

Summer 6-26-2005

III. Optical and mechanical properties of transparent glass-ceramics

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Federal University of Sao Carlos

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LAMAV

Vitreous Materials
Laboratory



**Optical and mechanical properties of
mature and new transparent
*glass - ceramics***

Edgar D. Zanotto

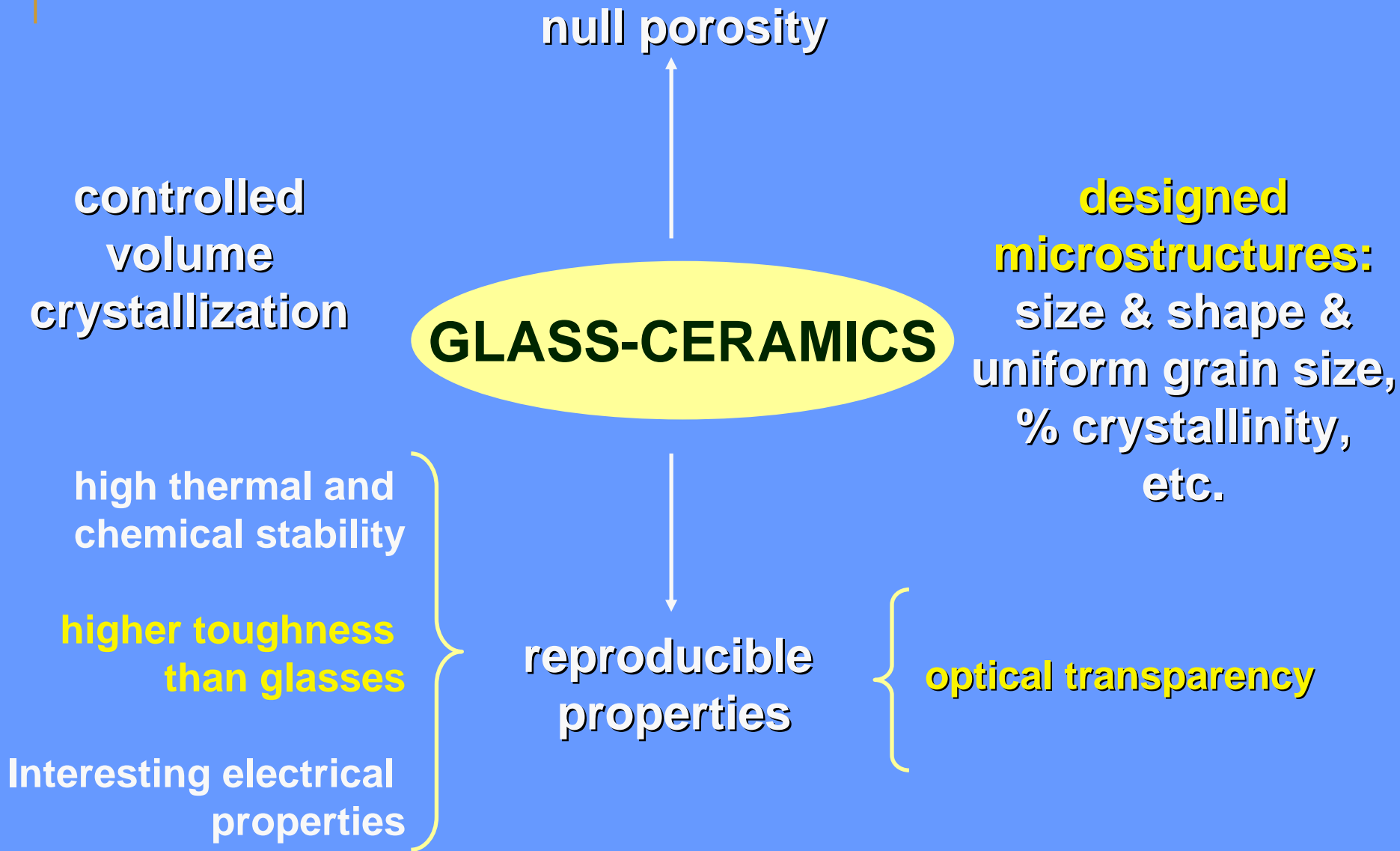
Federal University of São Carlos, Brazil

I IMI Meeting, State College, June 2005

OUTLINE

- Introduction to glass-ceramics
- Brief literature review on **TGC**
- Potential applications of **TGC**
- Conditions for transparency
- Mature TGC – nanocrystals
- **New TGC:** **Properties**
 - **Sintered aluminate GC** Opt & Mech
 - **IR transmitting CG** Opt & Mech
 - **Ce: YAG GC for lighting** Opt
 - **Laser crystallized GC** Opt
 - **PTR GC** Opt & Mech
 - **LGHC GC** Opt & Mech
- Surprise....
- Conclusions

INTRODUCTION



INTRODUCTION

Applications of transparent glass-ceramics

Thermo-mechanical

Cooking ware
Fire resistant plates
Security windows
Telescope mirrors...

Optical (potential)

saturable absorber media; illumination devices using IR; heat-resistant materials that absorb UV, that reflect infrared and are transparent to visible light; that absorb UV and fluoresce in red/IR; second harmonics generating; substrates for LCD devices; optical amplifiers for up-conver; substrates for arrayed waveguide grating (AWG); radiation sources of lamps; Laser pumps; Laser media; Materials for precision photolithography; ring laser gyroscopes; solar collectors; printed optical circuits; etc.

The inventor of GLASS-CERAMICS

S.D. Stookey discovering GC
in the middle 1950s



LITERATURE REVIEW- PIONEERS OF TGC

STOOKEY, S.D.

V Int. Congress on Glass, pp. V/1-8 **1959**

BORRELLI, N.F. ELECTRO-OPTIC EFFECT IN
TRANSPARENT NIOBATE GLASS-CERAMIC SYSTEMS

Journal of Applied Physics, 38 (11): 4243 **1967**

BEALL, G.H.; DUKE, D.A. TRANSPARENT GLASS-
CERAMICS

Journal of Materials Science, 4 (4): 340 **1969**

Recent articles in the next slide

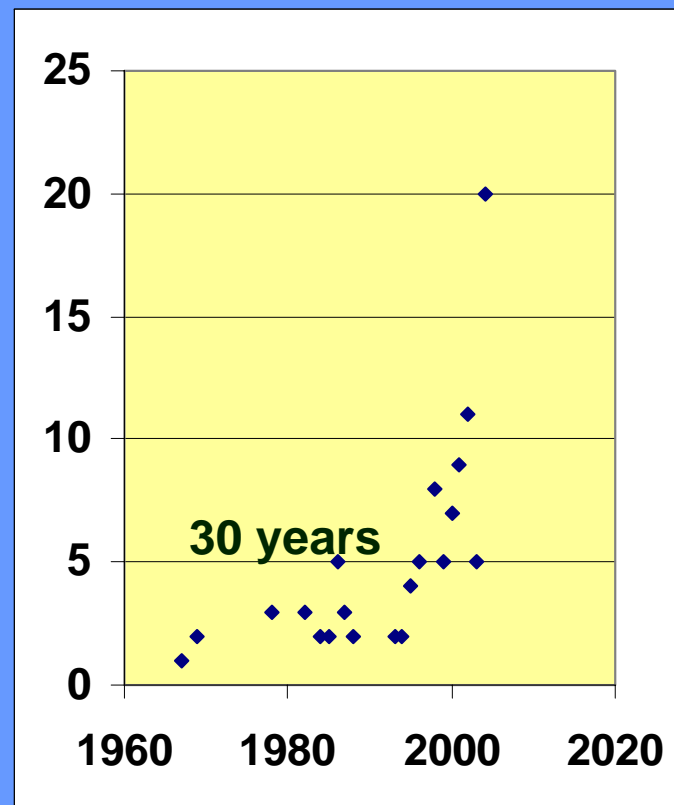
LITERATURE REVIEW (TGC in the title)

<u>YEAR</u>	<u>112 ISI papers</u>
1967	1
1969	2
1978	3
1982	3
1984	2
1985	2
1986	5
1987	3
1988	2
1993	2
1994	2
1995	4
1996	5
1998	8
1999	5
2000	7
2001	9
2002	11
2003	5
2004	20



Derwent II
72 patents

Corning
Schott
Nippon
Others



Crystalline phases in TGC

- B-quartz ss
- B-eucryptite
- Mullite
- Spinel
- Willemite
- Ghanite
- Forsterite
- β -BBO
- LiNbO_3
- NaNbO_3
- PbF_2
- LaF_3
- ZnO
- Etc.

Most TGC have
nanosize crystals &
small crystallized
volume fraction
(~ 50% or less)

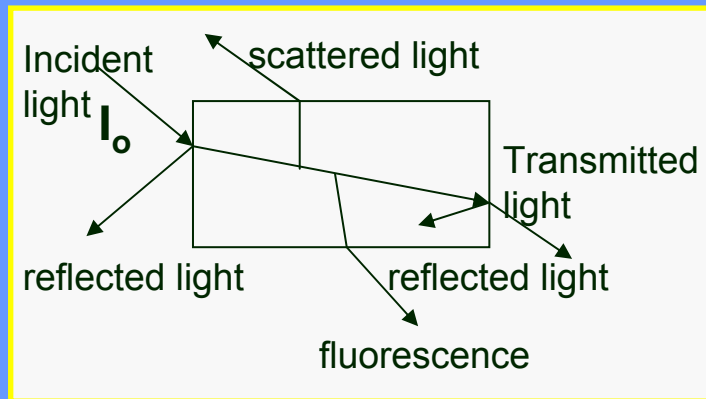
THEORY

Light attenuation

atomic absorption (β)

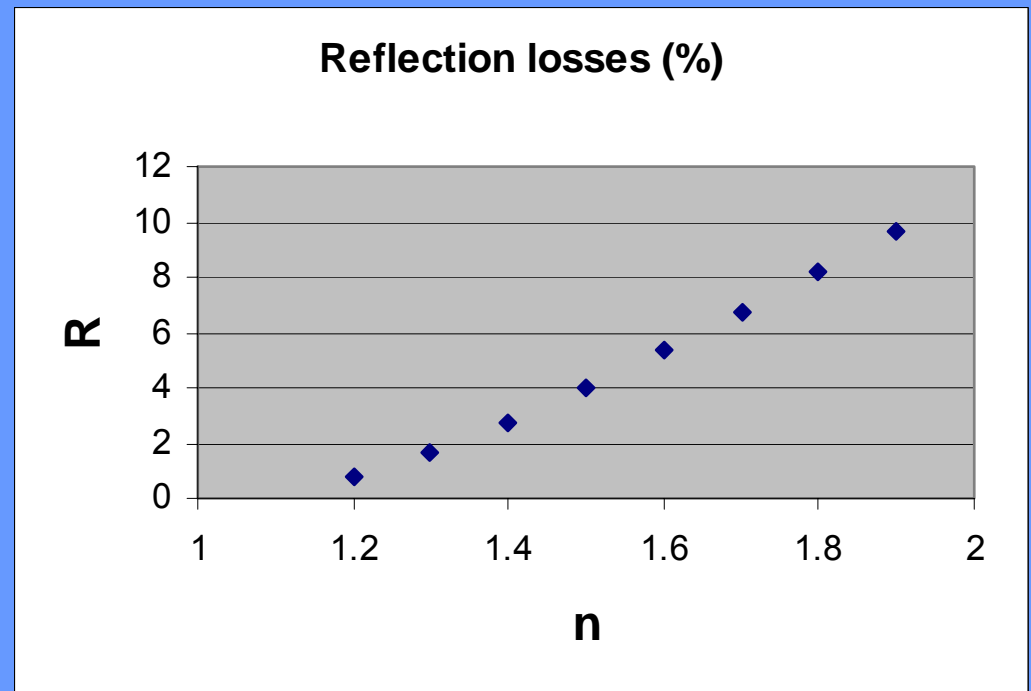
scattering (S)

+ surface reflection (R)



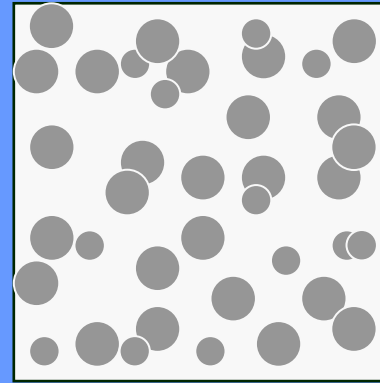
$$I = I_0(1 - R)^2 \exp(-(\beta + S)x)$$

$$R = \left(\frac{n-1}{n+1} \right)^2$$



Conditions for transparency

Transparent glass-ceramics



crystal size \ll wavelength of light

Basic requirements

low birefringence

$$n_{\text{glass}} \cong n_{\text{crystal}}$$

Examples of commercially mature TGC

Corning's VISION

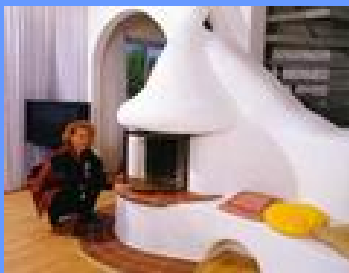


VLT 8.2 m Zerodur mirror on its way to Paranal Observatory, Chile, Dec. 97/ Schott



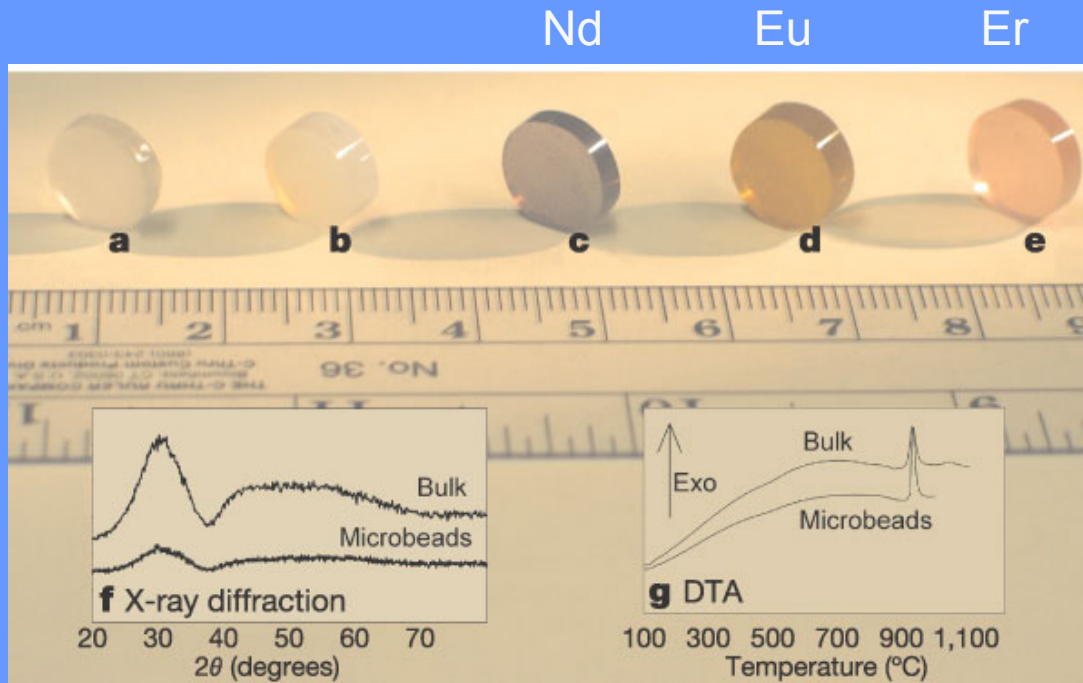
ROBAX – Schott
NEOCERAM – NIPPON
KERAGLASS- Corning/ St. Gobain

CERAN- Schott



NEW
TRANSPARENT GC
(yet on the development stage)

Bulk glasses and ultrahard nanoceramics based on alumina and rare-earth oxides by A. Rosenflanz et al. *Nature* **430**, 761 - 764 (August 2004).



