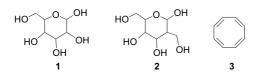
## From cyclic polyenes to carbohydrates: synthesis of the hexose sugar β-allose and its 2C-branched homologue from cyclooctatetraene

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Abstract—In an unconventional but interesting synthetic enterprise, the commercially available hydrocarbon cyclooctatetraene (COT) has been elaborated to the rare hexose sugar (DL)- $\beta$ -allose and its 2C-branched analogue. The synthetic sequence delineated here is notable for its high regio- and stereoselectivity and is flexible enough to enable access to polyoxygenated systems, hexose sugars, and their siblings from a cyclic polyene precursor.

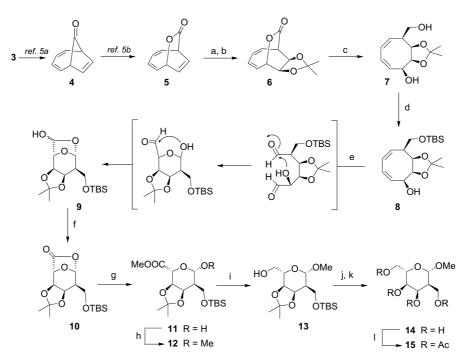
Hexose sugars 1 are among nature's premier and ubiquitous building blocks that are essential for the sustenance of diverse biological systems and processes.<sup>1</sup> Though many hexose sugars are readily available and are plentiful, their architecture with a network of hydroxyl functionalities, and stereochemical nuances has always posed an attractive ongoing challenge to synthetic chemists.<sup>2</sup> During the last few decades, a variety of new and interesting strategies have been developed for the synthesis of hexose sugars and their Cbranched siblings. Branched hexose sugars (e.g., 2) are interesting in their own right as they constitute the glycosidic component of many antibiotics and have also received a great deal of attention from synthetic chemists.<sup>3</sup> While many synthetic approaches to hexose sugars and their branched analogues have been explored,<sup>2,3</sup> the possibility of employing a cyclic polyene like cyclooctatetraene 3 (COT) for their synthesis appealed to us as an esoteric and interesting proposition. Herein, we report the transformation of COT 3 into a rare hexose sugar (DL)-β-allose and its 2C-branched sibling.<sup>4</sup>



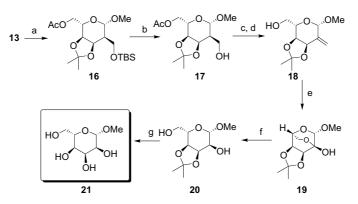
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Our synthetic approach to hexose sugars emanated from bicyclo[4.2.1]nona-2,4,7-trien-9-one 4, a 'functionally locked' cyclooctatetraene, readily available from 3 in a single pot operation as described by Shechter and coworkers.<sup>5a</sup> Baeyer-Villiger oxidation of 4 led to the lactone 5 and further catalytic OsO<sub>4</sub> dihydroxylation, and acetonide protection led to 6 with complete regioand stereocontrol, Scheme 1.<sup>5b,6</sup> LAH reduction of **6** led to the cyclooctadienediol **7** in which the primary hydroxyl group was selectively protected as the TBS derivative 8. Ozonolysis of 8 and PCC oxidation of the resulting product led to the bicyclic lactone 10 through the intermediacy of the lactol 9 as depicted in Scheme 1.<sup>6</sup> Methoxide mediated lactone opening in 10 furnished 11 and the anomeric hydroxyl group was protected as the methyl ether 12. LAH reduction of 12 revealed the branched sugar 13 and further deprotections led to (DL)methyl-2-deoxy-2C-hydroxymethyl- $\beta$ -allose 14. The branched hexose 14 was transformed to the tetraacetate 15 and its X-ray crystal structure determination<sup>7</sup> unambiguously secured its formulation.

