

A Facile Synthesis of 3-Cyclopropyl- and 5-Cyclopropyl-isoxazoles†

Okram Mukherjee Singh,^{*a} H. Junjappa^b and H. Ila^b

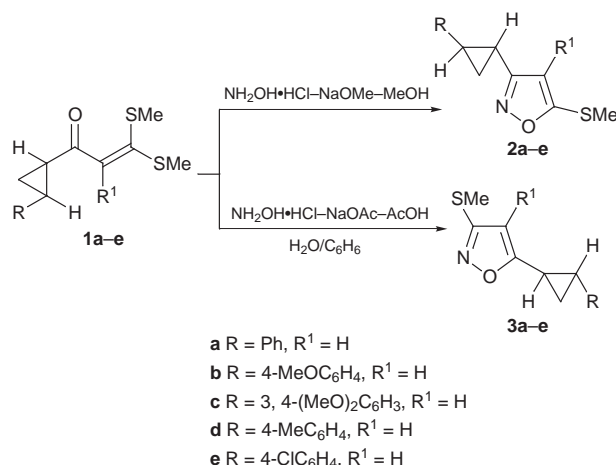
^a Department of Chemistry, Manipur University, Canchipur – 795003, Manipur, India

^b Department of Chemistry, IIT Kanpur-208016, U.P, India

J. Chem. Research (S),
1999, 398–399†

The regioselective synthesis of isomeric isoxazoles 3-(2-arylcyclopropyl)-5-methylthio- and 5-(2-arylcyclopropyl)-3-methylthio-isoxazoles is described.

Cyclopropyl ketones **1** which can be prepared in quantitative yields by the addition of dimethylloxosulfonium methylide to the corresponding α -cinnamoyl ketene dithioacetals in the presence of phase transfer catalyst¹ have been reported as useful precursors for functionalized cyclopentanones,^{2a,b} cyclopent[*a*]indenes^{2c} and 11-oxosteroids.^{2d} Their synthetic applications as 1,3-dielectrophilic intermediates to obtain various heterocycles by reacting with various binucleophiles have also been reported.³ In continuation of these studies we now report a highly regioselective synthesis of both 3-cyclopropyl and 5-cyclopropyl isoxazoles **2** and **3** by reacting cyclopropyl ketones **1** with hydroxylamine hydrochloride under different reaction conditions.



Scheme 1

When α -bis(methylthio)methylene cyclopropyl ketones **1a–e** were reacted with hydroxylamine hydrochloride (4 equiv.) in the presence of NaOMe (4–6 equiv., pH 7–9)⁴ and refluxed in methanol the corresponding 3-(2-arylcyclopropyl)-5-methylthioisoxazoles **2a–e** were obtained in 80–90% overall yields as colourless needles (CHCl₃–hexane). The structures of **2a–e** were confirmed with the help of spectral and analytical data (see Experimental section). In these reactions regioisomeric products **3** were not detected even in traces. On the other hand, the α -bis(methylthio)methylene cyclopropyl ketones **1a–e** when reacted with NH₂OH in sodium acetate–acetic acid–ethanol–water and refluxed with benzene (pH 2–3),⁵ gave the corresponding isomeric isoxazoles 5-(2-arylcyclopropyl)-3-methylthioisoxazoles **3a–e** in 50–60% overall yields. In these reactions small quantities of regioisomeric isoxazoles **2** (< 10%) were also detected. The isomeric

isoxazoles **2** and **3** have very similar *R_f* values (EtOAc–hexane, 1 : 4) and their separation was achieved by column chromatography.

Isomers **2** and **3** were clearly distinguished by comparing their melting points, IR and NMR spectral data. However, the firmest distinction between the isomers was obtained^{4a} from the mass spectrum fragments arising from loss of the substituents at the 5-position of the isoxazole ring.

Experimental

All melting points are uncorrected. The IR spectra were obtained (KBr disk) on a Perkin–Elmer-297, ¹H NMR spectra were measured on a Varian EM-390 spectrometer, mass on a JEOL D-300 mass spectrometer and elemental analytical data were obtained from a Heraeus CHN-O-Rapid analyzer.

3-(2-Arylcyclopropyl)-5-methylthioisoxazoles 2a–e.—Hydroxylamine hydrochloride (0.04 mol) was added to NaOCH₃ (0.06 mol) in absolute methanol (30 ml) and stirred for 10 min. Cyclopropyl ketone **1** (0.01 mol) was added and the mixture was refluxed for 10–12 h. Methanol was evaporated under reduced pressure and the residue was poured into ice-cold water. It was extracted with chloroform (100 ml), washed with water (200 ml), dried (Na₂SO₄) and evaporated to yield the cyclopropyl isoxazoles **2a** as pale coloured solids. Recrystallization from ethanol gave the analytically pure products.