IR transparent

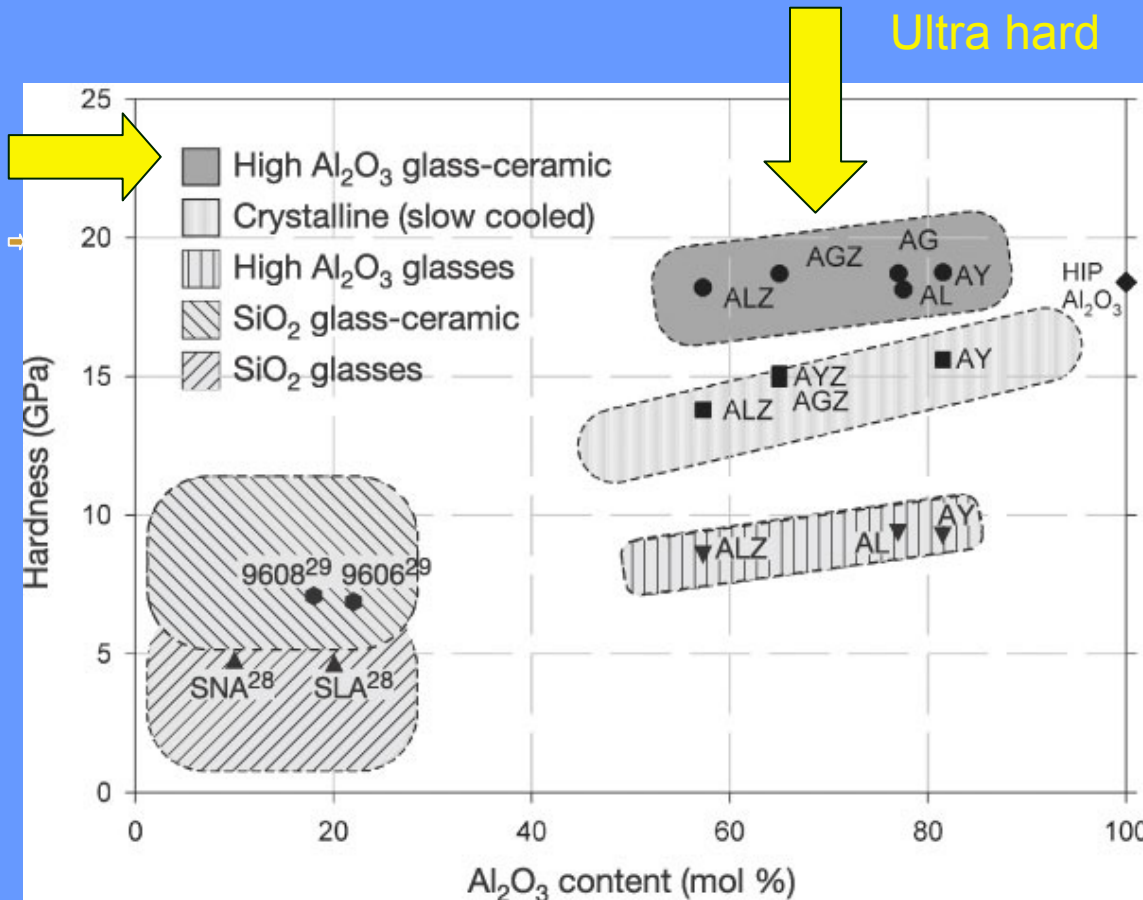


50-90% Al₂O₃

a, b: no dopants; c 5wt% Nd₂O₃; d 5wt% Eu₂O₃; e 5wt% Er₂O₃.
All except b were hot-pressed at 905 °C at 34 MPa for 360 s.

Material b was hot-pressed for 1,200 s inducing **partial crystallization**, giving the opalescent appearance.

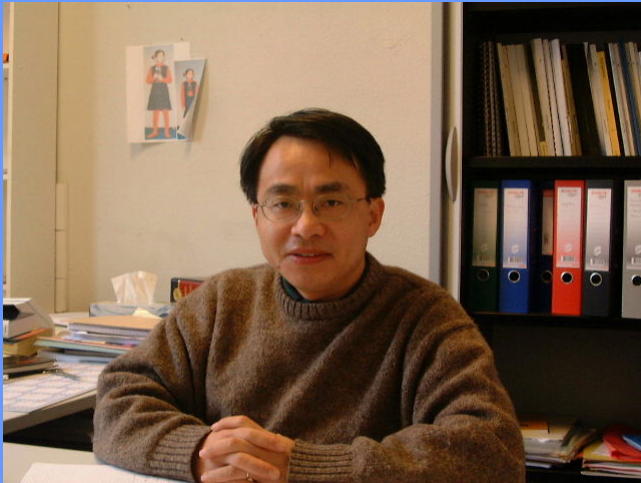
High alumina glasses and GC



Hardness against Al₂O₃ content. High-alumina glasses and **glass-ceramics** surpass other oxides : BeO, MgO, Y₂O₃, ZrO₂, TiO₂, Y₃Al₅O₁₂, Corning 9606 and 9608 GC, and are comparable to pure *a*-Al₂O₃ and *b*-Si₃N₄.

These compositions were also crystallized directly from the melt during slow cooling.

IR transmitting chalco-sulfide glass-ceramics



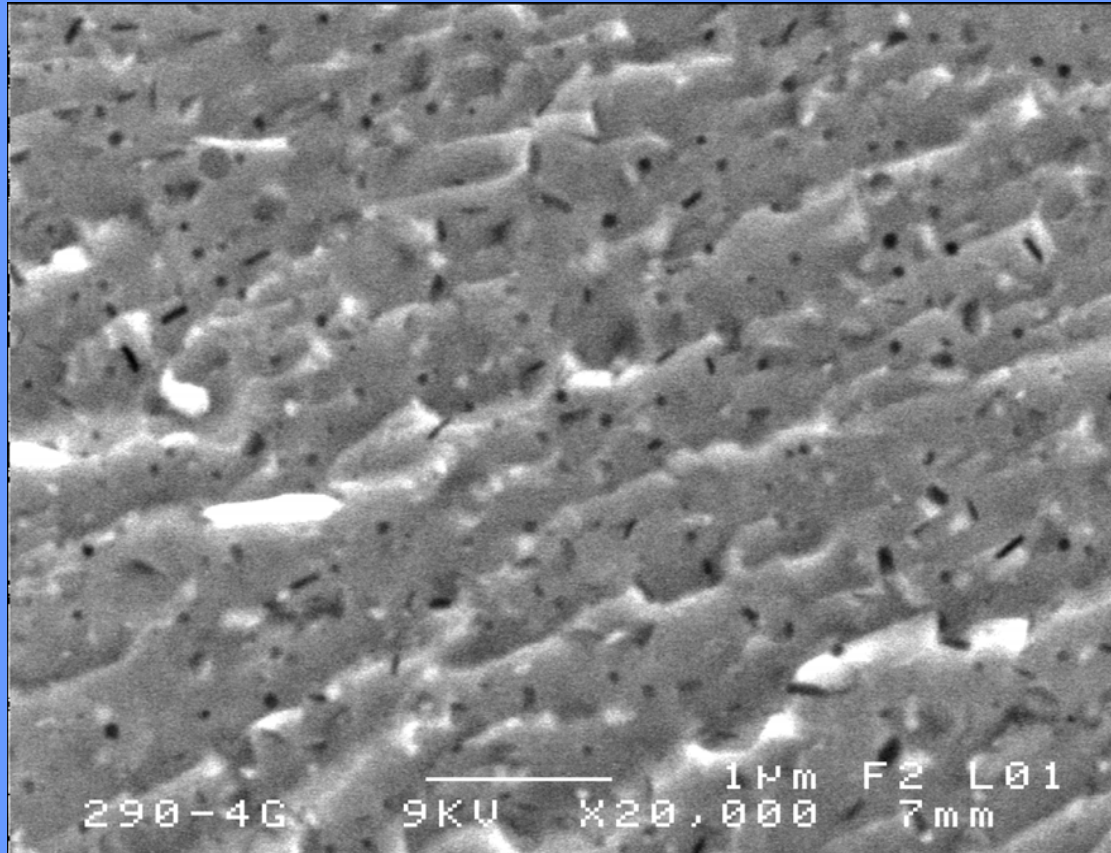
Ge-Sb-S-Cs-Cl glass with
CsCl crystals

X. Zhang et. al. , J. Non-
crystalline Solids 337 (2004)
130

**Lab. glasses and ceramics,
University of Rennes,
France**

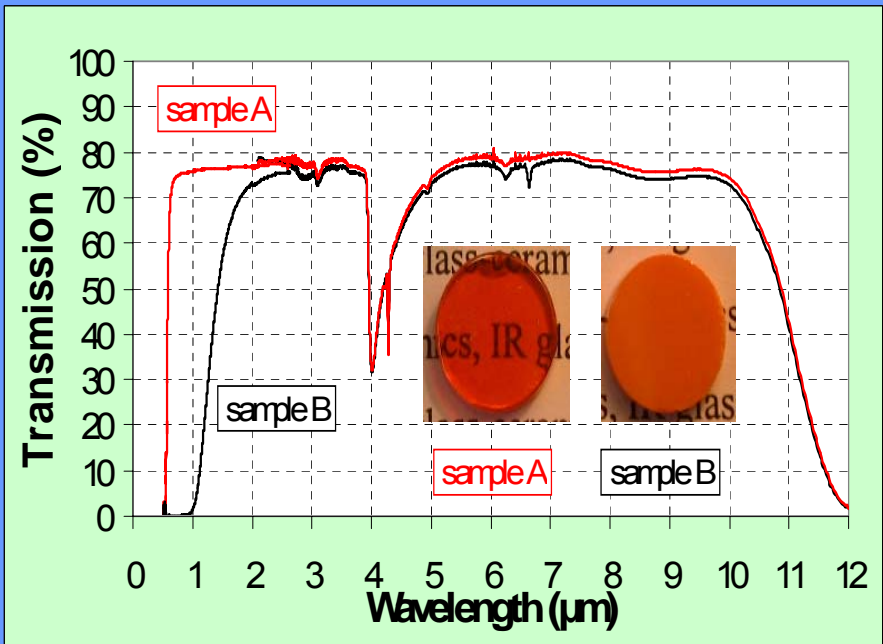
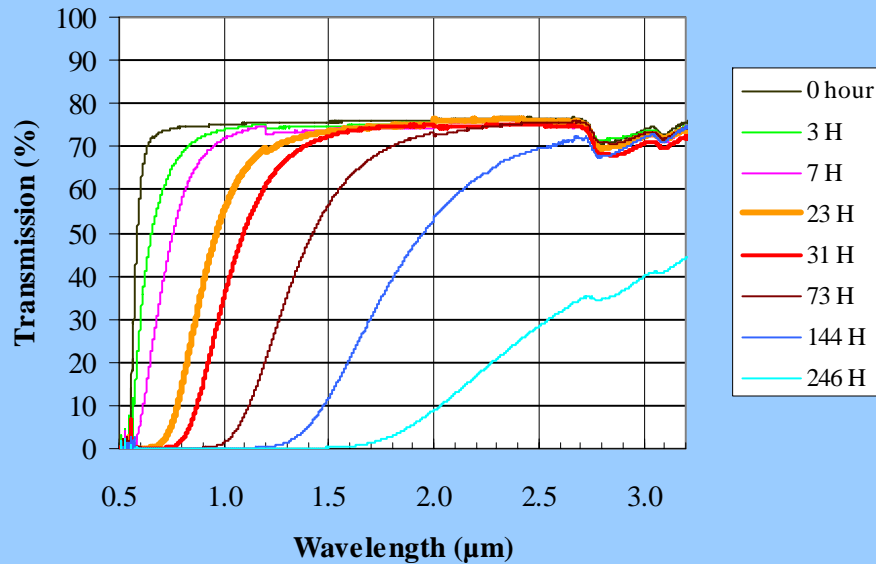
Typical microstructure of IR glass-ceramics

100nm CsCl
crystals



Zhang et. al.

