factors explain the high material efficiency observed during fabrication?**

466 Tungsten's high economic value tends to ensure its efficient use during manufacturing of
467 intermediate and consumer products (WBH, 2013). As a result, intermediate processing losses are
468 significantly lower than beneficiation losses, ranging from 1 to 4% (BGS, 2012b; WBH, 2013;
469 Smith 1994; Weiss, 1985), as shown in Figure 1. Although there is no available data on the
470 generation and recycling of internal scrap during the manufacturing of intermediate and consumer
471 products, Smith (1994) and WBH (2013) have suggested that there is a high and efficient reuse of
472 this type of scrap in manufacturing facilities (closed-loop recycling). In addition, between 70 and
473 80% of tungsten products are manufactured through powder metallurgy (Lassner and Schubert,
474 1999), which is a highly efficient and controlled technique in which losses are minimised and waste
475 material is efficiently recovered and reused.

476

477 **4.4 What are the main barriers towards achieving higher recycling rates?**

478 Tungsten recycling is not significantly constrained by technological availability, but rather by its
479 use in some applications where recycling is not possible due to dispersion or dilution and by the
480 lack of appropriate post-consumer collection systems.

481

482 Acharyulu and Rama Rao (1996) indicate that tungsten scrap is commonly available in four main
483 forms: pure tungsten metal scrap, heavy alloy scrap, tungsten carbide scrap and tungsten-containing
484 steels. Available tungsten recycling processes and the type of scrap they can process are discussed
485 in the work of Acharyulu and Rama Rao (1996), while schematic views of these recycling methods
486 can be seen in a report from Smith (1994). High-grade scrap such as cemented carbides, which
487 account for the largest fraction of scrap resources, can be recycled by direct physical re-use methods
488 or by semi-direct and indirect chemical processing (Reuter et al., 2013). The two most common
489 direct re-use methods include the zinc and coldstream processes, with the former being preferred

490 over any other method due to its high efficiency (>95%), lower cost and higher energy efficiency
491 (Acharyulu and Rama Rao, 1996). Heavy alloy scrap (over 90% tungsten content) and tungsten mill
492 products (considered as pure tungsten metal scrap) are valuable raw materials due to their high
493 tungsten fraction. These materials can be recycled through chemical digestion, coldstream/crush,
494 oxidation/reduction and chemical separation routes.

495

496 In contrast, tungsten contained in ferrous and non-ferrous alloys with low tungsten fractions (cast
497 steels, high-speed steels, tool steels, etc.) is not commonly recycled, as most of it is diluted during
498 the recycling of steel, as suggested by Smith (1994). In a similar way, tungsten used for chemical
499 and specialist applications is generally not recycled due to the high dispersion of the material in
500 these applications (BGS, 2012b; Smith, 1994).

501

502 Even more significant than dispersion and dilution, there seems to be a general consensus among
503 industry members that the biggest limitation affecting tungsten recycling is the lack of appropriate
504 post-consumer collection systems for open-loop recycling (WBH, 2013; BGS, 2012b). The
505 industry's tendency towards vertical integration could help to ensure an efficient recycling of
506 internal scrap (BGS, 2012b). However, more awareness is required among end-product consumers
507 with respect to the potential economic advantages of implementing efficient collection systems and
508 strategies. Practical initiatives such as take back schemes between manufacturers and end-users
509 have already been reported (WBH, 2013). These represent a move towards the achievement of
510 higher recycling rates. Additionally, current high prices of tungsten are already pushing end-users to
511 pursue alternative material efficiency strategies such as re-designing the same products with less
512 tungsten. Examples have been reported where the tungsten content of specific carbide tools has
513 been reduced by as much as 90% through product re-design (WBH, 2013). In summary, dispersion,
514 dilution and lack of collection infrastructure are the key factors limiting global tungsten recycling.

