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Carbon monoxide conversion by thermophilic sulfate-reducing bacteria in pure culture and in co-culture with *Carboxydothemus hydrogenoformans*

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Abstract Biological sulfate (SO₄) reduction with carbon monoxide (CO) as electron donor was investigated. Four thermophilic SO₄-reducing bacteria, *Desulfotomaculum thermoacetoxidans* (DSM 5813), *Thermodesulfovibrio yellowstonii* (ATCC 51303), *Desulfotomaculum kuznetsovii* (DSM 6115; VKM B-1805), and *Desulfotomaculum thermobenzoicum* subsp. *thermosyntrophicum* (DSM 14055), were studied in pure culture and in co-culture with the thermophilic carboxydrotrophic bacterium *Carboxydothemus hydrogenoformans* (DSM 6008). *D. thermoacetoxidans* and *T. yellowstonii* were extremely sensitive to CO: their growth on pyruvate was completely inhibited at CO concentrations above 2% in the gas phase. *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* were less sensitive to CO. In pure culture, *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* were able to grow on CO as the only electron donor and, in particular in the presence

of hydrogen/carbon dioxide, at CO concentrations as high as 50–70%. The latter SO₄ reducers coupled CO oxidation to SO₄ reduction, but a large part of the CO was converted to acetate. In co-culture with *C. hydrogenoformans*, *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* could even grow with 100% CO ($P_{CO}=120$ kPa).

Introduction

A mesophilic process that applies biological reduction of sulfate (SO₄)/sulfite (SO₃) to hydrogen sulfide (H₂S), and subsequent biological conversion of the produced H₂S to elemental sulfur (S), is suggested as a cost-effective method for the removal of S compounds from waste streams (Maree et al. 1987; Lens et al. 1998). Thermophilic treatment of SO₄/SO₃ rich wastewater is an attractive alternative for the currently employed mesophilic treatment of hot wastewater of paper and pulp industries or for the conventional process of flue gas desulfurization.

Many SO₄-rich wastewaters are poor in organic matter. Therefore, a supply of an appropriate electron donor is essential to reduce SO₄. Hydrogen (H₂) is an excellent electron donor for SO₄ reduction (Widdel and Hansen 1992; van Houten et al. 1994, 1997). Synthesis gas is a cheap source of H₂-rich gas. It is produced by steam reforming of natural gas or by thermal gasification of coal, oil, biomass, or other organic matter (Graboski 1984). Synthesis gas, depending on its origin, typically contains H₂ (30–76 vol%), carbon monoxide [(CO) 15–59 vol%], carbon dioxide [(CO₂) 8–27 vol%], and traces of methane, nitrogen (N₂), and hydrogen sulfide (H₂S) (Perry et al. 1997). The major restriction of synthesis gas utilization for biological S removal is the relative high percentage of CO.

Some SO₄-reducing bacteria can use CO as an electron donor (Mörsdorf et al. 1992; Davidova et al. 1994). Nevertheless, they are also strongly inhibited by CO (Davidova et al. 1994). *Desulfotomaculum orientis* and *Desulfotomaculum nigrificans* grow slowly on CO up to 20% in the gas phase (Klemps et al. 1985), as does *Desulfovibrio desulfuricans* (Karpilova et al. 1983). *Desulfovibrio vul-*

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garis oxidizes CO (maximum 4.5%) to CO₂ coupled to H₂ formation, which is subsequently used as an electron donor for SO₄ reduction (Lupton et al. 1984). Biological SO₄ reduction with a H₂/CO mixture as electron donor was studied in mesophilic lab-scale gas-lift reactors (van Houten et al. 1996). SO₄ reduction was observed with 20% of CO in the feed gas. However, 5% CO already resulted in lower rates of SO₄ reduction. The microbial population of the CO-fed reactor mainly consisted of *Desulfovibrio* and *Acetobacterium* species. The authors speculated that a main part of the CO was converted by homoacetogens, preventing CO toxicity for SO₄-reducing bacteria. Several anaerobic bioreactor sludges at 55°C were able to convert 100% CO in SO₄-free media to H₂ or to methane via H₂ as intermediate (Sipma et al. 2003). Recently, it was demonstrated that in anaerobic bioreactor sludges, both CO and H₂ were used by SO₄-reducing bacteria that tolerated and used high CO ($P_{CO} > 1.6$ bar) concentrations (Sipma et al. 2004).