IR transmission versus crystallinity

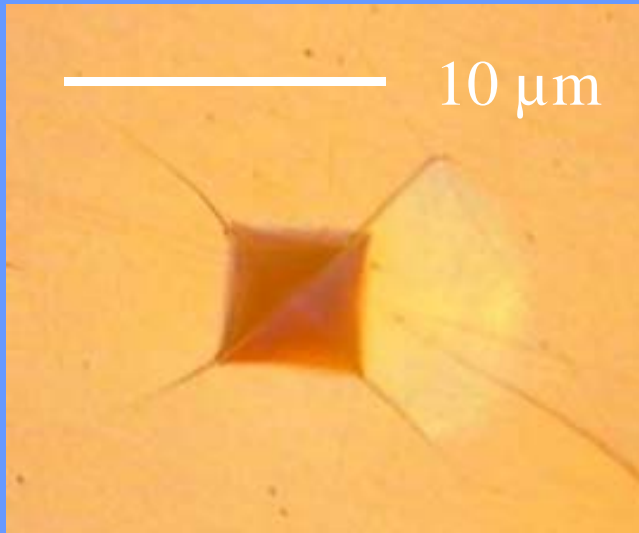


Zhang et. al.

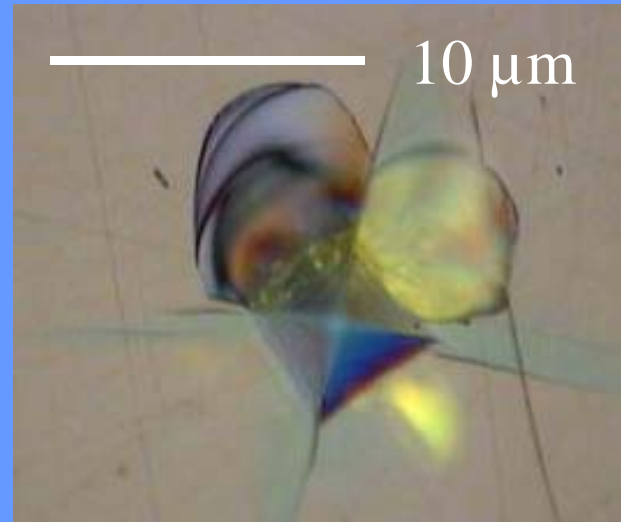
Night vision



Resistance to fracture propagation



GC



Glass

Zhang et. al.

Glass-Ceramic for Solid State Lighting - *White LED*

Ce:YAG-GC

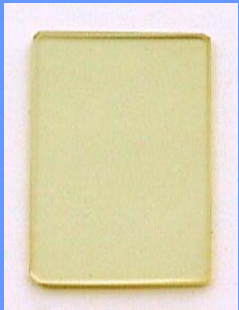
Setsuhisa Tanabe
Kyoto University, Kyoto, Japan

Shunsuke Fujita, Akihiko Sakamoto, Shigeru Yamamoto
Nippon Electric Glass, Otsu, Japan

Presented at the ACerS meeting, Baltimore, April 2005

YAG-GC from glass- microstructure

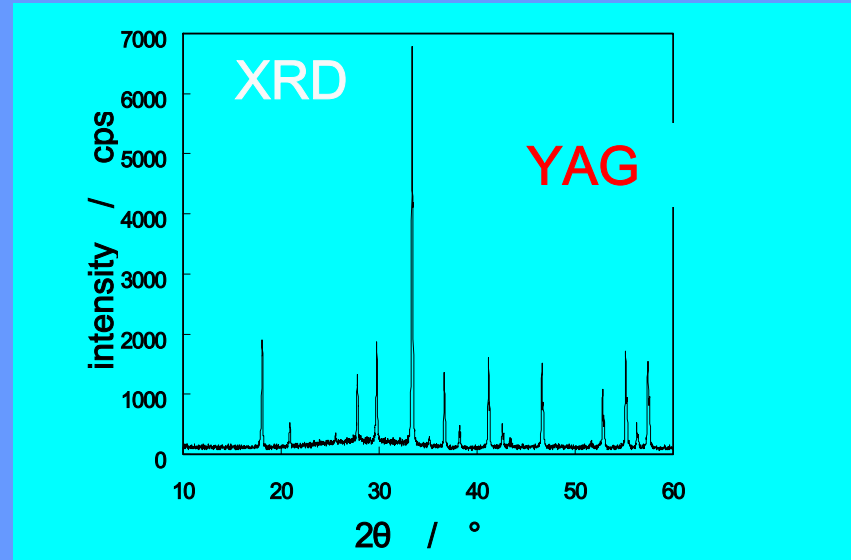
As-made



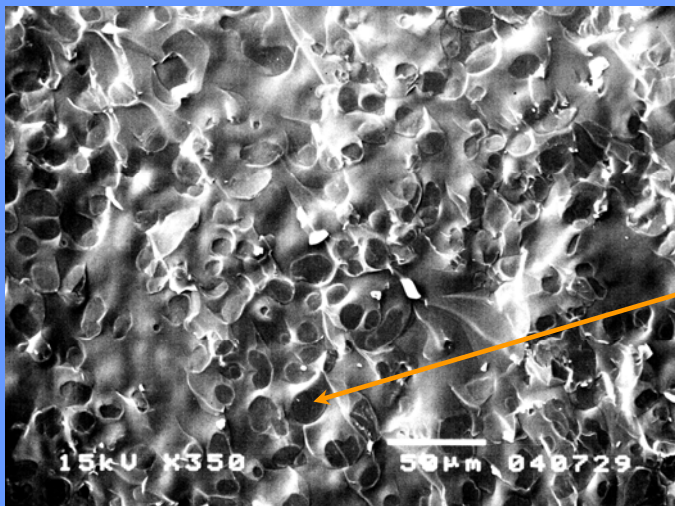
Cerammed



1cm



SEM

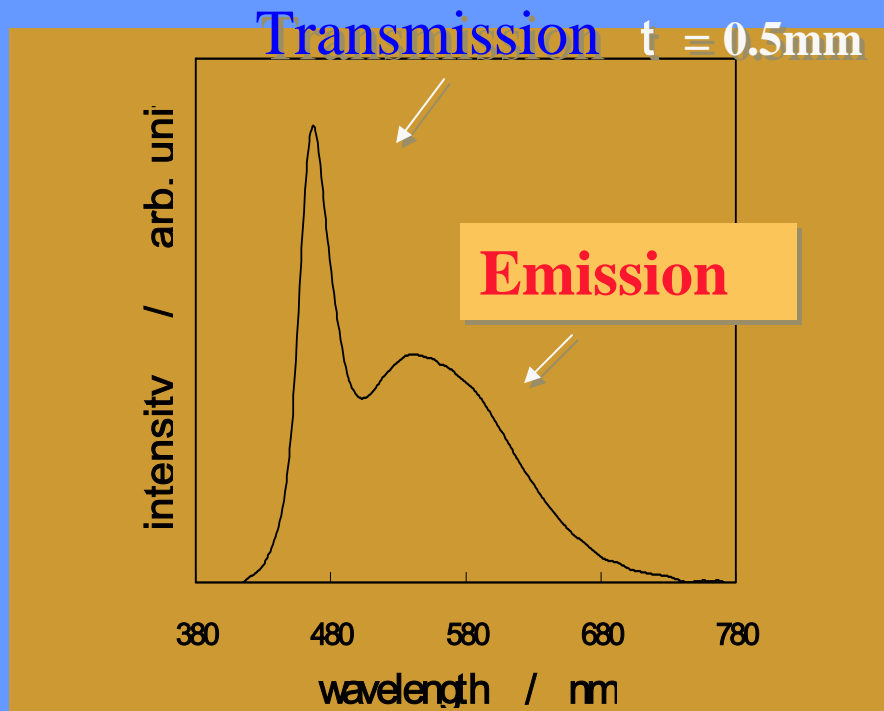


YAG
20µm

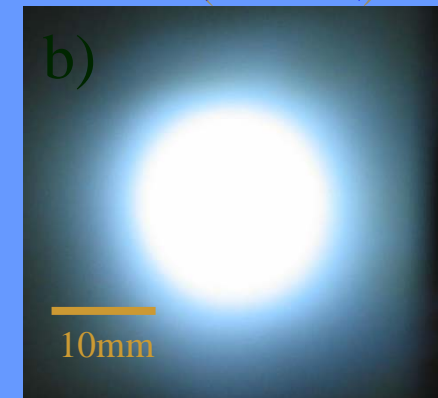
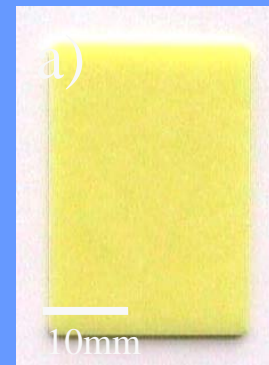
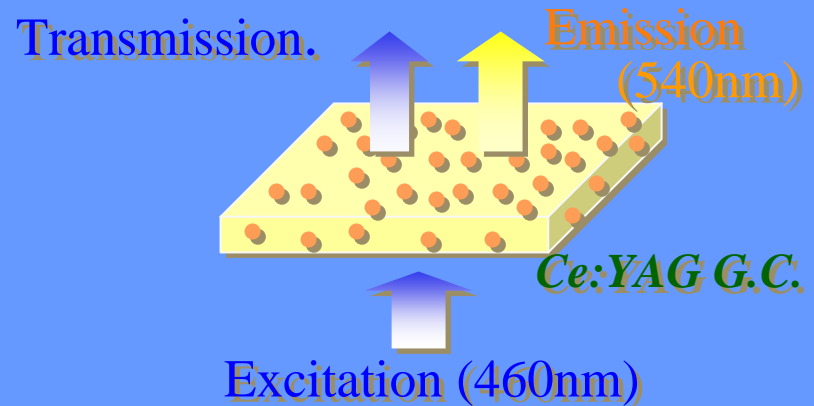
60%
Crystallinity
(wt%)

S. Tanabe et al.

Moderate transmission of blue Yellow fluorescence



White light



a) Ce:YAG GC

b) White light emission from Ce:YAG G.C.

Solid-State Lighting (future)

Promise of LEDs for illumination

	Efficiency	Life
Incandescent Light Bulb	16 <i>lm / W</i>	1000h
Fluorescent Lamp	80 <i>lm / W</i>	10,000h
Today's white LED	60 <i>lm / W</i>	20,000h
Future white LED	200 <i>lm / W</i>	100,000h

Efficiently bright, broad spectrum, long-lifetime...

S. Tanabe et al.

Laser crystallization in Nagaoka

Takayuki Komatsu & collaborators
(Benino, Ihara, Fujiwara, et al.)

Department of Chemistry
Nagaoka University of Technology
Japan

Example: Appl. Phys. Lett., 82 (2003) 892, 83 (2003) 2796.

Laser crystallization in glass

Rare-earth (Sm or Dy) atom heat processing

1. CW Nd:YAG laser irradiation to Sm_2O_3 or Dy_2O_3 containing glasses

2. Absorption and non-radiative relaxation

Irradiated region is heated \longrightarrow Crystallization

Writing of nonlinear optical/ferroelectric crystal dots and lines

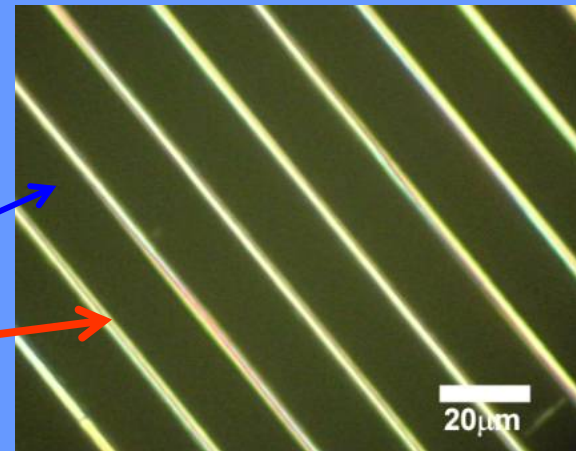
- $\text{Sm}_2\text{O}_3\text{-BaO-B}_2\text{O}_3 \rightarrow \beta\text{-BaB}_2\text{O}_4$
- $\text{Sm}_2\text{O}_3\text{-Bi}_2\text{O}_3\text{-B}_2\text{O}_3 \rightarrow \text{Sm}_x\text{Bi}_{1-x}\text{BO}_3$
- $\text{Sm}_2\text{O}_3\text{-MoO}_3\text{-B}_2\text{O}_3 \rightarrow \beta'\text{-Sm}_2(\text{MoO}_4)_3$
- $\text{Sm}_2\text{O}_3\text{-K}_2\text{O-P}_2\text{O}_5 \rightarrow \text{KSm}(\text{PO}_3)_4$

