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Ultrafast Charge Separation Dynamics of Twisted Intramolecular Charge Transfer State (TICT) in Coumarin Dye Sensitized TiO₂ Film: A New Route to Achieve Higher Efficient Dye-Sensitized Solar Cell

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Abstract. Ultrafast transient spectroscopy of 7-diethyl amino coumarin 3-carboxylic acid (D-1421) sensitized TiO_2 film reveals that TICT states facilitate higher charge separation and slow recombination and proved to be new route to design higher efficient solar cell.

Introduction

Interfacial electron transfer (ET) between molecular adsorbate and semiconductor nanoparticles is an intense area of current research work [1]. Investigations in this area are strongly motivated by both its fundamental importance and practical applications such as solar energy conversion [2]. Development of higher efficient dye-sensitized solar cell (DSSC) through designing proper sensitizing molecule is still a challenging task. It is strongly believed designing a proper sensitizer molecule can do the wonder. The photophysical properties of coumarin dyes are very sensitive towards rotational hindrance of 7-amino derivatives. The rotation of amine functionality results in excited state conformers viz. near planar and near perpendicular twisted conformers3. Photoexcitation to locally excited (LE) states leads to intramolecular charge transfer (ICT) and twisted intramolecular charge transfer (TICT) states which depend upon rotating ability of 7-amino moiety in local environment [2]. D1421 and C343 dyes are known to form TICT and ICT excited states, respectively [3]. The rotating ability of $-NEt_2$ group stabilizes intramolecular charge separation of TICT state more than that of ICT state [4]. Thus, TICT states can be useful in achieving an efficient charge separation across dye/TiO₂ system. So, in order to explore the role of TICT and ICT excited state on interfacial charge separation (CS) in dye/TiO₂ system, femtosecond transient absorption studies has been carried out on TiO₂ film sensitized with D1421 and C343 dyes (Scheme 1) in acetonitrile (ACN) medium under neutral and acidic condition. The kinetic overdrive of electron injection with respect to LE—ICT/TICT internal conversion and then secondary electron donating effects of NEt₂ on back electron transfer (BET) reactions are addressed in present studies.

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Scheme 1: Molecular structure of Coumarin D1421 and Coumarin 343

Results and Discussion

Ultrafast transient absorption studies have been carried out in D-1421 and C-343 sensitized TiO₂ film in acetonitrile solvent environment exciting the samples at 400 nm laser pulse and shown in Figure 1 and Figure 2 respectively. Transient spectra clearly shown absorption band at 550 nm to 700 nm regions which can be attributed to cation radical of coumarin dyes and broad absorption band at 700 nm to 1000 nm attributed to injected electron in TiO₂. The strong electronic coupling due to β-keto carboxylate binding of coumarin dyes on TiO₂ NPs surface also known for its adiabatic electron injection process. Transient absorption studies clearly confirmed pulse width limited (<100fs) electron injection kinetics in both C343/TiO2 and D1421/TiO2 systems. Separate ultrafast transient absorption studies [5] have been carried out for free C-343 and D-1421 system in both low polar (methyl cyclohexane) and high polar (acetonitrile) solvent, which revealed that the formation of TICT states takes place only in the excited state of D-1421. However ultrafast electron injection kinetics for both the systems establishes that the strong electronic interaction facilitates electron injection from LE states prior to LE \rightarrow TICT internal conversion. The coumarin^{•+} cation of D1421 dye is then stabilized in TICT state (Scheme 2) which is evidenced by growth of 630nm TA signal in D1421/TiO₂ system as compared to C343/TiO₂ system (Figure 3A). Back electron transfer (BET) kinetics has been monitored by following the kinetics at both 640 nm (cation radical of coumarin) and at 1000 nm (injected electron in TiO_2). Interestingly it has been observed much slower BET kinetics in D1421/TiO₂ film (CS is 65% after 1ns) as compared to C343/TiO₂ system



Fig. 1.Transient absorption spectra of D 1421 sensitized TiO_2 film at different time delay after excitation at 400 nm laser light.

Fig 2. Transient absorption spectra of C-343 sensitized TiO_2 film at different time delay after excitation at 400 nm laser light.

(CS is 18% after 1ns) (Figure 3A). To confirm the effect of TICT state in D-1421/TiO₂ system, transient studies have been carried out in acidic condition in both the systems (Figure 3B).

Interestingly exactly similar transient kinetics has been observed in both the systems in acidic condition. In acidic condition, the protonation of NEt₂ group of D1421 dye suppress the TICT state formation and both the dye comprises intrinsic ICT states only. As a result no difference in kinetics has been observed. The remarkable improvisation of BET process in D1421/TiO₂ film as compared to C343/TiO₂ film in ACN solvent is solely attributed to TICT behaviour of D1421 dye. Comparison of IPCE and IV characteristics of D1421/TiO₂ and C343/TiO₂ system clearly demonstrated that D1421 is better sensitizer for higher photocurrent and photo-voltage as compared to C-343 [5].



Fig. 3. Electron injection and BET kinetics monitored by coumarin^{\bullet^+} cation at 630nm in (a) D1421/TiO₂ and (b) C343/TiO₂ film; (A) Left Panel: ACN solvent; (B) Right Panel: ACN-HNO₃ solvent.

Scheme 2: Schematic diagram of IET process in $D1421/TiO_2$ system.

Conclusions

Ultrafast transient absorption spectroscopy has been carried out to study the effect of molecular structure in photoinduced electron-transfer dynamics in coumarin (D-1421 and C-343) dye-sensitized TiO_2 film, where photoexcited D-1421 molecule exhibits TICT state. Electron injection dynamics found to pulse-width limited in both cases however back ET reaction found to be much slower in D-1421/TiO₂ system as compared to that in D-1421/TiO₂ system. On protonation both the systems show exactly same BET dynamics which clearly suggest that TICT state of D-1421 is responsible for higher electron injection yield and slower back ET reaction. This observation is also reflected in IPCE and IV characteristics of the above systems.

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