

# Observation of atmospheric aerosols at Mt. Hua and Mt. Tai in central and east China during spring 2009 – Part 1: EC, OC and inorganic ions

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**Abstract.** PM<sub>10</sub> and size-segregated samples were simultaneously collected at Mt. Hua (2060 m a.s.l.) and Mt. Tai (1545 m a.s.l.) in central and east coastal China during spring, 2009 including an intensive dust storm event occurring on 24 April, and determined for EC, OC and inorganic ions. During the non-dust storm period particles, EC, OC and ions except for SO<sub>4</sub><sup>2-</sup> were 2–10 times more abundant at Mt. Tai than at Mt. Hua. SO<sub>4</sub><sup>2-</sup> (13 ± 7.1 μg m<sup>-3</sup>) at Mt. Hua was the dominant ion, followed by NO<sub>3</sub><sup>-</sup> (5.0 ± 3.9 μg m<sup>-3</sup>), NH<sub>4</sub><sup>+</sup> (2.5 ± 1.3 μg m<sup>-3</sup>) and Ca<sup>2+</sup> (1.6 ± 0.8 μg m<sup>-3</sup>). In contrast, at Mt. Tai NO<sub>3</sub><sup>-</sup> was most abundant (20 ± 14 μg m<sup>-3</sup>), followed by SO<sub>4</sub><sup>2-</sup> (16 ± 13 μg m<sup>-3</sup>), NH<sub>4</sub><sup>+</sup> (12 ± 8.9 μg m<sup>-3</sup>) and Ca<sup>2+</sup> (3.9 ± 2.1 μg m<sup>-3</sup>). The fact of NO<sub>3</sub><sup>-</sup> exceeding over SO<sub>4</sub><sup>2-</sup> at Mt. Tai may suggest the changes in chemical composition of the atmosphere over east China due to sharply increasing vehicle emission. pH values of the water-extracts of PM<sub>10</sub> samples indicate that at the two mountain sites aerosols transported from the south regions are more acidic than those from the north and more acidic at Mt. Tai than at Mt. Hua during the non-dust storm period. During the dust storm event particle mass, OC, Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup> and Ca<sup>2+</sup> at both sites increased by a factor of 1–9, while EC, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> decreased by 20–80%. However, SO<sub>4</sub><sup>2-</sup> concentrations (13 ± 7.7 μg m<sup>-3</sup> at Mt. Hua and 15 ± 5.6 μg m<sup>-3</sup> at Mt. Tai, respectively) at the two sites during the episode were compa-

rable and did not change significantly compared to those in the non-dust storm period, probably due to a similar level of free tropospheric SO<sub>2</sub> in central and east China.

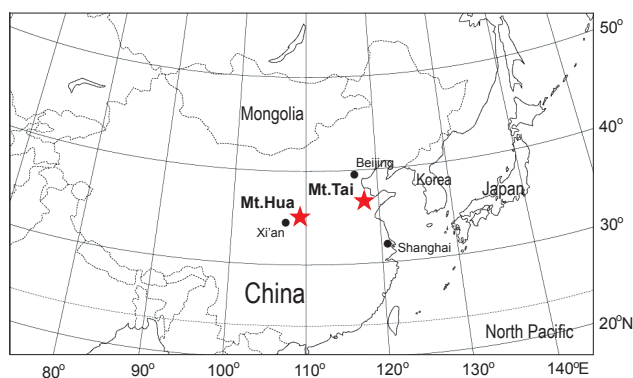
Compared with those at Mt. Hua the coarse modes (>2.1 μm) of K<sup>+</sup> and SO<sub>4</sub><sup>2-</sup> at Mt. Tai during the non-event period were more abundant and the coarse mode of NO<sub>3</sub><sup>-</sup> was less abundant. When the dust storm was present all ions significantly moved toward coarse particles, except for NH<sub>4</sub><sup>+</sup>, with a disappeared peak in fine mode (<2.1 μm) for NO<sub>3</sub><sup>-</sup>. Linear regression for ion equivalents in fine particles indicates that ammonium exists in the forms of NH<sub>4</sub>NO<sub>3</sub> and NH<sub>4</sub>HSO<sub>4</sub> at Mt. Hua and NH<sub>4</sub>NO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> at Mt. Tai during both the nonevent and the event periods. While the regression for coarse mode of Ca<sup>2+</sup> suggests a close coupling of dust with nitrate during the nonevent time and with sulfate during the dust-storm period. pH values of the size-resolved samples further suggest that during the nonevent period most acidic particles at Mt. Hua are in the range of 0.7–1.1 μm, while those at Mt. Tai are in the range of 1.1–2.1 μm. Aerosols at both sites became alkaline during the event, but the Mt. Tai particles still showed a lower pH value.

## 1 Introduction

Asian dust storms, which often occur in spring, are lofted up by strong winds from central Asia and Gobi deserts and can be transported into East Asia, North America (Arimoto et al., 2006; Heald et al., 2006; Leaitch et al., 2009;



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**Fig. 1.** Sampling sites of Mt. Hua and Mt. Tai in East Asia.

Parrington et al., 1983; VanCuren and Cahill, 2002; Wilkening et al., 2000), and even be transported more than one full circuit around the globe within two weeks (Uno et al., 2009). During the long-range transport, dust particles react with a diversity of chemical species, coagulate with other particles, and/or provide reaction sites in the atmosphere, exerting a significant impact on the atmospheric environment (Huebert et al., 2003; Sun et al., 2010; Tobo et al., 2010; Trochkin et al., 2003) and human health (Chiu et al., 2008; Hong et al., 2010; Singh et al., 2009) of the downwind regions.

Numerous studies on changes in the chemical properties of Asian dust particles during long-range transport have been documented (Kanayama et al., 2002; Parrington et al., 1983; Zhang et al., 2003), but most of them were performed in urban (Huang et al., 2010b; Sun et al., 2004; Wang et al., 2002b, 2003; Yuan et al., 2008; Zhang et al., 2009), rural and marine areas (Decesari et al., 2005; Geng et al., 2009; Kanayama et al., 2002; Wang et al., 2009a). In contrast, only a limited number of observations on aerosol chemistry were conducted at alpine sites, most of which are situated in the eastern part of China (Gao et al., 2005; Wang et al., 2009b, 2009c). Atmospheric environment in mountain area is unique because of lower temperature, higher relative humidity (RH), and stronger solar radiation. Tropospheric aerosols over mountain areas are derived mostly from long-range transport, and are thus representative of atmospheric characters in a larger scale. Recently Rosenfeld et al. (2007) found that the decreasing precipitation in mountain area of inland China is deeply linked with the increasing air pollution. Satellite observations (Richter et al., 2005; van Donkelaar et al., 2010) also pointed out that  $\text{NO}_2$  and particle levels in east China are the highest in the world. These increasing anthropogenic pollutants, together with the frequent occurrence of dust storm, may have been changing the physicochemical properties of the downwind atmosphere (Huebert et al., 2003; IPCC, 2007; Liu and Diamond, 2005; Mori et al., 2003; Seinfeld et al., 2004).

**Table 1.** Meteorological parameters during the sampling period.

	$T^a$ , °C	RH <sup>b</sup> , %	$V^c$ , km
Mt. Hua	$5.8 \pm 5.5$	$58 \pm 21$	$15 \pm 4.6$
Mt. Tai	$4.9 \pm 5.2$	$63 \pm 20$	$16 \pm 9.2$

<sup>a</sup> Temperature, <sup>b</sup> relative humidity, <sup>c</sup> visibility.

The purpose of this study is to recognize the difference in composition, concentration and size distribution of inorganic ions and elemental and organic carbon in airborne particles between Mt. Hua and Mt. Tai, two mountains located in central and east China, and investigate the impact of dust storm on aerosol chemistry of the mountain atmospheres.

## 2 Experimental section

### 2.1 Sample collection

Sampling sites of Mt. Hua ( $34.48^\circ \text{N}$ ,  $110.08^\circ \text{E}$ , 2060 m a.s.l.) and Mt. Tai ( $36.27^\circ \text{N}$ ,  $117.10^\circ \text{E}$ , 1545 m a.s.l.) are situated in mid and east China, respectively (Fig. 1). TSP and  $\text{PM}_{10}$  samples were collected in a 24-h interval at an airflow rate of  $100 \text{ l min}^{-1}$  by using two mid-volume samplers (KC-120H, Qingdao Laoshan Company, China) fixed on a rooftop of the Meteorological Observation Station at the summits of Mt. Hua and Mt. Tai. Size-segregated particles were simultaneously collected at the same sites using a 9-stage sampler (Andersen, USA) at an airflow rate of  $28.3 \text{ l min}^{-1}$  with cut-off points as 0.4, 0.7, 1.1, 2.1, 3.3, 4.7, 5.8, and  $9.0 \mu\text{m}$ . Each set of the size-segregated samples was continuously collected for 4–6 days depending on aerosol loading. All the samples were collected onto pre-combusted ( $450^\circ \text{C}$ , 6 h) quartz filters and the air inlets were around 5–10 m above the ground. The Mt. Hua sampling was performed from 25 March to 29 April 2009, while the sampling at Mt. Tai was conducted from 27 March to 29 April 2009. On 24 April a massive dust storm originating from Gobi desert arrived in Mt. Hua and Mt. Tai simultaneously, during which TSP and  $\text{PM}_{10}$  sampling time was changed into 3–6 h. Sampling at Mt. Tai for size-segregated particles was stopped during the event due to instrument problem of the 9-stage sampler, thus the size-resolved data for the dust storm period are unavailable. Field blank was collected before and after the sampling by mounting a filter onto the sampler for about 10 min without sucking any air. After sampling, the sample and the blank filters were sealed in an aluminum foil, transported into the lab and stored in a freezer under  $-20^\circ \text{C}$  prior to analysis. Meteorological parameters at the two sites during the sampling periods are given in Table 1.