515 **4.5 Are energy efficiency benefits sufficient to incentivise recycling?**

516 The energy estimates presented in Figure 1 (a detailed breakdown of these numbers is shown in
517 Table 2) indicate that virgin material mining and handling plus beneficiation accounts for roughly
518 2,700–58,800 kWh per tonne of W content. This large range is related to variation in ore grade of
519 0.06–1.2% W (0.08–1.5% WO₃), although cut-off grades below 0.2% are uncommon (WBH, 2013;
520 Weiss, 1985) (the average energy consumption in Mittersill is about 10,500 kWh per tonne of W
521 content for an ore grade of 0.3% WO₃). In comparison, recycling through the chemical route leads
522 to a lower energy consumption of 2,000 kWh per tonne of W content. Similarly, tungsten powder
523 production from virgin ore involves four steps: mining and handling, beneficiation, chemical
524 processing to APT and powder production. Adding these processes leads to roughly 16,300–80,400
525 kWh per tonne of W content, which is a much higher figure than the 4,000–6,000 kWh per tonne of
526 W content employed during direct scrap re-use through the zinc process.

527

528 Despite these energy savings, tungsten recycling is not necessarily cheaper than buying ore
529 concentrate. Depending on market conditions, product fabrication through the virgin and recycled
530 routes may have similar costs, as the cost of tungsten scrap may be even higher than that of tungsten
531 concentrate (for example, ~US\$15,000 per tonne of carbide scrap [Tungsten Carbide Recycling,
532 2015] versus ~US\$14,000 per tonne of Chinese concentrate [MetalBulletin, 2015]). However,
533 recycling benefits are not only measured in terms of energy and cash savings, but also in terms of
534 material efficiency, supply security and reduced environmental impacts. Tungsten mining and
535 beneficiation are processes with high losses (10–40%), while recycling routes have high yields
536 (>95%) (WBH, 2013; Weiss, 1985). Although primary production can never be entirely substituted
537 by recycling as demand grows from year to year (Tercero Espinoza, 2012), recycling could help to
538 secure an efficient secondary supply of tungsten that requires lower processing energies, generates
539 lower carbon emissions and avoids rock waste and leachates from mining operations. In summary,

540 energy efficiency is not enough to incentivise recycling because the economic benefits are not
 541 sufficiently obvious. However, the positive environmental benefits associated with recycling could
 542 help to offset the negative environmental impacts related to material losses in primary production
 543 and improve supply security, hence representing an extra incentive for the organisations involved.
 544

Processing step	Energy (kWh/t)	Comments	Source
Ore mining & extraction			
▪ Drilling	0.4 *	USA best practice across the whole metals sector	(USDOE, 2007)
▪ Blasting	2.2 *		(USDOE, 2007)
▪ Digging	1.5 *		(USDOE, 2007)
▪ Ventilation	1.3 *		(USDOE, 2007)
▪ Dewatering	0.2 *		(USDOE, 2007)
▪ Materials handling	14.7 *		(USDOE, 2007)
Beneficiation and processing			
▪ Crushing	0.4 *	USA best practice	(USDOE, 2007)
▪ Grinding	14.4 *	USA best practice	(USDOE, 2007)
▪ Beneficiation general	12.8 *	Wolframite ore in India	(Krishna Rao, 1996)
Ore mining & extraction plus beneficiation and processing in Mittersill mine, Austria	31.7 *	Scheelite concentrate production from ore in Mittersill, Austria	(WBH, 2013)
APT production from concentrate	1600	Solvent extraction method (99.87% efficiency)	(De Wang et al. 1995)
APT production from scrap	2000	Anodic dissolution method	(Hairunnisha et al. 2007)
Powder from concentrate	12,000-20,000	Chemical and electrolytic methods	(Acharyulu and Rama Rao, 1996 [for 12,000 kWh/t value]; Suchkov et al. 1971 [for 20,000 kWh/t value])
Powder from WC recycling	4,000-6,000	Zinc process (>95% efficiency)	(Acharyulu and Rama Rao, 1996; Gürmen and Friedrich, 2004)

545 Table 2: Energy estimates presented in Figure 1 (*units in kWh/t ore), including both electricity and fuel
 546 consumption converted to kWh units.
 547