Next, the 2C-branched precursor 13 was elaborated to the rare hexose  $\beta$ -allose. Protection of the C<sub>5</sub>-hydroxylmethyl as an acetate 16 and TBS deprotection furnished 17, Scheme 2.<sup>6</sup> The primary hydroxyl group in 17 was transformed to the terminal olefin 18 via mesylate formation and base mediated elimination. Ozonolysis of 18 furnished the intermediate hemiacetal 19, which was reduced with sodium borohydride to furnish 20, Scheme 2.<sup>6</sup> Acetonide deprotection in 20 delivered



Scheme 1. Reagents and conditions: (a) OsO<sub>4</sub>, NMMO, 75%; (b) 2,2-dimethoxypropane, acetone, CSA, 65%; (c) LiAlH<sub>4</sub>, THF, 80%; (d) TBSCl, Im, DMF, 54%; (e) O<sub>3</sub>, DCM–MeOH, DMS; (f) PCC, NaOAc, DCM, 40% for two steps; (g) NaOMe, MeOH; (h) MeI, Ag<sub>2</sub>O, 73% for two steps; (i) LiAlH<sub>4</sub>, THF, 85%; (j) TBAF, THF, 70%; (k) Amberlyst-15, MeOH, 65%; (l) Ac<sub>2</sub>O, py, DMAP, 90%.



Scheme 2. Reagents and conditions: (a) Ac<sub>2</sub>O, DMAP, DCM, 92%; (b) TBAF, THF, 74%; (c) MsCl, Et<sub>3</sub>N, DCM, 65%; (d) KO'Bu, DMSO, 70%; (e) O<sub>3</sub>, DCM, DMS, 75%; (f) NaBH<sub>4</sub>, MeOH, 80%; (g) Amberlyst-15, MeOH, 60%.

(DL)-methyl- $\beta$ -allopyranoside **21** whose spectral characteristics were identical with those reported in the literature.<sup>8</sup>

In short, we have accomplished an interesting elaboration of a commercially available polyene (COT) into  $\beta$ allose and a 2C-branched congener through a strategy that should be amenable for adaptation to access other hexose sugars and some densely oxygenated systems.

## Acknowledgements

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## **References and notes**

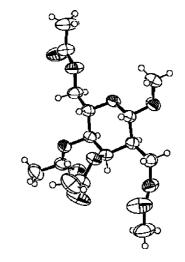
- (a) Collins, P. M.; Ferrier, R. J. Monosaccharides: Their Chemistry and Their Roles in Natural Products. 1st ed.; Wiley, 1995 (b) Bashkin, J. K. Chem. Rev. 2000, 100, 4265; (c) Dwek, R. A.; Butters, T. D. Chem. Rev. 2002, 102, 283.
- (a) Ko, S. Y.; Lee, A. W. M.; Masamune, S.; Reed, L. A., III; Sharpless, K. B.; Walker, F. J. Science 1983, 220, 949; (b) Vogel, P.; Fattori, D.; Gasparini, F.; Le Drian, C. Synlett 1990, 173; (c) Hudlicky, T.; Entwistle, D. A.; Pitzer, K. K.; Thorpe, A. Chem. Rev. 1996, 96, 1195; (d) Takeuchi, M.; Taniguchi, T.; Ogasawara, K. Synthesis 1999, 341; (e) Taniguchi, T.; Takeuchi, M.; Kadota, K.; ElAzab, A. S.; Ogasawara, K. Synthesis 1999, 1325; (f) Honzumi, M.; Taniguchi, T.; Ogasawara, K. Org. Lett. 2001, 3, 1355.
- (a) Yoshimura, J. Adv. Carbohydr. Chem. Biochem. 1984, 42, 69; (b) Wagner, J.; Vogel, P. Tetrahedron 1991, 47, 9641; (c) Beyer, J.; Madsen, R. J. Am. Chem. Soc. 1998,

120, 12137; (d) Kato, K.; Yamamura, S. Nucleosides Nucleotides 1999, 18, 657.