## 2.2 Sample analysis

### 2.2.1 Inorganic ions

Detailed analytical method for inorganic ions has been reported elsewhere (Wang et al., 2002a, 2010). Here we only give a brief introduction. Filter aliquots from PM<sub>10</sub> and the size-segregated samples were cut in pieces and extracted with 5 ml pure water for 3 times each in 10 min by ultrasonication, respectively. Then the combined water-extracts were filtered through a PTFE filter to remove the particles and filter debris, and determined for pH using a pH meter (HANNA HI8424 pH meter, US) at an ambient temperature of 25 °C and inorganic ions using an ion chromatograph (Dionex 500, Dionex, US). Particle mass on the filter was gravimetrically measured using an electronic balance (Sartorius E5, Germany) under a relative humidity of 40 ± 5 % at ambient temperature.

### 2.2.2 Organic carbon (OC) and elemental Carbon (EC)

OC and EC in the TSP and PM<sub>10</sub> samples were analyzed using DRI Model 2001 Carbon Analyzer following the Interagency Monitoring of Protected Visual Environments (IMPROVE) thermal/optical reflectance (TOR) protocol (Chow et al., 2004, 2007). Briefly, a size of 0.526 cm<sup>2</sup> sample filter was put in a quartz boat inside the analyzer and progressively heated to temperatures of 120°, 250°, 450°, and 550° in a non-oxidizing helium (He) atmosphere, and 550°, 700°, and 800° in an oxidizing atmosphere of 2 % oxygen in helium.

Inorganic ions, EC and OC in the field blanks were less than 10 % of those in real samples. Data reported here were all subtracted by the blanks. An intercomparison was made by comparing the species determined in PM<sub>10</sub> with those in the corresponding size-segregated samples. A good linear correlation was obtained for the two data sets (Fig. 2), demonstrating the consistency between the two samplers.

## 3 Results and discussion

### 3.1 Chemical compositions

#### 3.1.1 General description

Figure 3 shows the temporal variations of TSP and PM<sub>10</sub> during the campaign, while concentrations of chemical species in the samples are presented in Table 2. TSP and PM<sub>10</sub> during the non-dust storm periods were 103 ± 42 and 71 ± 28 μg m<sup>-3</sup> at Mt. Hua, accounting for about 50 % of those at Mt. Tai. The massive dust storm event on 24 April significantly affected the atmospheres over both mountains with the highest 3-h TSP and PM<sub>10</sub> concentrations being 991 and 740 μg m<sup>-3</sup> at Mt. Hua and 2280 and 1797 μg m<sup>-3</sup> at Mt. Tai (Fig. 3), 6–8 times more abundant than those in the nonevent time. Although the lower altitude of Mt. Tai is more accessible to pollutants from lowland sources, the higher levels

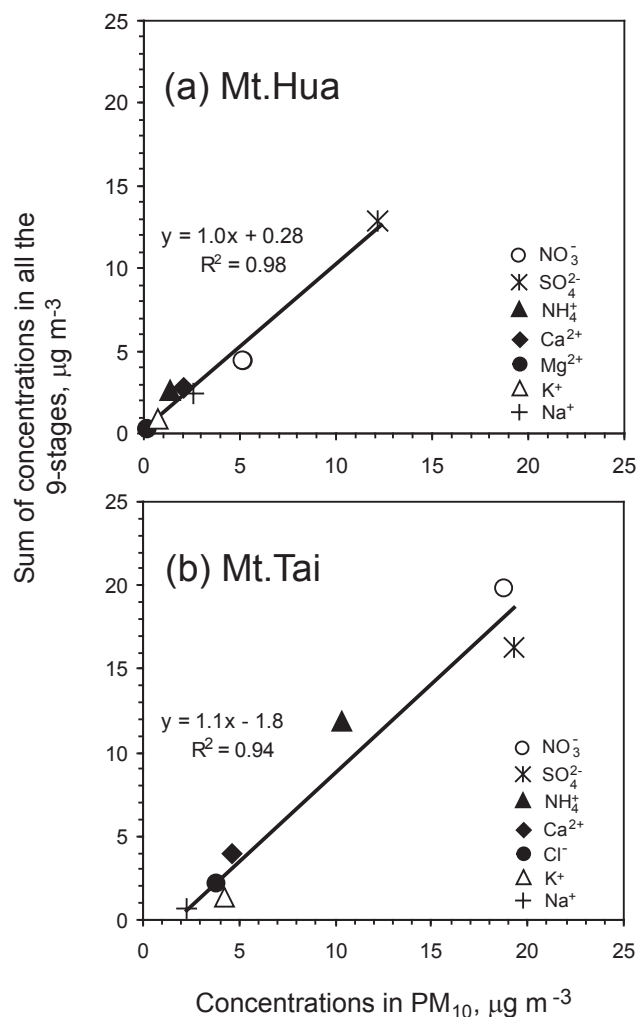


Fig. 2. An intercomparison of major components measured by the PM<sub>10</sub> and 9-stage samplers.

of TSP and PM<sub>10</sub> at Mt. Tai in both the nonevent and event periods clearly demonstrate that aerosol pollution at Mt. Tai is more serious than at Mt. Hua, which is consistent with results observed by satellite and aircraft measurements for NO<sub>2</sub> (Richter et al., 2005) and PM<sub>2.5</sub> (van Donkelaar et al., 2010).

#### 3.1.2 Non-dust storm period

As shown in Fig. 4 and Table 2, concentrations of SO<sub>4</sub><sup>2-</sup> and Na<sup>+</sup> of PM<sub>10</sub> during the non-dust storm period in the Mt. Tai atmosphere are comparable to those in the air of Mt. Hua, but PM<sub>10</sub>, EC, OC and other ions at Mt. Tai are 2–10 times more abundant than those at Mt. Hua especially for Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, and K<sup>+</sup>. Potassium ion has been considered a key tracer for biomass burning emissions (Engling and Gelencser, 2010; Li et al., 2003). Several studies on vegetation fire emissions from Africa savanna found that the smoke was enriched in Cl<sup>-</sup>, and almost the entire amount released

**Table 2.** Concentrations of OC, EC and inorganic ions in aerosols from the atmospheres of Mt. Hua and Mt. Tai in China during the spring of 2009,  $\mu\text{g m}^{-3}$ .

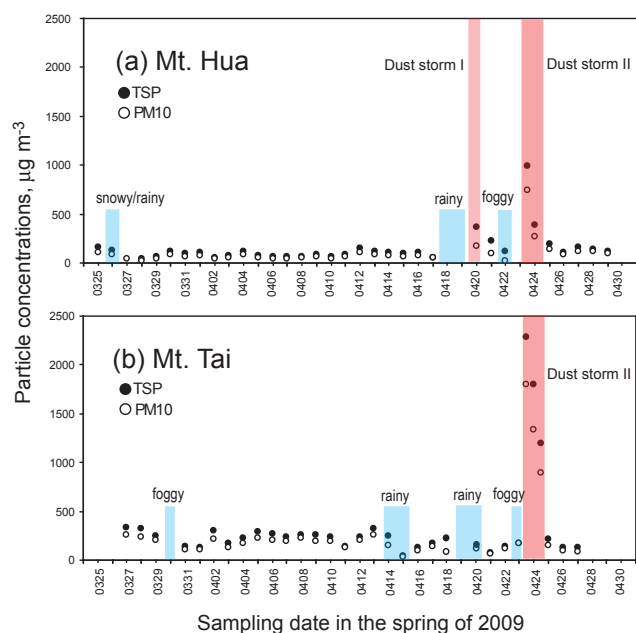
	Mt. Hua			Mt. Tai	
	Non-dust storm (n = 31)	Dust storm I (n = 1)	Dust storm II (n = 2)	Non-dust storm (n = 29)	Dust storm II (n = 3)
Total suspended particles (TSP)					
PM <sup>a</sup>	103 ± 42	365	689 ± 355	205 ± 77	1759 ± 542
OC	5.9 ± 1.6	15	28 ± 12	13 ± 5.8	49 ± 12
EC	1.4 ± 0.5	1.4	0.6 ± 0.7	3.3 ± 1.9	Nd <sup>b</sup>
Particles in diameter less than 10 $\mu\text{m}$ (PM <sub>10</sub> )					
PM <sup>a</sup>	71 ± 28	173	506 ± 303	159 ± 60	1343 ± 450
OC	4.9 ± 1.4	8	23 ± 11	13 ± 5.9	40 ± 14
EC	1.2 ± 0.5	1.0	0.6 ± 0.6	2.8 ± 2.1	Nd <sup>b</sup>
pH <sup>c</sup>	−0.0 ± 0.6	0.9	0.8 ± 0.2	−1.2 ± 0.7	0.2 ± 0.0
F <sup>−</sup>	0.1 ± 0.1	0.1	0.0 ± 0.1	0.1 ± 0.1	0.3 ± 0.1
Cl <sup>−</sup>	0.2 ± 0.2	0.7	0.8 ± 0.8	2.2 ± 1.7	1.9 ± 0.7
NO <sub>3</sub> <sup>−</sup>	5.0 ± 3.9	4.1	3.9 ± 1.5	20 ± 14	13 ± 3.1
SO <sub>4</sub> <sup>2−</sup>	13 ± 7.1	6.1	13 ± 7.7	16 ± 13	15 ± 5.6
Na <sup>+</sup>	0.7 ± 0.8	2.4	5.7 ± 2.3	0.7 ± 0.4	3.2 ± 1.2
NH <sub>4</sub> <sup>+</sup>	2.5 ± 1.3	1.0	1.2 ± 0.2	12 ± 8.9	4.2 ± 1.5
K <sup>+</sup>	0.3 ± 0.2	0.3	0.5 ± 0.3	1.3 ± 0.8	1.8 ± 0.7
Mg <sup>2+</sup>	0.1 ± 0.1	0.3	0.4 ± 0.2	0.4 ± 0.4	1.1 ± 0.3
Ca <sup>2+</sup>	1.6 ± 0.8	2.0	4.7 ± 2.0	3.9 ± 2.1	25 ± 7.0
Total ions	23 ± 11	17	30 ± 12	57 ± 39	64 ± 19

<sup>a</sup> PM: particle mass; <sup>b</sup> Nd: not detected; <sup>c</sup> pH = pH of water-extracts of the samples – pH of water-extracts of the filed blanks.