548

549

550 **5. Conclusions**

551 This paper has mapped the global mass flows of tungsten, from mining to end-use sectors, for the
552 year 2010 and identified key areas where intervention would be beneficial to broaden the supply
553 base and increase the material efficiency of the system. The evidence gathered in this analysis
554 suggests that, although tungsten is susceptible to real risk factors and bottlenecks, there are also
555 options for change. Future R&D work to improve tungsten's material efficiency should focus on
556 two main priority areas. The first should investigate ways of avoiding tungsten losses as fine
557 particles during beneficiation (both by the optimisation of comminution and recovery from tailings)
558 and improving the economics of the process. Considering that tungsten recovery rates normally
559 range between 60–90%, as reported by Lassner and Schubert (1999) and Smith (1994), this could
560 potentially lead to a 10–40% recovery improvement. The second priority area should investigate
561 how to increase awareness of tungsten's recycling value, examine what are the current limitations
562 of recycling collection systems and recovery technologies and evaluate alternatives for
563 improvement. Considering that the supply chain presented in Figure 1 has an overall mass
564 efficiency varying from 55% to 83% for worst- and best-case scenarios, the potential impact of
565 improved recycling could oscillate between 17% and 45% more recovered tungsten.

566

567 As a complement to these lines of research, it is necessary to complete a detailed analysis of
568 finished products that contain tungsten, which is missing from this study. An analysis of such type
569 could be extended to cover historical tungsten consumption to provide a model output of end-of-life
570 scrap that could give a valuable estimation on the current and future size of the tungsten stock
571 available for recycling from end-of-life products. In addition, an economic analysis (cost-benefit
572 analysis) which identifies the factors that promote and/or hinder tungsten beneficiation optimisation
573 and recovery is missing from this study. This should explore the cost of current practices and the
574 cost of new technologies against market trends at global level. The economic analysis should

575 investigate geographical differences (i.e. China vs Europe) or sector scale variance (small vs big
576 deposits) and identify if any of the above could result in process optimisation and recovery from
577 mine waste.

578

579 Future policy efforts to ensure a secure supply of tungsten should consider promoting two main
580 strategies. The first should investigate and evaluate the potential benefits of providing economic
581 incentives for investors and companies willing to explore and develop new tungsten resources
582 and/or re-evaluate known resources outside China to reduce or eliminate dependence on Chinese
583 exports and reduce Chinese influence over prices. This could not only include mining activities but
584 also fabrication and manufacturing of intermediate tungsten products outside of China, where
585 current skills and infrastructure are mainly located. The second approach should investigate the
586 applicability of alternative material efficiency strategies at product level, including the use of less
587 material by design and lifetime extensions for specific key products (e.g. tungsten carbide tools). As
588 with recycling, this line of research would require a detailed analysis of finished products that
589 contain tungsten and available manufacturing technologies to identify available efficiency
590 opportunities.

591

592 Finally, various knowledge gaps have been identified through the development of the global mass
593 flow analysis for tungsten, including the amount of material contained in both closed and open
594 recycling loops, the resource intensity of different processing routes (including up to date energy
595 consumption figures), material losses along the entire supply chain, data on the fabrication of
596 intermediate products and mine production data discrepancies. Initiatives to fill in these data gaps in
597 future work could include two strategies. The first is to improve stakeholder engagement by
598 academics. Direct and extensive communication with stakeholders, as attempted in this paper
599 through the organisation of a stakeholder workshop and a visit to a tungsten mine, is recommended

600 as a suitable approach to obtain non-commercially sensitive data that would otherwise remain
601 unknown. The second strategy is to increase sustainability reporting by industry. Although it might
602 be difficult to convince companies to report their sustainability indicators without direct regulation
603 by governments, voluntary reporting initiatives such as that promoted by the Global Reporting
604 Initiative (GRI, 2015) have demonstrated their value to help organisations become more sustainable
605 and improve their reputation with the wider public. In the case of tungsten, this initiative should be
606 championed by the International Tungsten Industry Association and the International Council on
607 Mining and Metals across different countries and supply chain stakeholders, to improve the
608 transparency of its supply chain.

609

610 The priority areas, strategies and initiatives outlined above, especially if combined together,
611 would result in much greater material efficiency and supply chain transparency for tungsten and
612 may eventually lead to a reduction in the supply security concerns identified at the start of this
613 paper.

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632

633 SUPPORTING INFORMATION

634 A detailed analysis of mineral supply security literature together with a breakdown of the
635 methods, data and assumptions used to create the mass flow analysis presented in Figure 1 are
636 contained in the supporting information.