$\text{Sm}_2\text{O}_3\text{-Bi}_2\text{O}_3\text{-B}_2\text{O}_3$ glass

$\text{Sm}_x\text{Bi}_{1-x}\text{BO}_3$ crystal

Power: 0.66W

Scanning speed: $10\mu\text{m/s}$



$20,000$
 J/cm^2

Polarization optical microscope

10Sm₂O₃.35Bi₂O₃.55B₂O₃ glass

T_g=474°C, T_x=574°C

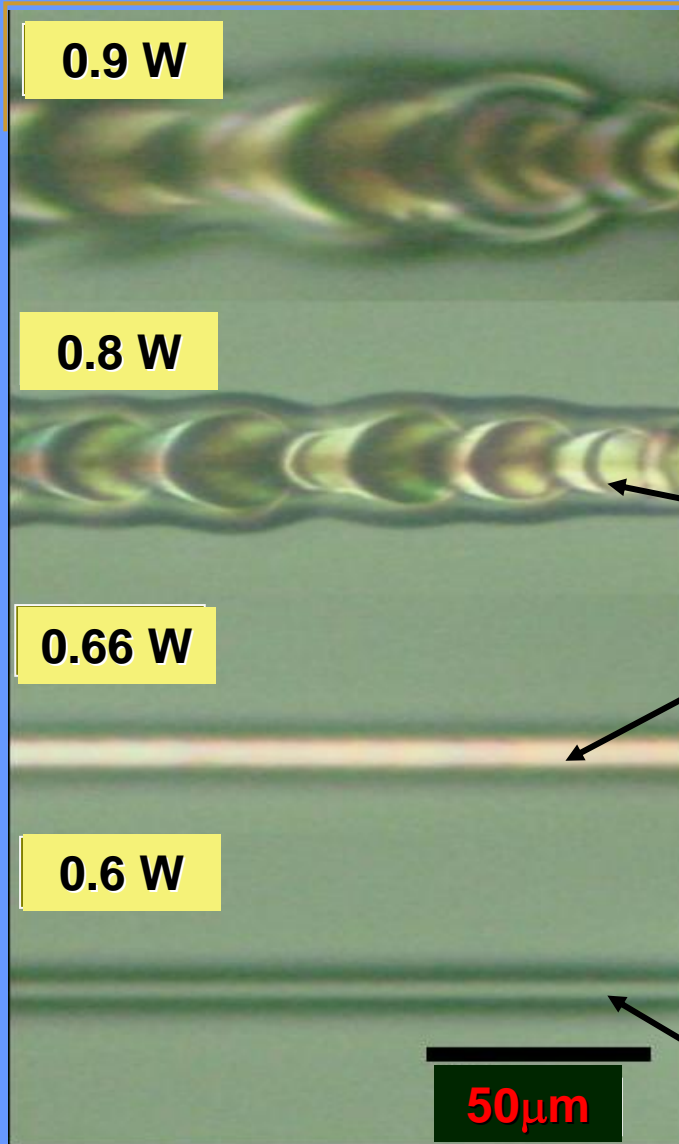
Sm_xBi_{1-x}BO₃

crystal

Temp. >> T_x

Temp. < T_x

Refractive index change



Laser crystallization in São Carlos

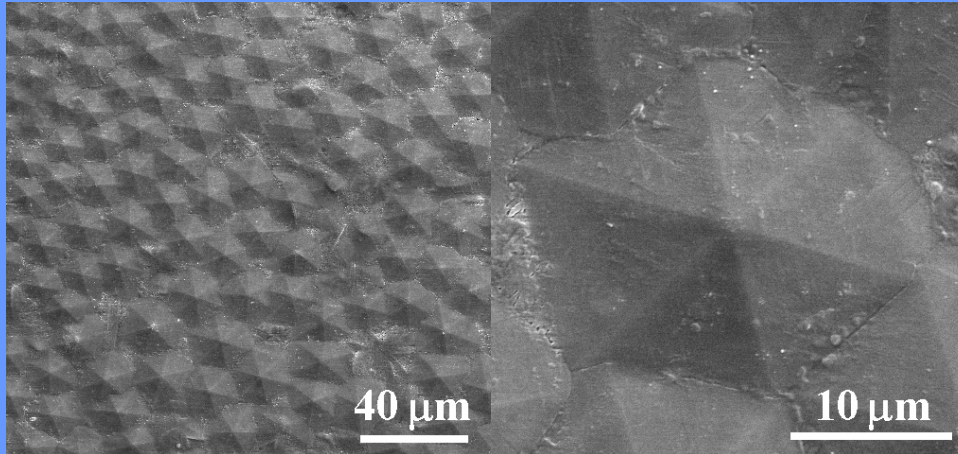
C. A. C. Feitosa, L. J. Q. Maia, A. L. Martinez,
A. C. Hernandez, **Valmor R. Mastelaro**,

IFQSC,
University of
São Paulo, São
Carlos, Brazil

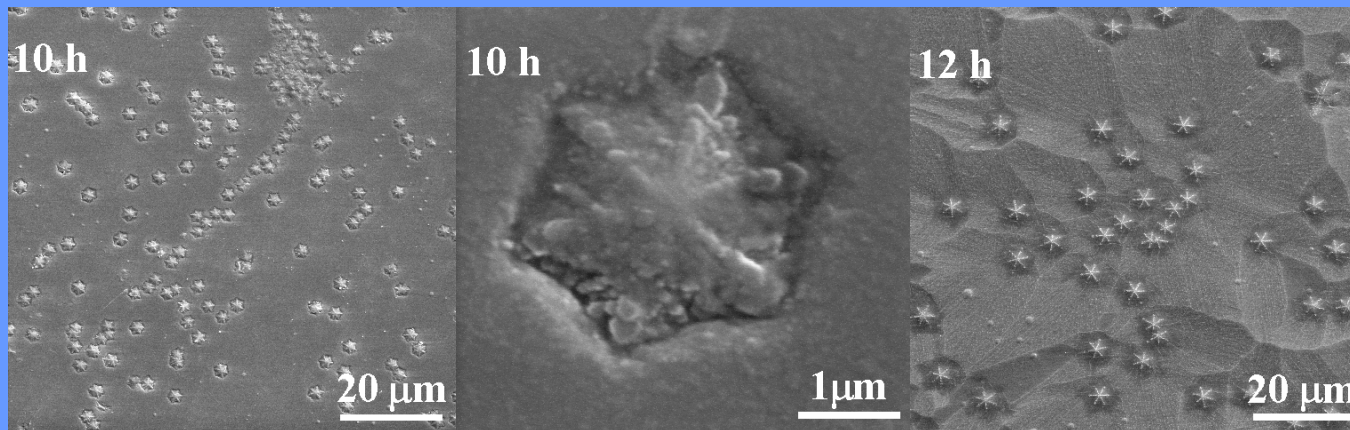


40BaO - 45B₂O₃ - 15 TiO₂ (BBT)

Microstructures from two crystallization processes

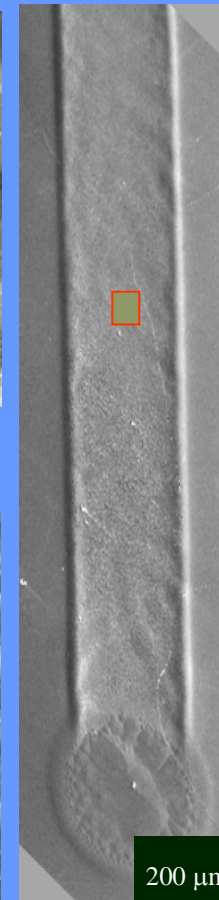
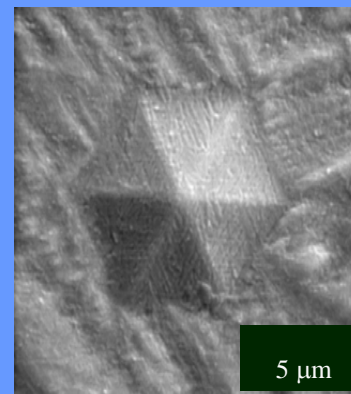
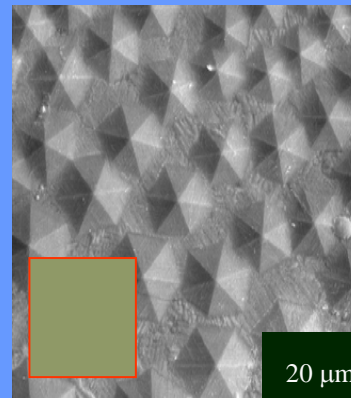
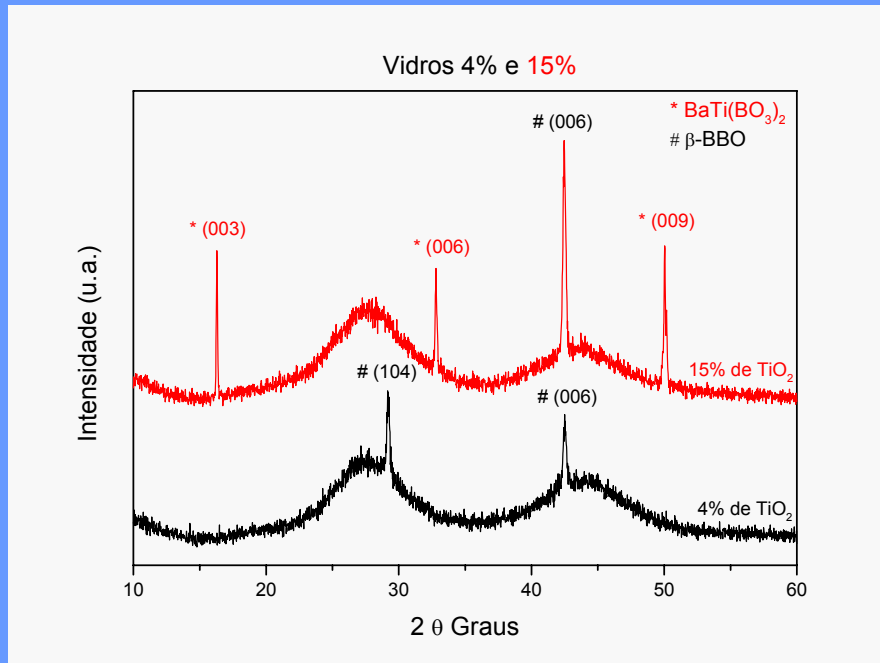


BBT glass after irradiation with CO₂ laser ($\lambda = 10.6 \mu\text{m}$) 4 min, 40 W/cm².
= 10,000 J/cm²
Glass at 300°C ($T_g = 580 \text{ }^\circ\text{C}$)



BBT GC in resistive furnace at 620°C.

Surface crystallization of BBT glass

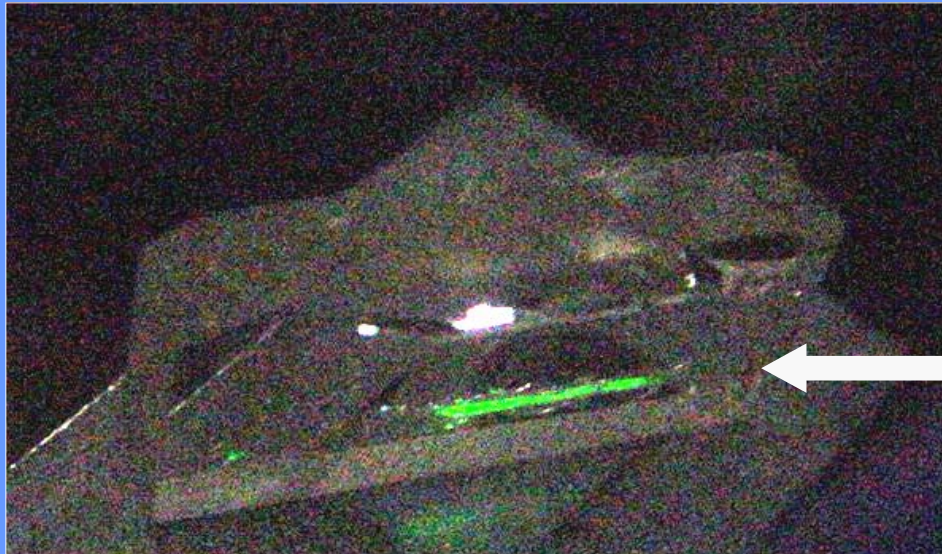


It is possible to produce polycrystalline lines.

Details; crystals within the line and diffraction pattern

SHG in partially crystallized BBT glass

Mastelaro et. al.



Laser beam
Nd:YAG ($\lambda = 1064 \text{ nm}$)



Second harmonic generation

PTR Glasses

Oxy fluor bromide glasses

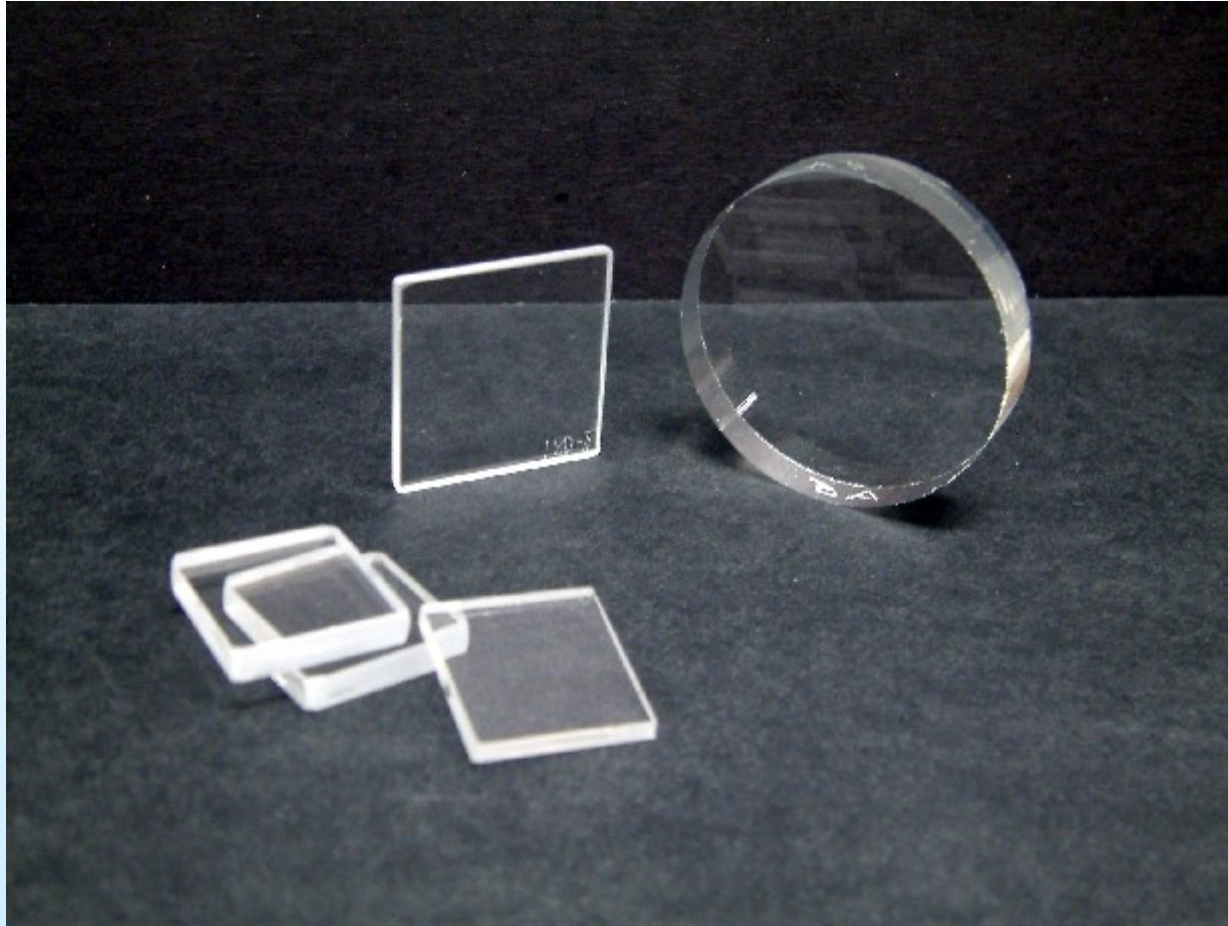
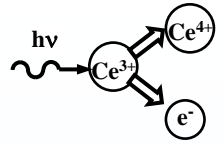
S.D. Stookey et al. (1954) – Corning, USA