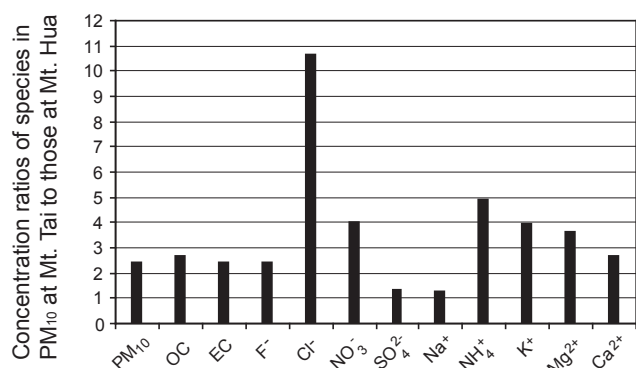
during the fires was present in particle phase (Andreae et al., 1998; Pósfai et al., 2003). Thus, a strong linear correlation ( $r^2 = 0.79$ , Fig. 5b) found between Cl<sup>−</sup> and K<sup>+</sup> for the non-dust PM<sub>10</sub> samples indicates that the high levels of Cl<sup>−</sup> and K<sup>+</sup> at Mt. Tai were mostly derived from biomass burning emissions in the North China Plain (NCP). However, such a significant correlation was not observed for the related samples at Mt. Hua (Fig. 5a). The equivalent ratio of Cl<sup>−</sup>/K<sup>+</sup> during the non-dust storm period was  $0.5 \pm 0.8$  and  $1.7 \pm 0.7$  at Mt. Hua and Mt. Tai, respectively, suggesting that KCl salt is not the only form by which Cl<sup>−</sup> and K<sup>+</sup> exist in the samples, and they may have other origins or experienced additional atmospheric processes during the transport of biomass burning smoke such as deposition and/or reaction with acidic gases. Chinese loess and dust from Gobi desert also contain certain amount of Cl<sup>−</sup> and K<sup>+</sup> (Cao et al., 2008). Arimoto et al. (2004) reported that springtime TSP aerosols at Zhenbeitai, a site located in inland China, contain some amount of Cl<sup>−</sup> ( $0.5 \pm 0.5 \mu\text{g m}^{-3}$ ) and K<sup>+</sup> ( $0.3 \pm 0.2 \mu\text{g m}^{-3}$ ) during non-dust storm periods. Size distribution of particles at both mountain sites showed that a significant amount of Cl<sup>−</sup> and K<sup>+</sup> was enriched in particles with a diameter larger than 3  $\mu\text{m}$ , especially when dust was present (see detailed discussion in Sect. 3.2), indicating an important contribution of

soil/dust to the particulate Cl<sup>−</sup> and K<sup>+</sup> in the mountain atmospheres in addition to the contributions from biomass burning, dried salt lakes and sea salt, which results in the equivalent ratio of Cl<sup>−</sup> and K<sup>+</sup> being different from unity. Recent studies on biomass burning events have reported that KCl salt abundantly existing in young smoke can readily be converted into K<sub>2</sub>SO<sub>4</sub> and KNO<sub>3</sub> during the smoke aging process by reacting with acid gases HNO<sub>3</sub> and H<sub>2</sub>SO<sub>4</sub> (Hand et al., 2005; Ikegami et al., 2001; Li et al., 2003; Pósfai et al., 2003). HNO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> have been confirmed to be of ability to react with NaCl and release HCl, latter may further react with atmospheric alkaline particles like CaCO<sub>3</sub> (Finlayson-Pitts et al., 1989; Tobo et al., 2010). High levels of HNO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> in mega-cities in east coastal China have been documented recently (Pathak et al., 2009, 2010; Wang et al., 2011), thus it is possible that KCl salt in the atmosphere of Mt. Tai may react with HNO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> in the same manner as does NaCl, which is another factor leading to the equivalent ratio of particulate Cl<sup>−</sup> and K<sup>+</sup> not be unity.

The high concentrations of NO<sub>3</sub><sup>−</sup> and NH<sub>4</sub><sup>+</sup> at Mt. Tai are caused by high emissions of NO<sub>x</sub> and NH<sub>3</sub> because of the sharp increase in vehicle numbers and the more active agricultural practice in NCP. Atmospheric aerosols at Mt. Hua during the normal days, i.e., the non-dust storm period,



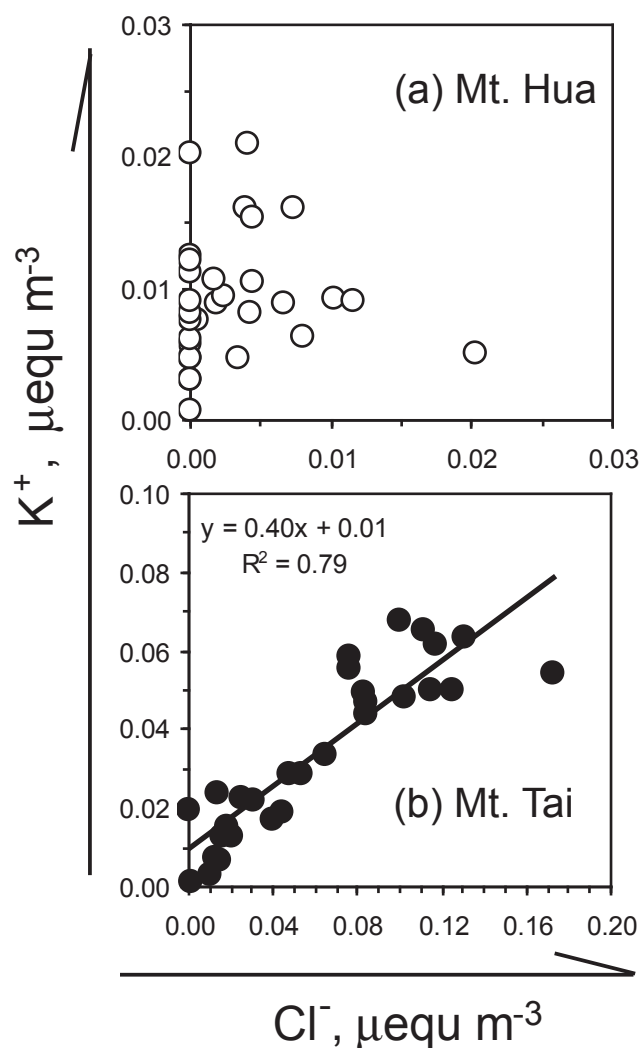
**Fig. 3.** Temporal variations of TSP and PM<sub>10</sub> concentrations in the atmospheres over Mt. Hua and Mt. Tai during spring 2009.



**Fig. 4.** Difference in the concentrations of species in PM<sub>10</sub> between Mt. Hua and Mt. Tai during the non-dust storm period in spring 2009.

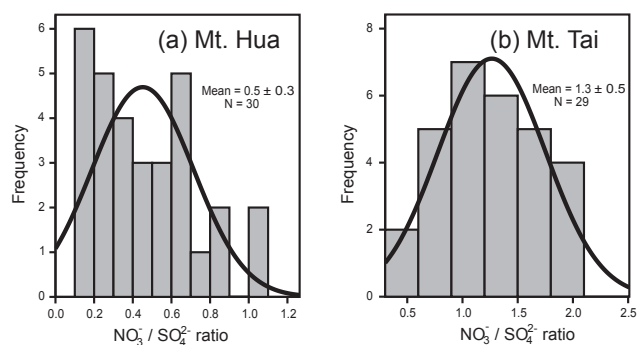
were dominated by  $\text{SO}_4^{2-}$  ( $13 \pm 7.1 \mu\text{g m}^{-3}$ , Table 2), followed by  $\text{NO}_3^-$  ( $5.0 \pm 3.9 \mu\text{g m}^{-3}$ ),  $\text{NH}_4^+$  ( $2.5 \pm 1.3 \mu\text{g m}^{-3}$ ),  $\text{Ca}^{2+}$  ( $1.6 \pm 0.8 \mu\text{g m}^{-3}$ ) and  $\text{Na}^+$  ( $0.7 \pm 0.8 \mu\text{g m}^{-3}$ ), in contrast to those at Mt. Tai, where  $\text{NO}_3^-$  ( $20 \pm 14 \mu\text{g m}^{-3}$ ) was the most abundant, followed by  $\text{SO}_4^{2-}$  ( $16 \pm 13 \mu\text{g m}^{-3}$ ),  $\text{NH}_4^+$  ( $12 \pm 8.9 \mu\text{g m}^{-3}$ ),  $\text{Ca}^{2+}$  ( $3.9 \pm 2.1 \mu\text{g m}^{-3}$ ),  $\text{Cl}^-$  ( $2.2 \pm 1.7 \mu\text{g m}^{-3}$ ),  $\text{K}^+$  ( $1.3 \pm 0.8 \mu\text{g m}^{-3}$ ), and  $\text{Na}^+$  ( $0.7 \pm 0.4 \mu\text{g m}^{-3}$ ).