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649 REFERENCES

- 650 Acharyulu, S.L.N. and P. Rama Rao. 1996. "An integrated approach to the optimum utilization of
651 national tungsten resources: technology gaps." (Bulletin of Materials Science) 19, no. 2: 79–
652 199.
- 653 Almonty. 2012. Almonty Industries Inc.
654 http://almonty.com/_resources/presentation/Almonty_05_12_Presentation.pdf (accessed 07
655 12, 2013).
- 656 BGS. 2011. Tungsten Profile. Keyworth, UK: British Geological Survey.
- 657 BGS. 2012a. Risk List 2012. Keyworth, UK: British Geological Survey.
- 658 BGS. 2012b. "Understanding the tungsten lifecycle in Europe." Workshop organised by the British
659 Geological Survey. London, UK, December 5, 2012.
- 660 BGS. 2013. World Mineral Statistics Dataset. Keyworth, UK: British Geological Survey.
- 661 Brown, T. 2012. "Tungsten – Primary Production and Trade." Presentation given at the BGS
662 "Understanding the Tungsten Life Cycle in Europe" workshop organised by the British
663 Geological Survey, with data from historical BGS World Mineral Production publications.
664 London, UK, 5 December, 2012.
- 665 Brown, T. and P.Pitfield. 2014. "Tungsten" in G.Gunn (Ed) "Critical Metal Handbook" John Wiley
666 & Sons
- 667 Chanturiya, V. A. 2008. "Innovation processes in technologies for the processing of refractory
668 mineral raw materials." (Geology of Ore Deposits) 50, no. 6: 491–501.
- 669 Clemente, D., P. Newling, A. Botelho de Sousa, G. LeJeune, S.P. Barber and P. Tucker. 1993.
670 "Reprocessing slimes tailings from a tungsten mine." (Minerals Engineering) 6, no. 8-10:
671 831–839.
- 672 Cullen, J.M. and J.M. Allwood. 2013. "Mapping the Global Flow of Aluminum: From Liquid
673 Aluminum to End-Use Goods." (Environmental Science and Technology) 47: 3057-3064.
- 674 Cullen, J.M., J.M. Allwood and M.D. Bambach. 2012. "Mapping the Global Flow of Steel: From
675 Steelmaking to End-Use Goods." (Environmental Science and Technology) 46: 13048-
676 13055.
- 677 De Wang, X., B. Zhu, Y.H. Wan and X.J. Zhang. 1995. "A new technique for quick production of
678 ammonium paratungstate (APT) crystals by a liquid membrane." (Journal of membrane
679 science) 105: 55-62.
- 680 Du, X.Y. and T. E. Graedel. 2011a. "Global in-use stocks of the rare earth elements: A first
681 estimate." (Environmental Science and Technology) 45, no. 9: 4096-4101.
- 682 Du, X.Y. and T.E. Graedel. 2011b. "Uncovering the global life cycles of the rare earth elements."
683 (Scientific Reports).
- 684 European Commission. 2014. Report on critical raw materials for the EU. Report of the Ad-hoc
685 working group on defining critical raw materials. European Commission.