L.B. Glebov et al. (1990) - Vavilov SOI, Russia + Creol/ UCF, USA

- **Composition**
- **Major: SiO_2 , Na_2O , ZnO , Al_2O_3 , K_2O**
- **Minor: **F, Br****
- **Dopants (~100 ppm): **Ag, Ce, Sb, Sn****
- **Impurities (< 2 ppm): **transition metals****



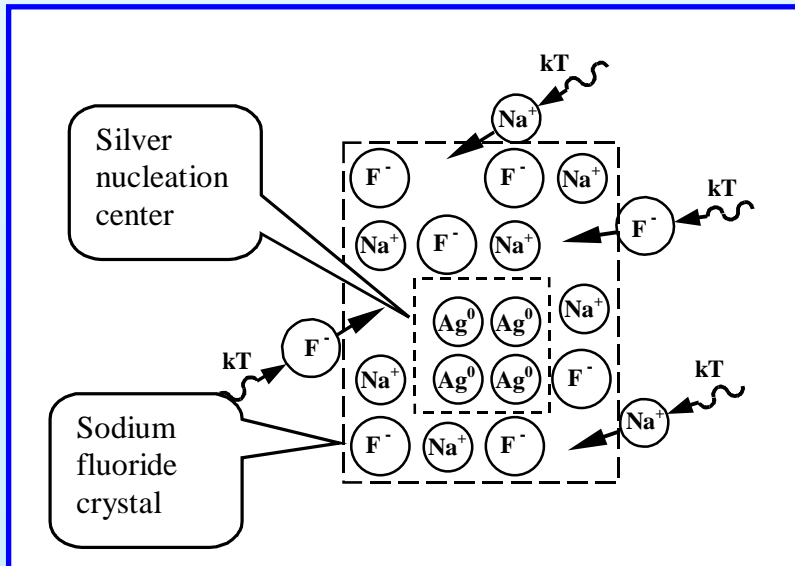
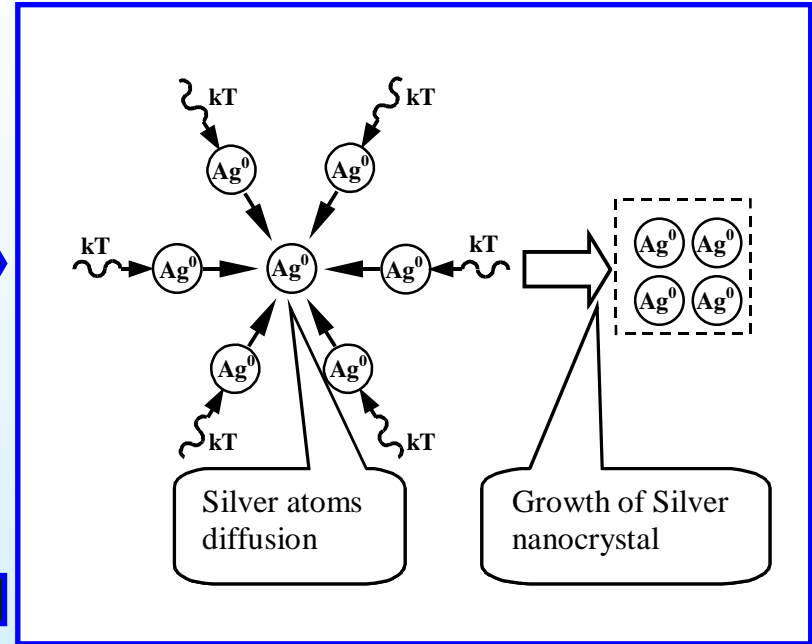
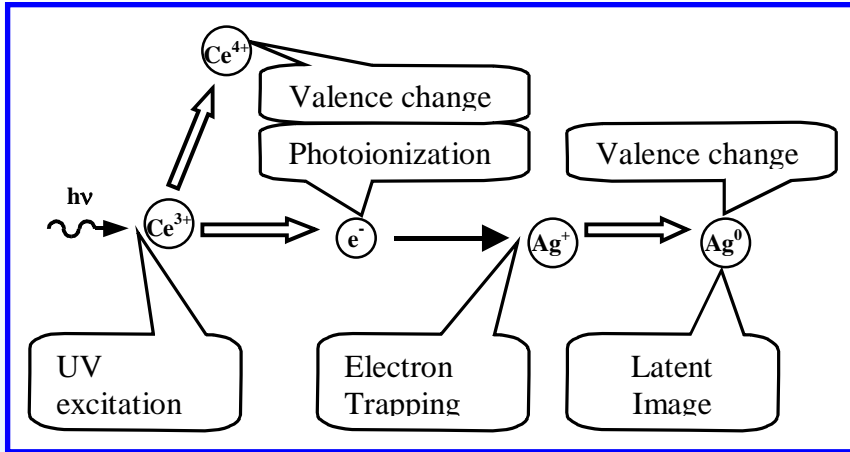
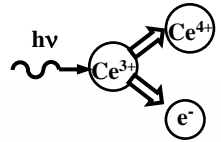
PTR glass is a F-Br sodium-zinc-aluminum-silicate glass doped with Ag, Ce, Sn and Sb



Current technology at UCF/CREOL - optical quality PTR glasses with aperture up to 50 mm.

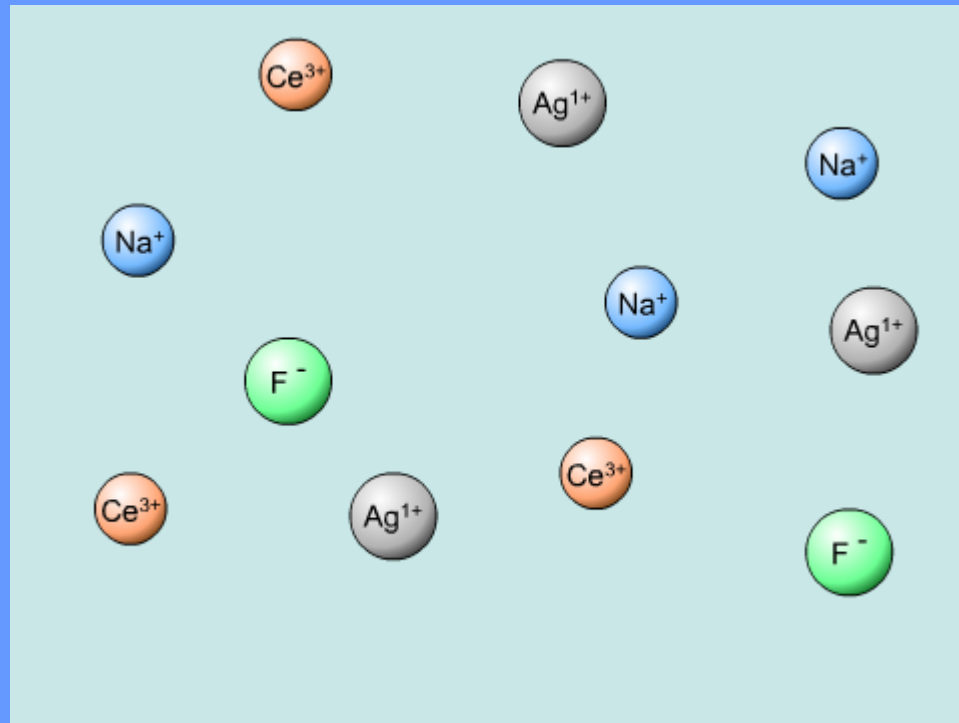


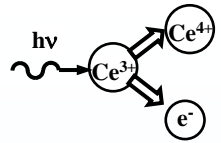
Mechanism of photo-thermo-crystallization



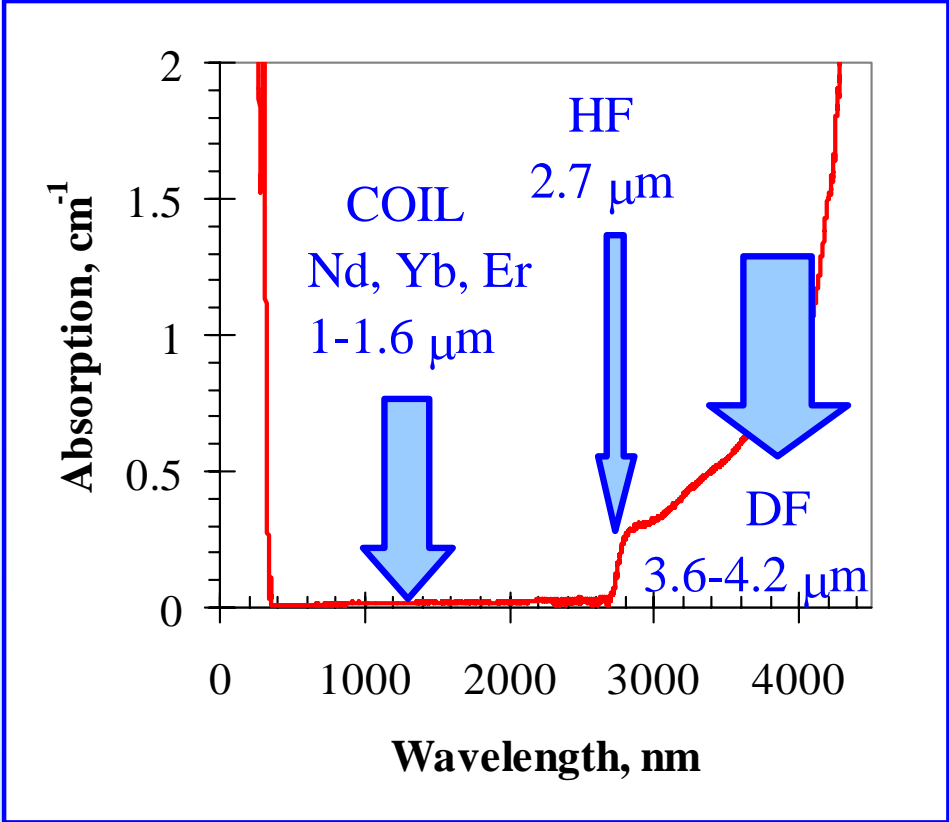
3D image (hologram) of object is transformed to the phase pattern (refractive index variations) caused by selective NaF crystal distribution in accordance with the UV intensity distribution in glass interior.

PTRG (only the active ions are shown) Proposed mechanism of photo induced crystallization





Absorption spectrum of photo-thermo-refractive glass



No detectable absorption in the range of 1 μm
Absorption of hydroxyl in the range of 4 μm

PTR glasses

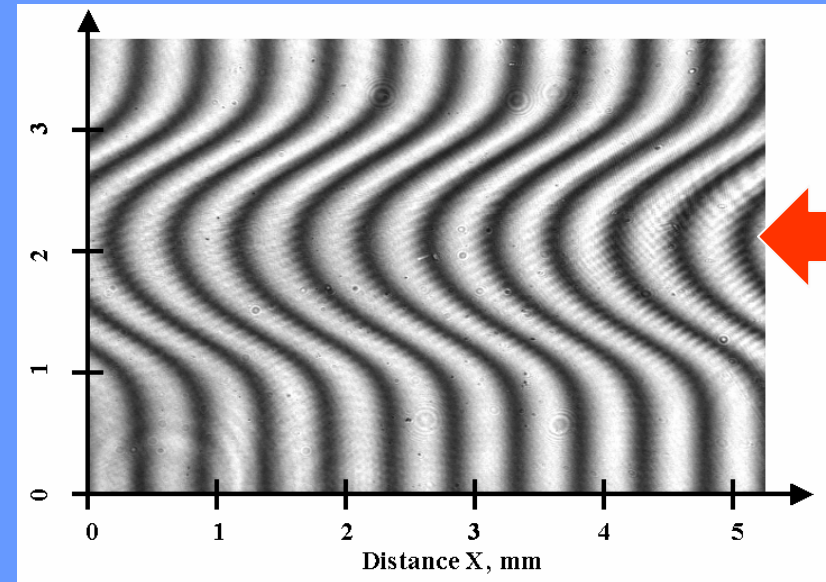
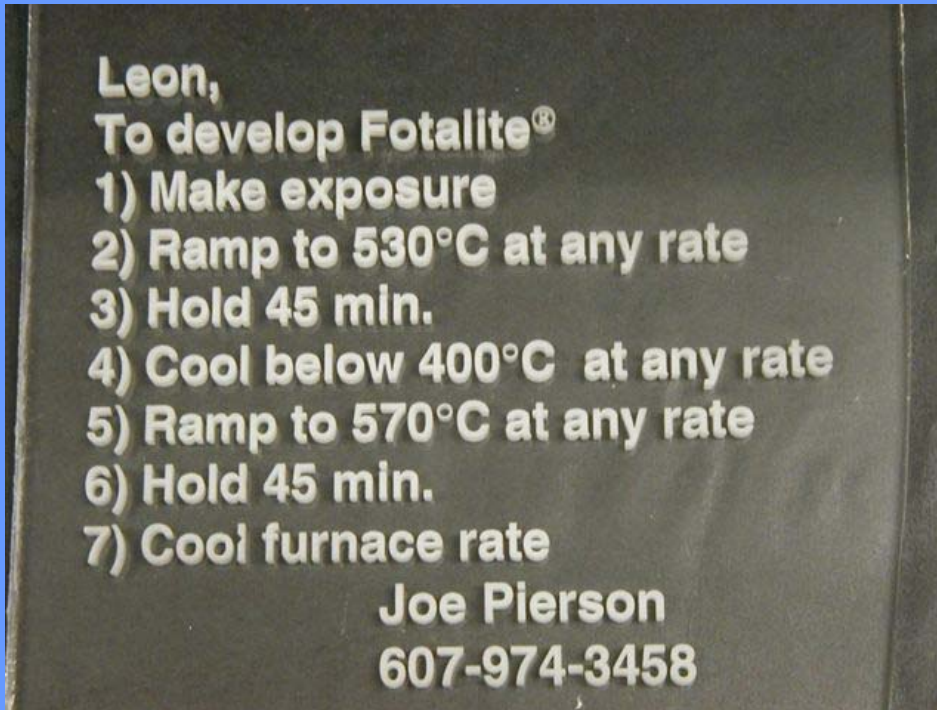
S.D. Stookey et. al.