During the nonevent period more than 90 % of PM<sub>10</sub> samples collected at Mt. Hua showed that nitrate is lower than sulfate (Fig. 6a). However, around 80 % of PM<sub>10</sub> samples collected at Mt. Tai showed that nitrate is higher than sulfate

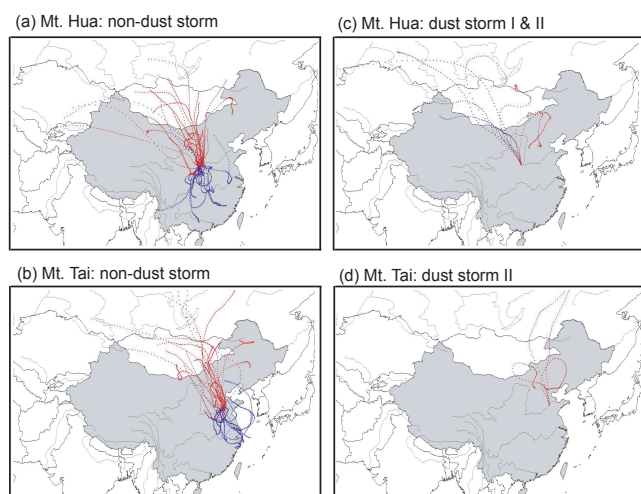


**Fig. 5.** Correlation of  $\text{K}^+$  and  $\text{Cl}^-$  in PM<sub>10</sub> at Mt. Hua and Mt. Tai during the non-dust storm period.

(Fig. 6b). Coal is the major energy source in the country with more than 2.7 billion tons of coal being burned in 2003 (Aldhous, 2005). To improve the air quality Chinese government promulgated a strict law to reduce  $\text{SO}_2$  emission in 2005. Thus the increasing rate of  $\text{SO}_2$  emission is expected to decrease. Concentrations of nitrogen oxides, on the other hand, have been increasing due to economic growth (Richter et al., 2005), leading to an enhanced nitrate aerosol in many urban areas especially in mega-cities such as Beijing and Shanghai with nitrate concentration being comparable and even higher than sulfate (Huang et al., 2010a, b; Pathak et al., 2009, 2010; Wang et al., 2006b). Compared to sulfate the more abundant nitrate in the Mt. Tai atmosphere are not only due to the high level of nitrogen oxides in the NCP region but also probably due to the unique atmosphere environment, because higher humidity and lower temperature of the mountaintop atmosphere are favorable for the transformation of  $\text{HNO}_3$  from



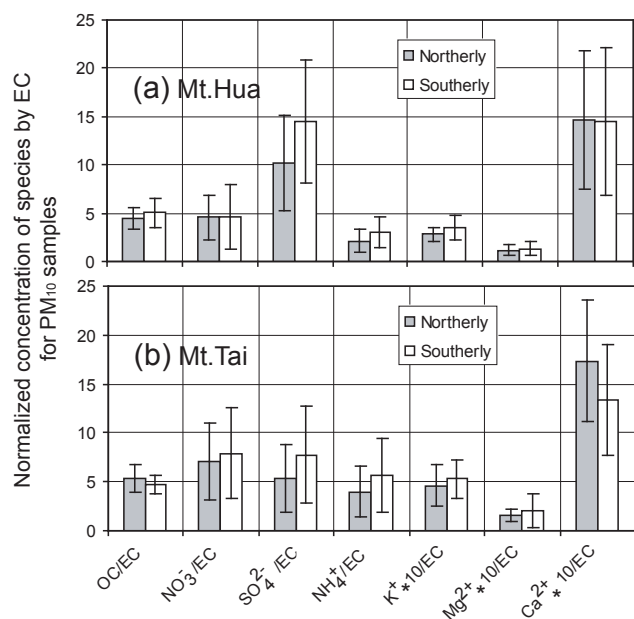
**Fig. 6.** Frequency of PM<sub>10</sub> samples containing nitrate with a concentration higher than that of sulfate.



**Fig. 7.** 72-h backward trajectories of air masses arriving in Mt. Hua and Mt. Tai during the non-dust and the dust-storm periods (The blue trajectories in (c) represent the dust storm I (DSI) while the red trajectories in (c) and (d) represent the dust storm II (DSII)).

gas to solid phase. However, the fact that concentration of  $\text{NO}_3^-$  exceeds over  $\text{SO}_4^{2-}$  clearly demonstrates that the atmospheric environment in east China is significantly changing.

Figure 7 plots the 72-h backward trajectories of air masses reaching the two sampling sites. Air masses at both sites during the non-dust storm period were transported from the south and north directions, respectively (Fig. 7a and b). Thus the corresponding samples can be classified as two groups, i.e., southerly and northerly. At the Mt. Hua site concentrations of  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{NH}_4^+$  and  $\text{K}^+$  are higher in the southerly air masses than in the northerly (Table 3), while EC, OC and other major ions in both directions of air masses are comparable. In contrast, at Mt. Tai all species except for  $\text{F}^-$  are equal or more abundant in the southerly air masses than in the northerly. Moreover, pH values showed that aerosols from south China are more acidic than those from north China especially in the Mt. Tai region (Table 3).



**Fig. 8.** Concentrations of species in PM<sub>10</sub> normalized by EC at (a) Mt. Hua and (b) Mt. Tai during the non-dust storm period.

EC is chemically stable, thus the normalized concentrations of species by EC can be useful for recognizing the changes in chemical compositions of aerosols during transport. As shown in Fig. 8a and b, relative abundance of OC and  $\text{Ca}^{2+}$  to EC in the two mountain regions are similar. However, the ratios of  $\text{NO}_3^-$ ,  $\text{NH}_4^+$  and  $\text{K}^+$  to EC at Mt. Hua are lower than those at Mt. Tai, again suggesting the high levels of nitrogen oxides, ammonia and biomass burning emission in east China. In contrast, the ratio of  $\text{SO}_4^{2-}$  to EC is higher at Mt. Hua than at Mt. Tai. The lower ratio of  $\text{SO}_4^{2-}/\text{EC}$  at Mt. Tai (Fig. 8) can be explained by more vehicle exhausts in the east coastal region, which contain more EC and less  $\text{SO}_2$  compared to coal burning emissions (Gaffney and Marley, 2009; Wang et al., 2007; Xie et al., 2010). Sulfate strongly reflects solar radiation, whereas EC strongly absorbs solar radiation. Thus the net radiative forcing is determined by the relative amounts of sulfate and EC (Ramana et al., 2010). The lower ratio of  $\text{SO}_4^{2-}$  to EC suggests that climate-warming effect caused by aerosols may be more significant in the Mt. Tai area. A recent aircraft measurement (Xue et al., 2010) showed that boundary level  $\text{SO}_2$  in east coastal China is about 10 times higher than that in the inland region, but the free tropospheric level (>1.5 km a.s.l.) of  $\text{SO}_2$  in the east and inland parts of China are comparable, which could be one of the reasons why  $\text{SO}_4^{2-}$  presented a similar concentration in the two mountain areas. The ratios of  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{NH}_4^+$  and  $\text{K}^+$  to EC at both sites are generally higher in the southerly air mass than in the northerly air mass, which is consistent with the observation for winter-time aerosols at Mt. Hua (Li et al., 2011). Figure 9 plots the

**Table 3.** Concentrations of species in PM<sub>10</sub> transported southerly and northerly at Mt. Hua and Mt. Tai during the non-dust storm period,  $\mu\text{g m}^{-3}$ .

	Mt. Hua			Mt. Tai		
	Southerly (n = 11)	Northerly (n = 19)	S/N <sup>a</sup>	Southerly (n = 10)	Northerly (n = 19)	S/N <sup>a</sup>
PM <sub>10</sub>	68 ± 30	74 ± 27	0.9	175 ± 50	151 ± 65	1.2
OC	4.7 ± 2.0	5.1 ± 1.0	0.9	12 ± 4.7	13 ± 6.5	1.0
EC	1.1 ± 0.6	1.2 ± 0.4	0.9	2.7 ± 1.2	2.8 ± 2.4	1.0
$\Delta\text{pH}^{\text{b}}$	-0.2 ± 0.6	0.1 ± 0.5	-	-1.5 ± 0.8	-1.0 ± 0.6	-
F <sup>-</sup>	0.1 ± 0.1	0.1 ± 0.0	0.8	0.1 ± 0.1	0.2 ± 0.1	0.6
Cl <sup>-</sup>	0.1 ± 0.0	0.2 ± 0.2	0.6	2.6 ± 1.8	2.0 ± 1.6	1.3
NO <sub>3</sub> <sup>-</sup>	4.8 ± 4.5	5.6 ± 3.6	0.9	23 ± 16	18 ± 13	1.3
SO <sub>4</sub> <sup>2-</sup>	13 ± 7.2	12 ± 6.8	1.1	21 ± 15	14 ± 11	1.6
Na <sup>+</sup>	1.1 ± 0.8	0.5 ± 0.7	2.0	0.7 ± 0.5	0.7 ± 0.3	1.0
NH <sub>4</sub> <sup>+</sup>	2.7 ± 1.3	2.4 ± 1.2	1.1	16 ± 11	10 ± 7.4	1.6
K <sup>+</sup>	0.4 ± 0.2	0.3 ± 0.2	1.1	1.5 ± 0.7	1.3 ± 0.9	1.2
Mg <sup>2+</sup>	0.1 ± 0.1	0.1 ± 0.1	0.9	0.6 ± 0.5	0.4 ± 0.3	1.4
Ca <sup>2+</sup>	1.5 ± 1.0	1.7 ± 0.7	0.8	3.6 ± 1.9	4.1 ± 2.2	0.9
Total ions	23 ± 13	23 ± 9.9	1.0	69 ± 45	50 ± 34	1.4

<sup>a</sup> Ratio of average concentration of southerly to that of northerly; <sup>b</sup> pH = pH of water-extracts of the samples – pH of water-extracts of the filed blanks.

