51st Lunar and Planetary Science Conference (2020)

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INVESTIGATING RELATIONSHIPS BETWEEN GEOCHEMISTRY AND PHYSICAL GRAIN CHARACTERISTICS ALONG A GLACIO-FLUVIAL-AEOLIAN SEDIMENT TRANSPORT PATHWAY USING μXRF. E. Champion¹, R.C. Ewing¹, M. Nachon¹, E.B. Rampe², B. Horgan⁵, M.G.A., Lapotre⁴, M.T. Thorpe², C.C. Bedford³, P. Sinha⁵, K. Mason¹, M. Tice¹, ¹Texas A&M University, ²NASA Johnson Space Center, ³LPI/USRA/JSC, ⁴Stanford University, ⁵Purdue University, (emily.shae8@gmail.com)

Introduction: Iceland's basaltic volcanic rocks and glacial, fluvial, and aeolian landscapes resemble those studied on Mars, which makes it an ideal location to study the evolution of basaltic landscapes, the weathering and alteration of basaltic sediments in cold and wet environments, and the generation of a basaltic sedimentary record [1]. The SAND-E: Semi-Autonomous Navigation for Detrital Environments project examines physical and chemical changes in sediments transported through basaltic fluvial and aeolian environments, and tests operational scenarios, which include a drone and a robotic rover instrumented with autonomous terrain analysis software [2]. As part of the SAND-E project, we examined a glacial outwash plain at Skjaldbreiðauhraun, in SW Iceland.

This study uses micro X-ray fluorescence (μ XRF) to examine the chemical and physical properties of unconsolidated sediment-size fractions from 710 μ m to < 63 μ m along a downstream transect from a glacial sourced watershed. μ XRF is ideally suited for this task because it maps elemental distributions at sub-grain scales thereby allowing a direct correlation between grainsize, grain shape, and chemistry. It is also a good analog technique for the Mars 2020 mission equipped with the PIXL (Planetary Instrument for X-ray Lithochemistry) that will be deployed at Jezero Crater, Mars [3].



Figure 1: Sediments analyzed in the μ XRF. (left) Sediments were sieved and different size fractions placed on tape for analysis in the μ XRF. The μ XRF scanned a 5.12 by 5.12 mm area of the sample. (middle) Fe element map of the 710 μ m fraction. (right) Optical stereo microscope image of sample spatially correlated to the Fe element map.

Methods: Preliminary analysis of six out of 217 sediment samples were analyzed by μ XRF for this study. The samples were collected from the þórisjökull glacial runoff system in SW Iceland. Samples were selected based on transport environment and location. Six fluvial and aeolian samples were chosen at distances 6.3 km (proximal), 11.3km (medial), and 14.4 km (distal) to the þórisjökull glacier. Samples were taken from wind ripples and Samples were taken from the bed of a braided river channel.

Samples were sieved to five grainsize separates at 710 μ m, 250 μ m, 125 μ m, 63 μ m, and <63 μ m. Non-separated bulk samples were analyzed using the CAMSIZER to produce grainsize and shape characteristics, such as sphericity, elongation, and angularity (Fig. 2).



Figure 2: Cumulative grainsize distribution of six samples in this study. The 10^{th} (d10), 50^{th} (d50) and 90^{th} (d90) percentiles are reported in the table.

Each grainsize separate was analyzed using a μ XRF benchtop Horiba XGT7000 μ XRF with a 10 μ m probe head over a 5.12 x 5.12 mm area generating a 512 x 512-pixel image. Pixel brightness on the maps represents pseudo-intensity, which is the uncalibrated concentration of that element (Fig. 1, middle). Composite color maps show relationships among different elements.

Results: Bulk grain size analysis shows that the median grainsize of these aeolian samples is finer than the fluvial, though variability exists in the system (e.g., [4]), Furthermore, the sediment size fines farther from the glacial source area (Fig. 2). The fluvial samples are more poorly sorted than the aeolian samples, as indicated by the less steep cumulative distribution curve (Fig. 2).

The μ XRF color maps were used to determine where are concentrated in specific grains. In Figure 3, Fe (displayed in yellow) stands out as bright patches on the grains. Certain areas of the grains contain higher concentrations of Ca (blue). Mg (red) does not appear strongly in these grains. This lack of Mg is unexpected due to the high concentration of Mg in the sediments and sediment source rocks [6] which could be due to the insensitivity of the XRF to lighter elements.



Figure 3: The 250 μ m sample is shown with a highresolution camera on the ZEISS Stemi508 microscope (A). The color maps are produced from μ XRF (B,C,D). Map B is the combination of maps C and D.



Figure 4: µXRF color maps (16-bit images) of five grain sizes of the proximal aeolian sample. Fe is yellow, Ti is blue, and Mg is red. All maps are 5.12 mm by 5.12 mm.

Figure 4 shows the elemental variability of each grainsize fraction within one aeolian sample. In all 5 maps, the Fe signal is strong and, similar to Fig. 3, presents as bright patches on the grains, however the Mg and Ti signal is weak. Ti has similar overall abundance to Mg, but as indicated by the brightness is slightly more concentrated in some grains rather than others.

Discussion and Conclusions: Relationships between geochemistry and physical characteristics can be made with μ XRF scans and manual measurements. The high intensity of FE, Mg, Ca, and Ti are typical of basaltic minerals like those found in this system [see 6 for XRD analysis]. The patchy presentation of the Fe may be due to multimineralic grains or surficial weathering products. In some grains, the Fe patches correlate to the vesicles in the grains.

Future work will develop a method to effectively correlate μ XRF elemental abundances to individual particle grain characteristics. This would be useful determining grain-scale physical and chemical processes affecting sedimentary particles within the system. Further, such methods could be applied to future PIXL Mars2020 rover data.

References: [1] Mangold N. et al. (2011) Earth and Planetary Science Letters 310.3-4: 233-243. [2] Ewing et al. (2019), SAND-E: Semi-Autonomous Navigation for Detrital Environments First Results, AGU Fall Meeting Abstract EP24A-05 [3] Allwood A. et al. (2013) AGU Fall Meeting Abstracts, P51F-1786. [4] Mason et al. *this meeting*. [5] Bedford et al. *this meeting*. [6] Rampe et al. *this meeting*.