Predicting the Congener-Specific Environmental Behaviour of Perfluorinated Acid **Contaminants Using Semi-Empirical Computational Methods**



Sierra Rayne^{a,b,*}, Kaya Forest^{c,} and Ken J. Friesen^{a,b}

^aDepartment of Chemistry. The University of Winnipeg, Winnipeg, MB, Canada, R3B 2E9; ^bThe Richardson College for the Environment, The University of Winnipeg. Winnipeg, MB, Canada, R3B 2E9: Department of Chemistry, Okanagan College, Penticton, BC, Canada, V2A 8E1: * E-mail: rayne.sierra@gmail.com

Results

Introduction

- Perfluerinated acids (PEAs) are contaminants detected worldwide in a range of abiotic and biotic environmental matrices
- Two man or classes of PEAs include the perfluorinated sulfonic acids (PFCAs) and perfluorinated carboxylic acids (PFCAs)

$F = (CF_2)_a = S = OH F = (CF_2)_a$ DECAG PESAs

- Both Gesses are considered persistent, bioaccumulative, and subject o long-range transport
- Both menally have low rates of hydrolysis, direct photolysis, and biode dation/metabolism in environmental systems (1.2)
- Current esearch and regulatory efforts focussed on the straightchain members of each PEA class and homologue group. primarity because these congeners are the major components of technical mixtures and are also available as pure standards
- Nume Sus potential branched congeners in each PEA class are a poorly inderstood family of contaminants whose environmental and toxicological properties may be more important than the straight chain members
- Current work applied established quantitative structure and reactive (QSAR) models for other multi-class emerging and legacy ontaminants to estimate key parameters related to the congener-specific environmental fates of PFSAs and PFCAs

Methods

- Number of PFA congeners increases exponentially with perfluerinated chain length
- Practigal to compute properties for all congeners only up to ~C. (there are 161 C1 through C8 congeners)
- Then to be the statistical relations between chain length and C₁ through C_8 properties to extend our understanding to $\geq C_9$
- Semiempirical molecular descriptors were calculated for all C. through C_o PFSA and PFCA and the straight-chain C_o, C₁₀, and C₁₁ PFCA congeners
- Initial geometry optimization using MM2 followed by energy minimization and computations using AM1 and PM3 basis sets



Figure 1. Relationship between number of congeners in each homologue group and the perfluoroalkyl chain length (PCL). The actual number of congeners in each homologue group was determined for C, through C, and extrapolated for C₉ though C₁₄ A second-order polynomial ion equation of the form log_{10} (# of congeners) = -0.5 + 0.248 × PCL -0.00758 × PCL2 (r2=0.9999) is shown.

1. Bioconcentration Factors (BCEs) Model developed with regressions between computed molecular areas and volumes and BCEs reported for straight-chain C. to C. PESAs and C₂ to C₄, PECAs (3) Used to estimate BCFs for all C, through C, PFSA and PFCA congeners ⇒ PESAs ~10-fold higher BCEs than PECAs ous C. to C. BCE distribu loque mixing versus volume-based approach = homologue isolation (a) DESA more & branching & less (b) PFCAs ในสมาริการ เกมส์เห็นสี่เห็นสาย 5 10 15 20 25 30 35 1 5 10 15 20 25 30 35 40 45 50 55 60 oner number within each homologue gro Figure 2. Estimated BCFs of all C4 through C8 (a) PFSA and (b) PFCA congeners based on correlations with the calculated Connelly accessible area (CAA), Connelly molecular area (CMA), and Connelly solvent excluded volume (CSEV) using the PM3 semiempirical basis set (AMI) data not shown for brevity). Values represent the mean BCF value from the correlations Error bars signify lower and upper error limits based on the regression relationships. Summary statistics for the PFSA and PFCA C₄ through C₈ BCF training sets can be reliably predicted allows reliable esti DESA Figure 3. Predicted mean, minimu maximum and standard deviation of stimated BCFs for C4 through C11 (a) PESA and (b) PECA congene using the calculated Connelly accessible areas (CAA; (i) and (iv)), onnelly molecular areas (CMA; (ii,

Results

and (v)) and Connelly solvent excluded volumes (CSEV: (iii) and vi)) with the 4MI and PM3 miempirical basis sets Regression ines were developed based on calculated BCFs for all C₄ through C PFSA congeners Perfluoroalkyl chain length (# carbons)