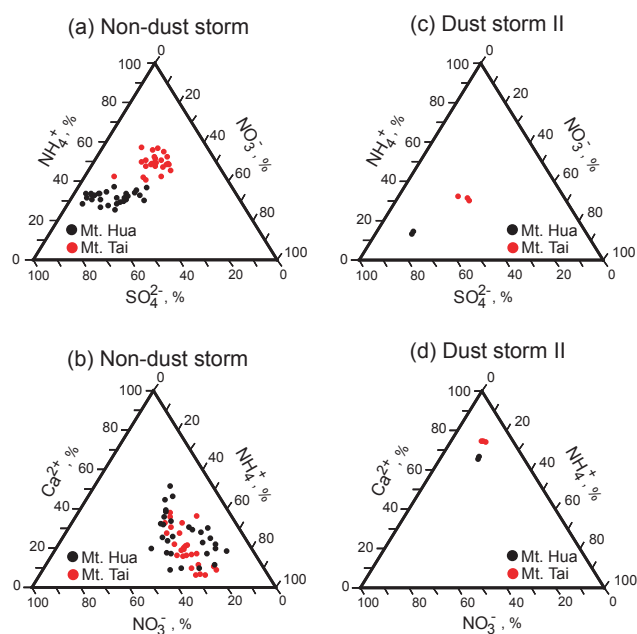
equivalent ratio of each species to the total of the three ions. The percentage of NH<sub>4</sub><sup>+</sup> is higher at Mt. Tai than at Mt. Hua during the non-dust periods (Fig. 9a), indicating that in the Mt. Hua atmosphere more NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> are neutralized by soil dust derived alkaline ions such as Ca<sup>2+</sup> and Mg<sup>2+</sup>, in contrast to Mt. Tai, where more NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> are neutralized by NH<sub>4</sub><sup>+</sup>.

To further recognize the difference of aerosol composition between the two mountain atmospheres, an EC tracer method as follow was used to approximately estimate primary organic carbon (POC) and secondary organic carbon (SOC) in the PM<sub>10</sub> samples (Castro et al., 1999; Chu, 2005; Yu et al., 2009).

$$\text{POC} = \text{EC} \times \left( \frac{\text{OC}}{\text{EC}} \right)_{\min} \quad (1)$$

$$\text{SOC} = \text{OC} - \text{POC} \quad (2)$$

Where the (OC/EC)<sub>min</sub> is the minimum ratio of OC/EC for the samples at each site. Results based on the calculation showed that throughout the non-dust storm period SOC/POC ratio was 0.5 ± 0.4 at Mt. Hua and 0.9 ± 0.5 at Mt. Tai, while SOC/EC ratio at the sites was 1.5 ± 1.3 and 2.4 ± 1.3, respectively, suggesting that carbonaceous aerosols in east coastal China is more oxidized, which is consistent with the enhanced concentration of glyoxal, a photo-oxidation product of volatile organic compounds (VOCs), over the NCP region as reported by satellite observation (Wittrock et al., 2006) and is ascribed to the high levels of oxidants such as O<sub>3</sub>



**Fig. 9.** Ternary diagrams for the equivalent ratio of major ions in PM<sub>10</sub> samples during (a), (b) the non-dust and (c), (d) the dust storm II periods.

and NO<sub>x</sub> and VOCs in the region (Akimoto, 2003; Fu et al., 2008; Hatakeyama et al., 2005; Streets and Waldhoff, 2000; Takiguchi et al., 2008). The ratio of SOC/POC at Mt. Hua is also slightly higher than that (0.4 ± 0.4) in Baoji

(Wang et al., 2010), a city nearby Mt. Hua, further indicating that mountain aerosols are more aged compared to those on ground surface.

### 3.1.3 Dust storm II on 24 April

As seen in Table 2, a moderate dust storm event, named as Dust storm I (DSI), occurred on 20 April at Mt. Hua with concentrations of 365 and 173  $\mu\text{g m}^{-3}$  for TSP and  $\text{PM}_{10}$ , respectively. Four days later an intensive dust storm event (Dust storm II, DSII), originating from Gobi desert, simultaneously reached Mt. Hua and Mt. Tai with averaged 3-h concentrations of TSP and  $\text{PM}_{10}$  being  $689 \pm 355$  and  $506 \pm 303 \mu\text{g m}^{-3}$  at Mt. Hua and  $1759 \pm 542$  and  $1343 \pm 450 \mu\text{g m}^{-3}$  at Mt. Tai, respectively. Here we only compare the composition of DSII samples to discuss difference in the impact of the event on the two alpine atmospheres. Compositions of chemicals in the highest loading of 3-h  $\text{PM}_{10}$  samples collected during DSII are shown in Table 4.  $\Delta\text{pH}$  values showed that even in the dust period aerosols at Mt. Tai are still more acidic than those at Mt. Hua. EC in the 3-h sample at Mt. Hua was 1.1  $\mu\text{g m}^{-3}$  but undetectable at Mt. Tai, whereas  $\text{SO}_4^{2-}$  and  $\text{Na}^+$  showed a similar concentration at both sites (Table 4).  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ ,  $\text{Ca}^{2+}$  and other ions in the DSII samples at Mt. Tai were 1–5 times more than those at Mt. Hua. Based on the characterization of single Asian dust particle, Sullivan et al. (2007) found that Fe-rich Asian dust particles were closely associated with secondary sulfate whereas Ca-rich particles could contain secondary sulfate and nitrate. Therefore, high concentration of  $\text{Ca}^{2+}$  observed during the event at Mt. Tai was probably caused by enhanced heterogeneous reactions of  $\text{H}_2\text{SO}_4$ ,  $\text{HNO}_3$  and their precursors (e.g.,  $\text{NO}_2$ ,  $\text{N}_2\text{O}_5$ , and  $\text{SO}_2$ ) with dust particles.

Compared to those during the non-dust storm period, the average concentrations of particle mass, OC,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  of  $\text{PM}_{10}$  at both sites increased by a factor of 1–9 during the DSII period with EC,  $\text{NO}_3^-$  and  $\text{NH}_4^+$  decreasing by 20–80%. Interestingly, during both the event and the non-event time  $\text{SO}_4^{2-}$  did not change significantly at the two alpine sites, further suggesting a homogeneous distribution of free tropospheric  $\text{SO}_2$  in the DSII pathways. Due to an additional input of  $\text{Ca}^{2+}$  the relative abundance of  $\text{NH}_4^+$  to the total equivalent of  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  significantly decreased (Fig. 9c), compared with those during the non-dust period (Fig. 9a), and the abundance of  $\text{Ca}^{2+}$  relative to the total of  $\text{Ca}^{2+}$ ,  $\text{NH}_4^+$ , and  $\text{NO}_3^-$  sharply increased, especially at Mt. Tai (Fig. 9b and d), indicating that the existence of  $\text{Ca}^{2+}$  is unfavorable for the formation of  $\text{NH}_4^+$ , because mineral dust shifts ammonia from the particle to gas phase by changing the aerosol from a cation- to anion-limited state due to the presence of alkaline species (Song and Carmichael, 1999).

**Table 4.** Concentrations of species in the highest loading of 3-h  $\text{PM}_{10}$  sample during the dust storm II event occurring on 24 April,  $\mu\text{g m}^{-3}$ .

	Mt. Hua	Mt. Tai	Mt. Tai/Mt. Hua
$\text{PM}_{10}$	740	1797	2.4
OC	30	52	1.7
EC	1.1	0.0	0.0
pH <sup>a</sup>	1.0	0.3	0.2
$\text{F}^-$	0.1	0.4	4.6
$\text{Cl}^-$	1.7	2.7	1.6
$\text{NO}_3^-$	5.4	15	2.8
$\text{SO}_4^{2-}$	20	20	1.0
$\text{Na}^+$	4.4	4.1	0.9
$\text{NH}_4^+$	1.3	5.7	4.2
$\text{K}^+$	0.8	2.5	3.3
$\text{Mg}^{2+}$	0.6	1.4	2.3
$\text{Ca}^{2+}$	6.0	31	5.2
Total ions	40	84	2.1

<sup>a</sup> pH = pH of water-extracts of the samples – pH of water-extracts of the filed blanks.

### 3.1.4 Comparison with urban aerosols

As shown in Table 5, the springtime concentrations of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  in the Mt. Hua air are 2–5 times lower than those in Xi'an, Beijing and Shanghai, three mega-cities in China, while concentrations of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  at Mt. Tai are similar to those in Shanghai and about 30–50% lower than those in Xi'an and Beijing. Compared to the three cities  $\text{NH}_4^+$  is much lower at Mt. Hua but comparable and even higher at Mt. Tai. Concentration ratio of  $\text{NO}_3^-/\text{SO}_4^{2-}$  is highest at Mt. Tai, followed by Beijing, Shanghai, Xi'an and Mt. Hua, being coincident with a higher level of nitrogen oxides in east China (Richter et al., 2005). The ratio of  $\text{NO}_3^-$  to  $\text{SO}_4^{2-}$  at Mt. Hua is 50% lower than that in Xi'an, a nearby city (see Fig. 1). In contrast, the ratio at Mt. Tai is 30% higher than that in Beijing and Shanghai (Table 5), two mega-cities located in east coastal China. Such an opposite vertical pattern between the mountains and nearby cities suggests that the atmospheric environment in the Mt. Tai area is favorable to form nitrate aerosol, largely due to the high level of nitrogen oxides in the NCP region and the advantaged meteorological conditions of lower temperature and higher humidity of the mountain air.