- 686 GRI. 2015. Global Reporting Initiative. [https://www.globalreporting.org/Information/about-](https://www.globalreporting.org/Information/about-gri/Pages/default.aspx)
687 [gri/Pages/default.aspx](https://www.globalreporting.org/Information/about-gri/Pages/default.aspx) (accessed January 30, 2015).
- 688 GTP. 2015. "Global Tungsten and Powders Corp." Tungsten Chemicals.
689 http://www.globaltungsten.com/en/products-services_tungsten-chemicals.htm (accessed
690 January 30, 2015).
- 691 Gürmen, S. and B. Friedrich. 2004. Recovery of Cobalt Powder and Tungsten Carbide from
692 Cemented Carbide Scrap – Part I: Kinetics of Cobalt Acid Leaching. Aachen, Germany:
693 IME Process Metallurgy and Metal Recycling, RWTH Aachen University.
- 694 Hairunnisha, S., G.K. Sendil, J. Prabhakar Rethinaraj, G.N. Srinivasan, P. Adaikkalam and S.
695 Kulandaisamy. 2007. "Studies on the preparation of pure ammonium paratungstate from
696 tungsten alloy scrap." (Hydrometallurgy) 85: 67-71.
- 697 Harper, E.M. 2008. "A Product-Level Approach to Historical Material Flow Analysis: Tungsten as
698 a Case Study." (Journal of Industrial Ecology) 12, no. 5/6.
- 699 Harper, E.M. and T.E. Graedel. 2008. "Illuminating Tungsten's Life Cycle in the United States:
700 1975-2000." (Environmental Science and Technology) 42: 3835–3842.
- 701 Harper, E.M., G. Kavlak and T.E. Graedel. 2012. "Tracking the metal of the goblins: Cobalt's cycle
702 of use." (Environmental Science and Technology) 46, no. 2: 1079–1086.
- 703 Haynes. 2013. Haynes International Inc. <http://www.haynesintl.com/CRAAlloys.htm> (accessed 03
704 04, 2014).
- 705 Hinde, C. (Ed). 2008. Tungsten. Mining Journal Special Publication (June 2008) pp16.
- 706 IDC. 2013. "Strong demand for smartphones and heated vendor competition characterize the
707 worldwide mobile phone market at the end of 2012." International Data Corporation.
708 <https://www.idc.com/getdoc.jsp?containerId=prUS23916413> (accessed 02 10, 2013).
- 709 ITIA. 2009. Tungsten Brochure. International Tungsten Industry Association
- 710 ITIA. 2010. Cemented Carbides - A Success Story. International Tungsten Industry Association.
711 June 2010. http://www.itia.info/assets/files/Newsletter_2010_06.pdf (accessed 01 14, 2013).
- 712 ITIA. 2011a. Tungsten supply and demand. International Tungsten Industry Association.
713 <http://www.itia.info/supply-and-demand.html> (accessed 01 14, 2013).
- 714 ITIA. 2011b. Primary Uses of Tungsten. International Tungsten Industry Association.
715 <http://www.itia.info/tungsten-primary-uses.html> (accessed 01 14, 2013).
- 716 ITIA. 2011c. Tungsten Chemicals and their Applications. International Tungsten Industry
717 Association. June 2011.
718 http://www.itia.info/assets/files/newsletters/Newsletter_2011_06.pdf (accessed 01 14,
719 2013).
- 720 ITIA. 2011d. Ferro-Tungsten and Melting Base. International Tungsten Industry Association.
721 <http://www.itia.info/ferro-tungsten-melting-base.html> (accessed January 30, 2015).
- 722 ITIA. 2011e. Tungsten in Steel. International Tungsten Industry Association.
723 <http://www.itia.info/tungsten-in-steel.html> (accessed 01 14, 2013).

