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Accessing the Cyclopenta[c]pyridine Structure: Development of an Enamine/Enal Cycloaddition Pyridine Formation Methodology

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Abstract

The synthesis of 6,7-dihydro-5H-cyclopenta[c]pyridine was completed in six steps culminating in a tandem enamine-enal cycloaddition/pyridine formation. While the tandem reaction gave rise to target pyridine with consistently low yield (5%), an analogous sequential method allowed for the unadorned cyclopenta[c]pyridine structure to be prepared with improved albeit modest yield (31%). To examine if higher yields would be obtained for cyclopenta[c]pyridine structures with lower volatility, (*E*)-2,6-dimethyl-2octenedial was prepared and converted to actinidine using the tandem protocol with 60% yield. Using (E)-2,6-dimethyl-2-octenedial as the model substrate, the reaction conditions for this transformation were optimized. It was found that the optimal reaction temperature is 50 °C and that the presence of an ptoluenesulfonic acid (*p*-TsOH) substantially improved the rate of the reaction without sacrificing yield or product purity.

Background

- Semiochemicals are a class a small organic molecules that mediate interactions between animal species. The Hofferberth Lab has endeavored to prepare macroscopic quantities of terpene-derived semiochemicals to study the nature of a unique ant-butterfly mutualism in nature.¹
- Over the course of this program, we have discovered a novel synthesis of actinidine that allows for macroscopic quantities of the insect semiochemical to be prepared in enantiomerically pure form in two synthetic steps from commercially available citronellal. The key step in this reaction is a tandem enamine/enal cycloaddition-pyridine formation that results in the bicyclic, cyclopenta[c]pyridine framework.
- Many natural products contain the cyclopenta[c]pyridine framework, but few methodologies have detailed the synthesis of such molecules.² Improved access to the cyclopenta[c]pyridine core will allow for the preparation both novel compounds and chemicals with known biological activity (See Below).

		R	(
S-(-)	R	R-(+)	Lc (R = OF
(+)-Deoxyrhexifoline	CO ₂ Me		· ·
(+)-Boschniakinic acid	CO ₂ H	(-)-Plantagonine	
(+)-Boshniakine	СНО	(-)-Indicaine	
(+)-Actinidine	CH_3	(-)-Actinidine	

• The purpose of this study is to develop a methodology for the key cycloaddition-pyridine formation and to evaluate the limitations of the reaction beginning with a simplified model substrate, (E)-2-octenedial.³

Target and Retrosynthetic Analysis







Accessing the Cyclopenta[c]pyridine Structure: Development of an **Enamine/Enal Cycloaddition Pyridine Formation Methodology** Ben Kester¹, John Hofferberth²

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ouisianines A - D H, H; R' = allyl, isoallyl)

(-)-Oxerine





Preparation of Model Substrate





 NH_2OH (HCI) (1.5 eq) THF, 70 °C, Overnight <5%

•Alternate Sequential Method:

 \checkmark `CHO

4Å Sieves, rt,



Optimization of Reaction Conditions

• Model reaction for optimization:



 NH_2OH (HCI) (X eq) -----Solvent, Temp, Time, **Additives**

•Equivalents of reactants and additives:

Reactant	Equivalents	Result
NH ₂ OH	1.3-3	No Change
TsOH	0.2-1.2	Increased Rate
H ₂ O	0.1	Increased Rate
NaHCO ₃	1	No Effect

•Solvent:

Solvent	Result*	Temperature (°C)	Time to Completion
Benzene	No Reaction	RT	>24h
Et ₂ O	No Reaction	30	>24h
EtOH	No Reaction	40	Overnight
DMF	Trace	50	Hours
DMSO	Present**	60	Hours
DCM	No Reaction	70	Decomposition <1h
THF	Present		

(EtO)₂POCH₂COMe, **\cetone, 1N HC**

COOEt



Pyridine Formation



33%

•Temperature (Rxn in DMSO):

*Pyridine formation as determined by TLC **Difficult isolation of pyridine constituent

Preparation of Actinidine Using Optimized Conditions



• Due to difficulty in isolating 6,7-dihydro-5H-cyclopenta[c]pyridine, the previously optimized reaction parameters were further optimized using (E)-2,6-dimethyl-2-octenedial as the model substrate.

Current and Future Work

•The next phase of the project will explore how the presence and location of substituents on the substrate affects the reaction.

• Targets:



• Synthesis of substrate for preparation of A:



- route to the cyclopenta[c]pyridine framework.
- volatility of the target and consequent difficulty in its isolation.
- cyclopenta[c]pyridine framework.



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 NH_2OH (HCI) (1.5 eq) TsOH (1 eq), THF, 50 °C Overnight 60%



N	Ph	N
Α	В	, C

Conclusions

• The tandem enamine/enal cycloaddition-pyridine formation represents a novel

• The unadorned cyclopenta[c]pyridine structure can be prepared under numerous reaction conditions. The consistently low yield is believed to be due to the

• Future substrates will be evaluated using the optimized reaction conditions to define the scope and limitations of this novel preparation of the

Acknowledgements

References