## 3.2 Size distribution

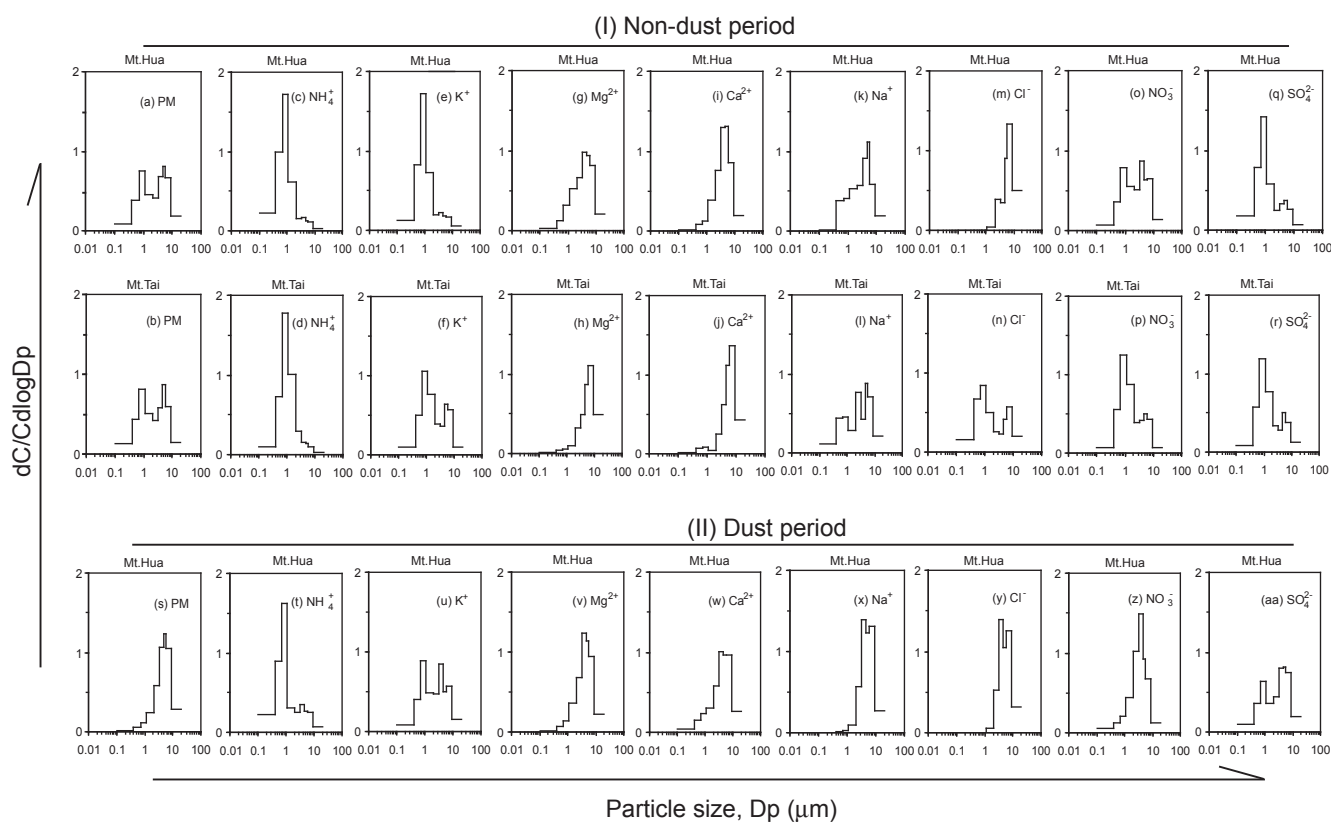
### 3.2.1 Non-dust storm period

As seen in Fig. 10a and b, particle mass size distribution during the non-dust storm period at Mt. Hua was the same as that at Mt. Tai, showing a bimodal size distribution with two equivalent peaks in the fine (<2.1  $\mu\text{m}$ ) and coarse ( $\geq 2.1 \mu\text{m}$ )



**Table 5.** Concentrations of inorganic ions in the mountain and urban atmosphere over China,  $\mu\text{g m}^{-3}$ .

Year	Size	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{NH}_4^+$	$\text{Ca}^{2+}$	$\text{K}^+$	$\text{NO}_3^-/\text{SO}_4^{2-}$	Reference	
I. Mountains									
Mt. Hua	Spring, 2009	$\text{PM}_{10}$	5.0	13	2.5	1.6	0.3	0.4	This study
Mt. Tai	Spring 2009	$\text{PM}_{10}$	20	16	12	3.9	1.3	1.3	This study
II. Mega-cities									
Xi'an	Spring 2009	$\text{PM}_{10}$	27	32	11	8.0	2.0	0.8	Wang et al. (2010b)
Beijing	2001–2004	TSP	30	31	17	8.4	2.0	1.0	Wang et al. (2006a)
Beijing	12–13 July 2005	$\text{PM}_{2.5}$	20–50	$50 \pm 10$	na <sup>a</sup>	na <sup>a</sup>	na <sup>a</sup>	0.4–1.0	Pathak et al. (2010)
Shanghai	Spring 2006	TSP	22	21	7.0	5.9	0.73	1.0	Wang et al. (2006b)
Shanghai	2–3 July 2005	$\text{PM}_{2.5}$	20–70	$50 \pm 10$	na <sup>a</sup>	na <sup>a</sup>	na <sup>a</sup>	0.4–1.2	Pathak et al. (2010)

<sup>a</sup> na: not available.**Fig. 10.** Size distributions of particles and inorganic ions during (I) non-dust storm period and (II) dust storm period.

ranges. Ammonium presented a similar unimodal pattern at both sites, dominating in the fine mode (Fig. 10c and d).  $\text{K}^+$  showed a bimodal size distribution at the sites (Fig. 10e and f), but coarse mode of potassium ion at Mt. Tai was much more pronounced. A similar pattern with  $\text{K}^+$  can be seen for  $\text{Cl}^-$  at Mt. Tai (Fig. 10n), together with the strong correlation of  $\text{K}^+$  and  $\text{Cl}^-$  in  $\text{PM}_{10}$  mentioned above, again demonstrating biomass-burning emission as their ma-

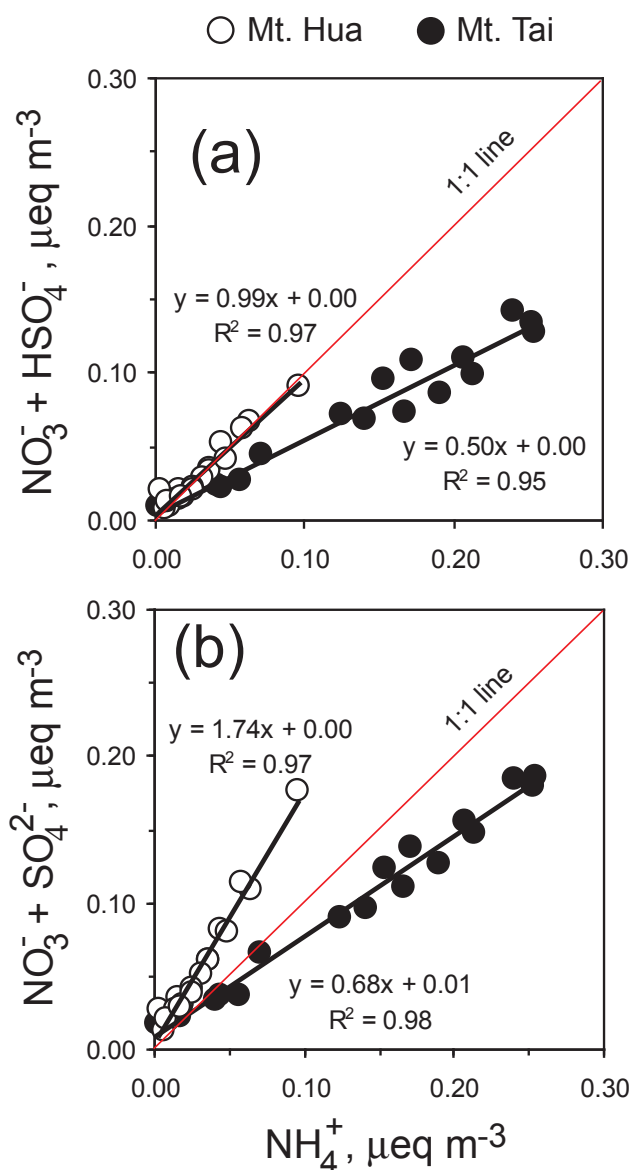
major source.  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$  and  $\text{Na}^+$  at both sites dominated in coarse mode (Fig. 10g–l), because they are mostly derived from soil/dust. But the significant amount of  $\text{Na}^+$  presenting in the fine size range indicates an importance of additional sources other than soil/dust (Fig. 10k and l).  $\text{Cl}^-$  at Mt. Hua was almost entirely present in coarse particles, in contrast to the case at Mt. Tai, where  $\text{Cl}^-$  largely occurs in the fine size range (Fig. 10m and n).  $\text{Cl}^-$  at Mt. Hua, along with

other cations such as  $\text{Na}^+$  and  $\text{Mg}^{2+}$ , probably originated in part from dried salt lakes in north and northwest China. Nitrate showed a bimodal distribution with two comparable peaks at Mt. Hua in the size ranges of 0.7–1.1 and  $>2.1 \mu\text{m}$  (Fig. 10o), and with a predominant peak in the fine mode and a small peak in the coarse mode at Mt. Tai (Fig. 10p). 64 % of  $\text{NO}_3^-$  was present in the coarse particles at Mt. Hua, while only 39 % of  $\text{NO}_3^-$  was present in the coarse particles at Mt. Tai (Table 6). The increased coarse fraction of nitrate at Mt. Hua can be explained by an increased adsorption of gaseous  $\text{HNO}_3$  onto alkaline particles due to more dust in the region (Takiguchi et al., 2008). As seen in Table 2,  $\text{NH}_4^+$  ( $12 \pm 8.9 \mu\text{g m}^{-3}$ ) at Mt. Tai is 4 times higher than that ( $2.5 \pm 1.3 \mu\text{g m}^{-3}$ ) at Mt. Hua, which means  $\text{NH}_3$  is also much more abundant at Mt. Tai than at Mt. Hua, because in China large amount of  $\text{NH}_3$  is emitted through agricultural activities such as fertilizer application (Streets et al., 2003; Streets and Waldhoff, 2000) in addition to biomass burning emissions (Andreae et al., 1998). Particulate  $\text{NO}_3^-$  can be formed by a gas phase reaction of  $\text{HNO}_3$  with  $\text{NH}_3$  and enriched in fine particles (Seinfeld and Pandis, 1998), which is more significant at Mt. Tai due to the high level of  $\text{NH}_3$ . However, due to the relatively lower level of  $\text{NH}_3$ , gaseous  $\text{HNO}_3$  in the Mt. Hua atmosphere has more chance to adsorb onto alkaline dust. As a result, the coarse mode of  $\text{NO}_3^-$  at Mt. Hua is more abundant than that at Mt. Tai. Sulfate at the two alpine regions exhibited a similar bimodal pattern with a large peak in the fine mode and a small peak in the coarse mode (Fig. 10q and r).