Corning's Fotalite



Creol's PTRG

Hologram Leon Glebov et. al.



LARGE GRAIN, HIGHLY CRYSTALLINE, HIGHLY TRANSPARENT GC

T. Berthier, V.M. Fokin, E.D. Zanotto
LaMav- Federal University São Carlos, Brazil



Vlad Fokin



Thiana
Berthier



STRATEGY

Simultaneous compositional variation

**of solid solution crystals
and glassy matrix**

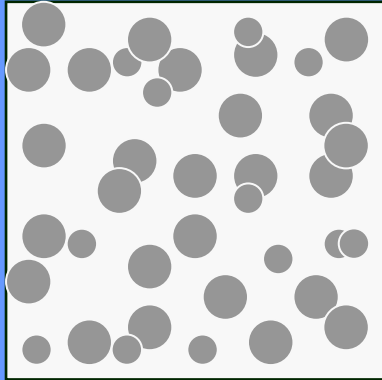
decreases Δn

New type of transparent glass-ceramic

small or large grain size

high crystallized
volume fraction

OPTICAL PROPERTIES



Transmission Spectra

200 nm – 1100 nm

Crystal morphology

Grain size

Degree of crystallinity OM

Transmittance measured for different sample thicknesses

Estimated parameters (P_1 and P_2):

$$\frac{I}{I_0} = P_1 \exp(-P_2 x)$$

$$P_1 = (1-R)^2$$

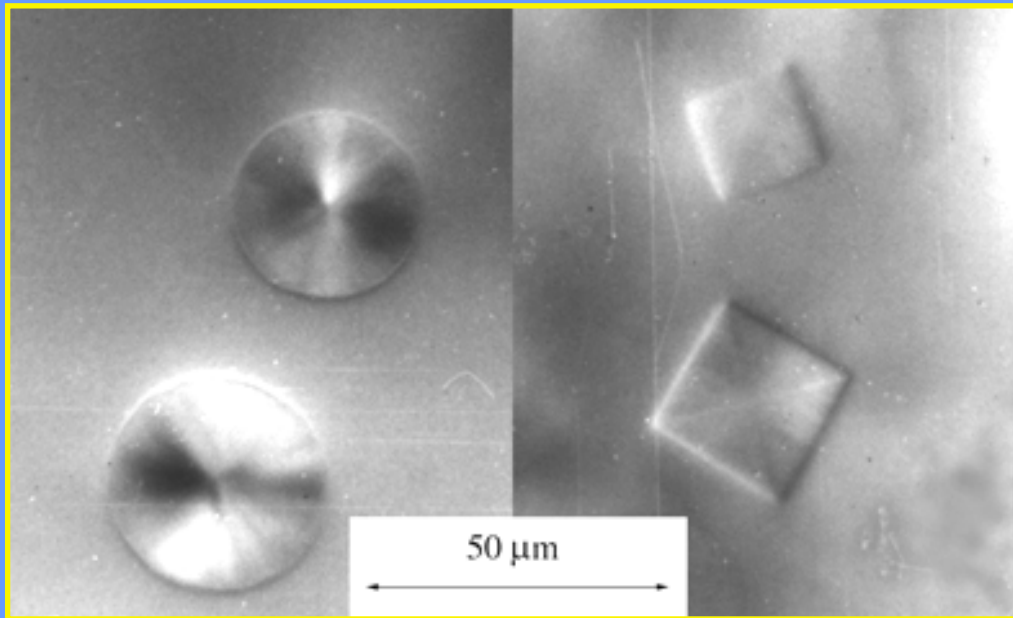
$$P_2 = (\beta+S)$$

MICROSTRUCTURES

The crystals are solid solutions: $TA_{4+2x}AE_{4-x}[GF_6O_{18}]$ ($0 \leq x \leq 1$)

Their morphology can vary from

J, spherical to **V8**, cubic



T = trace element

A = alkali

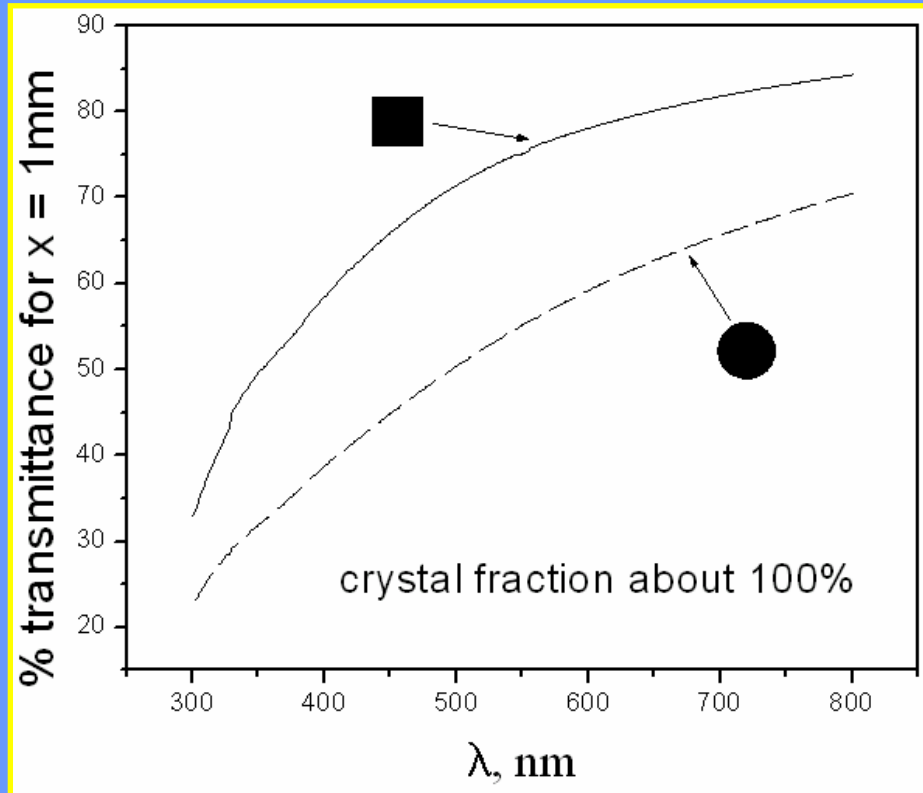
AE = alkaline earth

GF = Si, P, B

TRANSMITTANCE

Morphology

Distinct crystal shapes \longrightarrow Different transmittances



V8, cubic 5-6 μm
J, spherical 7-8 μm

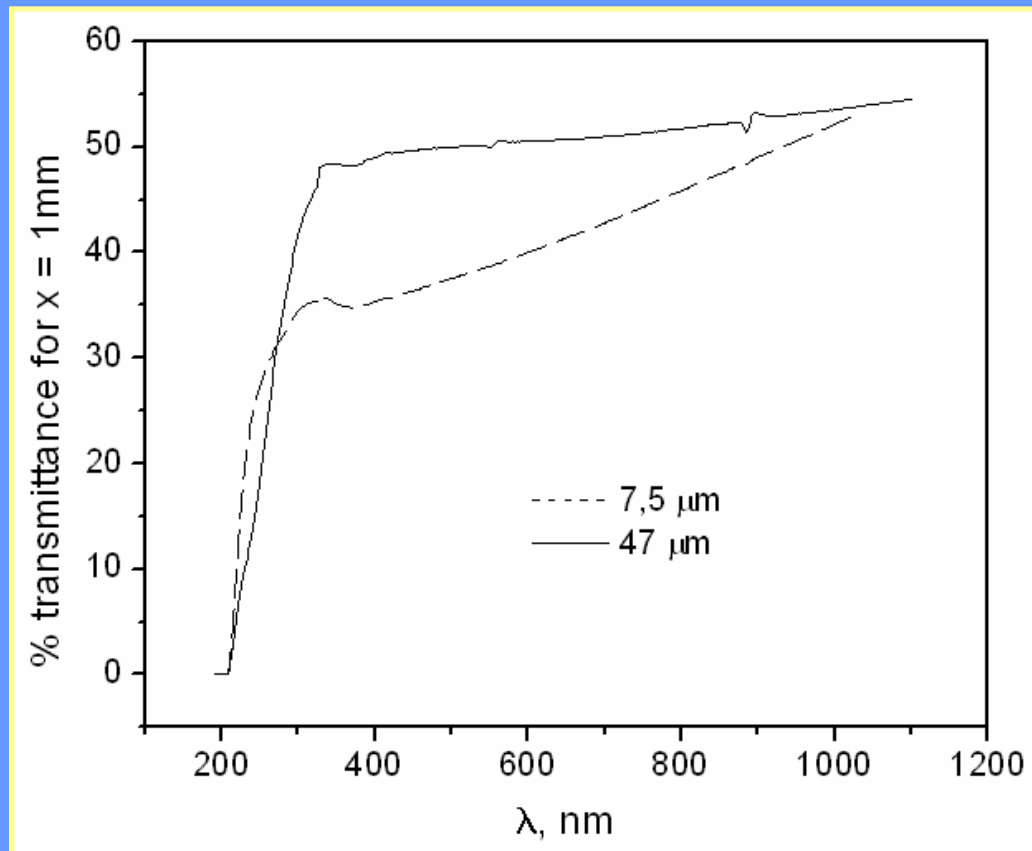
crystal/crystal
Interfaces are
quite different for
spherical and
cubic crystals