Thus, current knowledge indicates that synthesis gas is poorly suitable as electron donor for SO₄ reduction under mesophilic conditions due to the sensitivity of SO₄-reducing bacteria towards CO. The mechanism of CO inhibition of SO₄ reduction is poorly understood. Here, we aim to avoid or reduce the effect of CO inhibition on SO₄ reduction, starting from two perspectives. Recent findings indicate that CO is better tolerated under thermophilic conditions, as illustrated above. We selected four thermophilic SO₄-reducing bacteria capable of chemolithoautotrophic growth with SO₄: *Desulfotomaculum thermoacetoxidans* (Min and Zinder 1990), *Desulfotomaculum thermobenzoicum* subsp. *thermosyntrophicum* (Plugge et al. 2002), *Desulfotomaculum kuznetsovii* (Nazina et al. 1988), and *Thermodesulfovibrio yellowstonii* (Henry et al. 1994). *D. kuznetsovii* (Nazina et al. 1988) and *D. thermoacetoxidans* (Min and Zinder 1990) were characterized as a SO₄ reducers able to convert organic substrates completely to CO₂ coupled to SO₄ reduction. *D. thermobenzoicum* subsp. *thermosyntrophicum* (Plugge et al. 2002) and *T. yellowstonii* (Henry et al. 1994) oxidize organic substrates incompletely to acetate coupled to SO₄ reduction. *D. thermobenzoicum* subsp. *thermosyntrophicum* (Plugge et al. 2002) and *D. thermoacetoxidans* (Min and Zinder 1990) produce acetate and SO₄ during growth on H₂/CO₂ plus SO₄. These four strains have not been tested previously with CO. Conversion of CO to H₂ by thermophilic anaerobes may provide additional means to avoid CO inhibition. An increasing number of anaerobes grow by the conversion of CO to H₂ but do not reduce SO₄ (Fardeau 2004; Sokolova et al. 2001, 2002, 2004a, b; Svetlichnyi et al. 1991, 1994). Formation of H₂ and removal of CO by these bacteria may remove inhibition and thus stimulate SO₄ reduction. For this purpose, *Carboxydotherrmus hydrogenoformans* was selected and used in co-culture with the selected SO₄-reducing bacteria. Our results demonstrate the potential of thermophilic SO₄ reduction with synthesis gas. Further insight in CO inhibition on SO₄-reducing bacteria is provided as well.