- 724 ITIA. 2011f. Tungsten Heavy Metal Alloys. International Tungsten Industry Association.
725 <http://www.itia.info/tungsten-heavy-metal-alloys.html> (accessed 01 14, 2013).
- 726 Jakhu, M.R. and S. Ray. 1996. "Degana tungsten project—present plant practice and future
727 scenario." (Bulletin of Materials Science) 19, no. 2: 313–329.
- 728 Krishna Rao, N. 1996. "Beneficiation of tungsten ores in India: A review." (Bulletin of Materials
729 Science) 19, no. 2: 201–265.
- 730 Lassner, E. 1995. "From tungsten concentrates and scrap to highly pure ammonium paratungstate
731 (APT)." (Int. J. of Refractory Metals & Hard Materials) 13: 35-44.
- 732 Lassner, E. and W. Schubert. 1999. Tungsten. New York, USA: Kluwer Academic/Plenum
733 Publishers.
- 734 Marinakis, K.I. and G.H. Kelsall. 1987. "The surface chemical properties of scheelite (CaWO₄) –
735 II. Collector adsorption and recovery of fine scheelite particles at the Iso-Octane/Water
736 interface." (Colloids and surfaces) 26: 243-255.
- 737 MetalBulletin. 2015. Chinese tungsten concentrates price rise on SRB rumour. MetalBulletin.
738 [http://www.metalbulletin.com/Article/3424596/Chinese-tungsten-concentrates-prices-rise-
739 on-SRB-rumour.html#axzz3ay4wLDtP](http://www.metalbulletin.com/Article/3424596/Chinese-tungsten-concentrates-prices-rise-on-SRB-rumour.html#axzz3ay4wLDtP) (accessed May 10, 2015).
- 740 Morley, N. and D. Eatherley. 2008. Material Security - Ensuring Resource Availability for the UK
741 Economy. Report prepared by Oakdene Hollins Ltd for the Resource Efficiency Knowledge
742 Transfer Network.
- 743 Nansai, K., K. Nakajima, S. Kagawa, Y. Kondo, S. Suh, Y. Shigetomi, and Y. Oshita. 2014.
744 "Global flows of critical metals necessary for low-carbon technologies: the case of
745 neodymium, cobalt and platinum." Environmental Science and Technology 48, no. 3: 1391-
746 1400.NATC. 2013. North American Tungsten Corporation Ltd.
747 <http://www.natungsten.com/s/Cantung.asp> (accessed 07 12, 2013).
- 748 Reuter, M.A., C. Hudson, A. van Schaik, K. Heiskanen, C. Meskers and C. Hagelüken. 2013. Metal
749 recycling: opportunities, limits, infrastructure. A report of the working group on the global
750 metal flows to the international resource panel. Nairobi, Kenya: UNEP.
- 751 RPA. 2012. Study on data needs for a full raw materials flow analysis. Prepared for Directorate-
752 General Enterprise and Industry. London, UK: Risk and Policy Analysis Limited.
- 753 Schmidt, M. 2008. "The Sankey Diagram in Energy and Material Flow Management". (Journal of
754 Industrial Ecology) 12, no. 2.
- 755 Shedd, K. 2015. Tungsten Mineral Commodity Summary. United States Geological Survey.
- 756 Smith, G.R. 1994. Material Flow of Tungsten in the United States. US Department of the Interior.
- 757 Suchkov, A.B., G.V. Rummyantseva, A.R. Demachev and N.V. Zhukova. 1971. "Electrolytic
758 preparation of tungsten powder." (Poroshkovaya Metallurgiya) 12: 1-3.
- 759 Tercero Espinoza, L.A. 2012. "The contribution of recycling to the supply of metals and minerals".
760 POLINARES working paper number 20. European Commission.

761 THPP. 2014. Tungsten Heavy Powder and Parts. Tungsten powder specs.
762 http://www.tungstenheavypowder.com/Tungsten_Heavy_Powder/Tungsten_Heavy_Powder
763 [_Products/tungsten_heavy_powder_products.html](http://www.tungstenheavypowder.com/Tungsten_Heavy_Powder/Tungsten_Heavy_Powder_Products/tungsten_heavy_powder_products.html) (accessed January 30, 2015).

764 Tungsten Carbide Recycling. 2015. Tungsten Carbide Scrap Price UK. Tungsten Carbide
765 Recycling. <http://tungstencarbidescrap.co.uk/tungsten-carbide-scrap-price-uk/> (accessed
766 May 10, 2015).

767 USDOE. 2007. Mining industry energy bandwidth study. US Department of Energy, Industrial
768 Technologies Program.

769 USGS. 2011. Tungsten recycling in the United States in 2000. US Geological Survey Circular
770 1196-R. <http://pubs.usgs.gov/circ/circ1196-R> (accessed 02 02, 2013).

771 USGS. 2013. Metal prices in the United States through 2010. Reston, VA: US Geological Survey,
772 National Minerals Information Center.

773 WBH. 2013. "Visit to Wolfram Bergbau und Hütten, Mittersill tungsten mine." Mittersill, Austria.

774 Weiss, N. L. 1985. SME Mineral Processing Hand book. New York, USA: Society of Mining
775 Engineers (SME).

776 Wills, B. A. 1988. "Mineral Processing Technology: an Introduction to the Practical Aspects of Ore
777 Treatment and Mineral Recovery" Oxford, England: Pergamon Press.

778 Yoshimura, A., I. Daigo and Y. Matsuno. 2011. "Construction of global scale substance flow of
779 indium from mining." (J. Japan Inst. Metals) 75, no. 9: 493-501.

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