Compound 2a. Needles, mp 99 °C, yield 78%; $\nu_{\max}/\text{cm}^{-1}$ (KBr): 1602, 1546, 1413; δ_{H} (CDCl₃): 1.23–1.51 (2H, m, CH₂), 2.15–2.39 (2H, m, CH), 2.50 (3H, s, SCH₃), 5.85 (1H, s, H-4), 7.20–7.40 (5H, m, ArH); *m/z*: 231 (M⁺, 50%), 184 (M⁺ – 47, 100), 156 (M⁺ – 75, 25) (Found; C, 67.6; H, 5.5; N, 6.16. C₁₃H₁₃NSO requires C, 67.53; H, 5.62; N, 6.06%).

Compound 2b. Mp 150 °C, yield 80%; $\nu_{\max}/\text{cm}^{-1}$ (KBr): 1600, 1500, 1430; δ_{H} (CDCl₃): 1.10–1.32 (2H, m, CH₂), 1.85–2.20 (2H, m, CH), 2.40 (3H, s, SCH₃), 3.65 (3H, s, OCH₃), 5.72 (1H, s, 4-H); 6.70 (2H, d, *J* = 9 Hz, ArH), 6.95 (2H, d, *J* = 9 Hz, ArH); *m/z*: 261 (M⁺, 65%), 214 (M⁺ – 47, 100), 186 (M⁺ – 75, 30) (Found; C, 64.21; H, 5.6; N, 5.52. C₁₄H₁₅NO₂S requires C, 64.36; H, 5.74; N, 5.36%).

Compound 2c. Mp 100 °C, yield 72%; $\nu_{\max}/\text{cm}^{-1}$ (KBr): 1600, 1530, 1420; δ_{H} (CDCl₃): 1.25–1.50 (2H, m, CH₂), 2.10–2.36 (2H, m, CH), 2.65 (3H, s, SCH₃), 3.85 (3H, s, OCH₃), 3.90 (3H, s, OCH₃), 5.90 (1H, s, 4-H), 6.75–6.85 (3H, m, ArH); (Found; C, 61.60; H, 5.72; N, 4.81. C₁₅H₁₇NO₃S requires C, 61.83; H, 5.88; N, 4.81%).

Compound 2d. Mp 105 °C, yield 70%; $\nu_{\max}/\text{cm}^{-1}$ (KBr): 1601, 1500, 1450; δ_{H} (CDCl₃): 1.15–1.30 (2H, m, CH₂), 1.87–2.26 (2H, m, CH), 2.48 (3H, s, SCH₃), 2.51 (3H, s, CH₃), 5.75 (1H, s, H-4), 6.90–7.01 (4H, m, ArH); *m/z*: 245 (M⁺, 35), 198 (M⁺ – 47, 75), 170 (M⁺ – 75, 17) (Found; C, 68.42; H, 6.1; N, 5.65. C₁₄H₁₅NSO requires C, 68.57; H, 6.12; N, 5.71%).

Compound 2e. Mp 120 °C, yield 75%; $\nu_{\max}/\text{cm}^{-1}$ (KBr): 1602, 1545, 1415; δ_{H} (CDCl₃): 1.21–1.45 (2H, m, CH₂), 2.10–2.35 (2H, m, CH), 2.50 (3H, s, SCH₃), 5.80 (1H, s, H-4), 7.05 (2H, d, *J* = 9 Hz, ArH), 7.25 (2H, d, *J* = 9 Hz, ArH); *m/z*: 265 (M⁺, 65%), 218 (M⁺ – 47, 100), 190 (M⁺ – 75, 41) (Found; C, 58.82; H, 4.6; N, 5.15. C₁₃H₁₂ClNO₂S requires C, 58.86; H, 4.52; N, 5.28%).

5-(2-Arylcyclopropyl)-3-methylthioisoxazoles 3a–e.—Cyclopropyl ketone **1** (0.01 mol) was dissolved in a mixture containing benzene (100 ml) and acetic acid (100 ml). After stirring for 10 min, a mixture of hydroxylamine hydrochloride (0.04 mol), NaOAc (0.03 mol) in ethanol (55 ml) and water (10 ml) was added. The reaction mixture was refluxed for 8–10 h. After evaporating the organic solvents under reduced pressure, the residue was dissolved in water (100 ml). It was extracted with CHCl₃ (100 ml), dried (Na₂SO₄) and concentrated

* To receive any correspondence.

† This is a **Short Paper** as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research (S)*, 1999, Issue 1]; there is therefore no corresponding material in *J. Chem. Research (M)*.

under reduced pressure to give crude products of **3a–e**. The crude products were purified by passing through column chromatography using EtOAc–hexane (1 : 1) as eluent.

Compound 3a. Mp 60 °C, yield 58%; $\nu_{\max}/\text{cm}^{-1}$ (KBr); 1620, 1537, 1400; δ_{H} (CDCl₃): 1.32–1.51 (2H, m, CH₂), 2.15–2.39 (2H, m, CH), 2.50 (3H, s, SCH₃), 5.88 (1H, s, H-4), 7.30–7.50 (5H, m, ArH); m/z : 231 (M⁺, 75), 105(100) (Found; C, 67.3; H, 5.7; N, 6.01. C₁₃H₁₃NSO requires C, 67.53; H, 5.62; N, 6.06%).