Materials and methods

Bacterial strains and growth conditions

The following bacterial strains were used in the experiments: *C. hydrogenoformans* (DSM 6008) (Svetlichnyi et al. 1991), *D. thermoacetoxidans* (DSM 5813) (Min and Zinder 1990), *D. thermobenzoicum* subsp. *thermosyntrophicum* (DSM 14055) (Plugge et al. 2002), *D. kuznetsovii* (DSM 6115; VKM B-1805) (Nazina et al. 1988), and *T. yellowstonii* (ATCC 51303) (Henry et al. 1994). Bacteria were grown anaerobically in a basal mineral bicarbonate-phosphate buffered medium that contained (in g/l of demineralized water) KH₂PO₄ (0.38), Na₂HPO₄ (0.54), NH₄Cl (0.3), NaCl (0.3), CaCl₂·2H₂O (0.11), MgCl₂·6H₂O (0.1), NaHCO₃ (2.4), Na₂S·9H₂O (0.29), resazurin (0.0005), yeast extract (0.5 g/l), trace elements (1 ml), and vitamins (1 ml). Trace elements and vitamins were prepared as described by Stams et al. (1993). Sodium sulfate (20 mM) was added for the cultivation of SO₄-reducing bacteria. When indicated, 10 mM sodium pyruvate and 10 mM sodium sulfate were supplied. Bacteria were grown in 250-ml or 120-ml serum bottles that contained 50 ml medium and were sealed with butyl rubber stoppers and aluminum caps. The precultures were grown on H₂/CO₂ plus SO₄, for experiments with pure CO precultures were grown on 20% CO plus SO₄. In all experiments, the inoculum size was 5% of each species of bacteria. The headspace was flushed with H₂/CO₂ (80:20), CO, or different ratios of CO/N₂, CO/H₂, and CO/H₂/CO₂. The protocol for making the gas phase was as follows: 120-ml bottles with 50 ml of medium were flushed with N₂. Vacuum (0.2 bar) was created in the bottles, and CO was added to give a volume percentage (vol%) in the gas phase of 5, 20, 50 or 70 vol%. Then, N₂, H₂, or H₂/CO₂ was added to a pressure of 120 kPa (100 kPa=1 bar). For cultivation of *T. yellowstonii* with H₂/CO₂, 2 mM acetate was supplied as additional carbon source. When CO₂ was not present in the gas phase, bicarbonate was omitted from the medium and a two- to threefold higher concentration of phosphate buffer was added. Media were maintained at a pH of about 7.0. Bacteria were incubated at 60°C standing or shaken (150 rpm). For growth on CO, 2–100% of CO was initially used.

Substrates and products analyses

H₂ and CO were analyzed by a Chrompack gas chromatograph (CP9001) equipped with a TCD-detector. The capillary column was filled with fused silica (Molsieve 5A, 30 m × 0.53 mm). The oven temperature was 50°C, and the temperature of the TCD-detector was 100°C; argon was the carrier gas. Volatile free fatty acids were analyzed by HPLC as described by Stams et al. (1993). H₂S was analyzed according to Trüper and Schlegel (1964). Concentration of gaseous and liquid compounds after the analyses was expressed in mmol per liter of medium.

Results

Growth of pure cultures on pyruvate in the presence of CO

The thermophilic SO_4 -reducing bacteria used in this research have different temperature optima and temperature limits. *T. yellowstonii* grows up to 70°C , and *D. kuznetsovii* is able to grow up to 85°C , while the other tested SO_4 -reducing bacteria cannot grow above 65°C . *C. hydrogenoformans* has an optimum growth temperature of 70°C , but it grows well at 60°C . Therefore, all experiments were performed at 60°C . All selected SO_4 -reducing bacteria can grow on H_2/CO_2 with SO_4 as electron acceptor. In standing cultures at 60°C all H_2 was converted within 12 days (data not shown).

Some mesophilic SO_4 -reducing bacteria can grow with CO (or their growth is improved) only in the presence of organic carbon sources (Karpilova et al. 1983; Lupton et al. 1984). The CO sensitivity of the selected thermophilic SO_4 -reducing bacteria was tested by cultivation (without shaking) in a medium that contained pyruvate (10 mM) and SO_4 (10 mM) under an atmosphere of 0, 2, 5, 20, or 50% CO. All tested bacteria were able to ferment pyruvate without a CO-containing gas phase. Nevertheless, no vis-

ible growth of *D. thermoacetoxidans* and *T. yellowstonii* was observed when 2% CO was added. *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* grew with pyruvate under all tested CO concentrations.

Chemolithoautotrophic CO conversion by co-cultures

C. hydrogenoformans converts CO to H_2 and CO_2 and may, thus, relieve the CO toxicity for thermophilic SO_4 -reducing bacteria in the binary cultures. The two SO_4 -reducing bacteria, that were the most sensitive for CO, *D. thermoacetoxidans* and *T. yellowstonii*, were incubated with 20% and 50% CO as the only electron donor in co-culture with *C. hydrogenoformans*. CO was consumed, but no SO_4 reduction was observed (data not shown).