2. Atmospheric Reaction with Hydroxyl Radicals (kow)

- · Model developed with well-established regressions between ionization potentials (IPs) and kow values for a wide range of aliphatic compounds (4-6)
- Calibrated for PEAs using published experimental kou data for the C₁-C₄ straight chain PFCAs (7), C₁-C₀ fluorotelomer alcohols/aldehvdes (8-10), and C4 perfluoroalkyl sulfonamide derivatives (11,12)
- Used to estimate atmospheric lifetimes (τ_{OH}) for all C₁ through C₈ PFSA and PFCA and the straight-chain C₉, C₁₀, and C₁₁ PFCA straight-chain members of each homologu



1 5 10 15 20 25 30 35 1 5 10 15 20 25 30 35 40 45 50 55 60 6 Congener number within each homologue group hing within each homologue group increases to

aing offect more important than chain length for a Figure 4. Estimated atmospheric lifetimes of the C, through C, PFSA and PFCA and C, C, and C.

raight-chain PECA congeners using the AMI ((a) and (b)) and PM3 ((c) and (d)) basis sets

3. Abjotic Reductive Defluorination Rates (kmd)

- Recent work shows qualitative proof-of-principle that C₂-C₂ PFSAs can be reductively defluorinated (13.14)
- Model developed with established relationships between the energy of the lowest unoccupied molecular orbital (eLUMO) and surface-area normalized reduction rate constants (k_{red}) using zero-valent metal reductants (e.g., Fe, Mg) across a range of halogenated aliphatic and aromatic contaminants (15-17)
- In absence of quantitative PFA kred "benchmarks", values were normalized to the minimum estimated kred within each basis set among all C1 through Ca PFSA and PFCA congeners (kred/kred.min)

PESAs about 0.5-2 orders of

magnitude more amenable to reduction than PFCAs

PESA AM1 PFSA-AMI
 PFCA-AMI
 PFSA-PM3
 PFCA-PM3
 PFCA-PM3 Figure 5. Estimated average relative rates of abiotic reductive defluorination within each C, through C, homologue group for PFSAs and PFCAs. Regression lines were developed based on calculated BCFs for all C, through C. PFSA congeners. Error bars show standard deviations in knowing within each homologue group



Figure 6. Estimated relative rate constants for the abiotic reductive defluorination (k. /k. of the C1 through C8 PFSA and PFCA and C9, C10, and C11 straight-chain PFCA conges using the AM1 ((a) and (b)) and PM3 ((c) and (d)) basis sets

Conclusions and Recommendations

· Findings provide guidance for:

Results

- > developing new analytical methods for separating and identifying PFAs in environmental and technical mixtures
- > prioritizing efforts on synthesizing authentic standards Focussing toxicological studies on the congeners most likely to be of concern
- Molecular area approach to estimating BCFs suggests wide variability within homologue groups ⇒ while volume approach indicates low variability:
 - need to determine which model best applies via targeted synthesis of selected congeners and BCF testing
 - e.g., (C, PFSA-23) perfluoro-1,1',2,2'tetramethylbutylsulfonic acid predicted to have BCF >120-
- fold higher using volume versus area » also, (C₈ PFCA-28) perfluoro-2,2,3,3'-
- tetramethylbutylcarboxylic acid predicted to have BCF >100fold higher using volume versus area

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Metadata, citation and similar

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Perfluoroalkyl chain length (# carbon Vanden Heuvel et al. (1991) Toxicol Appl / harmacol 107, 450-459.² Goecke et al. (1992) Chem Res Toxicol 5, 512-519.³ Martin et al. (2003) Environ Sci Technol 40, 524-519.³ Valinquon et al. (1994) J Almos Chem 2, 83-93.¹ Gatliney and Levine (1979) Int J Chem Ken 11, 1197-1209.⁶ Alkinson (1985) Chem Rev 85, 69-201.⁷ Hurley et al. (2003) Environ Sci Technol 40, 524-519.³ 2 Seleva et al. (2004) Phys Chem Chem Phys 6. 1243-1252. 10 Eon et al. (2005) Environ Sci Technol 40. 1862-1868. 1240-1252. 14 Control 40. 1862-1868. 14 Contro