Best transmittance \longrightarrow Cubic crystals

TRANSMITTANCE

Grain size

glass J, spherical crystals, ~42% crystallized



$I(\lambda)$ dependence



Crystal size



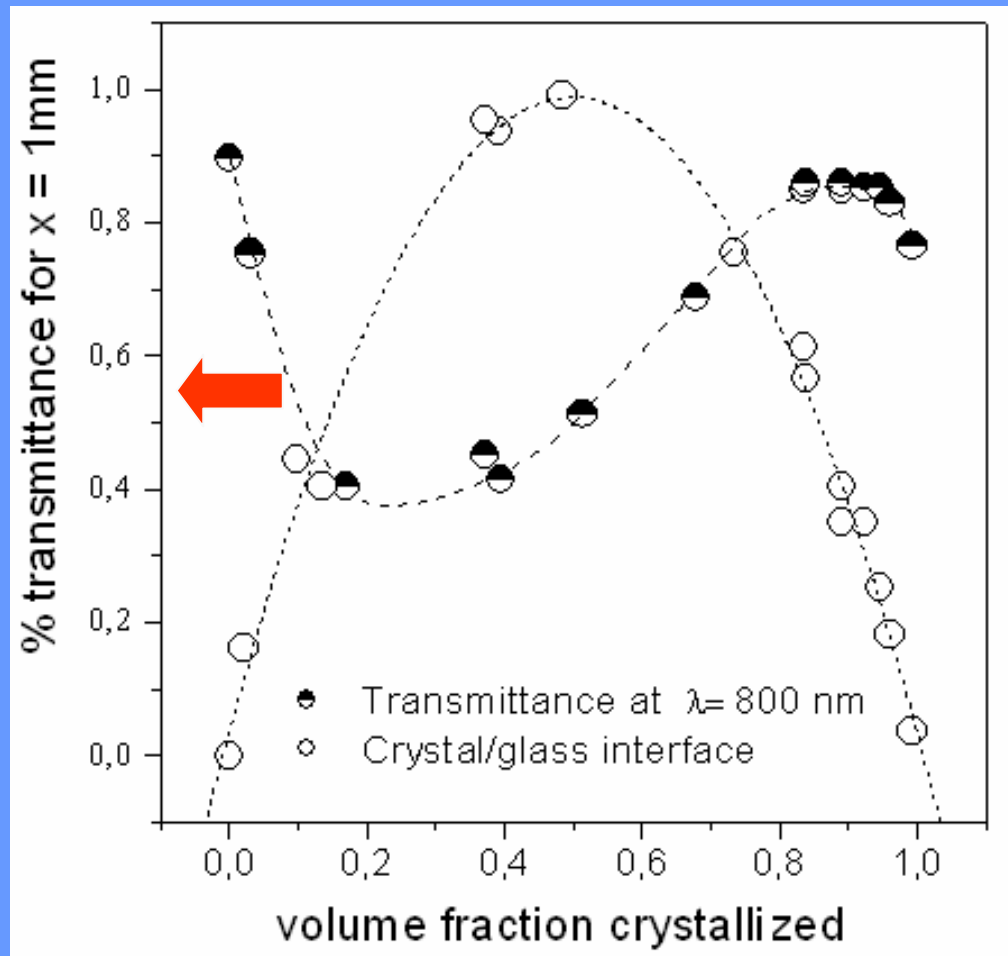
Affects P_2

Importance of thermal history

TRANSMITTANCE

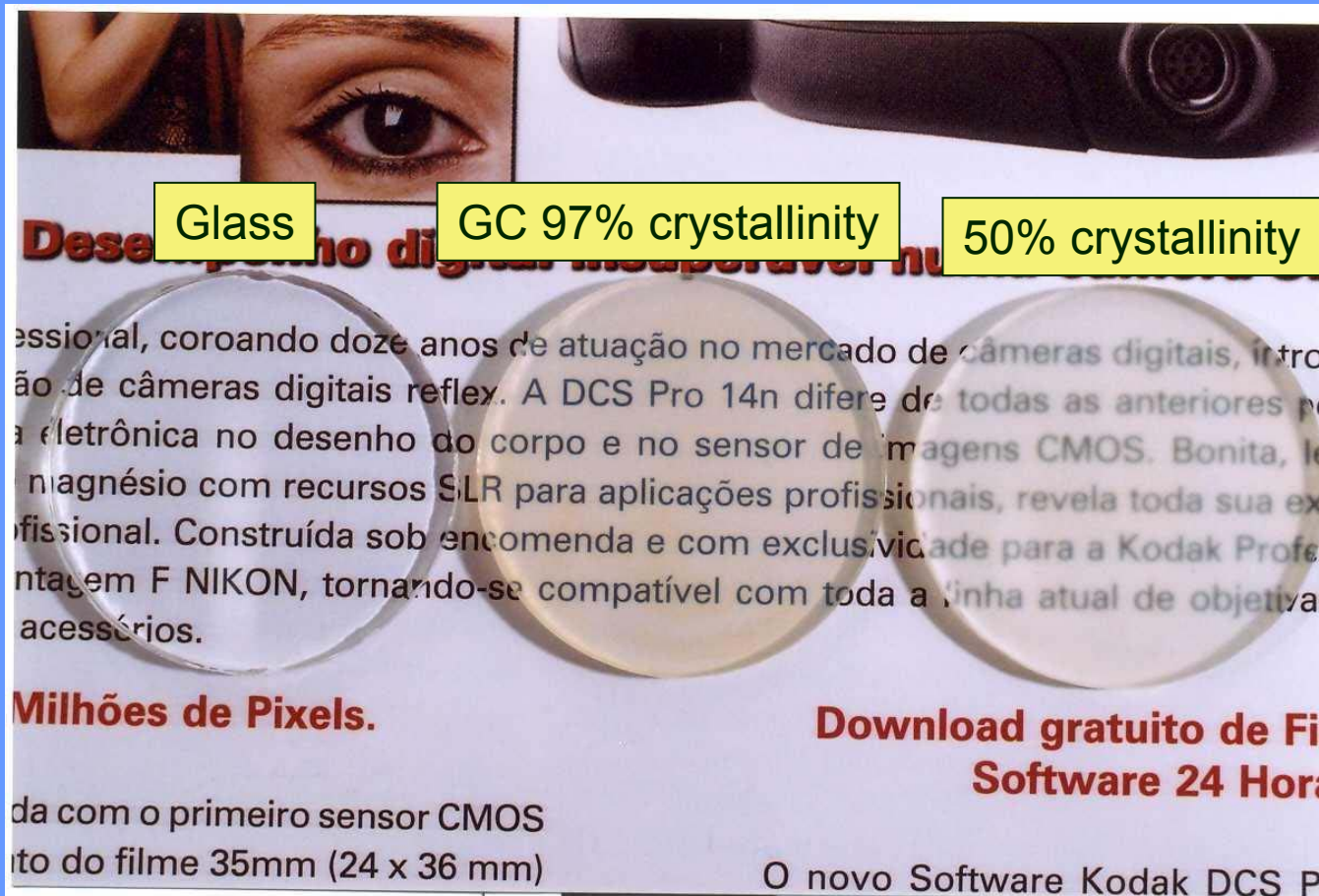
Degree of crystallinity

glass V8, cubic crystals (3-5 μm)

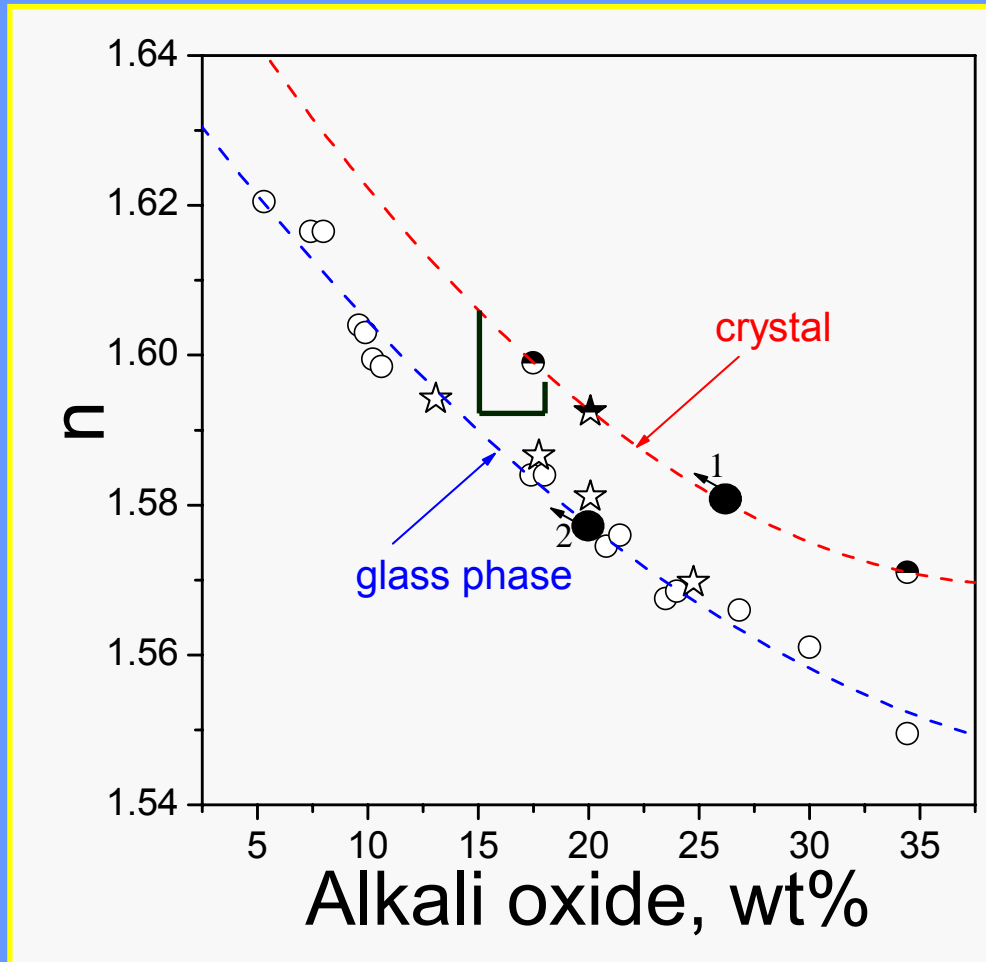


Glass V8 & T6, maximum transmission for ~ 95-97% OM crystallinity

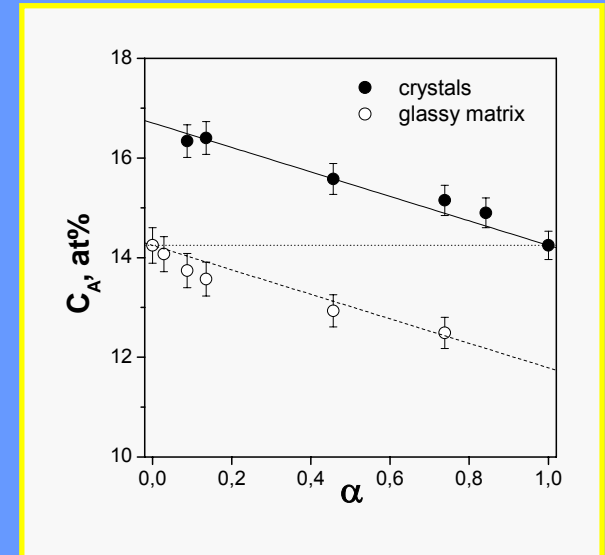
The beasts! Transparency of 4 mm thick specimens



DISCUSSION



EDS measurements



alkali content in
crystals

30% > glassy matrix

DISCUSSION

**High crystallized
vol. fraction**



reduced
crystal / glass
interface



Simultaneous variations of the
glass-matrix and s/s-crystal compositions
during crystallization



**refractive indexes
of crystal and glass
verge**

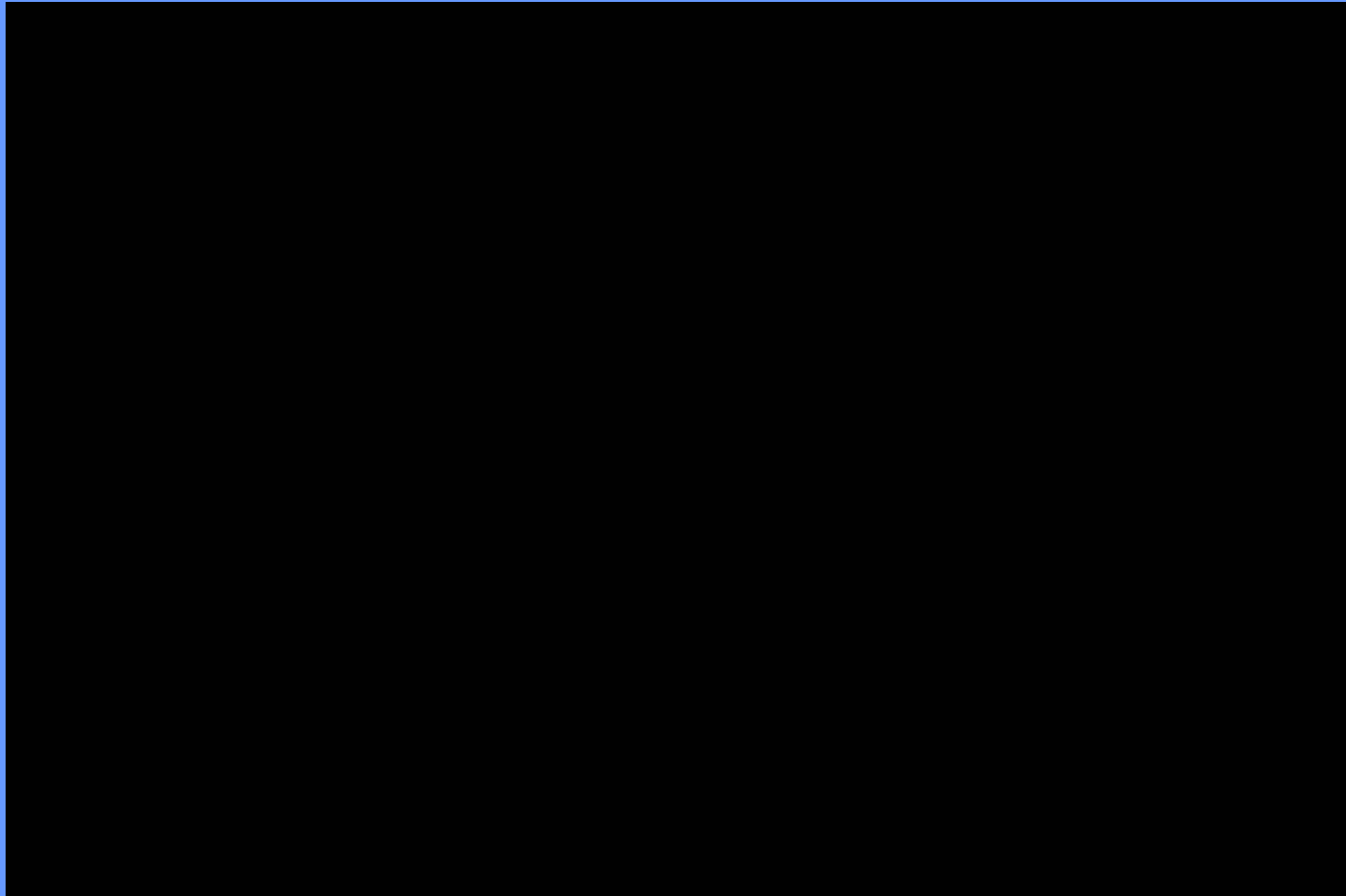


**main reasons for improved
transparency of these new TGC**

Mechanical behaviour of HCHT-GC

A new, specially designed,
method of impact testing!

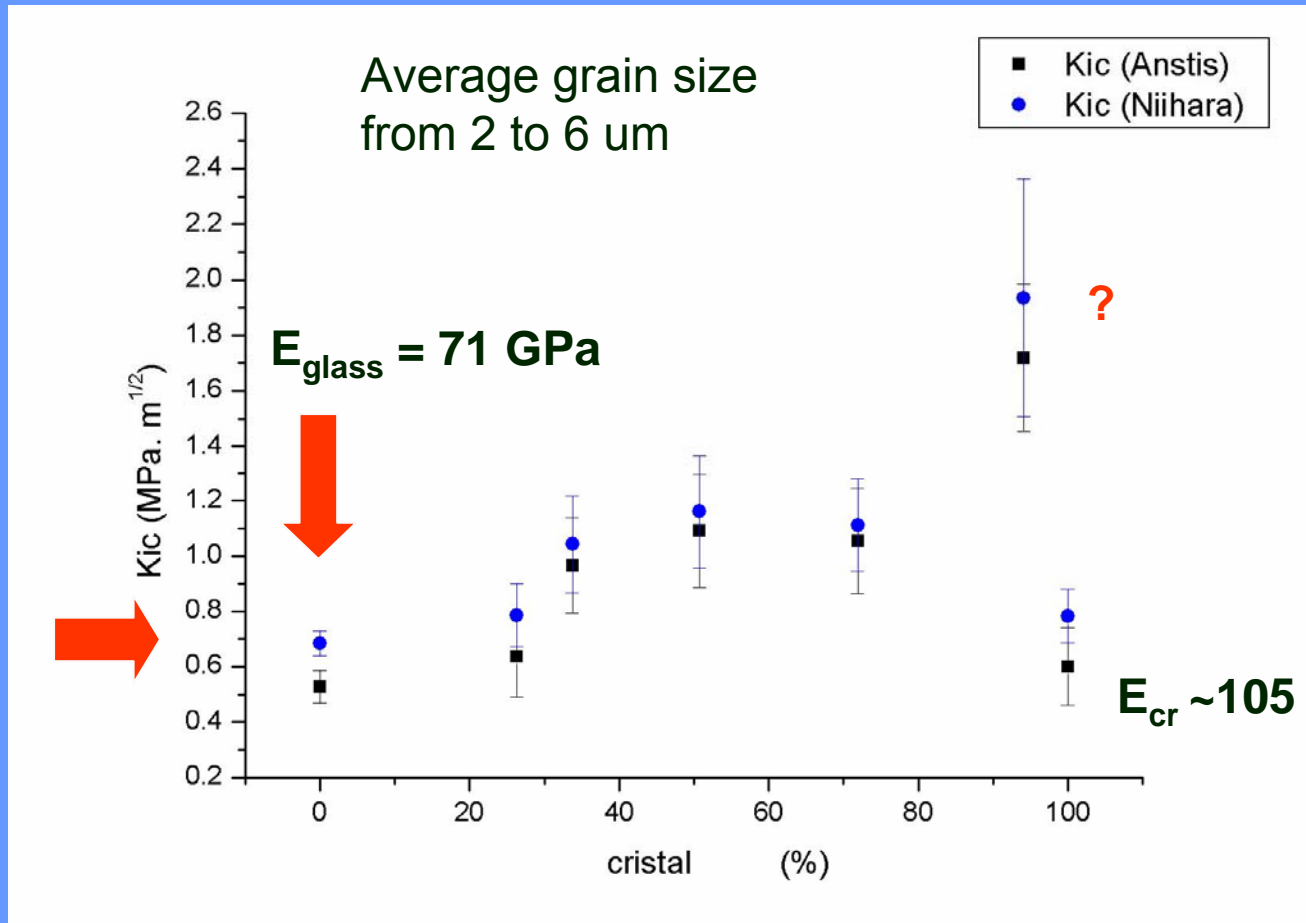
Impact testing of glass



Courtesy of Leo Siiman, Creol/ UCF

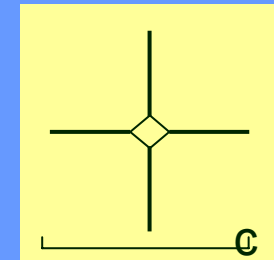
Don't try this in your
labs!