The co-cultures of *C. hydrogenoformans* and *D. kuznetsovii* or *D. thermobenzoicum* subsp. *thermosyntrophicum*, grown with 100% CO as sole carbon and energy source in standing cultures, converted CO and reduced SO_4 (Fig. 1a, b). When shaken, CO conversion and formation of H_2 by both co-cultures (Fig. 1a, b) occurred faster, but the fate of H_2 was different. When *C. hydrogenoformans* was cultivated with *D. kuznetsovii* without shaking (Fig. 1a), H_2 was formed gradually, and it was also consumed gradually. Overall, the H_2 concentration remained low. H_2 consumption coincided with H_2S formation. At the end of the experiment, 4.3 mM acetate was formed. Under shaken conditions, H_2 accumulated rapidly, and its further conversion occurred slowly (Fig. 1a). SO_4 reduction was inhibited (only 0.4 mM H_2S was formed) and H_2 was not consumed completely. More acetate (6.6 mM) was formed compared with the standing cultures. When *C. hydrogenoformans* was cultivated with *D. thermobenzoicum* subsp. *thermosyntrophicum* in standing cultures (Fig. 1b), the rates of H_2 formation and SO_4 reduction were similar to the rates in the co-culture with *D. kuznetsovii* (Fig. 1a). Under shaken conditions, H_2 was formed fast, but only after all CO was converted, the H_2 concentration decreased and SO_4 was reduced (Fig. 1b). At the end of the experiment, the acetate concentration was 4 mM in standing cultures and 7.5 mM in shaken cultures.

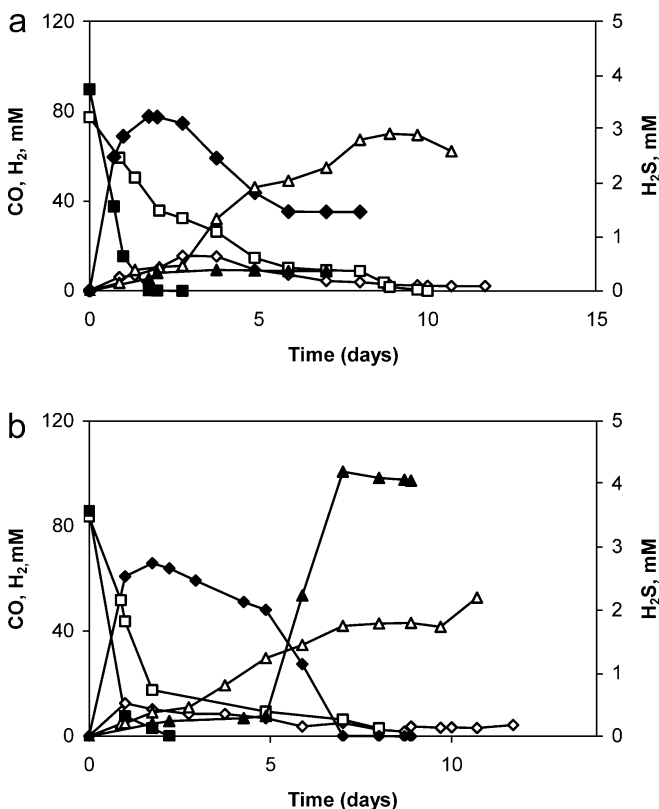


Fig. 1 Conversion of 100% carbon monoxide (CO) by binary culture of *Carboxydotherrmus hydrogenoformans* plus SO_4 -reducing bacteria at standing (open symbols) and shaking (closed symbols) conditions. Squares carbon monoxide (CO), rhombuses hydrogen (H_2), triangles hydrogen sulfide (H_2S). **a** *C. hydrogenoformans* plus *Desulfotomaculum kuznetsovii*. **b** *C. hydrogenoformans* plus *Desulfotomaculum thermobenzoicum* subsp. *thermosyntrophicum*

Conversion of mixtures of H_2 and CO by pure cultures and co-cultures

D. kuznetsovii and *D. thermobenzoicum* subsp. *thermosyntrophicum* were cultivated in the presence of SO_4 , shaken with 0, 5, 20, 50 and 70% CO in the H_2/CO_2 gas phase (Table 1). Both *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* were able to convert CO, H_2 , and CO_2 and reduce SO_4 (Table 1). Conversion of CO and a mixture of CO and H_2/CO_2 by *D. kuznetsovii* (Table 1) occurred slower than by *D. thermobenzoicum* subsp. *thermosyntrophicum* (Table 1). Data on 50% and 70% of CO in the H_2/CO_2 gas phase conversion by *D. kuznetsovii* are not shown. In our experiments, both SO_4 -reducing bacteria formed acetate. At higher CO concentra-