- (a) Fukami, H.; Koh, H.-S.; Sakata, T.; Nakajima, M. *Tetrahedron Lett.* **1967**, 4771; (b) Brimacombe, J. S.; Hunedy, F.; Husain, A. J. Chem. Soc. C. **1970**, 1273; (c) Singh, U. P.; Brown, R. K. Can. J. Chem. **1971**, 49, 1179; (d) Dondoni, A.; Fantin, G.; Fogagnolo, M.; Medici, A. *Angew. Chem., Int. Ed. Engl.* **1986**, 25, 835; (e) Vogel, P.; Auberson, Y. *Helv. Chim. Acta* **1989**, 72, 278; (f) Malleron, A.; David, S. New J. Chem. **1996**, 20, 153.
- (a) Antkowiak, T. A.; Sanders, D. C.; Trimitsis, G. B.; Press, J. B.; Shechter, H. J. Am. Chem. Soc. 1972, 94, 5366;
  (b) Mehta, G.; Pallavi, K. Chem. Commun. 2002, 2828.
- 6. All compounds reported here are racemic and all new compounds were characterized on the basis of their IR, <sup>1</sup>H and <sup>13</sup>C NMR and mass spectral data. Selected spectral data: Compound 10: IR (cm<sup>-1</sup>) 1802; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 6.04 (s, 1H), 4.56–4.51 (m, 2H), 4.24 (d, J = 7.2 Hz, 1H), 3.90 (dd, J = 10.8, 5.7 Hz, 1H), 3.72 (t, J = 10.5 Hz, 1H), 2.34–2.26 (m, 1H), 1.47 (s, 3H), 1.30 (s, 3H), 0.89 (s, 9H), 0.06 (s, 6H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  107.6, 109.6, 104.5, 73.8, 69.8, 69.4, 60.1, 40.9, 25.9, 25.8 (3C), 24.5, 18.2, -5.5, -5.4; Mass (EI, 70 eV): m/z 345  $(M+1)^+$ . Compound 12: <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ 4.58 (dd, J = 5.5, 4.0 Hz, 1H), 4.40 (d, J = 9.0 Hz, 1H), 4.34 (dd, J = 7.5, 5.5 Hz, 1H), 4.06 (d, J = 7.0 Hz, 1H), 3.83-3.79 (m, 1H), 3.79 (s, 3H), 3.70 (t, J = 10.0 Hz, 1H), 3.47 (s, 3H), 2.12–2.09 (m, 1H), 1.50 (s, 3H), 1.36 (s, 3H), 0.88 (s, 9H), 0.05 (s, 6H);  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ 170.2, 109.5, 100.6, 74.7, 72.5, 72.2, 59.5, 56.5, 52.4, 43.5, 27.9, 25.8 (3C), 25.6, 18.3, -5.6, -5.5. Compound 14: <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD)  $\delta$  4.58 (d, J = 9.0 Hz, 1H), 4.17 (t, J = 3.0 Hz, 1H), 3.90-3.86 (m, 1H), 3.75-3.69 (m, 4H),3.44 (s, 3H), 3.46 (dd, J = 9.5, 3.0 Hz, 1H), 1.73–1.68 (m, 1H); <sup>13</sup>C NMR (75 MHz, CD<sub>3</sub>OD) δ 101.7, 75.3, 70.0, 69.7, 63.3, 60.2, 56.9, 48.9; LRMS: *m*/*z* 231.1 (M+Na)<sup>+</sup>; HRMS for C<sub>8</sub>H<sub>16</sub>O<sub>6</sub>Na. Calcd: 231.0845. Found: 231.0864. Compound 18: IR (cm<sup>-1</sup>) 3470, 937; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) & 5.51 (s, 1H), 5.43 (s, 1H), 5.08 (s, 1H), 4.80 (d,  $J = 6.0 \,\text{Hz}, 1 \text{H}$ , 4.16 (t,  $J = 7.2 \,\text{Hz}, 1 \text{H}$ ), 3.84 (br d, J = 10.8 Hz, 1H), 3.71 (br d, J = 5.4 Hz, 1H), 3.65–3.60 (m, 1H), 3.53 (s, 3H), 1.50 (s, 3H), 1.38 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) & 139.0, 117.6, 109.9, 100.3, 75.9, 75.4, 73.3, 63.1, 56.2, 27.7, 25.7; LRMS: *m/z* 253.1052  $(M+Na)^+$ ; HRMS for  $C_{11}H_{18}O_5Na$ . Calcd: 253.1053. Found: 253.1091. Compound 19: IR (cm<sup>-1</sup>): 3412; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 4.93 (s, 1H), 4.41 (d 1/2 ABq, J = 7.8, 1.2 Hz, 1H), 4.37 (1/2 ABq, J = 8.1 Hz, 1H), 4.27 (dd, J = 9.9, 1.8 Hz, 1H), 4.04 (d, J = 1.5 Hz, 1H), 3.81 (d, J = 1.J = 9.9 Hz, 1H), 3.54 (s, 3H), 1.53 (s, 3H), 1.38 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 110.9, 98.9, 92.3, 75.7, 75.4, 68.5, 65.0, 55.9, 25.7, 24.5; LRMS: *m*/*z* 255.0563