K_{IC} versus volume fraction crystallized



Anstis

$$K_{IC} = 0.016 \left(\frac{E}{H} \right)^{\frac{1}{2}} \frac{F}{c^{\frac{3}{2}}}$$

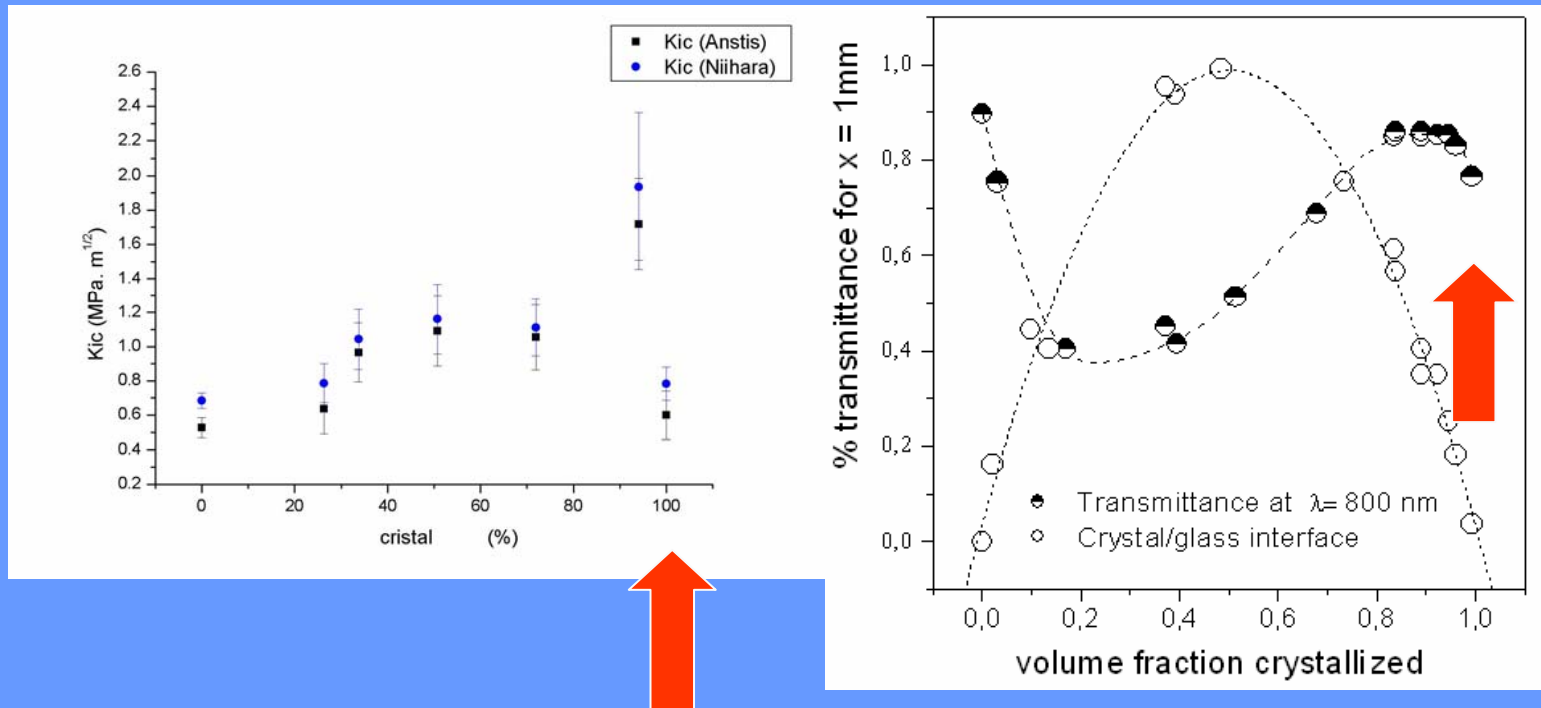


Nihara

$$k_{ic} = 0.035 \left(\frac{l}{a} \right)^{\frac{1}{2}} \left(\frac{H}{E\phi} \right)^{\frac{2}{5}} \left(\frac{H\sqrt{a}}{\phi} \right)$$

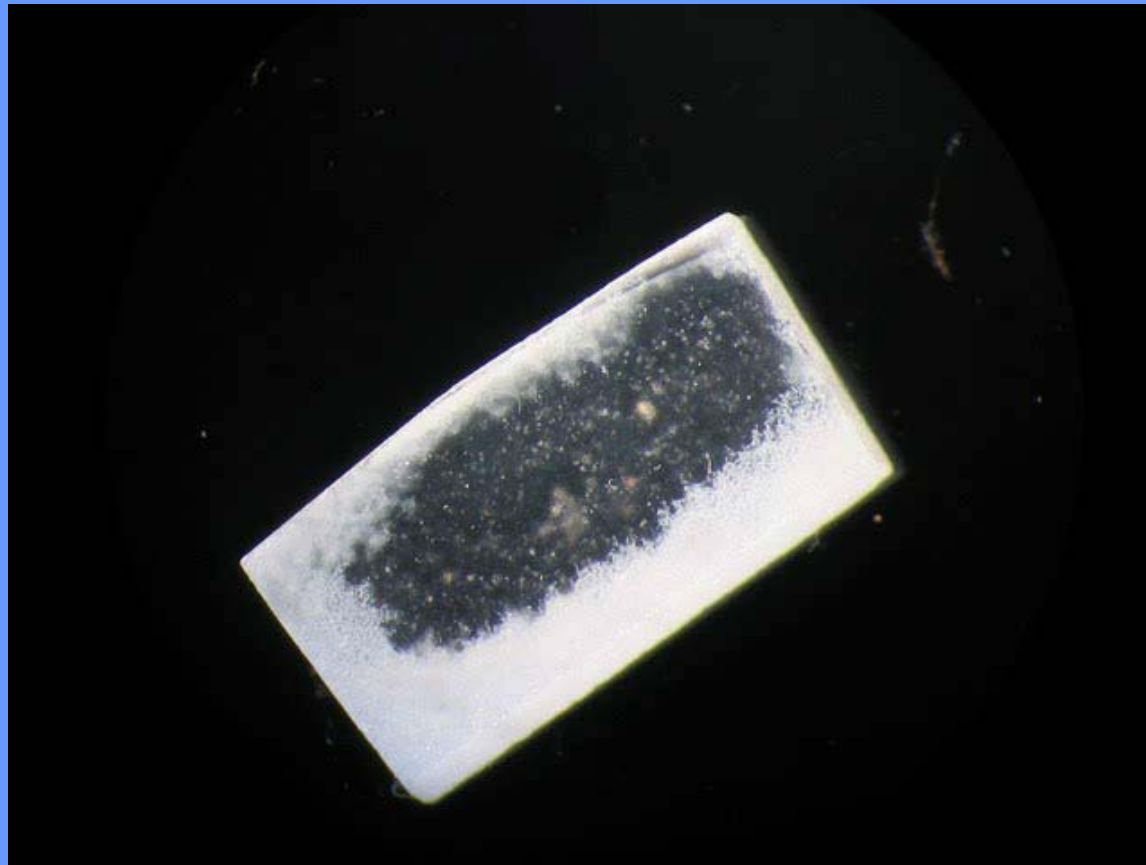
$\phi \sim 3$,
a, l, c [um]

Why do the transparency and impact strength drop significantly when crystallinity > 97%?



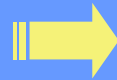
SPONTANEOUS CRACKING for $> 97\%$ crystallinity!

accelerated 300X



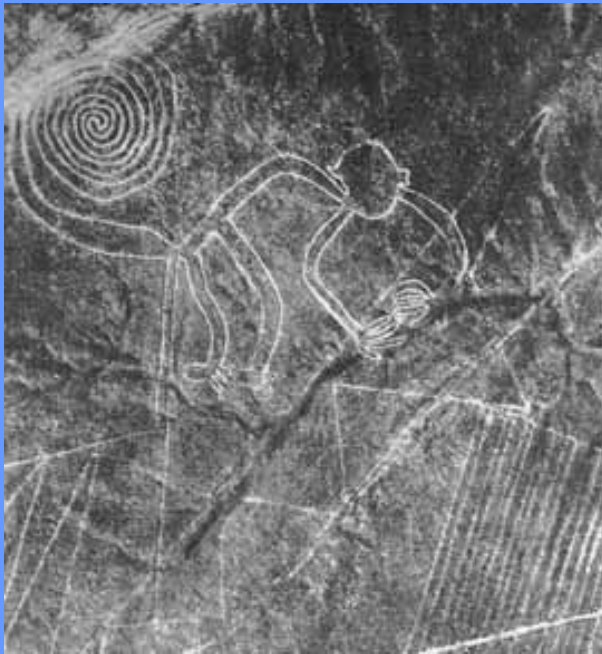
CONCLUSIONS

New
type of TGC



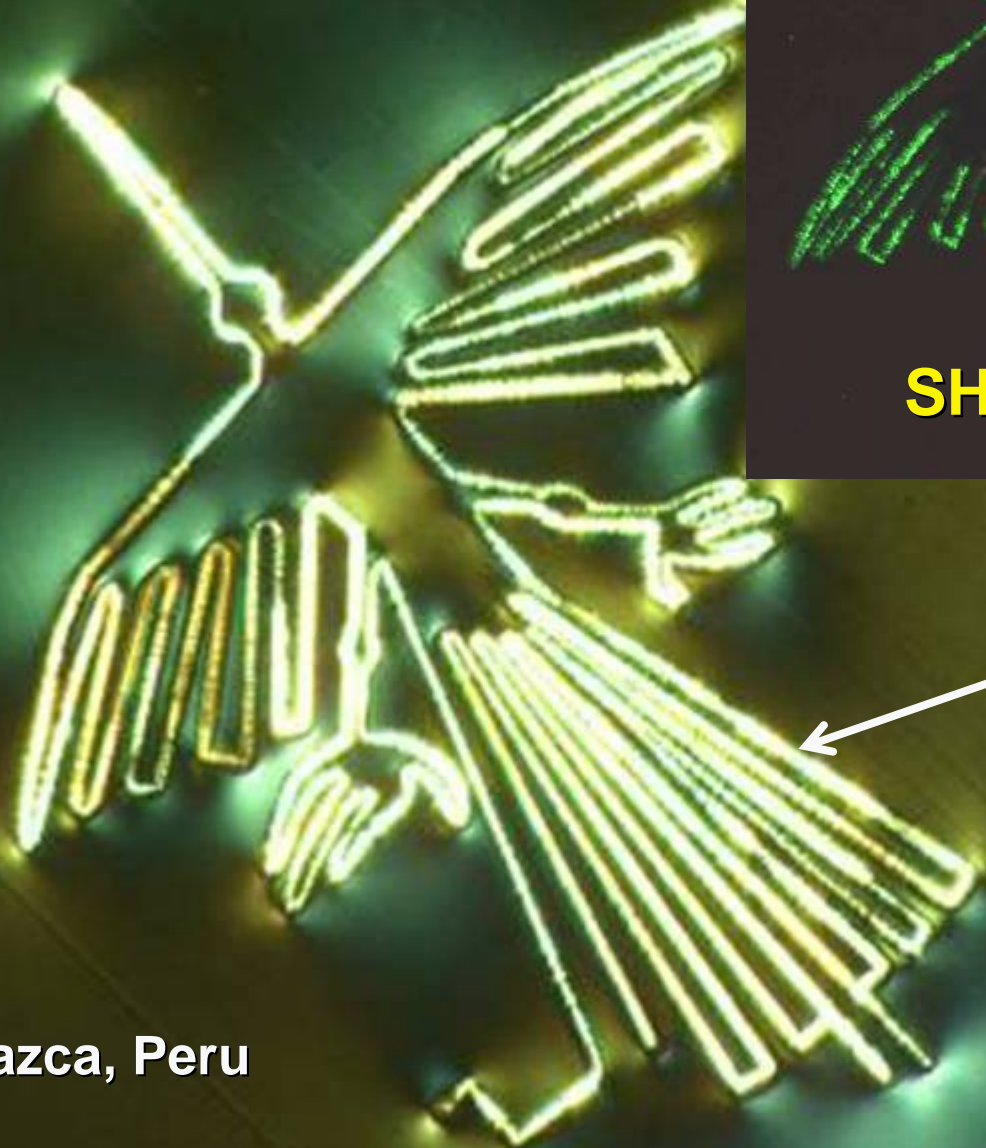
- highly transparent in the visible ~ 90% for 1mm
- nm to μm grain size
- up to 97% crystallized volume fraction
- good mechanical properties, which can probably be much improved by ion-exchange.
- good chemical durability
- can be drawn into fibers
- luminescence ? doping with TM and RE ions should be tested...

On the origem of misterious biomorphs and geoglyphs in Nazca, Peru, 200 B.C.



$\text{Sm}_2\text{O}_3\text{-Bi}_2\text{O}_3\text{-B}_2\text{O}_3$ glass

$\text{Sm}_x\text{Bi}_{1-x}\text{BO}_3$ crystal



SHG

Crystals



Bird in Nazca, Peru

Courtesy of
T. Komatsu

300 μm

VITREOUS MATERIALS LAB, UFSCar, São Carlos BRAZIL



Thank you!

Our hard working group