Table 1 Conversion of different concentrations of carbon monoxide (CO) plus hydrogen/carbon dioxide (H₂/CO₂) plus sulfate (SO₄) by *Desulfotomaculum kuznetsovii* and *Desulfotomaculum thermobenzoicum* subsp. *thermosyntrophicum*. H₂S Hydrogen sulfide

Gas phase	Time needed to complete degradation (days)		CO consumed (mmol/l)	H ₂ consumed (mmol/l)	H ₂ S formed (mmol/l)	Acetate formed (mmol/l)
	CO	H ₂				
<i>Desulfotomaculum kuznetsovii</i>						
0 CO + H ₂ /CO ₂	–	9	–	52.8	9.5	0.7
5% CO + H ₂ /CO ₂	8	14	5.0	51.2	7.5	3.6
20% CO + H ₂ /CO ₂	30	34	17.6	41.6	2.9	4.5
<i>Desulfotomaculum thermobenzoicum</i> subsp. <i>thermosyntrophicum</i>						
0 CO + H ₂ /CO ₂	–	5	–	53.1	8.3	3.5
20% CO + H ₂ /CO ₂	7	18	16.1	43.0	7.0	7.8
50% CO + H ₂ /CO ₂	10	27	39.0	49.3	8.0	7.7
70% CO + H ₂ /CO ₂	14	18	47.0	25.0	2.6	5.0

tions, more CO was used for acetate production and less for SO₄ reduction (Table 1).

Substrate conversion by *C. hydrogenoformans* alone, *C. hydrogenoformans* in co-culture with *D. thermobenzoicum* subsp. *thermosyntrophicum* grown with 50% CO (Fig. 2a,

b; Table 2) and *D. thermobenzoicum* subsp. *thermosyntrophicum* alone with N₂ plus H₂/CO₂, CO plus N₂, or CO plus H₂ in the presence of SO₄ is shown (Fig. 2c, d, e; Table 2). The co-culture converted 50% CO in the same time as a pure culture of *C. hydrogenoformans* (Fig. 2a, b).

Fig. 2 Conversion of gases by *C. hydrogenoformans* and *D. thermobenzoicum* subsp. *thermosyntrophicum* at shaking conditions. **a** *C. hydrogenoformans* on 50% CO [plus nitrogen (N₂) plus SO₄]. **b** *C. hydrogenoformans* plus *D. thermobenzoicum* subsp. *thermosyntrophicum* on 50% CO (plus N₂ plus SO₄). **c** *D. thermobenzoicum* subsp. *thermosyntrophicum* on 50% N₂ (plus H₂/CO₂ plus SO₄). **d** *D. thermobenzoicum* subsp. *thermosyntrophicum* on 50% CO (plus N₂ plus SO₄). **e** *D. thermobenzoicum* subsp. *thermosyntrophicum* on 50% CO (plus H₂ plus SO₄). Squares CO, rhombuses H₂