(M<sup>+</sup>+Na<sup>+</sup>); HRMS for C<sub>10</sub>H<sub>16</sub>O<sub>6</sub>Na. Calcd: 255.0845. Found: 255.0846. Compound **21**: IR (cm<sup>-1</sup>) 3371; <sup>1</sup>H NMR (300 MHz, D<sub>2</sub>O)  $\delta$  4.42 (d, J = 8.7 Hz, 1H), 3.96 (t, J = 3.3 Hz, 1H), 3.72 (dd, J = 12.0, 1.8 Hz, 1H), 3.63–3.58 (m, 1H), 3.49 (dd as t, J = 6.3 Hz, 1H), 3.42 (dd, J = 9.6, 2.7 Hz, 1H), 3.37 (s, 3H), 3.25 (dd, J = 8.4, 3.3 Hz, 1H); <sup>13</sup>C NMR (75 MHz, D<sub>2</sub>O)  $\delta$  101.7, 74.1, 71.6, 70.8, 67.4, 61.7, 57.6; LRMS: m/z 217.0528 (M+Na)<sup>+</sup>; HRMS for C<sub>7</sub>H<sub>14</sub>O<sub>6</sub>Na. Calcd: 217.0688. Found: 217.0704.

X-ray data for 15: X-ray data were collected at 293 K on a 7. SMART CCD-BRUKER diffractometer with graphite monochromated Mo K $\alpha$  radiation ( $\lambda = 0.7107$  Å). Structure was solved by direct methods (SIR92). Refinement was by full-matrix least-squares procedures on F<sup>2</sup> using SHELXL-97. The nonhydrogen atoms were refined anisotropically whereas hydrogen atoms were refined isotropically.  $C_{16}H_{24}O_{10}$ , MW = 376.36, colorless crystal, crystal system: monoclinic, space group: P2(1)/n, cell parameters: *a* = 8.9525 (4) Å, *b* = 20.7773 (10) Å, *c* = 11.2568 (5) Å,  $\beta$  = 111.037 (1), *V* = 1954.30 Å<sup>3</sup>, *Z* = 4, *D<sub>c</sub>* = 1.279 g cm<sup>-3</sup>, F(0 0 0) = 800.0,  $\mu$  = 0.11 mm<sup>-1</sup>. Total number of l.s. parameters = 331, R1 = 0.0472 for 3179 Fo > 4sig(Fo) and 0.0576 for all 3981 data. wR2 = 0.1400, GOF = 1.024, restrained GOF = 1.024 for all data. Crystallographic data has been deposited with Cambridge Crystallographic Data Centre. CCDC 232049. ORTEP diagram of 15 is shown below:



8. Bock, K.; Pedersen, C. Adv. Carbohydr. Chem. Biochem. 1983, 41, 27.