### 3.2.2 Dust storm on 24 April

As shown in Fig. 10s–aa, the size distribution patterns of particles and all ions were altered during the dust storm event on 24 April with an increase in the coarse range, especially for PM,  $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ . Size distribution of PM and  $\text{NO}_3^-$  during the event changed from a bimodal pattern (Fig. 10a, and o) into a unimodal one in the coarse mode (Fig. 11s and z). Sulfate still exhibited a bimodal pattern, but its fine fraction significantly decreased compared to that in the non-dust storm period (Fig. 10q and aa). Such changes in size distribution pattern indicate that gas-to-particle conversion of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  affect both sub- and supermicrometer aerosol modes by redirecting much of the deposition that would normally occur on the accumulation mode to the larger dust particles during the event (Seinfeld et al., 2004; Takiguchi et al., 2008). Other ions also showed an increase in the coarse mode when the dust was present, especially for  $\text{K}^+$ ,  $\text{Na}^+$  and  $\text{Cl}^-$ , with a significantly diminished and even disappeared mass peak in the fine range during the episode (see Fig. 10u, x and y).

Different compounds are of different solubility, for example,  $\text{Ca}(\text{NO}_3)_2$  and  $\text{CaCl}_2$  are highly soluble and much more hygroscopic than other insoluble or slightly soluble calcium salts like  $\text{CaCO}_3$ ,  $\text{CaSO}_4$ , and  $\text{CaC}_2\text{O}_4$  (Sullivan et al., 2009;



**Fig. 11.** Equivalent ratios of ammonium to (a) the sum of nitrate and bisulfate and (b) the sum of nitrate and sulfate in particles with a diameter less than  $2.1 \mu\text{m}$  at Mt. Hua and Mt. Tai during the whole sampling time including the non-dust storm and dust storm periods.

Tobo et al., 2010), affecting a particle's ability to activate as cloud condensation nuclei (Tobo et al., 2010). Thus, it is indispensable to investigate the chemical forms in which major ions exit in the particles. As shown in Table 6, during the nonevent and the event periods  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  and  $\text{NH}_4^+$  are the three major ions in the fine range ( $<2.1 \mu\text{m}$ ), although only 22 % of nitrate centered within this mode during the dust storm period. In contrast,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  and  $\text{Ca}^{2+}$  are the three most abundant ions in the coarse size range ( $>2.1 \mu\text{m}$ ) in both nonevent and event time. Here, therefore, we investigate the specific chemical forms in fine particles for  $\text{NO}_3^-$ ,

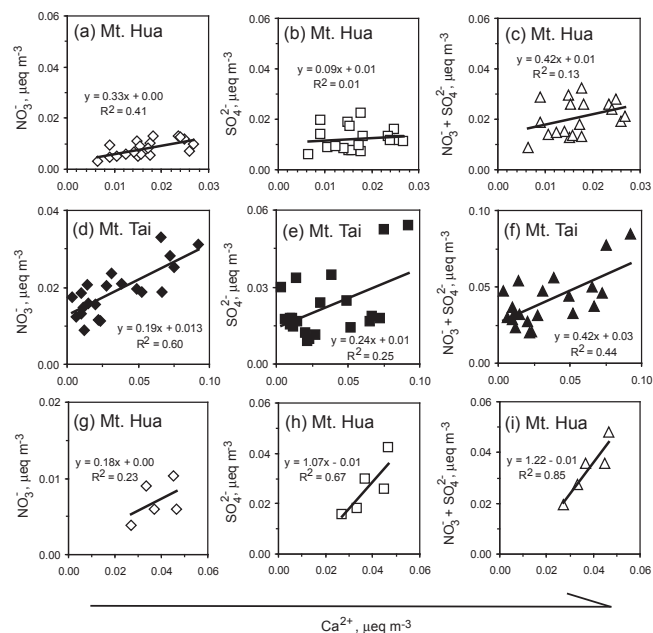
**Table 6.** Averaged concentration of inorganic ions in each stage at Mt. Hua and Mt. Tai during the non-dust storm and dust storm periods in spring 2009,  $\mu\text{g m}^{-3}$ .

Size, $\mu\text{m}$	F <sup>-</sup>	Cl <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>	SO <sub>4</sub> <sup>2-</sup>	Na <sup>+</sup>	NH <sub>4</sub> <sup>+</sup>	K <sup>+</sup>	Mg <sup>2+</sup>	Ca <sup>2+</sup>
I. Non-dust storm									
(a) Mt. Hua (n = 4)									
> 9.0	0.01 (33) <sup>a</sup>	0.06 (55)	0.72 (14)	1.01 (7)	0.10 (27)	0.09 (3)	0.02 (5)	0.04 (22)	0.56 (25)
5.8–9.0	0.01 (25)	0.02 (26)	0.60 (14)	0.65 (5)	0.05 (17)	0.06 (2)	0.01 (3)	0.02 (16)	0.42 (19)
4.7–5.8	0.00 (6)	0.01 (6)	0.29 (7)	0.44 (3)	0.05 (9)	0.03 (1)	0.01 (2)	0.01 (9)	0.23 (10)
3.3–4.7	0.01 (21)	0.01 (6)	0.67 (15)	0.63 (5)	0.05 (14)	0.07 (3)	0.01 (3)	0.02 (15)	0.36 (16)
2.1–3.3	0.00 (8)	0.01 (6)	0.50 (11)	0.66 (5)	0.04 (8)	0.09 (3)	0.02 (4)	0.02 (13)	0.26 (12)
1.1–2.1	0.00 (5)	0.00 (1)	0.71 (14)	2.46 (16)	0.05 (11)	0.57 (17)	0.08 (21)	0.02 (15)	0.15 (7)
0.7–1.1	0.00 (1)	0.00 (0)	0.72 (14)	4.07 (28)	0.03 (6)	1.00 (34)	0.13 (34)	0.01 (6)	0.08 (4)
0.4–0.7	0.00 (0)	0.00 (0)	0.39 (8)	2.47 (19)	0.03 (7)	0.63 (24)	0.07 (20)	0.00 (3)	0.07 (3)
< 0.4	0.00 (0)	0.00 (0)	0.16 (4)	1.27 (11)	0.00 (1)	0.32 (13)	0.03 (8)	0.00 (2)	0.05 (3)
(b) Mt. Tai (n = 5)									
> 9.0	0.07 (40)	0.42 (30)	1.21 (7)	2.35 (12)	0.31 (20)	0.10 (1)	0.19 (10)	0.25 (51)	2.06 (44)
5.8–9.0	0.04 (24)	0.22 (11)	1.50 (7)	1.29 (7)	0.18 (13)	0.24 (2)	0.20 (10)	0.10 (21)	1.21 (27)
4.7–5.8	0.02 (10)	0.08 (2)	0.81 (4)	0.81 (4)	0.11 (7)	0.16 (1)	0.11 (5)	0.04 (8)	0.47 (10)
3.3–4.7	0.02 (13)	0.07 (2)	1.15 (6)	0.78 (4)	0.09 (7)	0.31 (3)	0.10 (5)	0.04 (8)	0.45 (10)
2.1–3.3	0.01 (4)	0.11 (3)	1.35 (7)	1.19 (7)	0.20 (17)	0.73 (6)	0.14 (8)	0.03 (6)	0.23 (5)
1.1–2.1	0.01 (4)	0.29 (14)	4.43 (24)	3.90 (22)	0.11 (10)	3.67 (30)	0.39 (23)	0.01 (3)	0.06 (1)
0.7–1.1	0.00 (3)	0.35 (17)	4.40 (24)	4.17 (24)	0.12 (9)	3.92 (32)	0.38 (21)	0.01 (1)	0.07 (1)
0.4–0.7	0.00 (2)	0.34 (15)	2.41 (13)	2.49 (14)	0.14 (11)	2.27 (18)	0.22 (12)	0.01 (1)	0.07 (1)
< 0.4	0.00 (0)	0.20 (6)	0.77 (4)	0.97 (5)	0.09 (7)	0.72 (6)	0.10 (5)	0.00 (1)	0.03 (1)
II. Dust storm									
(a) Mt. Hua (n = 1)									
> 9.0	0.01 (23)	0.22 (33)	0.37 (13)	2.03 (20)	0.36 (28)	0.05 (7)	0.04 (16)	0.08 (23)	0.93 (21)
5.8–9.0	0.01 (18)	0.10 (24)	0.37 (13)	1.43 (14)	0.32 (25)	0.04 (5)	0.03 (11)	0.06 (18)	0.74 (16)
4.7–5.8	0.00 (8)	0.06 (10)	0.24 (8)	0.75 (7)	0.14 (11)	0.02 (2)	0.01 (5)	0.03 (10)	0.54 (12)
3.3–4.7	0.01 (30)	0.14 (21)	0.64 (23)	1.23 (12)	0.27 (21)	0.04 (5)	0.03 (13)	0.06 (19)	0.90 (20)
2.1–3.3	0.01 (22)	0.07 (10)	0.56 (20)	0.87 (9)	0.14 (11)	0.04 (5)	0.02 (9)	0.04 (13)	0.67 (15)
1.1–2.1	0.00 (0)	0.01 (2)	0.35 (12)	1.02 (10)	0.03 (3)	0.07 (9)	0.04 (14)	0.03 (10)	0.48 (11)
0.7–1.1	0.00 (0)	0.00 (0)	0.12 (4)	1.23 (12)	0.01 (0)	0.25 (32)	0.05 (17)	0.01 (3)	0.11 (2)
0.4–0.7	0.00 (0)	0.00 (0)	0.08 (3)	0.87 (9)	0.00 (0)	0.17 (22)	0.03 (10)	0.01 (2)	0.09 (2)
< 0.4	0.00 (0)	0.00 (0)	0.09 (3)	0.57 (6)	0.00 (0)	0.10 (13)	0.01 (5)	0.00 (1)	0.05 (1)

<sup>a</sup> Number in parenthesis is the mass percentage in each stage (%).