Compound 3b. Mp 100 °C, yield 60%; $\nu_{\max}/\text{cm}^{-1}$ (KBr); 1612, 1520, 1430; δ_{H} (CDCl₃): 1.10–1.32 (2H, m, CH₂), 1.85–2.20 (2H, m, CH), 2.35 (3H, s, SCH₃), 3.60 (3H, s, OCH₃), 5.85 (1H, s, 4-H); 6.70 (2H, d, $J = 9$ Hz, ArH), 6.90 (2H, d, $J = 9$ Hz, ArH); m/z : 261 (M⁺, 70), 135(100) (Found; C, 64.61; H, 5.0; N, 5.6. C₁₄H₁₅NO₂S requires C, 64.36; H, 5.74; N, 5.36%).

Compound 3c. Mp 70 °C, yield 52%; $\nu_{\max}/\text{cm}^{-1}$ (KBr); 1613, 1540, 1420; δ_{H} (CDCl₃): 1.35–1.50 (2H, m, CH₂), 2.10–2.36 (2H, m, CH), 2.55 (3H, s, SCH₃), 3.80 (3H, s, OCH₃), 3.85 (3H, s, OCH₃), 5.80 (1H, s, 4-H), 6.70–6.80 (3H, m, ArH); (Found; C, 61.45; H, 5.86; N, 4.9. C₁₅H₁₇NO₃S requires C, 61.83; H, 5.88; N, 4.81%).

Compound 3d. Mp 60 °C, yield 50%; $\nu_{\max}/\text{cm}^{-1}$ (KBr); 1615, 1532, 1435; δ_{H} (CDCl₃): 1.25–1.30 (2H, m, CH₂), 1.87–2.20 (2H, m, CH), 2.40 (3H, s, CH₃), 2.55 (3H, s, SCH₃), 5.85 (1H, s, 4-H), 6.90–7.0 (4H, m, ArH); m/z : 245 (M⁺, 70), 119(100) (Found; C, 68.30; H, 5.9; N, 5.66. C₁₄H₁₅NSO requires C, 68.57; H, 6.0; N, 5.71%).

Compound 3e. Mp 67 °C, yield 57%; $\nu_{\max}/\text{cm}^{-1}$; 1615, 1535, 1425; δ_{H} (CDCl₃): 1.15–1.25 (2H, m, CH₂), 2.05–2.20 (2H, m, CH), 2.46 (3H, s, SCH₃), 5.85 (1H, s, 4-H), 7.0 (2H, d, $J = 9$ Hz, ArH), 7.20 (2H, d, $J = 9$ Hz, ArH); m/z : 265 (M⁺, 75), 139(100) (Found; C, 58.6; H, 4.75; N, 5.0. C₁₃H₁₂ClNOS requires C, 58.86; H, 4.52; N, 5.28%).

O.M.S. thanks MASTEC (DST) for the award of a young scientist fellowship.

Received, 16th November 1998; Accepted, 16th March 1999
Paper E/8/08909D

References

- 1 A. Merz and G. Mark, *Angew. Chem., Int. Ed. Engl.*, 1973, **12**, 845.
- 2 (a) B. Deb, C. V. Asokan, H. Ila and H. Junjappa, *Tetrahedron Lett.*, 1988, **29**, 2111; (b) B. Deb, H. Ila and H. Junjappa, *J. Chem. Res.*, 1990, (S) 356; (M) 2728; (c) P. K. Patra, V. Sriram, H. Ila and H. Junjappa, *Tetrahedron*, 1998, **54**, 531; (d) B. Patro, B. Deb, H. Ila and H. Junjappa, *J. Org. Chem.*, 1992, **57**, 2257.
- 3 Okram, M. Singh, H. Ila and H. Junjappa, *Indian J. Chem., Sect. B*, 1997, **36**, 1056; 1123.
- 4 (a) S. A. Lang, Y.-i Lin, in *Comprehensive Heterocyclic Chemistry*, ed. A. R. Katritzky and C. W. Rees, vol. 6, ed. K. T. Potts, Pergamon Press, New York, 1984, pp. 6–7, 61–66; (b) A. R. Katritzky, P. Barczyski, D. L. Ostercamp and T. I. Yousaf, *J. Org. Chem.*, 1986, **51**, 4037; (c) C. Kashima, N. Yoshiwara, S. I. Shirai and Y. Omote, *Chem. Lett.*, 1982, 1455; (d) M. L. Purkayastha, H. Ila and H. Junjappa, *Synthesis*, 1989, 20.
- 5 T. J. Doorenbos and L. Milewich, *J. Org. Chem.*, 1966, **31**, 3193.