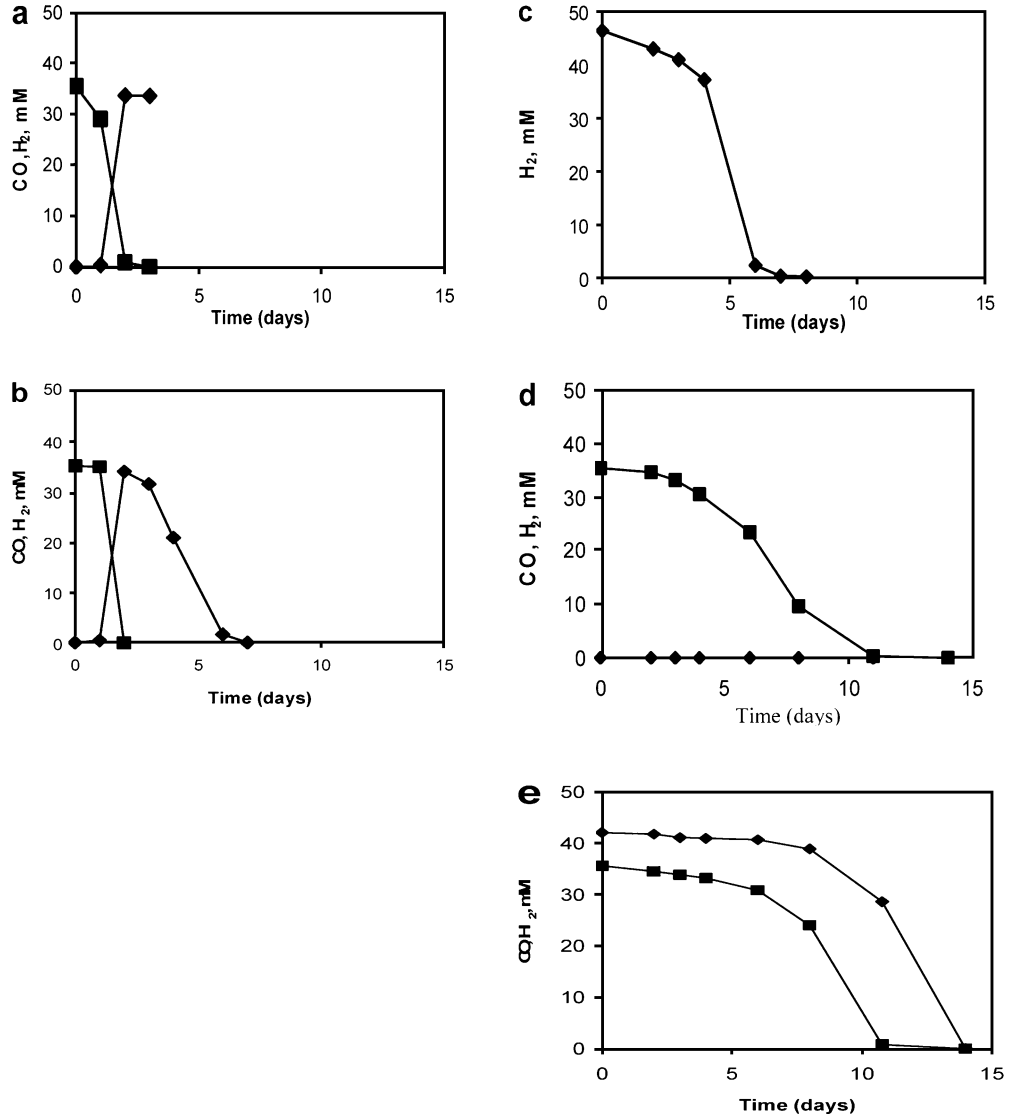


Table 2 Growth of *Carboxydotherrmus hydrogenoformans* and *D. thermobenzoicum* subsp. *thermosyntrophicum* in the atmosphere of different gases and products formation (supplement to Fig. 2)

Bacteria	Gases	Start concentration (mmol/l)	OD ^a end	H ₂ S end (mmol/l)	Acetate formed (mmol/l)
<i>C. hydrogenoformans</i>	50% CO + N ₂ + SO ₄	CO-37	nd ^b	0	0
<i>C. hydrogenoformans</i> + <i>D. thermobenzoicum</i> subsp. <i>thermosyntrophicum</i>	50% CO + N ₂ + SO ₄	CO-36	nd ^b	4.7	3.3
<i>D. thermobenzoicum</i> subsp. <i>thermosyntrophicum</i>	50% N ₂ + H ₂ /CO ₂ + SO ₄	H ₂ -48	0.19	9.2	2.5
<i>D. thermobenzoicum</i> subsp. <i>thermosyntrophicum</i>	50% CO + N ₂ + SO ₄	CO-37	0.21	5.0	4.2
<i>D. thermobenzoicum</i> subsp. <i>thermosyntrophicum</i>	50% CO + H ₂ + SO ₄	CO-36 H ₂ -42	0.31	8.2	9.5