SO<sub>4</sub><sup>2-</sup> and NH<sub>4</sub><sup>+</sup> and in coarse particles for NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup> and Ca<sup>2+</sup>, respectively, based on their equivalent ratios. It can be seen that equivalent ratio of [NH<sub>4</sub><sup>+</sup>] to [NO<sub>3</sub><sup>-</sup> + HSO<sub>4</sub><sup>-</sup>] is very close to unity with a high correlation efficient for fine particles at the site of Mt. Hua ( $r^2 = 0.97$ ,  $p = 0.01$ , Fig. 11a), indicating that NH<sub>4</sub><sup>+</sup> exists in the forms of NH<sub>4</sub>NO<sub>3</sub> and NH<sub>4</sub>HSO<sub>4</sub>. For the fine particles at the site of Mt. Tai, equivalent ratio of [NH<sub>4</sub><sup>+</sup>] to [NO<sub>3</sub><sup>-</sup> + HSO<sub>4</sub><sup>-</sup>] is lower than that of [NH<sub>4</sub><sup>+</sup>] to [NO<sub>3</sub><sup>-</sup> + SO<sub>4</sub><sup>2-</sup>] (Fig. 11a and b), suggesting that most of NH<sub>4</sub><sup>+</sup> exists in the forms of NH<sub>4</sub>NO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> with certain amount of NH<sub>4</sub>HSO<sub>4</sub>. Such a difference in the chemical forms of NH<sub>4</sub><sup>+</sup> is attributed to the high level of NH<sub>3</sub> in the Mt. Tai atmosphere, rendering more neu-

tralizing matter for sulfate. During the non-dust period Ca<sup>2+</sup> in the coarse size range (>2.1  $\mu\text{m}$ ) showed a stronger correlation with NO<sub>3</sub><sup>-</sup> at Mt. Hua ( $r^2 = 0.41$ ,  $p = 0.01$ , Fig. 12a) and Mt. Tai ( $r^2 = 0.60$ ,  $p = 0.01$ , Fig. 12d) than with SO<sub>4</sub><sup>2-</sup> (Fig. 12b and e) and [NO<sub>3</sub><sup>-</sup> + SO<sub>4</sub><sup>2-</sup>] (Fig. 12c and f), indicating that more Ca<sup>2+</sup> exists as Ca(NO<sub>3</sub>)<sub>2</sub> rather than CaSO<sub>4</sub>, especially in the Mt. Tai air (Fig. 12b–f). For the dust samples, Ca<sup>2+</sup> displayed a stronger correlation with SO<sub>4</sub><sup>2-</sup> and a weaker correlation with NO<sub>3</sub><sup>-</sup> (Fig. 12g and h), suggesting that most of Ca<sup>2+</sup> exists in the form of CaSO<sub>4</sub> with certain amount of Ca(NO<sub>3</sub>)<sub>2</sub>. When nitrate is combined, the linear regression could further be improved ([Ca<sup>2+</sup>] versus [NO<sub>3</sub><sup>-</sup> + SO<sub>4</sub><sup>2-</sup>],  $r^2 = 0.85$ ,  $p = 0.01$ , Fig. 12i), may imply

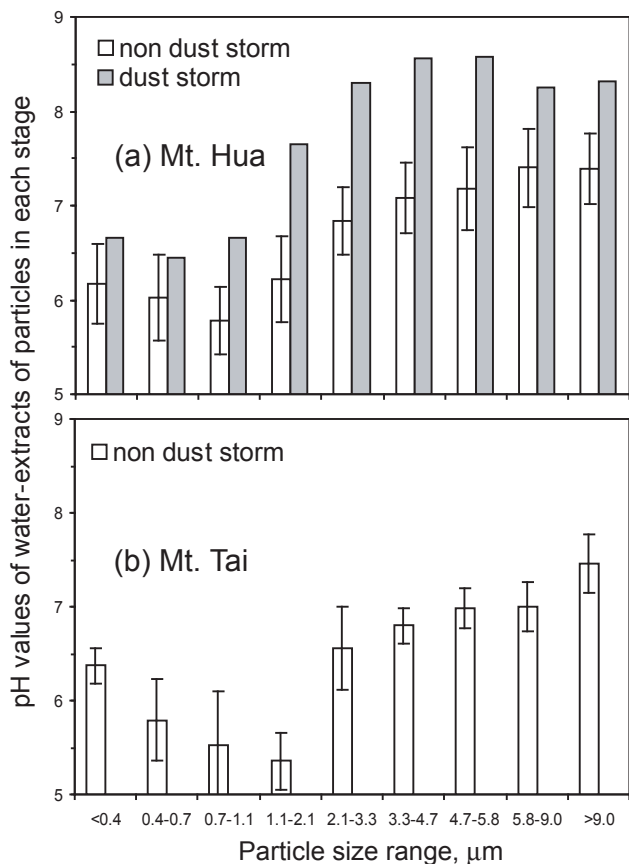


**Fig. 12.** Equivalent ratios of calcium to nitrate, sulfate and the sum of nitrate and sulfate in particles with a diameter larger than  $2.1 \mu\text{m}$  during (a–f) the non-dust storm and (g–i) dust storm periods.

an internal mixing state of  $\text{CaSO}_4$  with  $\text{Ca}(\text{NO}_3)_2$  during the dust storm period. Such results are in agreement with the 1996 to 2008 record from Mauna Loa Observatory in Hawaii, showing that 19 % of  $\text{CaCO}_3$  in Asian dust has reacted to form  $\text{CaSO}_4$  and 7 % of the  $\text{CaCO}_3$  has reacted to form  $\text{Ca}(\text{NO}_3)_2$  (McNaughton et al., 2009) pH values of the water-extracts showed that compared with those in the non-dust period particle acidity during the event decreased significantly even for the particles with a diameter less than  $0.4 \mu\text{m}$  (Fig. 13a). During the non-event period particles with the strongest acidity are in the size of  $0.7\text{--}1.1 \mu\text{m}$  at Mt. Hua while those at Mt. Tai are in the size range of  $1.1\text{--}2.1 \mu\text{m}$  (Fig. 13a and b).

#### 4 Summary and conclusions

Atmospheric aerosols were simultaneously collected during the spring of 2009 at Mt. Hua and Mt. Tai in central and east coastal China, during which an intensive dust storm event occurred on 24 April. TSP and  $\text{PM}_{10}$  at Mt. Tai were two times higher than those at Mt. Hua in both the event and non-event periods. Concentrations of  $\text{SO}_4^{2-}$  and  $\text{Na}^+$  in  $\text{PM}_{10}$  during the non-dust storm period at Mt. Tai were comparable to those at Mt. Hua, but EC, OC and other ions at Mt. Tai were 2–10 times higher than those at Mt. Hua especially for  $\text{Cl}^-$ ,  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ , and  $\text{K}^+$ , suggesting more serious air pollution in the NCP region. During the non-dust storm period  $\text{NO}_3^-$  was the dominant ion at Mt. Tai, followed by  $\text{SO}_4^{2-}$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$ , in contrast to those at Mt. Hua, where  $\text{SO}_4^{2-}$  was the



**Fig. 13.** Acidity of water-extracts of particles in each stage at (a) Mt. Hua and (b) Mt. Tai during the non-dust storm and dust storm periods.

highest, followed by  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$ . The exceeding of  $\text{NO}_3^-$  over  $\text{SO}_4^{2-}$  at Mt. Tai indicates the changes in chemical composition of the atmosphere over East Asia due to the increasing nitrogen oxides caused by vehicle exhausts and industrial emissions. SOC/POC and SOC/EC ratios demonstrate that aerosol at Mt. Tai is more oxidized than that at Mt. Hua owing to high levels of precursors and oxidants in east coastal China. At both sites air masses transported from the south region contained higher levels of  $\text{SO}_4^{2-}$  and  $\text{NH}_4^+$  compared to those transported from the north region, although EC and OC in the two directions were similar.

During the DSII period, concentrations of particles, OC,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  in the two mountain atmospheres increased by a factor of 1–9, while EC,  $\text{NO}_3^-$  and  $\text{NH}_4^+$  decreased by 20–80 %. However, there was no significant difference for  $\text{SO}_4^{2-}$  at both sites in either the dust storm or the non-dust storm periods, most likely due to a homogeneous spatial distribution of free tropospheric  $\text{SO}_2$ .

During the non-event period  $\text{K}^+$ ,  $\text{Cl}^-$  and  $\text{NO}_3^-$  showed a significant difference in size distribution between the two mountains.  $\text{K}^+$  at both sites was dominated in fine mode, but the coarse mode of  $\text{K}^+$  was much more pronounced at

Mt. Tai than at Mt. Hua.  $\text{Cl}^-$  at Mt. Tai showed a similar pattern to  $\text{K}^+$ , while  $\text{Cl}^-$  at Mt. Hua was largely present in coarse mode, because  $\text{Cl}^-$  at Mt. Hua is largely derived from dust or dried lakes in north and northwest China while  $\text{K}^+$  and  $\text{Cl}^-$  at Mt. Tai are mostly originated from biomass burning. Compared to that in Mt. Tai nitrate in Mt. Hua during the nonevent period was more enriched in coarse particles. When the dust storm was present, particles and all ions significantly shifted toward coarse particles, except for  $\text{NH}_4^+$ , with a diminished and even disappeared peak in the fine mode. Equivalent ratios indicate that during the whole campaign ammonium exists largely as  $\text{NH}_4\text{NO}_3$  and  $\text{NH}_4\text{HSO}_4$  at Mt. Hua and  $\text{NH}_4\text{NO}_3$  and  $(\text{NH}_4)_2\text{SO}_4$  at Mt. Tai, while calcium ion exists mostly as  $\text{Ca}(\text{NO}_3)_2$  during the non-event and as  $\text{CaSO}_4$  during the event.

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