^aOD Optical density (wavelength 660 nm)

^bnd Not determined

Products of CO and H₂ conversion are listed in Table 2. H₂S and a low amount of acetate (Table 2) were detected after complete conversion of H₂. When *D. thermobenzoicum* subsp. *thermosyntrophicum* was cultivated under a N₂/H₂/CO₂ gas phase (Fig. 2c), H₂S and a small amount of acetate were detected at the end of the experiment (Table 2). *D. thermobenzoicum* subsp. *thermosyntrophicum* could also grow with 50% CO and formed H₂S and acetate (Fig. 2d; Table 2). When grown on CO/H₂, more acetate was formed (Table 2). In this experiment H₂ consumption followed CO conversion (Fig. 2e) similar to the experiment with CO/H₂/CO₂ (Table 1).

Discussion

CO tolerance of selected SO₄-reducing bacteria

Experiments with *D. thermoacetoxidans* and *T. yellowstonii* with pyruvate/CO and co-cultivation of these strains with *C. hydrogenoformans* on 100% CO indicate that these bacteria are highly sensitive towards CO and are not applicable for biological SO₄ reduction with synthesis gas. The two other bacteria, *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum*, were remarkably tolerant to high CO concentrations. In pure culture, both bacteria tolerated 70% of CO during chemolithotrophic growth in the presence of SO₄. In co-culture, at standing conditions (with 100% CO), both *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* were able to reduce SO₄ with the H₂ formed by *C. hydrogenoformans*. However, when the co-cultures were shaken, SO₄ reduction of *D. kuznetsovii* was inhibited and only acetate was formed. Improved gas-to-liquid mass transfer of CO during shaking most likely resulted in inhibiting CO concentration in the liquid phase. In most cases, H₂ was used for acetate formation in addition to SO₄ reduction. Only one of the four investigated bacteria, *D. thermobenzoicum* subsp. *thermosyntrophicum*, was capable of SO₄ reduction under 100% CO in shaken co-culture with *C. hydrogenoformans*.

Effect of CO on the ratio of sulfidogenesis and acetogenesis

D. thermobenzoicum subsp. *thermosyntrophicum* coupled the oxidation of organic substrates to acetate formation and to SO₄ reduction. Acetate was formed during growth on H₂/CO₂ as well (Plugge et al. 2002). In our experiments, *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* formed acetate during growth on H₂/CO₂ and different CO concentrations (Table 1). Initial amounts of H₂ were less in the bottles with higher CO concentrations. Nevertheless, more acetate was formed at high CO concentrations. At CO concentrations higher than 20% for *D. kuznetsovii* (data not shown) and higher than 50–70% for *D. thermobenzoicum* subsp. *thermosyntrophicum*, SO₄ reduction was partially inhibited. Thus, more CO was used for acetate formation than for SO₄ reduction. In our experiments the electron recovery was not complete in the cultures with CO/H₂/CO₂ mixtures. We can not exclude that other organic products could have been formed.

It has been postulated that *D. vulgaris* and *D. desulfuricans* first convert CO with H₂O to H₂ and CO₂, and then use H₂ for SO₄ reduction (Karpilova et al. 1983; Lupton et al. 1984). In our experiments with pure cultures of *D. kuznetsovii* (data not shown) and *D. thermobenzoicum* subsp. *thermosyntrophicum* grown on CO (Fig. 2d), H₂ was never detected as an intermediate of CO conversion. Instead, CO₂, acetate, and H₂S were formed as products of CO conversion. When SO₄-reducing bacteria were co-cultivated with *C. hydrogenoformans*, formation of H₂ from CO by *C. hydrogenoformans* was faster than the subsequent consumption of H₂ by SO₄-reducing bacteria (Fig. 1a, b). The absence of H₂ in the gas phase (Fig. 2d) suggests a direct conversion of CO coupled to SO₄ reduction by SO₄-reducing bacteria.

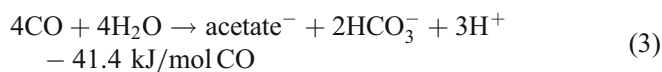
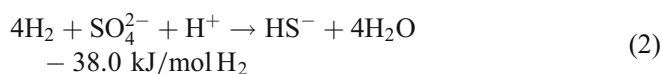
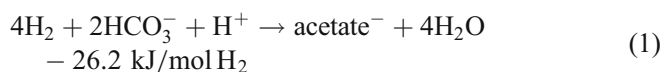
Inhibition of H₂ utilization and sulfidogenesis by CO

A typical curve of CO and H₂ conversion by *D. thermobenzoicum* subsp. *thermosyntrophicum* is shown in Fig. 2e.

Remarkably, H₂ consumption started later than CO consumption. In another experiment, the time needed to consume all H₂ became longer at higher CO concentrations (Table 1). It is generally reasoned that hydrogenase is inhibited by CO. Observations supporting this fact are omnipresent. When the acetogenic bacterium *Eubacterium limosum* was grown with a mixture of CO and H₂ (Sharak Genthner and Bryant 1982), H₂ consumption started after the CO concentration in the gas phase had decreased to values below 5%. This observation was explained by the possible inhibition of hydrogenase by CO. A similar hydrogenase inhibition was found for other anaerobic bacteria cultivated on CO (Daniels et al. 1977; Pankhania et al. 1986; Berlier et al. 1987; Adams 1990; Bennett et al. 2000).

SO₄-reducing bacteria are generally more sensitive towards CO than acetogens. CO partial pressures of 20% are the maximum tolerated by SO₄-reducing bacteria reported so far (Klemps et al. 1985; Karpilova et al. 1983). It is unlikely that inhibition of hydrogenase by CO is the sole reason of the sensitivity of SO₄-reducing bacteria towards CO. As discussed above, a shift towards acetogenesis occurs with increasing CO concentrations. Hydrogenase plays a central role in acetogenesis as well. Recently, Rother and Metcalf (2004) demonstrated growth of a *Methanosarcina acetivorans* strain on CO. Methanogenesis was largely inhibited by high CO concentrations; acetate was formed instead. It is unclear which step of methanogenesis is inhibited. We support the authors in their statement that further investigation on the physiological mechanism of inhibition by CO is necessary and like to extend this to the SO₄-reducing bacteria.

From our data we conclude that besides SO₄ reduction with H₂ and acetogenesis from H₂ and CO₂, *D. kuznetsovii* and *D. thermobenzoicum* subsp. *thermosyntrophicum* can convert CO to acetate and can couple CO oxidation directly to SO₄ reduction. Thus, we assume that these strains can perform the following reactions:



This is the first report that provides evidence that thermophilic SO₄-reducing bacteria can grow at a high concentration (50–70%) of CO. In co-culture with *C. hydrogeniformans* growth and SO₄ reduction of *D. thermobenzoicum* subsp. *thermosyntrophicum* is even possible

with 100% CO. In the latter case, CO is first converted to H₂ and CO₂, which is subsequently used by the SO₄ reducer. Our results show clearly that under moderately thermophilic conditions synthesis gas with high amounts of CO is an excellent electron donor for biotechnological SO₄ reduction at a high temperature. In particular, when the SO₄ reducers are co-cultivated with carboxydophilic bacteria, high CO concentrations are tolerated. Thus, purification of the gas to reduce the CO content is not needed. It is not certain that in the bioreactor sludges the tested bacteria are dominant. However, as moderately thermophilic carboxydophilic bacteria and SO₄-reducing bacteria can be easily enriched from different sources, it is not unlikely that they occur in bioreactors operated with CO as well. We show here unknown capacities of SO₄-reducing bacteria and possible pathways of CO conversion in bioreactors.

Our observations are important for the application of synthesis gas for biological S removal from flue gas or wastewater discharged at a high temperature.

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